

Multi-Stimulus Responsive Polyoxometalate-Viologen Hybrid Ionic Liquids for Rewritable Inkless Printing and Environmental Monitoring

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(1) C₃BPY·Br

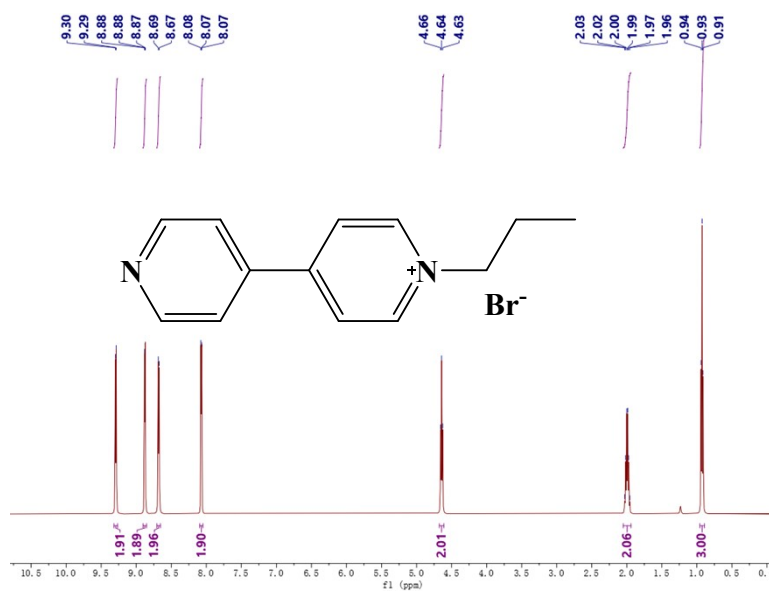


Fig. S1. ¹H NMR of C₃BPY·Br.

¹H NMR (500 MHz, DMSO-d₆), δ ppm: 9.29 (d, *J* = 6.4 Hz, 2H), 8.90 – 8.85 (m, 2H), 8.68 (d, *J* = 6.3 Hz, 2H), 8.10 – 8.05 (m, 2H), 4.64 (t, *J* = 7.3 Hz, 2H), 1.99 (h, *J* = 7.3 Hz, 2H), 0.93 (t, *J* = 7.3 Hz, 3H).

(2) C₄BPY·Br

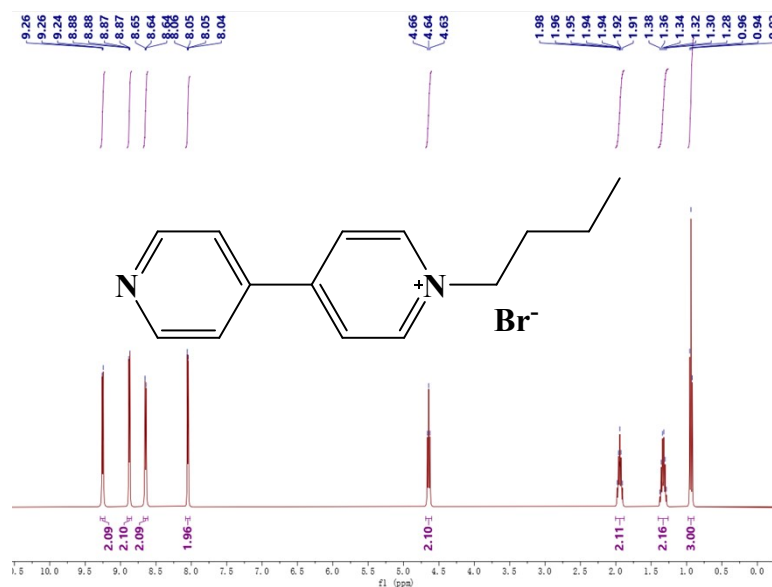


Fig. S2. 1H NMR of $C_4BPY \cdot Br$.

1H NMR (500 MHz, $DMSO-d_6$), δ ppm: 9.28 – 9.22 (m, 2H), 8.91 – 8.84 (m, 2H), 8.65 (d, $J = 6.7$ Hz, 2H), 8.08 – 8.02 (m, 2H), 4.64 (t, $J = 7.4$ Hz, 2H), 1.94 (p, $J = 7.5$ Hz, 2H), 1.33 (h, $J = 7.4$ Hz, 2H), 0.94 (t, $J = 7.3$ Hz, 3H).

(3) $C_5BPY \cdot Br$

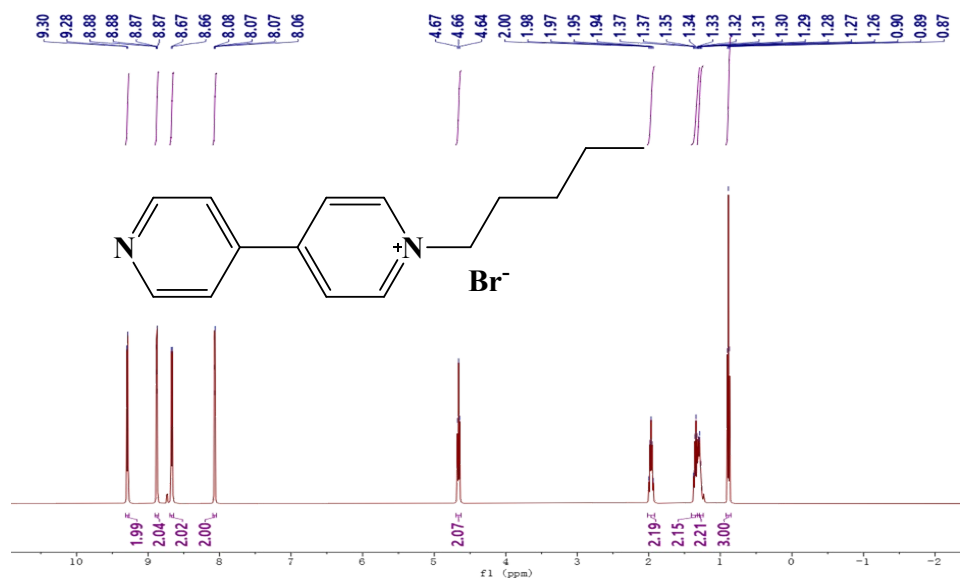


Fig. S3. 1H NMR of $C_5BPY \cdot Br$.

1H NMR (500 MHz, $DMSO-d_6$), δ ppm: 9.29 (d, $J = 6.5$ Hz, 2H), 8.90 – 8.85 (m, 2H), 8.67 (d, $J = 6.5$ Hz, 2H), 8.09 – 8.04 (m, 2H), 4.66 (t, $J = 7.4$ Hz, 2H), 1.97 (p, $J = 7.5$ Hz, 2H), 1.40 – 1.29 (m, 2H), 1.29 (dd, $J = 10.4, 5.5$ Hz, 2H), 0.89 (t, $J = 7.0$ Hz, 3H).

(4) $C_6BPY \cdot Br$

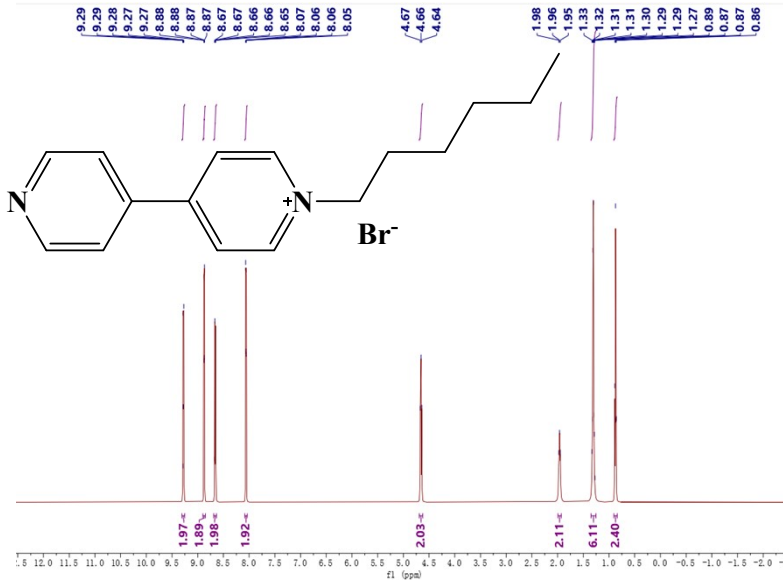


Fig. S4. ^1H NMR of $\text{C}_6\text{BPY}\cdot\text{Br}$.

¹H NMR (500 MHz, DMSO-d₆), δ ppm: 9.31 – 9.25 (m, 2H), 8.90 – 8.85 (m, 2H), 8.69 – 8.63 (m, 2H), 8.09 – 8.04 (m, 2H), 4.66 (t, J = 7.5 Hz, 2H), 1.96 (t, J = 7.2 Hz, 2H), 1.30 (q, J = 8.6, 4.3, 3.7 Hz, 6H), 0.90 – 0.84 (m, 2H).

(5) C₁₂BPY·Br

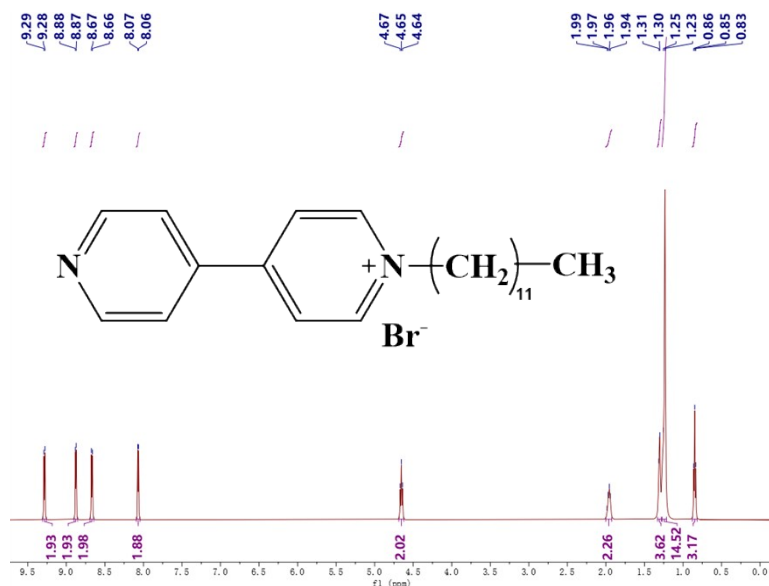


Fig. S5. ^1H NMR of $\text{C}_{12}\text{BPY}\cdot\text{Br}$.

¹H NMR (500 MHz, DMSO-d₆), δ ppm: 9.28 (d, J = 6.3 Hz, 2H), 8.88 (d, J = 5.1 Hz, 2H), 8.67 (d, J = 6.2 Hz, 2H), 8.07 (d, J = 5.2 Hz, 2H), 4.65 (t, J = 7.4 Hz, 2H), 1.96 (q, J = 7.3 Hz, 2H), 1.31 (d, J = 6.0 Hz, 4H), 1.24 (d, J = 10.2 Hz, 15H), 0.85 (t, J = 6.7 Hz, 3H).

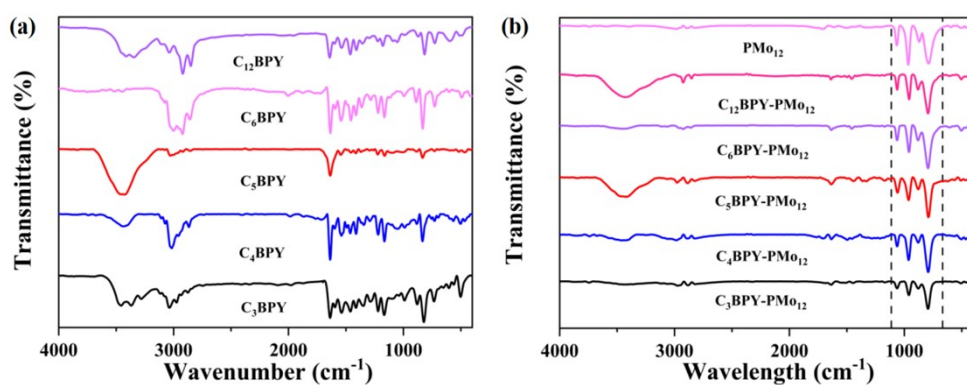
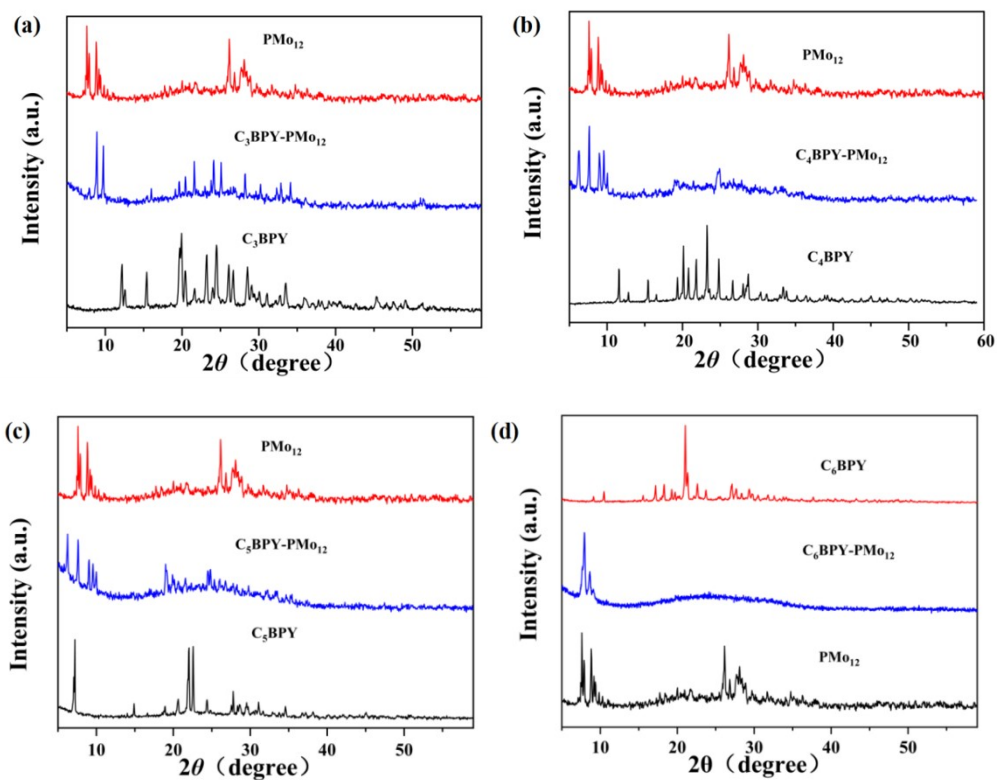


Fig. S6. The FT-IR spectra of C_nbpy (a) and C_n-PMo₁₂ (b).



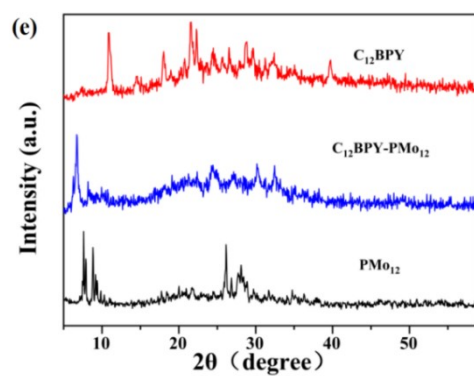
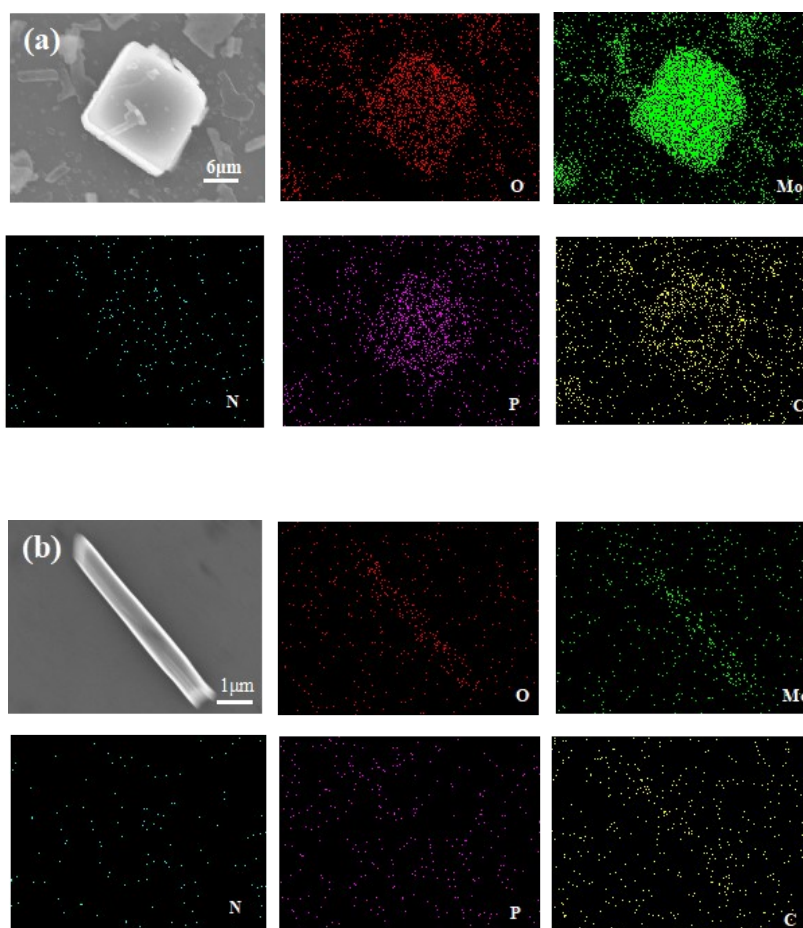


Fig. S7. (a-e) The XRD spectra of (a) C_3 - PMo_{12} , (b) C_4 - PMo_{12} , (c) C_5 - PMo_{12} , (d) C_6 - PMo_{12} , (e) C_{12} - PMo_{12} .



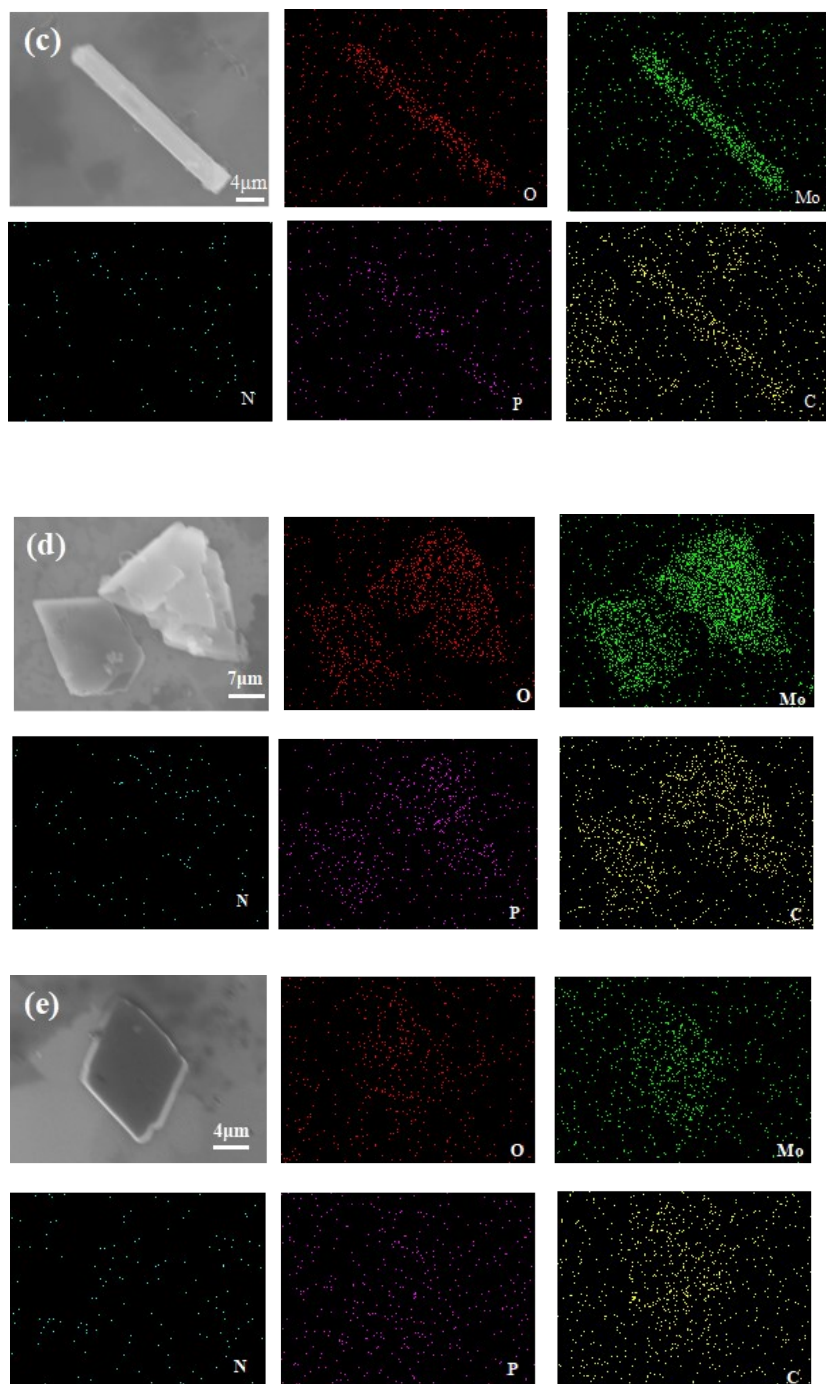


Fig. S8. (a-e) The elemental mapping of O, N, Mo, C, and P for (a) C₃-PMo₁₂, (b) C₄-PMo₁₂, (c) C₅-PMo₁₂, (d) C₆-PMo₁₂, and (e) C₁₂-PMo₁₂.

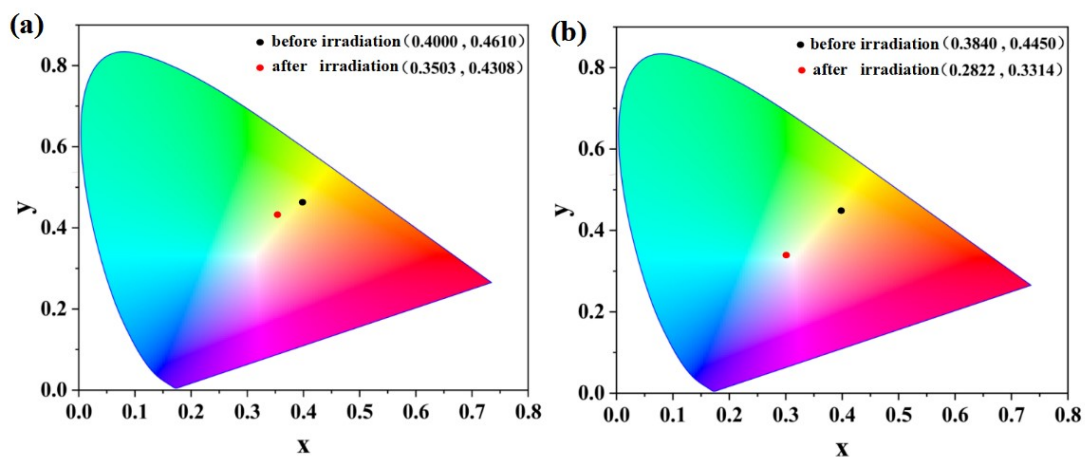


Fig. S9. (a, b) CIE chromaticity diagrams of $C_3\text{-PMo}_{12}$, $C_6\text{-PMo}_{12}$, before and after irradiation with a Xe lamp.

Table S1. L^* , a^* , b^* , and ΔE values of $C_3\text{-PMo}_{12}$, $C_6\text{-PMo}_{12}$, and $C_{12}\text{-PMo}_{12}$ before and after irradiation with a Xe lamp.

Sample	Before irradiation			After irradiation			ΔE
	L^*	a^*	b^*	$L^{*'}$	$a^{*'}$	$b^{*'}$	
$C_3\text{-PMo}_{12}$	82.891	-12.430	58.740	60.540	-16.440	29.120	29.960
$C_6\text{-PMo}_{12}$	83.450	-13.230	49.730	33.650	-7.590	-2.210	72.360
$C_{12}\text{-PMo}_{12}$	85.630	-13.238	6.912	24.910	-3.320	-6.940	85.040

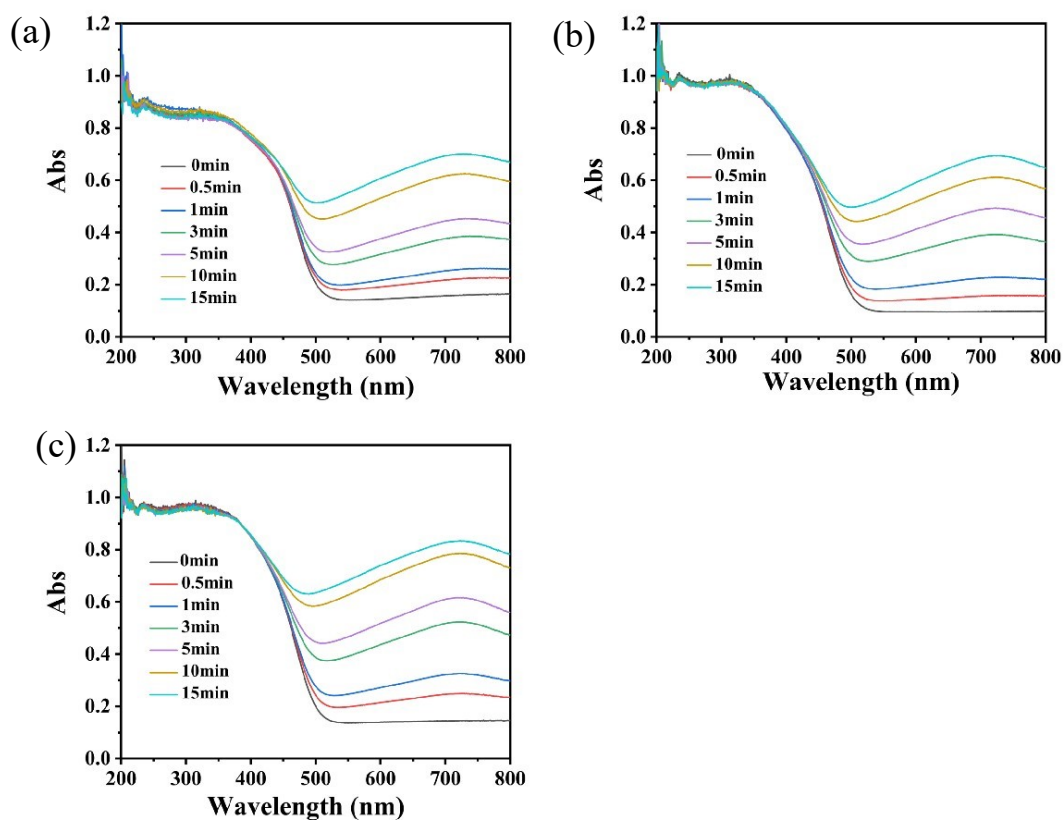


Fig. S10. (a-c) UV-vis spectra of compound $C_3\text{-PMo}_{12}$, $C_4\text{-PMo}_{12}$, and $C_5\text{-PMo}_{12}$ before and after irradiation with a Xe lamp.

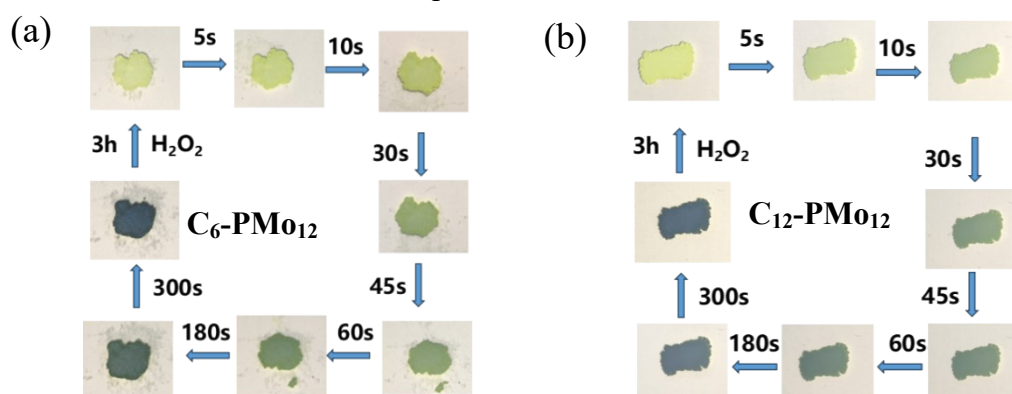
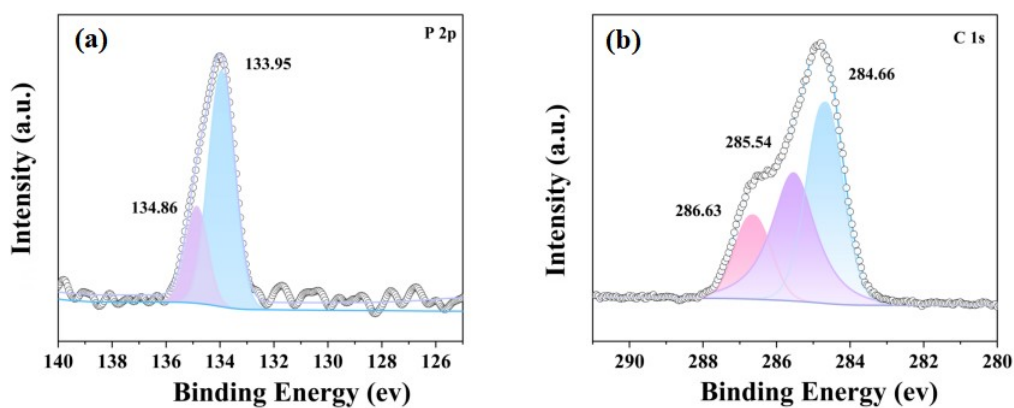


Fig. S11. (a, b) Compound $C_6\text{-PMo}_{12}$, $C_{12}\text{-PMo}_{12}$ with a Xe lamp and in H_2O_2 atmosphere.

Before irradiation:



After irradiation:

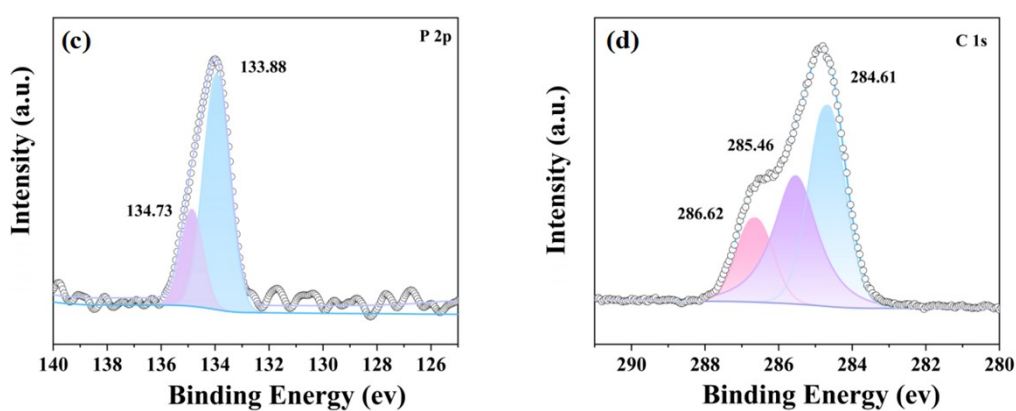


Fig. S12. (a-b) High resolution XPS spectra for C_{12} -PMo $_{12}$ P2p (a), C1s (b) before irradiation. (c-d) High resolution XPS spectra for C_{12} -PMo $_{12}$ P2p (c), N1s (d) after irradiation.

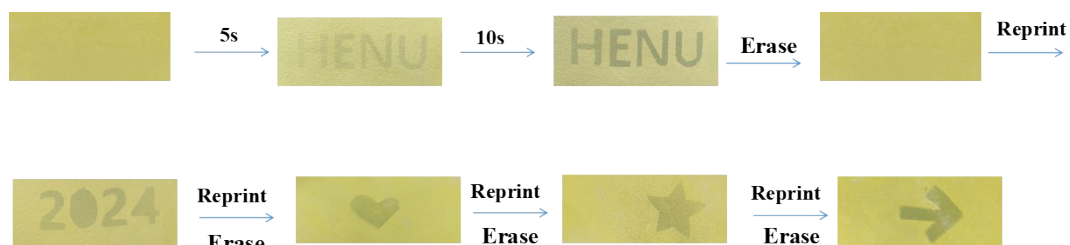


Fig. S13. Photographs of a pattern printed on C_6 -PMo $_{12}$ without ink under light irradiation.

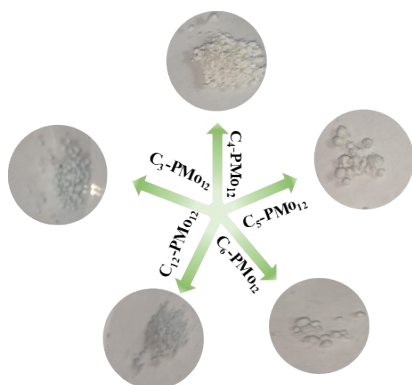


Fig. S14. Color change of $C_n\text{-PMo}_{12}$ after detection for 3min of organic amine.

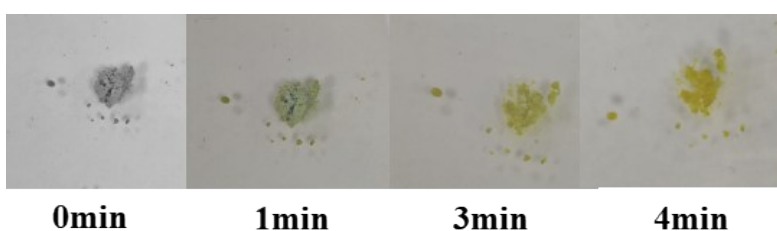


Fig. S15. Color change of $C_{12}\text{-PMo}_{12}$ (after detection of organic amine) in HCl.

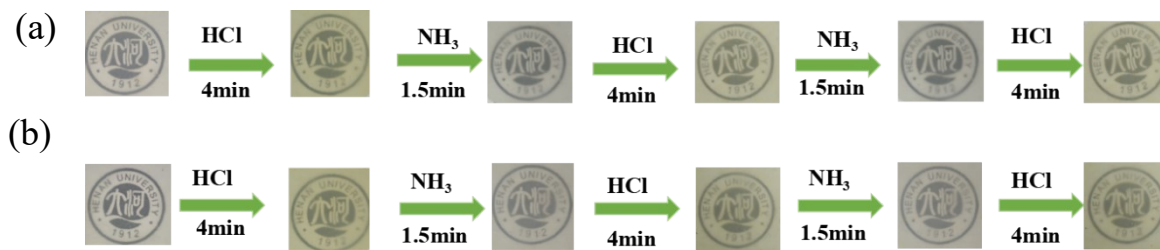


Fig. S16. (a-b) The Color change of (a) $C_6\text{-PMo}_{12}\text{-a}$, (b) $C_{12}\text{-PMo}_{12}\text{-b}$ in HCl and organic amine.

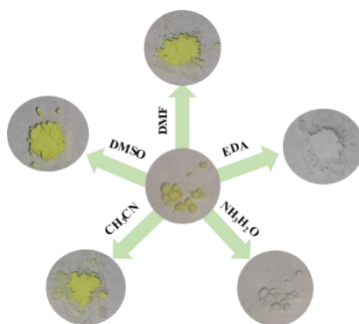


Fig. S17. Color change of $C_{12}\text{-PMo}_{12}$ after detection of different solvent atmospheres of organic amine.