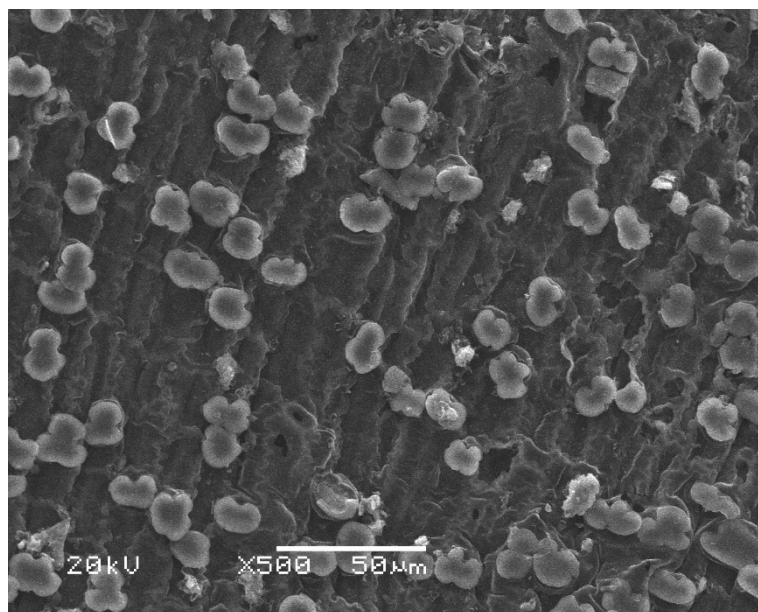
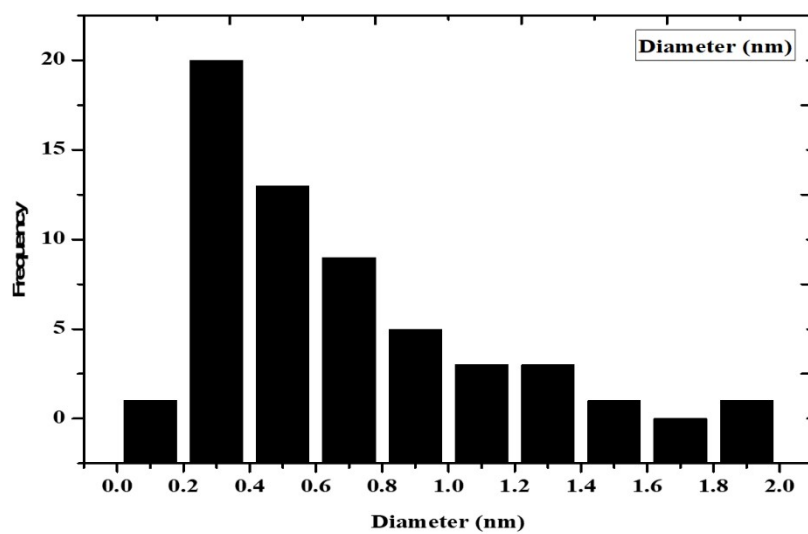


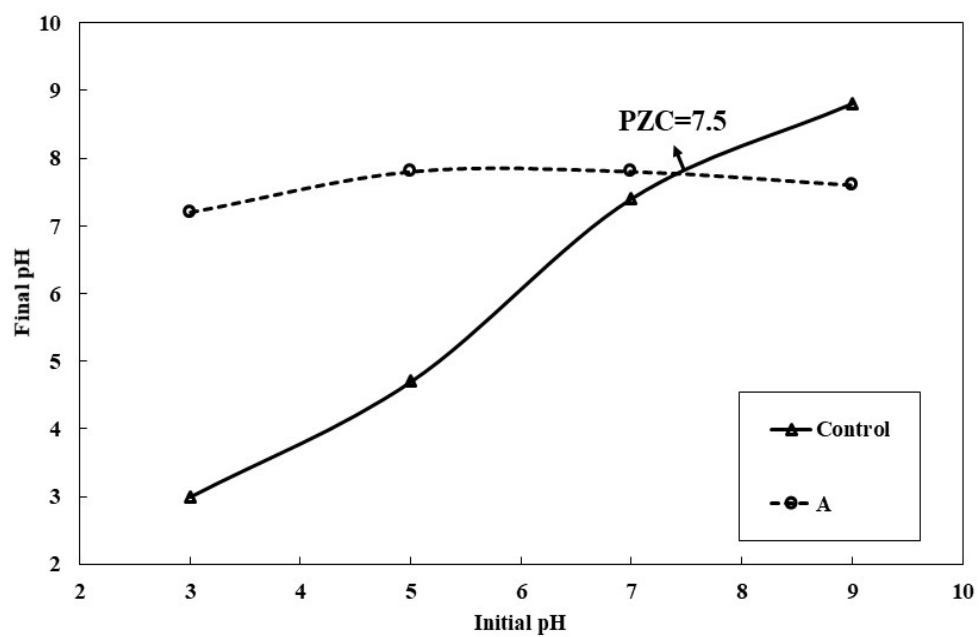
SUPPLEMENTARY DATA



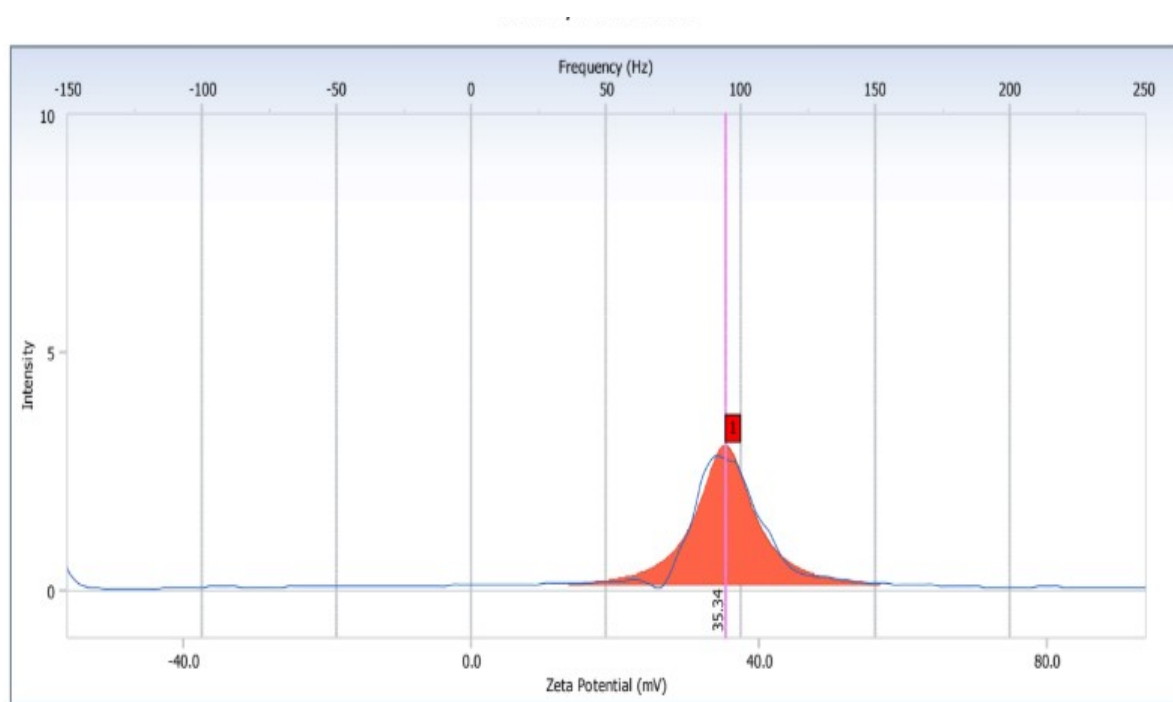
S1. SEM images of Biochar



S2. Average particle size histogram

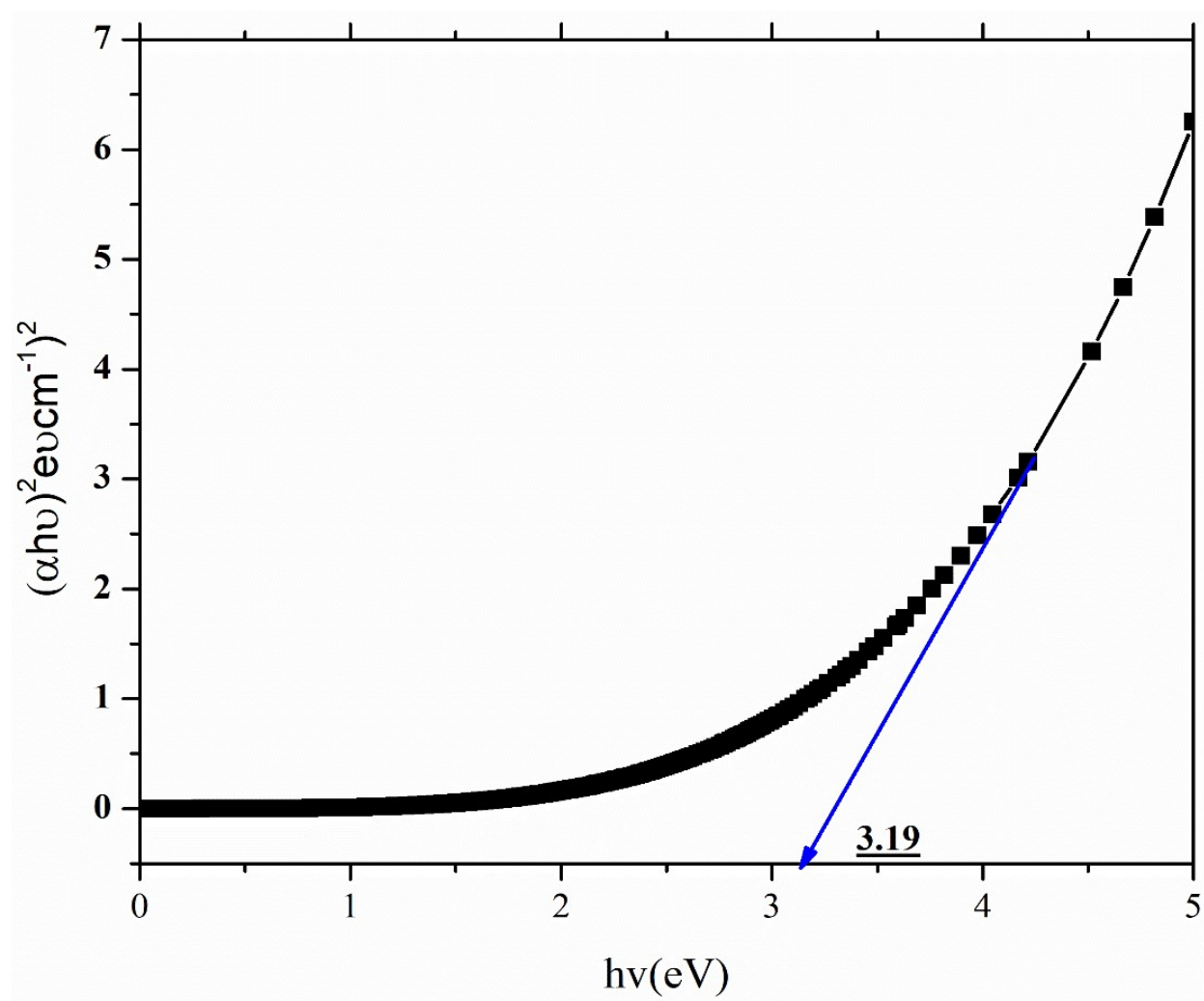


S5. Point of zero charge of nZVMn/PBC

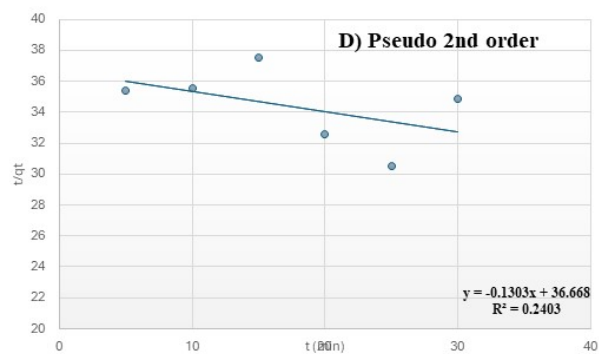
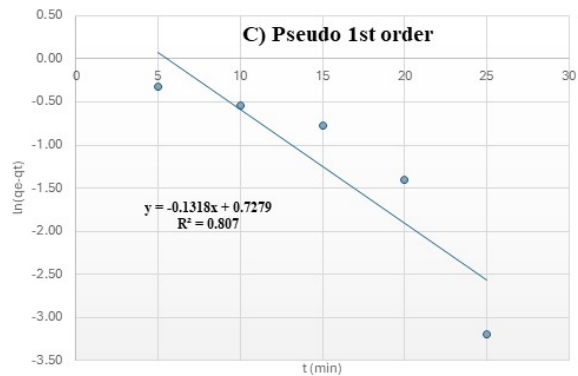
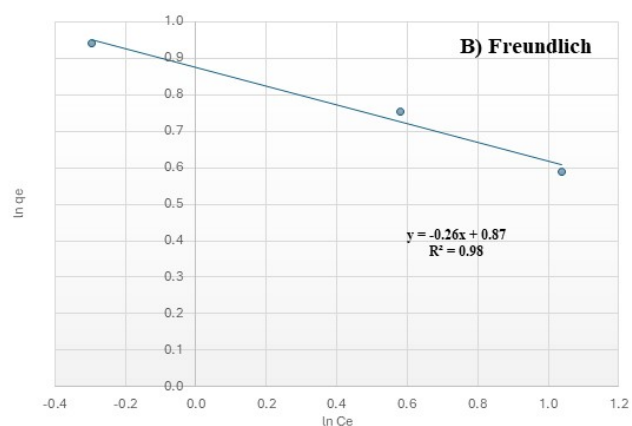
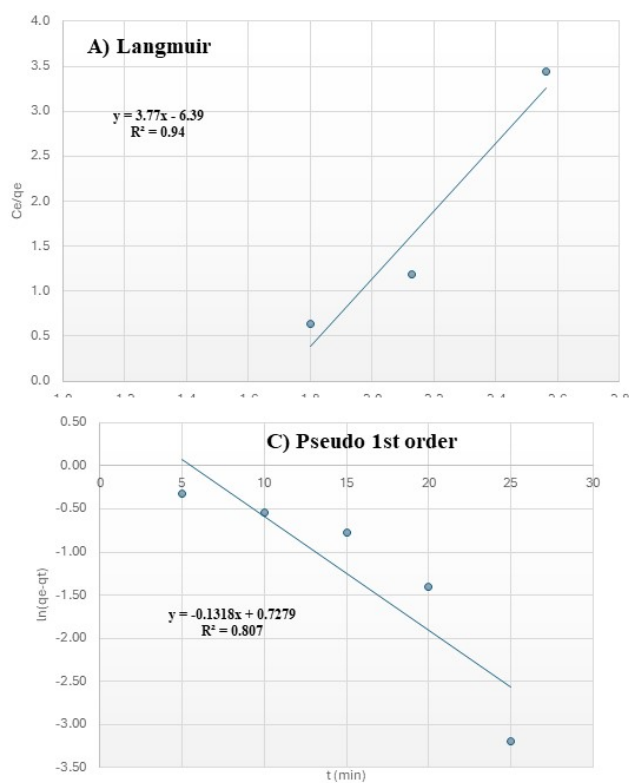


Measurement Results

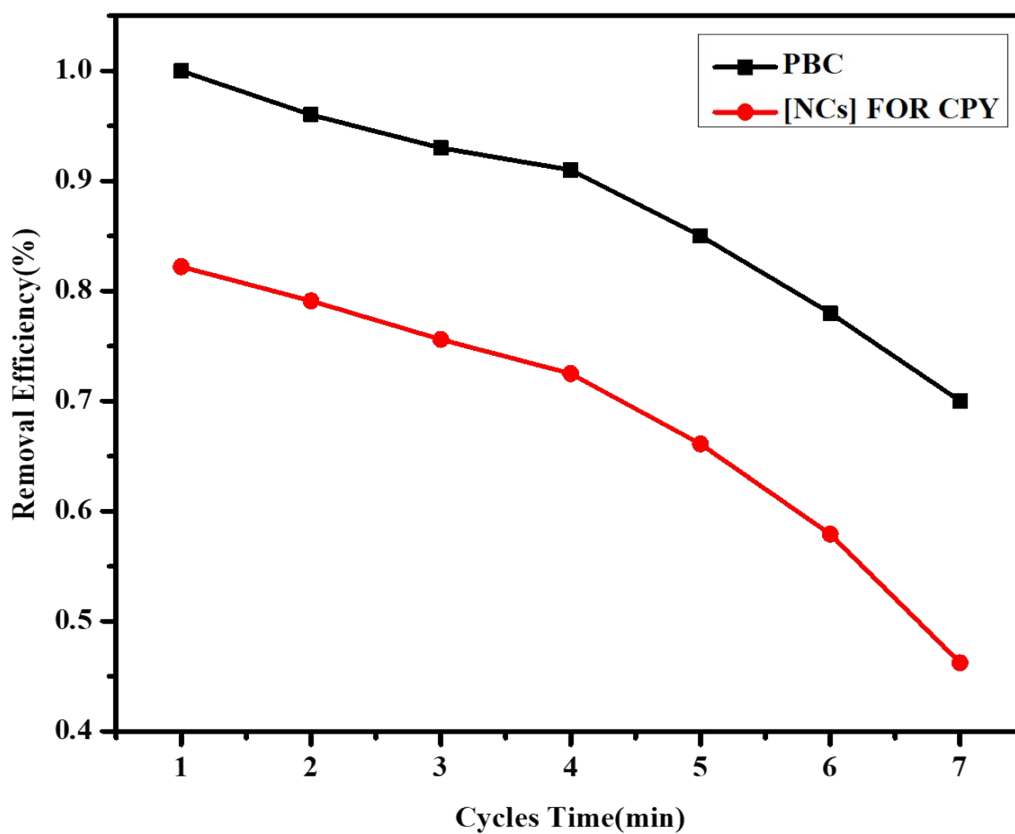
S6. Zeta Potential of nZVMn/PBC



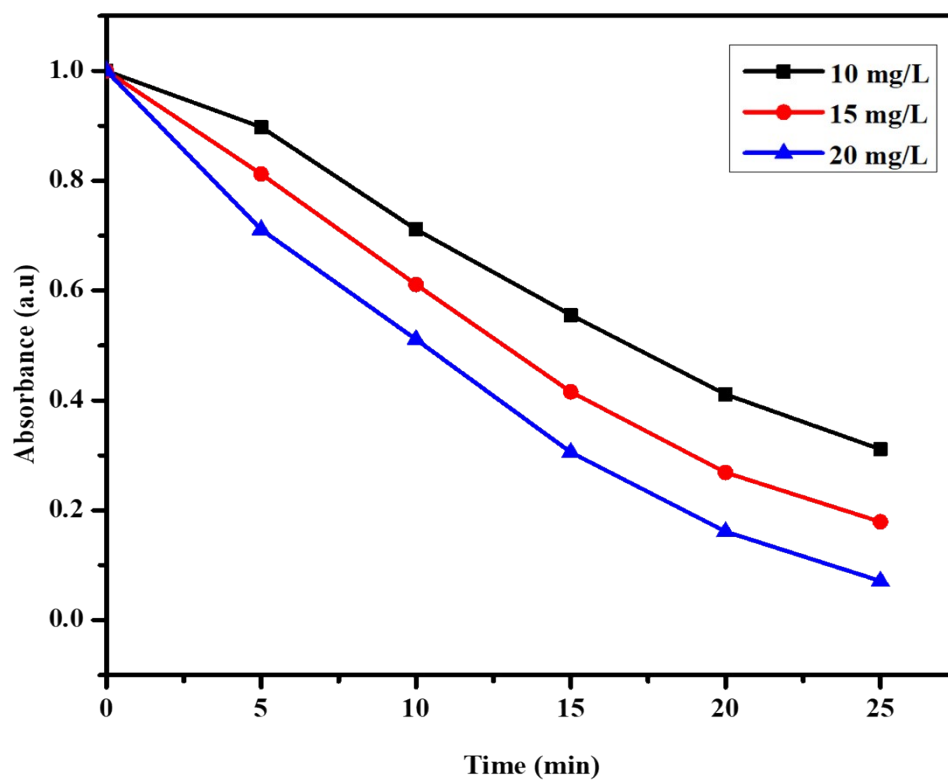
S7: UV-Vis band gap of *n*ZVMn@PBC



S8. Adsorption and kinetic modeling A) Langmuir B) Freundlich C) pseudo-first- order D) pseudo-second-Order) for pesticide adsorption by nZVMn/PBCcomposite



S9. Reusability evaluation of the synthesized PBC and nZVMn/PBC for CPF removal. [Experimental conditions, UV light, $[nZVMn/PBC]_0 = 1000$ mg/L, $[CPF]_0 = 100$ mg/L, $[H_2O_2]_0 = 10$ mg/L, Temp = 25°C, Contact time 20-120 min, pH = 8.0



S10. Effect of $[\text{H}_2\text{O}_2]_0$ concentration on degradation of CPF. [Experimental conditions, UV light, $[\text{nZVMN/PBC}]_0 = 1000 \text{ mg/L}$, $[\text{CPF}]_0 = 100 \text{ mg/L}$, $[\text{H}_2\text{O}_2]_0 = 10, 15, \text{ and } 20 \text{ mg/L}$, $\text{pH} = 8.0$]