

Supplementary Information

High-Efficiency and Environmentally Friendly Copper Recovery from Waste Copper-Clad Laminate and Its Pyrolysis Behavior through a Novel Low-Temperature Rotary Furnace Process

Submitted to

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Table S1. Kinetic parameters of the C-R method for the pyrolysis of PL-WCCL

raw material	n	fitting equation	r^2	$E(\text{KJ/mol})$	$A(\text{min}^{-1})$
PL-NMWCCCL	0.5	$Y=-9.951-2495.239X$	0.886	20.745	0.303
	1	$Y=-8.624-3048.758X$	0.957	25.347	1.018×10^1
	1.5	$Y=-6.748-3842.899X$	0.981	31.949	1.303×10^3
	2	$Y=-4.438-4826.710X$	0.953	40.129	4.508×10^5
	2.5	$Y=-1.866-5922.628X$	0.909	49.240	2.692×10^8
	3	$Y=0.860-7083.236X$	0.870	58.890	2.163×10^{11}
	3.5	$Y=3.682-8283.827X$	0.838	68.871	2.056×10^{14}
PL-WCCL	4	$Y=6.569-9511.024X$	0.464	79.074	2.183×10^{17}
	0.5	$Y=-10.375-2232.583X$	0.882	18.561	0.088
	1	$Y=-9.038-2779.418X$	0.959	23.108	3.177
	1.5	$Y=-7.093-3588.637X$	0.973	29.835	5.035×10^2
	2	$Y=-4.683-4596.337X$	0.927	38.213	2.286×10^5
	2.5	$Y=-2.013-5714.559X$	0.872	47.510	1.766×10^8
	3	$Y=0.809-6894.776X$	0.828	57.323	1.807×10^{11}
	3.5	$Y=3.725-8112.889X$	0.795	67.450	2.168×10^{14}
	4	$Y=6.703-9356.099X$	0.770	77.786	2.864×10^{17}

Table S2. Processing conditions and the metal separation rate in large-sized of PL-WCCL and WPCB

	dosage/ type/size	temperat ure/ atmosphe re	reaction device	retent ion time	subseque nt method of separatio n	Metal separati on rate	Refere nce
1	WPCB of household appliance 3×3 cm²	200°C	fixed furnace	120 min	crushing	0.3–2 mm 85.66%	1
2	5 pieces FR4- WPCB 5cm²	300°C air	electric furnace	30 min	crushing	710–1000 μm ≥90% 2000– 2800 μm >99%	2
3	PL-WCCL 25 × 45 mm²	300°C air	fixed furnace	30min for 3 times in a row	peeling off manually	100%	3
4	50g FR4- WPCB 4cm²	330°C N ₂	fixed-bed reactor	30min	crushing	92.38%	1
5	60g FR4- WPCB 4*5cm²	500°C N ₂	fixed-bed reactor	60min	peeling off manually	100%	4
6	3kg WPCB 5×5cm²	550°C vacuum	fixed-bed reactor	120mi n	crushing 、 gravity force	99%	5
7	4kg FR4 WPCB 2×1.5cm²	800°C N ₂	a fixed bed reactor with vertical stirring function	105mi n 18r/m in	automati cally	100%	6
8	3 pieces of WPCB 170×30 mm	Air	ultrasonic treatment	16h		99.99%	7

	dosage/ type/size	temperat ure/ atmosphe re	reaction device	retent ion time	subseque nt method of separatio n	Metal separati on rate	Refere nce
9	60 mm² NMP	180°C 30min	1000rar/min	30min	Overnight pre- swelling for 17 hours	99.9%	8

Table S3 Relative content (%) of main component screened by GC-MS in liquid products of pyrolysis with a retention time of 60 min

Chemical Name	Formula	PW-R-150-L	PW-R-200-L	PW-R-250-L	PW-R-300-L	PW-R-350-L
Phenol	C ₆ H ₅ OH	8.665	8.435	34.75	36.88	28.27
Phenol,2-methyl	C ₇ H ₈ O	10.045	9.93	12.52	14.83	11.32
p-Cresol	C ₇ H ₈ O	-	10.36	8.26	10.03	7.03
Phenol,2,6-dimethyl-	C ₈ H ₁₀ O	-	10.795	8.26	2.32	1.68
Phenol,2,4-dimethyl-	C ₈ H ₁₀ O	-	-	5.24	2.32	5.75
Phenol,4-ethyl-	C ₈ H ₁₀ O	-	11.985	-	1.25	1.14
Phenol,2,4,5-trimethyl-	C ₉ H ₁₂ O	-	12.505	-	-	1.4
Benzofuran, 4,7-dimethyl-	C ₁₀ H ₁₀ O	-	-	1.18	-	1.08
Phenol,4-(1-methylethyl)-	C ₁₅ H ₁₆ O	-	-	3.86	-	6.12
Phenol,2-(1-methylethyl)-	C ₉ H ₁₂ O	-	-	-	-	0.76
Phenol,4-bromo-2-chloro-	C ₆ H ₄ BrClO	-	-	-	-	0.24
Propiophenone	C ₉ H ₁₀ O	-	-	-	-	0.55
Phenol,2,4-dibromo-	C ₆ H ₄ Br ₂ O	-	-	-	-	0.31
Fluorene	C ₁₃ H ₁₀	-	-	-	-	1.22
Not detected: -						
Detection Threshold (Peak Area): ≥ 0.034						

Table S4 The relative elemental contents of XPS overall spectrum of the copper foil surface obtained after pyrolysis in a rotary furnace at 150 °C

Element name	Peak Position BE	Atomic%
C1s	283.91	53.32
O1s	530.96	14.37
Cu2p	932.66	4.06
Al2p	75.36	21.98
Cl2p ³	96.94	2.78
Si2p	101.57	3.50

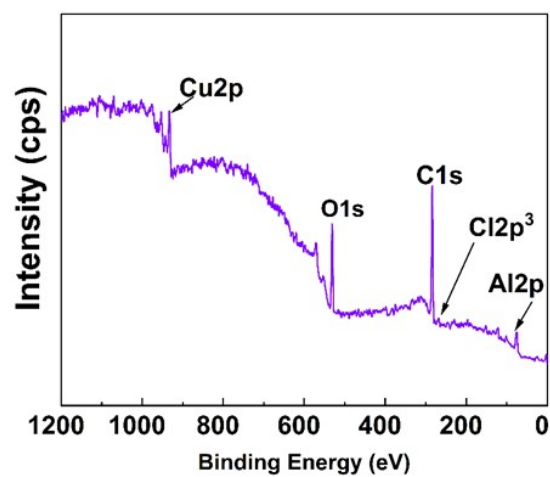


Fig. S1 The XPS overall spectrum of the copper foil surface obtained after pyrolysis in a rotary furnace at 150 °C

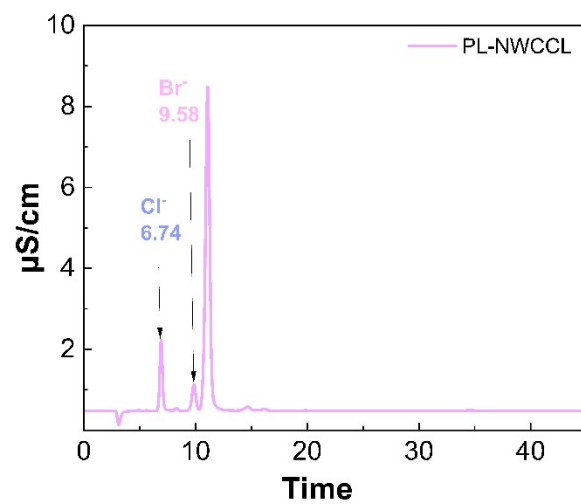


Fig. S2 The Ion chromatography analysis of PL-NWCCL.

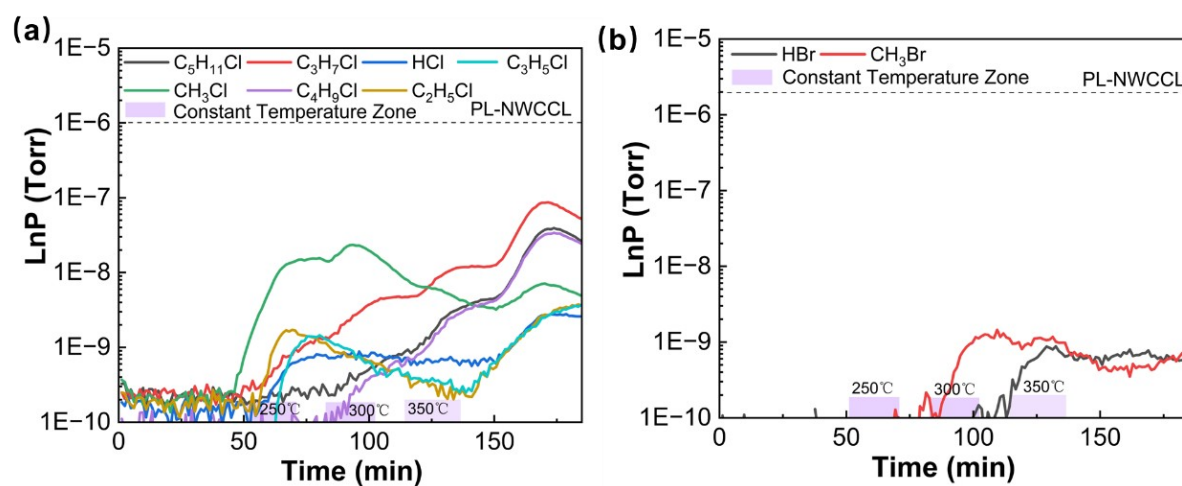


Fig. S3 Pyrolysis products of halogenated flame retardants of PL-NWCCL: (a) Chlorinated compounds, (b) Brominated compounds

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