

Supporting Information

**NiS_x/NiO/MoS₂ Heterostructure for Enhanced Hydrogen Evolution in
Acidic Media and Supercapacitor Electrode Applications**

Sumaiya Saleem^{1, #}, Muhammad Salman^{1, #}, Ilham Noor², Baseena Sardar¹, Majid Khan^{1, *}

¹ Department of Physics, Abdul Wali Khan University Mardan, Mardan 23200, Pakistan

² Department of Physics, University of Malakand, Chakdara, 18800, Pakistan

[#] These two authors equally contributed to this work

^{*} Corresponding author: Dr. Majid Khan, majidkhan@awkum.edu.pk

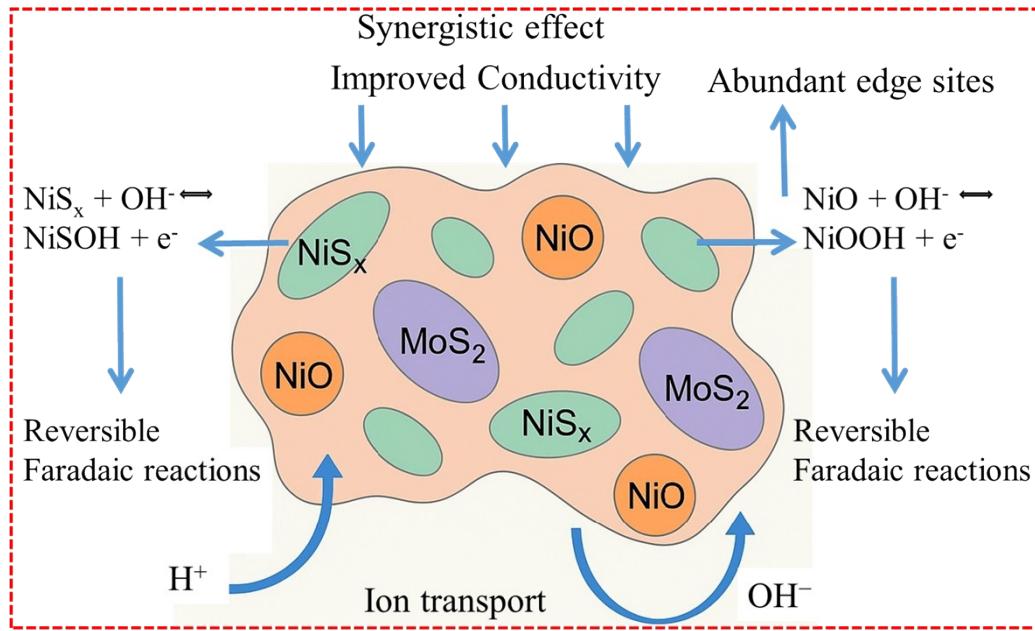


Figure S1: Mechanism of the selected electrode materials based on their redox-active components.

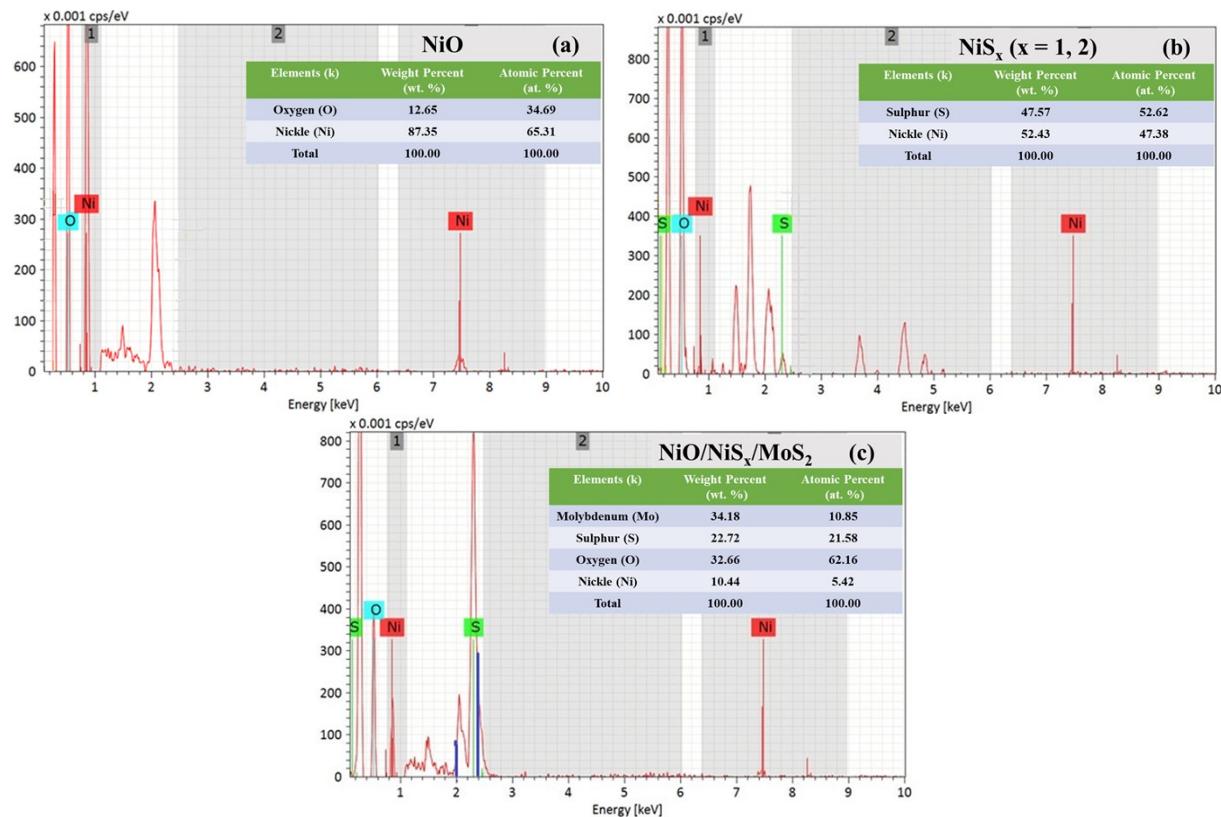


Figure S2: EDX results of (a) NiO, (b) NiS_x, and (c) NiS_x/NiO/MoS₂ (x = 1, 2) heterostructure.

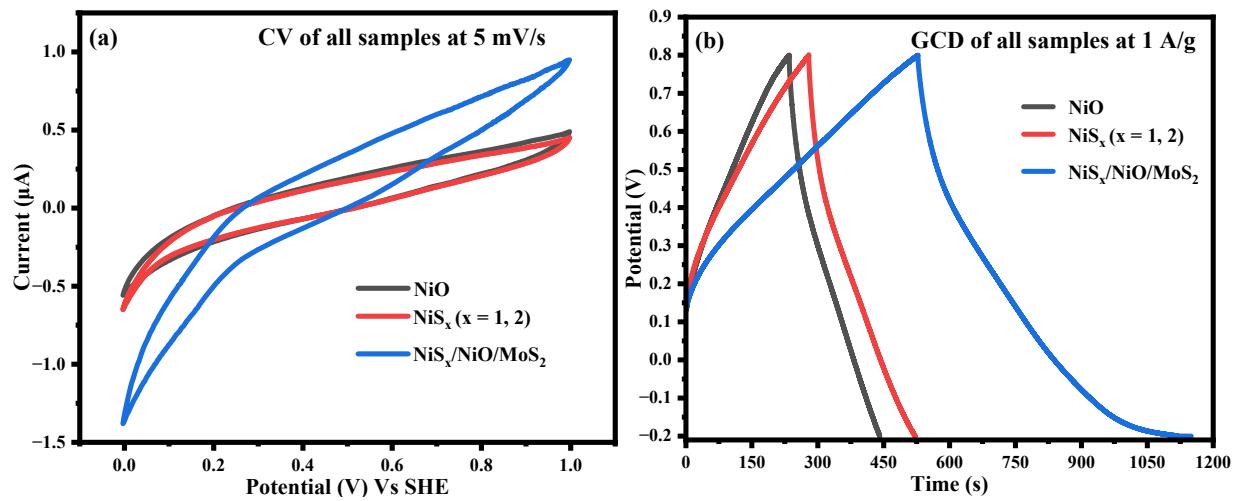


Figure S3: (a) CV curves at 5 mV/s, and (b) GCD plots at 1 A/g of all NiO, NiS_x, and NiS_x/NiO/MoS₂ (x = 1, 2) heterostructure.

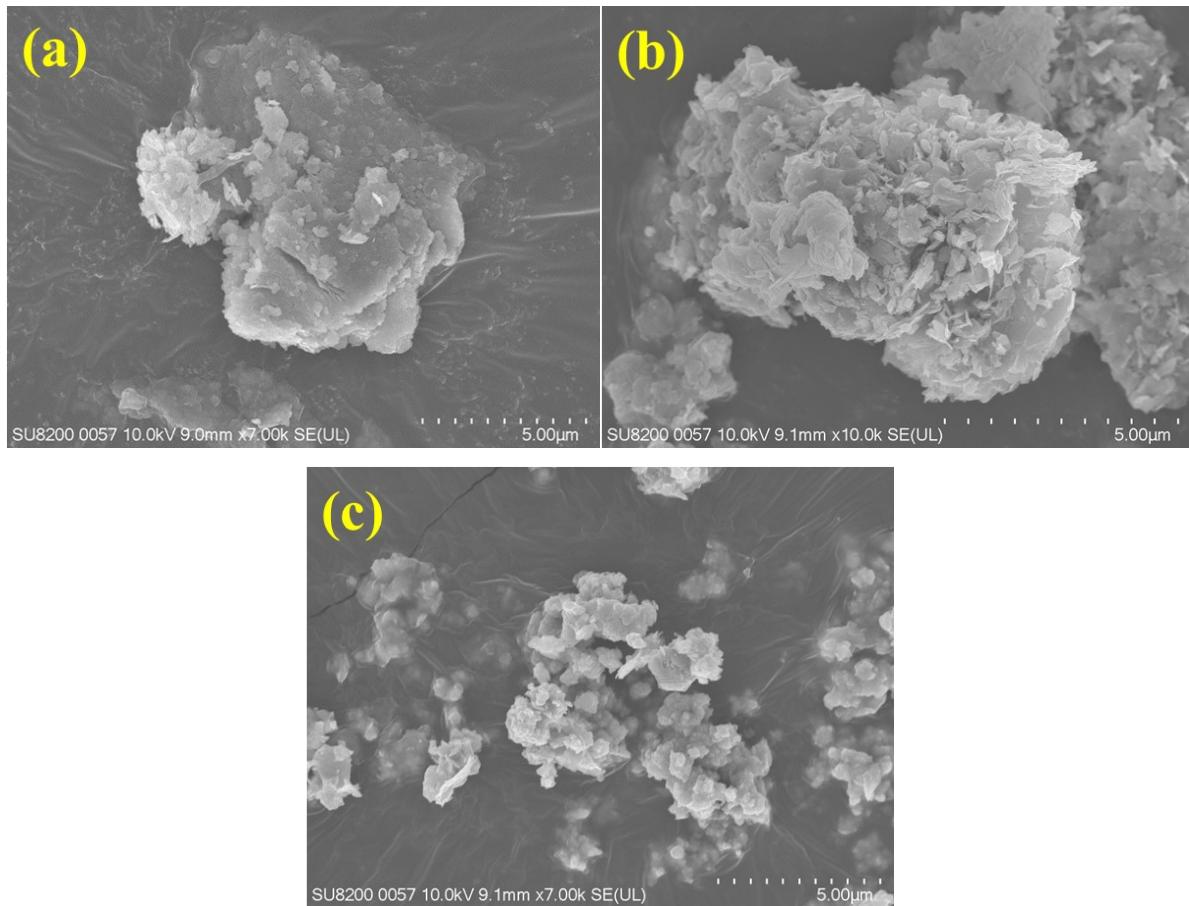


Figure S4: SEM images of (a) NiO, (b) NiS_x, and (c) NiS_x/NiO/MoS₂ (x = 1, 2) heterostructure.

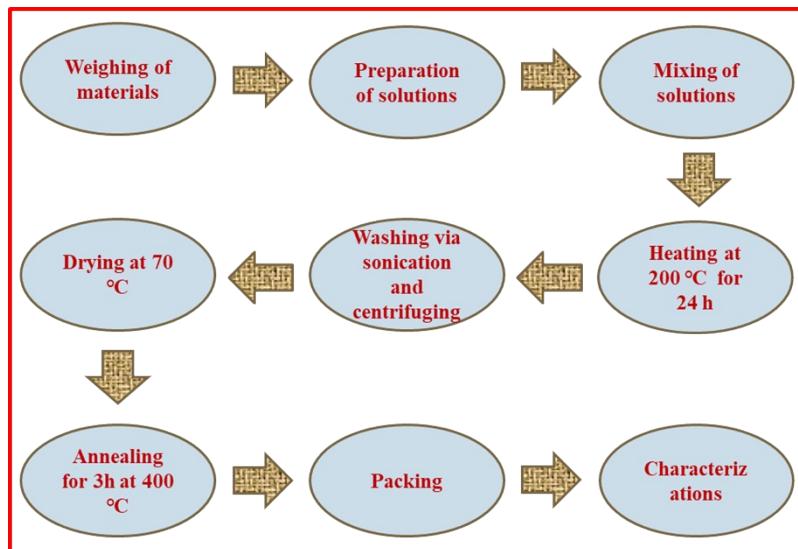


Figure S5: Schematic illustration of the synthesis route of the $\text{NiS}_x/\text{NiO}/\text{MoS}_2$ ($x = 1, 2$) heterostructure.

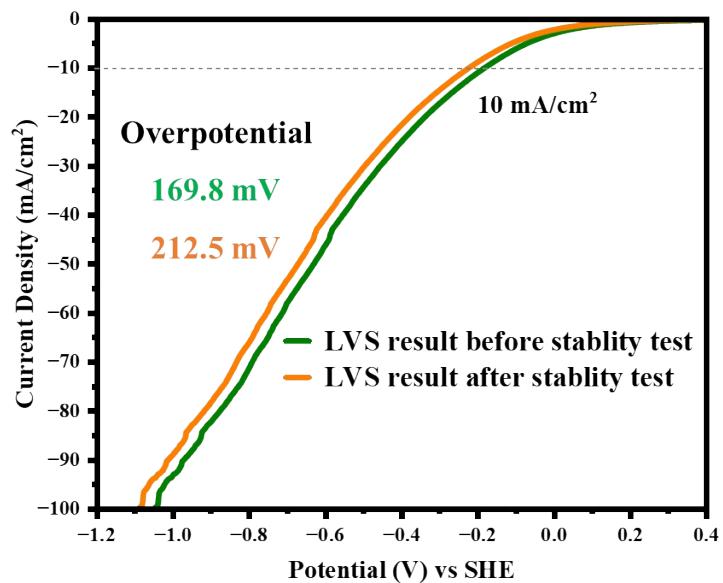


Figure S6: LVS results of the prepared composite before and after the stability test of $\text{NiS}_x/\text{NiO}/\text{MoS}_2$ ($x = 1, 2$) heterostructure.

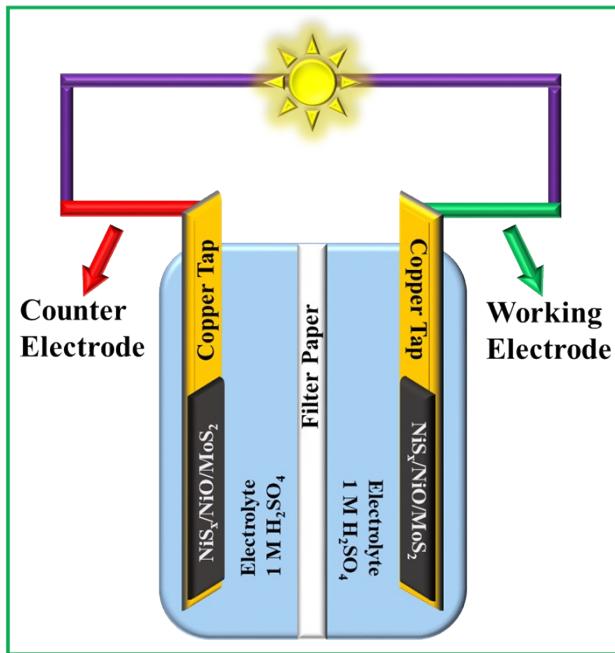


Figure S7: Schematic diagram of the symmetric supercapacitor device

Table S1: A comparative table that highlights the key features of H_2SO_4 and other commonly used hydrogen-based electrolytes

Electrolyte	pH	Proton conductivity (S/cm)	Redox kinetics	Corrosiveness	Common use in
1 M H_2SO_4	~ 0.3	~ 0.08	Fast	Moderate	Supercapacitors, HER/OER
1 M HCl	~ 0.1	~ 0.09	Fast	High	HER, acid corrosion studies
1 M HNO_3	~ 0.3	~ 0.07	Moderate	High	Catalysis, oxidation studies
1 M CH_3COOH	~ 2.4	~ 0.002	Slow	Low	Organic/biochemical systems

Table S2: Comparison of the overpotential, specific capacitance, synthesis method, and performance notes of $\text{NiS}_x/\text{NiO}/\text{MoS}_2$ ($x = 1, 2$) heterostructure with literature.

Material	Overpotential at 10 mA/cm ² (mV)	Specific Capacitance (F/g)	Synthesis Method	Performance Notes	Reference
NiS/NiS_2 composite with layered structure	180	95	Hydrothermal	NiS/NiS_2 composite exhibits excellent HER activity and stability	[1]
$\text{NiO}/\text{NiS}_2/\text{MoS}_2$ ternary composite with tunable sulfur vacancies	120	115	Hydrothermal	$\text{NiO}/\text{NiS}_2/\text{MoS}_2$ composite provides high stability and low overpotential	[2]
NiS/MoS_2 composite with MoS_2 edges exposed	150	125	Chemical bath deposition	Enhanced HER efficiency due to the exposed MoS_2 edges	[3]
$\text{NiS}_x/\text{NiO}/\text{MoS}_2$	166	428.2425	Hydrothermal		Our work

References

1. Dhakal, G., Sahoo, S., Sharma, K. P. A Review on the Recent Advancements of Ni-Based Sulfides and Mixed Sulfides for Supercapacitors and Electrocatalysis (Oxygen Evolution Reaction). *Molecules*, 30 (13), 2025.
2. Shah, S. A., Sayyar, R., Xu, L., Sun, H., Khan, I., Guo, J., ... & Ullah, H. In-situ synthesis of NiS_2 nanoparticles/ MoS_2 nanosheets hierarchical sphere anchored on reduced graphene oxide for enhanced electrocatalytic hydrogen evolution reaction. *Journal of Colloid and Interface Science*, 624 (18), 2022.
3. Wang, W., Li, L., Tan, S., Wu, K., Zhu, G., Liu, Y., ... & Yang, Y. Preparation of $\text{NiS}_2/\text{MoS}_2$ catalysts by two-step hydrothermal method and their enhanced activity for hydrodeoxygenation of p-cresol. *Fuel*, 179 (37) 2016.