

Supporting information

Spectrochemical, Medicinal, and Toxicological Studies of Moxifloxacin and its Novel Analog: A Quantum Chemistry and Drug Discovery Approach

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Table S1 Chemical structures and IUPAC name of MOX and its analogs.

Name	Structure	IUPAC Name
MOX		1-cyclopropyl-6-fluoro-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid
MOX1		1-cyclopropyl-8-methoxy-6-methyl-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid
MOX2		6-amino-1-cyclopropyl-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid
MOX3		1-cyclopropyl-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-6-(trifluoromethoxy)-1,4-dihydroquinoline-3-carboxylic acid
MOX4		1-cyclopropyl-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-6-(trifluoromethyl)-1,4-dihydroquinoline-3-carboxylic acid
MOX5		1-cyclopropyl-6-fluoro-3-glycyl-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)quinolin-4(1H)-one
MOX6		1-cyclopropyl-6-fluoro-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carbonyl chloride

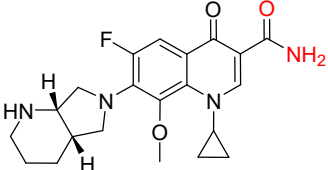
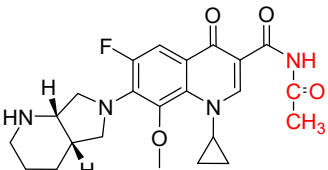
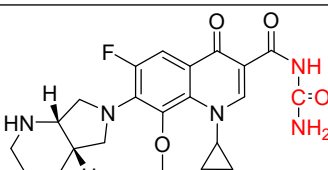
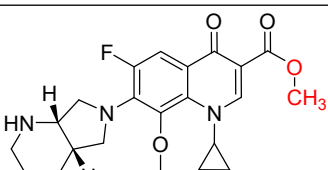
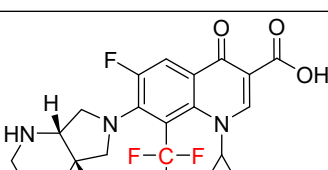
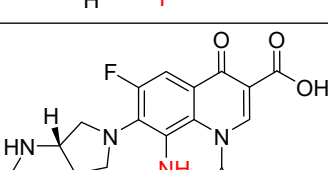
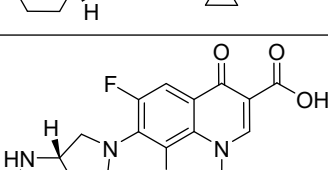
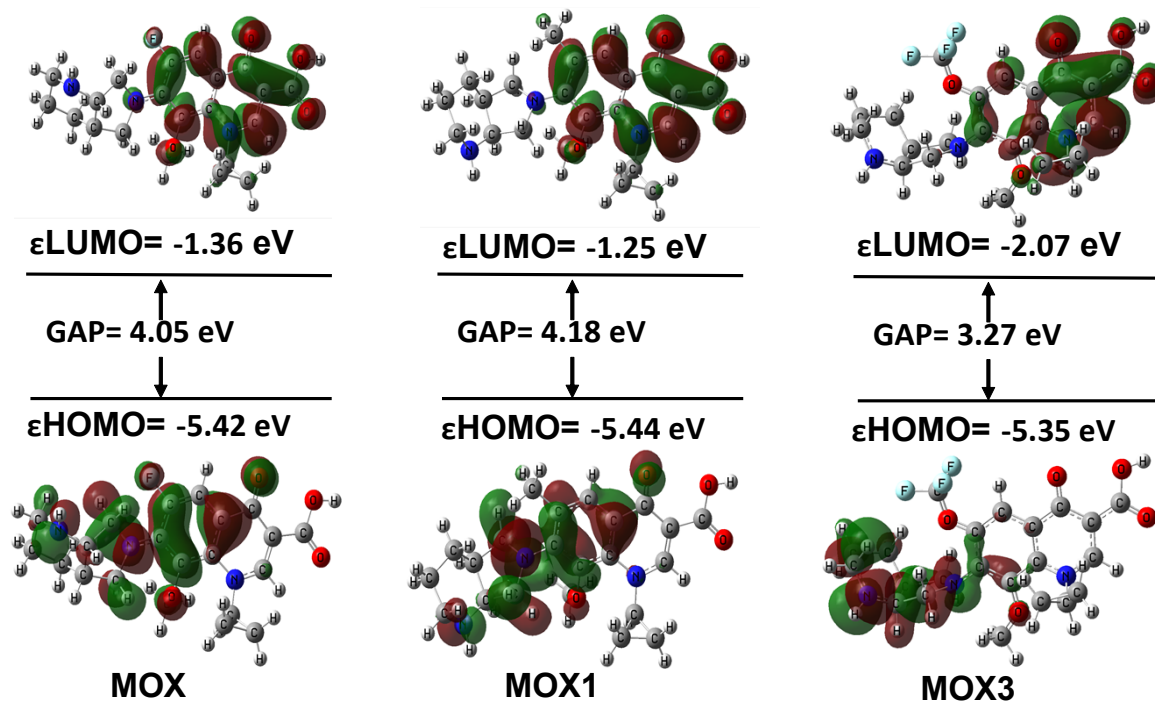
MOX7		1-cyclopropyl-6-fluoro-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxamide
MOX8		N-acetyl-1-cyclopropyl-6-fluoro-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxamide
MOX9		N-carbamoyl-1-cyclopropyl-6-fluoro-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxamide
MOX10		Methyl 1-cyclopropyl-6-fluoro-8-methoxy-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylate
MOX11		1-cyclopropyl-6-fluoro-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-8-(trifluoromethyl)-1,4-dihydroquinoline-3-carboxylic acid
MOX12		8-amino-1-cyclopropyl-6-fluoro-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid
MOX13		1-cyclopropyl-6-fluoro-7-((4aS,7aS)-octahydro-6H-pyrrolo[3,4-b]pyridin-6-yl)-4-oxo-8-ureido-1,4-dihydroquinoline-3-carboxylic acid

Table S2 Energy (eV) of HOMO-LUMO, gap, hardness (η), softness (S), and chemical potential (μ) of MOX and its analogs.

Name	ϵ_{HOMO} (eV)	ϵ_{LUMO} (eV)	Gap (eV)	η	S	μ
MOX	-5.42	-1.36	4.05	2.02	0.24	-3.39
MOX1	-5.44	-1.25	4.18	2.09	0.23	-3.35
MOX2	-5.12	-1.12	4.00	2.00	0.24	-3.12
MOX3	-5.35	-2.07	3.27	1.63	0.30	-3.71
MOX4	-5.86	-2.26	3.59	1.79	0.27	-4.06
MOX5	-5.51	-1.52	3.99	1.99	0.25	-3.52
MOX6	-5.82	-1.93	3.89	1.94	0.25	-3.80
MOX7	-5.67	-1.48	4.18	2.09	0.23	-3.58
MOX8	-5.68	-1.63	4.04	2.02	0.24	-3.66
MOX9	-5.36	-1.21	4.14	2.07	0.24	-3.29
MOX10	-5.36	-1.27	4.09	2.04	0.24	-3.32
MOX11	-5.96	-2.42	3.54	1.77	0.28	-4.19
MOX12	-5.44	-1.30	4.14	2.07	0.24	-3.37
MOX13	-5.31	-1.70	3.61	1.80	0.27	-3.51



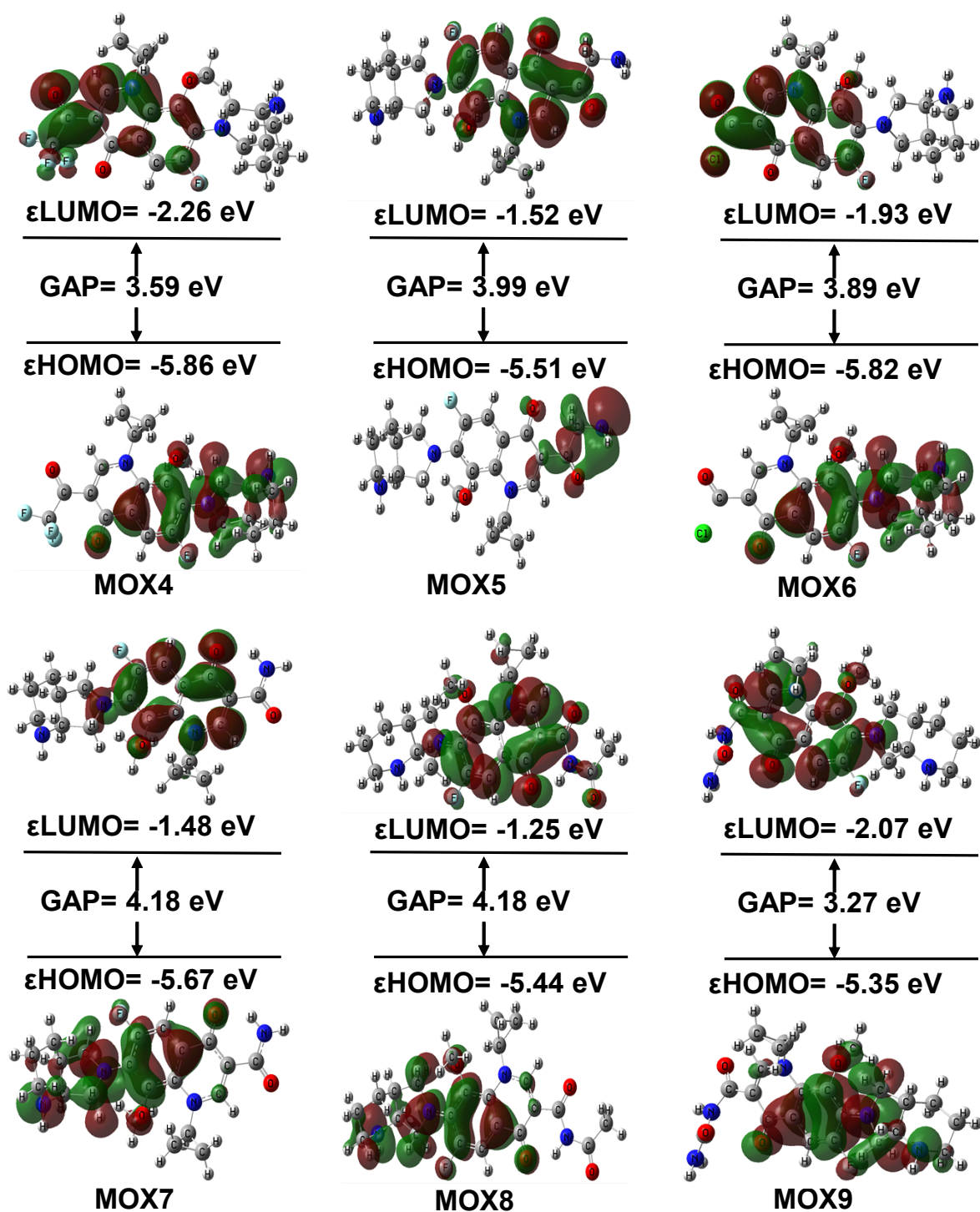


Fig. S1 Frontier molecular orbital (HOMO-LUMO energy gap) of MOX and its analogs.

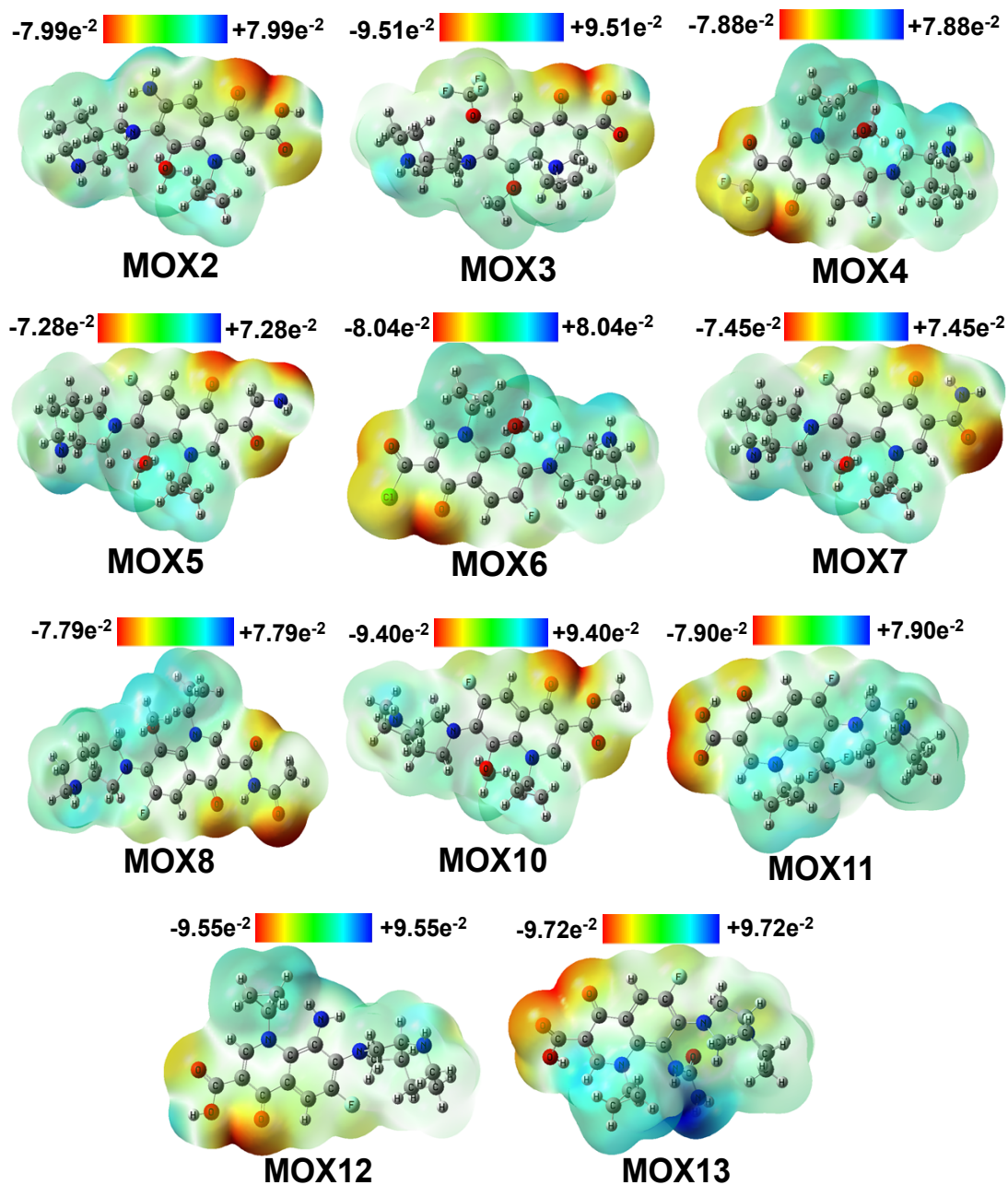
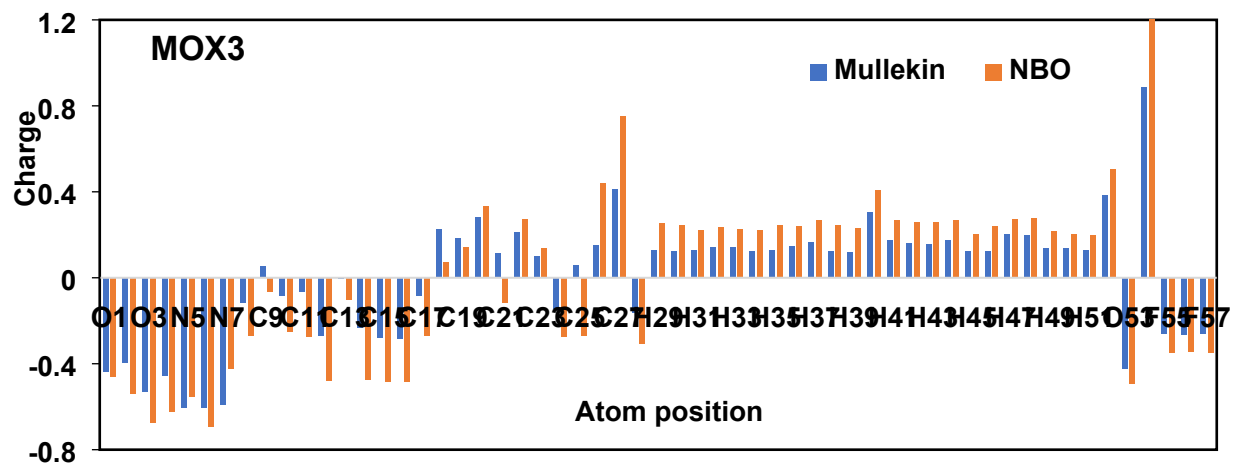
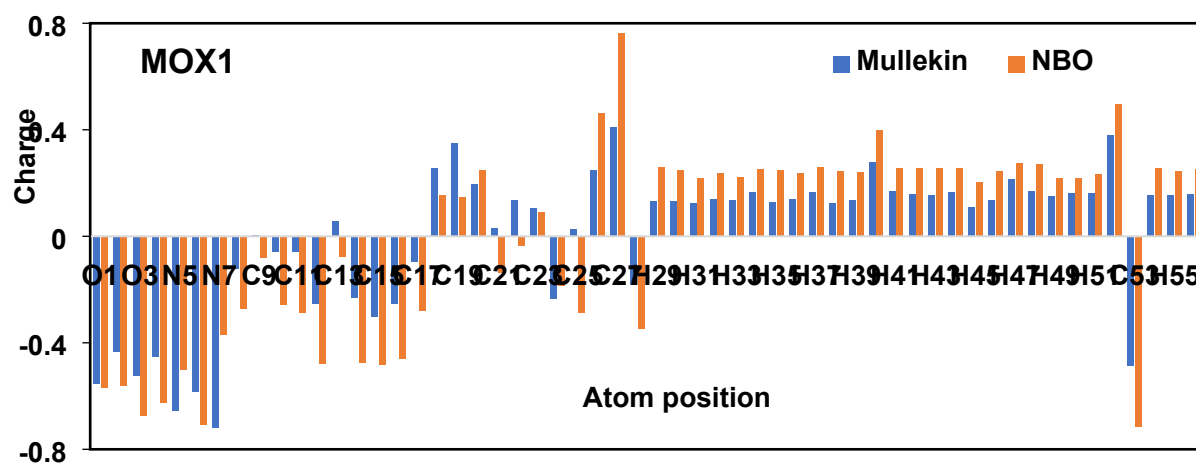
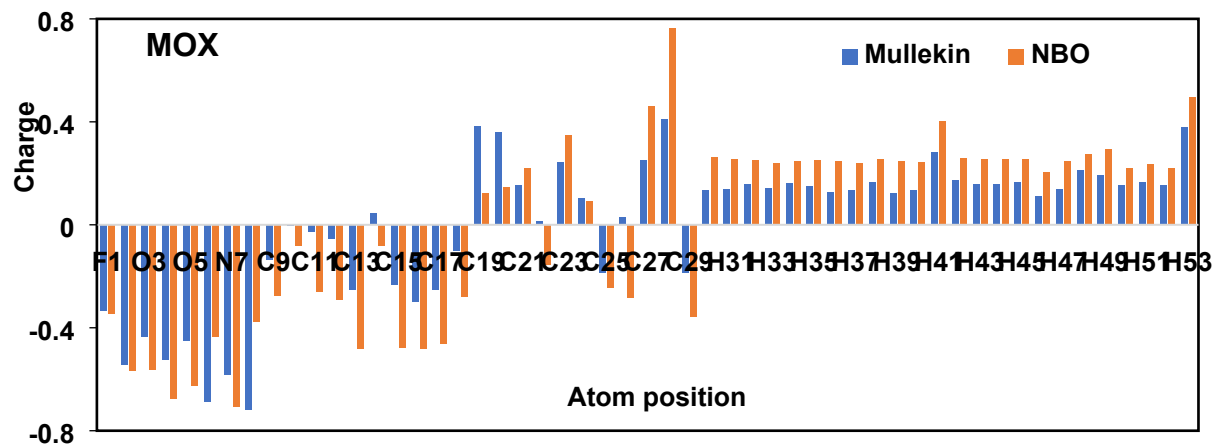
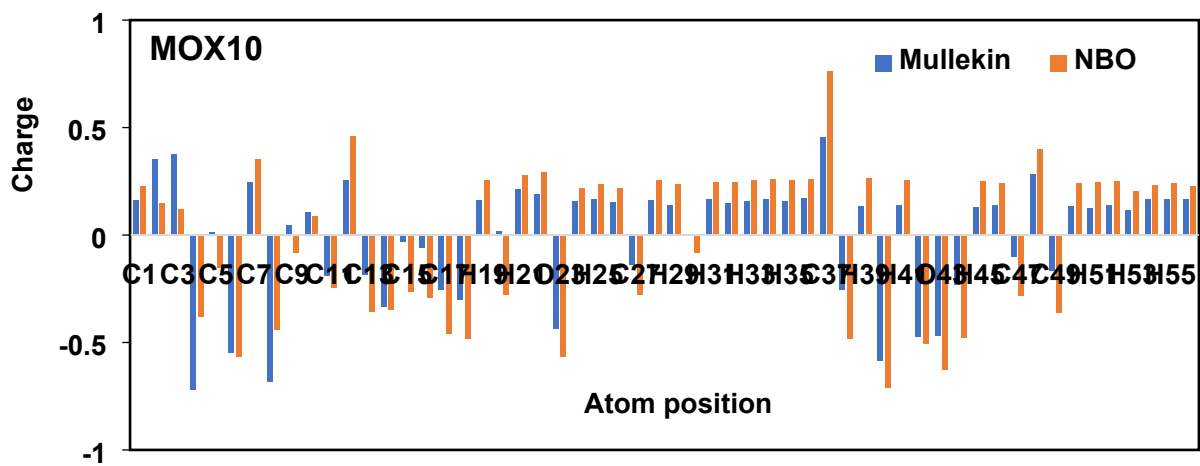
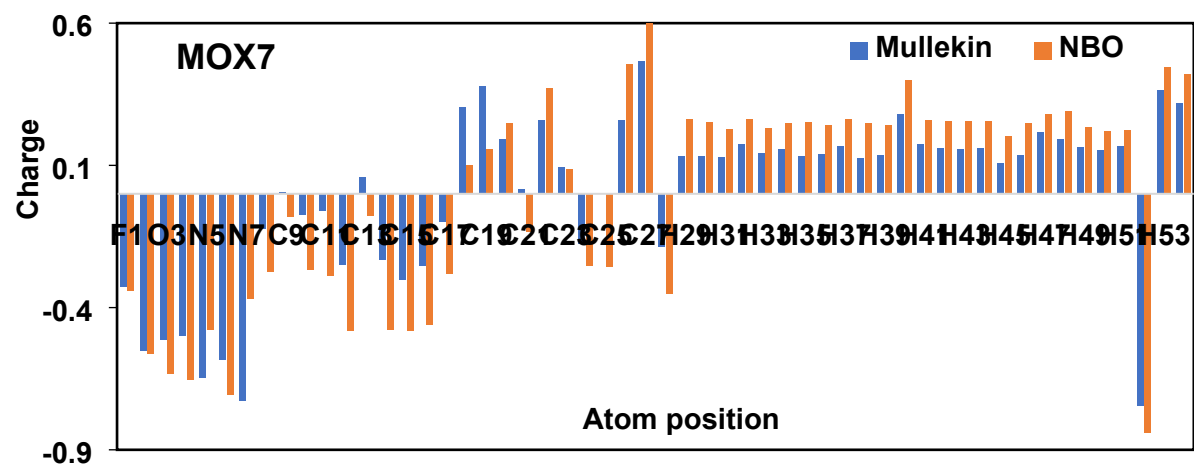
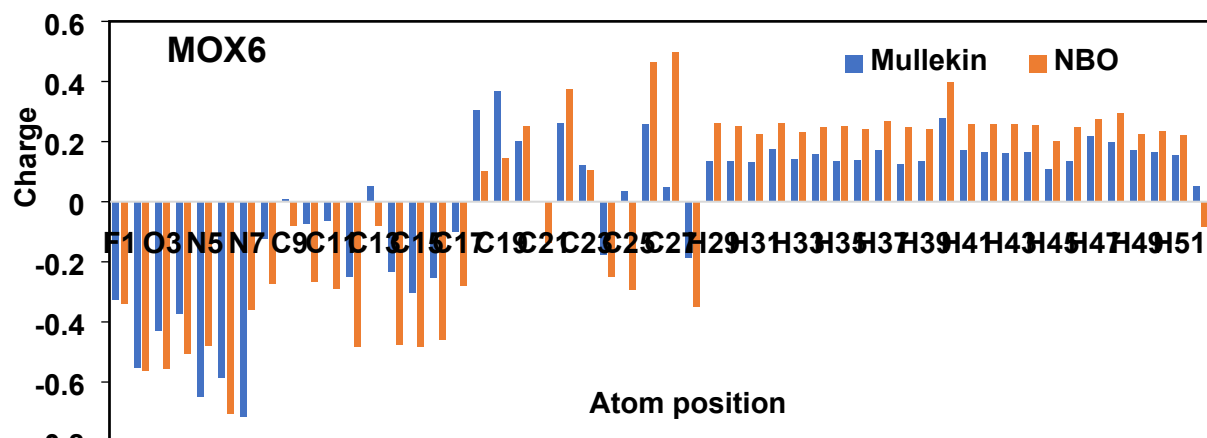


Fig. S2 ESP map of MOX derivatives.





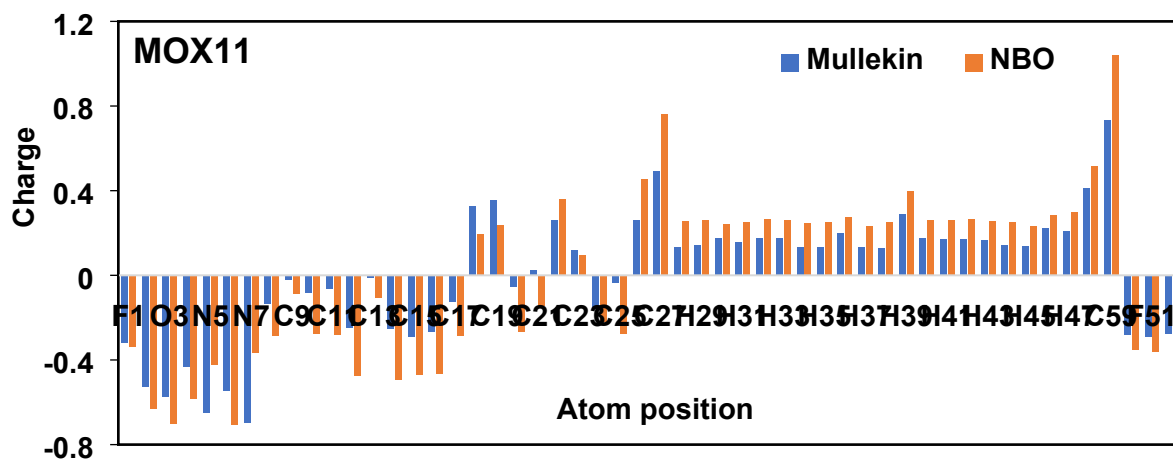


Fig S3 Atomic partial charge of MOX and its analogs.

Table S3 Electronic absorption of MOX and its analogs, calculated at TD-DFT/6-31g level of theory.

Name	Excited state	Wavelength (nm)	Excitation energy(eV)	Configurations composition	Oscillator strength
MOX	$S_0 \rightarrow S_1$	550.51	2.2522	(0.70022) H $\rightarrow$ L	0.0300
MOX1	$S_0 \rightarrow S_1$	520.72	2.3810	(0.69947) H $\rightarrow$ L	0.0256
MOX2	$S_0 \rightarrow S_1$	499.38	2.4828	(-0.10149) H-2 $\rightarrow$ L, (-0.13749) H-1 $\rightarrow$ L, (0.68020) H $\rightarrow$ L	0.0477
MOX3	$S_0 \rightarrow S_1$	429.01	2.8900	(-0.11053) H-3 $\rightarrow$ L, (0.17674) H-2 $\rightarrow$ L, (0.66155) H $\rightarrow$ L	0.0055
MOX4	$S_0 \rightarrow S_1$	619.59	2.0011	(0.69975) H $\rightarrow$ L	0.0002
MOX5	$S_0 \rightarrow S_1$	442.51	2.8018	(0.62801) H-1 $\rightarrow$ L, (-0.12600) H-1 $\rightarrow$ L+1, (-0.27400) H $\rightarrow$ L	0.0035
MOX6	$S_0 \rightarrow S_1$	597.22	2.0760	(0.70104) H $\rightarrow$ L	0.0222
MOX7	$S_0 \rightarrow S_1$	423.00	2.9310	(0.61348) H-2 $\rightarrow$ L, (0.10793) H-2 $\rightarrow$ L+1, (-0.16872) H-1 $\rightarrow$ L	0.0043
MOX8	$S_0 \rightarrow S_1$	534.24	2.3208	(0.70015) H $\rightarrow$ L	0.0441
MOX9	$S_0 \rightarrow S_1$	552.99	2.2421	(0.69991) H $\rightarrow$ L	0.0166
MOX10	$S_0 \rightarrow S_1$	561.28	2.2089	(0.69993) H $\rightarrow$ L	0.0308
MOX11	$S_0 \rightarrow S_1$	629.82	1.9686	(0.70569) H $\rightarrow$ L	0.0153
MOX12	$S_0 \rightarrow S_1$	440.68	2.8135	(0.48324) H-1 $\rightarrow$ L, (0.50153) H $\rightarrow$ L	0.0136

MOX13	$S_0 \rightarrow S_1$	438.05	2.8304	(0.68072) H-2 $\rightarrow$ L, (0.11928) H-1 $\rightarrow$ L	0.0003
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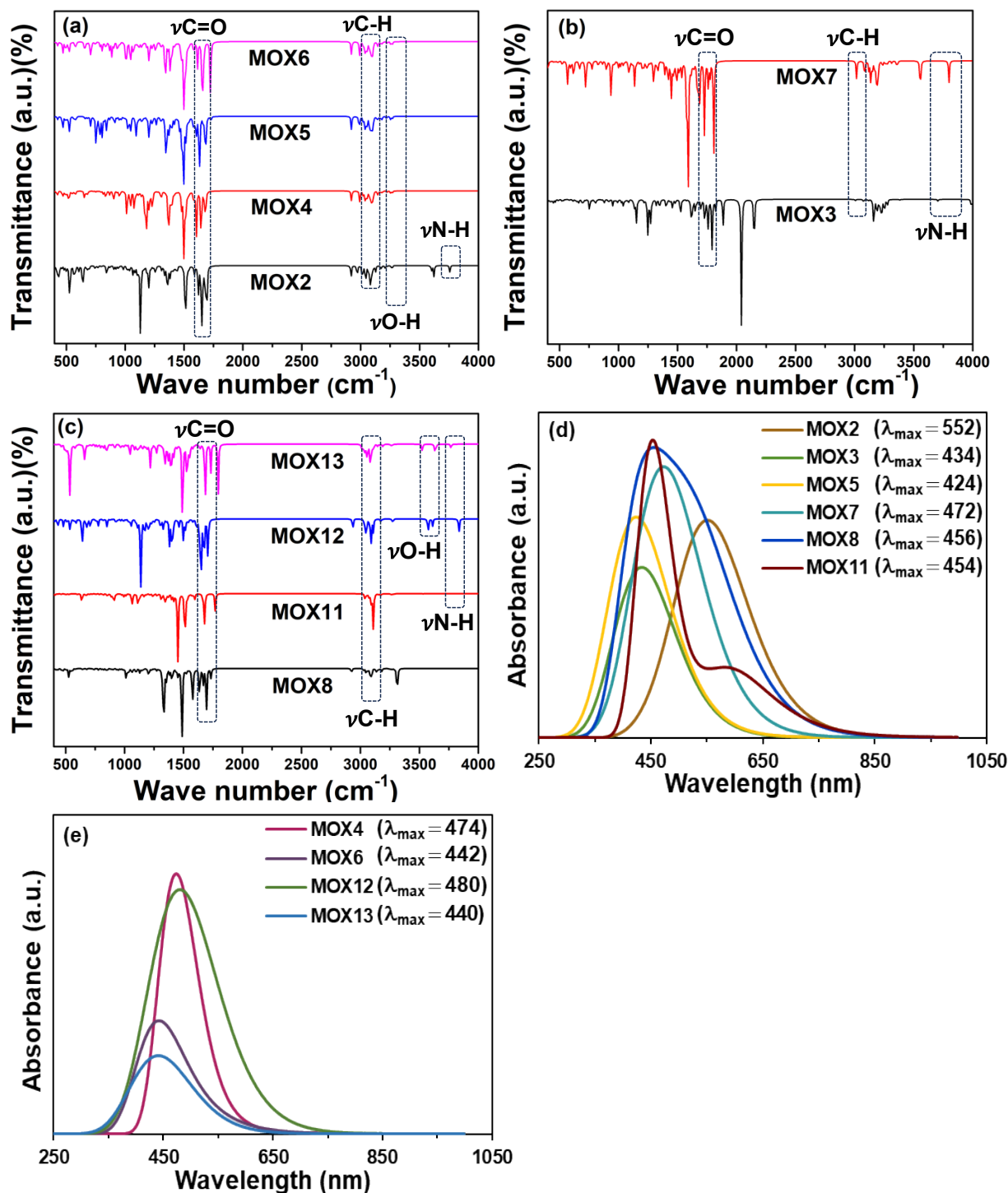


Fig. S4 (a), (b), (c) FT-IR spectra, and (d), (e) UV-visible spectra of MOX derivatives

Table S4 Vibrational frequencies of MOX and derivatives

Name	Assignment	Vibrational frequency (cm ⁻¹)
MOX	ν C-H	3210
	ν C=O	1697
	ν C-C	1436
	ν O-H	3602
	ν C=C ^a	1623
MOX1	ν C-H	3085
	ν C=O	1696
	ν C=C ^a	1624
	ν N-H	3573
	ν O-H	3602
MOX2	ν C-H	3070
	ν N-H	1622
	ν C=C ^a	1506
	ν O-H	3648
	ν C=O	1692
MOX3	ν C-H	3063
	ν C=C ^a	1608
	ν C-F	1174
	ν C=O	1695
MOX4	ν C-H	3176
	ν C=C ^a	1511
	ν C=O	1645
	ν N-H	3578
	ν C-F	1182
MOX5	ν C-H ^a	3096
	ν N-H	3575
	ν C=C ^a	1664
	ν C=O	1630
	ν C-F	1220
MOX6	ν C-H ^a	3254
	ν C=C ^a	1654
	ν C=O	1655
	ν C-F	1299
MOX7	ν N-H	3460
	ν C-H ^a	3096
	ν C=C ^a	1664
	ν C=O	1710
	ν C-F	1297
MOX8	ν N-H	3580
	ν C-H ^a	3255
	ν C=C ^a	1636
	ν C=O	1694
	ν C-F	1291
MOX9	ν N-H	3610

	$\nu\text{C-H}^{\text{a}}$	3087
	$\nu\text{C}=\text{C}$	1660
	$\nu\text{C}=\text{O}$	1692
MOX10	$\nu\text{C-H}^{\text{a}}$	3087
	$\nu\text{N-H}$	3575
	$\nu\text{C}=\text{C}^{\text{a}}$	1644
	$\nu\text{C}=\text{O}$	1686
MOX11	$\nu\text{N-H}$	3522
	$\nu\text{C}=\text{C}^{\text{a}}$	1670
	$\nu\text{C}=\text{O}$	1585
	$\nu\text{C-F}$	1058
MOX12	$\nu\text{N-H}$	3565
	$\nu\text{C-H}$	3087
	$\nu\text{C}=\text{O}$	1663
	$\nu\text{C-F}$	1101
MOX13	$\nu\text{N-H}$	3620
	$\nu\text{C-H}$	3073
	$\nu\text{C}=\text{O}$	1722
	$\nu\text{C}=\text{C}^{\text{a}}$	1483
	$\nu\text{C}=\text{F}$	1149

a = aromatic

Table S5 The binding affinity and nonbonding interactions of MOX and its analogues with the receptor protein.

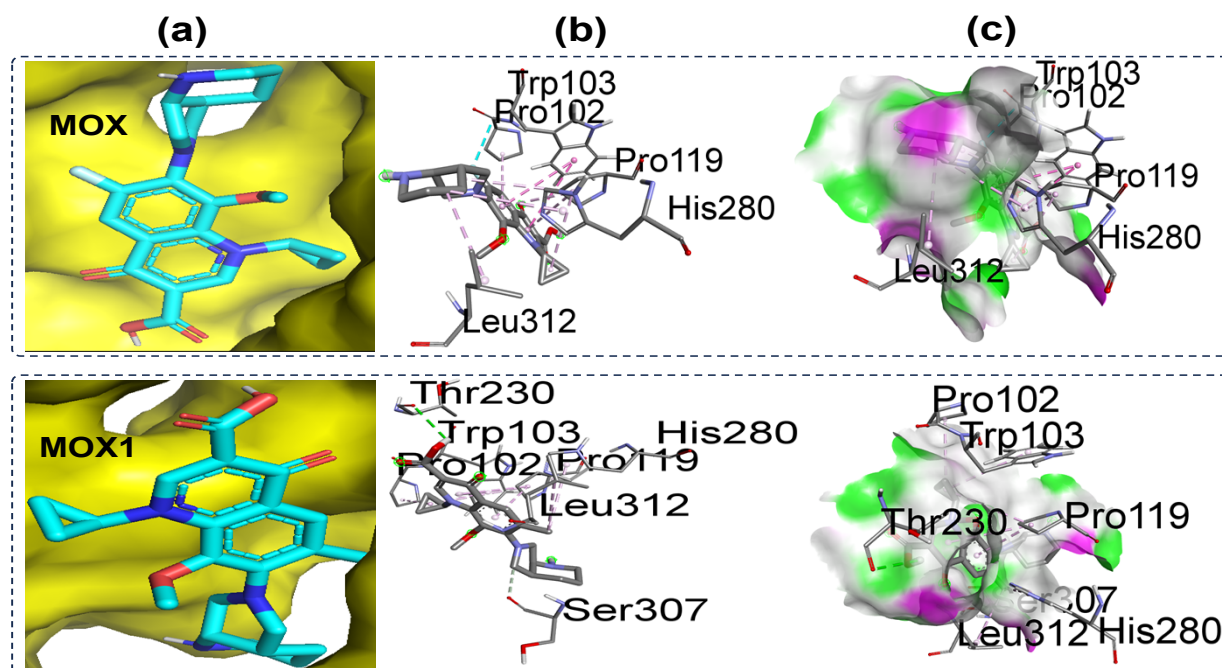
Name	Binding affinity (kcal/mol)	Residues in contact	Interaction's type	Bond distance (Å)
MOX	-7.2	PRO102	HF	3.62
		TRP103	PPT	5.14
		TRP103	PPT	5.10
		PRO119	A	5.29
		LEU312	A	4.71
		PRO119	A	4.95
		PRO119	PA	5.33
		PRO102	PA	5.24
		PRO119	PA	4.74
		HIS280	PA	4.94
MOX1	-8.4	TRP103	H	2.11
		ASP94	H	2.61
		TRP103	H	1.91
		SER104	H	3.09
		GLY120	H	3.75
		GLN101	PDH	2.96
		PRO119	A	5.18
		PRO124	A	4.53
		ARG98	PA	5.11
		ARG98	PA	5.08
		TRP103	PA	5.07
		TRP103	PA	5.02
		TRP103	PA	4.90
		TRP103	H	2.11
MOX2	-7.0	TRP103	PPT	5.18

		TRP103	PPT	5.14
		PRO119	A	5.34
		LEU312	A	4.73
		PRO119	A	4.88
		PRO119	PA	5.26
		PRO102	PA	5.34
		PRO119	PA	4.68
		HIS280	PA	4.99
MOX3	-7.7	ARG98	H	2.60
		ARG98	H	2.29
		GLN101	H	2.20
		SER104	H	2.72
		GLN277	H	2.22
		GLN277	H	2.10
		ASP94	HF	3.51
		GLN101	HF	3.23
		PRO123	A	5.41
		PRO124	A	4.73
		PRO123	A	4.78
		VAL97	A	4.76
		PRO124	A	4.73
		TRP103	PA	5.47
MOX4	-7.8	PRO102	HF	3.21
		GLY117	HF	3.44
		GLY117	HF	3.19
		TRP103	PPT	5.18
		TRP103	PPT	5.17
		LEU312	A	4.43
		PRO119	A	4.57
		PRO119	PA	5.09
		PRO102	PA	5.16
		PRO119	PA	4.78
		HIS280	PA	4.91
MOX5	-7.4	GLY117	H	2.41
		PRO102	HF	3.21
		TRP103	PPT	5.21
		TRP103	PPT	5.16
		TRP103	PPT	5.65
		LEU312	A	4.44
		PRO119	A	4.59
		PRO119	PA	5.03
		PRO102	PA	5.22
		PRO119	PA	4.73
		HIS280	PA	4.92
MOX6	-7.5	PRO102	HF	3.28
		TRP103	PPT	5.36
		TRP103	PPT	5.18
		LEU312	A	4.47
		PRO119	A	4.64
		PRO119	PA	5.00
		PRO102	PA	5.36
		PRO119	PA	4.65
		HIS280	PA	4.90

MOX7	-7.6	ASP308	H	2.91
		PRO102	HF	3.46
		PRO119	A	5.40
		LEU312	A	4.53
		PRO119	A	4.89
		PRO119	PA	4.88
		PRO119	PA	4.43
		HIS280	PA	4.93
MOX8	-8.3	GLY117	H	2.98
		PHE116	H	2.00
		TRP103	PPT	5.07
		TRP103	PPT	5.09
		TRP103	PPT	5.73
		PRO119	A	5.29
		LEU312	A	4.78
		PRO119	A	4.85
		PRO119	PA	5.10
		PRO102	PA	5.40
		PRO119	PA	4.60
		HIS280	PA	5.02
MOX9	-8.3	ASP122	H	2.04
		PRO229	H	2.44
		PRO102	H	2.68
		HIS280	HF	3.26
		PRO119	PA	4.99
		PRO119	PA	4.22
		TRP103	PA	5.34
MOX10	-7.9	TRP103	H	2.03
		ARG98	H	2.66
		TRP103	H	2.08
		GLN277	H	2.66
		SER104	C	2.94
		GLY120	C	3.75
		GLN101	PDH	3.04
		PRO124	A	4.53
		ARG98	PA	5.03
		ARG98	PA	5.28
		PRO124	PA	5.43
		TRP103	PA	5.17
		TRP103	PA	5.18
		TRP103	PA	4.94
MOX11	-7.8	GLY311	C; HF	3.27
		GLY311	C; HF	3.16
		GLY311	HF	3.68
		PRO119	A	5.23
		PRO119	A	5.12
		LEU312	A	5.06
		PRO119	PA	4.68
		PRO119	PA	4.16
		HIS280	PA	4.73
MOX12	-7.2	THR230	H	2.43
		HIS280	C	3.66
		PRO102	A	4.61

MOX13	-8.5	PRO119	PA	5.48
		PRO119	PA	4.50
		GLY120	H	2.45
		ARG98	H; HF	2.27
		TRP103	H	1.96
		SER104	C	3.06
		PRO119	C	3.61
		ARG98	HF	3.50
		PRO124	A	4.64
		PRO124	PA	5.26
		ARG98	PA	4.98
		TRP103	PA	5.31
		TRP103	PA	4.64

H=Conventional hydrogen bond, A=Alkyl, =Pi-Alkyl, HF=Halogen (fluorine), PPT=Pi-Pi T-shaped, PDH=Pi-donor hydrogen bond, C=Conventional carbon bond



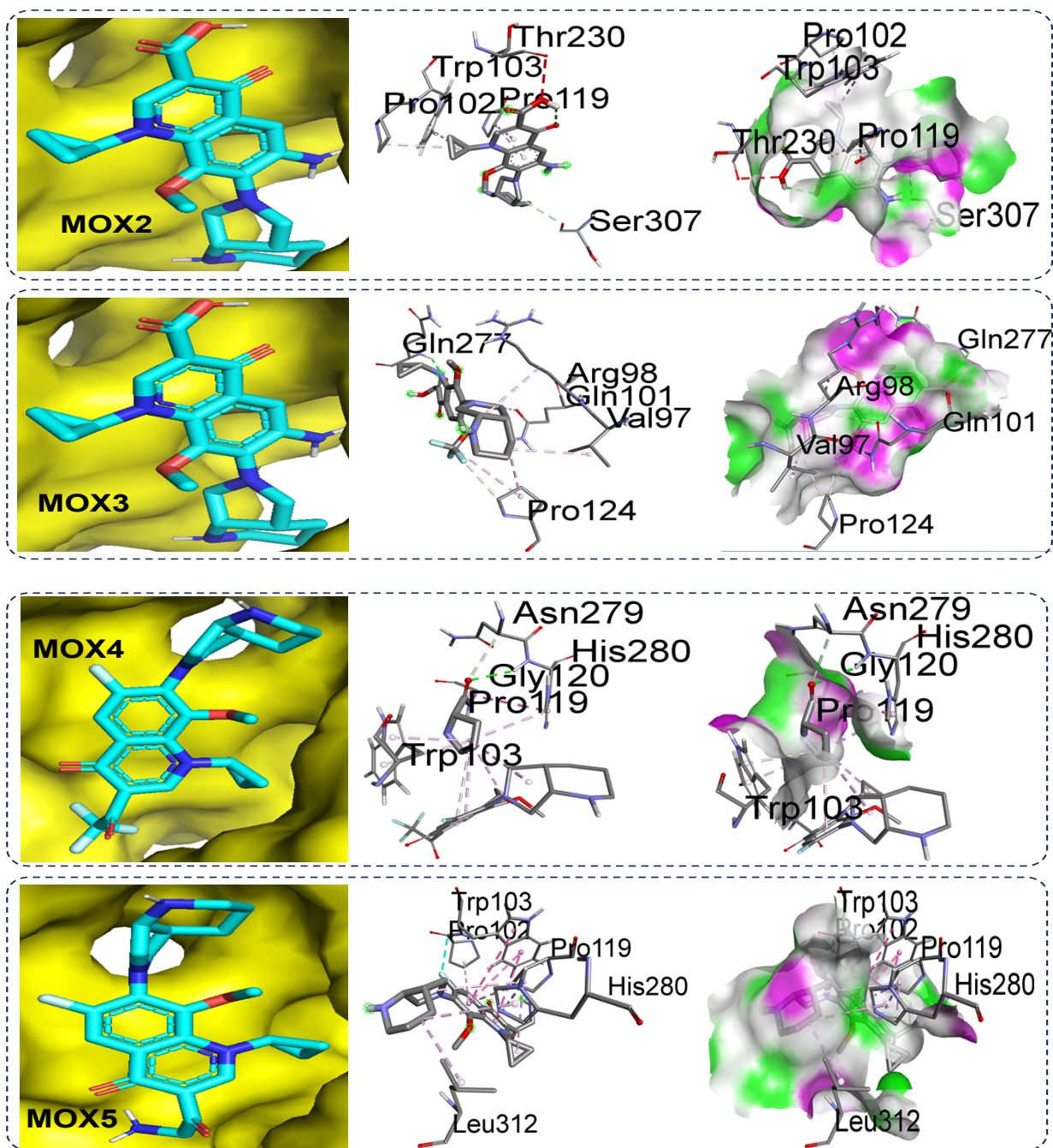


Fig. S5 (a) Docked conformation of MOX and its derivatives at the inhibition binding site of receptor protein 5BS8 (b) Non-bonding interactions and, (c) Hydrogen bond surface area of MOX and its derivatives with the receptor protein.

Table S6 Prediction of biological activities of MOX and some of its derivatives using PASS

Name	AB	AM	AA	TI	DSI	ST	TDP	BC	TC	AS	HT	QTP	HG	KP	TR
MOX	0.66	0.70	0.63	0.82	0.81	0.88	0.84	0.78	0.66	0.73	0.62	0.75	0.80	0.73	0.72
MOX1	0.57	0.63	0.47	0.61	0.67	0.93	0.91	0.89	0.84	0.93	0.86	0.52	0.66	0.92	0.78
MOX2	0.61	0.64	0.33	0.63	0.69	0.92	0.91	0.88	0.84	0.92	0.83	0.54	0.64	0.91	0.83
MOX3	0.58	0.70	0.70	0.55	0.52	0.92	0.80	0.88	0.80	0.94	0.80	0.31	0.43	0.80	0.74
MOX4	0.46	0.48	0.37	0.73	0.54	0.85	0.94	0.85	0.70	0.83	0.67	0.62	0.51	0.90	0.65
MOX5	0.51	0.54	0.40	0.72	0.57	0.90	0.94	0.87	0.76	0.86	0.68	0.62	0.66	0.90	0.76
MOX6	0.48	0.57	0.37	0.68	0.56	0.83	0.93	0.79	0.63	0.76	0.60	0.53	0.53	0.83	0.61
MOX7	0.58	0.66	0.52	0.67	0.59	0.92	0.93	0.89	0.76	0.88	0.76	0.52	0.58	0.81	0.79
MOX8	0.56	0.62	0.62	0.61	0.55	0.90	0.91	0.85	0.72	0.85	0.67	0.50	0.47	0.74	0.75
MOX9	0.53	0.57	0.31	0.60	0.60	0.83	0.83	0.76	0.56	0.71	0.50	0.39	0.54	0.70	0.55
MOX10	0.61	0.67	0.50	0.83	0.70	0.93	0.95	0.95	0.88	0.93	0.88	0.76	0.75	0.95	0.91
MOX11	0.63	0.56	0.63	0.68	0.67	0.90	0.93	0.84	0.85	0.86	0.70	0.49	0.56	0.91	0.59
MOX12	0.55	0.50	0.54	0.56	0.66	0.92	0.92	0.86	0.77	0.88	0.76	0.51	0.58	0.92	0.72
MOX13	0.54	0.40	0.49	0.51	0.62	0.87	0.79	0.73	0.57	0.73	0.57	0.29	0.59	0.78	0.47

AB = Antibacterial, AM = Antimycobacteria, AA = Anti amyloidogenic, TI = Topoisomerase II inhibitor, DSI = DNA synthesis inhibitor, ST = Stomatitis, TDP = Torsades de pointes, BC = Bradycardic, TC = Tachycardic, AS = Asthma, HT = Hepatitis, QTIP = QT interval prolongation, HG = Hyperglycemic, KP = Keratopathy, TR = Tremor, Hyp = Hypertensive.

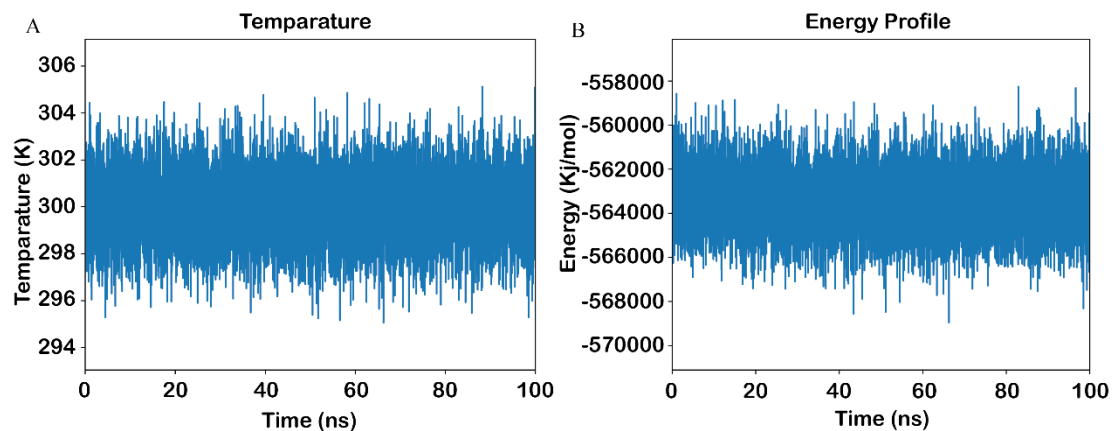


Fig. S6 Potential energy and temperature plots from 100 ns molecular dynamics simulation. Number them as (A) Temperature Plot, and (B) Energy File.