

Supplementary Information

Accelerating the “one-pot” melt polycondensation for thermotropic liquid crystalline polymers by introducing a second acetylating agent

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Supplementary text for characterization.

Supplementary figure-1 ^1H NMR of PGAc, PO, and their mixture at room temperature.

Supplementary figure-2 ^1H NMR of POAc.

Supplementary figure-3 IR absorption spectrum of Vectra A950.

Supplementary figure-4 (a) TGA, and (b) DTG of TLCPs prepared by adding PGAc, HQAc, RCAC, and BPAC without catalyst.

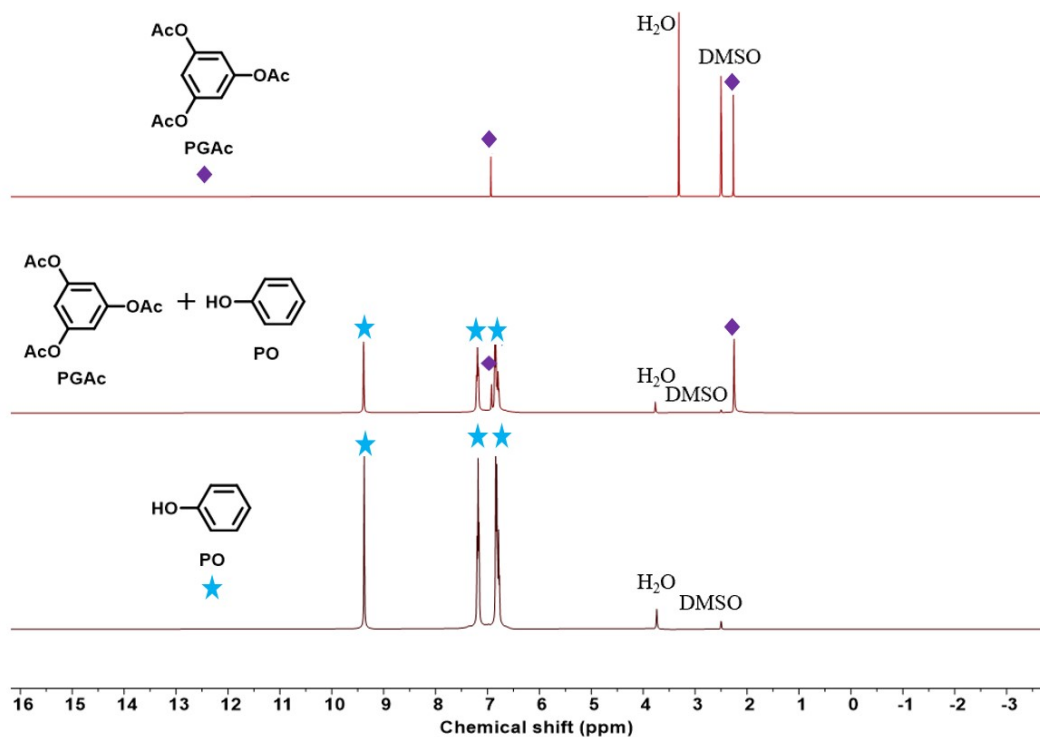
Supplementary figure-5 (a) the second heating curves, and (b) the first cooling curves from DSC of TLCPs prepared by adding PGAc, HQAc, RCAC, and BPAC without catalyst.

Some related ^1H NMR spectra.

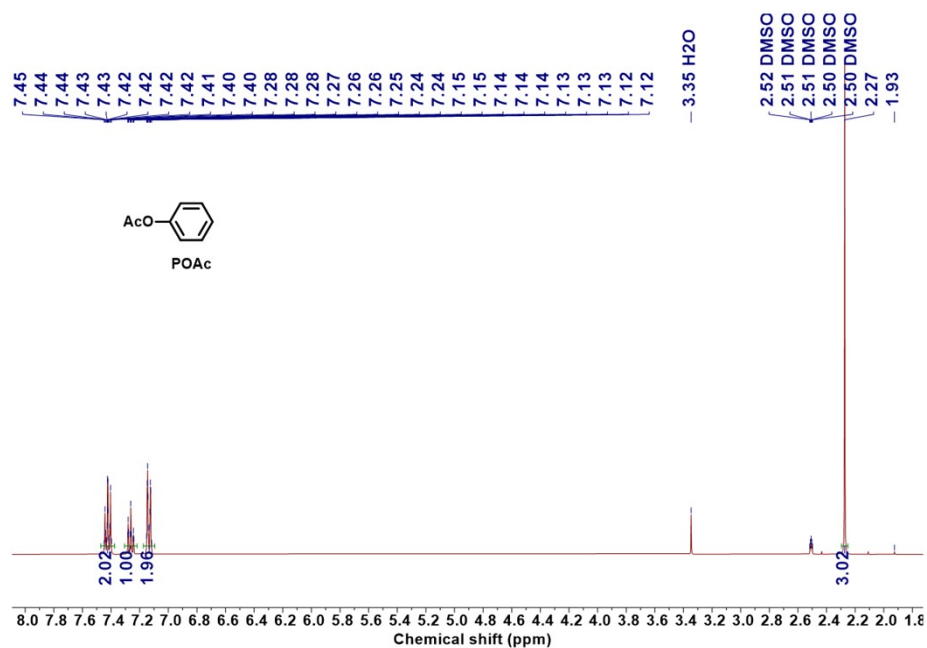
Supplementary text for characterization.

^1H nuclear magnetic resonance (^1H NMR) spectra and fourier transform infrared (FTIR) spectroscopy were used to analyze the structure of products. $\text{DMSO-}d^6$ was used as the testing solvent. FTIR was obtained by total internal reflection with 64 scans and a wavenumber range of 4000 cm^{-1} to 400 cm^{-1} . 100 mg of TLCP was added to a three-necked round-bottom flask containing 50 mL of pentafluorophenol and dissolved at $80\text{ }^\circ\text{C}$ for 4 h to obtain a homogeneous solution. The solution was subjected to hot filtration, and the filtrate was placed in an Ubbelohde viscometer and maintained at $60\text{ }^\circ\text{C}$ for 30 min. The outflow time of the solution was tested and compared with that of pure pentafluorophenol solvent to calculate the intrinsic viscosity ($[\eta]$) of TLCP.

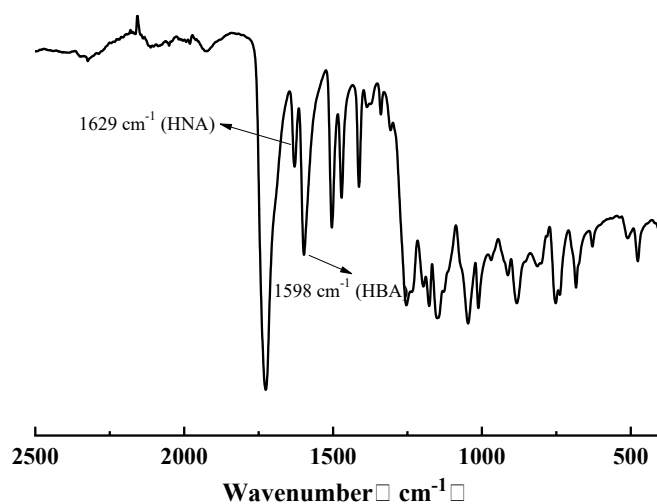
Thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC) analysis were used to study thermal performance of TLCP. After drying in vacuum, TLCP was heated from $100\text{ }^\circ\text{C}$ to $800\text{ }^\circ\text{C}$ at a rate of $20\text{ }^\circ\text{C}/\text{min}$ under nitrogen, and TGA, DTG curves as well as related data were obtained. TLCP was heated to $350\text{ }^\circ\text{C}$ at a rate of $10\text{ }^\circ\text{C}/\text{min}$ in a nitrogen atmosphere. After 5 min of constant temperature, the temperature was reduced to $50\text{ }^\circ\text{C}$ at a rate of $10\text{ }^\circ\text{C}/\text{min}$ to obtain crystallization point (T_c). Maintain a constant temperature for 5 min, the temperature was raised at a rate of $10\text{ }^\circ\text{C}/\text{min}$ to $350\text{ }^\circ\text{C}$ to obtain melting point (T_m).



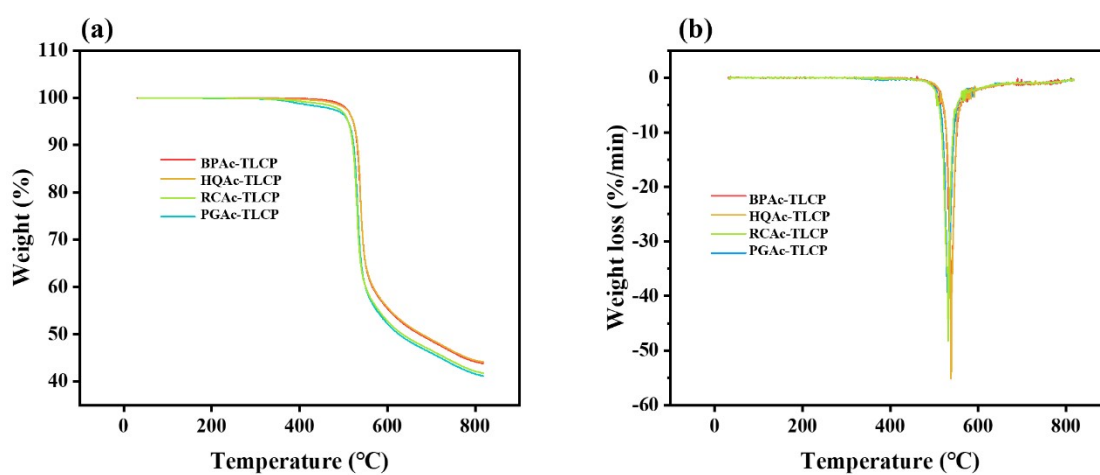
Supplementary figure-1 ^1H NMR of PGAc, PO, and their mixture at room temperature (400 MHz, DMSO- d^6).



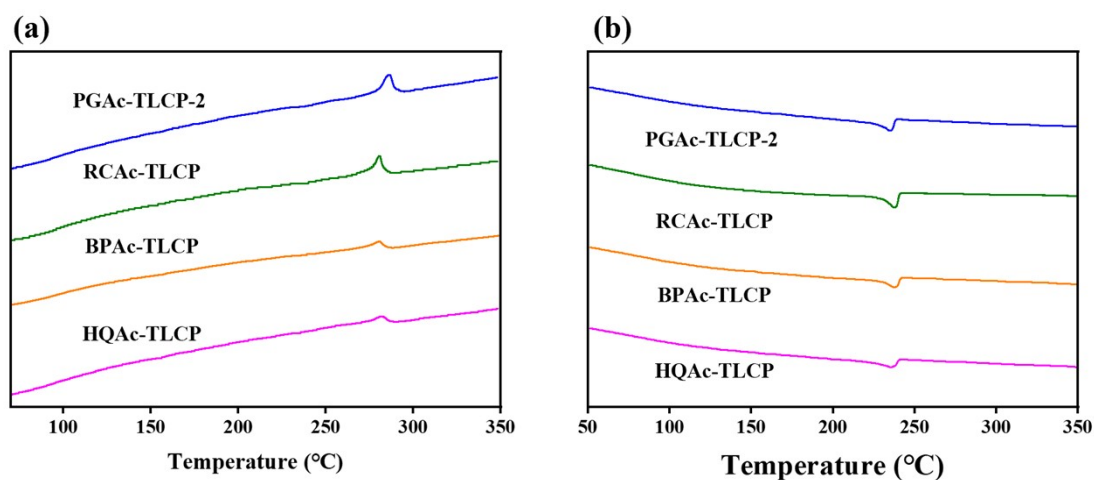
Supplementary figure-2 ^1H NMR of POAc (400 MHz, DMSO- d^6).



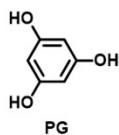
Supplementary figure-3 IR absorption spectrum of Vectra A950.



Supplementary figure-4 (a) TGA, and (b) DTG of TLCPs prepared by adding PGAc, HQAc, RCAC, and BPAC without catalyst.



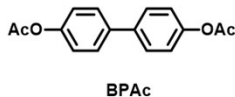
Supplementary figure-5 (a) the second heating curves, and (b) the first cooling curves from DSC of TLCPs prepared by adding PGAc, HQAc, RCAC, and BPAC without catalyst.

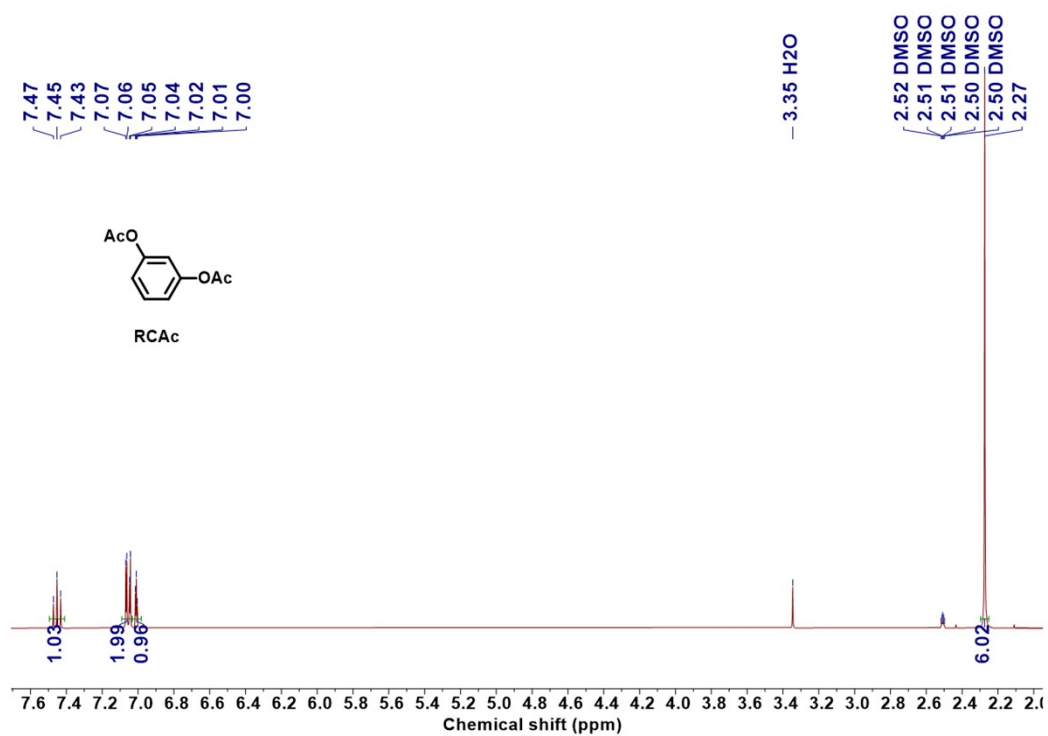
PGAc (400 MHz, DMSO-*d*⁶)

7.72
7.71
7.71
7.69
7.69
7.68
7.25
7.24
7.23
7.22
7.22
7.21

— 3.34 H₂O

2.52 DMSO
2.51 DMSO
2.51 DMSO
2.50 DMSO
2.50 DMSO
2.30

RCAc (400 MHz, DMSO-*d*⁶)



HQAc (400 MHz, DMSO-*d*⁶)

