

Doxorubicin drug delivery using an electrospun nanofiber membrane of chitosan–polycaprolactone with metal-organic framework: Box-Behnken optimization, anticancer treatment, and antimicrobial activity

Table S1. Chemical name, formula, and company.

Chemical name	Formula	Company
Chitosan	$C_{56}H_{103}N_9O_3$	Sigma-Aldrich, Germany
Polycaprolactone	$(C_6H_{10}O_2)_n$	Sigma-Aldrich, Germany
Lanthanum nitrate hexahydrate	$La(NO_3)_3 \cdot 6H_2O$	Sigma-Aldrich, Germany
benzene-1,3,5-tricarboxylic acid	$C_6H_3(CO_2H)_3$	Sigma-Aldrich, Germany
Methanol	CH_3OH	LOBA CHEMIE PVT.LTD, India
Ethanol	C_2H_6O	Sigma-Aldrich, Germany
Sodium hydroxide (99%, AR)	$NaOH$	Chimmed, Russia
Hydrochloric acid (37%, AR)	HCl	LOBA CHEMIE PVT.LTD, India

Table S2. Instruments and equipments.

Test name	Abbreviation	Instrument name	Company	Illustration
Fourier transformer infrared	FT-IR	A Nicolet IS10 Fourier transform infrared (FTIR) spectrometer	Thermo Fisher Scientific, Waltham, MA, USA	equipped with an attenuated total reflectance accessory and which ran in the 4000-400 cm^{-1} range was used to gather FTIR spectra
Powered X-ray diffraction	PXRD	Siemens diffractometer (model D500, Germany)	Germany	patterns were captured from powder samples through the use of a Siemens diffractometer (model D500, Germany) that was fitted with a Cu-K radiation source (wavelength 1.54 Angstroms (\AA)) operating at 30 kV and 20 mA.
Scanning Electron Microscope	SEM	(JSM-6510LV, JEOL Ltd., Tokyo, Japan)	JEOL Ltd., Tokyo, Japan	The morphology of the investigated sorbents was analyzed with the use of a scanning electron microscope
X-ray photoelectron spectroscopy	XPS	K-ALPHA (Thermo Fisher Scientific, USA)	Thermo Fisher Scientific, USA	Used for determination the elemental analysis for the compound
Braunnar Emmet Teller	BET	Quantachrome Instruments, Anton Paar Inc., Quanta Tec, Inc., Boynton Beach, FL, USA	Quanta Tec, Inc., Boynton Beach, FL, USA	was utilised for surface and pore analysis (Brunauer Emmett-Teller (BET) surface area, porous volume, and pore size), and NovaWin Software (v11.0) was used for data interpretation.

		USA			The BET surface area of material adsorbents was obtained by the application of nitrogen adsorption-desorption isotherms at 77K through the use of a specific analyser (Quadratorb-EVO, Quantachrome, USA).
UV-visible spectrophotometer	UV spectrophotometer	Perkin-Elmer AA800 spectrophotometer Double beam, with 1 cm cell length.			Measuring the concentration of the adsorbate solution via using Beer-Lambert law
Energy Dispersive X-ray	EDX	Leo1430VP microscope	Carl Zeiss AG, Jena, Germany		Elemental analysis of the material
pH meter	pH	HANNA (model 211)	USA		Measuring the acidity or basicity of the solution
Sonication	Ultrasonic	Elmasonic ultrasonic bath, continuous mode, power 380 W	P300H Elma Schmidbauer GmbH, Singen, Germany		Sonication of the material as well as used to disperse material on the solution as it decreases the particle size of the material
Water bath	Shaking	GFL Orbital Shaker 3017			

Table S3. Summary of DPPH Antioxidant Activity Evaluation.

Parameter	Description
Location of Analysis	Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University, Egypt.
Assay Type	DPPH (2,2-diphenyl-1-picrylhydrazyl) free radical scavenging assay.
Replicates	Performed in triplicate and averaged.
DPPH Solution	0.004% (w/v) DPPH in methanol; stored at 10°C in the dark.
Sample Preparation	Test compound dissolved in methanol; 40 µL of this solution added to 3 mL DPPH solution.
Reference Compound	Ascorbic acid.
Instrument Used	UV–Visible spectrophotometer (Milton Roy, Spectronic 1201).
Measurement Wavelength	515 nm.
Measurement Time	Absorbance recorded every 1 min for 16 min until stable.
Control	Absorbance of DPPH without antioxidant.
% Inhibition Formula	$PI = \{[(AC - AT) / AC] \times 100\}$ where AC = control absorbance at t=0, AT = sample absorbance at t=16 min.
IC ₅₀ Determination	Calculated from plotted dose-response curve.

Table S4. Cytotoxicity evaluation for MCF-7 and HepG-2.

Parameter	Description
Location of Analysis	Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University, Egypt.
Cell Lines Used	MCF-7 (human breast carcinoma), HepG-2 (human liver carcinoma).
Source	American Type Culture Collection (ATCC, Rockville, MD)
Chemicals & Reagents	DMSO, MTT, Trypan blue (Sigma, USA); RPMI-1640, HEPES buffer, L-glutamine, gentamycin, 0.25% Trypsin-EDTA, Fetal Bovine Serum (Lonza, Belgium).
Culture Medium	RPMI-1640 medium supplemented with 10% inactivated fetal calf serum and 50 µg/mL gentamycin.
Incubation Conditions	Maintained at 37°C in a humidified atmosphere with 5% CO ₂ ; subcultured 2–3 times per week.
Plating Density	5×10^4 cells per well in 96-well plates.
Treatment Setup	Tested compounds added in 8 concentrations; each concentration in triplicates.
Controls	Six vehicle controls per plate using media or 0.5% DMSO.
Viability Assay	Cell viability determined using MTT assay after 24 hours incubation.
MTT Procedure	Replace media with 100 µL RPMI-1640 (no phenol red) + 10 µL MTT (12 mM); incubate 4 h at 37°C/5% CO ₂ .
Solubilization	Remove 85 µL media, add 50 µL DMSO, mix, and incubate 10 min at

Step	37°C.
Measurement	Optical density measured at 590 nm using a SunRise TECAN microplate reader.
Cell Viability Calculation	$\text{Viability (\%)} = (\text{OD}_t / \text{OD}_c) \times 100$
Data Interpretation	Dose–response plotted and IC ₅₀ calculated using GraphPad Prism software.

Table S5. Cytotoxicity Evaluation Table for A-431.

Parameter	Description
Location of Analysis	Regional Center for Mycology and Biotechnology (RCMB), Al-Azhar University, Egypt.
Cell Line Used	A-431 cells (human skin carcinoma cell line).
Source	American Type Culture Collection (ATCC, Rockville, MD).
Chemicals & Reagents	DMSO, MTT, trypan blue (Sigma, USA); RPMI-1640, HEPES buffer, L-glutamine, gentamycin, 0.25% Trypsin-EDTA, Fetal Bovine Serum (Lonza, Belgium).
Culture Medium	RPMI-1640 supplemented with 10% inactivated fetal calf serum and 50 µg/mL gentamycin.
Incubation Conditions	Maintained at 37°C in a humidified 5% CO ₂ atmosphere; subcultured 2–3 times/week.
Sample Preparation	Nanofiber sample (10 mg) soaked in 1 mL RPMI-1640 medium for 24 h before use.
Plating Density	5×10^4 cells/well in 96-well plates; incubated 24 h before treatment.
Treatment Setup	Tested compounds added in 8 concentrations; each in triplicate.
Controls	Six vehicle controls per plate using media or 0.5% DMSO .
Incubation Period Before MTT	48 hours after compound addition.
MTT Procedure	Replace media with 100 µL RPMI-1640 (no phenol red) + 10 µL MTT (12 mM); incubate 4 h at 37°C/5% CO ₂ .
Solubilization Step	Remove 85 µL media; add 50 µL DMSO; mix and incubate 10 min at 37°C.
Measurement	Optical density measured at 590 nm using SunRise TECAN microplate reader.
Viability Calculation	$\% \text{ viability} = (\text{OD}_t / \text{OD}_c) \times 100$
Data Interpretation	Plot survival curve; IC ₅₀ calculated using GraphPad Prism software.

Table S6. True variables, codes, and their BBD levels.

Code	Variables	-1	0	+1
A	pH	5	6.2	7.4
B	Temperature (°C)	25	33.5	42
C	Time (h.)	5	52.5	100

Table S7. Equations used in this work to fit the data of adsorption experiments.

Serial	Equation	Nmae	Description	Ref.
1	$Q_0^{1/3} - Q_t^{1/3} = K_{HC} \cdot t$	Hixson–Crowell model	Q_0 = Initial amount of drug Q_t = Remaining amount of drug at time t K_{HC} = Hixson–Crowell dissolution rate constant t = Time	[1]
2	$Q_t = Q_0 + K_0 \cdot t$	Zero-Order	Q_t = Amount of drug released at time ttt Q_0 = Initial amount of drug in the solution (often 0) K_0 = Zero-order release constant (units: concentration/time) t = Time	[2]
3	$\ln Q_t = \ln Q_0 - K_1 \cdot t$	First order	Q_0 = Initial amount of drug Q_t = Amount of drug remaining at time ttt K_1 = First-order rate constant (1/time) t = Time	[3]
4	$\frac{M_t}{M_\infty} = K \cdot t^n$	Korsmeyer–Peppas	M_t = Amount of drug released at time ttt M_∞ = Total amount of drug released at infinite time (i.e., final amount) $\frac{M_t}{M_\infty}$ = Fraction of drug released at time t K = Kinetic constant incorporating structural and geometric characteristics n = Release exponent that indicates the mechanism of drug release	[4]
5	$Q_t = K_H \sqrt{t}$	Higuchi	Q_t = Cumulative amount of drug released at time t K_H = Higuchi dissolution constant (units: amount/time ^{1/2}) t = Time	[5]

Table S8. The parameter of the kinetic models of DOX release from La-MOF nanofiber membrane

Kinetic model	Value of parameters	
Zero-order	K_o (h)	1.39
	Reduced Chi-Sqr	675.61915
	Residual Sum of Squares	0.95326
	R-Square (COD)	0.9087
	R^2	0.90363
First-order	K_F (h^{-1})	0.4
	Reduced Chi-Sqr	1.40883
	Residual Sum of Squares	0.97737
	R-Square (COD)	0.95526
	R^2	0.95277
Hexson-crowell	K_{HC} (h^{-1})	0.289
	Reduced Chi-Sqr	231.45299
	Residual Sum of Squares	0.92608
	R-Square (COD)	0.85763
	R^2	0.84972
Kosmeyer-peppas	K_F (h^{-1})	0.23
	n	0.00915
	Reduced Chi-Sqr	0.2313
	Residual Sum of Squares	0.92608
	R-Square (COD)	0.85763
	R^2	0.84972
Higuchi	K_H	1.74
	Reduced Chi-Sqr	2083.07687
	Residual Sum of Squares	0.92608
	R-Square (COD)	0.85763
	R^2	0.84972

Table S9. Comparison of different nanocarriers based on MOFs for DOX delivery.

Materials	Drug release	Cell lines	Cell viability	Concentration	Treatment	Ref
LDH-Fe ₃ O ₄ /Cu MOF-DOX-CS@CAR CS (chitosan), CAR (carrageenan hydrogel)	pH 5.5 / 72h / 60% pH 7.4 / 72h / 23%	L929 MCF-7	95% 50%	62.5 µg/mL	48h	[6]
DOX- CS/Fe ₃ O ₄ /Cu-MOF CS (chitosan)	pH 4.5 / 96h / 60% pH 7.4 / 96h / 20%	MCF-7	65%	16 µg/mL	48h	[7]
CS/DOX@Ti-MOF Cs (chitosan)	pH 6.5 / 48h / 76% pH 7.4 / 48h / 10%	MNNG/HOS MDA-MB-231	30% 20%	6 µg/mL	48h	[8]
SiO ₂ @Fe ₃ O ₄ -HA-MIL-100-GQDs-DOX HA (hydroxyapatite), GQDs (graphene quantum dots)	pH 5 / 70h / 67% pH 7.4 / 70h / 29%	MCF-7	5%	32 µg/mL	72h	[9]
Alg-DOX-Cu MOF-LDH Alg (alginate)	pH 5 / 72h / 69% pH 6.8 / 72h / 39% pH 7.4 / 72h / 29%	L929 MCF-7	90% 10%	60 µg/mL	48h	[10]
UiO-66 @P @ DOX P (porphyrin)	pH 4.5/ 200h /90% pH 5.5 / 200h /70% pH 7.4 / 200h /85%	HEK-293 HT-29 MCF-7 MCF-10A	40% 60% 20% 60%	50 µg/mL	48h	[11]

UiO-66 @P @ DOX@RO P (porphyrin), RO (<i>Rosmarinus officinalis</i>)	pH 4.5 / 200h / 40%	HEK-293	80%	50 µg/mL	48h	[11]
		HT-29	80%			
	pH 5.5 / 200h / 60%	MCF-7	65%			
	pH 7.4 / 200h / 50%	MCF-10A	80%			
A520@L@DOX	pH 4.5/ 200h / 94%	HEK-293	95%	50 µg/mL	48h	[12]
		HeLa	65%			
	pH 5.5 / 200h / 97%	MCF-7	76%			
	pH 7.4 / 200h / 96%	PC12	70%			
A520@L@DOX@L	pH 4.5 / 150h / 36%	HEK-293	96%	50 µg/mL	48h	[12]
		HeLa	90%			
	pH 5.5 / 150h / 49%	MCF-7	90%			
	pH 7.4 / 150h / 88%	PC12	83%			
DOX@La-MOF nanofiber membrane	pH 5 / 10h / 94.9%	HepG-2	95.2	94.6 µg/mL	50	This study
		A431	98.4			
	pH 6.2 / 100h / 78.8%					
	pH 7.4 / 100h / 53.48%					

Table S10. Using different MOFs with different coating agents on different cell lines.

MOFs	Coating agents	Cell lines	Ref
Silver-Based MOF	Chitosan	L929	[13]
BioMOF	Chitosan	HUVEC	[14]
UiO-66	Fe ₃ O ₄ Nanoparticles	HeLa, NIH/3T3	[15]
UiO-66	Aloe vera Biopolymer	HFFF2	[16]
UiO-66	PEG	MCF-7	[17]
UiO-68	Aptamer	MDA-MB-23 , MCF-10A	[18]

Cu-MOF	L-lysine	MCF-7 , MCF-10A	[19]
Cu-MOF	Aptamer	Aptamer	[20]
MIL-100(Fe)	Silica	MCF-7 , MCF-10A	[21]
MIL-100(Fe)	PEG	MCF-7	[22]
ZIF-8	Chitosan & Folic acid	MCF-7	[23]
MIL-88B	Chitosan & Folic acid	M109	[24]
Ni/Ta-MOF	Chitosan & Folic acid	MCF-7 , HepG2	[25]
Zn-N-MOF	Chitosan & Folic acid	HCT116	[26]
MOF-5	Chitosan & Alginate	HEK-293 , PC12 , HepG2	[27]
MOF-5	Carboxymethylcellulose, Aptamer	HeLa , 4TA	[28]
UiO-66-NH ₂	Porphyrin	MCF-7 , HT-29	[11]
beta- CD-MOF	Glutamine	MCF-7 , AGS	[29]
Bio-MOF-11	Pectin Biopolymer	SW489	[30]
Fe-BTC MOF	Liposome	MCF-7	[31]
A520	Tp Extract	MCF-7 , HeLa, HEK-293, PC12	[12]
DOX@La-MOF nanofiber membrane	Chitosan and polycaprolactone	HepG-2, A431	This study

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