

## **Supplementary data**

### **Novel pH Responsive Pectin-Based Hybrid Smart Hydrogels for In-Vitro Drug Release and In-Vivo Wound Healing Applications**

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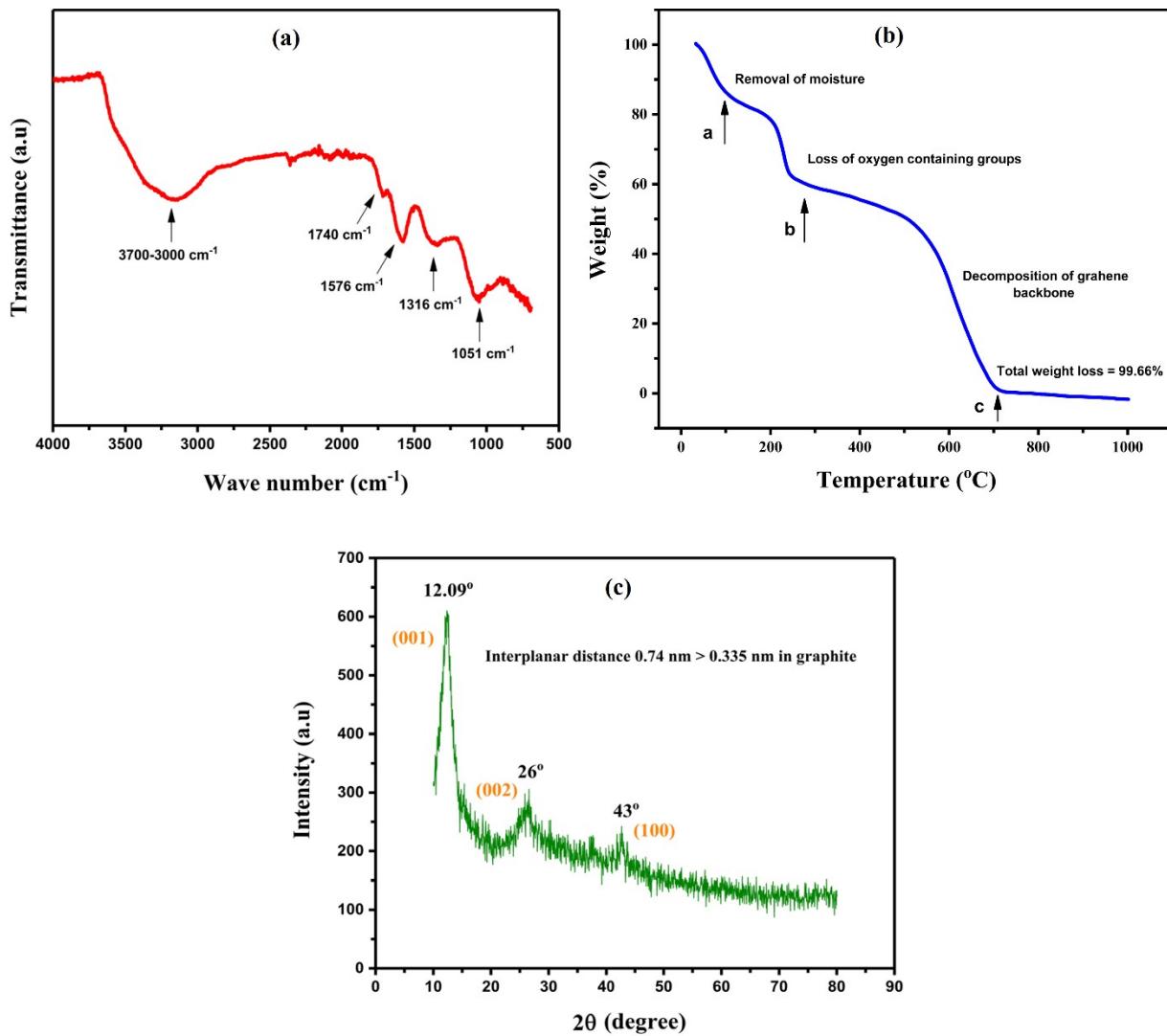
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## Graphene Oxide (GO) Characterization

FTIR spectrum of GO is shown in Fig. S1 (a). A prominent and wide -OH stretching peak was seen at roughly  $3154\text{ cm}^{-1}$ , indicating that the produced GO contained hydroxyl groups. The strong C=O stretching peak was also detected at  $1740\text{ cm}^{-1}$ , while the existence of C=C and C-O-C (epoxy) groups is indicated by the peaks at  $1576\text{ cm}^{-1}$  and  $1051\text{ cm}^{-1}$ , respectively. The peak at  $1316\text{ cm}^{-1}$  exhibits C-O stretching bond [1].

The graphene oxide thermogram, shown in Fig. S1 (b), reflected three-step decomposition. Up to  $100\text{ }^{\circ}\text{C}$ , the elimination of bound water is thought to be responsible for the early deterioration in GO. In phase two, 22.27% of GO was broken down. For the second phase, oxygen-containing functionalities (-COOH, -OH, -C=O) are responsible for rapid breakdown between the temperature range of  $100$  to  $250\text{ }^{\circ}\text{C}$ . In the third step, a gradual decrease in weight was seen at temperatures between  $250$  to  $700\text{ }^{\circ}\text{C}$ . The thermally stable graphitic part was confirmed to exist by this third stage of disintegration, which was decomposed completely upto the temperature of  $700\text{ }^{\circ}\text{C}$  [2].

Fig. S1 (c) displays the XRD results of Graphene oxide (GO). GO showed a strong and intense XRD peak at a  $2\theta$  value of  $12.09^{\circ}$ , which corresponded to the (001) with an interplanar spacing of  $0.74\text{ nm}$ . The presence of reactive functional groups with oxygen was confirmed by the increase in interplanar spacing of graphene oxide ( $0.74\text{ nm}$ ) as compared to that of pure graphite ( $0.335\text{ nm}$ ).  $26^{\circ}$  peak shows interlayer stacking of graphene sheets and  $43^{\circ}$  confirms in plane structure of graphene sheets. This confirms the oxidation of graphite to graphene oxide [3].



**Fig. S1.** (a) FTIR spectrum (b) TGA thermogram and (c) XRD spectra of graphene oxide.

## References

[1] P. Viprya, D. Kumar & S. Kowshik, Study of different properties of graphene oxide (GO) and reduced graphene oxide (rGO), Eng. Proc. 59(1) (2023) 84. <https://doi.org/10.3390/engproc2023059084>

[2] D. Losic, F. Farivar & P. L. Yap, Refining and Validating Thermogravimetric Analysis (TGA) for Robust Characterization and Quality Assurance of Graphene-Related Two-

[3] S. M. Alduwaib & M. M. Abd, Investigating the Structural, Optical and Antibacterial Properties of Go, Go: Ag, GO: ZnO Thin Layers and Go: ZnO/GO: Ag Bilayers Synthesized by Spray Pyrolysis Method. *Iran. J. Mater. Sci. Eng.* 17(4) (2020) 170. <http://dx.doi.org/10.22068/ijmse.17.4.170>

**Table S1**

Residual weight of GO reinforced Pectin/PVA/APTES/GO hydrogels in PBS solution with respect to time.

Time (days)	Weight (mg)				
	PPG (control)	PPG-2.5	PPG-5	PPG-7.5	PPG-10
0	30	30	30	30	30
1	19.02	22.21	23.5	25.17	26.33
3	9.98	13.46	16.5	15.38	17.86
7	8.17	10.99	12.11	13.29	15
10	6.22	9.05	9.86	11.47	12.73
14	4.08	6.36	8.14	9.19	10.05
21	2.06	3.28	4.54	5.02	6.6

**Table S2**

Residual weight of GO reinforced Pectin/PVA/APTES/GO hydrogels in Proteinase-K solution with respect to time.

Time (days)	Weight (mg)				
	PPG (control)	PPG-2.5	PPG-5	PPG-7.5	PPG-10
0	30	30	30	30	30
1	9.7	12.59	12	14.40	16.88
3	5.11	6.73	8.5	11.34	12.16
7	2.20	3.42	4.59	5.10	6.27

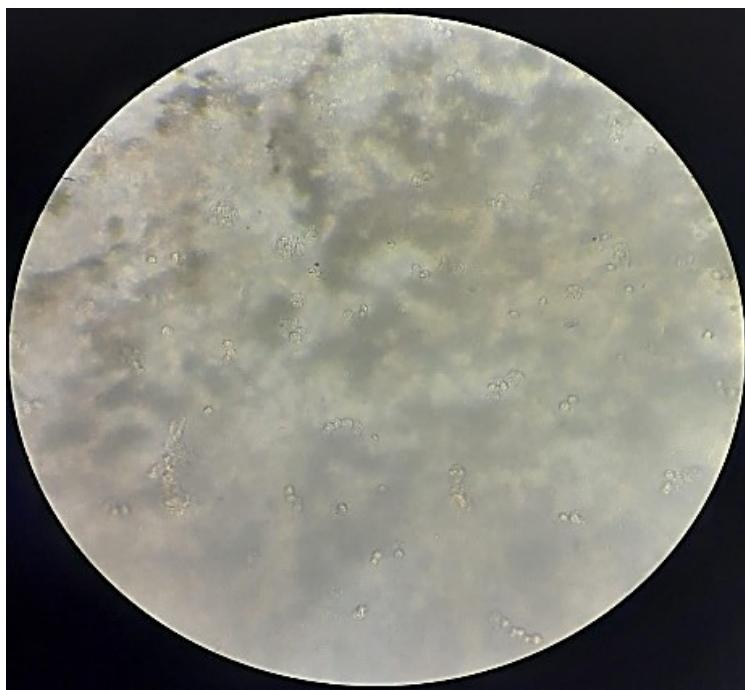
**Table S3**

Drug release data of GO reinforced Pectin/PVA/APTES/GO hydrogel in PBS solution at pH 7.4.

Time (mins)	Concentration (mg)	Drug release (%)	Square root of T	$M_t/M_o$	Log T	Log release	drug
10	12.234	24.469	3.162	0.244	1.000	1.388	
20	15.967	31.934	4.472	0.319	1.301	1.504	
30	18.552	37.104	5.477	0.371	1.477	1.569	
40	21.031	42.063	6.324	0.420	1.602	1.623	
50	22.812	45.624	7.071	0.456	1.698	1.659	
60	24.988	49.977	7.745	0.499	1.778	1.698	
70	26.452	52.905	8.366	0.529	1.845	1.723	
80	28.945	57.890	8.944	0.578	1.903	1.762	
90	30.409	60.818	9.486	0.608	1.954	1.784	
100	32.110	64.221	10.00	0.642	2.000	1.807	
110	33.522	67.044	10.488	0.670	2.041	1.826	
120	35.988	71.976	10.954	0.719	2.079	1.857	
130	37.043	74.087	11.401	0.740	2.113	1.869	
140	39.496	78.993	11.832	0.789	2.146	1.897	
150	40.947	81.895	12.247	0.818	2.176	1.913	
160	41.725	83.451	12.649	0.834	2.204	1.921	
170	43.057	86.115	13.038	0.861	2.230	1.935	
180	43.453	86.906	13.416	0.869	2.255	1.939	
190	43.572	87.144	13.784	0.871	2.278	1.940	
200	45.049	90.098	14.142	0.900	2.301	1.954	
210	45.721	91.443	14.491	0.914	2.322	1.961	
220	45.721	91.443	14.832	0.914	2.342	1.961	
230	45.721	91.443	15.165	0.914	2.361	1.961	



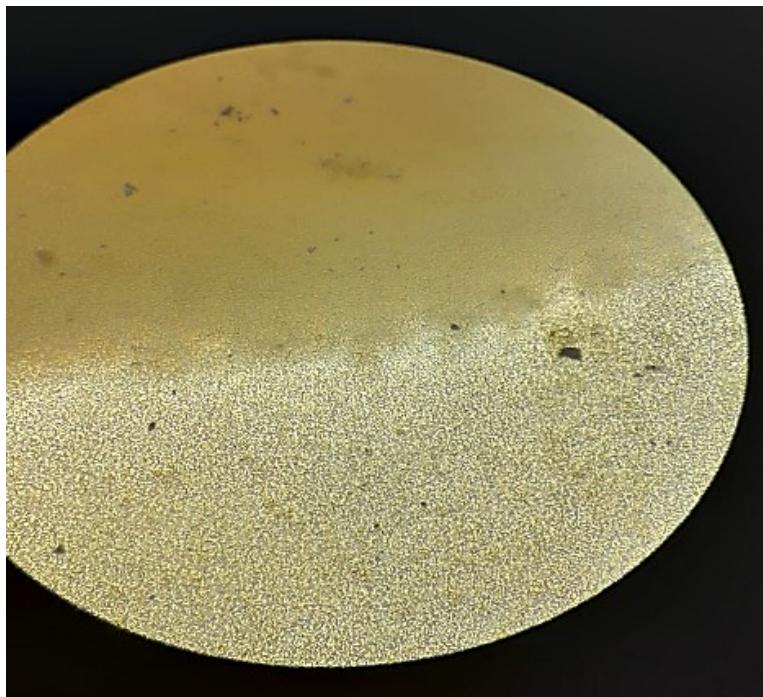
**Fig. S2.** Picture of cell viability analysis of the Reference.



**Fig. S3.** Picture of cell viability analysis of PPG (control).



**Fig. S4.** Picture of cell viability analysis of PPG-2.5.



**Fig. S5.** Picture of cell viability analysis of PPG-5.



**Fig. S6.** Picture of cell viability analysis of PPG-7.5.



**Fig. S7.** Picture of cell viability analysis of PPG-10.