

## Supplementary Information

### **Fly Ash Upcycling to Functional Silica Nanomaterials: Insights into Synthetic Strategies towards Efficient adsorbents for Water Purification**

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**Table S1.** Elemental analysis of FA. The sample underwent microwave-assisted digestion in a mixture of nitric acid, hydrochloric acid, and hydrofluoric acid at elevated temperature. The total element content in the solution was then measured with Inductively Coupled Plasma Sector Field Mass Spectrometry (ICP-SFMS; Thermo Scientific Element XR)

<b>Element</b>	<b>Amount (mg/kg dry FA)</b>
<b>Al</b>	31900
<b>As</b>	16.6
<b>Ba</b>	984
<b>Be</b>	1.16
<b>Ca</b>	89700
<b>Cd</b>	2.55
<b>Co</b>	16.4
<b>Cr</b>	99.6
<b>Cu</b>	663
<b>Fe</b>	99600
<b>Hg</b>	0.0536
<b>K</b>	15200
<b>Mg</b>	21700
<b>Mn</b>	566
<b>Mo</b>	8.09
<b>Na</b>	7460
<b>Nb</b>	7.72
<b>Ni</b>	76.3
<b>P</b>	85000
<b>Pb</b>	142
<b>S</b>	3300
<b>Sb</b>	9.72
<b>Sc</b>	4.77
<b>Si</b>	165000
<b>Sn</b>	49.7
<b>Sr</b>	1510
<b>Ti</b>	4920
<b>V</b>	62.2
<b>W</b>	18.9
<b>Y</b>	15.3
<b>Zn</b>	2600
<b>Zr</b>	148
LOI 1000°C	1.52 %

**Table S2.** Elemental analysis of FA residue after acid treatment. The sample underwent microwave-assisted digestion in a mixture of nitric acid, hydrochloric acid, and hydrofluoric acid at elevated temperature. The total element content in the solution was then measured with Inductively Coupled Plasma Sector Field Mass Spectrometry (ICP-SFMS; Thermo Scientific Element XR)

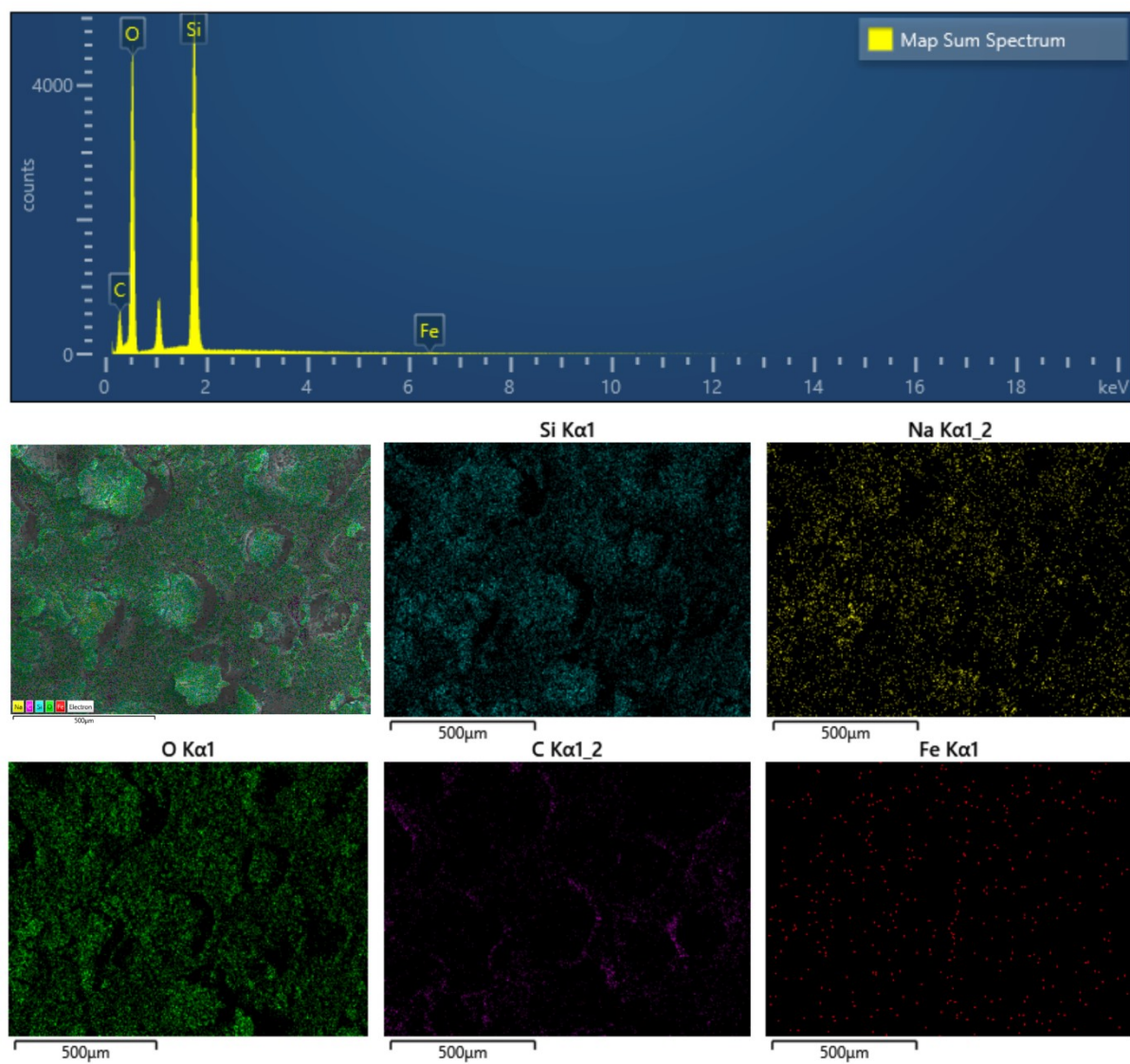
<b>Element</b>	<b>Amount (mg/kg dry FA)</b>
<b>Al</b>	38400
<b>As</b>	3.16
<b>Ba</b>	2350
<b>Be</b>	1.08
<b>Ca</b>	33400
<b>Cd</b>	0.145
<b>Co</b>	4.1
<b>Cr</b>	123
<b>Cu</b>	132
<b>Fe</b>	15800
<b>Hg</b>	<0.05
<b>K</b>	21900
<b>Mg</b>	3550
<b>Mn</b>	119
<b>Mo</b>	1.66
<b>Na</b>	8850
<b>Nb</b>	14.4
<b>Ni</b>	20.2
<b>P</b>	7810
<b>Pb</b>	47.2
<b>S</b>	1620
<b>Sb</b>	9.15
<b>Sc</b>	2.79
<b>Si</b>	343000
<b>Sn</b>	19.4
<b>Sr</b>	550
<b>Ti</b>	9350
<b>V</b>	37
<b>W</b>	19
<b>Y</b>	9.67
<b>Zn</b>	316
<b>Zr</b>	238
LOI 1000°C	2.84 %

**Table S3.** FA composition before and after Si extraction measured by EDS analysis

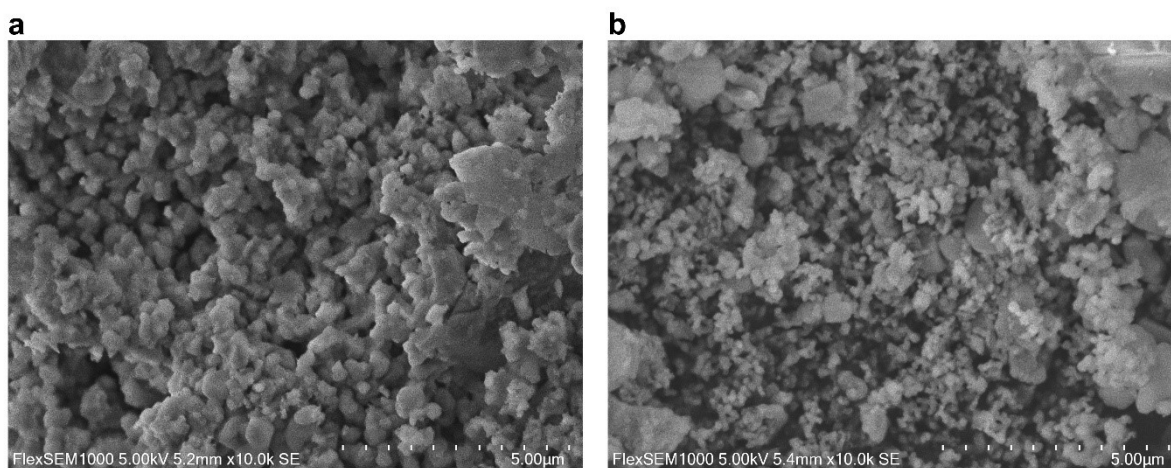
Element	FA	Si extracted FA
	Atomic %	Atomic %
O	58.88	57.12
Si	17.96	7.63
K	0.99	0.69
Ti	0.74	1.22
Fe	0.68	1.12
P	0.55	0.02
Na	0.48	13.47
Mg	0.30	0.19
Al	2.41	1.67
Ca	0.14	0.14
C	15.03	15.86
N	1.71	0.85
S	0.11	0.00
Total	100.00	100.00

**Table S4.** Elemental composition of SSS measured by EDS analysis. The SSS solution was dried on the carbon tape prior the measurement

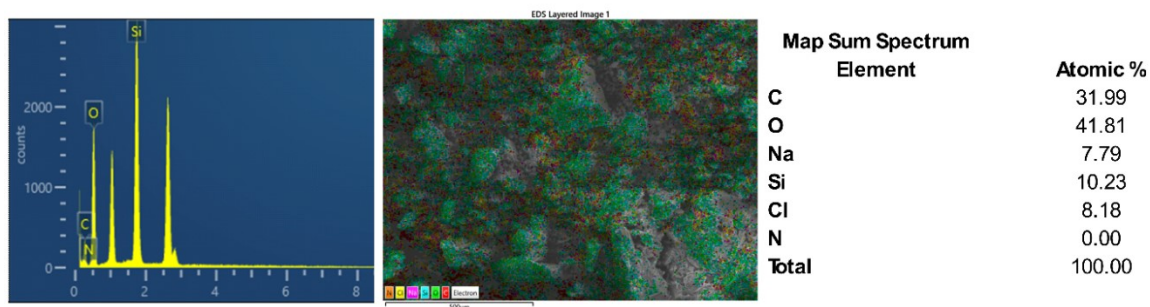
Element	SSS solution
	Atomic %
C	23.87
O	53.80
Na	20.54
Si	1.29
P	0.15
N	0.34
Total	100.00



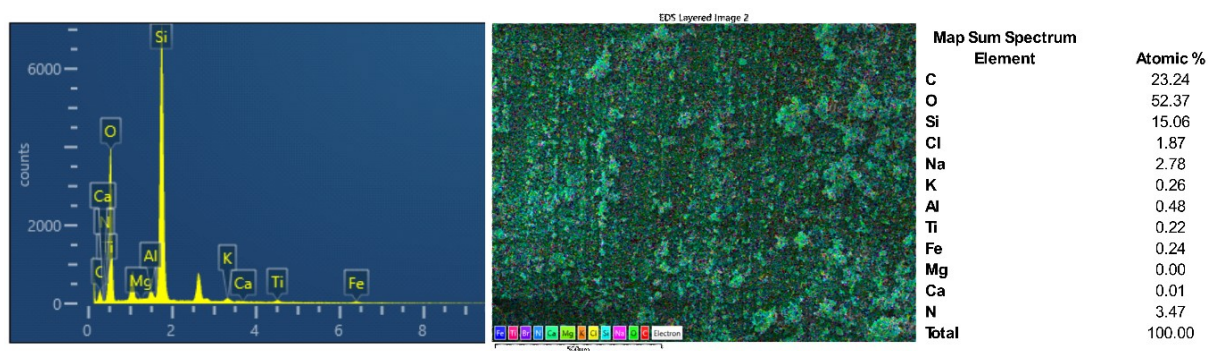
**Fig. S1** EDS analysis of dense SiO<sub>2</sub> NPs derived from FA



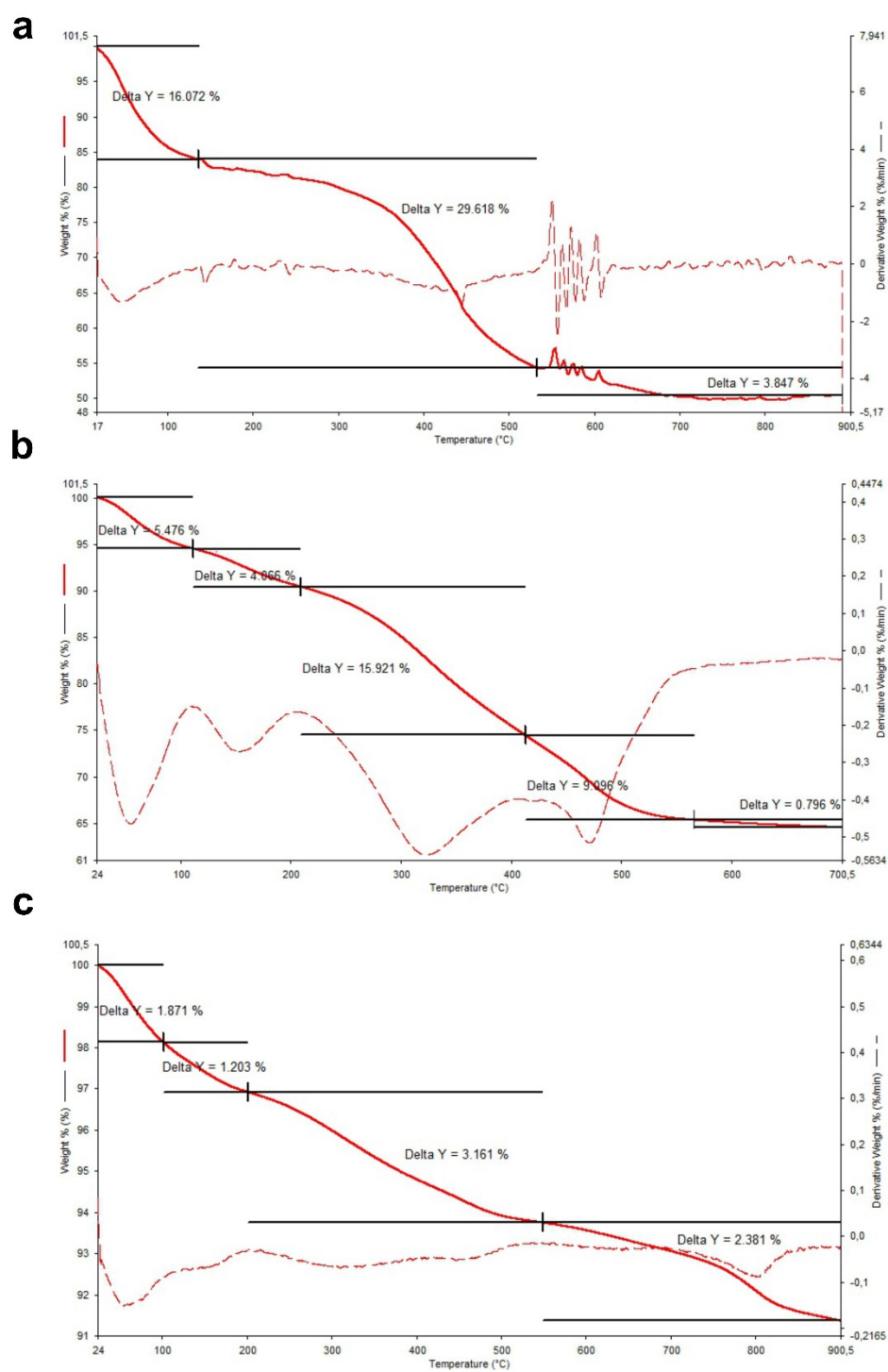
**Fig. S2** SEM images of synthesized  $M\_SiO_2$  NPs (a) and  $1\ pot\_M\_SiO_2$  (b)



**Fig. S3** EDS analysis of M\_SiO<sub>2</sub> NPs derived from FA

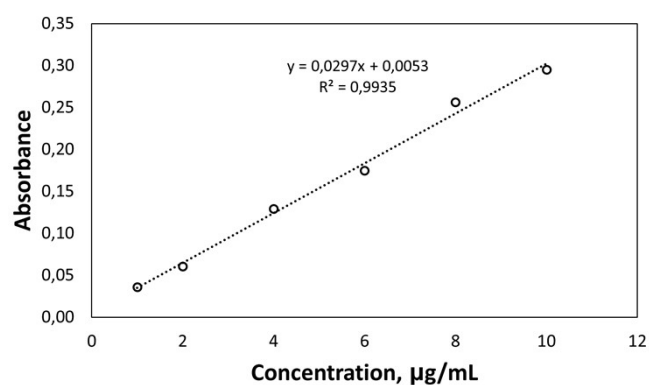


**Fig. S4** EDS analysis of 1 pot\_M\_SiO<sub>2</sub> NPs derived from FA



**Fig. S5** TGA curves of synthesized SiO<sub>2</sub> nanomaterials grafted with TMSPDETA ligand: (a) M\_SiO<sub>2</sub>, (b) D\_SiO<sub>2</sub> and (c) 1 pot\_M\_SiO<sub>2</sub> NPs

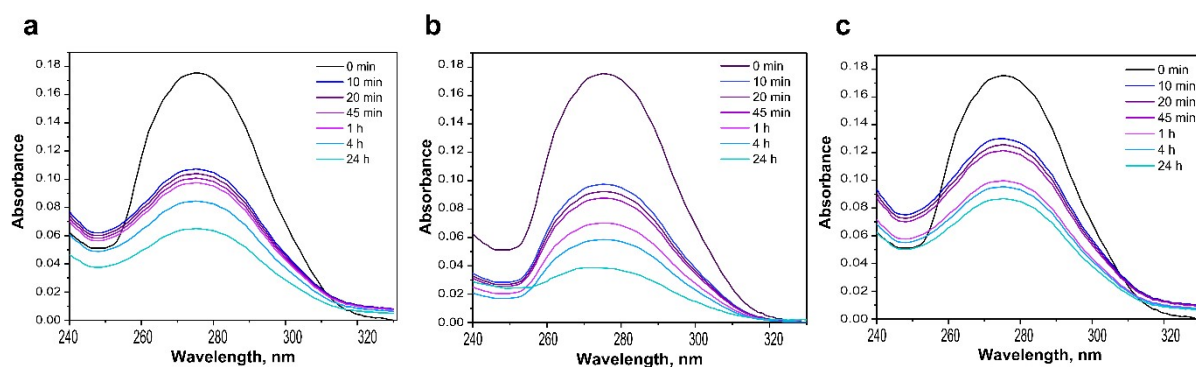




**Fig. S6** Calibration curve DFC measured by UV-Vis with DFC standard concentrations varying between 1-10 µg/mL

**Table S5.** Amount of grafted TMSPDETA ligand on synthesized SiO<sub>2</sub> NPs

Sample	Grafted TMSPDETA (mmol/g)
D_SiO <sub>2</sub>	0.94
M_SiO <sub>2</sub>	1.12
1 pot_M_SiO <sub>2</sub>	0.2



**Fig. S7** Adsorption kinetics of DFC on SiO<sub>2</sub> nanomaterials: (A) D\_SiO<sub>2</sub>, (B) M\_SiO<sub>2</sub> and (C) 1 pot\_M\_SiO<sub>2</sub> NPs