

Supplementary information for:

**From Structure to Function: A One-Dimensional Palladium(II)  
Hybrid for Real-Sample Hydroquinone Detection**

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**Table S1.** Crystal data and refinement parameters of  $(\text{C}_4\text{H}_7\text{N}_2)[\text{PdCl}_3(\text{H}_2\text{O})]$ .

Formula	$(\text{C}_4\text{H}_7\text{N}_2)[\text{PdCl}_3(\text{H}_2\text{O})]$
$D_{\text{calc.}}/\text{g cm}^{-3}$	2.126
$\mu/\text{mm}^{-1}$	2.658
Formula Weight	313.88
Colour	colourless
Shape	Block-shaped
Size/ $\text{mm}^3$	0.59×0.23×0.18
$T/\text{K}$	100.00(10)
Crystal System	monoclinic
Space Group	$P2_1/c$
$a/\text{Å}$	9.0007(2)
$b/\text{Å}$	14.7508(4)
$c/\text{Å}$	7.4059(2)
$\alpha^\circ$	90
$\beta^\circ$	94.084(2)
$\gamma^\circ$	90
$V/\text{Å}^3$	980.77(4)
$Z$	4
$Z'$	1
Wavelength/Å	0.71073
Radiation type	Mo $K_\alpha$
$\theta_{\text{min}}^\circ$	5.393
$\theta_{\text{max}}^\circ$	30.208
Measured Refl's.	14007
Indep't Refl's	2517
Refl's $I \geq 2 \sigma(I)$	2320
$R_{\text{int}}$	0.0916
Parameters	103
Restraints	0
Largest Peak	0.872
Deepest Hole	-0.894
GooF	1.044
$wR_2$	0.1063
$R_1$	0.0389

**Table S2.** Molecular geometric parameters (Å, °) of (C<sub>4</sub>H<sub>7</sub>N<sub>2</sub>)[PdCl<sub>3</sub>(H<sub>2</sub>O)].

<b>Bond length (Å)</b>			
Pd1—Cl3	2.6143 (8)	N2—H2	0.8800
Pd1—Cl3i	2.6508 (9)	N2—C2	1.387 (4)
Pd1—Cl2	2.5762 (8)	N2—C1	1.327 (4)
Pd1—Cl2ii	2.5886 (9)	C2—H2A	0.9500
Pd1—Cl1	2.5984 (8)	C2—C3	1.348 (5)
Pd1—O1	2.391 (2)	C4—H4A	0.9800
O1—H1A	0.8514	C4—H4B	0.9800
O1—H1B	0.8516	C4—H4C	0.9800
N1—H1	0.8800	C4—C1	1.478 (5)
N1—C1	1.335 (4)	C3—H3	0.9500
N1—C3	1.375 (4)		
<b>Bond angles (°)</b>			
Cl3—Pd1—Cl3i	89.62 (3)	C1—N1—C3	110.6 (3)
Cl2ii—Pd1—Cl3i	172.16 (3)	C3—N1—H1	124.7
Cl2—Pd1—Cl3	169.09 (3)	C2—N2—H2	124.8
Cl2—Pd1—Cl3i	88.10 (3)	C1—N2—H2	124.8
Cl2ii—Pd1—Cl3	88.62 (3)	C1—N2—C2	110.4 (3)
Cl2—Pd1—Cl2ii	92.21 (3)	N2—C2—H2A	126.9
Cl2—Pd1—Cl1	96.01 (3)	C3—C2—N2	106.3 (3)
Cl2ii—Pd1—Cl1	97.67 (3)	C3—C2—H2A	126.9
Cl1—Pd1—Cl3	94.66 (2)	H4A—C4—H4B	109.5
Cl1—Pd1—Cl3i	90.09 (3)	H4A—C4—H4C	109.5
O1—Pd1—Cl3	81.11 (6)	H4B—C4—H4C	109.5
O1—Pd1—Cl3i	84.79 (6)	C1—C4—H4A	109.5
O1—Pd1—Cl2	88.06 (6)	C1—C4—H4B	109.5
O1—Pd1—Cl2ii	87.40 (6)	C1—C4—H4C	109.5
O1—Pd1—Cl1	173.36 (6)	N1—C1—C4	126.8 (3)
Pd1—Cl3—Pd1ii	90.12 (3)	N2—C1—N1	106.2 (3)
Pd1—Cl2—Pd1i	92.37 (3)	N2—C1—C4	127.0 (3)
Pd1—O1—H1A	109.5	N1—C3—H3	126.7
Pd1—O1—H1B	109.3	C2—C3—N1	106.5 (3)
H1A—O1—H1B	104.4	C2—C3—H3	126.7
C1—N1—H1	124.7		
N2—C2—C3—N1	0.3 (4)	C1—N2—C2—C3	0.0 (4)
C2—N2—C1—N1	-0.3 (4)	C3—N1—C1—N2	0.5 (4)
C2—N2—C1—C4	179.8 (3)	C3—N1—C1—C4	-179.7 (3)
C1—N1—C3—C2	-0.5 (4)		
Symmetry codes : (i) x, -y+1/2, z-1/2; (ii) x, -y+1/2, z+1/2.			

**Table S3.** Hydrogen-bond geometry (Å, °) of (C<sub>4</sub>H<sub>7</sub>N<sub>2</sub>)[PdCl<sub>3</sub>(H<sub>2</sub>O)].

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O1—H1 <i>A</i> ···Cl1 <sup>i</sup>	0.85	2.36	3.194 (3)	165
O1—H1 <i>B</i> ···Cl1 <sup>iii</sup>	0.85	2.40	3.146 (2)	147
N1—H1···Cl3 <sup>ii</sup>	0.88	2.89	3.452 (3)	124
N1—H1···O1	0.88	2.05	2.845 (4)	150
N2—H2···Cl1 <sup>iv</sup>	0.88	2.28	3.144 (3)	166
C3—H3···Cl2 <sup>v</sup>	0.95	2.87	3.550 (4)	129
C3—H3···Cl1 <sup>iii</sup>	0.95	2.89	3.713 (3)	145
<b>Symmetry codes:</b> (i) <i>x</i> , $-y+1/2$ , $z+1/2$ ; (ii) <i>x</i> , $-y+1/2$ , $z-1/2$ ; (iii) $-x+1$ , $y+1/2$ , $-z+1/2$ ; (iv) $-x+2$ , $y+1/2$ , $-z+1/2$ ; (v) $-x+1$ , $-y+1$ , $-z$ .				