

Supporting information

P-N Heterojunction-Enhanced Structural Stability of Tungsten Disulfide Nanosheets for Highly Efficient, Stable Alkaline Water Electrolysis

Kumasser Kusse Kuchayita,^a Yohannes Asmare Fesseha,^a Chih-Wei Chiu,^c Jem-Kun Chen,^c Ai-

Wei Lee^{d} and Chih-Chia Cheng^{ab*}*

- a. Graduate Institute of Applied Science and Technology, National Taiwan University of Science and Technology, Taipei 10607, Taiwan. E-mail: cccheng@mail.ntust.edu.tw
- b. Advanced Membrane Materials Research Center, National Taiwan University of Science and Technology, Taipei 10607, Taiwan.
- c. Department of Materials Science and Engineering, National Taiwan University of Science and Technology, Taipei 10607, Taiwan.
- d. Department of Anatomy and Cell Biology, School of Medicine, College of Medicine, Taipei Medical University, Taipei, 11031, Taiwan. E-mail: ammielee@tmu.edu.tw

* Corresponding author is marked with an asterisk (*) in the complete list of authors.

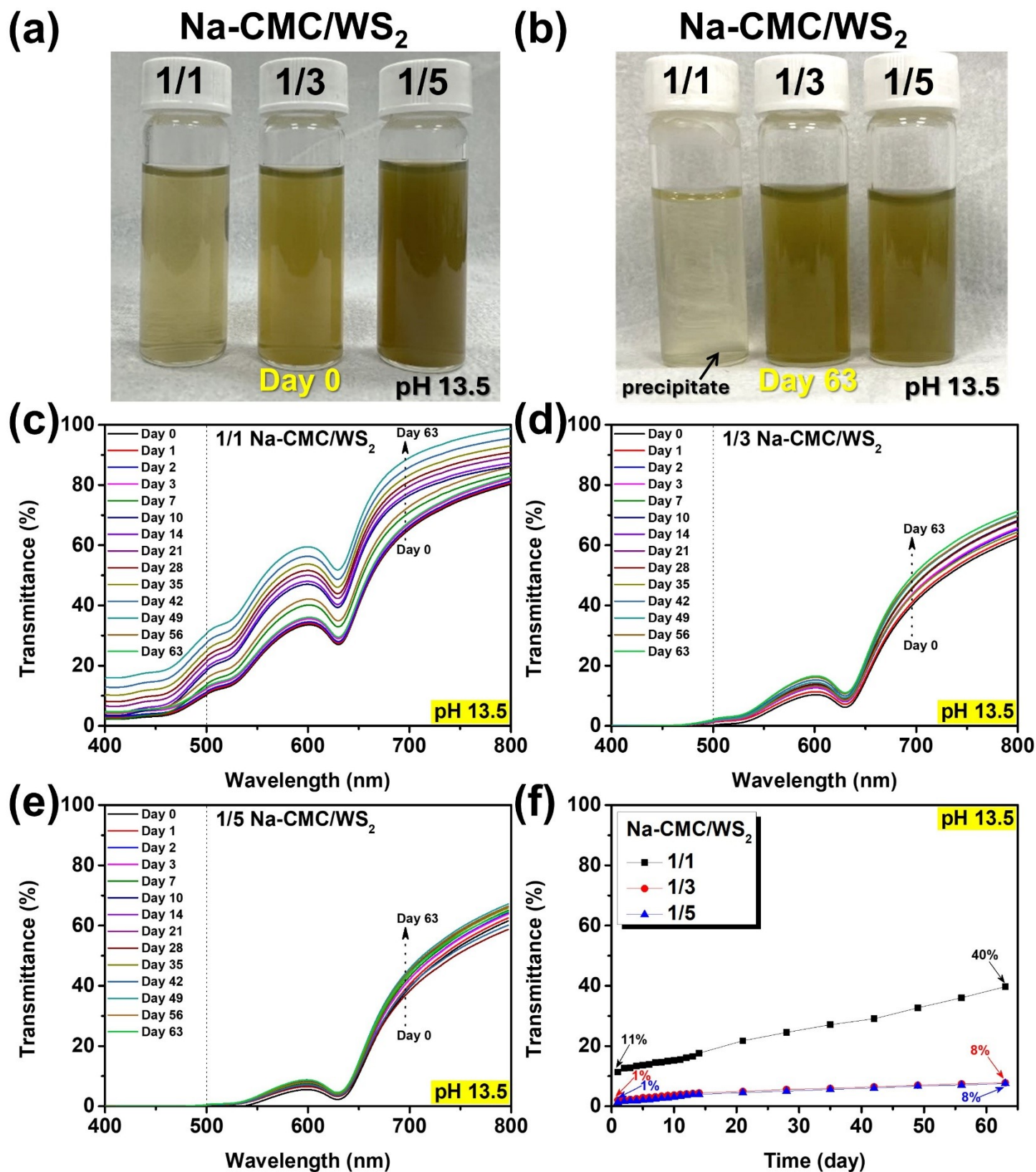


Fig. S1: Photographs of 1/1, 1/3, and 1/5 Na-CMC/WS₂ nanosheets in water at pH 13.5 on (a) day 0 and (b) day 63. Time-dependent transmittance (%) at 500 nm of (c) 1/1, (d) 1/3, and (e) 1/5 Na-CMC/WS₂ nanosheets in water at pH 13.5 recorded at 25 °C. (f) Summary plot of time-dependent transmittance at 500 nm for 1/1, 1/3, and 1/5 Na-CMC/WS₂ nanosheets in water at pH 13.5 over a 63-day monitoring period at 25 °C.

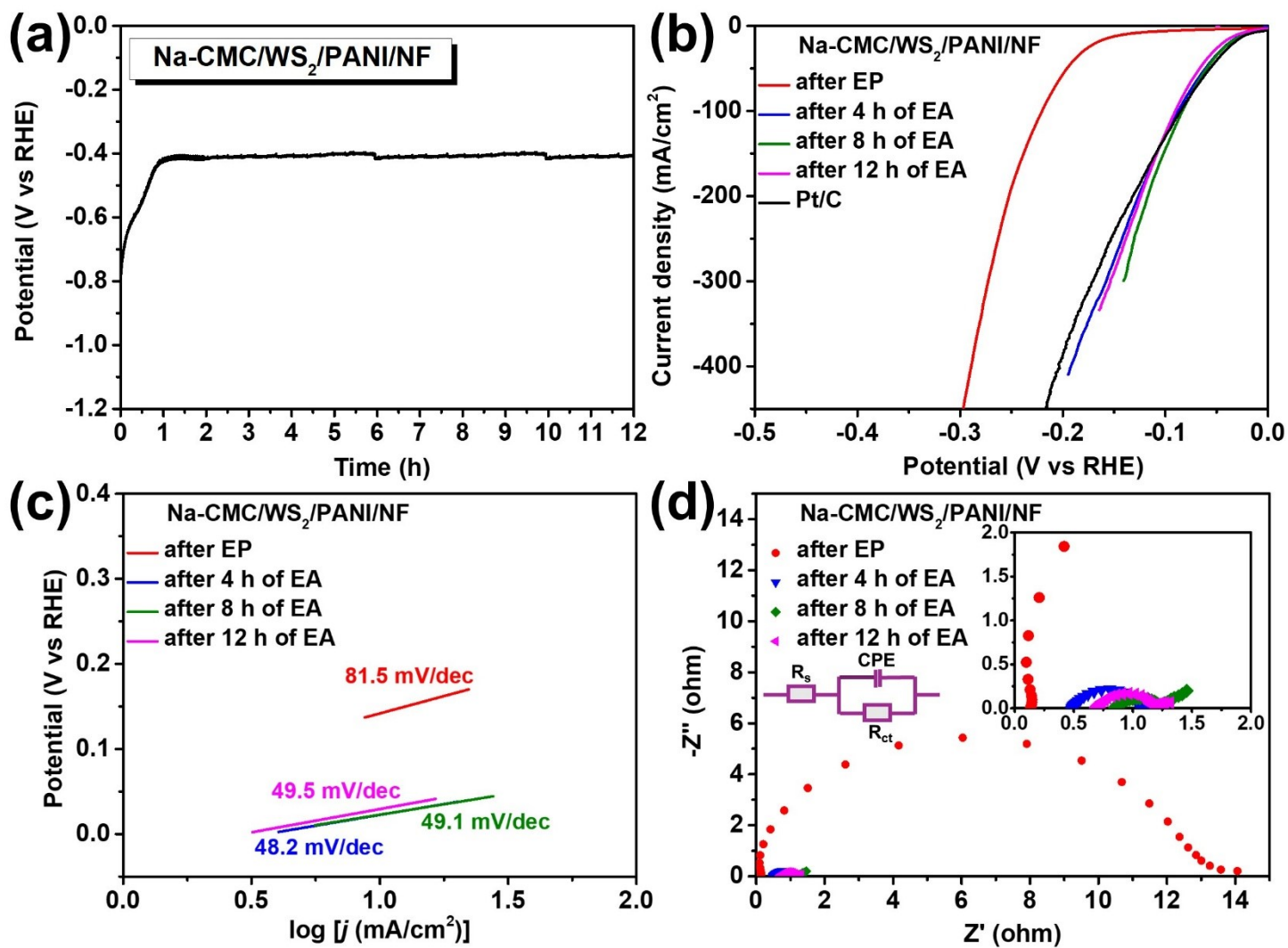


Fig. S2: (a) Chronopotentiometric curve for the EP-treated Na-CMC/WS₂/PANI/NF electrode at a constant current density of 500 mA cm⁻² in 1.0 M KOH electrolyte for 12 h. (b) LSV curves, (c) Tafel plots, and (d) EIS Nyquist plots of Pt/C, EP-treated Na-CMC/WS₂/PANI/NF, and Na-CMC/WS₂/PANI/NF electrodes subjected to EA treatment for various durations in 1.0 M KOH electrolyte. (d) The inset on the left shows the equivalent circuit model used to fit the experimental data.

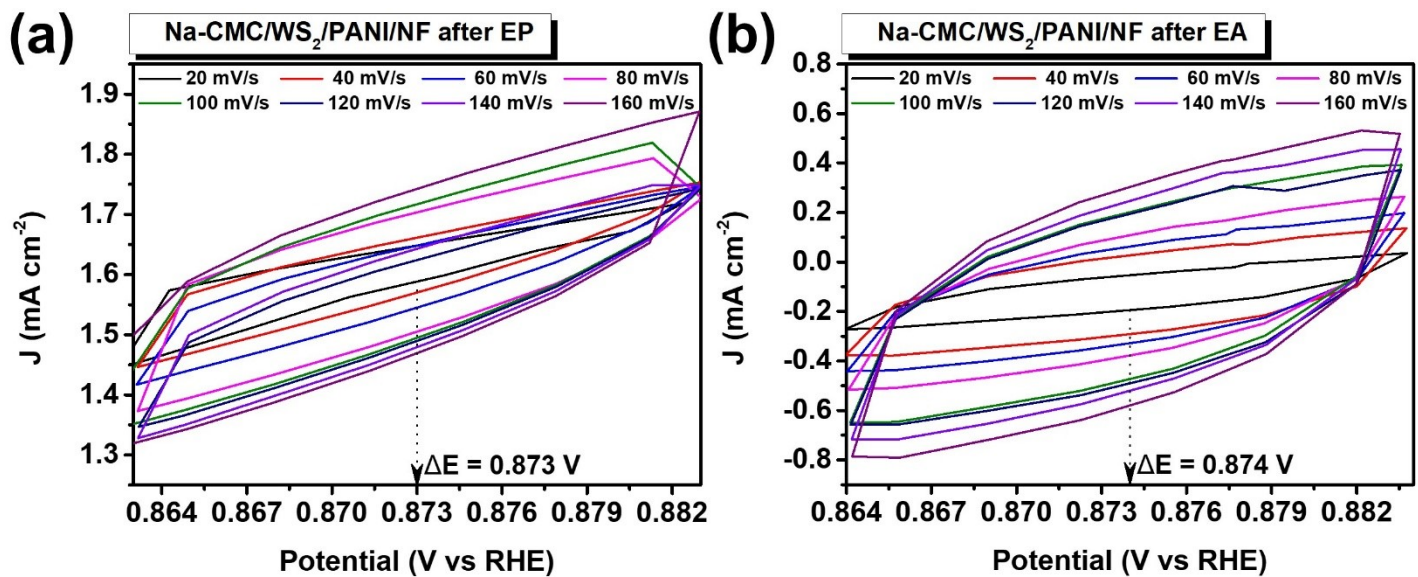
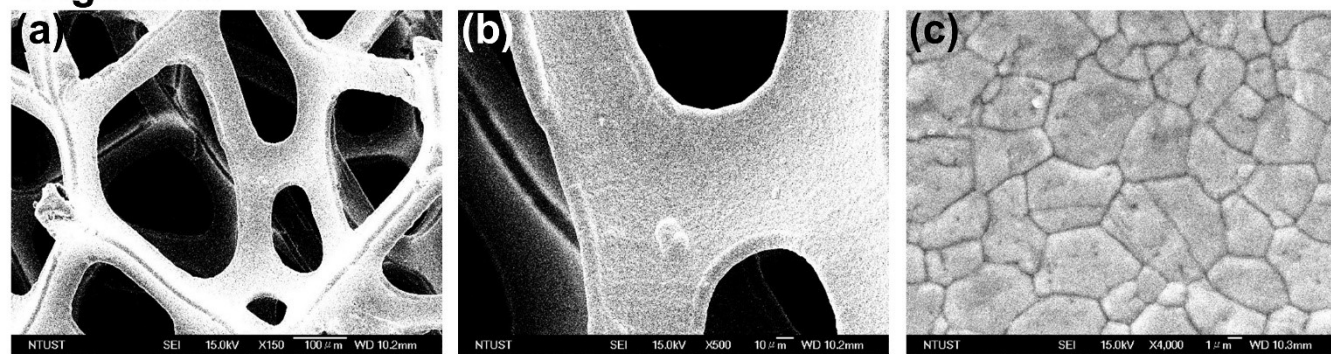


Fig. S3: Measurement of C_{dl} CV curves of the (a) EP-treated and (b) EA-treated Na-CMC/WS₂/PANI/NF electrodes in 1.0 M KOH aqueous solution at different scan rates.

Original NF



Na-CMC/WS₂/PANI/NF after EP

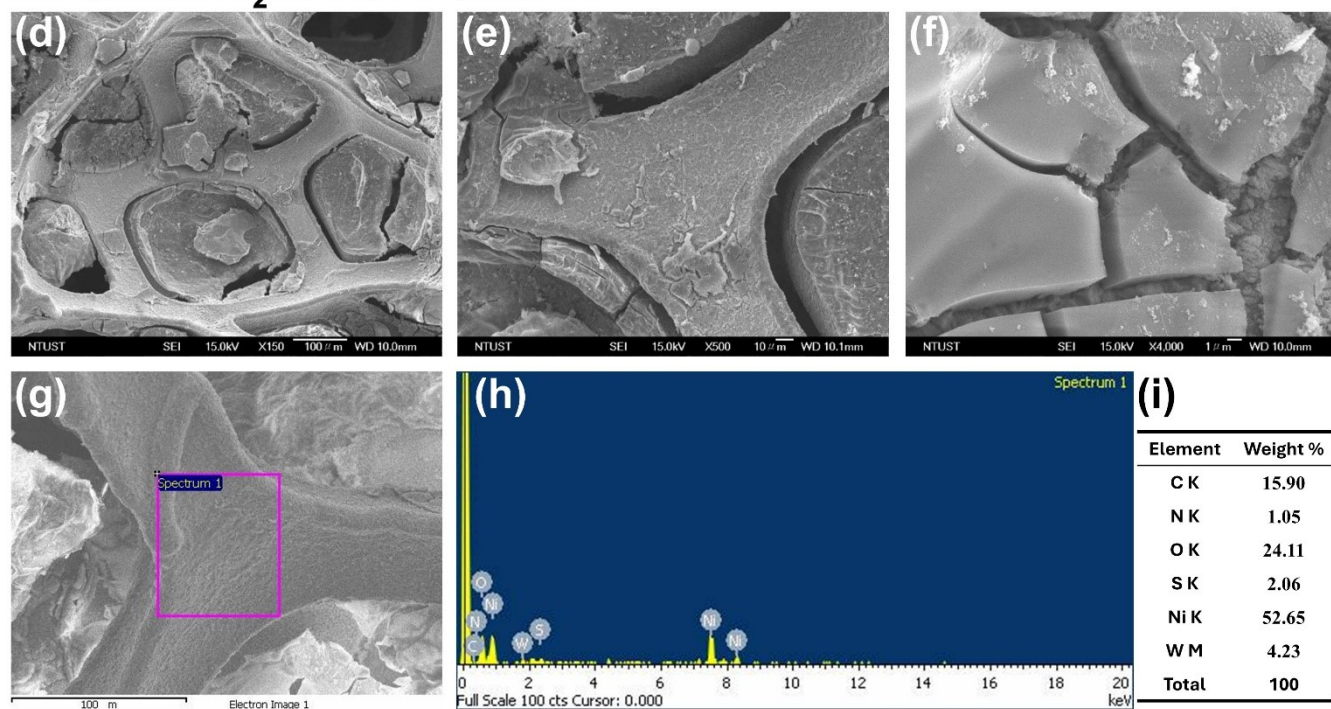


Fig. S4: SEM images of pristine NF at magnifications of (a) $\times 150$, (b) $\times 500$, and (c) $\times 4,000$. SEM images of the EP-treated Na-CMC/WS₂/PANI/NF electrode at magnifications of (d) $\times 150$, (e) $\times 500$, and (f) $\times 4,000$. (g) SEM image of the EP-treated Na-CMC/WS₂/PANI/NF electrode, along with (h) the corresponding EDX spectrum and (i) the elemental weight percentage composition.

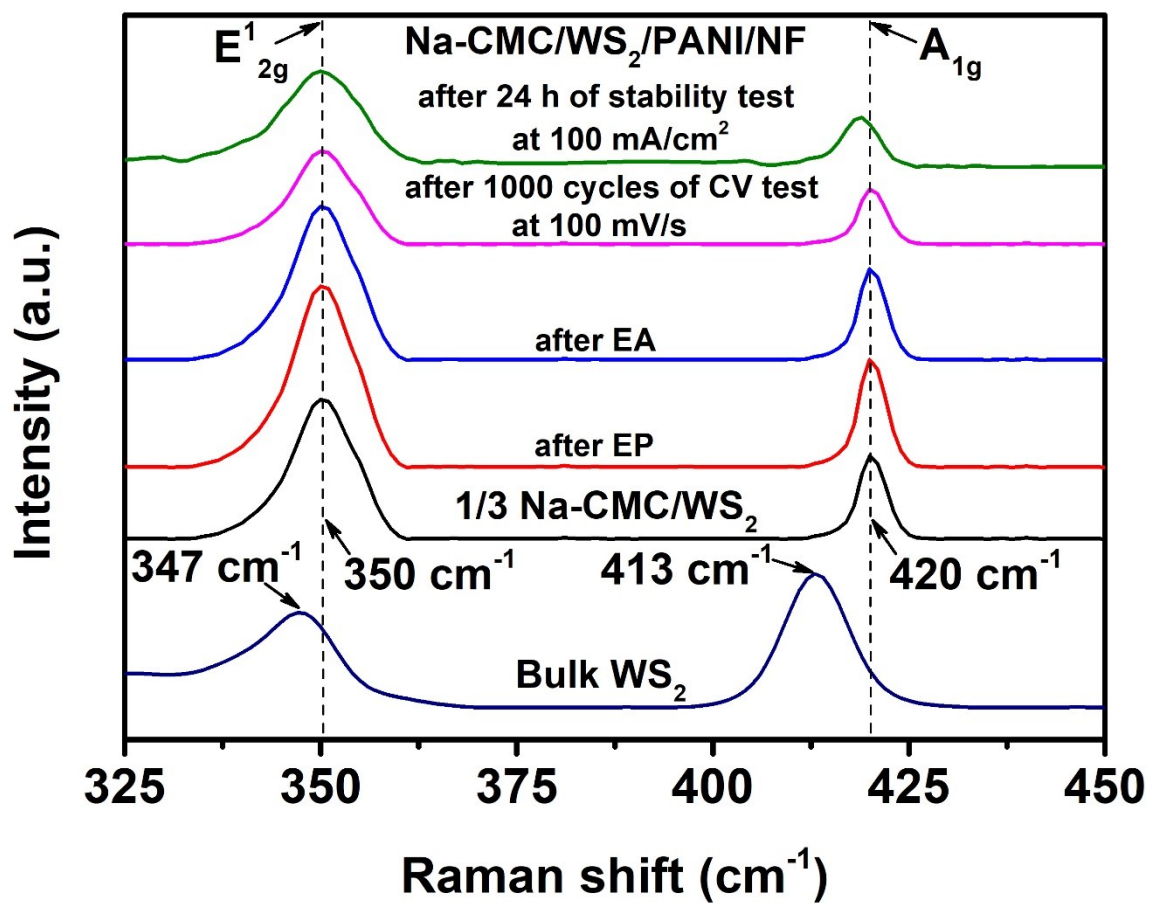


Fig. S5: Raman spectra recorded at 25 °C for bulk WS₂ crystals, the 1/3 Na-CMC/WS₂ composite, the EP-treated electrode, the EA-treated electrode, and the EA-treated Na-CMC/WS₂/PANI/NF electrode after a 24-h stability test at 100 mA cm⁻² and 1000 CV cycles in 1 M KOH solution.

Table S1: Comparison of the HER performance of representative WS₂-based electrocatalysts in 1 M KOH solution.

Catalyst	Electrolyte	η at -10 mA cm^{-2} (mV vs. RHE)	Tafel slope (mV dec ⁻¹)	Ref.
WS ₂ @UiO-66/NF	1 M KOH	121	83	[28]
Exfoliated WS ₂ /NF	1 M KOH	153	283	[28]
WS ₂ /CC	1 M KOH	235	174	[36]
1T-WS ₂ P-5@CFP	1 M KOH	190	92	[37]
WS ₂ /CoS ₂ /CC	1 M KOH	146	64	[38]
WS ₂ /CoS ₂ /Cu	1 M KOH	252	70	[39]
5% WS ₂ -MXene/NF	1 M KOH	66	47	[40]
CeO ₂ /WS ₂ /CC	1 M KOH	276	200	[41]
WS ₂ /CC	1 M KOH	340	174	[41]
EP-treated Na-CMC/WS ₂ /PANI/NF	1 M KOH	142	82	This work
EA-treated Na-CMC/WS ₂ /PANI/NF	1 M KOH	25	48	This work

CC: carbon close; CFP: carbon fiber paper; GC: glassy carbon; Cu: copper; NF: nickel foam; GF: graphite foam