

Efficient Regeneration of Pd-SSZ-13 catalysts deactivated by alkali
metals for Passive NO_x Adsorption

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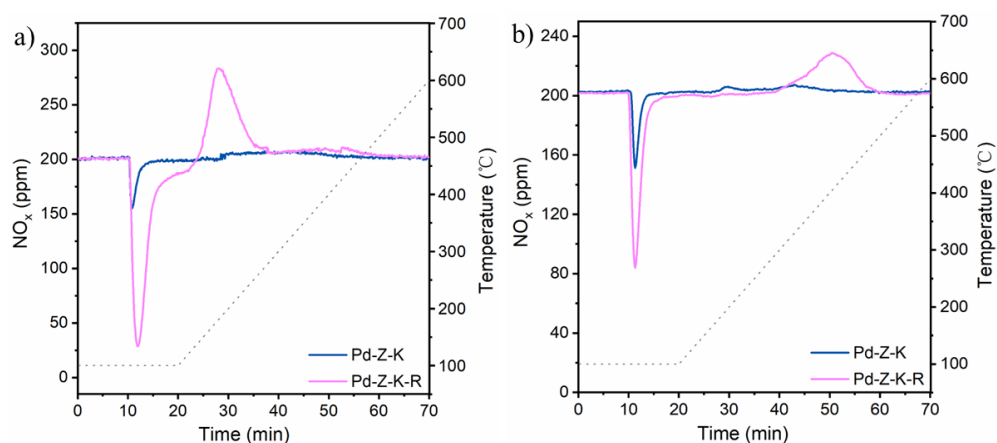


Figure S1. PNA performances of K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts under different conditions. a) Gas feed: 200 ppm of NO_x and 10 vol.% O_2 in N_2 ; b) Gas feed: 200 ppm of NO_x , 200 ppm of CO , 3 vol.% H_2O and 10 vol.% O_2 in N_2 . Samples were treated at 500 °C in 10 vol% O_2 for one hour before all the tests, and a total flow rate of $200 \text{ mL} \cdot \text{min}^{-1}$ was used.

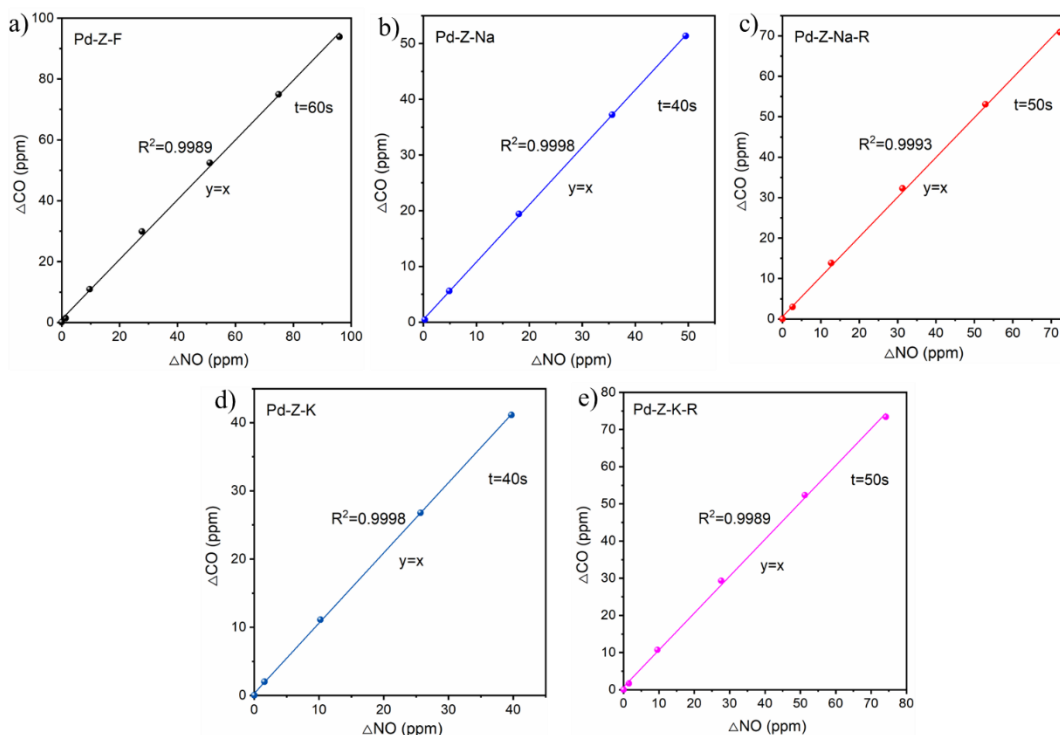


Figure S2. The correlation between the reduced concentrations of NO and CO (labeled as ΔNO and ΔCO) during the initial stage of NO_x adsorption in PNA tests over Pd-Z-F (a), alkali-metal poisoned Pd-Z-Na and Pd-Z-K (b and d) and the regenerated Pd-Z-Na and Pd-Z-K catalysts (c and e). ΔNO (ΔCO) = $C_0 - C_t$ (C_0 and C_t represent initial and real-time concentrations of NO or CO, respectively). Gas feed: gas feed: 200 ppm of NO_x , 200 ppm of CO, 3 vol.% H_2O and 10 vol.% O_2 in N_2 , with a total flow rate of 200 mL min^{-1} . All the samples were pretreated at 500°C in 10 vol.% O_2 before the tests.

Note: he concentrations of CO and NO exhibited a parallel decline during the initial stages of the PNA test, attributed to the formation of the $\text{Pd}^{2+}(\text{NO})(\text{CO})$ intermediate. Notably, a significantly greater quantity of CO and NO is consumed during the PNA tests over Pd-SSZ-13, suggesting the presence of a higher density of isolated Pd^{2+} sites within the material. The introduction of alkali metals leads to a diminution in the number of active Pd^{2+} centers, thereby reducing the formation of the $\text{Pd}^{2+}(\text{NO})(\text{CO})$ intermediate. Conversely, regeneration with NH_4Cl solution alleviates the deactivating impact of alkali metals on the Pd sites, resulting in a notably higher production of the $\text{Pd}^{2+}(\text{NO})(\text{CO})$ intermediate after washing, in comparison to the poisoned Pd-SSZ-13 catalysts.

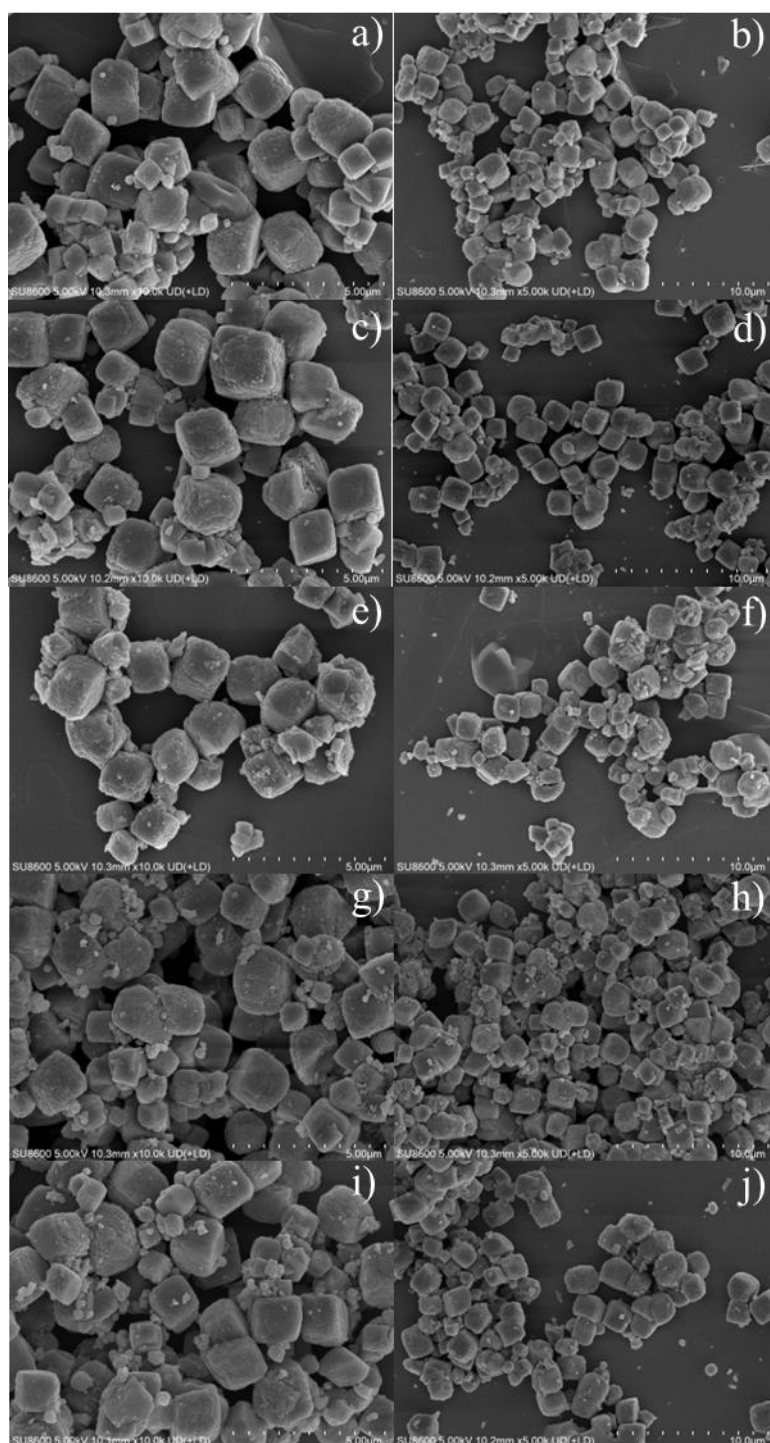


Figure S3. Representative FE-SEM images of Pd-Z-F (a and b), Pd-Z-Na (c and d), Pd-Z-Na-R (e and f), Pd-Z-K (g and h) and Pd-Z-K-R (i and j) catalysts.

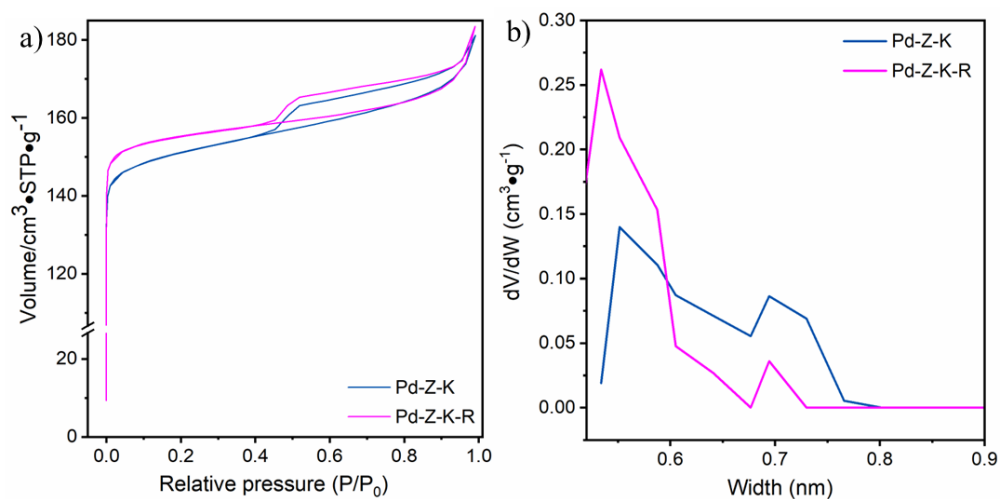


Figure S4. N₂-physisorption curves (a) and micropore size distributions (b) analyzed by the NLDT method for K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts.

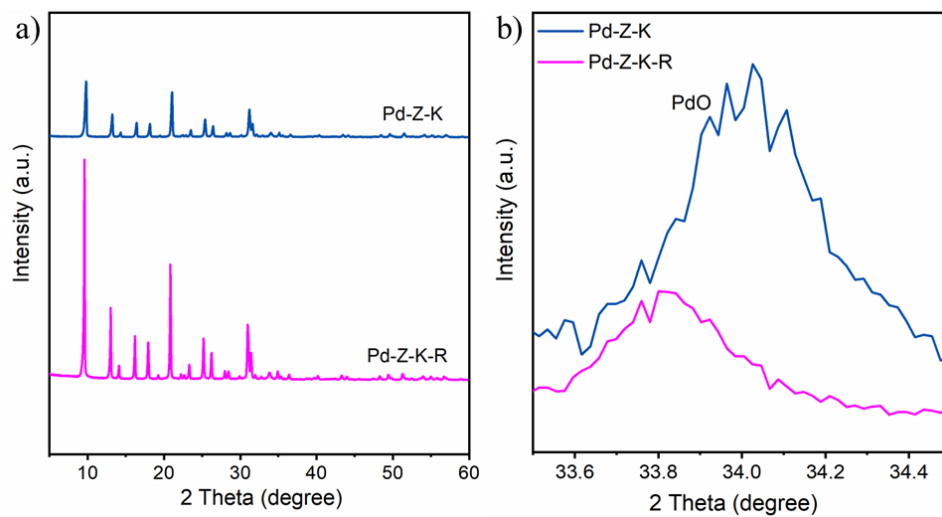


Figure S5. (a) XRD patterns of K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts. (b) Enlarged XRD patterns (normalized to the intensity of main band at $2\theta = 9.6^\circ$ in the respective spectrum) in $33.5\text{--}34.5^\circ$.

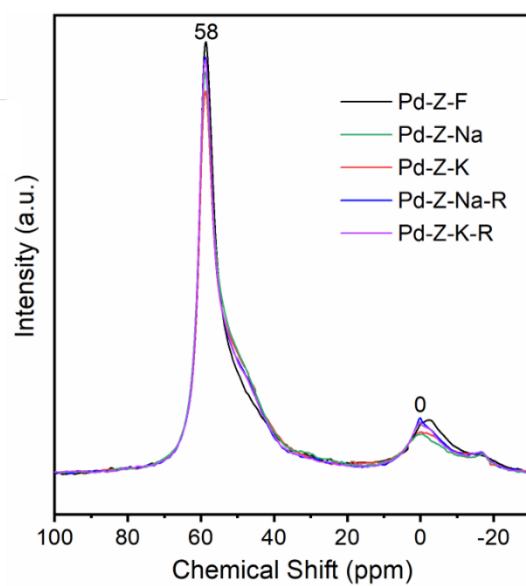


Figure S6. ^{27}Al MAS NMR spectra of fresh, poisoned and regenerated Pd-SSZ-13 catalysts. Pd-Z-F (black), Pd-Z-Na (green), Pd-Z-K (red), Pd-Z-Na-R (blue) and Pd-Z-F (violet).

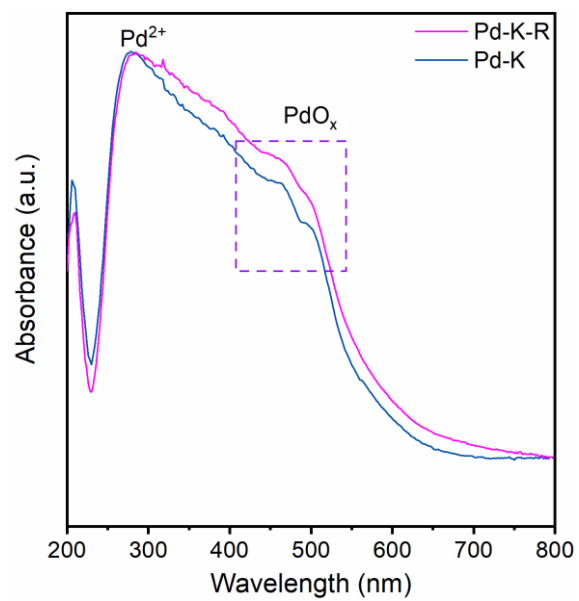


Figure S7. DR UV/Vis spectra of K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts.

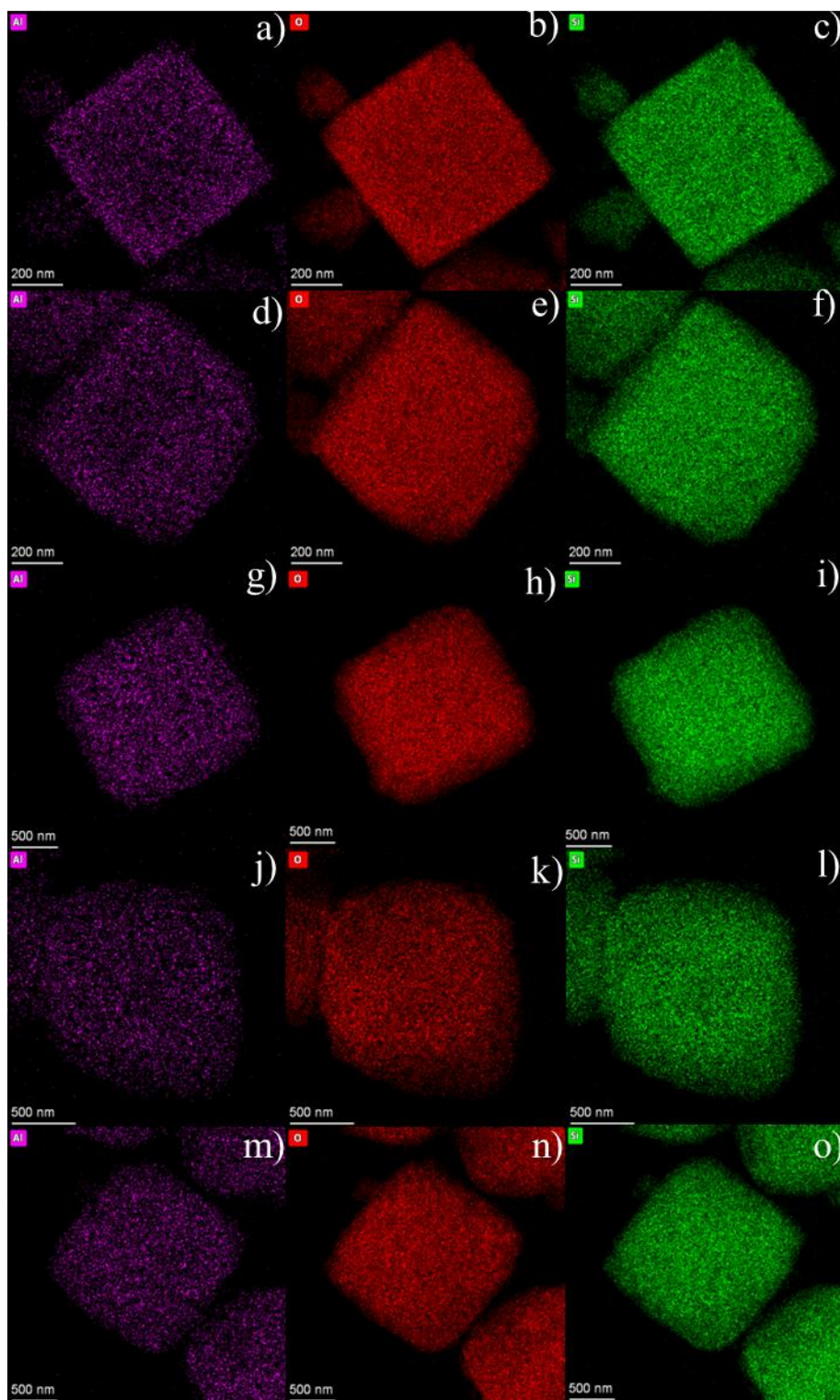


Figure S8. EDX mappings of for Pd-Z-F (panels (a)–(c)), the poisoned Pd-Z-Na (panels (d)–(f)) and Pd-Z-K (panels (g)–(i)), the regenerated Pd-Z-Na-R (panels (j)–(l)) and Pd-Z-K-R (panels (m)–(o)) catalysts.

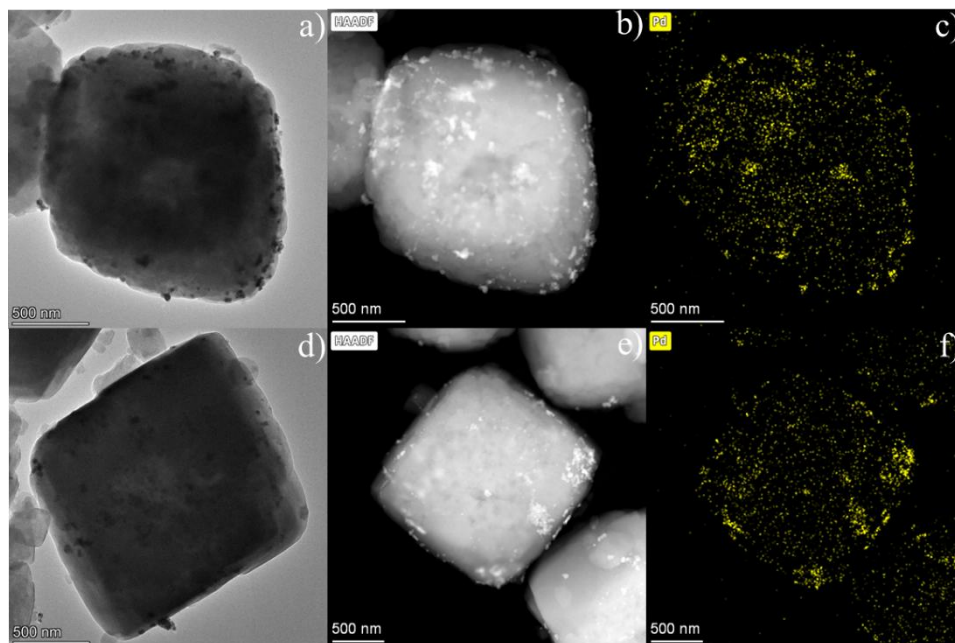


Figure S9. (a, d) TEM, (b, e) HAADF-STEM images, and (c, f) EDX mapping of Pd for Pd-Z-K (panels (a)–(c)) and Pd-Z-K-R (panels (d)–(f)) catalysts.

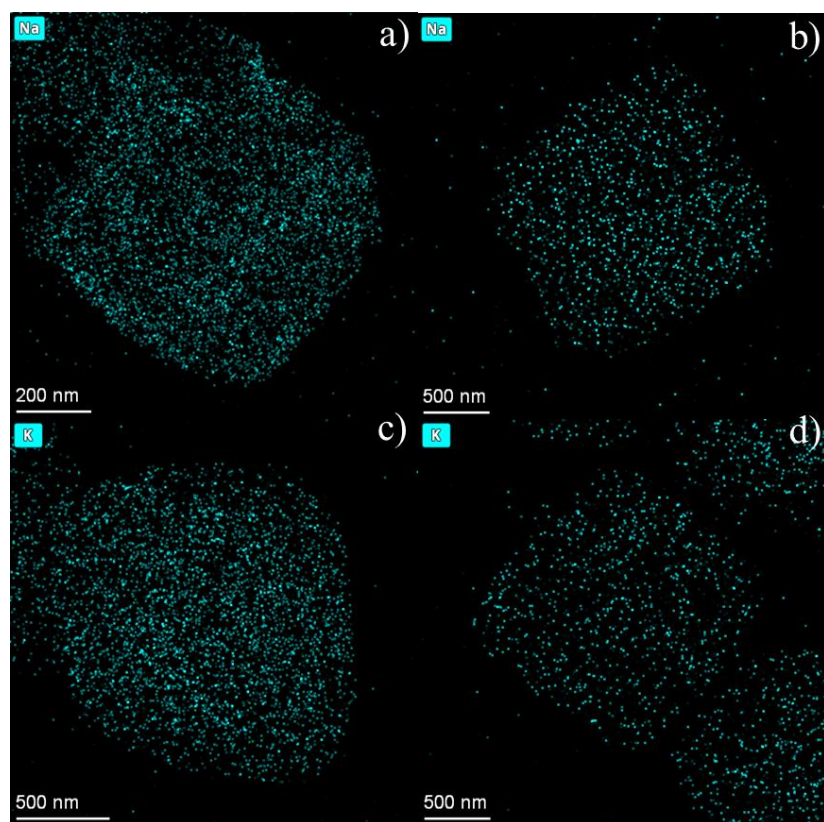


Figure S10. EDX mappings of alkali metals for Pd-Z-Na (a), the regenerated Pd-Z-Na-R (b), Pd-Z-K (C) and the regenerated Pd-Z-K-R (d) catalysts.

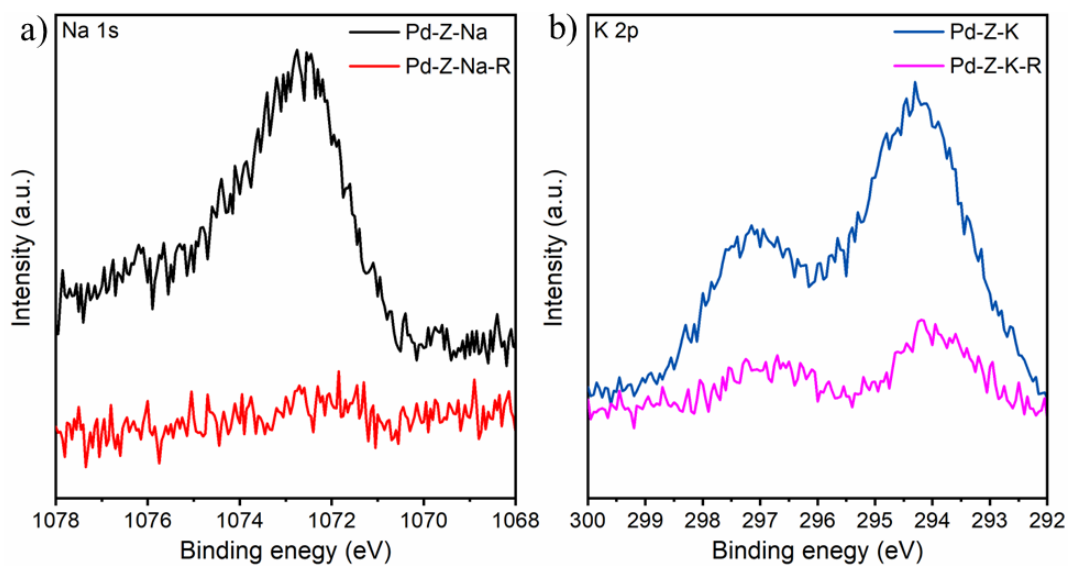


Figure S11. (a) High-resolution XPS Na 1s spectra of the poisoned Pd-Z-Na and the regenerated Pd-Z-Na-R catalysts; (b) High-resolution XPS K 2p spectra of the poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts.

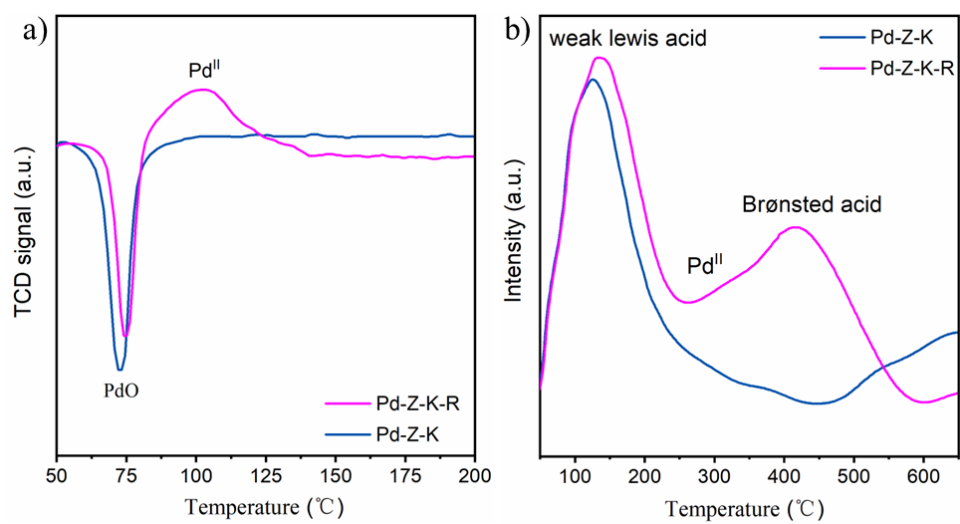


Figure S12. H_2 -TPR curves (a) and NH_3 -TPD profiles (b) of K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts.

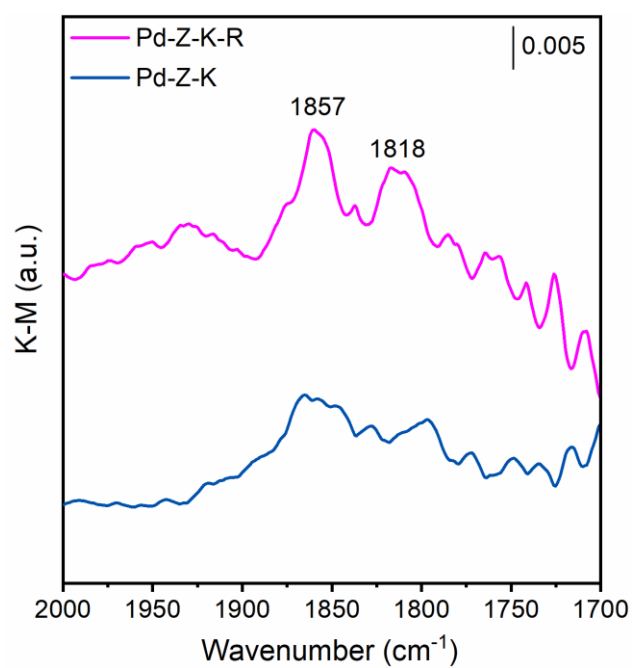


Figure S13. NO-DRIFTS profiles of K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts.

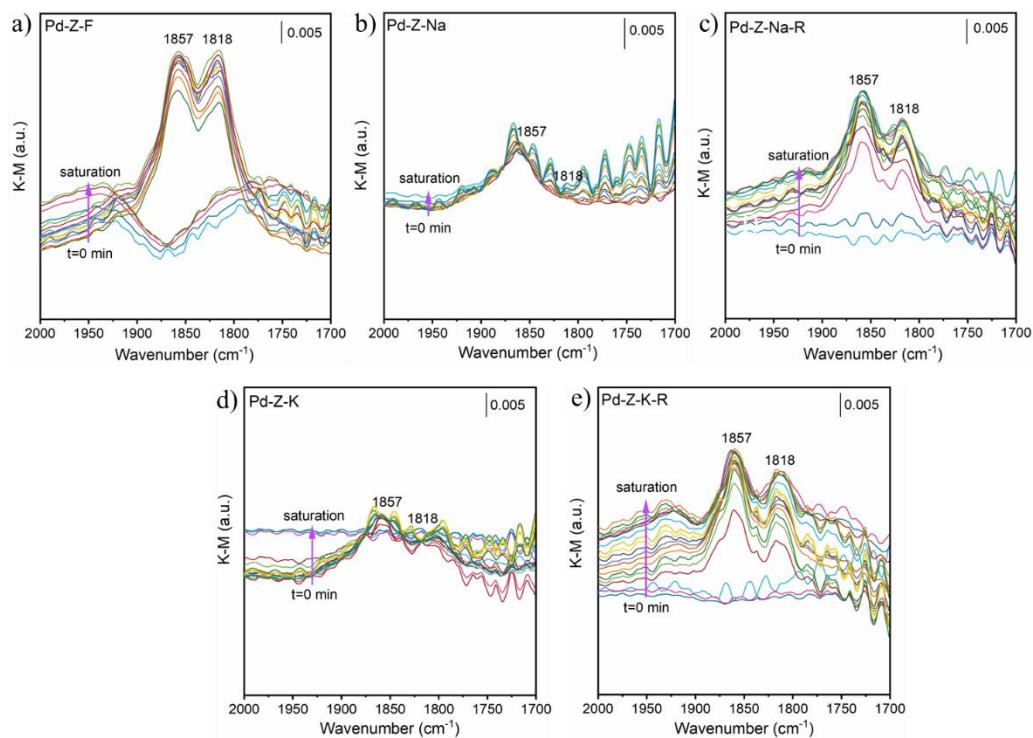


Figure S14. Time-resolved *in situ* NO-DRIFT spectra for the Pd-Z-F(a), alkali-metal poisoned Pd-Z-Na and Pd-Z-K (b and d) and the regenerated Pd-Z-Na-R and Pd-Z-K-R (c and e) catalysts. The measurements were performed at 100 °C after pretreatment in flowing dry air at 500 °C for about 1 h. An initial background spectrum was acquired in a continuous flow of N₂ and subsequently subtracted from all spectra recorded in a NO-enriched atmosphere, all conducted at a consistent temperature of 100 °C.

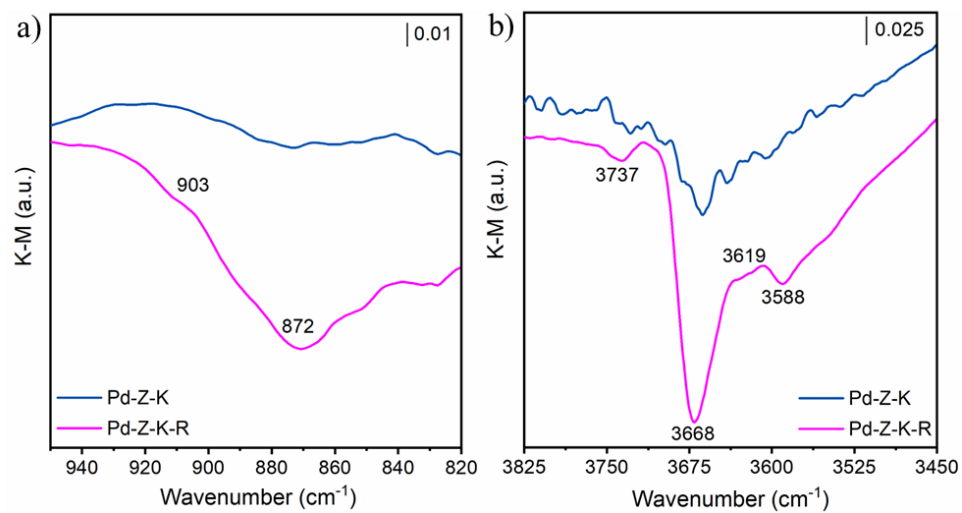


Figure S15. NH_3 -DRIFTS profiles of K-poisoned Pd-Z-K and the regenerated Pd-Z-K-R catalysts.

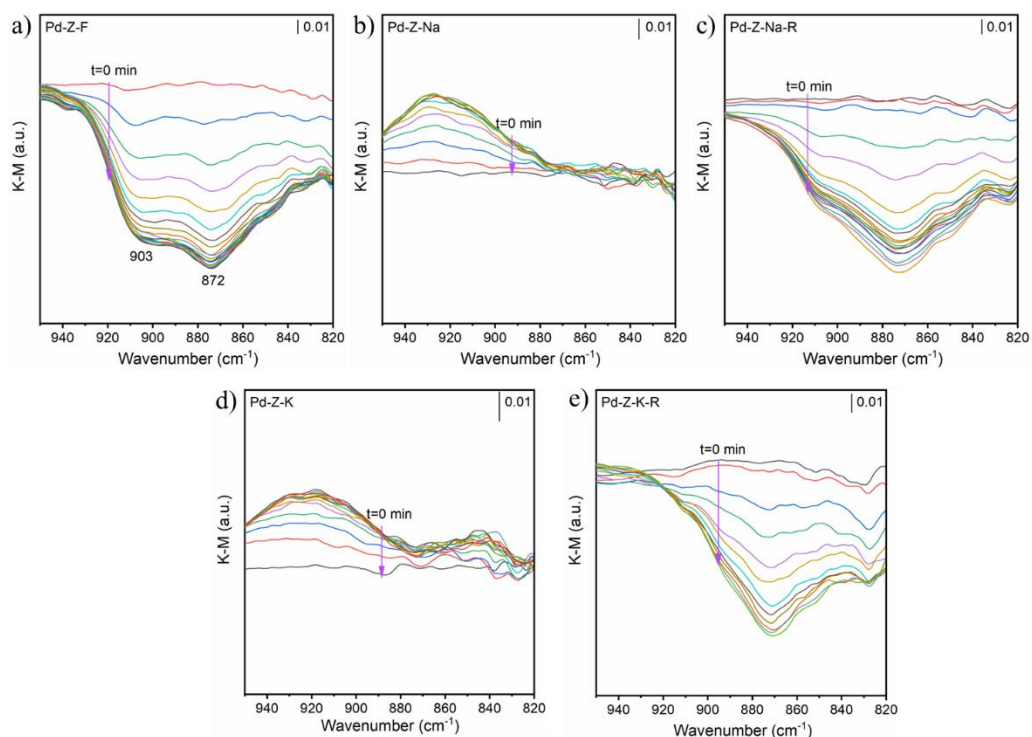


Figure S16. Time-resolved *in situ* NH₃-DRIFT spectra (950-820 cm⁻¹) for the Pd-Z-F(a), alkali-metal poisoned Pd-Z-Na and Pd-Z-K (b and d) and the regenerated Pd-Z-Na-R and Pd-Z-K-R (c and e) catalysts. The measurements were performed at 25 °C after pretreatment in flowing dry air at 500 °C for about 1 h. An initial background spectrum was acquired in a continuous flow of N₂ and subsequently subtracted from all spectra recorded in a NH₃-enriched atmosphere, all conducted at a consistent temperature of 25 °C.

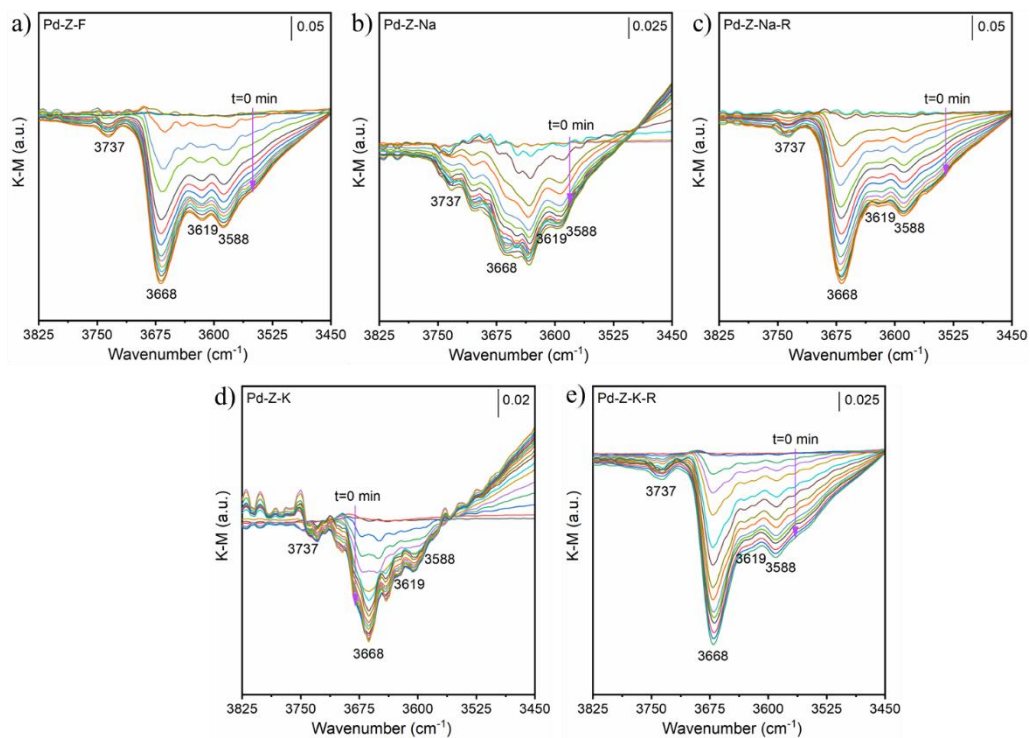


Figure S17. Time-resolved *in situ* NH₃-DRIFT spectra (3825-3450 cm⁻¹) for the Pd-Z-F(a), alkali-metal poisoned Pd-Z-Na and Pd-Z-K (b and d) and the regenerated Pd-Z-Na-R and Pd-Z-K-R (c and e) catalysts. The measurements were performed at 25 °C after pretreatment in flowing dry air at 500 °C for about 1 h. An initial background spectrum was acquired in a continuous flow of N₂ and subsequently subtracted from all spectra recorded in a NH₃-enriched atmosphere, all conducted at a consistent temperature of 25 °C.

Table S1. Textural properties of fresh, alkali-metal-poisoned and Pd-SSZ-13 catalysts.

Sample	S_{BET} ($\text{m}^2 \text{g}^{-1}$)	V_{micro} ($\text{cm}^3 \text{g}^{-1}$)
Pd-Z-F	536.05	0.233
Pd-Z-Na	497.89	0.205
Pd-Z-K	487.14	0.209
Pd-Z-Na-R	505.74	0.216
Pd-Z-K-R	498.28	0.224

Table S2. Weight percentage of alkali metals before and after NH₄Cl washing determined by ICP-OES.

Sample	Weight (g)	analyte	mass fraction
Pd-Z-Na	0.0405	Na	1.5488%
Pd-Z-Na-R	0.0393	Na	0.1066%
Pd-Z-K	0.0435	K	1.9911%
Pd-Z-K-R	0.0402	K	0.4964%

Table S3. Acid Content ($\mu\text{mol/g}$) of fresh, alkali-metal-poisoned and Pd-SSZ-13 catalysts

Calculated based on Gaussian deconvolution and area integral of the NH_3 -TPD curves.

Sample	Weak acid	Moderate acid	Strong acid	total acid
Pd-Z-F	230.2	149.8	134.9	514.9
Pd-Z-Na	194.8	36.5	0	231.3
Pd-Z-K	158.9	28.0	0	186.9
Pd-Z-Na-R	179.3	67.5	126	372.8
Pd-Z-K-R	156.8	62.1	113.2	332.1