

**Chemical Reactivity of 2-Methylchromone with some Nitrogen-Based Nucleophiles:
Reactions, DFT Analysis, Anticancer Activity and Molecular Docking Studies**

**Mohamed Abdel-Megid,¹ Najla A. Alshaye,² Al-Shimaa Badran^{*3} and Magdy A.
Ibrahim³**

¹ *Department of Chemistry, College of Science, Imam Mohammad Ibn Saud Islamic University (IMSIU), Riyadh, Saudi Arabia. moabmohamed@imamu.edu.sa*

² *Department of Chemistry, College of Science, Princess Nourah bint Abdulrahman University, P.O. Box 84428, Riyadh 11671, Saudi Arabia. naalshaye@pnu.edu.sa*

³ *Department of Chemistry, Faculty of Education, Ain Shams University, Roxy, 11711, Cairo, Egypt*

** E-mail: badran.shimaa@yahoo.com,*

Tel.: +2 01011444940 fax: +2 022581243

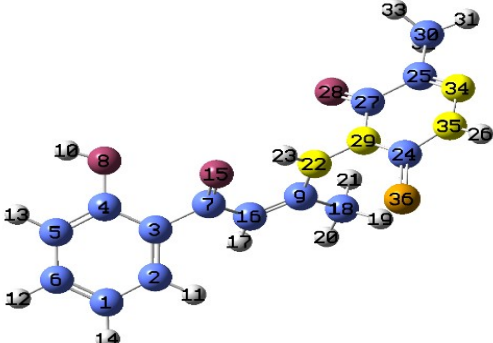
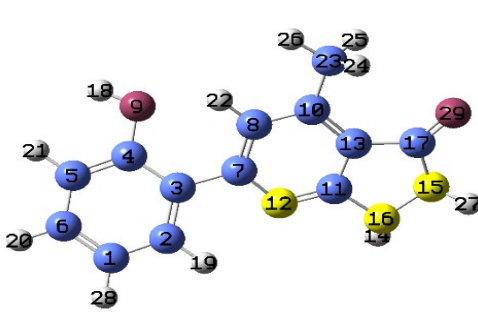
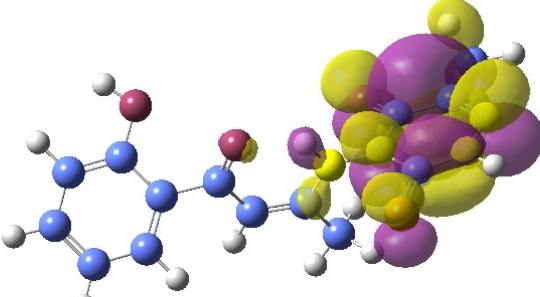
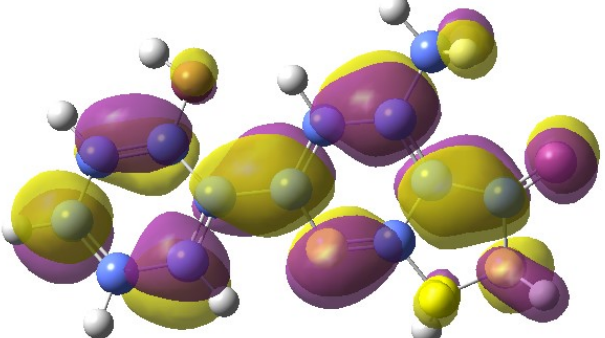
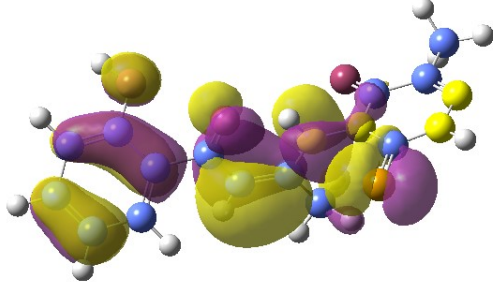
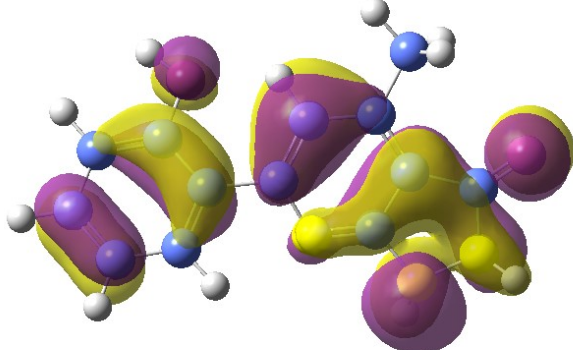
Compound	9	10
Optimized Structures	 Optimized structure of compound 9, showing a complex organic molecule with numbered atoms (1-36) and various functional groups.	 Optimized structure of compound 10, showing a complex organic molecule with numbered atoms (1-29) and various functional groups.
LUMO	 LUMO orbital of compound 9, showing the electron density distribution. The energy is $E_{\text{LUMO}} = -2.407$.	 LUMO orbital of compound 10, showing the electron density distribution. The energy is $E_{\text{LUMO}} = -1.808$.
HOMO	 HOMO orbital of compound 9, showing the electron density distribution. The energy is $E_{\text{HOMO}} = -6.072$.	 HOMO orbital of compound 10, showing the electron density distribution. The energy is $E_{\text{HOMO}} = -6.125$.

Fig. S1. Molecular modeling and the electron density of HOMO and LUMO of compounds 9 and 10

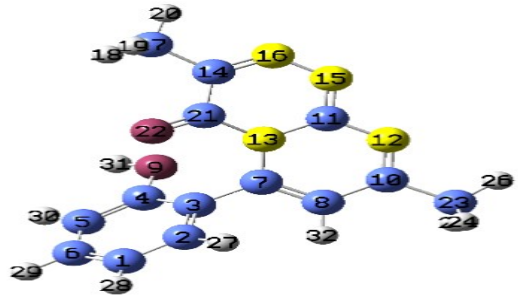
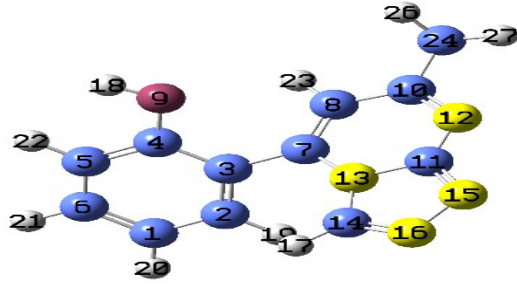
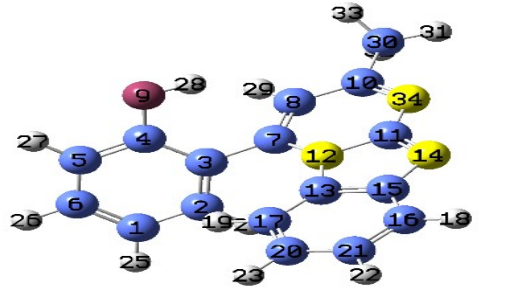
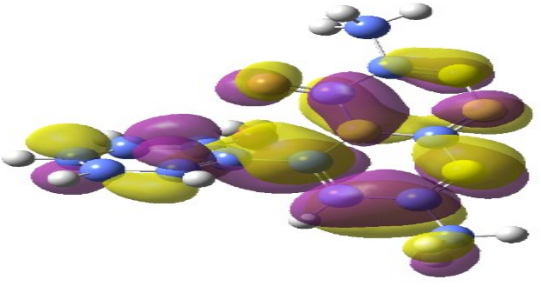
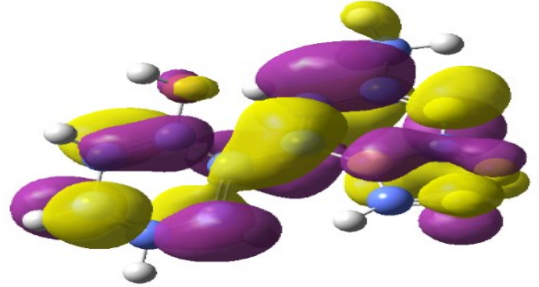
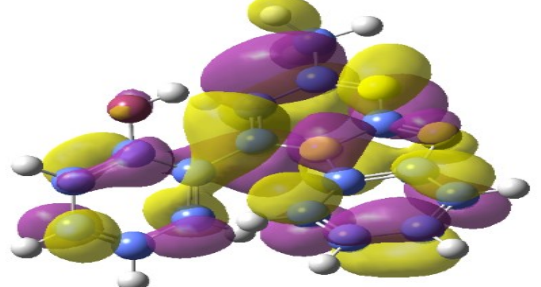
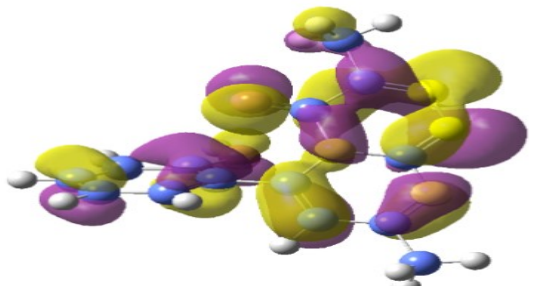
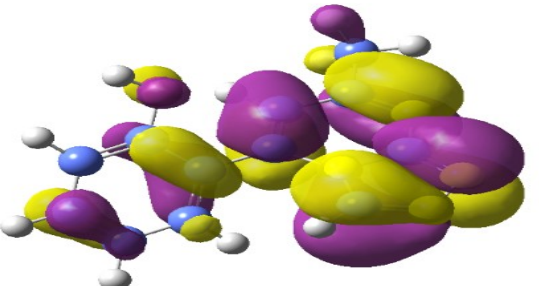
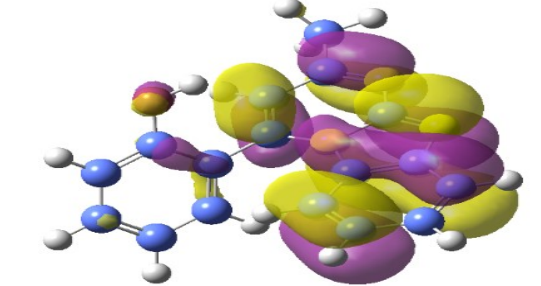
Compound 12	Compound 13	Compound 14
 Optimized Structures	 Optimized Structures	 Optimized Structures
 $E_{\text{HOMO}} = -6.236$	 $E_{\text{HOMO}} = -6.350$	 $E_{\text{HOMO}} = -5.965$
 $E_{\text{LUMO}} = -2.295$	 $E_{\text{LUMO}} = -2.046$	 $E_{\text{LUMO}} = -2.074$

Fig. S2. Molecular modeling and the electron density of HOMO and LUMO of compounds 12-14

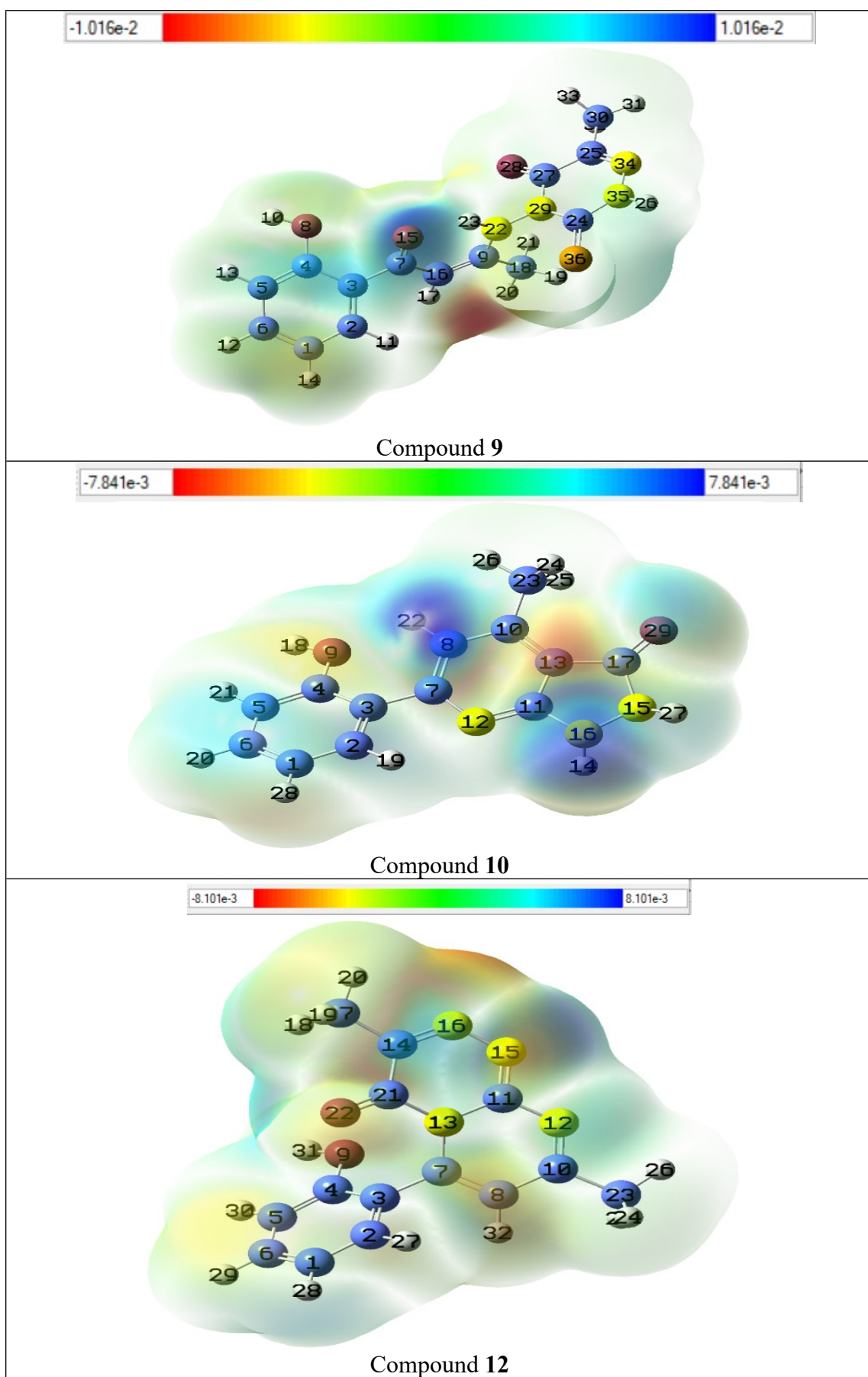


Fig. S3. Molecular electrostatic potential of compounds 9, 10 and 12

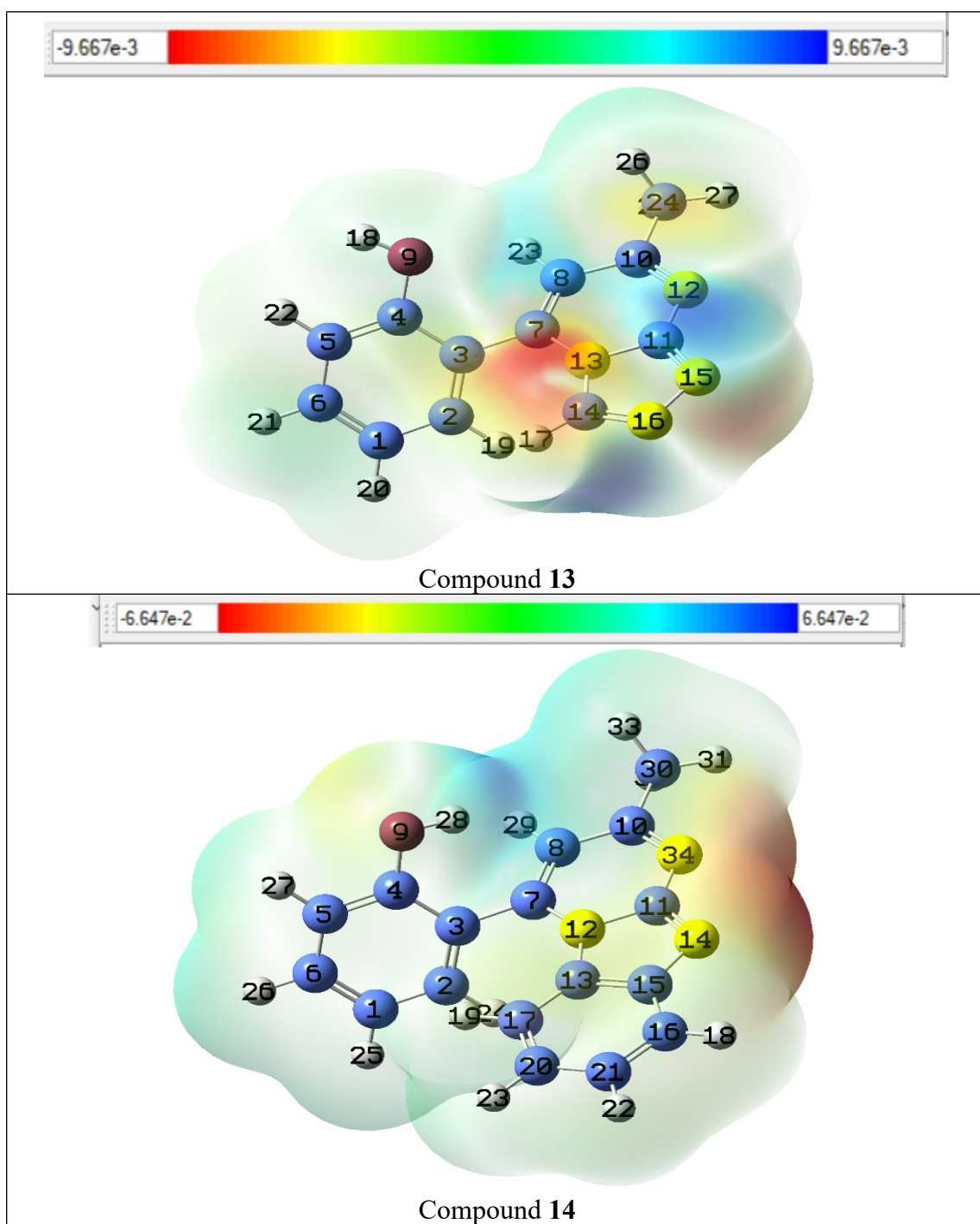


Fig. S4. Molecular electrostatic potential of compounds **13** and **14**

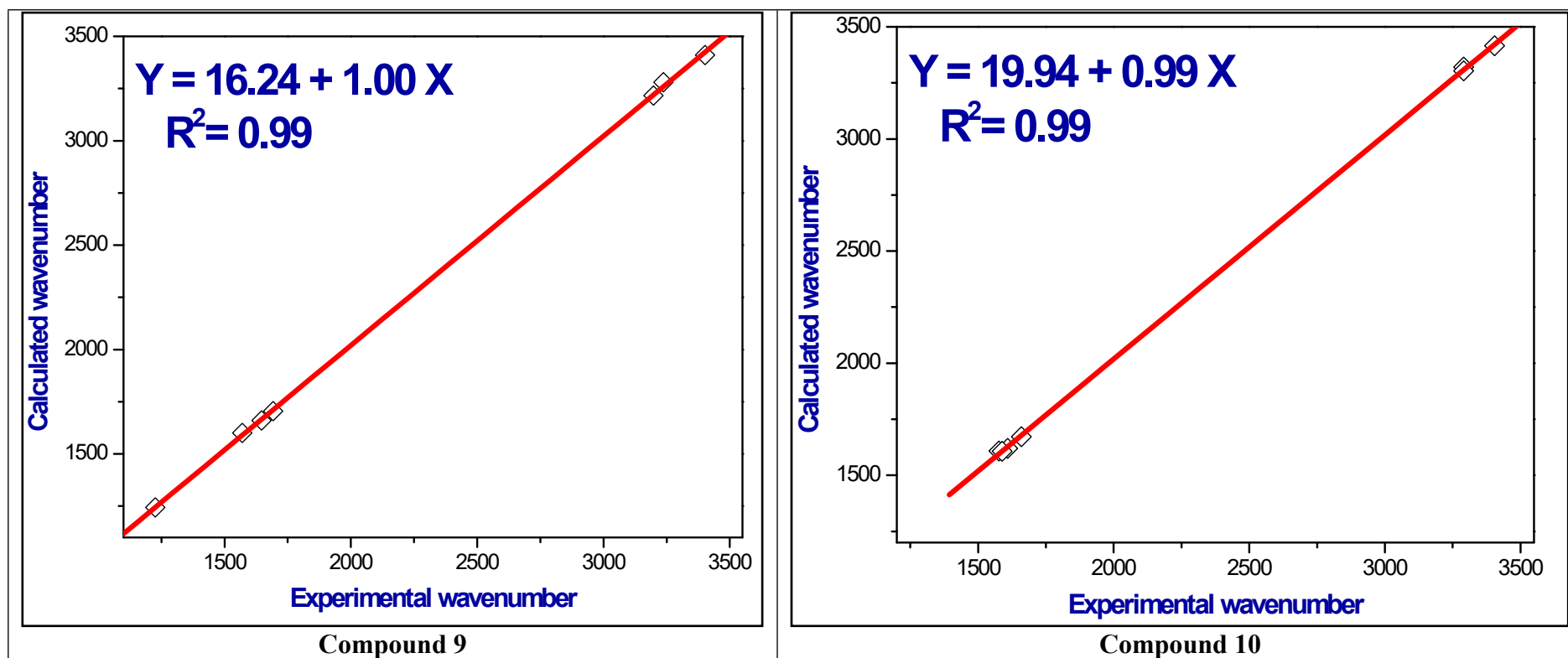


Fig. S5. The correlation relationships of the experimental *versus* calculated IR wavenumbers of compounds **9** and **10**.

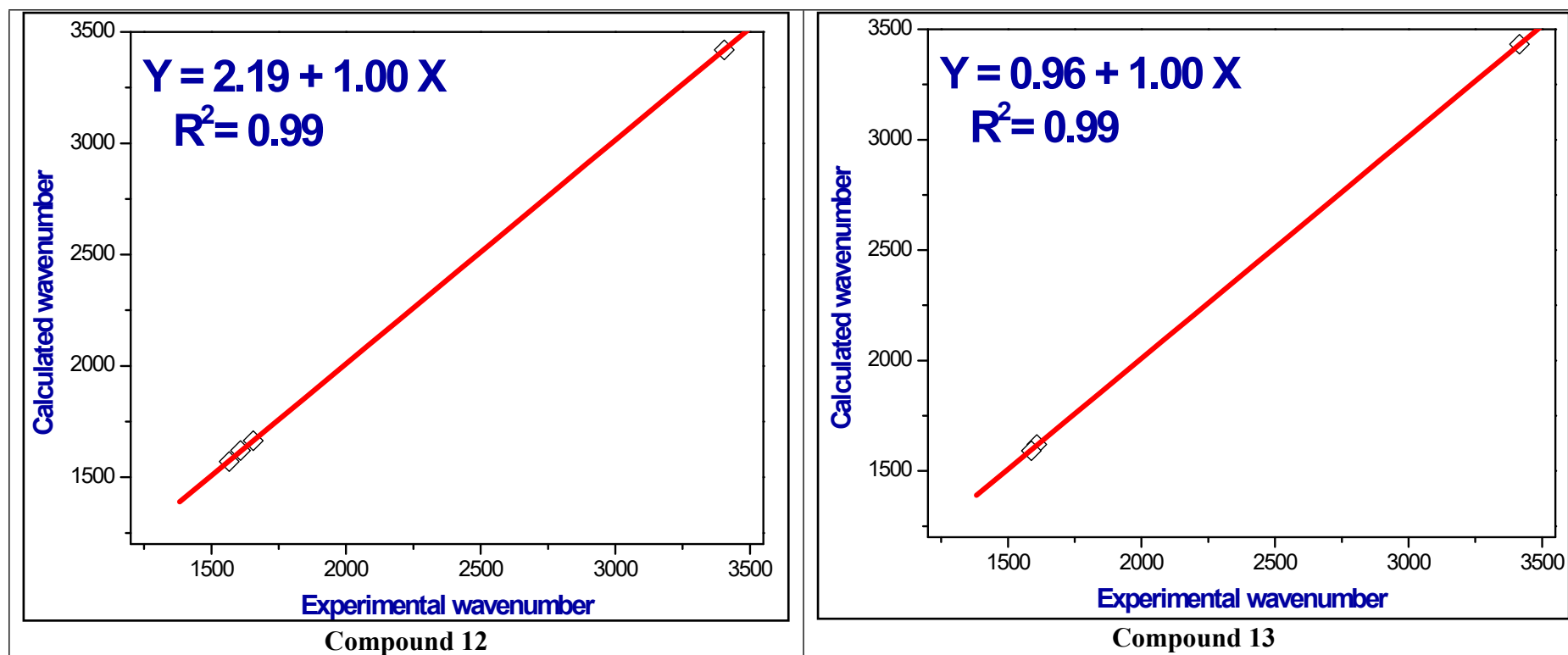


Fig. S6. The correlation relationships of the experimental *versus* calculated IR wavenumbers of compounds **12** and **13**.

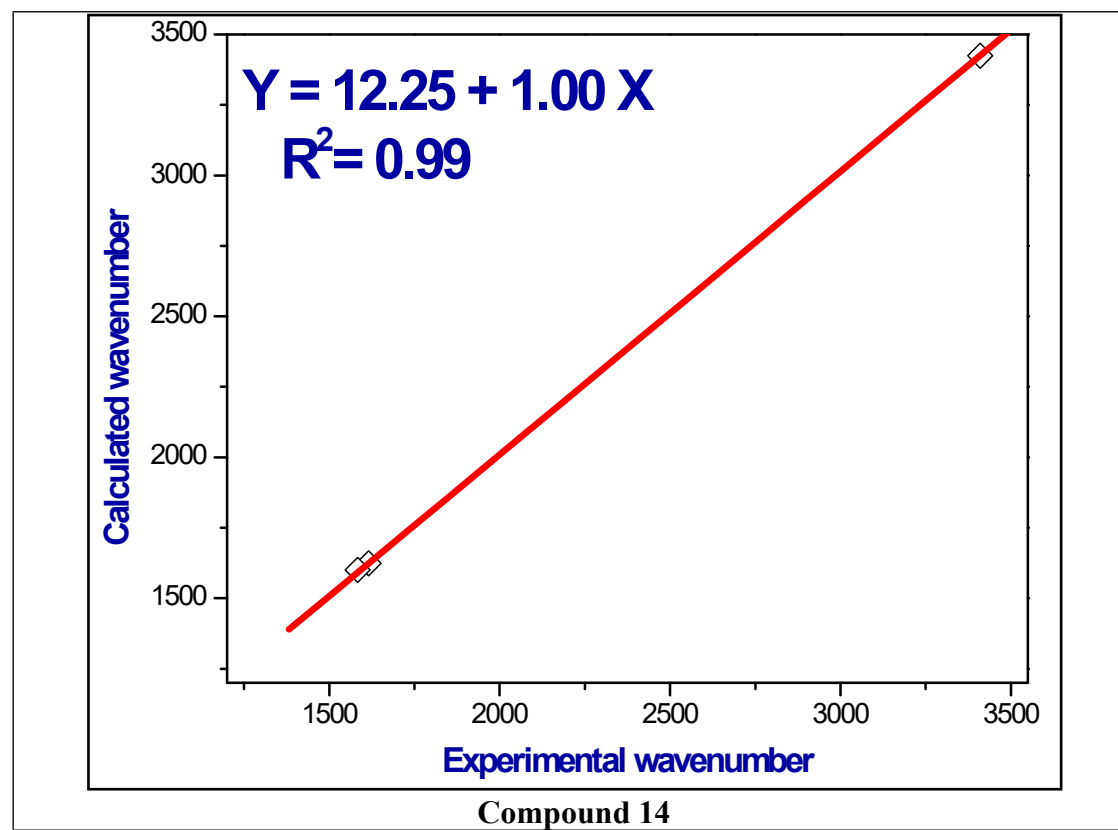


Fig. S7. The correlation relationships of the experimental *versus* calculated IR wavenumbers of compound 14.

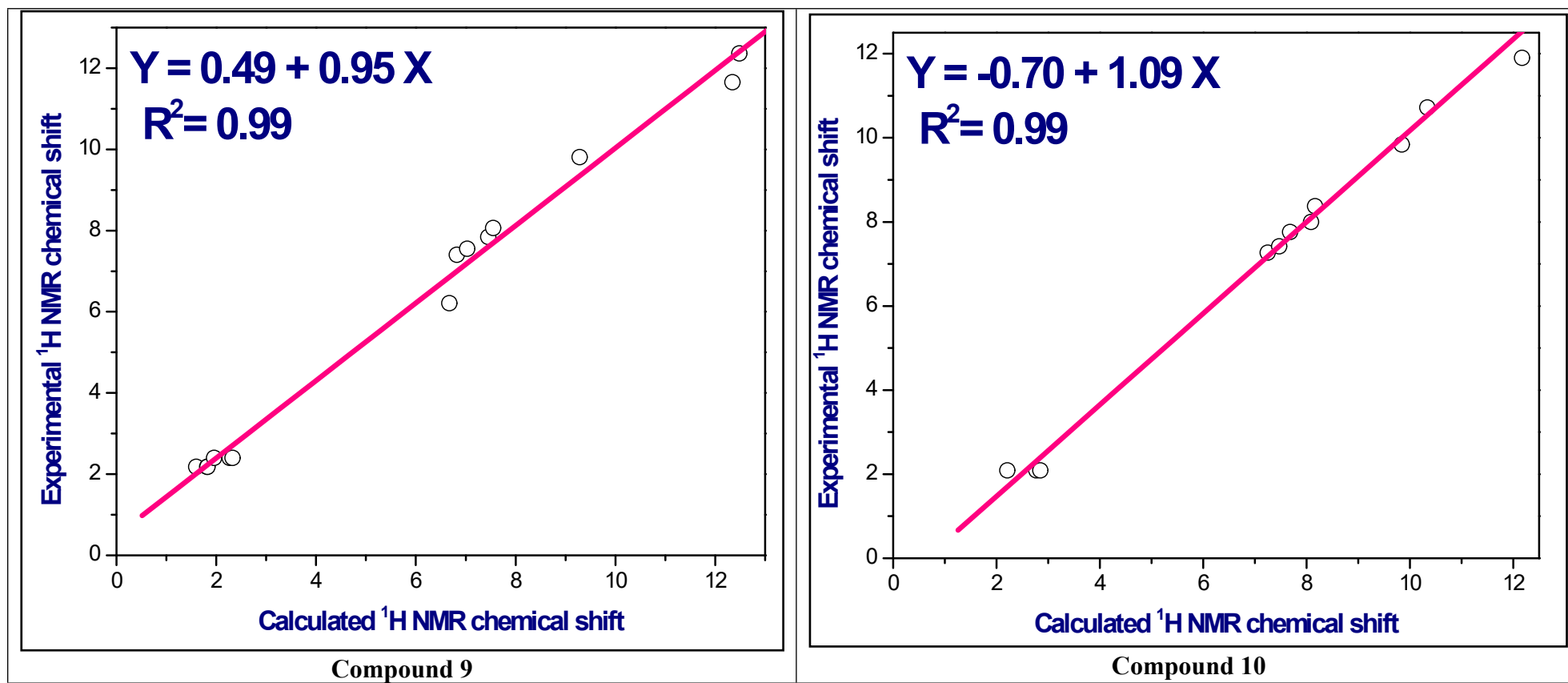


Fig. S8. The correlation relationships of the calculated *versus* experimental ^1H NMR chemical shifts of compounds **9** and **10**.

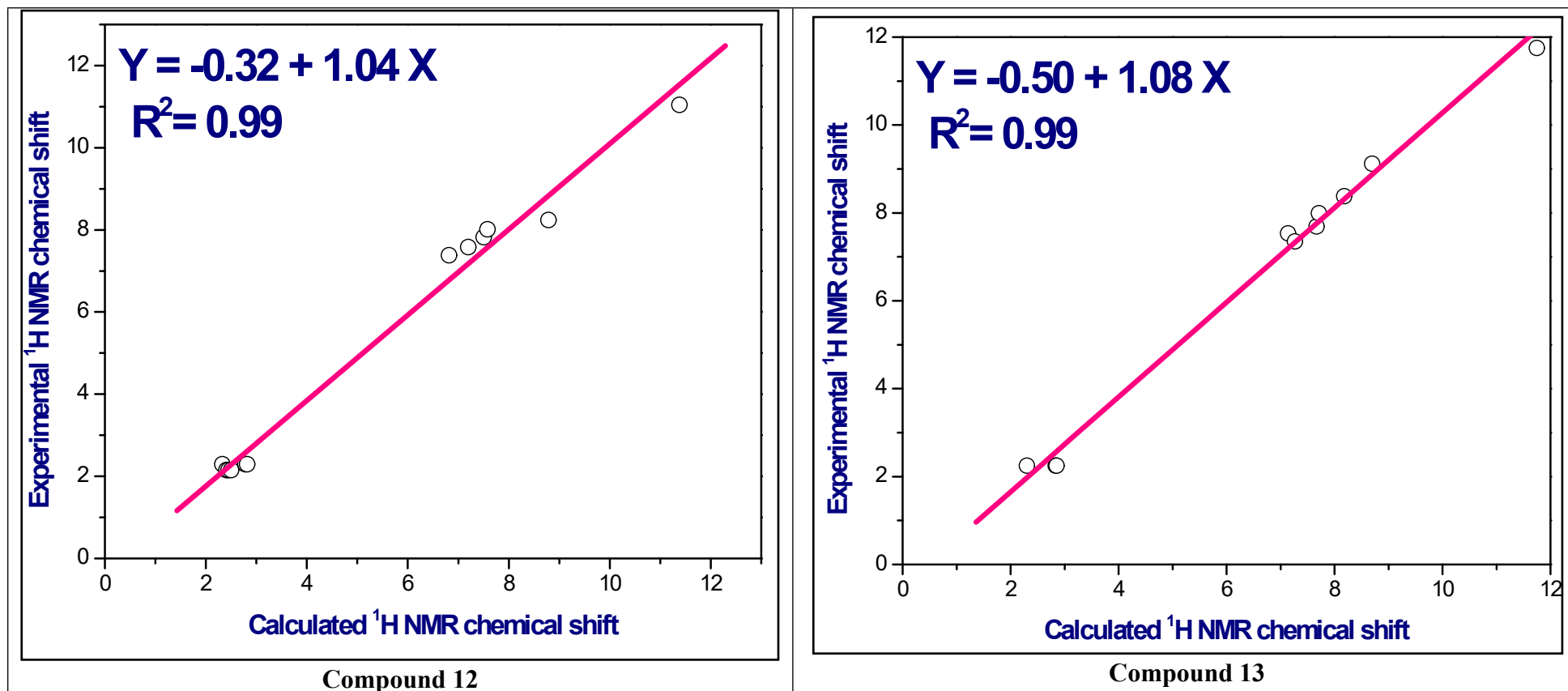


Fig. S9. The correlation relationships of the calculated *versus* experimental ^1H NMR chemical shifts of compounds **12** and **13**.

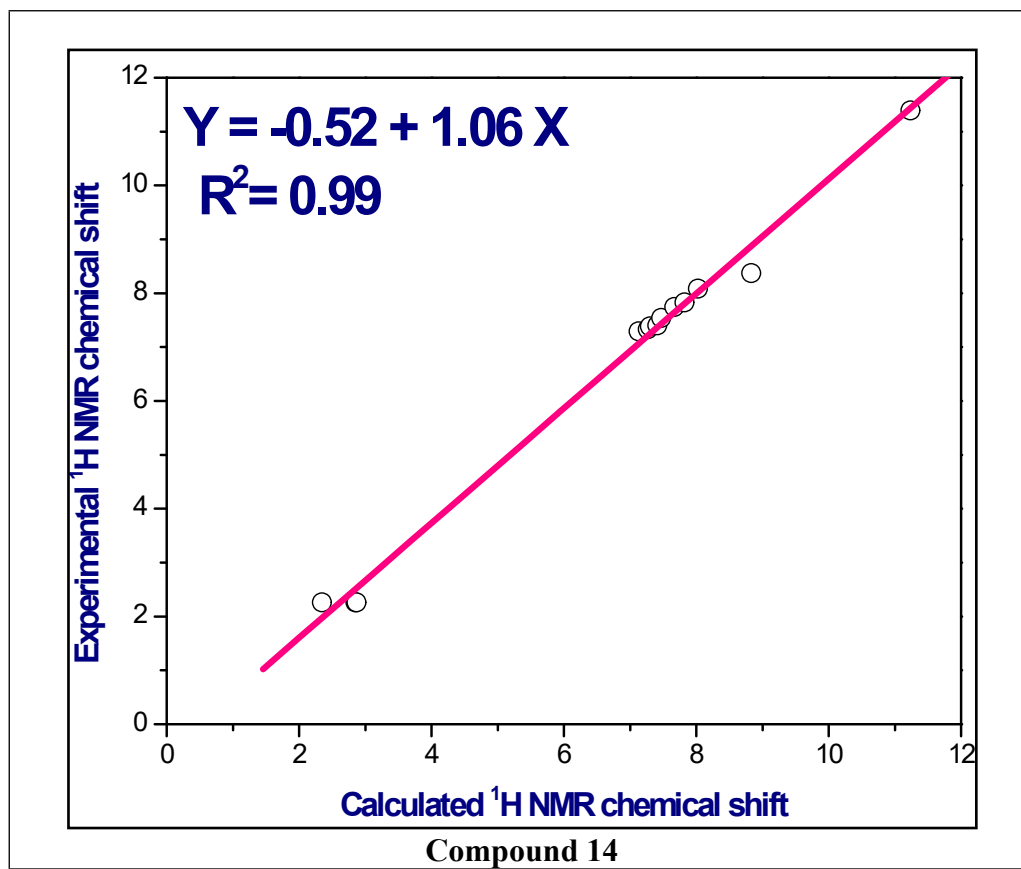


Fig. S10. The correlation relationships of the calculated *versus* experimental ^1H NMR chemical shifts of compound **14**.

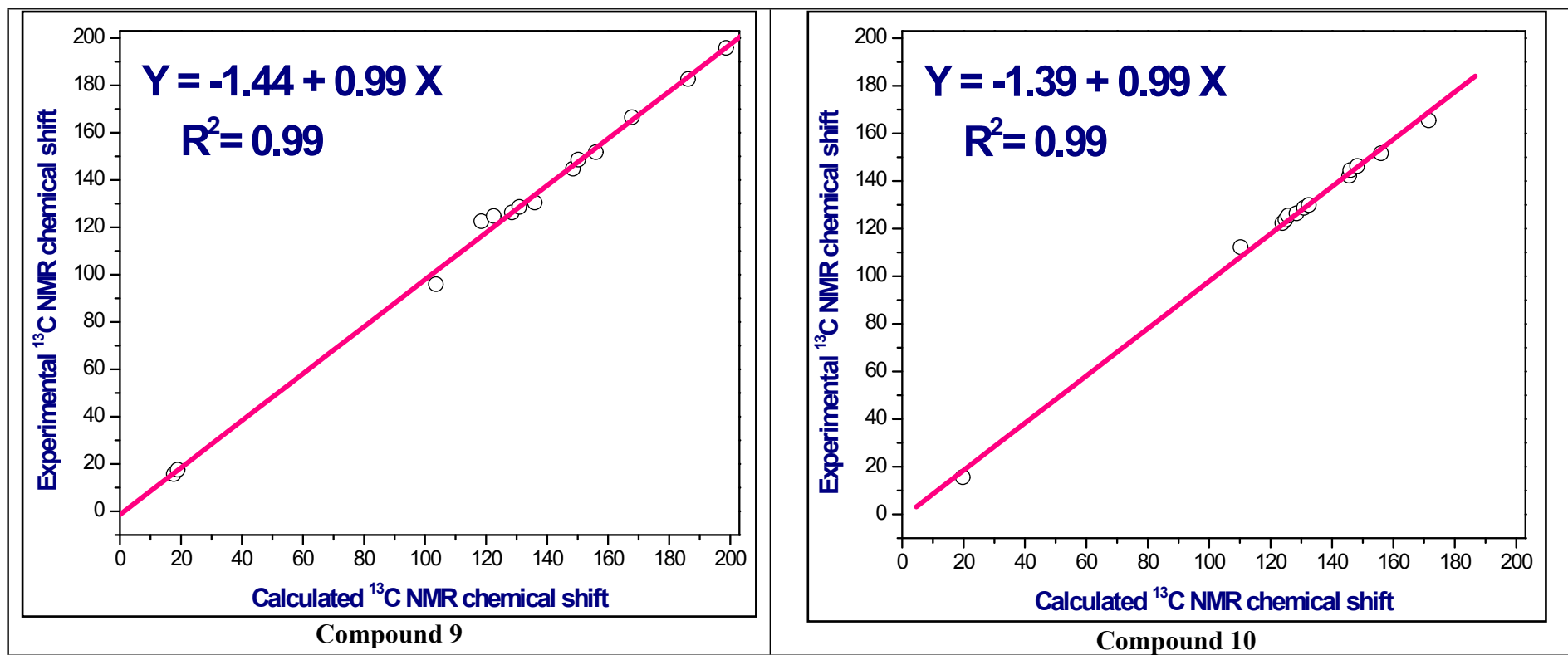


Fig. S11. The correlation relationships of the calculated *versus* experimental ^{13}C NMR chemical shifts of compounds **9** and **10**.

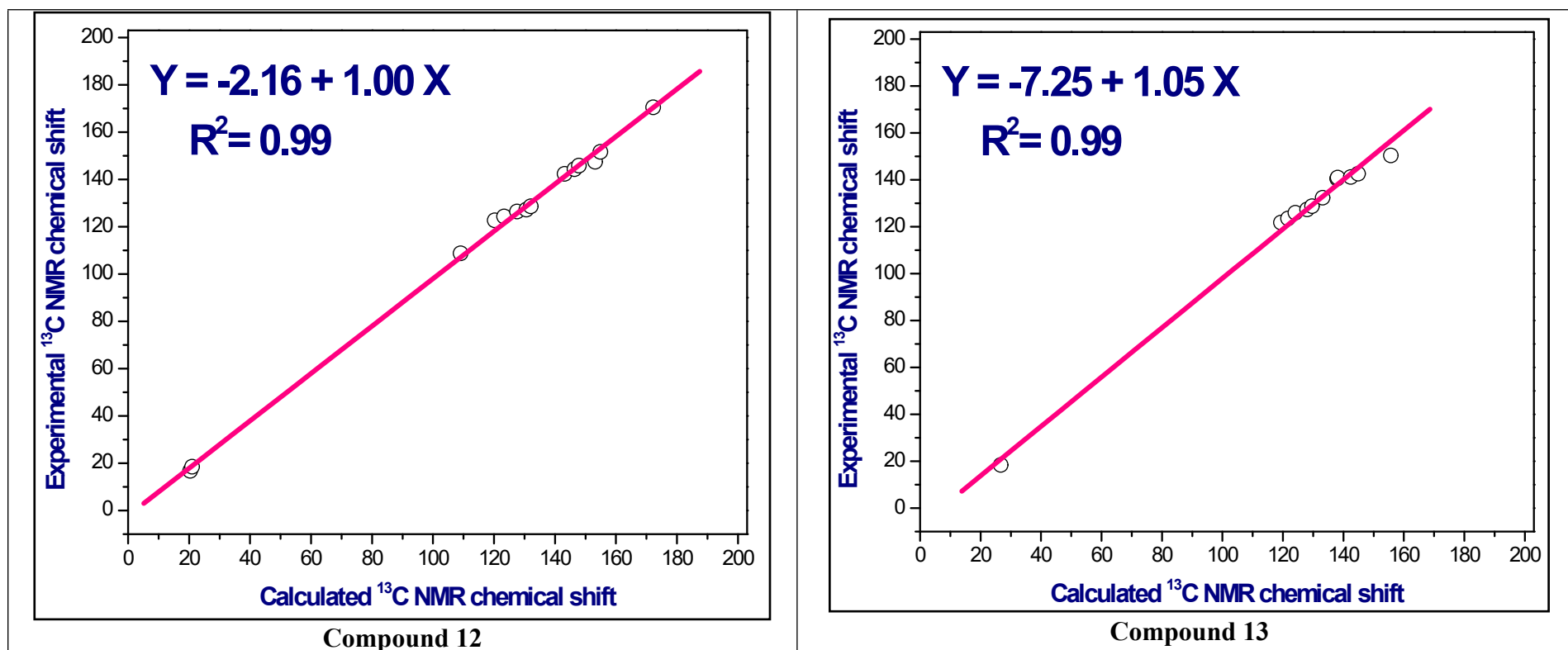


Fig. S12. The correlation relationships of the calculated *versus* experimental ^{13}C NMR chemical shifts of compounds **12** and **13**.

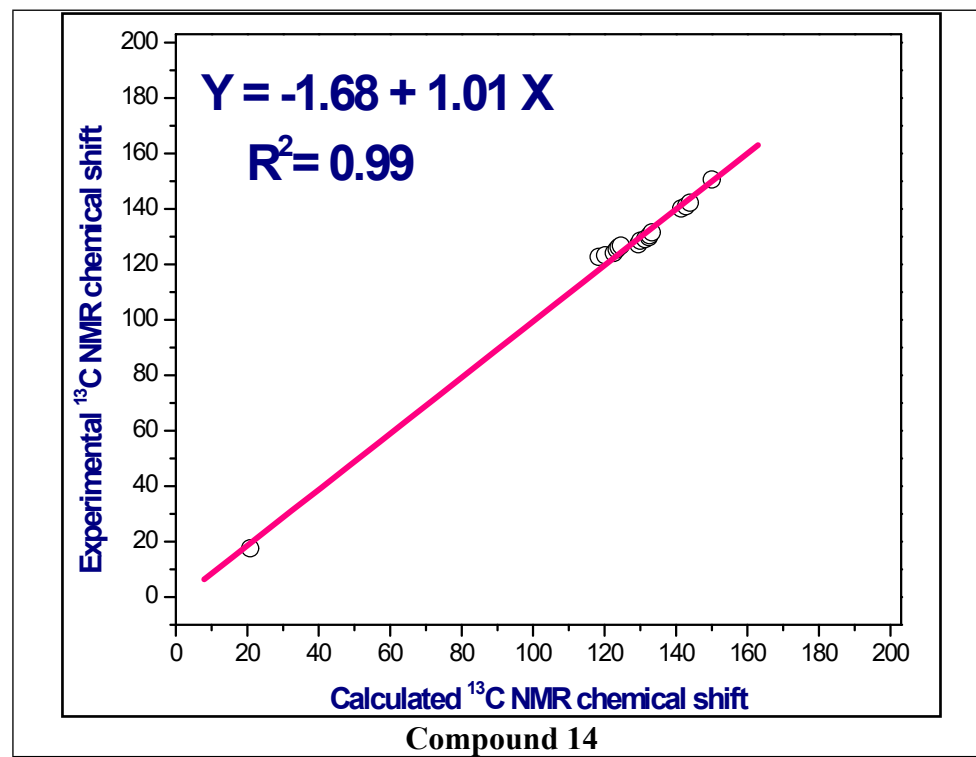


Fig. S13. The correlation relationships of the calculated *versus* experimental ¹³C NMR chemical shifts of compound **14**.

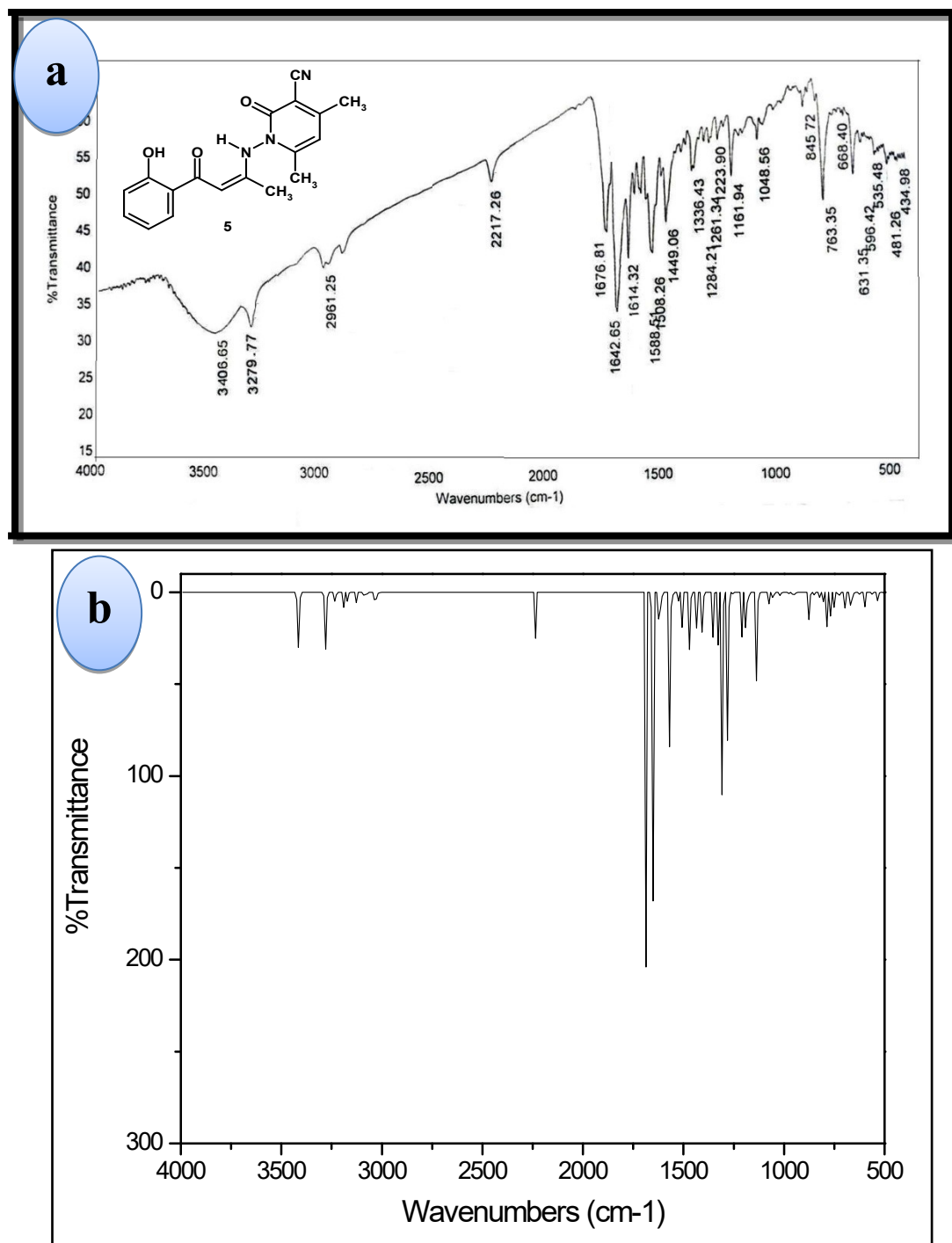


Fig. S14. (a) Experimental and (b) Calculated IR spectra of compound **5** at B3LYP/6-311++G(d,p).

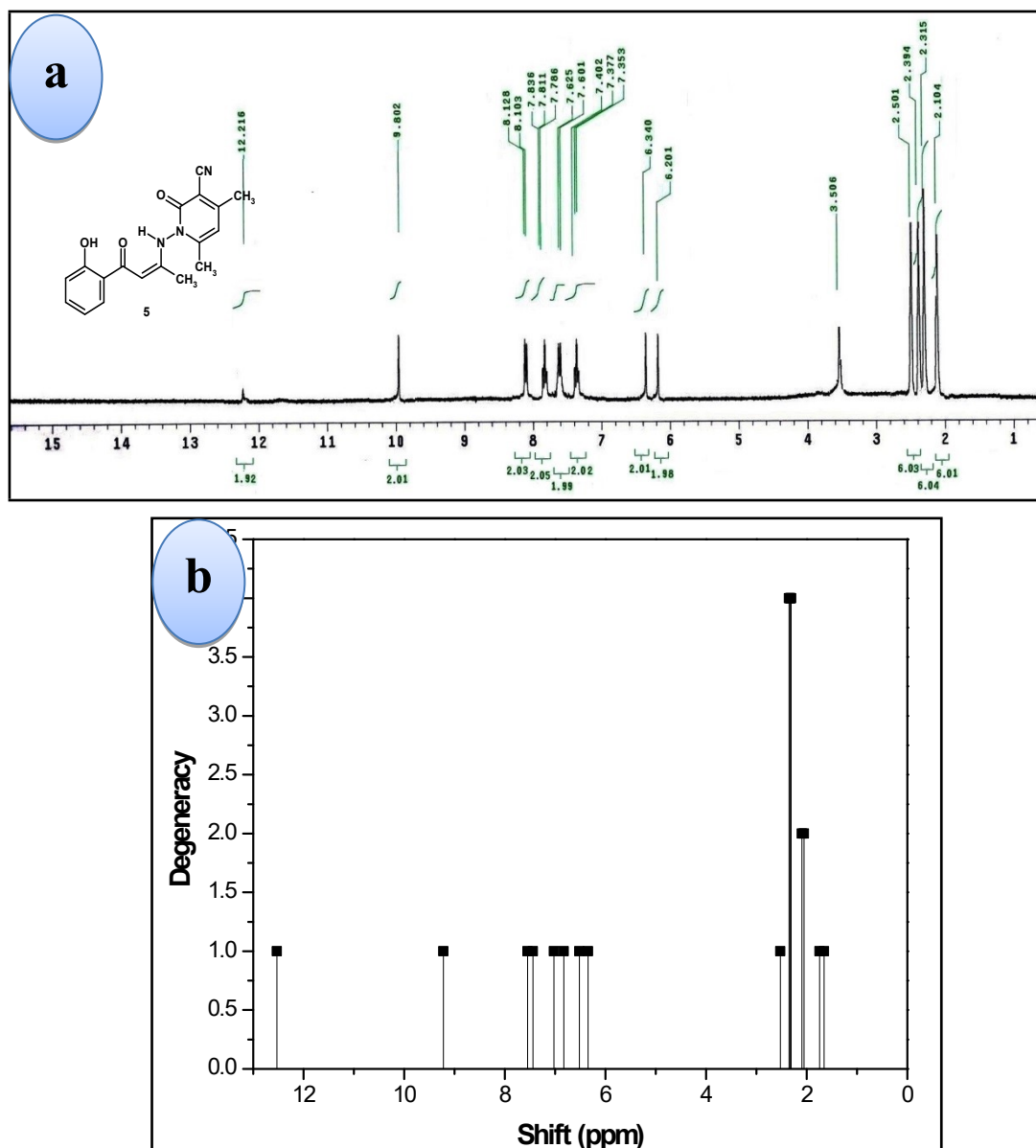


Fig. S15. (a) Experimental and (b) Calculated ^1H NMR spectra of compound **5** at B3LYP/6-311++G(d,p).

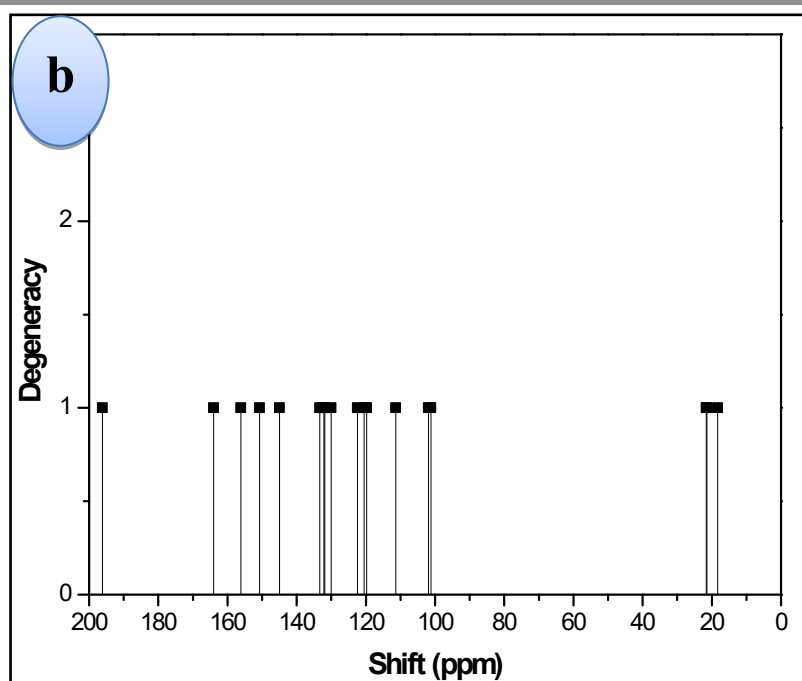
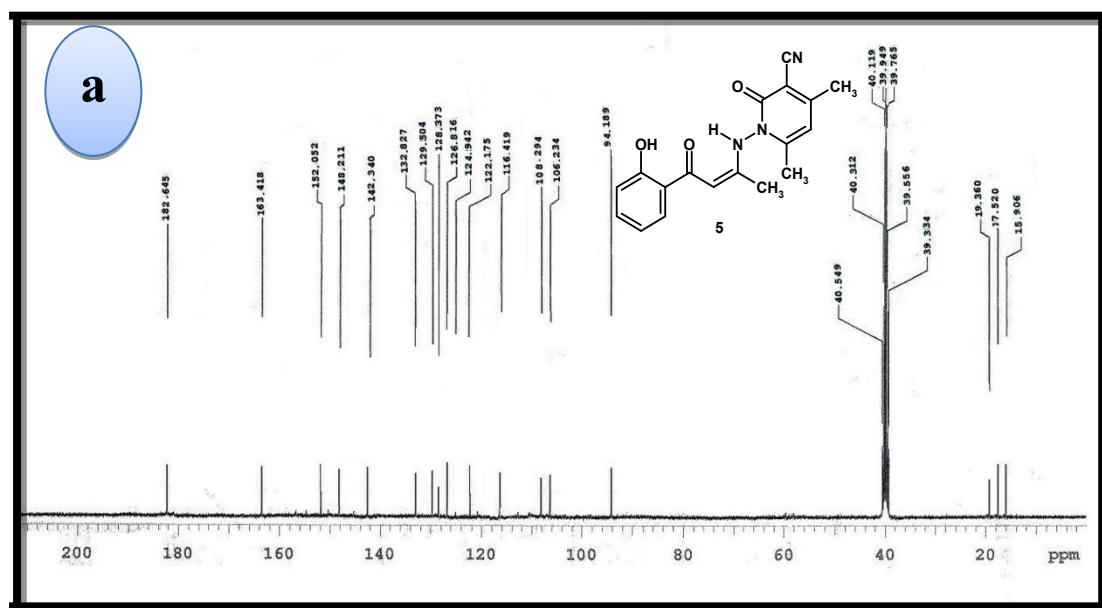


Fig. S16. (a) Experimental and (b) Calculated ^{13}C NMR spectra of compound **5** at B3LYP/6-311++G(d,p).

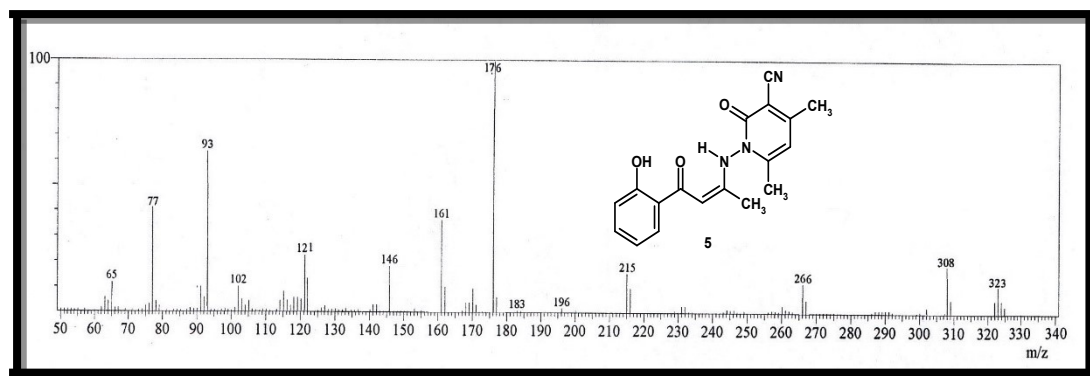


Fig. S17. Mass spectrum of compound **5**

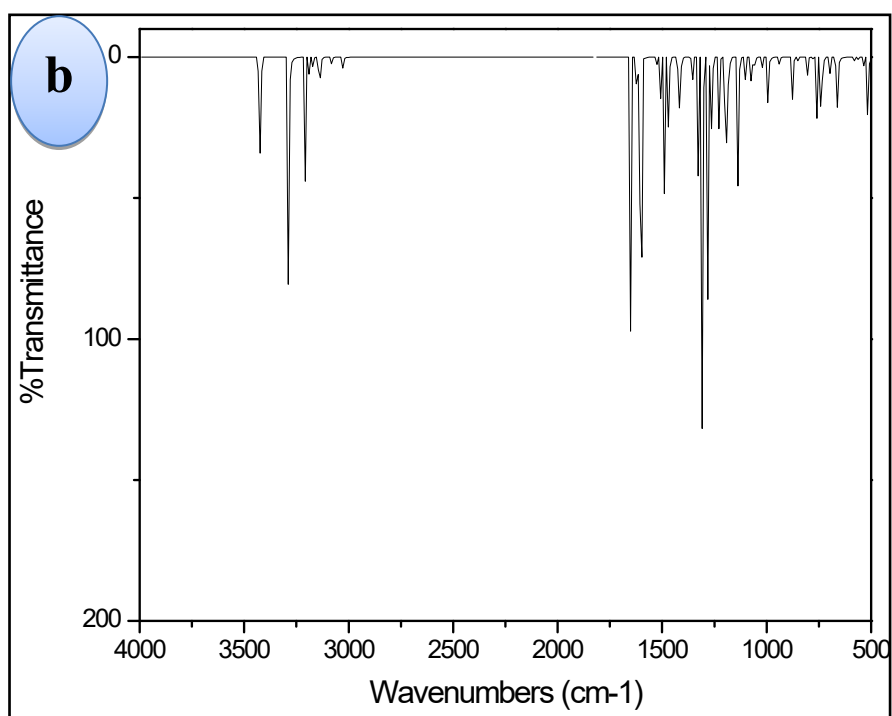
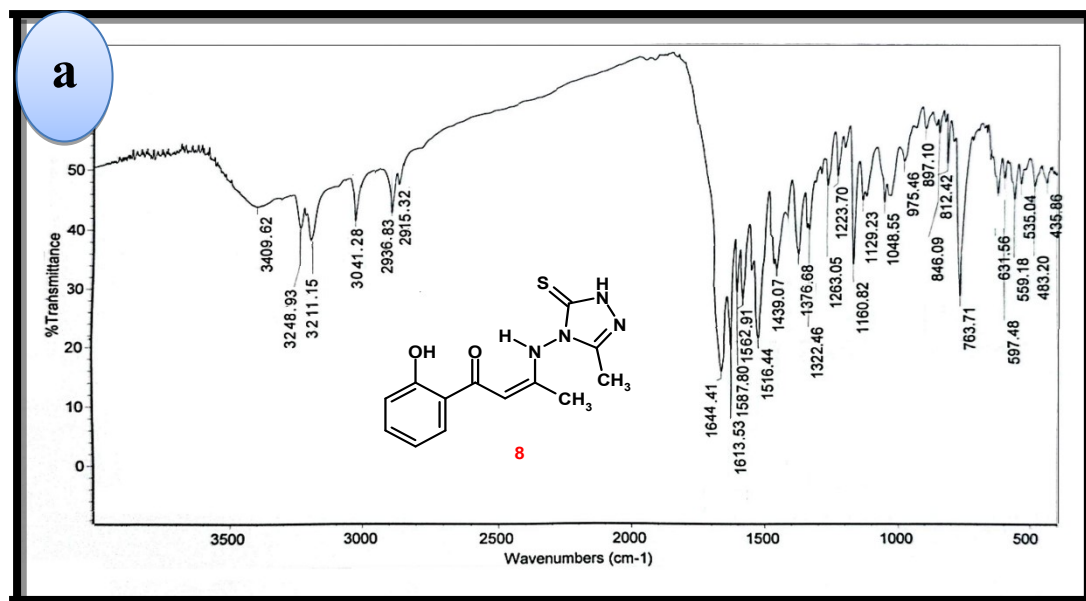


Fig. S18. (a) Experimental and (b) Calculated IR spectra of compound **8** at B3LYP/6-311++G(d,p).

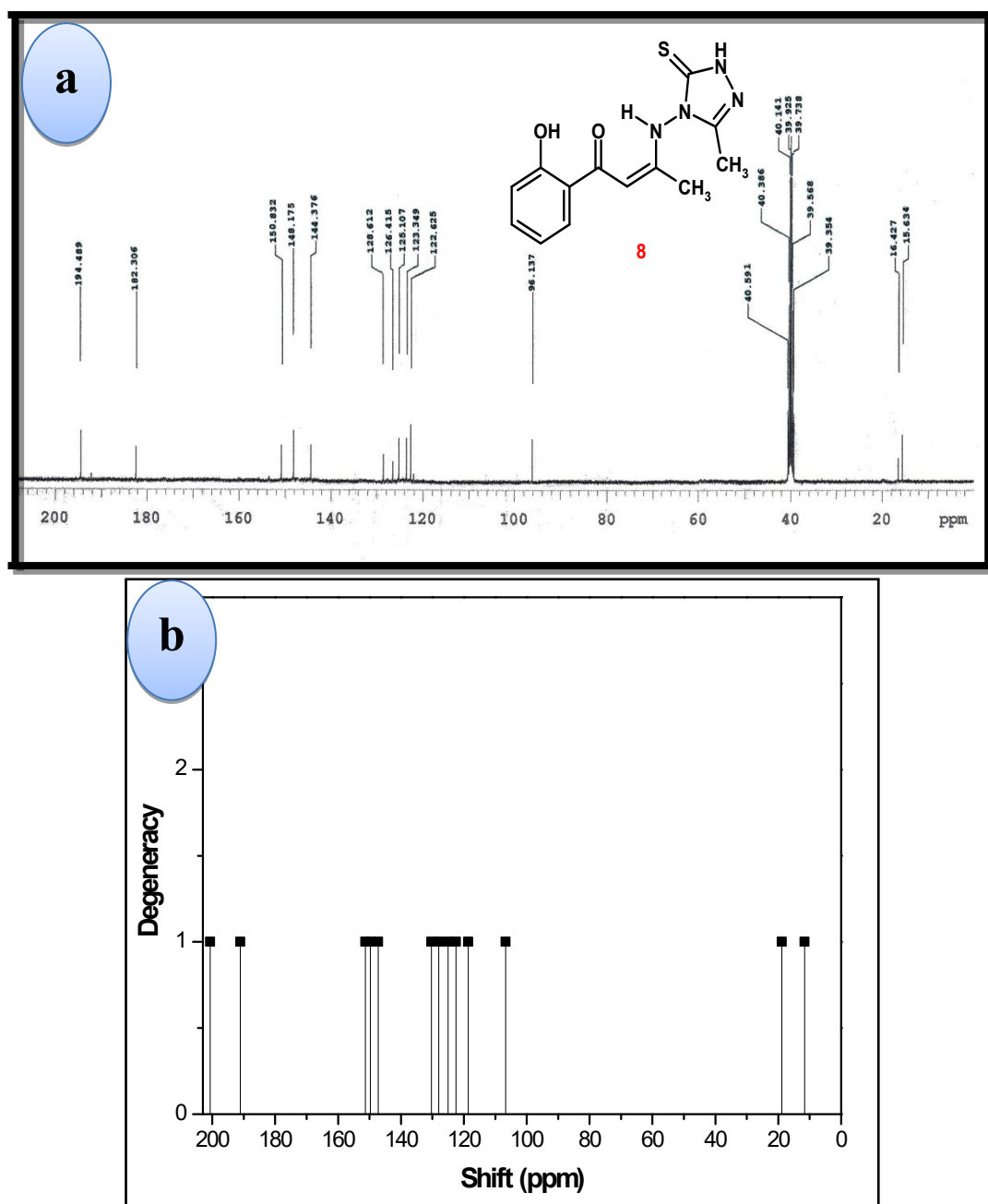


Fig. S20. (a) Experimental and (b) Calculated ^{13}C NMR spectra of compound **8** at B3LYP/6-311++G(d,p).

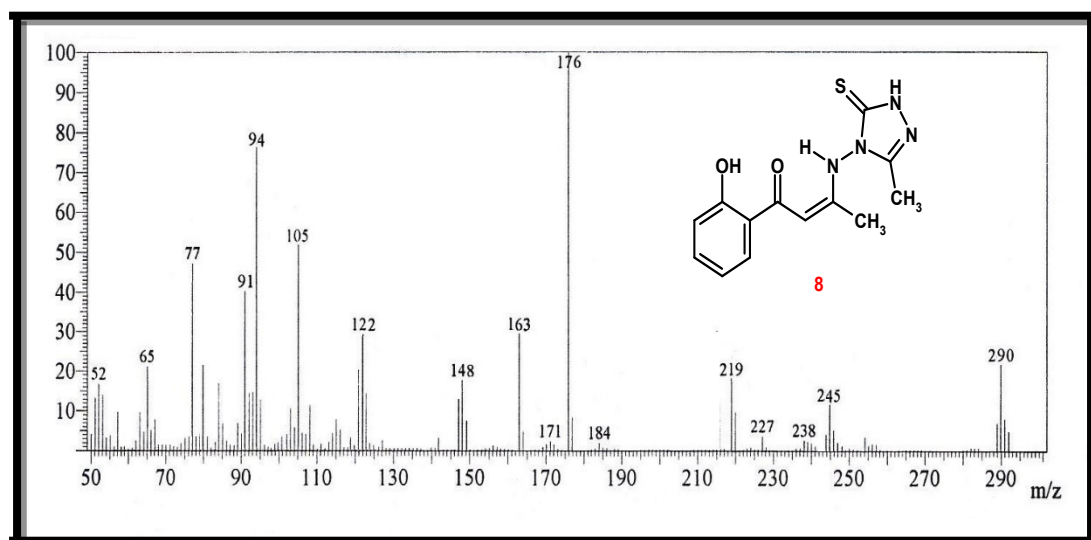


Fig. S21. Mass spectrum of compound **8**

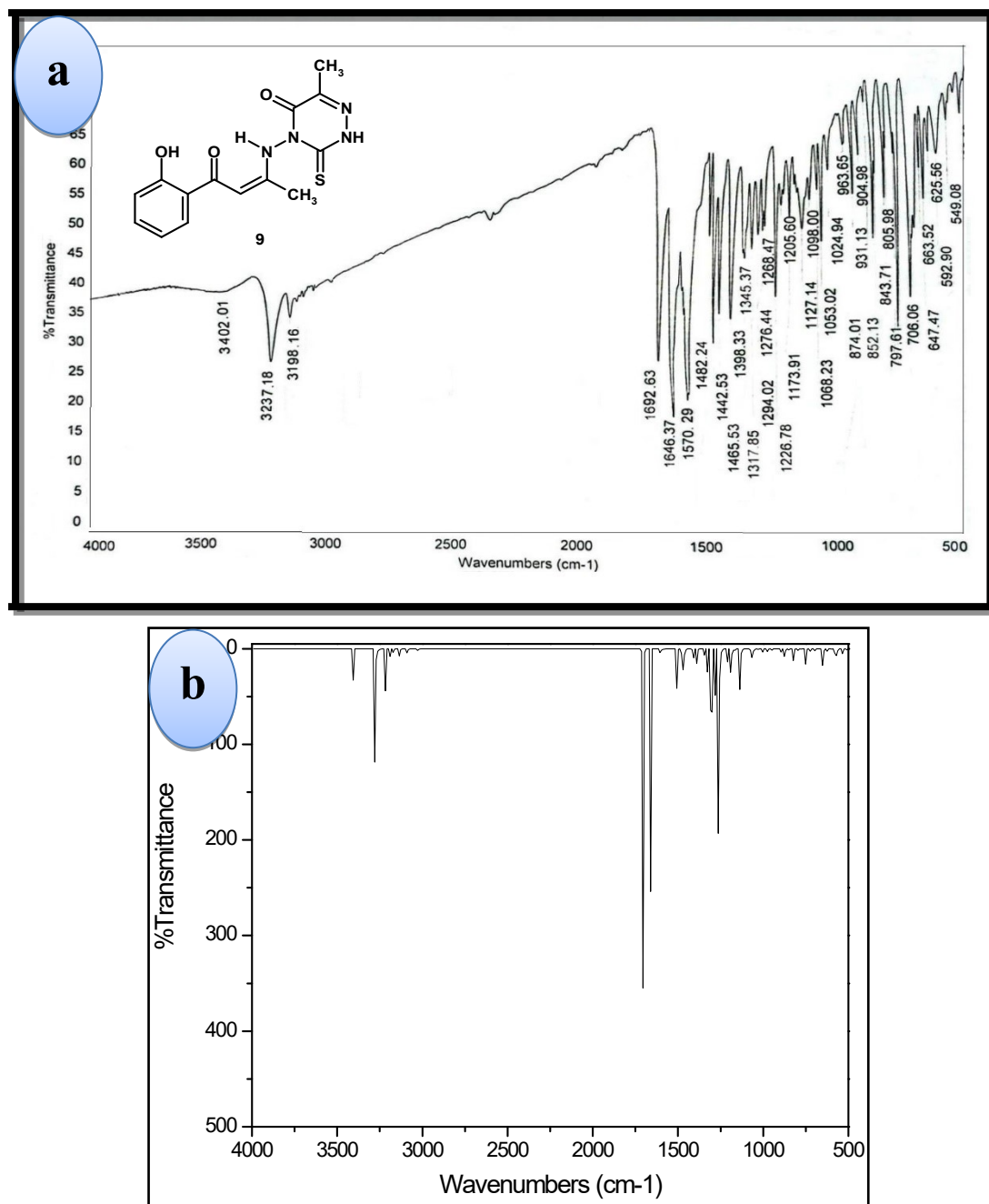


Fig. S22. (a) Experimental and (b) Calculated IR spectra of compound **9** at B3LYP/6-311++G(d,p).

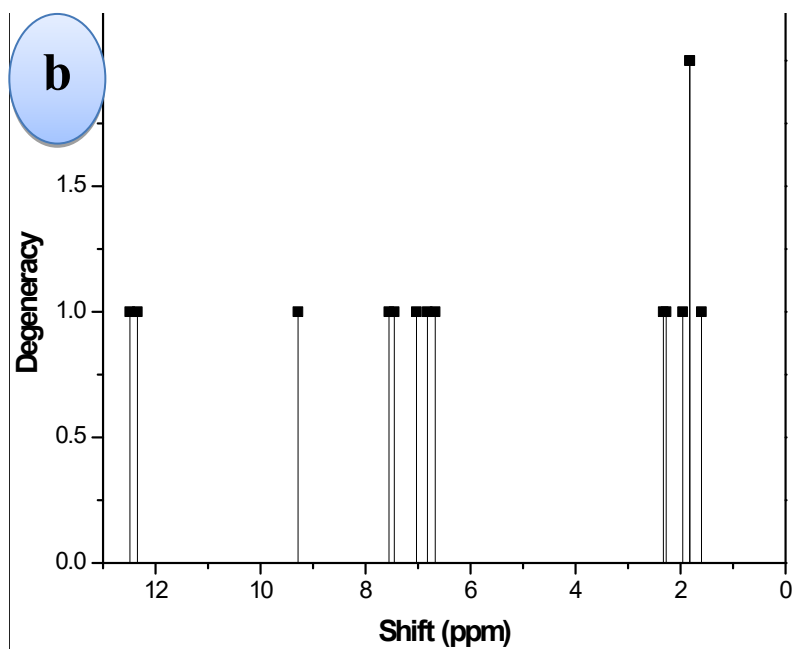
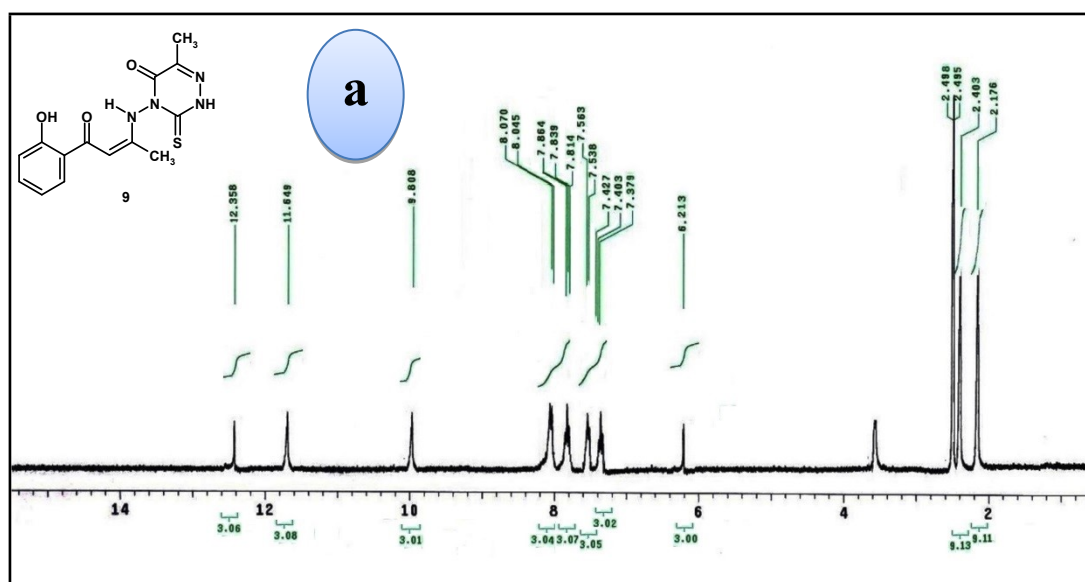


Fig. S23. (a) Experimental and (b) Calculated ^1H NMR spectra of compound **9** at B3LYP/6-311++G(d,p).

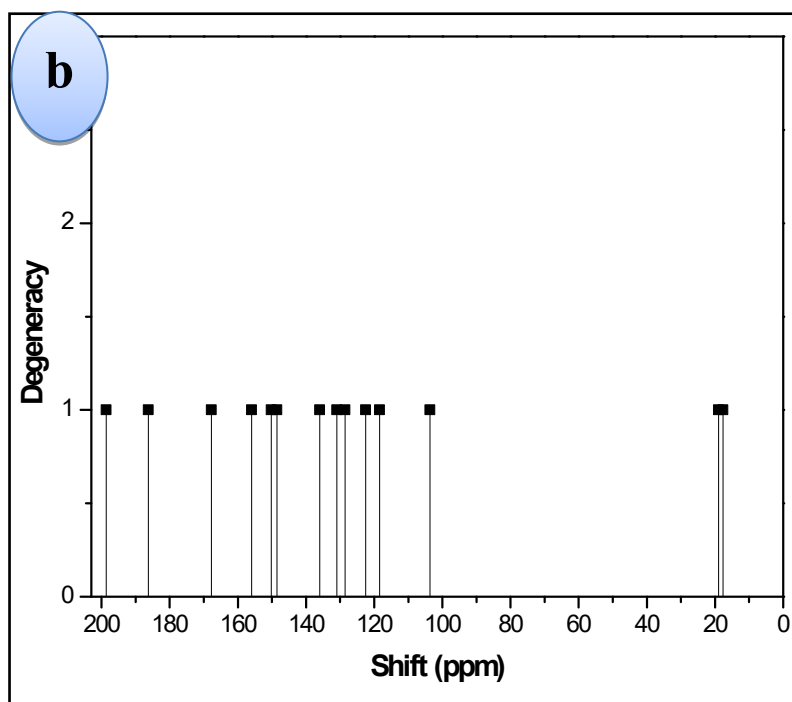
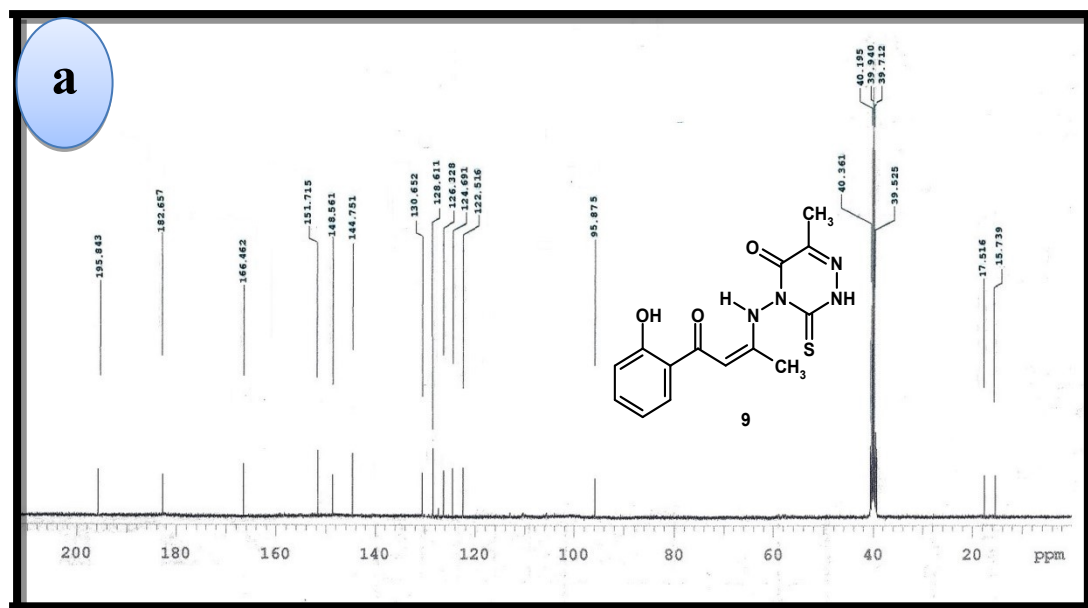


Fig. S24. (a) Experimental and (b) Calculated ^{13}C NMR spectra of compound **9** at B3LYP/6-311++G(d,p).

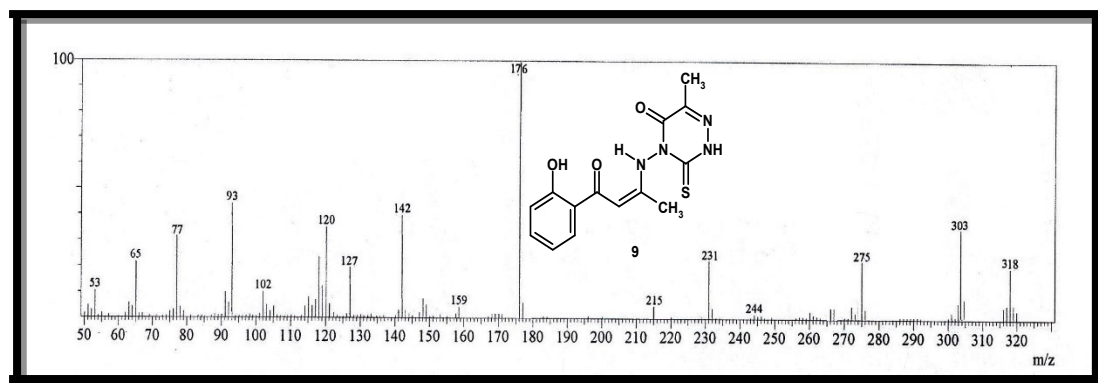


Fig. S25. Mass spectrum of compound 9

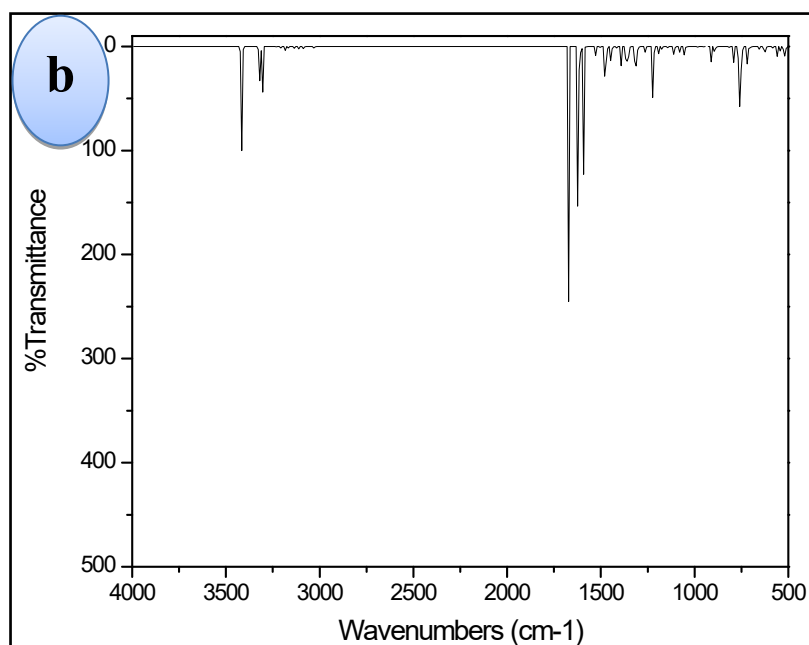
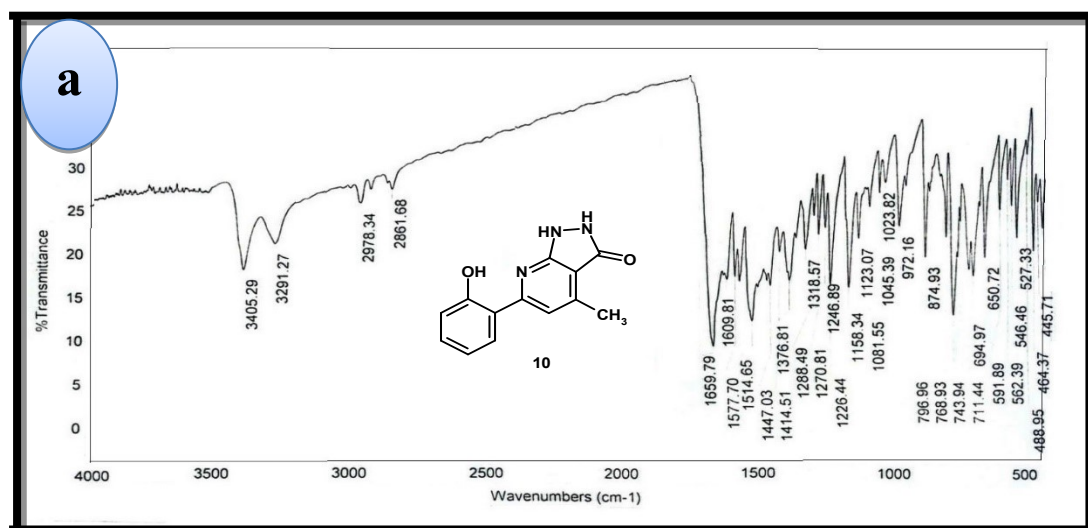


Fig. S26. (a) Experimental and (b) Calculated IR spectra of compound **10** at B3LYP/6-311++G(d,p).

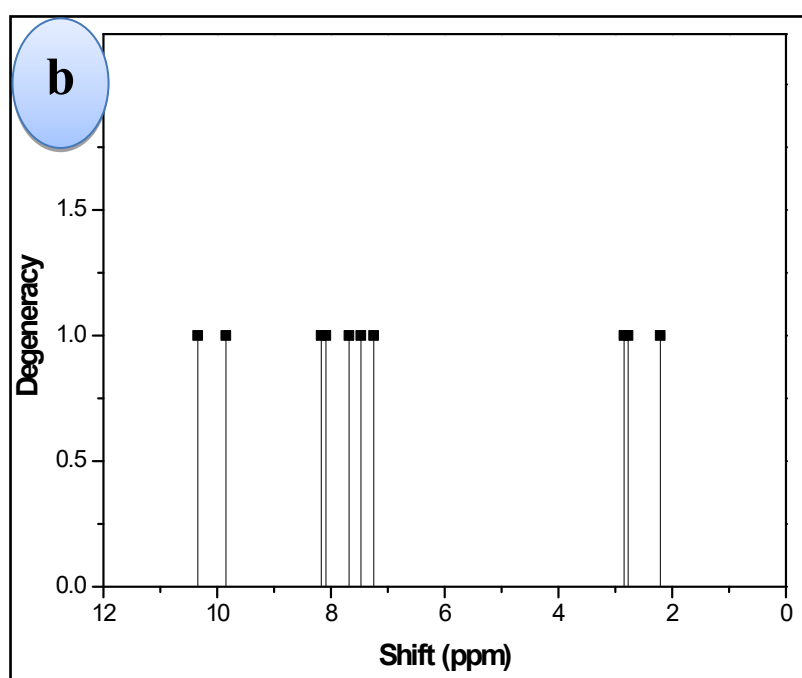
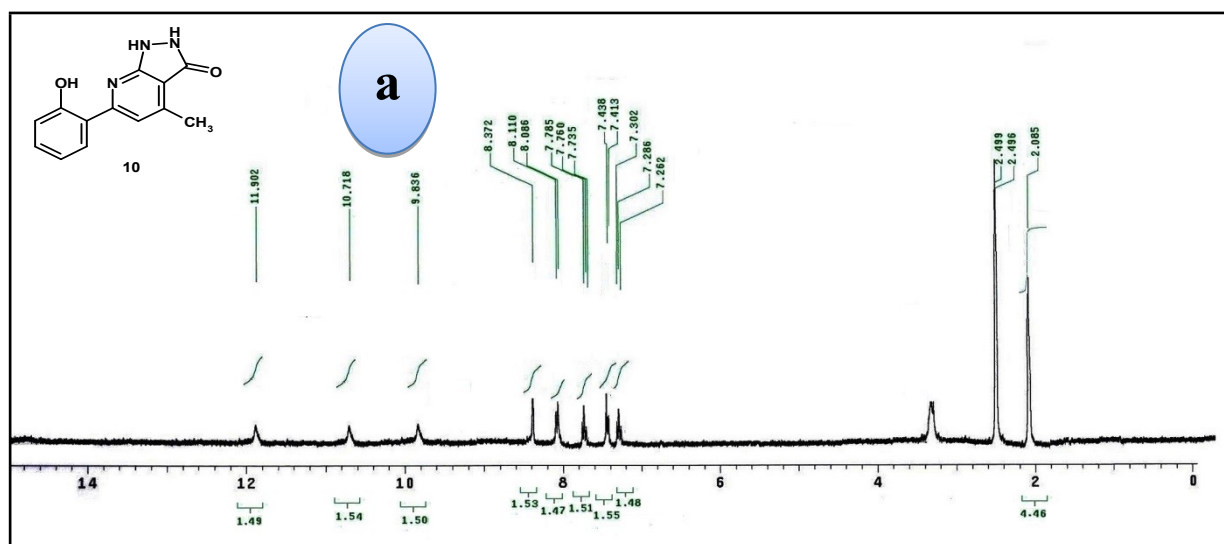


Fig. S27. (a) Experimental and (b) Calculated ^1H NMR spectra of compound **10** at B3LYP/6-311++G(d,p).

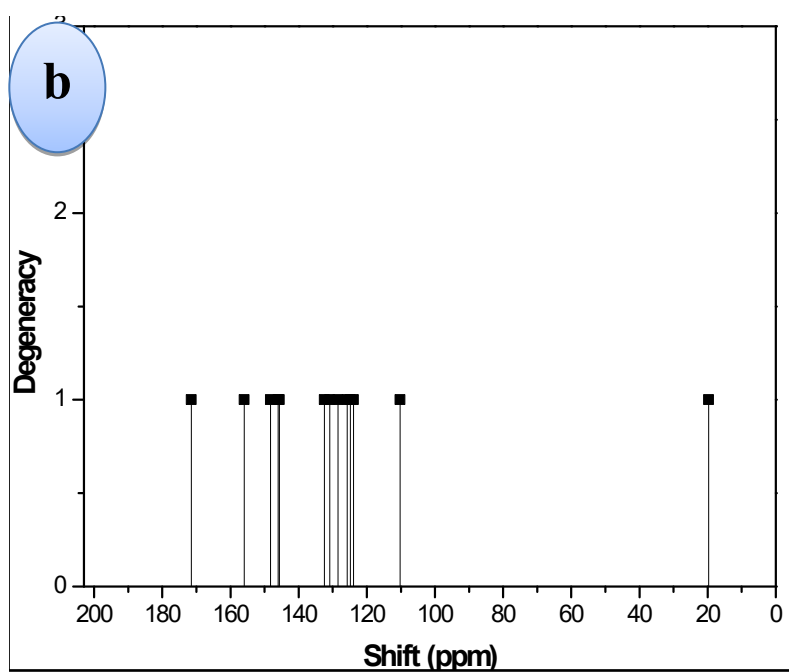
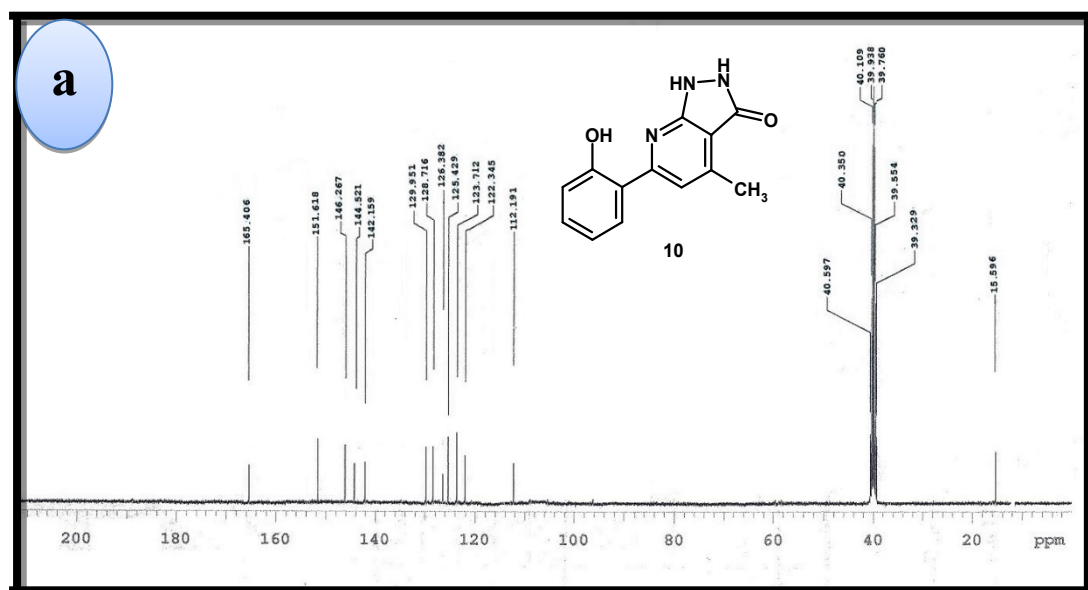


Fig. S28. Experimental and Calculated ^{13}C NMR spectra of compound **10** at B3LYP/6-311++G(d,p).

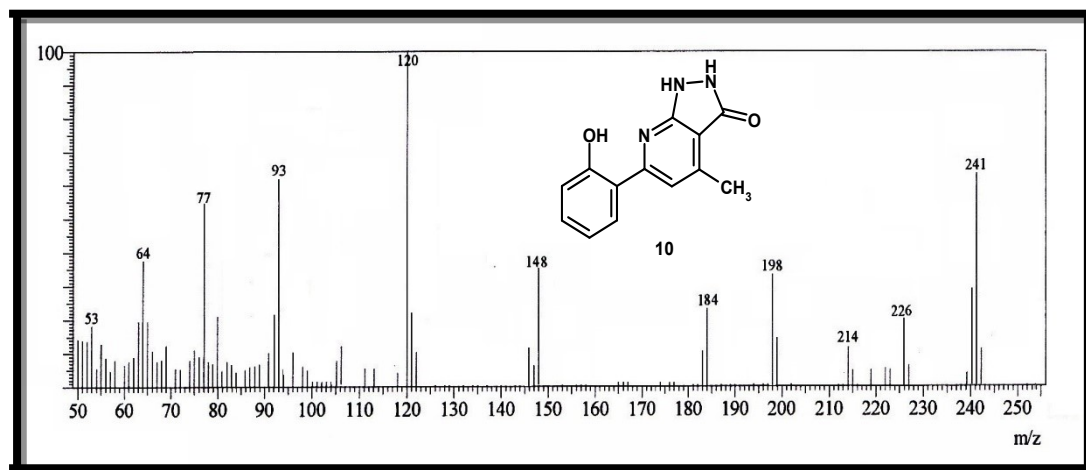


Fig. S29. Mass spectrum of compound **10**

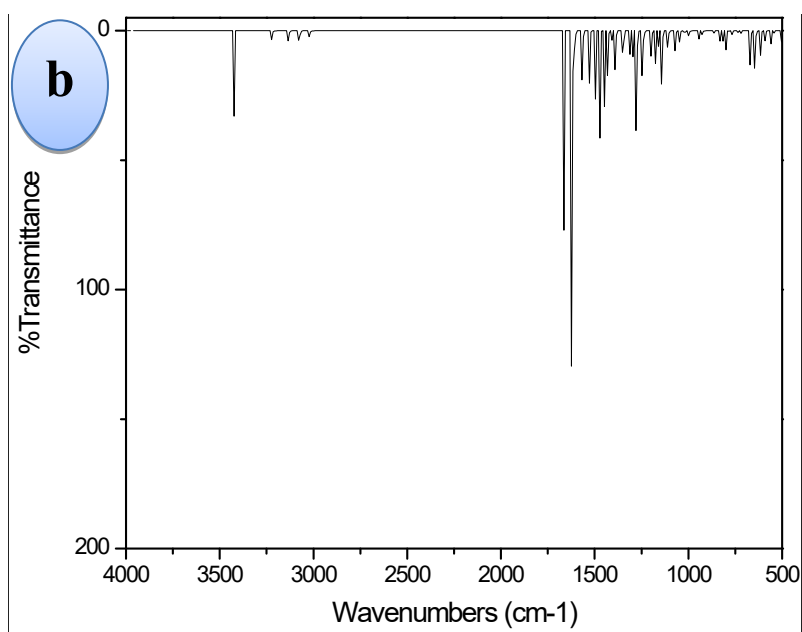
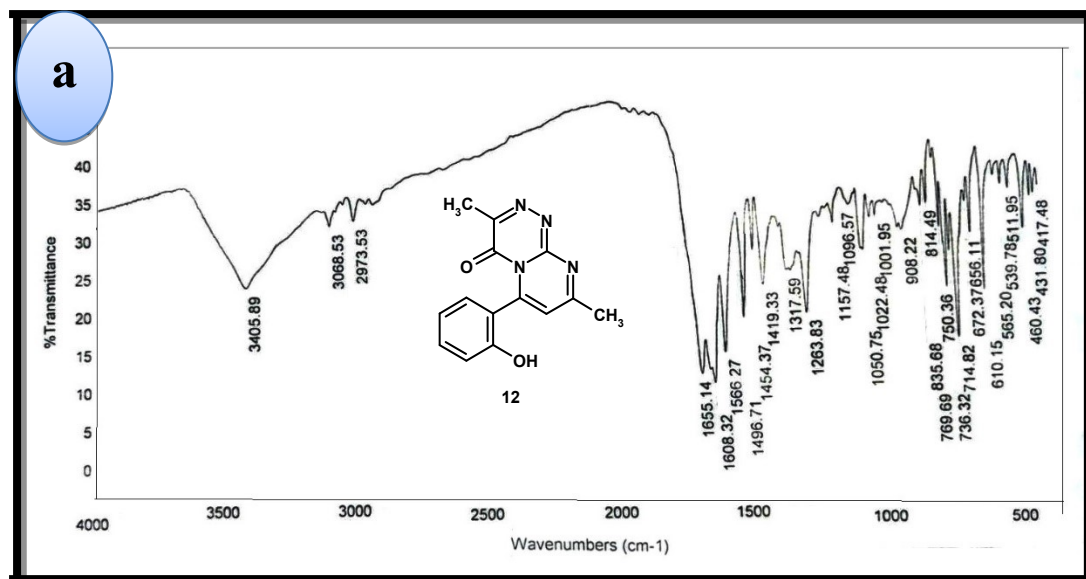


Fig. S30. (a) Experimental and (b) Calculated IR spectra of compound **12** at B3LYP/6-311++G(d,p).

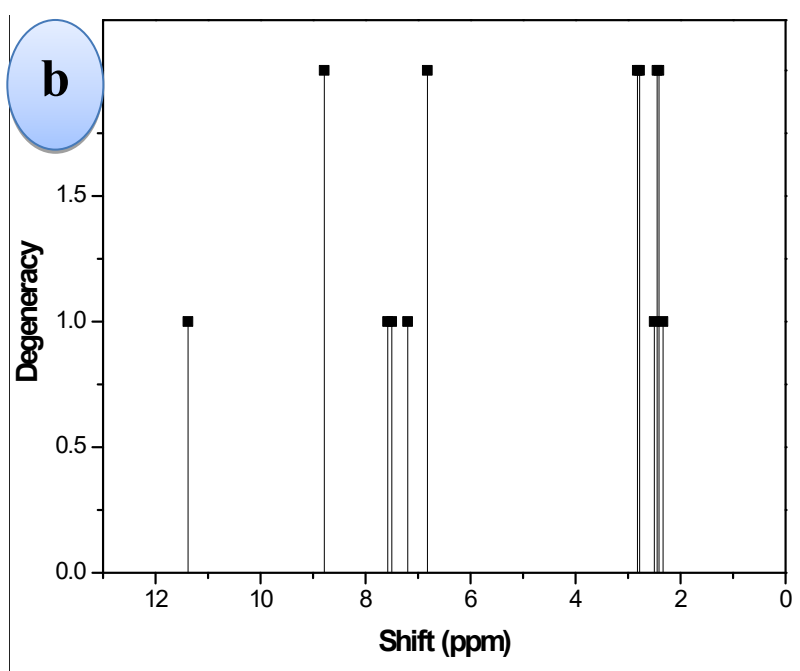
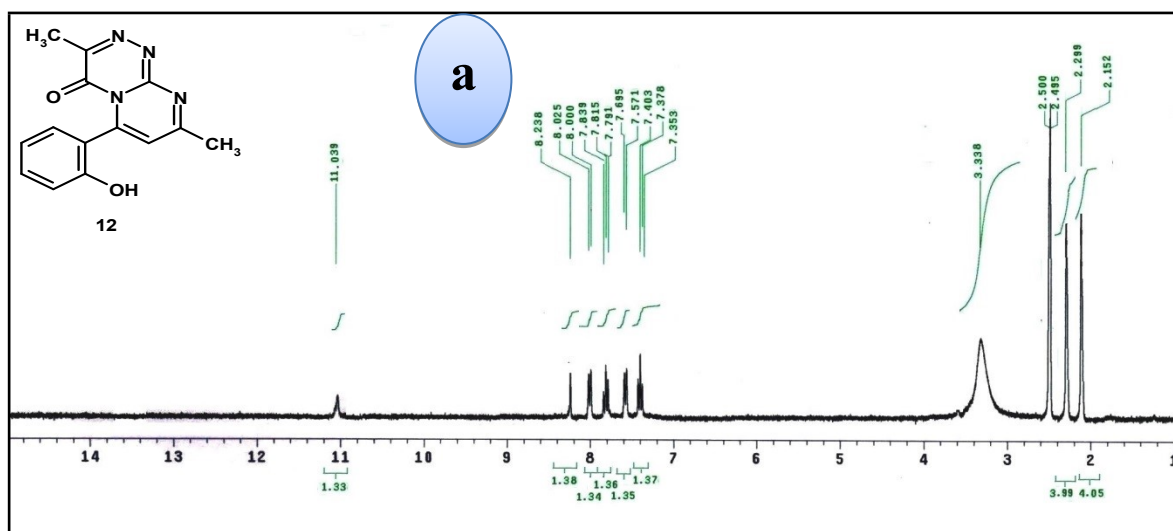


Fig. S31. (a) Experimental and (b) Calculated ^1H NMR spectra of compound **12** at B3LYP/6-311++G(d,p).

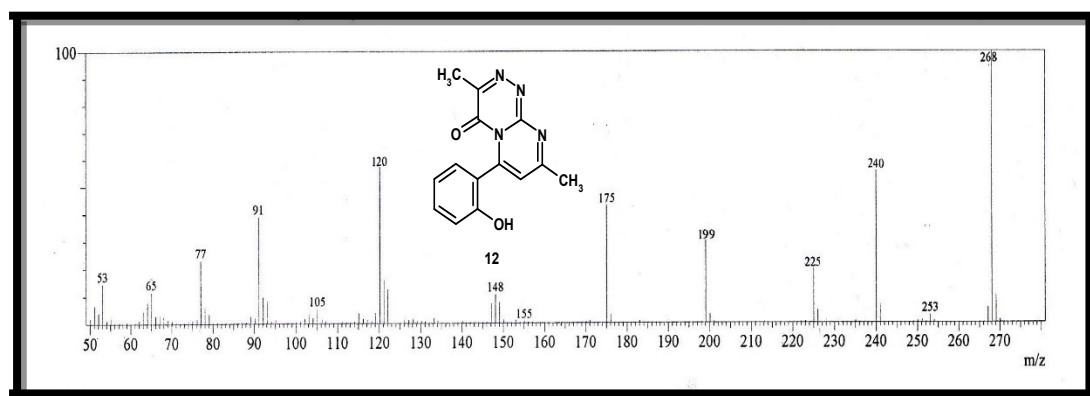


Fig. S33. Mass spectrum of compound **12**

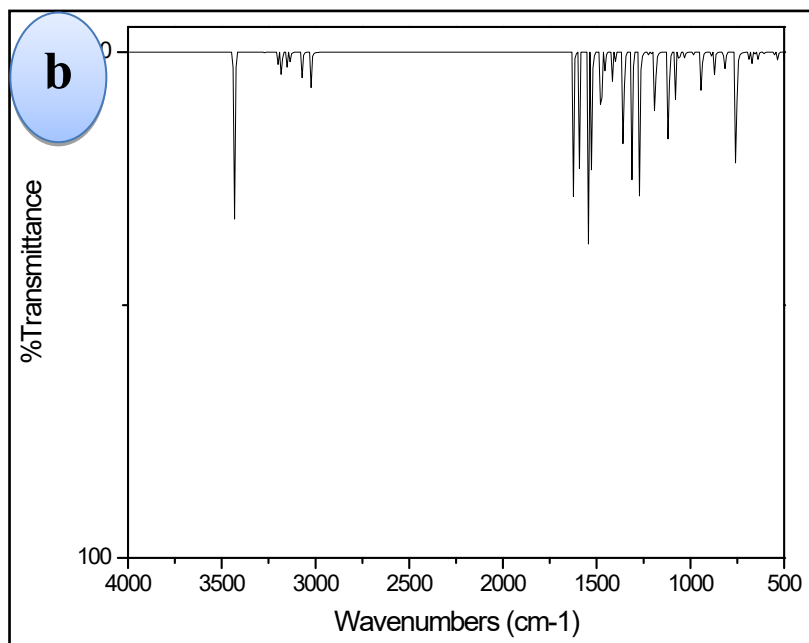
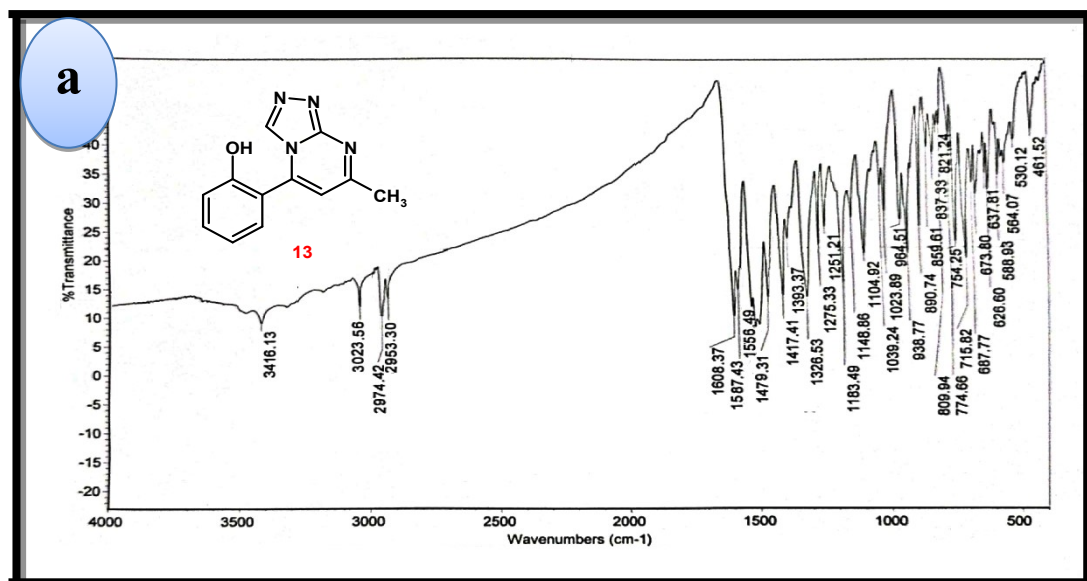


Fig. S34. (a) Experimental and (b) Calculated IR spectra of compound **13** at B3LYP/6-311++G(d,p).

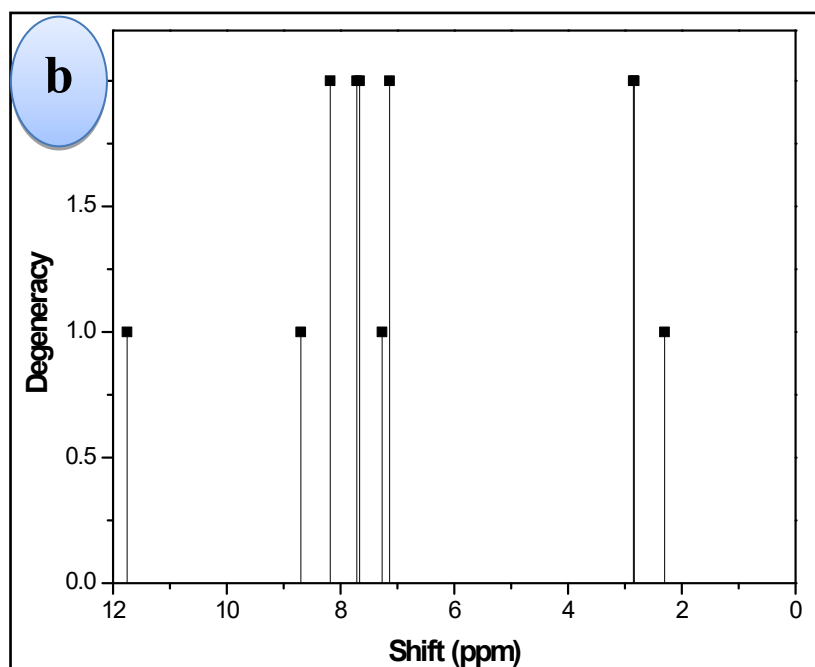
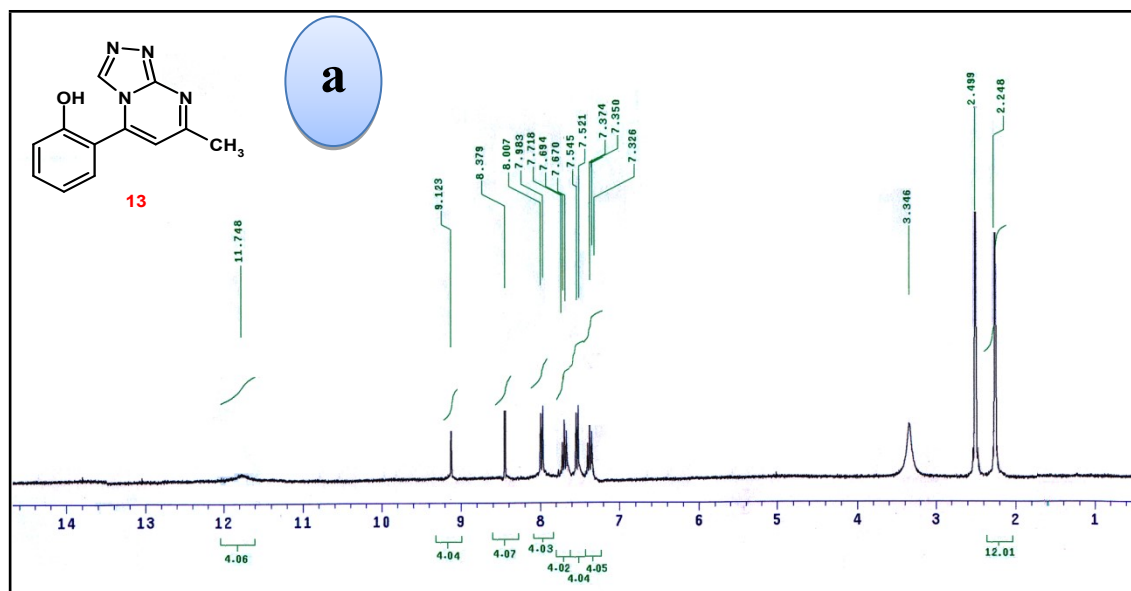


Fig. S35. (a) Experimental and (b) Calculated ^1H NMR spectra of compound **13** at B3LYP/6-311++G(d,p).

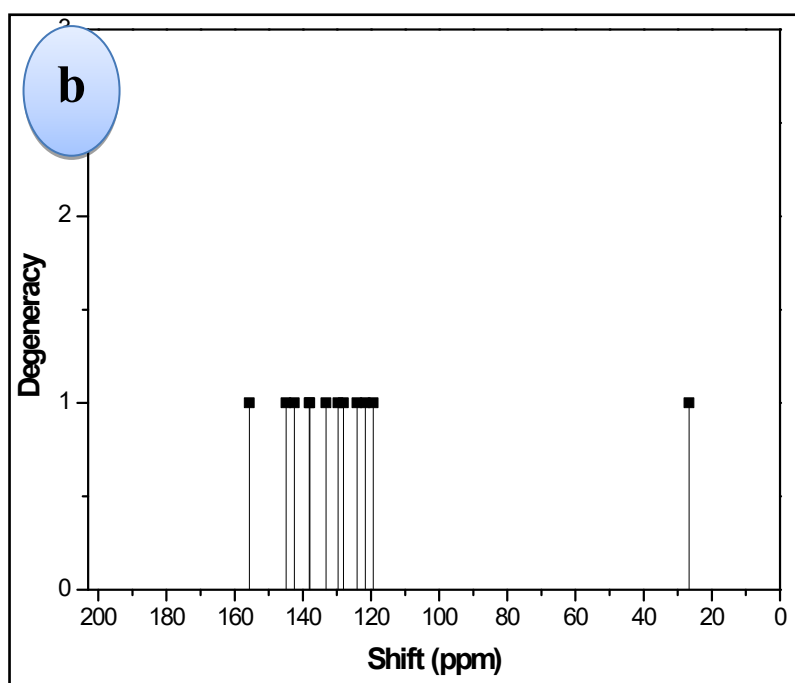
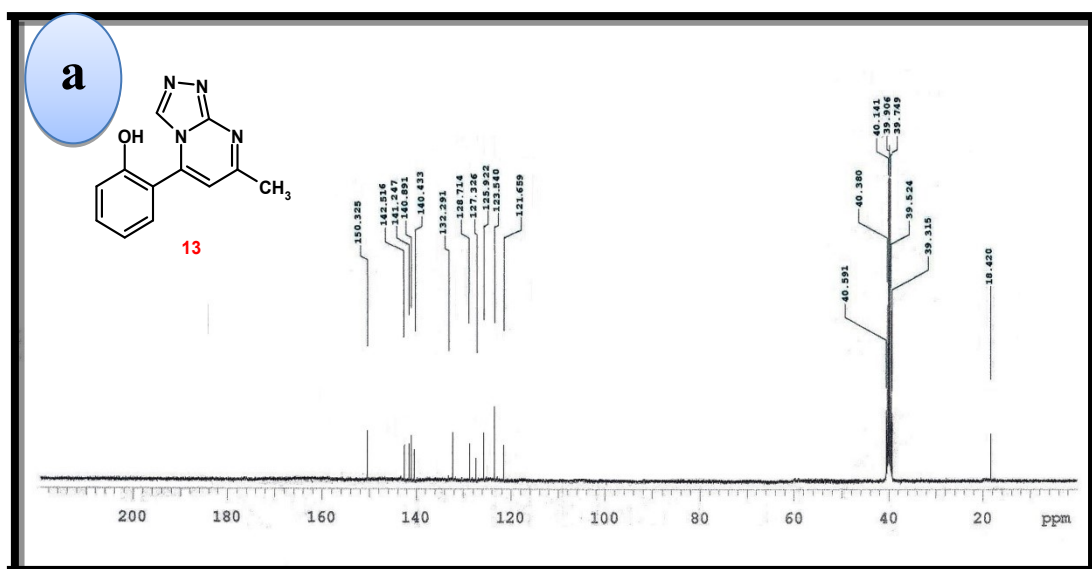


Fig. S36. Experimental and Calculated ¹³C NMR spectra of compound **13** at B3LYP/6-311++G(d,p).

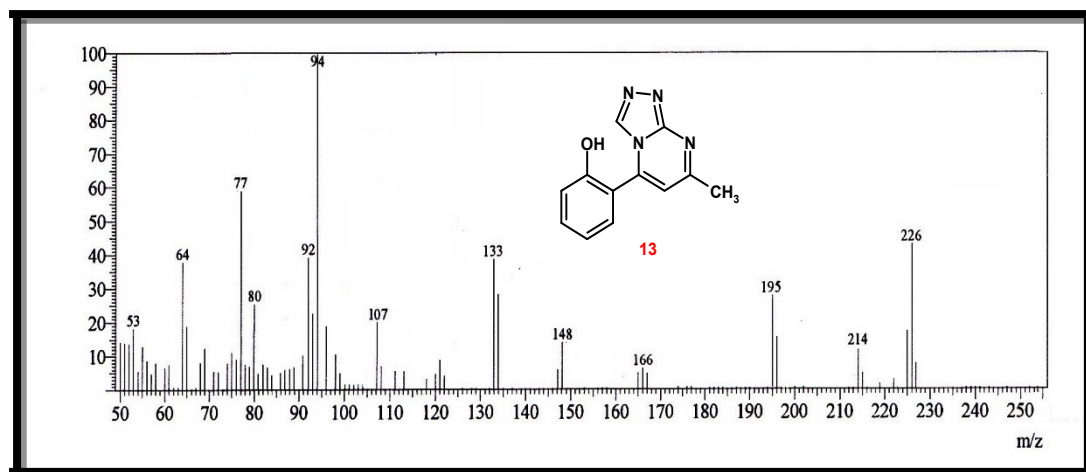


Fig. S37. Mass spectrum of compound **13**

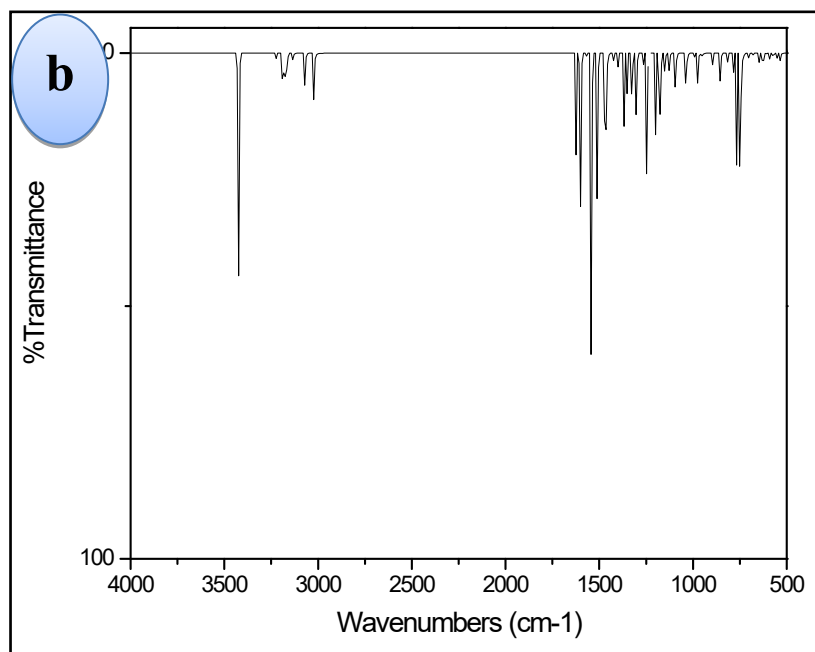
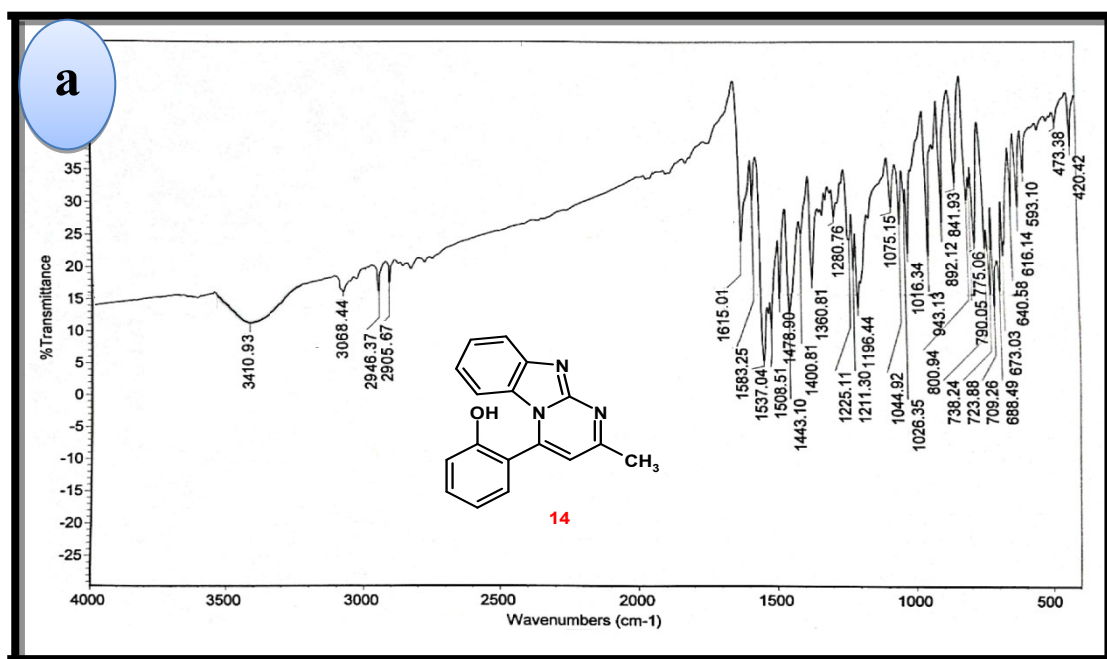


Fig. S38. (a) Experimental and (b) Calculated IR spectra of compound **14** at B3LYP/6-311++G(d,p).

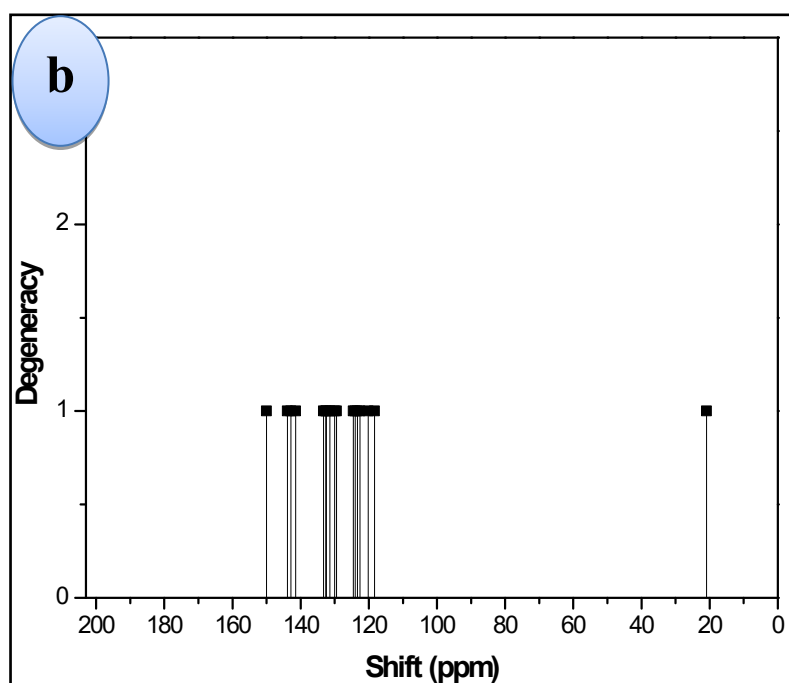
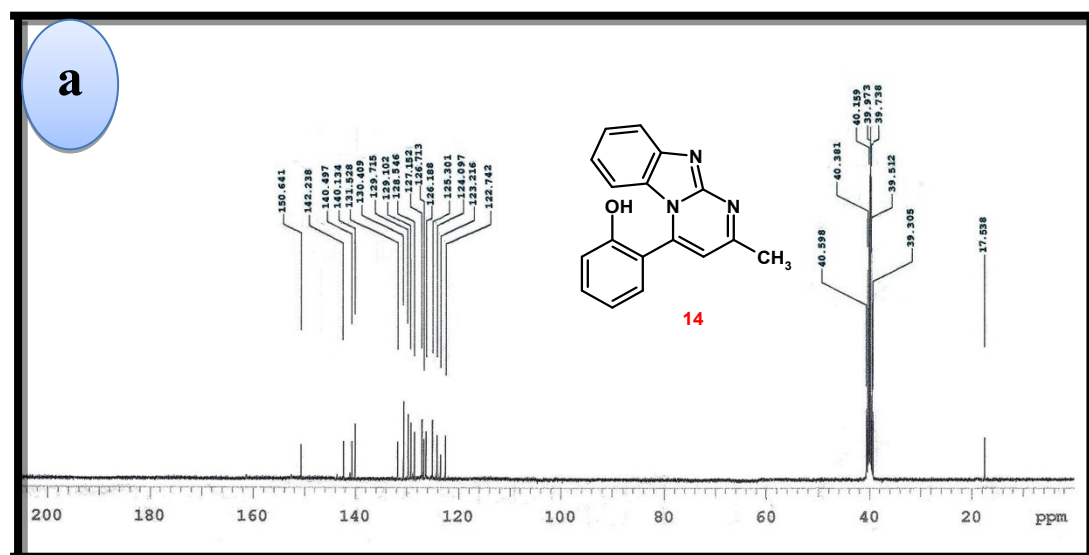


Fig. S40. Experimental and Calculated ^{13}C NMR spectra of compound **14** at B3LYP/6-311++G(d,p).

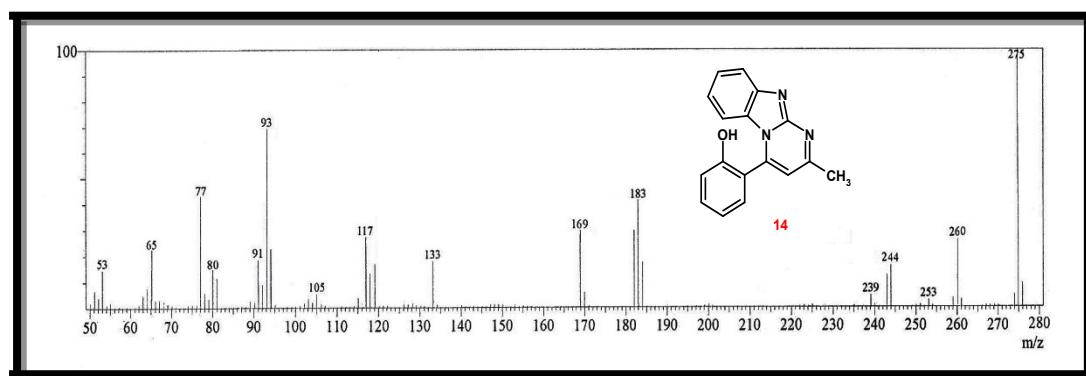


Fig. S41. Mass spectrum of compound **14**



Fig. S42. The bioavailability radars of compounds 9-10 and 12-14

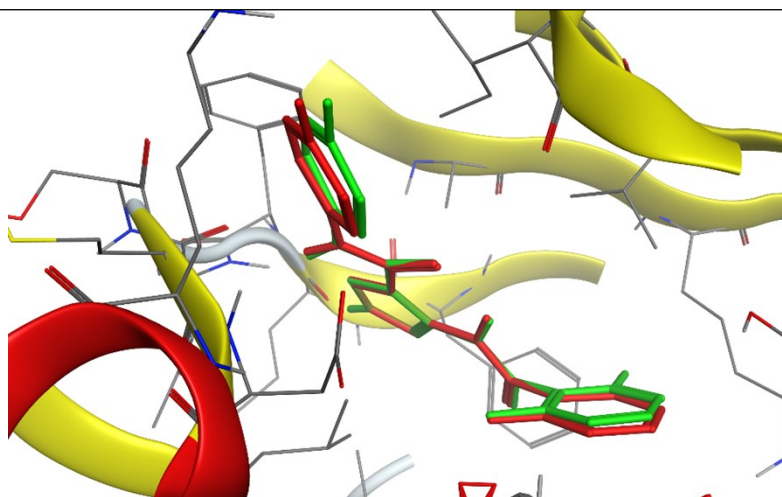


Fig. S43. 3D representation of the superimposition of the co-crystallized (red) and the redocking pose (green) of the NC(=O)Nc1cc(C(=O)Nc2cc(F)c(F)cc2)nn1 ligand in target protein (pdb ID: 4Y72).

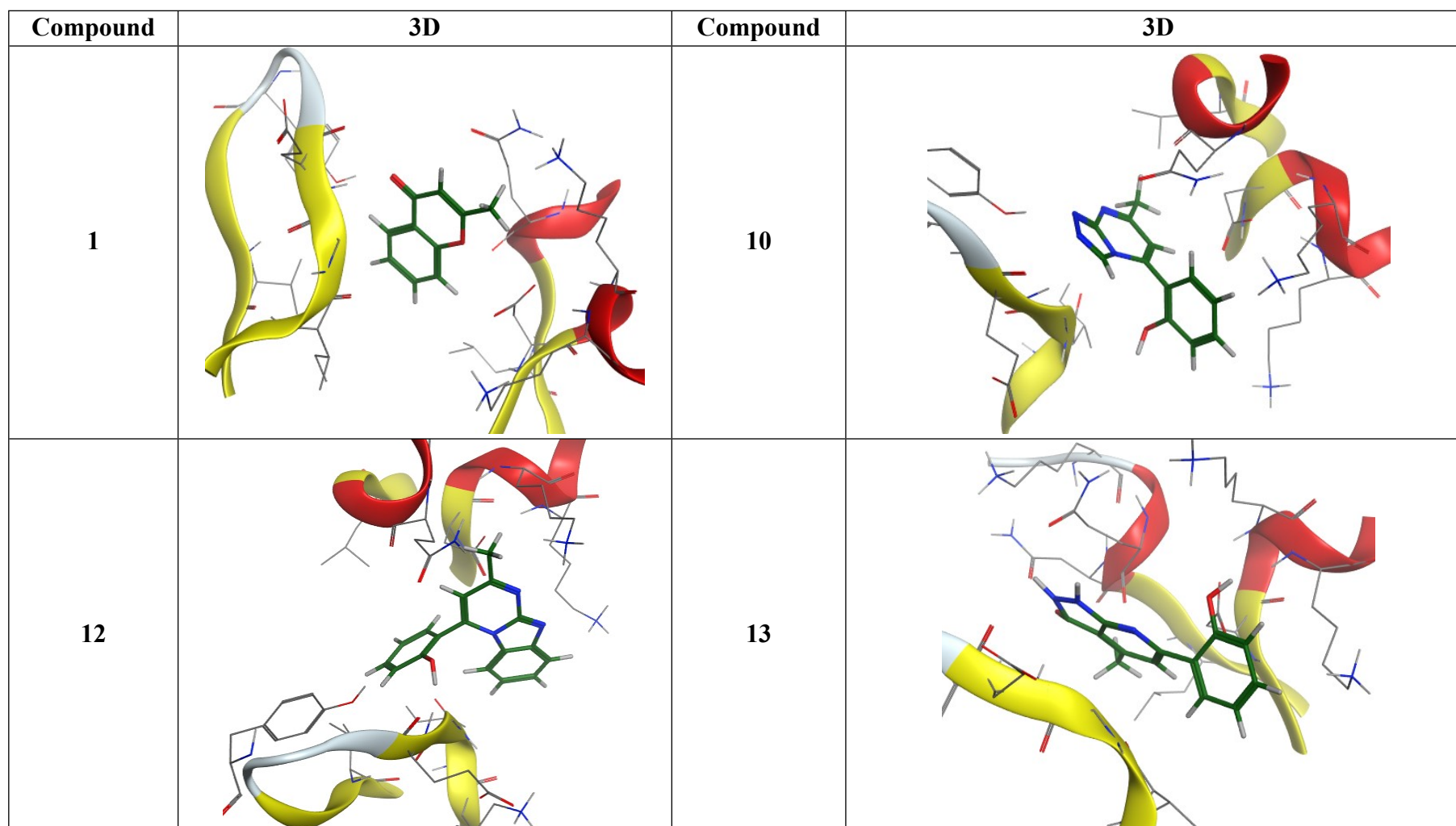


Fig. S44. 3D representation of the hydrogen bonding between the studied compounds (**1**, **10**, **12** and **13**) with the amino acids residues of the target protein (PDB ID: 4Y72).

Table S1. Various interactions between the studied compounds **1**, **10**, **12** and **13** with the amino acids residues of the target protein (pdb ID: 4Y72).

Compound 1 (Binding energy -7.6 kcal/mol)			Compound 10 (Binding energy -7.80 kcal/mol)		
Receptor	Distance	Interaction Type	Receptor	Distance	Interaction Type
GLU12	2.07	H-Bond	GLU12	2.94	H-Bond
LYS88	4.31	Alkyl	ASP86	4.60	Pi-Anion
			GLU12	2.70	Pi-Donor
Compound 12 (Binding energy -7.60 kcal/mol)			Compound 13 (Binding energy -8.10 kcal/mol)		
ASP86	3.45	Pi-Anion	GLN132	2.20	H-Bond
ASP86	3.71	Pi-Anion	ASP86	4.21	Pi-Anion
ASP86	4.43	Pi-Anion	ASP86	4.93	Pi-Anion
GLY11	3.83	Pi-Sigma	LYS89	4.98	Pi-Alkyl
ILE10	2.82	Pi-Lone Pair			
LYS88	5.34	Pi-Alkyl			
LYS89	5.38	Pi-Alkyl			
VAL18	5.10	Pi-Alkyl			
Co-crystallized ligand (Binding energy -10.60 kcal/mol)					
LEU83	2.1229	H-Bond			
GLU81	1.66193	H-Bond			
ASP146	4.75511	Pi-Anion			
VAL18	3.99529	Pi-Sigma			
LEU135	3.50724	Pi-Sigma			
ILE10	4.07235	Pi-Alkyl			
ALA31	3.38275	Pi-Alkyl			
VAL64	5.26978	Pi-Alkyl			
LEU83	5.45921	Pi-Alkyl			
ALA145	5.06422	Pi-Alkyl			