

Supplementary Information

**Fluorescent-state switching of a 10-hydroxybezo[*h*]quinoline skeleton
through electronic nature of substituents**

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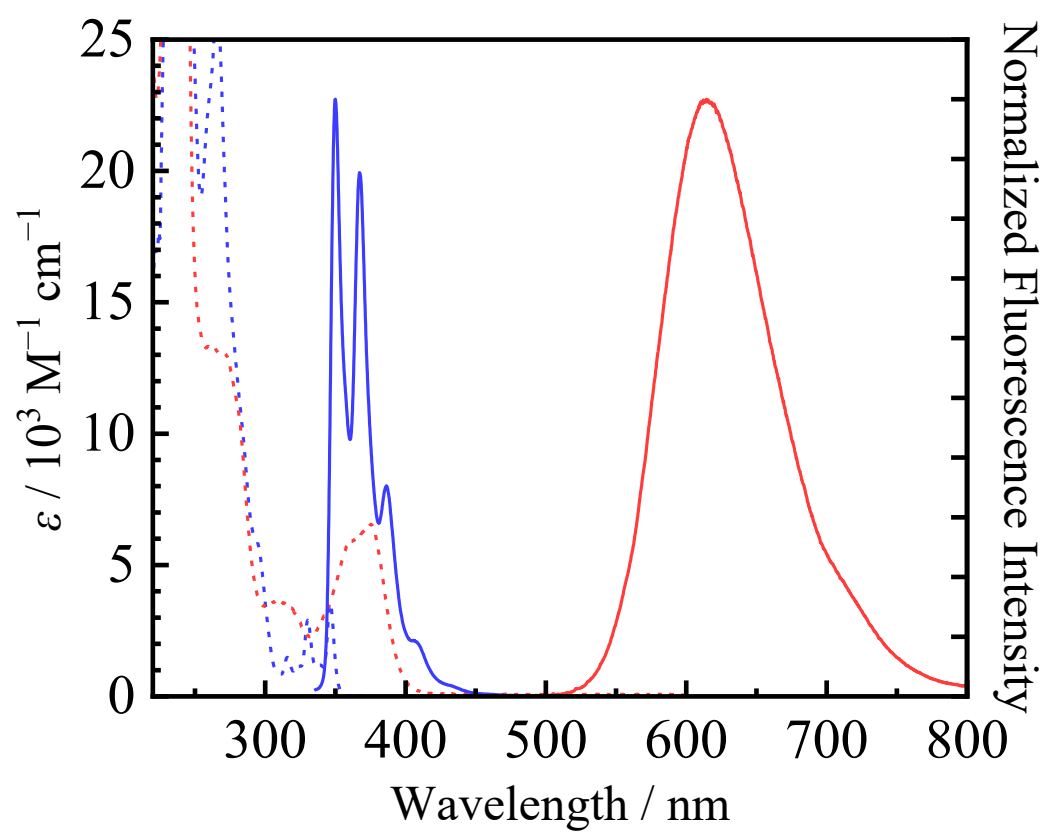


Fig. S1 Absorption (broken curves) and fluorescence spectra (solid curves) of HBq (red) and benzo[h]quinoline (blue) in THF.

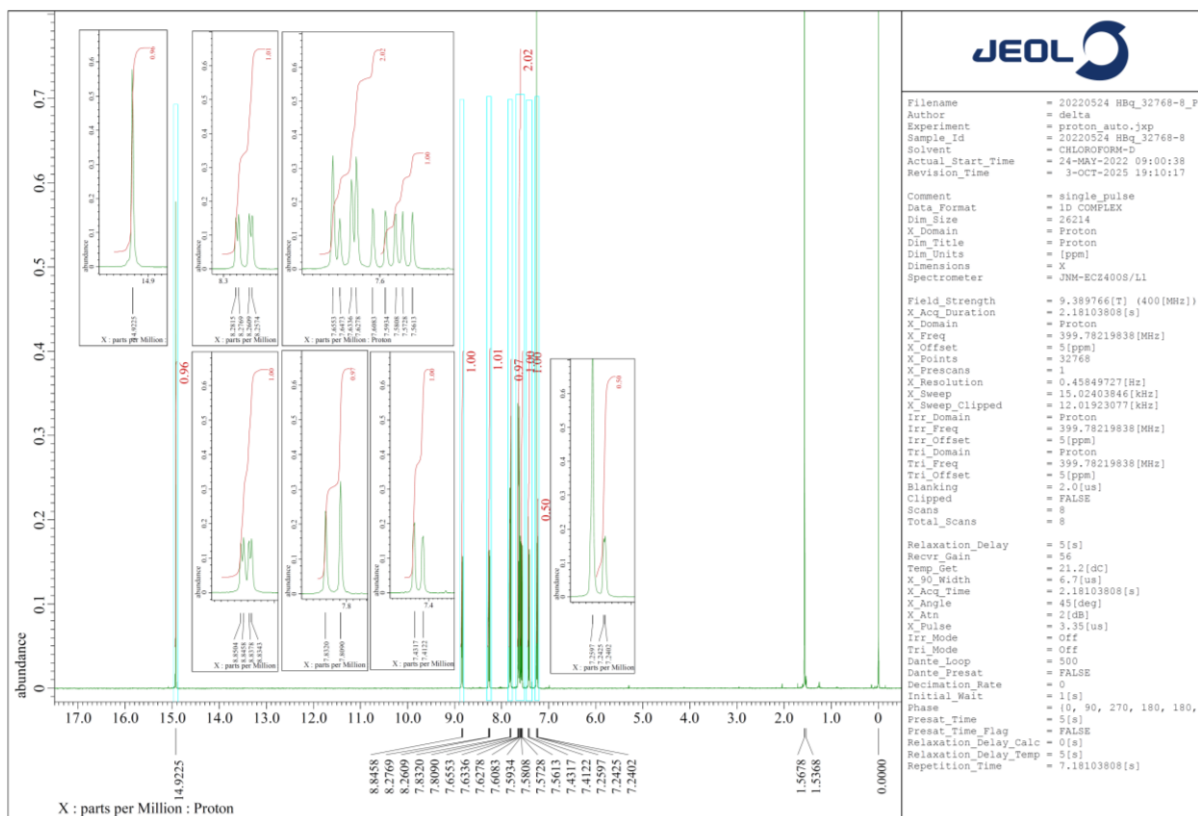


Fig. S2 ^1H -NMR spectrum of HBq in CDCl_3 .

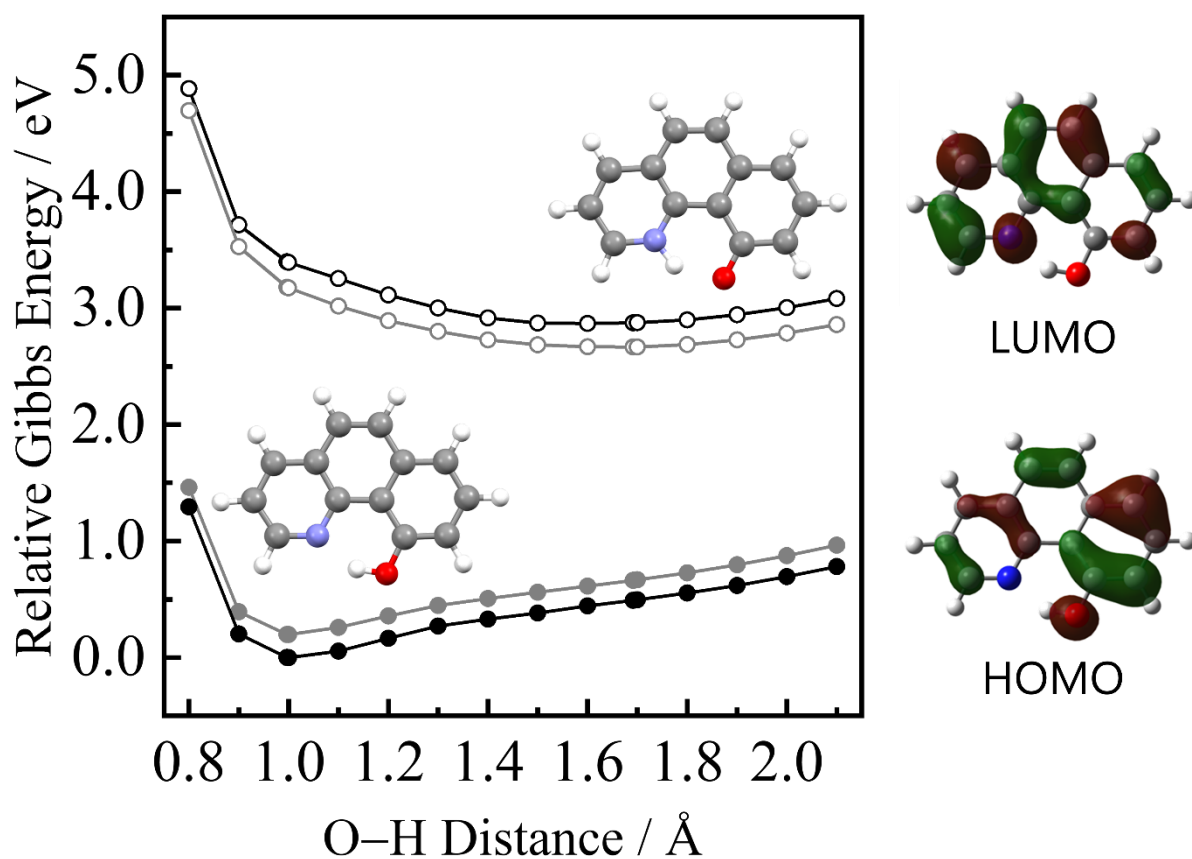


Fig. S3 Results of (TD-)DFT calculations for HBq in vacuum (B3LYP/6-31G(d,p) level): dependences of the O-H distances on the relative Gibbs energies of ground (close circles) and S_1 states (open circles) in the ground- (black) and S_1 -state-optimized geometries (gray) with Kohn-Sham orbitals in the ground-state-optimized geometries.

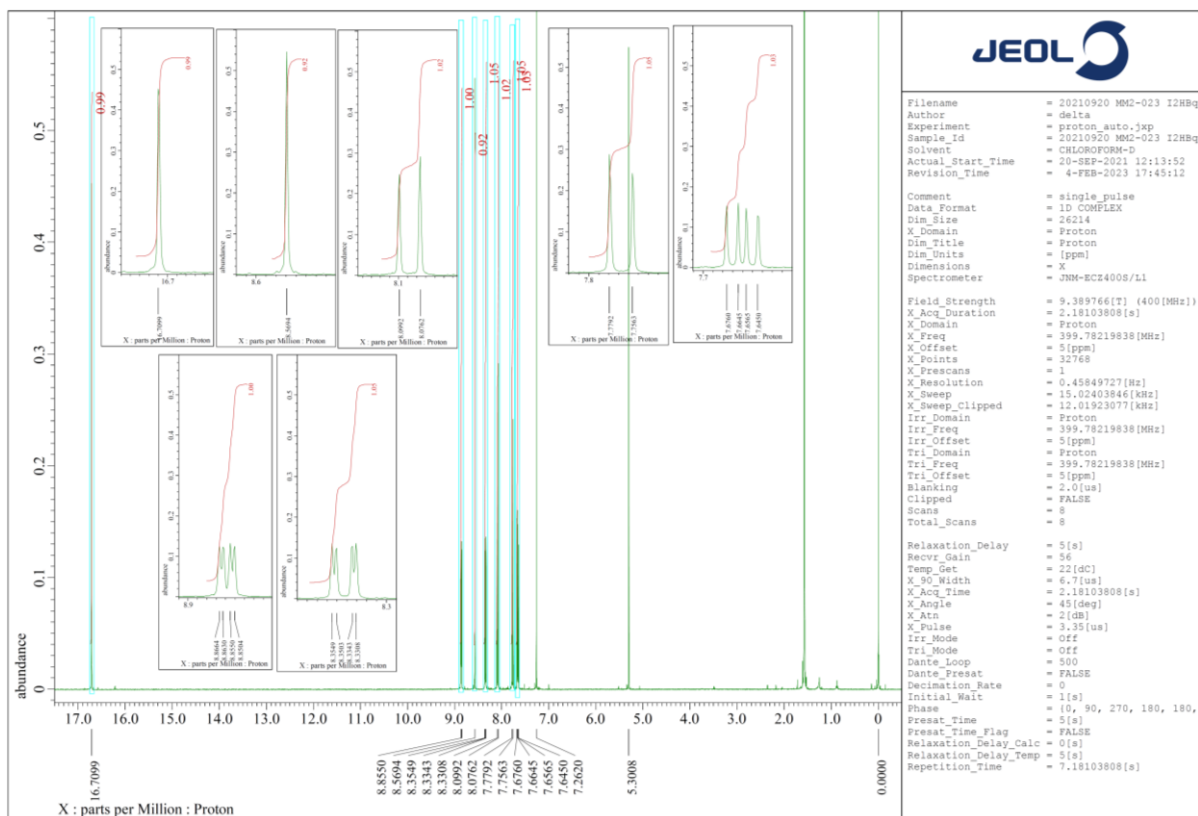


Fig. S4 ^1H -NMR spectrum of HBql_2 in CDCl_3 .

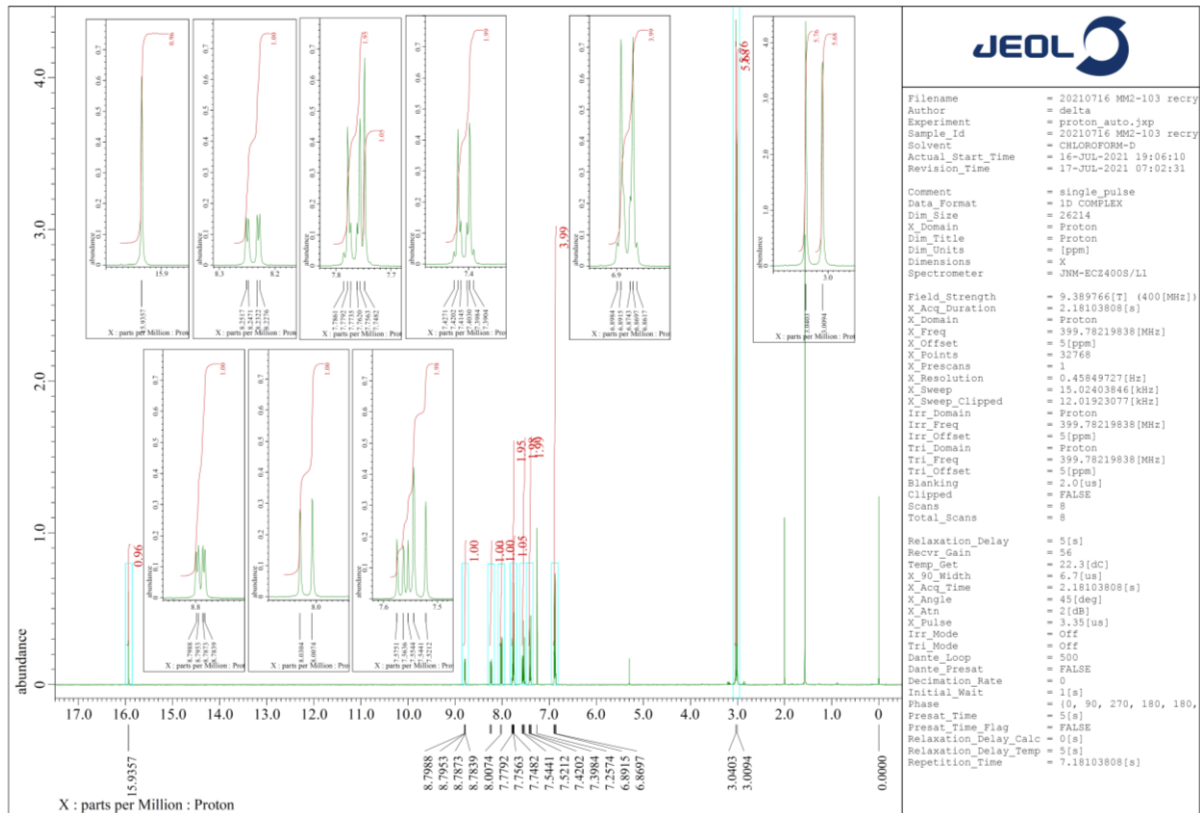


Fig. S5 ^1H -NMR spectrum of **1NMe₂** in CDCl_3 .

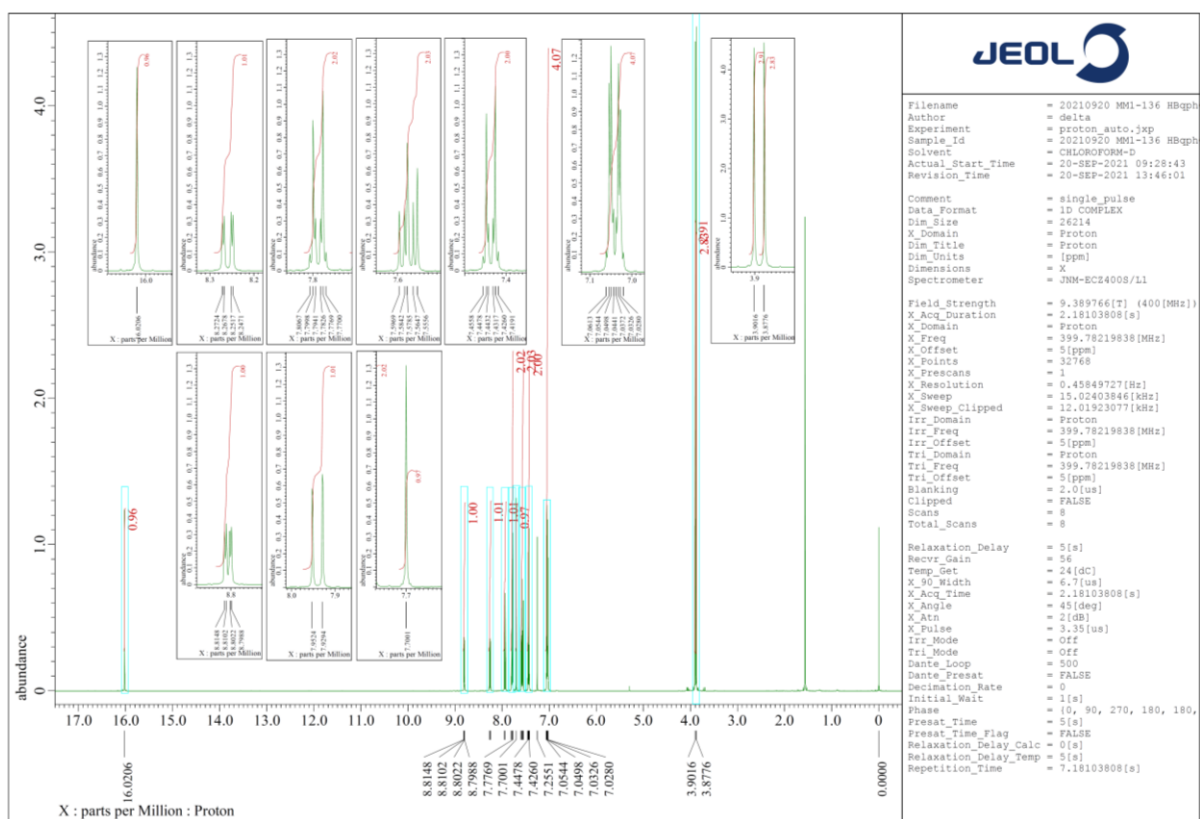


Fig. S6 ^1H -NMR spectrum of **1OMe** in CDCl_3 .

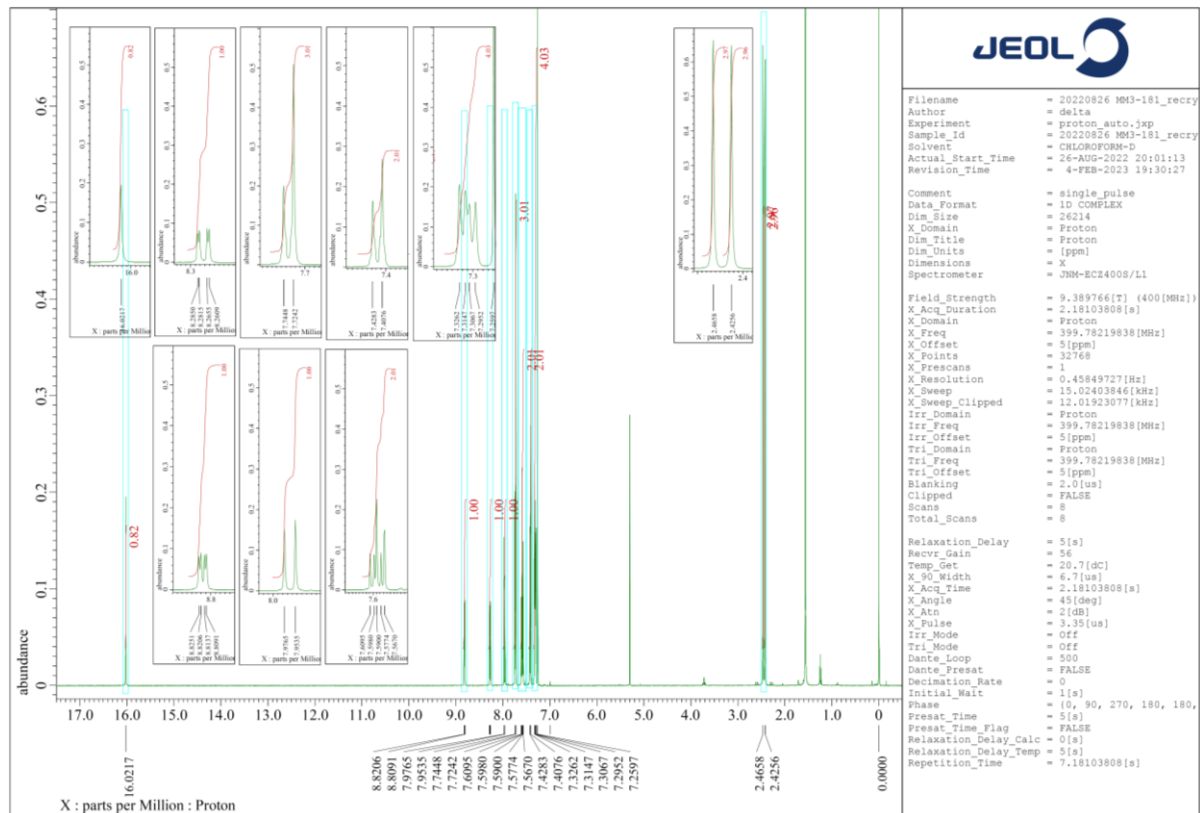


Fig. S7 ^1H -NMR spectrum of **1Me** in CDCl_3 .

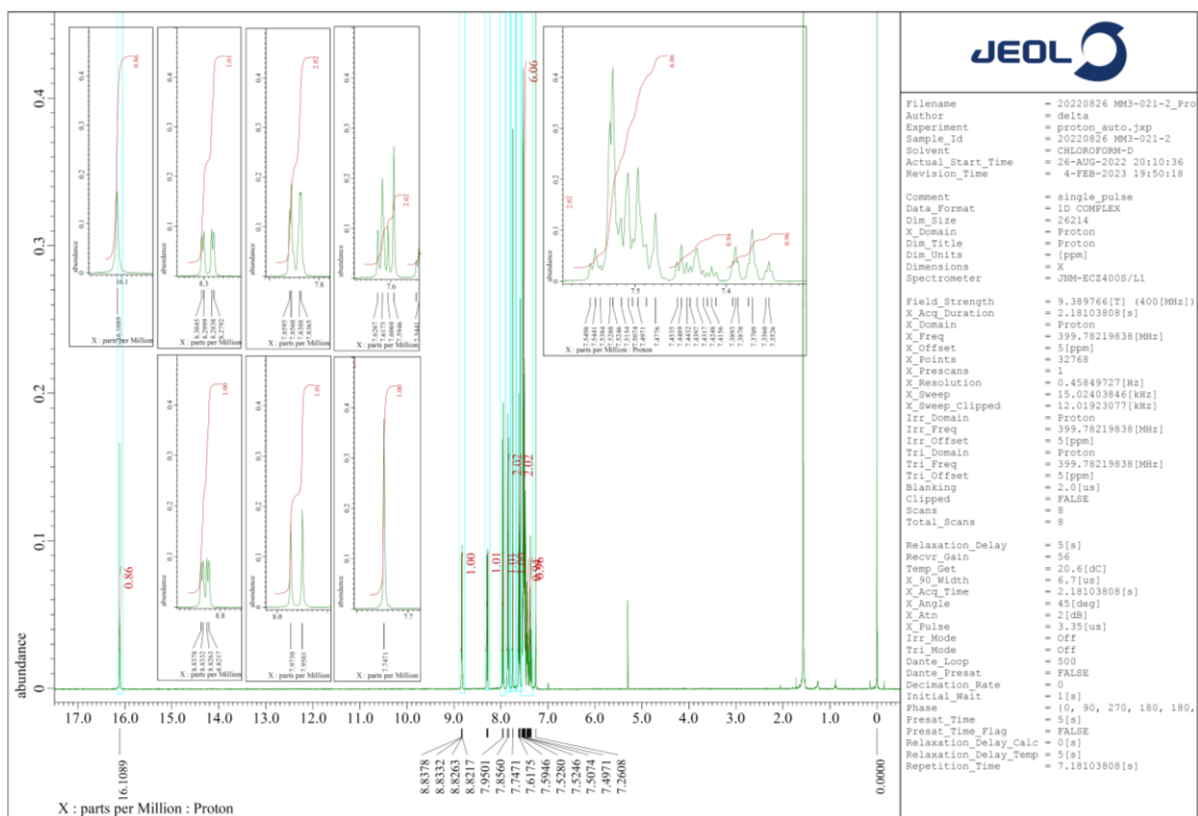


Fig. S8 ^1H -NMR spectrum of **1H** in CDCl_3 .

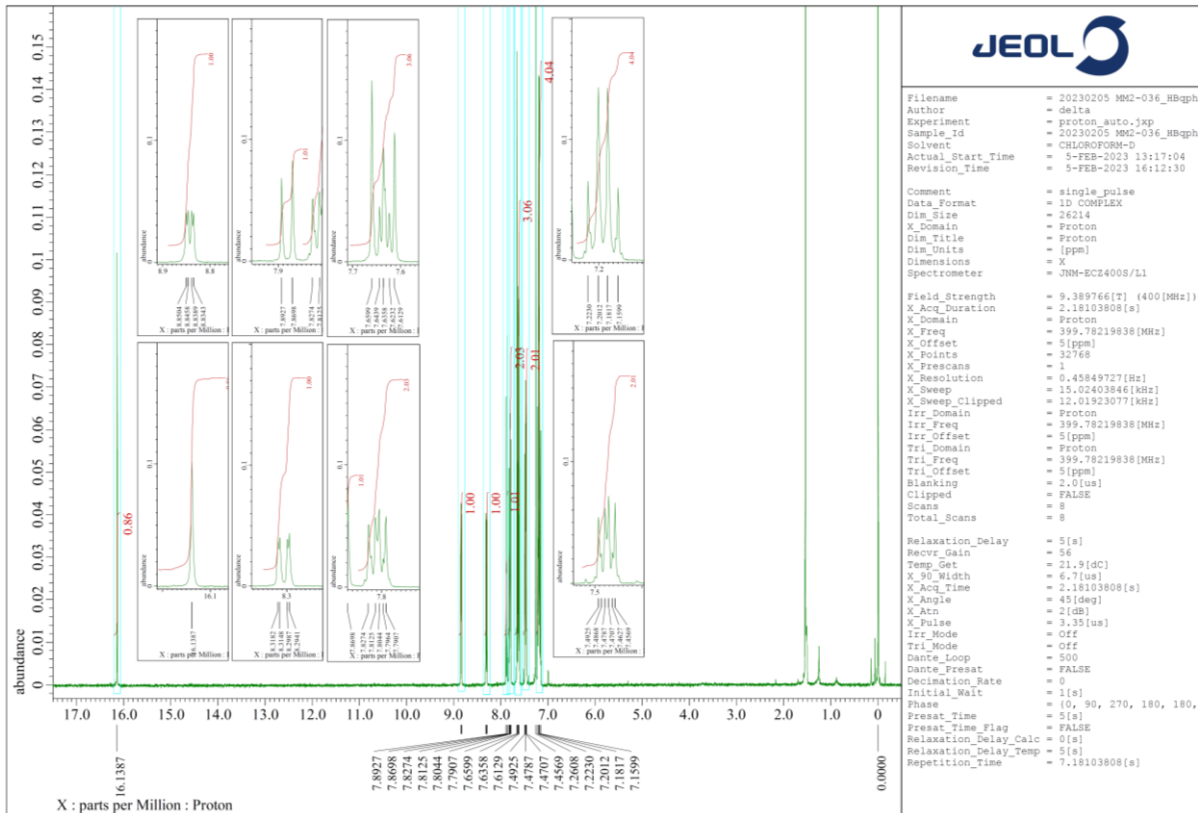


Fig. S9 ^1H -NMR spectrum of **1F** in CDCl_3 .

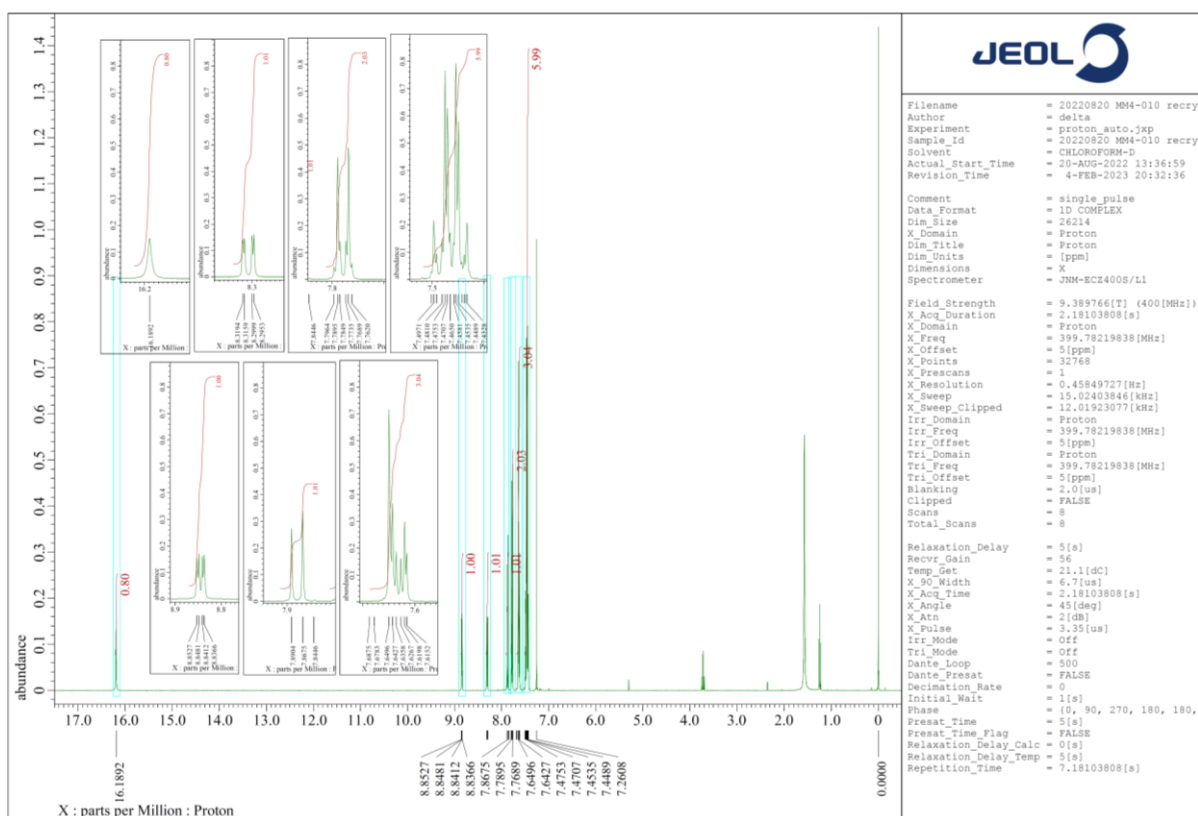


Fig. S10 ^1H -NMR spectrum of **1Cl** in CDCl_3 .

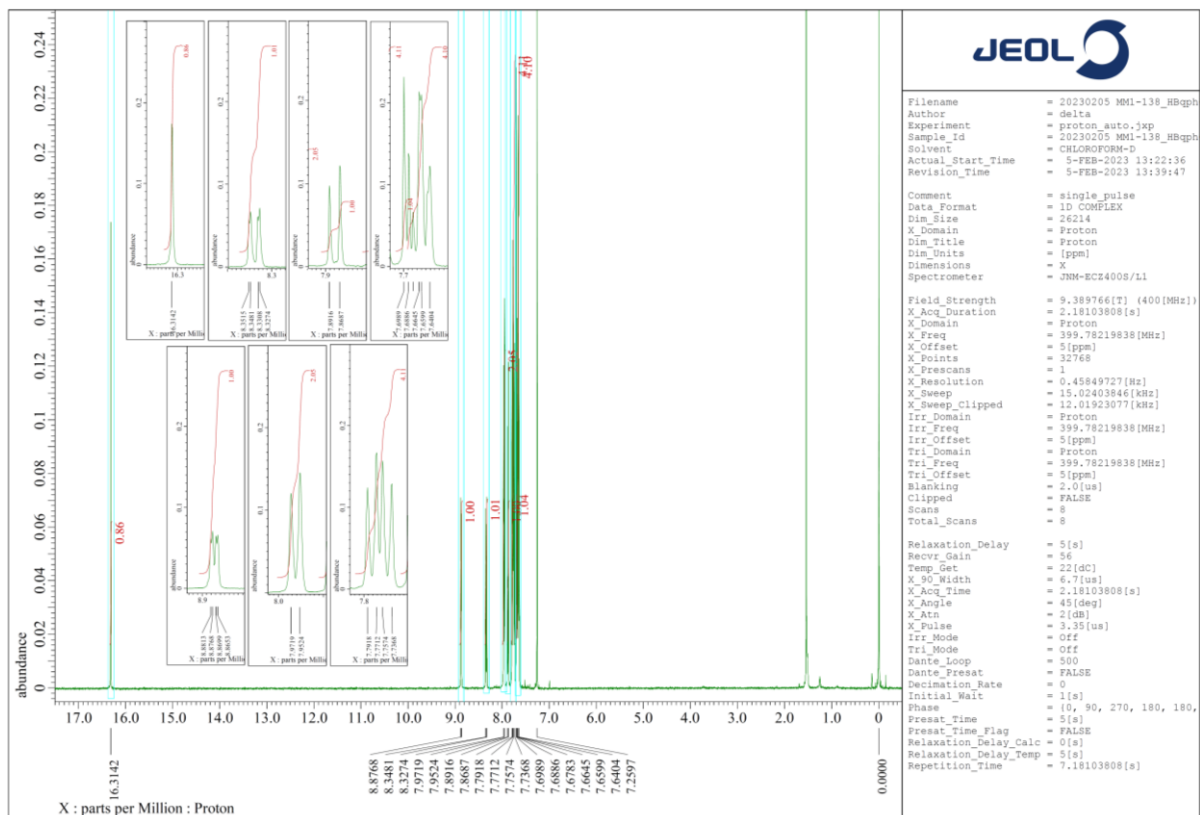


Fig. S11 ^1H -NMR spectrum of **1CF₃** in CDCl_3 .

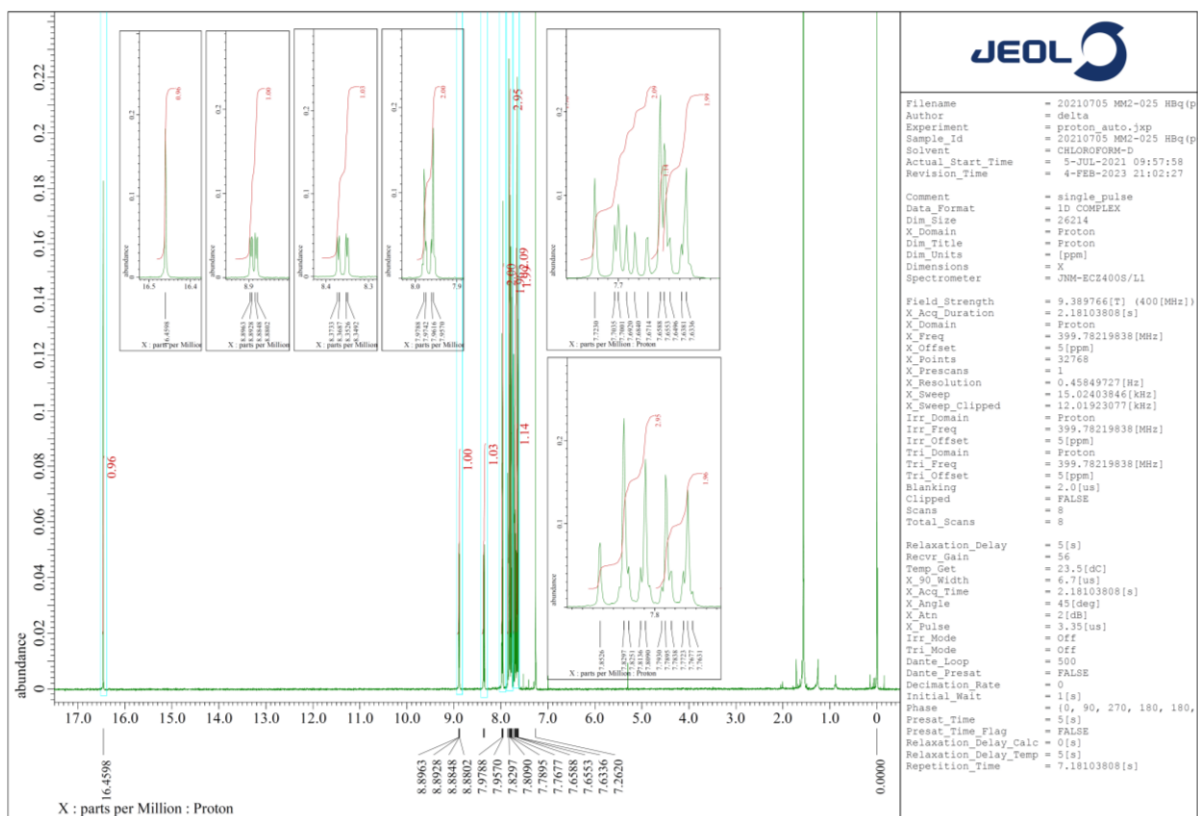


Fig. S12 ^1H -NMR spectrum of 1CN in CDCl_3 .

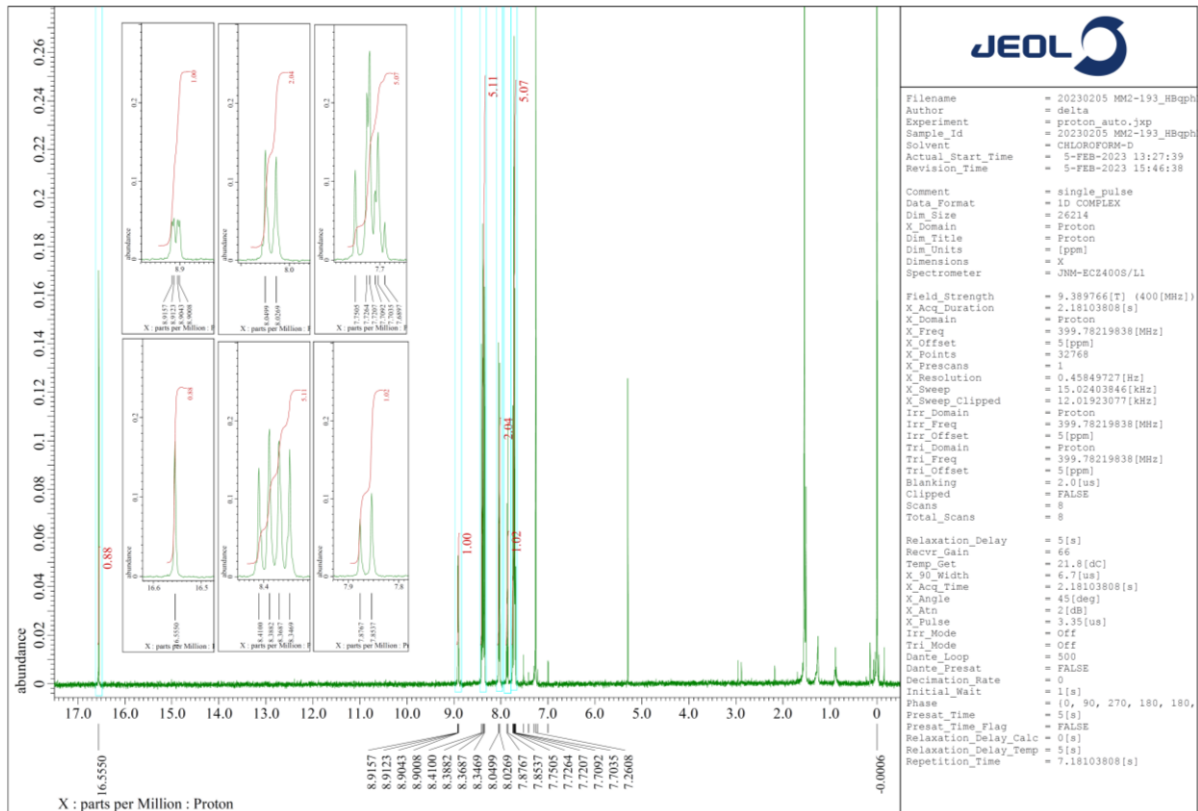


Fig. S13 ^1H -NMR spectrum of 1NO_2 in CDCl_3 .

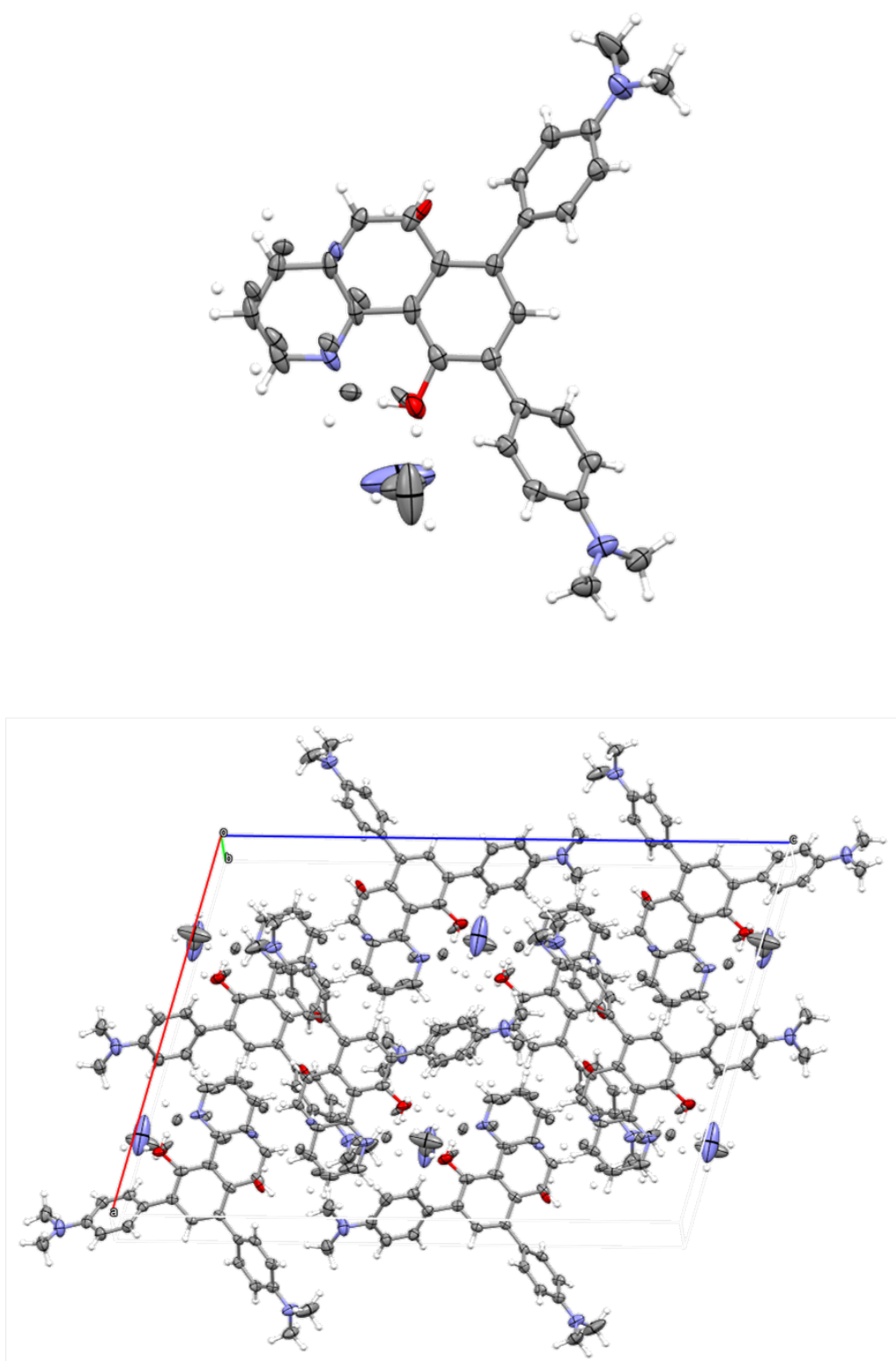


Fig. S16 Crystal structure of 1NMe₂.

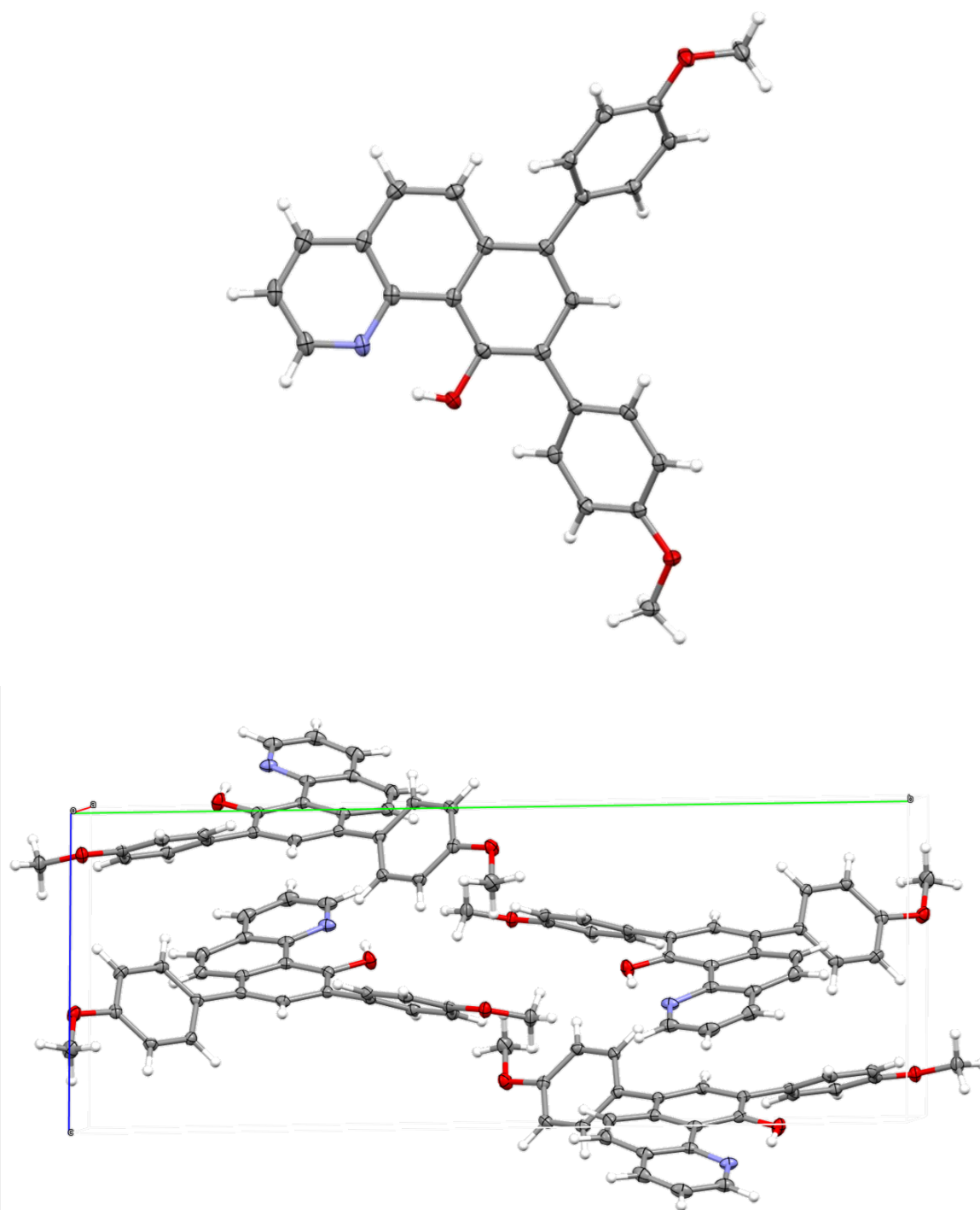


Fig. S17 Crystal structure of 1OMe.

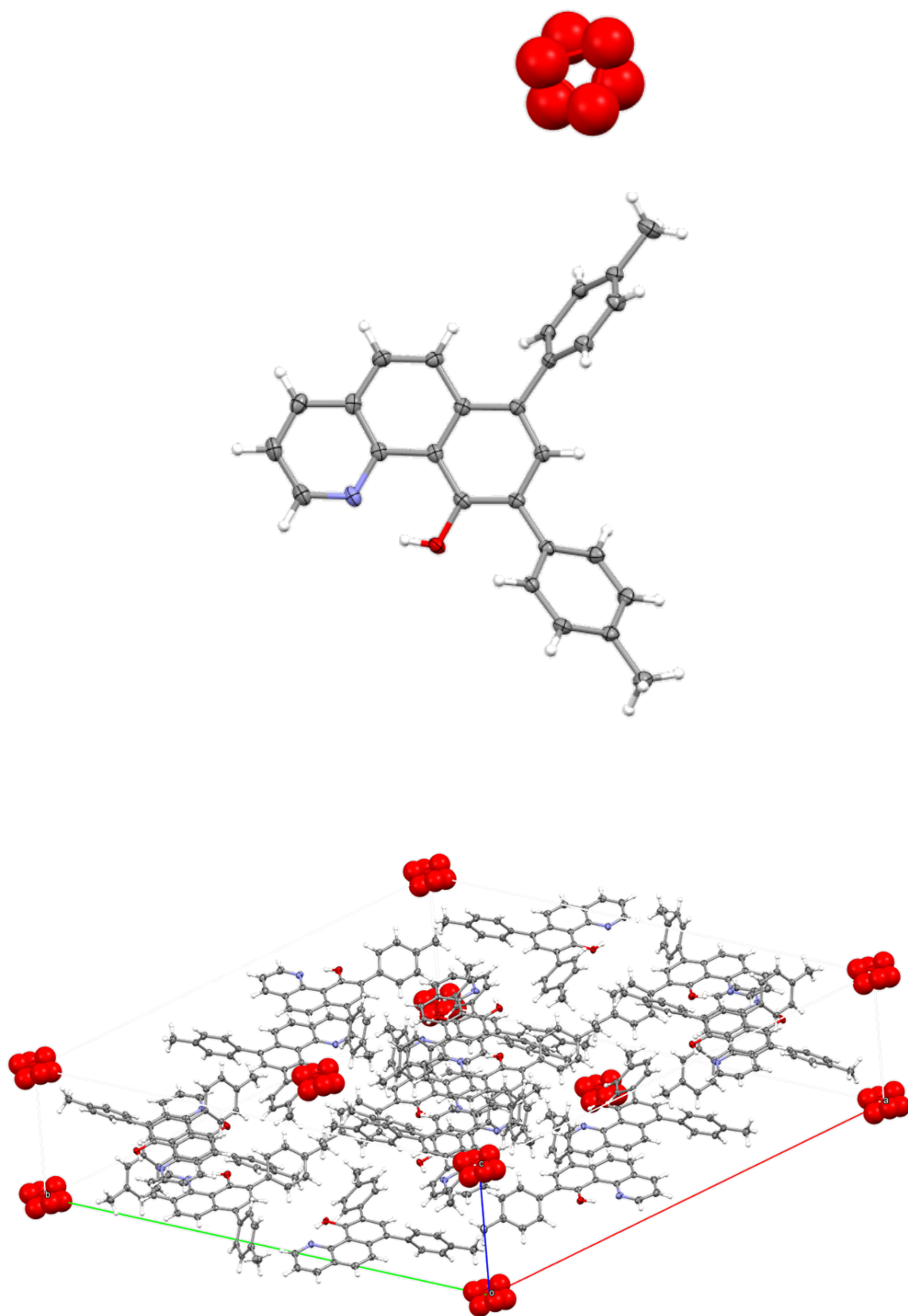


Fig. S18 Crystal structure of 1Me.

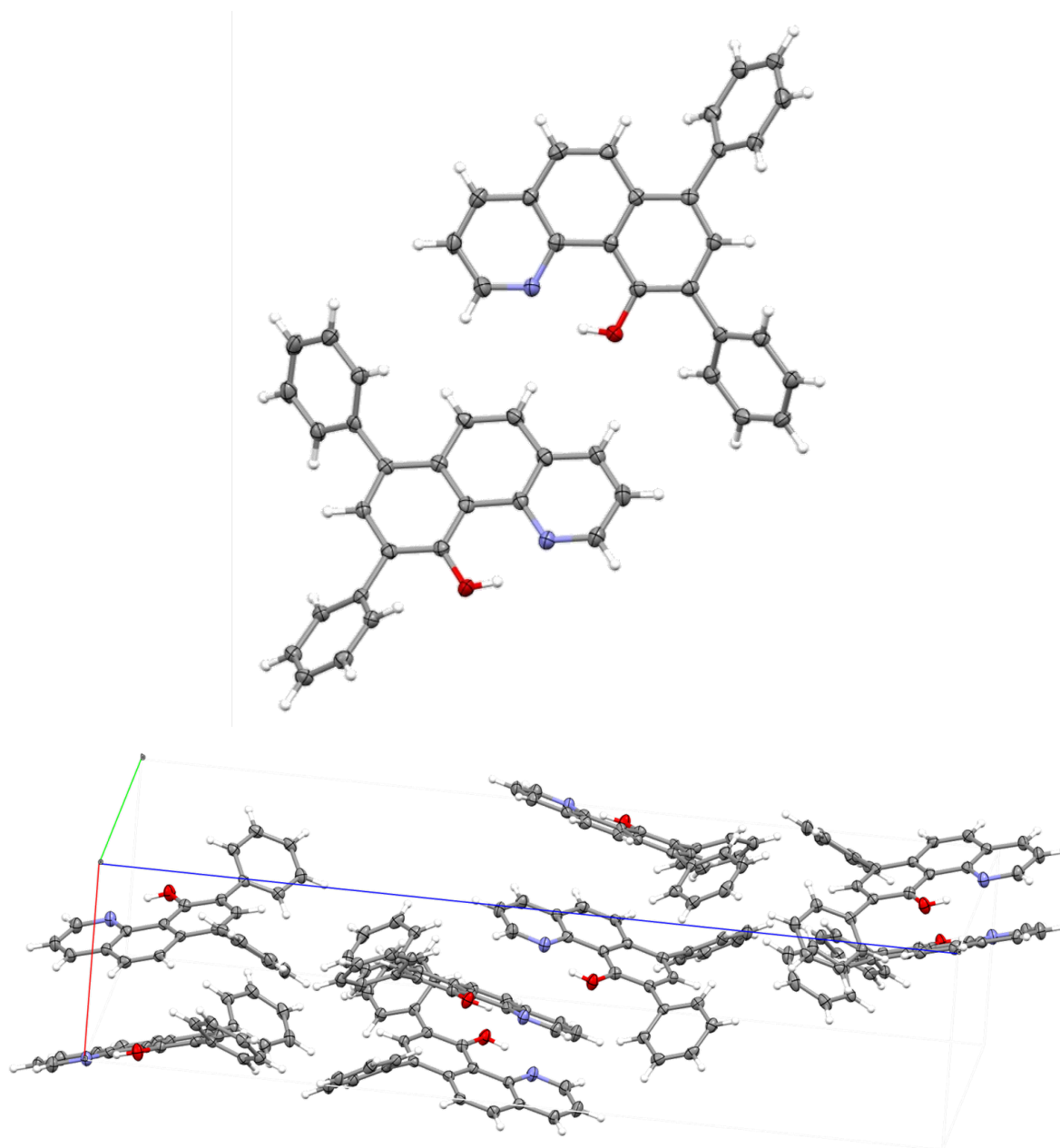


Fig. S19 Crystal structure of 1H.

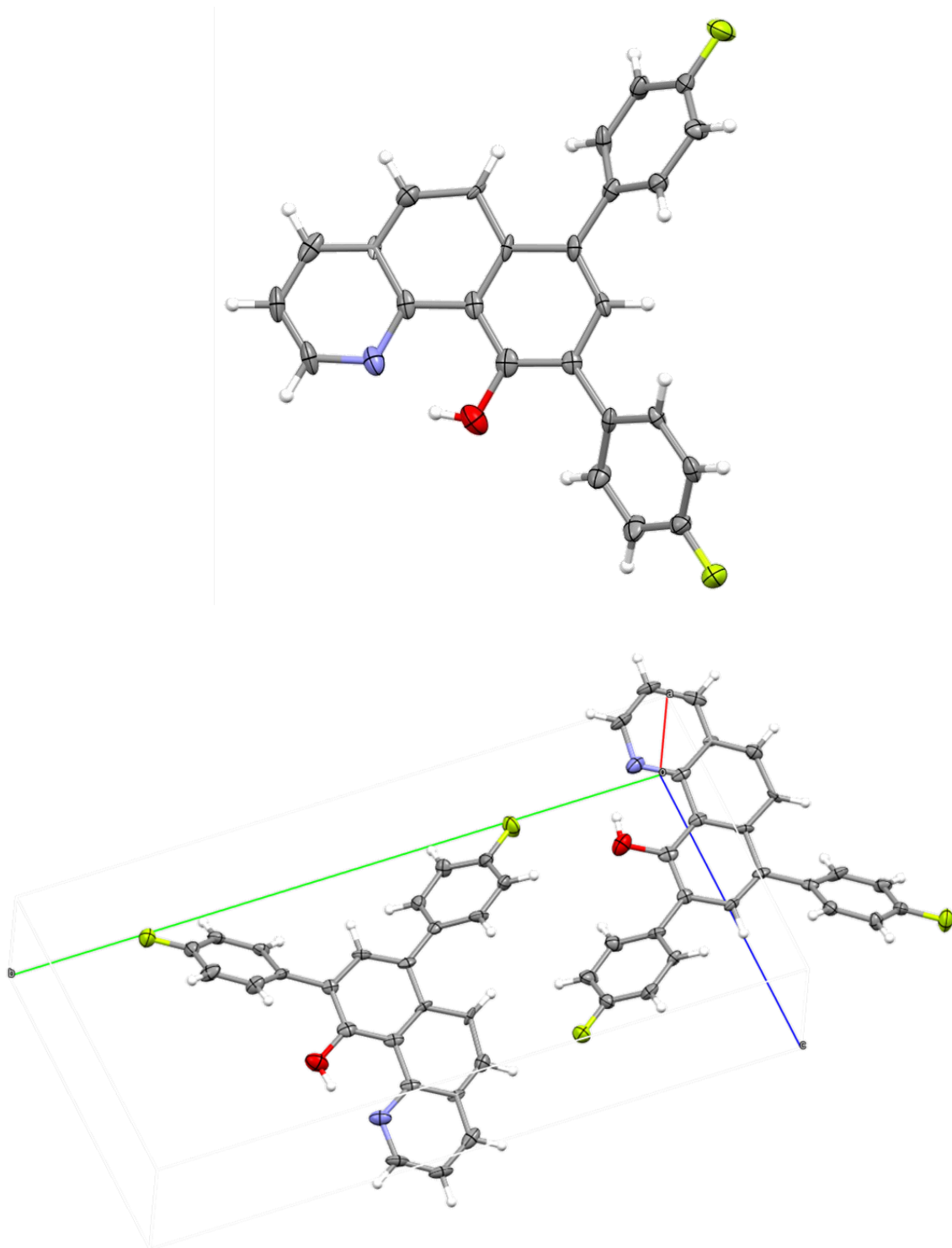


Fig. S20 Crystal structure of 1F.

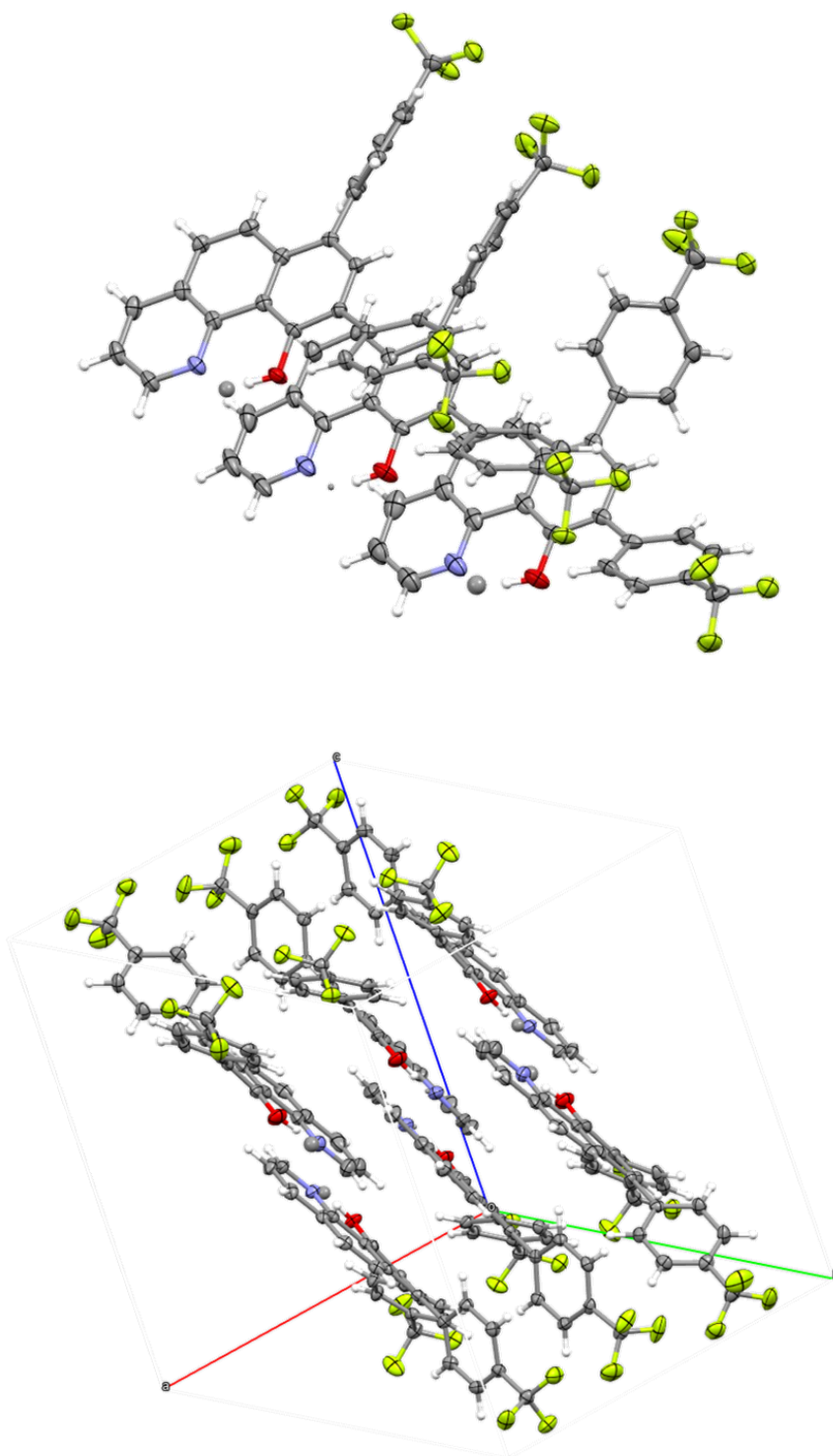


Fig. S21 Crystal structure of 1CF₃.

Table S1 Crystallographic parameters of **1NMe₂**, **1OMe** and **1Me**.

	1NMe₂	1OMe	1Me
Empirical formula	C _{29.5} H _{27.75} N _{3.25} O	C ₂₇ H ₂₁ NO ₃	C ₂₇ H ₂₁ NO _{1.17}
Formula weight	443.80	407.45	378.11
Crystal system	monoclinic	monoclinic	trigonal
Space group	<i>I</i> 2/ <i>a</i>	<i>P</i> 2 ₁ / <i>c</i>	<i>R</i> $\bar{3}$
<i>a</i> / Å	18.9713(7)	8.7730(4)	31.2623(8)
<i>b</i> / Å	9.4358(3)	24.1239(8)	31.2623(8)
<i>c</i> / Å	27.1606(11)	9.6591(4)	10.4233(3)
α / deg	90	90	90
β / deg	105.179(4)	108.576(4)	90
γ / deg	90	90	120
Volume / Å ³	4692.4(3)	1937.74(14)	8822.2(5)
<i>Z</i>	8	4	18
<i>Z'</i>	1	1	1
ρ_{calc} / g cm ⁻³	1.256	1.397	1.281
μ / mm ⁻¹	0.077	0.091	0.078
<i>F</i> (000)	1884.0	856.0	3588.0
Crystal size / mm ³	0.552 × 0.27 × 0.213	0.964 × 0.147 × 0.102	0.62 × 0.17 × 0.129
2 θ range / deg	4.714–52.74	4.758–52.738	4.188–52.73
Index ranges	–22 ≤ <i>h</i> ≤ 23 –11 ≤ <i>k</i> ≤ 8 –33 ≤ <i>l</i> ≤ 33	–9 ≤ <i>h</i> ≤ 10 –30 ≤ <i>k</i> ≤ 30 11 ≤ <i>l</i> ≤ 12	–29 ≤ <i>h</i> ≤ 39 –38 ≤ <i>k</i> ≤ 39 –12 ≤ <i>l</i> ≤ 12
Reflections collected	13150	11936	12446
Independent reflections	4765	3954	3995
Goodness-of-fit on <i>F</i> ₂	1.026	1.030	1.018
Final <i>R</i> indexes ^a (<i>I</i> ≥ 2 σ (<i>I</i>))	<i>R</i> ₁ = 0.0466 w <i>R</i> ₂ = 0.1129	<i>R</i> ₁ = 0.0366 w <i>R</i> ₂ = 0.0944	<i>R</i> ₁ = 0.0471 w <i>R</i> ₂ = 0.1244

$$a) R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|, wR_2 = \left\{ \sum w (F_o^2 - F_c^2)^2 / \sum w (F_o^2)^2 \right\}^{1/2}.$$

Table S2 Crystallographic parameters of **1H**, **1F** and **1CF₃**.

	1H	1F	1CF₃
Empirical formula	C ₂₅ H ₁₇ NO	C ₂₅ H ₁₅ F ₂ NO	C ₂₇ H ₁₅ F ₆ NO
Formula weight	347.39	383.38	483.40
Crystal system	orthorhombic	monoclinic	triclinic
Space group	<i>Pna</i> 2 ₁	<i>P</i> 2 ₁	<i>P</i> $\bar{1}$
<i>a</i> / Å	7.7291(5)	3.8336(2)	14.0330(3)
<i>b</i> / Å	13.5968(9)	20.9976(9)	14.9961(3)
<i>c</i> / Å	32.5171(19)	10.5840(4)	15.6184(3)
α / deg	90	90	106.652(2)
β / deg	90	92.115(4)	94.045(2)
γ / deg	90	90	99.828(2)
Volume / Å ³	3417.3(4)	851.39(7)	3077.75(11)
<i>Z</i>	8	2	6
<i>Z'</i>	2	1	3
ρ_{calc} / g cm ⁻³	1.350	1.495	1.565
μ / mm ⁻¹	0.082	0.106	0.133
<i>F</i> (000)	1456.0	396.0	1476.0
Crystal size / mm ³	0.954 × 0.107 × 0.057	0.887 × 0.158 × 0.142	0.86 × 0.085 × 0.067
2 θ range / deg	4.806–52.73	4.312–52.736	3.708–52.744
Index ranges	–9 ≤ <i>h</i> ≤ 6 –16 ≤ <i>k</i> ≤ 15 –31 ≤ <i>l</i> ≤ 40	–3 ≤ <i>h</i> ≤ 4 –25 ≤ <i>k</i> ≤ 26 –13 ≤ <i>l</i> ≤ 12	–17 ≤ <i>h</i> ≤ 17 –18 ≤ <i>k</i> ≤ 18 –19 ≤ <i>l</i> ≤ 19
Reflections collected	14274	4865	37387
Independent reflections	6378	3141	12595
Goodness-of-fit on <i>F</i> ₂	1.049	1.082	1.022
Final <i>R</i> indexes ^a (<i>I</i> ≥ 2 σ (<i>I</i>))	<i>R</i> ₁ = 0.0399 w <i>R</i> ₂ = 0.1025	<i>R</i> ₁ = 0.0544 w <i>R</i> ₂ = 0.1434	<i>R</i> ₁ = 0.0694 w <i>R</i> ₂ = 0.1856

$$a) R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|, wR_2 = \left\{ \sum w (F_o^2 - F_c^2)^2 / \sum w (F_o^2)^2 \right\}^{1/2}.$$

Table S3 Selected bond lengths and dihedral angles of **1R** in crystalline phases.

Derivative	C ₍₁₀₎ –O	C ₍₈₎ –C ₍₇₎ –C _(1') –C _(2')	C ₍₈₎ –C ₍₉₎ –C _(1'') –C _(2'')
1NMe₂	1.404(3) Å	–47.9(2)°	–46.6(2)°
1OMe	1.354(2) Å	+42.1(2)°	+29.4(2)°
1Me	1.355(2) Å	–57.5(2)°	–34.9(2)°
1H	1.351(3) Å	–51.7(4)°	+43.2(4)°
1F	1.359(3) Å	–50.9(4)°	+41.7(4)°
	1.409(7) Å	–40.5(6)°	+43.6(6)°
	1.359(3) Å	–44.0(3)°	–23.2(3)°
1CF₃	1.363(4) Å	–49.7(3)°	–39.5(4)°
	1.368(4) Å	+49.4(4)°	–28.9(4)°

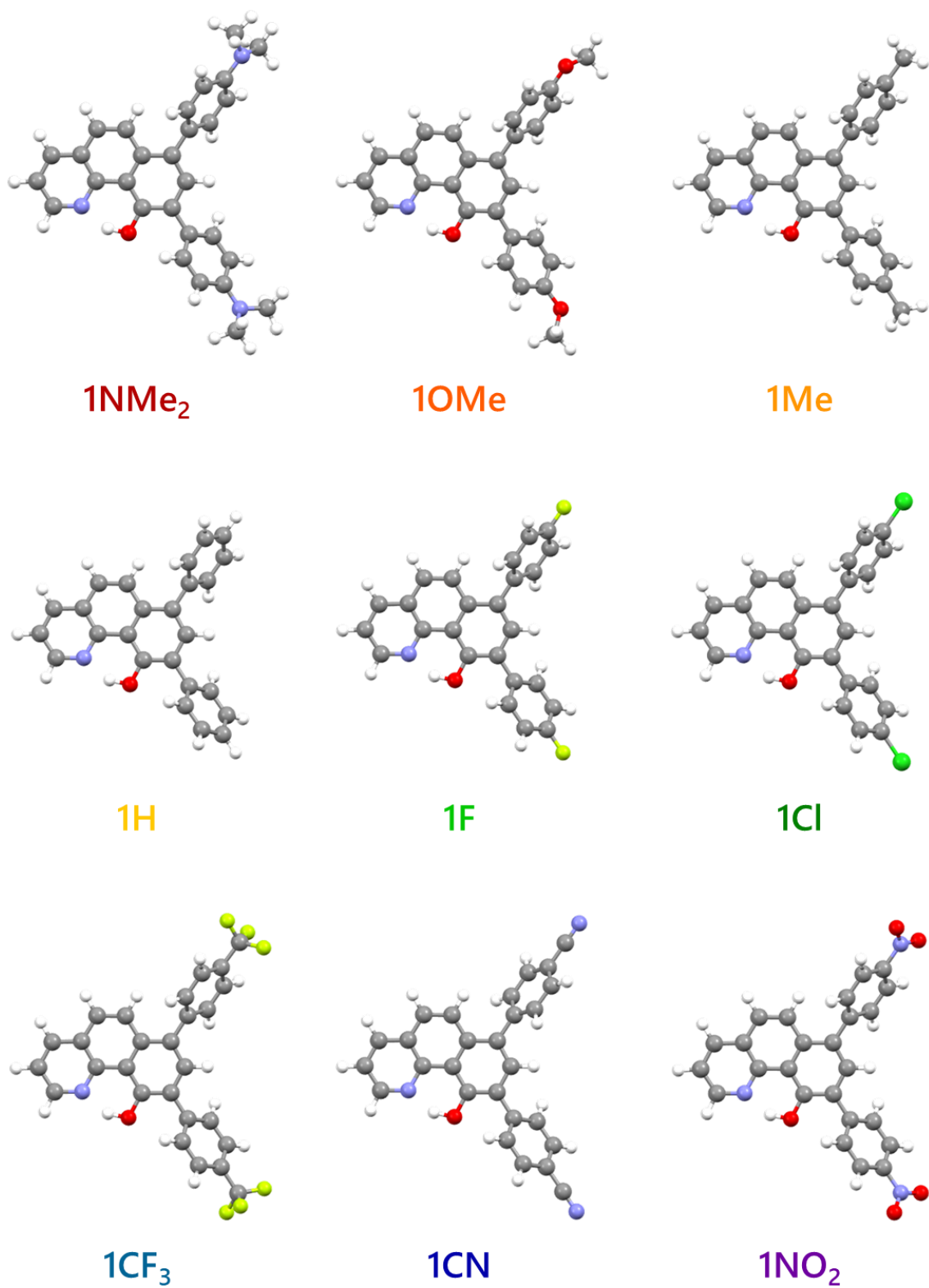


Fig. S22 Optimized geometries of **1R** in the ground states in dichloromethane.

Table S4 Selected lengths and dihedral angles of optimized **1R** in dichloromethane.

Derivative	State	C ₍₁₀₎ –O	O–H (O···H)	H–N ₍₁₎ (H···N ₍₁₎)	C ₍₈₎ –C ₍₇₎ –C _(1') –C _(2') C ₍₈₎ –C ₍₉₎ –C _(1'') –C _(2'')
1NMe₂	GS	1.3508 Å	1.0025 Å	1.6331 Å	+57.06° –42.12°
	S ₁	1.2769 Å	1.6710 Å	1.0379 Å	+43.61° –27.91°
1OMe	GS	1.3501 Å	1.0035 Å	1.6322 Å	+58.58° –44.55°
	S ₁	1.2716 Å	1.7033 Å	1.0327 Å	+44.53° –29.39°
1Me	GS	1.3487 Å	1.0031 Å	1.6318 Å	+58.64° –45.01°
	S ₁	1.2702 Å	1.7145 Å	1.0313 Å	+45.07° –30.74°
1H	GS	1.3477 Å	1.0038 Å	1.6314 Å	+58.38° –46.34°
	S ₁	1.2698 Å	1.7221 Å	1.0311 Å	+45.74° –31.95°
1F	GS	1.3484 Å	1.0040 Å	1.6300 Å	+59.44° –45.28°
	S ₁	1.2694 Å	1.7229 Å	1.0304 Å	+45.74° –31.10°
1Cl	GS	1.3461 Å	1.0052 Å	1.6277 Å	+58.67° –44.79°
	S ₁	1.2678 Å	1.7311 Å	1.0286 Å	+45.49° –30.57°
1CF₃	GS	1.3445 Å	1.0062 Å	1.6242 Å	+57.55° –44.64°
	S ₁	1.2660 Å	1.7388 Å	1.0278 Å	+46.02° –31.19°
1CN	GS	1.3442 Å	1.0067 Å	1.6211 Å	–55.68° –43.10°
	S ₁	1.2635 Å	1.7476 Å	1.0268 Å	–44.87° –28.30°
1NO₂	GS	1.3417 Å	1.0088 Å	1.6140 Å	+54.42° –42.41°
	S ₁	1.3039 Å	1.0711 Å	1.4718 Å	–89.76° –38.33°

Table S5 Calculated excited states of **1NMe₂** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO115 → MO116	2.7829 eV (445.52 nm)	0.2988
S ₂	MO114 → MO116	3.0257 eV (409.77 nm)	0.0866
S ₃	MO115 → MO117	3.2209 eV (384.94 nm)	0.0592
S ₄	MO114 → MO117	3.4918 eV (355.07 nm)	0.0878
S ₅	MO113 → MO116 (86%) MO114 → MO117 (14%)	3.6680 eV (338.02 nm)	0.1376
S ₆	MO115 → MO118	3.9205 eV (316.25 nm)	0.3222
S ₇	MO112 → MO116 (28%) MO113 → MO117 (54%) MO115 → MO119 (18%)	4.0700 eV (304.63 nm)	0.0130
S ₈	MO114 → MO118 (14%) MO114 → MO120 (12%) MO115 → MO119 (50%) MO115 → MO120 (24%)	4.1225 eV (300.75 nm)	0.0075
S ₉	MO114 → MO119 (19%) MO114 → MO120 (18%) MO115 → MO120 (63%)	4.1450 eV (299.12 nm)	0.0473
S ₁₀	MO114 → MO118 (66%) MO114 → MO119 (14%) MO115 → MO120 (20%)	4.1787 eV (296.71 nm)	0.1994
S ₁₁	MO112 → MO116	4.3140 eV (287.40 nm)	0.1227
S ₁₂	MO112 → MO116 (16%) MO114 → MO119 (29%) MO115 → MO119 (22%) MO115 → MO121 (33%)	4.3784 eV (283.17 nm)	0.1980
S ₁₃	MO115 → MO121	4.3983 eV (281.89 nm)	0.2380
S ₁₄	MO111 → MO116 (21%) MO115 → MO122 (79%)	4.4950 eV (275.83 nm)	0.0181
S ₁₅	MO114 → MO120 (70%) MO115 → MO122 (30%)	4.5269 eV (273.88 nm)	0.0112
S ₁₆	MO108 → MO116 (31%) MO111 → MO116 (37%) MO114 → MO120 (32%)	4.5571 eV (272.07 nm)	0.0228
S ₁₇	MO114 → MO120 (11%) MO114 → MO121 (27%) MO115 → MO121 (11%) MO115 → MO122 (12%) MO115 → MO123 (39%)	4.5839 eV (270.48 nm)	0.0376
S ₁₈	MO110 → MO116 (38%) MO111 → MO116 (15%) MO112 → MO117 (25%) MO113 → MO118 (9%) MO114 → MO121 (13%)	4.6058 eV (269.19 nm)	0.1380
S ₁₉	MO109 → MO116 (26%) MO110 → MO116 (55%) MO111 → MO116 (19%)	4.6402 eV (267.19 nm)	0.2436

S ₂₀	MO108 → MO116 (72%) MO110 → MO116 (28%)	4.6791 eV (264.98 nm)	0.0553
S ₂₁	MO109 → MO116 (18%) MO114 → MO121 (51%) MO114 → MO122 (17%) MO115 → MO124 (14%)	4.7093 eV (263.27 nm)	0.1746
S ₂₂	MO115 → MO124	4.7793 eV (259.42 nm)	0.0386
S ₂₃	MO109 → MO116 (21%) MO113 → MO118 (79%)	4.7891 eV (258.89 nm)	0.0383
S ₂₄	MO109 → MO116 (37%) MO113 → MO118 (17%) MO115 → MO125 (46%)	4.8829 eV (253.91 nm)	0.0199
S ₂₅	MO109 → MO116 (17%) MO114 → MO122 (19%) MO114 → MO123 (10%) MO115 → MO122 (8%) MO115 → MO123 (13%) MO115 → MO124 (12%) MO115 → MO125 (21%)	4.9264 eV (251.67 nm)	0.0882
S ₂₆	MO109 → MO116 (10%) MO112 → MO117 (8%) MO113 → MO119 (18%) MO114 → MO123 (25%) MO114 → MO124 (8%) MO115 → MO124 (17%) MO115 → MO126 (14%)	4.9817 eV (248.88 nm)	0.0596
S ₂₇	MO109 → MO116 (16%) MO112 → MO117 (17%) MO113 → MO119 (52%) MO114 → MO122 (15%)	4.9986 eV (248.04 nm)	0.1256
S ₂₈	MO108 → MO117 (30%) MO111 → MO117 (70%)	5.0136 eV (247.30 nm)	0.0087
S ₂₉	MO115 → MO126 (65%) MO115 → MO127 (35%)	5.0471 eV (245.66 nm)	0.0042
S ₃₀	MO113 → MO118 (17%) MO113 → MO119 (39%) MO114 → MO123 (19%) MO115 → MO125 (25%)	5.0689 eV (244.60 nm)	0.2562

HOMO: MO115, LUMO: MO116

Table S6 Calculated excited states of **1OMe** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO107 → MO108	3.0683 eV (404.08 nm)	0.3267
S ₂	MO107 → MO109	3.4868 eV (355.58 nm)	0.0388
S ₃	MO106 → MO108 (79%) MO107 → MO109 (21%)	3.7464 eV (330.94 nm)	0.1532
S ₄	MO104 → MO109 (15%) MO105 → MO108 (73%) MO107 → MO109 (12%)	4.0579 eV (305.54 nm)	0.0469
S ₅	MO105 → MO108 (21%) MO106 → MO109 (23%) MO107 → MO110 (56%)	4.1615 eV (297.93 nm)	0.0776
S ₆	MO106 → MO109	4.2227 eV (293.62 nm)	0.3329
S ₇	MO105 → MO109 (12%) MO107 → MO111 (76%) MO107 → MO112 (12%)	4.3546 eV (284.72 nm)	0.0280
S ₈	MO104 → MO108 (63%) MO106 → MO109 (14%) MO107 → MO111 (23%)	4.3732 eV (283.51 nm)	0.0297
S ₉	MO103 → MO108 (16%) MO104 → MO108 (18%) MO107 → MO112 (66%)	4.3930 eV (282.23 nm)	0.0232
S ₁₀	MO104 → MO108 (14%) MO105 → MO109 (23%) MO107 → MO113 (63%)	4.5070 eV (275.10 nm)	0.2288
S ₁₁	MO100 → MO108 (48%) MO103 → MO108 (29%) MO107 → MO113 (23%)	4.5901 eV (270.11 nm)	0.1265
S ₁₂	MO100 → MO108 (32%) MO103 → MO108 (19%) MO104 → MO108 (14%) MO105 → MO109 (35%)	4.6011 eV (269.47 nm)	0.1177
S ₁₃	MO100 → MO108 (22%) MO102 → MO108 (51%) MO104 → MO109 (15%) MO107 → MO112 (12%)	4.7316 eV (262.04 nm)	0.0077
S ₁₄	MO102 → MO108 (31%) MO103 → MO108 (36%) MO104 → MO109 (33%)	4.7385 eV (261.65 nm)	0.0785
S ₁₅	MO101 → MO108 (28%) MO102 → MO108 (43%) MO103 → MO108 (15%) MO107 → MO115 (14%)	4.7676 eV (260.06 nm)	0.1747
S ₁₆	MO101 → MO108 (21%) MO106 → MO110 (79%)	4.8311 eV (256.64 nm)	0.1458
S ₁₇	MO107 → MO114 (87%) MO107 → MO119 (13%)	4.9165 eV (252.18 nm)	0.0013
S ₁₈	MO106 → MO111 (85%) MO107 → MO111 (15%)	4.9659 eV (249.67 nm)	0.0796

S ₁₉	MO103 → MO110 (16%) MO105 → MO112 (17%) MO106 → MO112 (67%)	5.0223 eV (246.87 nm)	0.0060
S ₂₀	MO102 → MO109 (13%) MO104 → MO110 (20%) MO106 → MO110 (21%) MO107 → MO115 (46%)	5.0599 eV (245.03 nm)	0.2088
S ₂₁	MO100 → MO109 (49%) MO103 → MO109 (31%) MO106 → MO112 (20%)	5.0897 eV (243.60 nm)	0.0070
S ₂₂	MO101 → MO108 (16%) MO104 → MO109 (20%) MO105 → MO110 (21%) MO106 → MO113 (11%) MO107 → MO115 (32%)	5.1251 eV (241.92 nm)	0.3734
S ₂₃	MO102 → MO109 (32%) MO105 → MO110 (37%) MO105 → MO111 (12%) MO106 → MO111 (19%)	5.2026 eV (238.31 nm)	0.0783
S ₂₄	MO103 → MO109 (22%) MO105 → MO110 (23%) MO107 → MO116 (45%) MO107 → MO117 (10%)	5.2083 eV (238.05 nm)	0.0442
S ₂₅	MO100 → MO109 (34%) MO107 → MO116 (46%) MO107 → MO117 (20%)	5.2114 eV (237.91 nm)	0.0293
S ₂₆	MO101 → MO108 (13%) MO102 → MO109 (37%) MO103 → MO109 (27%) MO107 → MO116 (23%)	5.2293 eV (237.10 nm)	0.0709
S ₂₇	MO107 → MO117	5.2551 eV (235.93 nm)	0.0058
S ₂₈	MO105 → MO111 (28%) MO106 → MO113 (54%) MO107 → MO117 (18%)	5.2765 eV (234.97 nm)	0.0153
S ₂₉	MO102 → MO113 (12%) MO104 → MO109 (13%) MO105 → MO111 (55%) MO106 → MO111 (20%)	5.3354 eV (232.38 nm)	0.0170
S ₃₀	MO105 → MO111 (28%) MO107 → MO118 (72%)	5.3758 eV (230.64 nm)	0.0325

HOMO: MO107, LUMO: MO108

Table S7 Calculated excited states of **1Me** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO099 → MO100	3.1478 eV (393.88 nm)	0.3394
S ₂	MO098 → MO100 (22%) MO099 → MO101 (78%)	3.5688 eV (347.41 nm)	0.0465
S ₃	MO098 → MO100	3.9981 eV (310.10 nm)	0.1821
S ₄	MO098 → MO100 (15%) MO098 → MO101 (14%) MO099 → MO102 (71%)	4.1945 eV (295.59 nm)	0.1713
S ₅	MO095 → MO101 (14%) MO097 → MO100 (72%) MO099 → MO102 (14%)	4.3611 eV (284.30 nm)	0.0675
S ₆	MO095 → MO100 (17%) MO099 → MO102 (15%) MO099 → MO103 (68%)	4.4000 eV (281.78 nm)	0.2499
S ₇	MO097 → MO100 (19%) MO098 → MO101 (46%) MO099 → MO103 (35%)	4.4394 eV (279.28 nm)	0.2430
S ₈	MO095 → MO100 (18%) MO099 → MO104 (82%)	4.4691 eV (277.43 nm)	0.0030
S ₉	MO092 → MO100 (12%) MO096 → MO100 (31%) MO099 → MO104 (33%) MO099 → MO105 (24%)	4.4978 eV (275.66 nm)	0.0190
S ₁₀	MO092 → MO100 (12%) MO095 → MO100 (41%) MO096 → MO100 (14%) MO098 → MO101 (33%)	4.5669 eV (271.48 nm)	0.2296
S ₁₁	MO095 → MO100 (32%) MO099 → MO105 (68%)	4.5883 eV (270.22 nm)	0.0108
S ₁₂	MO094 → MO100 (75%) MO099 → MO105 (25%)	4.6323 eV (267.65 nm)	0.0189
S ₁₃	MO092 → MO100 (69%) MO099 → MO105 (31%)	4.6815 eV (264.84 nm)	0.0083
S ₁₄	MO095 → MO100 (19%) MO097 → MO100 (18%) MO097 → MO101 (63%)	4.8377 eV (256.29 nm)	0.1105
S ₁₅	MO093 → MO100 (25%) MO095 → MO101 (20%) MO097 → MO101 (46%) MO098 → MO102 (9%)	4.8555 eV (255.35 nm)	0.0913
S ₁₆	MO092 → MO101 (23%) MO096 → MO101 (64%) MO099 → MO107 (13%)	4.9831 eV (248.81 nm)	0.0234
S ₁₇	MO093 → MO100 (19%) MO096 → MO101 (25%) MO098 → MO102 (56%)	4.9949 eV (248.22 nm)	0.0625
S ₁₈	MO099 → MO107	5.0203 eV (246.96 nm)	0.0021

S ₁₉	MO094 → MO101 (61%) MO094 → MO103 (12%) MO097 → MO104 (14%) MO099 → MO107 (13%)	5.0591 eV (245.07 nm)	0.0159
S ₂₀	MO094 → MO101 (22%) MO098 → MO102 (16%) MO099 → MO106 (52%) MO099 → MO107 (10%)	5.0828 eV (243.93 nm)	0.1248
S ₂₁	MO092 → MO101	5.1456 eV (240.95 nm)	0.0056
S ₂₂	MO093 → MO100 (80%) MO099 → MO106 (20%)	5.2086 eV (238.04 nm)	0.6087
S ₂₃	MO092 → MO101 (12%) MO094 → MO103 (11%) MO096 → MO102 (28%) MO096 → MO106 (12%) MO098 → MO105 (37%)	5.2490 eV (236.21 nm)	0.0040
S ₂₄	MO093 → MO100 (9%) MO094 → MO101 (14%) MO096 → MO102 (14%) MO097 → MO103 (12%) MO098 → MO104 (27%) MO098 → MO105 (24%)	5.2738 eV (235.09 nm)	0.0425
S ₂₅	MO098 → MO103	5.3238 eV (232.88 nm)	0.0382
S ₂₆	MO099 → MO108 (83%) MO099 → MO112 (17%)	5.3465 eV (231.90 nm)	0.0008
S ₂₇	MO099 → MO109	5.3864 eV (230.18 nm)	0.0100
S ₂₈	MO097 → MO102 (20%) MO099 → MO110 (80%)	5.4356 eV (228.10 nm)	0.0848
S ₂₉	MO093 → MO101 (25%) MO097 → MO102 (47%) MO097 → MO104 (9%) MO098 → MO104 (19%)	5.4656 eV (226.84 nm)	0.0007
S ₃₀	MO094 → MO103 (19%) MO097 → MO104 (37%) MO098 → MO104 (44%)	5.5112 eV (224.97 nm)	0.0336

HOMO: MO099, LUMO: MO100

Table S8 Calculated excited states of **1H** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO091 → MO092	3.1959 eV (387.95 nm)	0.3325
S ₂	MO090 → MO092 (24%) MO091 → MO093 (76%)	3.6175 eV (342.73 nm)	0.0481
S ₃	MO090 → MO092	4.1006 eV (302.35 nm)	0.1300
S ₄	MO090 → MO092 (28%) MO091 → MO094 (72%)	4.2210 eV (293.74 nm)	0.2221
S ₅	MO091 → MO095	4.3682 eV (283.83 nm)	0.1873
S ₆	MO086 → MO092 (16%) MO088 → MO092 (22%) MO090 → MO093 (36%) MO091 → MO094 (9%) MO091 → MO096 (17%)	4.4889 eV (276.20 nm)	0.2486
S ₇	MO091 → MO096	4.5171 eV (274.48 nm)	0.0512
S ₈	MO084 → MO092 (15%) MO088 → MO092 (10%) MO089 → MO092 (27%) MO090 → MO093 (7%) MO091 → MO096 (22%) MO091 → MO097 (19%)	4.5311 eV (273.63 nm)	0.0293
S ₉	MO087 → MO092 (61%) MO089 → MO092 (13%) MO091 → MO096 (12%) MO091 → MO097 (14%)	4.5774 eV (270.86 nm)	0.0184
S ₁₀	MO085 → MO092 (19%) MO091 → MO097 (81%)	4.6146 eV (268.68 nm)	0.0141
S ₁₁	MO088 → MO092	4.6525 eV (266.49 nm)	0.1878
S ₁₂	MO086 → MO092 (25%) MO088 → MO092 (23%) MO089 → MO092 (52%)	4.6981 eV (263.90 nm)	0.0224
S ₁₃	MO084 → MO092 (26%) MO086 → MO092 (58%) MO091 → MO098 (16%)	4.7312 eV (262.05 nm)	0.1645
S ₁₄	MO084 → MO092 (8%) MO085 → MO092 (29%) MO086 → MO092 (7%) MO086 → MO093 (11%) MO088 → MO093 (16%) MO090 → MO094 (12%) MO091 → MO098 (17%)	4.9321 eV (251.38 nm)	0.0327
S ₁₅	MO084 → MO093 (30%) MO089 → MO093 (59%) MO090 → MO094 (11%)	5.0094 eV (247.50 nm)	0.0043
S ₁₆	MO085 → MO092 (13%) MO086 → MO093 (11%) MO087 → MO093 (45%) MO089 → MO093 (12%) MO090 → MO094 (19%)	5.0381 eV (246.10 nm)	0.0408

S ₁₇	MO090 → MO094 (22%) MO091 → MO099 (78%)	5.0720 eV (244.45 nm)	0.0052
S ₁₈	MO087 → MO093 (15%) MO088 → MO093 (46%) MO091 → MO099 (39%)	5.0820 eV (243.97 nm)	0.0152
S ₁₉	MO086 → MO093 (79%) MO089 → MO093 (21%)	5.1124 eV (242.52 nm)	0.0887
S ₂₀	MO085 → MO093 (12%) MO086 → MO093 (16%) MO087 → MO093 (14%) MO089 → MO093 (16%) MO090 → MO093 (9%) MO091 → MO098 (33%)	5.1185 eV (242.23 nm)	0.1752
S ₂₁	MO084 → MO093 (54%) MO086 → MO093 (26%) MO087 → MO093 (20%)	5.1669 eV (239.96 nm)	0.0490
S ₂₂	MO085 → MO092 (56%) MO087 → MO093 (18%) MO090 → MO095 (26%)	5.2693 eV (235.30 nm)	0.3812
S ₂₃	MO086 → MO097 (11%) MO089 → MO094 (38%) MO090 → MO095 (11%) MO090 → MO097 (40%)	5.3273 eV (232.73 nm)	0.0109
S ₂₄	MO088 → MO095 (16%) MO090 → MO095 (84%)	5.3444 eV (231.99 nm)	0.0859
S ₂₅	MO086 → MO095 (22%) MO088 → MO093 (14%) MO090 → MO095 (24%) MO090 → MO096 (40%)	5.3766 eV (230.60 nm)	0.0296
S ₂₆	MO091 → MO100	5.4166 eV (228.90 nm)	0.0016
S ₂₇	MO091 → MO100 (17%) MO091 → MO101 (71%) MO091 → MO102 (12%)	5.4587 eV (227.13 nm)	0.0169
S ₂₈	MO088 → MO094 (21%) MO091 → MO102 (79%)	5.4914 eV (225.78 nm)	0.0738
S ₂₉	MO086 → MO096 (10%) MO087 → MO096 (23%) MO088 → MO095 (19%) MO090 → MO096 (48%)	5.5645 eV (222.81 nm)	0.0201
S ₃₀	MO083 → MO092 (18%) MO085 → MO093 (37%) MO090 → MO094 (11%) MO091 → MO102 (34%)	5.6139 eV (220.85 nm)	0.1155

HOMO: MO091, LUMO: MO092

Table S9 Calculated excited states of **1F** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO099 → MO100	3.2056 eV (386.77 nm)	0.3313
S ₂	MO098 → MO100 (24%) MO099 → MO101 (76%)	3.6265 eV (341.88 nm)	0.0459
S ₃	MO098 → MO100 (75%) MO099 → MO102 (25%)	4.0961 eV (302.69 nm)	0.1439
S ₄	MO099 → MO102	4.2364 eV (292.66 nm)	0.1826
S ₅	MO099 → MO103	4.3179 eV (287.14 nm)	0.0216
S ₆	MO099 → MO104	4.3861 eV (282.68 nm)	0.0061
S ₇	MO098 → MO101 (14%) MO099 → MO105 (86%)	4.4424 eV (279.10 nm)	0.1779
S ₈	MO096 → MO100 (34%) MO096 → MO101 (11%) MO098 → MO101 (42%) MO099 → MO102 (13%)	4.4871 eV (276.31 nm)	0.2517
S ₉	MO092 → MO100 (10%) MO096 → MO100 (20%) MO097 → MO100 (51%) MO098 → MO101 (19%)	4.5363 eV (273.31 nm)	0.0781
S ₁₀	MO092 → MO100 (62%) MO095 → MO100 (26%) MO098 → MO101 (12%)	4.6099 eV (268.95 nm)	0.0273
S ₁₁	MO092 → MO100 (16%) MO096 → MO100 (63%) MO099 → MO106 (21%)	4.6898 eV (264.37 nm)	0.3445
S ₁₂	MO095 → MO100	4.8638 eV (254.91 nm)	0.0059
S ₁₃	MO094 → MO100 (60%) MO096 → MO101 (27%) MO098 → MO102 (13%)	4.9072 eV (252.66 nm)	0.0051
S ₁₄	MO093 → MO100 (37%) MO094 → MO100 (48%) MO099 → MO106 (15%)	4.9377 eV (251.10 nm)	0.0329
S ₁₅	MO092 → MO101 (18%) MO097 → MO101 (82%)	4.9966 eV (248.14 nm)	0.0958
S ₁₆	MO092 → MO101 (33%) MO095 → MO101 (16%) MO098 → MO102 (51%)	5.0664 eV (244.72 nm)	0.0209
S ₁₇	MO099 → MO107	5.1001 eV (243.10 nm)	0.0013
S ₁₈	MO092 → MO101 (50%) MO095 → MO101 (27%) MO096 → MO101 (23%)	5.1074 eV (242.75 nm)	0.0064
S ₁₉	MO096 → MO101 (11%) MO097 → MO101 (19%) MO098 → MO101 (12%) MO099 → MO106 (58%)	5.1291 eV (241.73 nm)	0.1838

S ₂₀	MO093 → MO100 (12%) MO094 → MO101 (11%) MO096 → MO101 (12%) MO098 → MO102 (18%) MO098 → MO103 (37%) MO099 → MO106 (10%)	5.1910 eV (238.84 nm)	0.1232
S ₂₁	MO095 → MO100 (17%) MO095 → MO101 (22%) MO095 → MO106 (11%) MO098 → MO104 (50%)	5.2142 eV (237.78 nm)	0.0080
S ₂₂	MO093 → MO100 (28%) MO094 → MO105 (10%) MO096 → MO101 (22%) MO097 → MO103 (16%) MO098 → MO102 (24%)	5.2804 eV (234.80 nm)	0.3357
S ₂₃	MO094 → MO105 (13%) MO097 → MO102 (14%) MO097 → MO103 (27%) MO098 → MO103 (46%)	5.3407 eV (232.15 nm)	0.0668
S ₂₄	MO095 → MO101	5.3592 eV (231.35 nm)	0.0059
S ₂₅	MO093 → MO101 (19%) MO098 → MO105 (81%)	5.3874 eV (230.14 nm)	0.0463
S ₂₆	MO093 → MO101 (22%) MO096 → MO102 (12%) MO099 → MO108 (50%) MO099 → MO110 (16%)	5.4637 eV (226.92 nm)	0.0277
S ₂₇	MO097 → MO103 (11%) MO098 → MO105 (16%) MO099 → MO108 (55%) MO099 → MO109 (18%)	5.4761 eV (226.41 nm)	0.0557
S ₂₈	MO093 → MO101 (12%) MO094 → MO101 (19%) MO094 → MO105 (12%) MO097 → MO102 (11%) MO097 → MO103 (16%) MO098 → MO103 (7%) MO098 → MO105 (8%) MO099 → MO108 (6%) MO099 → MO110 (9%)	5.4862 eV (225.99 nm)	0.0148
S ₂₉	MO093 → MO101 (10%) MO099 → MO108 (13%) MO099 → MO109 (42%) MO099 → MO110 (35%)	5.5260 eV (224.37 nm)	0.0072
S ₃₀	MO091 → MO100 (15%) MO093 → MO101 (36%) MO096 → MO102 (11%) MO097 → MO103 (18%) MO098 → MO102 (10%) MO099 → MO109 (10%)	5.5977 eV (221.49 nm)	0.0862

HOMO: MO099, LUMO: MO100

Table S10 Calculated excited states of **1Cl** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO107 → MO108	3.2028 eV (387.11 nm)	0.3669
S ₂	MO106 → MO108 (24%) MO107 → MO109 (76%)	3.6302 eV (341.54 nm)	0.0541
S ₃	MO106 → MO108	4.0747 eV (304.27 nm)	0.1038
S ₄	MO106 → MO108 (35%) MO107 → MO110 (65%)	4.1638 eV (297.77 nm)	0.4090
S ₅	MO107 → MO110 (17%) MO107 → MO111 (83%)	4.2360 eV (292.69 nm)	0.1399
S ₆	MO107 → MO112	4.3248 eV (286.68 nm)	0.0076
S ₇	MO107 → MO113	4.3952 eV (282.09 nm)	0.0084
S ₈	MO104 → MO108 (23%) MO104 → MO109 (10%) MO105 → MO108 (33%) MO106 → MO109 (34%)	4.4522 eV (278.48 nm)	0.1940
S ₉	MO105 → MO108	4.4916 eV (276.04 nm)	0.1167
S ₁₀	MO100 → MO108 (59%) MO103 → MO108 (26%) MO106 → MO109 (15%)	4.6012 eV (269.46 nm)	0.0428
S ₁₁	MO100 → MO108 (28%) MO104 → MO108 (72%)	4.6481 eV (266.74 nm)	0.2478
S ₁₂	MO103 → MO108 (66%) MO106 → MO113 (19%) MO107 → MO113 (15%)	4.8367 eV (256.34 nm)	0.0113
S ₁₃	MO100 → MO108 (20%) MO101 → MO108 (55%) MO106 → MO110 (25%)	4.8570 eV (255.27 nm)	0.0246
S ₁₄	MO101 → MO108 (28%) MO102 → MO108 (72%)	4.9093 eV (252.55 nm)	0.0286
S ₁₅	MO105 → MO109	4.9471 eV (250.62 nm)	0.0917
S ₁₆	MO104 → MO109 (40%) MO105 → MO109 (13%) MO106 → MO110 (47%)	5.0118 eV (247.39 nm)	0.0404
S ₁₇	MO101 → MO108 (12%) MO101 → MO109 (8%) MO104 → MO108 (10%) MO105 → MO109 (14%) MO106 → MO111 (13%) MO107 → MO114 (35%) MO107 → MO115 (8%)	5.0459 eV (245.71 nm)	0.2197
S ₁₈	MO107 → MO115	5.0763 eV (244.24 nm)	0.0062
S ₁₉	MO100 → MO109 (56%) MO103 → MO109 (31%) MO105 → MO109 (13%)	5.0870 eV (243.73 nm)	0.0219

S ₂₀	MO102 → MO108 (7%) MO102 → MO109 (16%) MO102 → MO111 (17%) MO105 → MO112 (19%) MO106 → MO111 (9%) MO106 → MO112 (32%)	5.1480 eV (240.84 nm)	0.0531
S ₂₁	MO100 → MO109 (12%) MO101 → MO108 (9%) MO104 → MO109 (19%) MO106 → MO111 (41%) MO106 → MO113 (19%)	5.1642 eV (240.08 nm)	0.0158
S ₂₂	MO100 → MO109 (23%) MO106 → MO110 (15%) MO106 → MO113 (62%)	5.1834 eV (239.19 nm)	0.0089
S ₂₃	MO101 → MO108 (60%) MO104 → MO109 (40%)	5.2197 eV (237.53 nm)	0.5407
S ₂₄	MO106 → MO112	5.3273 eV (232.73 nm)	0.0109
S ₂₅	MO103 → MO109 (70%) MO106 → MO113 (30%)	5.3371 eV (232.31 nm)	0.0042
S ₂₆	MO107 → MO116	5.3873 eV (230.14 nm)	0.0030
S ₂₇	MO104 → MO110 (11%) MO107 → MO116 (35%) MO107 → MO117 (43%) MO107 → MO119 (11%)	5.3977 eV (229.70 nm)	0.0417
S ₂₈	MO102 → MO109 (50%) MO105 → MO110 (18%) MO106 → MO112 (16%) MO107 → MO118 (16%)	5.4331 eV (228.20 nm)	0.0029
S ₂₉	MO104 → MO110 (18%) MO107 → MO118 (55%) MO107 → MO119 (27%)	5.4704 eV (226.64 nm)	0.0280
S ₃₀	MO105 → MO110 (72%) MO105 → MO111 (28%)	5.5106 eV (224.99 nm)	0.0319

HOMO: MO107, LUMO: MO108

Table S11 Calculated excited states of **1CF₃** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO123 → MO124	3.2370 eV (383.02 nm)	0.3880
S ₂	MO122 → MO124 (26%) MO123 → MO125 (74%)	3.6695 eV (337.88 nm)	0.0765
S ₃	MO123 → MO126	3.9368 eV (314.93 nm)	0.1946
S ₄	MO123 → MO125 (21%) MO123 → MO126 (13%) MO123 → MO127 (66%)	4.0331 eV (307.42 nm)	0.0507
S ₅	MO122 → MO124 (72%) MO123 → MO127 (28%)	4.2078 eV (294.65 nm)	0.3526
S ₆	MO123 → MO128	4.3651 eV (284.03 nm)	0.0031
S ₇	MO123 → MO129	4.4509 eV (278.56 nm)	0.0016
S ₈	MO122 → MO125 (76%) MO123 → MO130 (24%)	4.5172 eV (274.47 nm)	0.3001
S ₉	MO117 → MO124 (54%) MO118 → MO124 (13%) MO120 → MO124 (23%) MO122 → MO125 (10%)	4.5910 eV (270.06 nm)	0.0243
S ₁₀	MO119 → MO124 (8%) MO121 → MO124 (35%) MO122 → MO125 (22%) MO122 → MO127 (24%) MO123 → MO130 (11%)	4.7242 eV (262.44 nm)	0.1730
S ₁₁	MO118 → MO124 (44%) MO119 → MO124 (39%) MO122 → MO127 (17%)	4.8164 eV (257.42 nm)	0.0565
S ₁₂	MO118 → MO124 (15%) MO119 → MO124 (22%) MO120 → MO124 (42%) MO122 → MO127 (9%) MO123 → MO129 (12%)	4.8505 eV (255.61 nm)	0.0289
S ₁₃	MO122 → MO126	4.9057 eV (252.74 nm)	0.1111
S ₁₄	MO119 → MO124	4.9235 eV (251.82 nm)	0.0082
S ₁₅	MO116 → MO124 (17%) MO119 → MO124 (19%) MO122 → MO126 (22%) MO123 → MO130 (42%)	4.9482 eV (250.57 nm)	0.0535
S ₁₆	MO117 → MO124 (9%) MO117 → MO125 (41%) MO118 → MO125 (12%) MO120 → MO125 (19%) MO121 → MO125 (19%)	5.0506 eV (245.49 nm)	0.0011
S ₁₇	MO116 → MO125 (13%) MO122 → MO127 (58%) MO123 → MO130 (29%)	5.1095 eV (242.65 nm)	0.5175

S ₁₈	MO116 → MO124 (24%) MO118 → MO124 (12%) MO121 → MO125 (51%) MO122 → MO127 (13%)	5.1286 eV (241.75 nm)	0.0144
S ₁₉	MO118 → MO124 (15%) MO118 → MO126 (20%) MO119 → MO125 (28%) MO123 → MO131 (37%)	5.2008 eV (238.40 nm)	0.0338
S ₂₀	MO118 → MO125 (14%) MO122 → MO128 (16%) MO123 → MO131 (70%)	5.2172 eV (237.65 nm)	0.0009
S ₂₁	MO120 → MO125 (48%) MO120 → MO126 (26%) MO120 → MO127 (14%) MO121 → MO129 (12%)	5.2439 eV (236.44 nm)	0.0214
S ₂₂	MO116 → MO124 (52%) MO119 → MO125 (24%) MO120 → MO125 (11%) MO122 → MO127 (13%)	5.2876 eV (234.48 nm)	0.1224
S ₂₃	MO118 → MO125 (27%) MO119 → MO125 (36%) MO120 → MO126 (20%) MO122 → MO128 (17%)	5.3384 eV (232.25 nm)	0.0875
S ₂₄	MO118 → MO125 (23%) MO119 → MO125 (31%) MO120 → MO125 (28%) MO122 → MO129 (18%)	5.3600 eV (231.32 nm)	0.0733
S ₂₅	MO119 → MO125 (19%) MO122 → MO128 (81%)	5.3732 eV (230.74 nm)	0.0073
S ₂₆	MO118 → MO127 (13%) MO118 → MO128 (9%) MO119 → MO125 (18%) MO119 → MO126 (9%) MO119 → MO128 (9%) MO121 → MO126 (33%) MO123 → MO133 (9%)	5.4861 eV (226.00 nm)	0.0298
S ₂₇	MO118 → MO125 (16%) MO118 → MO126 (16%) MO121 → MO126 (15%) MO123 → MO133 (53%)	5.5183 eV (224.68 nm)	0.0474
S ₂₈	MO118 → MO125 (10%) MO118 → MO126 (20%) MO119 → MO127 (14%) MO121 → MO126 (46%) MO123 → MO132 (10%)	5.5482 eV (223.47 nm)	0.0053
S ₂₉	MO123 → MO132 (56%) MO123 → MO133 (24%) MO123 → MO134 (10%) MO123 → MO135 (10%)	5.5779 eV (222.28 nm)	0.0013

S ₃₀	MO116 → MO129 (7%)	5.6090 eV (221.05 nm)	0.0232
	MO117 → MO126 (7%)		
	MO117 → MO127 (10%)		
	MO120 → MO126 (21%)		
	MO120 → MO127 (9%)		
	MO121 → MO127 (8%)		
	MO121 → MO129 (9%)		
	MO122 → MO129 (29%)		
HOMO: MO123, LUMO: MO124			

Table S12 Calculated excited states of **1CN** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO103 → MO104	3.1737 eV (390.66 nm)	0.4744
S ₂	MO102 → MO104 (15%) MO103 → MO105 (85%)	3.5272 eV (351.51 nm)	0.2189
S ₃	MO103 → MO106 (81%) MO103 → MO107 (19%)	3.6901 eV (335.99 nm)	0.1145
S ₄	MO103 → MO105 (15%) MO103 → MO107 (85%)	3.8159 eV (324.92 nm)	0.0025
S ₅	MO102 → MO104 (70%) MO103 → MO107 (30%)	4.1321 eV (300.05 nm)	0.5646
S ₆	MO103 → MO108	4.3281 eV (286.46 nm)	0.0091
S ₇	MO099 → MO104 (24%) MO103 → MO109 (76%)	4.4054 eV (281.44 nm)	0.0007
S ₈	MO101 → MO104 (23%) MO102 → MO105 (77%)	4.4188 eV (280.59 nm)	0.1544
S ₉	MO101 → MO104	4.5455 eV (272.77 nm)	0.0138
S ₁₀	MO097 → MO104 (37%) MO097 → MO106 (12%) MO099 → MO104 (18%) MO100 → MO104 (33%)	4.5718 eV (271.20 nm)	0.0332
S ₁₁	MO100 → MO104 (18%) MO101 → MO104 (11%) MO102 → MO105 (13%) MO102 → MO106 (58%)	4.5987 eV (269.61 nm)	0.1355
S ₁₂	MO097 → MO104 (24%) MO099 → MO104 (14%) MO101 → MO104 (37%) MO102 → MO107 (10%) MO103 → MO110 (15%)	4.6541 eV (266.40 nm)	0.0580
S ₁₃	MO099 → MO104 (64%) MO101 → MO109 (15%) MO102 → MO109 (21%)	4.7795 eV (259.41 nm)	0.0311
S ₁₄	MO099 → MO104 (15%) MO100 → MO104 (28%) MO102 → MO107 (13%) MO103 → MO110 (44%)	4.7970 eV (258.46 nm)	0.0379
S ₁₅	MO098 → MO104 (42%) MO098 → MO105 (14%) MO100 → MO104 (11%) MO100 → MO105 (10%) MO100 → MO108 (11%) MO101 → MO105 (12%)	4.8778 eV (254.18 nm)	0.0168
S ₁₆	MO100 → MO104 (24%) MO101 → MO105 (76%)	4.9345 eV (251.26 nm)	0.0872

S ₁₇	MO096 → MO104 (9%) MO097 → MO105 (13%) MO099 → MO105 (9%) MO100 → MO104 (14%) MO101 → MO104 (9%) MO101 → MO106 (11%) MO102 → MO107 (35%)	4.9788 eV (249.02 nm)	0.4082
S ₁₈	MO097 → MO104 (13%) MO097 → MO105 (30%) MO098 → MO105 (12%) MO099 → MO105 (18%) MO100 → MO105 (27%)	5.0169 eV (247.14 nm)	0.2593
S ₁₉	MO096 → MO104 (15%) MO097 → MO107 (8%) MO098 → MO105 (31%) MO100 → MO105 (9%) MO100 → MO108 (12%) MO101 → MO106 (12%) MO102 → MO107 (13%)	5.0545 eV (245.30 nm)	0.0444
S ₂₀	MO097 → MO106 (15%) MO097 → MO107 (15%) MO100 → MO105 (70%)	5.0986 eV (243.17 nm)	0.3000
S ₂₁	MO096 → MO104 (44%) MO100 → MO105 (29%) MO101 → MO107 (15%) MO103 → MO110 (12%)	5.1512 eV (240.69 nm)	0.0994
S ₂₂	MO100 → MO106 (19%) MO101 → MO105 (12%) MO101 → MO106 (51%) MO102 → MO109 (18%)	5.1928 eV (238.76 nm)	0.0473
S ₂₃	MO099 → MO104 (19%) MO099 → MO106 (38%) MO100 → MO106 (13%) MO101 → MO106 (30%)	5.2098 eV (237.98 nm)	0.0151
S ₂₄	MO099 → MO105 (14%) MO103 → MO111 (74%) MO103 → MO118 (12%)	5.2322 eV (236.96 nm)	0.0055
S ₂₅	MO097 → MO107 (12%) MO099 → MO105 (71%) MO099 → MO106 (17%)	5.2400 eV (236.61 nm)	0.0046
S ₂₆	MO098 → MO105 (19%) MO100 → MO108 (11%) MO102 → MO108 (70%)	5.3203 eV (233.04 nm)	0.0207
S ₂₇	MO100 → MO106 (61%) MO101 → MO107 (39%)	5.3526 eV (231.63 nm)	0.0257
S ₂₈	MO101 → MO107	5.4058 eV (229.36 nm)	0.0931
S ₂₉	MO100 → MO107 (26%) MO102 → MO110 (20%) MO103 → MO113 (54%)	5.4430 eV (227.79 nm)	0.0521

S ₃₀	MO097 → MO104 (14%)	5.4657 eV (226.84 nm)	0.0288
	MO097 → MO107 (35%)		
	MO097 → MO110 (9%)		
	MO098 → MO106 (11%)		
	MO099 → MO107 (18%)		
	MO101 → MO107 (13%)		
HOMO: MO103, LUMO: MO104			

Table S13 Calculated excited states of **1NO₂** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO113 → MO114 (69%)	2.5492 eV (486.37 nm)	0.2502
	MO113 → MO115 (31%)		
S ₂	MO113 → MO115	2.6025 eV (476.40 nm)	0.2570
S ₃	MO112 → MO115 (20%)	3.3702 eV (367.89 nm)	0.1311
	MO112 → MO117 (13%)		
	MO113 → MO116 (67%)		
S ₄	MO112 → MO115 (84%)	3.4781 eV (356.47 nm)	0.3003
	MO113 → MO117 (16%)		
S ₅	MO112 → MO114 (57%)	3.4952 eV (354.73 nm)	0.0176
	MO112 → MO115 (28%)		
	MO113 → MO116 (15%)		
S ₆	MO105 → MO114 (12%)	3.7711 eV (328.78 nm)	0.0298
	MO105 → MO115 (53%)		
	MO113 → MO117 (35%)		
S ₇	MO104 → MO114	3.7869 eV (327.40 nm)	0.0002
S ₈	MO113 → MO117	3.7872 eV (327.38 nm)	0.1054
S ₉	MO110 → MO114 (16%)	3.9133 eV (316.83 nm)	0.0184
	MO110 → MO115 (70%)		
	MO111 → MO115 (14%)		
S ₁₀	MO108 → MO114 (68%)	3.9596 eV (313.12 nm)	0.0177
	MO109 → MO114 (32%)		
S ₁₁	MO108 → MO114 (22%)	4.0788 eV (303.97 nm)	0.1142
	MO111 → MO114 (39%)		
	MO111 → MO115 (39%)		
S ₁₂	MO111 → MO115	4.0806 eV (303.84 nm)	0.0159
S ₁₃	MO109 → MO114	4.2469 eV (291.94 nm)	0.4008
S ₁₄	MO110 → MO114	4.2673 eV (290.54 nm)	0.0004
S ₁₅	MO109 → MO114 (27%)	4.3105 eV (287.63 nm)	0.0211
	MO112 → MO116 (54%)		
	MO113 → MO117 (19%)		
S ₁₆	MO107 → MO115 (49%)	4.3325 eV (286.17 nm)	0.0121
	MO108 → MO115 (37%)		
	MO110 → MO114 (14%)		
S ₁₇	MO107 → MO115 (14%)	4.3572 eV (284.55 nm)	0.0646
	MO109 → MO114 (14%)		
	MO109 → MO115 (31%)		
	MO112 → MO116 (11%)		
	MO113 → MO118 (30%)		
S ₁₈	MO112 → MO116 (22%)	4.3723 eV (283.57 nm)	0.0050
	MO113 → MO117 (11%)		
	MO113 → MO118 (47%)		
	MO113 → MO119 (20%)		
S ₁₉	MO102 → MO114 (15%)	4.3888 eV (282.50 nm)	0.0025
	MO102 → MO115 (63%)		
	MO108 → MO115 (22%)		
S ₂₀	MO099 → MO114 (26%)	4.3925 eV (282.27 nm)	0.0004
	MO100 → MO114 (74%)		

S ₂₁	MO108 → MO114 (16%) MO108 → MO115 (64%) MO113 → MO119 (20%)	4.4148 eV (280.84 nm)	0.0014
S ₂₂	MO113 → MO119 (55%) MO113 → MO120 (45%)	4.4587 eV (278.07 nm)	0.0121
S ₂₃	MO107 → MO114 (17%) MO112 → MO117 (16%) MO113 → MO120 (67%)	4.4711 eV (277.30 nm)	0.0123
S ₂₄	MO107 → MO114 (87%) MO108 → MO114 (13%)	4.4885 eV (276.23 nm)	0.0020
S ₂₅	MO106 → MO114 (19%) MO106 → MO115 (63%) MO112 → MO117 (18%)	4.6574 eV (266.21 nm)	0.1739
S ₂₆	MO106 → MO114 (9%) MO107 → MO116 (12%) MO108 → MO116 (6%) MO111 → MO116 (22%) MO111 → MO117 (8%) MO112 → MO117 (23%) MO113 → MO119 (13%) MO113 → MO121 (7%)	4.6793 eV (264.96 nm)	0.1185
S ₂₇	MO106 → MO114	4.6933 eV (264.17 nm)	0.0803
S ₂₈	MO106 → MO115 (10%) MO107 → MO115 (13%) MO107 → MO116 (38%) MO107 → MO117 (9%) MO108 → MO116 (10%) MO110 → MO116 (20%)	4.7140 eV (263.01 nm)	0.0277
S ₂₉	MO113 → MO121	4.7799 eV (259.39 nm)	0.0767
S ₃₀	MO111 → MO116 (80%) MO113 → MO121 (20%)	4.9353 eV (251.22 nm)	0.5087

HOMO: MO113, LUMO: MO114

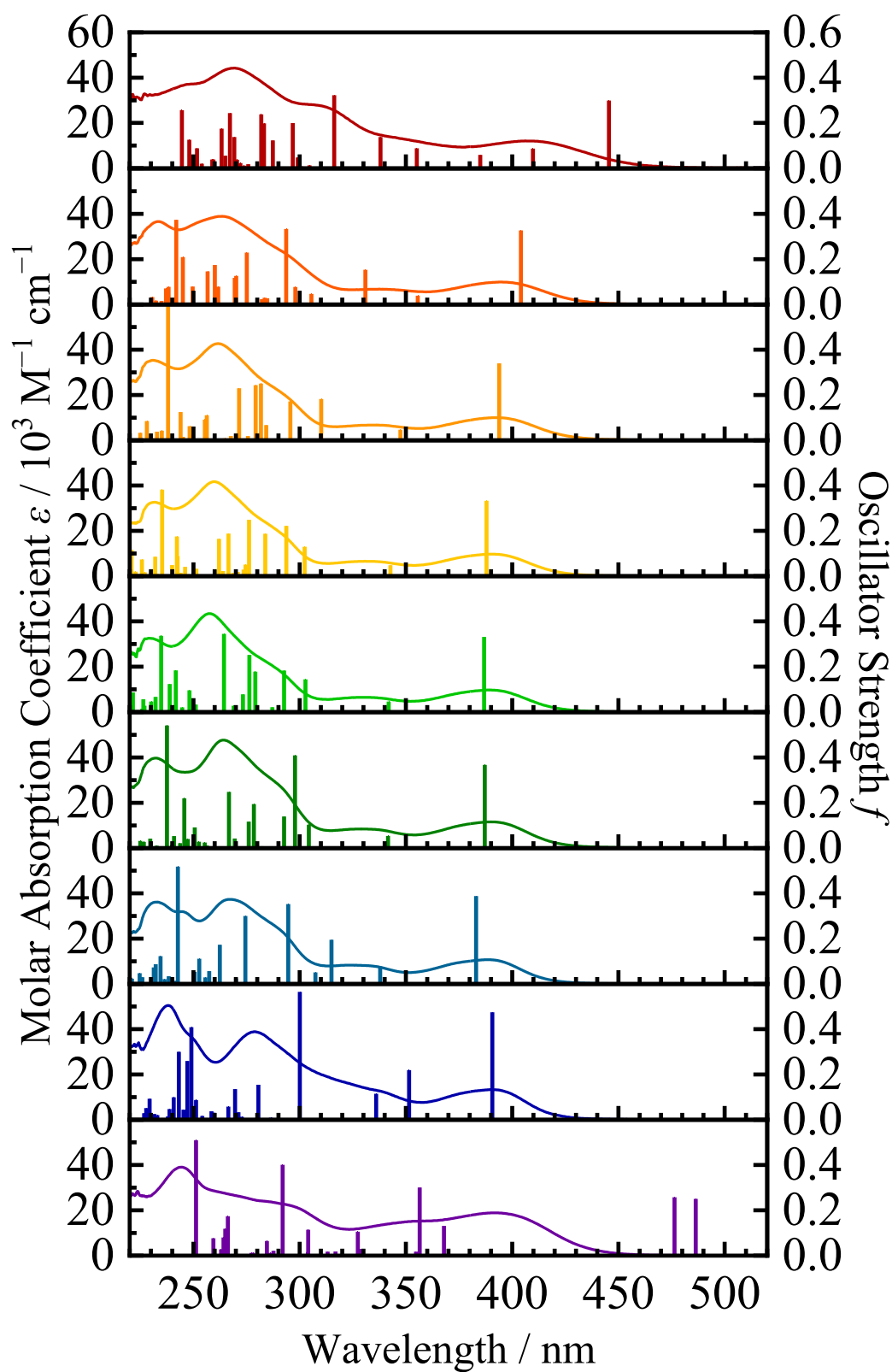


Fig. S23 Comparisons of the absorption spectra and oscillator strengths calculated by TD-DFT (perpendicular bars) of **1R** in dichloromethane. The colors correspond to those indicated in Fig. 1.

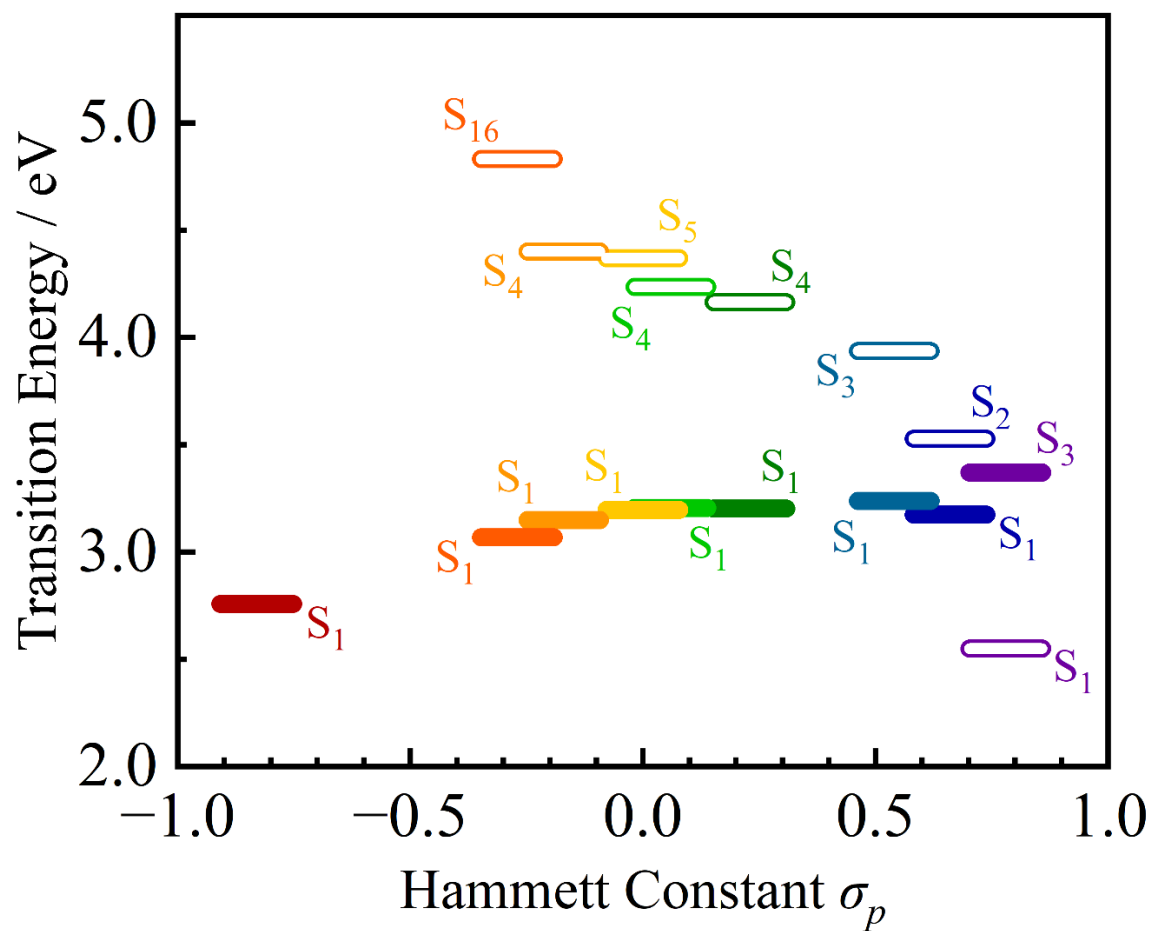


Fig. S24 Transition energies of the charge-transfer excited state in the HBq moiety (closed) and excited state ascribed to the charge-transfer transition from the HBq moiety to the 4-Rphenyl group at 7-position (open) against the σ_p value of the substituent R.

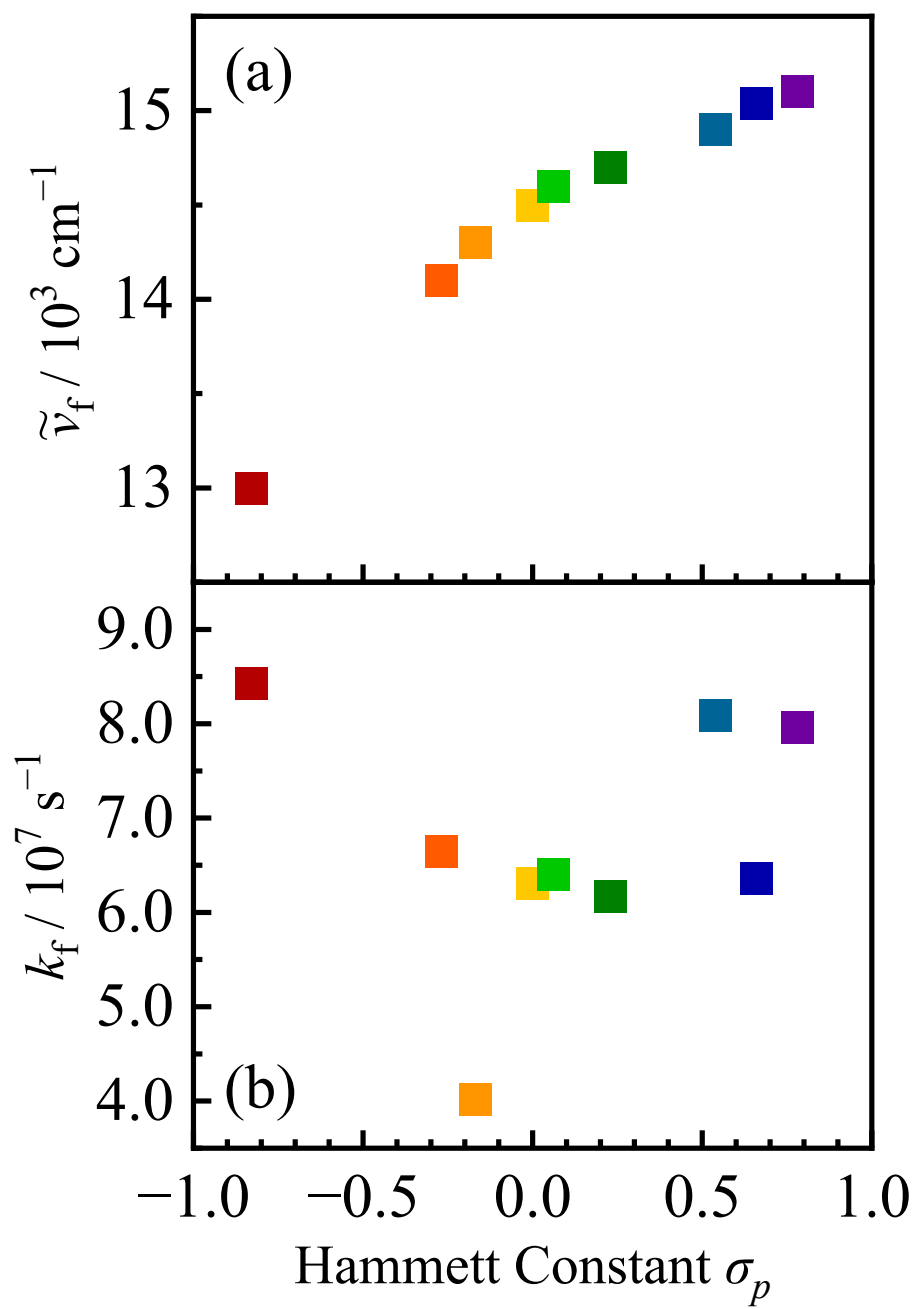


Fig. S25 Dependences of σ_p on the fluorescence maximum wavenumber (a) and fluorescence rate constant (b) of **1R** in dichloromethane.

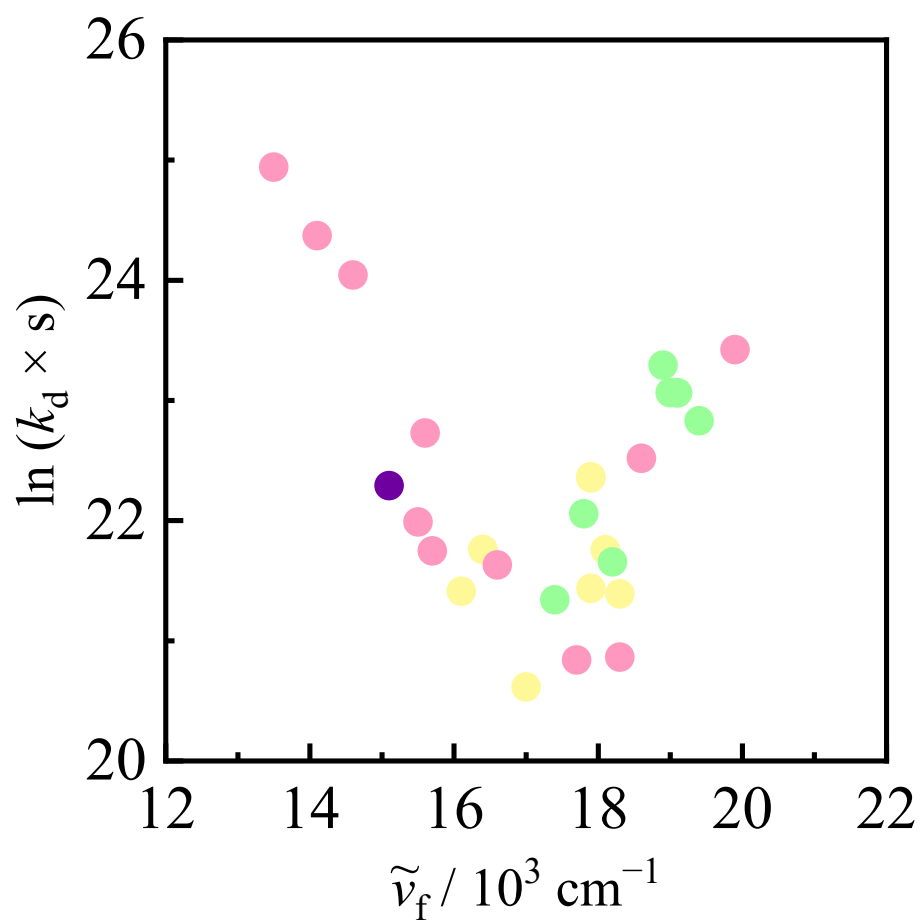


Fig. S26 Energy gap plots for **1NO₂** in dichloromethane (purple) with solvent dependences of 1-methoxy-4-(4-nitrophenyl)naphthalene (pink), 1-(4-nitrophenyl)naphthalene (yellow) and 4-methoxy-4'-nitrobiphenyl (green). Data of the reference derivatives were compiled from reference 48 in the main text.

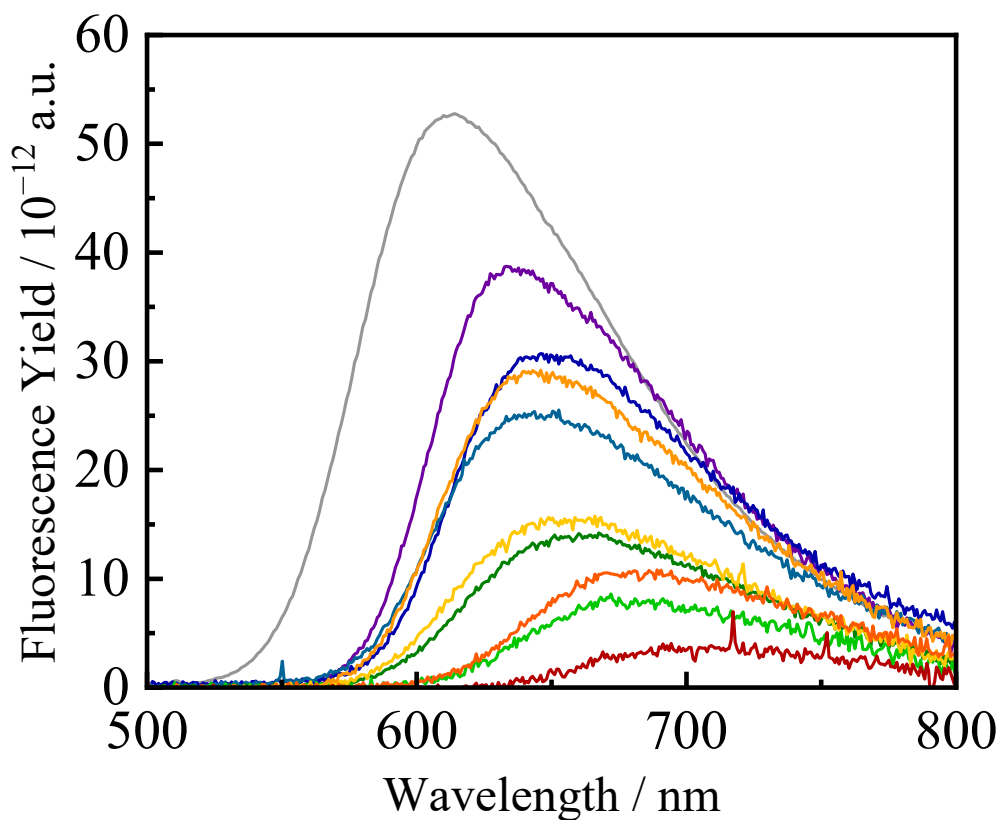


Fig. S27 Fluorescence spectra of **1R** and HBq in crystalline phase. The colors correspond to those indicated in Fig. 1.

Table S14 Photophysical properties of **1R** in crystalline phase.

Derivative	Φ_f^a	τ_f^a / ns	k_f^b / 10^7 s $^{-1}$	k_d^b / 10^9 s $^{-1}$	$k_{d,dcm}/k_{d,crys}^c$
1NMe₂	0.0041	0.13	3.2	7.7	2.8
1OMe	0.014	0.28	5.0	3.6	2.8
1Me	0.037	0.070 (44%)	5.5	1.4	4.1
		0.72 (56%)			
1H	0.020	0.53	3.9	1.9	3.1
1F	0.010	0.28	3.6	3.5	1.6
1Cl	0.018	0.38 (39%)	3.2	1.7	3.1
		0.65 (61%)			
1CF₃	0.034	0.73	4.7	1.3	2.6
1CN	0.042	0.80	5.2	1.2	2.2
1NO₂	0.047	0.76 (43%)	3.2	0.66	7.2
		1.7 (57%)			

a) λ_{ex} = 395 nm b) $\Phi_f = k_f / (k_f + k_d)$ and $\langle \tau_f \rangle = \Sigma(A_i \tau_i^2) / \Sigma(A_i \tau_i)$

c) ratio of k_d in dichloromethane to that in crystalline phase.

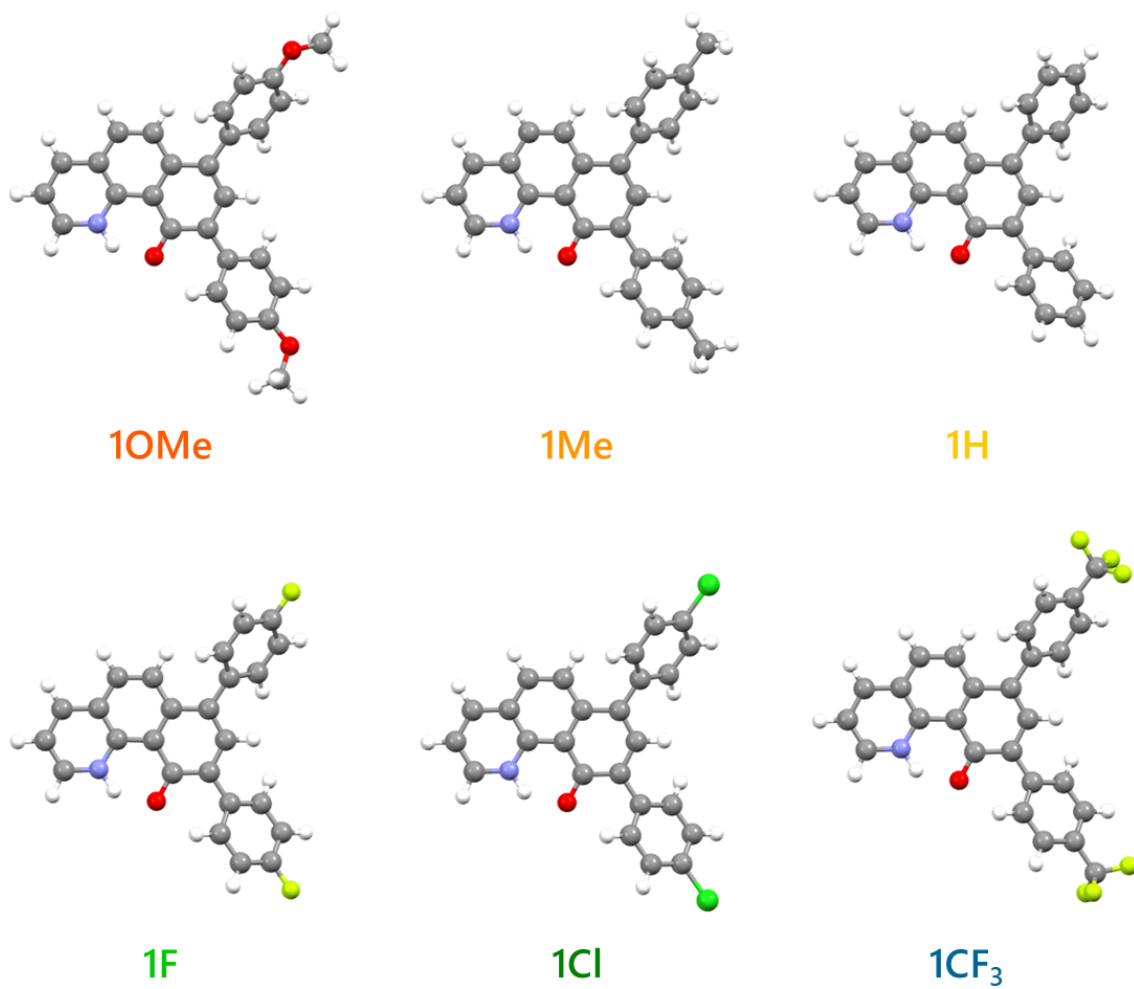


Fig. S28 Optimized geometries of **1R** in the S₁ states in dichloromethane.

Table S15 Calculated excited states of S₁-optimized geometry of **1NMe₂** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO113 → MO116 (16%) MO115 → MO116 (84%)	1.0967 eV (1130.51 nm)	0.0000
S ₁	MO115 → MO116	1.7490 eV (708.88 nm)	0.3440
T ₂	MO111 → MO116 (13%) MO114 → MO116 (72%) MO115 → MO118 (15%)	2.1372 eV (580.11 nm)	0.0000
T ₃	MO113 → MO117 (13%) MO114 → MO116 (28%) MO115 → MO117 (59%)	2.1819 eV (568.23 nm)	0.0000
S ₂	MO114 → MO116	2.2711 eV (545.93 nm)	0.1022
S ₃	MO115 → MO117	2.5677 eV (482.85 nm)	0.0293

HOMO: MO115, LUMO: MO116

Table S16 Calculated excited states of S₁-optimized geometry of **1OMe** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO107 → MO108	1.2132 eV (1022.00 nm)	0.0000
S ₁	MO107 → MO108	1.9079 eV (649.84 nm)	0.3271
T ₂	MO103 → MO108 (12%) MO107 → MO109 (71%) MO107 → MO110 (17%)	2.2590 eV (548.85 nm)	0.0000
T ₃	MO106 → MO108 (75%) MO107 → MO109 (25%)	2.6108 eV (474.90 nm)	0.0000
S ₂	MO107 → MO109	2.7179 eV (456.18 nm)	0.0121
S ₃	MO106 → MO108 (81%) MO107 → MO109 (19%)	2.9018 eV (427.27 nm)	0.1138

HOMO: MO107, LUMO: MO108

Table S17 Calculated excited states of S₁-optimized geometry of **1Me** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO099 → MO100	1.2544 eV (988.37 nm)	0.0000
S ₁	MO099 → MO100	1.9587 eV (632.98 nm)	0.3246
T ₂	MO099 → MO101	2.2921 eV (540.93 nm)	0.0000
T ₃	MO096 → MO100 (15%) MO099 → MO101 (24%) MO099 → MO102 (61%)	2.6924 eV (460.49 nm)	0.0000
S ₂	MO098 → MO100 (14%) MO099 → MO101 (86%)	2.7805 eV (445.91 nm)	0.0220
S ₃	MO097 → MO100 (18%) MO098 → MO100 (70%) MO099 → MO102 (12%)	3.1297 eV (396.15 nm)	0.0734

HOMO: MO099, LUMO: MO100

Table S18 Calculated excited states of S₁-optimized geometry of **1H** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO91 → MO92	1.2937 eV (958.37 nm)	0.0000
S ₁	MO91 → MO92	1.9961 eV (621.13 nm)	0.3128
T ₂	MO91 → MO93	2.3208 eV (534.23 nm)	0.0000
T ₃	MO91 → MO93 (27%) MO91 → MO94 (73%)	2.7386 eV (452.73 nm)	0.0000
S ₂	MO90 → MO92 (14%) MO91 → MO93 (86%)	2.8269 eV (438.58 nm)	0.0246
S ₃	MO88 → MO92 (13%) MO89 → MO92 (53%) MO90 → MO92 (34%)	3.1921 eV (388.41 nm)	0.0117

HOMO: MO91, LUMO: MO92

Table S19 Calculated excited states of S₁-optimized geometry of **1F** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO099 → MO100	1.3040 eV (950.80 nm)	0.0000
S ₁	MO099 → MO100	2.0012 eV (619.56 nm)	0.3103
T ₂	MO099 → MO101 (81%) MO099 → MO102 (19%)	2.3239 eV (533.52 nm)	0.0000
T ₃	MO092 → MO100 (10%) MO098 → MO100 (33%) MO098 → MO102 (9%) MO099 → MO102 (48%)	2.7356 eV (453.22 nm)	0.0000
S ₂	MO098 → MO100 (14%) MO099 → MO101 (86%)	2.8289 eV (438.27 nm)	0.0238
S ₃	MO097 → MO100 (48%) MO098 → MO100 (52%)	3.2126 eV (385.94 nm)	0.0299

HOMO: MO099, LUMO: MO100

Table S20 Calculated excited states of S₁-optimized geometry of **1Cl** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO107 → MO108	1.3200 eV (939.30 nm)	0.0000
S ₁	MO107 → MO108	2.0041 eV (618.64 nm)	0.3336
T ₂	MO107 → MO109	2.3315 eV (531.79 nm)	0.0000
T ₃	MO107 → MO109 (26%) MO107 → MO110 (74%)	2.7111 eV (457.32 nm)	0.0000
S ₂	MO106 → MO108 (14%) MO107 → MO109 (86%)	2.8477 eV (435.39 nm)	0.0259
S ₃	MO105 → MO108 (40%) MO106 → MO108 (47%) MO107 → MO110 (13%)	3.2081 eV (386.47 nm)	0.0272

HOMO: MO107, LUMO: MO108

Table S21 Calculated excited states of S₁-optimized geometry of **1CF₃** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO123 → MO124	1.3870 eV (893.87 nm)	0.0000
S ₁	MO123 → MO124	2.0556 eV (603.15 nm)	0.3341
T ₂	MO123 → MO125 (75%)	2.3677 eV (523.66 nm)	0.0000
	MO123 → MO126 (13%)		
	MO123 → MO127 (12%)		
T ₃	MO123 → MO126	2.7154 eV (456.59 nm)	0.0000
S ₂	MO123 → MO125	2.9208 eV (424.49 nm)	0.0337
S ₃	MO121 → MO124 (33%)	3.2305 eV (383.80 nm)	0.0716
	MO122 → MO124 (32%)		
	MO123 → MO126 (35%)		

HOMO: MO123, LUMO: MO124

Table S22 Calculated excited states of S₁-optimized geometry of **1CN** in dichloromethane.

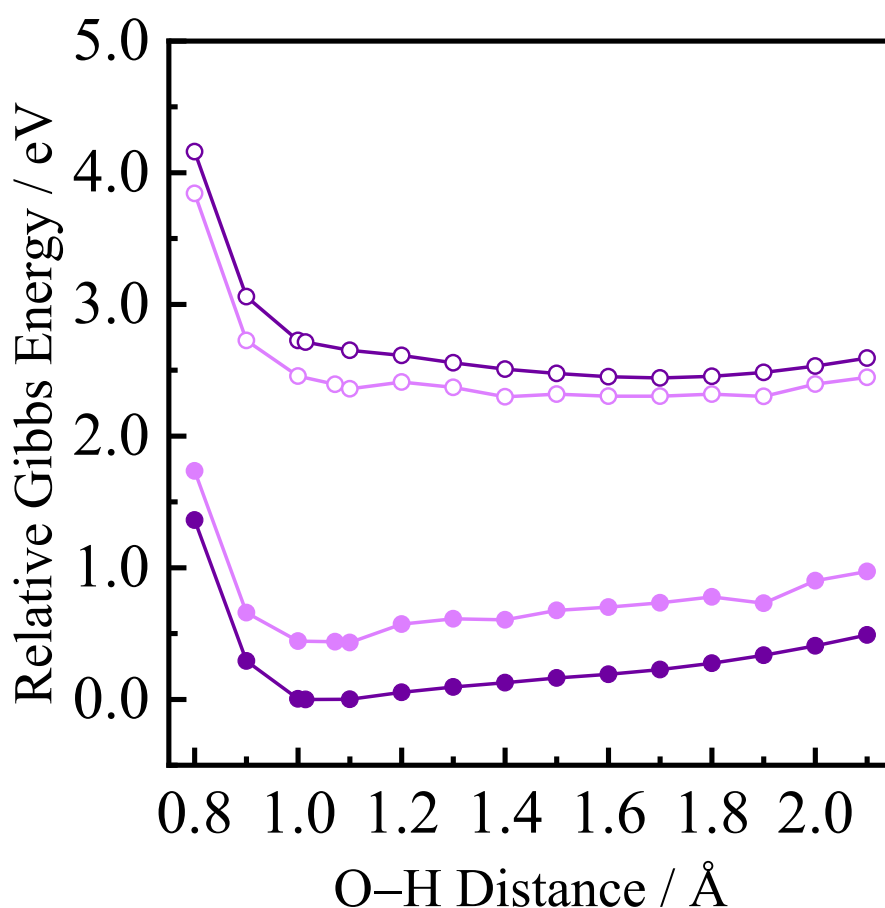
Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO103 → MO104	1.4129 eV (877.49 nm)	0.0000
S ₁	MO103 → MO104	2.0527 eV (604.00 nm)	0.3848
T ₂	MO103 → MO105	2.3358 eV (530.81 nm)	0.0000
T ₃	MO102 → MO104 (12%)	2.5387 eV (488.38 nm)	0.0000
	MO102 → MO106 (10%)		
	MO103 → MO105 (34%)		
	MO103 → MO106 (44%)		
S ₂	MO103 → MO105 (81%)	2.9243 eV (423.98 nm)	0.3183
	MO103 → MO106 (19%)		
S ₃	MO103 → MO106 (58%)	2.9824 eV (415.72 nm)	0.1591
	MO103 → MO107 (42%)		

HOMO: MO103, LUMO: MO104

Table S23 Calculated excited states of S₁-optimized geometry of **1NO₂** in dichloromethane.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO113 → MO114	1.7545 eV (706.68 nm)	0.0000
S ₁	MO113 → MO114	1.7623 eV (703.52 nm)	0.0004
T ₂	MO113 → MO115 (74%) MO113 → MO116 (26%)	2.0320 eV (610.17 nm)	0.0000
S ₂	MO113 → MO115	2.4380 eV (508.54 nm)	0.3602
T ₃	MO111 → MO115 (13%) MO111 → MO116 (14%) MO112 → MO116 (11%) MO113 → MO116 (62%)	2.5164 eV (492.70 nm)	0.0000
S ₃	MO112 → MO114	2.9235 eV (424.10 nm)	0.0006

HOMO: MO113, LUMO: MO114

**Fig. S29** Dependences of the O–H distances on the relative Gibbs energies of ground (close circles) and S₁ states (open circles) of the ground- (deep color) and S₁-state-optimized (pale color) **1NO₂** in dichloromethane by the (TD-)DFT calculations (B3LYP/6-31G(d,p) level).

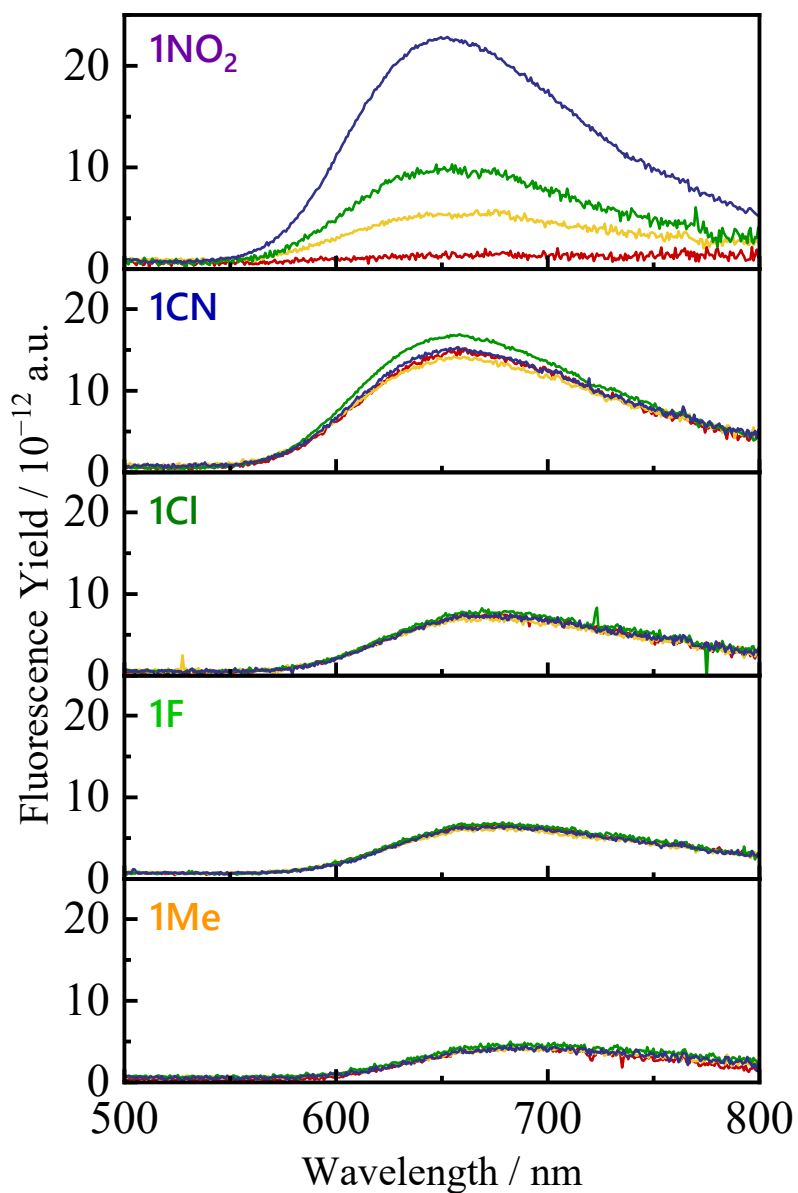


Fig. S30 Fluorescence spectra of several **1R** in ethyl acetate (dark blue), dichloromethane (green), acetone (yellow) and DMF (red).

Table S24 Solvent dependences of Φ_f of **1R**.

Derivative	Ethyl Acetate	Acetone	DMF
1Me	0.006	0.006	0.006
1F	0.01	0.01	0.01
1Cl	0.01	0.01	0.01
1CN	0.02	0.02	0.02
1NO₂	0.04	0.01	0.004

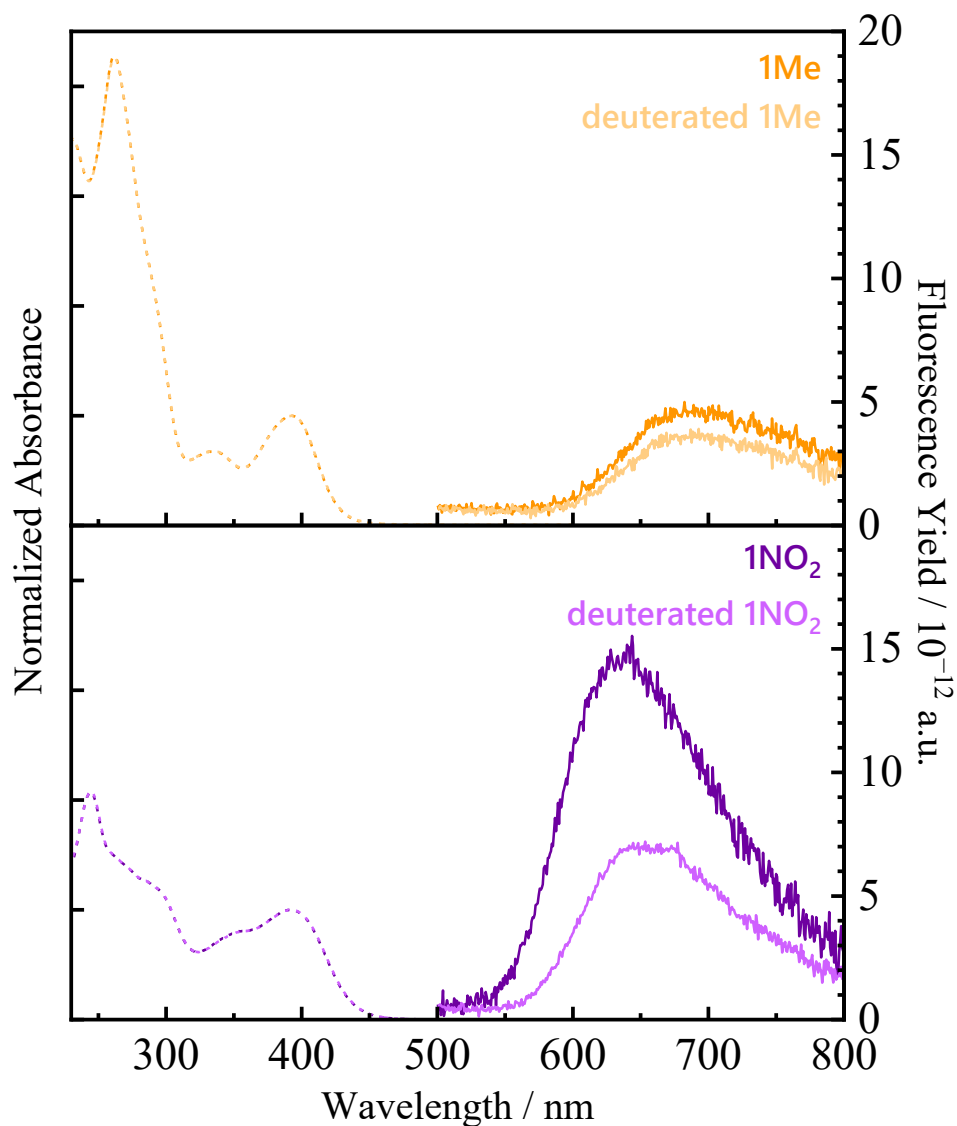


Fig. S31 Absorption (broken curves) and fluorescence spectra (solid curves, $\lambda_{\text{ex}} = 390$ or 400 nm) of **1Me** (top, $\lambda_{\text{ex}} = 390$ nm) and **1NO₂** (bottom, $\lambda_{\text{ex}} = 400$ nm) before/after deuteration in dichloromethane at room temperature.

Table S25 Photophysical properties of **1Me** and **1NO₂** before/after deuteration in dichloromethane at room temperature.

Derivative	Φ_{f}	$\tau_{\text{f}}^{\text{c}} / \text{ns}$	$k_{\text{f}}^{\text{d}} / 10^7 \text{ s}^{-1}$	$k_{\text{d}}^{\text{d}} / 10^9 \text{ s}^{-1}$
1Me	0.007 ^a	0.17	4.0	5.8
deuterated 1Me	0.007 ^a	0.12	5.4	8.1
1NO₂	0.02 ^b	0.21	8.0	4.8
deuterated 1NO₂	0.01 ^b	0.16	6.9	6.3

a) $\lambda_{\text{ex}} = 390$ nm b) $\lambda_{\text{ex}} = 400$ nm c) $\lambda_{\text{ex}} = 395$ nm d) $\Phi_{\text{f}} = k_{\text{f}}\tau_{\text{f}} = k_{\text{f}} / (k_{\text{f}} + k_{\text{d}})$

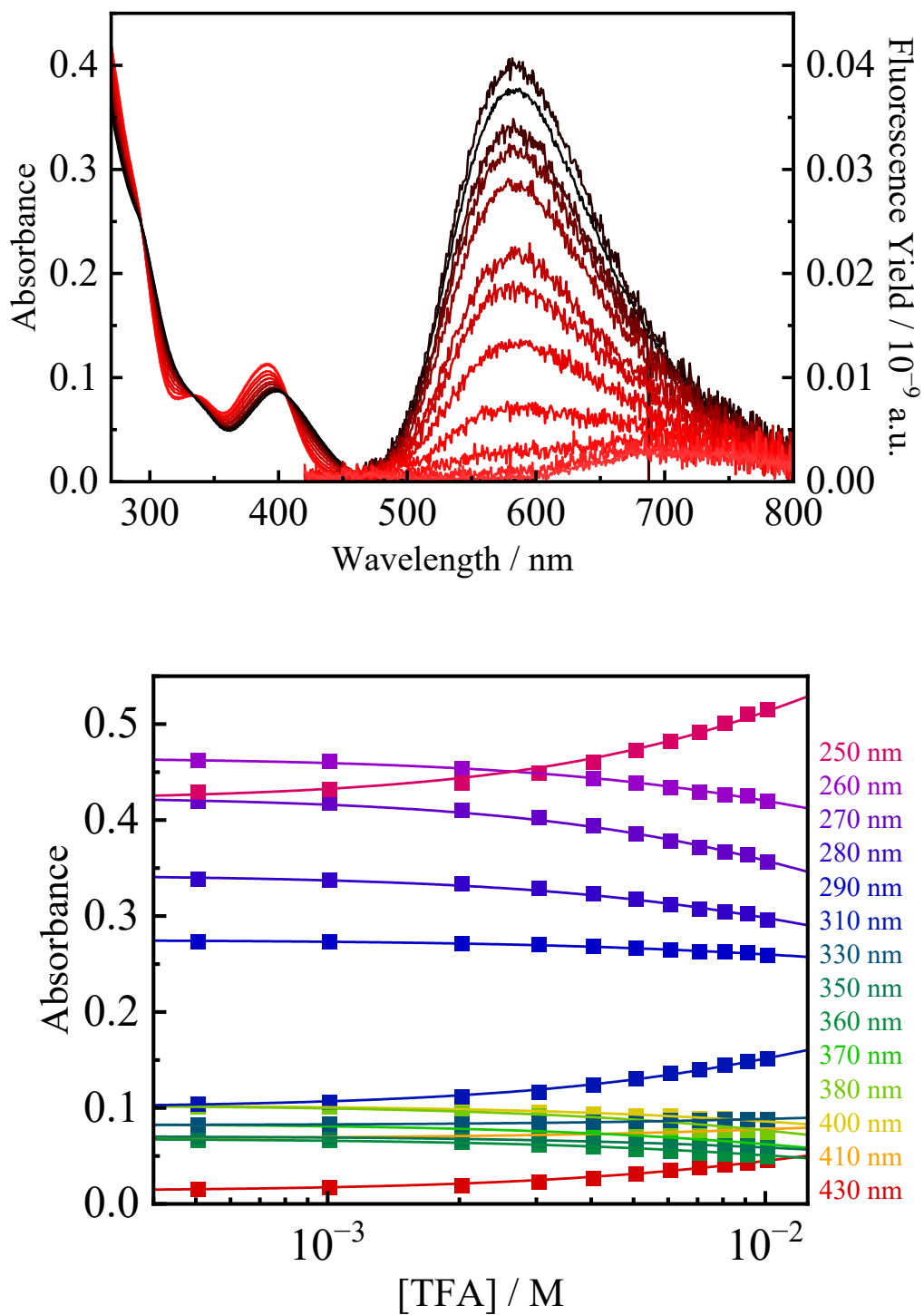


Fig. S32 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 410$ nm) and TFA titrations of the absorbances (bottom) of **1OMe** in acetonitrile ($[\text{1OMe}] = 1.06 \times 10^{-5}$ M, red $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $\text{p}K_{\text{b}} = -1.54 \pm 0.09$.

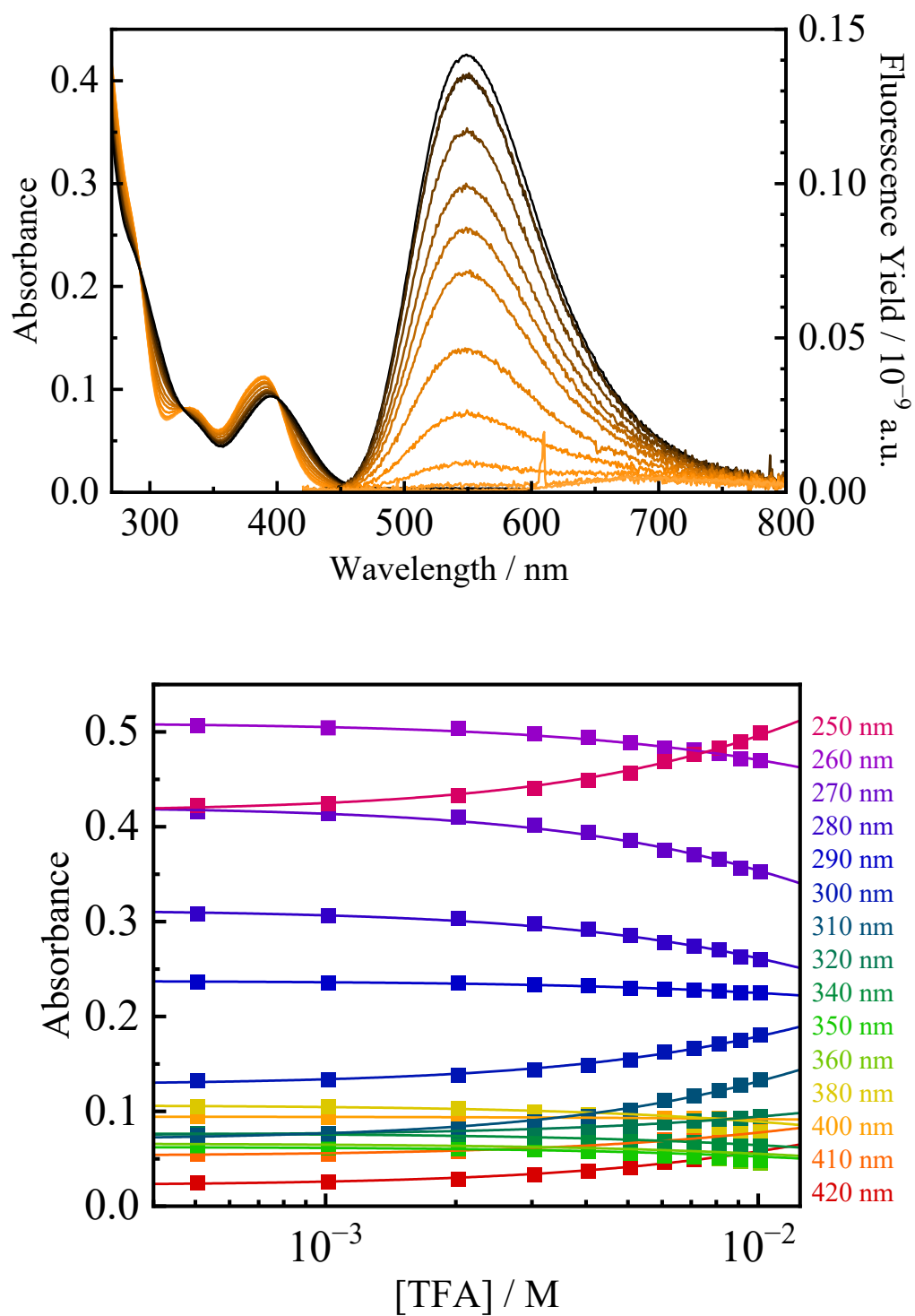


Fig. S33 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 400$ nm) and TFA titrations of the absorbances (bottom) of **1Me** in acetonitrile ($[\mathbf{1Me}] = 1.06 \times 10^{-5}$ M, orange $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $\text{p}K_{\text{b}} = -1.29 \pm 0.16$.

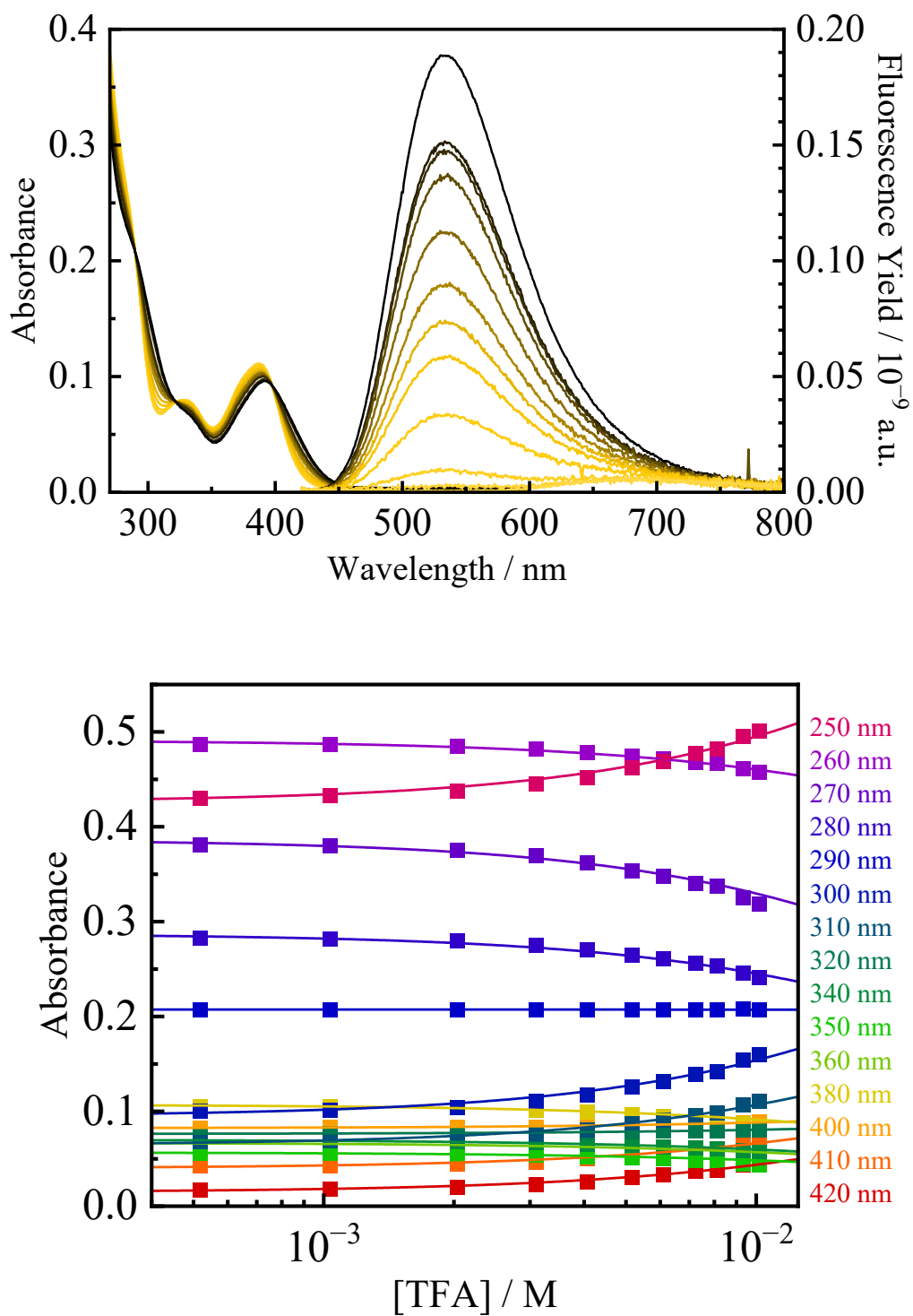


Fig. S34 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 400$ nm) and TFA titrations of the absorbances (bottom) of **1H** in acetonitrile ($[\mathbf{1H}] = 1.04 \times 10^{-5}$ M, yellow $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $\text{p}K_{\text{b}} = -1.24 \pm 0.17$.

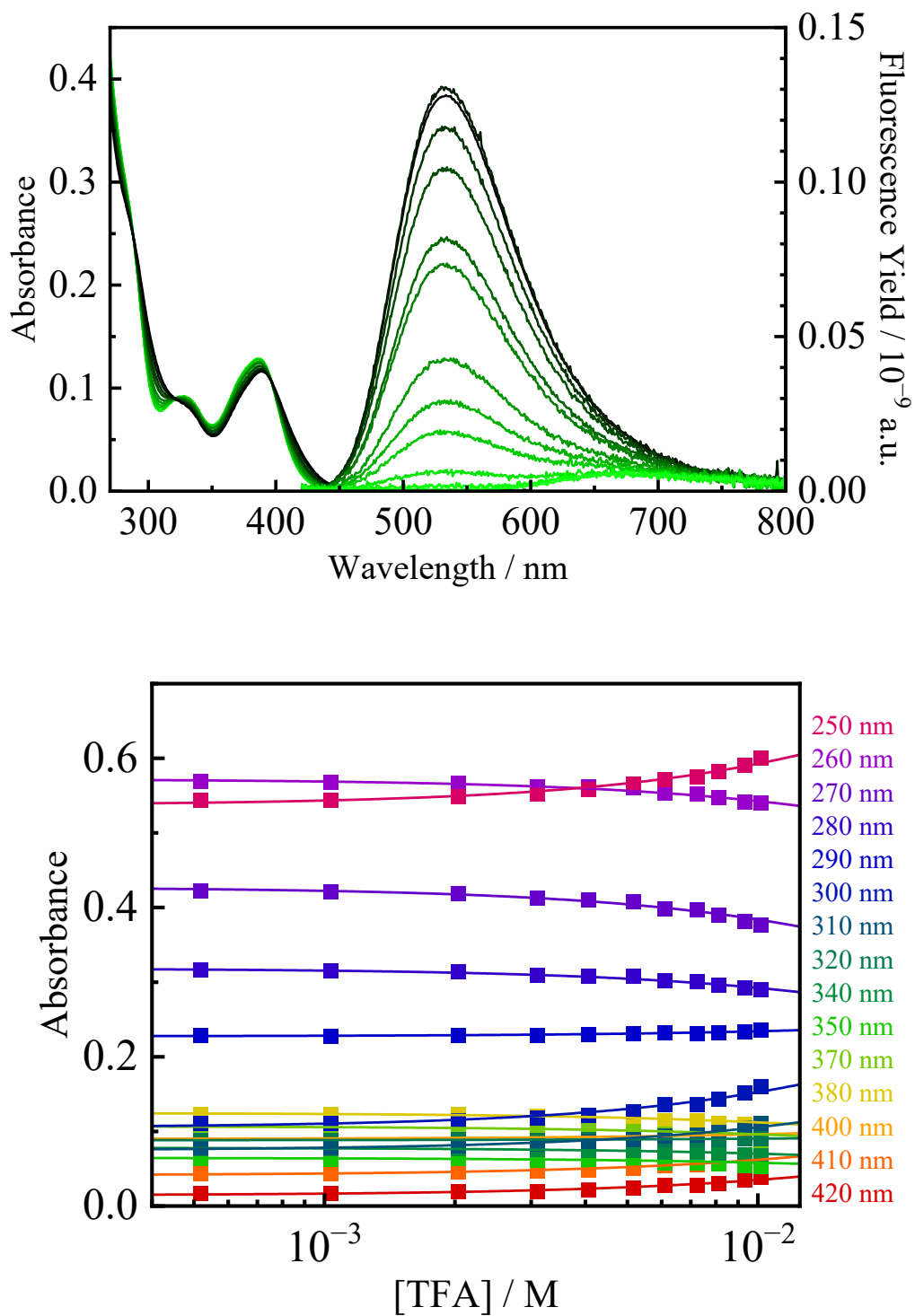


Fig. S35 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 400$ nm) and TFA titrations of the absorbances (bottom) of **1F** in acetonitrile ($[\mathbf{1F}] = 1.03 \times 10^{-5}$ M, light green $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $\text{p}K_{\text{b}} = -1.05 \pm 0.33$.

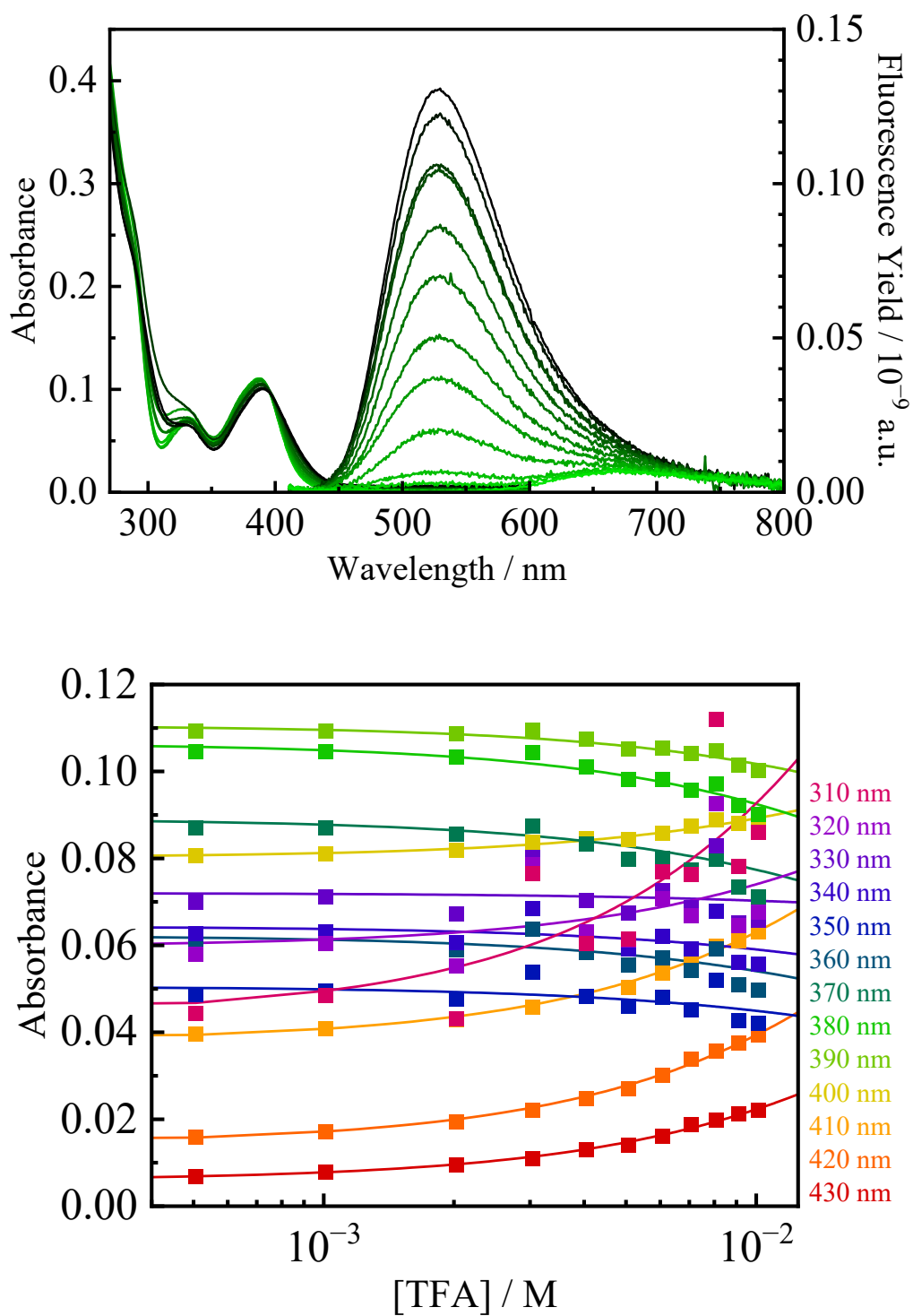


Fig. S36 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 395$ nm) and TFA titrations of the absorbances (bottom) of **1Cl** in acetonitrile ($[\mathbf{1Cl}] = 1.00 \times 10^{-5}$ M, green $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $pK_b = -1.18 \pm 0.78$.

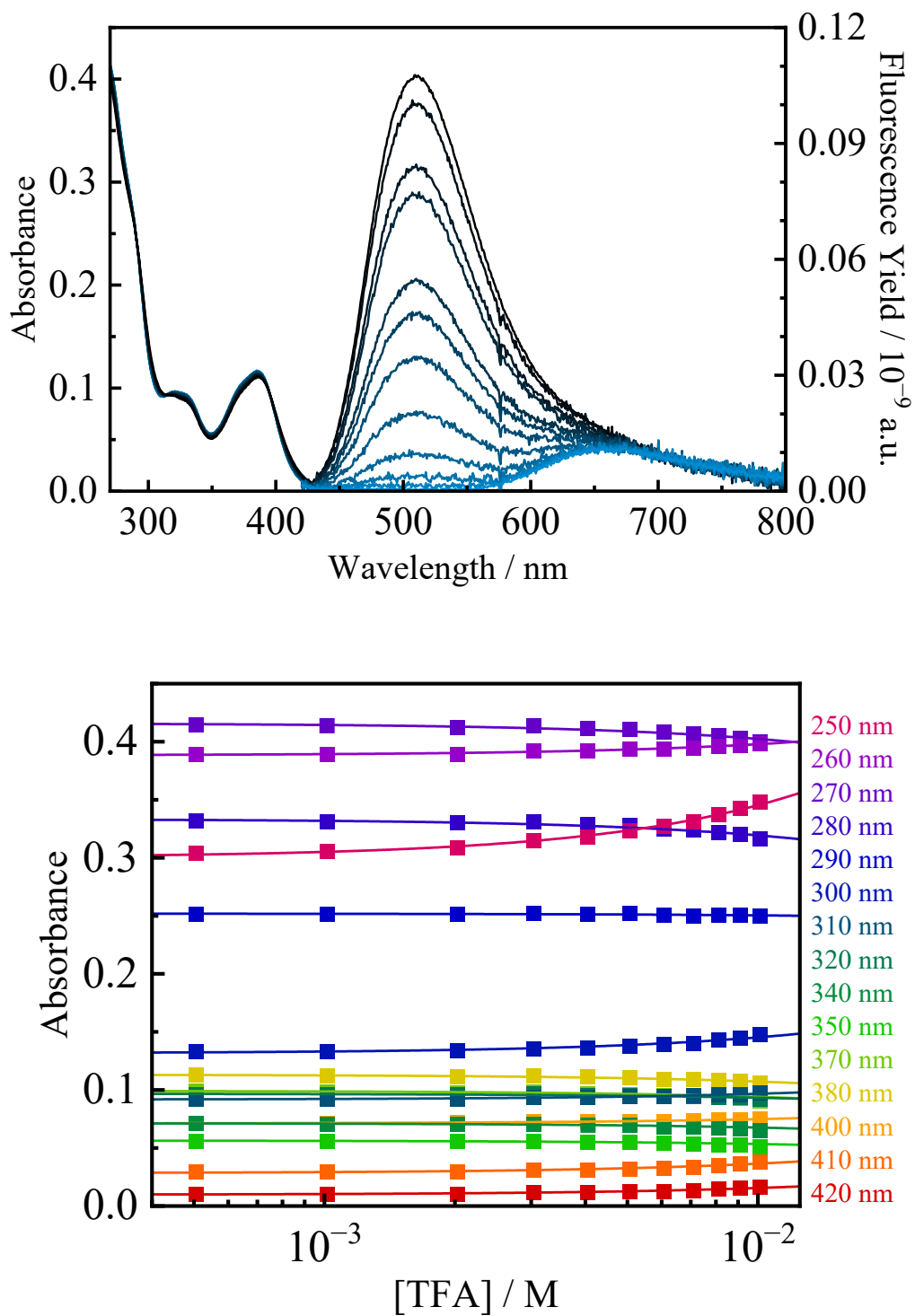


Fig. S37 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 400$ nm) and TFA titrations of the absorbances (bottom) of $1CF_3$ in acetonitrile ($[1CF_3] = 1.03 \times 10^{-5}$ M, light blue $\rightarrow$ black: $[TFA] = 0$ –10 mM). The curves represent global fits using equation (1) and $pK_b = -0.743 \pm 0.487$.

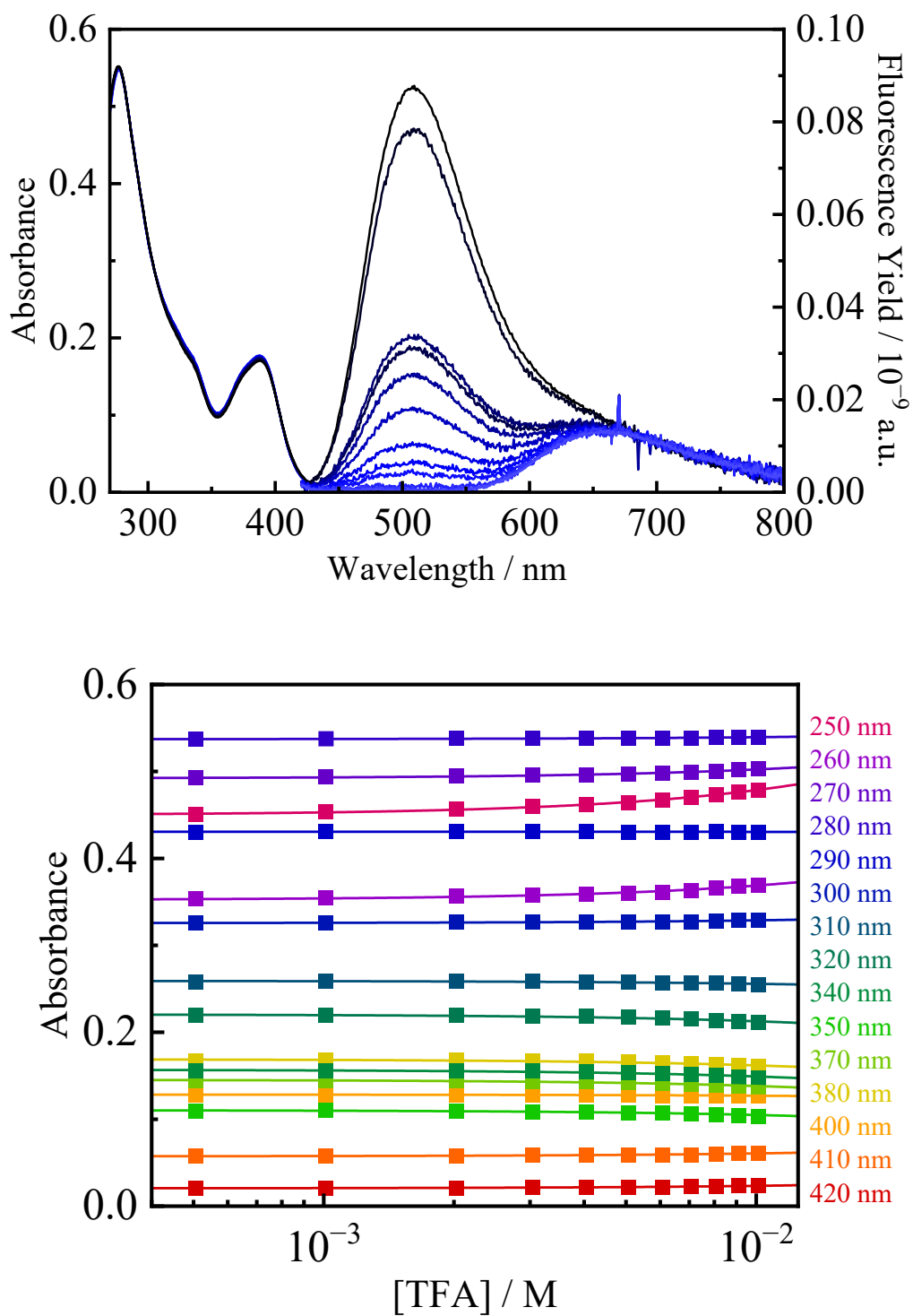


Fig. S38 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 400$ nm) and TFA titrations of the absorbances (bottom) of **1CN** in acetonitrile ($[\text{1CN}] = 1.04 \times 10^{-5}$ M, blue $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $\text{p}K_{\text{b}} = -0.715 \pm 0.507$.

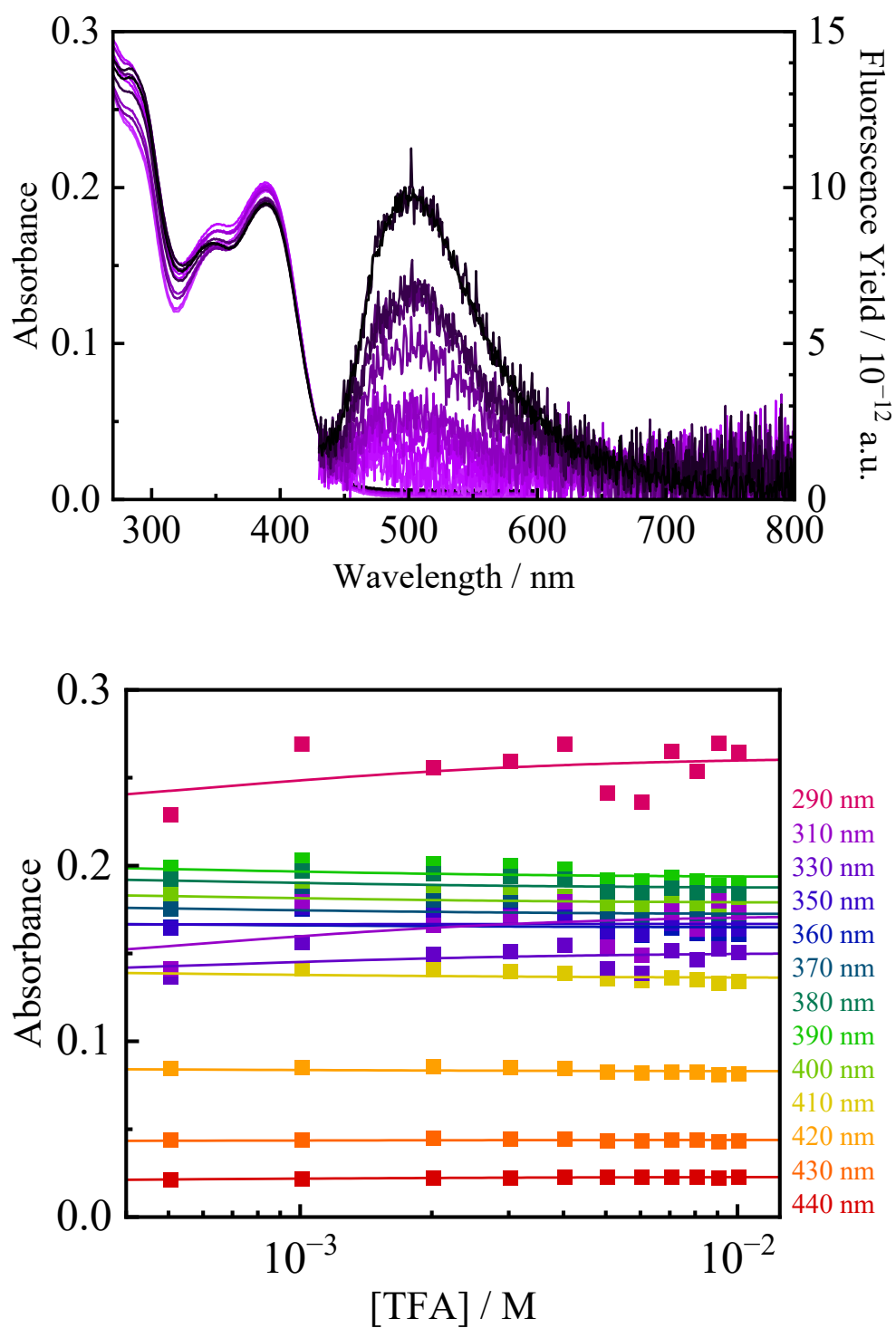


Fig. S39 TFA-concentration dependences of absorption/fluorescence spectra (top, $\lambda_{\text{ex}} = 415$ nm) and TFA titrations of the absorbances (bottom) of **1NO₂** in acetonitrile ($[\mathbf{1NO_2}] = 1.00 \times 10^{-5}$ M, purple $\rightarrow$ black: $[\text{TFA}] = 0\text{--}10$ mM). The curves represent global fits using equation (1) and $\text{p}K_{\text{b}} = -3.21 \pm 0.21$.

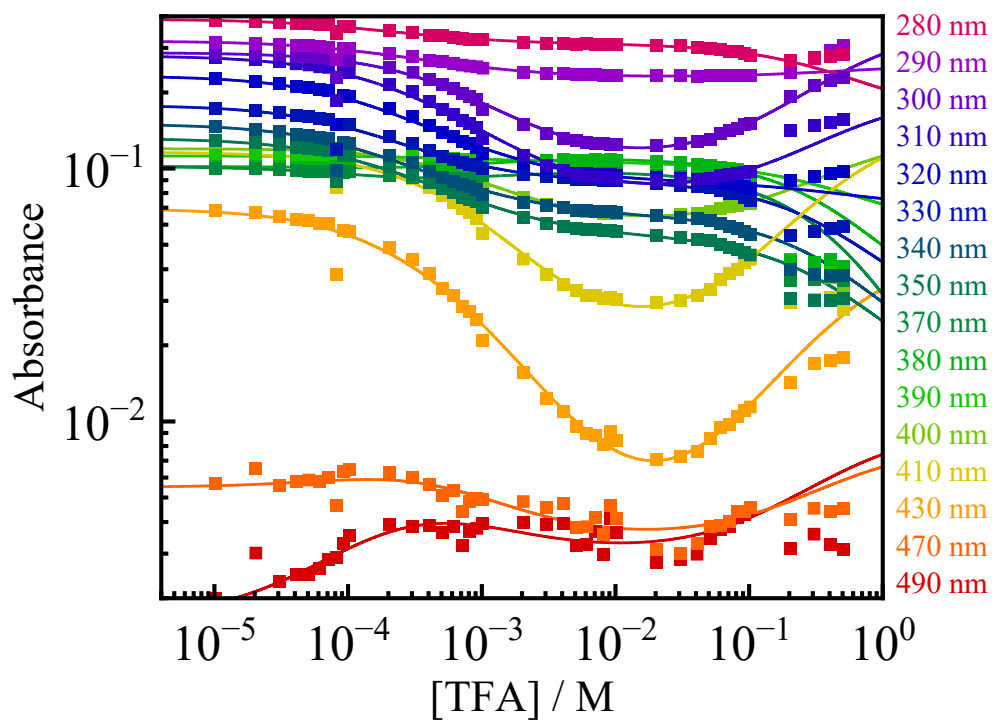


Fig. S40 TFA-concentration dependences of the absorbances of **1NMe₂** ($[1NMe_2] = 1.02 \times 10^{-5}$ M) in acetonitrile. The curves represent a global fit using equation (2) and $pK_{b1} = -3.89 \pm 0.07$, $pK_{b2} = -3.04 \pm 0.06$, $pK_{b3} = -0.323 \pm 0.622$.

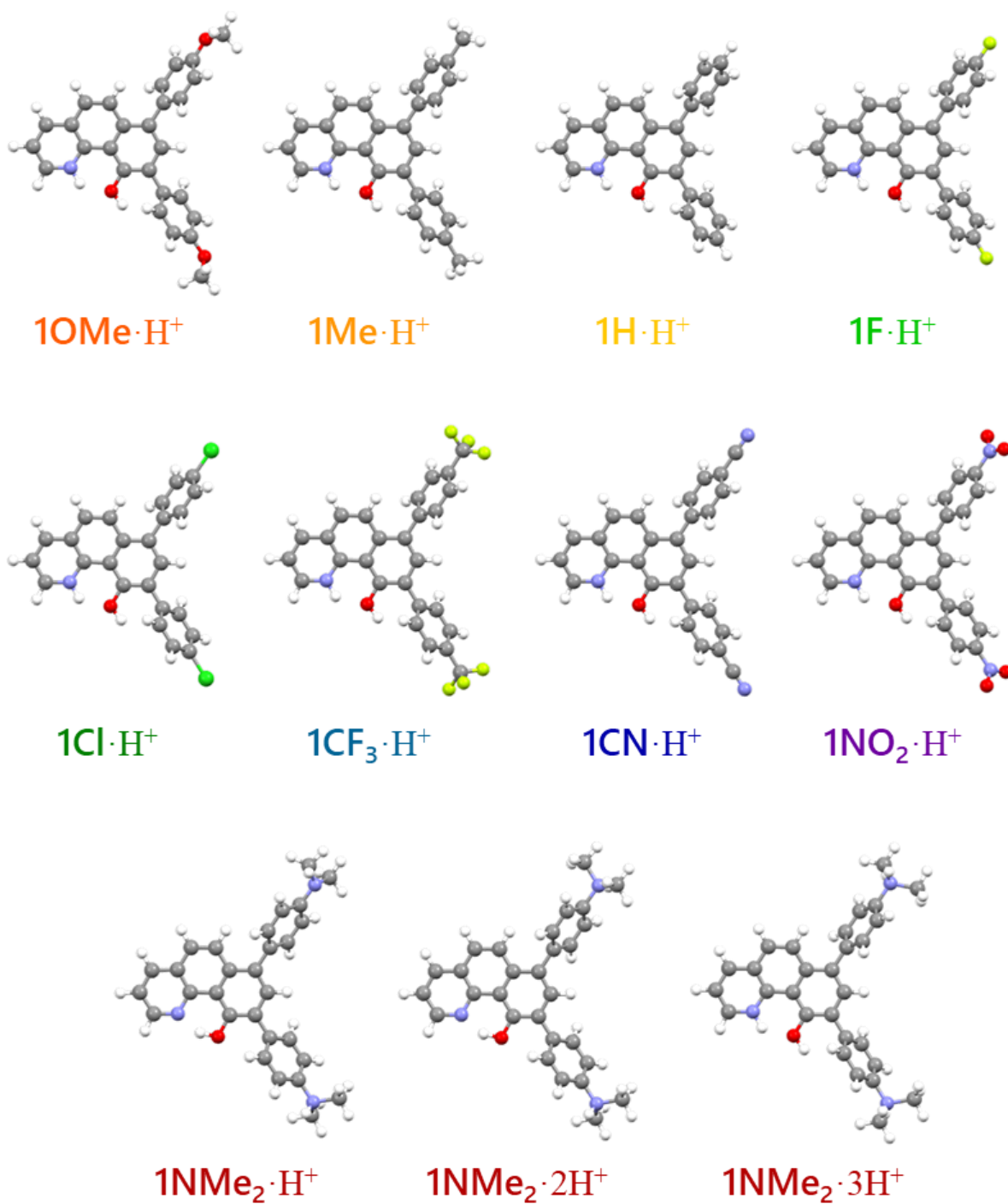


Fig. S41 Optimized geometries of **1R**·H⁺ in the ground states in acetonitrile.

Table S26 Thermodynamic parameters of **1R** and its protonated forms in acetonitrile at 298.15 K.

Compound	G_{tot}^a / a.u.	ΔG_{tot}^b / a.u.	K_b	pK_b
HBqNO ₂	-1502.044554	+0.006238	0.001351	2.869
HBqNO ₂ ·H ⁺	-1502.465088			
HBqCN	-1277.453456	+0.004946	0.005309	2.275
HBqCN·H ⁺	-1277.875282			
HBqCF ₃	-1767.208804	+0.003674	0.02042	1.690
HBqCF ₃ ·H ⁺	-1767.631902			
HBqCl	-2012.171437	+0.002396	0.07905	1.102
HBqCl·H ⁺	-2012.595813			
HBqF	-1291.459088	+0.001839	0.1426	0.8459
HBqF·H ⁺	-1291.884021			
HBqH	-1092.901199	+0.000891	0.389	0.410
HBqH·H ⁺	-1093.327080			
HBqMe	-1171.508753	+0.000097	0.90	0.0045
HBqMe·H ⁺	-1171.935428			
HBqOMe	-1321.960675	-0.000263	1.32	-0.121
HBqOMe·H ⁺	-1322.387710			
HBqNMe ₂	-1360.778667	-0.000264	1.32	-0.121
HBqNMe·H ⁺	-1361.205703			
		+0.002139	0.1038	0.9839
HBqNMe·2H ⁺	-1361.630336	+0.008456	0.0001290	3.889
HBqNMe·3H ⁺	-1362.048652			

^a G_{tot} : sum of total electronic energy (E_0) and thermal correction to Gibbs free energy (G_{corr}).

^b calculated using $G_{\text{tot}}(\text{CF}_3\text{COOH}) = -526.966398$ a.u. and $G_{\text{tot}}(\text{CF}_3\text{COO}^-) = -526.539626$ a.u.

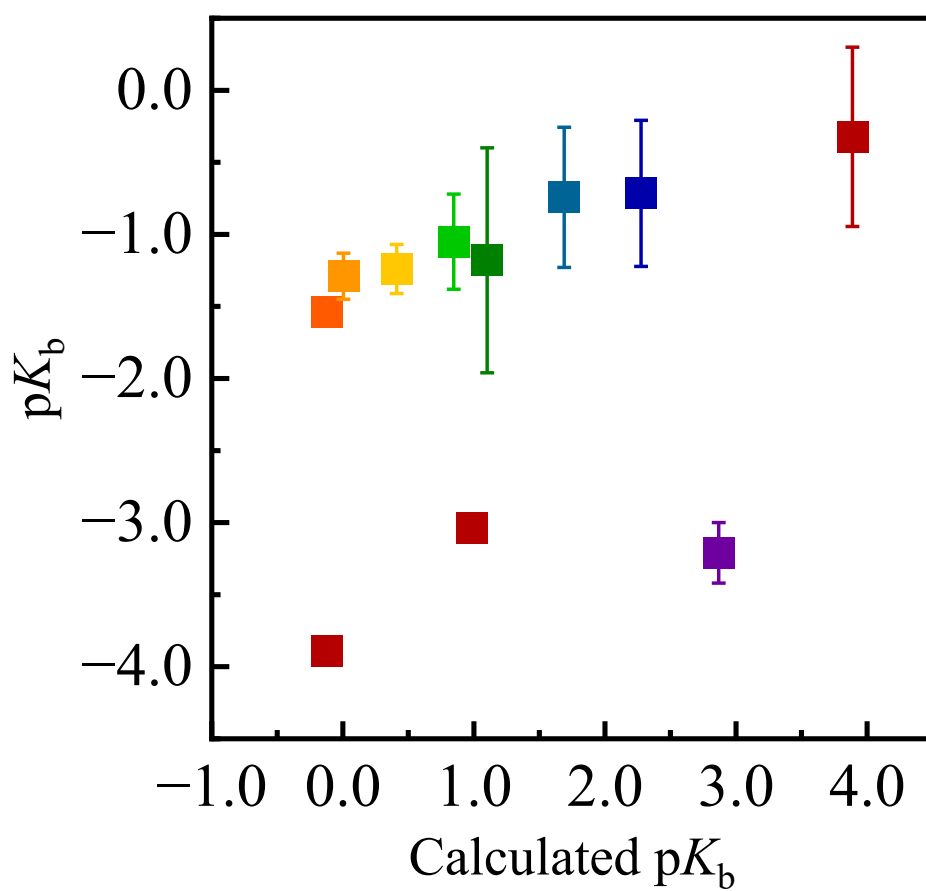


Fig. S42 Plot for comparison between calculated and experimentally-determined pK_b of **1R** in acetonitrile. The colors correspond to those indicated in Fig. 1.

Table S27 Calculated excited states of **1NMe₂** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO115 → MO116	2.8013 eV (442.60 nm)	0.2757
S ₂	MO114 → MO116	3.0265 eV (409.66 nm)	0.0886
S ₃	MO115 → MO117	3.2320 eV (383.61 nm)	0.0585
S ₄	MO113 → MO116 (14%) MO114 → MO117 (86%)	3.4834 eV (355.93 nm)	0.0829
S ₅	MO112 → MO117 (17%) MO113 → MO116 (83%)	3.6770 eV (337.19 nm)	0.1372
S ₆	MO115 → MO118	3.9352 eV (315.07 nm)	0.2933
S ₇	MO113 → MO117	4.0718 eV (304.49 nm)	0.0121
S ₈	MO113 → MO117 (12%) MO114 → MO118 (9%) MO114 → MO120 (13%) MO115 → MO119 (39%) MO115 → MO120 (27%)	4.1222 eV (300.77 nm)	0.0044
S ₉	MO114 → MO119 (26%) MO114 → MO120 (16%) MO115 → MO120 (58%)	4.1422 eV (299.32 nm)	0.0514
S ₁₀	MO114 → MO118 (67%) MO114 → MO119 (15%) MO115 → MO120 (18%)	4.1764 eV (296.87 nm)	0.1999
S ₁₁	MO112 → MO116 (66%) MO113 → MO117 (34%)	4.3292 eV (286.39 nm)	0.1200
S ₁₂	MO112 → MO116 (15%) MO113 → MO117 (11%) MO114 → MO119 (30%) MO115 → MO119 (21%) MO115 → MO121 (23%)	4.3898 eV (282.43 nm)	0.1405
S ₁₃	MO115 → MO121	4.4108 eV (281.09 nm)	0.2976
S ₁₄	MO114 → MO120 (83%) MO114 → MO123 (17%)	4.4957 eV (275.79 nm)	0.0166
S ₁₅	MO114 → MO120 (39%) MO115 → MO122 (61%)	4.5205 eV (274.27 nm)	0.0185
S ₁₆	MO108 → MO116 (35%) MO111 → MO116 (41%) MO112 → MO117 (10%) MO114 → MO120 (14%)	4.5690 eV (271.36 nm)	0.0183
S ₁₇	MO114 → MO121 (30%) MO115 → MO121 (13%) MO115 → MO122 (11%) MO115 → MO123 (46%)	4.5985 eV (269.62 nm)	0.0334
S ₁₈	MO108 → MO116 (13%) MO110 → MO116 (43%) MO111 → MO116 (20%) MO113 → MO118 (10%) MO114 → MO121 (14%)	4.6148 eV (268.67 nm)	0.1153

S ₁₉	MO109 → MO116 (19%) MO110 → MO116 (37%) MO111 → MO116 (12%) MO112 → MO117 (32%)	4.6475 eV (266.78 nm)	0.2573
S ₂₀	MO108 → MO116 (78%) MO110 → MO116 (22%)	4.6957 eV (264.04 nm)	0.0472
S ₂₁	MO109 → MO116 (16%) MO112 → MO117 (14%) MO114 → MO121 (51%) MO114 → MO122 (19%)	4.7150 eV (262.95 nm)	0.1755
S ₂₂	MO109 → MO116 (9%) MO113 → MO118 (41%) MO114 → MO121 (14%) MO114 → MO122 (12%) MO114 → MO123 (11%) MO115 → MO123 (13%)	4.7901 eV (258.83 nm)	0.0569
S ₂₃	MO109 → MO116 (12%) MO113 → MO118 (33%) MO115 → MO124 (44%) MO115 → MO128 (11%)	4.7991 eV (258.35 nm)	0.0076
S ₂₄	MO109 → MO116 (33%) MO109 → MO117 (13%) MO113 → MO118 (16%) MO115 → MO125 (38%)	4.8922 eV (253.43 nm)	0.0243
S ₂₅	MO109 → MO116 (14%) MO114 → MO122 (22%) MO114 → MO123 (9%) MO115 → MO122 (6%) MO115 → MO123 (15%) MO115 → MO124 (15%) MO115 → MO125 (19%)	4.9414 eV (250.91 nm)	0.0530
S ₂₆	MO109 → MO116 (9%) MO113 → MO119 (17%) MO114 → MO123 (25%) MO114 → MO124 (11%) MO115 → MO124 (17%) MO115 → MO126 (13%) MO115 → MO129 (8%)	4.9936 eV (248.29 nm)	0.0561
S ₂₇	MO109 → MO116 (20%) MO113 → MO119 (62%) MO114 → MO122 (18%)	5.0034 eV (247.80 nm)	0.1312
S ₂₈	MO108 → MO117 (33%) MO111 → MO117 (67%)	5.0300 eV (246.49 nm)	0.0053
S ₂₉	MO109 → MO116 (17%) MO114 → MO124 (65%) MO114 → MO127 (18%)	5.0622 eV (244.92 nm)	0.0515

S ₃₀	MO110 → MO117 (26%)	5.0765 eV (244.23 nm)	0.1180
	MO111 → MO117 (8%)		
	MO112 → MO117 (14%)		
	MO113 → MO118 (9%)		
	MO113 → MO119 (16%)		
	MO114 → MO124 (14%)		
	MO115 → MO125 (13%)		
HOMO: MO115, LUMO: MO116			

Table S28 Calculated excited states of **1NMe₂·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO115 → MO116	2.7790 eV (446.15 nm)	0.1270
S ₂	MO115 → MO117 (86%) MO115 → MO118 (14%)	3.2883 eV (377.04 nm)	0.0601
S ₃	MO114 → MO116	3.4348 eV (360.97 nm)	0.2656
S ₄	MO115 → MO118	3.5758 eV (346.73 nm)	0.0527
S ₅	MO115 → MO119	3.7937 eV (326.82 nm)	0.0005
S ₆	MO113 → MO116 (31%) MO114 → MO117 (59%) MO114 → MO118 (10%)	3.8743 eV (320.01 nm)	0.0422
S ₇	MO115 → MO120 (54%) MO115 → MO121 (46%)	4.1525 eV (298.58 nm)	0.0284
S ₈	MO113 → MO116 (29%) MO114 → MO118 (48%) MO115 → MO118 (9%) MO115 → MO120 (14%)	4.1684 eV (297.44 nm)	0.0027
S ₉	MO113 → MO116 (70%) MO115 → MO121 (30%)	4.2798 eV (289.70 nm)	0.4667
S ₁₀	MO114 → MO118 (24%) MO115 → MO121 (76%)	4.3329 eV (286.15 nm)	0.2463
S ₁₁	MO114 → MO119 (84%) MO115 → MO119 (16%)	4.3899 eV (282.43 nm)	0.0100
S ₁₂	MO112 → MO116 (87%) MO115 → MO123 (13%)	4.4775 eV (276.90 nm)	0.0163
S ₁₃	MO111 → MO116 (16%) MO115 → MO122 (20%) MO115 → MO123 (46%) MO115 → MO124 (18%)	4.5208 eV (274.25 nm)	0.2713
S ₁₄	MO109 → MO116 (11%) MO110 → MO116 (11%) MO113 → MO117 (20%) MO115 → MO120 (6%) MO115 → MO122 (12%) MO115 → MO123 (10%) MO115 → MO124 (22%) MO115 → MO125 (8%)	4.5907 eV (270.08 nm)	0.1977
S ₁₅	MO109 → MO116 (47%) MO110 → MO116 (53%)	4.6203 eV (268.34 nm)	0.0843
S ₁₆	MO113 → MO117 (27%) MO115 → MO123 (73%)	4.6761 eV (265.15 nm)	0.1841
S ₁₇	MO113 → MO118 (19%) MO115 → MO122 (81%)	4.7596 eV (260.49 nm)	0.0299
S ₁₈	MO108 → MO116 (13%) MO111 → MO117 (15%) MO113 → MO118 (54%) MO115 → MO124 (18%)	4.7716 eV (259.84 nm)	0.1809

S ₁₉	MO114 → MO120 (45%) MO114 → MO121 (32%) MO115 → MO122 (10%) MO115 → MO125 (13%)	4.8814 eV (253.99 nm)	0.0064
S ₂₀	MO111 → MO116 (18%) MO113 → MO118 (21%) MO114 → MO120 (40%) MO115 → MO124 (21%)	4.9462 eV (250.67 nm)	0.0335
S ₂₁	MO111 → MO116 (18%) MO113 → MO118 (22%) MO114 → MO124 (12%) MO115 → MO125 (48%)	4.9677 eV (249.58 nm)	0.0840
S ₂₂	MO109 → MO116 (10%) MO111 → MO116 (11%) MO112 → MO117 (66%) MO112 → MO118 (13%)	4.9937 eV (248.28 nm)	0.0091
S ₂₃	MO109 → MO116 (34%) MO110 → MO118 (15%) MO113 → MO119 (51%)	5.0046 eV (247.74 nm)	0.0030
S ₂₄	MO111 → MO116 (17%) MO113 → MO117 (12%) MO113 → MO118 (21%) MO114 → MO121 (15%) MO114 → MO123 (9%) MO115 → MO126 (26%)	5.0775 eV (244.18 nm)	0.1784
S ₂₅	MO109 → MO117 (39%) MO110 → MO117 (40%) MO110 → MO118 (10%) MO113 → MO119 (11%)	5.1014 eV (243.04 nm)	0.0036
S ₂₆	MO114 → MO120 (12%) MO115 → MO126 (52%) MO115 → MO128 (13%) MO115 → MO130 (23%)	5.1123 eV (242.52 nm)	0.0986
S ₂₇	MO107 → MO116 (7%) MO108 → MO116 (30%) MO109 → MO116 (13%) MO109 → MO117 (10%) MO110 → MO117 (13%) MO111 → MO117 (27%)	5.1627 eV (240.15 nm)	0.1005
S ₂₈	MO110 → MO116 (10%) MO111 → MO117 (8%) MO113 → MO119 (14%) MO114 → MO125 (10%) MO115 → MO126 (8%) MO115 → MO127 (39%) MO115 → MO128 (11%)	5.2277 eV (237.17 nm)	0.0084
S ₂₉	MO108 → MO116 (19%) MO110 → MO116 (34%) MO113 → MO119 (47%)	5.2309 eV (237.02 nm)	0.0161

S ₃₀	MO108 → MO116 (15%) MO114 → MO122 (22%) MO114 → MO123 (63%)	5.2521 eV (236.07 nm)	0.1904
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HOMO: MO115, LUMO: MO116

Table S29 Calculated excited states of **1NMe₂·2H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO114 → MO117 (15%) MO115 → MO116 (85%)	3.2785 eV (378.18 nm)	0.3820
S ₂	MO115 → MO117	3.7124 eV (333.97 nm)	0.0696
S ₃	MO115 → MO118 (83%) MO115 → MO119 (17%)	3.9781 eV (311.67 nm)	0.1861
S ₄	MO114 → MO116 (21%) MO115 → MO117 (19%) MO115 → MO119 (60%)	4.0745 eV (304.29 nm)	0.0459
S ₅	MO115 → MO120	4.1701 eV (297.32 nm)	0.0172
S ₆	MO115 → MO121	4.2432 eV (292.19 nm)	0.0079
S ₇	MO114 → MO116 (65%) MO115 → MO117 (21%) MO115 → MO120 (14%)	4.2579 eV (291.19 nm)	0.3340
S ₈	MO113 → MO116 (41%) MO114 → MO117 (59%)	4.5428 eV (272.92 nm)	0.2882
S ₉	MO109 → MO116	4.6024 eV (269.39 nm)	0.0306
S ₁₀	MO113 → MO116 (62%) MO114 → MO118 (22%) MO115 → MO122 (16%)	4.7499 eV (261.03 nm)	0.1979
S ₁₁	MO114 → MO118	4.9150 eV (252.26 nm)	0.1082
S ₁₂	MO111 → MO116 (31%) MO113 → MO116 (12%) MO113 → MO117 (9%) MO114 → MO118 (32%) MO114 → MO119 (16%)	4.9450 eV (250.73 nm)	0.0461
S ₁₃	MO109 → MO116 (15%) MO112 → MO116 (23%) MO113 → MO121 (7%) MO114 → MO118 (8%) MO114 → MO119 (13%) MO114 → MO121 (18%) MO115 → MO122 (16%)	4.9868 eV (248.62 nm)	0.0174
S ₁₄	MO114 → MO117 (22%) MO114 → MO119 (20%) MO115 → MO122 (58%)	5.0073 eV (247.60 nm)	0.0540
S ₁₅	MO109 → MO116 (17%) MO109 → MO117 (59%) MO113 → MO117 (24%)	5.0623 eV (244.92 nm)	0.0003
S ₁₆	MO111 → MO116 (38%) MO114 → MO120 (62%)	5.1067 eV (242.79 nm)	0.1008
S ₁₇	MO109 → MO116 (18%) MO110 → MO116 (35%) MO114 → MO120 (47%)	5.1415 eV (241.14 nm)	0.0163

S ₁₈	MO111 → MO120 (11%) MO113 → MO116 (24%) MO113 → MO117 (12%) MO114 → MO119 (40%) MO114 → MO120 (13%)	5.1666 eV (239.97 nm)	0.4054
S ₁₉	MO112 → MO117 (12%) MO113 → MO117 (59%) MO115 → MO122 (14%) MO115 → MO123 (15%)	5.1835 eV (239.19 nm)	0.1112
S ₂₀	MO114 → MO119 (11%) MO115 → MO123 (55%) MO115 → MO125 (34%)	5.1972 eV (238.56 nm)	0.0372
S ₂₁	MO110 → MO116 (20%) MO114 → MO121 (80%)	5.2448 eV (236.40 nm)	0.0075
S ₂₂	MO108 → MO116 (29%) MO110 → MO117 (16%) MO110 → MO118 (12%) MO110 → MO119 (10%) MO113 → MO117 (12%) MO113 → MO119 (11%) MO114 → MO120 (10%)	5.3305 eV (232.59 nm)	0.0147
S ₂₃	MO108 → MO116 (18%) MO110 → MO117 (10%) MO115 → MO123 (16%) MO115 → MO124 (56%)	5.3390 eV (232.22 nm)	0.0015
S ₂₄	MO108 → MO116 (43%) MO111 → MO117 (18%) MO111 → MO120 (27%) MO113 → MO117 (12%)	5.3959 eV (229.78 nm)	0.0476
S ₂₅	MO115 → MO124 (18%) MO115 → MO125 (82%)	5.4328 eV (228.21 nm)	0.0005
S ₂₆	MO111 → MO117	5.4628 eV (226.96 nm)	0.1626
S ₂₇	MO109 → MO117 (30%) MO111 → MO117 (15%) MO112 → MO117 (39%) MO113 → MO121 (16%)	5.5222 eV (224.52 nm)	0.0117
S ₂₈	MO113 → MO118 (63%) MO113 → MO119 (13%) MO115 → MO126 (24%)	5.5475 eV (223.50 nm)	0.0387
S ₂₉	MO107 → MO116 (14%) MO108 → MO116 (18%) MO115 → MO126 (68%)	5.5634 eV (222.86 nm)	0.0281
S ₃₀	MO108 → MO121 (8%) MO109 → MO116 (10%) MO109 → MO117 (12%) MO109 → MO119 (8%) MO112 → MO117 (18%) MO112 → MO118 (34%) MO114 → MO121 (10%)	5.6144 eV (220.83 nm)	0.0047

HOMO: MO115, LUMO: MO116

Table S30 Calculated excited states of **1NMe₂·3H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO114 → MO117 (14%) MO115 → MO116 (86%)	3.1134 eV (398.22 nm)	0.2147
S ₂	MO115 → MO117	3.7160 eV (333.65 nm)	0.0289
S ₃	MO114 → MO116 (55%) MO115 → MO117 (35%) MO115 → MO118 (10%)	4.0164 eV (308.70 nm)	0.2817
S ₄	MO112 → MO116 (25%) MO115 → MO118 (75%)	4.2636 eV (290.80 nm)	0.0425
S ₅	MO112 → MO116	4.3780 eV (283.20 nm)	0.0155
S ₆	MO112 → MO116 (31%) MO113 → MO116 (69%)	4.4037 eV (281.54 nm)	0.0399
S ₇	MO110 → MO116 (29%) MO115 → MO119 (45%) MO115 → MO121 (14%) MO115 → MO122 (12%)	4.5020 eV (275.40 nm)	0.1142
S ₈	MO114 → MO117 (21%) MO115 → MO119 (79%)	4.5213 eV (274.23 nm)	0.1362
S ₉	MO111 → MO116	4.5621 eV (271.77 nm)	0.0163
S ₁₀	MO110 → MO116 (16%) MO111 → MO116 (16%) MO114 → MO117 (11%) MO115 → MO119 (15%) MO115 → MO120 (42%)	4.6137 eV (268.73 nm)	0.0287
S ₁₁	MO115 → MO120 (39%) MO115 → MO121 (43%) MO115 → MO122 (18%)	4.6391 eV (267.26 nm)	0.2432
S ₁₂	MO114 → MO117 (35%) MO115 → MO121 (65%)	4.7406 eV (261.54 nm)	0.1887
S ₁₃	MO109 → MO116 (45%) MO110 → MO117 (9%) MO113 → MO117 (11%) MO114 → MO117 (9%) MO114 → MO118 (13%) MO115 → MO122 (13%)	4.9201 eV (252.00 nm)	0.1019
S ₁₄	MO109 → MO116 (31%) MO112 → MO117 (69%)	4.9763 eV (249.15 nm)	0.0603
S ₁₅	MO112 → MO117 (18%) MO113 → MO117 (70%) MO115 → MO121 (12%)	5.0135 eV (247.30 nm)	0.0074
S ₁₆	MO112 → MO117 (32%) MO114 → MO117 (21%) MO115 → MO122 (47%)	5.1096 eV (242.65 nm)	0.5153
S ₁₇	MO110 → MO120 (8%) MO111 → MO117 (38%) MO111 → MO118 (13%) MO114 → MO118 (12%) MO114 → MO119 (9%) MO114 → MO120 (11%) MO115 → MO122 (9%)	5.1230 eV (242.02 nm)	0.0548

S ₁₈	MO110 → MO117 (76%) MO111 → MO117 (24%)	5.2091 eV (238.01 nm)	0.0134
S ₁₉	MO110 → MO117 (47%) MO114 → MO118 (53%)	5.3084 eV (233.56 nm)	0.4864
S ₂₀	MO111 → MO117 (75%) MO111 → MO121 (25%)	5.3664 eV (231.04 nm)	0.0206
S ₂₁	MO112 → MO118 (13%) MO112 → MO121 (31%) MO113 → MO117 (13%) MO113 → MO119 (43%)	5.4030 eV (229.47 nm)	0.0101
S ₂₂	MO109 → MO117 (23%) MO114 → MO119 (77%)	5.4893 eV (225.87 nm)	0.0175
S ₂₃	MO109 → MO117 (66%) MO115 → MO124 (34%)	5.5801 eV (222.19 nm)	0.0630
S ₂₄	MO109 → MO117 (22%) MO112 → MO118 (45%) MO115 → MO123 (33%)	5.6430 eV (219.71 nm)	0.0428
S ₂₅	MO115 → MO123 (67%) MO115 → MO124 (33%)	5.6621 eV (218.97 nm)	0.0053
S ₂₆	MO108 → MO116 (9%) MO112 → MO118 (9%) MO112 → MO120 (9%) MO113 → MO118 (16%) MO114 → MO120 (32%) MO115 → MO123 (9%) MO115 → MO124 (16%)	5.6699 eV (218.67 nm)	0.0150
S ₂₇	MO108 → MO116 (15%) MO111 → MO118 (13%) MO113 → MO118 (60%) MO113 → MO119 (12%)	5.6866 eV (218.03 nm)	0.0128
S ₂₈	MO108 → MO116 (16%) MO110 → MO118 (16%) MO112 → MO118 (8%) MO112 → MO121 (9%) MO114 → MO121 (34%) MO115 → MO124 (17%)	5.7140 eV (216.98 nm)	0.1040
S ₂₉	MO108 → MO116 (28%) MO111 → MO118 (17%) MO112 → MO118 (33%) MO115 → MO124 (22%)	5.7194 eV (216.78 nm)	0.0115
S ₃₀	MO108 → MO116 (19%) MO110 → MO118 (50%) MO114 → MO122 (31%)	5.8336 eV (212.53 nm)	0.0233

HOMO: MO115, LUMO: MO116

Table S31 Calculated excited states of **1OMe** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO107 → MO108	3.1083 eV (398.88 nm)	0.3148
S ₂	MO105 → MO108 (13%) MO107 → MO109 (87%)	3.5177 eV (352.46 nm)	0.0355
S ₃	MO106 → MO108 (77%) MO107 → MO109 (23%)	3.7226 eV (333.05 nm)	0.1496
S ₄	MO105 → MO108 (64%) MO106 → MO109 (36%)	4.0907 eV (303.09 nm)	0.0444
S ₅	MO106 → MO109	4.1750 eV (296.97 nm)	0.0290
S ₆	MO106 → MO109 (39%) MO107 → MO110 (61%)	4.2074 eV (294.68 nm)	0.3333
S ₇	MO107 → MO111	4.3614 eV (284.28 nm)	0.0062
S ₈	MO104 → MO108 (69%) MO105 → MO109 (31%)	4.3834 eV (282.85 nm)	0.0080
S ₉	MO103 → MO108 (12%) MO107 → MO111 (25%) MO107 → MO112 (63%)	4.4148 eV (280.84 nm)	0.0419
S ₁₀	MO104 → MO108 (18%) MO107 → MO113 (82%)	4.5608 eV (271.85 nm)	0.2299
S ₁₁	MO100 → MO108 (57%) MO102 → MO108 (13%) MO103 → MO108 (30%)	4.6095 eV (268.97 nm)	0.0074
S ₁₂	MO105 → MO109 (62%) MO107 → MO113 (38%)	4.6511 eV (266.57 nm)	0.2901
S ₁₃	MO100 → MO108 (18%) MO102 → MO108 (82%)	4.7391 eV (261.62 nm)	0.0362
S ₁₄	MO100 → MO108 (27%) MO101 → MO108 (31%) MO104 → MO109 (42%)	4.7664 eV (260.12 nm)	0.0926
S ₁₅	MO101 → MO108 (16%) MO102 → MO108 (41%) MO103 → MO108 (18%) MO104 → MO109 (25%)	4.7730 eV (259.76 nm)	0.1065
S ₁₆	MO101 → MO108 (19%) MO106 → MO110 (81%)	4.7983 eV (258.39 nm)	0.1443
S ₁₇	MO102 → MO108 (13%) MO102 → MO113 (17%) MO105 → MO111 (14%) MO106 → MO111 (56%)	4.9382 eV (251.07 nm)	0.0606
S ₁₈	MO106 → MO111 (38%) MO107 → MO114 (62%)	4.9454 eV (250.71 nm)	0.0283
S ₁₉	MO106 → MO112	4.9946 eV (248.23 nm)	0.0103
S ₂₀	MO101 → MO108 (37%) MO101 → MO109 (13%) MO107 → MO115 (50%)	5.0483 eV (245.60 nm)	0.0790

S ₂₁	MO100 → MO109 (48%) MO102 → MO109 (12%) MO103 → MO109 (30%) MO106 → MO112 (10%)	5.0938 eV (243.40 nm)	0.0494
S ₂₂	MO100 → MO109 (13%) MO104 → MO109 (22%) MO105 → MO110 (19%) MO106 → MO110 (9%) MO106 → MO113 (10%) MO107 → MO115 (27%)	5.1106 eV (242.60 nm)	0.4621
S ₂₃	MO102 → MO109	5.2106 eV (237.95 nm)	0.0051
S ₂₄	MO100 → MO109 (26%) MO101 → MO108 (19%) MO105 → MO110 (42%) MO107 → MO117 (13%)	5.2322 eV (236.97 nm)	0.0789
S ₂₅	MO101 → MO108 (12%) MO102 → MO109 (23%) MO103 → MO109 (23%) MO105 → MO110 (29%) MO107 → MO116 (13%)	5.2447 eV (236.40 nm)	0.1257
S ₂₆	MO100 → MO109 (19%) MO106 → MO114 (17%) MO107 → MO116 (64%)	5.2512 eV (236.11 nm)	0.0208
S ₂₇	MO101 → MO109 (15%) MO105 → MO111 (23%) MO107 → MO117 (62%)	5.2897 eV (234.39 nm)	0.0020
S ₂₈	MO106 → MO113 (43%) MO107 → MO116 (17%) MO107 → MO117 (40%)	5.2946 eV (234.17 nm)	0.0148
S ₂₉	MO102 → MO113 (9%) MO105 → MO111 (41%) MO105 → MO112 (11%) MO106 → MO113 (19%) MO107 → MO118 (20%)	5.3786 eV (230.51 nm)	0.0107
S ₃₀	MO101 → MO109 (35%) MO104 → MO110 (24%) MO107 → MO118 (41%)	5.4030 eV (229.47 nm)	0.0305

HOMO: MO107, LUMO: MO108

Table S32 Calculated excited states of **1OMe**·H⁺ in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO107 → MO108	2.7226 eV (455.40 nm)	0.1378
S ₂	MO106 → MO108	3.0478 eV (406.79 nm)	0.0263
S ₃	MO105 → MO108 (22%) MO107 → MO109 (78%)	3.4013 eV (364.52 nm)	0.0604
S ₄	MO105 → MO108	3.5581 eV (348.46 nm)	0.1254
S ₅	MO102 → MO108 (17%) MO105 → MO108 (13%) MO106 → MO109 (70%)	3.7563 eV (330.07 nm)	0.0897
S ₆	MO104 → MO108	3.9353 eV (315.06 nm)	0.0413
S ₇	MO102 → MO108 (53%) MO104 → MO108 (21%) MO105 → MO109 (17%) MO107 → MO109 (9%)	3.9819 eV (311.37 nm)	0.1246
S ₈	MO103 → MO108	4.1117 eV (301.54 nm)	0.0018
S ₉	MO102 → MO109 (17%) MO107 → MO110 (83%)	4.1969 eV (295.42 nm)	0.0898
S ₁₀	MO101 → MO108 (13%) MO105 → MO109 (70%) MO107 → MO110 (17%)	4.2972 eV (288.53 nm)	0.0680
S ₁₁	MO106 → MO110 (86%) MO107 → MO111 (14%)	4.5368 eV (273.28 nm)	0.1816
S ₁₂	MO104 → MO109	4.6047 eV (269.26 nm)	0.0050
S ₁₃	MO101 → MO108 (13%) MO102 → MO109 (13%) MO104 → MO109 (21%) MO107 → MO111 (53%)	4.6362 eV (267.43 nm)	0.0199
S ₁₄	MO101 → MO108 (43%) MO102 → MO109 (39%) MO106 → MO110 (18%)	4.6789 eV (264.99 nm)	0.2030
S ₁₅	MO104 → MO109 (22%) MO106 → MO112 (11%) MO107 → MO112 (53%) MO107 → MO113 (14%)	4.7566 eV (260.66 nm)	0.0467
S ₁₆	MO101 → MO108 (17%) MO103 → MO109 (33%) MO105 → MO110 (8%) MO106 → MO111 (13%) MO107 → MO111 (17%) MO107 → MO113 (12%)	4.8056 eV (258.00 nm)	0.2385
S ₁₇	MO101 → MO108 (45%) MO105 → MO110 (22%) MO107 → MO113 (33%)	4.8307 eV (256.66 nm)	0.3914
S ₁₈	MO102 → MO109 (14%) MO106 → MO111 (40%) MO107 → MO111 (15%) MO107 → MO113 (31%)	4.9310 eV (251.44 nm)	0.1525

S ₁₉	MO102 → MO109 (16%) MO103 → MO109 (23%) MO107 → MO113 (61%)	4.9598 eV (249.98 nm)	0.1053
S ₂₀	MO102 → MO109 (17%) MO105 → MO110 (35%) MO107 → MO113 (13%) MO107 → MO114 (35%)	5.0715 eV (244.47 nm)	0.2357
S ₂₁	MO102 → MO109 (14%) MO105 → MO110 (13%) MO105 → MO112 (16%) MO106 → MO112 (57%)	5.1503 eV (240.73 nm)	0.1044
S ₂₂	MO106 → MO112 (22%) MO106 → MO113 (29%) MO107 → MO114 (49%)	5.2355 eV (236.81 nm)	0.3168
S ₂₃	MO106 → MO112 (27%) MO107 → MO115 (73%)	5.2840 eV (234.64 nm)	0.0311
S ₂₄	MO105 → MO110 (11%) MO106 → MO113 (42%) MO107 → MO115 (47%)	5.3058 eV (233.68 nm)	0.0308
S ₂₅	MO100 → MO108 (29%) MO101 → MO109 (31%) MO104 → MO110 (40%)	5.4406 eV (227.89 nm)	0.0049
S ₂₆	MO104 → MO110 (62%) MO105 → MO111 (16%) MO106 → MO114 (22%)	5.4542 eV (227.32 nm)	0.0582
S ₂₇	MO105 → MO111 (65%) MO106 → MO111 (21%) MO106 → MO114 (14%)	5.4723 eV (226.57 nm)	0.0106
S ₂₈	MO100 → MO108 (20%) MO102 → MO110 (27%) MO105 → MO111 (16%) MO107 → MO116 (37%)	5.5066 eV (225.15 nm)	0.0157
S ₂₉	MO102 → MO110 (15%) MO104 → MO110 (17%) MO106 → MO117 (14%) MO107 → MO117 (54%)	5.5434 eV (223.66 nm)	0.0359
S ₃₀	MO100 → MO108 (17%) MO103 → MO111 (13%) MO106 → MO114 (39%) MO107 → MO117 (31%)	5.5651 eV (222.79 nm)	0.0278

HOMO: MO107, LUMO: MO108

Table S33 Calculated excited states of **1Me** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO099 → MO100	3.1719 eV (390.88 nm)	0.3245
S ₂	MO098 → MO100 (22%) MO099 → MO101 (78%)	3.5860 eV (345.74 nm)	0.0470
S ₃	MO098 → MO100	4.0200 eV (308.42 nm)	0.1713
S ₄	MO098 → MO100 (17%) MO098 → MO101 (14%) MO099 → MO102 (69%)	4.2056 eV (294.80 nm)	0.1644
S ₅	MO097 → MO100 (82%) MO099 → MO102 (18%)	4.3677 eV (283.86 nm)	0.0703
S ₆	MO099 → MO102 (17%) MO099 → MO103 (83%)	4.4161 eV (280.76 nm)	0.2349
S ₇	MO096 → MO100 (29%) MO097 → MO100 (14%) MO098 → MO101 (33%) MO099 → MO103 (24%)	4.4551 eV (278.30 nm)	0.2379
S ₈	MO095 → MO100 (28%) MO096 → MO100 (13%) MO099 → MO104 (59%)	4.4774 eV (276.91 nm)	0.0034
S ₉	MO092 → MO100 (14%) MO099 → MO104 (44%) MO099 → MO105 (42%)	4.5116 eV (274.81 nm)	0.0172
S ₁₀	MO096 → MO100 (52%) MO097 → MO101 (14%) MO099 → MO105 (20%) MO099 → MO106 (14%)	4.5800 eV (270.71 nm)	0.2105
S ₁₁	MO092 → MO100 (62%) MO094 → MO100 (38%)	4.5936 eV (269.90 nm)	0.0161
S ₁₂	MO094 → MO100 (50%) MO095 → MO100 (18%) MO096 → MO100 (9%) MO098 → MO101 (10%) MO099 → MO105 (13%)	4.6409 eV (267.16 nm)	0.0204
S ₁₃	MO092 → MO100 (41%) MO095 → MO100 (23%) MO096 → MO100 (19%) MO099 → MO105 (17%)	4.6960 eV (264.02 nm)	0.0082
S ₁₄	MO096 → MO101 (33%) MO097 → MO101 (67%)	4.8455 eV (255.87 nm)	0.1390
S ₁₅	MO093 → MO100 (32%) MO095 → MO101 (16%) MO097 → MO101 (41%) MO098 → MO102 (11%)	4.8599 eV (255.11 nm)	0.0623
S ₁₆	MO093 → MO100 (15%) MO095 → MO101 (13%) MO096 → MO101 (37%) MO098 → MO102 (35%)	4.9985 eV (248.04 nm)	0.0519

S ₁₇	MO095 → MO101 (78%) MO096 → MO101 (22%)	5.0084 eV (247.55 nm)	0.0248
S ₁₈	MO095 → MO101 (15%) MO099 → MO107 (73%) MO099 → MO111 (12%)	5.0283 eV (246.57 nm)	0.0015
S ₁₉	MO094 → MO101 (56%) MO096 → MO101 (13%) MO097 → MO104 (12%) MO099 → MO107 (19%)	5.0603 eV (245.01 nm)	0.0125
S ₂₀	MO094 → MO101 (17%) MO096 → MO101 (17%) MO098 → MO102 (16%) MO099 → MO106 (50%)	5.0885 eV (243.66 nm)	0.1452
S ₂₁	MO092 → MO101 (69%) MO095 → MO101 (18%) MO096 → MO101 (13%)	5.1517 eV (240.66 nm)	0.0041
S ₂₂	MO093 → MO100 (54%) MO096 → MO101 (31%) MO099 → MO106 (15%)	5.2251 eV (237.29 nm)	0.5622
S ₂₃	MO092 → MO101 (18%) MO095 → MO101 (18%) MO096 → MO101 (16%) MO098 → MO105 (48%)	5.2530 eV (236.03 nm)	0.0108
S ₂₄	MO093 → MO100 (10%) MO094 → MO101 (14%) MO096 → MO101 (11%) MO097 → MO103 (13%) MO098 → MO104 (28%) MO098 → MO105 (24%)	5.2803 eV (234.80 nm)	0.0423
S ₂₅	MO098 → MO103	5.3367 eV (232.32 nm)	0.0531
S ₂₆	MO099 → MO108 (82%) MO099 → MO112 (18%)	5.3680 eV (230.97 nm)	0.0005
S ₂₇	MO099 → MO109	5.4136 eV (229.03 nm)	0.0108
S ₂₈	MO096 → MO102 (21%) MO097 → MO102 (13%) MO099 → MO110 (66%)	5.4440 eV (227.74 nm)	0.0808
S ₂₉	MO093 → MO101 (27%) MO097 → MO102 (55%) MO098 → MO104 (18%)	5.4663 eV (226.81 nm)	0.0004
S ₃₀	MO094 → MO103 (16%) MO097 → MO104 (33%) MO098 → MO104 (40%) MO098 → MO105 (11%)	5.5207 eV (224.58 nm)	0.0299

HOMO: MO099, LUMO: MO100

Table S34 Calculated excited states of **1Me**·H⁺ in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO099 → MO100	2.8701 eV (431.98 nm)	0.1789
S ₂	MO098 → MO100 (76%) MO099 → MO101 (24%)	3.4203 eV (362.50 nm)	0.0137
S ₃	MO096 → MO100 (16%) MO097 → MO100 (13%) MO099 → MO101 (71%)	3.5549 eV (348.77 nm)	0.0856
S ₄	MO097 → MO100	3.8048 eV (325.86 nm)	0.0542
S ₅	MO096 → MO100 (83%) MO098 → MO101 (17%)	3.8876 eV (318.92 nm)	0.0982
S ₆	MO095 → MO100	3.9923 eV (310.56 nm)	0.0085
S ₇	MO094 → MO100 (66%) MO096 → MO100 (11%) MO099 → MO101 (11%) MO099 → MO102 (12%)	4.0463 eV (306.41 nm)	0.0496
S ₈	MO094 → MO100 (30%) MO098 → MO101 (70%)	4.1371 eV (299.69 nm)	0.1357
S ₉	MO094 → MO101 (14%) MO098 → MO101 (24%) MO099 → MO102 (62%)	4.3183 eV (287.11 nm)	0.1630
S ₁₀	MO097 → MO101 (87%) MO099 → MO102 (13%)	4.4738 eV (277.14 nm)	0.0310
S ₁₁	MO093 → MO100 (13%) MO096 → MO101 (74%) MO099 → MO102 (13%)	4.5819 eV (270.59 nm)	0.1471
S ₁₂	MO095 → MO101 (87%) MO095 → MO102 (13%)	4.6725 eV (265.35 nm)	0.0081
S ₁₃	MO093 → MO100 (49%) MO094 → MO101 (31%) MO098 → MO102 (9%) MO099 → MO104 (11%)	4.7193 eV (262.72 nm)	0.0429
S ₁₄	MO093 → MO100 (16%) MO095 → MO101 (14%) MO098 → MO102 (17%) MO099 → MO103 (53%)	4.8244 eV (256.99 nm)	0.2081
S ₁₅	MO094 → MO101 (20%) MO095 → MO101 (16%) MO099 → MO103 (25%) MO099 → MO104 (39%)	4.8468 eV (255.81 nm)	0.1257
S ₁₆	MO098 → MO102 (60%) MO099 → MO104 (40%)	4.9306 eV (251.46 nm)	0.4629
S ₁₇	MO094 → MO101 (37%) MO097 → MO103 (14%) MO098 → MO105 (12%) MO099 → MO106 (37%)	4.9471 eV (250.62 nm)	0.2696
S ₁₈	MO094 → MO101 (25%) MO098 → MO102 (21%) MO099 → MO105 (54%)	4.9862 eV (248.66 nm)	0.1181

S ₁₉	MO096 → MO102 (16%) MO097 → MO102 (19%) MO099 → MO104 (10%) MO099 → MO105 (18%) MO099 → MO106 (37%)	5.1587 eV (240.34 nm)	0.1628
S ₂₀	MO095 → MO102 (55%) MO099 → MO103 (25%) MO099 → MO104 (20%)	5.1971 eV (238.56 nm)	0.0113
S ₂₁	MO097 → MO102	5.2858 eV (234.56 nm)	0.0433
S ₂₂	MO094 → MO101 (13%) MO096 → MO102 (38%) MO097 → MO102 (15%) MO097 → MO104 (11%) MO098 → MO103 (14%) MO098 → MO106 (9%)	5.3380 eV (232.27 nm)	0.2246
S ₂₃	MO096 → MO102 (21%) MO097 → MO103 (11%) MO098 → MO105 (16%) MO099 → MO107 (52%)	5.4198 eV (228.76 nm)	0.0496
S ₂₄	MO093 → MO101 (11%) MO096 → MO105 (10%) MO097 → MO104 (15%) MO098 → MO106 (13%) MO099 → MO106 (17%) MO099 → MO107 (34%)	5.4640 eV (226.91 nm)	0.0249
S ₂₅	MO092 → MO100 (21%) MO093 → MO101 (62%) MO098 → MO103 (17%)	5.4934 eV (225.70 nm)	0.0023
S ₂₆	MO095 → MO102 (31%) MO098 → MO103 (52%) MO099 → MO108 (17%)	5.5310 eV (224.16 nm)	0.0736
S ₂₇	MO092 → MO100 (22%) MO094 → MO102 (19%) MO095 → MO102 (11%) MO096 → MO102 (8%) MO098 → MO103 (8%) MO098 → MO105 (7%) MO099 → MO108 (25%)	5.5650 eV (222.79 nm)	0.0066
S ₂₈	MO095 → MO102 (36%) MO095 → MO105 (11%) MO095 → MO106 (10%) MO098 → MO104 (33%) MO098 → MO106 (10%)	5.6046 eV (221.22 nm)	0.0529
S ₂₉	MO093 → MO101 (11%) MO094 → MO102 (33%) MO098 → MO105 (27%) MO098 → MO106 (29%)	5.7087 eV (217.18 nm)	0.1897

S ₃₀	MO092 → MO100 (10%)	5.7824 eV (214.42 nm)	0.1250
	MO094 → MO102 (28%)		
	MO095 → MO103 (16%)		
	MO095 → MO104 (11%)		
	MO096 → MO103 (19%)		
	MO097 → MO103 (16%)		
HOMO: MO099, LUMO: MO100			

Table S35 Calculated excited states of **1H** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO091 → MO092	3.2194 eV (385.12 nm)	0.3188
S ₂	MO090 → MO092 (24%) MO091 → MO093 (76%)	3.6350 eV (341.08 nm)	0.0473
S ₃	MO090 → MO092 (68%) MO091 → MO094 (32%)	4.1186 eV (301.03 nm)	0.1083
S ₄	MO091 → MO094	4.2310 eV (293.04 nm)	0.2329
S ₅	MO091 → MO095	4.3834 eV (282.85 nm)	0.1746
S ₆	MO086 → MO092 (14%) MO088 → MO092 (19%) MO089 → MO092 (19%) MO090 → MO093 (30%) MO091 → MO096 (18%)	4.5003 eV (275.50 nm)	0.2253
S ₇	MO089 → MO092 (23%) MO091 → MO096 (77%)	4.5251 eV (273.99 nm)	0.0549
S ₈	MO084 → MO092 (20%) MO088 → MO092 (13%) MO091 → MO096 (34%) MO091 → MO097 (33%)	4.5441 eV (272.85 nm)	0.0176
S ₉	MO087 → MO092 (66%) MO091 → MO096 (14%) MO091 → MO097 (20%)	4.5874 eV (270.27 nm)	0.0370
S ₁₀	MO085 → MO092 (18%) MO089 → MO092 (29%) MO091 → MO097 (53%)	4.6167 eV (268.56 nm)	0.0094
S ₁₁	MO088 → MO092 (74%) MO089 → MO092 (26%)	4.6608 eV (266.02 nm)	0.1723
S ₁₂	MO084 → MO092 (26%) MO087 → MO092 (15%) MO089 → MO092 (30%) MO090 → MO093 (8%) MO091 → MO097 (21%)	4.7157 eV (262.92 nm)	0.0181
S ₁₃	MO084 → MO092 (17%) MO086 → MO092 (41%) MO087 → MO093 (9%) MO089 → MO092 (11%) MO090 → MO094 (9%) MO091 → MO098 (13%)	4.7455 eV (261.27 nm)	0.1784
S ₁₄	MO084 → MO092 (11%) MO085 → MO092 (28%) MO086 → MO092 (7%) MO086 → MO093 (12%) MO088 → MO093 (13%) MO089 → MO093 (11%) MO091 → MO098 (18%)	4.9448 eV (250.73 nm)	0.0380
S ₁₅	MO085 → MO093 (13%) MO087 → MO093 (31%) MO089 → MO093 (56%)	5.0258 eV (246.70 nm)	0.0021

S ₁₆	MO085 → MO092 (17%) MO087 → MO093 (56%) MO088 → MO093 (27%)	5.0442 eV (245.79 nm)	0.0382
S ₁₇	MO088 → MO093 (68%) MO090 → MO094 (32%)	5.0778 eV (244.17 nm)	0.0070
S ₁₈	MO088 → MO093 (30%) MO090 → MO094 (24%) MO091 → MO099 (46%)	5.0894 eV (243.61 nm)	0.0086
S ₁₉	MO086 → MO093 (62%) MO090 → MO094 (38%)	5.1196 eV (242.18 nm)	0.0834
S ₂₀	MO085 → MO093 (15%) MO086 → MO093 (18%) MO087 → MO093 (18%) MO090 → MO093 (12%) MO091 → MO098 (37%)	5.1299 eV (241.69 nm)	0.1790
S ₂₁	MO084 → MO093 (38%) MO086 → MO093 (15%) MO087 → MO093 (17%) MO089 → MO093 (14%) MO090 → MO094 (8%) MO091 → MO098 (8%)	5.1745 eV (239.61 nm)	0.0500
S ₂₂	MO084 → MO092 (9%) MO085 → MO092 (30%) MO087 → MO093 (12%) MO090 → MO094 (30%) MO090 → MO095 (19%)	5.2792 eV (234.85 nm)	0.3233
S ₂₃	MO086 → MO097 (9%) MO089 → MO092 (7%) MO089 → MO093 (11%) MO089 → MO094 (26%) MO089 → MO095 (9%) MO089 → MO098 (9%) MO090 → MO097 (29%)	5.3323 eV (232.51 nm)	0.0087
S ₂₄	MO086 → MO093 (11%) MO087 → MO096 (13%) MO088 → MO095 (17%) MO090 → MO095 (59%)	5.3551 eV (231.53 nm)	0.1081
S ₂₅	MO086 → MO095 (19%) MO088 → MO093 (12%) MO090 → MO095 (27%) MO090 → MO096 (32%) MO090 → MO097 (10%)	5.3844 eV (230.26 nm)	0.0477
S ₂₆	MO091 → MO100	5.4369 eV (228.04 nm)	0.0006
S ₂₇	MO091 → MO101 (74%) MO091 → MO102 (26%)	5.4832 eV (226.12 nm)	0.0324
S ₂₈	MO091 → MO102	5.5000 eV (225.43 nm)	0.0510
S ₂₉	MO087 → MO094 (9%) MO087 → MO096 (23%) MO088 → MO095 (19%) MO090 → MO096 (49%)	5.5702 eV (222.59 nm)	0.0168

S ₃₀	MO083 → MO092 (11%)	5.6230 eV (220.50 nm)	0.1103
	MO084 → MO093 (7%)		
	MO085 → MO093 (22%)		
	MO086 → MO094 (7%)		
	MO087 → MO094 (19%)		
	MO088 → MO094 (13%)		
	MO091 → MO102 (21%)		
HOMO: MO091, LUMO: MO092			

Table S36 Calculated excited states of $1\text{H}\cdot\text{H}^+$ in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO091 → MO092	2.9416 eV (421.48 nm)	0.1851
S ₂	MO090 → MO092 (37%) MO091 → MO093 (63%)	3.5757 eV (346.74 nm)	0.0194
S ₃	MO087 → MO092 (15%) MO090 → MO092 (85%)	3.7679 eV (329.06 nm)	0.1253
S ₄	MO089 → MO092 (87%) MO090 → MO092 (13%)	3.8635 eV (320.91 nm)	0.0140
S ₅	MO088 → MO092	4.0022 eV (309.79 nm)	0.1224
S ₆	MO087 → MO092 (50%) MO088 → MO092 (50%)	4.0429 eV (306.67 nm)	0.0144
S ₇	MO086 → MO092	4.1818 eV (296.49 nm)	0.0280
S ₈	MO090 → MO093 (41%) MO091 → MO094 (59%)	4.2946 eV (288.70 nm)	0.0094
S ₉	MO087 → MO093 (16%) MO089 → MO093 (18%) MO090 → MO093 (66%)	4.5024 eV (275.37 nm)	0.3602
S ₁₀	MO089 → MO093	4.5301 eV (273.69 nm)	0.0376
S ₁₁	MO085 → MO092 (20%) MO088 → MO093 (80%)	4.6825 eV (264.78 nm)	0.2228
S ₁₂	MO087 → MO093 (44%) MO088 → MO093 (43%) MO091 → MO095 (13%)	4.7174 eV (262.82 nm)	0.0348
S ₁₃	MO085 → MO092 (59%) MO086 → MO093 (28%) MO091 → MO094 (13%)	4.7651 eV (260.19 nm)	0.0330
S ₁₄	MO091 → MO095	4.8369 eV (256.33 nm)	0.2032
S ₁₅	MO087 → MO093 (12%) MO088 → MO093 (16%) MO091 → MO096 (72%)	4.9020 eV (252.92 nm)	0.0165
S ₁₆	MO086 → MO093 (44%) MO091 → MO097 (56%)	4.9566 eV (250.14 nm)	0.1623
S ₁₇	MO086 → MO093 (68%) MO089 → MO095 (32%)	5.0212 eV (246.92 nm)	0.2276
S ₁₈	MO090 → MO094 (77%) MO091 → MO097 (23%)	5.2145 eV (237.77 nm)	0.2461
S ₁₉	MO086 → MO093 (24%) MO087 → MO094 (13%) MO090 → MO094 (19%) MO091 → MO098 (44%)	5.2360 eV (236.79 nm)	0.2780
S ₂₀	MO086 → MO096 (10%) MO087 → MO094 (15%) MO088 → MO094 (34%) MO088 → MO097 (8%) MO090 → MO096 (20%) MO091 → MO096 (13%)	5.3026 eV (233.82 nm)	0.0061
S ₂₁	MO089 → MO094 (81%) MO091 → MO098 (19%)	5.3302 eV (232.61 nm)	0.0247

S ₂₂	MO087 → MO094 (36%) MO089 → MO095 (24%) MO091 → MO097 (12%) MO091 → MO098 (12%) MO091 → MO099 (16%)	5.4252 eV (228.53 nm)	0.1373
S ₂₃	MO085 → MO093 (10%) MO088 → MO094 (13%) MO089 → MO095 (12%) MO090 → MO095 (9%) MO091 → MO099 (46%) MO091 → MO100 (10%)	5.4912 eV (225.79 nm)	0.0426
S ₂₄	MO084 → MO092 (27%) MO085 → MO093 (59%) MO086 → MO094 (14%)	5.5244 eV (224.43 nm)	0.0271
S ₂₅	MO085 → MO093 (8%) MO086 → MO094 (8%) MO087 → MO094 (14%) MO087 → MO097 (9%) MO087 → MO098 (9%) MO090 → MO097 (12%) MO090 → MO098 (9%) MO091 → MO099 (31%)	5.5305 eV (224.18 nm)	0.0398
S ₂₆	MO084 → MO092 (42%) MO087 → MO094 (14%) MO090 → MO097 (10%) MO091 → MO100 (34%)	5.6240 eV (220.45 nm)	0.0219
S ₂₇	MO086 → MO094 (8%) MO087 → MO094 (15%) MO087 → MO096 (9%) MO088 → MO094 (23%) MO088 → MO098 (8%) MO090 → MO095 (37%)	5.6846 eV (218.11 nm)	0.0319
S ₂₈	MO084 → MO092 (10%) MO086 → MO094 (21%) MO088 → MO097 (11%) MO090 → MO095 (27%) MO090 → MO096 (31%)	5.7320 eV (216.30 nm)	0.0036
S ₂₉	MO086 → MO094 (41%) MO088 → MO095 (15%) MO090 → MO098 (15%) MO091 → MO100 (29%)	5.7881 eV (214.20 nm)	0.2507
S ₃₀	MO084 → MO092 (47%) MO086 → MO094 (14%) MO087 → MO095 (39%)	5.8800 eV (210.86 nm)	0.0301

HOMO: MO091, LUMO: MO092

Table S37 Calculated excited states of **1F** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO098 → MO101 (14%) MO099 → MO100 (86%)	3.2372 eV (383.00 nm)	0.3162
S ₂	MO099 → MO101	3.6512 eV (339.57 nm)	0.0451
S ₃	MO098 → MO100 (73%) MO099 → MO101 (27%)	4.1075 eV (301.85 nm)	0.1509
S ₄	MO098 → MO100 (26%) MO099 → MO102 (74%)	4.2417 eV (292.30 nm)	0.1486
S ₅	MO099 → MO103	4.3269 eV (286.54 nm)	0.0165
S ₆	MO099 → MO104	4.3801 eV (283.06 nm)	0.0116
S ₇	MO099 → MO105	4.4802 eV (276.74 nm)	0.2067
S ₈	MO092 → MO100 (16%) MO097 → MO100 (68%) MO099 → MO105 (16%)	4.5017 eV (275.42 nm)	0.0227
S ₉	MO096 → MO100 (29%) MO098 → MO101 (30%) MO099 → MO105 (41%)	4.5370 eV (273.27 nm)	0.2845
S ₁₀	MO092 → MO100 (71%) MO095 → MO100 (29%)	4.6293 eV (267.82 nm)	0.0063
S ₁₁	MO096 → MO100 (66%) MO098 → MO102 (15%) MO099 → MO106 (19%)	4.6992 eV (263.84 nm)	0.2329
S ₁₂	MO094 → MO100 (14%) MO095 → MO100 (65%) MO098 → MO104 (21%)	4.8896 eV (253.57 nm)	0.0072
S ₁₃	MO092 → MO100 (11%) MO094 → MO100 (45%) MO096 → MO100 (8%) MO097 → MO101 (16%) MO099 → MO106 (20%)	4.9270 eV (251.64 nm)	0.0010
S ₁₄	MO096 → MO100 (18%) MO096 → MO101 (11%) MO097 → MO101 (43%) MO099 → MO106 (28%)	4.9332 eV (251.33 nm)	0.0804
S ₁₅	MO092 → MO101 (9%) MO093 → MO100 (31%) MO097 → MO101 (30%) MO098 → MO102 (30%)	4.9885 eV (248.54 nm)	0.1452
S ₁₆	MO096 → MO100 (11%) MO096 → MO101 (44%) MO098 → MO102 (33%) MO099 → MO107 (12%)	5.0950 eV (243.34 nm)	0.0745
S ₁₇	MO092 → MO101 (58%) MO095 → MO101 (29%) MO099 → MO107 (13%)	5.1035 eV (242.94 nm)	0.0096
S ₁₈	MO097 → MO101 (17%) MO099 → MO107 (83%)	5.1160 eV (242.35 nm)	0.0223

S ₁₉	MO092 → MO101 (7%) MO093 → MO100 (7%) MO096 → MO101 (14%) MO098 → MO101 (8%) MO098 → MO102 (11%) MO098 → MO103 (9%) MO099 → MO106 (21%) MO099 → MO107 (23%)	5.1320 eV (241.59 nm)	0.1860
S ₂₀	MO093 → MO100 (12%) MO094 → MO100 (10%) MO096 → MO101 (10%) MO097 → MO101 (11%) MO097 → MO103 (10%) MO098 → MO103 (47%)	5.1784 eV (239.42 nm)	0.0988
S ₂₁	MO095 → MO101 (23%) MO095 → MO106 (14%) MO098 → MO104 (63%)	5.2015 eV (238.36 nm)	0.0044
S ₂₂	MO093 → MO100 (43%) MO094 → MO101 (15%) MO096 → MO101 (30%) MO097 → MO101 (12%)	5.2897 eV (234.39 nm)	0.3768
S ₂₃	MO094 → MO101 (39%) MO094 → MO105 (9%) MO096 → MO103 (10%) MO097 → MO102 (11%) MO098 → MO103 (31%)	5.3500 eV (231.75 nm)	0.0136
S ₂₄	MO095 → MO101	5.3814 eV (230.39 nm)	0.0071
S ₂₅	MO094 → MO101 (35%) MO097 → MO103 (16%) MO098 → MO105 (49%)	5.4271 eV (228.46 nm)	0.0314
S ₂₆	MO093 → MO101 (19%) MO094 → MO101 (10%) MO098 → MO105 (11%) MO099 → MO108 (25%) MO099 → MO109 (35%)	5.4650 eV (226.87 nm)	0.0577
S ₂₇	MO094 → MO105 (10%) MO097 → MO102 (15%) MO098 → MO105 (23%) MO099 → MO108 (42%) MO099 → MO110 (10%)	5.4965 eV (225.57 nm)	0.0110
S ₂₈	MO094 → MO101 (23%) MO094 → MO102 (12%) MO097 → MO103 (28%) MO099 → MO108 (37%)	5.5152 eV (224.80 nm)	0.0080
S ₂₉	MO096 → MO102 (13%) MO099 → MO109 (37%) MO099 → MO110 (50%)	5.5705 eV (222.57 nm)	0.0313

S ₃₀	MO093 → MO100 (10%)	5.5913 eV (221.75 nm)	0.0027
	MO093 → MO101 (25%)		
	MO097 → MO102 (37%)		
	MO097 → MO103 (16%)		
	MO099 → MO110 (12%)		
HOMO: MO099, LUMO: MO100			

Table S38 Calculated excited states of **1F·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO099 → MO100	2.9671 eV (417.86 nm)	0.1856
S ₂	MO098 → MO100	3.5755 eV (346.76 nm)	0.0069
S ₃	MO097 → MO100 (14%) MO098 → MO100 (45%) MO099 → MO101 (41%)	3.6954 eV (335.51 nm)	0.0974
S ₄	MO097 → MO100	4.0073 eV (309.40 nm)	0.1918
S ₅	MO095 → MO100 (45%) MO096 → MO100 (55%)	4.1222 eV (300.77 nm)	0.0113
S ₆	MO094 → MO100 (32%) MO096 → MO100 (68%)	4.1491 eV (298.82 nm)	0.0131
S ₇	MO097 → MO100 (20%) MO098 → MO101 (80%)	4.2778 eV (289.83 nm)	0.0385
S ₈	MO094 → MO100 (54%) MO095 → MO100 (33%) MO098 → MO101 (13%)	4.2985 eV (288.43 nm)	0.0068
S ₉	MO095 → MO101 (9%) MO097 → MO101 (15%) MO098 → MO101 (35%) MO099 → MO102 (41%)	4.4288 eV (279.95 nm)	0.2192
S ₁₀	MO093 → MO100 (16%) MO097 → MO101 (84%)	4.6832 eV (264.75 nm)	0.2638
S ₁₁	MO097 → MO101 (22%) MO099 → MO103 (78%)	4.7029 eV (263.64 nm)	0.0152
S ₁₂	MO093 → MO100 (54%) MO094 → MO101 (17%) MO099 → MO102 (15%) MO099 → MO104 (14%)	4.7621 eV (260.35 nm)	0.0222
S ₁₃	MO096 → MO101 (69%) MO099 → MO104 (31%)	4.7780 eV (259.49 nm)	0.0474
S ₁₄	MO097 → MO101 (25%) MO099 → MO104 (75%)	4.8455 eV (255.88 nm)	0.0550
S ₁₅	MO093 → MO100 (25%) MO095 → MO101 (40%) MO098 → MO102 (12%) MO099 → MO106 (23%)	4.9418 eV (250.89 nm)	0.3346
S ₁₆	MO094 → MO101 (39%) MO095 → MO101 (29%) MO098 → MO103 (10%) MO099 → MO103 (10%) MO099 → MO105 (12%)	4.9659 eV (249.67 nm)	0.0129
S ₁₇	MO093 → MO100 (10%) MO095 → MO101 (12%) MO098 → MO102 (9%) MO099 → MO104 (9%) MO099 → MO105 (44%) MO099 → MO106 (16%)	5.0162 eV (247.17 nm)	0.2272

S ₁₈	MO094 → MO101 (16%) MO098 → MO102 (84%)	5.1153 eV (242.38 nm)	0.2261
S ₁₉	MO094 → MO101 (19%) MO097 → MO102 (18%) MO099 → MO106 (63%)	5.2107 eV (237.94 nm)	0.3310
S ₂₀	MO094 → MO105 (11%) MO094 → MO106 (14%) MO095 → MO106 (11%) MO098 → MO103 (53%) MO099 → MO103 (11%)	5.2563 eV (235.88 nm)	0.0139
S ₂₁	MO097 → MO104 (32%) MO098 → MO104 (55%) MO099 → MO105 (13%)	5.3814 eV (230.39 nm)	0.0572
S ₂₂	MO097 → MO102 (85%) MO098 → MO104 (15%)	5.4634 eV (226.94 nm)	0.1674
S ₂₃	MO093 → MO101 (64%) MO094 → MO102 (17%) MO099 → MO108 (19%)	5.5248 eV (224.42 nm)	0.0289
S ₂₄	MO099 → MO107	5.5308 eV (224.17 nm)	0.0015
S ₂₅	MO095 → MO102 (25%) MO096 → MO102 (58%) MO097 → MO104 (17%)	5.5988 eV (221.45 nm)	0.0027
S ₂₆	MO092 → MO100 (23%) MO094 → MO102 (10%) MO096 → MO102 (38%) MO098 → MO104 (16%) MO099 → MO108 (13%)	5.6369 eV (219.95 nm)	0.0254
S ₂₇	MO094 → MO102 (10%) MO095 → MO102 (16%) MO097 → MO103 (8%) MO098 → MO103 (20%) MO098 → MO104 (19%) MO098 → MO105 (27%)	5.6778 eV (218.36 nm)	0.0229
S ₂₈	MO092 → MO100 (10%) MO095 → MO102 (28%) MO097 → MO103 (10%) MO098 → MO103 (16%) MO098 → MO106 (17%) MO099 → MO108 (19%)	5.7430 eV (215.89 nm)	0.1244
S ₂₉	MO094 → MO102 (44%) MO097 → MO103 (23%) MO097 → MO104 (17%) MO098 → MO103 (16%)	5.7685 eV (214.93 nm)	0.1520
S ₃₀	MO092 → MO100 (20%) MO093 → MO101 (9%) MO096 → MO104 (9%) MO097 → MO104 (27%) MO098 → MO105 (25%) MO098 → MO106 (10%)	5.8495 eV (211.96 nm)	0.0189

HOMO: MO099, LUMO: MO100

Table S39 Calculated excited states of **1Cl** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO107 → MO108	3.2250 eV (384.45 nm)	0.3516
S ₂	MO106 → MO108 (25%) MO107 → MO109 (75%)	3.6462 eV (340.04 nm)	0.0544
S ₃	MO106 → MO108 (57%) MO107 → MO110 (43%)	4.0908 eV (303.08 nm)	0.0631
S ₄	MO107 → MO110 (79%) MO107 → MO111 (21%)	4.1739 eV (297.05 nm)	0.4402
S ₅	MO107 → MO111	4.2455 eV (292.04 nm)	0.1264
S ₆	MO107 → MO112	4.3311 eV (286.26 nm)	0.0048
S ₇	MO107 → MO112 (14%) MO107 → MO113 (86%)	4.3821 eV (282.93 nm)	0.0092
S ₈	MO104 → MO108 (23%) MO104 → MO109 (10%) MO105 → MO108 (34%) MO106 → MO109 (33%)	4.4644 eV (277.71 nm)	0.1802
S ₉	MO105 → MO108	4.5095 eV (274.94 nm)	0.1268
S ₁₀	MO100 → MO108 (61%) MO103 → MO108 (26%) MO106 → MO109 (13%)	4.6119 eV (268.84 nm)	0.0306
S ₁₁	MO100 → MO108 (21%) MO104 → MO108 (65%) MO106 → MO110 (14%)	4.6668 eV (265.67 nm)	0.2440
S ₁₂	MO101 → MO108 (20%) MO103 → MO108 (52%) MO106 → MO113 (17%) MO107 → MO113 (11%)	4.8599 eV (255.12 nm)	0.0175
S ₁₃	MO100 → MO108 (37%) MO101 → MO108 (63%)	4.8722 eV (254.47 nm)	0.0252
S ₁₄	MO101 → MO108 (28%) MO102 → MO108 (72%)	4.9168 eV (252.16 nm)	0.0268
S ₁₅	MO105 → MO109 (81%) MO106 → MO110 (19%)	4.9561 eV (250.16 nm)	0.0758
S ₁₆	MO100 → MO109 (20%) MO106 → MO110 (80%)	5.0209 eV (246.94 nm)	0.0447
S ₁₇	MO101 → MO108 (13%) MO101 → MO109 (9%) MO104 → MO108 (10%) MO105 → MO109 (15%) MO106 → MO111 (13%) MO107 → MO114 (32%) MO107 → MO115 (8%)	5.0555 eV (245.25 nm)	0.2155
S ₁₈	MO107 → MO115	5.0818 eV (243.98 nm)	0.0073
S ₁₉	MO100 → MO109 (51%) MO102 → MO109 (9%) MO103 → MO109 (27%) MO105 → MO109 (13%)	5.0938 eV (243.40 nm)	0.0269

S ₂₀	MO102 → MO108 (8%) MO102 → MO109 (17%) MO102 → MO111 (17%) MO105 → MO112 (19%) MO106 → MO111 (7%) MO106 → MO112 (32%)	5.1558 eV (240.47 nm)	0.0481
S ₂₁	MO100 → MO109 (9%) MO101 → MO108 (8%) MO104 → MO109 (17%) MO106 → MO110 (16%) MO106 → MO111 (35%) MO106 → MO113 (15%)	5.1678 eV (239.92 nm)	0.0193
S ₂₂	MO100 → MO109 (18%) MO103 → MO110 (25%) MO106 → MO113 (57%)	5.1870 eV (239.03 nm)	0.0114
S ₂₃	MO101 → MO108 (45%) MO104 → MO109 (30%) MO106 → MO110 (25%)	5.2383 eV (236.69 nm)	0.5165
S ₂₄	MO106 → MO112 (81%) MO106 → MO113 (19%)	5.3323 eV (232.51 nm)	0.0097
S ₂₅	MO103 → MO109 (73%) MO106 → MO113 (27%)	5.3611 eV (231.27 nm)	0.0038
S ₂₆	MO107 → MO116	5.3882 eV (230.10 nm)	0.0020
S ₂₇	MO107 → MO116 (22%) MO107 → MO117 (58%) MO107 → MO120 (20%)	5.4054 eV (229.37 nm)	0.0373
S ₂₈	MO102 → MO109 (64%) MO106 → MO112 (19%) MO107 → MO118 (17%)	5.4405 eV (227.89 nm)	0.0025
S ₂₉	MO107 → MO118 (39%) MO107 → MO119 (34%) MO107 → MO120 (27%)	5.4830 eV (226.12 nm)	0.0269
S ₃₀	MO101 → MO109 (28%) MO105 → MO110 (49%) MO107 → MO119 (13%) MO107 → MO120 (10%)	5.5155 eV (224.79 nm)	0.0248

HOMO: MO107, LUMO: MO108

Table S40 Calculated excited states of **1Cl·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO107 → MO108	2.9715 eV (417.24 nm)	0.1950
S ₂	MO106 → MO108 (59%) MO107 → MO109 (41%)	3.5610 eV (348.17 nm)	0.0081
S ₃	MO105 → MO108 (25%) MO107 → MO109 (75%)	3.6671 eV (338.10 nm)	0.1003
S ₄	MO105 → MO108 (85%) MO106 → MO109 (15%)	3.9596 eV (313.13 nm)	0.1757
S ₅	MO102 → MO108 (14%) MO103 → MO108 (62%) MO105 → MO108 (12%) MO107 → MO109 (12%)	4.0953 eV (302.75 nm)	0.0666
S ₆	MO104 → MO108	4.1310 eV (300.13 nm)	0.0039
S ₇	MO103 → MO108 (11%) MO106 → MO109 (52%) MO107 → MO110 (37%)	4.2409 eV (292.35 nm)	0.0527
S ₈	MO102 → MO108	4.2926 eV (288.83 nm)	0.0021
S ₉	MO105 → MO109 (20%) MO107 → MO110 (80%)	4.3563 eV (284.61 nm)	0.2746
S ₁₀	MO105 → MO109	4.6142 eV (268.70 nm)	0.2231
S ₁₁	MO103 → MO109 (13%) MO105 → MO109 (22%) MO107 → MO111 (65%)	4.6798 eV (264.93 nm)	0.0482
S ₁₂	MO101 → MO108	4.7136 eV (263.04 nm)	0.0060
S ₁₃	MO101 → MO108 (21%) MO104 → MO109 (35%) MO105 → MO109 (8%) MO107 → MO112 (15%) MO107 → MO113 (10%) MO107 → MO114 (11%)	4.7456 eV (261.26 nm)	0.0547
S ₁₄	MO101 → MO108 (25%) MO105 → MO109 (14%) MO107 → MO112 (61%)	4.7754 eV (259.63 nm)	0.2158
S ₁₅	MO101 → MO108 (19%) MO106 → MO110 (15%) MO107 → MO113 (66%)	4.8521 eV (255.53 nm)	0.0382
S ₁₆	MO101 → MO108 (28%) MO103 → MO109 (38%) MO104 → MO109 (15%) MO106 → MO110 (19%)	4.9005 eV (253.00 nm)	0.3091
S ₁₇	MO102 → MO109 (66%) MO106 → MO110 (16%) MO107 → MO111 (18%)	4.9389 eV (251.04 nm)	0.0054
S ₁₈	MO106 → MO110 (79%) MO107 → MO114 (21%)	5.0160 eV (247.18 nm)	0.2679

S ₁₉	MO102 → MO109 (10%) MO103 → MO109 (27%) MO105 → MO110 (22%) MO107 → MO114 (41%)	5.1308 eV (241.65 nm)	0.4515
S ₂₀	MO102 → MO109 (21%) MO102 → MO110 (20%) MO102 → MO113 (11%) MO105 → MO111 (14%) MO106 → MO111 (34%)	5.1809 eV (239.31 nm)	0.0105
S ₂₁	MO104 → MO110 (10%) MO104 → MO111 (7%) MO104 → MO112 (10%) MO105 → MO110 (12%) MO106 → MO111 (7%) MO106 → MO112 (12%) MO106 → MO113 (17%) MO106 → MO114 (9%) MO107 → MO113 (7%) MO107 → MO114 (9%)	5.2911 eV (234.33 nm)	0.0663
S ₂₂	MO105 → MO110 (83%) MO107 → MO116 (17%)	5.3737 eV (230.72 nm)	0.2048
S ₂₃	MO101 → MO109 (56%) MO106 → MO112 (24%) MO107 → MO116 (20%)	5.4710 eV (226.62 nm)	0.0081
S ₂₄	MO098 → MO108 (19%) MO104 → MO110 (17%) MO106 → MO111 (25%) MO106 → MO112 (39%)	5.5102 eV (225.01 nm)	0.0440
S ₂₅	MO106 → MO111 (12%) MO106 → MO112 (14%) MO107 → MO115 (74%)	5.5135 eV (224.87 nm)	0.0064
S ₂₆	MO101 → MO109 (27%) MO103 → MO110 (53%) MO106 → MO112 (20%)	5.5433 eV (223.66 nm)	0.0016
S ₂₇	MO104 → MO110	5.5576 eV (223.09 nm)	0.0283
S ₂₈	MO100 → MO108 (53%) MO101 → MO109 (10%) MO103 → MO110 (12%) MO104 → MO110 (25%)	5.5633 eV (222.86 nm)	0.0147
S ₂₉	MO099 → MO108	5.6316 eV (220.16 nm)	0.0106
S ₃₀	MO098 → MO108 (22%) MO099 → MO108 (10%) MO103 → MO110 (22%) MO106 → MO114 (25%) MO107 → MO116 (21%)	5.6486 eV (219.49 nm)	0.1348

HOMO: MO107, LUMO: MO108

Table S41 Calculated excited states of **1CF₃** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO123 → MO124	3.2547 eV (380.94 nm)	0.3731
S ₂	MO122 → MO124 (27%) MO123 → MO125 (73%)	3.6811 eV (336.81 nm)	0.0758
S ₃	MO123 → MO126	3.9336 eV (315.19 nm)	0.1862
S ₄	MO123 → MO125 (21%) MO123 → MO126 (12%) MO123 → MO127 (67%)	4.0364 eV (307.17 nm)	0.0487
S ₅	MO122 → MO124 (72%) MO123 → MO127 (28%)	4.2243 eV (293.50 nm)	0.3410
S ₆	MO123 → MO128	4.3718 eV (283.60 nm)	0.0041
S ₇	MO120 → MO124 (17%) MO123 → MO129 (83%)	4.4353 eV (279.54 nm)	0.0016
S ₈	MO122 → MO125 (75%) MO123 → MO130 (25%)	4.5253 eV (273.98 nm)	0.2866
S ₉	MO117 → MO124 (70%) MO118 → MO124 (17%) MO122 → MO125 (13%)	4.5996 eV (269.56 nm)	0.0236
S ₁₀	MO121 → MO124 (35%) MO122 → MO125 (22%) MO122 → MO126 (8%) MO122 → MO127 (25%) MO123 → MO130 (10%)	4.7278 eV (262.25 nm)	0.1568
S ₁₁	MO118 → MO124 (33%) MO119 → MO124 (24%) MO120 → MO124 (21%) MO122 → MO126 (12%) MO122 → MO127 (10%)	4.8313 eV (256.63 nm)	0.0655
S ₁₂	MO117 → MO124 (25%) MO120 → MO124 (51%) MO122 → MO129 (12%) MO123 → MO130 (12%)	4.8697 eV (254.60 nm)	0.0276
S ₁₃	MO122 → MO126	4.8971 eV (253.18 nm)	0.0758
S ₁₄	MO119 → MO124	4.9290 eV (251.54 nm)	0.0089
S ₁₅	MO116 → MO124 (17%) MO119 → MO124 (19%) MO122 → MO126 (18%) MO123 → MO130 (46%)	4.9519 eV (250.38 nm)	0.0630
S ₁₆	MO117 → MO124 (12%) MO117 → MO125 (52%) MO118 → MO125 (14%) MO121 → MO125 (22%)	5.0549 eV (245.28 nm)	0.0010
S ₁₇	MO116 → MO125 (14%) MO122 → MO127 (57%) MO123 → MO130 (29%)	5.1243 eV (241.95 nm)	0.4883

S ₁₈	MO116 → MO124 (18%) MO118 → MO124 (9%) MO120 → MO125 (11%) MO121 → MO125 (45%) MO122 → MO127 (17%)	5.1345 eV (241.47 nm)	0.0628
S ₁₉	MO118 → MO124 (16%) MO118 → MO126 (19%) MO119 → MO125 (31%) MO123 → MO131 (34%)	5.2040 eV (238.25 nm)	0.0434
S ₂₀	MO118 → MO125 (13%) MO122 → MO128 (15%) MO123 → MO131 (72%)	5.2216 eV (237.44 nm)	0.0008
S ₂₁	MO117 → MO125 (14%) MO118 → MO125 (8%) MO120 → MO125 (27%) MO120 → MO126 (19%) MO120 → MO127 (8%) MO122 → MO129 (24%)	5.2566 eV (235.86 nm)	0.0160
S ₂₂	MO116 → MO124 (60%) MO119 → MO125 (26%) MO122 → MO127 (14%)	5.2989 eV (233.98 nm)	0.0940
S ₂₃	MO118 → MO125 (29%) MO119 → MO125 (34%) MO120 → MO125 (22%) MO122 → MO128 (15%)	5.3472 eV (231.87 nm)	0.1346
S ₂₄	MO116 → MO124 (11%) MO117 → MO125 (24%) MO120 → MO125 (37%) MO122 → MO128 (28%)	5.3723 eV (230.79 nm)	0.0328
S ₂₅	MO119 → MO125 (22%) MO120 → MO126 (13%) MO122 → MO128 (47%) MO122 → MO129 (18%)	5.3772 eV (230.57 nm)	0.0143
S ₂₆	MO118 → MO127 (14%) MO118 → MO128 (11%) MO119 → MO125 (25%) MO121 → MO126 (50%)	5.4850 eV (226.04 nm)	0.0273
S ₂₇	MO118 → MO125 (18%) MO118 → MO126 (19%) MO121 → MO126 (24%) MO123 → MO133 (39%)	5.5256 eV (224.38 nm)	0.0438
S ₂₈	MO116 → MO125 (14%) MO118 → MO126 (15%) MO119 → MO127 (13%) MO121 → MO126 (34%) MO122 → MO129 (10%) MO123 → MO132 (14%)	5.5444 eV (223.62 nm)	0.0055
S ₂₉	MO123 → MO132 (61%) MO123 → MO133 (28%) MO123 → MO135 (11%)	5.5868 eV (221.92 nm)	0.0016

S ₃₀	MO116 → MO129 (12%)	5.5970 eV (221.52 nm)	0.0233
	MO117 → MO126 (11%)		
	MO117 → MO127 (14%)		
	MO121 → MO129 (16%)		
	MO122 → MO129 (47%)		
HOMO: MO123, LUMO: MO124			

Table S42 Calculated excited states of **1CF₃·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO123 → MO124	3.0545 eV (405.91 nm)	0.2112
S ₂	MO119 → MO124 (10%) MO122 → MO124 (32%) MO123 → MO125 (58%)	3.6625 eV (338.52 nm)	0.0332
S ₃	MO122 → MO124	3.9417 eV (314.54 nm)	0.2516
S ₄	MO121 → MO124	4.1737 eV (297.06 nm)	0.0060
S ₅	MO122 → MO124 (15%) MO122 → MO125 (20%) MO123 → MO126 (65%)	4.1932 eV (295.68 nm)	0.0568
S ₆	MO119 → MO124 (47%) MO120 → MO124 (22%) MO123 → MO126 (31%)	4.2902 eV (289.00 nm)	0.0593
S ₇	MO120 → MO124	4.3407 eV (285.63 nm)	0.0008
S ₈	MO118 → MO124 (63%) MO119 → MO124 (15%) MO123 → MO126 (22%)	4.4293 eV (279.92 nm)	0.0970
S ₉	MO123 → MO127	4.4848 eV (276.46 nm)	0.1831
S ₁₀	MO118 → MO124 (20%) MO122 → MO125 (41%) MO123 → MO127 (27%) MO123 → MO128 (12%)	4.5686 eV (271.39 nm)	0.4026
S ₁₁	MO120 → MO125 (18%) MO120 → MO126 (11%) MO123 → MO128 (58%) MO123 → MO129 (13%)	4.7363 eV (261.77 nm)	0.0170
S ₁₂	MO121 → MO125 (69%) MO121 → MO127 (12%) MO123 → MO130 (19%)	4.7649 eV (260.20 nm)	0.0214
S ₁₃	MO118 → MO125 (13%) MO121 → MO125 (18%) MO122 → MO125 (23%) MO123 → MO129 (46%)	4.8319 eV (256.59 nm)	0.0668
S ₁₄	MO117 → MO124 (78%) MO123 → MO129 (22%)	4.8957 eV (253.25 nm)	0.0950
S ₁₅	MO119 → MO125 (36%) MO120 → MO125 (18%) MO121 → MO125 (20%) MO123 → MO129 (26%)	4.9208 eV (251.96 nm)	0.1024
S ₁₆	MO120 → MO125	4.9515 eV (250.40 nm)	0.0078
S ₁₇	MO117 → MO124 (8%) MO117 → MO125 (8%) MO118 → MO125 (13%) MO119 → MO125 (15%) MO121 → MO127 (6%) MO122 → MO125 (9%) MO123 → MO129 (12%) MO123 → MO130 (29%)	5.0788 eV (244.12 nm)	0.2880

S ₁₈	MO122 → MO126 (81%) MO123 → MO129 (19%)	5.1271 eV (241.82 nm)	0.1126
S ₁₉	MO117 → MO124 (20%) MO118 → MO125 (38%) MO122 → MO126 (42%)	5.2607 eV (235.68 nm)	0.6173
S ₂₀	MO120 → MO126	5.2866 eV (234.53 nm)	0.0071
S ₂₁	MO119 → MO129 (14%) MO121 → MO126 (27%) MO121 → MO127 (36%) MO122 → MO127 (11%) MO122 → MO130 (12%)	5.3345 eV (232.42 nm)	0.0169
S ₂₂	MO117 → MO125 (23%) MO122 → MO127 (77%)	5.4306 eV (228.31 nm)	0.0099
S ₂₃	MO119 → MO126 (17%) MO121 → MO126 (56%) MO122 → MO127 (27%)	5.4744 eV (226.48 nm)	0.0125
S ₂₄	MO119 → MO126 (49%) MO120 → MO126 (20%) MO121 → MO127 (12%) MO123 → MO131 (19%)	5.5516 eV (223.33 nm)	0.0688
S ₂₅	MO116 → MO124 (16%) MO117 → MO125 (58%) MO123 → MO131 (26%)	5.5599 eV (223.00 nm)	0.0417
S ₂₆	MO123 → MO131 (38%) MO123 → MO132 (62%)	5.6341 eV (220.06 nm)	0.0041
S ₂₇	MO116 → MO124 (34%) MO117 → MO125 (10%) MO118 → MO126 (15%) MO119 → MO126 (14%) MO123 → MO132 (27%)	5.6708 eV (218.64 nm)	0.0713
S ₂₈	MO118 → MO126 (42%) MO120 → MO126 (11%) MO122 → MO128 (29%) MO123 → MO131 (18%)	5.7037 eV (217.38 nm)	0.0011
S ₂₉	MO120 → MO126 (26%) MO120 → MO127 (11%) MO122 → MO128 (37%) MO122 → MO129 (26%)	5.7810 eV (214.47 nm)	0.0255
S ₃₀	MO116 → MO124 (24%) MO119 → MO127 (42%) MO120 → MO127 (21%) MO122 → MO130 (13%)	5.8555 eV (211.74 nm)	0.0393

HOMO: MO123, LUMO: MO124

Table S43 Calculated excited states of **1CN** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO103 → MO104	3.1806 eV (389.81 nm)	0.4577
S ₂	MO102 → MO104 (15%) MO103 → MO105 (85%)	3.5247 eV (351.76 nm)	0.2204
S ₃	MO102 → MO104 (18%) MO103 → MO106 (82%)	3.6814 eV (336.78 nm)	0.1011
S ₄	MO103 → MO106 (22%) MO103 → MO107 (78%)	3.8218 eV (324.41 nm)	0.0010
S ₅	MO102 → MO104 (68%) MO103 → MO107 (32%)	4.1405 eV (299.44 nm)	0.5622
S ₆	MO103 → MO108	4.3356 eV (285.97 nm)	0.0096
S ₇	MO099 → MO104 (20%) MO099 → MO106 (11%) MO103 → MO109 (69%)	4.3909 eV (282.37 nm)	0.0003
S ₈	MO101 → MO104 (17%) MO102 → MO105 (61%) MO102 → MO106 (22%)	4.4198 eV (280.52 nm)	0.1285
S ₉	MO101 → MO104 (73%) MO102 → MO106 (27%)	4.5465 eV (272.70 nm)	0.0279
S ₁₀	MO097 → MO104 (31%) MO099 → MO104 (17%) MO100 → MO104 (19%) MO102 → MO106 (33%)	4.5714 eV (271.22 nm)	0.0815
S ₁₁	MO102 → MO106 (58%) MO102 → MO107 (28%) MO103 → MO110 (14%)	4.5818 eV (270.60 nm)	0.0578
S ₁₂	MO097 → MO104 (24%) MO099 → MO104 (14%) MO101 → MO104 (37%) MO102 → MO107 (10%) MO103 → MO110 (15%)	4.6640 eV (265.83 nm)	0.0610
S ₁₃	MO099 → MO104 (52%) MO099 → MO106 (17%) MO101 → MO109 (13%) MO102 → MO109 (18%)	4.7862 eV (259.05 nm)	0.0304
S ₁₄	MO099 → MO104 (15%) MO100 → MO104 (28%) MO102 → MO107 (13%) MO103 → MO110 (44%)	4.8027 eV (258.16 nm)	0.0346
S ₁₅	MO098 → MO104 (42%) MO098 → MO105 (14%) MO100 → MO104 (11%) MO100 → MO105 (10%) MO100 → MO108 (11%) MO101 → MO105 (12%)	4.8733 eV (254.41 nm)	0.0170
S ₁₆	MO100 → MO104 (21%) MO101 → MO105 (66%) MO102 → MO107 (13%)	4.9337 eV (251.30 nm)	0.0790

S ₁₇	MO097 → MO104 (10%) MO097 → MO105 (15%) MO097 → MO106 (13%) MO099 → MO105 (10%) MO100 → MO104 (10%) MO100 → MO106 (8%) MO101 → MO104 (8%) MO102 → MO107 (26%)	4.9886 eV (248.53 nm)	0.2668
S ₁₈	MO097 → MO104 (8%) MO097 → MO105 (18%) MO097 → MO106 (13%) MO098 → MO105 (15%) MO099 → MO105 (11%) MO100 → MO105 (20%) MO100 → MO108 (8%) MO103 → MO110 (7%)	5.0178 eV (247.09 nm)	0.3287
S ₁₉	MO096 → MO104 (20%) MO097 → MO107 (10%) MO098 → MO105 (35%) MO100 → MO108 (13%) MO102 → MO107 (22%)	5.0563 eV (245.21 nm)	0.1109
S ₂₀	MO097 → MO107 (14%) MO100 → MO105 (69%) MO100 → MO106 (17%)	5.1055 eV (242.85 nm)	0.3396
S ₂₁	MO096 → MO104 (40%) MO100 → MO105 (20%) MO101 → MO106 (17%) MO101 → MO107 (13%) MO103 → MO110 (10%)	5.1594 eV (240.31 nm)	0.0803
S ₂₂	MO098 → MO105 (16%) MO100 → MO106 (25%) MO101 → MO106 (59%)	5.1817 eV (239.27 nm)	0.0579
S ₂₃	MO097 → MO104 (11%) MO099 → MO106 (47%) MO101 → MO109 (11%) MO102 → MO109 (31%)	5.2225 eV (237.41 nm)	0.0020
S ₂₄	MO099 → MO106 (10%) MO103 → MO111 (66%) MO103 → MO114 (13%) MO103 → MO117 (11%)	5.2349 eV (236.84 nm)	0.0050
S ₂₅	MO097 → MO107 (14%) MO099 → MO105 (86%)	5.2543 eV (235.97 nm)	0.0017
S ₂₆	MO098 → MO105 (20%) MO102 → MO108 (80%)	5.3243 eV (232.86 nm)	0.0176
S ₂₇	MO096 → MO104 (25%) MO100 → MO106 (62%) MO100 → MO107 (13%)	5.3470 eV (231.88 nm)	0.0109
S ₂₈	MO100 → MO106 (42%) MO101 → MO107 (58%)	5.4142 eV (229.00 nm)	0.0711

S ₂₉	MO096 → MO105 (17%)	5.4468 eV (227.63 nm)	0.0714
	MO097 → MO106 (14%)		
	MO097 → MO107 (16%)		
	MO101 → MO107 (11%)		
	MO103 → MO113 (42%)		
S ₃₀	MO097 → MO104 (10%)	5.4543 eV (227.32 nm)	0.0482
	MO097 → MO106 (21%)		
	MO097 → MO107 (25%)		
	MO099 → MO106 (8%)		
	MO099 → MO107 (12%)		
	MO101 → MO107 (16%)		
	MO102 → MO110 (8%)		
HOMO: MO103, LUMO: MO104			

Table S44 Calculated excited states of **1CN·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO103 → MO104	3.0586 eV (405.36 nm)	0.2351
S ₂	MO102 → MO104 (30%) MO103 → MO105 (70%)	3.6338 eV (341.20 nm)	0.0757
S ₃	MO102 → MO104	3.8640 eV (320.87 nm)	0.1695
S ₄	MO102 → MO104 (35%) MO103 → MO106 (65%)	4.0048 eV (309.59 nm)	0.2206
S ₅	MO102 → MO104 (15%) MO103 → MO107 (85%)	4.1115 eV (301.56 nm)	0.1603
S ₆	MO098 → MO104 (11%) MO100 → MO104 (11%) MO101 → MO104 (44%) MO103 → MO106 (13%) MO103 → MO107 (21%)	4.2210 eV (293.73 nm)	0.0528
S ₇	MO098 → MO104 (19%) MO100 → MO104 (69%) MO103 → MO107 (12%)	4.2500 eV (291.72 nm)	0.0101
S ₈	MO098 → MO104 (58%) MO102 → MO105 (42%)	4.3412 eV (285.60 nm)	0.1422
S ₉	MO098 → MO104 (20%) MO099 → MO104 (56%) MO102 → MO105 (24%)	4.4068 eV (281.35 nm)	0.0657
S ₁₀	MO100 → MO104 (15%) MO102 → MO105 (66%) MO103 → MO109 (19%)	4.4338 eV (279.63 nm)	0.2874
S ₁₁	MO099 → MO105 (18%) MO102 → MO105 (20%) MO103 → MO108 (51%) MO103 → MO109 (11%)	4.6551 eV (266.34 nm)	0.0951
S ₁₂	MO098 → MO105 (13%) MO099 → MO105 (10%) MO099 → MO106 (10%) MO101 → MO105 (28%) MO103 → MO109 (39%)	4.6989 eV (263.86 nm)	0.0315
S ₁₃	MO098 → MO105 (13%) MO100 → MO105 (44%) MO103 → MO109 (19%) MO103 → MO110 (24%)	4.7539 eV (260.81 nm)	0.0402
S ₁₄	MO097 → MO104 (26%) MO100 → MO105 (25%) MO101 → MO105 (31%) MO102 → MO106 (18%)	4.7915 eV (258.76 nm)	0.2169
S ₁₅	MO101 → MO105 (43%) MO103 → MO108 (31%) MO103 → MO110 (26%)	4.8781 eV (254.16 nm)	0.0531
S ₁₆	MO098 → MO105 (45%) MO102 → MO106 (55%)	4.9038 eV (252.83 nm)	0.0487

S ₁₇	MO098 → MO105 (16%) MO099 → MO105 (65%) MO103 → MO110 (19%)	4.9065 eV (252.70 nm)	0.0275
S ₁₈	MO097 → MO104 (17%) MO098 → MO105 (22%) MO100 → MO107 (13%) MO103 → MO109 (12%) MO103 → MO110 (36%)	4.9714 eV (249.39 nm)	0.1216
S ₁₉	MO101 → MO107 (14%) MO102 → MO107 (86%)	5.0305 eV (246.47 nm)	0.1917
S ₂₀	MO097 → MO104 (14%) MO098 → MO105 (31%) MO100 → MO105 (15%) MO101 → MO107 (28%) MO103 → MO108 (12%)	5.1150 eV (242.39 nm)	0.5182
S ₂₁	MO098 → MO105 (17%) MO100 → MO107 (43%) MO101 → MO106 (26%) MO102 → MO110 (14%)	5.1824 eV (239.24 nm)	0.0128
S ₂₂	MO098 → MO106 (29%) MO098 → MO108 (14%) MO099 → MO106 (57%)	5.1896 eV (238.91 nm)	0.0111
S ₂₃	MO098 → MO106 (12%) MO100 → MO106 (13%) MO101 → MO106 (63%) MO103 → MO110 (12%)	5.2343 eV (236.87 nm)	0.1586
S ₂₄	MO100 → MO106	5.2907 eV (234.34 nm)	0.0731
S ₂₅	MO098 → MO106 (53%) MO100 → MO107 (20%) MO101 → MO107 (27%)	5.4130 eV (229.05 nm)	0.2269
S ₂₆	MO097 → MO105 (38%) MO100 → MO107 (28%) MO101 → MO107 (34%)	5.4291 eV (228.37 nm)	0.0667
S ₂₇	MO097 → MO105 (25%) MO098 → MO105 (10%) MO098 → MO106 (26%) MO102 → MO108 (20%) MO103 → MO111 (19%)	5.4827 eV (226.14 nm)	0.0020
S ₂₈	MO098 → MO107	5.5711 eV (222.55 nm)	0.0851
S ₂₉	MO096 → MO104 (26%) MO101 → MO107 (12%) MO102 → MO108 (27%) MO103 → MO111 (35%)	5.6004 eV (221.38 nm)	0.0016
S ₃₀	MO098 → MO107 (46%) MO099 → MO107 (54%)	5.6235 eV (220.47 nm)	0.0006

HOMO: MO103, LUMO: MO104

Table S45 Calculated excited states of **1NO₂** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO113 → MO114	2.5049 eV (494.96 nm)	0.2538
S ₂	MO113 → MO115	2.5611 eV (484.11 nm)	0.2283
S ₃	MO112 → MO115 (38%) MO112 → MO117 (9%) MO113 → MO116 (43%) MO113 → MO117 (10%)	3.3640 eV (368.57 nm)	0.0926
S ₄	MO112 → MO114 (55%) MO112 → MO115 (45%)	3.4303 eV (361.44 nm)	0.2586
S ₅	MO112 → MO114 (49%) MO113 → MO116 (51%)	3.4598 eV (358.36 nm)	0.0960
S ₆	MO105 → MO114 (46%) MO105 → MO115 (54%)	3.7748 eV (328.45 nm)	0.0106
S ₇	MO104 → MO114	3.7883 eV (327.29 nm)	0.0004
S ₈	MO105 → MO114 (17%) MO105 → MO115 (20%) MO113 → MO117 (63%)	3.7949 eV (326.72 nm)	0.1087
S ₉	MO110 → MO114 (41%) MO110 → MO115 (49%) MO111 → MO115 (10%)	3.8804 eV (319.51 nm)	0.0176
S ₁₀	MO108 → MO114 (64%) MO111 → MO115 (36%)	3.9193 eV (316.35 nm)	0.0179
S ₁₁	MO108 → MO114 (12%) MO111 → MO114 (48%) MO111 → MO115 (40%)	4.0220 eV (308.27 nm)	0.1183
S ₁₂	MO108 → MO115 (34%) MO111 → MO115 (66%)	4.0355 eV (307.23 nm)	0.0031
S ₁₃	MO109 → MO114	4.2243 eV (293.51 nm)	0.3828
S ₁₄	MO110 → MO114	4.2466 eV (291.96 nm)	0.0005
S ₁₅	MO107 → MO114 (16%) MO107 → MO115 (31%) MO108 → MO114 (14%) MO108 → MO115 (20%) MO110 → MO114 (19%)	4.2828 eV (289.50 nm)	0.0166
S ₁₆	MO107 → MO114 (7%) MO107 → MO115 (12%) MO108 → MO115 (15%) MO109 → MO114 (27%) MO109 → MO115 (20%) MO112 → MO116 (19%)	4.3058 eV (287.95 nm)	0.0729
S ₁₇	MO112 → MO116 (72%) MO113 → MO117 (28%)	4.3199 eV (287.01 nm)	0.0077
S ₁₈	MO108 → MO114 (43%) MO108 → MO115 (57%)	4.3557 eV (284.65 nm)	0.0022

S ₁₉	MO108 → MO114 (10%) MO112 → MO116 (22%) MO113 → MO117 (10%) MO113 → MO118 (44%) MO113 → MO119 (14%)	4.3719 eV (283.59 nm)	0.0123
S ₂₀	MO101 → MO114 (46%) MO101 → MO115 (54%)	4.4009 eV (281.73 nm)	0.0019
S ₂₁	MO099 → MO114 (68%) MO100 → MO115 (32%)	4.4052 eV (281.45 nm)	0.0003
S ₂₂	MO107 → MO114 (43%) MO113 → MO119 (19%) MO113 → MO120 (38%)	4.4388 eV (279.32 nm)	0.0021
S ₂₃	MO107 → MO115 (18%) MO113 → MO119 (33%) MO113 → MO120 (49%)	4.4418 eV (279.13 nm)	0.0066
S ₂₄	MO113 → MO119	4.4620 eV (277.87 nm)	0.0263
S ₂₅	MO106 → MO114 (44%) MO106 → MO115 (45%) MO112 → MO117 (11%)	4.6168 eV (268.55 nm)	0.1327
S ₂₆	MO106 → MO115	4.6651 eV (265.77 nm)	0.0253
S ₂₇	MO107 → MO116 (16%) MO110 → MO116 (9%) MO111 → MO116 (23%) MO111 → MO117 (7%) MO112 → MO117 (26%) MO113 → MO119 (12%) MO113 → MO121 (7%)	4.6841 eV (264.69 nm)	0.1653
S ₂₈	MO107 → MO115 (13%) MO107 → MO116 (43%) MO107 → MO117 (9%) MO108 → MO116 (9%) MO110 → MO116 (26%)	4.7107 eV (263.20 nm)	0.0640
S ₂₉	MO113 → MO121	4.7701 eV (259.92 nm)	0.0690
S ₃₀	MO111 → MO116 (69%) MO111 → MO117 (15%) MO113 → MO121 (16%)	4.9446 eV (250.75 nm)	0.4937

HOMO: MO113, LUMO: MO114

Table S46 Calculated excited states of **1NO₂·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
S ₁	MO113 → MO114	2.9572 eV (419.27 nm)	0.2574
S ₂	MO113 → MO115	3.0872 eV (401.61 nm)	0.1925
S ₃	MO113 → MO114 (16%) MO113 → MO116 (84%)	3.1618 eV (392.13 nm)	0.0403
S ₄	MO112 → MO114 (44%) MO113 → MO117 (56%)	3.7197 eV (333.32 nm)	0.0314
S ₅	MO105 → MO114 (56%) MO105 → MO116 (18%) MO106 → MO114 (14%) MO113 → MO117 (12%)	3.7798 eV (328.02 nm)	0.0098
S ₆	MO106 → MO114 (18%) MO106 → MO115 (54%) MO107 → MO115 (28%)	3.7926 eV (326.91 nm)	0.0002
S ₇	MO105 → MO114 (16%) MO112 → MO114 (60%) MO112 → MO116 (24%)	3.8730 eV (320.13 nm)	0.3339
S ₈	MO111 → MO114 (20%) MO111 → MO115 (45%) MO112 → MO115 (35%)	3.9583 eV (313.23 nm)	0.0204
S ₉	MO112 → MO115	4.0094 eV (309.23 nm)	0.1661
S ₁₀	MO109 → MO114 (17%) MO110 → MO114 (69%) MO110 → MO116 (14%)	4.0259 eV (307.96 nm)	0.0162
S ₁₁	MO112 → MO115 (25%) MO112 → MO116 (47%) MO113 → MO117 (28%)	4.0520 eV (305.98 nm)	0.1530
S ₁₂	MO111 → MO114	4.2229 eV (293.60 nm)	0.0087
S ₁₃	MO109 → MO114 (74%) MO109 → MO115 (13%) MO113 → MO118 (13%)	4.3126 eV (287.49 nm)	0.0972
S ₁₄	MO102 → MO114 (75%) MO102 → MO116 (25%)	4.3760 eV (283.33 nm)	0.0011
S ₁₅	MO104 → MO114 (24%) MO104 → MO115 (76%)	4.3885 eV (282.52 nm)	0.0624
S ₁₆	MO108 → MO116 (13%) MO111 → MO114 (20%) MO111 → MO116 (67%)	4.3960 eV (282.04 nm)	0.0069
S ₁₇	MO104 → MO114 (9%) MO104 → MO115 (28%) MO109 → MO115 (44%) MO109 → MO116 (19%)	4.4018 eV (281.67 nm)	0.1871
S ₁₈	MO108 → MO114 (38%) MO111 → MO116 (21%) MO112 → MO117 (23%) MO113 → MO118 (18%)	4.4236 eV (280.28 nm)	0.0450
S ₁₉	MO110 → MO116	4.4944 eV (275.86 nm)	0.0003

S ₂₀	MO109 → MO114 (26%) MO109 → MO116 (74%)	4.5011 eV (275.45 nm)	0.0078
S ₂₁	MO112 → MO117 (30%) MO113 → MO118 (70%)	4.5127 eV (274.75 nm)	0.0703
S ₂₂	MO109 → MO115 (10%) MO109 → MO116 (17%) MO110 → MO115 (38%) MO110 → MO116 (35%)	4.5174 eV (274.46 nm)	0.0141
S ₂₃	MO108 → MO114 (10%) MO108 → MO115 (56%) MO108 → MO116 (34%)	4.5967 eV (269.73 nm)	0.0124
S ₂₄	MO108 → MO116 (51%) MO109 → MO116 (20%) MO113 → MO118 (29%)	4.6281 eV (267.90 nm)	0.0286
S ₂₅	MO113 → MO119	4.7358 eV (261.80 nm)	0.0103
S ₂₆	MO105 → MO114 (14%) MO107 → MO114 (43%) MO108 → MO114 (11%) MO113 → MO118 (18%) MO113 → MO120 (14%)	4.8609 eV (255.07 nm)	0.4879
S ₂₇	MO106 → MO114 (18%) MO113 → MO118 (24%) MO113 → MO120 (58%)	4.8825 eV (253.94 nm)	0.0356
S ₂₈	MO107 → MO114 (19%) MO112 → MO117 (24%) MO113 → MO119 (11%) MO113 → MO120 (46%)	4.9110 eV (252.46 nm)	0.1490
S ₂₉	MO107 → MO116 (26%) MO111 → MO117 (74%)	5.0031 eV (247.81 nm)	0.0138
S ₃₀	MO105 → MO116 (16%) MO107 → MO116 (66%) MO108 → MO117 (18%)	5.0351 eV (246.24 nm)	0.0304

HOMO: MO113, LUMO: MO114

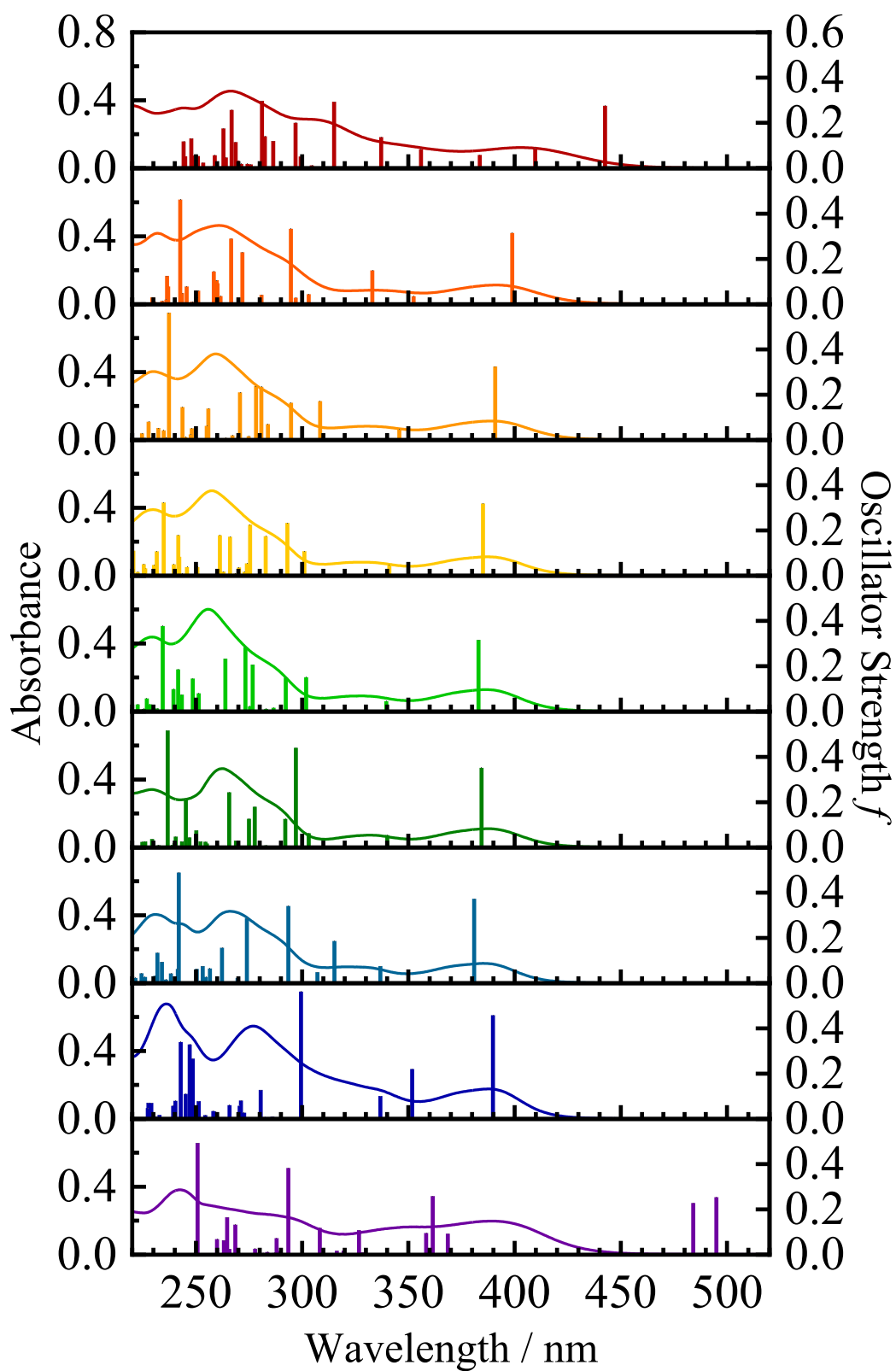


Fig. S43 Comparisons of the absorption spectra and oscillator strengths calculated by TD-DFT (perpendicular bars) of **1R** in acetonitrile. The colors correspond to those indicated in Fig. 1.

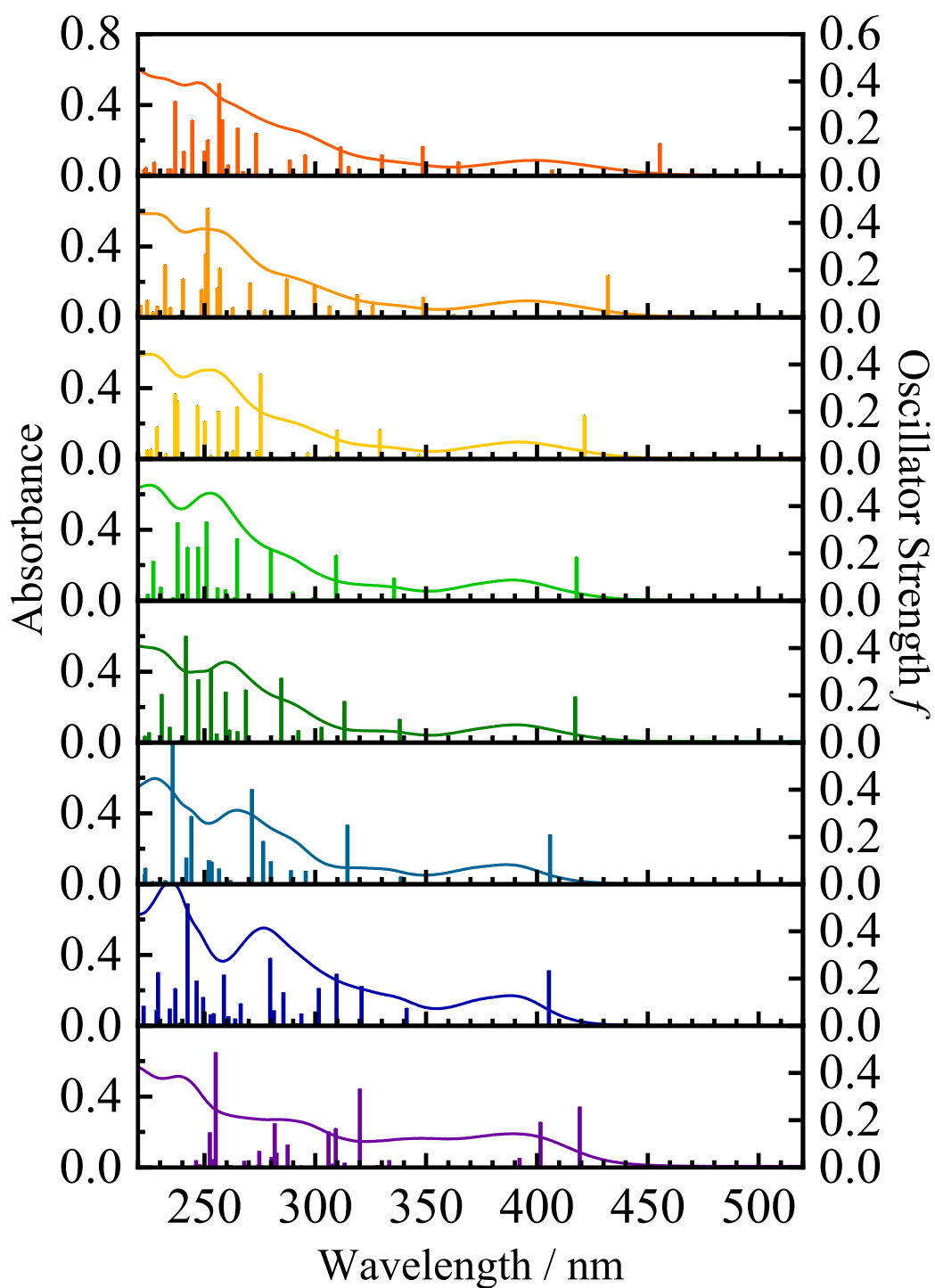


Fig. S44 Comparisons of the absorption spectra ($[\text{TFA}] = 10 \text{ mM}$) and oscillator strengths calculated by TD-DFT (perpendicular bars) of $1\text{R}\cdot\text{H}^+$ in acetonitrile. The colors correspond to those indicated in Fig. 1.

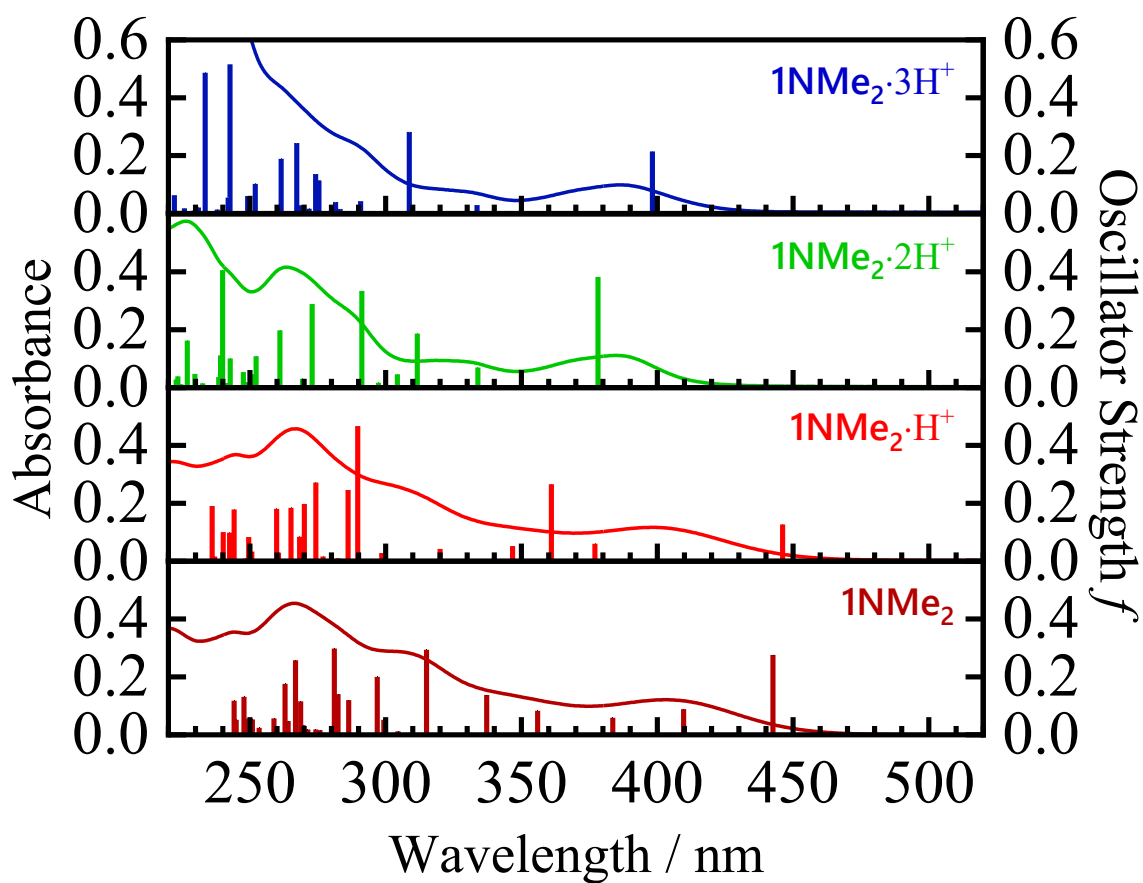


Fig. S45 Comparisons of the absorption spectra ($[\text{TFA}] = 0, 0.1, 10, 100 \text{ mM}$) and oscillator strengths calculated by TD-DFT (perpendicular bars) of $1\text{NMe}_2 \cdot \text{H}^+$ in acetonitrile.

Table S47 Calculated excited states of S₁-optimized geometry of **1NMe₂** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO115 → MO116	1.1328 eV (1094.50 nm)	0.0000
S ₁	MO115 → MO116	1.7810 eV (696.16 nm)	0.3308
T ₂	MO111 → MO116 (14%)	2.1502 eV (576.61 nm)	0.0000
	MO114 → MO116 (70%)		
	MO115 → MO118 (16%)		
T ₃	MO114 → MO116 (38%)	2.1932 eV (565.30 nm)	0.0000
	MO115 → MO117 (62%)		
S ₂	MO114 → MO116	2.2847 eV (542.68 nm)	0.0997
S ₃	MO115 → MO117	2.5918 eV (478.37 nm)	0.0306

HOMO: MO115, LUMO: MO116

Table S48 Calculated excited states of S₁-optimized geometry of **1NMe₂·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO115 → MO116	1.2383 eV (1001.21 nm)	0.0000
S ₁	MO115 → MO116	1.7993 eV (689.07 nm)	0.2584
T ₂	MO113 → MO116 (13%)	2.2105 eV (560.88 nm)	0.0000
	MO115 → MO117 (72%)		
	MO115 → MO120 (15%)		
T ₃	MO114 → MO116 (68%)	2.4193 eV (512.48 nm)	0.0000
	MO115 → MO116 (18%)		
	MO115 → MO117 (14%)		
S ₂	MO114 → MO116 (82%)	2.6488 eV (468.08 nm)	0.0571
	MO115 → MO118 (18%)		
S ₃	MO114 → MO116 (43%)	2.6926 eV (460.45 nm)	0.1179
	MO115 → MO117 (57%)		

HOMO: MO115, LUMO: MO116

Table S49 Calculated excited states of S₁-optimized geometry of **1NMe₂·2H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO115 → MO116	1.4636 eV (847.14 nm)	0.0000
S ₁	MO115 → MO116	2.1288 eV (582.42 nm)	0.3329
T ₂	MO115 → MO117	2.3992 eV (516.78 nm)	0.0000
T ₃	MO112 → MO116 (10%)	2.7444 eV (451.77 nm)	0.0000
	MO114 → MO116 (12%)		
	MO115 → MO117 (11%)		
	MO115 → MO118 (57%)		
	MO115 → MO119 (10%)		
S ₂	MO115 → MO117 (81%)	2.9854 eV (415.30 nm)	0.0371
	MO115 → MO119 (19%)		
S ₃	MO115 → MO118	3.2568 eV (380.70 nm)	0.2588

HOMO: MO115, LUMO: MO116

Table S50 Calculated excited states of S₁-optimized geometry of **1NMe₂·3H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO115 → MO116	2.0018 eV (619.37 nm)	0.0000
T ₂	MO115 → MO116 (25%)	2.3988 eV (516.87 nm)	0.0000
	MO115 → MO117 (75%)		
S ₁	MO115 → MO116	2.6281 eV (471.76 nm)	0.2286
T ₃	MO114 → MO116 (75%)	3.0466 eV (406.96 nm)	0.0000
	MO115 → MO117 (25%)		
S ₂	MO115 → MO117	3.3647 eV (368.49 nm)	0.1341
S ₃	MO114 → MO116 (72%)	3.6585 eV (338.90 nm)	0.1874
	MO115 → MO117 (28%)		

HOMO: MO115, LUMO: MO116

Table S51 Calculated excited states of S₁-optimized geometry of **1OMe** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO107 → MO108	1.2608 eV (983.35 nm)	0.0000
S ₁	MO107 → MO108	1.9547 eV (634.28 nm)	0.3134
T ₂	MO107 → MO109 (80%) MO107 → MO110 (20%)	2.2755 eV (544.88 nm)	0.0000
T ₃	MO103 → MO108 (15%) MO106 → MO108 (63%) MO107 → MO109 (22%)	2.6431 eV (469.09 nm)	0.0000
S ₂	MO107 → MO109	2.7623 eV (448.85 nm)	0.0111
S ₃	MO106 → MO108 (80%) MO107 → MO109 (20%)	2.9332 eV (422.69 nm)	0.1092

HOMO: MO107, LUMO: MO108

Table S52 Calculated excited states of S₁-optimized geometry of **1OMe**·H⁺ in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO107 → MO108	1.7498 eV (708.58 nm)	0.0000
S ₁	MO107 → MO108	2.1987 eV (563.90 nm)	0.1514
T ₂	MO102 → MO108 (20%) MO107 → MO109 (80%)	2.2923 eV (540.88 nm)	0.0000
T ₃	MO106 → MO108 (77%) MO107 → MO109 (23%)	2.6313 eV (471.18 nm)	0.0000
S ₂	MO106 → MO108	2.7717 eV (447.32 nm)	0.0414
S ₃	MO107 → MO109	3.0690 eV (403.99 nm)	0.1783

HOMO: MO107, LUMO: MO108

Table S53 Calculated excited states of S₁-optimized geometry of **1Me** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO099 → MO100	1.3028 eV (951.69 nm)	0.0000
S ₁	MO099 → MO100	2.0050 eV (618.37 nm)	0.3153
T ₂	MO099 → MO101 (81%) MO099 → MO102 (19%)	2.3070 eV (537.42 nm)	0.0000
T ₃	MO096 → MO100 (19%) MO099 → MO102 (81%)	2.7183 eV (456.11 nm)	0.0000
S ₂	MO099 → MO101	2.8186 eV (439.87 nm)	0.0212
S ₃	MO097 → MO100 (15%) MO098 → MO100 (61%) MO099 → MO101 (11%) MO099 → MO102 (13%)	3.1813 eV (389.73 nm)	0.0640

HOMO: MO099, LUMO: MO100

Table S54 Calculated excited states of S₁-optimized geometry of **1Me·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO099 → MO100	1.8275 eV (678.45 nm)	0.0000
T ₂	MO099 → MO101	2.3213 eV (534.12 nm)	0.0000
S ₁	MO099 → MO100	2.3553 eV (526.41 nm)	0.1827
T ₃	MO098 → MO100	2.8671 eV (432.44 nm)	0.0000
S ₂	MO098 → MO100 (70%) MO099 → MO101 (30%)	3.1057 eV (399.21 nm)	0.0232
S ₃	MO099 → MO101	3.2290 eV (383.97 nm)	0.2010

HOMO: MO099, LUMO: MO100

Table S55 Calculated excited states of S₁-optimized geometry of **1H** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO091 → MO092	1.3441 eV (922.41 nm)	0.0000
S ₁	MO091 → MO092	2.0446 eV (606.41 nm)	0.3041
T ₂	MO087 → MO092 (12%) MO091 → MO093 (72%) MO091 → MO094 (16%)	2.3367 eV (530.60 nm)	0.0000
T ₃	MO091 → MO094	2.7630 eV (448.73 nm)	0.0000
S ₂	MO091 → MO093	2.8673 eV (432.41 nm)	0.0233
S ₃	MO088 → MO092 (11%) MO089 → MO092 (53%) MO090 → MO092 (36%)	3.2487 eV (381.65 nm)	0.0114

HOMO: MO091, LUMO: MO092

Table S56 Calculated excited states of S₁-optimized geometry of **1H·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO087 → MO092 (13%) MO091 → MO092 (87%)	1.8700 eV (663.01 nm)	0.0000
T ₂	MO091 → MO092 (16%) MO091 → MO093 (84%)	2.3379 eV (530.32 nm)	0.0000
S ₁	MO091 → MO092	2.4212 eV (512.07 nm)	0.1819
T ₃	MO086 → MO092 (33%) MO090 → MO092 (49%) MO091 → MO093 (18%)	2.9579 eV (419.16 nm)	0.0000
S ₂	MO091 → MO093	3.2292 eV (383.95 nm)	0.0767
S ₃	MO090 → MO092 (66%) MO091 → MO093 (34%)	3.4004 eV (364.61 nm)	0.1678

HOMO: MO091, LUMO: MO092

Table S57 Calculated excited states of S₁-optimized geometry of **1F** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO099 → MO100	1.3522 eV (916.91 nm)	0.0000
S ₁	MO099 → MO100	2.0491 eV (605.06 nm)	0.2985
T ₂	MO096 → MO100 (12%)	2.3411 eV (529.60 nm)	0.0000
	MO099 → MO101 (71%)		
	MO099 → MO102 (17%)		
T ₃	MO096 → MO100 (14%)	2.7630 eV (448.74 nm)	0.0000
	MO098 → MO102 (14%)		
	MO099 → MO102 (72%)		
S ₂	MO099 → MO101	2.8750 eV (431.25 nm)	0.0234
S ₃	MO098 → MO100 (80%)	3.2606 eV (380.25 nm)	0.0331
	MO099 → MO102 (20%)		

HOMO: MO099, LUMO: MO100

Table S58 Calculated excited states of S₁-optimized geometry of **1F·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO099 → MO100	1.8859 eV (657.44 nm)	0.0000
T ₂	MO098 → MO100 (14%)	2.3462 eV (528.44 nm)	0.0000
	MO099 → MO101 (86%)		
S ₁	MO099 → MO100	2.4352 eV (509.12 nm)	0.1842
T ₃	MO098 → MO100	2.9432 eV (421.25 nm)	0.0000
S ₂	MO098 → MO100 (48%)	3.2255 eV (384.38 nm)	0.0404
	MO099 → MO101 (52%)		
S ₃	MO098 → MO100	3.3400 eV (371.21 nm)	0.1802

HOMO: MO099, LUMO: MO100

Table S59 Calculated excited states of S₁-optimized geometry of **1Cl** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO107 → MO108	1.3681 eV (906.26 nm)	0.0000
S ₁	MO107 → MO108	2.0522 eV (604.17 nm)	0.3253
T ₂	MO107 → MO109	2.3464 eV (528.41 nm)	0.0000
T ₃	MO107 → MO109 (25%) MO107 → MO110 (75%)	2.7316 eV (453.90 nm)	0.0000
S ₂	MO106 → MO108 (15%) MO107 → MO109 (85%)	2.8870 eV (429.46 nm)	0.0254
S ₃	MO105 → MO108 (32%) MO106 → MO108 (47%) MO107 → MO110 (21%)	3.2582 eV (380.53 nm)	0.0204

HOMO: MO107, LUMO: MO108

Table S60 Calculated excited states of S₁-optimized geometry of **1Cl·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO107 → MO108 (86%) MO107 → MO109 (14%)	1.9017 eV (651.95 nm)	0.0000
T ₂	MO107 → MO109	2.3422 eV (529.36 nm)	0.0000
S ₁	MO107 → MO108	2.4558 eV (504.87 nm)	0.1990
T ₃	MO106 → MO108	2.9425 eV (421.36 nm)	0.0000
S ₂	MO106 → MO108 (52%) MO107 → MO109 (48%)	3.2222 eV (384.78 nm)	0.0433
S ₃	MO107 → MO109	3.3234 eV (373.07 nm)	0.2117

HOMO: MO107, LUMO: MO108

Table S61 Calculated excited states of S₁-optimized geometry of **1CF₃** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO123 → MO124	1.4344 eV (864.39 nm)	0.0000
S ₁	MO123 → MO124	2.1035 eV (589.42 nm)	0.3281
T ₂	MO123 → MO125 (87%) MO123 → MO134 (13%)	2.3813 eV (520.67 nm)	0.0000
T ₃	MO123 → MO126	2.7231 eV (455.31 nm)	0.0000
S ₂	MO123 → MO125	2.9597 eV (418.90 nm)	0.0334
S ₃	MO122 → MO124 (20%) MO123 → MO126 (80%)	3.2333 eV (383.46 nm)	0.2354

HOMO: MO123, LUMO: MO124

Table S62 Calculated excited states of S₁-optimized geometry of **1CF₃·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO123 → MO124 (82%) MO123 → MO125 (18%)	1.9692 eV (629.60 nm)	0.0000
T ₂	MO122 → MO124 (14%) MO123 → MO125 (74%) MO123 → MO132 (12%)	2.3790 eV (521.16 nm)	0.0000
S ₁	MO123 → MO124	2.5727 eV (481.93 nm)	0.2158
T ₃	MO122 → MO124	3.0237 eV (410.04 nm)	0.0000
S ₂	MO122 → MO124 (27%) MO123 → MO125 (73%)	3.3326 eV (372.03 nm)	0.1224
S ₃	MO122 → MO124	3.5915 eV (345.22 nm)	0.1846

HOMO: MO123, LUMO: MO124

Table S63 Calculated excited states of S₁-optimized geometry of **1CN** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO103 → MO104	1.4596 eV (849.43 nm)	0.0000
S ₁	MO103 → MO104	2.1001 eV (590.38 nm)	0.3819
T ₂	MO102 → MO104 (14%) MO103 → MO106 (65%) MO103 → MO107 (21%)	2.3436 eV (529.03 nm)	0.0000
T ₃	MO102 → MO104 (11%) MO102 → MO106 (10%) MO103 → MO105 (43%) MO103 → MO106 (36%)	2.5253 eV (490.98 nm)	0.0000
S ₂	MO103 → MO105 (74%) MO103 → MO106 (26%)	2.9279 eV (423.46 nm)	0.4492
S ₃	MO103 → MO106 (52%) MO103 → MO107 (48%)	3.0007 eV (413.19 nm)	0.0260

HOMO: MO103, LUMO: MO104

Table S64 Calculated excited states of S₁-optimized geometry of **1CN·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO103 → MO104	1.9744 eV (627.94 nm)	0.0000
T ₂	MO100 → MO104 (11%) MO103 → MO104 (22%) MO103 → MO105 (67%)	2.3512 eV (527.33 nm)	0.0000
S ₁	MO103 → MO104	2.5877 eV (479.13 nm)	0.2521
T ₃	MO102 → MO104 (63%) MO103 → MO105 (15%) MO103 → MO107 (22%)	2.9880 eV (414.95 nm)	0.0000
S ₂	MO103 → MO105	3.2669 eV (379.52 nm)	0.1977
S ₃	MO102 → MO104 (50%) MO103 → MO105 (14%) MO103 → MO106 (27%) MO103 → MO107 (9%)	3.5238 eV (351.85 nm)	0.0611

HOMO: MO103, LUMO: MO104

Table S65 Calculated excited states of S₁-optimized geometry of **1NO₂** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO113 → MO114	1.7108 eV (724.71 nm)	0.0000
S ₁	MO113 → MO114	1.9272 eV (643.34 nm)	0.1767
T ₂	MO113 → MO115 (82%) MO113 → MO117 (18%)	2.0417 eV (607.25 nm)	0.0000
T ₃	MO110 → MO114	2.3578 eV (525.84 nm)	0.0000
S ₂	MO113 → MO115	2.4148 eV (513.44 nm)	0.3363
S ₃	MO105 → MO114 (15%) MO112 → MO114 (85%)	3.0821 eV (402.27 nm)	0.1708

HOMO: MO113, LUMO: MO114

Table S66 Calculated excited states of S₁-optimized geometry of **1NO₂·H⁺** in acetonitrile.

Excited State	Transition	Energy (Wavelength)	Oscillator Strength
T ₁	MO112 → MO114 (12%) MO113 → MO114 (60%) MO113 → MO115 (28%)	1.9435 eV (637.94 nm)	0.0000
T ₂	MO108 → MO114 (11%) MO113 → MO115 (47%) MO113 → MO116 (42%)	2.3215 eV (534.06 nm)	0.0000
S ₁	MO113 → MO114	2.3639 eV (524.50 nm)	0.2377
T ₃	MO113 → MO114 (19%) MO113 → MO116 (55%) MO113 → MO117 (26%)	2.3741 eV (522.23 nm)	0.0000
S ₂	MO113 → MO115 (83%) MO113 → MO116 (17%)	2.7869 eV (444.88 nm)	0.3750
S ₃	MO113 → MO116	2.9205 eV (424.53 nm)	0.1182

HOMO: MO113, LUMO: MO114

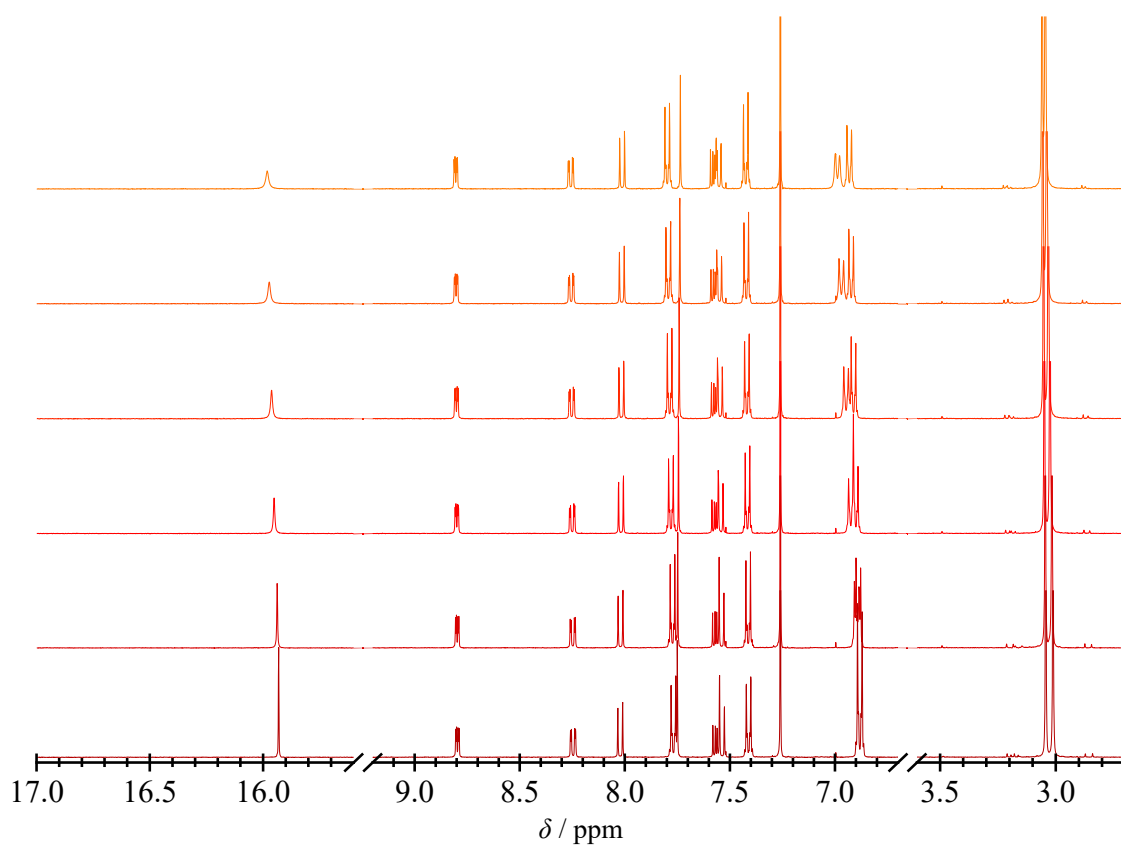


Fig. S46 ^1H -NMR spectra of **1NMe₂** in CDCl_3 ($[\text{1NMe}_2]_0 = 5.0 \text{ mM}$) upon an addition of 0, 0.040, 0.080, 0.12, 0.16 and 0.20 equivalents of TFA (dark red $\rightarrow$ orange).

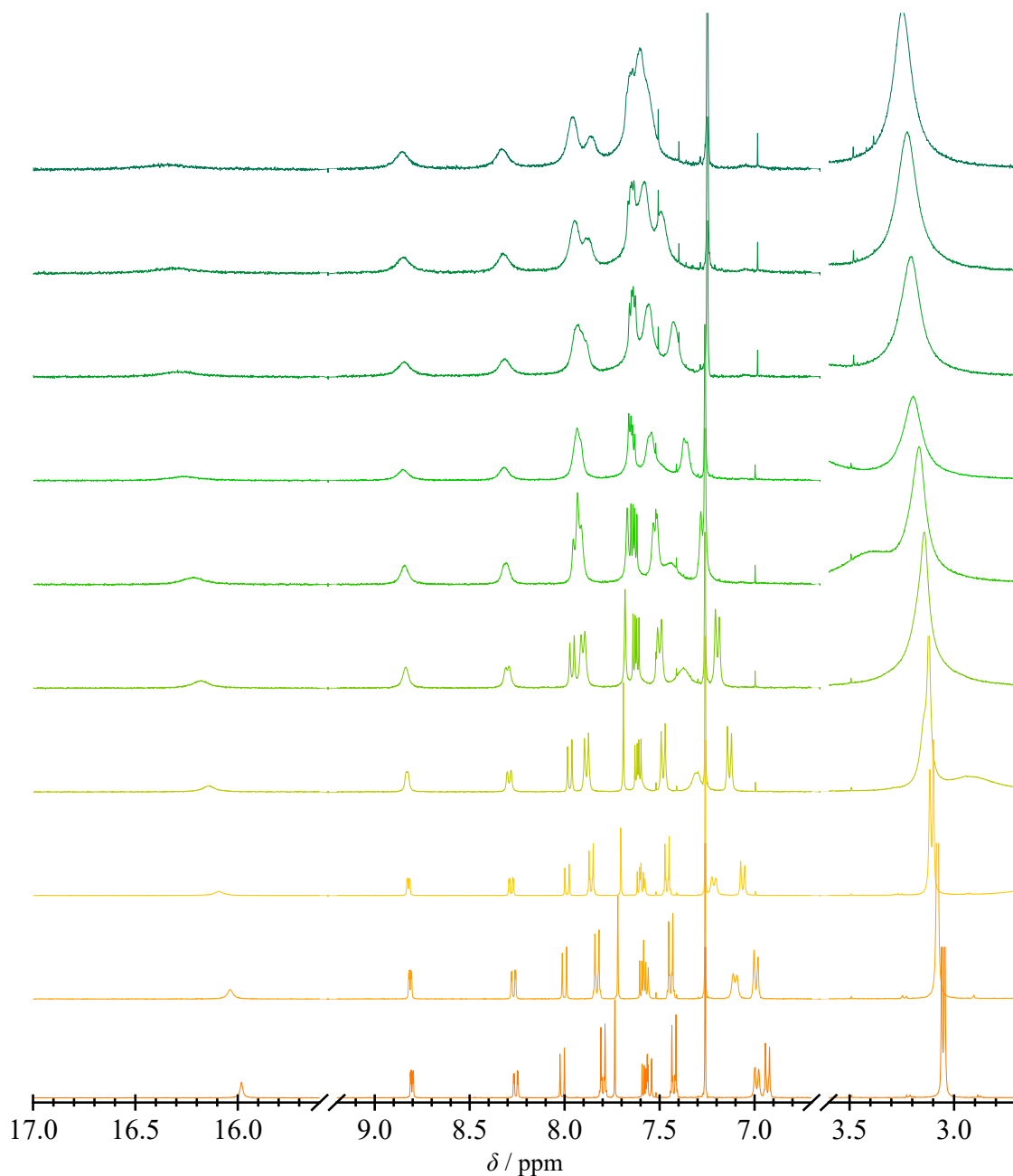


Fig. S47 ¹H-NMR spectra of **1NMe₂** in CDCl₃ ([**1NMe₂**]₀ = 5.0 mM) upon an addition of 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (orange → dark green).

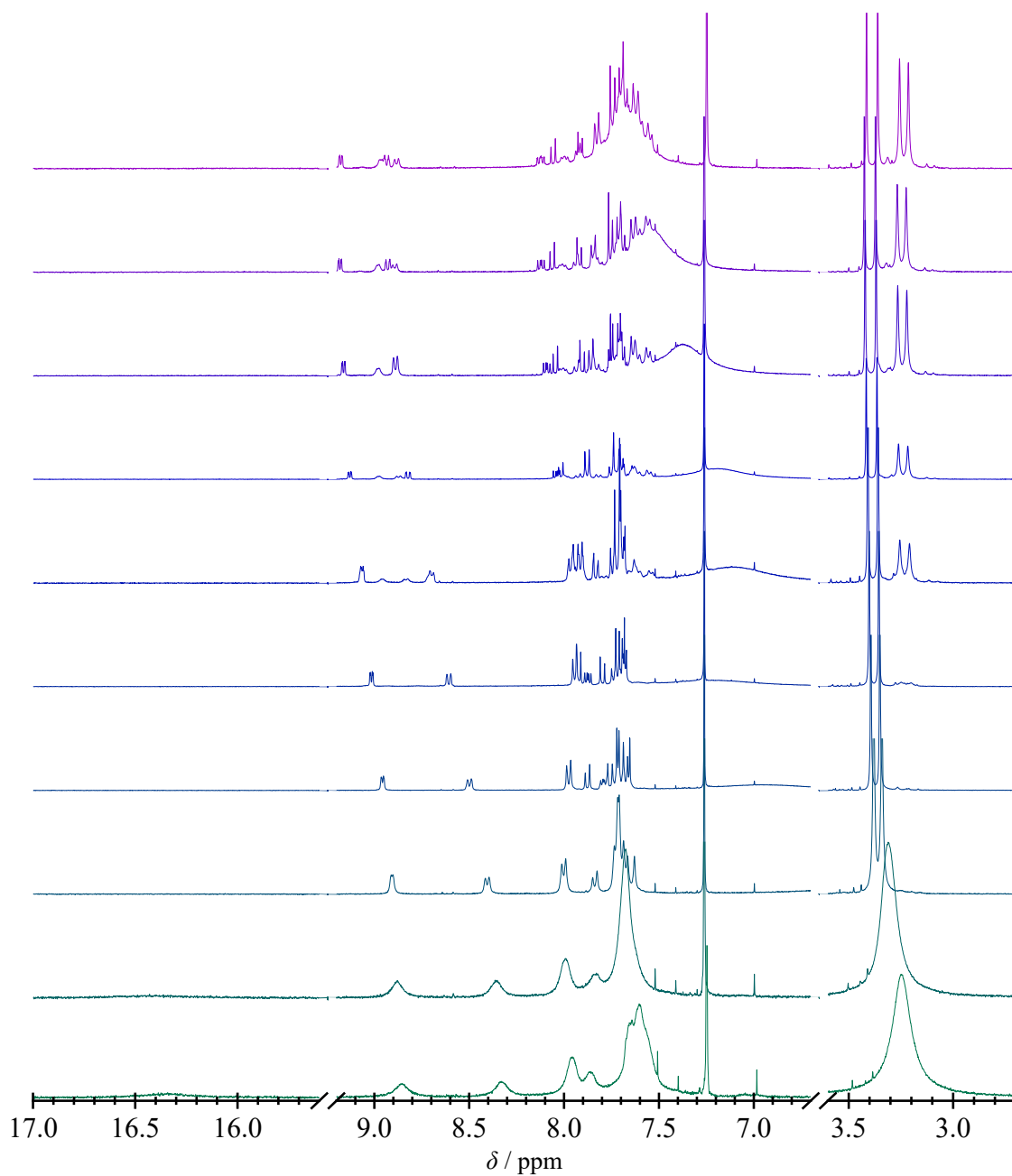


Fig. S48 ¹H-NMR spectra of **1NMe₂** in CDCl₃ ([**1NMe₂**]₀ = 5.0 mM) upon an addition of 2.0, 4.0, 6.0, 8.0, 10, 12, 14, 16, 18 and 20 equivalents of TFA (dark green → purple).

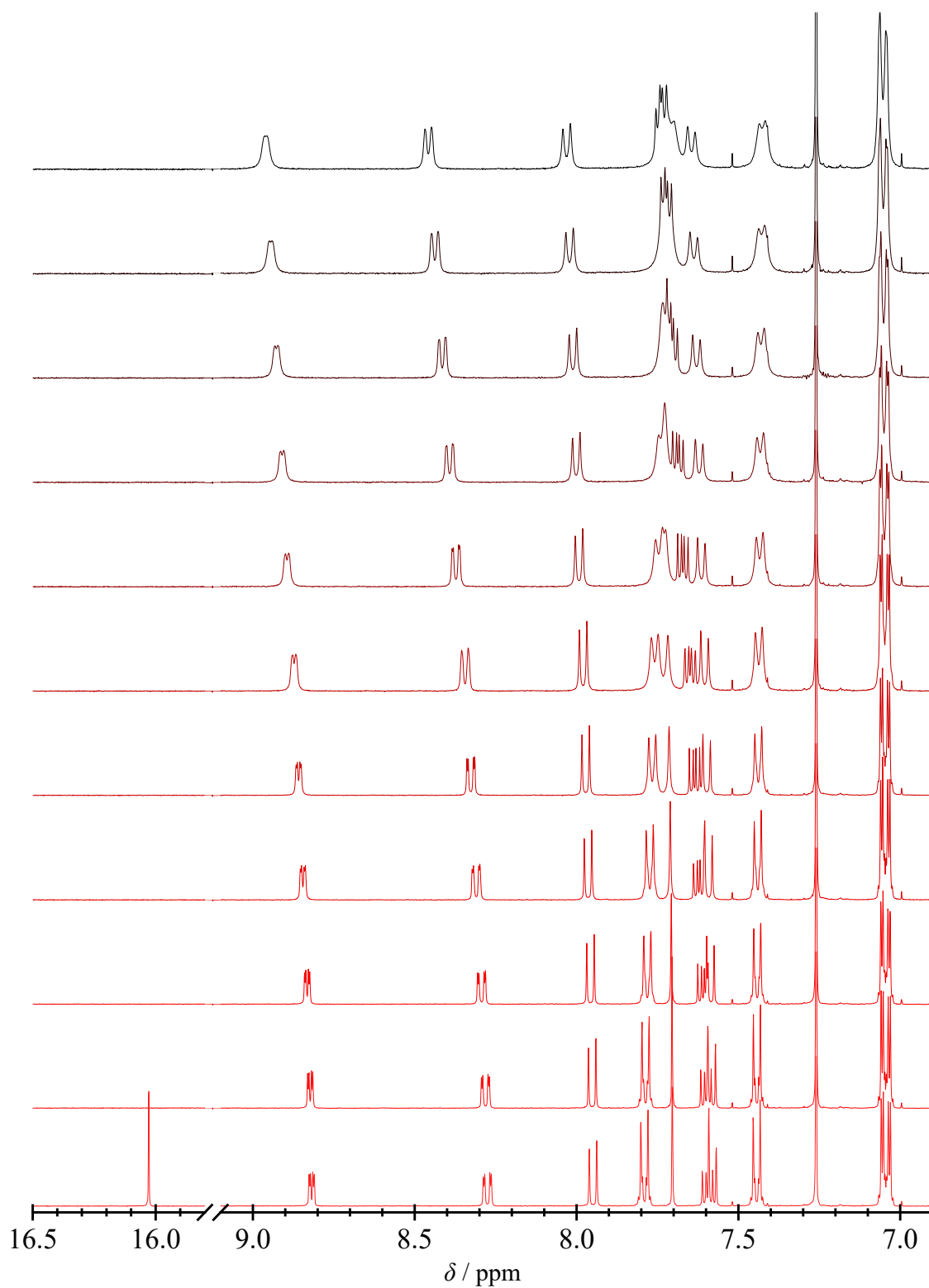


Fig. S49 ^1H -NMR spectra of **1OMe** in CDCl_3 ($[\mathbf{1OMe}]_0 = 5.0 \text{ mM}$) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (red $\rightarrow$ black).

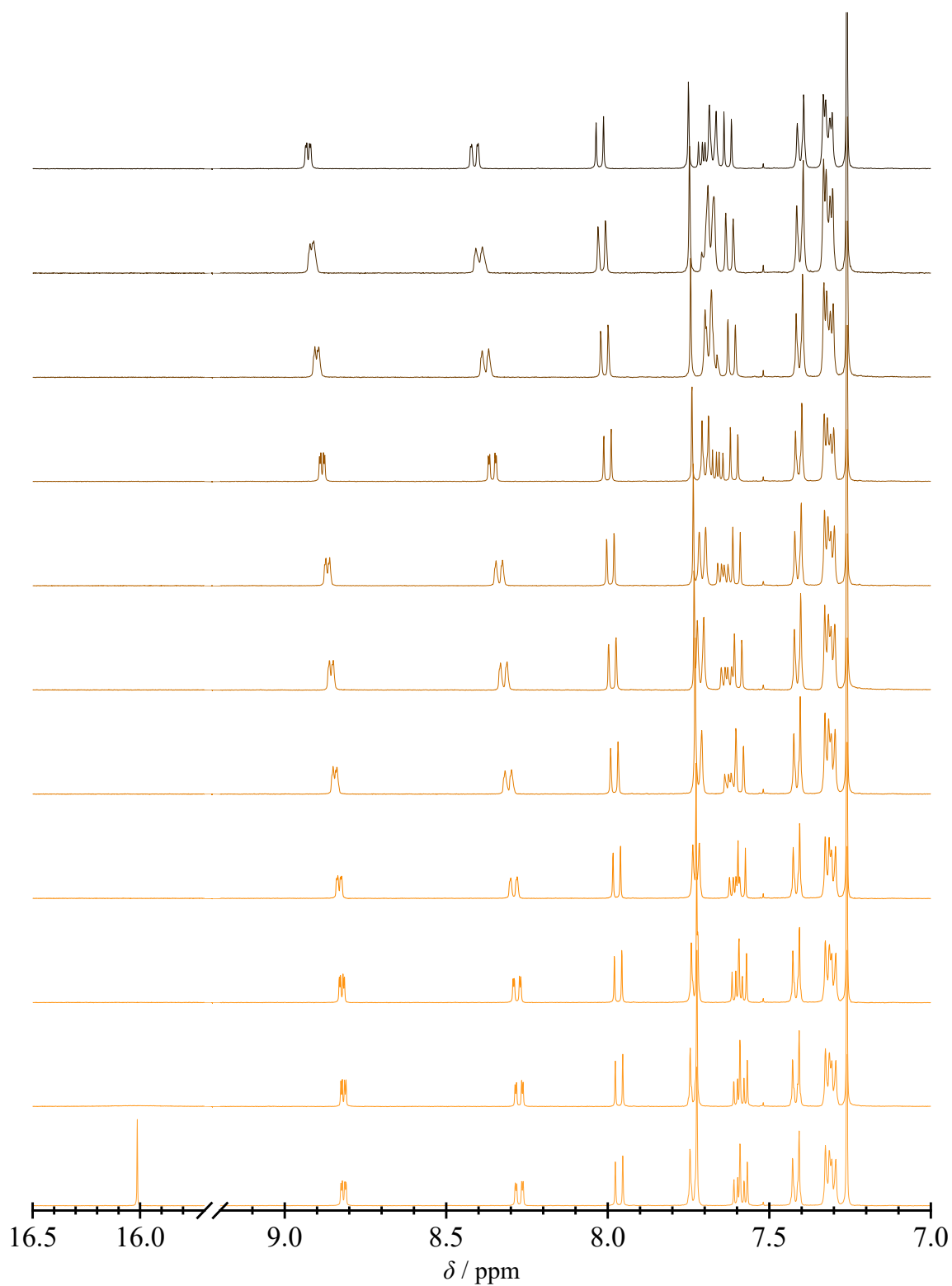


Fig. S50 ¹H-NMR spectra of **1Me** in CDCl₃ ([**1Me**]₀ = 5.3 mM) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (orange → black).

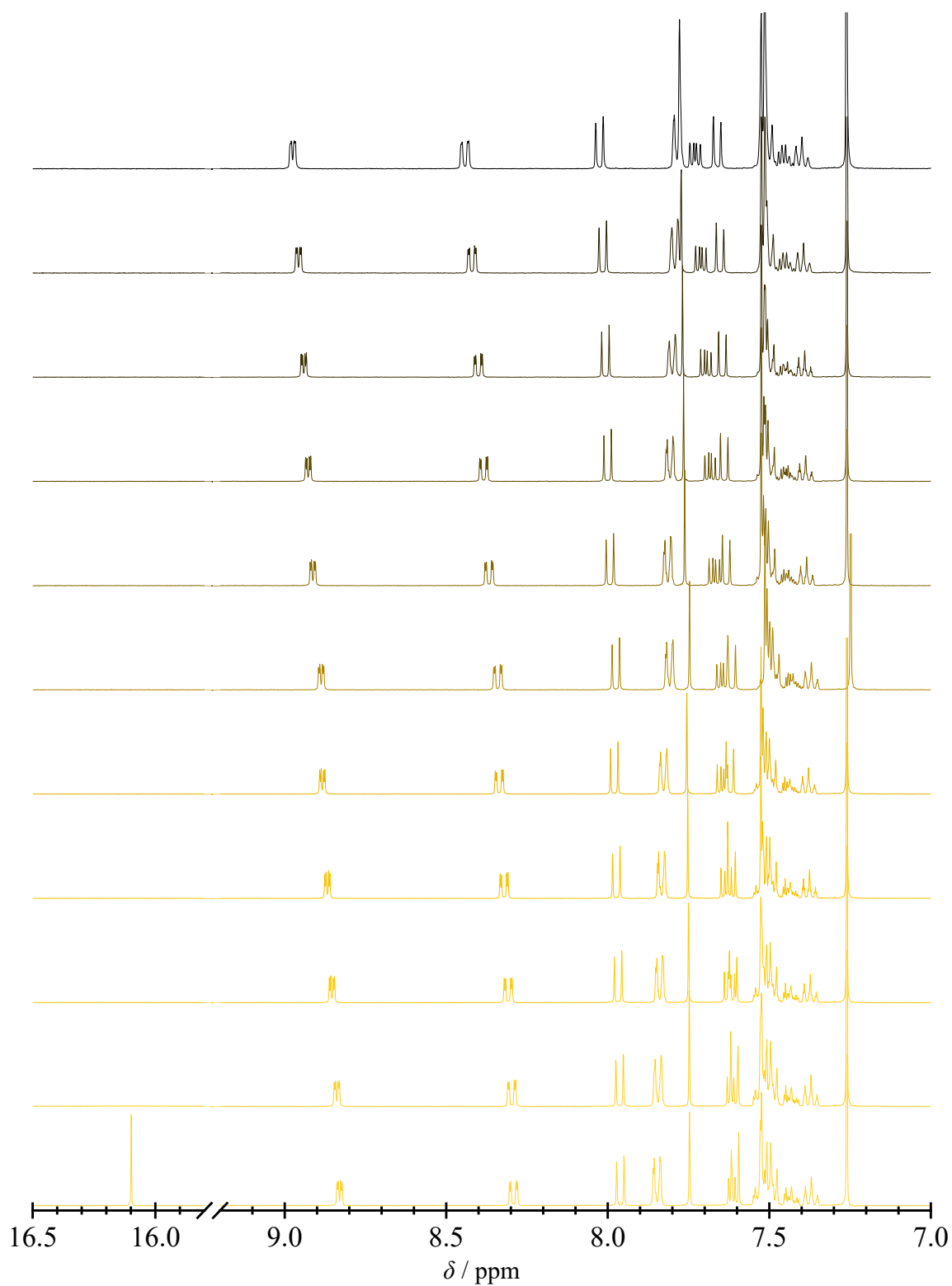


Fig. S51 ^1H -NMR spectra of **1H** in CDCl_3 ($[\text{1H}]_0 = 5.4 \text{ mM}$) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (yellow $\rightarrow$ black).

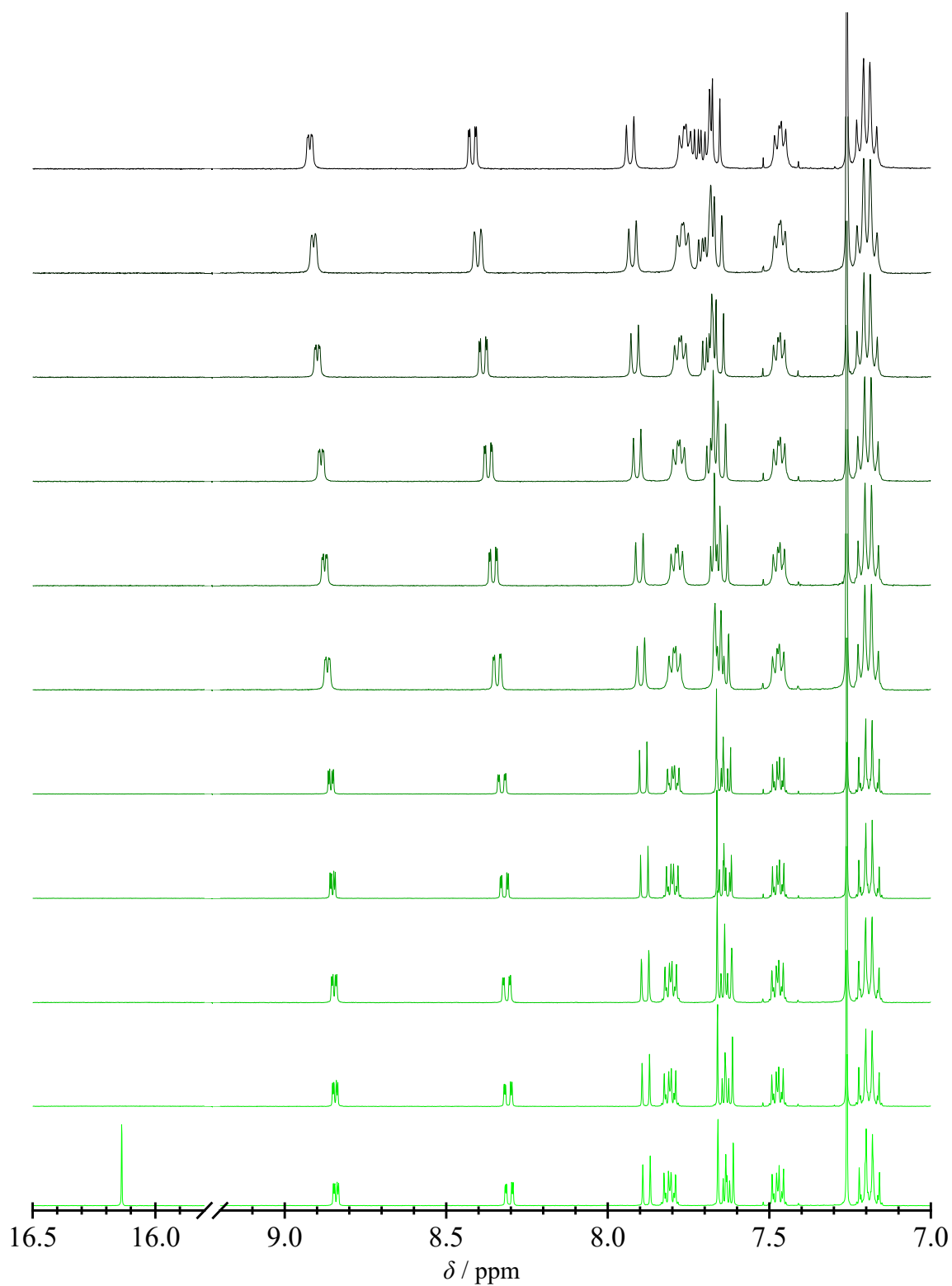


Fig. S52 ^1H -NMR spectra of **1F** in CDCl_3 ($[\text{1F}]_0 = 5.0 \text{ mM}$) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (light green $\rightarrow$ black).

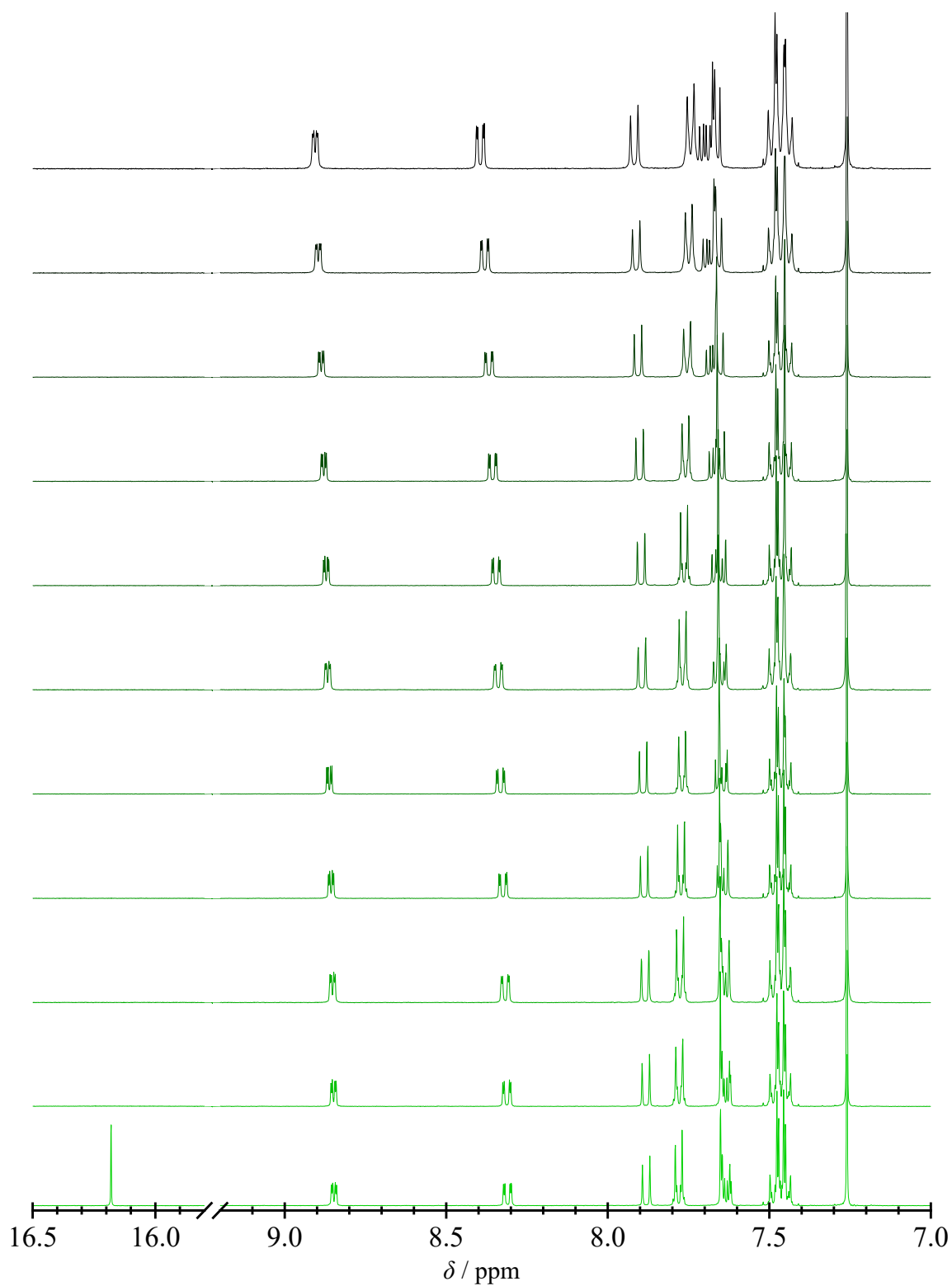


Fig. S53 ^1H -NMR spectra of **1Cl** in CDCl_3 ($[\text{1Cl}]_0 = 4.8 \text{ mM}$) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (green $\rightarrow$ black).

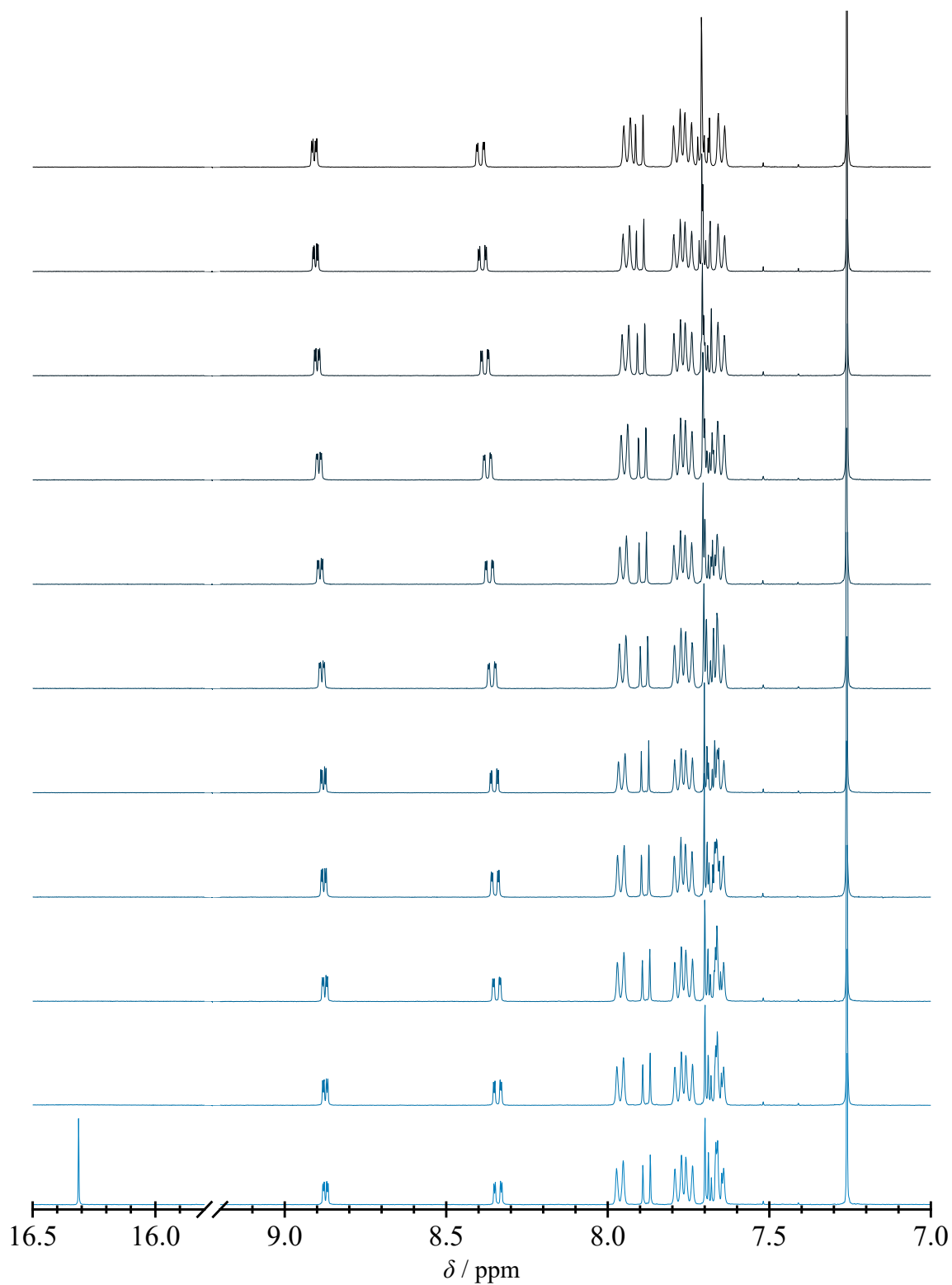


Fig. S54 ^1H -NMR spectra of 1CF_3 in CDCl_3 ($[1\text{CF}_3]_0 = 5.1 \text{ mM}$) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (light blue $\rightarrow$ black).

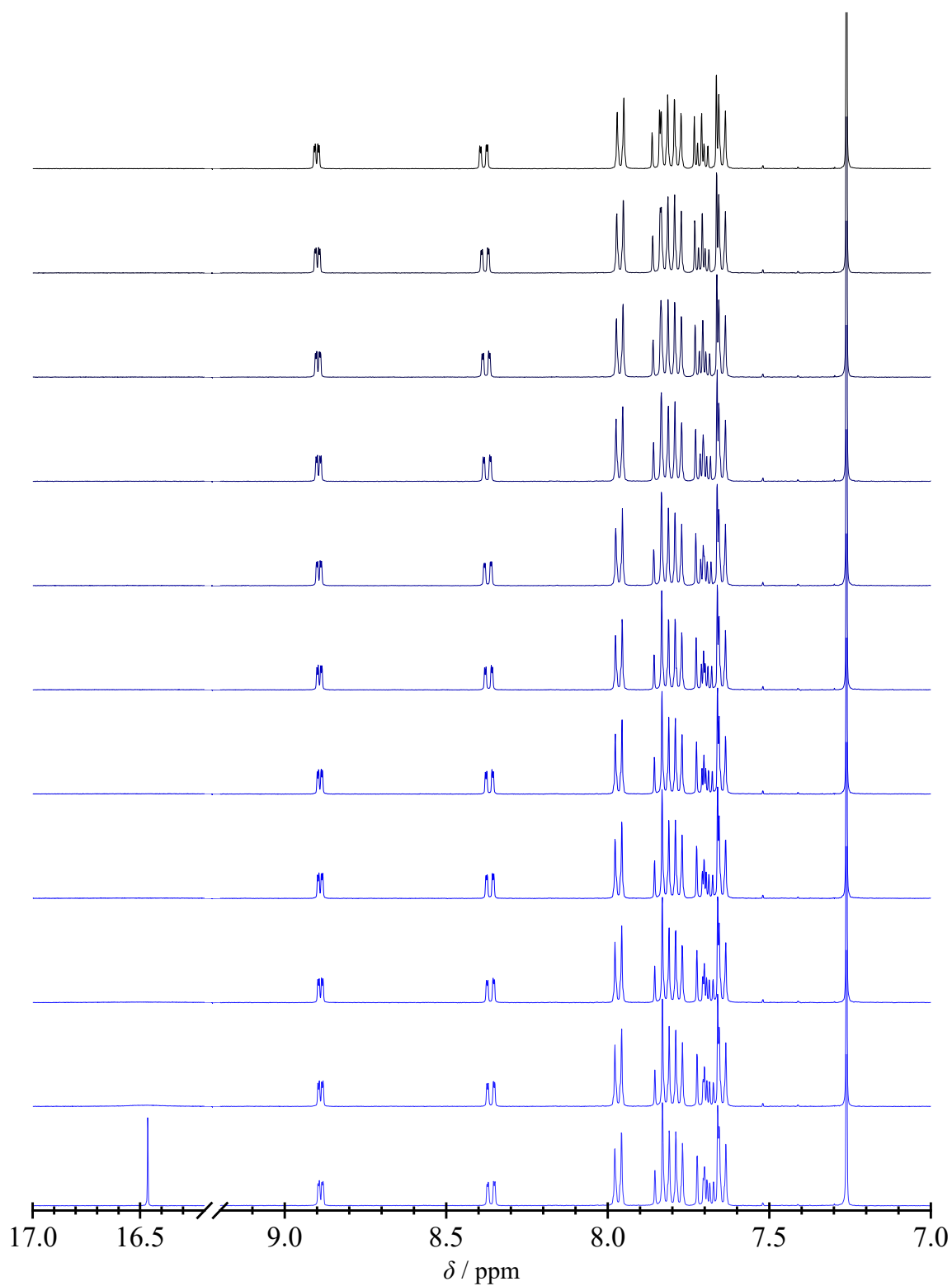


Fig. S55 ¹H-NMR spectra of **1CN** in CDCl₃ ([**1CN**]₀ = 5.1 mM) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (blue → black).

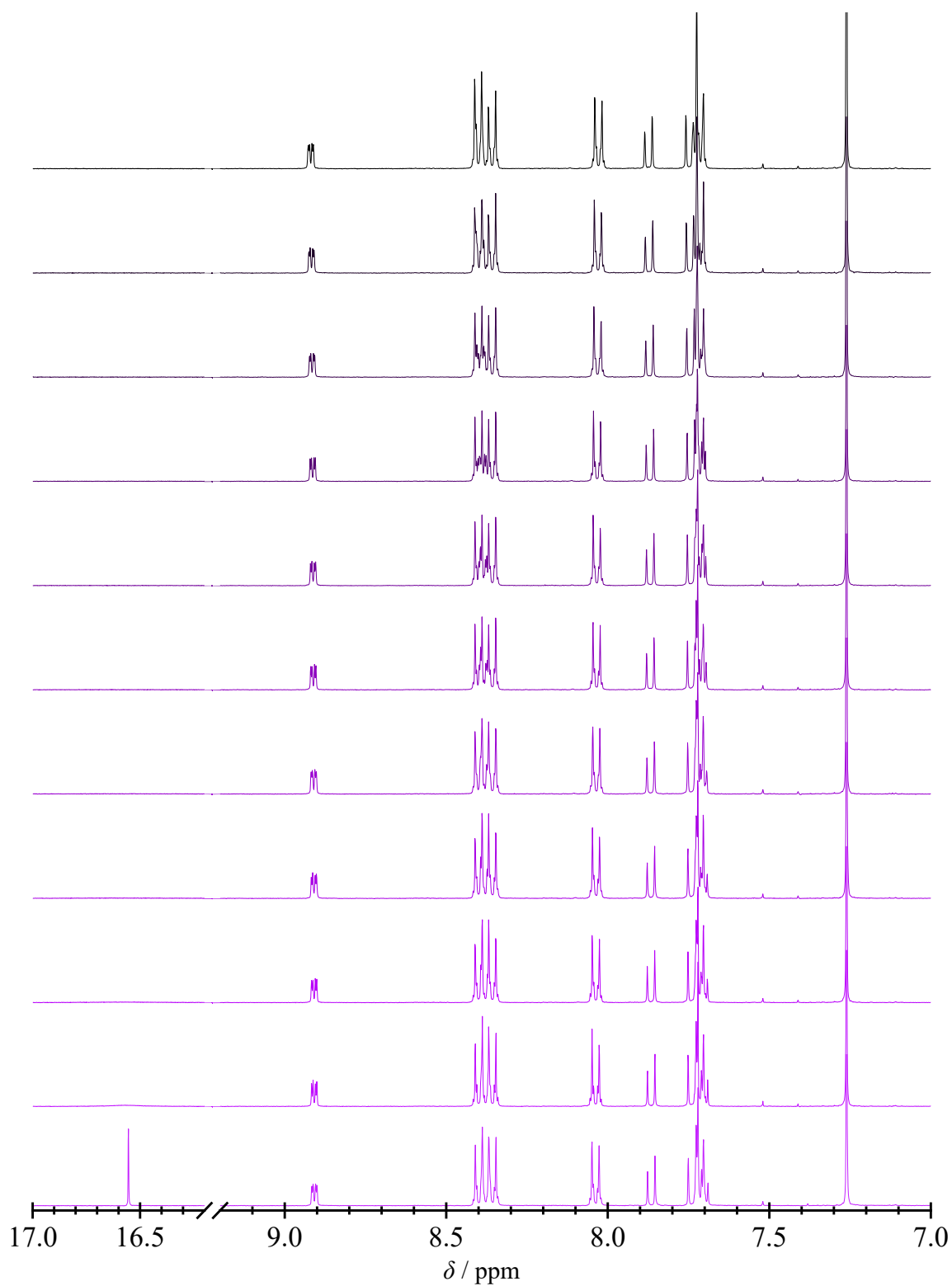


Fig. S56 ^1H -NMR spectra of **1NO₂** in CDCl_3 ($[\text{1NO}_2]_0 = 4.3 \text{ mM}$) upon an addition of 0, 0.20, 0.40, 0.60, 0.80, 1.0, 1.2, 1.4, 1.6, 1.8 and 2.0 equivalents of TFA (purple $\rightarrow$ black).

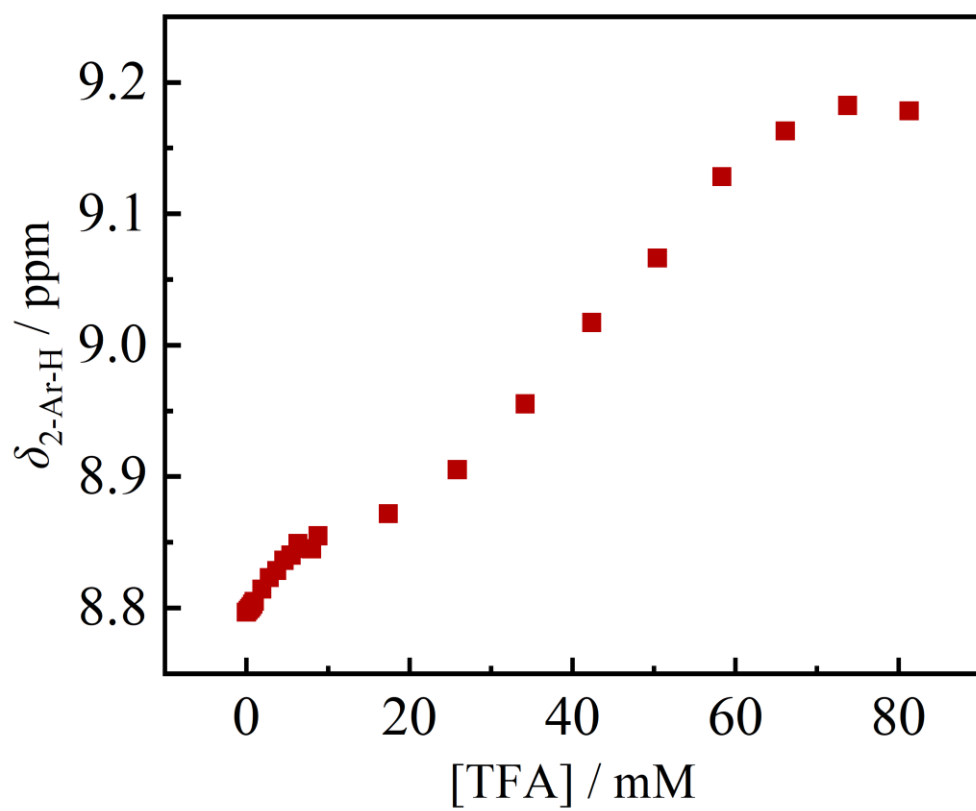
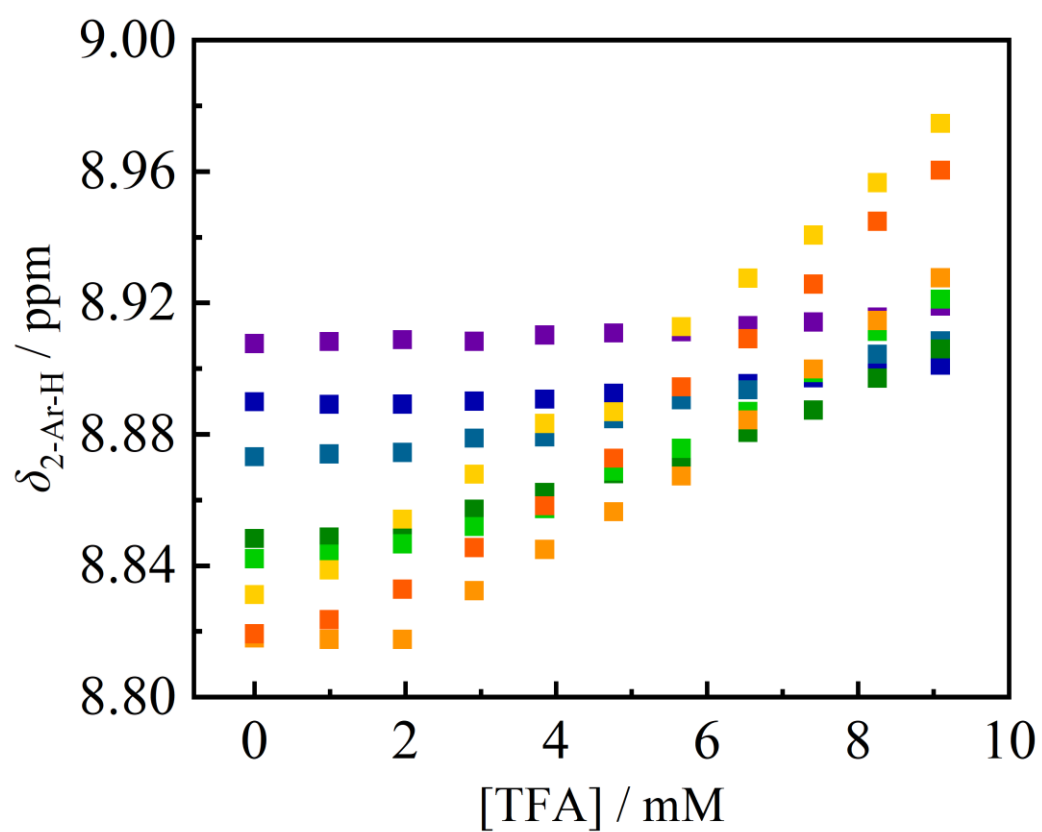


Fig. S57 TFA titrations of the chemical shift of 2-Ar-H signal of **1R** in CDCl_3 . The colors correspond to those indicated in Fig. 1.