

## Supplementary File

### Forest waste to Carbon Nanotubes: A Green Approach using NADES template for Reactive Orange 16 Dye Adsorption

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**Table S1: Table for comparison of synthesis of CNTs**

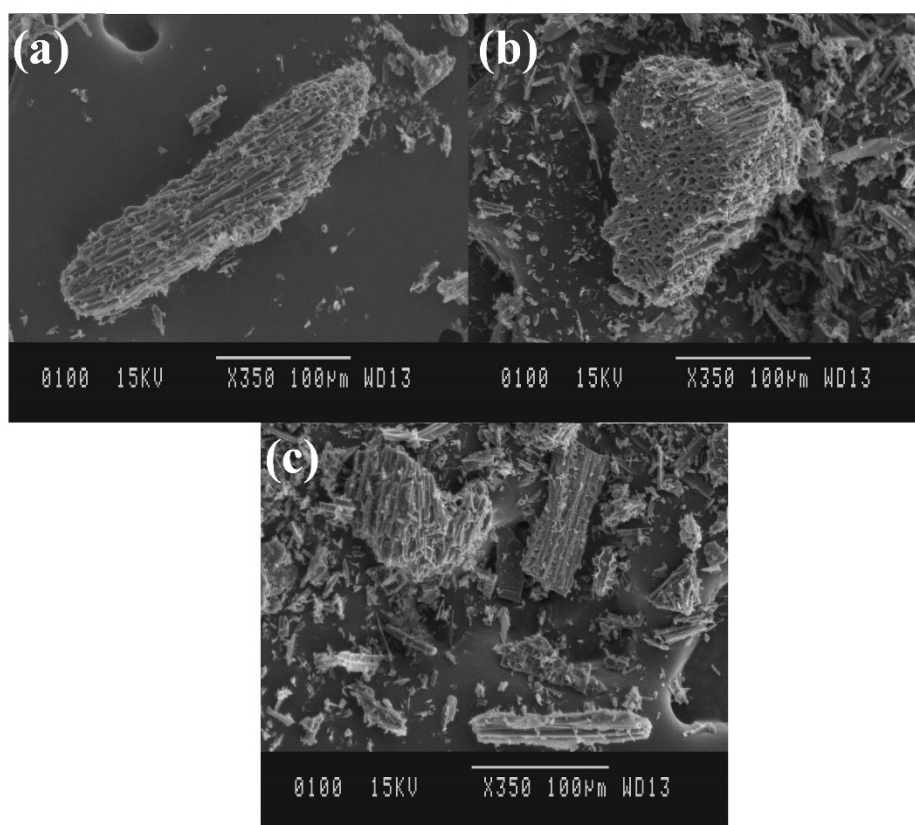
Reference	Carbon Source	Synthesis Technique	Synthesis Parameters	Catalyst	Disadvantages
<i>Zhang et al.</i> <sup>[1]</sup>	Methane	MW- assisted CVD		Nickel	<ul style="list-style-type: none"> <li>•Difficulty in scaling up</li> <li>•Require specialized microwave reactors</li> </ul>
<i>Paul et al.</i> <sup>[2]</sup>	Coconut Oil	CVD	850°C, N <sub>2</sub> atmosphere	Fe particles	<ul style="list-style-type: none"> <li>•High energy consumption</li> <li>•Difficulty in process optimization</li> </ul>
<i>Zhu et al.</i> <sup>[3]</sup>	Bamboo charcoal	CVD	1200-1500°C at 10°C min <sup>-1</sup> , Ar atmosphere	Mineral content in char	<ul style="list-style-type: none"> <li>•Variable mineral content can lead to variation in CNT quality.</li> </ul>
<i>Shi et al.</i> <sup>[4]</sup>	Gumwood	Microwave induced pyrolysis	500°C, 30 - min, N <sub>2</sub> atmosphere, 2.45 GHz multimode microwave cavity		<ul style="list-style-type: none"> <li>•Low yield</li> <li>•High energy consumption</li> </ul>
<i>Wang et al.</i> <sup>[5]</sup>	Rice Husk	MW- Chemical vapour deposition	2.45 GHz microwave power, 300 m Torr pressure, Ar and H <sub>2</sub> to generate plasma	Nickel	<ul style="list-style-type: none"> <li>•Method may result in high D/G ratio.</li> </ul>

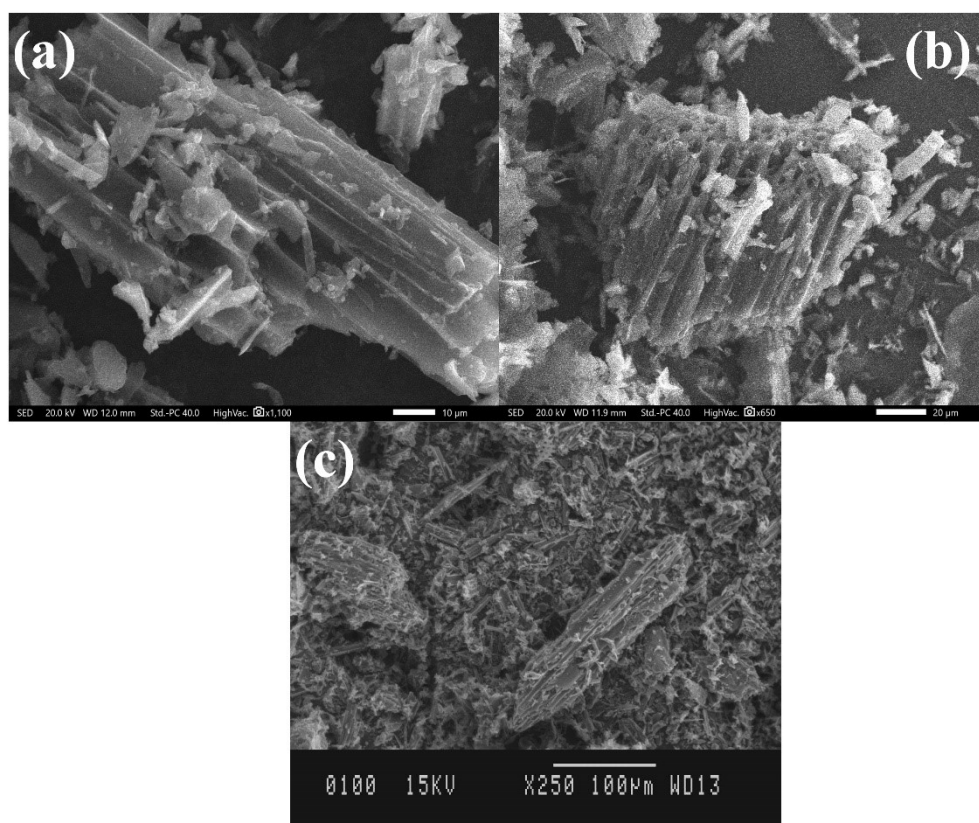
<i>Hamid et al.</i> <sup>[6]</sup>	Olive oil	Pyrolysis	900°C	NiCl <sub>2</sub>	<ul style="list-style-type: none"> <li>• High temperature requirement.</li> <li>• High economic barriers</li> </ul>
<i>Fathy et al.</i> <sup>[7]</sup>	Rice Straw	Conventional CVD	800°C	Ferrocene and Nickel	<ul style="list-style-type: none"> <li>• Complex Pre-treatment processing.</li> <li>• Risk of catalyst contamination.</li> <li>• High temperature</li> </ul>
<i>Debalina et al.</i> <sup>[8]</sup>	Sugarcane bagasse	MW pyrolysis	600 W, 500°C	Fe and Co	<ul style="list-style-type: none"> <li>• Difficulty in maintaining process parameters.</li> <li>• Difficulty in catalyst integration.</li> </ul>
<i>Zhang et al.</i> <sup>[9]</sup>	Pine nutshell	MW-Chemical vapour deposition	600°C, N <sub>2</sub> , Power: 2kW, 60 min	Nickel	<ul style="list-style-type: none"> <li>• Challenges in scaling up.</li> <li>• Plasma instability leads to defective product formation.</li> </ul>
<i>Omoriyekomwan et al.</i> <sup>[10]</sup>	Palm kernel shell cellulose	MW pyrolysis	Power output: - 2000W, Frequency: 2.45 GHz, 600°C, N <sub>2</sub> , 60 min		<ul style="list-style-type: none"> <li>• High energy consumption</li> </ul>
<i>Hildago-Oporto et al.</i> <sup>[11]</sup>	Biochar from waste of wheat straw, oat hulls, rapeseed cake and hazelnut hulls	Microwave Irradiation	600°C, Frequency: 2.45 GHz, Power: 200 W	Ferrocene	<ul style="list-style-type: none"> <li>• Low CNT quality and structural defects.</li> </ul>
<i>Roquia et al.</i> <sup>[12]</sup>	<i>Juglans regia</i> (walnut) shells	Pyrolysis	700°C, 4 h, Ar gas	Ferrocene and Benzene	<ul style="list-style-type: none"> <li>• Use of harmful catalysts.</li> <li>• Scale-up challenges.</li> </ul>
<i>Le et al.</i> <sup>[13]</sup>	Eucalyptus oil	Pyrolysis	800-900°C, 30 min, N <sub>2</sub> gas	Ferrocene	<ul style="list-style-type: none"> <li>• High temperature requirement.</li> <li>• Yield and purity challenges.</li> </ul>

MW: Microwave; CVD: Chemical Vapor Deposition

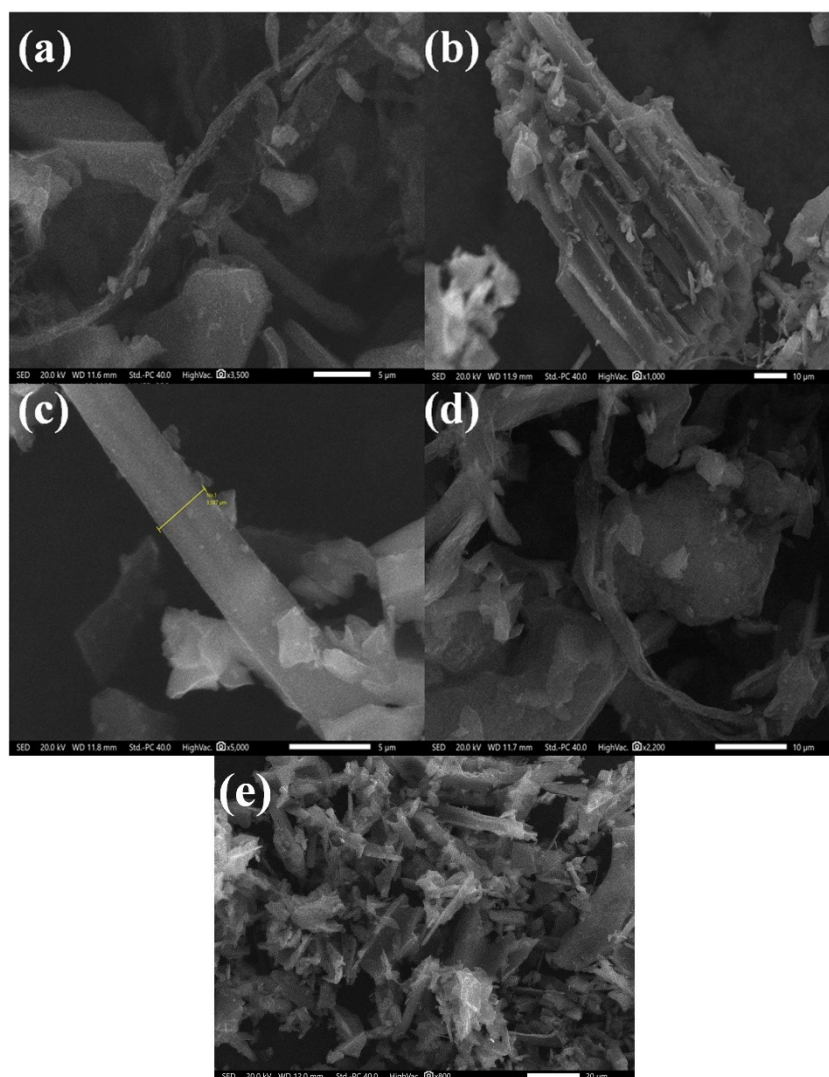
**Table S2: Composition of various NADES used**

Sr. No.	Hydrogen Acceptor	Bond	Hydrogen Bond Donor	Molar Ratio	Results
1.	Choline (ChCl)	Chloride	Oxalic Acid	1 : 1	Successful
2.	ChCl		Hydroquinone	1 : 1	Unsuccessful
3.	ChCl		Glucose	4 : 1	Unsuccessful
4.	ChCl		Citric Acid	1 : 1	Unsuccessful
5.	ChCl		Lactic acid	1 : 1	Unsuccessful
6.	ChCl		Oxalic Acid	1 : 2	Unsuccessful
7.	ChCl		Oxalic Acid	1 : 3	Unsuccessful

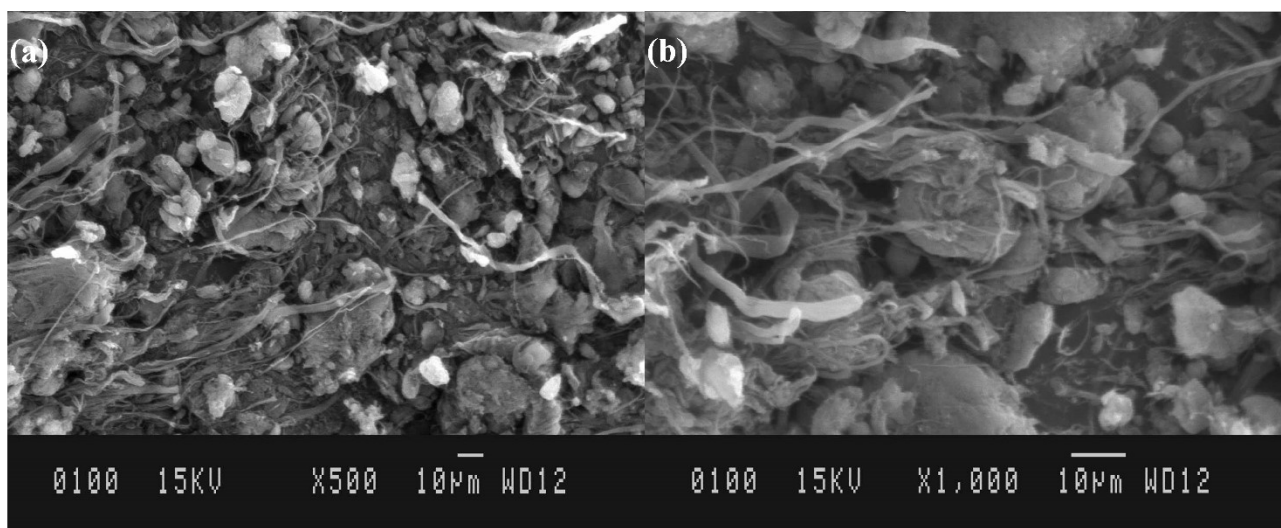
**Fig. S1. SEM images of carbon materials synthesized at (a) 1 h (b) 3 h (c) 8 h**



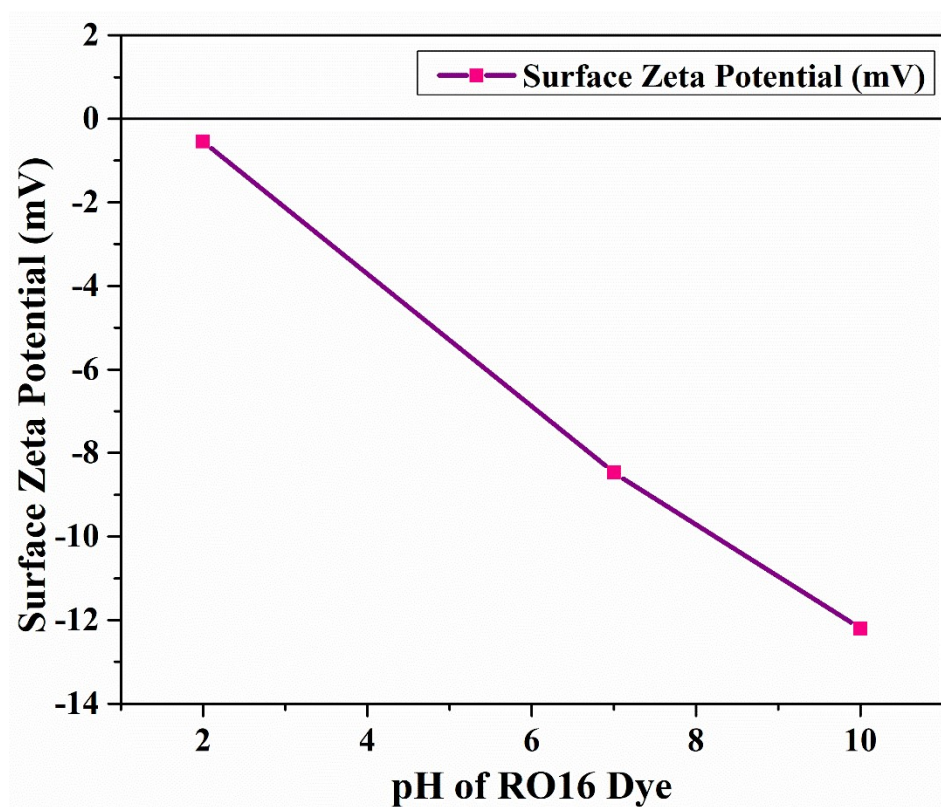
**Fig. S2. SEM images of carbon materials synthesized with different molar ratios of [ChCl]/Oxalic acid/NADES (a) 1 : 2 (b) 1 : 3 (c) 2 : 1**



**Fig. S3. SEM images of carbon materials synthesized with different NADES (a) [ChCl/Glucose] (b) [ChCl/Citric acid] (c) [ChCl/Hq] (d) [ChCl/Lactic acid] (e) Without using NADES**

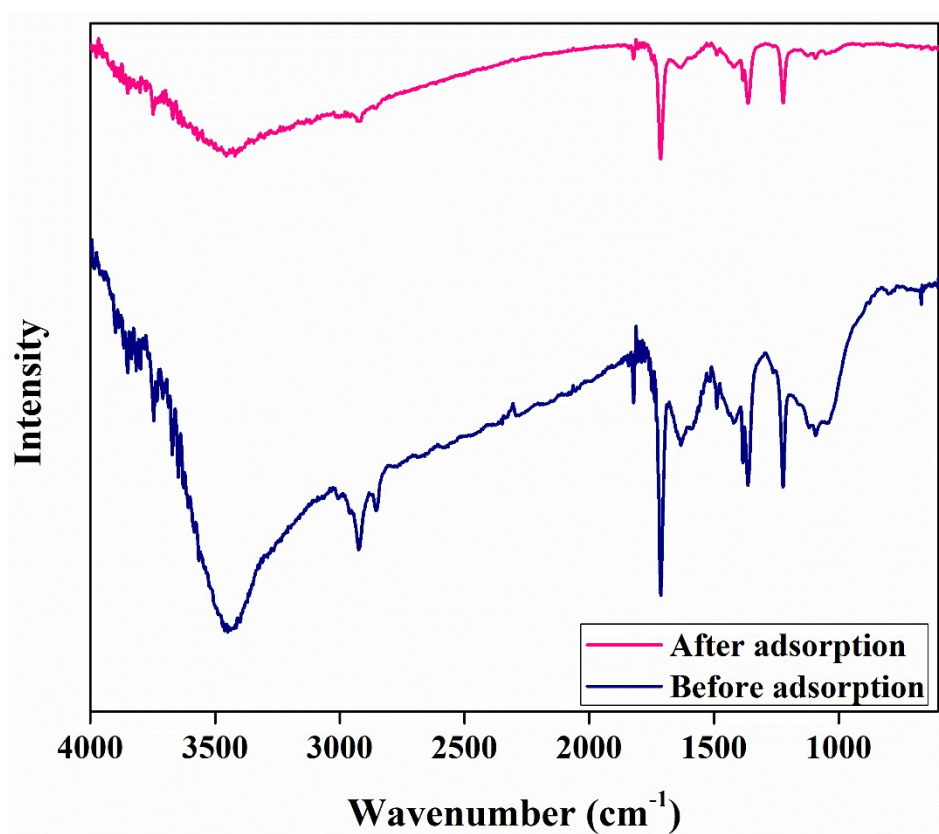


**Fig. S4. SEM images of CNT synthesized using [ChCl/Oxalic acid] NADES (experiment performed in duplicate)**



**Fig. S5. Surface Zeta Potential of RO16 Dye at pH 2, 7 and 10.**





**Fig. S6. FTIR of adsorbent before and after adsorption**

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