

## **Supplementary Material**

### **Ciprofloxacin Degradation in Electro-Fenton System with S-**

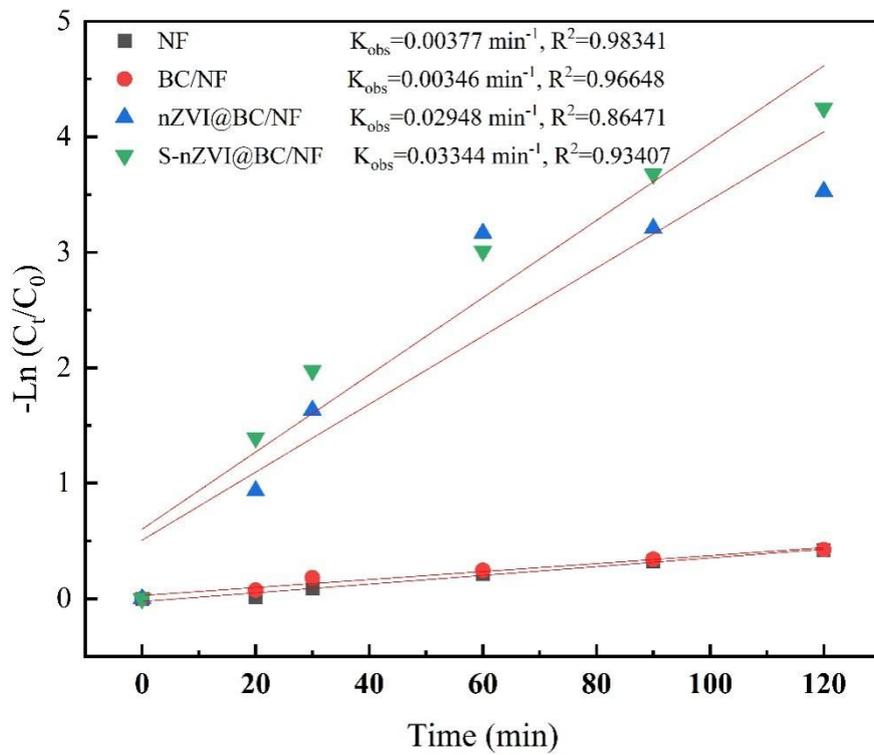
#### **nZVI@BC/NF Composite Electrode**

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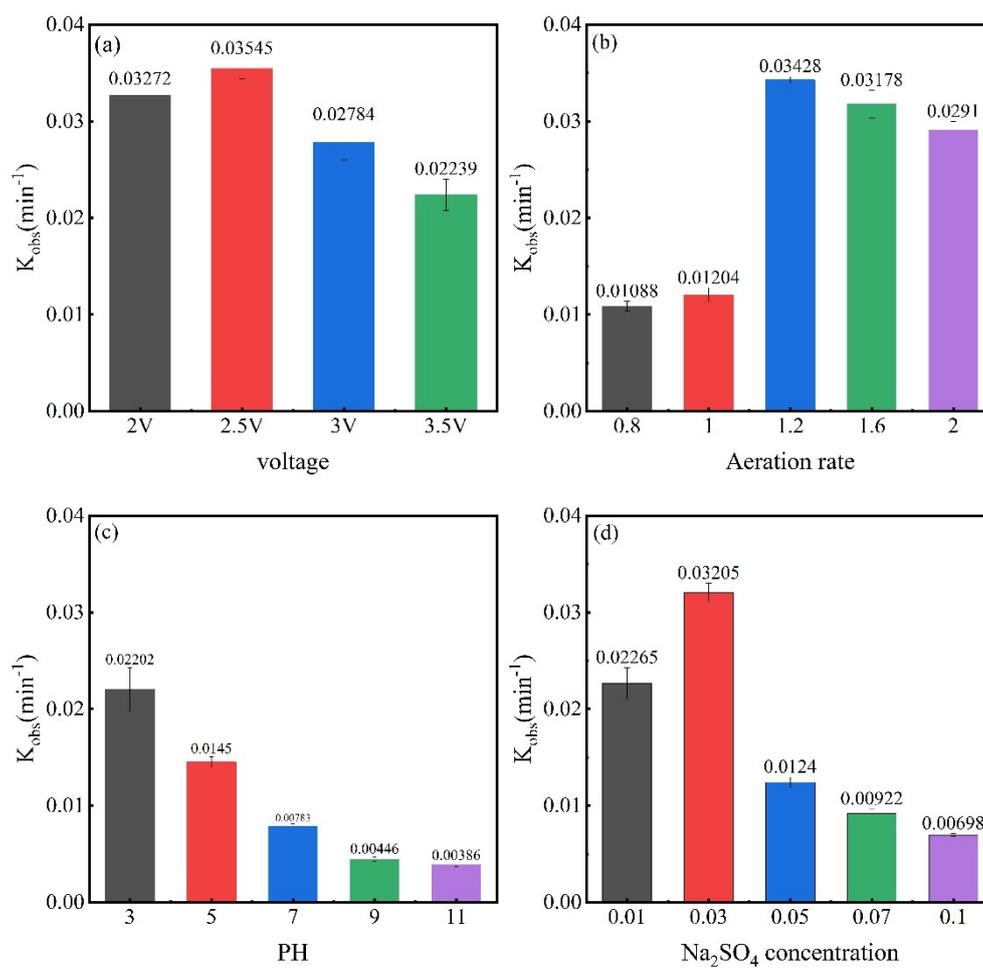
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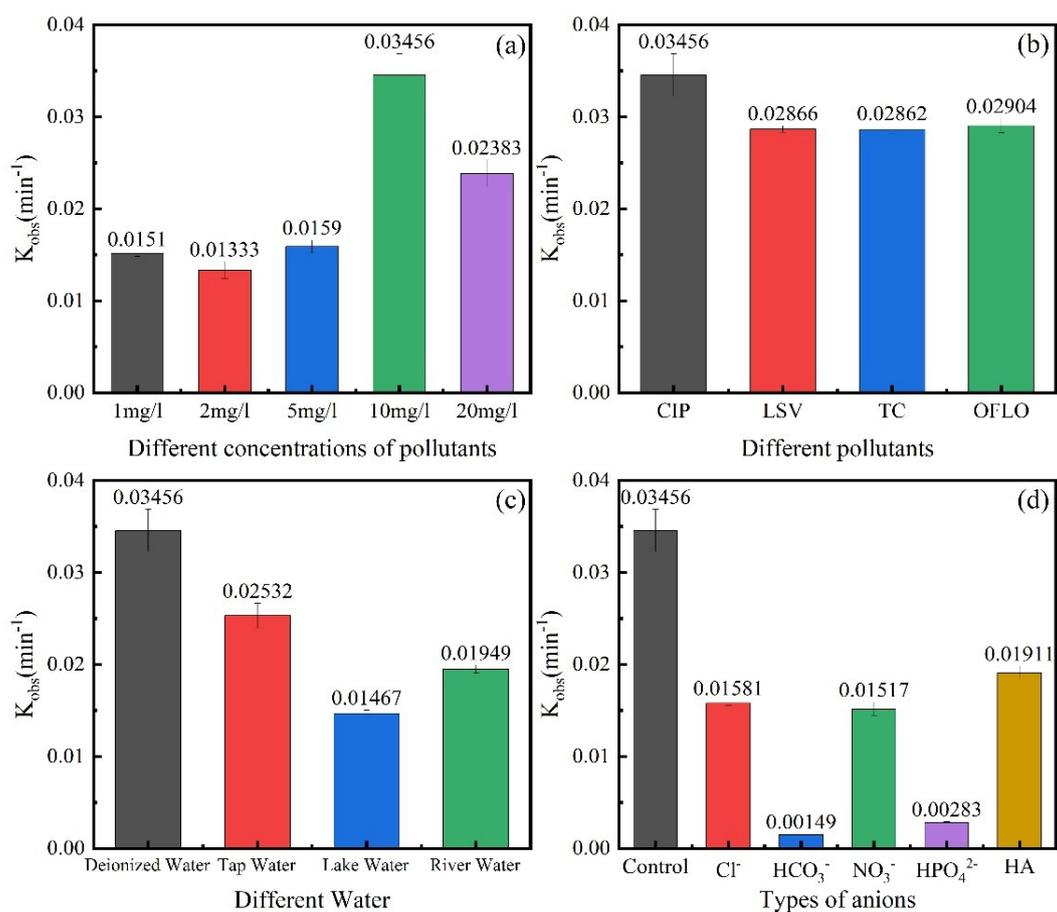
**Fig. S1 Preparation of S-nZVI@BC/NF**



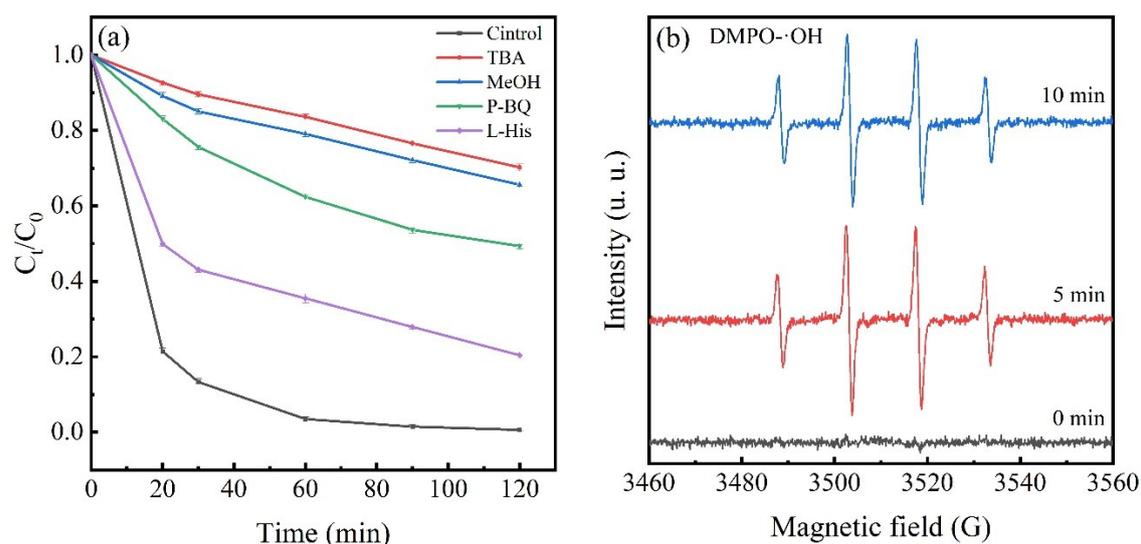
**Fig. S2 Kinetic analysis of different materials**



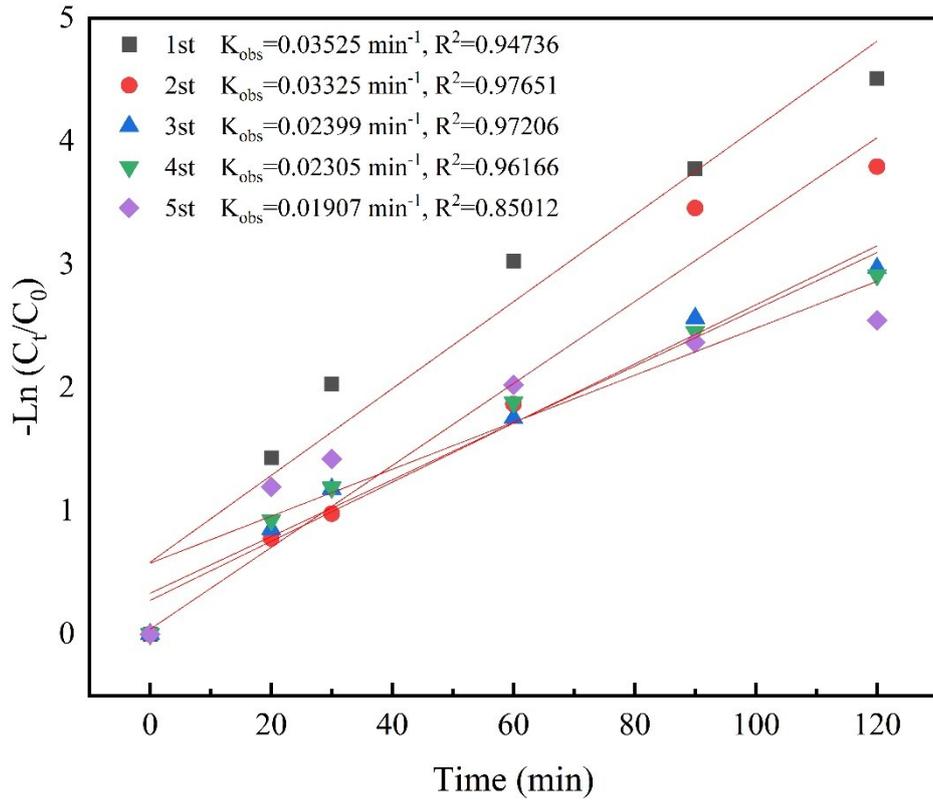
**Fig. S3. Kinetic constants of CIP removal by S-nZVI@BC/NF as a function of voltage (a), aeration rate (b), initial pH value (c), and  $\text{Na}_2\text{SO}_4$  concentration (d)**



**Fig. S4. Kinetic constants of the degradation process of S-nZVI@BC/NF system for different pollutant concentrations (a), different pollutants (b), different water bodies (c), and different anions (d)**



**Fig. S5 . Effect of different free radical scavengers on the removal of CIP in the S-nZVI@BC/NF-EF system. Reaction conditions: [CIP] = 10  $\text{mg}\cdot\text{L}^{-1}$ ,  $U = 2.5 \text{ V}$ , initial pH = 3.0,  $[\text{Na}_2\text{SO}_4] = 30 \text{ mM}$ , Aeration rate = 1.2  $\text{L}\cdot\text{min}^{-1}$ , and  $V = 250 \text{ mL}$ .**



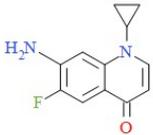
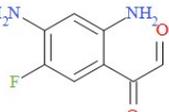
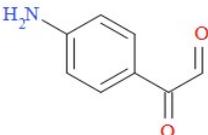
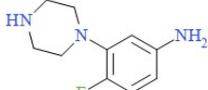
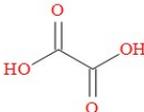
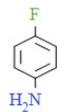
**Fig. S6 Dynamic analysis of CIP removal under different cycle times**

Type of the water	pH	COD (mg•L <sup>-1</sup> )	HCO <sub>3</sub> <sup>-</sup> (mg•L <sup>-1</sup> )	HPO <sub>4</sub> <sup>2-</sup> (mg•L <sup>-1</sup> )	Cl <sup>-</sup> (mg•L <sup>-1</sup> )	NO <sub>3</sub> <sup>-</sup> (mg•L <sup>-1</sup> )
Tap water	7.12	30.20	121.25	0.008	0.36	17.25
Lake water	7.44	110.10	1100.25	0.097	52.22	30.23
River water	8.35	165.20	227.76	0.054	113.82	20.74

**Table S1. Water quality characteristics of tap water, lake water, and drinking water**

**Table S2. Intermediates of CIP in the S-nZVI@BC/NF-EF system**

Code	m/z	Molecular formula	Chemical structure
CIP	332	C <sub>17</sub> H <sub>18</sub> FN <sub>3</sub> O <sub>3</sub>	
P1	361	C <sub>17</sub> H <sub>16</sub> FN <sub>3</sub> O <sub>5</sub>	
P2	321	C <sub>14</sub> H <sub>12</sub> FN <sub>3</sub> O <sub>5</sub>	
P3	265	C <sub>12</sub> H <sub>12</sub> FN <sub>3</sub> O <sub>3</sub>	
P4	207	C <sub>10</sub> H <sub>6</sub> FNO <sub>3</sub>	
P5	163	C <sub>9</sub> H <sub>6</sub> FNO	
P6	145	C <sub>9</sub> H <sub>7</sub> NO	
P7	60	C <sub>2</sub> H <sub>8</sub> N <sub>2</sub>	
P8	56	C <sub>4</sub> H <sub>8</sub>	
P9	303	C <sub>16</sub> H <sub>18</sub> O <sub>2</sub> FN <sub>3</sub>	

P10	218	$C_{12}H_{11}FN_2O$	
P11	178	$C_9H_7FN_2O$	
P12	186	$C_8H_7FN_2O_2$	
P13	149	$C_8H_7O_2N$	
P14	338	$C_{16}H_{18}FN_3O_4$	
P15	191	$C_{10}H_{14}FN_3$	
P16	90	$C_2H_2O_4$	
P17	111	$C_6H_6FN$	
P18	86	$C_4H_{10}N_2$	