## **Electronic Supplementary Information (ESI)**

Insights into the halogen-induced p-band center regulation promising highperformance lithium-sulfur batteries

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Fig. S1. XRD patterns of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$ .



Fig. S2. Lattice parameters of (a)  $Cs_2SnI_{6}$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$ .



Fig. S3. FESEM images of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$  before ballmilling.



**Fig. S4.** FESEM images of (a)  $Cs_2SnBr_6$  and (b)  $Cs_2SnCl_6$  after ball-milling, and corresponding particle size distribution diagrams (the insets).



Fig. S5. (a, b) HRTEM images and (c, d) SAED patterns of Cs<sub>2</sub>SnBr<sub>6</sub> and Cs<sub>2</sub>SnCl<sub>6</sub>.



Fig. S6. EDS mapping images of Cs, Sn, and halogen elements for (a - d) Cs<sub>2</sub>SnBr<sub>6</sub>, and (e - h) Cs<sub>2</sub>SnCl<sub>6</sub>.



Fig. S7. Typical Sn 3d spectra of (a)  $Cs_2SnBr_6$  and (b)  $Cs_2SnCl_6$ . (c) Br 3d spectra of  $Cs_2SnBr_6$  and (d) Cl 2p spectra of  $Cs_2SnCl_6$ .



Fig. S8. S 2p spectra of (a)  $Cs_2SnBr_6$  and (b)  $Cs_2SnCl_6$  absorbed with  $Li_2S_{6.}$ 



Fig. S9. Adsorption configurations of  $Li_2S_n$   $(1 \le n \le 8)$  on the surface of  $Cs_2SnX_6$ .



Fig. S10. Cross-sectional FESEM images of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$  modified separators.



Fig. S11. Top-view FESEM images of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$  modified separators.



Fig. S12. Thermogravimetric analysis plot of KB/S.



Fig. S13. (a – c) Enlarged CV curves (0.1 mV s<sup>-1</sup>) and (d – f) corresponding peak

current value of LSBs with  $Cs_2SnX_6$  modified separators.



Fig. S14. Tafel slopes of (a) peak 1 and (b) peak 2 of LSBs with  $Cs_2SnX_6$  modified separators.



Fig. S15. Charge-discharge curves of (a)  $Cs_2SnBr_6$  and (b)  $Cs_2SnCl_6$  batteries at different current rates.

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Fig. S16. Charge-discharge curves of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$  batteries at different cycle times.



Fig. S17. FESEM images of the cathode of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$ batteries after 500 cycles at 1 C.

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Fig. S18. FESEM images and digital photos (the insets) of lithium anodes of (a)  $Cs_2SnI_6$ , (b)  $Cs_2SnBr_6$ , and (c)  $Cs_2SnCl_6$  batteries after 500 cycles at 1 C.



Fig. S19. FESEM images of  $Cs_2SnX_6$  modified separators (a - c) before and (d - f) after 500 cycles at 1C.



Fig. S20. (a) The first charge-discharge curves of  $Cs_2SnI_6$  batteries under high-load and low E/S ratio conditions. (b – d) Corresponding charge-discharge curves at different cycle times as indicated.



Fig. S21. The first Charge-discharge curves of  $Cs_2SnI_6$  batteries in environments of 50°C (0.5 C) and -20 °C (0.1 C).

	Coated materials	$R_{s}\left(\Omega ight)$	$R_{sf}(\Omega)$	$R_{ct}\left(\Omega ight)$
Before cycling	$Cs_2SnCl_6$	1.6	-	73.5
	$Cs_2SnBr_6$	1.8	-	72.6
	$Cs_2SnI_6$	2.9	-	60.7
After cycling	$Cs_2SnCl_6$	3.9	20.6	7.6
	$\mathrm{Cs}_2\mathrm{SnBr}_6$	3.5	17.7	6.2
	$Cs_2SnI_6$	4.5	13.8	2.9

Table S1 EIS fitting results of  $Cs_2SnX_6$  batteries.

Coated materials	Current rate (C)	Cycle number	Capacity decay per cycle (%)
Cs <sub>2</sub> SnI <sub>6</sub> (This work)	2	500	0.068
VN@NG <sup>1</sup>	2	500	0.075
CQDs-PAN <sup>2</sup>	0.5	500	0.075
H-CMP <sup>3</sup>	1	500	0.066
Li-MOF/RGO <sup>4</sup>	1	600	0.089
Zwitterionic COF <sup>5</sup>	2	500	0.072
C <sub>3</sub> N <sub>4</sub> -CoSe <sub>2</sub> <sup>6</sup>	1	500	0.089
ZIF-8 <sup>7</sup>	1	500	0.100
Ni/SiO2/G <sup>8</sup>	1	300	0.086
Co/Mo <sub>2</sub> C <sup>9</sup>	1	600	0.072

 Table S2. Performance comparison of LSBs with various separators.

Coated materials	Sulfur loading (mg cm <sup>-2</sup> )	E/S ratio (µl mg <sup>-1</sup> )	Initial specific capacity (mAh $g^{-1}$ ) @ C rates
Cs <sub>2</sub> SnI <sub>6</sub> (This work)	6.10	5.5	768.8@0.2 C
NbB <sub>2</sub> /rGo <sup>10</sup>	7.06	10	590.7@0.1 C
Co-MoS <sub>2</sub> <sup>11</sup>	5.27	12	800.8@0.2 C
NS-MXene <sup>12</sup>	7.20	7.0	730.6@0.2 C
Fe-ZIF-8 <sup>13</sup>	5.00	5.0	517@0.05 C
TpPa-SO <sub>3</sub> H <sup>14</sup>	5.00	10	800.0@0.2 C
ZIF-67/SA- PAN <sup>15</sup>	5.45	10	797.5@0.1 C
ZrO <sub>2</sub> -SiO <sub>2</sub> <sup>16</sup>	4.00	10	757.0@0.2 C
GQDs-PAN <sup>17</sup>	5.10	15	633.3@0.1 C
RPM <sup>18</sup>	5.40	10	703.7@0.2 C

 Table S3 Performance comparison of LSBs with various separators under high-load

 and low-E/S-ratio conditions.

 Table S4. Performance comparison of LSBs with various separators under different

 operating temperatures

Coated materials	Temperature (°C)@Current rate (C)@Initial specific capacity (mAh g <sup>-1</sup> )		
Cs2SnI6 (This work)	-20 °C@0.1 C@ 912.7 mAh g <sup>-1</sup>		
	50 °C@0.5 C@1350 mAh g <sup>-1</sup>		
Go-CoNiP <sup>19</sup>	-20 °C@0.5 C@810.9 mAh g <sup>-1</sup>		
	60 °C@0.5 C@1064.8 mAh g <sup>-1</sup>		
SAF-3 <sup>20</sup>	-20 °C@0.1 C@870.0 mAh g <sup>-1</sup>		
	60 °C@0.5 C@1064.8 mAh g <sup>-1</sup>		
NbB <sub>2</sub> <sup>21</sup>	-10 °C@0.1 C@802.0 mAh g <sup>-1</sup>		
TPE <sup>22</sup>	-20 °C@0.1 C@802.0 mAh g <sup>-1</sup>		
FeCoNi <sup>23</sup>	0 °C@0.2 C@931.0 mAh g <sup>-1</sup>		
Fe/Ni-N@NC <sup>24</sup>	0 °C@0.5 C@741.0 mAh g <sup>-1</sup>		

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