

Supporting Information

Stable cationic nanobelts synthesized by chemical oxidation of methylene-bridged [6]cycloparaphenylene

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Table of Contents

Supplementary Methods	S2–S6
1. Materials and methods	S2–S3
2. Synthesis of [6]MCP ⁺ cations	S4–S7
Supplementary Figures and Tables	S8–S34
1. Half-life time of 1 , 2 , 3	S8–S9
2. X-ray crystallography	S10–S11
3. Comparison among [6]MCP ⁺ , [6]MCP ²⁺ , [6]CPP ²⁺	S12
4. Photophysical measurements	S13
5. DFT and TD-DFT calculations	S14–S34
6. NMR, ESR, and HRMS spectra	S35–S39
Supplementary References	S40

Supplementary Methods

1. Materials and methods

Materials: Unless otherwise noted, all reactants or reagents including dry solvents were obtained from commercial suppliers and used as received. Unless otherwise noted, all reactions were performed with dry solvents under an atmosphere of nitrogen in dried glassware using standard vacuum-line techniques. Unless otherwise noted, all work-up and purification procedures were carried out with reagent-grade solvents under air.

MS, NMR and ESR: The high-resolution mass spectra (HRMS) were obtained from Bruker Daltonics compact (ESI). Nuclear magnetic resonance (NMR) spectra were recorded on a JEOL JNM-ECA-600 (^1H 600 MHz, ^{13}C 150 MHz) spectrometer and a JEOL ECA 600 II spectrometer equipped with a UltraCOOL probe. Chemical shifts for ^1H NMR are expressed in parts per million (ppm) relative to residual CH_2Cl_2 (δ 5.32 ppm). ^{13}C NMR spectra were recorded using a proton-decoupled pulse sequence. Chemical shifts for ^{13}C NMR are expressed in ppm relative to CD_2Cl_2 (δ 54.0 ppm) or $\text{C}_2\text{D}_2\text{Cl}_4$ (δ 73.78 ppm). Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet), coupling constant (Hz), and integration. Electron paramagnetic resonance (EPR) spectra were recorded on JEOL ESR JES-X320 instruments using quartz Schlenk tube filled with argon.

X-ray crystallography: Details of the crystal data and a summary of the intensity data collection parameters for **2** are listed in Table S1. A suitable crystal was mounted with mineral oil on a MiTeGen MicroMounts and transferred to the goniometer of the kappa goniometer of a RIGAKU XtaLAB Synergy-S system with 1.2 kW MicroMax-007HF microfocus rotating anode (Graphite-monochromated Mo $\text{K}\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$)) and PILATUS200K hybrid photon-counting detector. Cell parameters were determined and refined, and raw frame data were integrated using CrysAlis^{Pro} (Agilent Technologies, 2010). The structures were solved by direct methods with SHELXT^{S1} and refined by full-matrix least-squares techniques against F^2 (SHELXL-2018/3)^{S2} by using Olex2 software package.^{S3} The intensities were corrected for Lorentz and polarization effects. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed using AFIX instructions. CCDC 2359239 contains the supplementary crystallographic data for **2**. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

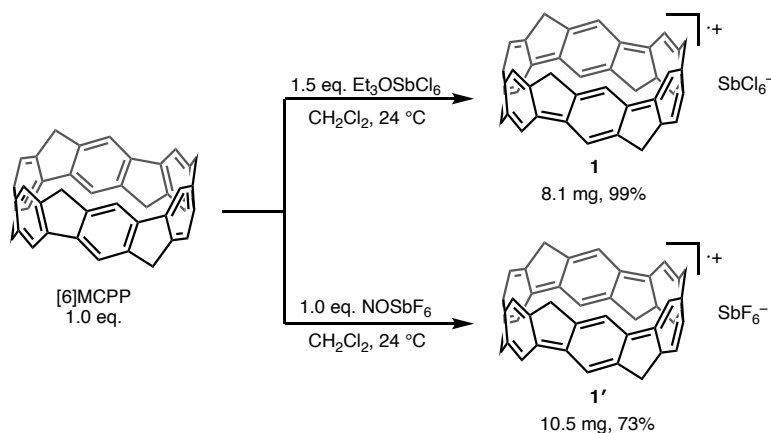
Photophysical measurements: UV-vis absorption spectra were recorded on a Shimadzu UV-3600 spectrometer with a resolution of 0.5 nm. Dilute solutions in spectral grade dichloromethane in a 1 cm square quartz cell were used for measurements. Emission spectra were recorded on a were recorded with a JASCO FP-6600 fluorescence spectrometer with a resolution of 0.4 nm. Absolute quantum yields (Φ) were determined with a Quantaurus-QY Plus C-13534-

02 (Hamamatsu Photonics) equipped with NIR PL measurement unit C13684-01 calibrated with an integrating sphere system.

Computational study: The Gaussian 16 program^{S4} running on a NEC LX 110Rh system was used for optimization (B3LYP/6-31G(d)).^{S5,S6} Structures were optimized without any symmetry assumptions. Zero-point energy, enthalpy, and Gibbs free energy at 298.15 K and 1 atm were estimated from the gas-phase studies. Harmonic vibration frequency calculation at the same level was performed to verify all stationary points as local minima (with no imaginary frequency). Also GIAO-NMR and CSGT-NMR were calculated using the B3LYP/6-311+G(2d,p) with SMD (CH₂Cl₂) model.

2. Synthesis of [6]MCP⁺ cations

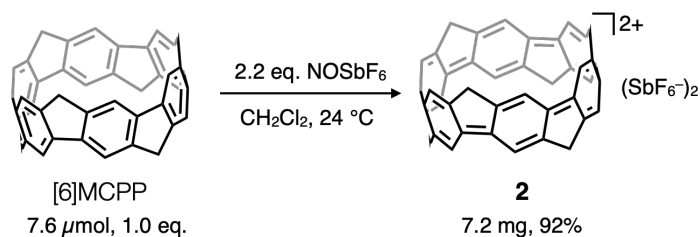
2.1 Synthesis of **1** and **1'**



To an oven dried 50-mL vial containing a magnetic stirring bar was added [6]MCP (9.6 μ mol, 5.1 mg, 1.0 eq.). The vial was degassed and placed inside the glove box and added dichloromethane 2.0 mL, which was degassed by freeze-pump-thaw cycles. Et₃OSbCl₆ (14 μ mol, 6.2 mg, 1.5 eq.) in dichloromethane 2.0 mL were added to the vial and stirred for 30 minutes at r.t. Hexane 4.0 mL was added to the reaction mixtures. The resulting precipitate was collected by filtration and washed with hexane/CH₂Cl₂ = 1:1 to give the oxidized product **1** (8.1 mg, 99%) as a black solid. Identification of **1** was concluded from its absorption spectrum and ESR spectrum (see Figure S5 and Figure S14, respectively).

To an oven dried 50-mL vial containing a magnetic stirring bar was added [6]MCP (19 μ mol, 10.1 mg, 1.0 eq.). The vial was degassed and placed inside the glove box and added dichloromethane 4.0 mL, which was degassed by freeze-pump-thaw cycles. NOSbF₆ (19 μ mol, 5.0 mg, 1.0 eq.) in dichloromethane 4.0 mL were added to the vial and stirred for 2 hours at 24 °C. Hexane 8.0 mL was added to the reaction mixtures. The resulting precipitate was collected by filtration and washed with hexane/CH₂Cl₂ = 1:1 to give the oxidized product **1'** (10.5 mg, 73%) as a black solid. Identification of **1'** was concluded from its ESR spectrum (see Figure S14).

2.2 Synthesis of 2



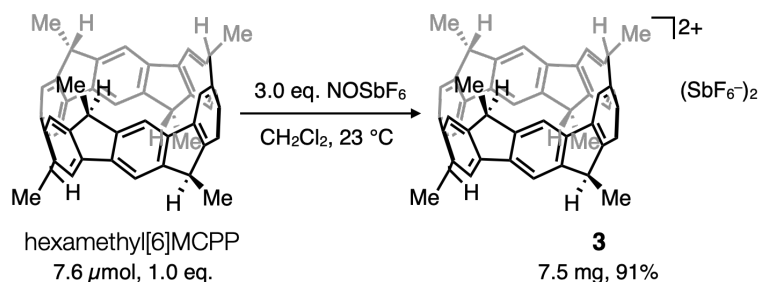
To a dried 10-mL two-neck round-bottom flask containing a magnetic stirring bar was added $[6]MCP$ (7.6 μ mol, 4.0 mg, 1.0 eq.). The vial was degassed and placed inside the glove box. $NOSbF_6$ (23 μ mol, 6.0 mg, 3.0 eq.) and dichloromethane 3.0 mL, which was degassed by freeze-pump-thaw cycles, were added to the vial and stirred for 2 hours at 24 °C. The vial was brought out to air and dark green colored reaction mixtures were filtered off by Celite[®] and then concentrated *in vacuo* to ca. 10 mL, and hexane (ca. 200 mL) was added. The resulting precipitate was collected by filtration to give the product **2** (7.2 mg, 92%) as a dark blue solid.

¹H NMR (600 MHz, CD_2Cl_2) δ (ppm) 4.18 (s, 12H), 3.61 (d, J = 18.0 Hz, 6H), -0.34 (d, J = 18.6 Hz, 6H).

¹³C NMR (150 MHz, $C_2D_2Cl_4$) δ (ppm) 164.91, 161.15, 123.69, 36.25.

HRMS (ESI, positive) m/z calcd for $C_{42}H_{24} [M]^{2+}$: 264.0934, found: 264.0923.

2.3 Synthesis of **3**



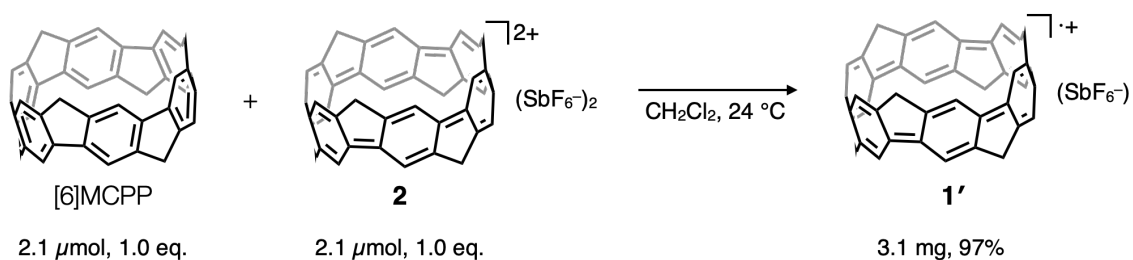
To a dried 10-mL two-neck round-bottom flask containing a magnetic stirring bar was added hexamethyl[6]MCP (7.6 μ mol, 4.6 mg, 1.0 eq., synthesized by following reported procedure^{S7}). The vial was degassed and placed inside the glove box. NOSbF₆ (23 μ mol, 6.0 mg, 3.0 eq.) and dichloromethane 3.0 mL, which was degassed by freeze-pump-thaw cycles, were added to the vial and stirred for 2 hours at 23 °C. The vial was brought out to air and dark green colored reaction mixtures were filtered off by Celite[®] and then concentrated *in vacuo* to ca. 10 mL, and hexane (ca. 200 mL) was added. The resulting precipitate was collected by filtration to give the oxidized products **3** (7.5 mg, 91%) as a dark blue solid.

¹H NMR (600 MHz, CD₂Cl₂) δ (ppm) 4.17 (s, 12H), 1.95 (d, J = 7.2 Hz, 18H), -0.12 (q, J = 13.8 Hz, 7.2 Hz, 6H).

¹³C NMR spectrum was not obtained. Even in a saturated solution, the concentration was insufficient.

HRMS (ESI, positive) m/z calcd for C₄₈H₃₆ [M]²⁺: 306.1403, found: 306.1414.

2.4 The comproportionating reaction by mixing **2** and neutral [6]MCP



To an oven dried 30-mL vial containing a magnetic stirring bar was added [6]MCP (2.1 μmol , 1.1 mg, 1.0 equiv). The vial was degassed and placed inside the glove box and added dichloromethane 1.0 mL, which was degassed by freeze-pump-thaw cycles. **2** (2.1 μmol , 2.1 mg, 1.0 equiv) in dichloromethane 1.0 mL were added to the vial and stirred for 30 minutes at r.t. Hexane 4.0 mL was added to the reaction mixtures. The resulting precipitate was collected by filtration and washed with hexane/ CH_2Cl_2 = 1:1 to give the oxidized product **1'** (3.1 mg, 97%) as a black solid. Identification of **1'** was concluded from its absorption spectrum.

Supplementally Figures and Tables

1. Half-life time of 1, 2, 3 (regarding Scheme 1 in the main text)

The stability of **1**, **2** and **3** in solution phase under air was respectively investigated by using the dichloromethane solution (5.68×10^{-5} M at 293 K). The UV-vis-NIR absorption spectra were measured every one hour (Figure S1a, S2a, S3a). The half-life time was elucidated from the scheme below based on the decrease of the absorption maximum peak (Figure S1b, S2b, S3b).

$$N = N_0 \left(\frac{1}{2} \right)^{\frac{t}{T}}$$

$$\log_{10} \frac{N}{N_0} = -\frac{\log 2}{T} \times t$$

N_0 : The maximum absorbance at starting time; N_t : The maximum absorbance at time t .

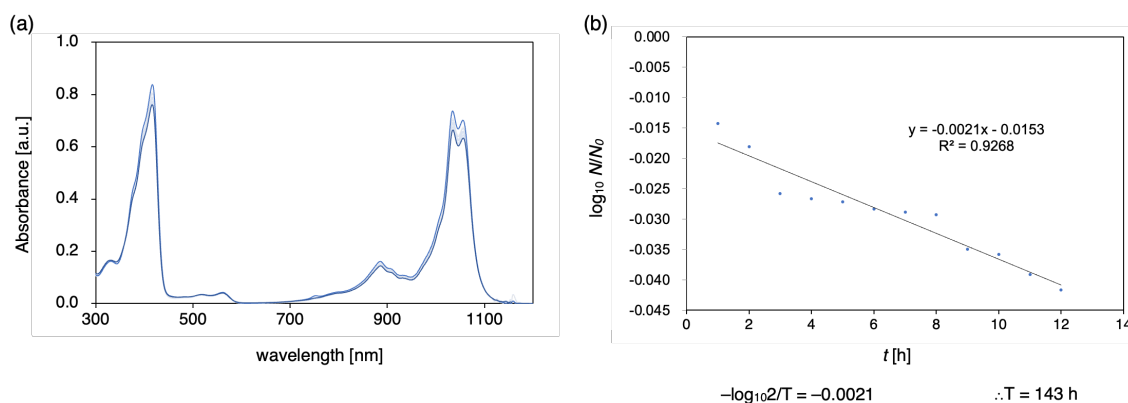


Figure S1. (a) Time-dependence of the absorption spectra of **1**. (b) Plots for half-life of **1** in CH_2Cl_2 solution (monitored at 415 nm).

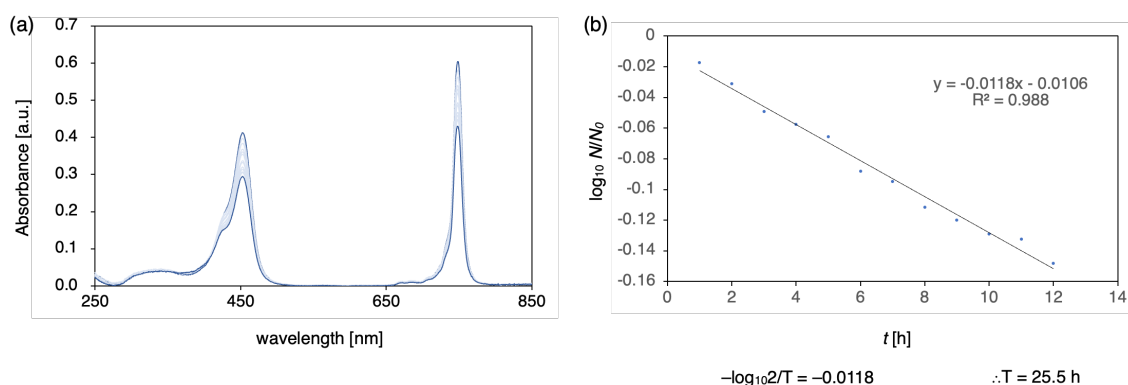


Figure S2. (a) Time-dependence of the absorption spectra of **2**. (b) Plots for half-life of **2** in CH_2Cl_2 solution (monitored at 745 nm).

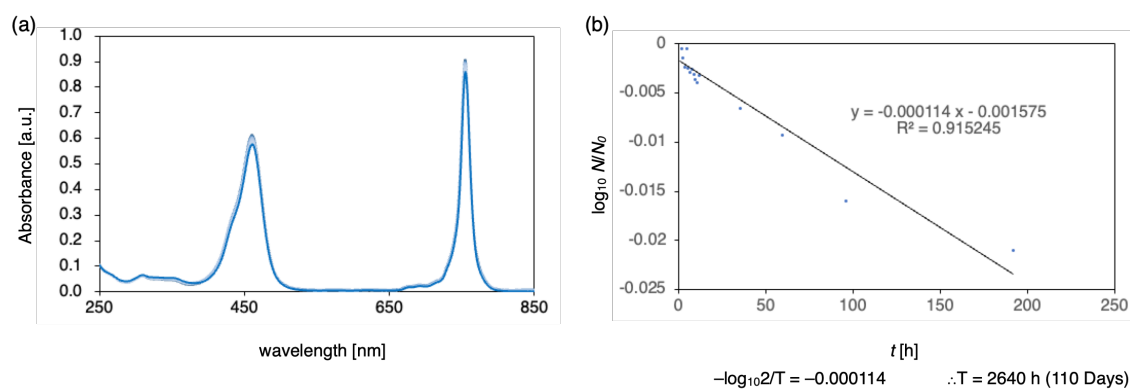


Figure S3. (a) Time-dependence of the absorption spectra of **3**. (b) Plots for half-life of **3** in CH_2Cl_2 solution (monitored at 755 nm).

2. X-ray crystallography (regarding Figure 2 in the main text)

Table S1. Crystallographic data and structure refinement details of **2**.

	2
CCDC No.	2359239
formula	C ₄₇ H ₃₄ Cl ₁₀ F ₁₂ Sb ₂
fw	1424.74
<i>T</i> (K)	123(2)
λ (Å)	0.71073
cryst syst	triclinic
space group	<i>P</i> -1
<i>a</i> (Å)	11.0560(3)
<i>b</i> (Å)	12.0361(2)
<i>c</i> (Å)	12.1017(3)
α (deg)	105.769(2)
β (deg)	104.054(2)
γ (deg)	116.448(2)
<i>V</i> (Å ³)	1257.78(6)
<i>Z</i>	1
<i>D</i> _{calc} (g·cm ⁻³)	1.881
μ (mm ⁻¹)	1.685
F(000)	696.0
cryst size (mm ³)	0.18 × 0.08 × 0.01
2 θ range (deg)	3.858–59.842
reflns collected	19722
indep reflns/ <i>R</i> _{int}	5972 / 0.0354
params	334
GOF on <i>F</i> ²	1.040
<i>R</i> ₁ , <i>wR</i> ₂ [<i>I</i> > 2 σ (<i>I</i>)]	0.0259, 0.0629
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0290, 0.0639

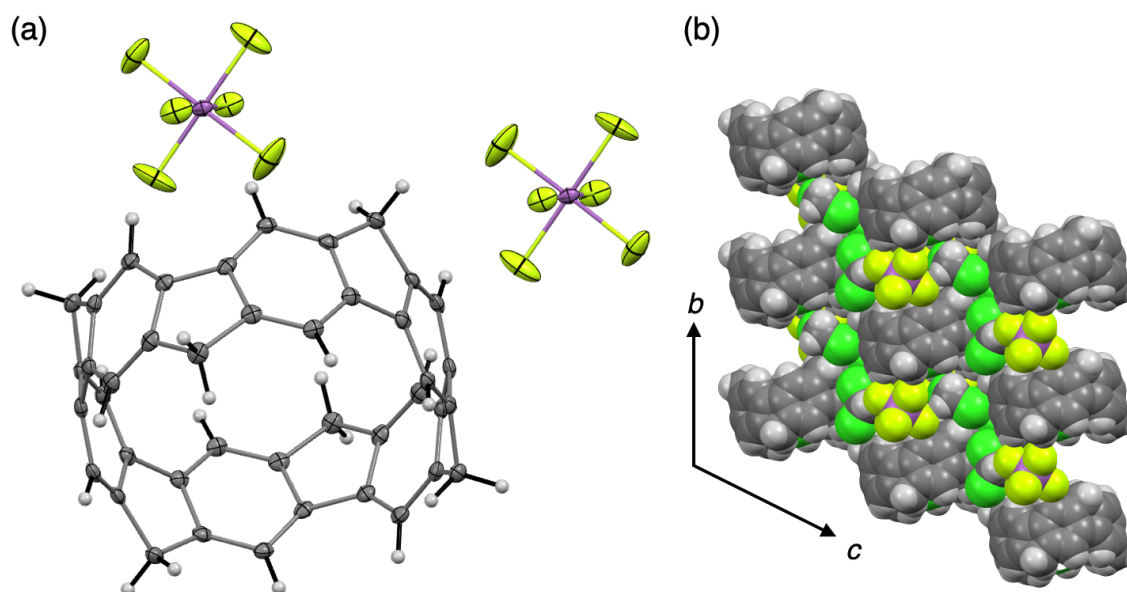
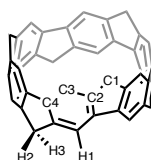


Figure S4. (a) X-ray crystal structures of **2** ($[6]MCP2^{+}(SbF_6^{-})_2$) with solvents omitted for clarity. (b) Packing structures viewed along the *a*-axis.

3. Comparison among [6]MCP, [6]MCP²⁺, [6]CPP²⁺ (regarding Table 1 in the main text)

Table S2. Comparison among [6]MCP^[S8], [6]MCP²⁺, [6]CPP²⁺ ^[S9].



	[6]MCP	[6]MCP ²⁺	[6]CPP ²⁺
Diameter [Å]	7.758	7.675	-
C1–C2 [Å]	1.518	1.514	-
C2–C3 [Å]	1.384	1.368	-
C2–C4 [Å]	1.408	1.436	-
C3–C4 [Å]	1.395	1.415	-
C4–C4 [Å]	1.478	1.428	-
HOMO [eV]	–4.40*	–11.38*	–11.62*
LUMO [eV]	–1.74*	–9.47*	–9.79*
Energy gap [eV]	2.66*	1.91*	1.83*
Abs. peak [nm]	347, 370	453, 745	464, 792
Flu. peak [nm]	-	677, 752	801
Fluorescence quantum yield	-	23%	1.8%
NMR (H1) [ppm]	7.88	4.01	5.30
NMR (H2) [ppm]	4.27	3.51	-
NMR (H3) [ppm]	4.01	–0.13	-
NMR (C1) [ppm]	41.2	36.2	-
NMR (C2) [ppm]	149.0	161.2	-
NMR (C3) [ppm]	122.4	123.7	-
NMR (C4) [ppm]	137.7	164.9	-
NICS(belt center)	–1.188**	–19.971**	–18.742**
NICS(-1)	–7.033**	–25.629**	–25.369**
NICS(0)	–5.679**	–9.013**	–8.964**
NICS(1)	–5.384**	–3.595**	–4.131**

*B3LYP/6-31G(d)level **GIAO-B3LYP/6-311+G(2d,p)

4. Photophysical measurements (regarding Figure 3 in the main text)

The measurement of fluorescence for **1** was unsuccessful in spite of our effort. The measurement of fluorescence in IR regions is often difficult, also because of depopulation of the excited state by electronic-vibrational energy transfer (see *e.g.* Monguzzi et al. *New J. Chem.*, **2009**, 33, 1542–1548). Moreover, according to Kasha rules the fluorescing state should be the S_1 state of symmetry E_u (derived from the $e_u \rightarrow a_{2g}$ single excitation), which can be expected to undergo an efficient non-radiative decay via dynamic Jahn-Teller effect. As a matter of fact, we find in a preliminary calculation that S_1 is second-order saddle and tends to lower symmetry following e_u vibrational modes.

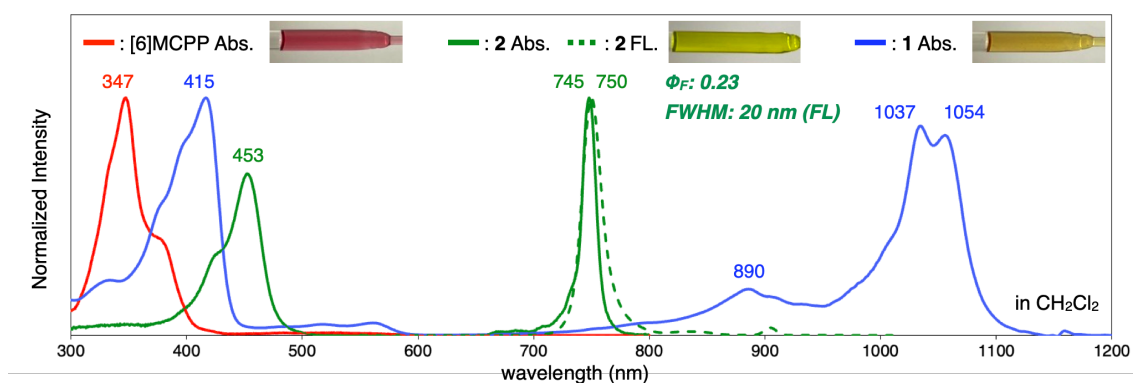


Figure S5. UV-vis-NIR absorption (solid line) and fluorescence (dashed line) spectra of CH_2Cl_2 solutions of [6]MCP, **1**, and **2**. Fluorescence spectrum of **2** was recorded upon excitation at 450 nm.

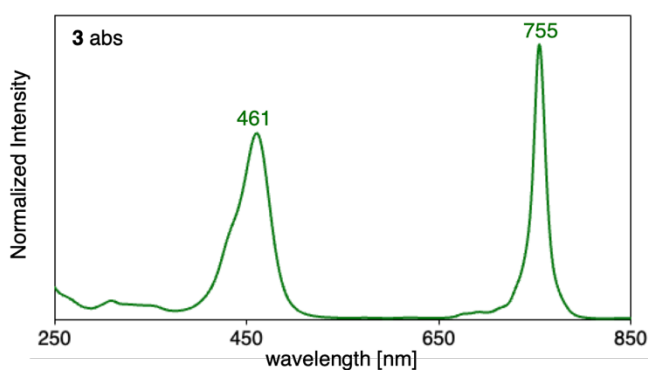
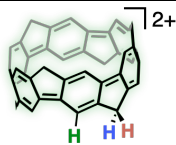


Figure S6. UV-vis-NIR absorption spectra of a CH_2Cl_2 solution of **3**.

5. DFT and TD-DFT calculations

Table S3. Simulated NMR chemical shifts (GIAO-B3LYP/6-311+G(2d,p)).

[6]MCP ²⁺		NMR chemical shifts [ppm]
	H	4.13
	H	3.92 (outer)
	H	-0.67 (inner)

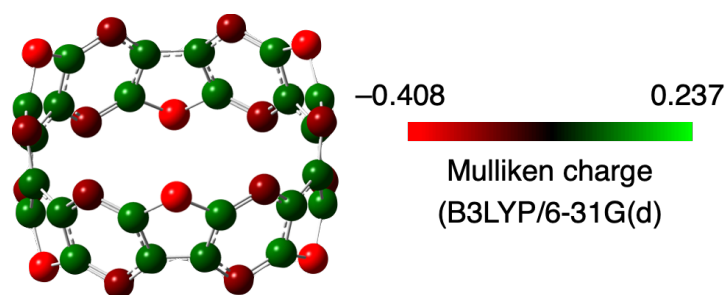
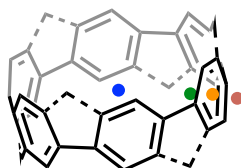


Figure S7. Mulliken charge of [6]MCP²⁺ (B3LYP/6-31G(d))

Table S4. NICS value of [6]MCP, [6]MCP²⁺, and [6]CPP²⁺ (GIAO-B3LYP/6-311+G(2d,p))



	[6]MCP	[6]MCP ²⁺	[6]CPP ²⁺
NICS(belt center)	-1.19	-19.97	-18.74
NICS(-1)	-7.03	-25.63	-25.37
NICS(0)	-5.68	-9.01	-8.96
NICS(1)	-5.38	-3.56	-4.13

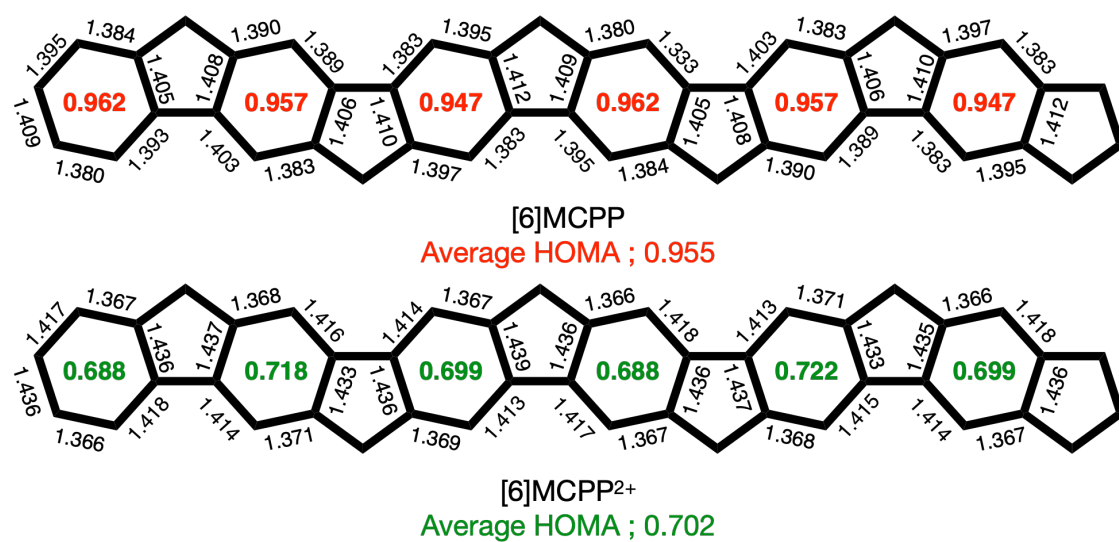


Figure S8. Bond length (Å) and HOMA value of [6]MCP and [6]MCP²⁺ (belt structure shown unfolded)

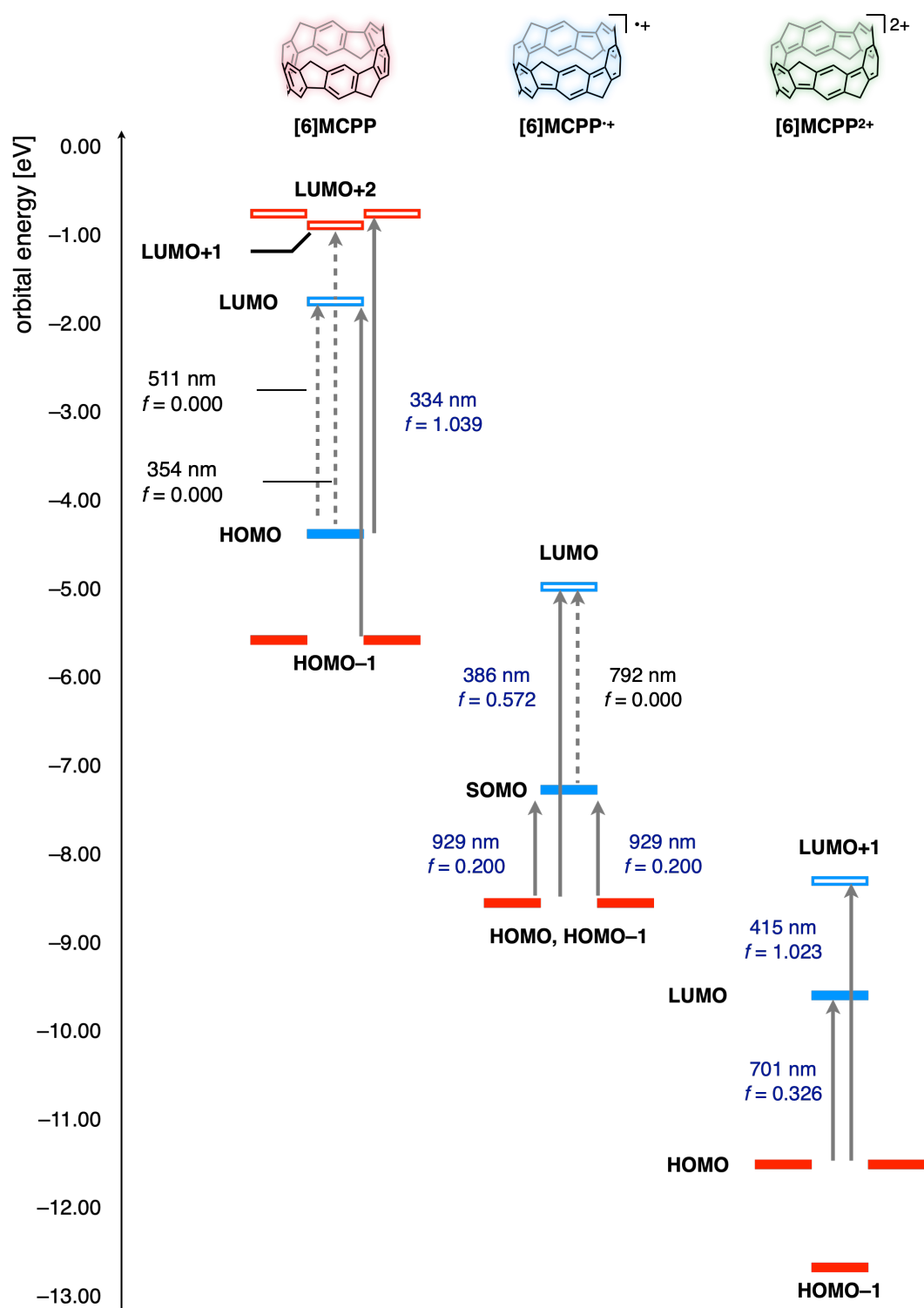


Figure S9. TD-DFT calculations of [6]MCP, [6]MCP⁺ and [6]MCP²⁺ ((U)B3LYP/6-31G(d)).

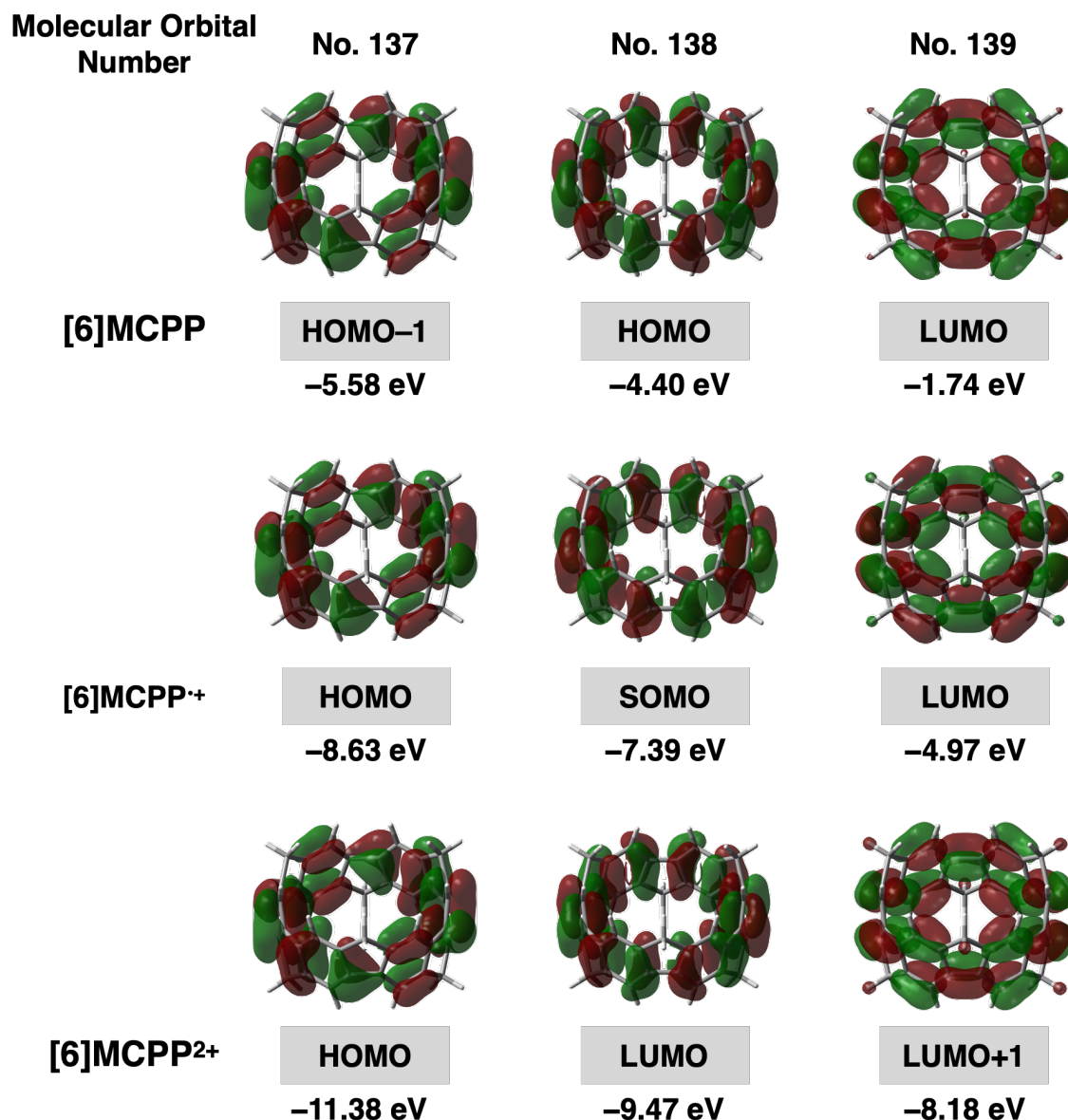


Figure S10. Orbital energies of [6]MCP, [6]MCP^{•+} and [6]MCP²⁺ (B3LYP/6-31G(d)).

In order to reevaluate the origin of the weak absorption (SOMO to LUMO transition) appearing around 890 nm in **1**, 15 different calculations were performed under multiple DFT/TD-DFT conditions. Since long-range-corrected and high-HF exchange-rate functionalities seem to be suitable to describe both electronic states and excited states of radical species, we combined UB3LYP, CAM-B3LYP, M06-2X, PBE0, or ω B97X- D with 6-31+G(d,p) or 6-311++G(d,p). Note that the aug-cc-pVTZ basis, which is expected to give more accurate results, was also investigated. However, the calculation has not been successful probably due to the very high computational cost. Therefore, we verified the oscillator strength of the SOMO to LUMO transition for 10 different calculation results. The results are as follows.

UB3LYP/6-31+G(d,p)

Excited State 1: 2.011-AU 1.3188 eV 940.10 nm f=0.2112 <S**2>=0.761
 136B ->139B 0.14122
 137B ->138B 0.98254

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1614.74188871

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.011-AU 1.3191 eV 939.91 nm f=0.2109 <S**2>=0.761
 136B ->138B 0.98254
 137B ->139B -0.14123

Excited State 3: 2.098-AG 1.5212 eV 815.05 nm **f=0.0000** <S**2>=0.851
 136A ->141A 0.11526
 137A ->140A -0.11535
138A ->139A 0.97534

Excited State 4: 2.022-AU 2.2637 eV 547.70 nm f=0.0000 <S**2>=0.772
 135B ->138B 0.99517

UB3LYP/6-311++G(d,p)

Excited State 1: 2.011-AU 1.3326 eV 930.42 nm f=0.2073 <S**2>=0.761
 136B -> 139B -0.14293
 137B -> 138B 0.98210

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1615.02243305

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.011-AU 1.3329 eV 930.19 nm f=0.2070 <S**2>=0.761
 136B -> 138B 0.98209
 137B -> 139B 0.14295

Excited State 3: 2.092-AG 1.5054 eV 823.58 nm **f=0.0000** <S**2>=0.844
 136A -> 141A -0.11228
 137A -> 140A 0.11238
138A -> 139A 0.97705

Excited State 4: 2.022-AU 2.2763 eV 544.67 nm f=0.0000 <S**2>=0.772
 135B -> 138B 0.99521

CAM-B3LYP/6-31+G(d,p)

Excited State 1: 2.065-AU 0.9790 eV 1266.49 nm f=0.2621 <S**2>=0.816
 136B ->139B 0.24808
 137B ->138B 0.99602
 136B <-139B 0.18817
 137B <-138B 0.20919

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1613.81408401

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State	2:	2.065-AU	0.9794 eV	1265.86 nm	f=0.2617	<S**2>=0.816
136B ->138B		0.99598				
137B ->139B		-0.24815				
136B <-138B		0.20907				
137B <-139B		-0.18818				

Excited State	3:	2.361-AG	1.8004 eV	688.66 nm	f=0.0000	<S**2>=1.144
136A ->141A		-0.24767				
137A ->140A		0.24787				
138A ->139A		0.89415				
136B ->142B		0.12631				
137B ->141B		-0.12644				

Excited State	4:	3.166-AU	2.3471 eV	528.23 nm	f=0.0000	<S**2>=2.257
137A ->139A		-0.42853				
138A ->140A		-0.44645				
128B ->141B		0.10398				
129B ->142B		0.10395				
136B ->138B		0.16876				
137B ->139B		0.67761				

CAM-B3LYP/6-311++G(d,p)

Excited State	1:	2.066-AU	0.9984 eV	1241.89 nm	f=0.2562	<S**2>=0.817
136B -> 139B		-0.25110				
137B -> 138B		0.99484				
136B <- 139B		-0.18787				
137B <- 138B		0.20740				

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1614.09974294

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State	2:	2.066-AU	0.9989 eV	1241.24 nm	f=0.2558	<S**2>=0.817
136B -> 138B		0.99479				
137B -> 139B		0.25118				
136B <- 138B		0.20728				
137B <- 139B		0.18787				

Excited State	3:	2.342-AG	1.7866 eV	693.97 nm	f=0.0000	<S**2>=1.121
136A -> 141A		-0.24344				
137A -> 140A		0.24365				
138A -> 139A		0.90062				
136B -> 142B		0.12084				

137B -> 141B -0.12096

Excited State 4: 3.164-AU 2.3525 eV 527.03 nm f=0.0000 <S**2>=2.253

137A -> 139A -0.42835

138A -> 140A -0.44625

128B -> 141B 0.10250

129B -> 142B 0.10246

136B -> 138B -0.17011

137B -> 139B 0.68318

M06-2X/6-31+G(d,p)

Excited State 1: 2.019-AU 1.1600 eV 1068.82 nm f=0.2555 <S**2>=0.769

136B ->139B -0.17391

137B ->138B 0.99250

136B <-139B -0.13980

137B <-138B 0.11797

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1614.07895124

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.019-AU 1.1602 eV 1068.69 nm f=0.2551 <S**2>=0.769

136B ->138B 0.99249

137B ->139B 0.17391

136B <-138B 0.11796

137B <-139B 0.13980

Excited State 3: 2.174-AG 1.9564 eV 633.75 nm **f=0.0000** <S**2>=0.931

136A ->141A -0.19296

137A ->140A 0.19310

138A ->139A 0.94838

Excited State 4: 3.170-AU 2.6232 eV 472.65 nm f=0.0003 <S**2>=2.263

137A ->139A -0.40897

138A ->140A -0.42536

137B ->139B 0.76280

M06-2X/6-311++G(d,p)

Excited State 1: 2.020-AU 1.1699 eV 1059.76 nm f=0.2505 <S**2>=0.770

136B -> 139B -0.17890

137B -> 138B 0.99193

136B <- 139B -0.14126

137B <- 138B 0.12042

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1614.38670142

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.020-AU 1.1704 eV 1059.33 nm f=0.2500 <S**2>=0.770
 136B -> 138B 0.99192
 137B -> 139B 0.17894
 136B <- 138B 0.12037
 137B <- 139B 0.14127

Excited State 3: 2.173-AG 1.9164 eV 646.97 nm **f=0.0000** <S**2>=0.930
 136A -> 141A -0.19236
 137A -> 140A 0.19255
138A -> 139A 0.94875

Excited State 4: 3.169-AU 2.6016 eV 476.56 nm f=0.0002 <S**2>=2.260
 137A -> 139A -0.41182
 138A -> 140A -0.42893
 137B -> 139B 0.75834

PBE0/6-31+G(d,p)

Excited State 1: 2.022-AU 1.3171 eV 941.37 nm f=0.2260 <S**2>=0.772
 136B ->139B -0.16033
 137B ->138B 0.98154

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1612.88700449

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.022-AU 1.3173 eV 941.20 nm f=0.2257 <S**2>=0.772
 136B ->138B 0.98154
 137B ->139B 0.16035

Excited State 3: 2.150-AG 1.5564 eV 796.62 nm **f=0.0000** <S**2>=0.905
 136A ->141A -0.14198
 137A ->140A 0.14208
138A ->139A 0.96178

Excited State 4: 3.154-AU 2.3133 eV 535.96 nm f=0.0029 <S**2>=2.236
 137A ->139A -0.43780
 138A ->140A -0.46519
 137B ->139B 0.70484

PBE0/6-311++G(d,p)

Excited State 1: 2.022-AU 1.3304 eV 931.90 nm f=0.2220 <S**2>=0.772
 136B -> 139B -0.16260
 137B -> 138B 0.98110

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1613.14475326

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.022-AU 1.3308 eV 931.66 nm f=0.2217 <S**2>=0.772
 136B -> 138B 0.98109
 137B -> 139B 0.16263

Excited State 3: 2.141-AG 1.5405 eV 804.83 nm **f=0.0000** <S**2>=0.895
 136A -> 141A -0.13855
 137A -> 140A 0.13867
138A -> 139A 0.96434

Excited State 4: 3.153-AU 2.3165 eV 535.23 nm f=0.0025 <S**2>=2.235
 137A -> 139A -0.43722
 138A -> 140A -0.46411
 137B -> 139B 0.70926

ω B97X-D/6-31+G(d,p)

Excited State 1: 2.065-A 0.9025 eV 1373.81 nm f=0.2587 <S**2>=0.816
 136B ->138B 1.00816
 137B ->139B 0.25636
 136B <-138B 0.27274
 137B <-139B 0.19895

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1614.21238351

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.056-A 1.0430 eV 1188.71 nm f=0.2607 <S**2>=0.806
 136B ->139B -0.20958
 137B ->138B 1.00176
 136B <-139B -0.17569
 137B <-138B 0.21735

Excited State 3: 2.439-A 1.9294 eV 642.59 nm **f=0.0065** <S**2>=1.237
 136A ->140A 0.25195
 137A ->141A 0.26671
138A ->139A 0.87635
 136B ->141B 0.11818
 137B ->142B -0.12080

Excited State 4: 3.098-A 2.4432 eV 507.47 nm f=0.0000 <S**2>=2.150
 136A ->139A -0.42306
 138A ->139A 0.12214
 138A ->140A -0.40941
 138A ->142A -0.12205
 128B ->141B 0.10787
 129B ->142B 0.10198
 133B ->138B 0.11334
 136B ->138B -0.14562

137B -> 139B	0.66169
137B -> 142B	0.11064

ω B97X-D/6-311++G(d,p)

Excited State 1: 2.067-A 0.9135 eV 1357.29 nm f=0.2530 <S**2>=0.818

136B -> 138B	1.00753
137B -> 139B	0.26126
136B <- 138B	0.27319
137B <- 139B	0.20107

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-DFT) = -1614.48114555

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: 2.057-A 1.0465 eV 1184.78 nm f=0.2557 <S**2>=0.808

136B -> 139B	-0.21669
137B -> 138B	1.00167
136B <- 139B	-0.17876
137B <- 138B	0.22106

Excited State 3: 2.417-A 1.9126 eV 648.24 nm **f=0.0060** <S**2>=1.210

136A -> 140A	0.24790
137A -> 141A	0.26351
138A -> 139A	0.88302
136B -> 142B	-0.11429
137B -> 141B	0.11711

Excited State 4: 3.114-A 2.4413 eV 507.87 nm f=0.0000 <S**2>=2.174

136A -> 139A	-0.42443
138A -> 139A	0.11212
138A -> 140A	-0.40945
138A -> 142A	0.13298
136B -> 138B	-0.15221
137B -> 139B	0.67106
137B -> 141B	-0.10741

Based on these calculations, the SOMO (138) $\rightarrow$ LUMO (139) did not clearly appear as a transition with large contribution. This result might indicate a limitation of the (TD)DFT calculation, particularly the inability of the standard TD-DFT with the linear response method to adequately reproduce the subtle oscillator strengths of the electronic states and forbidden transitions of radical species with complicated properties. We think it would be necessary to investigate more sophisticated methods and corrections with vibration effects for further discussion with experimental results.

Belt current effect on the ^{13}C NMR signals

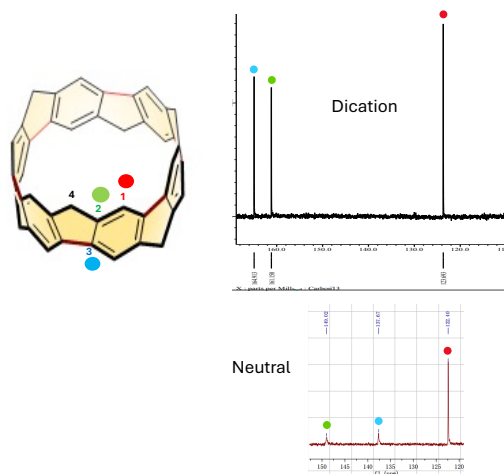


Table S5. [6]MCPD dication - calculated ^{13}C NMR chemical shifts for the free molecule in ppm, reference TMS. AvAD=Averaged Absolute Deviations with respect to experimental results: 123.69 C1, 161.15 C2, 164.91 C3, and 36.25 C4.

Functional	CSGT					GIAO				
	C1	C2	C3	C4	AvAD	C1	C2	C3	C4	AvAD
B3LYP	126.08	169.15	171.17	40.83	5.31	126.28	169.98	171.95	41.66	5.97
B97-2	122.42	163.55	165.48	39.56	1.89	122.12	164.13	166.01	40.18	2.39
BHandHLYP	127.05	170.77	173.12	37.96	5.72	127.10	171.35	173.68	38.51	6.16

Table S6. [6]MCPD dication in CD_2Cl_2 , see Table S5 for other details.

Functional	CSGT					GIAO				
	C1	C2	C3	C4	AvAD	C1	C2	C3	C4	AvAD
B3LYP	127.82	169.71	172.59	41.07	6.30	128.06	170.52	173.37	41.90	6.96
B972	124.14	164.17	166.92	39.82	2.26	124.15	164.67	167.36	40.40	2.65
BHandHLYP	128.73	171.46	174.60	38.22	6.75	128.65	171.95	175.17	38.70	7.12

Table S7. [6]MCPD neutral for the free molecule, see Table S5 for other details.

Functional	CSGT					GIAO				
	C1	C2	C3	C4	AvAD	C1	C2	C3	C4	AvAD
B3LYP	124.94	155.08	141.67	44.75	4.04	125.21	156.04	142.57	45.52	4.76
B97-2	122.07	150.16	137.36	43.51	1.02	121.68	150.87	138.04	44.09	1.46
BHandHLYP	126.85	156.24	144.42	42.46	4.92	126.87	156.94	145.11	43.03	5.41

Table S8. [6]MCPD neutral in CD₂Cl₂, see Table S5 for other details.

Functional	CSGT					GIAO				
	C1	C2	C3	C4	AvAD	C1	C2	C3	C4	AvAD
B3LYP	125.99	156.52	142.28	45.06	4.89	126.28	157.37	142.80	46.00	5.54
B97-2	123.12	151.58	137.91	43.81	1.53	123.06	152.22	138.48	44.33	1.95
BHandHLYP	127.91	157.62	144.96	42.73	5.73	127.93	158.23	145.55	43.22	6.16

Both molecules belong to the D_{3d} point group symmetry and display four signals in the decoupled ¹³C NMR spectra. These have been arbitrarily numerated as shown in the figure. A color has been attached to them: red C1, green C2, blue C3, and black C4.

Signal assignment is trivial for C1 and C4. The assignment of C2 and C3 can be made based on the calculation results. This is indicated by the small colored dots reported in the NMR spectra above. The NMR spectrum of the neutral species is from the supplementary information by Kono et al. *J. Am. Chem. Soc.* **2023**, *145*, 8939–8946. C4 is not shown.

The calculations were carried out using the 6-311+G(2d,p) basis set. Three functionals were adopted: B3LYP, B97-2, and BHandHLYP. Molecular geometries for neutral and dication, as well as TMS, were optimized at the same level of accuracy adopted for the NMR calculations. Chemical shifts were obtained for CSGT and GIAO methods for the free molecules and including solvent effect (dichloromethane).

Calculated chemical shifts are reported in four table. Table 1 for dication no solvent effects (free); Table 2 for dication in dichloromethane; Table 3 neutral free; Table 4 neutral in dichloromethane. In each table, the three functionals can be compared for CSGT e GIAO. To facilitate ranking the absolute deviation from experimental chemical shifts averaged over the four carbons is also reported under the heading AvAD. Without exceptions the combination B97-2/CSGT provides the smaller AvAD (best matching). The results are better for the free molecule with respect to the inclusion of the solvent effects. Remarkably, it turns out that CSGT is always a little better than GIAO.

The interesting fact is that, changing the oxidation state, two of the experimental carbon signals, namely C2 and C3, move considerably downfield (C2 by ~12 ppm and C3 by ~27 ppm); that of C4 moves ~5 ppm upfield; C1 signal changes very little remaining within the aromatic region. Theoretical results (see Table 1 and 3) reproduce the experimental shifts nicely.

Consequently, it seems quite likely that C2, C3, C4 shifts are related to the reversal of the belt current. To see whether this is the case, we considered the feature of the CSGT method of dissecting the induced current and related magnetic properties into orbital contributions and performed the following theoretical experiment. Since, in both molecules, the belt currents are induced by a magnetic field parallel to the main symmetry axis and are almost completely due to virtual transitions from the frontier orbitals, we have calculated the contributions to the parallel component of the magnetic shielding $\sigma_{\parallel}^N(\text{MFO})$ at each Nucleus (N) of interest due to the Molecular Frontier Orbitals (MFO). Then, we have taken the difference $\sigma_{\parallel}^N(\text{dication}) - \sigma_{\parallel}^N(\text{neutral})$, for N=C2, C3, and C4 (also H_{in} , H_{out} , and H_{ar}) changed the sign (from shielding to chemical shift) and divided by three (weight to average) to obtain the contributions to the relative chemical shifts. Results are shown in Table S9.

Table S9. Predicted change of the NMR chemical shifts due to the belt current.

N	$\sigma_{\parallel}^N(\text{MFO})$		Relative shifts	
	dication	neutral	Theor.	Expt.
H_{in}	13.27	-3.35	-5.54	-4.31
H_{out}	1.51	-0.38	-0.63	-0.62
H_{ar}	11.32	-2.74	-4.69	-3.67
C2	-13.34	6.41	6.58	12.13
C3	-33.33	17.04	16.79	27.24
C4	6.74	-2.52	-3.09	-4.95

As can be observed, the agreement between theoretical estimates and experimental relative shifts is encouraging, especially for the proton. For ^{13}C some other contribution is evidently necessary, but it is interesting to observe the concordance of the shift (down or up field) and the relative magnitude which are proportionate to the experimental ones.

Table S10. Cartesian coordinates of optimized structure of [6]MCP²⁺ in S₀.

C	-3.927363	-0.433079	1.373436	C	-1.588624	-3.617736	-1.373436
C	-3.927363	0.433079	-1.373436	C	-2.733225	-2.918795	-1.066388
C	-3.766034	1.345486	-0.298062	C	-3.048242	-2.588738	0.298062
C	-3.894363	0.907645	1.066388	C	-2.338739	-3.184657	1.373436
C	-3.048242	2.588738	-0.298062	C	-1.161138	-3.826440	1.066388
C	-2.338739	3.184657	-1.373436	C	0.000000	-4.137827	1.992961
C	-1.161138	3.826440	-1.066388	C	-3.583464	-2.068914	-1.992961
C	-0.717792	3.934224	0.298062	C	-3.766034	-1.345486	0.298062
C	-1.588624	3.617736	1.373436	C	-3.894363	-0.907645	-1.066388
C	-2.733225	2.918795	1.066388	H	-3.888407	-0.771182	2.404682
C	-3.583464	2.068914	1.992961	H	-3.888407	0.771182	-2.404682
C	0.000000	4.137827	-1.992961	H	-2.612066	2.981868	-2.404682
C	1.588624	3.617736	1.373436	H	-1.276340	3.753050	2.404682
C	0.717792	3.934224	0.298062	H	-4.499631	2.597863	2.291710
C	1.161138	3.826440	-1.066388	H	-3.064720	1.769417	2.907575
C	2.338739	3.184657	-1.373436	H	0.000000	3.538834	-2.907575
C	3.048242	2.588738	-0.298062	H	0.000000	5.195726	-2.291710
C	2.733225	2.918795	1.066388	H	1.276340	3.753050	2.404682
C	3.766034	1.345486	-0.298062	H	2.612066	2.981868	-2.404682
C	3.927363	0.433079	-1.373436	H	3.888407	0.771182	-2.404682
C	3.894363	-0.907645	-1.066388	H	3.888407	-0.771182	2.404682
C	3.766034	-1.345486	0.298062	H	4.499631	2.597863	2.291710
C	3.927363	-0.433079	1.373436	H	3.064720	1.769417	2.907575
C	3.894363	0.907645	1.066388	H	3.064720	-1.769417	-2.907575
C	3.583464	2.068914	1.992961	H	4.499631	-2.597863	-2.291710
C	3.583464	-2.068914	-1.992961	H	2.612066	-2.981868	2.404682
C	2.338739	-3.184657	1.373436	H	1.276340	-3.753050	-2.404682
C	3.048242	-2.588738	0.298062	H	-1.276340	-3.753050	-2.404682
C	2.733225	-2.918795	-1.066388	H	-2.612066	-2.981868	2.404682
C	1.588624	-3.617736	-1.373436	H	0.000000	-5.195726	2.291710
C	0.717792	-3.934224	-0.298062	H	0.000000	-3.538834	2.907575
C	1.161138	-3.826440	1.066388	H	-3.064720	-1.769417	-2.907575
C	-0.717792	-3.934224	-0.298062	H	-4.499631	-2.597863	-2.291710

Table S11. Cartesian coordinates of optimized structure of [6]MCP in S_0 .

C	-3.959872	-0.443140	1.350744	C	-1.596166	-3.650920	-1.350744
C	-3.959872	0.443140	-1.350744	C	-2.755141	-2.944057	-1.046616
C	-3.831978	1.358926	-0.293764	C	-3.092854	-2.639127	0.293764
C	-3.927199	0.913993	1.046616	C	-2.363706	-3.207780	1.350744
C	-3.092854	2.639127	-0.293764	C	-1.172058	-3.858051	1.046616
C	-2.363706	3.207780	-1.350744	C	0.000000	-4.101438	1.985672
C	-1.172058	3.858051	-1.046616	C	-3.551950	-2.050719	-1.985672
C	-0.739125	3.998054	0.293764	C	-3.831978	-1.358926	0.293764
C	-1.596166	3.650920	1.350744	C	-3.927199	-0.913993	-1.046616
C	-2.755141	2.944057	1.046616	H	-3.868157	-0.770805	2.383887
C	-3.551950	2.050719	1.985672	H	-3.868157	0.770805	-2.383887
C	0.000000	4.101438	-1.985672	H	-2.601615	2.964520	-2.383887
C	1.596166	3.650920	1.350744	H	-1.266542	3.735325	2.383887
C	0.739125	3.998054	0.293764	H	-4.442403	2.564822	2.378107
C	1.172058	3.858051	-1.046616	H	-2.968498	1.713863	2.849209
C	2.363706	3.207780	-1.350744	H	0.000000	3.427726	-2.849209
C	3.092854	2.639127	-0.293764	H	0.000000	5.129645	-2.378107
C	2.755141	2.944057	1.046616	H	1.266542	3.735325	2.383887
C	3.831978	1.358926	-0.293764	H	2.601615	2.964520	-2.383887
C	3.959872	0.443140	-1.350744	H	3.868157	0.770805	-2.383887
C	3.927199	-0.913993	-1.046616	H	3.868157	-0.770805	2.383887
C	3.831978	-1.358926	0.293764	H	4.442403	2.564822	2.378107
C	3.959872	-0.443140	1.350744	H	2.968498	1.713863	2.849209
C	3.927199	0.913993	1.046616	H	2.968498	-1.713863	-2.849209
C	3.551950	2.050719	1.985672	H	4.442403	-2.564822	-2.378107
C	3.551950	-2.050719	-1.985672	H	2.601615	-2.964520	2.383887
C	2.363706	-3.207780	1.350744	H	1.266542	-3.735325	-2.383887
C	3.092854	-2.639127	0.293764	H	-1.266542	-3.735325	-2.383887
C	2.755141	-2.944057	-1.046616	H	-2.601615	-2.964520	2.383887
C	1.596166	-3.650920	-1.350744	H	0.000000	-5.129645	2.378107
C	0.739125	-3.998054	-0.293764	H	0.000000	-3.427726	2.849209
C	1.172058	-3.858051	1.046616	H	-2.968498	-1.713863	-2.849209
C	-0.739125	-3.998054	-0.293764	H	-4.442403	-2.564822	-2.378107

Table S12. Cartesian coordinates of optimized structure of [6]CPP²⁺ in S₀.

C	4.156120	0.354290	1.210323	C	1.713879	-3.782190	0.000013
C	4.132728	-0.406913	-0.000015	C	2.384662	-3.421724	1.210396
C	4.155919	0.354240	-1.210388	C	3.367056	-2.461438	1.210399
C	3.815318	1.685122	-1.210390	C	3.742710	-1.799035	0.000016
C	3.429352	2.341570	-0.000016	C	3.367188	-2.461571	-1.210337
C	3.815521	1.685172	1.210324	C	2.384791	-3.421851	-1.210338
C	1.771084	3.775957	1.210394	H	4.334108	-0.133406	2.161798
C	2.418670	3.375304	0.000014	H	4.333689	-0.133507	-2.161876
C	1.771135	3.776142	-1.210334	H	3.736877	2.198288	-2.161885
C	0.448276	4.146666	-1.210336	H	3.737300	2.198400	2.161805
C	-0.313256	4.140540	0.000013	H	2.282356	3.685952	2.161887
C	0.448224	4.146486	1.210394	H	2.282469	3.686338	-2.161815
C	-1.713878	3.782191	-0.000014	H	-0.035269	4.335624	-2.161813
C	-2.384787	3.421852	1.210338	H	-0.035371	4.335257	2.161882
C	-3.367185	2.461571	1.210340	H	-2.051343	3.819769	2.161815
C	-3.742711	1.799036	-0.000012	H	-3.772568	2.137256	2.161828
C	-3.367060	2.461438	-1.210396	H	-3.772299	2.136978	-2.161896
C	-2.384665	3.421724	-1.210396	H	-2.051090	3.819503	-2.161884
C	-4.132729	0.406913	0.000019	H	-4.333687	0.133504	2.161881
C	-4.155918	-0.354242	1.210391	H	-3.736875	-2.198291	2.161886
C	-3.815317	-1.685124	1.210392	H	-3.737303	-2.198398	-2.161805
C	-3.429352	-2.341571	0.000016	H	-4.334112	0.133409	-2.161794
C	-3.815523	-1.685171	-1.210323	H	-2.282355	-3.685946	-2.161890
C	-4.156123	-0.354289	-1.210319	H	0.035373	-4.335251	-2.161886
C	-2.418670	-3.375305	-0.000016	H	0.035270	-4.335630	2.161809
C	-1.771083	-3.775954	-1.210398	H	-2.282469	-3.686345	2.161811
C	-0.448223	-4.146482	-1.210398	H	2.051084	-3.819503	2.161883
C	0.313257	-4.140540	-0.000016	H	3.772292	-2.136977	2.161900
C	-0.448276	-4.146669	1.210332	H	3.772574	-2.137255	-2.161824
C	-1.771135	-3.776146	1.210331	H	2.051349	-3.819769	-2.161815

Table S13. Cartesian coordinates of optimized structure of [6]MCPP⁺ in S₀.

C	3.633077	-1.590825	1.361586	C	3.194233	2.350924	-1.361586
C	3.194498	-2.351078	-1.361528	C	3.840049	1.166244	-1.056441
C	2.613637	-3.070214	-0.295258	C	3.965702	0.728370	0.295258
C	2.930021	-2.742458	1.056441	C	3.633342	1.590978	1.361528
C	1.352094	-3.798578	-0.295303	C	2.930177	2.742548	1.056400
C	0.438844	-3.941749	-1.361586	C	2.058263	3.564942	1.990236
C	-0.910027	-3.908702	-1.056441	C	4.116462	-0.000037	-1.990236
C	-1.352064	-3.798583	0.295258	C	3.965712	-0.728341	0.295303
C	-0.438844	-3.942056	1.361528	C	3.840205	-1.166333	-1.056400
C	0.910028	-3.908881	1.056400	H	3.742013	-1.268998	2.393727
C	2.058199	-3.564979	1.990236	H	2.970311	-2.606345	-2.393676
C	-2.058263	-3.564942	-1.990236	H	0.772021	-3.875177	-2.393727
C	-3.194233	-2.350924	1.361586	H	-0.772005	-3.875537	2.393676
C	-2.613618	-3.070236	0.295303	H	2.579890	-4.468378	2.337464
C	-2.930177	-2.742548	-1.056400	H	1.738929	-3.012069	2.878863
C	-3.633342	-1.590978	-1.361528	H	-1.739064	-3.011992	-2.878863
C	-3.965702	-0.728370	-0.295258	H	-2.579784	-4.468439	-2.337464
C	-3.840049	-1.166244	1.056441	H	-2.969991	-2.606179	2.393727
C	-3.965712	0.728341	-0.295303	H	-3.742316	-1.269193	-2.393676
C	-3.633077	1.590825	-1.361586	H	-3.742013	1.268998	-2.393727
C	-2.930021	2.742458	-1.056441	H	-2.970311	2.606345	2.393676
C	-2.613637	3.070214	0.295258	H	-5.159674	-0.000061	2.337464
C	-3.194498	2.351078	1.361528	H	-3.477993	0.000078	2.878863
C	-3.840205	1.166333	1.056400	H	-1.738929	3.012069	-2.878863
C	-4.116462	0.000037	1.990236	H	-2.579890	4.468378	-2.337464
C	-2.058199	3.564979	-1.990236	H	-0.772021	3.875177	2.393727
C	-0.438844	3.941749	1.361586	H	0.772005	3.875537	-2.393676
C	-1.352094	3.798578	0.295303	H	2.969991	2.606179	-2.393727
C	-0.910028	3.908881	-1.056400	H	3.742316	1.269193	2.393676
C	0.438844	3.942056	-1.361528	H	2.579784	4.468439	2.337464
C	1.352064	3.798583	-0.295258	H	1.739064	3.011992	2.878863
C	0.910027	3.908702	1.056441	H	3.477993	-0.000078	-2.878863
C	2.613618	3.070236	-0.295303	H	5.159674	0.000061	-2.337464

Table S14. Cartesian coordinates of optimized structure of [6]CPP⁺⁺ in S₀.

C	-0.011978	-3.974585	-1.254277	C	3.916985	-1.449117	0.040382
C	0.703359	-4.115980	-0.040020	C	3.737128	-2.252632	-1.117408
C	-0.082432	-4.362301	1.117718	C	2.845051	-3.308111	-1.117347
C	-1.442624	-4.117853	1.117657	C	2.082859	-3.619383	0.040510
C	-2.093380	-3.613286	-0.040134	C	2.544231	-3.054674	1.254752
C	-1.373434	-3.729803	-1.254331	C	3.437127	-1.998128	1.254681
C	-3.819756	-2.110027	-1.117376	H	0.522875	-3.881411	-2.192882
C	-3.213530	-2.667282	0.040429	H	0.394724	-4.653505	2.048188
C	-3.449126	-1.977131	1.254659	H	-1.991384	-4.224602	2.048099
C	-3.917892	-0.675665	1.254564	H	-1.842207	-3.456232	-2.193001
C	-4.176499	0.006086	0.040255	H	-3.833436	-2.668955	-2.047779
C	-4.288089	-0.809836	-1.117467	H	-3.101388	-2.393744	2.193404
C	-3.916985	1.449116	-0.040386	H	-3.915717	-0.132888	2.193237
C	-3.437124	1.998126	-1.254685	H	-4.655008	-0.388104	-2.047926
C	-2.544229	3.054671	-1.254757	H	-3.624160	1.488611	-2.193364
C	-2.082859	3.619383	-0.040515	H	-2.073099	3.324024	-2.193477
C	-2.845054	3.308114	1.117341	H	-2.662837	3.836550	2.047825
C	-3.737131	2.252635	1.117402	H	-4.227772	1.984954	2.047919
C	-0.703359	4.115980	0.040018	H	-0.394718	4.653507	-2.048188
C	0.082435	4.362301	-1.117717	H	1.991390	4.224603	-2.048094
C	1.442627	4.117854	-1.117652	H	1.842201	3.456230	2.193006
C	2.093380	3.613286	0.040139	H	-0.522881	3.881410	2.192880
C	1.373431	3.729801	1.254334	H	3.833434	2.668949	2.047785
C	0.011975	3.974585	1.254277	H	4.655006	0.388098	2.047927
C	3.213530	2.667282	-0.040423	H	3.915718	0.132893	-2.193237
C	3.819755	2.110024	1.117381	H	3.101390	2.393750	-2.193399
C	4.288089	0.809833	1.117468	H	4.227768	-1.984949	-2.047925
C	4.176499	-0.006086	-0.040256	H	2.662833	-3.836546	-2.047831
C	3.917892	0.675669	-1.254563	H	2.073104	-3.324029	2.193472
C	3.449127	1.977134	-1.254655	H	3.624164	-1.488616	2.193361

Table S15. Cartesian coordinates of optimized structure of [6]MCP²⁺ in S₀. (CAM-B3LYP/6-31+G(d,p)).

C	-1.369692	0.435172	3.898534	C	1.370236	3.600151	1.574123
C	1.369692	-0.435172	3.898534	C	1.067227	2.904007	2.713773
C	0.291915	-1.339286	3.743448	C	-0.291751	2.576794	3.031415
C	-1.066940	-0.899975	3.867342	C	-1.369374	3.165286	2.327337
C	0.291751	-2.576794	3.031415	C	-1.066365	3.807050	1.156123
C	1.369374	-3.165286	2.327337	C	-1.992464	4.106264	0.000000
C	1.066365	-3.807050	1.156123	C	1.993543	2.050090	3.548070
C	-0.292551	-3.918657	0.713824	C	-0.291915	1.339286	3.743448
C	-1.370236	-3.600151	1.574123	C	1.066940	0.899975	3.867342
C	-1.067227	-2.904007	2.713773	H	-2.397841	0.775204	3.839935
C	-1.993543	-2.050090	3.548070	H	2.397841	-0.775204	3.839935
C	1.992464	-4.106264	0.000000	H	2.397684	-2.944920	2.592442
C	-1.370236	-3.600151	-1.574123	H	-2.398333	-3.718773	1.249933
C	-0.292551	-3.918657	-0.713824	H	-2.304205	-2.574584	4.459388
C	1.066365	-3.807050	-1.156123	H	-2.892405	-1.741876	3.012359
C	1.369374	-3.165286	-2.327337	H	2.892705	-3.490341	0.000000
C	0.291751	-2.576794	-3.031415	H	2.300686	-5.158474	0.000000
C	-1.067227	-2.904007	-2.713773	H	-2.398333	-3.718773	-1.249933
C	0.291915	-1.339286	-3.743448	H	2.397684	-2.944920	-2.592442
C	1.369692	-0.435172	-3.898534	H	2.397841	-0.775204	-3.839935
C	1.066940	0.899975	-3.867342	H	-2.397841	0.775204	-3.839935
C	-0.291915	1.339286	-3.743448	H	-2.304205	-2.574584	-4.459388
C	-1.369692	0.435172	-3.898534	H	-2.892405	-1.741876	-3.012359
C	-1.066940	-0.899975	-3.867342	H	2.892405	1.741876	-3.012359
C	-1.993543	-2.050090	-3.548070	H	2.304205	2.574584	-4.459388
C	1.993543	2.050090	-3.548070	H	-2.397684	2.944920	-2.592442
C	-1.369374	3.165286	-2.327337	H	2.398333	3.718773	-1.249933
C	-0.291751	2.576794	-3.031415	H	2.398333	3.718773	1.249933
C	1.067227	2.904007	-2.713773	H	-2.397684	2.944920	2.592442
C	1.370236	3.600151	-1.574123	H	-2.300686	5.158474	0.000000
C	0.292551	3.918657	-0.713824	H	-2.892705	3.490341	0.000000
C	-1.066365	3.807050	-1.156123	H	2.892405	1.741876	3.012359
C	0.292551	3.918657	0.713824	H	2.304205	2.574584	4.459388

Table S16. Cartesian coordinates of optimized structure of [6]CPP²⁺ in S₁.

C	-2.30320	-3.34660	1.24120	C	3.44180	-2.35390	-0.01810
C	-1.04400	-3.90680	1.24010	C	2.90440	-2.86400	-1.24120
C	-0.33740	-4.11070	0.01870	C	1.77200	-3.64790	-1.23990
C	-1.12580	-4.08630	-1.17240	C	1.11270	-3.97520	-0.01820
C	-2.38150	-3.52230	-1.17350	C	1.88300	-3.80520	1.17360
C	-2.92820	-2.95160	0.01870	C	3.01070	-3.01690	1.17510
C	-3.77760	-1.77910	-0.01590	H	-2.75150	-3.09130	2.19420
C	-3.98140	-1.06510	-1.23990	H	-0.55180	-4.06620	2.19230
C	-4.11510	0.30230	-1.24030	H	-0.70930	-4.43040	-2.11270
C	-4.05200	1.04330	-0.01600	H	-2.91360	-3.43890	-2.11470
C	-4.28160	0.28570	1.18000	H	-3.90160	-1.57720	-2.19190
C	-4.14920	-1.08050	1.17980	H	-4.13680	0.81980	-2.19240
C	-3.44180	2.35390	0.01810	H	-4.47770	0.79180	2.11890
C	-3.01070	3.01690	-1.17510	H	-4.24480	-1.61520	2.11850
C	-1.88300	3.80520	-1.17360	H	-3.51820	2.83590	-2.11620
C	-1.11270	3.97520	0.01820	H	-1.53800	4.22190	-2.11340
C	-1.77200	3.64790	1.23990	H	-1.31980	3.90000	2.19200
C	-2.90440	2.86400	1.24120	H	-3.30090	2.53360	2.19410
C	0.33740	4.11070	-0.01870	H	0.55180	4.06620	-2.19230
C	1.04400	3.90680	-1.24010	H	2.75150	3.09130	-2.19420
C	2.30320	3.34660	-1.24120	H	2.91360	3.43890	2.11470
C	2.92820	2.95160	-0.01880	H	0.70940	4.43040	2.11270
C	2.38150	3.52230	1.17350	H	4.24480	1.61520	-2.11850
C	1.12580	4.08630	1.17240	H	4.47770	-0.79190	-2.11890
C	3.77760	1.77910	0.01590	H	4.13680	-0.81980	2.19240
C	4.14920	1.08050	-1.17980	H	3.90160	1.57730	2.19190
C	4.28160	-0.28570	-1.18000	H	3.30090	-2.53370	-2.19410
C	4.05200	-1.04330	0.01600	H	1.31980	-3.90000	-2.19200
C	4.11510	-0.30230	1.24030	H	1.53800	-4.22190	2.11340
C	3.98140	1.06520	1.23990	H	3.51820	-2.83590	2.11620

Excited State 1: Singlet-A 1.7324 eV 715.69 nm f=0.2078 <S**2>=0.000
118 ->121 -0.29878
119 ->120 0.64993

Table S17. Cartesian coordinates of optimized structure of [6]MCP²⁺ in S₁.

C	-3.07780	2.60580	0.29930	C	-1.57980	-3.60300	1.37210
C	-2.75710	2.93080	-1.06830	C	-3.79520	1.36270	0.29930
C	-1.60240	3.61680	-1.37570	C	-3.91610	0.92250	-1.06820
C	-0.73220	3.92940	-0.30060	C	-3.61640	2.08810	-1.99380
C	-1.17580	3.82790	1.06430	C	-3.94920	0.44480	1.37390
C	-2.36000	3.19850	1.37390	C	-3.90230	-0.89550	1.06430
C	0.71360	3.92900	-0.29970	C	-3.76850	-1.33040	-0.30070
C	1.14940	3.81630	1.06250	C	-3.93260	-0.42040	-1.37580
C	-0.01060	4.13220	1.98920	C	-3.58320	-2.05670	1.98910
C	1.57980	3.60300	-1.37210	H	-1.28820	3.74660	-2.40710
C	2.73000	2.90360	-1.06240	H	-2.63660	3.00340	2.40580
C	3.04550	2.58250	0.29980	H	-0.01340	3.53340	2.90400
C	2.33000	3.16970	1.37220	H	-0.00440	5.19010	2.28850
C	3.58320	2.05670	-1.98910	H	1.26890	3.73440	-2.40430
C	3.90230	0.89550	-1.06430	H	2.59930	2.96610	2.40440
C	3.76850	1.33040	0.30070	H	3.06580	1.75490	-2.90370
C	3.94920	-0.44480	-1.37390	H	4.49630	2.59080	-2.28880
C	3.79520	-1.36270	-0.29930	H	3.91840	-0.78180	-2.40580
C	3.91610	-0.92250	1.06820	H	3.88780	0.75740	2.40710
C	3.93260	0.42040	1.37580	H	3.10660	-1.79390	2.91500
C	3.07780	-2.60580	-0.29930	H	4.53590	-2.61880	2.27910
C	2.75710	-2.93080	1.06830	H	2.63660	-3.00340	-2.40580
C	3.61640	-2.08810	1.99380	H	1.28820	-3.74660	2.40710
C	2.36000	-3.19850	-1.37390	H	0.01340	-3.53340	-2.90400
C	1.17580	-3.82790	-1.06430	H	0.00440	-5.19010	-2.28850
C	0.73220	-3.92940	0.30060	H	-2.59930	-2.96610	-2.40440
C	1.60240	-3.61680	1.37570	H	-1.26890	-3.73440	2.40430
C	0.01060	-4.13220	-1.98920	H	-3.10660	1.79390	-2.91500
C	-1.14940	-3.81630	-1.06250	H	-4.53590	2.61880	-2.27910
C	-0.71360	-3.92900	0.29970	H	-3.91840	0.78180	2.40580
C	-2.33000	-3.16970	-1.37220	H	-3.88780	-0.75740	-2.40710
C	-3.04550	-2.58250	-0.29980	H	-4.49630	-2.59080	2.28880
C	-2.73000	-2.90360	1.06240	H	-3.06580	-1.75490	2.90370

Excited State 1: Singlet-AU 1.8063 eV 686.38 nm f=0.2164 <S**2>=0.000
136 ->139 -0.26628
137 ->138 0.66274

6. NMR, ESR, and HRMS spectra

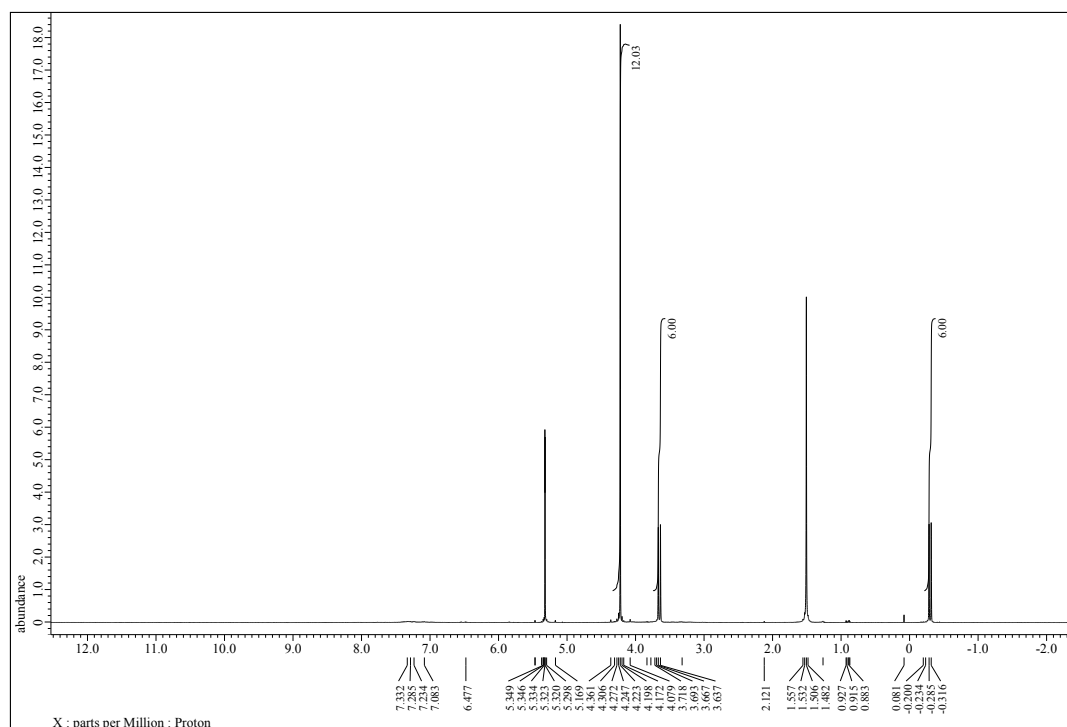


Figure S11. ^1H NMR spectrum of **2** (600 MHz, CD_2Cl_2). Reference: CDHCl_2 (5.32 ppm)

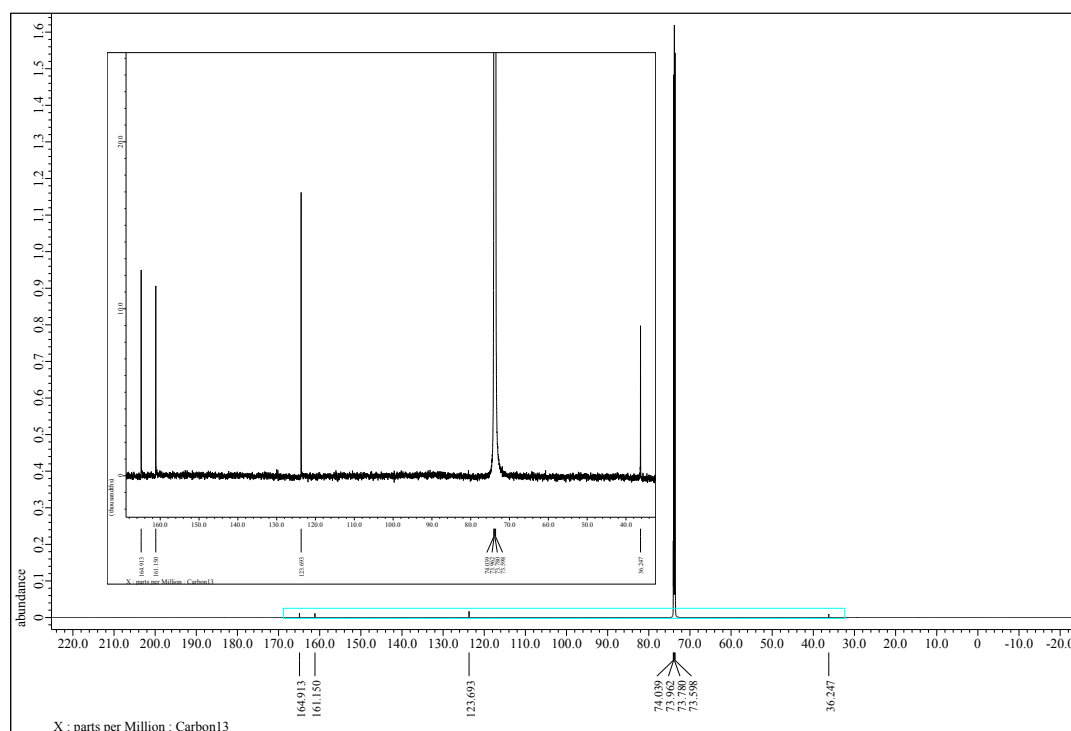


Figure S12. ^{13}C NMR spectrum of **2** (150 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$). Reference: $\text{C}_2\text{D}_2\text{Cl}_4$ (73.78 ppm).

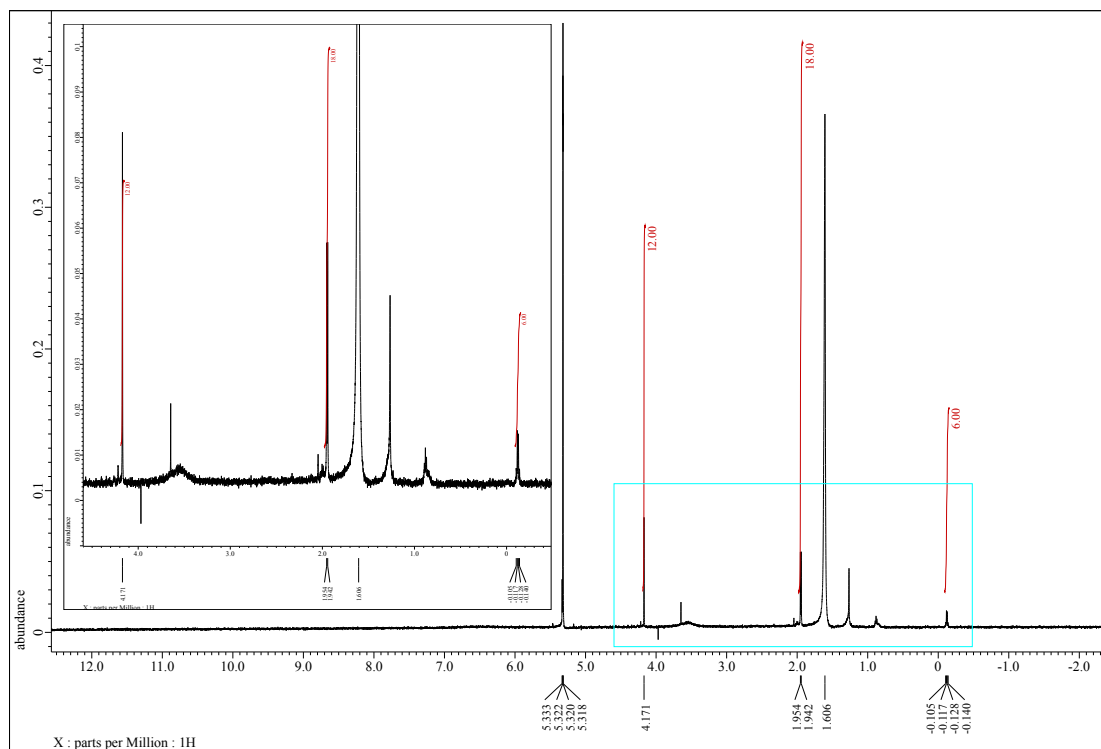


Figure S13. ^1H NMR spectrum of **3** (600 MHz, CD_2Cl_2). Reference: CDHCl_2 (5.32 ppm)

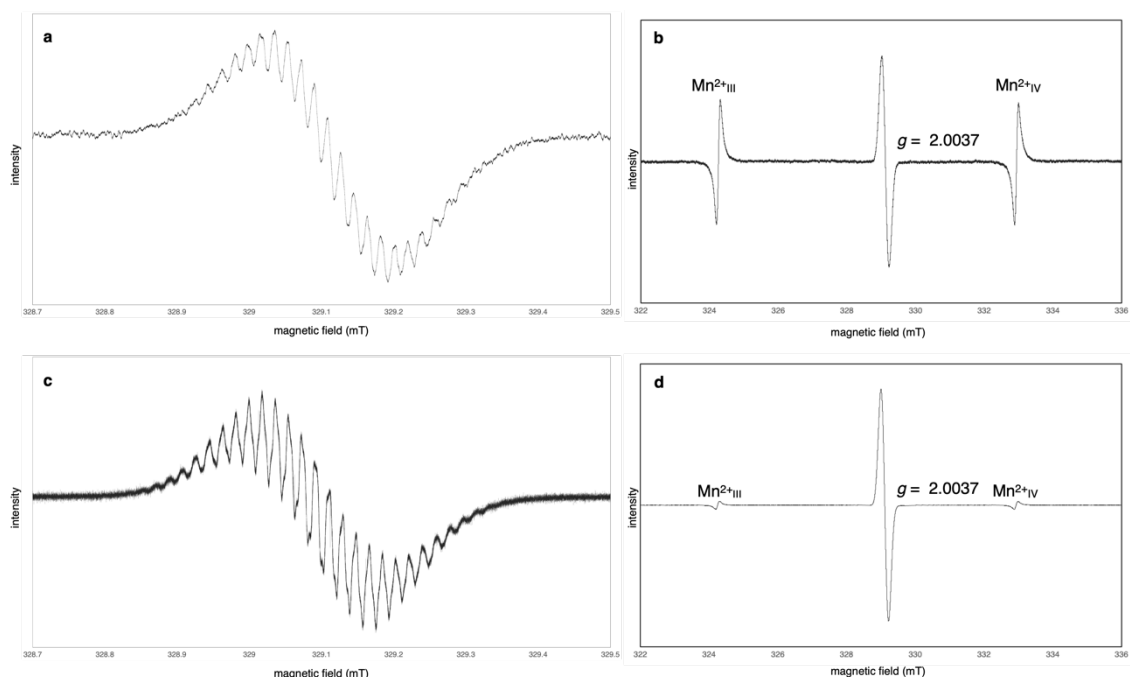
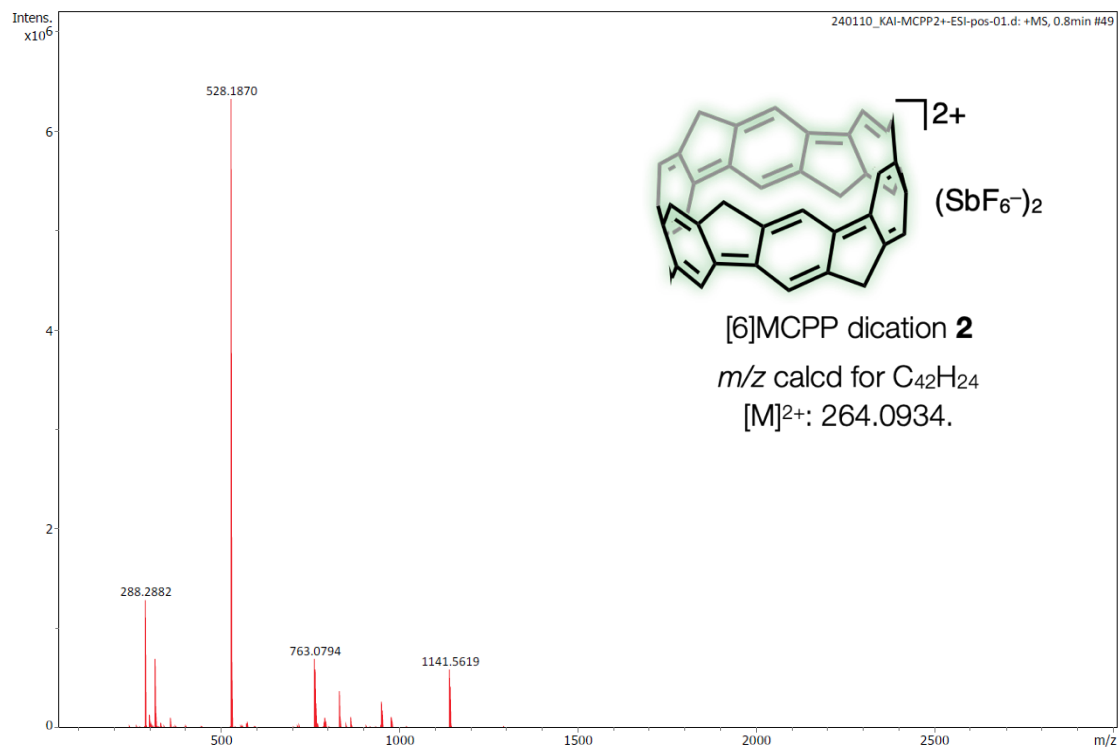


Figure S14. ESR spectra of $[\text{6}]\text{MCPH}^+\cdot\text{SbCl}_6^-$ (**1**) in (a) and (b), and $[\text{6}]\text{MCPH}^+\cdot\text{SbF}_6^-$ (**1'**) in (c) and (d). These samples were filtered using a syringe filter to prevent contamination by solids. Panels (a) and (b) were measured consecutively on the same $[\text{6}]\text{MCPH}^+\cdot\text{SbCl}_6^-$ sample, while (c) and (d) were measured consecutively on the same $[\text{6}]\text{MCPH}^+\cdot\text{SbF}_6^-$ sample. The spectrum (a) was integrated 30 times, whereas the spectra (b), (c), and (d) were not integrated. Additionally, (a) and (c) were not calibrated using standard substances. In contrast, the g -values (b) and (d) were determined using an external Mn^{2+} standard.

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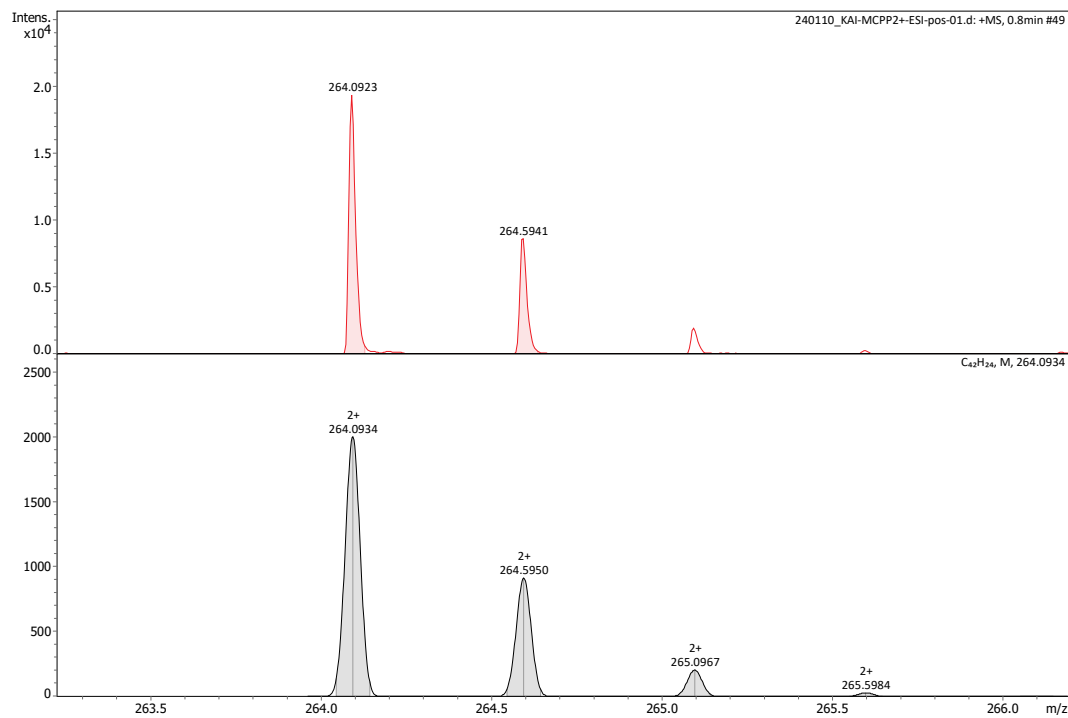
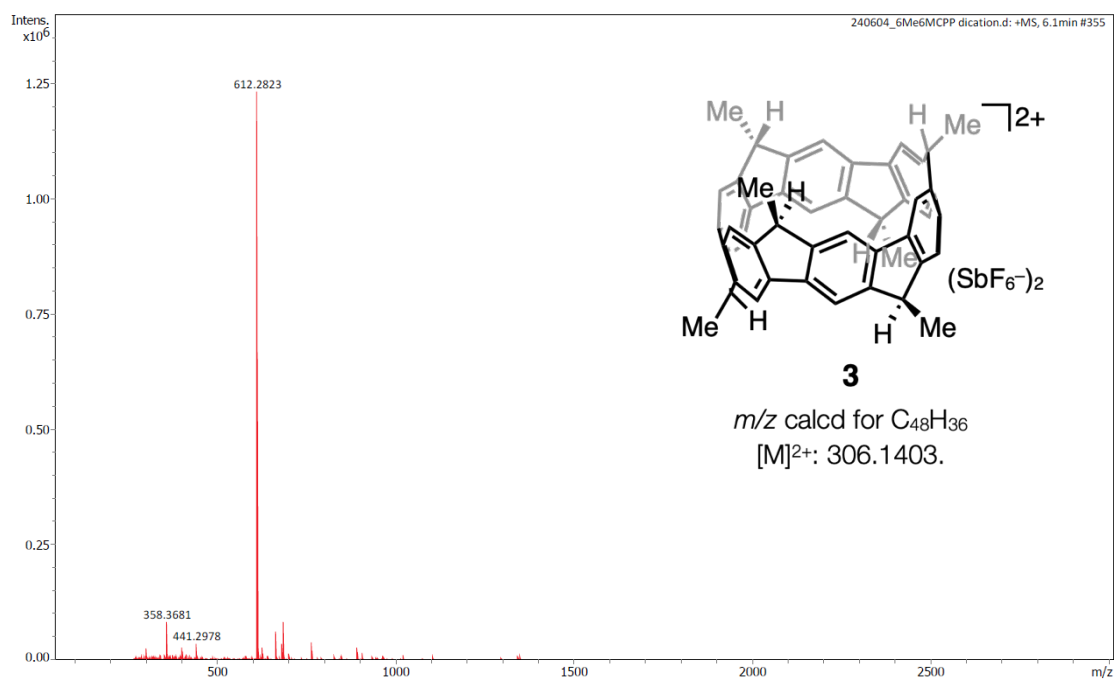


Figure S15. ESI mass spectrum of **2** (positive). Dissolving solvent: chloroform, no additives, type of detectors: Q/TOF, mass/charge ratios: +1, +2.

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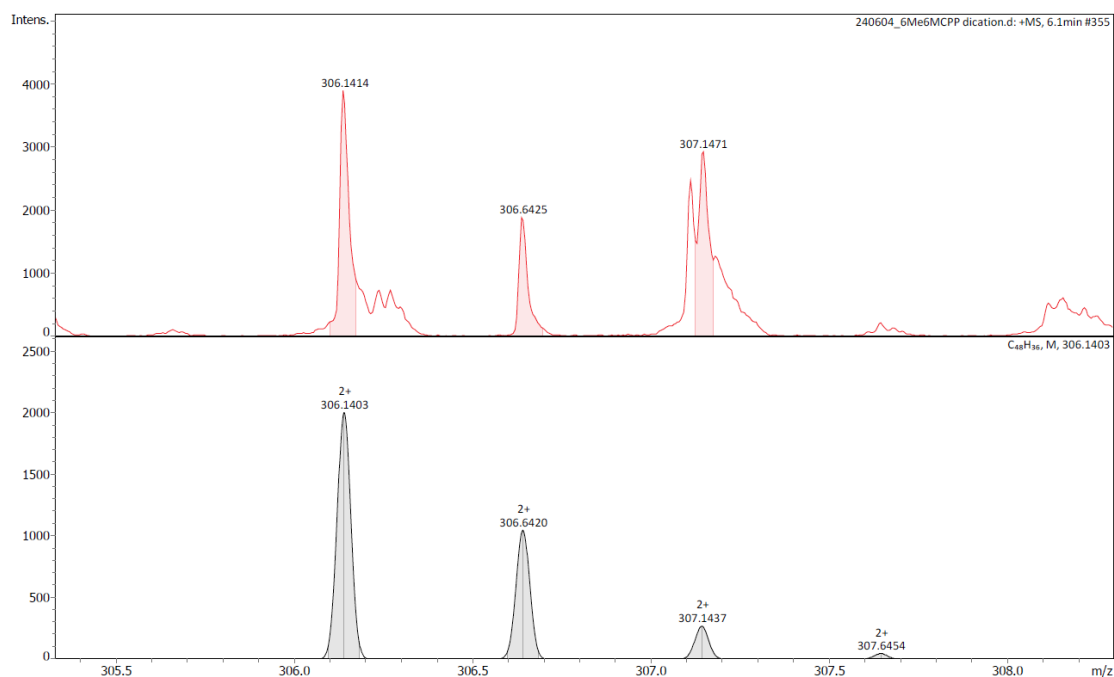


Figure S16. ESI mass spectrum of **3** (positive). Dissolving solvent: chloroform and methanol, no additives, type of detectors: Q/TOF, mass/charge ratios: +1, +2.

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