Electronic Supplementary Information – Experimental details

A Screening Approach Unveils an Unknown Mn²⁺-dependent Endopolyphosphatase Activity in Yeast

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Abbreviations:

BSA	bovine serium albumin			
CE	capillary electrophoresis			
Cy3	cyanine3			
Cy5	cyanine5			
DCI	4,5-dicyanoimidazole			
DMF	dimethylformamide			
DTT	dithiothreitol			
ESI	electrospray ionization or electronic supplementary information			
ETT	5-(ethylthio)-1 <i>H</i> -tetrazole			
HEPES	4-(2-hydroxyethyl)piperazine-1-ethanesulfonic acid			
HRMS	high resolution mass spectrometry			
<i>m</i> CPBA	meta-chloroperbenzoic acid			
MS	mass spectrometry			
m/z	mass-to-charge ratio			
NHS	N-hydroxysuccinimide			
NMR	nuclear magnetic resonance			
PMSF	phenylmethylsulfonyl fluoride			
PPi	pyrophosphate			
qTOF	quadrupole time-of-flight			
S. cerevisiae	Saccharomyces cerevisiae			
ТВА	tetrabutylammonium			
TFA	trifluoroacetic acid			
ТНРТА	tris[(1-hydroxy-propyl-1H-1,2,3-triazol-4-yl)methyl]amine			
TRIS	tris(hydroxymethyl)aminomethane			
UPLC	ultra-performance liquid chromatography			

1. Synthesis of FRET-polyP₈

1.1 General Methods and Materials

Reactions were performed in flame-dried glassware under inert gas atmosphere unless the solvent was water/buffer. Water was purified with a Milli-Q® lab water system. Reagents were purchased from commercial suppliers and were used without further purification. Sulfo-Cy3-NHS-ester and sulfo-Cy5-azide were purchased from Lumiprobe. Solvents were obtained in analytical grade and were used as received. Reaction control was done by ³¹P{¹H}-NMR.

Strong anion exchange chromatography was performed using an automated \Bar{A} KTA pureTM system and QSepharose® Fast Flow (Sigma-Aldrich). Crude products were loaded as aqueous solutions and eluted using increasing concentrations of either NH₄HCO₃ or NaClO₄ solutions.

Cation exchange for the preparation of TBA salts was performed with Dowex® 50WX8 H⁺, followed by neutralization with TBA hydroxide and subsequent lyophilization or with a Chelex® 100 column preconditioned with TBA(Br) (500 mM).

Lyophilization was performed using a Christ Alpha 1-4 LDplus.

Centrifugation was performed with an Eppendorf 5804R.

NMR-spectroscopy: The ¹H-, ¹³C-, ³¹P-NMR spectra were measured on a Bruker Avance Neo 400 MHz (101 MHz for ¹³C, 162 MHz for ³¹P) NMR spectrometer with broadband CryoProbe Prodigy and Bruker Avance Neo 700 MHz (176 MHz for ¹³C, 283 MHz for ³¹P) NMR spectrometer with broadband CryoProbe Prodigy. All signals were referred to an internal solvent signal (¹H-NMR: D₂O: δ = 4.79 ppm). The signals of ³¹P-NMR and ¹³C-NMR spectra were referenced to an external standard. The chemical shifts are quoted in ppm. The splitting patterns are labeled as: singlet (s), doublet (d), triplet (t), multiplet (m). The coupling constants *J* are given in Hertz (Hz). The evaluation of NMR-spectra was done using the software MestreNova from Mestrelab Research.

CE-ESI-MS experiments were performed on a bare-fused silica capillary, activated for 10 min with NaOH (1 M) before first measurement, with a length of 100 cm (50 μ m internal diameter and 365 μ m outer diameter) on an Agilent 7100 capillary

electrophoresis system coupled to a qTOF (6520, Agilent) equipped with a commercial CE-MS adapter and sprayer kit from Agilent. 35 mM ammonium acetate titrated by ammonia solution to pH 9.75 was background electrolyte. Samples were injected by applying 100 mbar pressure for 10 s, followed by an injection of a background electrolyte plug by applying 50 mbar for 5 s. For each analysis, a constant CE current of 23 μ A was established by applying 30 kV over the capillary. The sheath liquid was composed of a water-isopropanol (1:1) mixture spiked with mass references. It was introduced at a constant flowrate of 1.5 μ L/min. ESI-qTOF-MS was conducted in the negative ionization mode with published settings.¹ Automatic recalibration of each acquired spectrum was performed using reference masses of reference standards (TFA anion, [M-H]⁻, 112.9855), and (HP-0921, [M-H+CH₃COOH]⁻, 980.0163. Data were processed using the Agilent CE ChemStation Software.

UV/Vis absorption for concentration determination of FRET-polyP₈ (**6**) was measured on a Shimadzu UV-1900i UV-Vis spectrometer with quartz SUPRASIL® cuvettes (\emptyset = 10 mm). The extinction was measured at 548 nm and the concentration was calculated via Beer-Lambert law using ϵ (Cy3) = 162000 M⁻¹cm⁻¹ (Lumiprobe) (Table ESI-1).

Table ESI-1	Measured	extinction	of FRET	-polyP ₈ (6)	
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Compound	λ / nm ^[a]	A ^[b]	c _{sample} (µM)	dilution	с _{stock} (µМ)	solvent
FRET-P8	548	0.539	3.327	1:200	665	H ₂ O

[a] Wavelength of the absorption maximum of sulfo-Cy3. [b] Absorbance.

Absorption spectra and fluorescence spectra were recorded on a Tecan Spark® Plate Reader. Time course measurements were run according to the program in Table ESI-2.

Loop	Loop type	Duration [hh:mm:ss]	01:35:00
	Interval type	Fixed [hh:mm:ss]	00:05:00
	Fluorescence Intensity Scan	Excitation wavelength [nm]	500
		Emission wavelength [nm]	535-800
		Step size	5
	Shaking	Duration [s]	140
		Mode	Orbital
		Amplitude [mm]	1

Table ESI-2 Plate reader time course settings.

1.2 Synthesis of c-PyPA (1)



The compound was synthesized in two steps from pyrophosphate sodium salt as reported previously. Analytical data were identical to literature.² The compound was stored as stock solution (0.075 M in MeCN) at -20 °C.

1.3 Synthesis of propargylamido polyP₅ (3)



According to a reported procedure from J. Singh et al.³

PP_i × 2 TBA (0.075 M, 20 mL, 1.5 mmol, 1.0 eq.) was added to dry DCI (535 mg, 4.53 mmol, 3.0 eq., dried for one week in desiccator). The mixture was stirred for 5 min before it was cooled down to -20 °C and c-PyPA (0.075 M, 26 mL, 2.0 mmol, 1.3 eq.) was added. The reaction mixture was stirred for 15 min. Then *m*CPBA (70%, 555 mg,

2.25 mmol, 1.5 eq.) was added and the mixture was stirred at –20 °C. After 10 min the cooling bath was removed and the mixture was stirred for further 10 min. Subsequently, propargylamine (480 μ L, 413 mg, 7.50 mmol, 5.0 eq.) was added dropwise at 0 °C. After stirring for 60 min at r.t., the crude product was precipitated by cold NaClO₄ (0.5 M in acetone, 40 mL), centrifuged, washed with acetone (4×) and purified by strong anion exchange chromatography (Q Sepharose® Fast Flow, increasing concentrations of 1M NH₄HCO₃ in H₂O). Fractions eluted with 400-500 mM aqueous NH₄HCO₃ were combined and lyophilized. To remove buffer, the product was dissolved again in water and lyophilized (2×). The ammonium salt of **3** (411 mg, 738 μ mol, 49%) was obtained as a white solid.

Analytical data were identical to literature.³





ETT (18 mg, 0.14 mmol, 11 eq.) and propargylamido-polyP5 × 6 TBA (25 mg, 13 µmol, 1.0 eq.) were both separately coevaporated with dry MeCN (3 × 2 mL). Propargylamido-polyP5 × 6 TBA was dissolved in dry DMF (1 mL) and was added to a solution of ETT in dry DMF (0.5 mL). The mixture was stirred for 10 min before c-PyPA (0.075 M, 0.55 mL, 41 µmol, 3.2 eq.) was added. The reaction mixture was stirred at r.t. for 45 min. Then *m*CPBA (70%, 15 mg, 61 µmol, 4.7 eq.) was added at –15 °C. After 15 min the cooling bath was removed and the mixture was stirred for further 25 min. Subsequently, 1,3-diaminopropane (45 µL, 40 mg, 0.54 mmol, 40 eq.) was added dropwise at 0 °C. After stirring for 10 min at 0 °C the cooling bath was removed and the mixture was precipitated by NaClO₄ (0.5 M in acetone, 30 mL), centrifuged, washed with acetone (4×) and purified by strong anion exchange chromatography (Q Sepharose® Fast Flow, increasing concentrations of aqueous NH₄HCO₃ in H₂O). Product containing fractions were combined and lyophilized. To remove buffer, the product was dissolved again in water and lyophilized (2×) and purified again by strong anion exchange

chromatography (High Trap QFF, increasing concentration of aqueous NaClO₄solution (1 M) in H₂O). Fractions eluted with 10% aq. NaClO₄ were combined, lyophilized, precipitated with cold NaClO₄ acetone solution (0.5 M, 35 mL), washed with acetone (3 × 30 mL) and dried to afford **4** (1.2 mg, 1.3 µmol, 10%) as a white solid.

¹**H-NMR** (400 MHz, D₂O) δ = 3.75 (dd, *J* = 10.6, 2.5 Hz, 2H), 3.17 (t, *J* = 7.1 Hz, 2H), 3.10 (dt, *J* = 10.8, 6.6 Hz, 2H), 2.62 – 2.60 (m, 1H), 1.93 (tt, *J* = 6.9, 6.9 Hz, 2H) ppm. ³¹**P**{¹**H**}-**NMR** (162 MHz, D₂O) δ = -1.01 (d, *J* = 20.5 Hz, 1P), -2.28 (d, *J* = 20.0 Hz, 1P), -22.09 – -22.45 (m, 6P) ppm. ³¹**P-NMR** (162 MHz, D₂O) δ = -1.02 (dt, *J* = 21.0, 10.7 Hz, 1P), -2.28 (dt, *J* = 20.4, 10.5 Hz, 1P), -22.09 – -22.45 (m, 6P) ppm. ¹³**C-NMR** (101 MHz, CDCl₃) δ = 82.7, 71.5, 38.6, 37.4, 31.1, 28.4 (d, *J* = 7.0 Hz) ppm. **HRMS CE-ESI** calc. for C₆H₁₉N₃O₂₃P₆²⁻ [M-2H]²⁻: 374.4161, found: 374.4173.



Figure ESI-1 2D ¹H,³¹P-HMBC spectrum of **4** to assign the phosphor signals. Assignment of proton signals was performed with ¹H,¹H-COSY, ¹H,¹³C-HSQC and ¹H,¹³C-HMBC spectrum.

1.5 Synthesis of sulfo-cyanin3-polyP₈-alkyne (5)



4 (3.3 mg, 3.6 µmol, 1.0 eq.) was dissolved in 0.1 M NaHCO₃ buffer (450 µL) and was added to sulfo-cyanin3-NHS-ester (5.4 mg, 7.3 µmol, 2.0 eq.). The reaction mixture was stirred at r.t. for 3.5 h before it was diluted with water (20 mL) and directly injected to strong anion exchange chromatography system for purification (Q Sepharose® Fast Flow, increasing concentrations of 1 M NaClO₄ in H₂O). Fractions eluted with 12-28% aqueous NaClO₄ were combined and concentrated by lyophilization. Precipitation in cold NaClO₄ acetone solution (0.5 M, 35 mL), washing with cold acetone (3 × 15 mL) and drying afforded **5** (2.3 mg, 1.5 µmol, 42%) as a pink solid.

¹**H-NMR** (700 MHz, D₂O) δ = 8.62 (dd, *J* = 13.5, 13.5 Hz, 1H, H-β-Sulfo-Cy3), 7.95 (d, *J* = 1.7 Hz, 2H, 2x Ar-H), 7.91 (dd, *J* = 8.4, 1.8 Hz, 1H, Ar-H), 7.90 (dd, *J* = 8.4, 1.8 Hz, 1H, Ar-H), 7.45 (d, *J* = 7.9 Hz, 1H, Ar-H), 7.45 (d, *J* = 7.9 Hz, 1H, Ar-H), 6.46 (d, *J* = 13.9 Hz, 1H, H-α-Sulfo-Cy3), 6.44 (d, *J* = 13.9 Hz, 1H, H-α-Sulfo-Cy3), 4.18 (t, *J* = 7.5 Hz, 2H, Sulfo-Cy3-N⁺-CH₂), 3.76 (dd, *J* = 10.6, 2.5 Hz, 2H, P-NH-CH₂-C≡C), 3.70 (s, 3H, Sulfo-Cy3-N-CH₃), 3.39 (s, N-H), 3.39 (s, N-H), 3.27 (t, *J* = 6.8 Hz, 2H, CO-NH-CH₂), 2.99 (dt, *J* = 9.7, 7.0 Hz, 2H, P-NH-CH₂), 2.62 (dt, *J* = 3.0, 1.5 Hz, 1H, C≡C-H), 2.31 (t, *J* = 7.4 Hz, 2H, NH-CO-CH₂), 1.90 (tt, *J* = 7.7, 7.7 Hz, 2H, P-NH-CH₂-CH₂), 1.83 (s, 6H, 2x CH₃), 1.82 (s, 6H, 2x CH₃), 1.75 – 1.67 (m, 4H, 2x Sulfo-Cy3-CH₂), 1.51 – 1.45 (m, 2H, Sulfo-Cy3-CH₂) ppm. ³¹P{¹H}-NMR (283 MHz, D₂O) δ = -0.39 – -0.72 (m, 1P), -2.04 – 2.42 (m, 1P), -21.55 – -22.45 (m, 6P) ppm. ¹³C-NMR (176 MHz, D₂O) δ = 176.7, 176.6, 176.0, 152.2, 145.2, 144.5, 141.8, 141.7, 139.2, 139.2, 126.6, 119.8, 119.7, 111.5, 111.3, 103.6, 103.5, 83.3, 71.5, 49.4, 49.3, 44.2, 39.1, 37.0, 35.7, 31.3, 31.1, 30.4 (d, *J* = 8.2 Hz), 27.2, 27.0, 26.6, 25.7, 25.2 ppm. HRMS CE-ESI calc. for C₃₆H₅₃N₅O₃₀P₈S₂ [M-2H]²: 673.5064, found: 673.5064.

1.6 Synthesis of FRET-polyP₈ (6)



5 (2.3 mg, 1.49 µmol, 1.0 eq.) was dissolved in 200 mM triethylammonium acetate buffer (pH 7, 1.6 mL). Sulfo-cyanin-5-azide (1.73 mg, 2.27 µmol, 1.5 eq.) was dissolved in water (220 µL) and was added to the mixture. The solution was degassed with an argon stream for 5 min. Subsequently, 20 mM CuSO₄ • 5 H₂O solution (37 µL, 0.74 µmol, 0.5 eq.) and 50 mM THPTA solution (74 µL, 3.7 µmol, 2.5 eq.) were premixed and added. Then sodium ascorbate (3.0 mg, 15 µmol, 10 eg.) was added and the reaction mixture was stirred for 3.5 h at room temperature. The mixture was diluted with water (20 mL) and directly injected to strong anion exchange chromatography system for purification (Q Sepharose® Fast Flow, increasing concentrations of 1 M NaClO₄ in H₂O. Fractions eluted with 26-43% aqueous NaClO₄ were combined and concentrated by lyophilization. Precipitation in cold NaClO₄ acetone solution (0.5 M, 35 mL), washing with cold acetone (2 × 15 mL) and drying afforded a purple solid. The solid was dissolved in distilled H₂O (1 mL) and chelex (2.5 mg) was added. After stirring gently for 30 min, the mixture was filtered through a syringe filter. Addition of chelex, stirring and filtration were repeated twice before the sample was lyophilized. 6 (1.2 mg, 0.52 µmol, 35%) was obtained as a purple solid.

¹**H-NMR** (700 MHz, D₂O) δ = 8.46 (dd, J = 13.4, 13.4 Hz, 1H, **H**-β-Sulfo-Cy3), 8.12 – 8.05 (m, 1H, Ar-**H**), 7.97 (dd, J = 12.7, 12.7 Hz, 1H, **H**-β-Sulfo-Cy5), 7.95 (dd, J = 12.8, 12.8 Hz, 1H, **H**-β-Sulfo-Cy5), 7.89 – 7.81 (m, 7H, 7x Ar-**H**), 7.41 (d, J = 8.7 Hz, 1H, Ar-**H**), 7.40 (d, J = 8.3 Hz, 1H, Ar-**H**), 7.38 (d, J = 8,1 Hz, 1H, Ar-**H**), 7.36 (d, J = 8.3 Hz, 1H, Ar-**H**), 6.55 (dd, J = 12.4, 12.4 Hz, 1H, **H**-γ-Sulfo-Cy5), 6.30 (d, J = 13.5 Hz, 1H, **H**-α-Sulfo-Cy3/5), 6.28 (d, J = 13.3 Hz, 1H, **H**-α-Sulfo-Cy3/5), 6.25 (d, J = 13.6 Hz, 1H, **H**-α-Sulfo-Cy3/5), 6.22 (d, J = 13.8 Hz, 1H, **H**-α-Sulfo-Cy3/5), 4.42 (t, J = 6.9 Hz, N(triazole)-CH₂), 4.25 (d, J = 10.2 Hz, 2H, P-NH-CH₂-triazole-SulfoCy5), 4.10 (t, J = 7.2 Hz, 2H, Sulfo-Cy3/5-N⁺-CH₂), 4.00 (t, J = 7.8 Hz, 2H, Sulfo-Cy3/5-N⁺-CH₂), 3.62 (s, 3H, Sulfo-Cy3/5-N-CH₃), 3.60 (s, 3H, Sulfo-Cy3/5-N-CH₃), 3.39 (s, 2H, 2x N-H), 3.39 (s, 2H, 2x N-H), 3.28 (t, J = 6.7 Hz, 2H, CO-NH-CH₂), 3.16 (t, J = 6.6 Hz, 2H, CO-

NH-CH₂), 3.01 (dt, J = 10.0, 7.1 Hz, 2H, P-NH-CH₂-Linker-Sulfo-Cy3), 2.29 (t, J = 7.4 Hz, 2H, NH-CO-CH₂), 2.23 (t, J = 7.2 Hz, 2H, NH-CO-CH₂), 2.10 (tt, J = 6.8, 6.8 Hz, 2H, N-CH₂-CH₂), 1.87 – 1.30 (m, 38H, 7x CH₂, 8x CH₃) ppm. ³¹P{¹H}-NMR (283 MHz, D₂O) $\delta = -0.44 - -0.71$ (m, 1P), -1.66 – 1.97 (m, 1P), -21.51 – -22.48 (m, 6P) ppm. HRMS CE-ESI calc. for C₇₁H₉₆N₁₁O₃₇P₈S₄ [M-3H]³⁻: 690.0923, found: 690.0917.



Figure ESI-2 2D ¹H,³¹P-HMBC spectrum of 6.

2. UPLC-MS Analysis

UPLC-MS analysis was performed on a UHPLC (Agilent 1290 infinity II) coupled to a qTOF (Agilent 6546). Chromatographic separation was achieved using gradient elution on a C18 column (AdvanceBio peptide plus 2.1mm x 150 mm) at 20 °C. Mobile phase consisted of 10 mM ammonium formate (pH 8.5) (eluent A) and 100% acetonitrile (eluent B). Chromatograms of **5** and **6** are presented in Figure ESI-3 and ESI-4, respectively.



Figure ESI-3 UPLC-MS-chromatogram of **5**. Run commenced with a linear gradient from 2% B to 40% B in 8 min, followed by a linear gradient to 80% B in 2 min with a flow rate of 0.5 mL/min. **a**, 272 nm UV detection. **b**, Extracted ion chromatogram (EIC) of **5** (m/z=673.5064 with z=2).



Figure ESI-4 UPLC-MS-chromatogram of **6**. Run conducted with a linear gradient from 10% B to 40% B in 10 min with a flow rate of 0.4 mL/min. **a**, 272 nm UV detection. **b**, Extracted ion chromatogram of **6** (m/z=690.0923 with z=3).

3. Isolation of Yeast Cytosolic Extract

The generation of the yeast strains $ppx1\Delta ppn1\Delta ppn2\Delta$, $ppx1\Delta ppn1\Delta ppn2\Delta vtc4\Delta$, and $ppx1\Delta ppn1\Delta ppn2\Delta ddp1\Delta$, were previously described.^{4,5} Yeast cytosolic extract isolation protocol was adapted from Schmidt et al. (2010).⁶ BY4742 yeast strains were grown in YPD medium (1% (w/v) yeast extract, 2% (w/v) bacto peptone, 2% (w/v) glucose) overnight at 30 °C and 130 rpm to an OD₆₀₀ of 1. 100 mL yeast cells were harvested by centrifugation and washed with distilled water and pelleted again by centrifugation. Cell pellets were treated with DTT and 0.15 mg/mL zymolyase 20T to generate spheroblasts. Spheroblasts were washed with 1.2 M sorbitol buffer and pelleted by centrifugation. Spheroblast pellets were lysed in 1 mL lysis buffer (20 mM Tris/HCI, pH 7.2, 30 mM NaCl, 10 mM MgCl₂, 10% glycerol, 1 mg/mL BSA, 2 mM PMSF). The steps following spheroblast lysis were performed on ice or in the cold room. 1 mL of sterile glass beads (0.25-0.5 mm diameter) were added onto samples and cells were opened by FastPrep-24 5G homogenizer at 4 m/sec for 20 s. The cell homogenate was centrifuged for 10 min at 2,500 × g and 4 °C and the supernatant was collected into a pre-cooled tube. To remove the non-cytosolic cellular fractions, the supernatant was further centrifuged for 15 min at 16,000 × g and 4 °C. The supernatant containing yeast cytosolic extracts was collected and the protein concentration was determined by Bradford Assay.⁷ The cytosolic extracts were snap frozen in liquid nitrogen and stored at -80 °C.

4. FRET Endopolyphosphatase Activity Assay

4.1 With Isolated Proteins

A 3 μ M working concentration was prepared from a 665 μ M stock solution of FRETpolyP8 (**6**), whose concentration was determined by UV/Vis (Table ESI-1). **6** (100 nM end concentration) in the corresponding 1× reaction buffer (Table ESI-3) was supplemented with the corresponding enzyme (DDP1, PPX, NUDTs: 2 μ g, PPase: 16 mU). The total reaction volume was 60 μ L. The samples were either incubated in a thermoshaker (30 min, 37 °C, 450 rpm) before being transferred to a 96-well plate for fluorescence spectrum measurement, or directly incubated in a plate reader (90 min, 37 °C, orbital shaking) to monitor fluorescence over time, with measurements taken every 5 min. Enzymes (GST-PPX from E. coli, His-DDP1 from *S. cerevisiae* and human GST-NUDT3, GST-NUDT3E70A, GST-NUDT10, GST-NUDT11 were heterologously expressed and purified.^{8,9} Inorganic pyrophosphatase (PPase; EC 3.6.1.1) from baker's yeast was purchased from Sigma Aldrich.

enzyme	concentration	buffer
DDP1	4.4	
PPX	X	25 mm HEPES, 50 mm NaCi, 10 mm MgSO4, pH 6.8
NUDT3		
NUDT3	10×	without ion: 200 mM Tris pH 7.5, 1 M NH₄Ac
	- 10×	Mg ²⁺ : 200 mM Tris pH 7.5, 1 M NH₄Ac, 50 mM MgCl ₂
	- 10×	Zn ²⁺ : 200 mM Tris pH 7.5, 1 M NH ₄ Ac, 50 mM ZnSO ₄ x 7 H ₂ O
NUDT11		
PPase	1×	20 mM Tris pH 7.4, 30 mM NaCl, 10 mM MgCl ₂ , 10%
		glycerol, 2 mM PMSF

 Table ESI-3 Composition of buffer solutions for FRET endopolyphosphatase activity assay.



Figure ESI-5 Monitoring of the FRET ratio changes over 90 min during incubation of **6** (100 nm) with PPase in the absence or presence of $MnSO_4$. No Mn^{2+} -dependent endopolyphosphatase activity was observed for PPase.

To analyze the reaction products of **6** treated with PPX or DDP1, **6** (10 μ M end concentration) in the corresponding 0.1× reaction buffer (Table ESI-3) was supplemented with 8 μ g of either PPX or DDP1. The mixtures were incubated in a thermoshaker (30 min, 37 °C, 450 rpm) until the DDP1 reaction was complete (as

monitored by the plate reader assay). After incubation, 100 μ L of the reaction mixture were loaded on a 30% polyacrylamide gel (Figure ESI-5c), while the remaining 900 μ L were lyophilized, redissolved in 50 μ L water and analyzed by CE-MS (Figure ESI-5a,b).



Figure ESI-6 Analysis of the reaction products of treatment of **6** with PPX or DDP1. **a**, CE-MS analysis of **6** after treatment with PPX, showing no digestion by PPX. **b**, CE-MS analysis of **6** after treatment with DDP1, showing cleavage into a mixture of Cy3-polyP₂₋₅ and Cy5-polyP₂₋₄. **c**, PAGE analysis confirming no cleavage by PPX while DDP1 digestion produces Cy3 and Cy5 polyP fragments. The products are visually identifiable by color: Cy3 labeled polyPs are pink, Cy5 labeled polyPs blue and their combination in the FRET-polyP₈ (**6**) forms a purple band.

4.2 With Yeast Extracts

The reaction mixture was prepared by sequentially adding lysis buffer (20 mM Tris pH 7.4, 30 mM NaCl, 10 mM MgCl₂, 10% glycerol, 2 mM PMSF), polyP₁₀₀ (if applicable, from Kerafast), MnSO₄ (if applicable), **6** and yeast extract (1 mg) in a 96-well plate. The final concentrations in the 60 μ L reaction volume were 100 nM **6**, 100 mM polyP₁₀₀ and 1, 5 or 10 mM MnSO₄. Fluorescence specta were recorded at 5 min intervals over a 90 min period at 37 °C (Table ESI-2).



Figure ESI-7 Monitoring of the FRET ratio changes over 90 min during incubation of **6** (100 nm) with the yeast knockout strain $ppx1\Delta ppn1\Delta ppn2\Delta vtc4\Delta$ without or in the presence of polyP₁₀₀ (100 mM). In the absence of both endogenous polyP and added polyP₁₀₀, DDP1 cleaves FRET-polyP₈ **6**. When polyP₁₀₀ is present, DDP1 preferentially digests polyP₁₀₀ instead.



Figure ESI-8 Monitoring of the FRET ratio changes over 90 min during incubation of **6** (100 nm) with the yeast knockout strain $ppx1\Delta ppn2\Delta ddp1\Delta$ in the presence of different concentrations of MnSO₄. A concentration of 1 mM MnSO₄ is insufficient to activate the Mn²⁺-dependent endopolyphosphatase activity. A modest increase in activity is observed at 5 mM MnSO₄, while the most pronounced effect occurs at 10 mM MnSO₄ (Figure 3b).

5. FRET Inhibition Assay of DDP1

A solution of FRET-polyP8 (**6**) at a final concentration of 100 nM in 1× reaction buffer (25 mM HEPES, 50 mM NaCl, 10 mM MgSO₄, pH 6.8) was first supplemented with the corresponding inhibitor solution followed by the addition of 2 μ g of DDP1 or no enzyme (serving as decomposition control). The samples were incubated in a thermoshaker (30 min, 37 °C, 450 rpm) and then transferred to a 96-well plate for fluorescence spectrum analysis.

5.1 Stability Against Different Potential DDP1 Inhibitors



Figure ESI-9 Normalized fluorescence intensity spectra of **6** (100 nM) in 1× reaction buffer (25 mM HEPES, 50 mM NaCl, 10 mM MgSO₄, pH 6.8) after incubation with different salt concentrations of different potential DDP1 inhibitors at 37 °C for 30 min (heparin sodium salt: 0-1,000 nM (calculated with average molecular weight of 17 kDa), FeSO₄: 0-10 mM, others: 0-100 mM). Excitation at λ = 500 nm.

5.2. DDP1 Inhibition Screening



Figure ESI-10 Normalized fluorescence intensity spectra of **6** (100 nM) in 1× reaction buffer (25 mM HEPES, 50 mM NaCl, 10 mM MgSO₄, pH 6.8) after incubation with DDP1 (2 μ g) at 37 °C for 30 min in the presence of different potential inhibitors, tested across different concentrations. Excitation at λ = 500 nm.

6. PAGE Endopolyphosphatase Activity Assay in Yeast Extracts

The generation of the yeast strain $ppx1\Delta ppn1\Delta ppn2\Delta ddp1\Delta$ was previously described.^[4] To prepare protein extracts logarithmic growing yeast culture (20-50U of OD₆₀₀) were harvested by centrifugation at 5,000 × g for 3 min and washed once with ice-cold Milli-Q water. Proteins were extracted in ice-cold lysis buffer (50 mM Tris-HCl, pH 7.4, 100 mM NaCl with fresh added 5 mM DTT, and protease inhibitor mixture (SigmaAldrich, P8215)) by vortexing in the presence of acid washed glass beads for 5 min at 4 °C. The homogenates were centrifuged 15,000 × g for 5 min at 4 °C and the supernatants (protein extracts) were transferred to another tube and used immediately. Protein extracts (100 µg) were incubated in reaction buffer (25 mM Tris-HCl, pH 7.4, 100 mM NaCl, 1 mM DTT) alone or supplemented with 10 mM MgCl or 10 mM MnCl for 60 min at 37 °C. After the incubation, the reactions were resolved on a 30% polyacrylamide gel and polyP visualized by toluidine blue staining as previously described.¹⁰

7. Supporting Literature

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31P{1H}-NMR (162 MHz, D2O)



31P-NMR (162 MHz, D2O)







Residual amounts of acetone are marked with asterisks (*).





Residual amounts of acetone are marked with asterisks (*).





Residual amounts of acetone are marked with asterisks (*).







Residual amounts of acetone are marked with asterisks (*).



