

Supporting Information

What Distinguishes the Strength and the Effect of a Lewis Base: Insights with a Strong Chromogenic Silicon Lewis Acid

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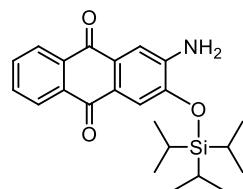
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1 Synthesis and Characterisation Data

1.1 General Considerations

All manipulations were carried out under air- and moisture-free conditions using standard Schlenk techniques under argon or nitrogen atmosphere. Sensitive compounds were stored in a nitrogen-filled *Sylatech* glovebox at -40 °C. All solvents were dried by standard procedures, stored over 3 Å or 4 Å molecular sieves and freshly degassed in three freeze-pump-thaw cycles prior to use. Commercially available starting materials were used as received. Identity and purity of newly synthesised compounds were confirmed by multinuclear NMR spectroscopy, as well as IR spectroscopy, mass spectrometry, and SCXRD if possible. ¹H, ¹³C, ¹⁹F, ³¹P and ²⁹Si NMR spectra were collected either by the NMR department of the Institute of Inorganic Chemistry of the University of Heidelberg on either a *Bruker Avance II 400* or *Bruker Avance III 600* spectrometer, or in house on a *Bruker Ascend 400* spectrometer at 298K. ¹³C and ²⁹Si spectra were recorded with broadband ¹H decoupling. NMR shifts are given as δ in ppm and referenced against solvent residual signals for ¹H and ¹³C spectra. Coupling constants J are given in Hz. NMR spectra were analysed and plotted using *MestReNova 14.3*. Electrospray ionisation mass spectra were recorded on a *Bruker ApexQe hybrid 9.4 T FT-ICR* spectrometer and LIFDI mass spectra were recorded on a *JEOL AccuTOF GCx* spectrometer by the mass spectrometry department of the Institute of Organic Chemistry at the University of Heidelberg. ATR-IR spectra were measured on an *Agilent Cary 630* spectrometer inside a nitrogen-filled glovebox. UV-vis spectra were measured on a JASCO V-570 UV/VIS/NIR spectrophotometer in a 2 mm quartz cuvette equipped with a J-Young valve.

1.2 2-Amino-3-(triisopropylsilyloxy)anthraquinone



2-Amino-3-hydroxyanthraquinone (3.00 g, 12.5 mmol, 1.00 eq), imidazole (1.28 g, 18.8 mmol, 1.50 eq) and triisopropylsilyl chloride (3.20 ml, 15.1 mmol, 1.20 eq) were stirred in DMF (20 ml) at room temperature for 1 h. Water (100 ml) was added, and the resulting suspension filtered. The solid was washed with water (4x50 ml), dried *in vacuo*, washed with *n*-pentane (3x20 ml) and dried again *in vacuo*. The product was isolated as a red powder (4.53 g, 12.3 mmol, 98%).

¹H NMR (600 MHz, CDCl₃) δ 8.26 – 8.18 (m, 2H), 7.74 – 7.68 (m, 2H), 7.61 (s, 1H), 7.50 (s, 1H), 4.54 (s, 2H), 1.43 (sept, *J* = 7.5 Hz, 3H), 1.16 (d, *J* = 7.5 Hz, 18H).

¹³C NMR (151 MHz, CDCl₃) δ 183.2, 182.2, 147.5, 144.2, 134.2, 133.8, 133.7, 133.4, 129.4, 126.9, 126.9, 125.7, 115.2, 111.6, 18.1, 12.9.

²⁹Si NMR (119 MHz, CDCl₃) δ 20.3.

HRMS (EI): [C₂₃H₂₉NO₃Si]⁺, calcd.: 395.1911, found: 395.1895.

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3481 (m, $\tilde{\nu}_{\text{NH}_2}$), 3356 (m, $\tilde{\nu}_{\text{NH}_2}$), 2947 (m, $\tilde{\nu}_{\text{CH}}$), 2867 (m, $\tilde{\nu}_{\text{CH}}$), 1665 (m, $\tilde{\nu}_{\text{C=O}}$), 1646 (m, $\tilde{\nu}_{\text{C=O}}$).

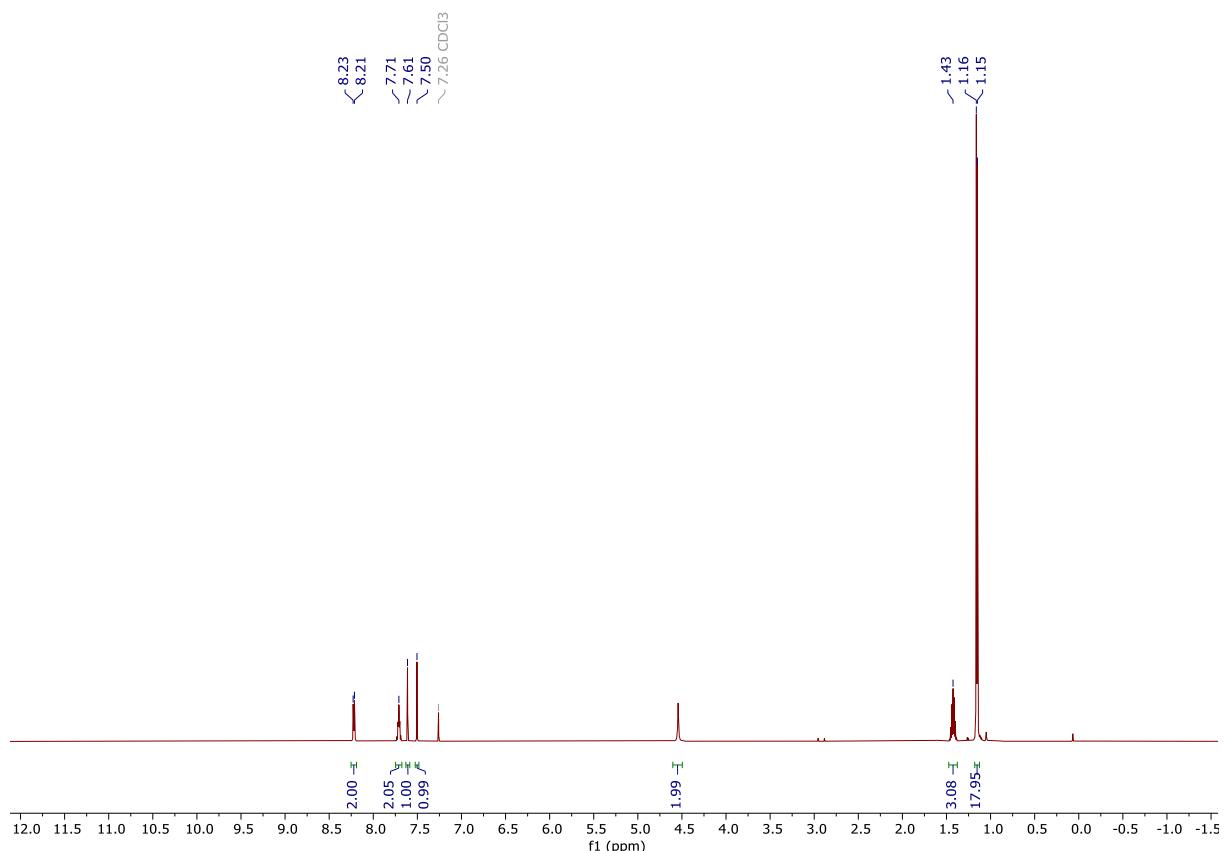


Figure S1.1. ¹H NMR spectrum (600 MHz, CDCl₃) of 2-amino-3-(triisopropylsilyloxy)anthraquinone.

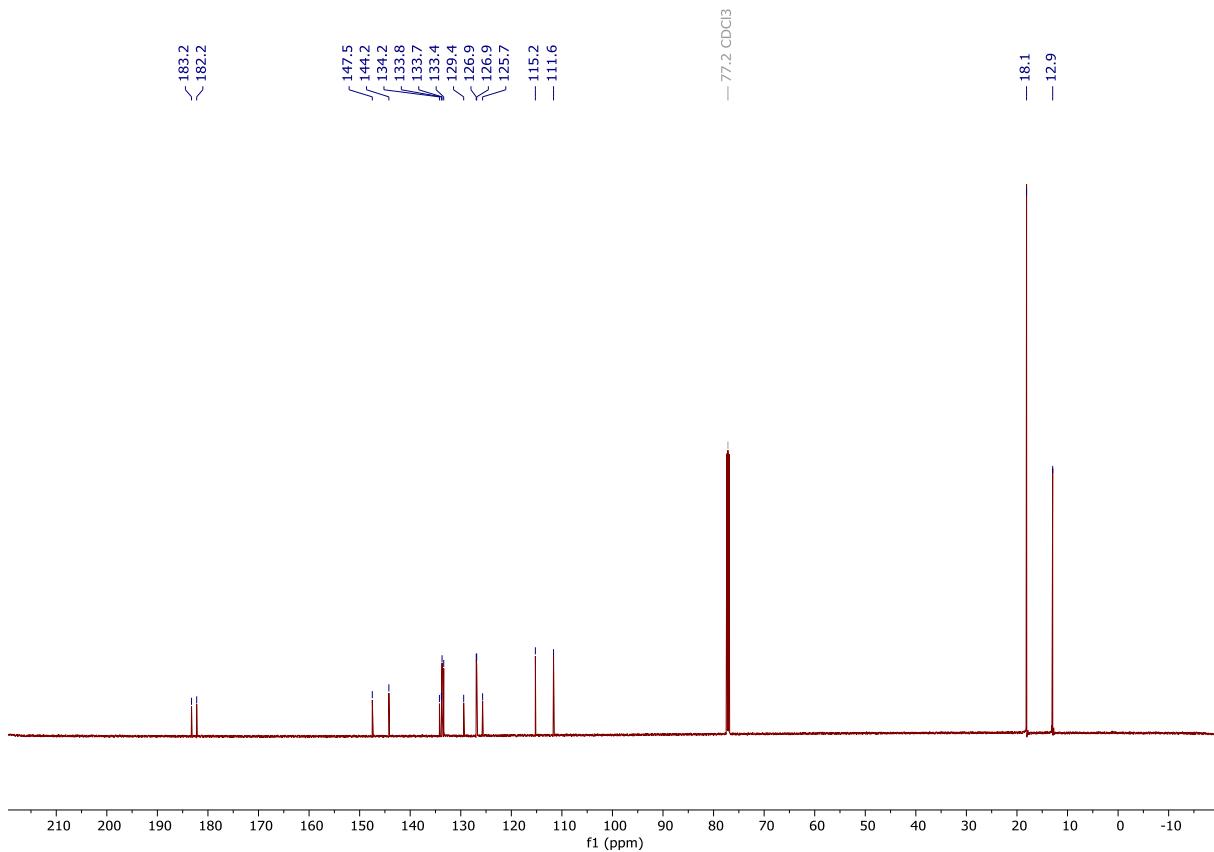


Figure S1.2. ¹³C NMR spectrum (151 MHz, CDCl₃) of 2-amino-3-(triisopropylsilyloxy)anthraquinone.

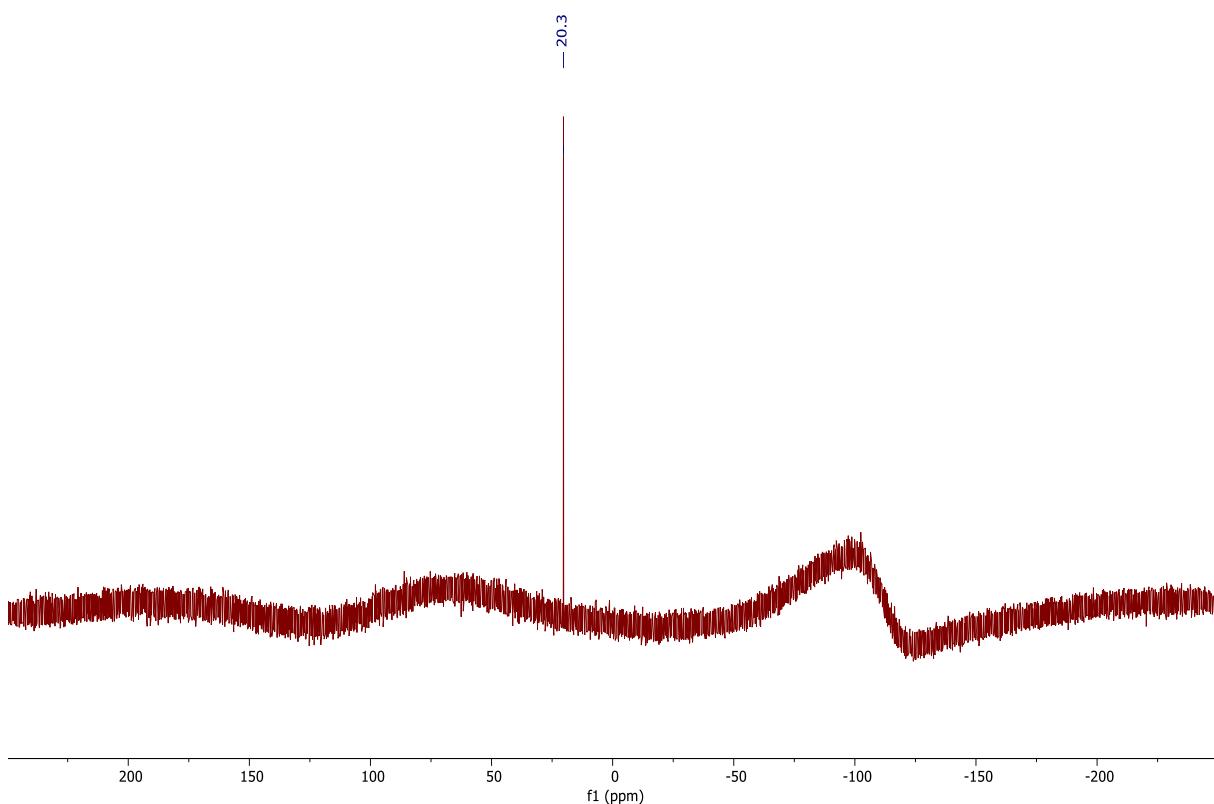
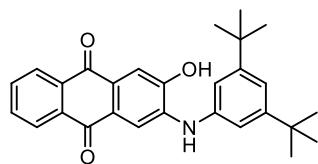


Figure S1.3. ²⁹Si NMR spectrum (119 MHz, CDCl₃) of 2-amino-3-(triisopropylsilyloxy)anthraquinone.

1.3 2-((3,5-Di-*tert*-butylphenyl)amino)-3-hydroxyanthraquinone (L₁**)**



2-Amino-3-(triisopropylsilyloxy)anthraquinone (5.78 g, 14.6 mmol, 1.00 eq), 1-bromo-3,5-di-*tert*-butylbenzene (4.13 g, 15.3 mmol, 1.05 eq), NaOtBu (3.51 g, 36.5 mmol, 2.50 eq), Pd₂(dba)₃ (268 mg, 0.29 mmol, 0.02 eq) and XPhos (348 mg, 0.35 mmol, 0.05 eq) were suspended in toluene (100 ml) and heated to 75 °C for 12 h. Full conversion of the starting material was ensured by TLC monitoring. The mixture was then heated to 110 °C for 16h to complete the silyl ether deprotection. The mixture was added to a separatory funnel containing ethyl acetate (200 ml). The organic phase was washed with 2M NaOH (3x200 ml), then with sat. NH₄Cl (200 ml), resulting in the precipitation of a red solid. THF was added until the solid dissolved (about 300 ml) and the phases were separated. The organic phase was washed with sat. NaCl (200 ml), dried over MgSO₄ and filtered. The solvent was removed *in vacuo*, the resulting solid suspended in *n*-hexane and filtered. The solid was washed with acetone (2x10 ml) and DCM (5x30 ml), then dried *in vacuo* overnight. The product was isolated as a red solid (5.67 g, 13.3 mmol, 91%).

¹H NMR (600 MHz, DMSO-d₆) δ 11.28 (s, 1H), 8.12 (s, 1H), 8.11 – 8.07 (m, 2H), 7.85 (s, 1H), 7.83 – 7.77 (m, 2H), 7.55 (s, 1H), 7.21 (s, 2H), 7.11 (s, 1H), 1.30 (s, 18H).

¹³C NMR (151 MHz, DMSO-d₆) δ 182.1, 181.0, 151.2, 150.3, 139.7, 139.3, 133.9, 133.6, 133.6, 133.1, 127.1, 126.4, 126.2, 124.9, 116.6, 115.9, 110.9, 108.0, 34.7, 31.3.

HRMS (EI): [C₂₈H₂₉NO₃]⁺, calcd.: 427.2142, found: 427.2113.

UV-vis (THF): λ_{max} (ε) = 479 nm (7200 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3413 (w, $\tilde{\nu}_{\text{NH}/\text{OH}}$), 3329 (m, $\tilde{\nu}_{\text{NH}/\text{OH}}$), 2954 (m, $\tilde{\nu}_{\text{CH}}$), 2863 (w, $\tilde{\nu}_{\text{CH}}$), 1658 (s, $\tilde{\nu}_{\text{C=O}}$), 1647 (m, $\tilde{\nu}_{\text{C=O}}$).

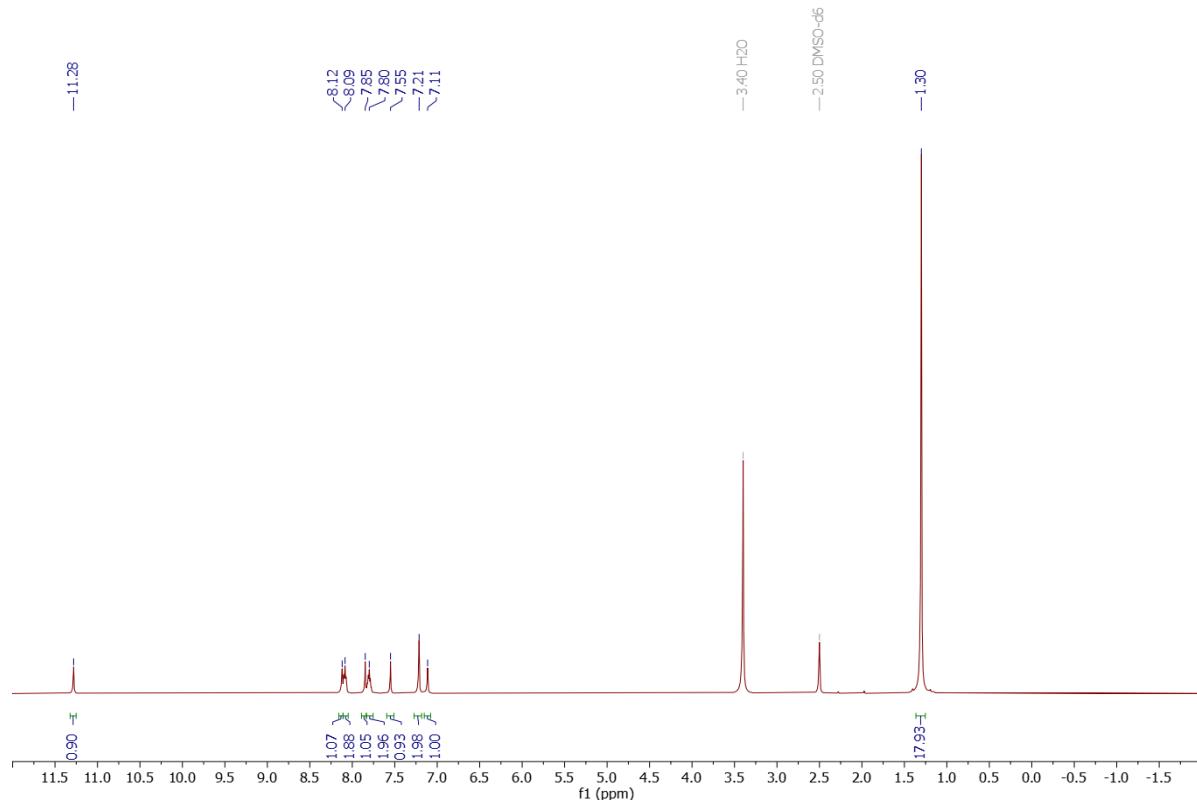


Figure S1.4. ¹H NMR spectrum (600 MHz, DMSO-d₆) of 2-((3,5-Di-*tert*-butylphenyl)amino)-3-hydroxyanthraquinone.

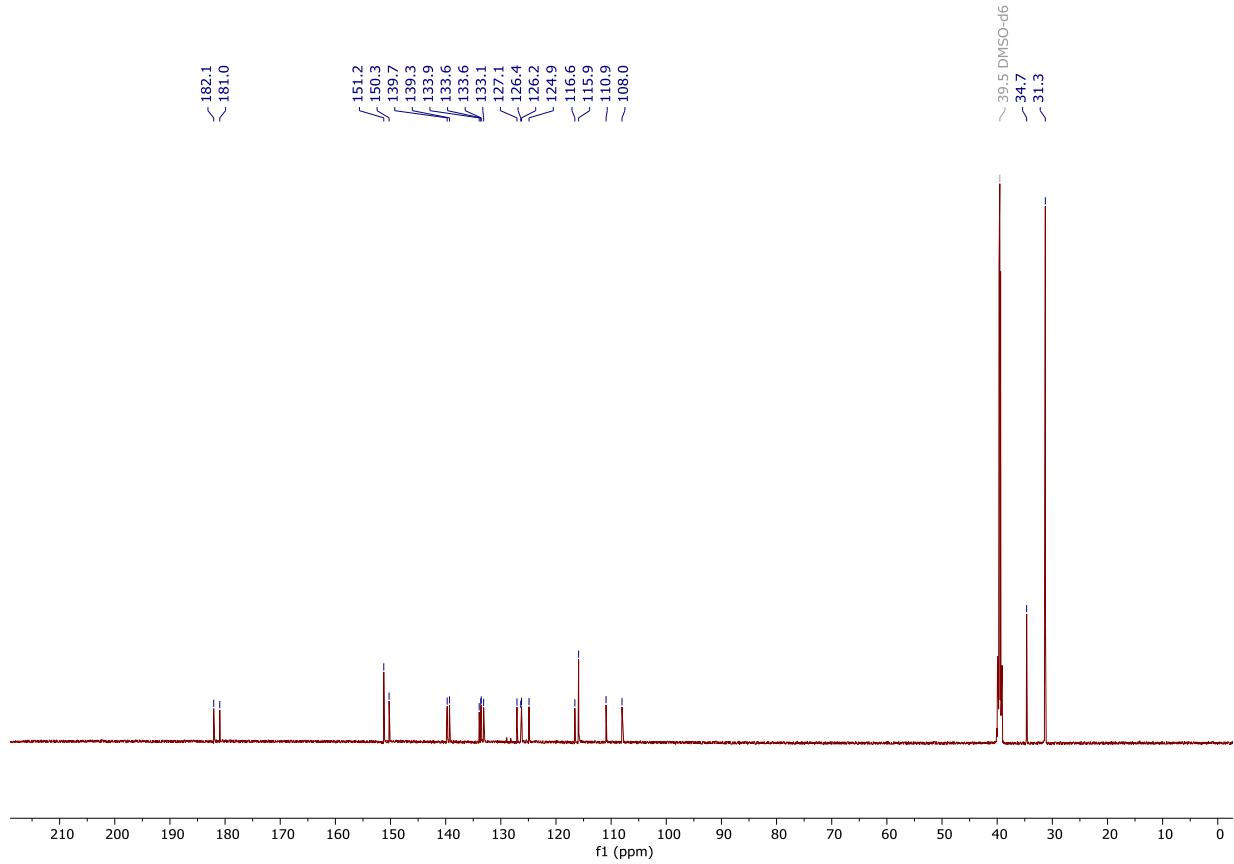
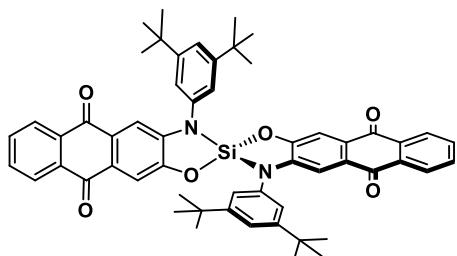


Figure S1.5. ^{13}C NMR spectrum (151 MHz, DMSO-d₆) of 2-((3,5-di-tert-butylphenyl)amino)-3-hydroxyanthraquinone.



2-((3,5-Di-*tert*-butylphenyl)amino)-3-hydroxyanthraquinone (5.22 g, 12.2 mmol, 2.00 eq), triethylamine (3.40 ml, 42.4 mmol, 4.00 eq) and silicon tetrachloride (0.77 ml, 6.7 mmol, 1.10 eq) were heated in toluene (100 ml) to 100 °C for 16 h. After cooling to room temperature, the mixture was filtered, and the filtrate concentrated *in vacuo*. The resulting dark yellow solid was suspended in *n*-hexane, filtered, and washed with *n*-hexane (3x10 ml). The solid was redissolved in DCM (approx. 100 ml) and the volume reduced to one third, accompanied by cooling. The product precipitated as a yellow microcrystalline solid, which was filtered off, washed with DCM (3x5 ml), and dried *in vacuo* overnight (2.74 g, 3.12 mmol, 51%). Single crystals suitable for SCXRD were grown by gaseous diffusion of *n*-pentane into a saturated DCM solution at room temperature.

The combined filtrates from the previous steps were stirred under ambient air for 2 days, followed by removal of the solvent and washing with DCM to recover the aminophenol ligand (1.98 g, 4.63 mmol, 38%), giving a yield of 89% based on recovered starting material.

¹H NMR (600 MHz, CDCl₃) δ 8.31 – 8.27 (m, 2H), 8.25 – 8.21 (m, 2H), 8.00 (s, 2H), 7.85 (s, 2H), 7.79 – 7.71 (m, 4H), 7.31 (s, 2H), 7.04 (s, 4H), 1.22 (s, 36H).

¹³C NMR (151 MHz, CDCl₃) δ 182.7, 182.2, 153.4, 149.3, 143.5, 136.1, 134.0, 133.8, 133.8, 133.6, 130.4, 128.4, 127.2, 127.2, 121.2, 118.4, 112.0, 109.0, 35.1, 31.5.

²⁹Si NMR (¹H-²⁹Si-HMBC) (119 MHz, CDCl₃) δ -39.0.

MS (LIFDI+): [C₅₆H₅₄N₂O₆Si]⁺, calcd.: 878.38, found: 878.45.

EA Anal. Calcd. For C₅₆H₅₄N₂O₆Si: C, 76.51; H, 6.19; N, 3.19; found C, 75.85; H, 6.08; N, 3.62.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon)$ = 407 nm (8200 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3070 (w, $\tilde{\nu}_{\text{CH}}$), 2963 (m, $\tilde{\nu}_{\text{CH}}$), 2869 (w, $\tilde{\nu}_{\text{CH}}$), 1675 (s, $\tilde{\nu}_{\text{C=O}}$).

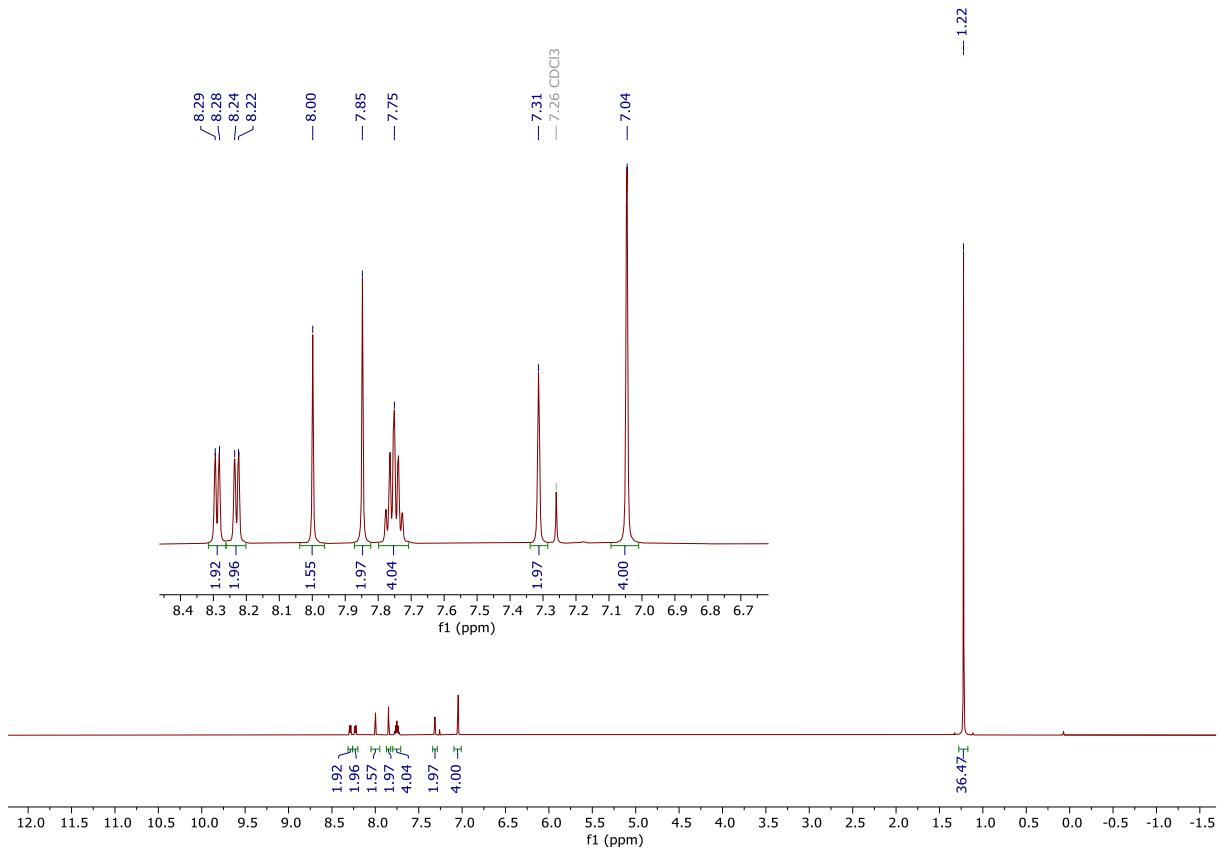


Figure S1.6. ^1H NMR spectrum (600 MHz, CDCl_3) of **1**.

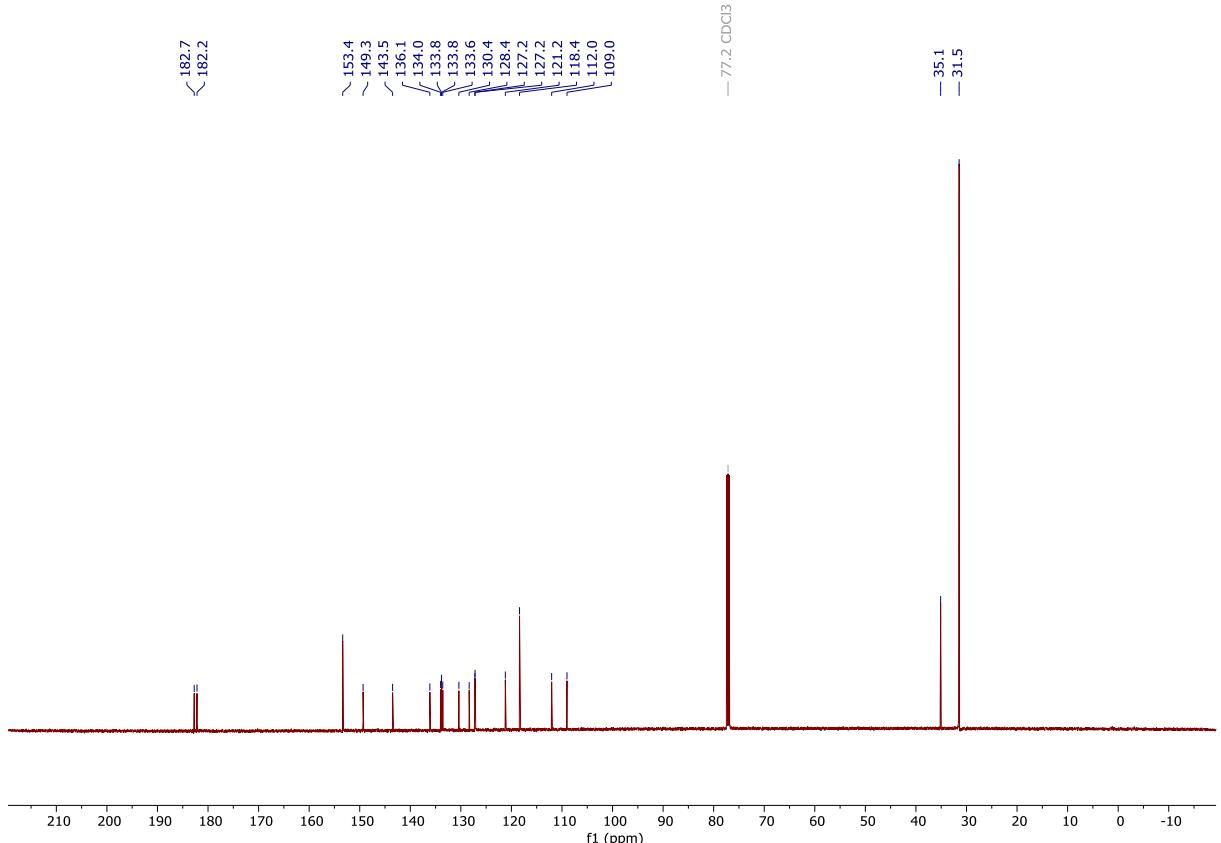


Figure S1.7. ^{13}C NMR spectrum (151 MHz, CDCl_3) of **1**.

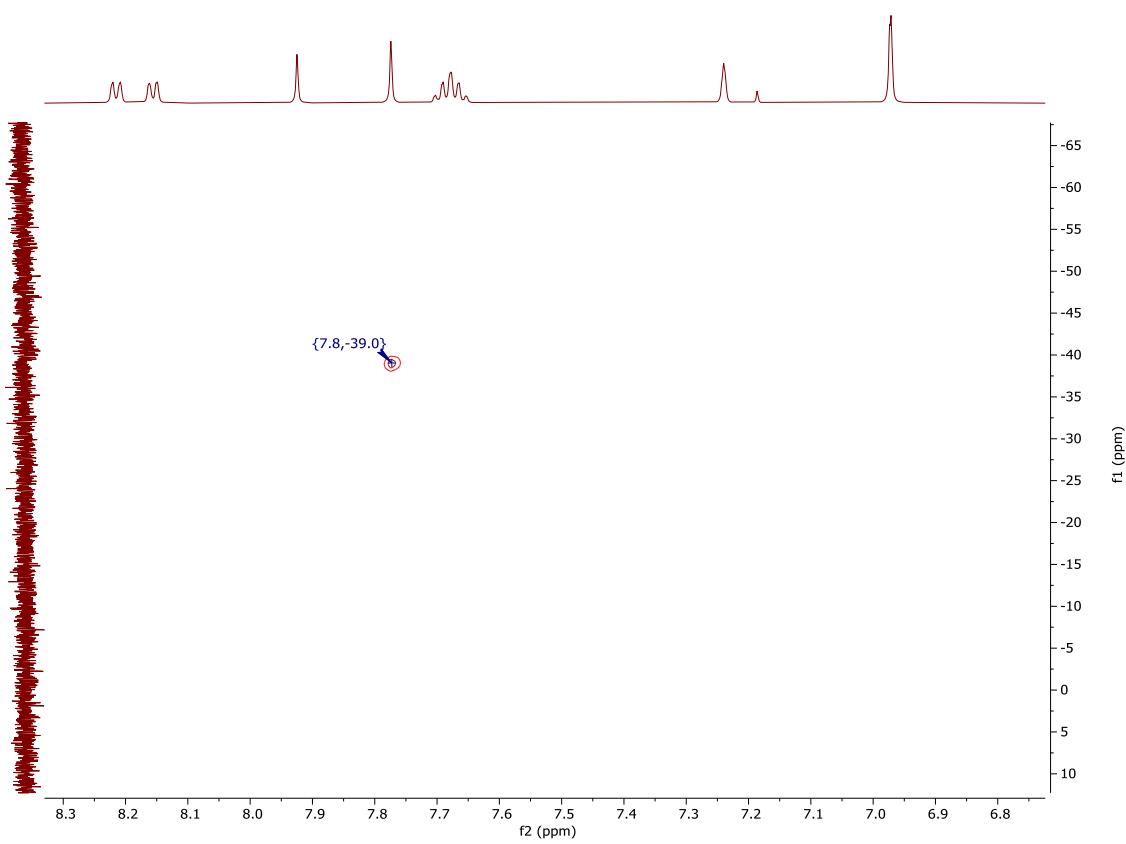
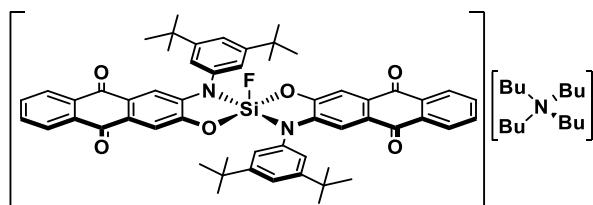


Figure S1.8. ^1H - ^{29}Si HMBC spectrum (119 MHz, CDCl_3) of **1**.

1.5 [1-F][NBu₄]



1 (20.0 mg, 22.8 µmol, 1.00 eq) was dissolved in a 1:2 benzene/*n*-hexane mixture (3 ml). TBAT (11.1 mg, 20.5 µmol, 0.90 eq) was added and the mixture stirred for 2 h. The red solution was passed through a syringe filter and the solvent was removed *in vacuo*. The residual sticky solid was triturated with *n*-hexane and filtered. The red solid was washed with *n*-hexane (3 x 2 ml), with a 3:1 *n*-hexane/benzene mixture (3 x 2 ml), and dried *in vacuo* (16.2 mg, 14.2 µmol, 62%).

¹H NMR (400 MHz, CDCl₃) δ 8.17 – 8.10 (m, 4H), 7.67 – 7.57 (m, 4H), 7.20 (s, 2H), 7.08 (s, 2H), 6.93 (s, 4H), 3.03 – 2.94 (m, 8H), 1.45 (p, *J* = 7.7 Hz, 8H), 1.25 (m, 44H, overlapping *t*Bu/NBu₄), 0.86 (t, *J* = 7.4 Hz, 12H).

Note: One signal with integral 2 is assumed to be underlying the solvent peak and the peak at 7.20 ppm.

¹³C NMR (101 MHz, CDCl₃) δ 183.6, 182.9, 154.4, 151.0, 148.6, 143.2, 134.4, 134.3, 132.9, 132.8, 127.2, 126.6, 126.4, 126.3, 122.7, 119.4, 107.1, 106.5, 58.8, 34.9, 31.7, 31.6, 23.9, 19.7, 13.7.

¹⁹F NMR (377 MHz, CDCl₃) δ -110.0 (s, 1F, ²⁹Si-satellites: *J* = 190 Hz).

²⁹Si NMR (80 MHz, CDCl₃) δ -107.9 (d, *J* = 190 Hz).

HRMS (ESI-): [C₅₆H₅₄FN₂O₆Si]⁻, calcd.: 897.3735, found: 897.3948.

HRMS (ESI+): [C₁₆H₃₆N]⁺, calcd.: 242.2842, found: 242.2830.

UV-vis (DCM): λ_{max} (ε) = 495 nm (6800 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3067 (w, $\tilde{\nu}_{\text{CH}}$), 2964 (m, $\tilde{\nu}_{\text{CH}}$), 2877 (w, $\tilde{\nu}_{\text{CH}}$), 1661 (m, $\tilde{\nu}_{\text{C=O}}$).

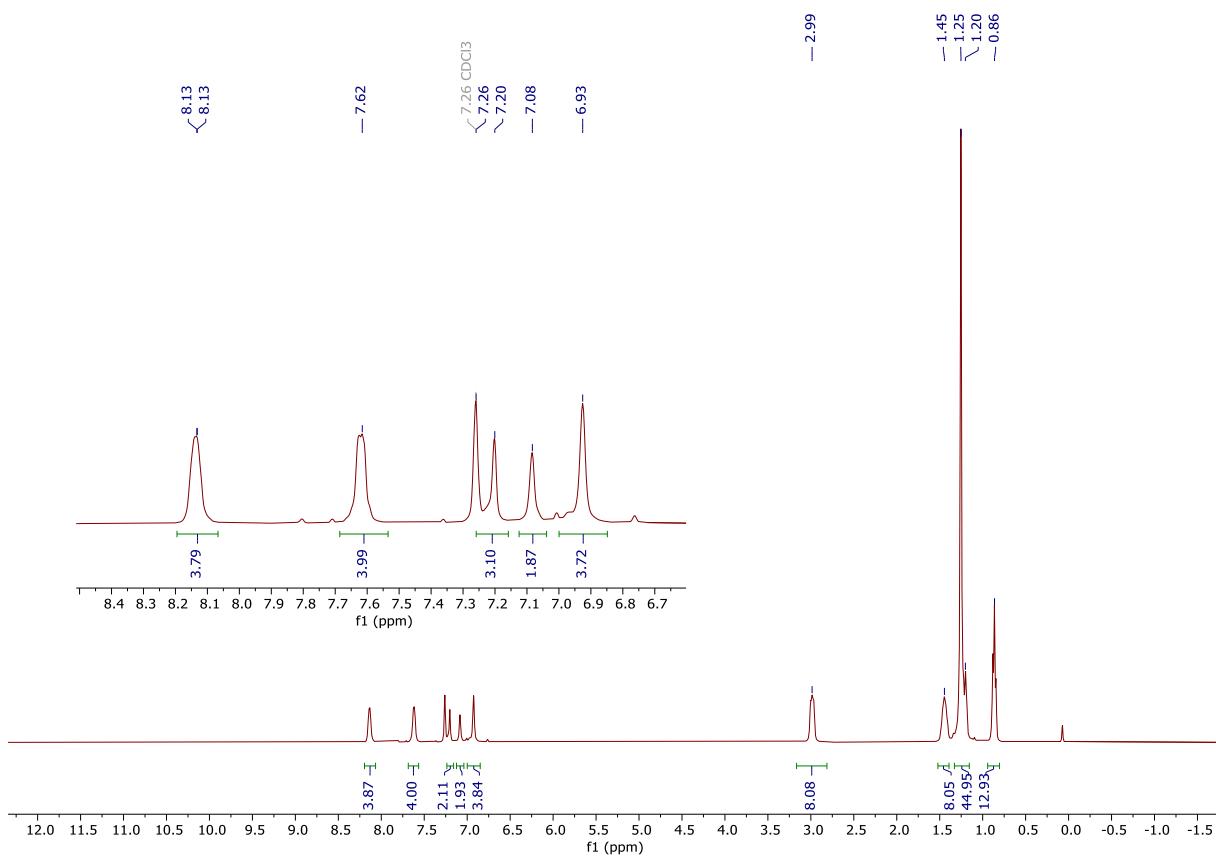


Figure S1.9. ¹H NMR spectrum (400 MHz, CDCl₃) of [1-F][NBu₄].

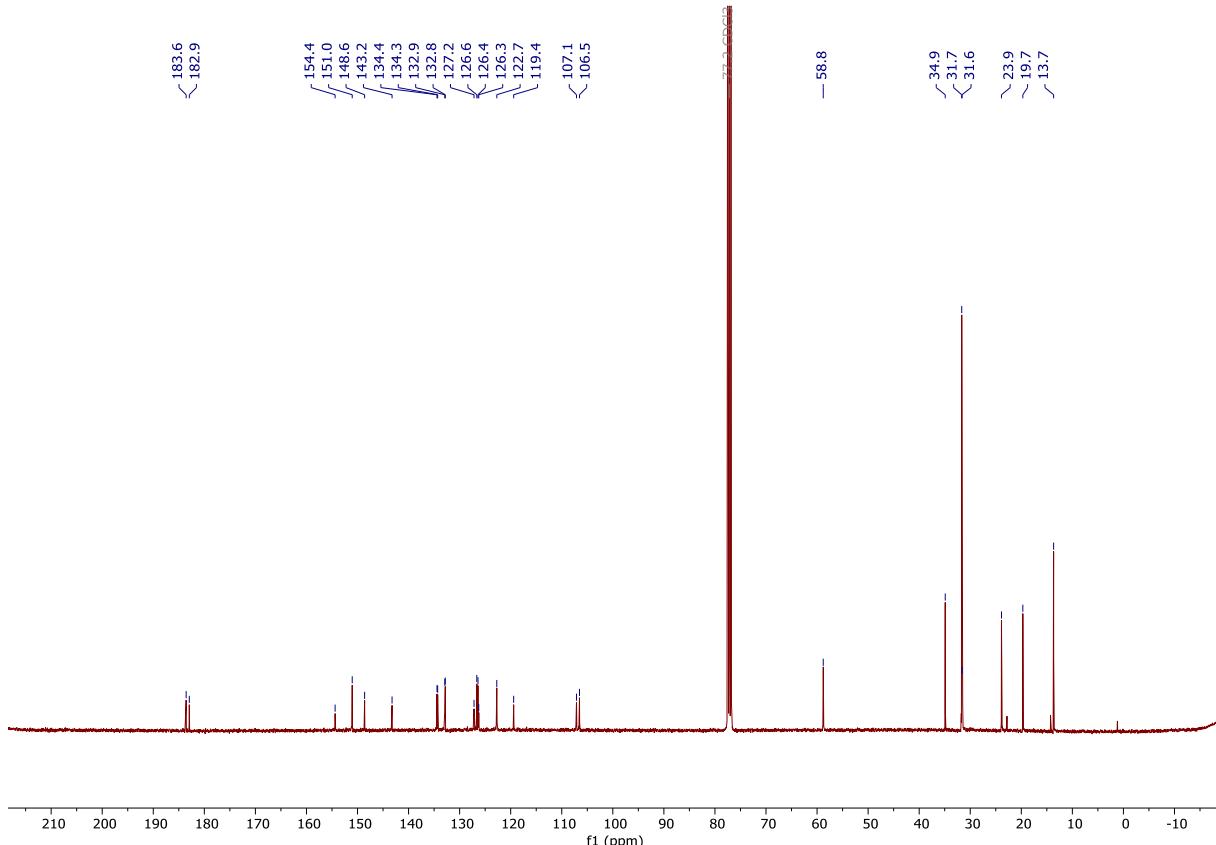


Figure S1.10. ¹³C NMR spectrum (101 MHz, CDCl₃) of [1-F][NBu₄].

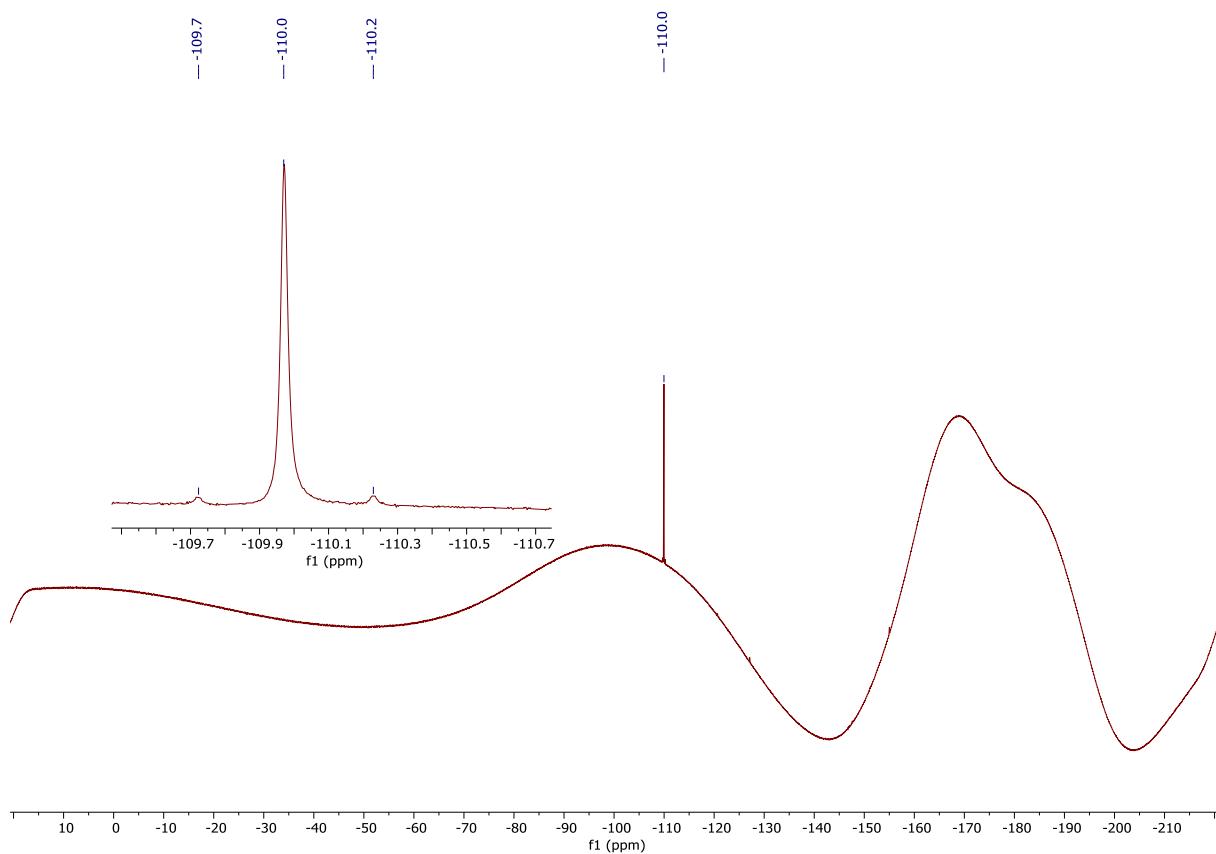


Figure S1.11. ^{19}F NMR spectrum (377 MHz, CDCl_3) of $[\mathbf{1-F}][\text{NBu}_4]$.

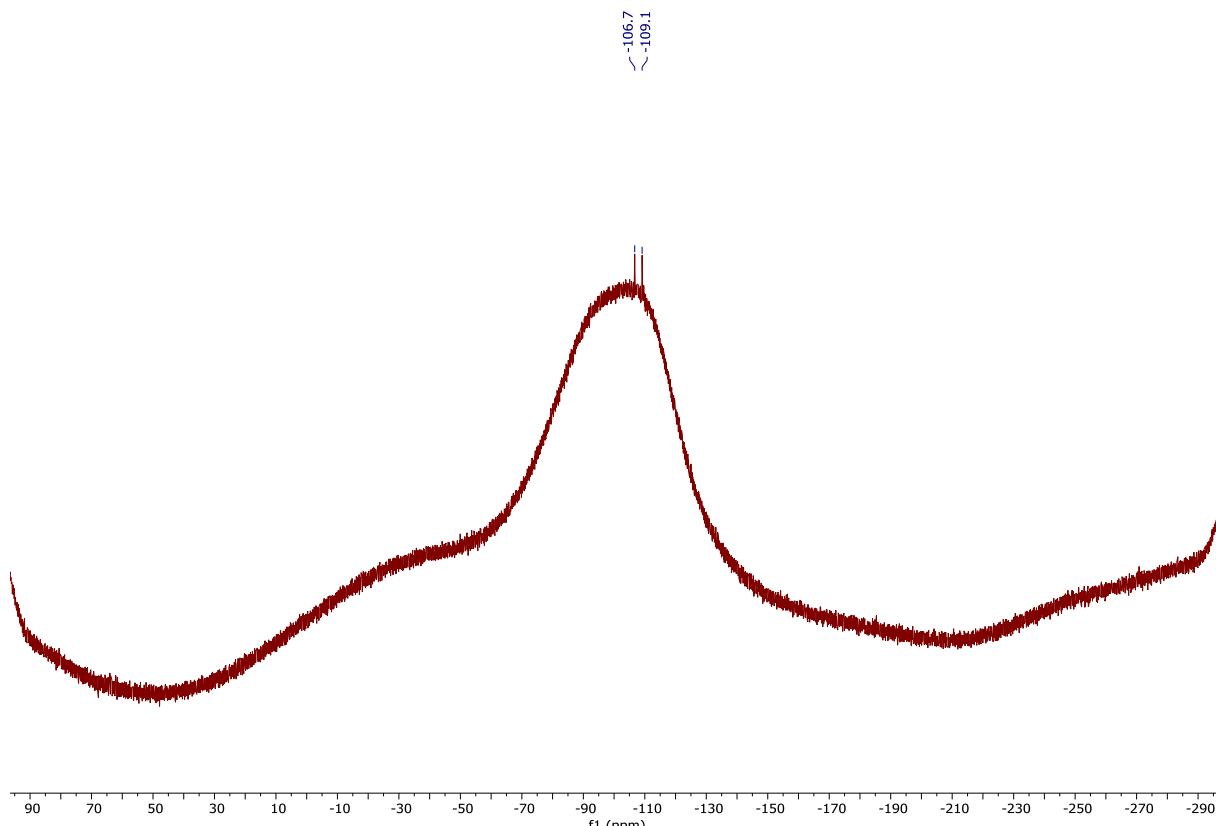
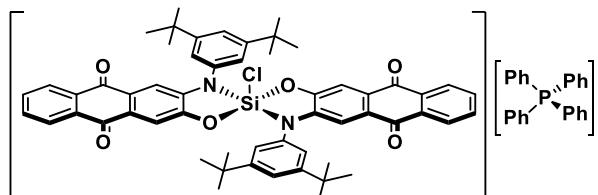


Figure S1.12. ^{29}Si NMR spectrum (80 MHz, CDCl_3) of $[\mathbf{1-F}][\text{NBu}_4]$.

1.6 [1-Cl][PPh₄]



1 (40.0 mg, 45.5 µmol, 1.00 eq) and PPh₄Cl (17.2 mg, 45.5 µmol, 1.00 eq) were dissolved in DCM (2 ml), resulting in a red solution. The solution was layered with *n*-pentane (4 ml) and stored at -40 °C for 72 h, resulting in the formation of bright red crystals, which were filtered off, washed once with DCM/*n*-pentane (1:2, 2 ml) and dried *in vacuo* (56.4 mg, 45.0 µmol, 99%). Single crystals suitable for SCXRD were grown by gaseous diffusion of *n*-pentane into a solution of [1-Cl][PPh₄] in DCM at -40 °C.

¹H NMR (600 MHz, CD₂Cl₂) δ 8.12 – 8.06 (m, 4H), 7.81 – 7.76 (m, 4H), 7.67 – 7.58 (m, 12H, overlapping PPh₄/anthraquinone), 7.55 – 7.49 (m, 8H), 7.39 (s, 2H), 7.12, 7.10 (br s, 4H and s, 2H, overlapping), 7.04 (s, 2H), 1.33 (s, 36H).

¹³C NMR (151 MHz, CD₂Cl₂) δ 183.3, 182.1, 154.1, 151.4, 148.1, 143.7, 136.0 (d, *J* = 3.1 Hz), 134.7 (d, *J* = 10.3 Hz), 134.4, 133.1, 132.9, 130.9 (d, *J* = 12.9 Hz), 127.4, 126.6, 126.5, 126.4, 123.4, 119.5, 117.8 (d, *J* = 89.6 Hz), 107.4, 106.2, 35.2, 31.7.

Note: There is one fewer signal in the ¹³C due to overlapping signals, with a shoulder peak underlying the doublet at 134.7 ppm.

²⁹Si NMR (119 MHz, CD₂Cl₂) δ -98.5.

³¹P NMR (243 MHz, CD₂Cl₂) δ 23.1.

HRMS (ESI-): [1-Cl] is not stable under ESI conditions. Ionisation of the ligand occurs. [C₂₈H₂₈NO₃]⁻, calcd.: 426.2075, found: 426.2002.

HRMS (ESI+): [C₂₄H₂₀P]⁺, calcd.: 339.1303, found: 339.1292.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon)$ = 489 nm (7600 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3065 (w, $\tilde{\nu}_{\text{CH}}$), 2953 (m, $\tilde{\nu}_{\text{CH}}$), 2902 (w, $\tilde{\nu}_{\text{CH}}$), 2864 (w, $\tilde{\nu}_{\text{CH}}$), 1659 (m, $\tilde{\nu}_{\text{C=O}}$).

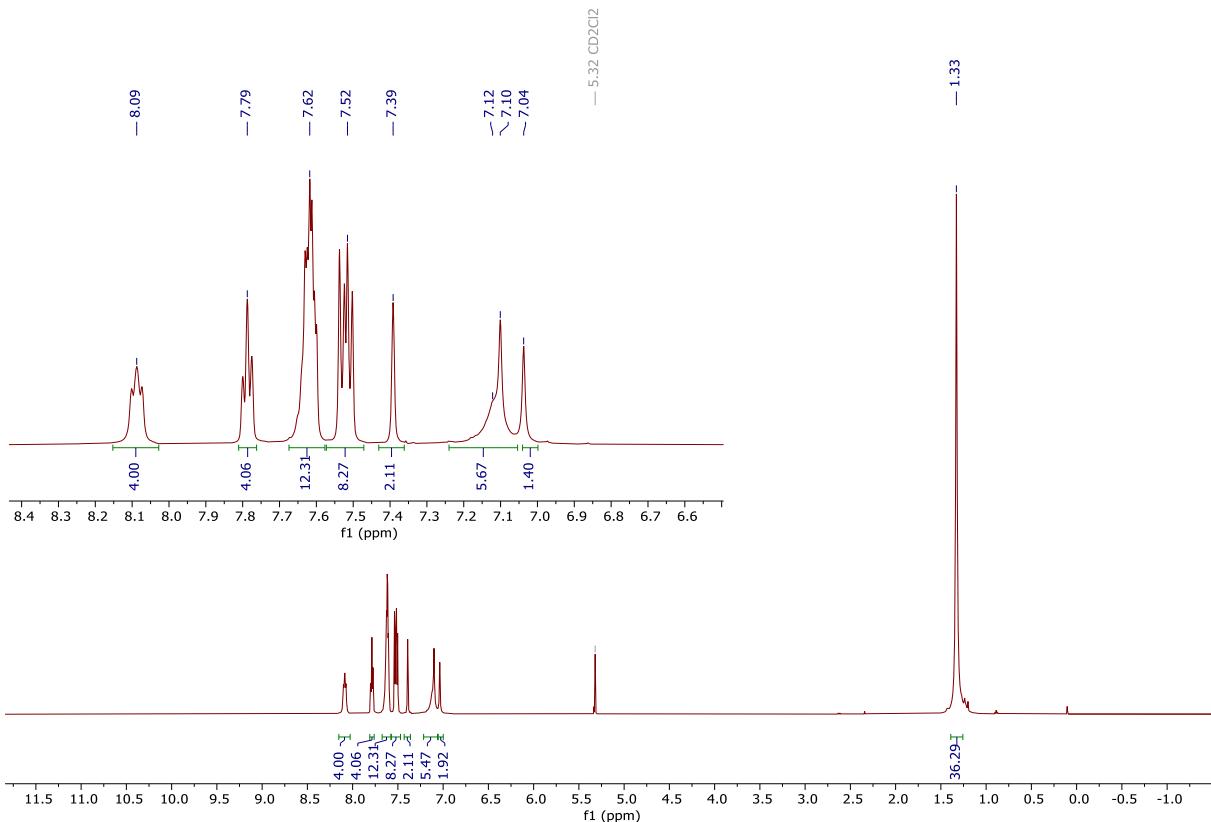


Figure S1.13. ¹H NMR spectrum (600 MHz, CD_2Cl_2) of **[1-Cl][PPh₄]**.

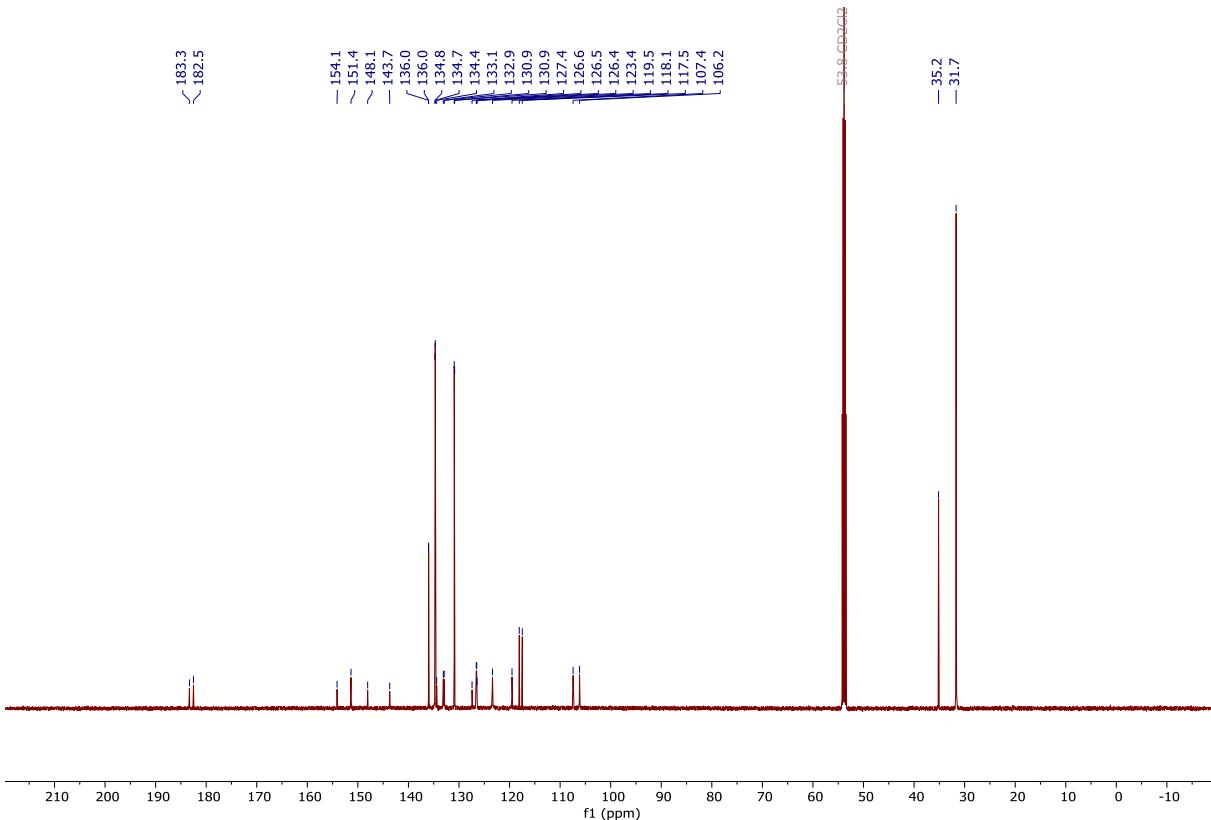


Figure S1.14. ¹³C NMR spectrum (151 MHz, CD_2Cl_2) of **[1-Cl][PPh₄]**.

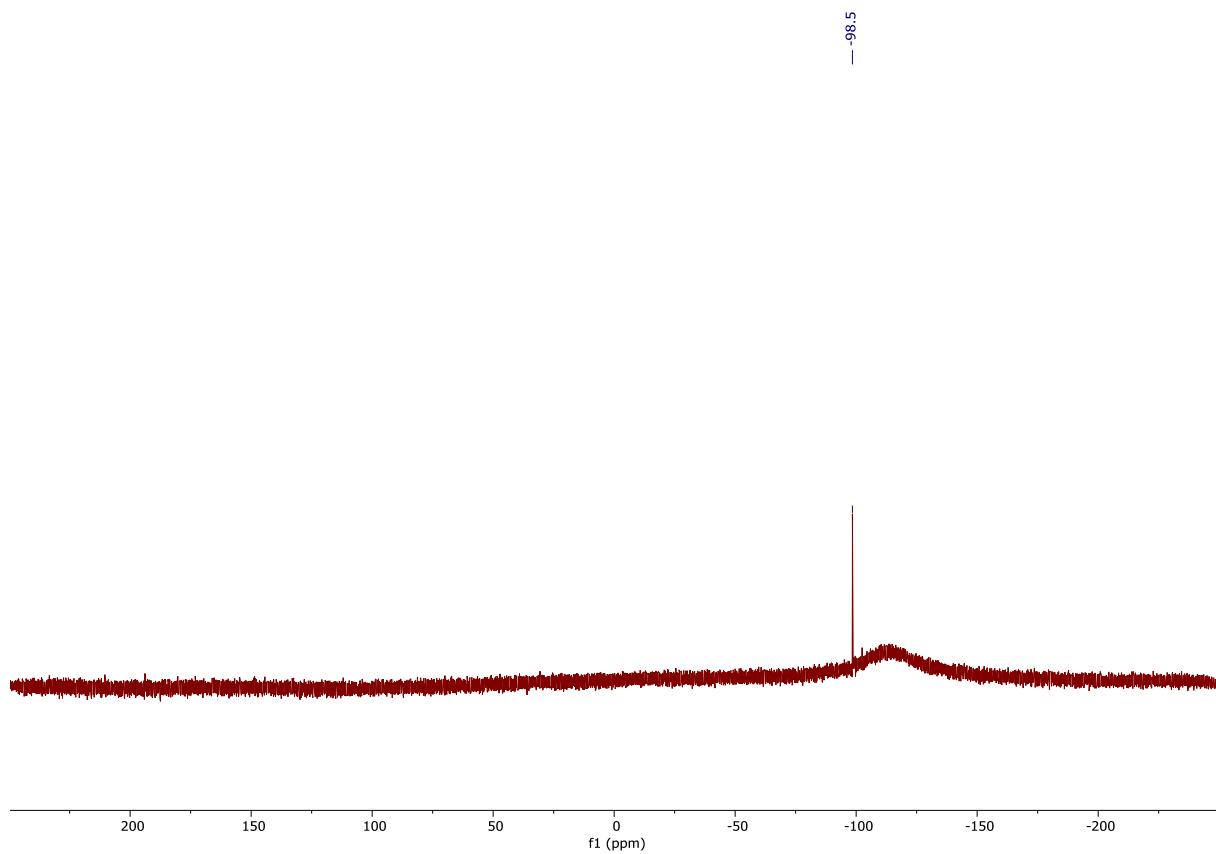


Figure S1.15. ^{29}Si NMR spectrum (119 MHz, CD_2Cl_2) of **[1-Cl][PPh₄]**.

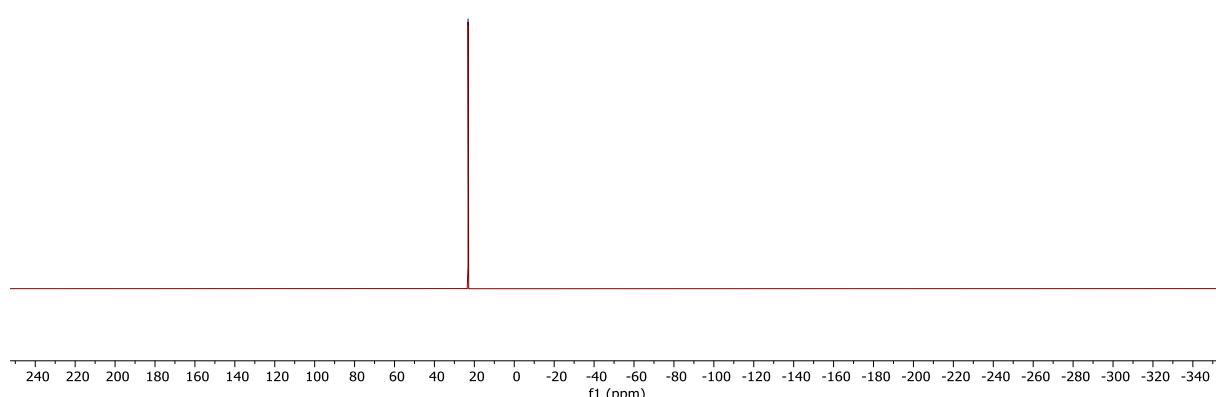
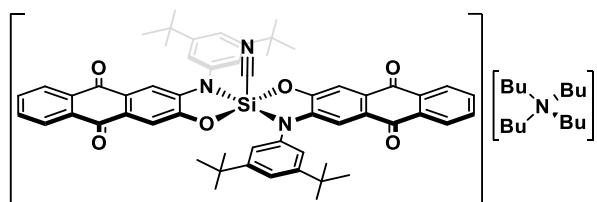


Figure S1.16. ^{31}P NMR spectrum (243 MHz, CD_2Cl_2) of **[1-Cl][PPh₄]**.

1.7 [1-CN][NBu₄]



1 (40.0 mg, 45.5 μmol , 1.00 eq) and NBu₄CN (17.2 mg, 45.5 μmol , 1.00 eq) were dissolved in DCM (2 ml), resulting in an orange solution. The solution was layered with *n*-pentane (4 ml) and stored at -40 °C for 72 h, resulting in the separation of an oily layer at the bottom of the vessel. The supernatant solution was decanted off, the residue dissolved in benzene (1 ml) and the solvent removed by freeze-drying overnight, resulting in an orange solid (29.8 mg, 26.0 μmol , 57%).

¹H NMR (600 MHz, CD₂Cl₂) δ 8.17 – 8.08 (m, 4H), 7.70 – 7.63 (m, 4H), 7.37 (s, 2H), 7.13 (s, 2H), 7.11 (s, 2H), 7.00 (br s, 4H), 1.56 – 1.47 (m, 8H), 1.36 – 1.22 (m, 44H, *overlapping* NBu₄/*tBu*), 0.92 (t, J = 7.3 Hz, 12H).

¹³C NMR (151 MHz, CD₂Cl₂) δ 183.2, 182.7, 155.4, 151.6, 148.0, 142.6, 134.6, 134.4, 133.3, 133.2, 129.2 (C≡N), 127.6, 126.9, 126.7, 126.6, 123.1, 120.0, 107.7, 106.8, 59.2, 35.1, 31.6, 24.1, 20.0, 13.7.

²⁹Si NMR (119 MHz, CD₂Cl₂) δ -113.7.

HRMS (ESI-): [C₅₇H₅₄N₃O₆Si]⁻, calcd.: 904.3787, found: 904.3792.

HRMS (ESI+): [C₁₆H₃₆N]⁺, calcd.: 242.2842, found: 242.2830.

UV-vis (DCM): λ_{max} (ε) = 481 nm (8200 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3065 (w, $\tilde{\nu}_{\text{CH}}$), 2961 (m, $\tilde{\nu}_{\text{CH}}$), 2873 (w, $\tilde{\nu}_{\text{CH}}$), 1661 (m, $\tilde{\nu}_{\text{C=O}}$).

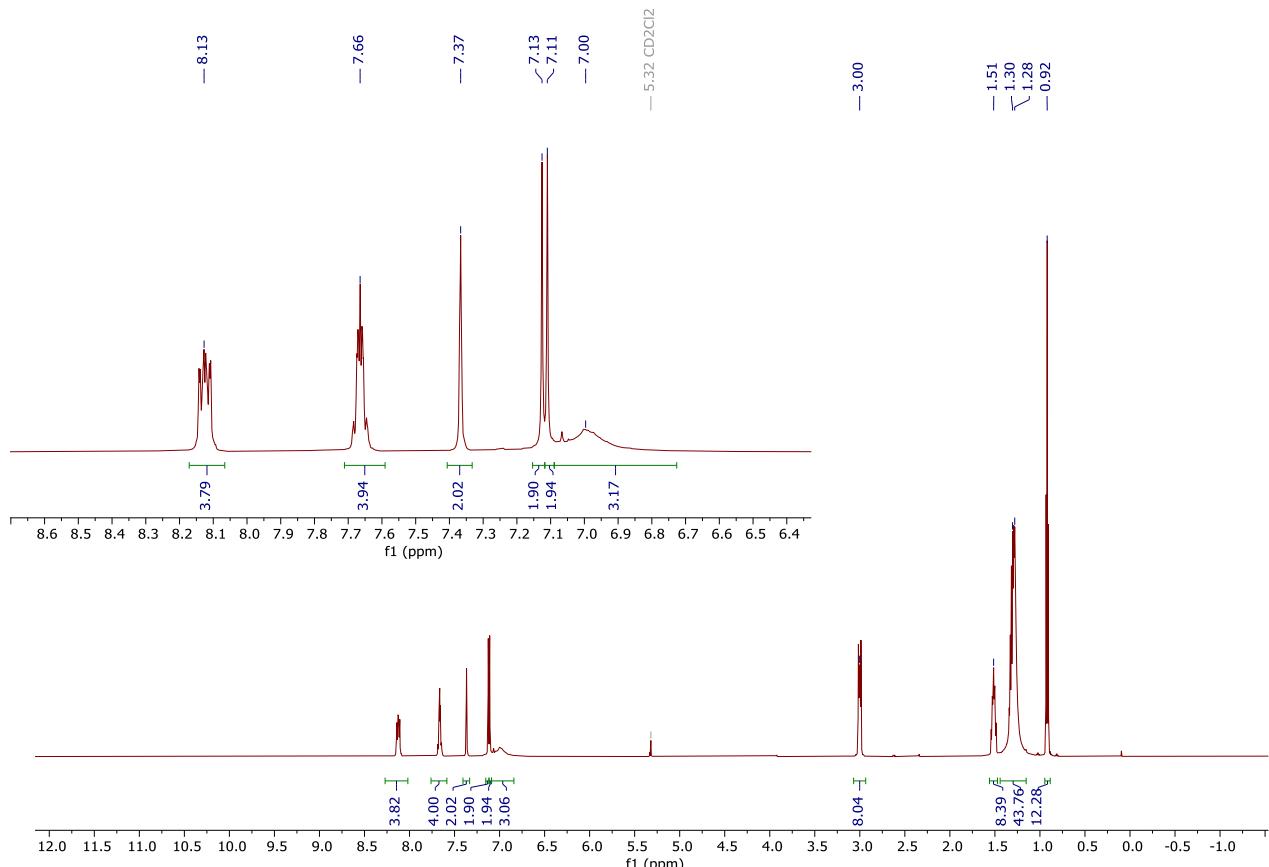


Figure S1.17. ¹H NMR spectrum (600 MHz, CD₂Cl₂) of [1-CN][NBu₄].

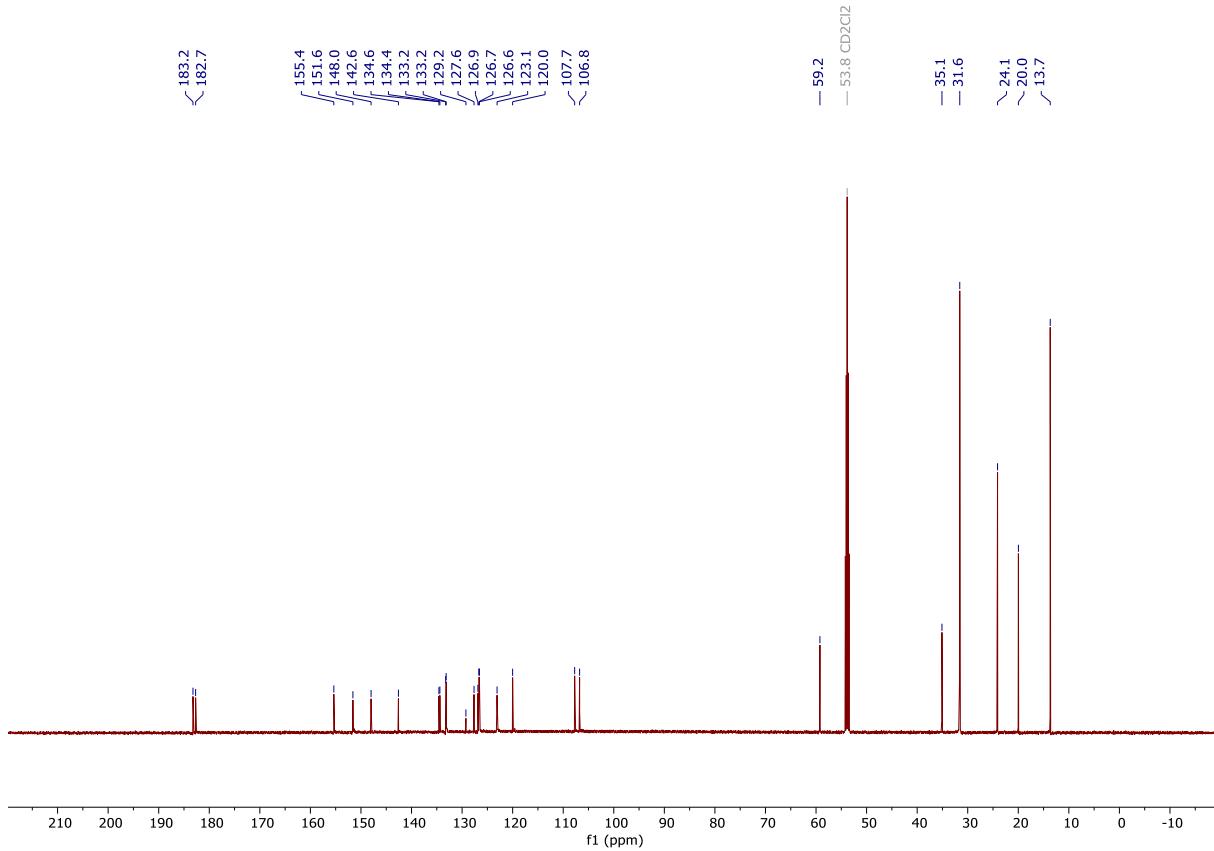


Figure S1.18. ^{13}C NMR spectrum (151 MHz, CD_2Cl_2) of $[\mathbf{1-CN}][\text{NBu}_4]$.

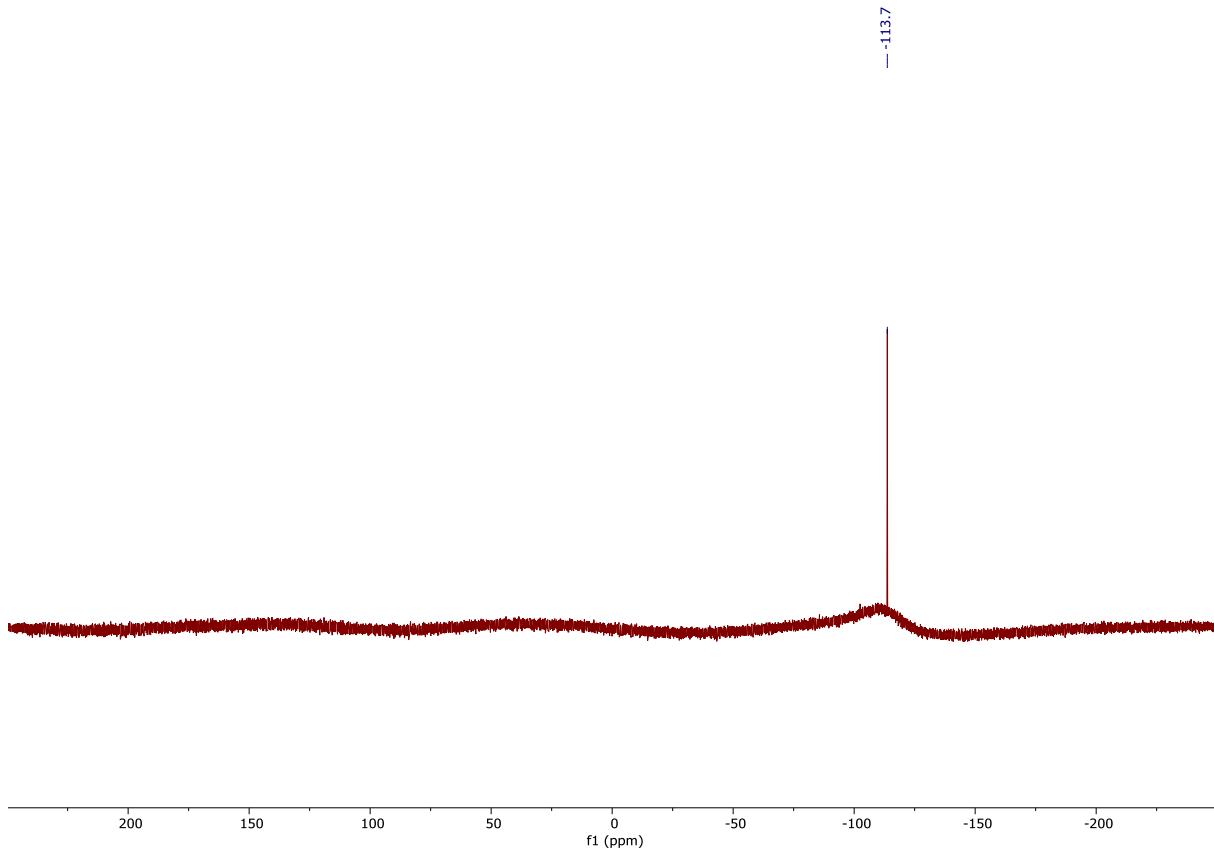
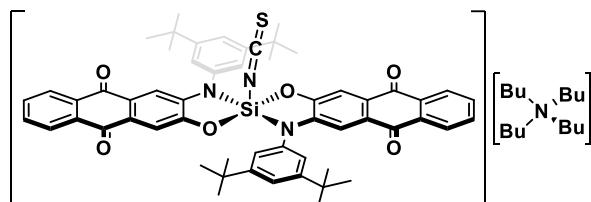


Figure S1.19. ^{29}Si NMR spectrum (119 MHz, CD_2Cl_2) of $[\mathbf{1-CN}][\text{NBu}_4]$.

1.8 [1-NCS][NBu₄]



1 (40.0 mg, 45.5 µmol, 1.00 eq) and NBu₄SCN (13.0 mg, 43.2 µmol, 0.95 eq) were dissolved in benzene (2 ml), resulting in a red solution. Scratching the glass surface of the vessel or adding a seed crystal from a previous run induced crystallisation, resulting in the formation of dark orange crystals, which were filtered off, washed with benzene (3x1 ml), and dried *in vacuo* overnight (48.8 mg, 41.4 µmol, 91%). The product contains some residual benzene which could not be removed even with prolonged drying *in vacuo*.

¹H NMR (600 MHz, CDCl₃) δ 8.17 – 8.10 (m, 4H), 7.67 – 7.59 (m, 4H), 7.27 (t, *J* = 1.8 Hz, 2H), 7.23 (s, 2H), 7.09 (s, 2H), 6.95 (br s, 4H), 2.99 – 2.93 (m, 8H), 1.49 – 1.40 (m, 8H), 1.30 – 1.16 (m, 44H, overlapping NBu₄/tBu), 0.86 (t, *J* = 7.3 Hz, 12H).

¹³C NMR (151 MHz, CDCl₃) δ 183.4, 182.9, 154.3, 151.1, 148.1, 142.4, 134.3, 134.2, 133.0, 133.0, 127.4, 126.7, 126.4, 126.2, 122.7, 119.6, 107.3, 106.9, 59.0, 34.9, 31.6, 23.9, 19.7, 13.7.

²⁹Si NMR (119 MHz, CD₂Cl₂) δ -116.0 (t, *J* = 27.0 Hz).

¹⁴N NMR (29 MHz, CD₂Cl₂) δ 88, 22.

HRMS (ESI-): **[1-NCS]** is not stable under ESI conditions. Instead, ionisation of the ligand occurs. [C₂₈H₂₈NO₃]⁻, calcd.: 426.2075, found: 426.2002.

HRMS (ESI+): [C₁₆H₃₆N]⁺, calcd.: 242.2842, found: 242.2830.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon)$ = 484 nm (6700 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3063 (w, $\tilde{\nu}_{\text{CH}}$), 2960 (m, $\tilde{\nu}_{\text{CH}}$), 2869 (w, $\tilde{\nu}_{\text{CH}}$), 2083 (s, $\tilde{\nu}_{\text{N=C=S}}$), 1662 (m, $\tilde{\nu}_{\text{C=O}}$).

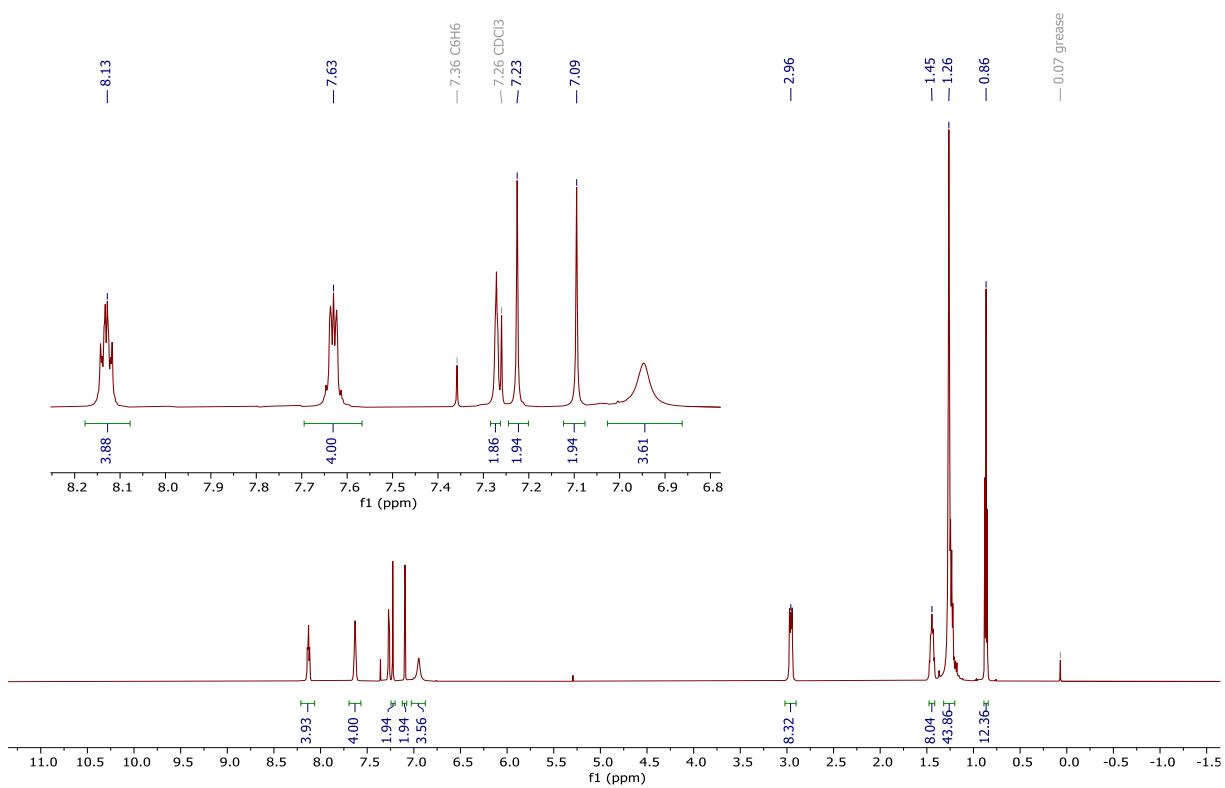


Figure S1.20. ^1H NMR spectrum (600 MHz, CDCl_3) of **[1-NCS][NBu₄]**.

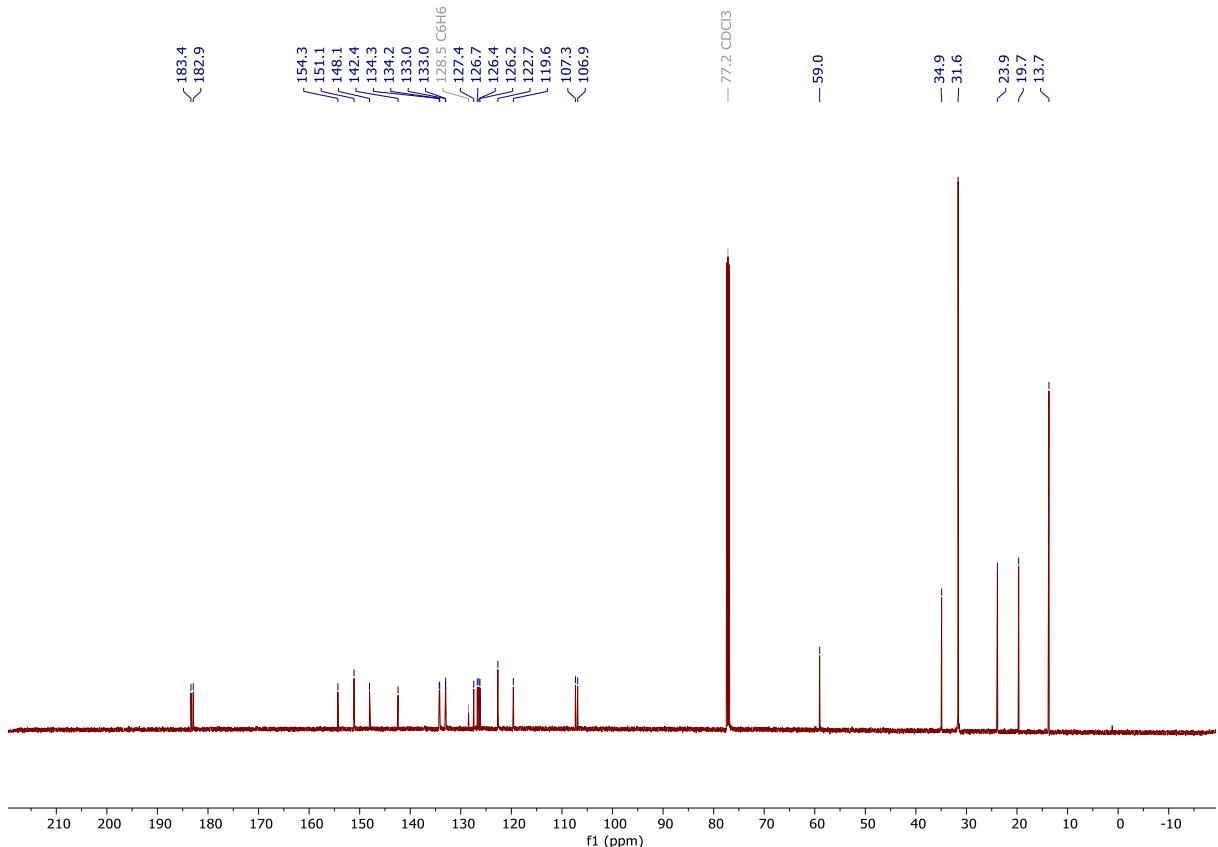


Figure S1.21. ^{13}C NMR spectrum (151 MHz, CDCl_3) of **[1-NCS][NBu₄]**.

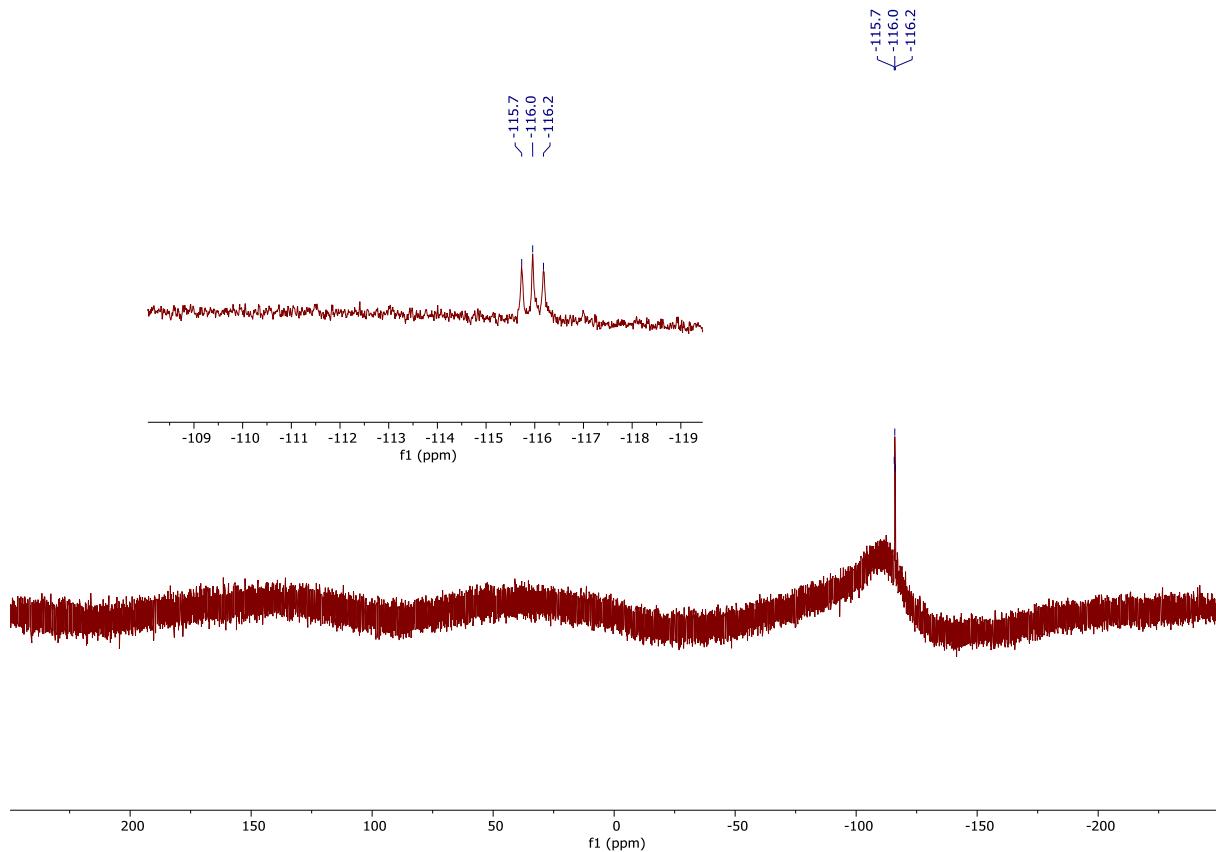


Figure S1.22. ^{29}Si NMR spectrum (119 MHz, CD_2Cl_2) of **[1-NCS][NBu₄]**.

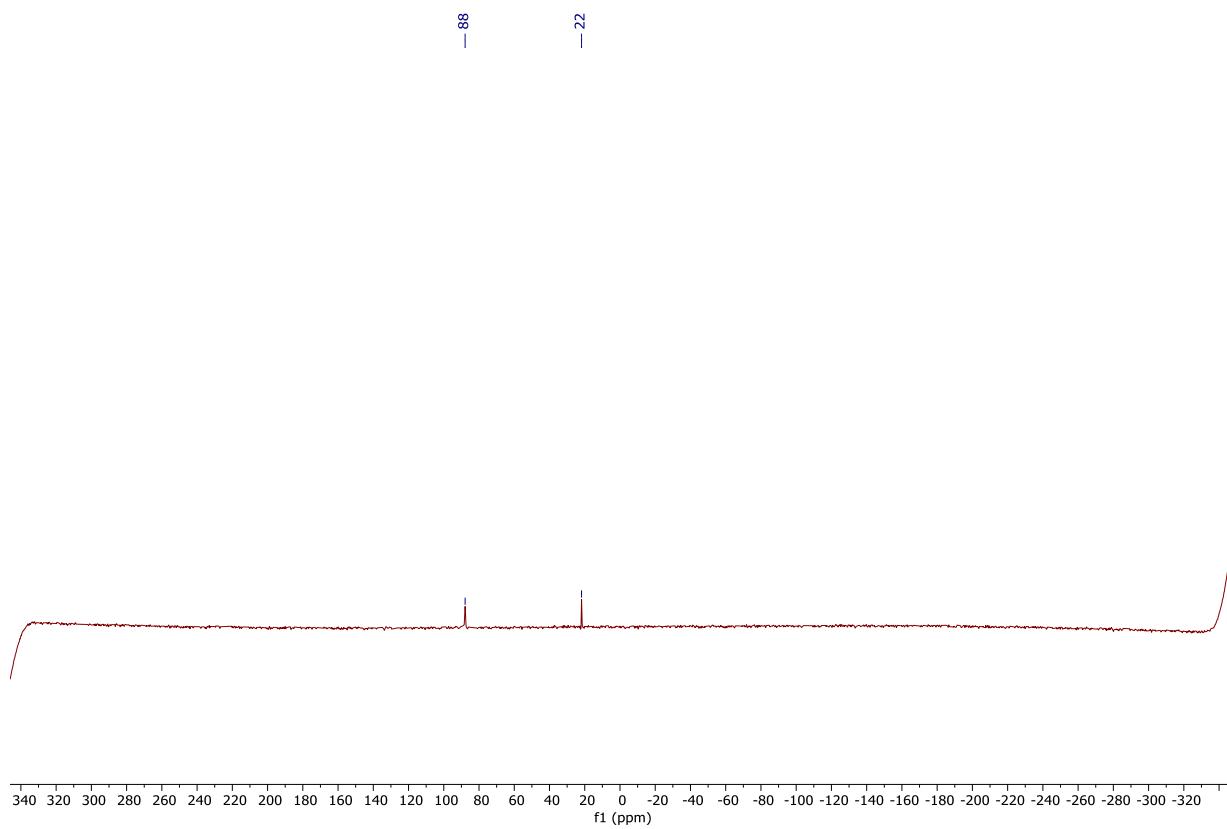
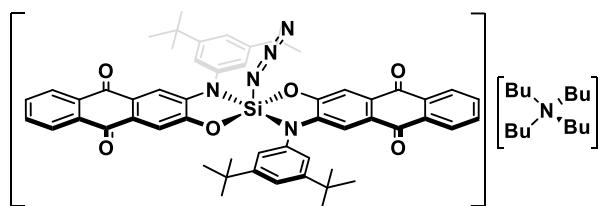


Figure S1.23. ^{14}N NMR spectrum (29 MHz, CD_2Cl_2) of **[1-NCS][NBu₄]**.

1.9 [1-N₃][NBu₄]



1 (40.0 mg, 45.5 µmol, 1.00 eq) and NBu₄N₃ (12.9 mg, 45.5 µmol, 1.00 eq) were dissolved in benzene (2 ml), resulting in a red solution. The solution was layered with *n*-pentane (4 ml) and stored at room temperature overnight, resulting in the separation of an oily layer at the bottom of the vessel. The supernatant solution was decanted off, the residue redissolved in benzene (2 ml) and the solvent removed by lyophilisation overnight. The product was isolated as a red powder (38.2 mg, 33.0 µmol, 73%). The product still contains benzene, even after drying *in vacuo* overnight.

¹H NMR (600 MHz, CDCl₃) δ 8.15 – 8.09 (m, 4H), 7.65 – 7.58 (m, 4H), 7.25 (t, *J* = 1.8 Hz, 2H), 7.19 (s, 2H), 7.05 (s, 2H), 6.95 (br s, 4H), 2.97 – 2.92 (m, 8H), 1.48 – 1.39 (m, 8H), 1.29 – 1.18 (m, 44H, overlapping NBu₄/tBu), 0.88 (t, *J* = 7.3 Hz, 12H).

¹³C NMR (151 MHz, CDCl₃) δ 183.5, 182.9, 155.0, 151.0, 148.6, 142.9, 134.4, 134.2, 132.9, 132.9, 127.3, 126.7, 126.3, 126.1, 122.7, 119.4, 107.1, 106.6, 58.8, 34.9, 31.6, 23.8, 19.7, 13.6.

²⁹Si NMR (80 MHz, CDCl₃) δ -107.6.

HRMS (ESI-): [C₅₆H₅₄N₅O₆Si]⁻, calcd.: 920.3843, found: 920.3850.

HRMS (ESI+): [C₁₆H₃₆N]⁺, calcd.: 242.2842, found: 242.2830.

UV-vis (DCM): λ_{max} (ε) = 496 nm (7300 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3066 (w, $\tilde{\nu}_{\text{CH}}$), 2961 (m, $\tilde{\nu}_{\text{CH}}$), 2874 (w, $\tilde{\nu}_{\text{CH}}$), 2121 (m, $\tilde{\nu}_{\text{N=N=N}}$), 1660 (m, $\tilde{\nu}_{\text{C=O}}$).

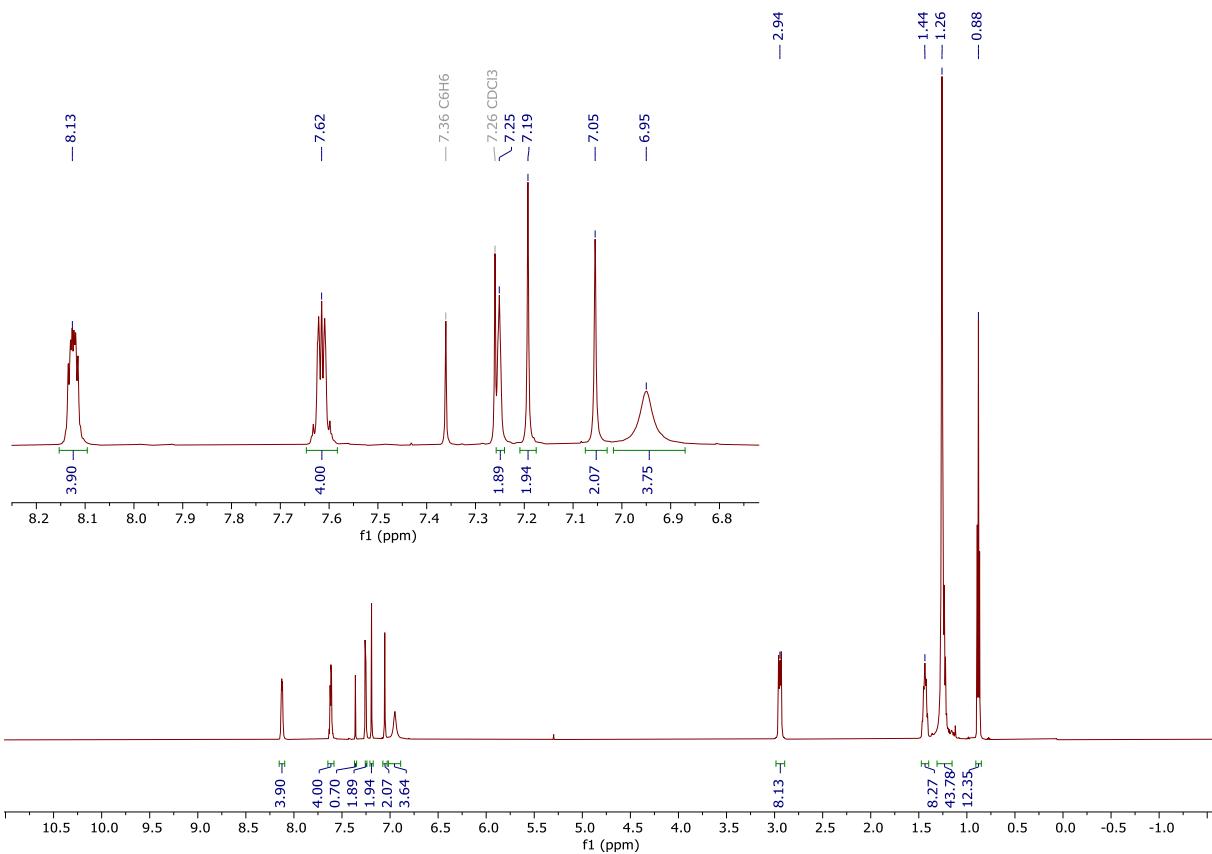


Figure S1.24. ^1H NMR spectrum (600 MHz, CDCl_3) of $[\mathbf{1-N}_3]\text{[NBu}_4]$.

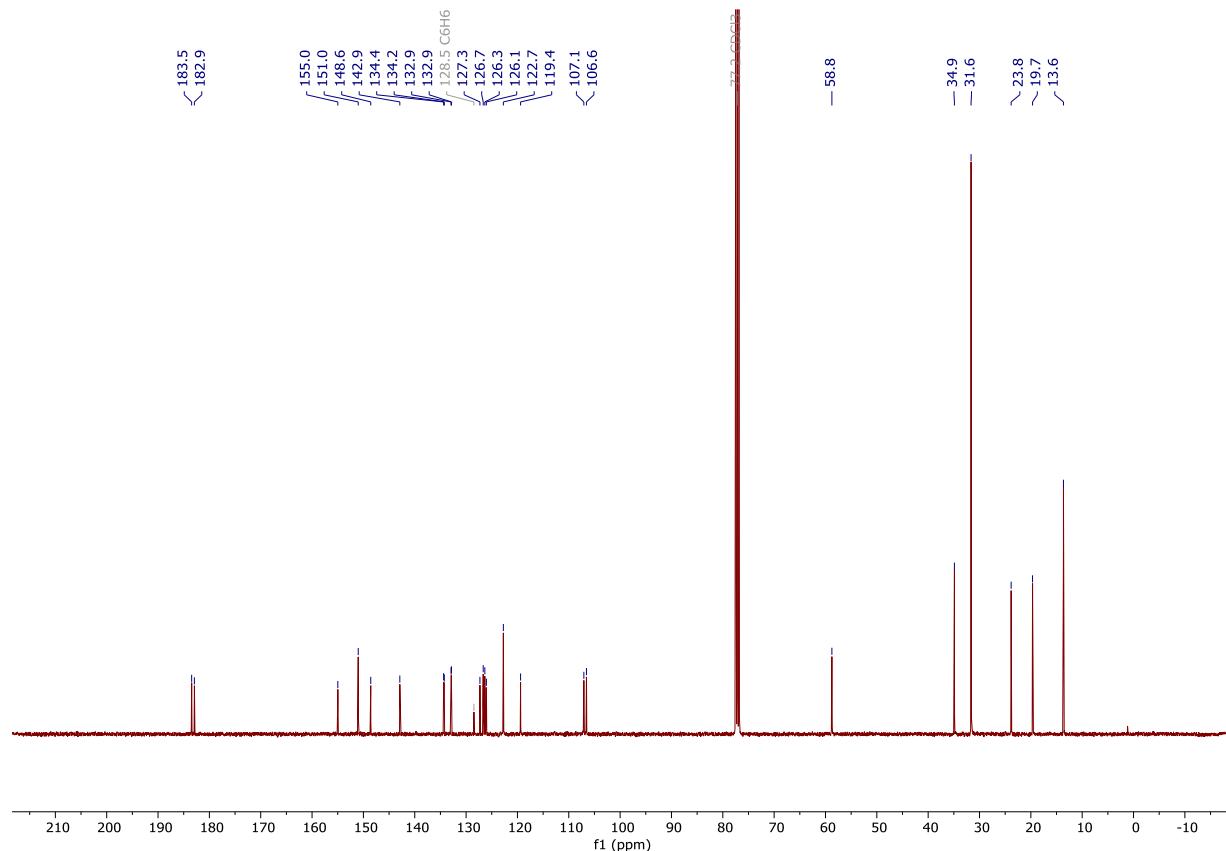


Figure S1.25. ^{13}C NMR spectrum (151 MHz, CDCl_3) of $[\mathbf{1-N}_3]\text{[NBu}_4]$.

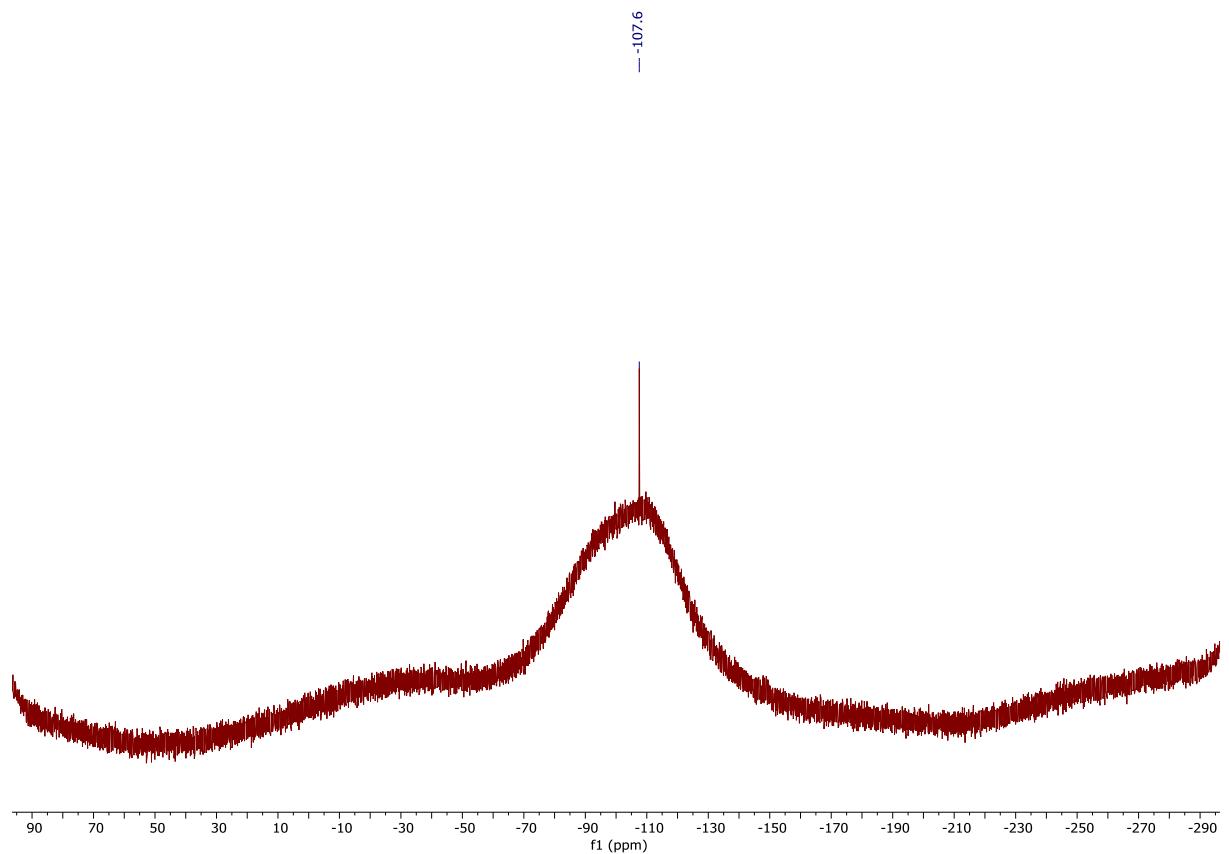
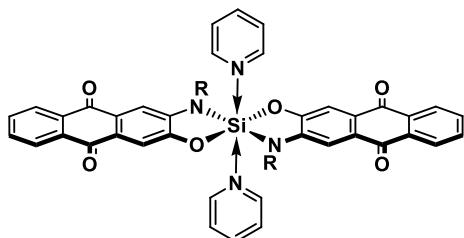


Figure S1.26. ^{29}Si NMR spectrum (80 MHz, CDCl_3) of **[1-N₃][NBu₄]**.

1.10 1-(pyridine)₂



1 (40.0 mg, 45.5 μ mol, 1.00 eq) and pyridine (8.1 μ l, 0.10 mmol, 2.2 eq) were dissolved in DCM (1 ml), resulting in an orange solution. The solution was layered with *n*-pentane (4 ml) and stored at -40 °C overnight, resulting in the precipitation of a dark red solid. The solid was filtered off, washed with a 2:1 *n*-pentane/DCM mixture (2x1 ml), and dried *in vacuo* (38.3 mg, 36.9 μ mol, 81%).

¹H NMR (600 MHz, CDCl₃) δ 8.75 – 8.71 (m, 4H), 8.22 – 8.13 (m, 4H), 7.93 (t, J = 7.6 Hz, 2H), 7.70 – 7.62 (m, 4H), 7.52 (t, J = 6.7 Hz, 4H), 7.40 (s, 2H), 7.38 (s, 2H), 7.35 (s, 2H), 7.00 (s, 4H), 1.33 (s, 36H).

¹³C NMR (101 MHz, CDCl₃) δ 183.3, 182.7, 152.3, 152.0, 147.9, 147.0, 141.8, 140.2, 134.1, 133.9, 133.3, 133.2, 128.6, 127.0, 126.9, 126.7, 125.3, 122.4, 120.2, 108.8, 107.8, 35.1, 31.7.

^{29}Si NMR (80 MHz, CDCl_3) δ -104.9.

MS (LFDI+): Due to the lability of binding pyridine, only the parent complex **1** was detected. $[C_{56}H_{54}N_2O_6Si]^+$, calcd.: 878.38, found: 878.34.

UV-vis (mono-adduct **1-pyridine**, DCM): $\lambda_{\text{max}} (\varepsilon) = 452 \text{ nm} (7300 \text{ M}^{-1}\text{cm}^{-1})$.

IR (ATR) [cm⁻¹] ν = 3063 (w, ν_{CH}), 2956 (m, ν_{CH}), 2863 (w, ν_{CH}), 1662 (m, ν_{C=O}), 1646 (w, ν_{C=O}).

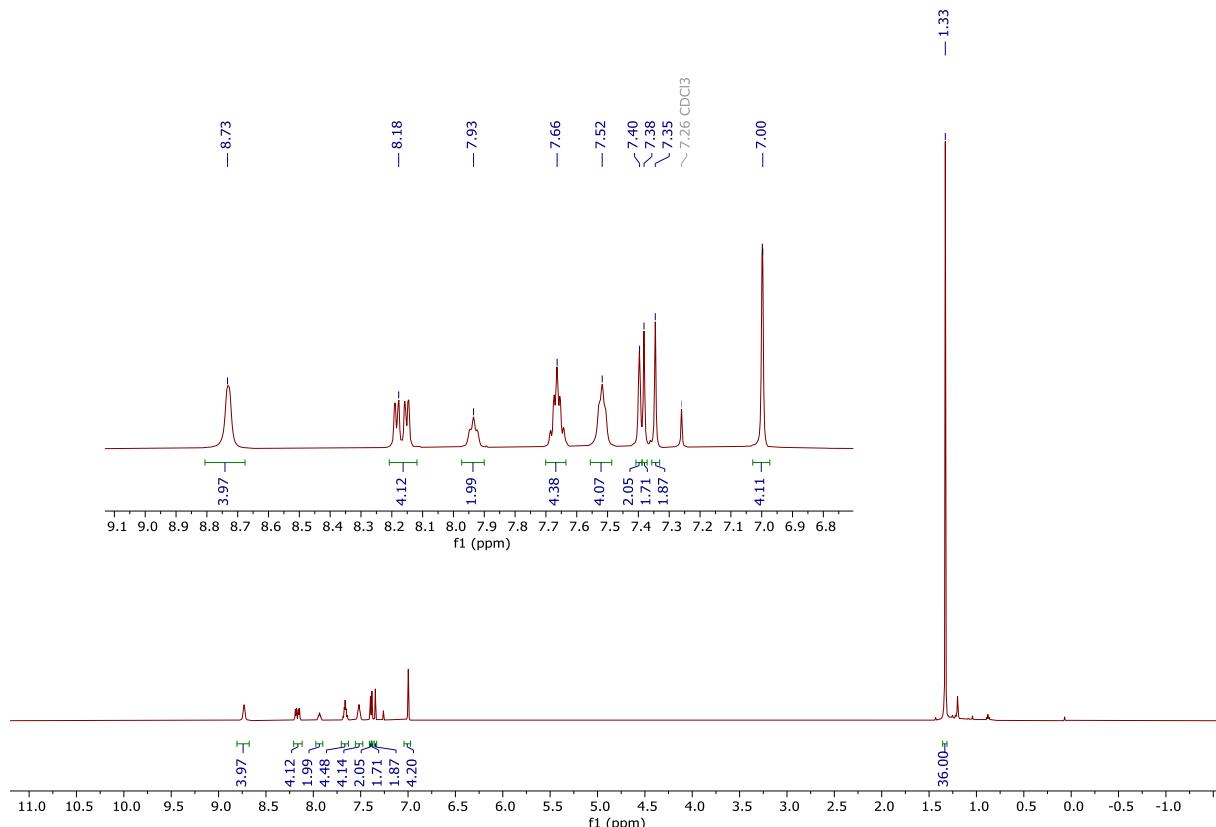
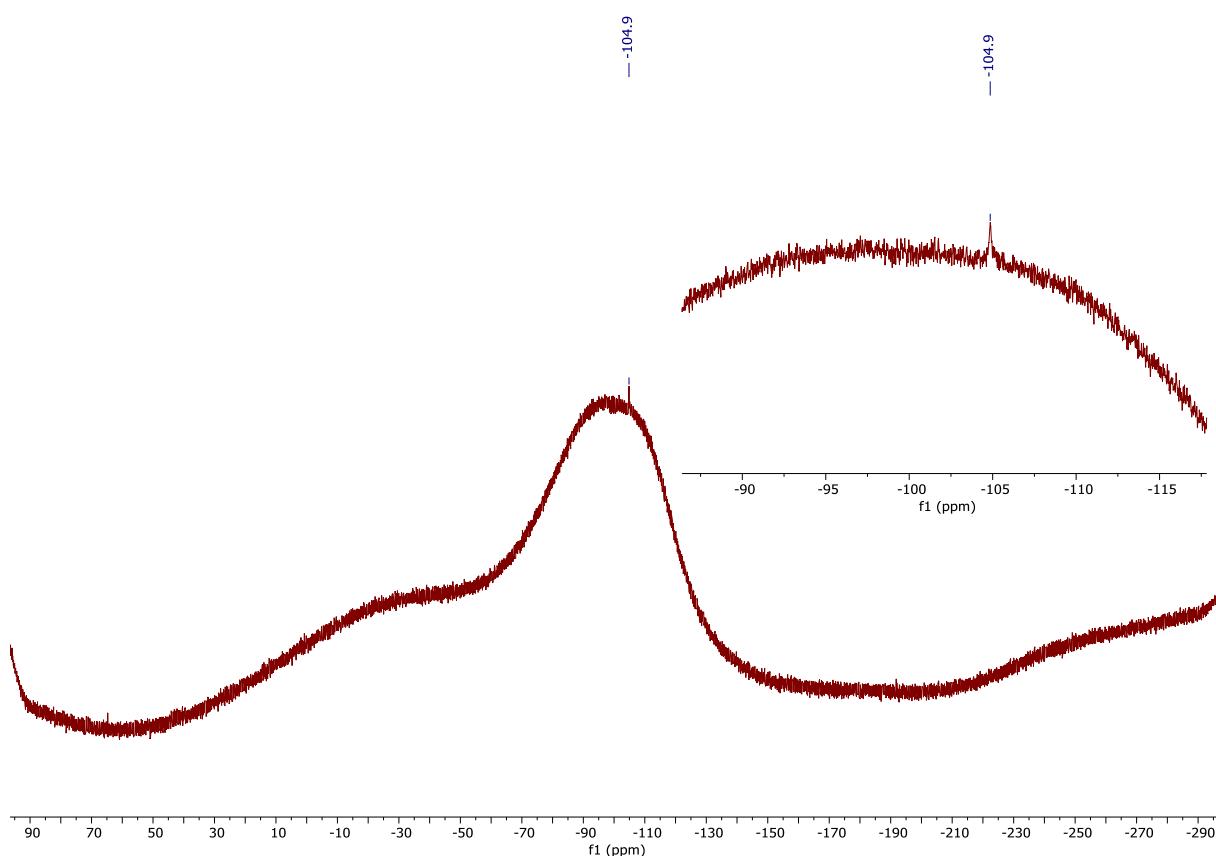
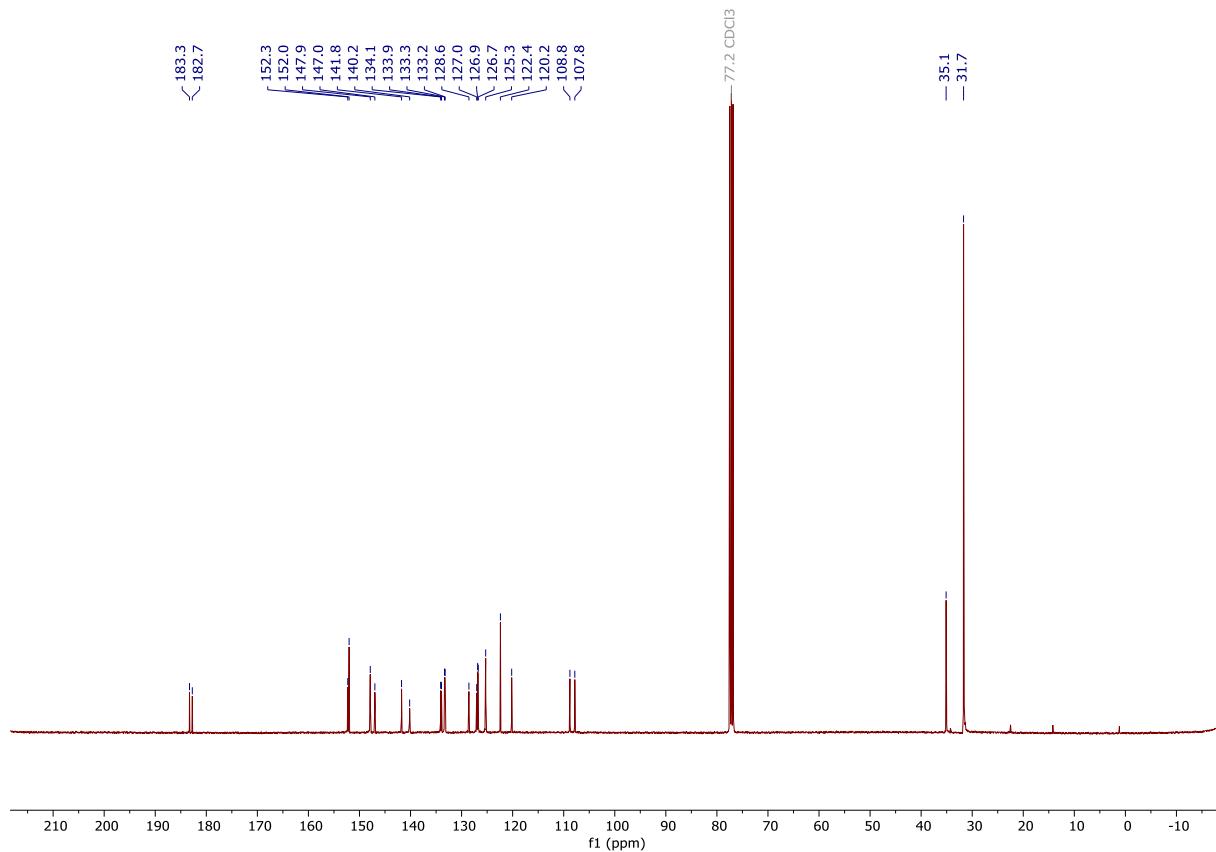


Figure S1.27. ^1H NMR spectrum (600 MHz, CDCl_3) of **1-(pyridine)₂**.



1.11 1-DMAP



1 (40.0 mg, 45.5 μmol , 1.00 eq) was dissolved in benzene (2 ml). DMAP (5.4 mg, 44 μmol , 0.97 eq) was added, resulting in the immediate precipitation of an orange solid. After stirring for 15 minutes, the suspension was filtered and the solid washed with benzene (2x1 ml). The product was isolated as an orange powder after drying *in vacuo* overnight (36.4 mg, 36.4 μmol , 80 %).

$^1\text{H NMR}$ (600 MHz, CDCl_3) δ 8.29 (d, $J = 7.2$ Hz, 2H), 8.22 – 8.16 (m, 2H), 8.16 – 8.13 (m, 2H), 7.70 – 7.61 (m, 4H), 7.39 – 7.37 (m, 2H), 7.31 (s, 2H), 7.25 (s, 2H), 6.99 (br s, 4H), 6.55 (d, $J = 7.2$ Hz, 2H), 3.14 (s, 6H), 1.34 (s, 36H).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 183.4, 182.9, 156.2, 152.9, 151.7, 147.8, 145.9, 142.5, 134.2, 134.1, 133.2, 133.0, 128.2, 126.8, 126.6, 122.8, 119.8, 108.2, 107.3, 106.6, 39.8, 35.1, 31.7.

Note: One less signal than expected is observed in the $^{13}\text{C NMR}$ spectrum. This is likely due to overlapping signals, with a shoulder peak next to the signal at 126.6 ppm.

$^{29}\text{Si NMR}$ (119 MHz, CD_2Cl_2) δ -107.0.

MS (LIFDI+): Due to the lability of binding DMAP, only the parent complex **1** was detected. $[\text{C}_{56}\text{H}_{54}\text{N}_2\text{O}_6\text{Si}]^+$, calcd.: 878.38, found: 878.36.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon) = 466 \text{ nm (}8100 \text{ M}^{-1}\text{cm}^{-1}\text{)}$.

IR (ATR) [cm $^{-1}$] $\tilde{\nu} = 3073$ (w, $\tilde{\nu}_{\text{CH}}$), 2963 (m, $\tilde{\nu}_{\text{CH}}$), 2869 (w, $\tilde{\nu}_{\text{CH}}$), 1660 (m, $\tilde{\nu}_{\text{C=O}}$).

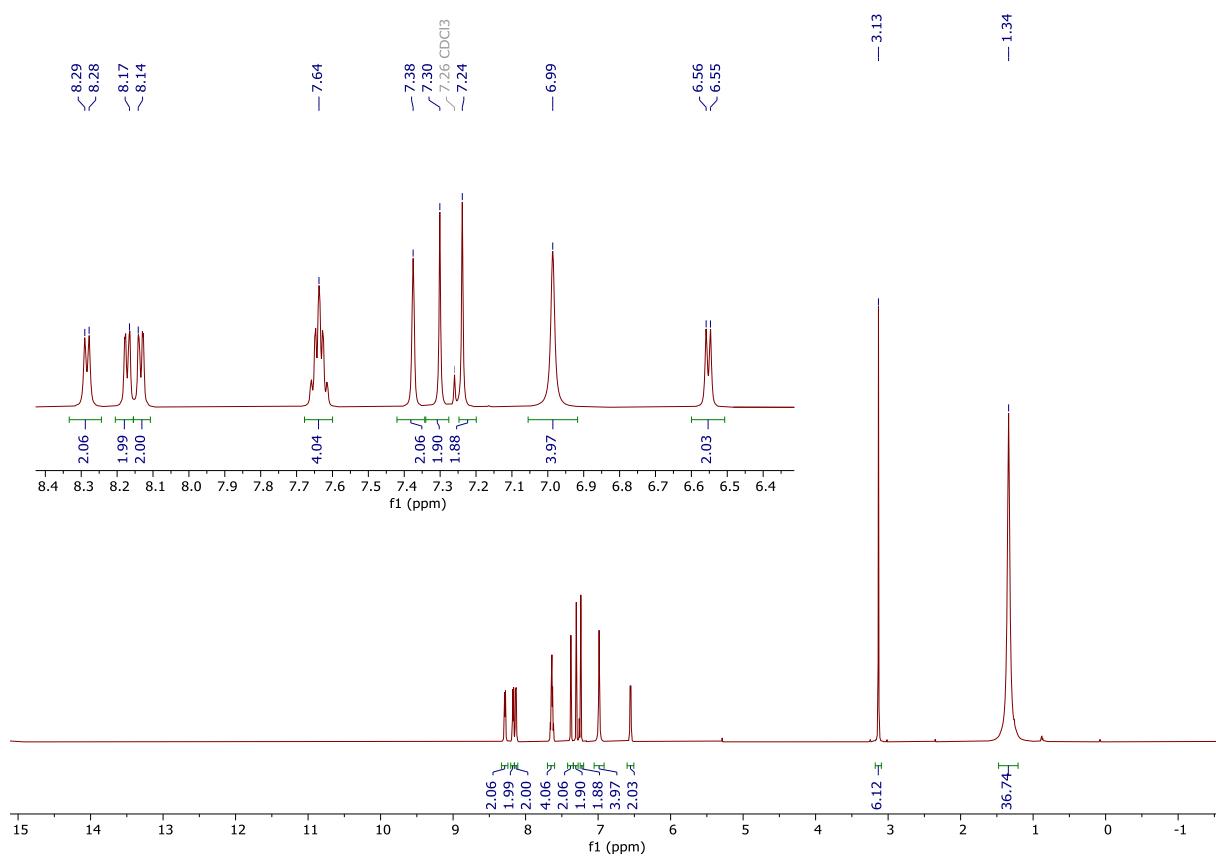


Figure S1.30. ¹H NMR spectrum (600 MHz, CDCl₃) of 1-DMAP.

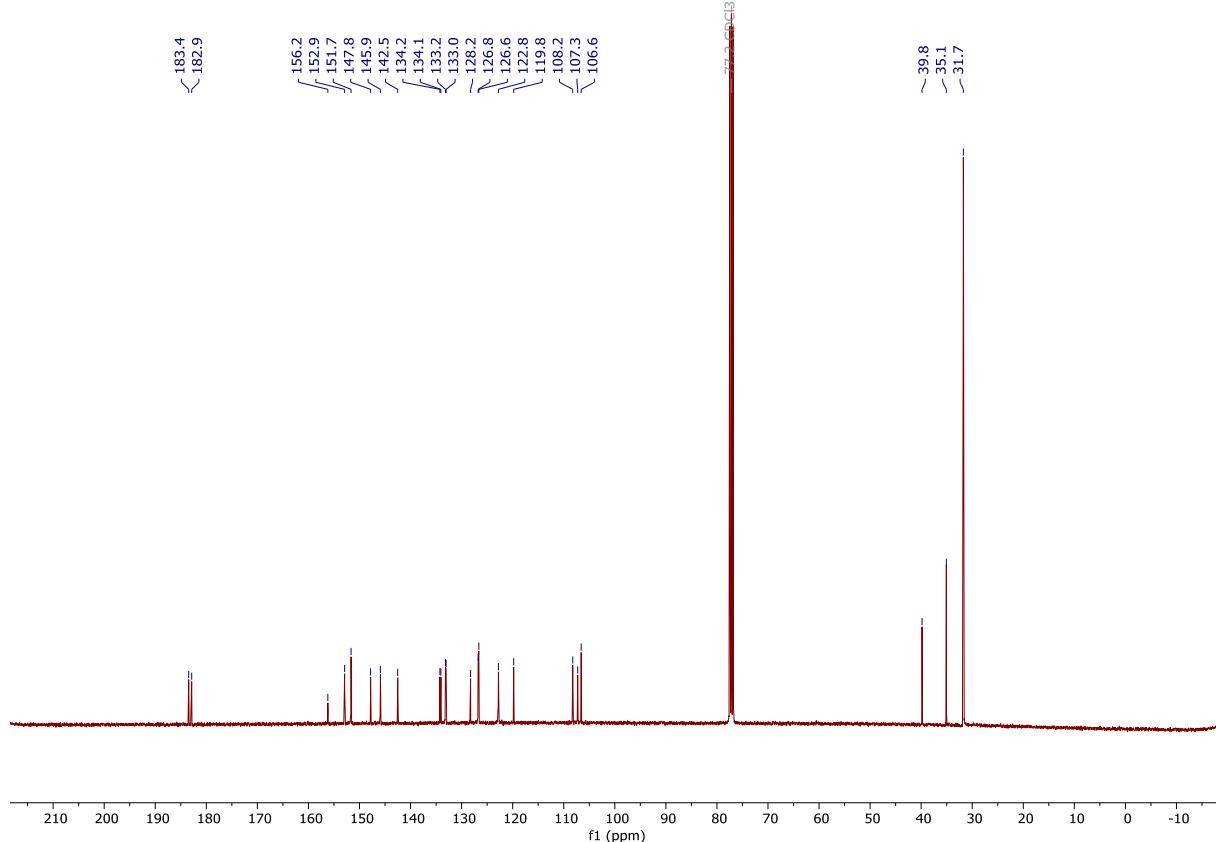


Figure S1.31. ¹³C NMR spectrum (101 MHz, CDCl₃) of 1-DMAP.

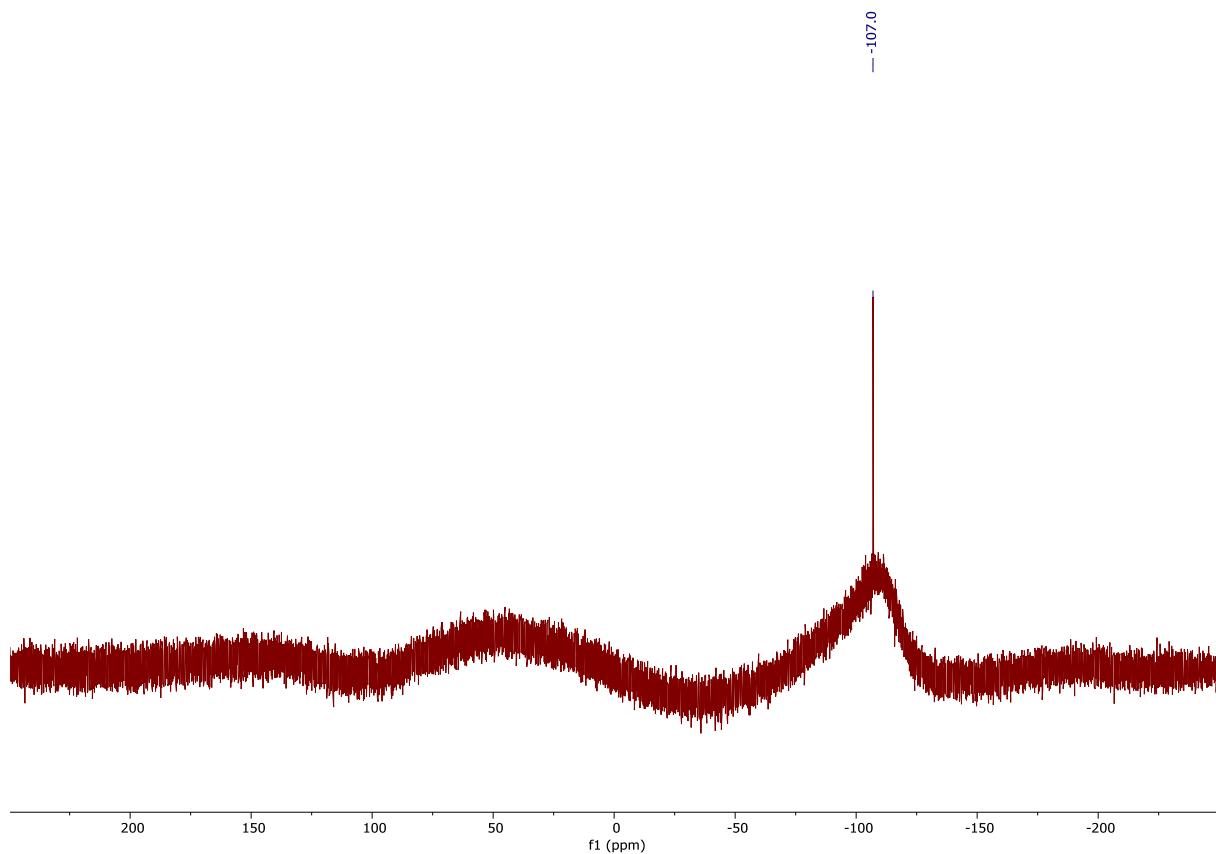
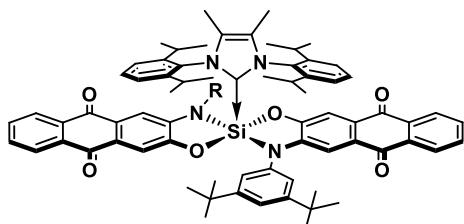


Figure S1.32. ^{29}Si NMR spectrum (119 MHz, CD_2Cl_2) of **1-DMAP**.

1.12 1-dippNHC



1 (40.0 mg, 45.5 µmol, 1.00 eq) was dissolved in DCM (2 ml). 4,5-Me₂-1,3-(dipp)₂-NHC (19.0 mg, 45.5 µmol, 1.00 eq) was added, resulting in a dark orange solution. Pentane (4 ml) was added, giving a clear solution, which was stored at -40 °C for 2 h. The precipitated solid was filtered off and washed with a 10:1 n-pentane/DCM mixture (3 x 1 ml). The product was isolated as an orange powder after drying in vacuo (43.2 mg, 33.3 µmol, 73 %).

¹H NMR (400 MHz, CDCl₃) δ 8.11 (m, 4H), 7.69 – 7.58 (m, 4H), 7.42 (d, J = 7.5 Hz, 2H), 7.20 (br, 2H), 7.13 (s, 2H), 6.82 (overlapping: br, 4H and s, 2H), 6.21 (overlapping: t, J = 1.9 Hz, 2H, and br, 2H), 3.06 (sept, J = 6.8 Hz, 2H), 2.39 (br sept, 2H), 1.78 (s, 6H), 1.37 (s, 18H), 1.20 – 1.04 (m, 18H), 0.93 (d, J = 5.5 Hz, 6H), 0.69 (s, 18H).

¹³C NMR (101 MHz, CDCl₃) δ 183.5 (overlapping C=O and carbene-C), 182.4, 151.9, 150.7, 148.3, 142.3, 134.2, 134.1, 133.0, 132.9, 130.2, 127.5, 126.6, 126.5, 125.4, 122.8, 121.9, 121.1, 120.6, 107.7, 107.3, 35.0 (tBu-C), 34.3 (tBu-C), 31.9 (tBu-CH₃), 30.8 (tBu-CH₃), 29.1 (iPr-CH), 28.9 ((iPr-CH), 24.4 (iPr-CH₃), 23.9 (iPr-CH₃), 11.1 (carbene-CH₃).

Note: Binding of the NHC induces asymmetry (two sets of tBu/iPr groups), which, together with line broadening in the ¹³C NMR spectrum, did not allow to account for all carbon atoms. Characteristic carbon shifts were assigned by HSQC and HMBC experiments and are denoted above.

²⁹Si NMR (119 MHz, CDCl₃) δ No signal observed due to limited signal intensity and solubility.

MS (LIFDI+): [C₈₅H₉₄N₄O₆Si]⁺, calcd.: 1294.69, found: 1294.65.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon)$ = 475 nm (5700 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3072 (w, $\tilde{\nu}_{\text{CH}}$), 2964 (m, $\tilde{\nu}_{\text{CH}}$), 2870 (w, $\tilde{\nu}_{\text{CH}}$), 1664 (m, $\tilde{\nu}_{\text{C=O}}$), 1652 (w, $\tilde{\nu}_{\text{C=O}}$).

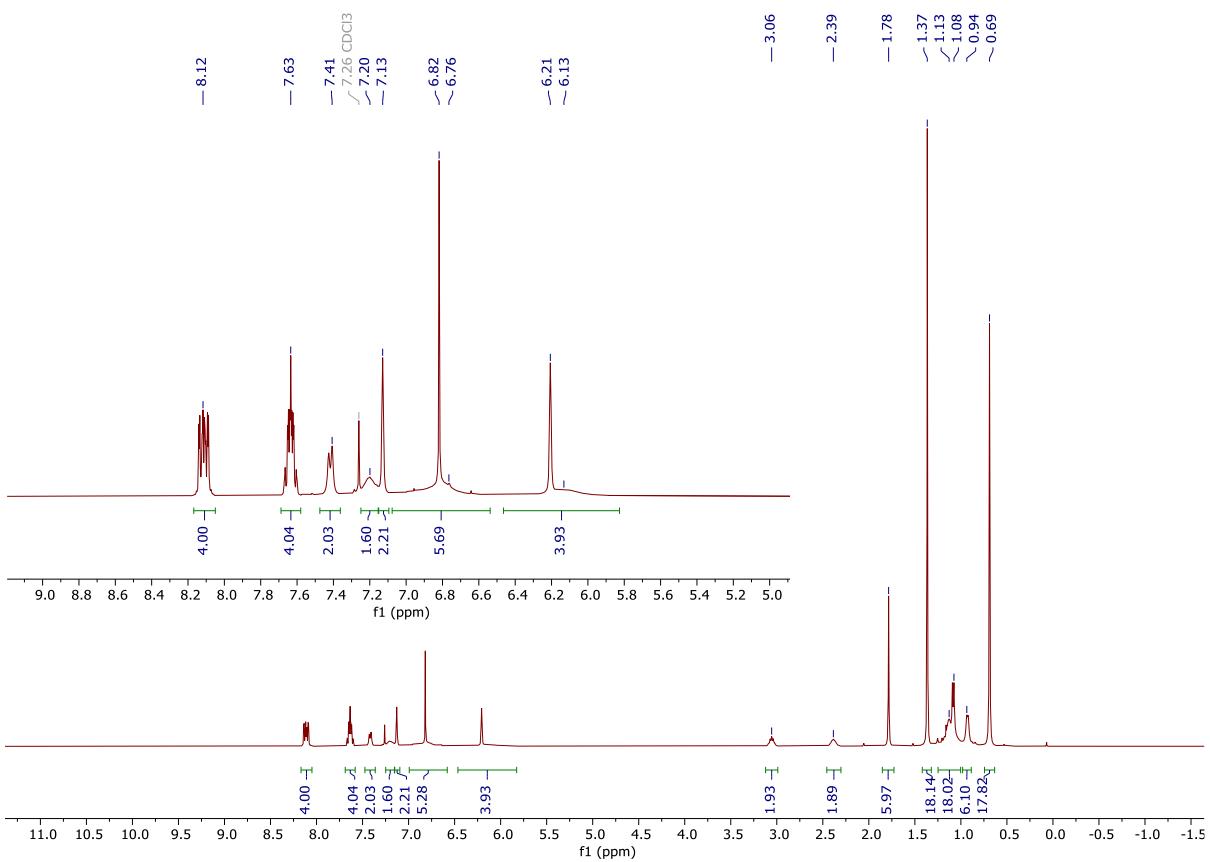


Figure S1.33. ¹H NMR spectrum (400 MHz, CDCl₃) of **1-dippNHC**.

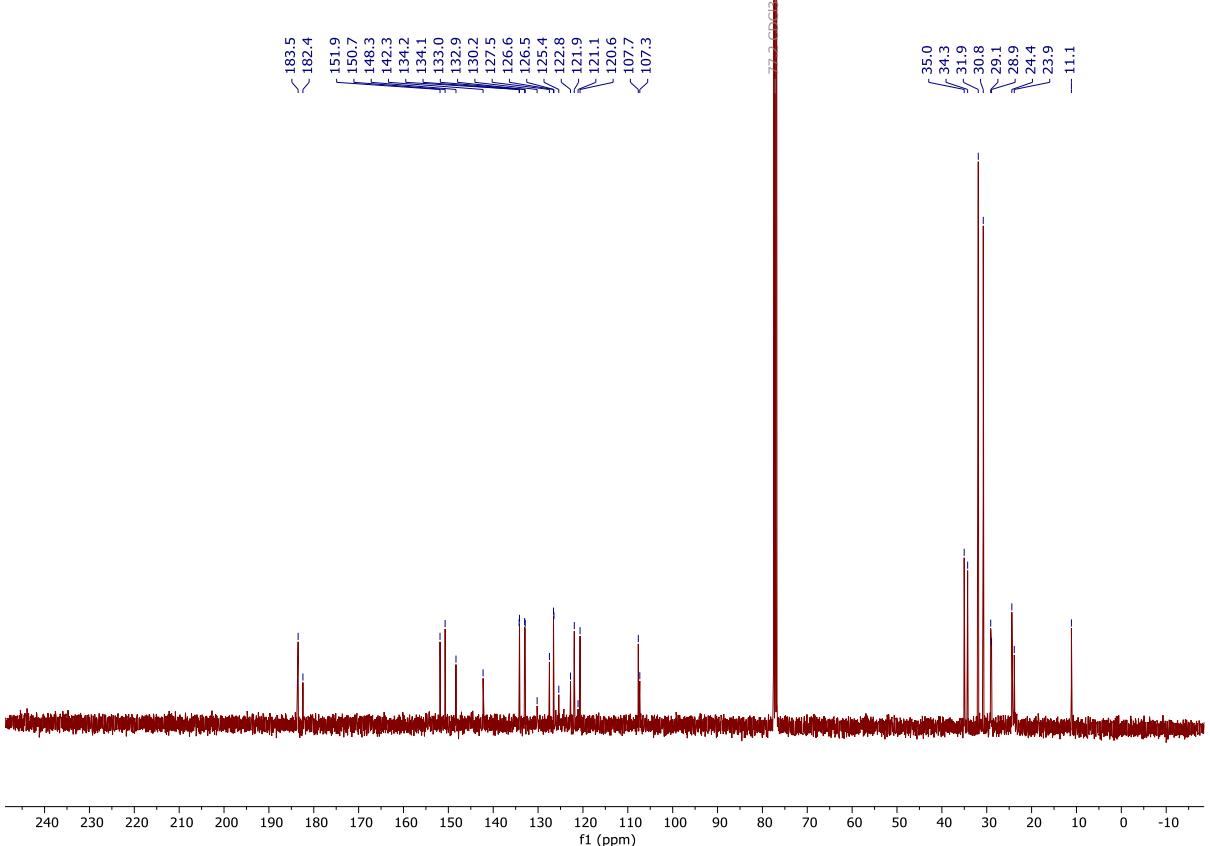


Figure S1.34. ¹³C NMR spectrum (101 MHz, CDCl₃) of **1-dippNHC**.

1.13 1-DMSO



1 (40.0 mg, 45.5 μmol , 1.00 eq) was dissolved in DCM (1 ml). DMSO (4.9 μl , 68 μmol , 1.5 eq) was added, resulting in an orange solution. The solution was layered with *n*-hexane (4 ml) and stored at -40°C overnight, resulting in the precipitation of an orange microcrystalline solid. The solid was filtered off, washed with a 4:1 *n*-hexane/DCM mixture (2x1 ml) and dried *in vacuo* (33.1 mg, 34.6 μmol , 76 %).

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.26 – 8.14 (m, 4H), 7.75 – 7.63 (m, 4H), 7.46 (s, 2H), 7.44 (s, 2H), 7.31 (t, $J = 2.1$ Hz, 2H), 6.93 (br s, 4H), 2.96 (s, 6H), 1.23 (s, 36H).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 183.2, 182.6, 152.2, 151.8, 146.1, 140.1, 134.1, 133.9, 133.5, 133.4, 128.9, 127.3, 126.9, 126.9, 121.3, 120.5, 109.3, 107.9, 37.8* (br, DMSO), 35.0, 31.5.

*This signal was identified as the DMSO shift by a ^1H - ^{13}C HSQC experiment.

$^{29}\text{Si NMR}$ (79 MHz, CDCl_3) No signal observed due to limited solubility and signal intensity

MS (LIFDI+): Due to the lability of binding DMSO, only the parent complex **1** was detected. $[\text{C}_{56}\text{H}_{54}\text{N}_2\text{O}_6\text{Si}]^+$, calcd.: 878.38, found: 878.38.

UV-vis (DCM): λ_{max} (ε) = 460 nm (8500 $\text{M}^{-1}\text{cm}^{-1}$).

IR (ATR) [cm^{-1}] $\tilde{\nu}$ = 3076 (w, $\tilde{\nu}_{\text{CH}}$), 3013 (w, $\tilde{\nu}_{\text{CH}}$), 2954 (w, $\tilde{\nu}_{\text{CH}}$), 2928 (w, $\tilde{\nu}_{\text{CH}}$), 2904 (w, $\tilde{\nu}_{\text{CH}}$), 2865 (w, $\tilde{\nu}_{\text{CH}}$), 1661 (m, $\tilde{\nu}_{\text{C=O}}$).

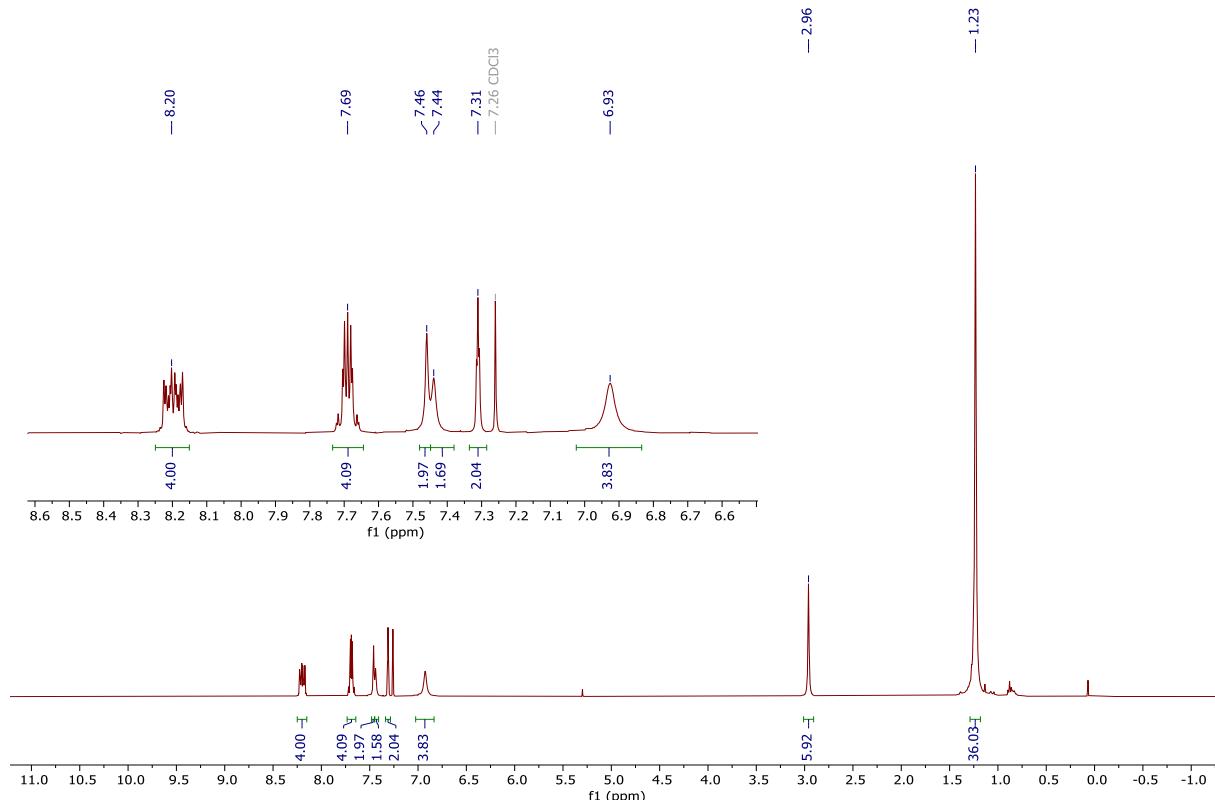


Figure S1.35. $^1\text{H NMR}$ spectrum (400 MHz, CDCl_3) of **1-DMSO**.

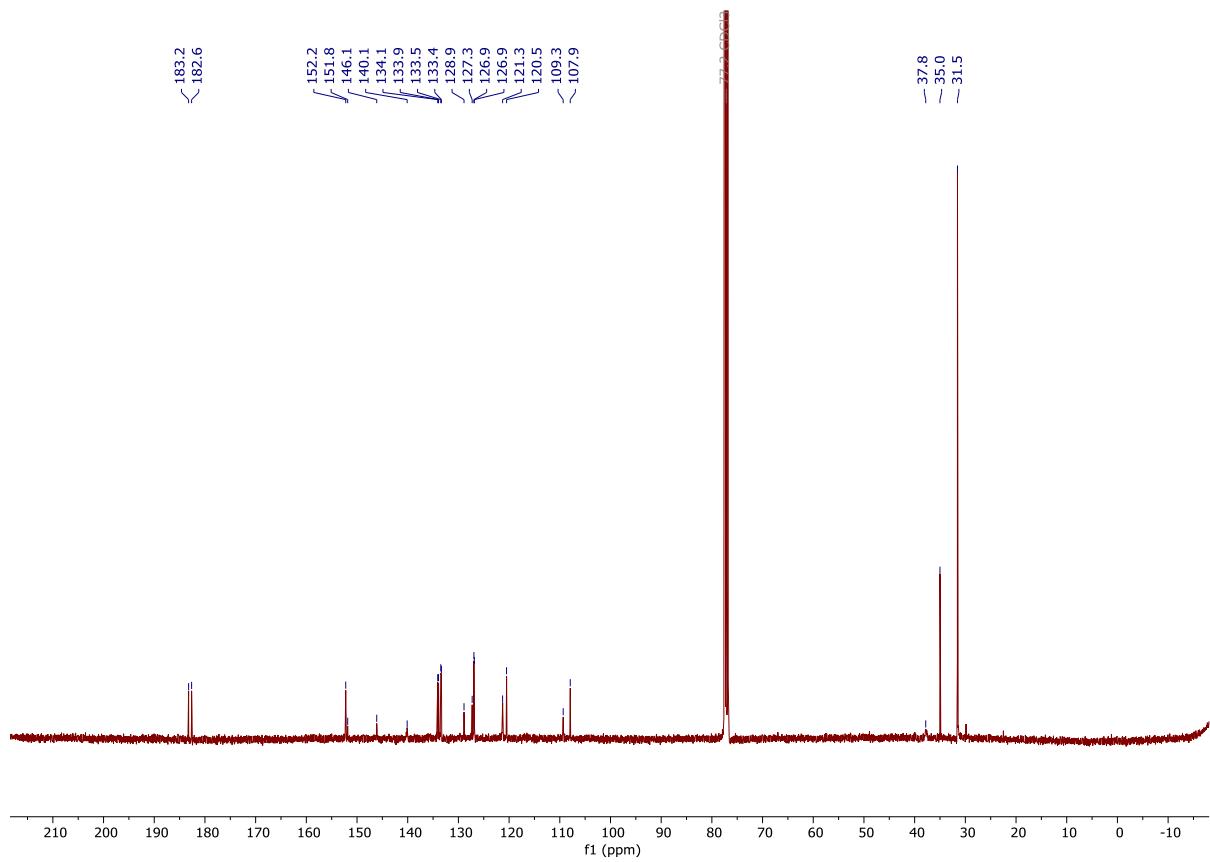
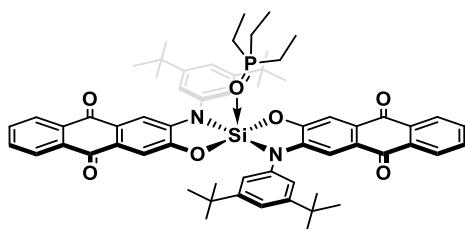


Figure S1.36. ¹³C NMR spectrum (101 MHz, CDCl₃) of **1-DMSO**.

1.14 1-OPEt₃



1 (40.0 mg, 45.5 µmol, 1.00 eq) was dissolved in DCM (1 ml). Et₃PO (5.1 mg, 46 µmol, 1.0 eq) was added, resulting in an orange solution. The solution was layered with *n*-hexane (4 ml) and stored at -40 °C overnight, resulting in the precipitation of an orange solid. The solid was filtered off, washed with a 2:1 *n*-hexane/DCM mixture (2x1 ml) and dried *in vacuo* (33.1 mg, 34.6 µmol, 76 %).

¹H NMR (400 MHz, CDCl₃) δ 8.22 – 8.12 (m, 4H), 7.71 – 7.60 (m, 4H), 7.30 (t, *J* = 1.8 Hz, 2H), 7.28 (s, 2H), 7.19 (s, 2H), 2.18 – 1.95 (m, 6H), 1.19 (overlapping: dt, *J* = 18.6, 7.7 Hz, 12H; br s, 36H; combined 48H).

¹³C NMR (101 MHz, CDCl₃) δ 183.5, 182.9, 153.1, 151.5, 147.2, 142.0, 134.2, 134.1, 133.2, 133.1, 127.9, 126.9, 126.8, 126.6, 122.4, 120.1, 107.8, 107.2, 34.9, 31.6, 17.4 (d, *J* = 64.6 Hz), 5.4 (d, *J* = 4.7 Hz).

³¹P NMR (162 MHz, CDCl₃) δ 80.8.

³¹P NMR (162 MHz, CD₂Cl₂) δ 81.6.

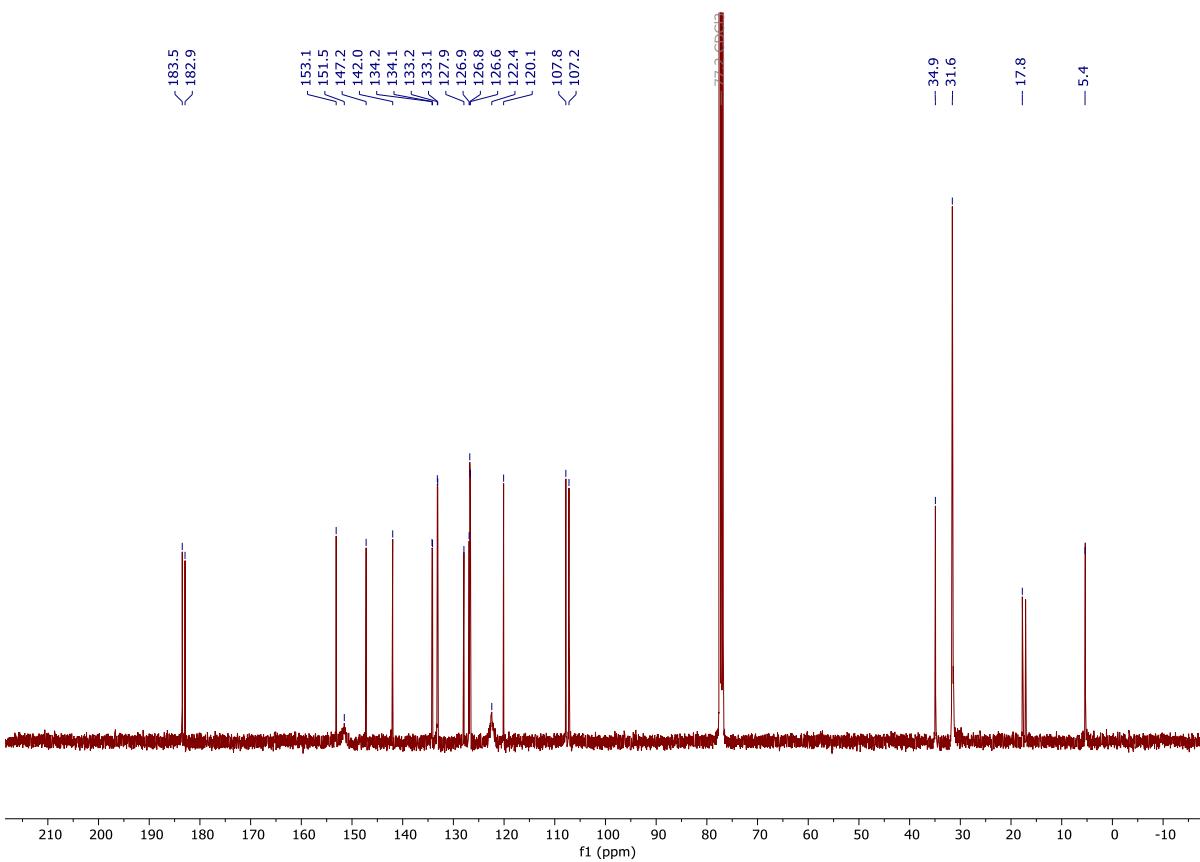
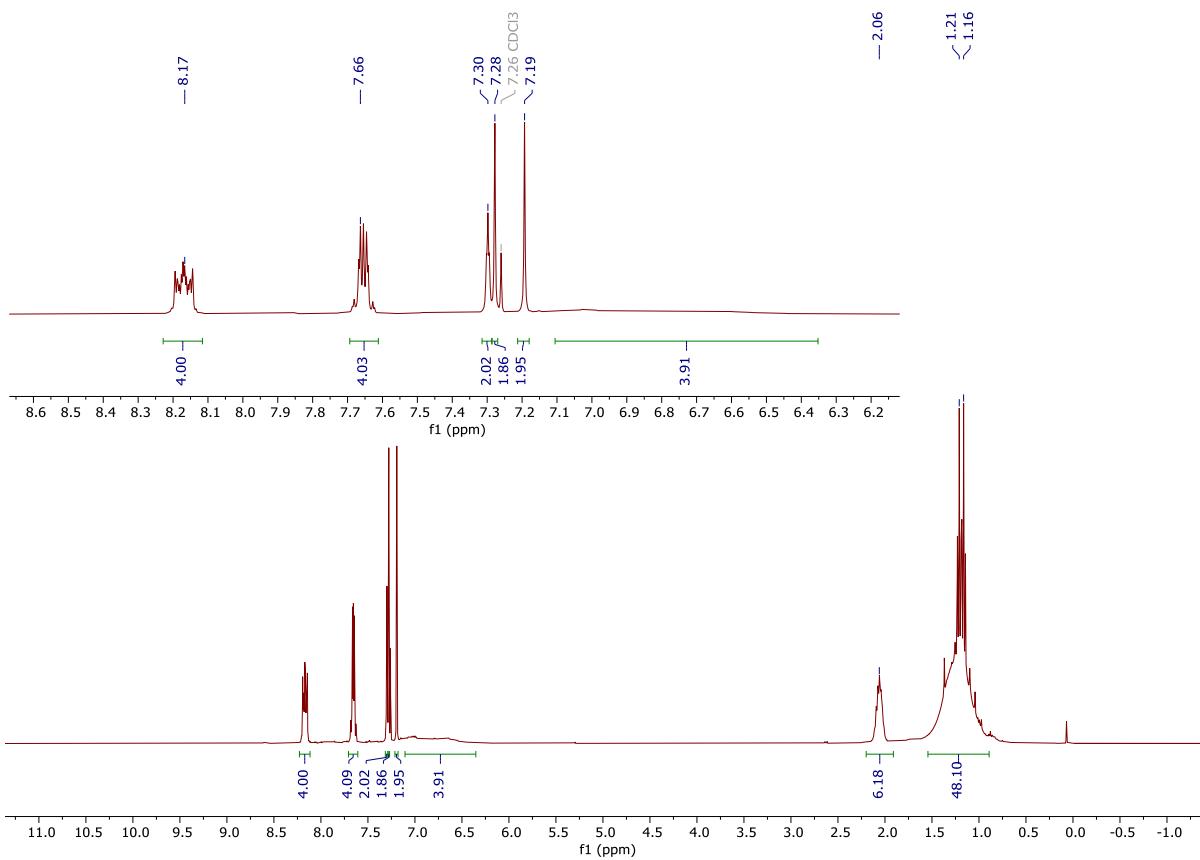
Note: The ³¹P NMR shift in DCM-d₂ is given for the evaluation of Lewis acidity by the Gutmann-Beckett method.

²⁹Si NMR (80 MHz, CDCl₃) δ -108.1.

MS (LIFDI+): Due to the lability of binding OPEt₃, only the parent complex **1** was detected. [C₅₆H₅₄N₂O₆Si]⁺, calcd.: 878.38, found: 878.39.

UV-vis (DCM): λ_{max} (ε) = 469 nm (9300 M⁻¹cm⁻¹).

IR (ATR) [cm⁻¹] $\tilde{\nu}$ = 3066 (w, $\tilde{\nu}_{\text{CH}}$), 2959 (m, $\tilde{\nu}_{\text{CH}}$), 2906 (w, $\tilde{\nu}_{\text{CH}}$), 2867 (w, $\tilde{\nu}_{\text{CH}}$), 1663 (m, $\tilde{\nu}_{\text{C=O}}$).



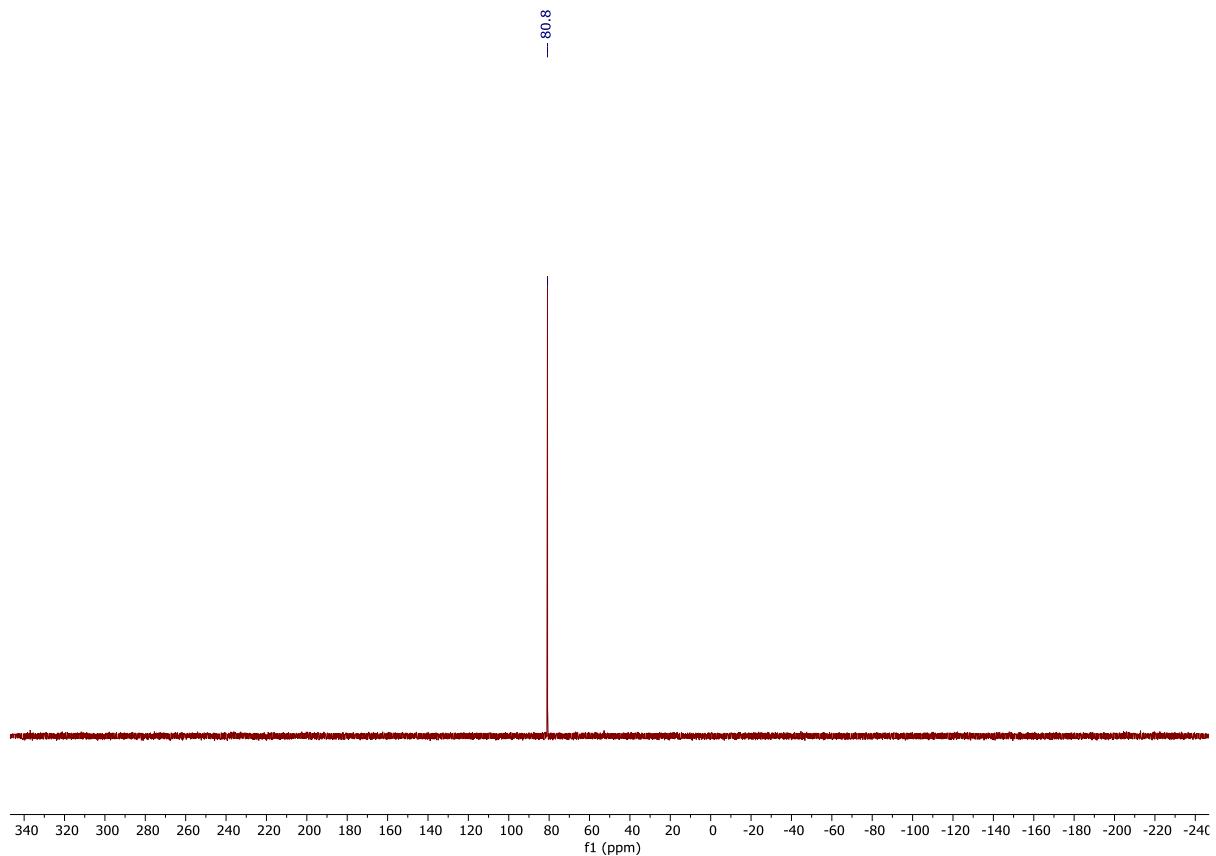


Figure S1.39. ^{31}P NMR spectrum (162 MHz, CDCl_3) of **1-OPEt₃**.

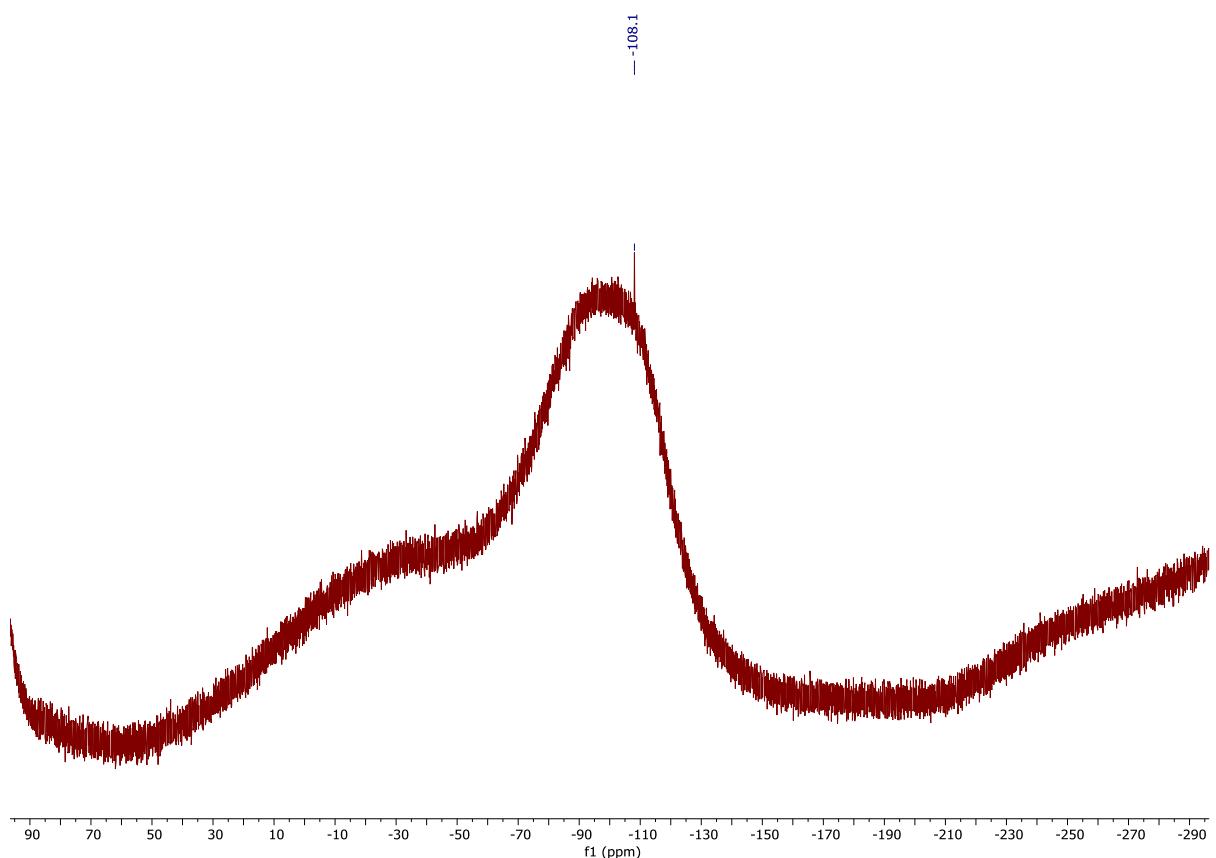


Figure S1.40. ^{29}Si NMR spectrum (80 MHz, CDCl_3) of **1-OPEt₃**.

1.15 1-DIBA



1 (40.0 mg, 45.5 μmol , 1.00 eq) was dissolved in DCM (1 ml). DIBA (9.3 mg, 46 μmol , 1.0 eq) was added, resulting in an orange solution. *n*-hexane (4 ml) was added and the mixture stored at -40°C for 2h, resulting in the precipitation of an orange microcrystalline solid. The solid was filtered off, washed with a 10:1 *n*-hexane/DCM mixture (4x1 ml) and dried *in vacuo* (34.9 mg, 34.6 μmol , 71 %).

$^1\text{H NMR}$ (600 MHz, CDCl_3) δ 8.25 – 8.15 (m, 4H), 7.72 – 7.65 (m, 4H), 7.36 (br, 7H), 7.25 (m, 4H), 6.80 (br s, 4H), 4.03 (br s, 1H, *iPr-H*), 3.74 (br s, 1H, *iPr-H*), 1.59 (br s, 6H, *iPr-CH_3*), 1.33 – 1.10 (m, 42H, overlapping *tBu/iPr-CH_3*).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 183.3, 182.8, 171.8 (amide C=O), 152.5, 151.9, 145.7, 140.2, 134.2, 134.0, 133.4, 133.2, 130.6, 128.7, 128.3, 127.1, 126.9, 126.8, 126.6, 121.1, 120.2, 109.3, 107.6, 53.6, 49.0, 34.9, 31.5, 20.7.

Note: Signals at 53.6 ppm and 49.0 ppm were assigned by a ^1H - ^{13}C HSQC experiment.

$^{29}\text{Si NMR}$ (80 MHz, CDCl_3) No signal observed due to limited solubility and signal intensity

MS (LIFDI+): Due to the lability of binding DIBA, only the parent complex **1** was detected. $[\text{C}_{56}\text{H}_{54}\text{N}_2\text{O}_6\text{Si}]^+$, calcd.: 878.38, found: 878.39. The mass of DIBA was also detected. $[\text{C}_{13}\text{H}_{19}\text{NO}]^+$, calcd.: 205.15, found: 205.15.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon) = 467 \text{ nm} (7500 \text{ M}^{-1}\text{cm}^{-1})$.

IR (ATR) [cm^{-1}] $\tilde{\nu} = 3067$ (w, $\tilde{\nu}_{\text{CH}}$), 2953 (m, $\tilde{\nu}_{\text{CH}}$), 2866 (w, $\tilde{\nu}_{\text{CH}}$), 1661 (m, $\tilde{\nu}_{\text{C=O}}$).

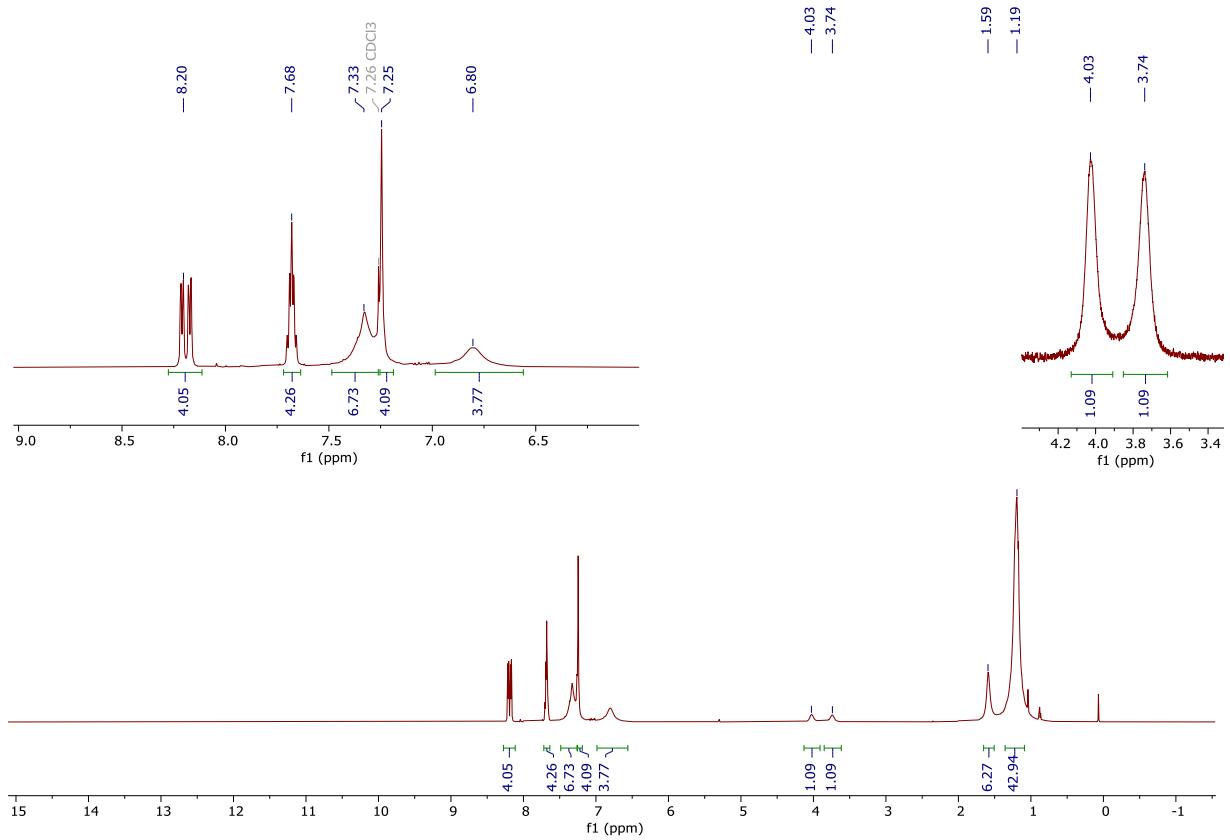


Figure S1.41. ^1H NMR spectrum (600 MHz, CDCl_3) of **1-DIBA**.

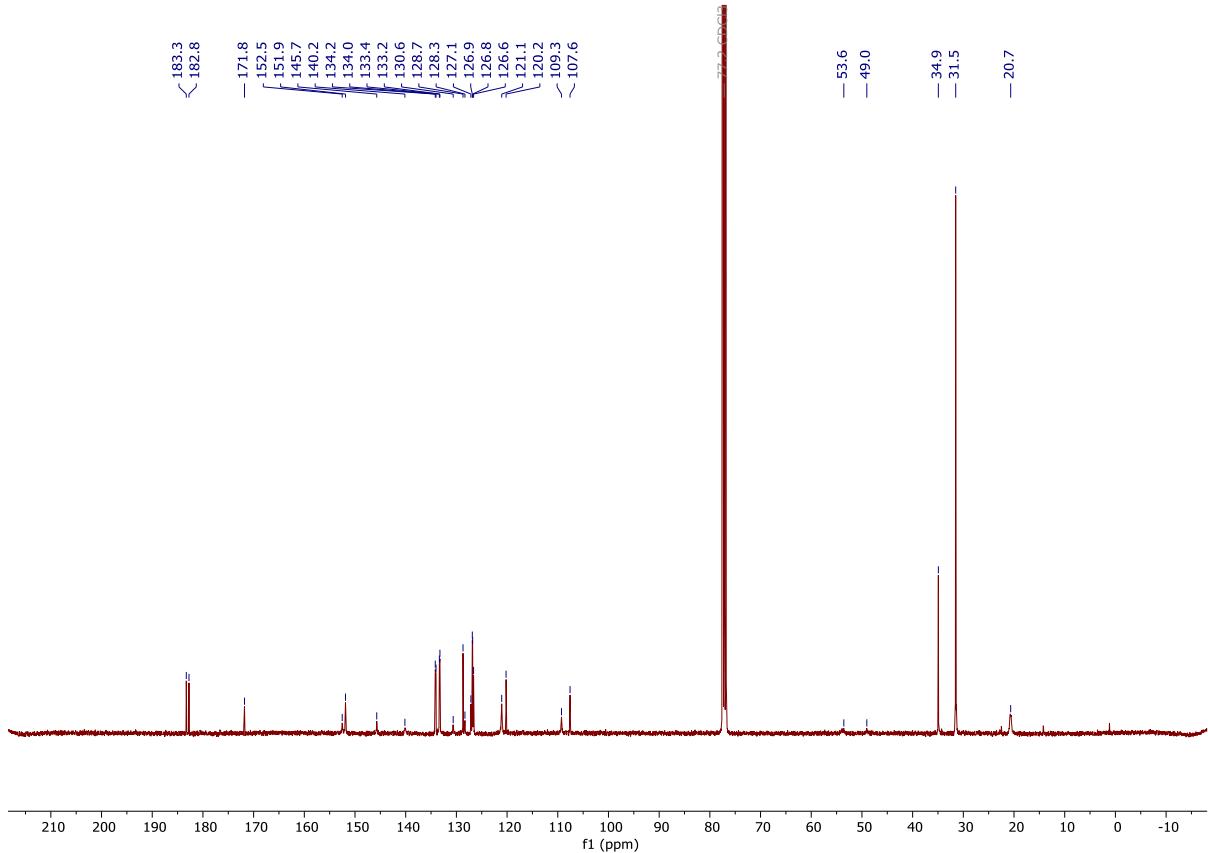
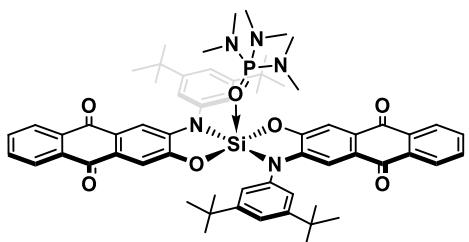


Figure S1.42. ^{13}C NMR spectrum (101 MHz, CDCl_3) of **1-DIBA**.

1.16 1-HMPA



1 (40.0 mg, 45.5 μmol , 1.00 eq) was dissolved in DCM (1 ml). HMPA (9.8 mg, 55 μmol , 1.2 eq) was added, resulting in an orange solution. The solution was layered with *n*-pentane (4 ml) and stored at -40°C for 1 h, resulting in the precipitation of an orange solid. The solid was filtered off, washed with a 2:1 *n*-pentane/DCM mixture (2x1 ml) and dried *in vacuo* (33.1 mg, 31.2 μmol , 69 %).

$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.22 – 8.09 (m, 4H), 7.69 – 7.59 (m, 4H), 7.29 (t, $J = 1.8\text{ Hz}$, 2H), 7.27 (s, 2H), 7.16 (s, 2H), 6.97 (br, 4H), 2.63 (d, $J = 10.3\text{ Hz}$, 18H), 1.26 (br s, 36H).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 183.5, 183.1, 153.8, 151.2, 147.5, 142.4, 134.3, 134.2, 133.1, 133.0, 127.5, 126.8, 126.7, 126.6, 122.7, 119.7, 107.6, 106.9, 36.9 (d, $J = 5.1\text{ Hz}$), 34.9, 31.6.

$^{31}\text{P NMR}$ (162 MHz, CDCl_3) δ 22.1.

$^{29}\text{Si NMR}$ (79 MHz, CDCl_3) δ -111.7.

MS (LIFDI+): Due to the instability of **1-HMPA** under mass spectrometry conditions, only the parent complex **1** was detected. $[\text{C}_{56}\text{H}_{54}\text{N}_2\text{O}_6\text{Si}]^+$, calcd.: 878.38, found: 878.34.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon) = 474\text{ nm}$ ($8200\text{ M}^{-1}\text{cm}^{-1}$).

IR (ATR) [cm^{-1}] $\tilde{\nu} = 3066$ (w, $\tilde{\nu}_{\text{CH}}$), 2956 (m, $\tilde{\nu}_{\text{CH}}$), 2905 (w, $\tilde{\nu}_{\text{CH}}$), 2865 (w, $\tilde{\nu}_{\text{CH}}$), 2824 (w, $\tilde{\nu}_{\text{CH}}$), 1663 (m, $\tilde{\nu}_{\text{C=O}}$).

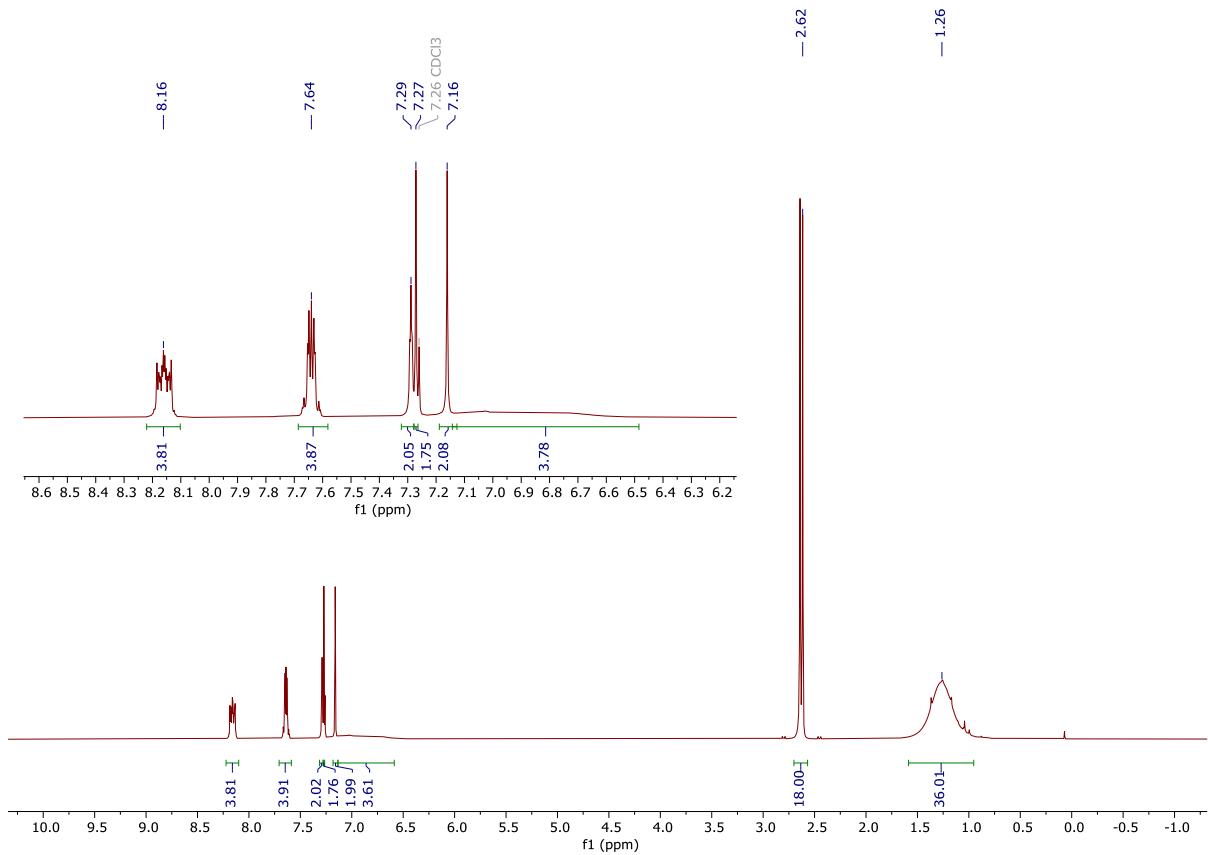


Figure S1.43. ^1H NMR spectrum (400 MHz, CDCl_3) of 1-HMPA.

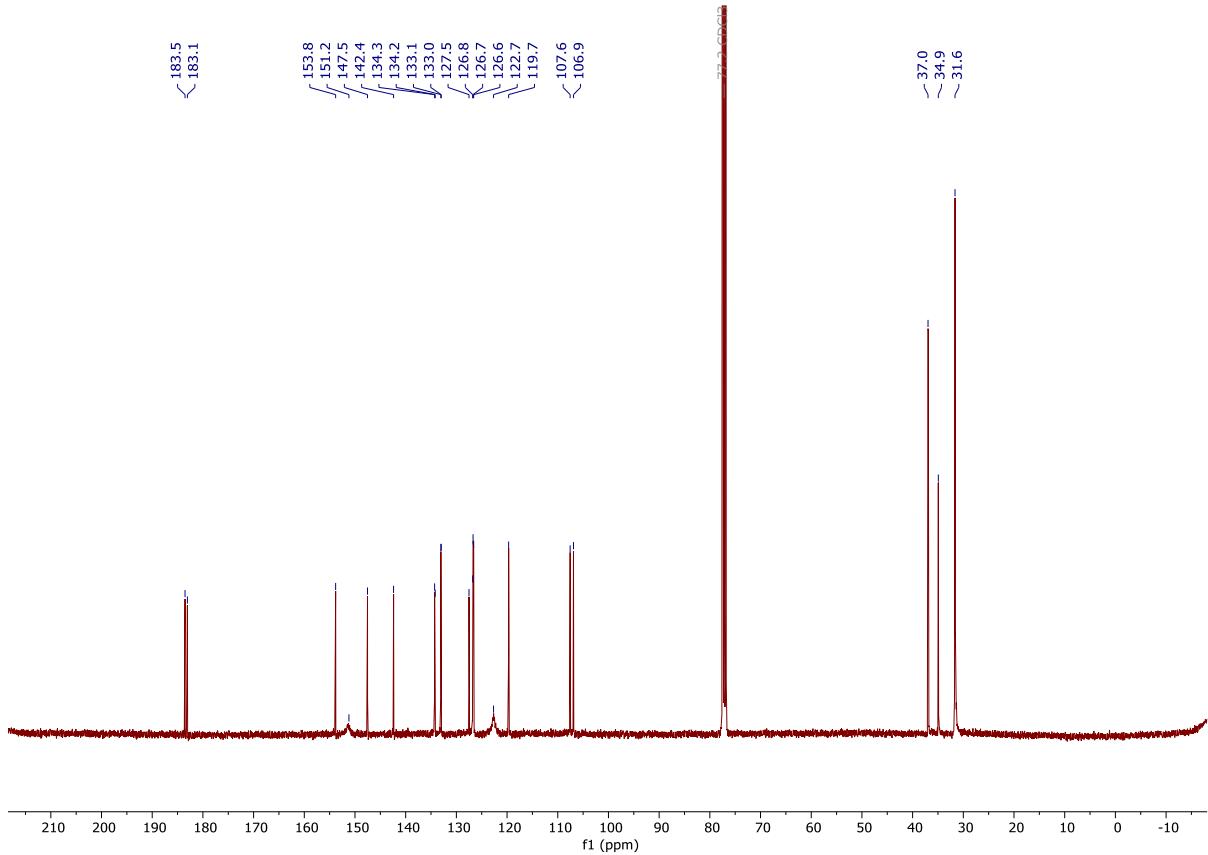


Figure S1.44. ^{13}C NMR spectrum (101 MHz, CDCl_3) of 1-HMPA.

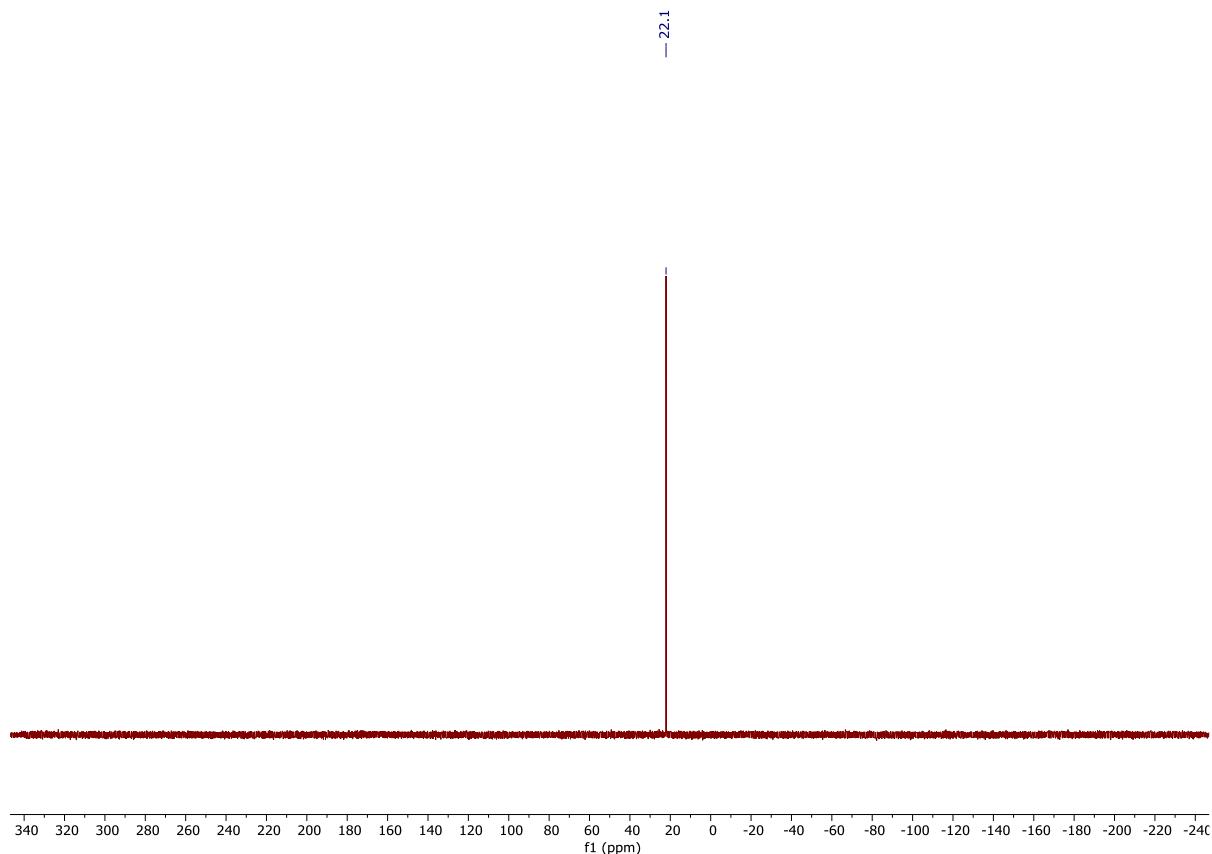


Figure S1.45. ^{31}P NMR spectrum (162 MHz, CDCl_3) of **1-HMPA**.

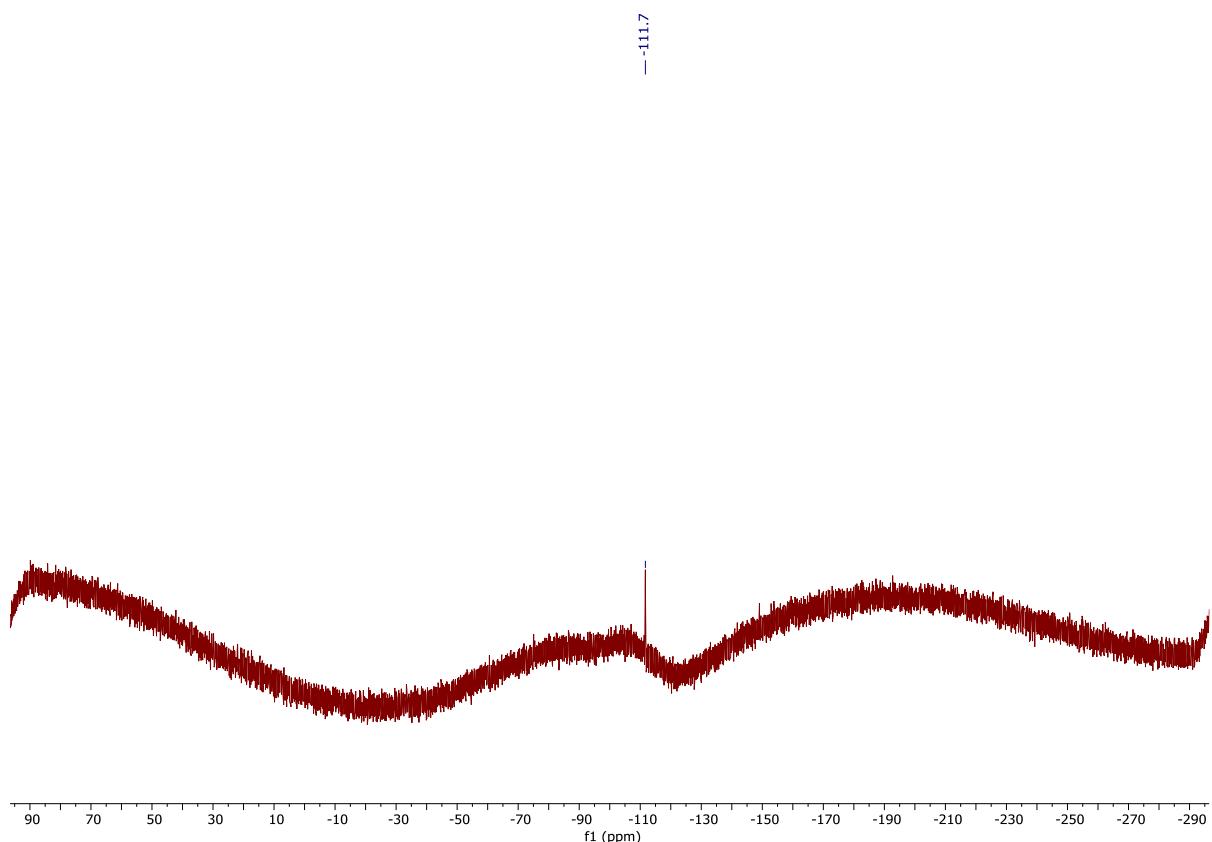
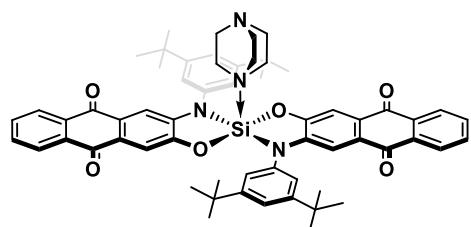


Figure S1.46. ^{29}Si NMR spectrum (79 MHz, CDCl_3) of **1-HMPA**.

1.17 1-DABCO



1 (40.0 mg, 45.5 μmol , 1.0 eq) was dissolved in DCM (1 ml). DABCO (6.6 mg, 59 μmol , 1.3 eq) was added, resulting in an orange solution. *n*-pentane (4 ml) was added and the solution was stored at -40°C overnight, resulting in the precipitation of an orange solid. The solid was filtered off, washed with *n*-pentane (3x1 ml) and dried at atmospheric pressure for 4 hours (35.2 mg, 35.5 μmol , 78 %).

*Note: Drying in vacuo leads to decomposition of the product, likely related to sublimation and loss of DABCO. Therefore, drying was not complete and small amounts of *n*-pentane are still present in the sample.*

$^1\text{H NMR}$ (400 MHz, CD_2Cl_2) δ 8.22 – 8.10 (m, 4H), 7.76 – 7.65 (m, 4H), 7.44 (s, 2H), 7.36 (s, 2H), 7.27 (s, 2H), 7.00 (s, 4H), 3.07 (s, 12H), 1.32 (s, 36H).

$^{13}\text{C NMR}$ (101 MHz, CD_2Cl_2) δ 183.1, 182.6, 152.6, 152.3, 147.1, 141.8, 134.4, 134.2, 133.7, 133.5, 128.7, 127.5, 126.9, 126.8, 122.9, 120.5, 108.7, 107.8, 47.4, 35.3, 31.6.

$^{29}\text{Si NMR}$ (80 MHz, CD_2Cl_2) δ -101.2.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon) = 449 \text{ nm} (8200 \text{ M}^{-1}\text{cm}^{-1})$.

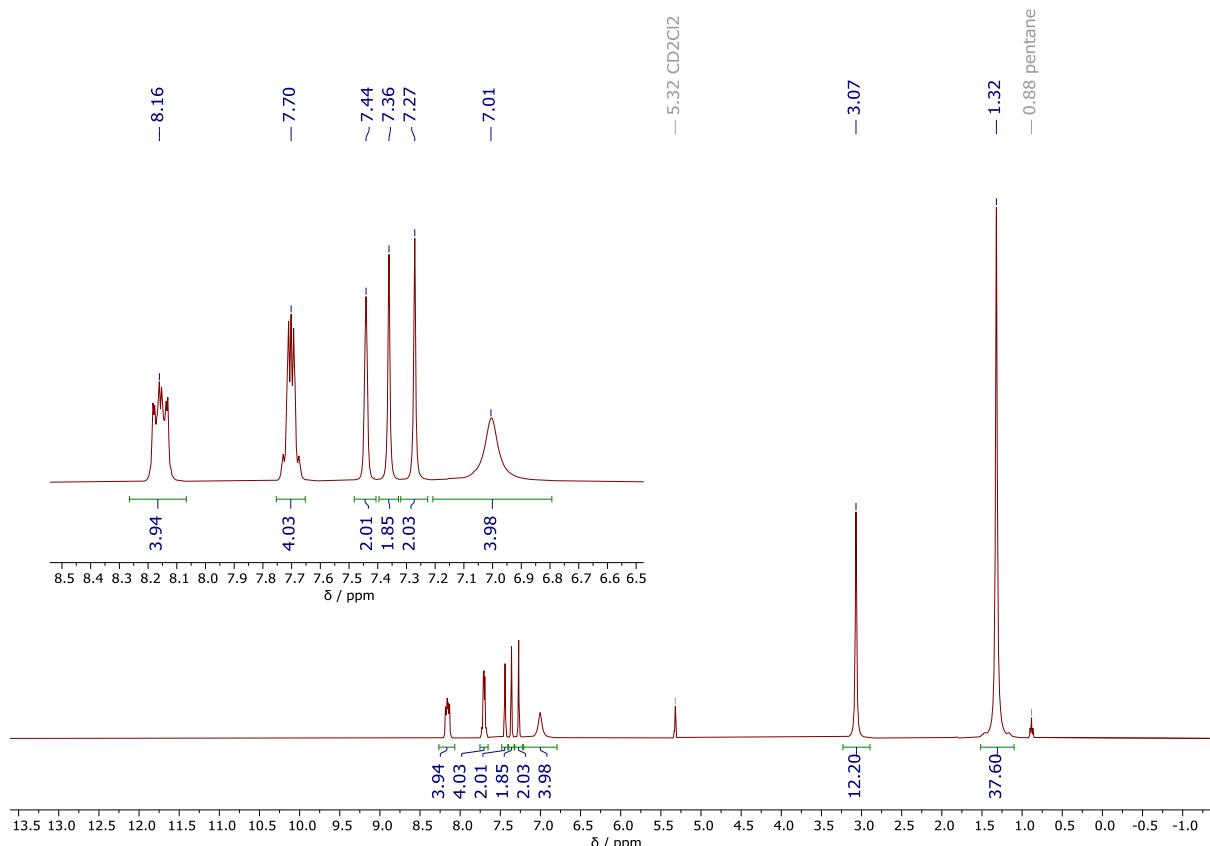


Figure S1.47. $^1\text{H NMR}$ spectrum (400 MHz, CD_2Cl_2) of **1-DABCO**.

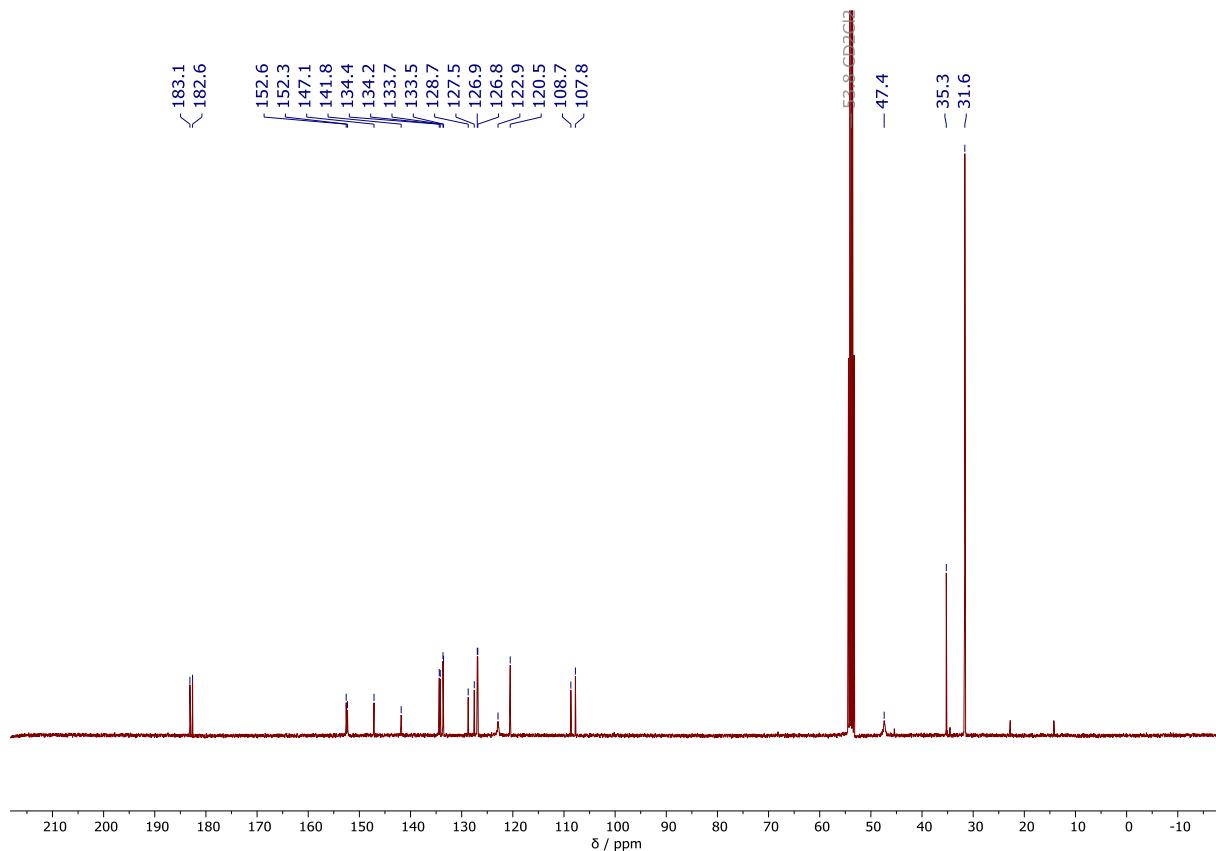


Figure S1.48. ^{13}C NMR spectrum (101 MHz, CD₂Cl₂) of **1-DABCO**.

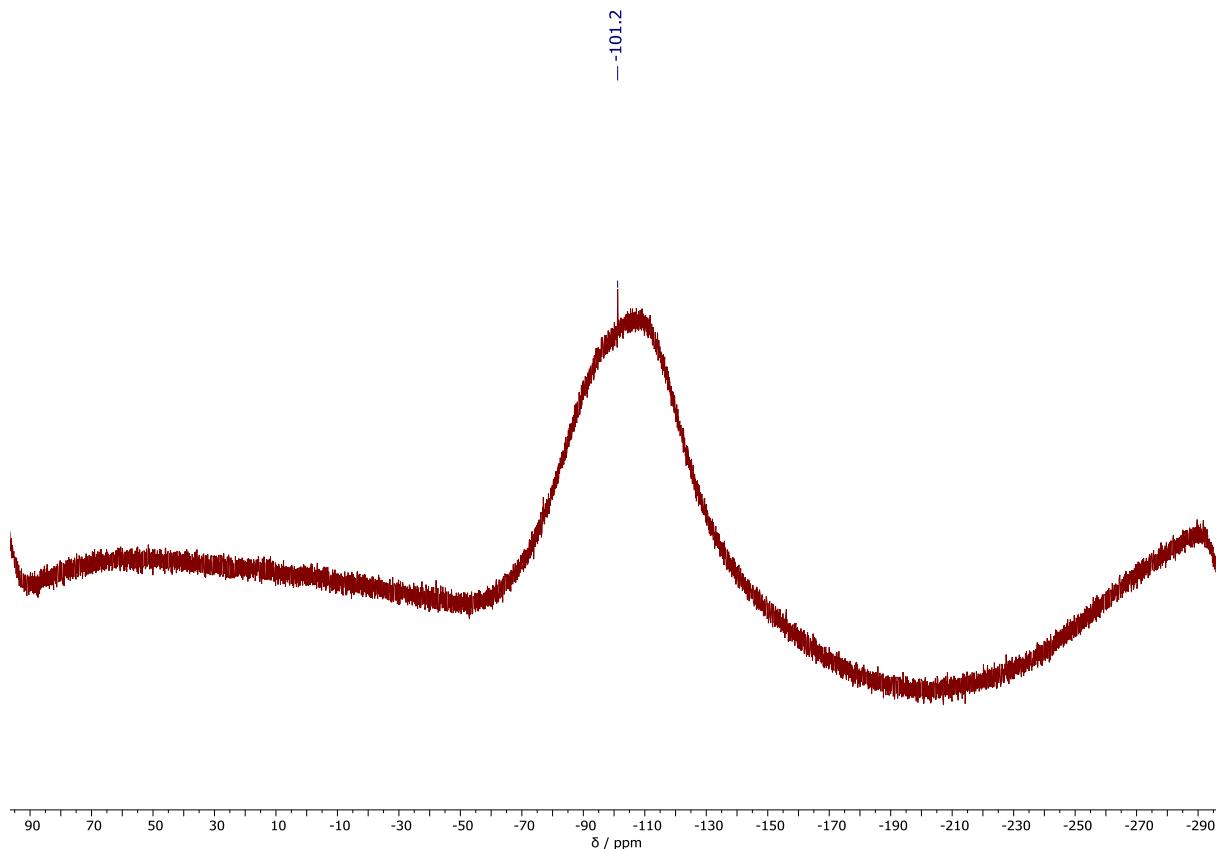
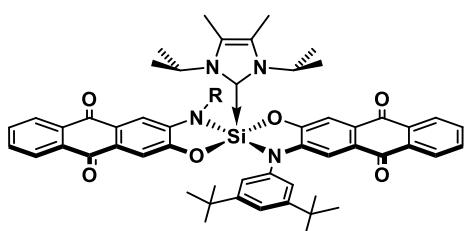


Figure S1.49. ^{29}Si NMR spectrum (80 MHz, CD₂Cl₂) of **1-DABCO**.

1.18 1-ⁱPrNHC



1 (40.0 mg, 45.5 μmol , 1.00 eq) was dissolved in DCM (2 ml). 4,5-Me₂-1,3-(iPr)₂-NHC (8.2 mg, 45.5 μmol , 1.00 eq) was added, resulting in an orange solution. Pentane (4 ml) was added, giving a clear solution, which was stored at -40°C overnight. The formed crystals were filtered off and washed with a 5:1 n-pentane/DCM mixture (3 x 2 ml). The product was isolated as an orange solid after drying *in vacuo* (33.6 mg, 31.7 μmol , 70 %).

¹H NMR (400 MHz, CD₂Cl₂) δ 8.25 – 8.06 (m, 4H), 7.71 – 7.62 (m, 4H), 7.38 (s, 2H), 7.22 (overlapping: s, 2H; br, 2H, combined 4H), 7.04 (s, 2H), 6.65 (br, 2H), 5.37 (sept, $J = 7.0$ Hz, 2H), 2.25 (s, 6H), 1.54 (d, $J = 6.9$ Hz, 6H), 1.48 (d, $J = 7.0$ Hz, 6H), 1.42 – 0.79 (br, 36H).

Note: Signals at 1.54, 1.48, and signals of residual pentane are overlapping with the broad signal of the tBu groups to give an overall integral of 53H.

¹³C NMR (101 MHz, CD₂Cl₂) δ 183.2, 182.9, 155.7, 155.4, 152.3, 147.1, 142.2, 134.5, 134.4, 133.3, 127.8, 127.5, 126.8, 126.8, 126.7, 123.2, 122.0, 120.6, 108.3, 107.1, 52.5, 35.1, 31.5, 22.8, 21.6, 10.9.

²⁹Si NMR (80 MHz, CD₂Cl₂) δ -102.3.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon) = 468 \text{ nm} (7000 \text{ M}^{-1}\text{cm}^{-1})$.

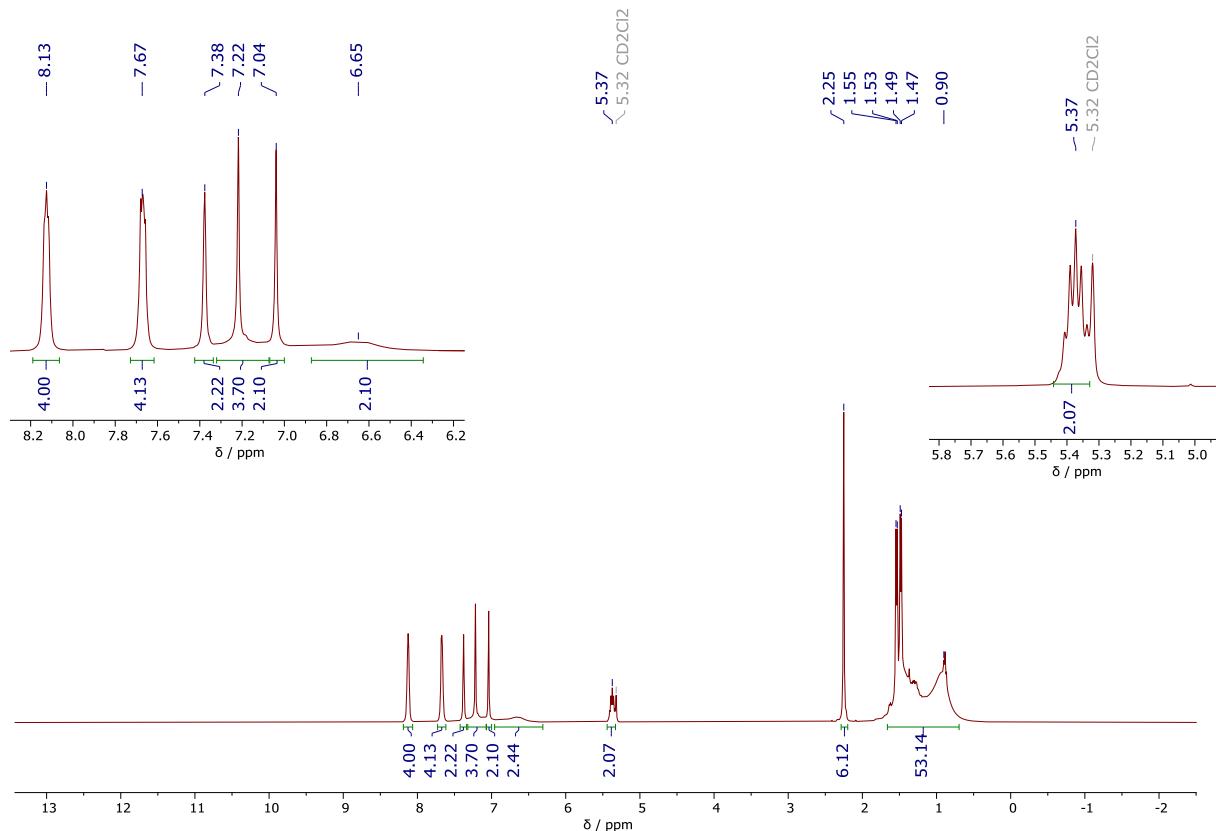


Figure S1.50. ¹H NMR spectrum (400 MHz, CD₂Cl₂) of **1-ⁱPrNHC**.

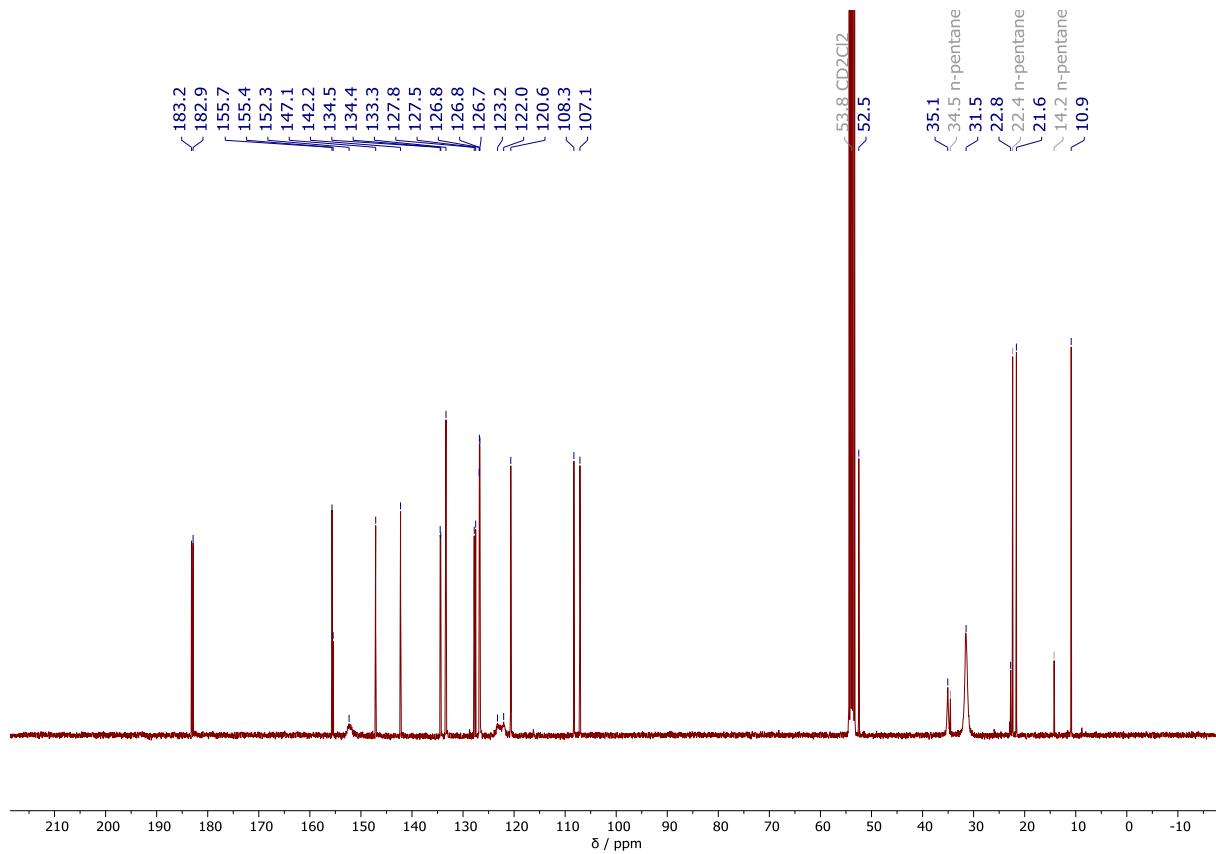


Figure S1.51. ^1H NMR spectrum (101 MHz, CD_2Cl_2) of $\mathbf{1-\text{i}PrNHC}$.

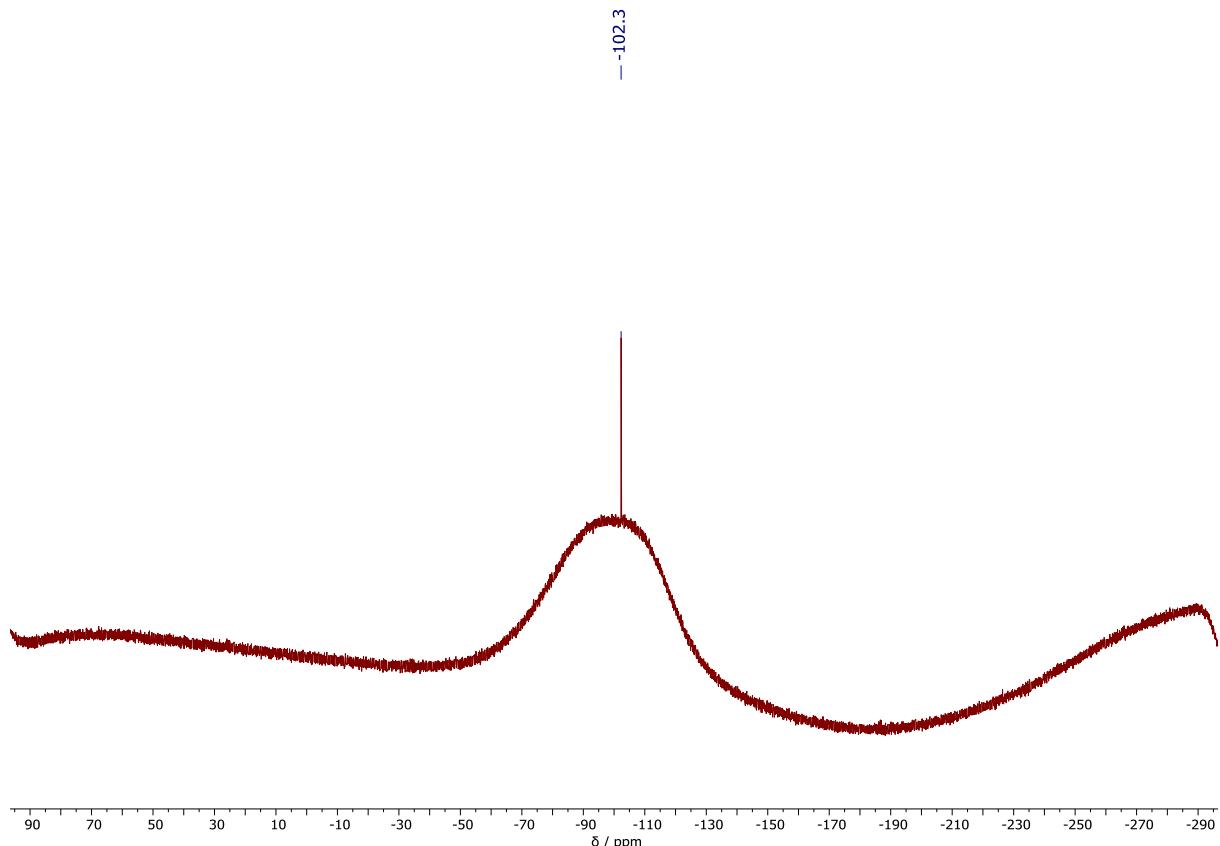
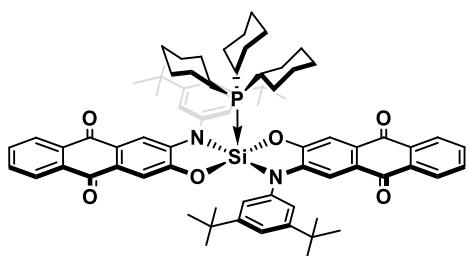


Figure S1.52. ^{29}Si NMR spectrum (80 MHz, CD_2Cl_2) of $\mathbf{1-\text{i}PrNHC}$.

1.19 1-*PCy*₃



1 (40.0 mg, 45.5 µmol, 1.00 eq) was suspended in DCM (1 ml). Tricyclohexylphosphine (12.8 mg, 45.5 µmol, 1.00 eq) was added, resulting in an orange solution. The solution was layered with *n*-pentane and stored at -40 °C for 7 days, resulting in the formation of orange crystals suitable for scXRD. The supernatant solution was decanted off and the orange solid dried *in vacuo* (26.3 mg, 24.3 µmol, 53 %).

¹H NMR (400 MHz, CD₂Cl₂) δ 8.27 – 8.07 (m, 4H), 7.75 – 7.67 (m, 4H), 7.45 (br, 4H), 7.34 (s, 2H), 6.97 (br, 4H), 2.25 – 0.79 (m, 69H, cyclohexyl, 33H, and *tBu*, 36H).

*Note: The dynamics of **1-PCy**₃ result in line broadening in the ¹H and ¹³C NMR spectra. Due to the broadening, one expected signal is not observed in the ¹³C NMR spectrum.*

¹³C NMR (101 MHz, CD₂Cl₂) δ 183.0, 182.5, 152.7, 145.5, 140.0, 134.3, 134.1, 133.8, 133.7, 129.3, 127.9, 127.0, 126.9, 121.5, 120.9, 109.6, 108.1, 35.1, 32.5 (d, *J* = 10.1 Hz), 31.4, 29.7, 27.9 (d, *J* = 9.8 Hz), 26.8.

²⁹Si NMR (80 MHz, CD₂Cl₂) No signal observed, likely due to line broadening resulting from dynamic binding of PCy₃.

³¹P NMR (162 MHz, CD₂Cl₂) δ 8.5.

UV-vis (DCM): λ_{max} (ε) = 461 nm (8000 M⁻¹cm⁻¹).

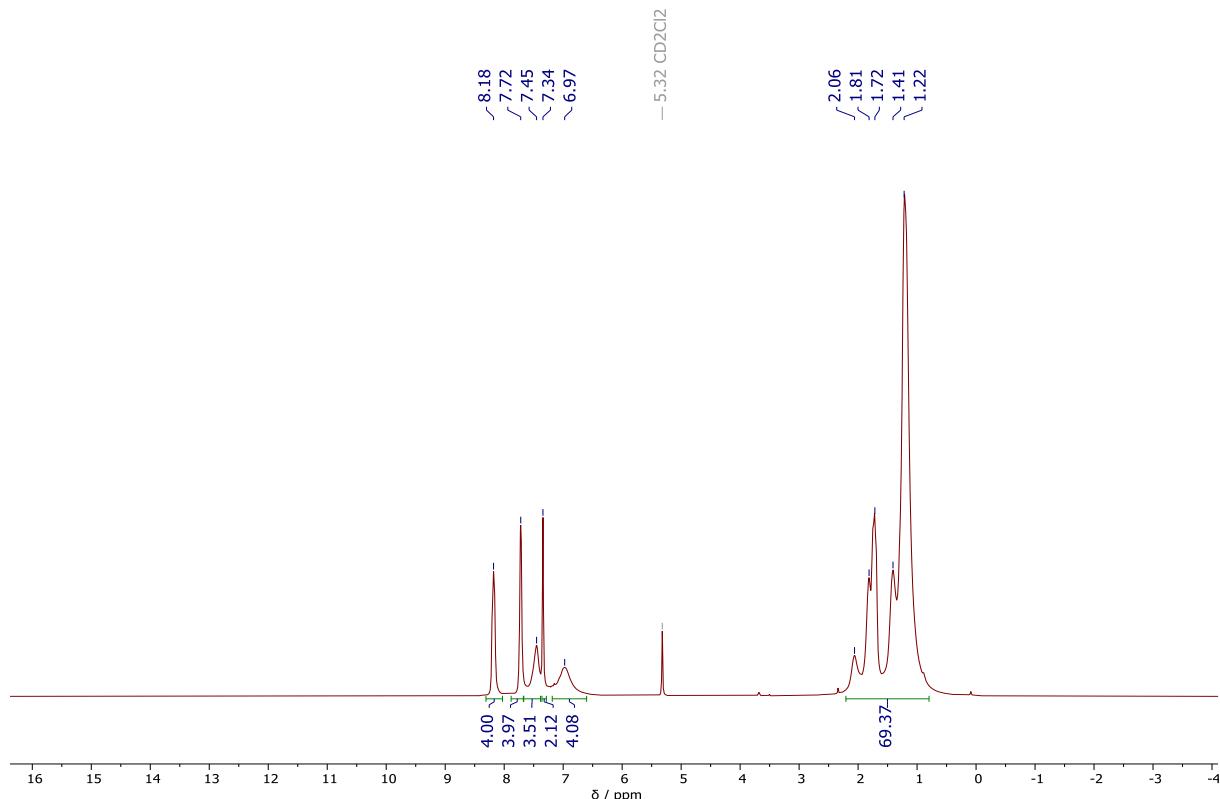


Figure S1.53. ^1H NMR spectrum (400 MHz, CD_2Cl_2) of **1-PCy₃**.

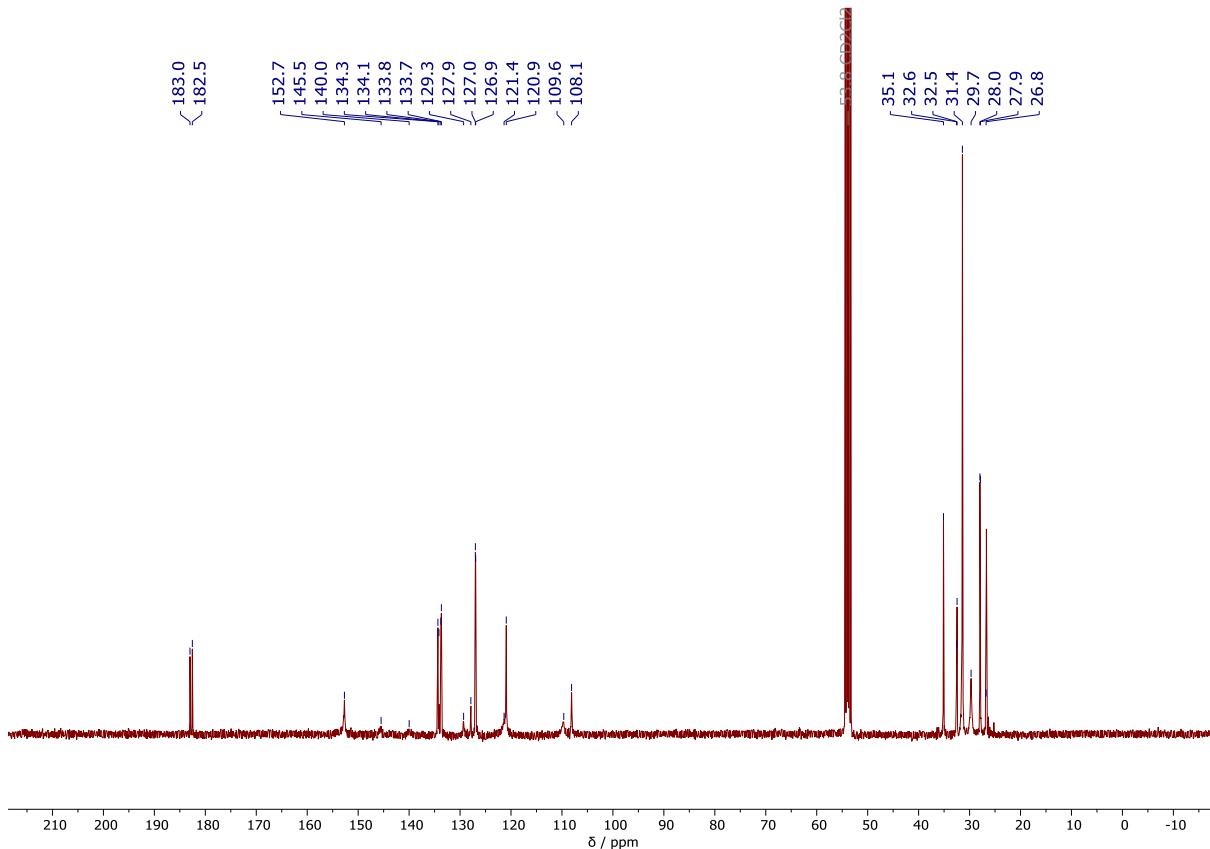


Figure S1.54. ^{13}C NMR spectrum (101 MHz, CD_2Cl_2) of **1-PCy₃**.

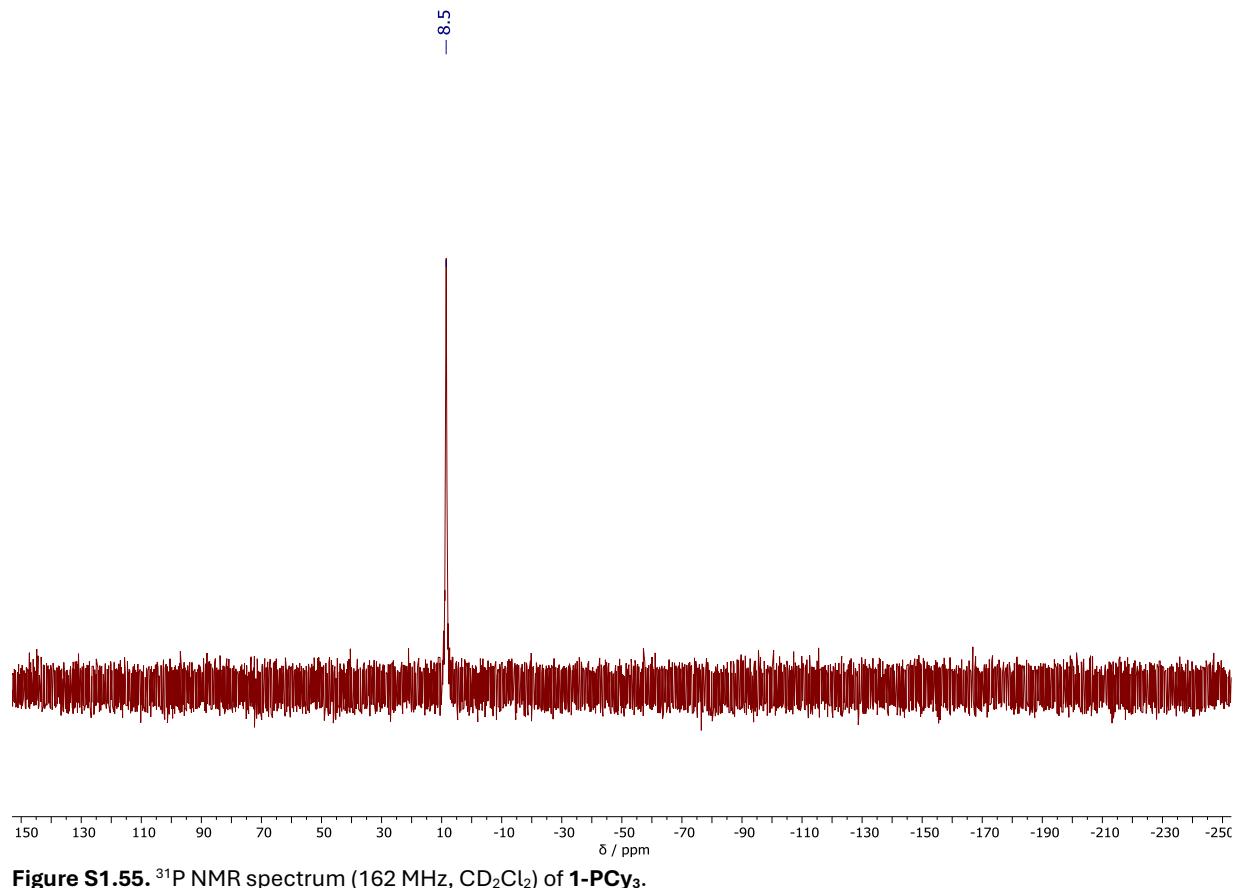
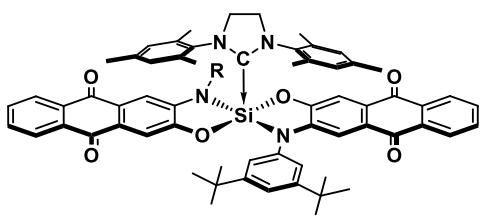


Figure S1.55. ^{31}P NMR spectrum (162 MHz, CD_2Cl_2) of **1-PCy₃**.

1.20 1-SIMes



1 (40.0 mg, 45.5 μmol , 1.00 eq) was dissolved in DCM (1.5 ml). 1,3-Dimesityl-imidazol-4,5-dihydro-2H-pyran-2-ylidene (SIMes) (13.9 mg, 45.5 μmol , 1.00 eq) was added, resulting in a dark orange solution. The solution was layered with *n*-pentane (4 ml) and stored at -40°C for 4 days. The supernatant was decanted from the formed crystals, which were washed with a 5:1 *n*-pentane/DCM mixture (2 x 2 ml). The obtained orange solid was dried in vacuo (27.5 mg, 23.2 μmol , 51 %).

$^1\text{H NMR}$ (600 MHz, CD_2Cl_2) δ 8.19 – 8.15 (m, 2H), 8.14 – 8.10 (m, 2H), 7.73 – 7.65 (m, 4H), 7.23 (s, 2H), 7.21 (s, 2H), 7.11 (s, 2H), 7.03 (s, 2H), 6.82 (s, 2H), 6.15 (s, 2H), 6.13 (s, 2H), 4.27 – 3.94 (m, 4H), 2.49 (s, 6H), 2.25 (s, 6H), 1.90 (s, 6H), 1.56 (s, 18H), 0.63 (s, 18H).

$^{13}\text{C NMR}$ (101 MHz, CD_2Cl_2) δ 183.0, 182.8, 181.8, 155.8, 152.1, 150.7, 146.1, 142.6, 140.2, 136.5, 136.4, 134.6, 134.3, 133.3, 133.3, 132.4, 129.4, 128.9, 127.2, 127.1, 126.8, 126.7, 122.5, 122.3, 119.9, 108.2, 106.8, 51.3, 35.5, 34.5, 32.0, 30.8, 20.8, 18.6, 18.1.

$^{29}\text{Si NMR}$ (80 MHz, CH_2Cl_2) δ -102.3.

UV-vis (DCM): $\lambda_{\text{max}} (\varepsilon) = 472 \text{ nm} (7000 \text{ M}^{-1}\text{cm}^{-1})$.

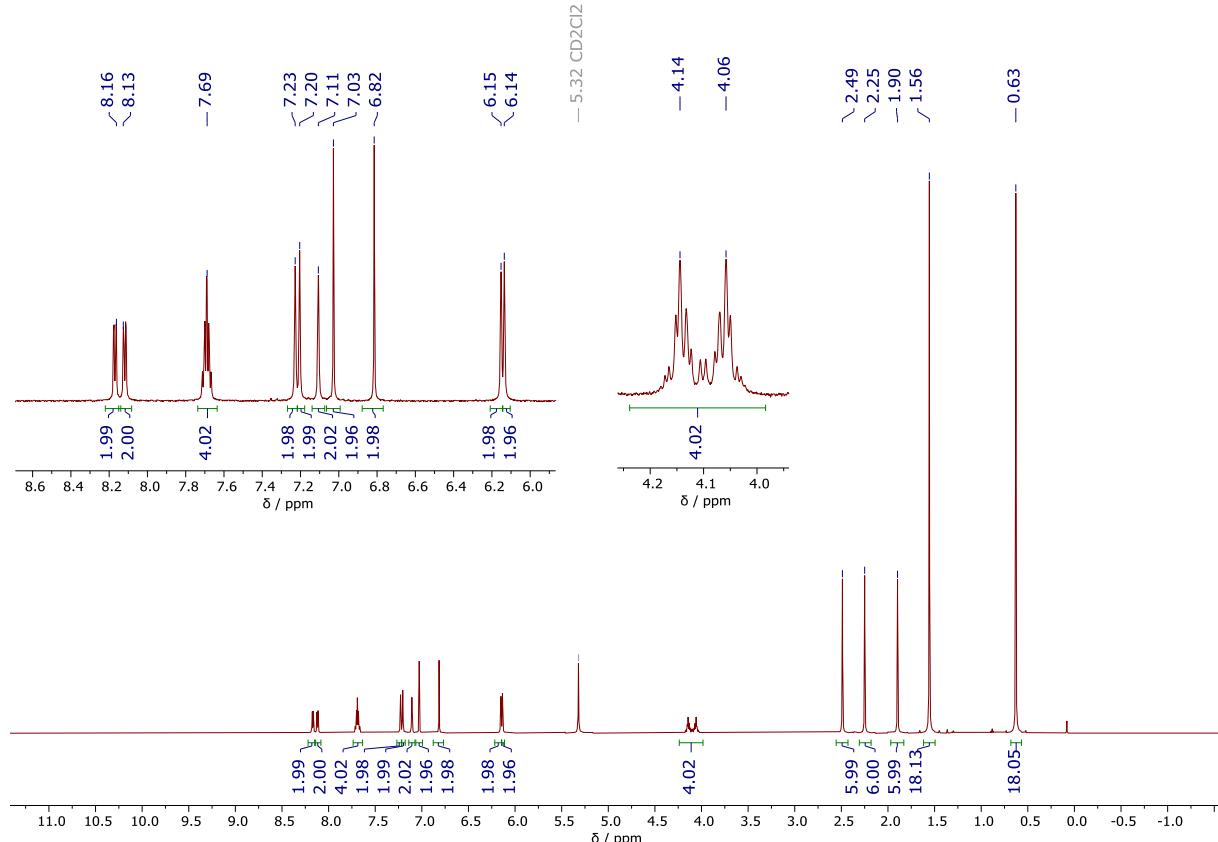
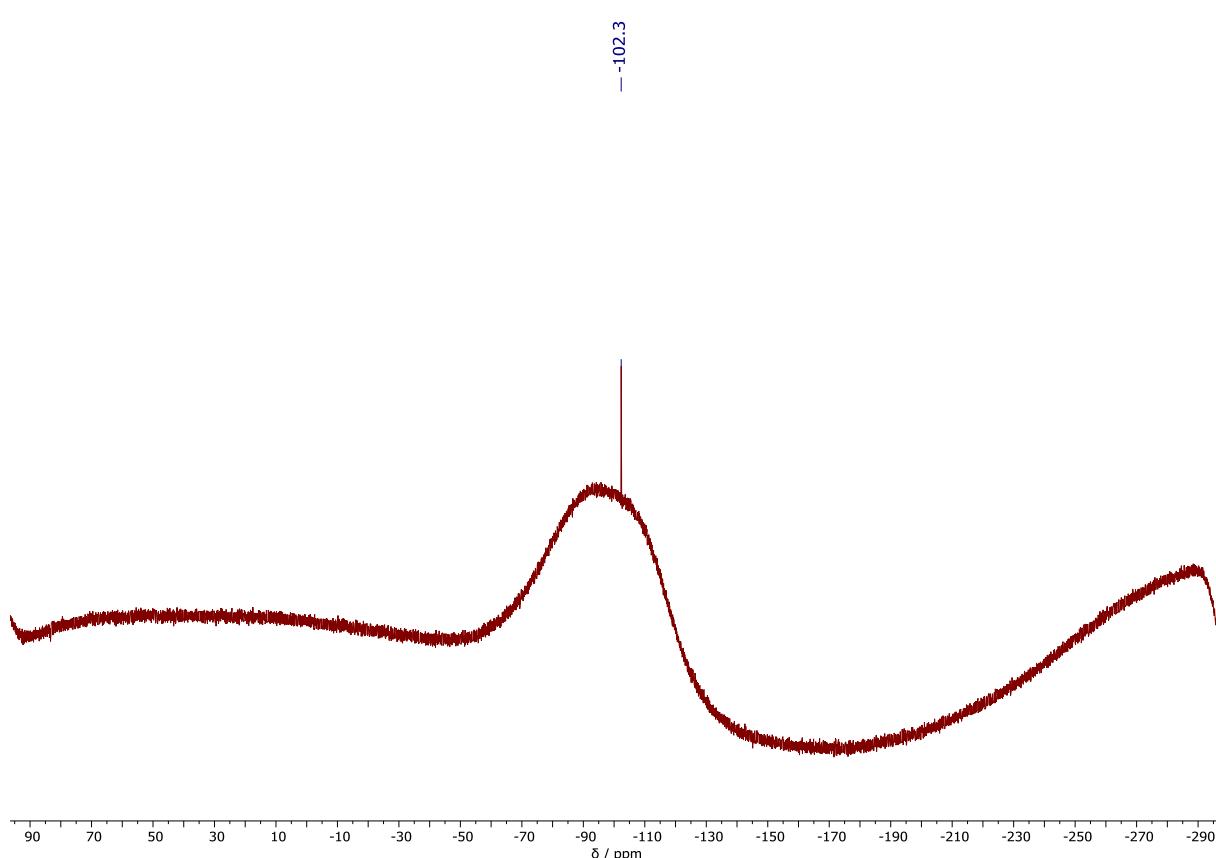
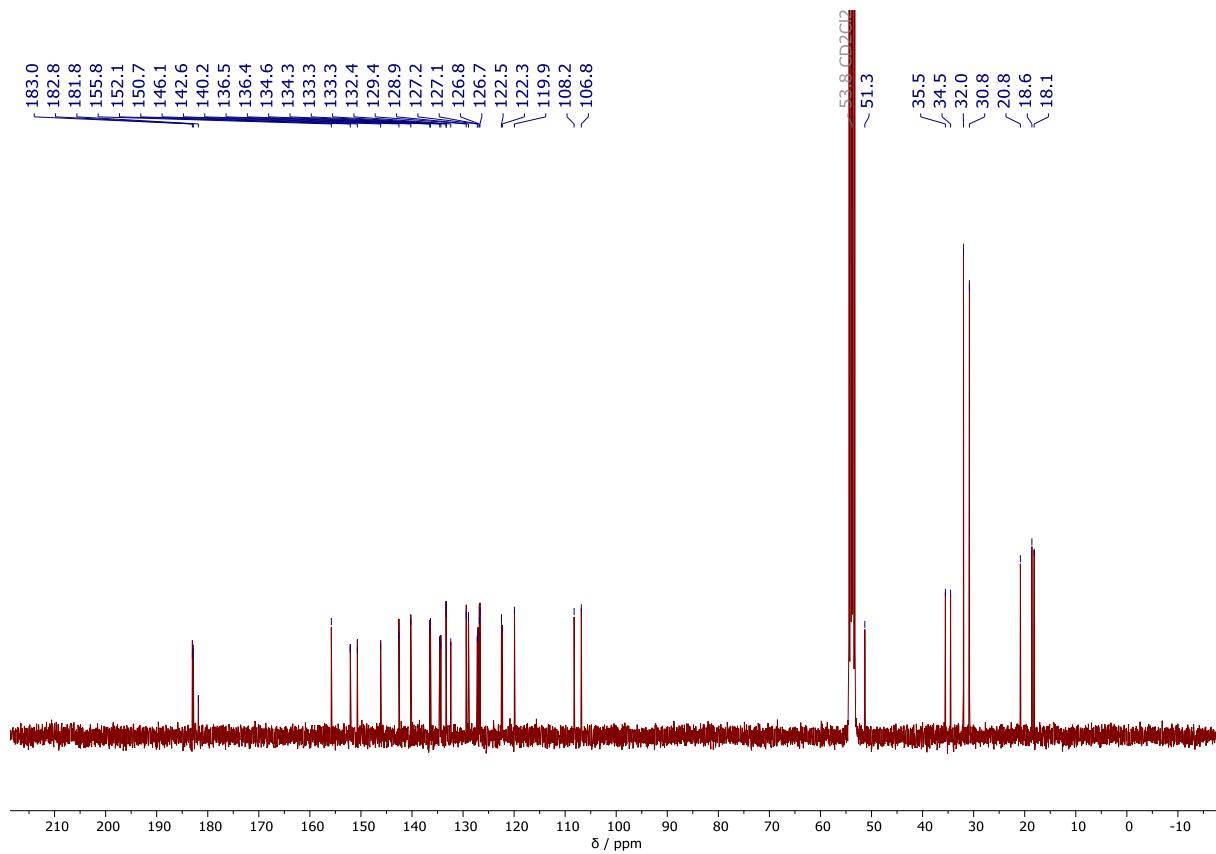


Figure S1.56. $^1\text{H NMR}$ spectrum (600 MHz, CD_2Cl_2) of **1-SIMes**.



1.21 Coordination State of 1-pyridine in Solution

When isolating the pyridine-coordinated silicon complex, two equivalents of pyridine are present in the bulk solid. This, along with the much darker red colour of the solid, indicated the formation of the pyridine bis-adduct in the solid state. Because the interpretations of the UV-vis data are based on the mono-adduct, it is crucial to ensure that only the mono-adduct is present in dilute solutions.

The first indication is the ^{29}Si NMR shift of -104.9 ppm, which is consistent with similar pentacoordinate silicon species in literature² and in this work.

Secondly, the UV-vis spectrum of 1-pyridine in solution gives an absorption spectrum in line with TD-DFT calculations on the mono-adduct. Given the good agreement of TD-DFT calculations with all other complexes (see computational section), they are reasonably good predictors of coordination states for these complexes. Adding a large excess of pyridine (200 equivalents) results in λ_{\max} redshifted by 50 nm, which corresponds to the calculated absorption spectrum of 1-(pyridine)₂. The ability to bind a second pyridine molecule is corroborated by calculated thermodynamics, which give a higher affinity for the binding of one pyridine equivalent, but a still slightly exergonic binding of the second pyridine ($\Delta G = -24$ kJ/mol and -6 kJ/mol, respectively, see computational section for further details).

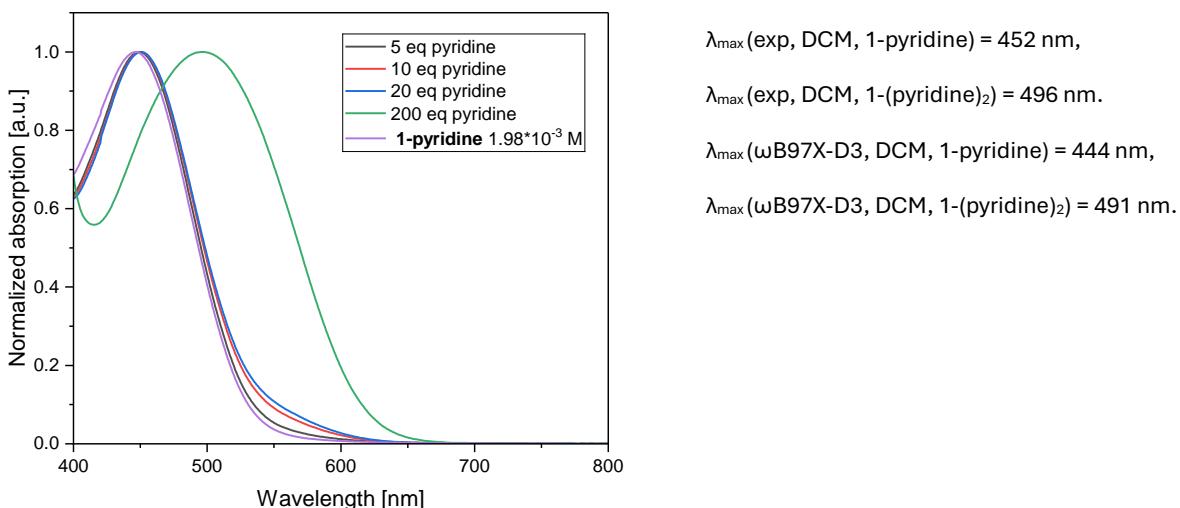


Figure S1.59. Normalised absorption spectrum of **1-pyridine** in DCM with additional equivalents of pyridine. λ_{\max} (1-pyridine) = 452 nm, λ_{\max} (1-(pyridine)₂) = 496 nm.

2 UV-vis Spectroscopy

UV-vis spectra were measured on a JASCO V-570 UV/VIS/NIR spectrophotometer. Measurements were carried out in 2 mm quartz cuvette equipped with a J-Young valve with a volume of 700 μL . A stock solution in 1000 μL DCM was prepared for all compounds. Different concentrations were used to obtain a more accurate estimate for the extinction coefficient of the CT band. Additionally, measurements at differing concentrations provided a method to determine if dissociation of the acid-base adducts occurs at low concentrations. If dissociation was observed, additional equivalents of base were added to obtain spectra of the fully associated complexes. All data analysis was done with OriginLab 2024b.

2.1 CT Absorption Bands of all Compounds

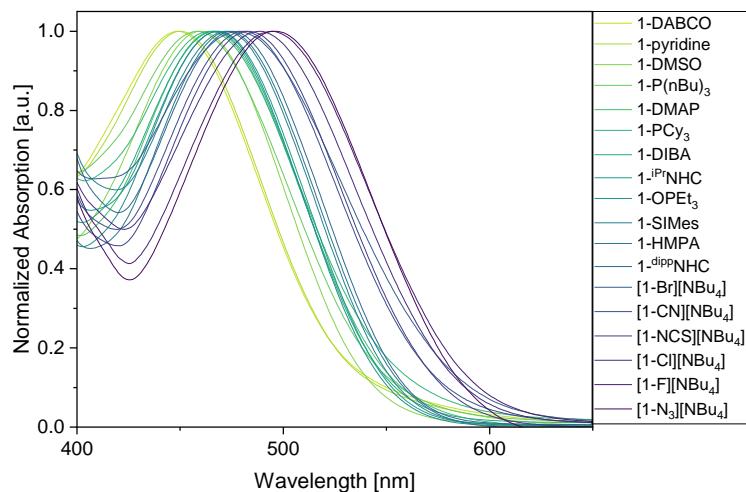


Figure S2.1. Normalised UV-vis absorption spectra of adducts **1-X** (CT band, DCM).

2.2 1

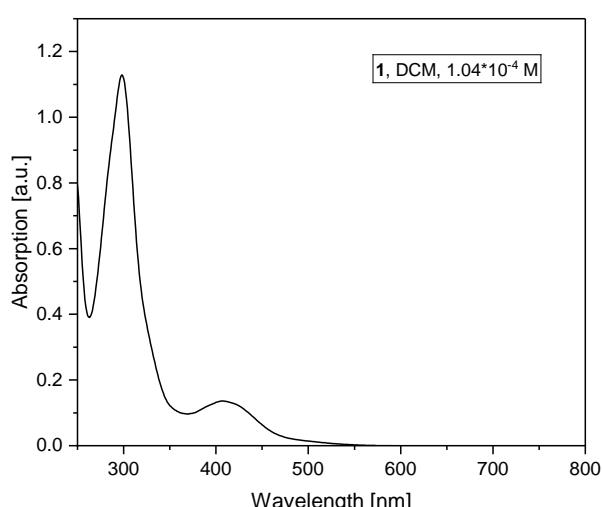


Figure S2.2. UV-vis absorption spectrum of **1** ($1.04 \cdot 10^{-4} \text{ M}$, DCM).

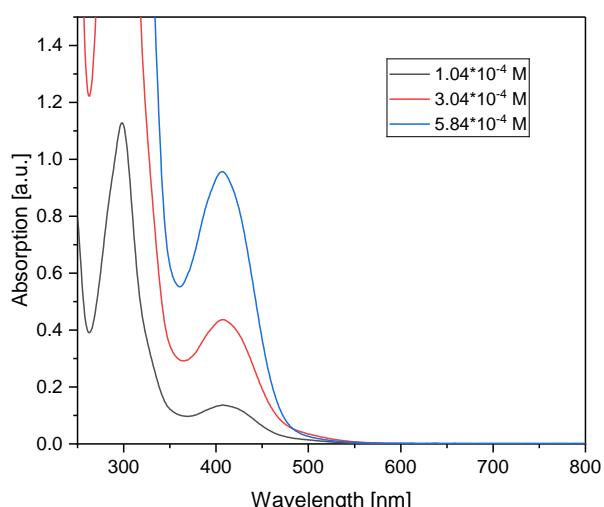


Figure S2.3. UV-vis absorption spectrum of **1** ($1.04 \cdot 10^{-4} \text{ M}$, $3.04 \cdot 10^{-4} \text{ M}$, $5.84 \cdot 10^{-4} \text{ M}$, DCM).

Stock solution:

6.5 mg **1** in 1000 μL DCM.

Cuvette: $d = 2 \text{ mm}$, $V = 700 \mu\text{L}$

Aliquots: 10 μL , 30 μL , 60 μL

Concentrations:

$$\text{i)} \left(\frac{\frac{0.0065 \text{ g}}{878 \text{ g/mol}}}{0.001 \text{ L}} \right) \cdot \frac{10 \mu\text{L}}{710 \mu\text{L}} = 1.043 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \left(\frac{\frac{0.0065 \text{ g}}{878 \text{ g/mol}}}{0.001 \text{ L}} \right) \cdot \frac{30 \mu\text{L}}{730 \mu\text{L}} = 3.042 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \left(\frac{\frac{0.0065 \text{ g}}{878 \text{ g/mol}}}{0.001 \text{ L}} \right) \cdot \frac{60 \mu\text{L}}{760 \mu\text{L}} = 5.844 \cdot 10^{-4} \text{ M}$$

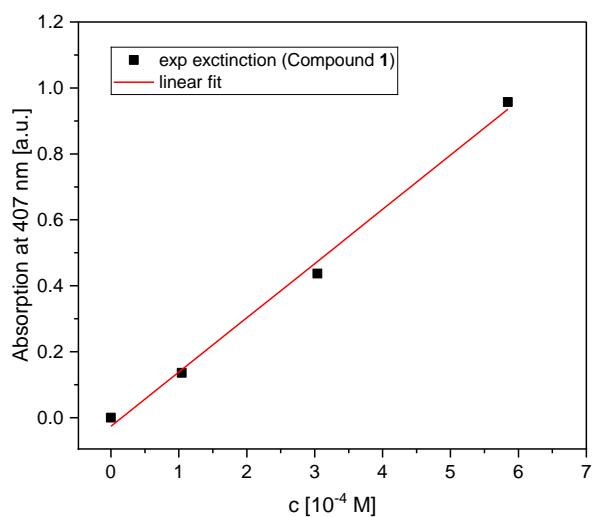


Figure S2.4. Extinction at 407 nm (CT band) of **1** (0 M, $1.04 \cdot 10^{-4} \text{ M}$, $3.04 \cdot 10^{-4} \text{ M}$, $5.84 \cdot 10^{-4} \text{ M}$, DCM).

Linear Fit: $y = 1644.8x - 0.02584$. $R^2 = 0.993$.

$$\varepsilon = \frac{1644.8 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 8200 \text{ M}^{-1}\text{cm}^{-1}$$

2.3 [1-CN][NBu₄]

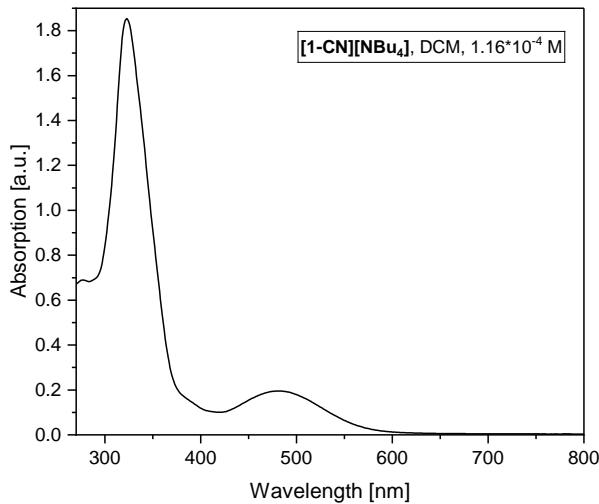


Figure S2.5. UV-vis absorption spectrum of [1-CN][NBu₄] (1.16·10⁻⁴ M, DCM).

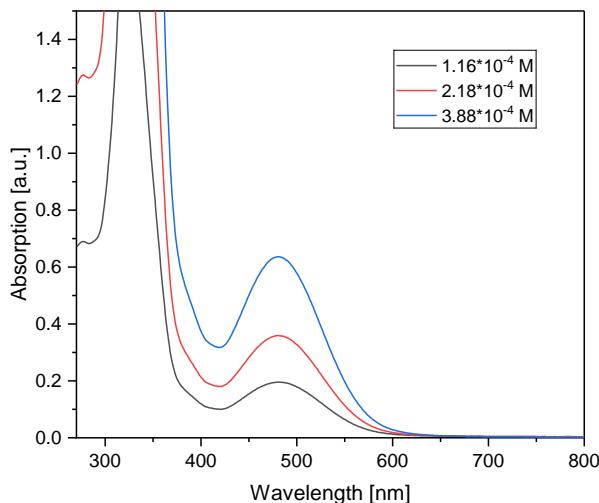


Figure S2.6. UV-vis absorption spectrum of [1-CN][NBu₄] (1.16·10⁻⁴ M, 2.18·10⁻⁴ M, 3.88·10⁻⁴ M, DCM). No change of λ_{max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

Stock solution:

2.0 mg [1-CN][NBu₄] in 1000 μ L DCM.

Cuvette: d = 2 mm, V = 700 μ L

Aliquots: 50 μ L, 100 μ L, 200 μ L

Concentrations:

$$\text{i)} \frac{\left(\frac{0.002 \text{ g}}{114.7 \text{ g/mol}}\right) \cdot \frac{50 \mu\text{L}}{0.001 \text{ L}}}{750 \mu\text{L}} = 1.162 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.002 \text{ g}}{114.7 \text{ g/mol}}\right) \cdot \frac{100 \mu\text{L}}{0.001 \text{ L}}}{800 \mu\text{L}} = 2.180 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.002 \text{ g}}{114.7 \text{ g/mol}}\right) \cdot \frac{200 \mu\text{L}}{0.001 \text{ L}}}{900 \mu\text{L}} = 3.875 \cdot 10^{-4} \text{ M}$$

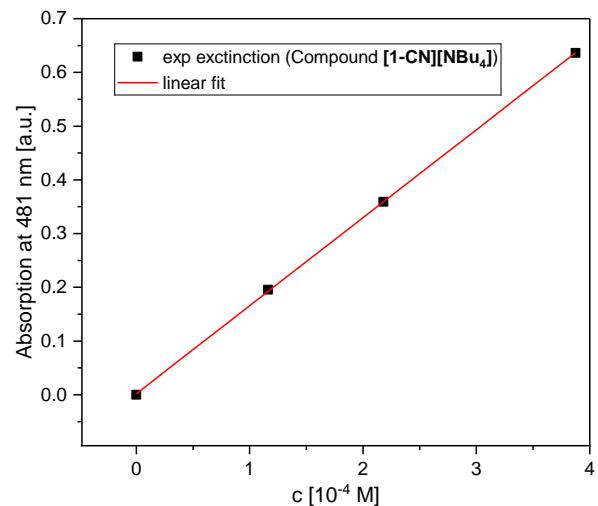


Figure S2.7. Extinction at 481 nm (CT band) of [1-CN][NBu₄] (0 M, 1.16·10⁻⁴ M, 2.18·10⁻⁴ M, 3.88·10⁻⁴ M, DCM). Linear Fit: $y = 1638.1x - 0.00205$. $R^2 = 0.999$. $\epsilon = \frac{1638.1 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 8200 \text{ M}^{-1}\text{cm}^{-1}$.

2.4 [1-Cl][PPh₄]

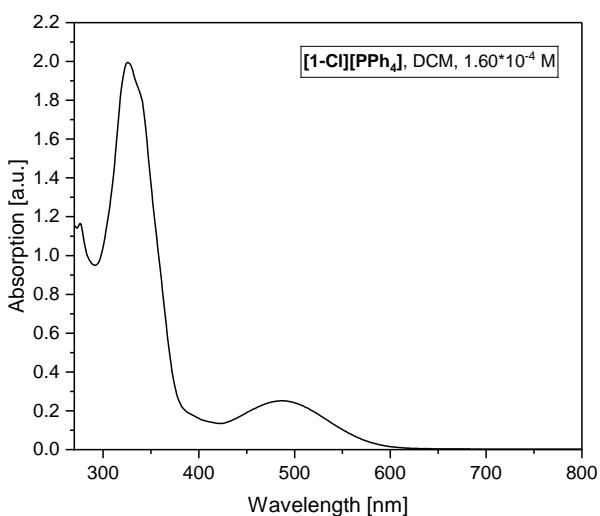


Figure S2.8. UV-vis absorption spectrum of [1-Cl][PPh₄] (1.60·10⁻⁴ M, DCM).

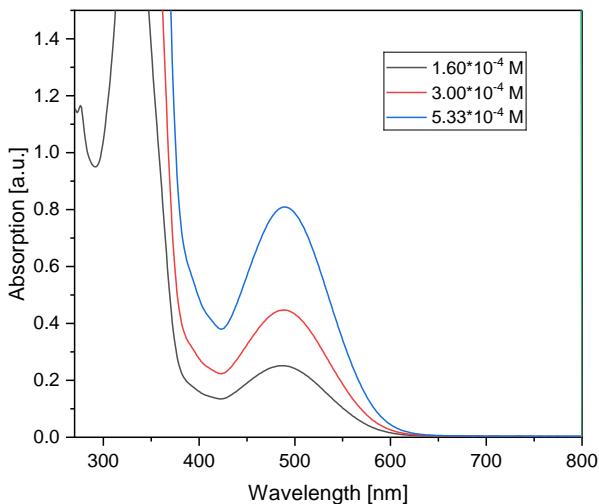


Figure S2.9. UV-vis absorption spectrum of [1-Cl][PPh₄] (1.60·10⁻⁴ M, 3.00·10⁻⁴ M, 5.33·10⁻⁴ M, DCM). No change of λ_{max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

Stock solution:

3.0 mg [1-Cl][PPh₄] in 1000 µL DCM.

Cuvette: d = 2 mm, V = 700 µL

Aliquots: 50 µL, 100 µL, 200 µL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.003\text{g}}{1252\text{g/mol}}\right) \cdot \frac{50\text{ }\mu\text{L}}{0.001\text{ l}}}{750\text{ }\mu\text{L}} = 1.597 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.003\text{g}}{1252\text{g/mol}}\right) \cdot \frac{100\text{ }\mu\text{L}}{0.001\text{ l}}}{800\text{ }\mu\text{L}} = 3.000 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.003\text{g}}{1252\text{g/mol}}\right) \cdot \frac{200\text{ }\mu\text{L}}{0.001\text{ l}}}{900\text{ }\mu\text{L}} = 5.325 \cdot 10^{-4} \text{ M}$$

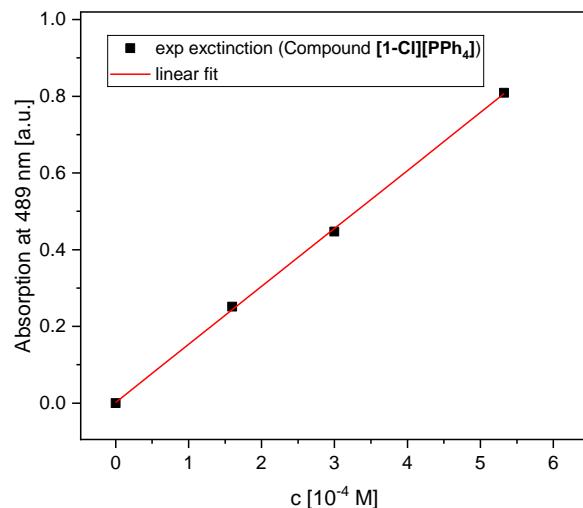


Figure S2.10. Extinction at 489 nm (CT band) of [1-Cl][PPh₄] (0 M, 1.60·10⁻⁴ M, 3.00·10⁻⁴ M, 5.33·10⁻⁴ M, DCM). Linear Fit: $y = 1511.1x - 0.00203$. $R^2 = 0.999$. $\epsilon(489\text{ nm}) = \frac{1511.1\text{ M}^{-1}}{0.2\text{ cm}} \approx 7600\text{ M}^{-1}\text{cm}^{-1}$.

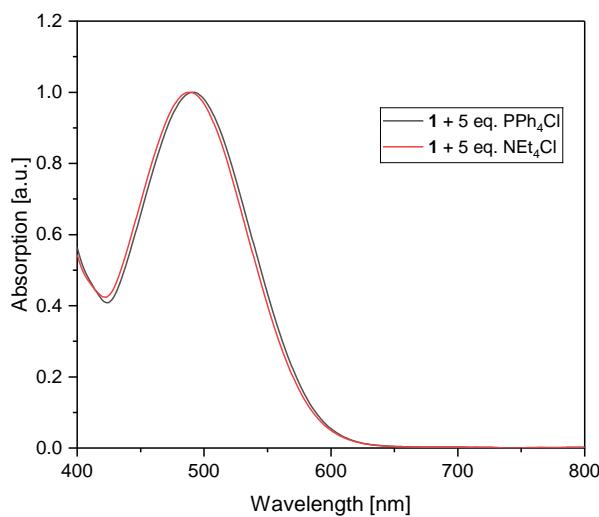


Figure S2.11. Normalised UV-vis absorption spectrum of **1** (ca. 10⁻³ M) with 5 eq. NEt₄Cl or PPh₄Cl added. $\lambda_{\text{max}}(\mathbf{1} + \text{NEt}_4\text{Cl}) = 489\text{ nm}$, $\lambda_{\text{max}}(\mathbf{1} + \text{PPh}_4\text{Cl}) = 490\text{ nm}$.

2.5 [1-NCS][NBu₄]

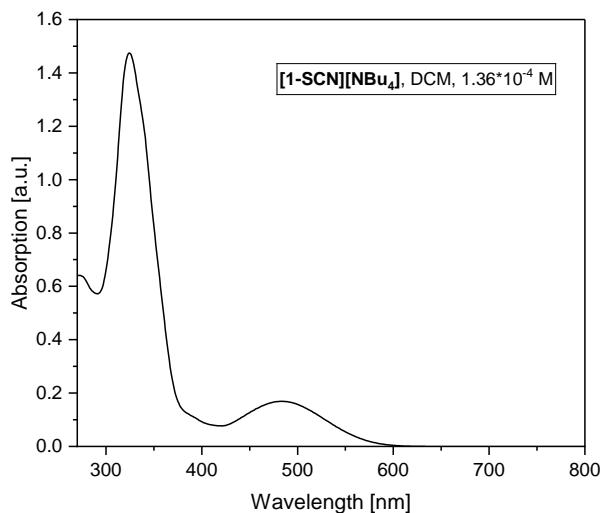


Figure S2.12. UV-vis absorption spectrum of [1-NCS][NBu₄] (1.36·10⁻⁴ M, DCM).

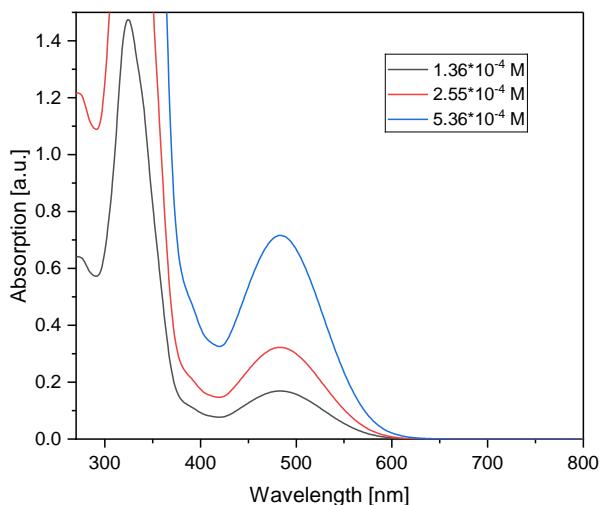


Figure S2.13. UV-vis absorption spectrum of [1-NCS][PPh₄] (1.36·10⁻⁴ M, 2.55·10⁻⁴ M, 5.36·10⁻⁴ M, DCM). No change of λ_{max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

Stock solution:

2.4 mg [1-NCS][NBu₄] in 1000 μL DCM.

Cuvette: d = 2 mm, V = 700 μL

Aliquots: 50 μL, 100 μL, 250 μL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0024 \text{ g}}{117.9 \text{ g/mol}}\right)}{0.001 \text{ L}} \cdot \frac{50 \mu\text{L}}{750 \mu\text{L}} = 1.357 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0024 \text{ g}}{117.9 \text{ g/mol}}\right)}{0.001 \text{ L}} \cdot \frac{100 \mu\text{L}}{800 \mu\text{L}} = 2.545 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0024 \text{ g}}{117.9 \text{ g/mol}}\right)}{0.001 \text{ L}} \cdot \frac{250 \mu\text{L}}{950 \mu\text{L}} = 5.357 \cdot 10^{-4} \text{ M}$$

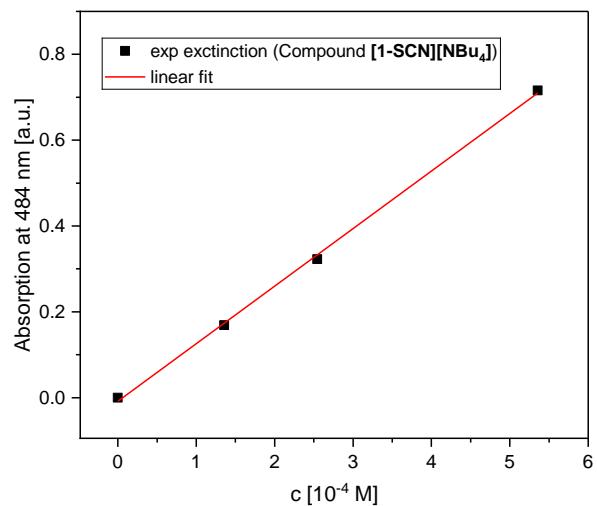
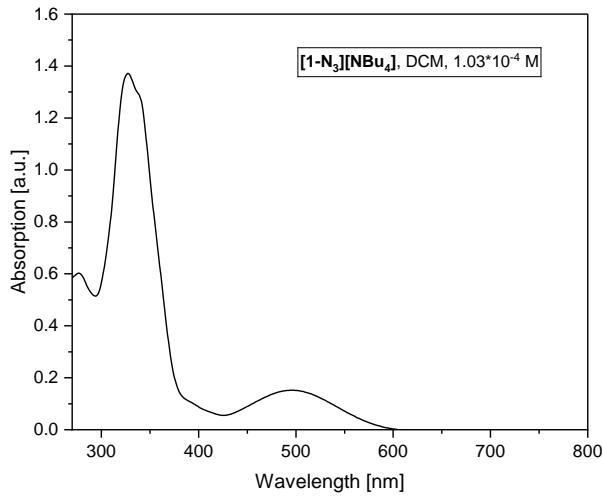


Figure S2.14. Extinction at 489 nm (CT band) of [1-NCS][PPh₄] (0 M, 1.36·10⁻⁴ M, 2.55·10⁻⁴ M, 5.36·10⁻⁴ M, DCM)
Linear Fit: $y = 1340.7x - 0.00862$. $R^2 = 0.999$.
 $\epsilon(484 \text{ nm}) = \frac{1340.7 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 6700 \text{ M}^{-1}\text{cm}^{-1}$.

2.6 [1-N₃][NBu₄]



Stock solution:

1.8 mg [1-N₃][NBu₄] in 1000 μL DCM.

Cuvette: d = 2 mm, V = 700 μL

Aliquots: 50 μL, 100 μL, 200 μL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0018\text{ g}}{1163\text{ g/mol}}\right) \cdot \frac{50\text{ }\mu\text{L}}{0.001\text{ l}}}{750\text{ }\mu\text{L}} = 1.032 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0018\text{ g}}{1163\text{ g/mol}}\right) \cdot \frac{100\text{ }\mu\text{L}}{0.001\text{ l}}}{800\text{ }\mu\text{L}} = 1.935 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0018\text{ g}}{1163\text{ g/mol}}\right) \cdot \frac{200\text{ }\mu\text{L}}{0.001\text{ l}}}{900\text{ }\mu\text{L}} = 3.439 \cdot 10^{-4} \text{ M}$$

Figure S2.15. UV-vis absorption spectrum of [1-N₃][NBu₄] (1.03·10⁻⁴ M, DCM).

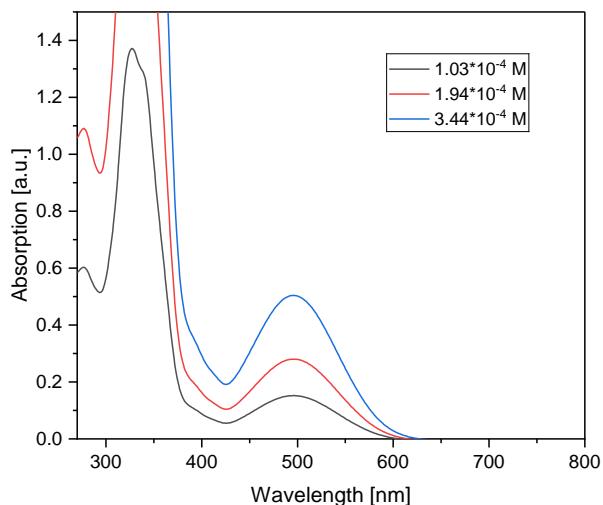


Figure S2.16. UV-vis absorption spectrum of [1-N₃][NBu₄] (1.03·10⁻⁴ M, 1.94·10⁻⁴ M, 3.44·10⁻⁴ M, DCM). No change of λ_{\max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

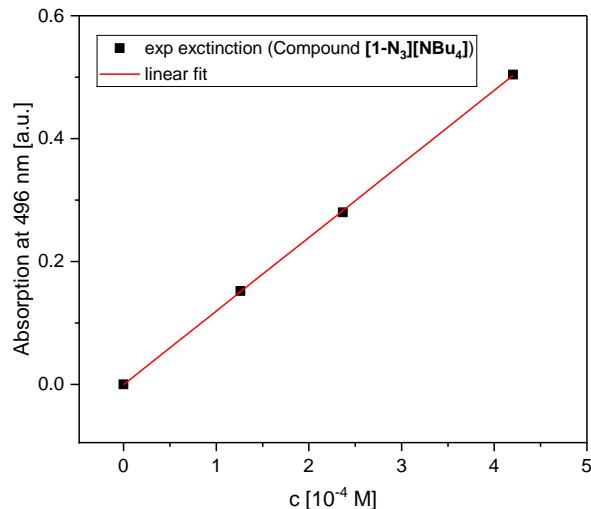


Figure S2.17. Extinction at 496 nm (CT band) of [1-N₃][NBu₄] (0 M, 1.03·10⁻⁴ M, 1.94·10⁻⁴ M, 3.44·10⁻⁴ M, DCM). Linear Fit: $y = 1463.3x - 0.00036$. $R^2 = 0.999$. $\epsilon(496\text{ nm}) = \frac{1463.3\text{ M}^{-1}}{0.2\text{ cm}} \approx 7300\text{ M}^{-1}\text{cm}^{-1}$.

2.7 [1-F][NBu₄]

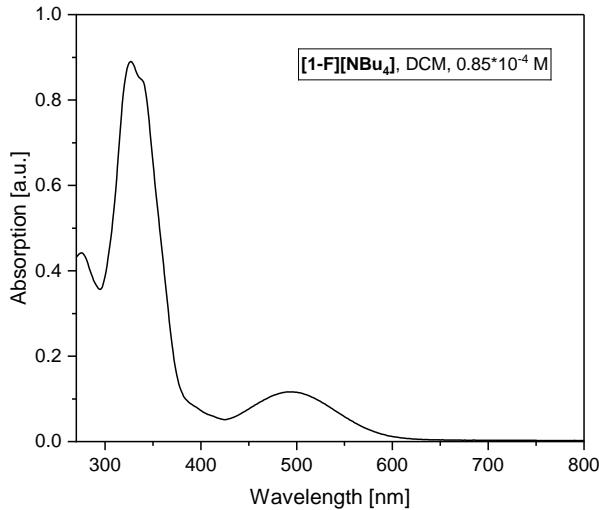


Figure S2.18. UV-vis absorption spectrum of [1-F][NBu₄] (0.85·10⁻⁴ M, DCM).

Stock solution:

1.0 mg [1-F][NBu₄] in 1000 μL DCM.

Cuvette: d = 2 mm, V = 700 μL

Aliquots: 75 μL, 200 μL, 400 μL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.001\text{g}}{1140\text{ g/mol}}\right)}{0.001\text{ L}} \cdot \frac{75\text{ }\mu\text{L}}{775\text{ }\mu\text{L}} = 8.498 \cdot 10^{-5} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.001\text{g}}{1140\text{ g/mol}}\right)}{0.001\text{ L}} \cdot \frac{200\text{ }\mu\text{L}}{900\text{ }\mu\text{L}} = 1.949 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.001\text{g}}{1140\text{ g/mol}}\right)}{0.001\text{ L}} \cdot \frac{400\text{ }\mu\text{L}}{1100\text{ }\mu\text{L}} = 3.190 \cdot 10^{-4} \text{ M}$$

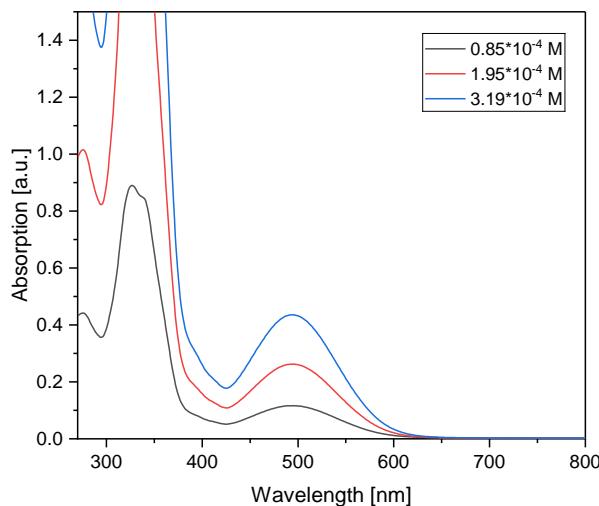


Figure S2.19. UV-vis absorption spectrum of [1-F][NBu₄] (0.85·10⁻⁴ M, 1.95·10⁻⁴ M, 3.19·10⁻⁴ M, DCM). No change of λ_{\max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

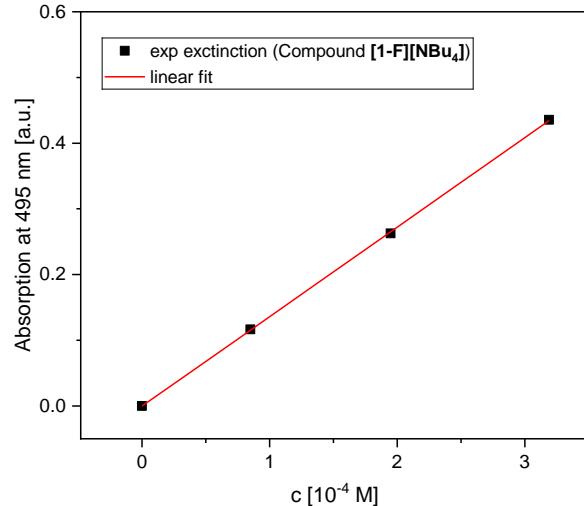


Figure S2.20. Extinction at 495 nm (CT band) of [1-F][NBu₄] (0 M, 0.85·10⁻⁴ M, 1.95·10⁻⁴ M, 3.19·10⁻⁴ M, DCM). Linear Fit: $y = 1362.1x - 0.00032$. $R^2 = 0.999$. $\epsilon(502\text{ nm}) = \frac{1362.1\text{ M}^{-1}}{0.2\text{ cm}} \approx 6800 \text{ M}^{-1}\text{cm}^{-1}$.

2.8 1-HMPA

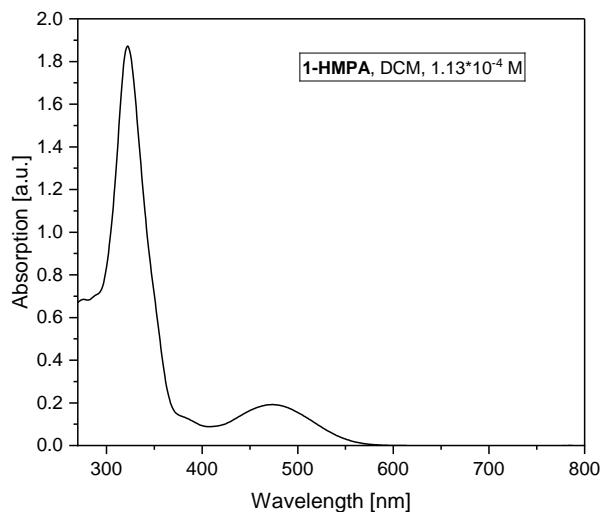


Figure S2.21. UV-vis absorption spectrum of **1-HMPA** ($1.13 \cdot 10^{-4} \text{ M}$, DCM).

Stock solution:

1.8 mg **1-HMPA** in 1000 μL DCM.

Cuvette: $d = 2 \text{ mm}$, $V = 700 \mu\text{L}$

Aliquots: 50 μL , 100 μL , 200 μL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0018 \text{ g}}{1058 \text{ g/mol}}\right)}{0.001 \text{ L}} \cdot \frac{50 \mu\text{L}}{750 \mu\text{L}} = 1.134 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0018 \text{ g}}{1058 \text{ g/mol}}\right)}{0.001 \text{ L}} \cdot \frac{100 \mu\text{L}}{800 \mu\text{L}} = 2.123 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0018 \text{ g}}{1058 \text{ g/mol}}\right)}{0.001 \text{ L}} \cdot \frac{200 \mu\text{L}}{900 \mu\text{L}} = 3.871 \cdot 10^{-4} \text{ M}$$

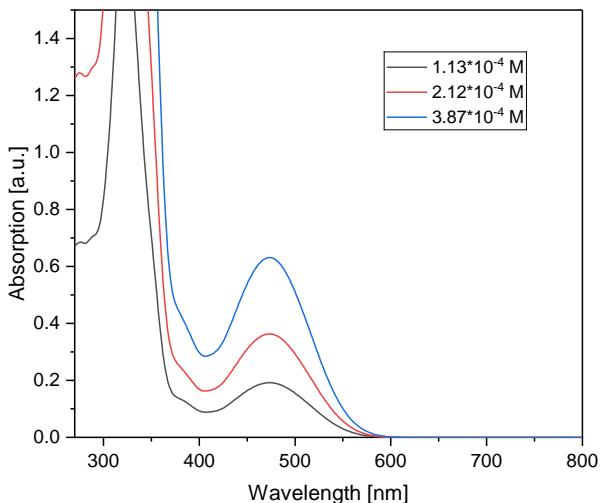


Figure S2.22. UV-vis absorption spectrum of **1-HMPA** ($1.13 \cdot 10^{-4} \text{ M}$, $2.12 \cdot 10^{-4} \text{ M}$, $3.87 \cdot 10^{-4} \text{ M}$, DCM). No change of λ_{max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

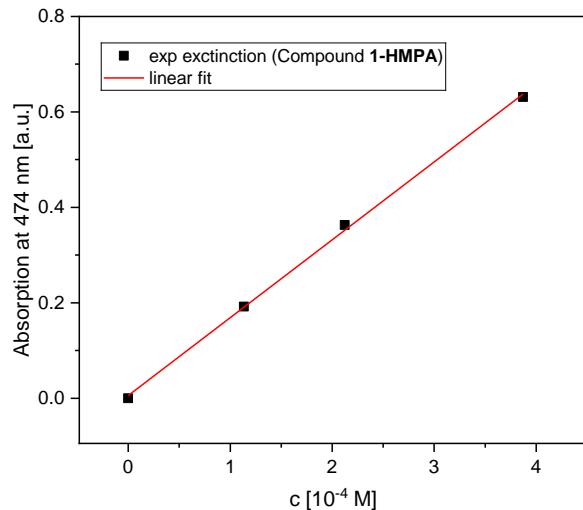


Figure S2.23. Extinction at 474 nm (CT band) of **1-HMPA** (0 M, $1.13 \cdot 10^{-4} \text{ M}$, $2.12 \cdot 10^{-4} \text{ M}$, $3.87 \cdot 10^{-4} \text{ M}$, DCM). Linear Fit: $y = 1631.9x + 0.00786$. $R^2 = 0.999$. $\epsilon(474 \text{ nm}) = \frac{1631.9 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 8200 \text{ M}^{-1}\text{cm}^{-1}$.

2.9 1-(pyridine)₂

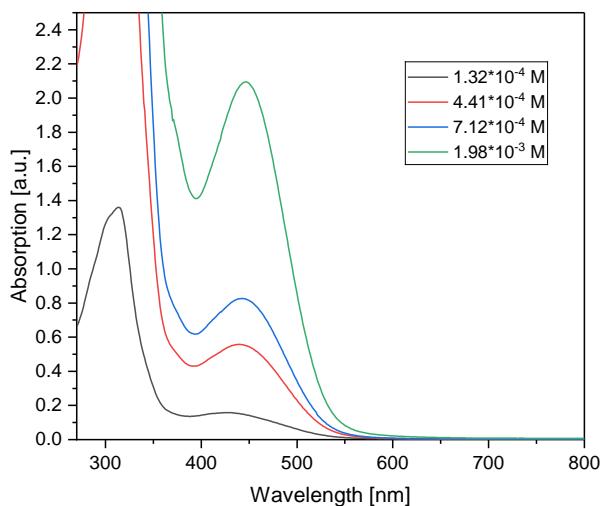


Figure S2.24. UV-vis absorption spectrum of **1-(pyridine)₂** (1.32·10⁻⁴ M, 4.41·10⁻⁴ M, 7.12·10⁻⁴ M, 1.98·10⁻³ M DCM). λ_{\max} (c = 1.32·10⁻⁴ M) = 426 nm, λ_{\max} (c = 4.41·10⁻⁴ M) = 439 nm, λ_{\max} (c = 7.12·10⁻⁴ M) = 443 nm, λ_{\max} (c = 1.98·10⁻³ M) = 448 nm.

The absorption spectrum (λ_{\max}) of **1-pyridine** shows significant concentration dependence (cf. figure S2.24), suggesting dissociation at low concentrations. Therefore, additional equivalents of pyridine were added to ensure complete conversion to the pentacoordinated silicon complex.

Due to partial formation of the bis-adduct, a low energy shoulder appears at high pyridine concentrations. However, the influence of this formation on λ_{\max} is minimal, as no change in λ_{\max} was observed between 10 and 20 equivalents of pyridine.

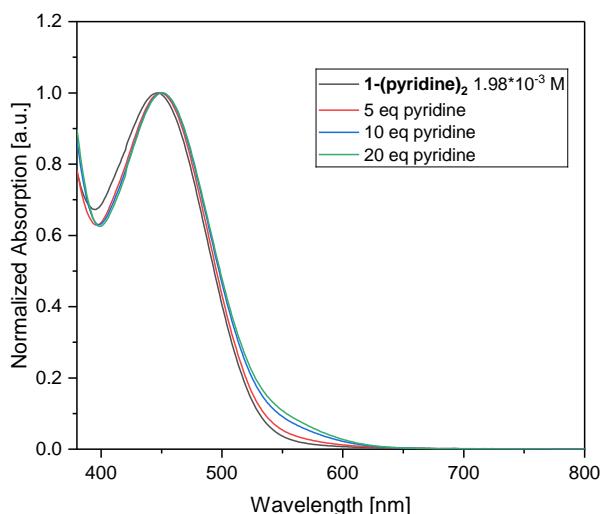


Figure S2.25. Normalised UV-vis absorption spectrum of **1-(pyridine)₂** (1.98·10⁻³ M; 6.26·10⁻⁴ M + 5/10/20 eq. pyridine, normalised). A shift of the spectrum is observed when adding 5 equivalents of pyridine, but no further change of λ_{\max} (452 nm) with 10 and 20 equivalents pyridine. A low-energy shoulder appears due to partial formation of the bis-adduct at high pyridine concentrations.

Stock solution:

1.9 mg **1-(pyridine)₂** in 1000 µL DCM.

Cuvette: d = 2 mm, V = 700 µL

Aliquots: 50 µL, 200 µL, 400 µL, stock solution

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0019\text{ g}}{958\text{ g/mol}}\right)}{0.001\text{ L}} \cdot \frac{50\text{ }\mu\text{L}}{750\text{ }\mu\text{L}} = 1.322 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0019\text{ g}}{958\text{ g/mol}}\right)}{0.001\text{ L}} \cdot \frac{200\text{ }\mu\text{L}}{900\text{ }\mu\text{L}} = 4.407 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0019\text{ g}}{958\text{ g/mol}}\right)}{0.001\text{ L}} \cdot \frac{400\text{ }\mu\text{L}}{1100\text{ }\mu\text{L}} = 7.120 \cdot 10^{-4} \text{ M}$$

$$\text{iv)} \frac{\left(\frac{0.0019\text{ g}}{958\text{ g/mol}}\right)}{0.001\text{ L}} = 1.983 \cdot 10^{-3} \text{ M}$$

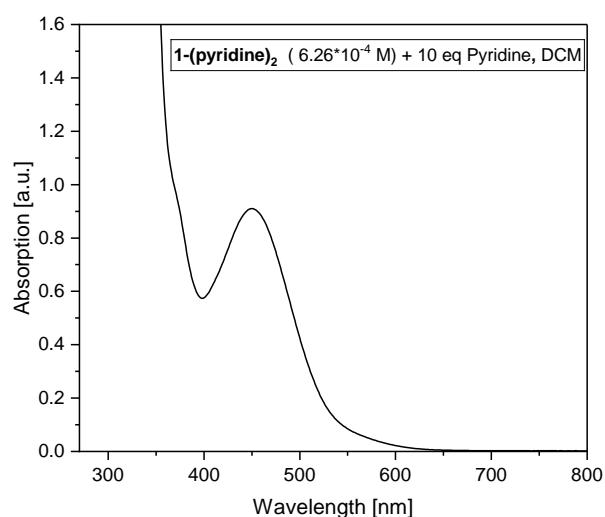


Figure S2.26. UV-vis absorption spectrum of **1-(pyridine)₂** (6.26·10⁻⁴ M + 10 eq pyridine, DCM). ϵ (452 nm) = $\frac{0.90998}{6.26 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 7300 \text{ M}^{-1}\text{cm}^{-1}$.

2.10 1-DMAP

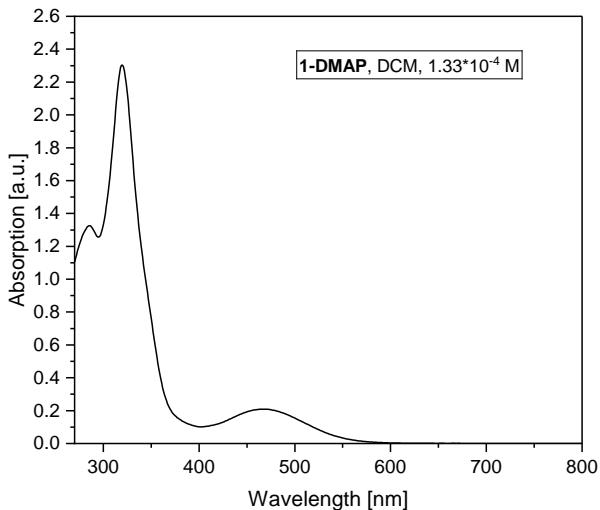


Figure S2.27. UV-vis absorption spectrum of **1-DMAP** ($1.33 \cdot 10^{-4} \text{ M}$, DCM).

Stock solution:

2.0 mg **1-DMAP** in 1000 μL DCM.

Cuvette: $d = 2 \text{ mm}$, $V = 700 \mu\text{L}$

Aliquots: 50 μL , 100 μL , 200 μL

Concentrations:

$$\begin{aligned} \text{i)} & \frac{\left(\frac{0.0020 \text{ g}}{1001 \text{ g/mol}}\right) \cdot \frac{50 \mu\text{L}}{0.001 \text{ L}}}{750 \mu\text{L}} = 1.332 \cdot 10^{-4} \text{ M} \\ \text{ii)} & \frac{\left(\frac{0.0020 \text{ g}}{1001 \text{ g/mol}}\right) \cdot \frac{100 \mu\text{L}}{0.001 \text{ L}}}{800 \mu\text{L}} = 2.498 \cdot 10^{-4} \text{ M} \\ \text{iii)} & \frac{\left(\frac{0.0020 \text{ g}}{1001 \text{ g/mol}}\right) \cdot \frac{200 \mu\text{L}}{0.001 \text{ L}}}{900 \mu\text{L}} = 4.440 \cdot 10^{-4} \text{ M} \end{aligned}$$

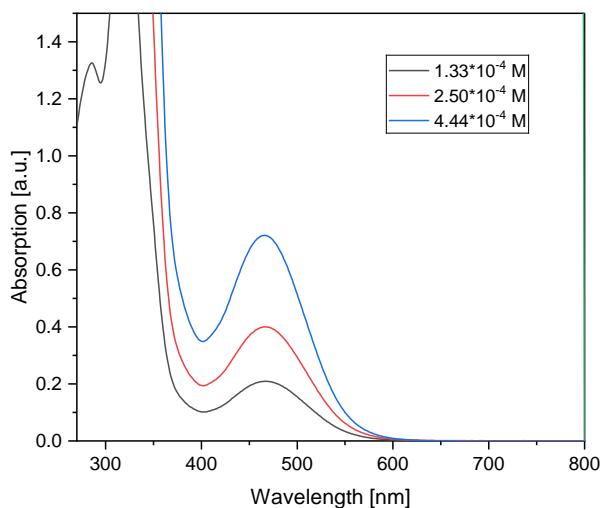


Figure S2.28. UV-vis absorption spectrum of **1-DMAP** ($1.33 \cdot 10^{-4} \text{ M}$, $2.50 \cdot 10^{-4} \text{ M}$, $4.44 \cdot 10^{-4} \text{ M}$, DCM). No change of λ_{max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

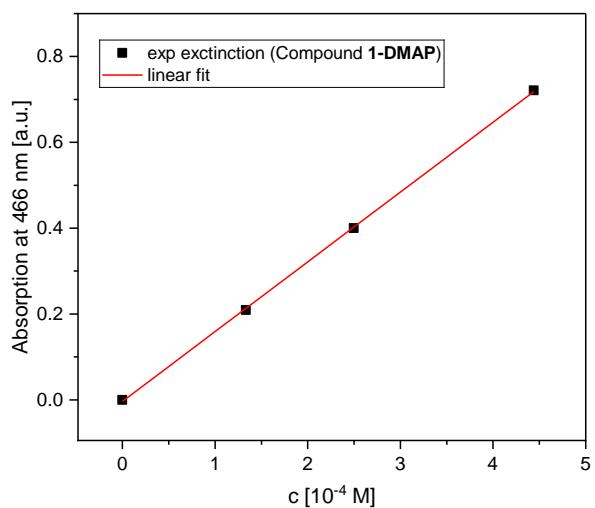


Figure S2.29. Extinction at 466 nm (CT band) of **1-DMAP** (0 M, $1.33 \cdot 10^{-4} \text{ M}$, $2.50 \cdot 10^{-4} \text{ M}$, $4.44 \cdot 10^{-4} \text{ M}$, DCM). Linear Fit: $y = 1626.2x - 0.00369$. $R^2 = 0.999$. $\epsilon(474 \text{ nm}) = \frac{1626.2 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 8100 \text{ M}^{-1}\text{cm}^{-1}$.

2.11 1-OPEt₃

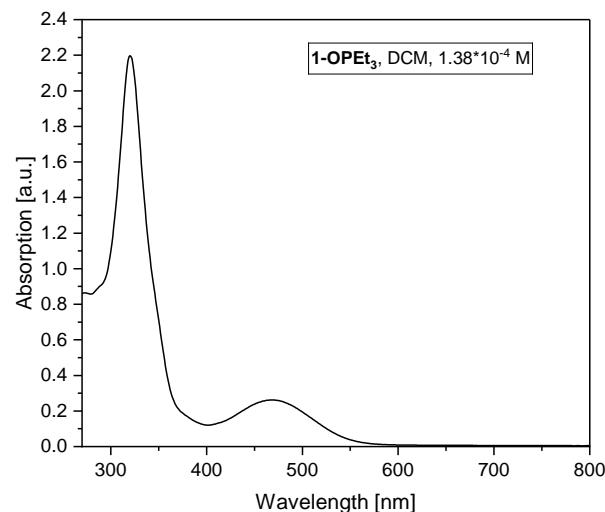


Figure S2.30. UV-vis absorption spectrum of **1-OPEt₃** (1.38·10⁻⁴ M, DCM).

Stock solution:

2.1 mg **1-OPEt₃** in 1000 μL DCM.

Cuvette: d = 2 mm, V = 700 μL

Aliquots: 50 μL, 100 μL, 200 μL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0021\text{ g}}{1012\text{ g/mol}}\right) \cdot \frac{50\text{ }\mu\text{L}}{0.001\text{ l}}}{750\text{ }\mu\text{L}} = 1.383 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0021\text{ g}}{1012\text{ g/mol}}\right) \cdot \frac{100\text{ }\mu\text{L}}{0.001\text{ l}}}{800\text{ }\mu\text{L}} = 2.594 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0021\text{ g}}{1012\text{ g/mol}}\right) \cdot \frac{200\text{ }\mu\text{L}}{0.001\text{ l}}}{900\text{ }\mu\text{L}} = 4.611 \cdot 10^{-4} \text{ M}$$

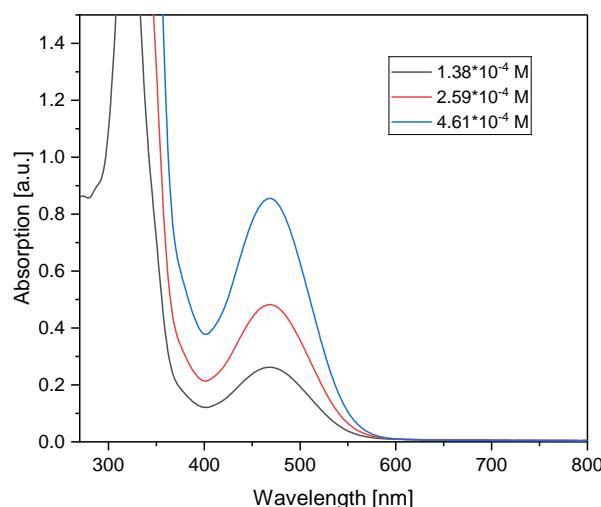


Figure S2.31. UV-vis absorption spectrum of **1-OPEt₃** (1.38·10⁻⁴ M, 2.59·10⁻⁴ M, 4.61·10⁻⁴ M, DCM). No change of λ_{\max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

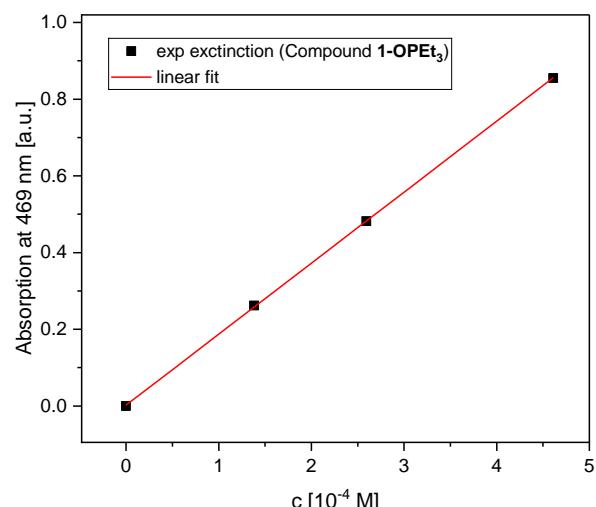


Figure S2.32. Extinction at 469 nm (CT band) of **1-OPEt₃** (0 M, 1.38·10⁻⁴ M, 2.59·10⁻⁴ M, 4.61·10⁻⁴ M, DCM). Linear Fit: $y = 1851.1x + 0.0023$. $R^2 = 0.999$. $\epsilon(474\text{ nm}) = \frac{1851.1\text{ M}^{-1}}{0.2\text{ cm}} \approx 9300 \text{ M}^{-1}\text{cm}^{-1}$.

2.12 1-DMSO

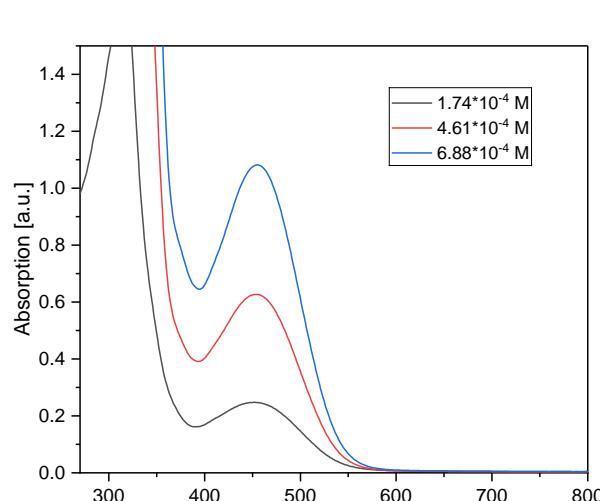


Figure S2.33. UV-vis absorption spectrum of **1-DMSO** ($1.74 \cdot 10^{-4}$ M, $4.61 \cdot 10^{-4}$ M, $6.88 \cdot 10^{-4}$ M, DCM). $\lambda_{\text{max}}(c = 1.74 \cdot 10^{-4}$ M) = 452 nm, $\lambda_{\text{max}}(c = 4.61 \cdot 10^{-4}$ M) = 454 nm, $\lambda_{\text{max}}(c = 6.88 \cdot 10^{-4}$ M) = 456 nm.

Stock solution:

2.5 mg **1-DMSO** in 1000 μ L DCM.

Cuvette: $d = 2$ mm, $V = 700 \mu$ L

Aliquots: 50 μ L, 150 μ L, 250 μ L

Concentrations:

$$\begin{aligned} \text{i)} & \frac{\left(\frac{0.0025\text{g}}{956\text{ g/mol}}\right)}{0.001\text{ l}} \cdot \frac{50\text{ }\mu\text{L}}{750\text{ }\mu\text{L}} = 1.743 \cdot 10^{-4} \text{ M} \\ \text{ii)} & \frac{\left(\frac{0.0025\text{g}}{956\text{ g/mol}}\right)}{0.001\text{ l}} \cdot \frac{150\text{ }\mu\text{L}}{850\text{ }\mu\text{L}} = 4.615 \cdot 10^{-4} \text{ M} \\ \text{iii)} & \frac{\left(\frac{0.0025\text{g}}{956\text{ g/mol}}\right)}{0.001\text{ l}} \cdot \frac{250\text{ }\mu\text{L}}{950\text{ }\mu\text{L}} = 6.881 \cdot 10^{-4} \text{ M} \end{aligned}$$

Due to dissociation at low concentrations (concentration dependence of λ_{max} , figure S2.33), additional equivalents of DMSO were added to obtain the spectrum of the fully associated complex.

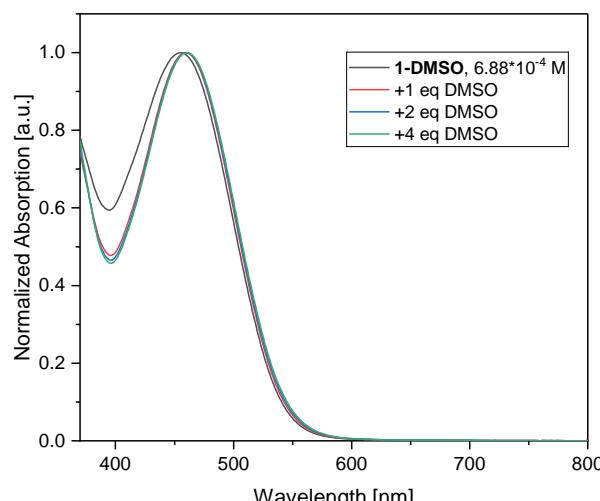


Figure S2.34. Normalised UV-vis absorption spectrum of **1-DMSO** ($6.88 \cdot 10^{-4}$ M, +1/2/4 eq. DMSO, DCM). No significant further change of the absorption spectrum observed between 2 and 4 equivalents of DMSO.

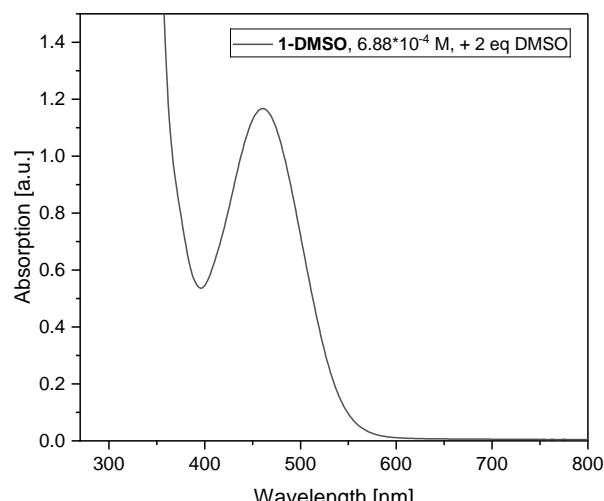


Figure S2.35. UV-vis absorption spectrum of **1-DMSO** ($6.88 \cdot 10^{-4}$ M + 2 eq DMSO, DCM). $\epsilon(460\text{ nm}) = \frac{1.16703}{6.88 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 8500 \text{ M}^{-1}\text{cm}^{-1}$.

2.13 1-DIBA

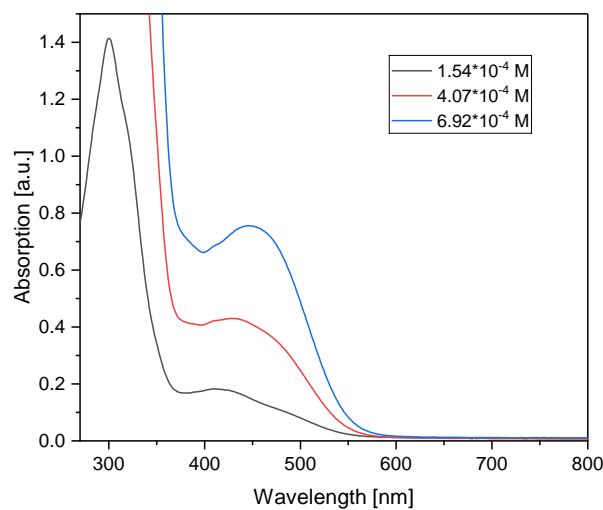


Figure S2.36. UV-vis absorption spectrum of **1-DIBA** ($1.54 \cdot 10^{-4}$ M, $4.07 \cdot 10^{-4}$ M, $6.92 \cdot 10^{-4}$ M, DCM). $\lambda_{\text{max}}(c = 1.54 \cdot 10^{-4}$ M) = 411 nm, $\lambda_{\text{max}}(c = 4.07 \cdot 10^{-4}$ M) = 430 nm, $\lambda_{\text{max}}(c = 6.92 \cdot 10^{-4}$ M) = 446 nm.

Stock solution:

2.5 mg **1-DIBA** in 1000 μ L DCM.

Cuvette: $d = 2$ mm, $V = 700 \mu$ L

Aliquots: 50 μ L, 150 μ L, 300 μ L

Concentrations:

$$\begin{aligned} \text{i)} & \left(\frac{0.0025 \text{ g}}{1084 \text{ g/mol}} \right) \cdot \frac{50 \mu\text{L}}{0.001 \text{ L}} \cdot \frac{50 \mu\text{L}}{750 \mu\text{L}} = 1.538 \cdot 10^{-4} \text{ M} \\ \text{ii)} & \left(\frac{0.0025 \text{ g}}{1084 \text{ g/mol}} \right) \cdot \frac{150 \mu\text{L}}{0.001 \text{ L}} \cdot \frac{150 \mu\text{L}}{850 \mu\text{L}} = 4.070 \cdot 10^{-4} \text{ M} \\ \text{iii)} & \left(\frac{0.0025 \text{ g}}{1084 \text{ g/mol}} \right) \cdot \frac{300 \mu\text{L}}{0.001 \text{ L}} \cdot \frac{300 \mu\text{L}}{1000 \mu\text{L}} = 6.919 \cdot 10^{-4} \text{ M} \end{aligned}$$

Due to significant dissociation at low concentrations (concentration dependence of λ_{max} , figure S2.36), additional equivalents of DIBA were added to obtain the spectrum of the fully associated complex.

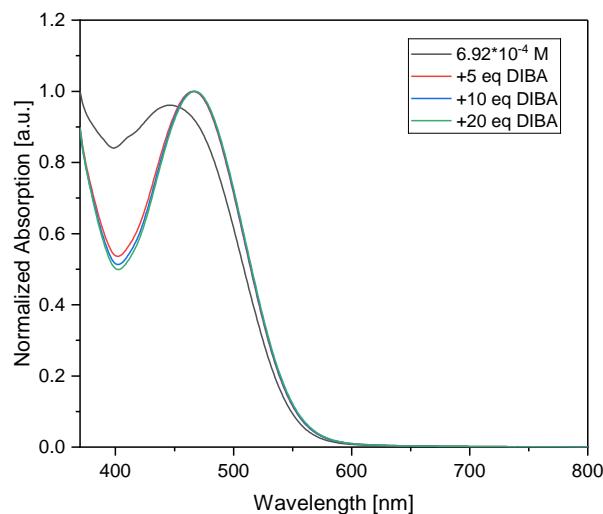


Figure S2.37. Normalised UV-vis absorption spectrum of **1-DIBA** ($6.92 \cdot 10^{-4}$ M, +5/10/20 eq. DIBA, DCM). No significant further change of the absorption spectrum observed between 10 and 20 equivalents of DIBA.

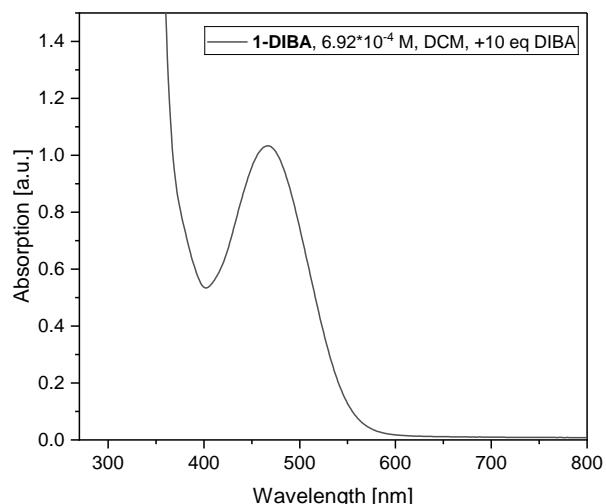


Figure S2.38. UV-vis absorption spectrum of **1-DIBA** ($6.92 \cdot 10^{-4}$ M + 10 eq DIBA, DCM). $\epsilon(467 \text{ nm}) = \frac{1.03329}{6.92 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 7500 \text{ M}^{-1}\text{cm}^{-1}$.

2.14 1-dippNHC

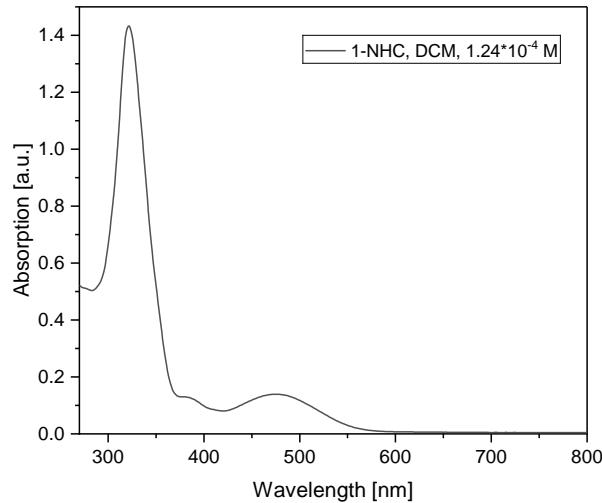


Figure S2.39. UV-vis absorption spectrum of **1-dippNHC** ($1.24 \cdot 10^{-4}$ M, DCM).

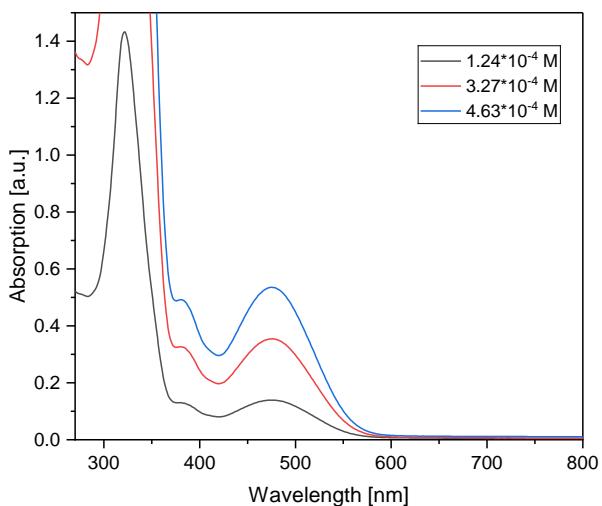


Figure S2.40. UV-vis absorption spectrum of **1-dippNHC** ($1.24 \cdot 10^{-4}$ M, $3.27 \cdot 10^{-4}$ M, $4.63 \cdot 10^{-4}$ M, DCM). No change of λ_{max} is observed at different concentrations, suggesting negligible dissociation at low concentrations.

Stock solution:

2.4 mg **1-dippNHC** in 1000 μL DCM.

Cuvette: $d = 2$ mm, $V = 700 \mu\text{L}$

Aliquots: 50 μL , 150 μL , 250 μL

Concentrations:

$$\begin{aligned} \text{i)} & \frac{\left(\frac{0.0024\text{g}}{1296\text{ g/mol}}\right) \cdot 50\text{ }\mu\text{L}}{0.001\text{ l} \cdot 750\text{ }\mu\text{L}} = 1.235 \cdot 10^{-4}\text{ M} \\ \text{ii)} & \frac{\left(\frac{0.0024\text{g}}{1296\text{ g/mol}}\right) \cdot 150\text{ }\mu\text{L}}{0.001\text{ l} \cdot 850\text{ }\mu\text{L}} = 3.268 \cdot 10^{-4}\text{ M} \\ \text{iii)} & \frac{\left(\frac{0.0024\text{g}}{1296\text{ g/mol}}\right) \cdot 250\text{ }\mu\text{L}}{0.001\text{ l} \cdot 1000\text{ }\mu\text{L}} = 4.630 \cdot 10^{-4}\text{ M} \end{aligned}$$

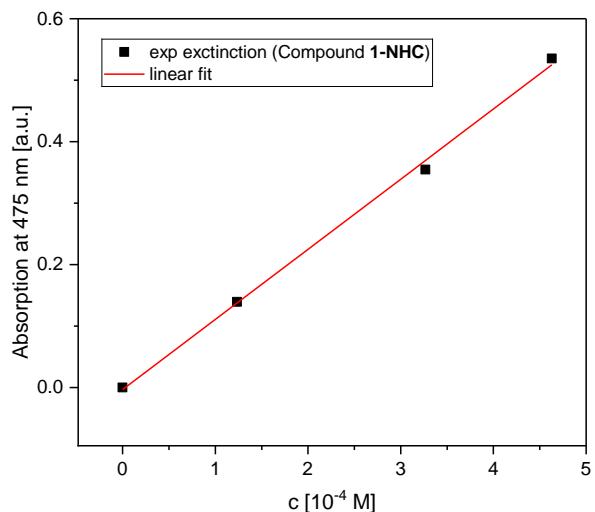


Figure S2.41. Extinction at 475 nm (CT band) of **1-dippNHC** (0 M, $1.24 \cdot 10^{-4}$ M, $3.27 \cdot 10^{-4}$ M, $4.63 \cdot 10^{-4}$ M, DCM). Linear Fit: $y = 1140.9x - 0.00323$. $R^2 = 0.997$. $\epsilon(475\text{ nm}) = \frac{1140.9\text{ M}^{-1}}{0.2\text{ cm}} \approx 5700\text{ M}^{-1}\text{cm}^{-1}$.

2.15 [1-Br][NBu₄]

The bromide adduct of **1** could not be isolated, but UV-vis spectra could be measured with a large excess of NBu₄Br. A 2.3·10⁻⁴ M of **1** was prepared, and NBu₄Br was added as a solid until no further changes were observed, indicating quantitative conversion to **[1-Br][NBu₄]**.

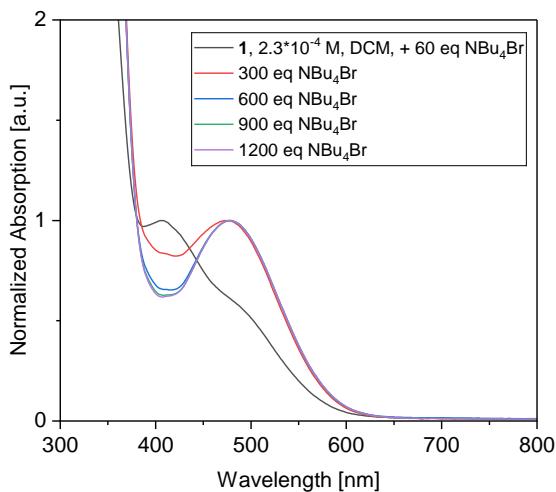


Figure S2.42. Normalised UV-vis absorption spectrum of **1** (2.3·10⁻⁴ M, +60/300/600/900/1200 eq. NBu₄Br, DCM). No significant further change of the absorption spectrum observed between 900 and 1200 equivalents of NBu₄Br. $\lambda_{\text{max}} = 479 \text{ nm}$.

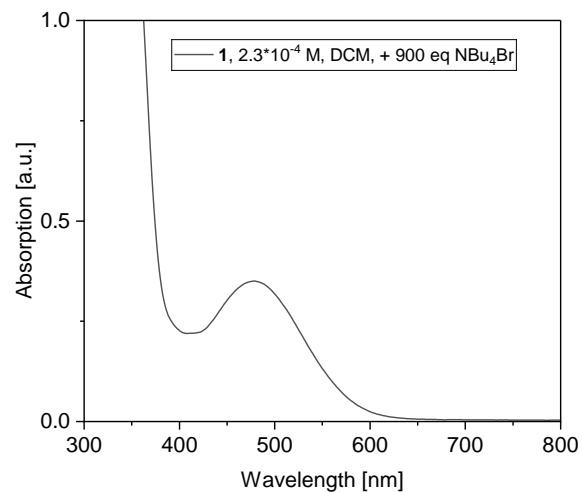


Figure S2.43. UV-vis absorption spectrum of **[1-Br][NBu₄]** (2.3·10⁻⁴ M + 900 eq NBu₄Br, DCM). ϵ (479 nm) = $\frac{0.3503}{2.3 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 7600 \text{ M}^{-1}\text{cm}^{-1}$.

2.16 1-P(*n*Bu)₃

1-P(*n*Bu)₃ could not be isolated. However, the UV-vis spectrum could be measured by using a large excess of tributylphosphine. To a $5.5 \cdot 10^{-4}$ M solution of **1** in DCM were added variable amounts of tributylphosphine. With 150 equivalents, quantitative conversion to 1-P(*n*Bu)₃ was observed.

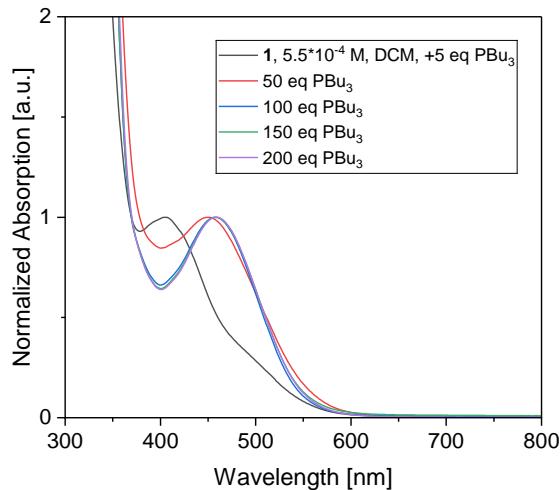


Figure S2.44. Normalised UV-vis absorption spectrum of **1** ($5.5 \cdot 10^{-4}$ M, +5/50/100/150/200 eq. P(*n*Bu)₃, DCM). No significant further change of the absorption spectrum observed between 150 and 200 equivalents of P(*n*Bu)₃. $\lambda_{\text{max}} = 458$ nm.

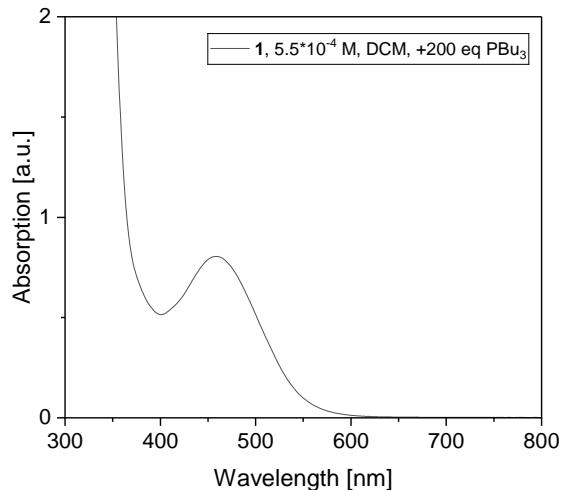


Figure S2.45. UV-vis absorption spectrum of 1-P(*n*Bu)₃ ($5.5 \cdot 10^{-4}$ M + 200 eq P(*n*Bu)₃, DCM). ϵ (458 nm) = $\frac{0.8052}{5.5 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 7300 \text{ M}^{-1}\text{cm}^{-1}$.

2.17 1-DABCO

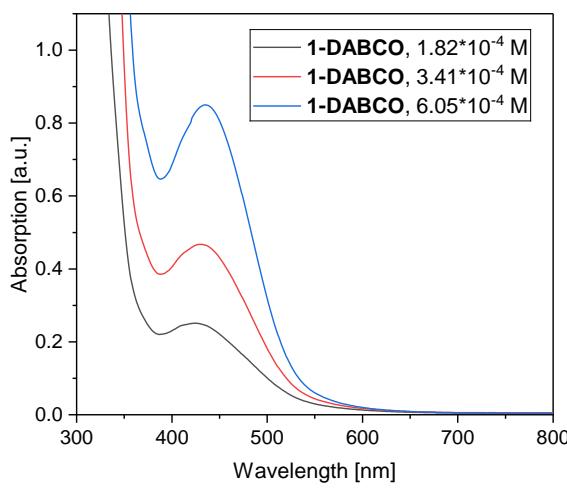


Figure S2.46. UV-vis absorption spectrum of **1-DABCO** ($1.82 \cdot 10^{-4} \text{ M}$, $3.41 \cdot 10^{-4} \text{ M}$, $6.05 \cdot 10^{-4} \text{ M}$, DCM). $\lambda_{\max}(c = 1.82 \cdot 10^{-4} \text{ M}) = 424 \text{ nm}$, $\lambda_{\max}(c = 3.41 \cdot 10^{-4} \text{ M}) = 430 \text{ nm}$, $\lambda_{\max}(c = 6.05 \cdot 10^{-4} \text{ M}) = 435 \text{ nm}$.

Due to dissociation at low concentrations (concentration dependence of λ_{\max} , figure S2.46), additional equivalents of DABCO were added to obtain the spectrum of the fully associated complex.

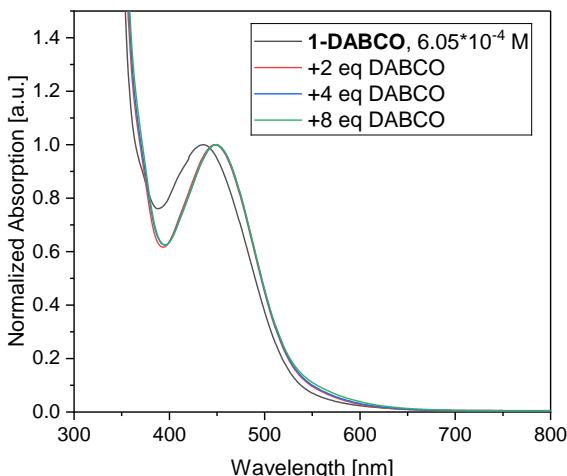


Figure S2.47. Normalised UV-vis absorption spectrum of **1-DABCO** ($6.05 \cdot 10^{-4} \text{ M}$, +2/4/8 eq. DABCO, DCM). No significant further change of the absorption spectrum observed between 4 and 8 equivalents of DABCO.

Stock solution:

2.7 mg **1-DABCO** in 1000 μL DCM.

Cuvette: $d = 2 \text{ mm}$, $V = 700 \mu\text{L}$

Aliquots: 50 μL , 100 μL , 200 μL

Concentrations:

$$\begin{aligned} \text{i)} & \frac{\left(\frac{0.0027 \text{ g}}{991 \text{ g/mol}}\right)}{0.001 \text{ l}} \cdot \frac{50 \mu\text{L}}{750 \mu\text{L}} = 1.816 \cdot 10^{-4} \text{ M} \\ \text{ii)} & \frac{\left(\frac{0.0027 \text{ g}}{991 \text{ g/mol}}\right)}{0.001 \text{ l}} \cdot \frac{100 \mu\text{L}}{800 \mu\text{L}} = 3.406 \cdot 10^{-4} \text{ M} \\ \text{iii)} & \frac{\left(\frac{0.0027 \text{ g}}{991 \text{ g/mol}}\right)}{0.001 \text{ l}} \cdot \frac{200 \mu\text{L}}{900 \mu\text{L}} = 6.054 \cdot 10^{-4} \text{ M} \end{aligned}$$

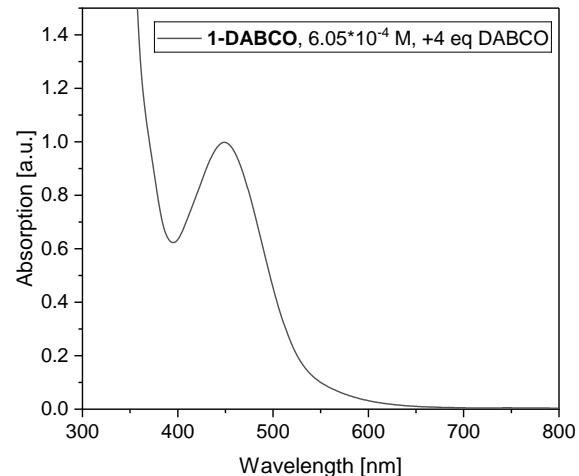


Figure S2.48. UV-vis absorption spectrum of **1-DABCO** ($6.05 \cdot 10^{-4} \text{ M} + 2 \text{ eq DMSO}$, DCM). $\epsilon(449 \text{ nm}) = \frac{0.99803}{6.05 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 8200 \text{ M}^{-1} \text{ cm}^{-1}$.

2.18 1-iPrNHC

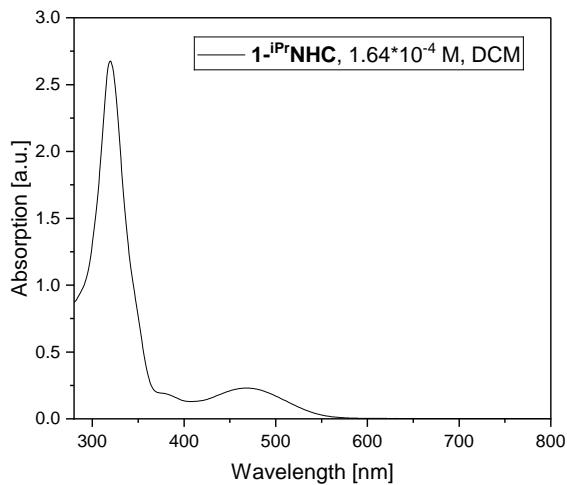


Figure S2.49. UV-vis absorption spectrum of **1-iPrNHC** ($1.64 \cdot 10^{-4}$ M, DCM).

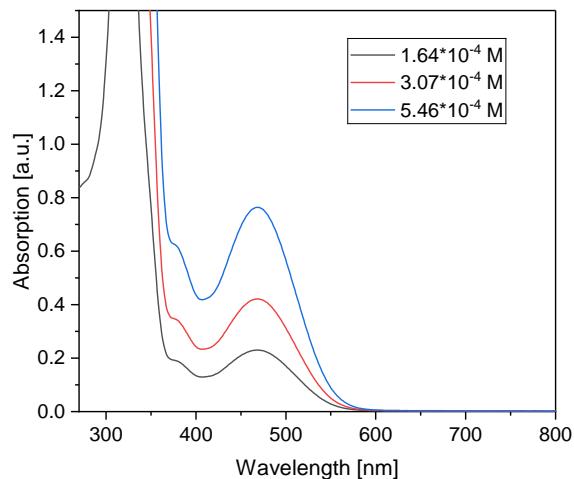


Figure S2.50. UV-vis absorption spectrum of **1-iPrNHC** ($1.64 \cdot 10^{-4}$ M, $3.07 \cdot 10^{-4}$ M, $5.46 \cdot 10^{-4}$ M, DCM). No concentration-dependant dissociation was observed.

Stock solution:

2.6 mg **1-iPrNHC** in 1000 μ L DCM.

Cuvette: d = 2 mm, V = 700 μ L

Aliquots: 50 μ L, 100 μ L, 200 μ L

Concentrations:

$$\text{i)} \quad \left(\frac{0.0026\text{g}}{1059\text{ g/mol}} \right) \cdot \frac{50\text{ }\mu\text{L}}{0.001\text{ l}} \cdot \frac{750\text{ }\mu\text{L}}{750\text{ }\mu\text{L}} = 1.637 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \quad \left(\frac{0.0026\text{g}}{1059\text{ g/mol}} \right) \cdot \frac{100\text{ }\mu\text{L}}{0.001\text{ l}} \cdot \frac{800\text{ }\mu\text{L}}{800\text{ }\mu\text{L}} = 3.069 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \quad \left(\frac{0.0026\text{g}}{1059\text{ g/mol}} \right) \cdot \frac{200\text{ }\mu\text{L}}{0.001\text{ l}} \cdot \frac{900\text{ }\mu\text{L}}{900\text{ }\mu\text{L}} = 5.456 \cdot 10^{-4} \text{ M}$$

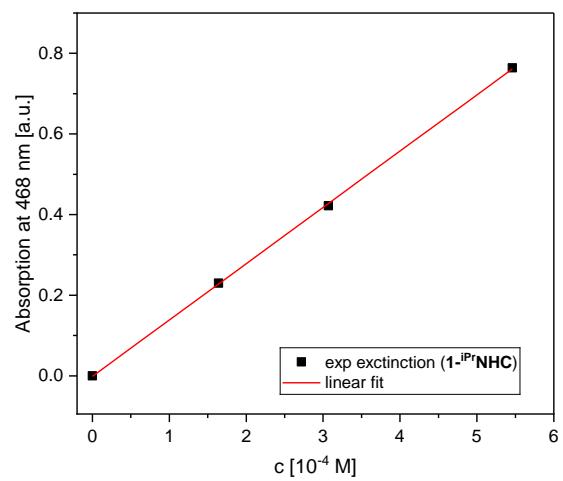


Figure S2.51. Extinction at 468 nm (CT band) of **1-iPrNHC** (0 M, $1.64 \cdot 10^{-4}$ M, $3.07 \cdot 10^{-4}$ M, $5.46 \cdot 10^{-4}$ M, DCM). Linear Fit: $y = 1396.5x - 0.0012$. $R^2 = 0.999$. $\epsilon(468 \text{ nm}) = \frac{1396.5 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 7000 \text{ M}^{-1}\text{cm}^{-1}$.

2.19 1-**PCy**₃

Due to dissociation of **1-PCy**₃ in solution, the UV-vis spectrum of was measured by using a large excess of **PCy**₃. To a $3.9 \cdot 10^{-4}$ M solution of **1-PCy**₃ in DCM were added increasing amounts of **PCy**₃. With 120 equivalents, quantitative conversion to **1-PCy**₃ was observed.

PCy₃ was purified by precipitation as its CS₂ adduct from diethylether, which was filtered off, washed with diethylether and heated to 70 °C under vacuum overnight to obtain oxide-free **PCy**₃. It was further purified by recrystallisation from diethylether at -40 °C. Even with these precautions, a low energy shoulder band appears due to the formation of the corresponding phosphine oxide adduct. However, this impurity was found to not impact the absorption maximum.

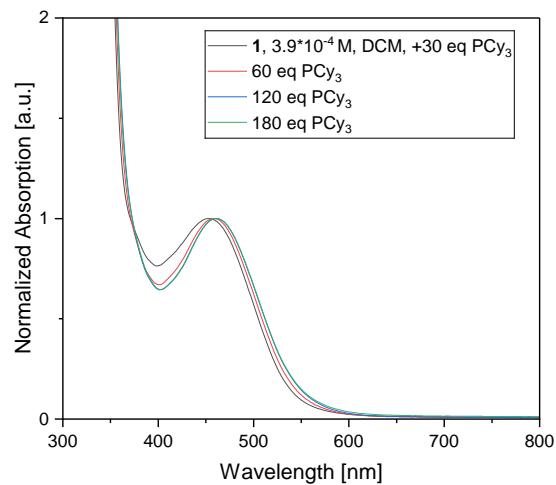


Figure S2.52. Normalised UV-vis absorption spectrum of **1-PCy**₃ ($3.9 \cdot 10^{-4}$ M, +30/60/120/180 eq. **PCy**₃, DCM). No significant further change of the absorption spectrum observed between 120 and 180 equivalents of **PCy**₃. $\lambda_{\text{max}} = 461$ nm.

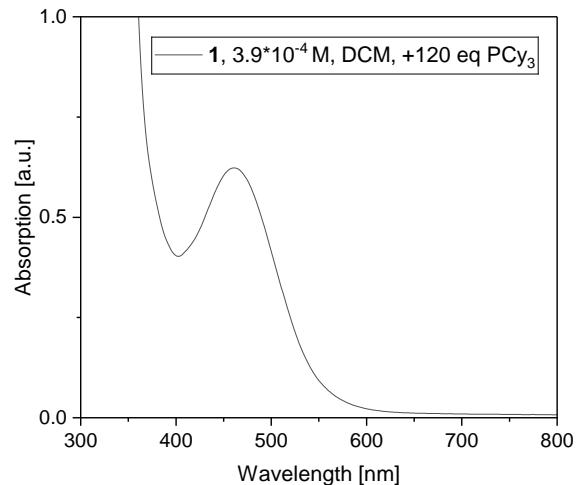


Figure S2.53. UV-vis absorption spectrum of **1-PCy**₃ ($3.9 \cdot 10^{-4}$ M + 120 eq **PCy**₃, DCM). $\epsilon(461 \text{ nm}) = \frac{0.62318}{3.9 \cdot 10^{-4} \text{ M} \cdot 0.2 \text{ cm}} \approx 8000 \text{ M}^{-1}\text{cm}^{-1}$.

2.20 1-SIMes

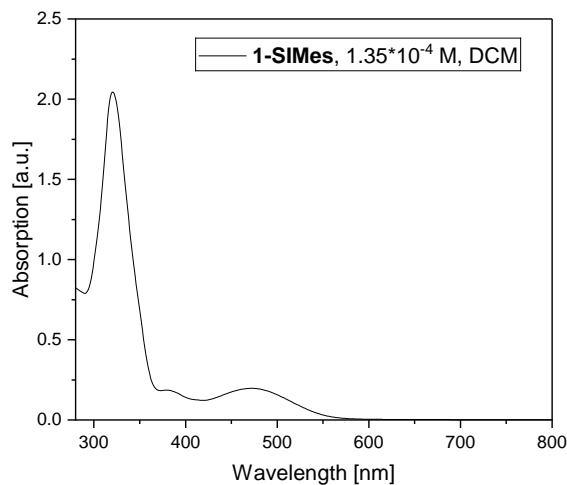


Figure S2.54. UV-vis absorption spectrum of **1-SIMes** ($1.35 \cdot 10^{-4}$ M, DCM).

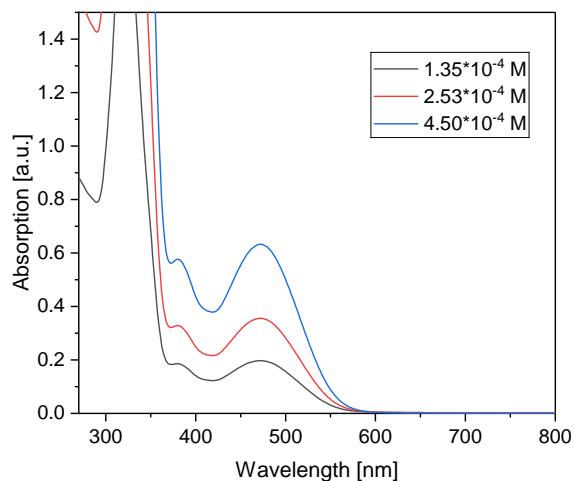


Figure S2.55. UV-vis absorption spectrum of **1-SIMes** (0 M, $1.35 \cdot 10^{-4}$ M, $2.53 \cdot 10^{-4}$ M, $4.50 \cdot 10^{-4}$ M, DCM). No concentration-dependant dissociation was observed.

Stock solution:

2.4 mg **1-SIMes** in 1000 μ L DCM.

Cuvette: d = 2 mm, V = 700 μ L

Aliquots: 50 μ L, 100 μ L, 200 μ L

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0024\text{g}}{1186\text{ g/mol}}\right)}{0.001\text{ l}} \cdot \frac{50\text{ }\mu\text{L}}{750\text{ }\mu\text{L}} = 1.349 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0024\text{g}}{1186\text{ g/mol}}\right)}{0.001\text{ l}} \cdot \frac{100\text{ }\mu\text{L}}{800\text{ }\mu\text{L}} = 2.530 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0024\text{g}}{1186\text{ g/mol}}\right)}{0.001\text{ l}} \cdot \frac{200\text{ }\mu\text{L}}{900\text{ }\mu\text{L}} = 4.497 \cdot 10^{-4} \text{ M}$$

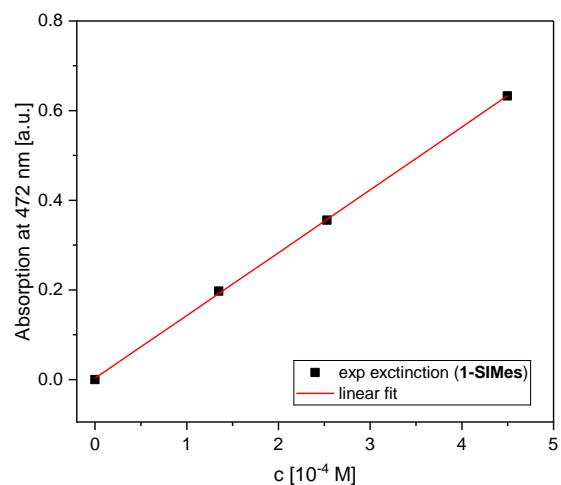


Figure S2.56. Extinction at 472 nm (CT band) of **1-SIMes** (0 M, $1.35 \cdot 10^{-4}$ M, $2.53 \cdot 10^{-4}$ M, $4.50 \cdot 10^{-4}$ M, DCM). Linear Fit: $y = 1401.0x - 0.0037$. $R^2 = 0.999$. $\varepsilon(468\text{ nm}) = \frac{1401.0\text{ M}^{-1}}{0.2\text{ cm}} \approx 7000\text{ M}^{-1}\text{cm}^{-1}$.

2.21 2-((3,5-di-*tert*-butylphenyl)amino)-3-hydroxyanthraquinone (**L₁**)

The absorption spectrum of ligand 2-((3,5-di-*tert*-butylphenyl)amino)-3-hydroxyanthraquinone (**L₁**) was measured in THF due to low solubility in DCM.

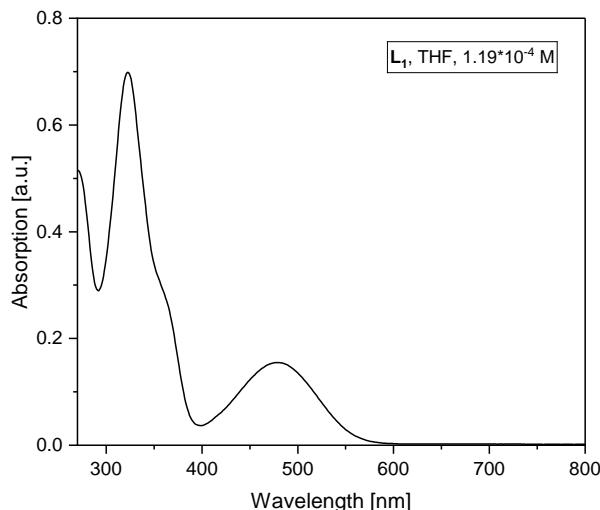


Figure S2.57. UV-vis absorption spectrum of **L₁** (1.19·10⁻⁴ M, THF).

Stock solution:

13.2 mg **L₁** in 20 ml THF.

Cuvette: d = 2 mm, V = 600 µL

Aliquots: 50 µL, 100 µL, 200 µL

Concentrations:

$$\text{i)} \frac{\left(\frac{0.0132\text{g}}{427\text{g/mol}}\right)}{0.02\text{l}} \cdot \frac{50\text{ }\mu\text{L}}{650\text{ }\mu\text{L}} = 1.189 \cdot 10^{-4} \text{ M}$$

$$\text{ii)} \frac{\left(\frac{0.0132\text{g}}{427\text{g/mol}}\right)}{0.02\text{l}} \cdot \frac{100\text{ }\mu\text{L}}{700\text{ }\mu\text{L}} = 2.208 \cdot 10^{-4} \text{ M}$$

$$\text{iii)} \frac{\left(\frac{0.0132\text{g}}{427\text{g/mol}}\right)}{0.02\text{l}} \cdot \frac{200\text{ }\mu\text{L}}{800\text{ }\mu\text{L}} = 3.864 \cdot 10^{-4} \text{ M}$$

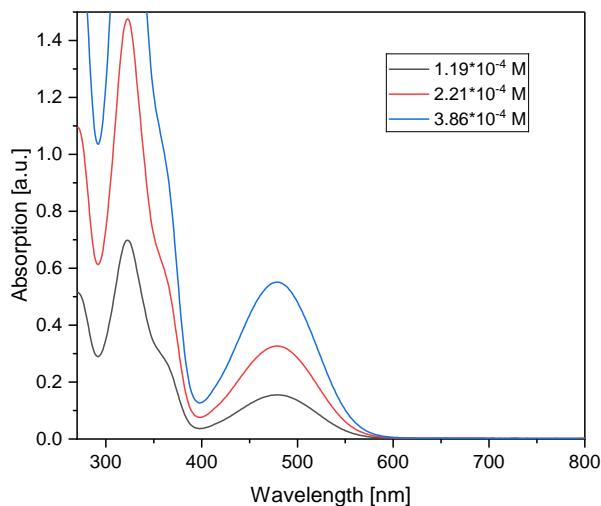


Figure S2.58. Normalised UV-vis absorption spectrum of **L₁** (1.19·10⁻⁴ M, 2.21·10⁻⁴ M, 3.86·10⁻⁴ M, THF).

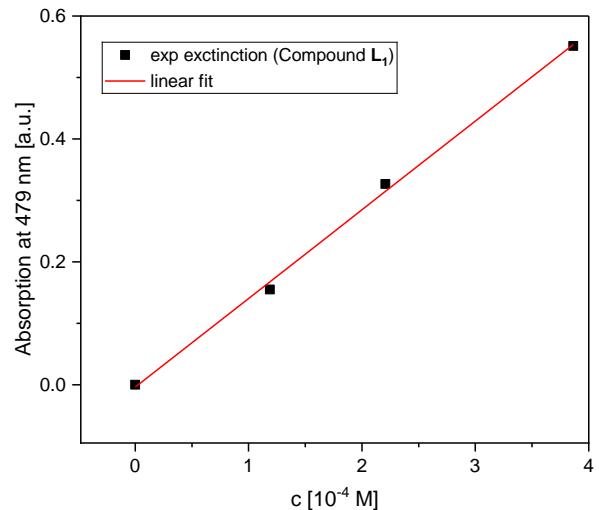


Figure S2.59. Extinction at 479 nm (CT band) of **L₁** (0 M, 1.19·10⁻⁴ M, 2.21·10⁻⁴ M, 3.86·10⁻⁴ M, THF). Linear Fit: $y = 1443.1x - 0.00386$. $R^2 = 0.997$. $\epsilon(479 \text{ nm}) = \frac{1443.1 \text{ M}^{-1}}{0.2 \text{ cm}} \approx 7200 \text{ M}^{-1}\text{cm}^{-1}$.

2.22 UV-vis titration of **1** with PPh_4Cl

To a solution of **1** in DCM (0.58 mM, 760 μL), 0.1 equivalents of PPh_4Cl were added in portions. A 12.5 mM solution of PPh_4Cl in DCM was prepared, of which 3.5 μL were added with each step. The absorption at 489 nm was fitted to a 1:1 binding isotherm with Musketeer 1.8.0,¹ which also provided the graphic for the fitted curve and the RMSD plot.

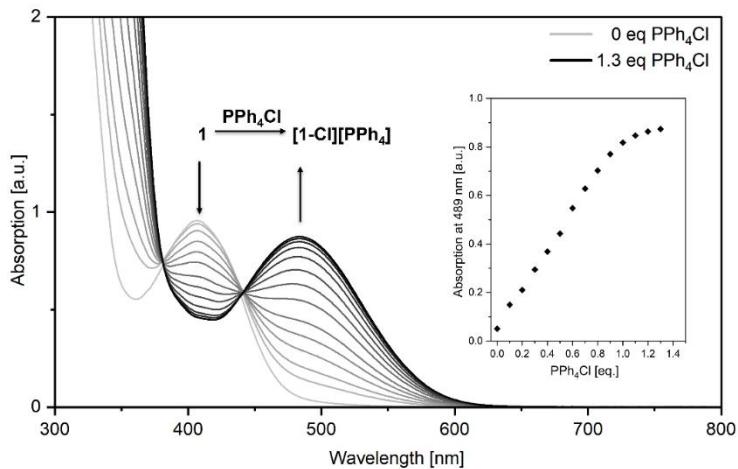


Figure S2.60. UV-vis titration of **1** (0.58 mM, DCM) with PPh_4Cl (12.5 mM, DCM) in steps of 0.1 equivalents PPh_4Cl (3.5 μL of stock solution).

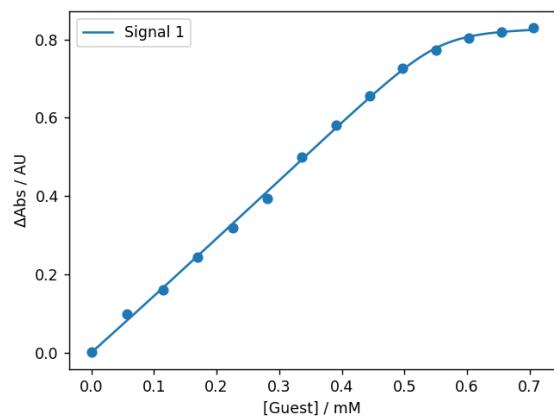


Figure S2.61. Binding isotherm for the absorption at 489 nm, fitted and plotted with *Musketeer* 1.8.0.¹ $K \approx 360000 \text{ M}^{-1}$.

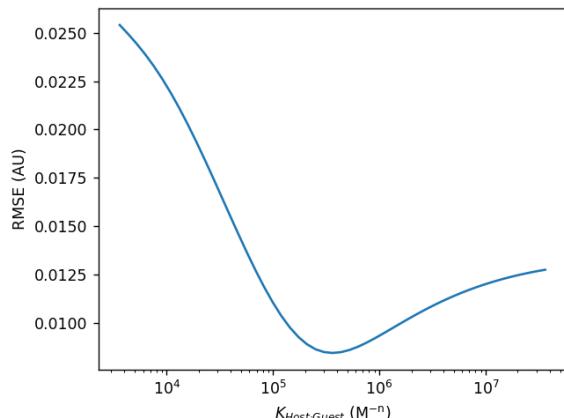
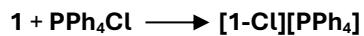


Figure S2.62. RMSE plot for the fitted curve.



$$K_c = \frac{[(\mathbf{1}-\text{Cl})(\text{PPh}_4)]}{[\mathbf{1}][\text{PPh}_4\text{Cl}]} \approx 360000 \text{ M}^{-1}$$

3 Computational Section

3.1 General Information

All quantum chemical calculations were carried out with the Orca 5.0.3 program package.³ The resources of JUSTUS2 (Ulm University) within the Baden-Württemberg High Performance Computing Program (bwHPC) were used. For all calculations, the RI approximation for the coulomb integral (RIJCOSX) together with the corresponding auxiliary basis sets were used.⁴ Geometry Optimisations and subsequent frequency calculations were performed with the r²scan-3c⁵ composite method, confirming the final structures as energetic minima by the absence of imaginary frequencies. If possible, experimental solid-state structures were used as the initial structures for geometry optimisations. Single Point calculations for the calculation of FIA and HIA were performed at the DSD-BLYP(D3BJ)/def2-QZVPP^{6,7} level of theory, with DCM solvation accounted for with the COSMO-RS solvent model.⁸ All other single point calculations were carried out at the DSD-BLYP(D3BJ)/def2-TZVPP⁹ level of theory. Solvent corrected values were obtained with the SMD solvent model and DCM as solvent.¹⁰ Gibbs free energy values were obtained from frequency calculations on the r²scan-3c level of theory using the rigid rotor harmonic oscillator (RRHO) approximation.¹¹ Time-dependant calculations with the Tamm-Danoff approximation (TDA)¹² were carried out with the long-range corrected hybrid functional wB97X-D3,¹³ the def2-TZVPP basis set, and DCM solvation correction with the SMD solvent model. The first 50 electronic transitions were calculated. Molecular Orbitals were visualised with IboView v20211019-RevA. NBO calculations were conducted on the PBE0¹⁴/def2-TZVPP⁹ level of theory with NBO6.0.18_i4¹⁵ as implemented in Orca 5.0.3. Natural Bond Orbitals were visualised with Chemcraft.

Inputs

Geometry Optimisation

```
! r2SCAN-3c RIJCOSX AutoAux TightSCF defgrid3 TightOPT FREQ
```

Single Point Calculation (DCM)

```
! DSD-BLYP D3BJ def2-TZVPP RIJCOSX AutoAux VeryTightSCF
```

```
%cpcm  
smd true  
SMDsolvent "CH2Cl2"  
End
```

NBO Calculation

```
! RKS PBE0 D4 def2-TZVPP NBO VeryTightSCF
```

TD-DFT (DCM)

```
! UKS wB97X-D3 def2-TZVPP RIJCOSX AutoAux TightSCF Engrad KeepDens Largeprint
```

```
%cpcm  
smd true  
SMDsolvent "CH2Cl2"  
End  
  
%tddft  
maxdim 5  
nroots 50  
end
```

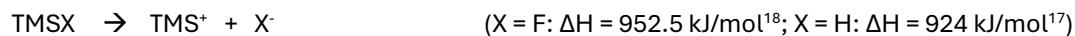
3.2 Fluoride and Hydride Ion Affinities

Fluoride and hydride ion affinities (FIA/HIA) were calculated utilizing isodesmic reactions with the TMS/TMSF and the TMS/TMSH anchor system, respectively, as proposed by *Krossing*¹⁶ and *Greb*.^{17,18} Solvent (DCM) enthalpy correction was applied using the COSMO-RS solvent model. The FIA and HIA values are given as the negative binding enthalpy (FIA/HIA = -ΔH(1 + X → 1-X)). For comparison, affinities for Si(am^Fph^F)₂,¹⁹ Si(cat^{Cl})₂,²⁰ and B(C₆F₅)₃ (BCF) were calculated in addition to compound **1**.

Reaction 1



Reaction 2



$$\text{FIA/HIA (vacuum)} = \Delta H(\text{reaction 2}) - \Delta H(\text{reaction 1})$$

$$\text{FIA/HIA (DCM)} = \text{FIA/HIA (vacuum)} - \Delta H_{\text{solv}}(\text{LA-X}^-) + \Delta H_{\text{solv}}(\text{LA}) + \Delta H_{\text{solv}}(\text{X}^-)$$

Table S3.1. Computed energies for the calculation of ion affinities in vacuum (DSD-BLYP(D3BJ)/def2-TZVPP//r²scan-3c) and in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/COSMO-RS(DCM)//r²scan-3c).

	Charge/multiplicity	Total Correction [kJ/mol] r ² scan-3c	Final Single Point Energy [kJ/mol] DSD-BLYP/def2-TZVPP	Final Enthalpy [kJ/mol] DSD-BLYP/def2-TZVPP/r ² scan-3c	Enthalpy Correction (DCM) [kJ/mol] COSMO-RS(DCM)
1	0/1	2691.90	-7915827.75	-7913133.38	-154.83
1-H	-1/1	2710.86	-7917617.58	-7914904.24	-272.83
1-F	-1/1	2699.12	-8178350.42	-8175648.82	-277.98
Si(am^Fph^F)₂	0/1	670.96	-8558437.37	-8557763.92	-57.35
Si(am^Fph^F)₂-H	-1/1	689.45	-8560283.03	-8559591.11	-158.96
Si(am^Fph^F)₂-F	-1/1	677.61	-8821016.78	-8820336.69	-163.18
Si(cat^{Cl})₂	0/1	324.35	-12411413.54	-12411086.71	-74.59
Si(cat^{Cl})₂-H	-1/1	345.27	-12413251.91	-12412904.17	-196.96
Si(cat^{Cl})₂-F	-1/1	333.40	-12673986.2	-12673650.32	-199.67
BCF	0/1	478.66	-5795523.71	-5795042.57	-46.37
BCF-H	-1/1	500.09	-5797387.46	-5796884.90	-159.45
BCF-F	-1/1	484.17	-6058042.92	-6057556.27	-167.06
TMS	1/1	305.17	-1073010.79	-1072703.15	
TMS-H	0/1	330.35	-1075326.72	-1074993.90	
TMS-F	0/1	318.21	-1336052.36	-1335731.66	
H	-1/1				-204.91
F	-1/1				-343.05

Table S3.2. Computed ion affinities for **1**, Si(am^Fph^F)₂, Si(cat^{Cl})₂, and BCF.

Lewis Acid	FIA(vacuum) [kJ/mol]	FIA(DCM) [kJ/mol]	HIA(vacuum) [kJ/mol]	HIA(DCM) [kJ/mol]
1	439	220	404	317
Si(am^Fph^F)₂	497	260	460	357
Si(cat^{Cl})₂	488	270	451	368
BCF	438	215	476	384

3.3 Thermodynamics

Structures of mono- and bis-adducts were optimised as described above, and single point calculations were carried out with DCM corrections by the SMD solvent model. Geometry optimisation of **1-I₂**, **1-(MeCN)₂**, **1-(ⁱPrNHC)₂**, **1-(SIMes)₂**, **1-(PBu₃)₂**, **1-DABCO** and **1-(PCy₃)₂** resulted in dissociation of one of the donors, suggesting unstable bis-adducts. Convergence failed for **1-(DIBA)₂** and **1-(^{dipp}NHC)₂**.

Table S3.3. Computed energies for the calculation of thermodynamics of mono- and bis-adduct formation in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r²scan-3c).

	Charge/multiplicity	Total Correction [kJ/mol] r ² scan-3c	Entropy Term (T*S) [kJ/mol] r ² scan-3c	Final Single Point Energy [kJ/mol] DSD-BLYP/def2-TZVPP/SMD(DCM)	Final Enthalpy [kJ/mol] DSD-BLYP/def2-TZVPP/SMD(DCM)//r ² scan-3c	Final Gibbs Free Energy [kJ/mol] DSD-BLYP/def2-TZVPP/SMD(DCM)/r ² scan-3c
1	0/1	2691.93	376.26	-7914781.73	-7912087.31	-7912463.57
1-F	-1/1	2699.03	379.85	-8177358.41	-8174656.90	-8175036.76
1-Cl	-1/1	2696.91	383.29	-9122968.37	-9120268.98	-9120652.27
1-Br	-1/1	2696.55	385.63	-14671962.86	-14669263.83	-14669649.46
1-I	-1/1	2696.27	388.32	-8696161.48	-8693462.74	-8693851.06
1-N₃	-1/1	2733.09	390.43	-8346033.55	-8343297.98	-8343688.42
1-NCS	-1/1	2728.38	393.46	-9203773.16	-9201042.29	-9201435.76
1-CN	-1/1	2717.19	386.42	-8158754.49	-8156034.82	-8156421.24
1-MeCN	0/1	2827.27	403.48	-8263073.81	-8260244.06	-8260647.54
1-Et₂O	0/1	3076.11	412.18	-8527759.11	-8524680.52	-8525092.69
1-THF	0/1	3019.52	405.30	-8524607.24	-8521585.23	-8521990.53
1-pyridine	0/1	2945.12	403.87	-8566161.20	-8563213.61	-8563617.48
1-DMAP	0/1	3147.53	427.36	-8917621.67	-8914471.66	-8914899.02
1-OPEt₃	0/1	3263.32	436.42	-9631538.71	-9628272.92	-9628709.34
1-HMPA	0/1	3406.33	459.28	-10067234.50	-10063825.69	-10064284.97
1-DMSO	0/1	2921.68	407.80	-9366471.80	-9363547.64	-9363955.44
1-DIBA	0/1	3517.51	454.65	-9585361.54	-9581841.55	-9582296.20
1-(^{dipp}NHC)	0/1	4422.81	535.51	-11164084.27	-11159658.98	-11160194.49
1-SIMes	0/1	3862.91	484.08	-10342140.89	-10338484.84	-10338968.92
1-(ⁱPrNHC)	0/1	3514.62	448.98	-9333247.78	-9329730.69	-9330179.67
1-P(nBu)₃	0/1	3715.98	463.75	-10052452.89	-10048924.30	-10049388.05
1-PCy₃	0/1	4013.84	470.01	-10661711.95	-10657903.00	-10658373.01
1-DABCO	0/1	3198.77	405.74	-8820534.58	-8817536.44	-8817942.18
1-F₂	-2/1	2701.60	386.58	-8439760.53	-8437056.45	-8437443.04
1-Cl₂	-2/1	2699.10	394.95	-10331036.73	-10328335.15	-10328730.10
1-Br₂	-2/1	2698.78	401.43	-21429036.42	-21426335.16	-21426736.59
1-I₂	-2/1					
1-(N₃)₂	-2/1	2769.67	409.78	-8777157.85	-8774385.70	-8774795.48
1-(NCS)₂	-2/1	2760.83	416.38	-10492655.43	-10489892.12	-10490308.50
1-(CN)₂	-2/1	2738.81	401.48	-8402614.55	-8399873.26	-8400274.74
1-(MeCN)₂	0/1					
1-(Et₂O)₂	0/1	3460.67	447.66	-9140713.18	-9137250.04	-9137697.70
1-(THF)₂	0/1	3346.18	434.43	-9134437.95	-9131089.29	-9131523.72
1-(pyridine)₂	0/1	3196.91	429.12	-9217523.30	-9214323.91	-9214753.04
1-(DMAP)₂	0/1	3598.69	471.51	-9920429.19	-9916828.02	-9917299.53
1-(OPEt₃)₂	0/1	3832.24	497.20	-11348211.94	-11344377.22	-11344874.42
1-(HMPA)₂	0/1	4118.13	532.80	-12219622.14	-12215501.53	-12216034.34
1-(DMSO)₂	0/1	3150.04	444.00	-10818122.56	-10814970.04	-10815414.04
1-(DIBA)₂	0/1					
1-(^{dipp}NHC)₂	0/1					
1-(SIMes)₂	0/1					
1-(ⁱPrNHC)₂	0/1					
1-(P(nBu)₃)₂	0/1					
1-(PCy₃)₂	0/1					
1-(DABCO)₂	0/1					
F	-1/1	3.72	43.37	-262338.92	-262332.73	-262376.10
Cl	-1/1	3.72	45.69	-1208090.35	-1208084.16	-1208129.85
Br	-1/1	3.72	48.71	-6757090.16	-6757083.96	-6757132.67
I	-1/1	3.72	50.43	-781357.71	-781351.51	-781401.94
N₃	-1/1	35.55	63.30	-431123.49	-431085.46	-431148.76
NCS	-1/1	28.34	51.40	-1288879.35	-1288848.53	-1288899.93
CN	-1/1	18.92	58.66	-243851.12	-243829.72	-243888.38
MeCN	0/1	127.66	72.33	-348270.81	-348140.67	-348213.00

Et₂O	0/1	373.99	98.93	-612936.36	-612559.89	-612658.82
THF	0/1	318.33	88.73	-609773.99	-609453.17	-609541.91
Pyridine	0/1	243.07	84.03	-651291.01	-651045.47	-651129.49
DMAP	0/1	444.67	111.79	-1002724.45	-1002277.29	-1002389.08
OPEt₃	0/1	561.89	127.63	-1716634.91	-1716070.54	-1716198.18
HMPA	0/1	703.96	148.44	-2152323.02	-2151616.58	-2151765.03
DMSO	0/1	221.42	92.02	-1451602.32	-1451378.42	-1451470.44
DIBA	0/1	816.62	149.08	-1670474.17	-1669655.06	-1669804.14
dippNHC	0/1	1720.25	242.01	-3249104.24	-3247381.51	-3247623.52
SiMes	0/1	656.73	131.33	-1211978.69	-2426170.76	-2426366.86
iPrNHC	0/1	811.34	146.62	-1418280.44	-1417466.63	-1417613.25
P(nBu)₃	0/1	1014.69	168.04	-2137702.66	-2136724.40	-2136892.44
PCy₃	0/1	1312.68	170.04	-2746943.38	-2745680.86	-2745850.90
DABCO	0/1	494.80	98.40	-905811.95	-905351.75	-905450.15

Table S3.4. Computed thermodynamics of mono- and bis-adduct formation of **1** with donors **X** in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r2scan-3c).

X	ΔH (DCM) [kJ/mol] 1 + X → 1-X	ΔG (DCM) [kJ/mol] 1 + X → 1-X	ΔH (DCM) [kJ/mol] 1-X + X → 1-X₂	ΔG (DCM) [kJ/mol] 1-X + X → 1-X₂
F-	-236.9	-197.1	-66.8	-30.2
Cl-	-97.5	-58.9	18.0	52.0
Br-	-92.6	-53.2	12.6	45.5
I-	-23.9	14.5		
N₃-	-125.2	-76.1	-2.3	41.7
NCS-	-106.5	-72.3	-1.3	27.2
CN-	-117.8	-69.3	-8.7	34.9
MeCN	-16.1	29.0		
Et₂O	-33.3	29.7	-9.6	53.8
THF	-44.7	14.9	-50.9	8.7
Pyridine	-80.8	-24.4	-64.8	-6.1
DMAP	-107.1	-46.4	-79.1	-11.4
OPEt₃	-115.1	-47.6	-33.8	33.1
HMPA	-121.8	-56.4	-59.3	15.7
DMSO	-81.9	-21.4	-44.0	11.8
DIBA	-99.2	-28.5		
dippNHC	-190.2	-107.4		
SiMes	-226.8	-138.5		
iPrNHC	-176.8	-102.9		
P(nBu)₃	-112.6	-32.0		
PCy₃	-134.8	-58.5		
DABCO	-97.4	-28.5		

Interaction Energy (E_{INT}) and Deformation Energy (E_{def}) for Experimentally Realised Complexes of 1

For the interaction energy, the Lewis acid and base fragments were taken from the optimised structures of the adducts and the single point energy was calculated in the gas phase for each fragment.

$$E_{\text{def}} = E(\text{deformed}) - E(\text{Opt})$$

$$E_{\text{INT}} = E(\text{adduct}) - E(\text{deformed base}) - E(\text{deformed acid}) = E(\text{adduct}) - E(\text{acid}) - E(\text{base}) - E_{\text{def}}(\text{acid}) - E_{\text{def}}(\text{base})$$

Table S3.5. Deformation energy (E_{def}) for acid/base fragments and interaction energy (E_{INT}) for Lewis adducts **1-X** (DSD-BLYP(D3BJ)/def2-TZVPP).

Adduct	SP energy (adduct) [kJ/mol]	SP energy (acid) [kJ/mol]	SP energy (base) [kJ/mol]	SP energy (deformed acid) [kJ/mol]	SP energy (deformed base) [kJ/mol]	E_{def} (acid) [kJ/mol]	E_{def} (base) [kJ/mol]	E_{INT} [kJ/mol]
1-F	-8177100.26	-7914624.21	-261991.10	-7914434.67	-261991.10	189.5	0.0	-674.5
1-Cl	-9122708.25	-7914624.21	-1207829.25	-7914439.88	-1207829.25	184.3	0.0	-439.1
1-Br	-14671706.02	-7914624.21	-6756878.62	-7914447.75	-6756878.62	176.5	0.0	-379.7
1-N₃	-8345790.97	-7914624.21	-430894.33	-7914437.06	-430878.53	187.2	15.8	-475.4
1-NCS	-9203523.74	-7914624.21	-1288665.14	-7914445.51	-1288658.44	178.7	6.7	-419.8
1-CN	-8158499.97	-7914624.21	-243593.83	-7914446.58	-243591.94	177.6	1.9	-461.5
1-pyridine	-8565960.65	-7914624.21	-651261.51	-7914490.80	-651257.62	133.4	3.9	-212.2
1-DMAP	-8917409.43	-7914624.21	-1002681.09	-7914482.39	-1002674.58	141.8	6.5	-252.5
1-OPEt₃	-9631327.09	-7914624.21	-1716591.55	-7914471.60	-1716548.73	152.6	42.8	-306.8
1-HMPA	-10067035.08	-7914624.21	-2152283.84	-7914468.70	-2152257.95	155.5	25.9	-308.4
1-DMSO	-9366273.53	-7914624.21	-1451569.82	-7914488.50	-1451549.56	135.7	20.3	-235.5
1-DIBA	-9585154.37	-7914624.21	-1670405.06	-7914468.07	-1670395.20	156.1	9.9	-291.1
1-dippNHC	-11163857.69	-7914624.21	-3249014.51	-7914455.15	-3248973.22	169.1	41.3	-429.3
1-SIMes	-10342140.89	-7914624.21	-2427239.34	-7914473.45	-2427228.31	150.8	11.0	-439.1
1-iPrNHC	-9333044.13	-7914624.21	-1418210.88	-7914468.39	-1418200.61	155.8	10.3	-375.1
1-P(nBu)₃	-10052452.89	-7914624.21	-2137702.66	-7914492.78	-2137686.87	131.4	15.8	-273.2
1-PCy₃	-10661711.95	-7914624.21	-2746943.38	-7914483.51	-2746927.28	140.7	16.1	-301.2
1-DABCO	-8820534.58	-7914624.21	-905811.95	-7914482.79	-905802.71	141.4	9.2	-249.1

Vacuum Affinities of 1

Table S3.6. Computed energies for the calculation of thermodynamics of mono-adduct formation in vacuum (DSD-BLYP(D3BJ)/def2-TZVPP//r²scan-3c).

	Charge/multiplicity	Total Correction [kJ/mol] r ² scan-3c	Entropy Term (T*S) [kJ/mol] r ² scan-3c	Final Single Point Energy [kJ/mol] DSD-BLYP/def2-TZVPP	Final Enthalpy [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c	Final Gibbs Free Energy [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c
1	0/1	2691.93	376.26	-7914623.05	-7911928.64	-7912304.90
1-F	-1/1	2699.03	379.85	-8177099.06	-8174397.55	-8174777.40
1-Cl	-1/1	2696.91	383.29	-9122706.91	-9120007.52	-9120390.81
1-N₃	-1/1	2733.09	390.43	-8345789.74	-8343054.17	-8343444.60
1-NCS	-1/1	2728.38	393.46	-9203522.38	-9200791.52	-9201184.98
1-CN	-1/1	2717.19	386.42	-8158498.77	-8155779.10	-8156165.52
1-Br	-1/1	2696.55	385.63	-14671706.02	-14669007.08	-14669391.29
1-pyridine	0/1	2945.12	403.87	-8565959.39	-8563011.80	-8563415.67
1-DMAP	0/1	3147.53	427.36	-8917408.12	-8914258.11	-8914685.47
1-OPEt₃	0/1	3263.32	436.42	-9631325.68	-9628059.88	-9628496.31
1-HMPA	0/1	3406.33	459.28	-10067033.60	-10063624.79	-10064084.07
1-DMSO	0/1	2921.68	407.80	-9366272.16	-9363348.00	-9363755.80
1-DIBA	0/1	3517.51	454.65	-9585152.96	-9581632.97	-9582087.62
1-dippNHC	0/1	4422.81	535.51	-11163872.38	-11159447.09	-11159982.61
1-SIMes	0/1	3862.91	484.08	-10342140.89	-10338275.5	-10338759.58
1-iPrNHC	0/1	3514.62	448.98	-9333044.13	-9329527.04	-9329976.02
1-P(nBu)₃	0/1	3715.98	463.75	-10052452.89	-10048734.44	-10049198.19
1-PCy₃	0/1	4013.84	470.01	-10661711.95	-10657695.63	-10658165.64
1-DABCO	0/1	3198.77	405.74	-8820534.58	-8817333.33	-8817739.07
F	-1/1	3.72	43.37	-261991.06	-261984.86	-262028.23
Cl	-1/1	3.72	45.69	-1207829.07	-1207822.88	-1207868.57
N₃	-1/1	35.55	63.30	-430894.27	-430856.24	-430919.54
NCS	-1/1	28.34	51.40	-1288664.95	-1288634.14	-1288685.53
CN	-1/1	18.92	58.66	-243593.80	-243572.40	-243631.06
Br	-1/1	3.72	48.71	-6756878.62	-6756872.42	-6756921.13
Pyridine	0/1	243.07	84.03	-651261.41	-651015.87	-651099.89
DMAP	0/1	444.67	111.79	-1002680.94	-1002233.79	-1002345.58
OPEt₃	0/1	561.89	127.63	-1716591.30	-1716026.93	-1716154.56
HMPA	0/1	703.96	148.44	-2152283.52	-2151577.08	-2151725.52
DMSO	0/1	221.42	92.02	-1451569.61	-1451345.71	-1451437.73
DIBA	0/1	816.62	149.08	-1670420.22	-1669601.12	-1669750.19
dippNHC	0/1	1720.25	242.01	-3249014.03	-3247291.31	-3247533.32
SIMes	0/1	656.73	131.33	-2427239.34	-2426078.67	-2426274.77
iPrNHC	0/1	811.34	146.62	-1418210.88	-1417397.07	-1417543.69
P(nBu)₃	0/1	1014.69	168.04	-2137702.66	-2136685.49	-2136853.53
PCy₃	0/1	1312.68	170.04	-2746943.38	-2745628.22	-2745798.26
DABCO	0/1	494.80	98.40	-905811.95	-905314.67	-905413.07

Table S3.7. Computed thermodynamics of mono-adduct (**1-X**) formation in vacuum (DSD-BLYP(D3BJ)/def2-TZVPP//r2scan-3c).

X	ΔH [kJ/mol] 1 + X → 1-X	ΔG [kJ/mol] 1 + X → 1-X
F-	-484.0	-444.3
Cl-	-256.0	-217.3
Br-	-206.0	-165.3
N₃-	-269.3	-220.2
NCS-	-228.7	-194.6
CN-	-278.1	-229.6
Pyridine	-67.3	-10.9
DMAP	-95.7	-35.0
OPEt₃	-104.3	-36.8
HMPA	-119.1	-53.7
DMSO	-73.7	-13.2
DIBA	-103.2	-32.5
dippNHC	-227.1	-144.4
SIMes	-268.2	-179.9
iPrNHC	-201.3	-127.4
P(nBu)₃	-120.3	-39.8
PCy₃	-138.8	-62.5
DABCO	-90.0	-21.1

Thermodynamics of Adduct Formation: COSMO-RS Solvation

Table S3.8. Computed energies for the calculation of thermodynamics of mono-adduct formation in vacuum (DSD-BLYP(D3BJ)/def2-TZVPP//r²scan-3c).

	Charge/multiplicity	Final Enthalpy (Vacuum) [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c	Final Gibbs Free Energy (Vacuum) [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c	Enthalpy Correction [kJ/mol] COSMO-RS (DCM)	Gibbs Free Energy Correction [kJ/mol] COSMO-RS (DCM)	Final Enthalpy (DCM) [kJ/mol] DSD-BLYP/def2-TZVPP/COSMO-RS(DCM)//r ² scan-3c	Final Gibbs Free Energy (Vacuum) [kJ/mol] DSD-BLYP/def2-TZVPP/COSMO-RS(DCM)//r ² scan-3c
1	0/1	-7911928.64	-7912304.90	-154.19	-141.28	-7912082.83	-7912446.18
1-F	-1/1	-8174397.55	-8174777.40	-277.70	-259.84	-8174675.25	-8175037.24
1-Cl	-1/1	-9120007.52	-9120390.81	-280.18	-262.94	-9120287.70	-9120653.75
1-N₃	-1/1	-8343054.17	-8343444.60	-274.76	-257.58	-8343328.93	-8343702.18
1-NCS	-1/1	-9200791.52	-9201184.98	-272.93	-258.29	-9201064.45	-9201443.27
1-CN	-1/1	-8155779.10	-8156165.52	-283.53	-264.83	-8156062.63	-8156430.35
1-pyridine	0/1	-8563011.80	-8563415.67	-198.26	-181.04	-8563210.06	-8563596.71
1-DMAP	0/1	-8914258.11	-8914685.47	-216.39	-198.65	-8914474.50	-8914884.12
1-OPEt₃	0/1	-9628059.88	-9628496.31	-215.10	-195.95	-9628274.98	-9628692.26
1-HMPA	0/1	-10063624.79	-10064084.07	-208.90	-191.40	-10063833.69	-10064275.47
1-DMSO	0/1	-9363348.00	-9363755.80	-205.10	-185.75	-9363553.10	-9363941.55
1-DIBA	0/1	-9581632.97	-9582087.62	-203.84	-186.76	-9581836.81	-9582274.38
1-dippNHC	0/1	-11159447.09	-11159982.61	-196.86	-180.33	-11159643.95	-11160162.94
F	-1/1	-261984.86	-262028.23	-343.05	-317.14	-262327.91	-262345.37
Cl	-1/1	-1207822.88	-1207868.57	-304.70	-277.22	-1208127.58	-1208145.79
N₃	-1/1	-430856.24	-430919.54	-269.02	-243.41	-431125.26	-431162.95
NCS	-1/1	-1288634.14	-1288685.53	-249.61	-226.64	-1288883.75	-1288912.17
CN	-1/1	-243572.40	-243631.06	-288.02	-261.14	-243860.42	-243892.20
Pyridine	0/1	-651015.87	-651099.89	-27.98	-18.91	-651043.85	-651118.80
DMAP	0/1	-1002233.79	-1002345.58	-45.57	-34.62	-1002279.36	-1002380.20
OPEt₃	0/1	-1716026.93	-1716154.56	-55.24	-42.51	-1716082.17	-1716197.07
HMPA	0/1	-2151577.08	-2151725.52	-54.20	-42.96	-2151631.28	-2151768.48
DMSO	0/1	-1451345.71	-1451437.73	-46.90	-33.25	-1451392.61	-1451470.98
DIBA	0/1	-1669601.12	-1669750.19	-58.09	-47.39	-1669659.21	-1669797.58
NHC	0/1	-3247291.31	-3247533.32	-76.31	-34.62	-3247367.62	-3247567.94

Table S3.9. Computed thermodynamics of mono-adduct (**1-X**) formation in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/COSMO-RS(DCM)//r²scan-3c).

X	ΔH [kJ/mol] 1 + X → 1-X	ΔG [kJ/mol] 1 + X → 1-X
F-	-264.5	-245.7
Cl-	-77.3	-61.8
N₃-	-120.8	-93.0
NCS-	-97.9	-84.9
CN-	-119.4	-92.0
Pyridine	-83.4	-31.7
DMAP	-112.3	-57.7
OPEt₃	-110.0	-49.0
HMPA	-119.6	-60.8
DMSO	-77.7	-24.4
DIBA	-94.8	-30.6
NHC	-193.5	-148.8

Thermodynamics of Adduct Formation: Explicit Cations

For geometry optimisations of ion pairs, SMD solvation modelling with DCM as solvent was used and numerical frequency analysis was performed. Not all imaginary frequencies could be removed, giving unreliable values for entropy and Gibbs free energy. However, imaginary frequencies were found sufficiently small to give estimates of reaction enthalpies.

Table S3.10. Computed energies for the calculation of thermodynamics of mono-adduct formation with explicit cations in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r²scan-3c/SMD(DCM)).

	Charge/multiplicity	Total Correction [kJ/mol] r ² scan-3c/SMD(DCM)	Entropy Term (T*S) [kJ/mol] r ² scan-3c/SMD(DCM)	Final Single Point Energy [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c/SMD(DCM)	Final Enthalpy [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c/SMD(DCM)	Final Gibbs Free Energy [kJ/mol] DSD-BLYP/def2-TZVPP//r ² scan-3c/SMD(DCM)
1	0/1	2679.3489	367.3269	-7914781.3586	-7912099.5308	-7912466.858
[1-F][NBu₄]	0/1	4074.7763	481.4077	-9976711.1285	-9972633.8732	-9973115.281
[1-Cl][PPh₄]	0/1	3711.7016	492.5844	-12448919.6054	-12445205.4249	-12445698.01
[1-N3][NBu₄]	0/1	4111.3101	501.1021	-10145390.9150	-10141277.1259	-10141778.23
[1-NCS][NBu₄]	0/1	4103.3370	513.6363	-11003129.3359	-10999023.5200	-10999537.16
[1-CN][NBu₄]	0/1	4092.7311	501.5246	-9958118.5382	-9954023.3282	-9954524.853
NBu₄F	0/1	1385.4121	200.4735	-2061692.1510	-2060304.2599	-2060504.733
PPh₄Cl	0/1	1021.1276	194.8645	-4534010.2040	-4532986.5975	-4533181.462
NBu₄N₃	0/1	1420.2356	213.4844	-2230468.8926	-2229046.1781	-2229259.663
NBu₄NCS	0/1	1414.5993	221.3629	-3088227.7025	-3086811.4526	-3087025.166
NBu₄CN	0/1	1402.8911	212.7395	-2043184.2791	-2041778.9090	-2041991.649

Table S3.11. Computed Enthalpy of mono-adduct formation with explicit cations (**[1-X][Cation]**) in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r²scan-3c/SMD(DCM)).

X	ΔH [kJ/mol] 1 + X → 1-X
NBu ₄ F	-230.1
PPh ₄ Cl	-119.3
NBu ₄ N ₃	-131.4
NBu ₄ NCS	-112.5
NBu ₄ CN	-144.9

Computed Structural Parameters

Table S3.12. Computed structural parameters of **1** and mono-adducts **1-X**, optimised on the r2scan-3c level of theory.

1-X	Average Si-N bond length [Å]	Average Si-O bond length [Å]	Si-X bond length [Å]	N-Si-N bond angle [°]
1	1.65974	1.69568		121.012
1-F	1.78286	1.78729	1.62247	118.157
1-Cl	1.77668	1.78368	2.13704	118.345
1-Br	1.79132	1.77168	2.32056	117.898
1-N₃	1.79306	1.78348	1.78935	117.107
1-NCS	1.77500	1.78285	1.76389	118.813
1-CN	1.77798	1.78236	1.88619	118.186
1-pyridine	1.75017	1.75758	1.93397	132.547
1-DMAP	1.75671	1.76237	1.91015	127.657
1-OPEt₃	1.76603	1.76779	1.76480	123.175
1-HMPA	1.76109	1.76954	1.76459	125.825
1-DMSO	1.75018	1.76060	1.82686	124.358
1-DIBA	1.75781	1.76870	1.78578	128.580
1-dippNHC	1.74881	1.77334	1.99259	120.275
1-SIMes	1.78141	1.76753	1.98452	119.357
1-iPr'NHC	1.77903	1.77559	1.93972	120.954
1-P(nBu)₃	1.77569	1.75407	2.41544	119.582
1-PCy₃	1.77766	1.76434	2.37355	120.330
1-DABCO	1.77022	1.74949	2.00216	126.368

Para-Fluorophenol Hydrogen Bonding Thermodynamics

Table S3.13. Computed energies for the calculation of thermodynamics of hydrogen bond complex formation between para-fluorophenol and Lewis bases in vacuum, DCM and CCl₄ (DSD-BLYP(D3BJ)/def2-TZVPP//r²scan-3c with corresponding SMD solvation correction).

	Charge/multiplicity	Total Correction [kJ/mol] r ² scan-3c	Entropy Term (T*S) [kJ/mol] r ² scan-3c	Final Single Point Energy Vacuum [kJ/mol] DSD-BLYP/def2-TZVPP	Final Single Point Energy DCM [kJ/mol] DSD-BLYP/def2-TZVPP/SMD(DCM)	Final Single Point Energy CCl ₄ [kJ/mol] DSD-BLYP/def2-TZVPP/SMD(CCl ₄)
p-F-PhOH	0/1	268.89	98.85	-1067013.1	-1067048.52	-1067040.21
p-F-PhOH-Br	-1/1	273.69	115.74	-7824001.21	-7824189.66	-7824127.68
p-F-PhOH-Cl	-1/1	273.17	112.73	-2274974.83	-2275185.37	-2275112.84
p-F-PhOH-CN	-1/1	288.02	118.26	-1310730.67	-1310928.07	-1310856.26
p-F-PhOH-F	-1/1	268.84	109.86	-1329264.37	-1329472.22	-1329397.54
p-F-PhOH-N ₃	-1/1	306.45	122.96	-1498024.52	-1498207.8	-1498138.01
p-F-PhOH-NCS	-1/1	303.76	127.44	-2355780.14	-2355963.98	-2355901.71
p-F-PhOH-DABCO	0/1	772.09	147.71	-1972877.95	-1972942.76	-1972930.16
p-F-PhOH-DIBA	0/1	1094.08	193.9	-2737495.33	-2737567.87	-2737556.89
p-F-PhOH-DMAP	0/1	721.93	163.72	-2069747.91	-2069815.56	-2069804.25
p-F-PhOH-DMSO	0/1	498.8	141.34	-2518641.35	-2518690.89	-2518679.69
p-F-PhOH-HMPA	0/1	981.6	196.75	-3219364.38	-3219420.24	-3219411.49
p-F-PhOH- ^{iPr} NHC	0/1	1086.2	192.66	-2485296.56	-2485371.7	-2485360.61
p-F-PhOH- ^{dipp} NHC	0/1	1995.42	285.06	-4316114.76	-4316214.98	-4316200.83
p-F-PhOH-OPEt ₃	0/1	839.66	172.7	-2783652.78	-2783716.49	-2783703.28
p-F-PhOH-PBu ₃	0/1	1290.85	213.01	-3204764.35	-3204825.51	-3204820.38
p-F-PhOH-PCy ₃	0/1	1589.22	214.85	-3814000.29	-3814074.18	-3814069.53
p-F-PhOH-Pyridine	0/1	520.46	138.51	-1718320.82	-1718376.75	-1718368.91
p-F-PhOH-SIMes	0/1	1433.63	235.08	-3494337.14	-3494434.58	-3494420.56
Br	-1/1	3.72	48.71	-6756878.62	-6757090.16	-6757014.72
Cl	-1/1	3.72	45.69	-1207829.25	-1208090.53	-1207994.25
CN	-1/1	18.92	58.66	-243593.83	-243851.29	-243751.68
F	-1/1	3.72	43.37	-261991.08	-262338.94	-262206.55
N3	-1/1	35.55	63.3	-430894.34	-431123.63	-431029.83
NCS	-1/1	28.34	51.4	-1288665.13	-1288879.67	-1288798.77
DABCO	0/1	494.8	98.4	-905811.95	-905849.03	-905839.78
DIBA	0/1	808.72	142.96	-1670405.03	-1670458.69	-1670450.46
DMAP	0/1	444.67	111.79	-1002681.15	-1002724.75	-1002714.11
DMSO	0/1	221.42	92.02	-1451569.8	-1451602.54	-1451591.47
HMPA	0/1	703.96	148.44	-2152283.78	-2152323.36	-2152314.8
^{iPr} NHC	0/1	811.34	146.62	-1418210.88	-1418280.44	-1418269.31
^{dipp} NHC	0/1	1720.25	242.01	-3249014.48	-3249104.77	-3249093.03
OPEt ₃	0/1	561.89	127.63	-1716591.55	-1716635.19	-1716623.33
PBu ₃	0/1	1014.69	168.04	-2137702.66	-2137741.57	-2137740.14
PCy ₃	0/1	1312.68	170.04	-2746943.38	-2746996.02	-2746993.9
Pyridine	0/1	243.07	84.03	-651261.52	-651291.1	-651285.03
SIMes	0/1	1158.2	196.1	-2427239.34	-2427331.43	-2427319.52
	Final Enthalpy (vacuum) DSD-BLYP/def2-TZVPP//r ² scan-3c	Final Gibbs Free Energy (vacuum) DSD-BLYP/def2-TZVPP//r ² scan-3c	Final Enthalpy (DCM) DSD-BLYP/def2-TZVPP//r ² scan-3c	Final Gibbs Free Energy (DCM) DSD-BLYP/def2-TZVPP/SMD(DCM)//r ² scan-3c	Final Enthalpy (CCl ₄) DSD-BLYP/def2-TZVPP/SMD(CCl ₄)//r ² scan-3c	Final Gibbs Free Energy (CCl ₄) DSD-BLYP/def2-TZVPP/SMD(CCl ₄)//r ² scan-3c
p-F-PhOH	-1066741.73	-1066840.59	-1066777.15	-1066876.01	-1066768.84	-1066867.7
p-F-PhOH-Br	-7823725.04	-7823840.78	-7823913.49	-7824029.23	-7823851.51	-7823967.25
p-F-PhOH-Cl	-2274699.18	-2274811.91	-2274909.72	-2275022.45	-2274837.19	-2274949.92
p-F-PhOH-CN	-1310440.18	-1310558.44	-1310637.58	-1310755.84	-1310565.77	-1310684.03
p-F-PhOH-F	-1328993.05	-1329102.91	-1329200.9	-1329310.76	-1329126.22	-1329236.08
p-F-PhOH-N ₃	-1497715.59	-1497838.54	-1497898.87	-1498021.82	-1497829.08	-1497952.03
p-F-PhOH-NCS	-2355473.9	-2355601.34	-2355657.74	-2355785.18	-2355595.47	-2355722.91
p-F-PhOH-DABCO	-1972103.38	-1972251.09	-1972168.19	-1972315.9	-1972155.59	-1972303.3
p-F-PhOH-DIBA	-2736398.77	-2736592.67	-2736471.31	-2736665.21	-2736460.33	-2736654.23
p-F-PhOH-DMAP	-2069023.5	-2069187.22	-2069091.15	-2069254.87	-2069079.84	-2069243.56
p-F-PhOH-DMSO	-2518140.08	-2518281.42	-2518189.62	-2518330.96	-2518178.42	-2518319.76
p-F-PhOH-HMPA	-3218380.3	-3218577.05	-3218436.16	-3218632.91	-3218427.41	-3218624.16
p-F-PhOH- ^{iPr} NHC	-2484207.88	-2484400.53	-2484283.02	-2484475.67	-2484271.93	-2484464.58
p-F-PhOH- ^{dipp} NHC	-4314116.86	-4314401.92	-4314217.08	-4314502.14	-4314202.93	-4314487.99
p-F-PhOH-OPEt ₃	-2782810.64	-2782983.34	-2782874.35	-2783047.05	-2782861.14	-2783033.84

p-F-PhOH-PBu ₃	-3203471.03	-3203684.03	-3203532.19	-3203745.19	-3203527.06	-3203740.06
p-F-PhOH- PCy ₃	-3812408.59	-3812623.44	-3812482.48	-3812697.33	-3812477.83	-3812692.68
p-F-PhOH-Pyridine	-1717797.89	-1717936.39	-1717853.82	-1717992.32	-1717845.98	-1717984.48
p-F-PhOH-SIMes	-3492901.02	-3493136.11	-3492998.46	-3493233.55	-3492984.44	-3493219.53
Br	-6756872.42	-6756921.13	-6757083.96	-6757132.67	-6757008.52	-6757057.23
Cl	-1207823.05	-1207868.74	-1208084.33	-1208130.02	-1207988.05	-1208033.74
CN	-243572.43	-243631.1	-243829.89	-243888.56	-243730.28	-243788.95
F	-261984.89	-262028.26	-262332.75	-262376.12	-262200.36	-262243.73
N3	-430856.31	-430919.61	-431085.6	-431148.9	-430991.8	-431055.1
NCS	-1288634.31	-1288685.71	-1288848.85	-1288900.25	-1288767.95	-1288819.35
DABCO	-905314.67	-905413.07	-905351.75	-905450.15	-905342.5	-905440.9
DIBA	-1669593.84	-1669736.8	-1669647.5	-1669790.46	-1669639.27	-1669782.23
DMAP	-1002234	-1002345.78	-1002277.6	-1002389.38	-1002266.96	-1002378.74
DMSO	-1451345.9	-1451437.92	-1451378.64	-1451470.66	-1451367.57	-1451459.59
HMPA	-2151577.34	-2151725.78	-2151616.92	-2151765.36	-2151608.36	-2151756.8
iPrNHC	-1417397.07	-1417543.69	-1417466.63	-1417613.25	-1417455.5	-1417602.12
dippNHC	-3247291.75	-3247533.76	-3247382.04	-3247624.05	-3247370.3	-3247612.31
OPEt ₃	-1716027.19	-1716154.82	-1716070.83	-1716198.46	-1716058.97	-1716186.6
PBu ₃	-2136685.49	-2136853.53	-2136724.4	-2136892.44	-2136722.97	-2136891.01
PCy ₃	-2745628.22	-2745798.26	-2745680.86	-2745850.9	-2745678.74	-2745848.78
Pyridine	-651015.98	-651100.01	-651045.56	-651129.59	-651039.49	-651123.52
SIMes	-2426078.67	-2426274.77	-2426170.76	-2426366.86	-2426158.85	-2426354.95

Table S3.14. Computed thermodynamics of hydrogen bond formation between Lewis bases and para-fluorophenol (**p-F-PhOH-X**) in vacuum, DCM, and CCl₄ (DSD-BLYP(D3BJ)/def2-TZVPP//r²scan-3c with corresponding SMD solvation correction).

X	ΔH (vacuum) [kJ/mol] p-F-PhOH + X → p-F-PhOH-X	ΔG (vacuum) [kJ/mol] p-F-PhOH + X → p-F-PhOH-X	ΔH (DCM) [kJ/mol] p-F-PhOH + X → p-F-PhOH-X	ΔG (DCM) [kJ/mol] p-F-PhOH + X → p-F-PhOH-X	ΔH (CCl ₄) [kJ/mol] p-F-PhOH + X → p-F-PhOH-X	ΔG (CCl ₄) [kJ/mol] p-F-PhOH + X → p-F-PhOH-X
Br	-110.9	-79.1	-52.4	-20.6	-74.2	-42.3
Cl	-134.4	-102.6	-48.2	-16.4	-80.3	-48.5
CN	-126.0	-86.7	-30.5	8.7	-66.6	-27.4
F	-266.4	-234.1	-91.0	-58.6	-157.0	-124.7
N3	-117.5	-78.3	-36.1	3.1	-68.4	-29.2
NCS	-97.9	-75.0	-31.7	-8.9	-58.7	-35.9
DABCO	-47.0	2.6	-39.3	10.3	-44.2	5.3
DIBA	-63.2	-15.3	-46.7	1.3	-52.2	-4.3
DMAP	-47.8	-0.8	-36.4	10.5	-44.0	2.9
DMSO	-52.4	-2.9	-33.8	15.7	-42.0	7.5
HMPA	-61.2	-10.7	-42.1	8.5	-50.2	0.3
iPrNHC	-69.1	-16.3	-39.2	13.6	-47.6	5.2
dippNHC	-83.4	-27.6	-57.9	-2.1	-63.8	-8.0
OPEt ₃	-41.7	12.1	-26.4	27.4	-33.3	20.5
PBu ₃	-43.8	10.1	-30.6	23.3	-35.2	18.6
PCy ₃	-38.6	15.4	-24.5	29.6	-30.2	23.8
Pyridine	-40.2	4.2	-31.1	13.3	-37.6	6.7
SIMes	-80.6	-20.8	-50.5	9.3	-56.7	3.1

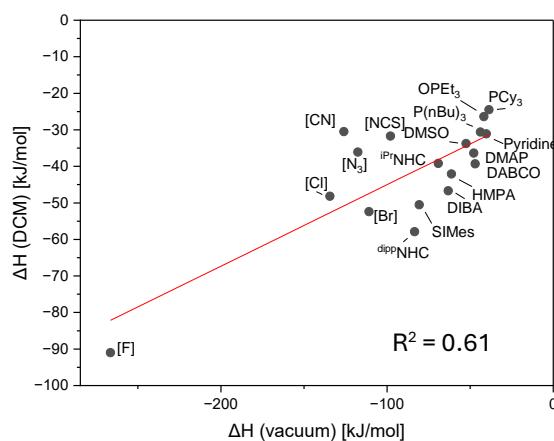


Figure S3.1. Correlation between vacuum and solution (DCM) enthalpies for hydrogen bond formation between Lewis bases and para-fluorophenol.

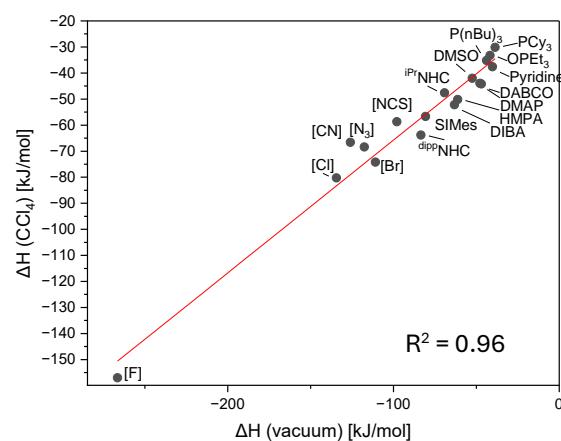


Figure S3.2. Correlation between vacuum and solution (CCl₄) enthalpies for hydrogen bond formation between Lewis bases and para-fluorophenol.

Relative Stability of Pyridine Bis-Adduct Conformations

For the conformer search, the relative Gibbs free energies of three conformations of the pyridine bis-adduct were calculated on the r²scan-3c level of theory.

Table S3.15. Conformer search of pyridine bis-adducts of **1** on the r2scan-3c level of theory.

	Total Correction [kJ/mol] r ² scan-3c	Entropy Term (T [*] S) [kJ/mol] r ² scan-3c	Single Point Energy [kJ/mol] r ² scan-3c	Enthalpy [kJ/mol] r ² scan-3c	Gibbs Free Energy [kJ/mol] r ² scan-3c
1-(pyridine)₂ trans	3196.91	429.12	-9223407.97	-9220208.58	-9220637.70
1-(pyridine)₂ cis	3196.05	422.58	-9223409.05	-9220210.53	-9220633.11
1-(pyridine)₂ cis 2	3196.79	426.25	-9223348.73	-9220149.45	-9220575.70

Relative energy

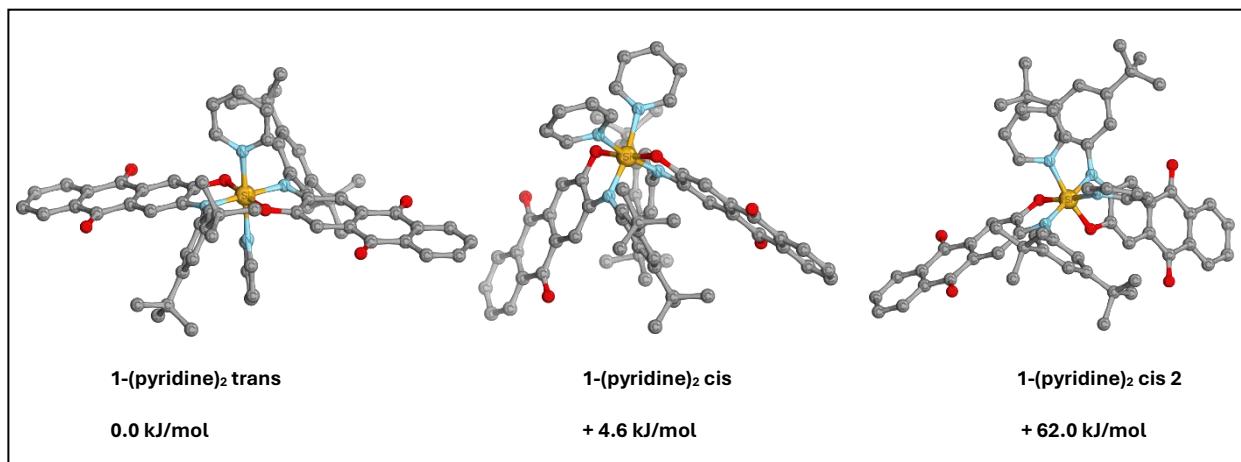


Figure S3.3. Optimised structures and relative Gibbs free energies of pyridine bis-adducts of **1**.

3.4 NBO Analysis of Negative Hyperconjugation in **1** and [**1**-Cl]

NBO analysis was done for **1** and the chloride adduct [**1**-Cl]. Only one set of orbital interactions of the symmetric complex is shown. Second order perturbation theory stabilisation energies of the LP(N/O) → LV(Si) negative hyperconjugation are given. Below, the visualisation of participating NBOs (Chemcraft) and occupancies (Occ.) are given.

1

Orbital Interaction	Stabilisation Energy [kcal/mol]
75. LP (2) O 23 → 234. LV (1)Si119	0.31
75. LP (2) O 23 → 235. LV (2)Si119	8.43
75. LP (2) O 23 → 236. LV (3)Si119	14.81
75. LP (2) O 23 → 237. LV (4)Si119	2.18
77. LP (1) N 24 → 234. LV (1)Si119	0.76
77. LP (1) N 24 → 235. LV (2)Si119	8.69
77. LP (1) N 24 → 236. LV (3)Si119	16.73
77. LP (1) N 24 → 237. LV (4)Si119	2.38
Total Stabilisation Energy [kcal/mol]	
2x LP(O/N) → LV (Si) negative hyperconjugation	108.58

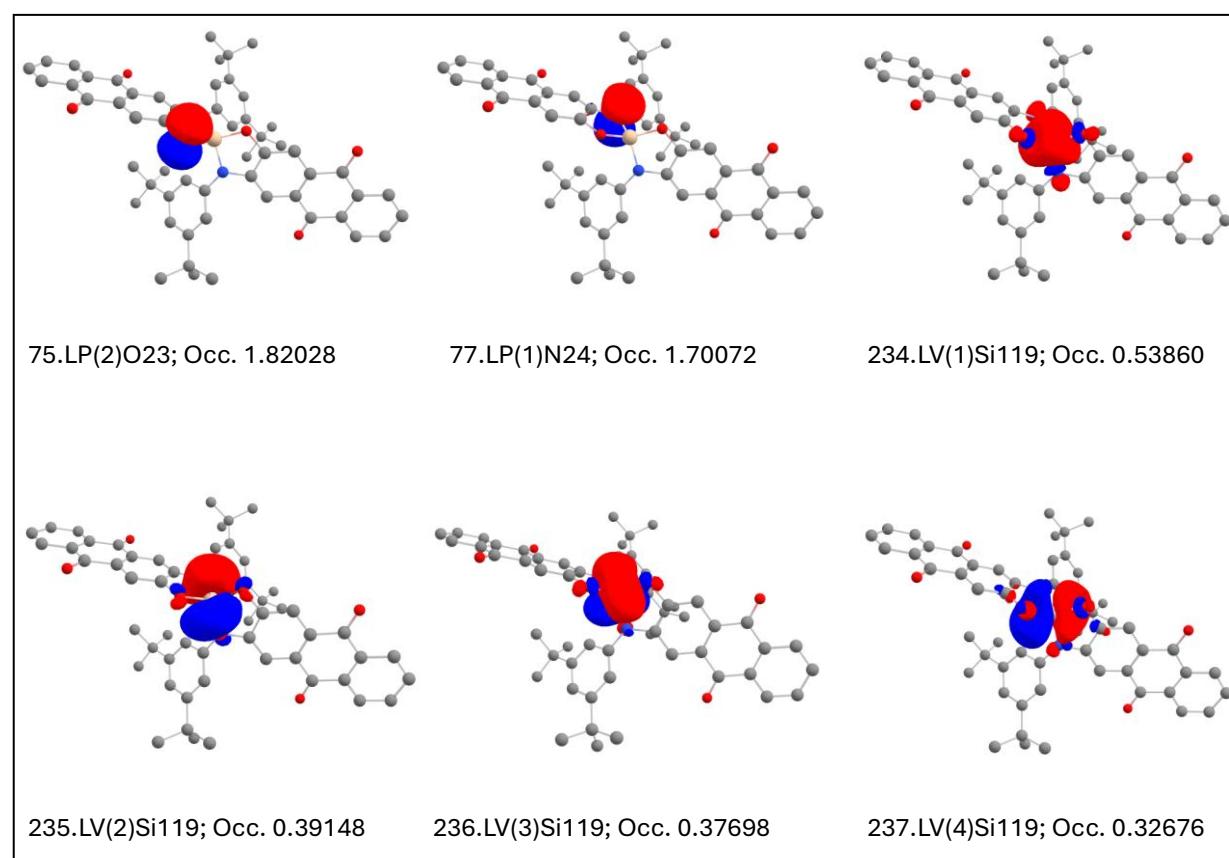


Figure S3.4. Natural bond orbitals involved in the LP(O/N) → LV(Si) negative hyperconjugation in **1**.

[1-Cl]

Orbital Interaction	Stabilisation Energy [kcal/mol]
80. LP (2) O 23 → 243. LV (1)Si119	5.91
80. LP (2) O 23 → 244. LV (2)Si119	3.05
80. LP (2) O 23 → 245. LV (3)Si119	0.61
80. LP (2) O 23 → 392. BD*(1)Si119-Cl120	3.21
82. LP (1) N 24 → 243. LV (1)Si119	21.26
82. LP (1) N 24 → 244. LV (2)Si119	1.57
82. LP (1) N 24 → 245. LV (3)Si119	0.37
82. LP (1) N 24 → 392. BD*(1)Si119-Cl120	6.71
Total Stabilisation Energy [kcal/mol]	
2x LP(O/N) → LV/BD*(Si-Cl) negative hyperconjugation	85.38

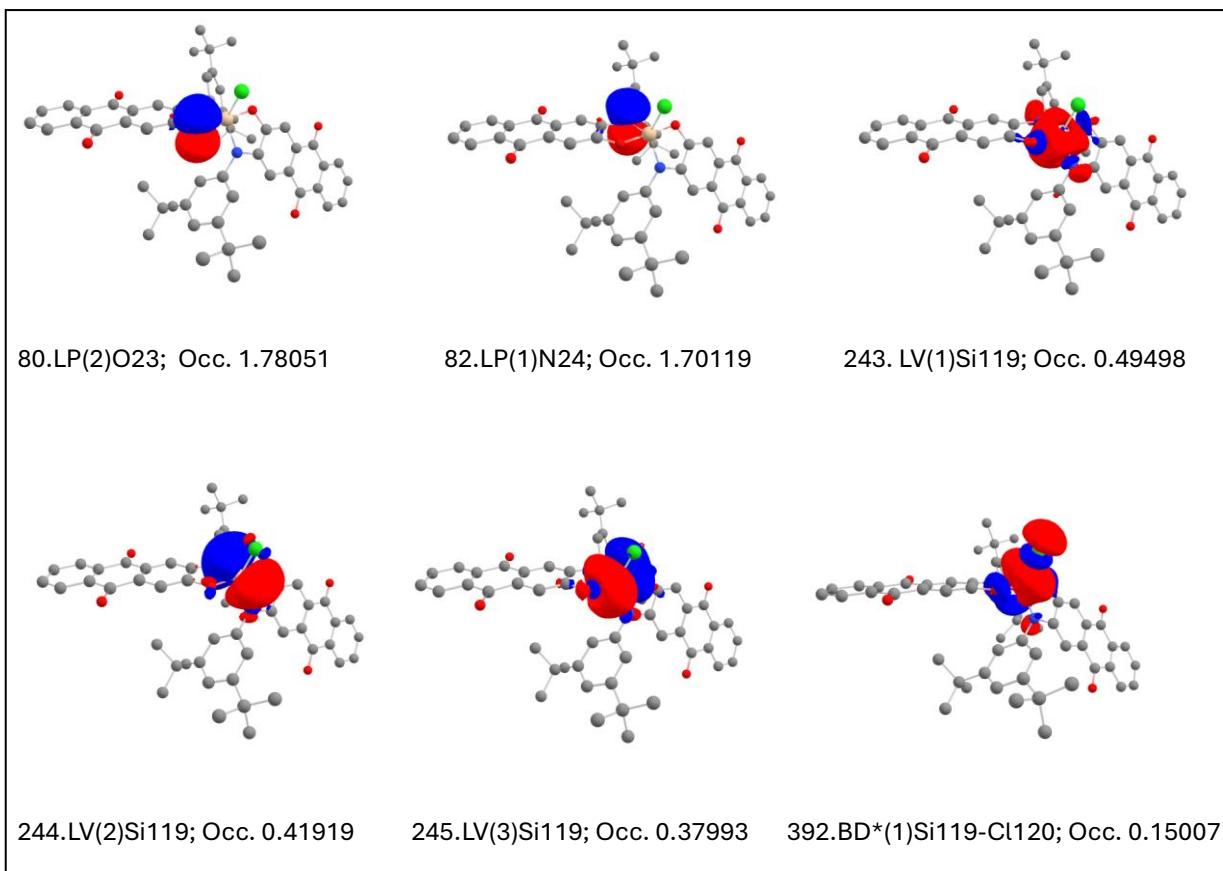


Figure S3.5. Natural bond orbitals involved in the LP(O/N) → LV(Si) negative hyperconjugation in **[1-Cl]**.

3.5 TD-DFT/TDA Calculated Excitations

Time-dependant calculations were carried out with structures optimised with the r2scan-3c composite method. The first 50 electronic transitions were calculated. For comparisons with experimental values, a systematic shift of -0.56 eV was applied to all energies given in the following section. The largest molecular orbital contributions for the low-energy excitations are visualised. For the simulated spectrum, a gaussian line broadening was applied as implemented in the Chemcraft software.

1

Table S3.16. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1**.

Wavelength [nm]	Oscillator strength (f)
441.5	0
441.1	0
404.3	0
403.8	0
374.2	0
373.8	0
367.6	0
367.3	0
354.4	4.8396E-5
354.1	7.4926E-5
347.9	0
347.7	0
344.2	0.295656916
343.2	0.087280005
330.3	0.0011334044
330.1	0.001123268
327.5	0
327.3	0
321.6	0
321.4	0
303.1	0
303	0
290.6	0.139986246
290.3	1.6962E-5
282.9	0
282.7	0
280.4	0
280.3	0
276.7	0.618652193
276.5	0.021707714
268.2	0
267.9	0
265.9	2.253580904
264.8	0
264.7	0
262.8	0.240132824
252.7	0
252.5	0
252	0.897900902
251.2	0.298566906
248.7	0
248	0
247.5	0
247.4	0
241.5	0
241.3	0.007951568
240.8	0.043093286
240.7	0
238.3	0.006908931
238.2	0.005599111

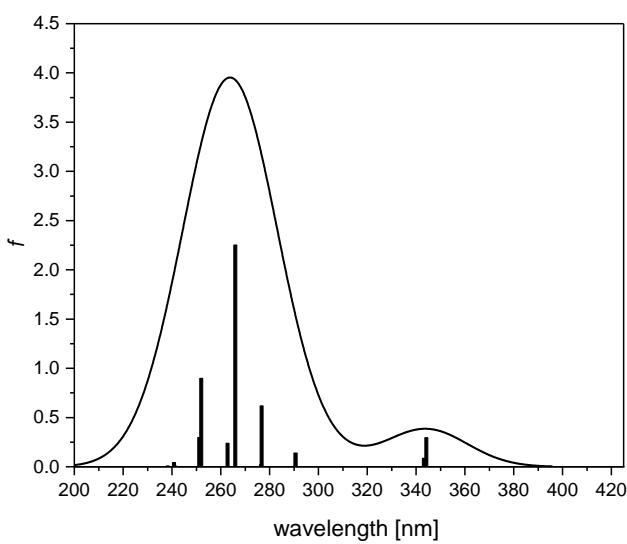


Figure S3.6. Simulated UV-vis spectrum of **1** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 40 nm FWHM.

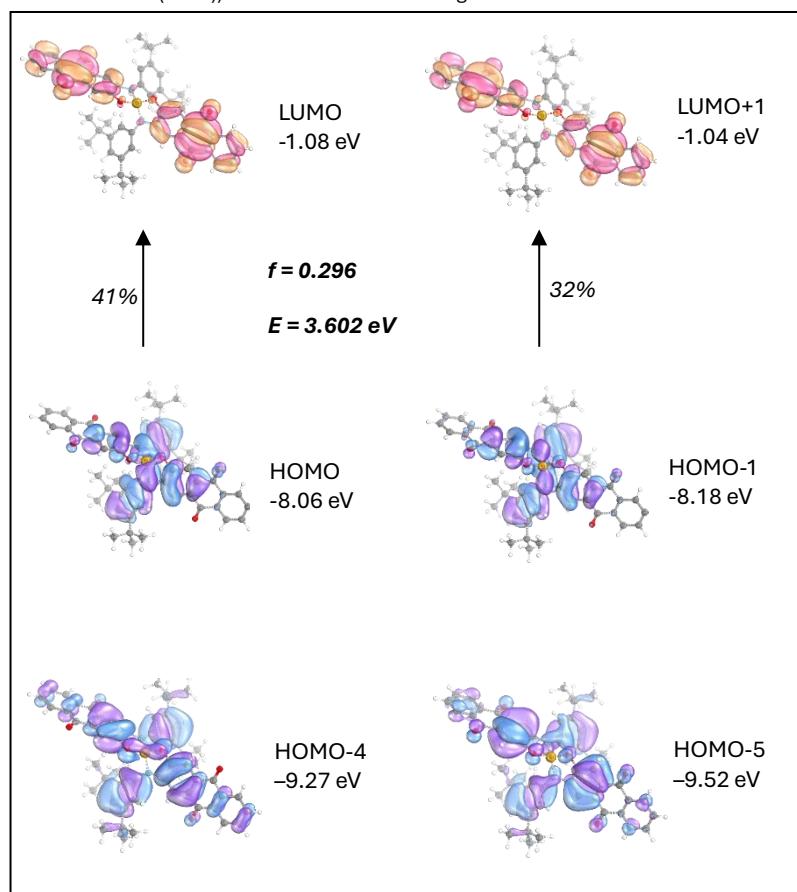


Figure S3.7. Molecular orbitals involved in the low energy CT transition of **1**. HOMO \rightarrow LUMO 41%, HOMO-1 \rightarrow LUMO+1 32%, HOMO-4 \rightarrow LUMO 6%, HOMO-5 \rightarrow LUMO+1 6%.

[1-F]

Table S3.17. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound [1-F].

Wavelength [nm]	Oscillator strength (f)
533.2	0
532.6	0
403.1	0
402.7	0
399.2	0.025802132
398.1	0.293075708
377.8	0
377.7	0
371.7	0
371.5	0
361.4	0
361.3	0
354.6	4.361E-5
354.4	4.8983E-5
350	0
349.8	0
330.4	0.001736554
330.2	0.00164204
323.6	0
323.4	0
311.3	0.094632135
311.2	0.037957755
302	0
302	0
289.1	0
289	0
278.7	0
278.6	0
278.3	3.140121082
272.3	0.107028194
271.8	0.018919125
268.5	0.185208177
263	0
261.5	0
261	0
260.9	0
259.5	0
259.4	0
258.8	0.220276148
258.6	0.128340056
256	0
254.7	0
253.5	0
253.4	0
251.9	0.134953974
251.1	0.571394804
244.7	0
244.2	0
234	0.001998585
233.5	0.004467563

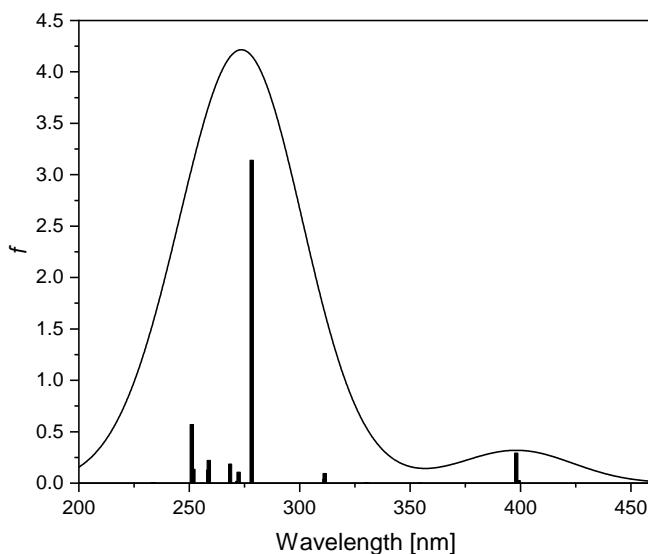


Figure S3.8. Simulated UV-vis spectrum of [1-F] in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 60 nm FWHM.

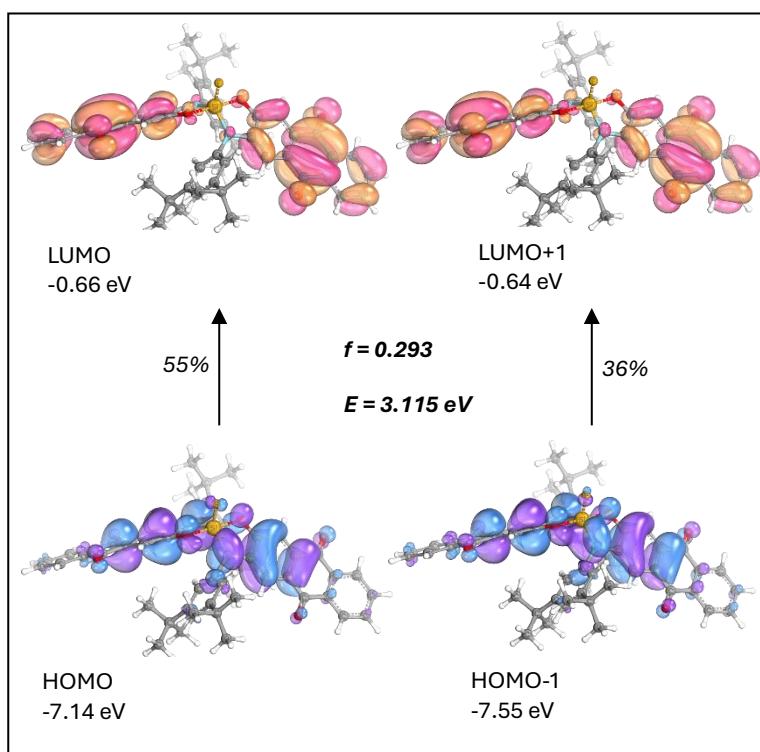


Figure S3.9. Molecular orbitals involved in the low energy CT transition of [1-F]. HOMO \rightarrow LUMO 55%, HOMO-1 \rightarrow LUMO+1 36%.

[1-Cl]

Table S3.18. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound [1-Cl].

Wavelength [nm]	Oscillator strength (f)
529.3	0
528.9	0
403.3	0
402.7	0
397.4	0.025805318
396.2	0.288521559
376.5	0
376.5	0
371.5	0
371.3	0
361.6	0
361.6	0
354.7	3.9946E-5
354.3	4.7219E-5
349	0
348.9	0
330.4	0.001686103
330.2	0.001638619
323.8	0
323.7	0
310.5	0.066185802
310.5	0.058732972
302.5	0
302.5	0
289.2	0
289.1	0
278.8	0
278.7	0
278.4	3.228361991
272.5	0.110562274
272	0.012497789
268.7	0.17760997
263	0
261.9	0
261.4	0
261.3	0
259.5	0
259.3	0
258.7	0.063301244
258.6	0.263570194
256.4	0
255.3	0
253.4	0
253.3	0
252.9	0.142052311
252.3	0.57727372
245.1	0
244.6	0
234.3	0.001504853
233.9	0.003991975

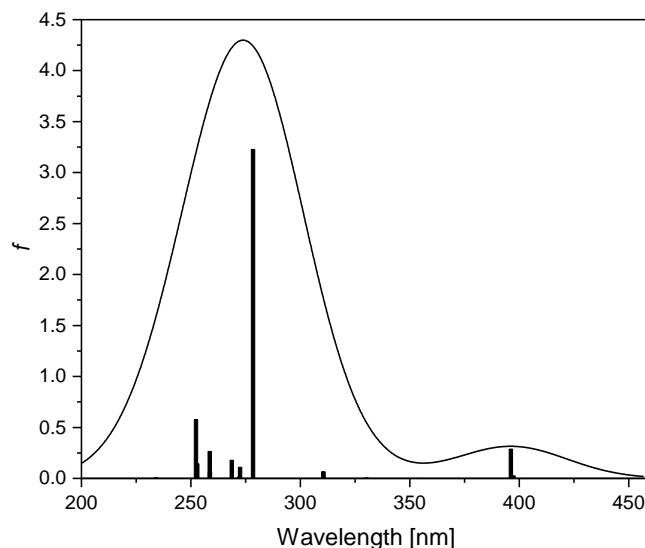


Figure S3.10. Simulated UV-vis spectrum of [1-Cl] in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 60 nm FWHM.

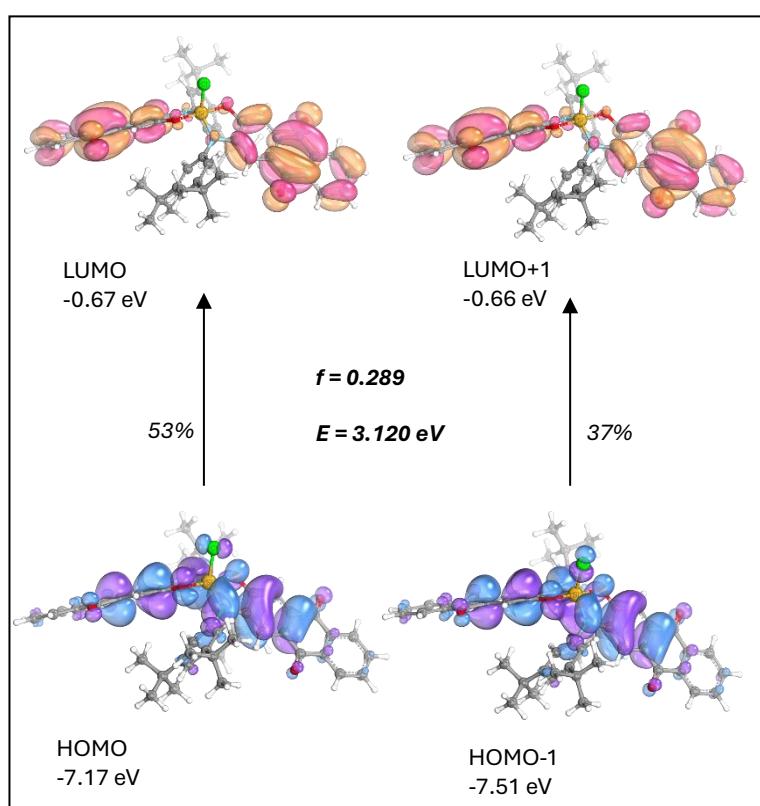


Figure S3.11. Molecular orbitals involved in the low energy CT transition of [1-Cl]. HOMO->LUMO 53%, HOMO-1->LUMO+1 37%.

[1-N₃]

Table S3.19. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound [1-N₃].

Wavelength [nm]	Oscillator strength (f)
532.6	0
529.9	0
403.1	0
402.8	0
398.6	0.073230648
397.1	0.246884016
376.6	0
375.9	0
371.5	0
371	0
361.6	0
361.5	0
354.6	3.002E-5
354.4	5.4681E-5
349.9	0
349.5	0
330.3	0.001633076
330.2	0.001556745
323.8	0
323.6	0
310.7	0.06372911
310.2	0.058074431
302.1	0
302	0
288.7	0
288.5	0
278.8	0
278.6	0
278.3	3.119970084
272.5	0.136738991
271.9	0.01725935
269	0.22031698
266.7	0
263.2	0
261.5	0
261.2	0
261.1	0
259.7	0
259.5	0
259	0.06136341
259	0.234703321
257.1	0
256.4	1E-9
255.6	0
253.4	0
253.4	0
253	0.160128394
252.3	0.603425892
245.3	1E-9
244.7	0

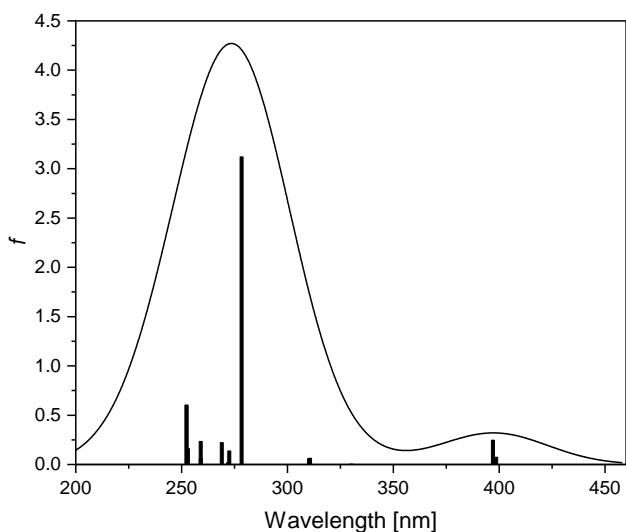


Figure S3.12. Simulated UV-vis spectrum of [1-N₃] in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 60 nm FWHM.

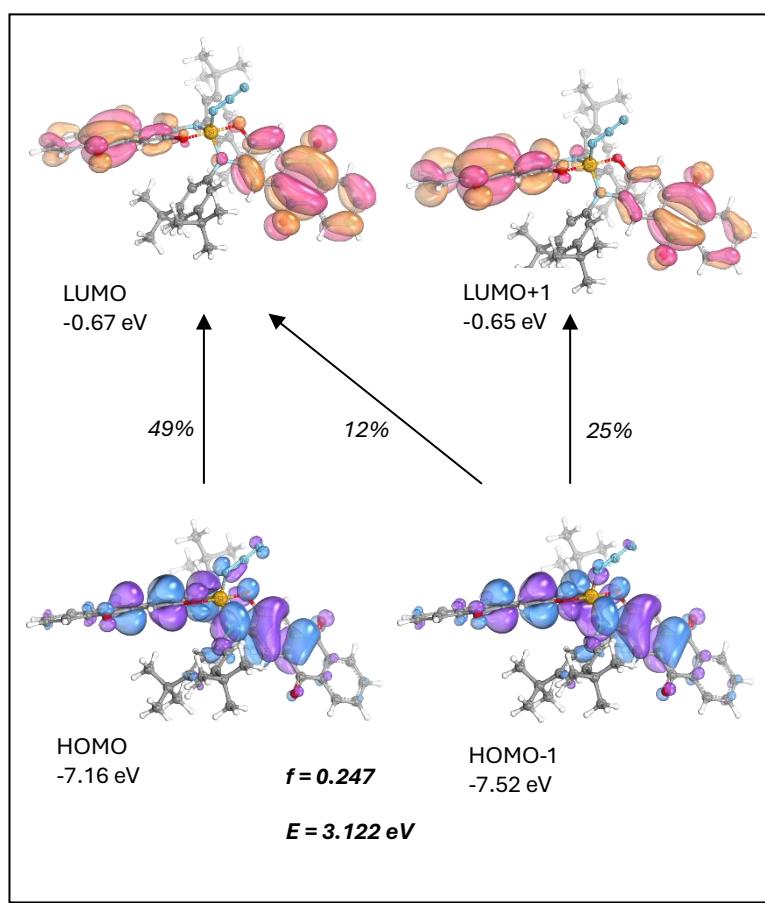


Figure S3.13. Molecular orbitals involved in the low energy CT transition of [1-N₃]. HOMO->LUMO 49%, HOMO-1->LUMO+1 25%, HOMO-1 -> LUMO 12%.

[1-NCS]

Table S3.20. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound [1-NCS].

Wavelength [nm]	Oscillator strength (f)
520.1	0
519.9	0
403.5	0
403.1	0
391.1	0.025159467
390	0.285337077
374.7	0
374.6	0
370.3	0
370.3	0
362	0
362	0
354.7	3.9076E-5
354.5	4.6615E-5
345.1	0
344.8	0
330.5	0.001442794
330.4	0.001371714
324.3	0
324.1	0
308.4	0.039288127
308.2	0.101147587
301.9	0
301.8	0
288.3	0
288.2	0
278.9	0
278.9	0
276.2	3.15977973
272.8	0.097839368
272	0.042361339
266.6	0.250136649
261.7	0
261.4	0
260.9	0
260.7	0
258.7	0
258.6	0
256.3	0.144279931
255.9	0.475212573
254.2	0
253.6	0
253.1	0
253	0
252.7	0.109810115
252.2	0.325587338
244.8	0
244.2	0
233.5	0.000995246
233	0.00143284

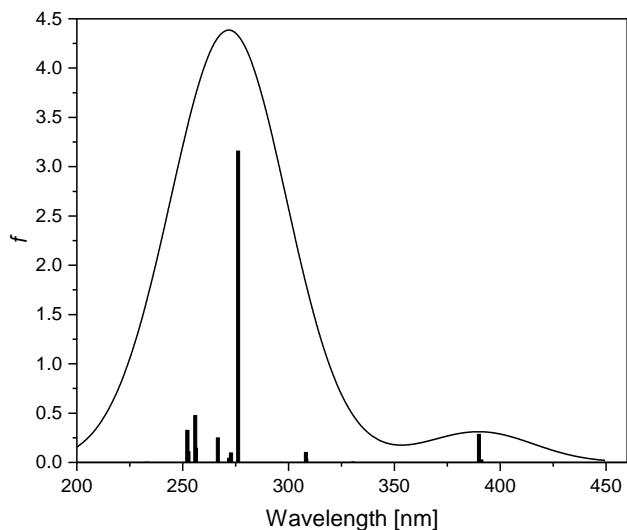


Figure S3.14. Simulated UV-vis spectrum of [1-NCS] in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 60 nm FWHM.

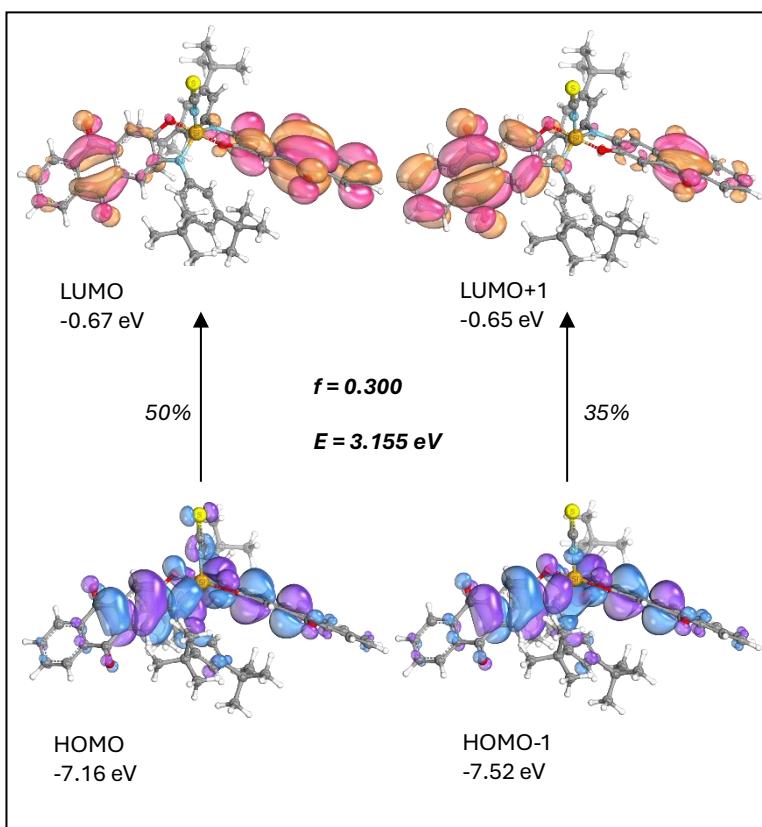


Figure S3.15. Molecular orbitals involved in the low energy CT transition of [1-NCS]. HOMO->LUMO 50%, HOMO-1->LUMO+1 35%.

[1-CN]

Table S3.21. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound [1-CN].

Wavelength [nm]	Oscillator strength (f)
520.1	0
519.9	0
403.5	0
403.1	0
391.1	0.025159467
390	0.285337077
374.7	0
374.6	0
370.3	0
370.3	0
362	0
362	0
354.7	3.9076E-5
354.5	4.6615E-5
345.1	0
344.8	0
330.5	0.001442794
330.4	0.001371714
324.3	0
324.1	0
308.4	0.039288127
308.2	0.101147587
301.9	0
301.8	0
288.3	0
288.2	0
278.9	0
278.9	0
276.2	3.15977973
272.8	0.097839368
272	0.042361339
266.6	0.250136649
261.7	0
261.4	0
260.9	0
260.7	0
258.7	0
258.6	0
256.3	0.144279931
255.9	0.475212573
254.2	0
253.6	0
253.1	0
253	0
252.7	0.109810115
252.2	0.325587338
244.8	0
244.2	0
233.5	0.000995246
233	0.00143284

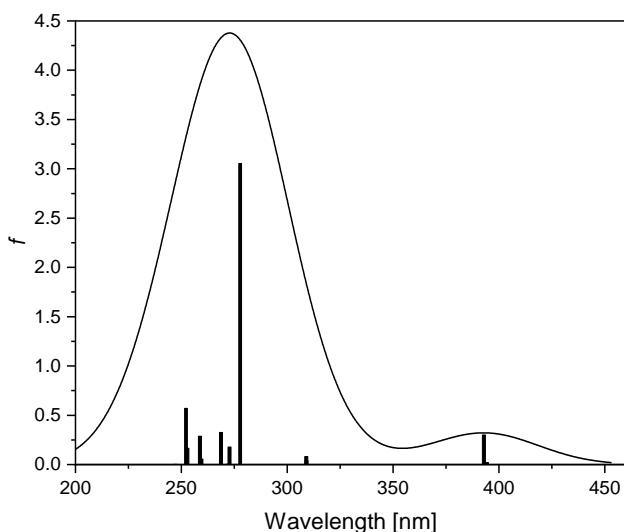


Figure S3.16. Simulated UV-vis spectrum of [1-CN] in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 60 nm FWHM.

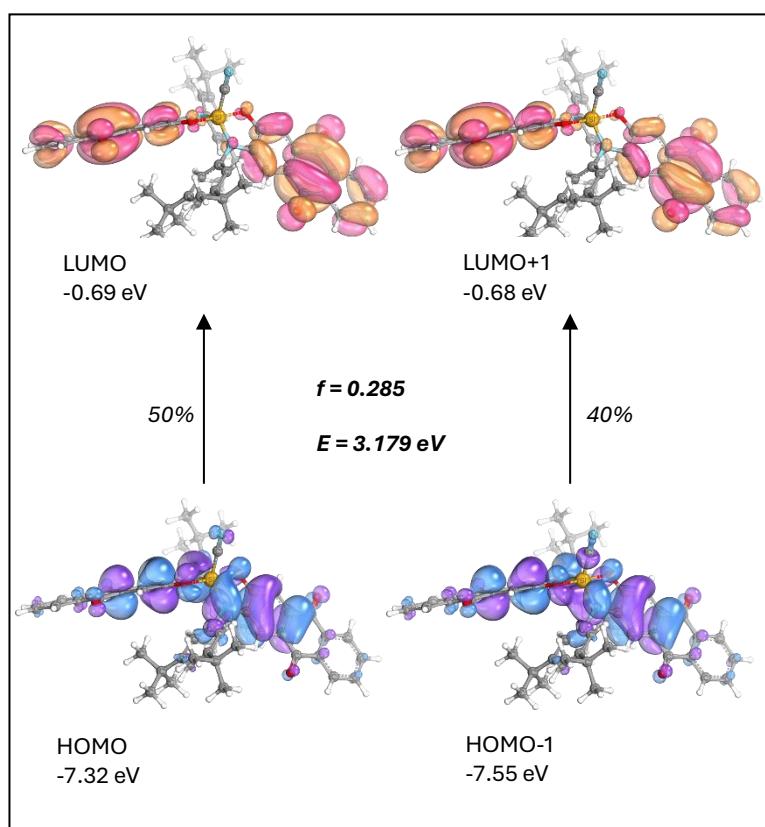


Figure S3.17. Molecular orbitals involved in the low energy CT transition of [1-CN]. HOMO->LUMO 50%, HOMO-1->LUMO+1 40%.

[1-Br]

Table S3.22. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **[1-Br]**.

Wavelength [nm]	Oscillator strength (f)
529.2	0
528.4	0
402.8	0
402.1	0
398	0.049951323
396.6	0.266287253
376.2	0
375.2	0
371	0
370.8	0
361.5	0
361.4	0
354.4	6.4963E-5
353.9	4.9516E-5
350.4	0
350.1	0
330.3	0.001250147
329.9	0.001670834
323.7	0
323.5	0
310.8	0.048901049
309.9	0.057584092
306.2	0
302.2	0
289.6	0
288.5	0
279.7	3.135070573
278.6	0
278.6	0
272.6	0.160496998
272	0.027552118
269.4	0.198909285
263.9	0
262.7	0
261.8	0
261.8	0.098893876
261.2	0
260	0
259.2	0
259	0.147851784
258.1	0
256.4	0
254.2	0.272268629
253.6	0
253.5	0
253.4	0.543782039
246	0
245	0
235	0.001353489
234.1	0.00285886

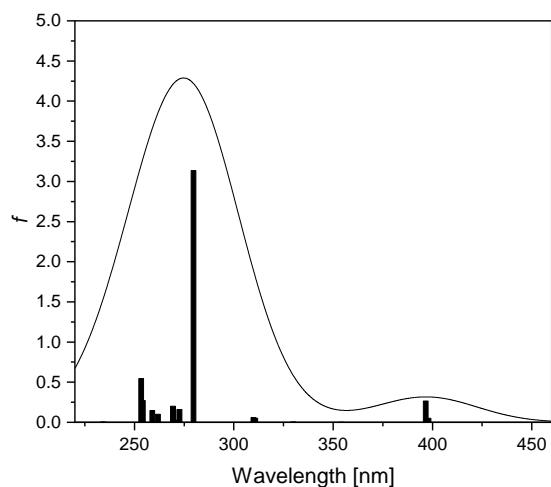


Figure S3.18. Simulated UV-vis spectrum of **[1-Br]** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 60 nm FWHM.

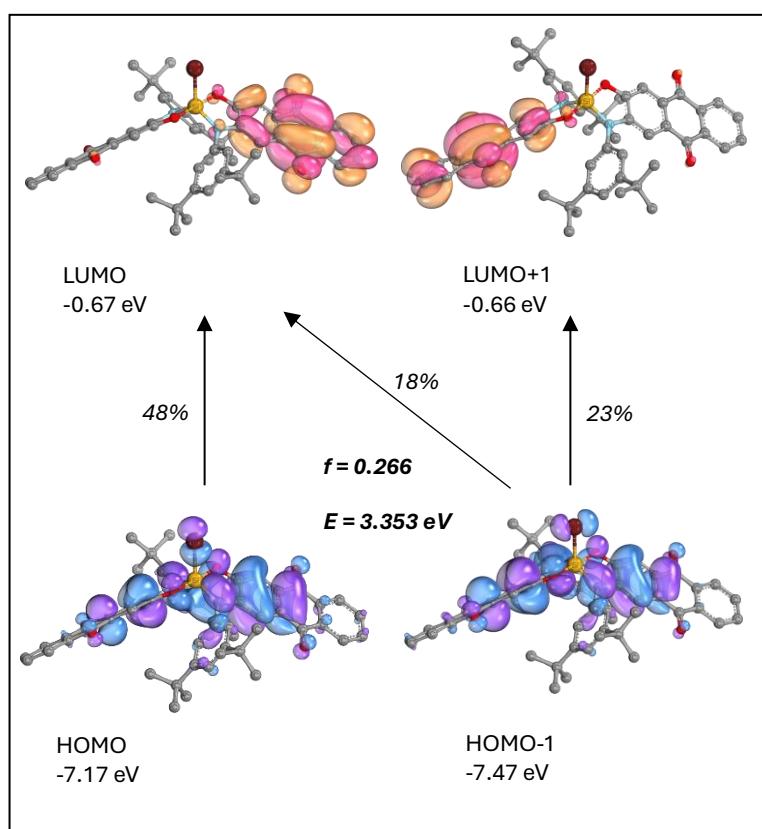


Figure S3.19. Molecular orbitals involved in the low energy CT transition of **[1-Br]**. HOMO->LUMO 48%, HOMO-1->LUMO+1 23%. HOMO-1->LUMO 18%.

1-pyridine

Table S3.23. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-pyridine**.

Wavelength [nm]	Oscillator strength (f)
483.3	0
483.1	0
402.7	0
402.3	0
372.5	0
372.2	0
371.1	0.016970852
369.8	0.274947103
364.2	0
364	0
360.8	0
360.8	0
353.7	0.000153788
353.5	0.00018907
334.3	0
333.9	0
329.5	0.000768816
329.3	0.000707551
326.2	0
326.2	0
304	0
303.9	0
301.6	0.123837256
301.5	0.008874665
293.7	0
288.8	0
288.7	0
287.5	0
279.4	0
279.4	0
274.8	1.681723192
274	0.045133681
270.5	1.694062133
264.8	0.135528855
263.2	0
263.1	0
262.7	0
262.5	0
260.1	0.033434062
259.9	0
255.4	0.14890743
255.3	0.809717086
253.8	0
253.7	0
252.7	0
252.7	0
249.3	0
246.2	9.35E-7
246	2.252E-6
245.9	0.008528666

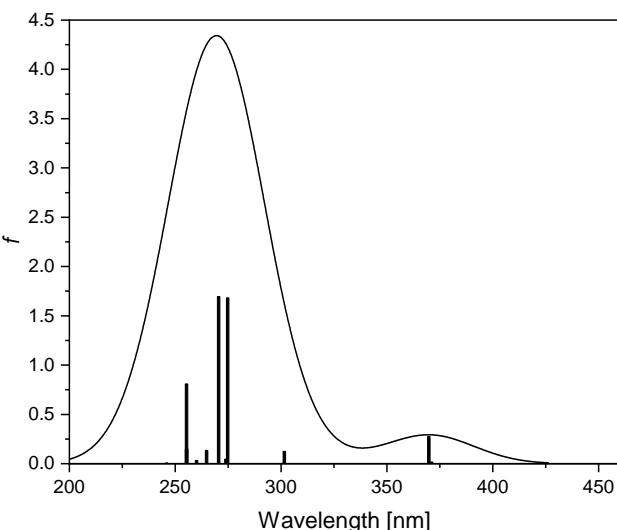


Figure S3.20. Simulated UV-vis spectrum of **1-pyridine** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM.

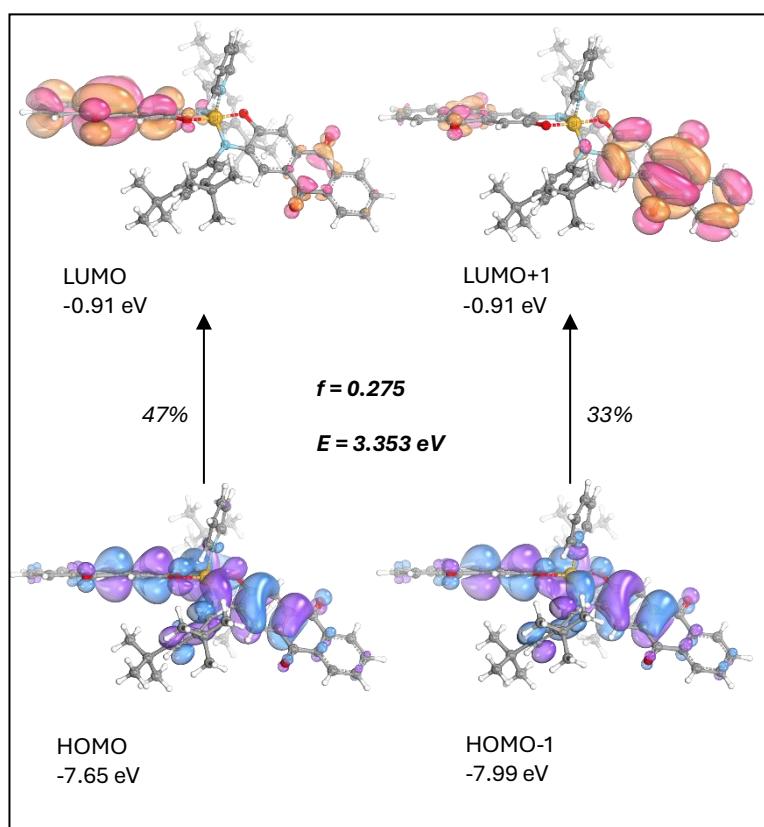


Figure S3.21. Molecular orbitals involved in the low energy CT transition of **1-pyridine**. HOMO->LUMO 47%, HOMO-1->LUMO+1 33%.

1-DMAP

Table S3.24. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-DMAP**.

Wavelength [nm]	Oscillator strength (f)
491.9	0
491.7	0
403.2	0
402.3	0
376	0.021298575
374.9	0.270769199
372.9	0
372.3	0
363.7	0
363.7	0
363	0
363	0
354.1	0.000148322
353.7	0.000169759
336.9	0
336.8	0
336	0
329.7	0.00088639
329.4	0.000830724
326	0
325.9	0
303.3	0
303.2	0
303.2	0.104842753
303.2	0.03152555
288.9	0
288.8	0
283.4	0
279.2	0
279.2	0
274.7	1.908440697
273.7	0.069481417
271	1.409159086
265.2	0.243351212
262.6	0
262.4	0
262.2	1E-9
261.9	1E-9
257.5	0
256	0.233883416
255.8	0.685577652
255.1	1E-9
254.7	0
252.8	1E-9
252.7	1E-9
247.2	4E-9
246.6	3E-9
246.1	0.46055255
245.5	0.186135296
245	0.289665931

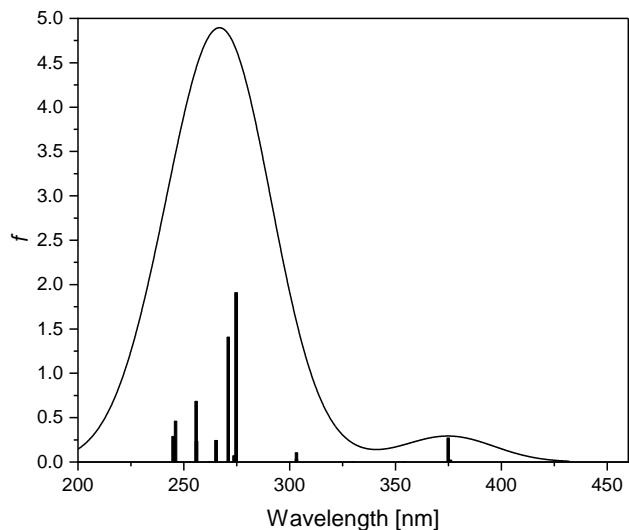


Figure S3.22. Simulated UV-vis spectrum of **1-DMAP** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM.

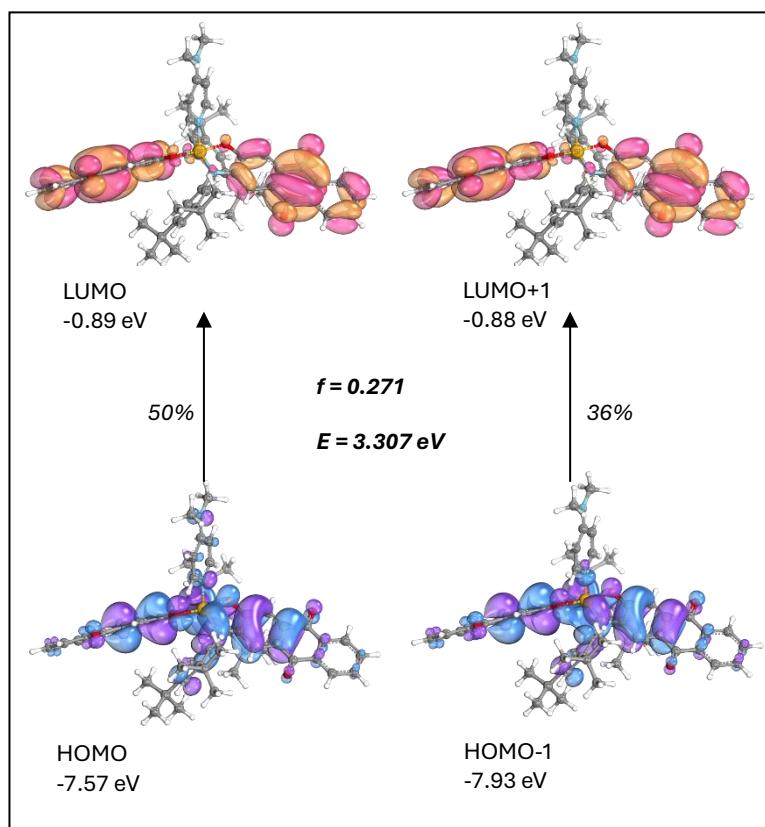


Figure S3.23. Molecular orbitals involved in the low energy CT transition of **1-DMAP**. HOMO \rightarrow LUMO 50%, HOMO-1 \rightarrow LUMO+1 36%.

1-OPEt₃

Table S3.25. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-OPEt₃**.

Wavelength [nm]	Oscillator strength (f)
497.9	0
494.4	0
403.2	0
402.6	0
378.9	0.102247235
377	0.206197776
372.6	0
372.5	0
364.6	0
364.1	0
363.3	0
362.8	0
354.3	0.000200295
353.8	8.8693E-5
339.5	0
338.3	0
329.7	0.000462897
329.5	0.001028722
326.3	0
325.7	0
304.5	0.065229517
303.4	0.064934387
303.1	0
302.2	0
288.8	0
288.5	0
279.1	0
279.1	0
274.8	1.986582888
273.7	0.156103364
271.3	1.143552997
264.6	0.135927957
262.7	0
262.4	0
261.2	0
260.9	0
256.1	0.450549779
256	0
255.4	0.443175597
255.1	0
252.9	0
252.7	0
249.7	0
248.2	0.154547549
245.6	0
244.8	0
244.2	0
243.4	0.076804383
234.4	0.002983195
233.2	0.001408145

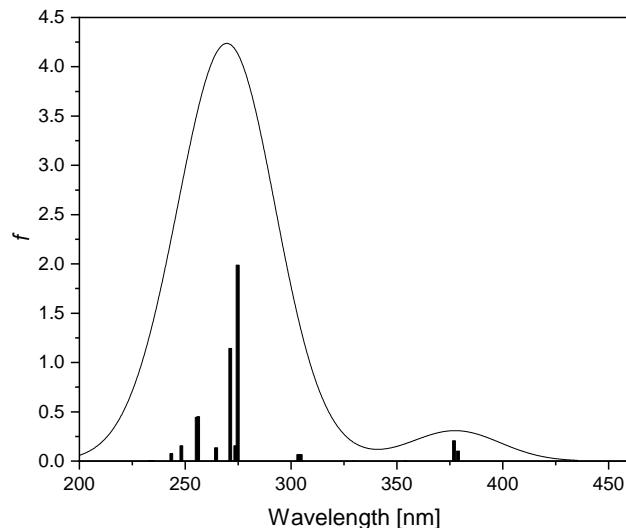


Figure S3.24. Simulated UV-vis spectrum of **1-OPEt₃** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. Two low-energy transitions have significant oscillator strength. The maximum was determined from the broadened spectrum to be $\lambda_{\text{max}} = 377.5$ nm.

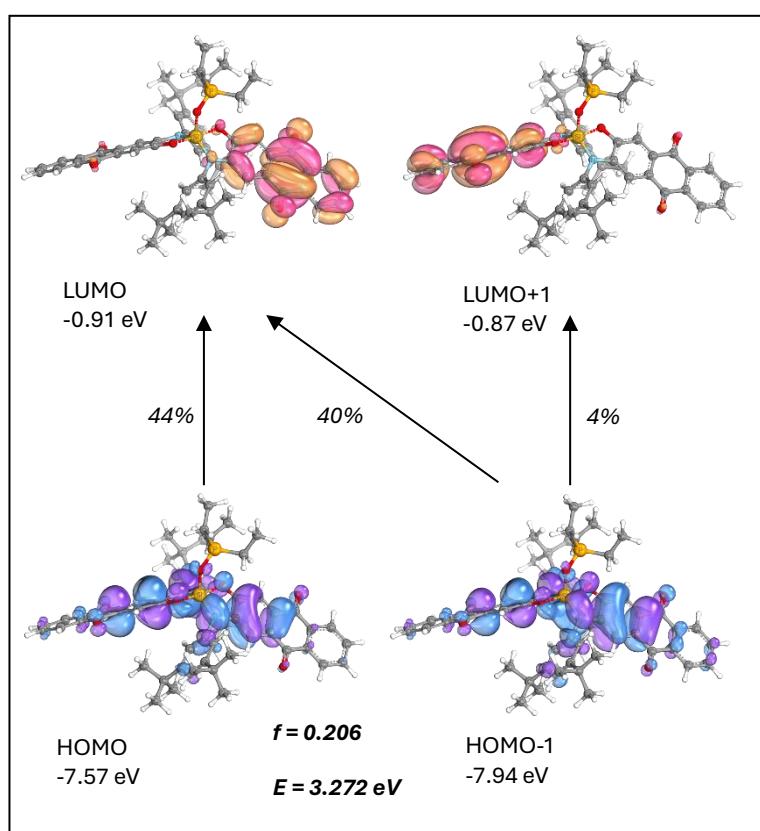


Figure S3.25. Molecular orbitals involved in the low energy CT transition of **1-OPEt₃**. HOMO \rightarrow LUMO 44%, HOMO-1 \rightarrow LUMO 40%, HOMO-1 \rightarrow LUMO+1 4%.

1-HMPA

Table S3.26. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-HMPA**.

Wavelength [nm]	Oscillator strength (f)
500	0
498.2	0
403.2	0
402.3	0
381.2	0.080370104
379.1	0.228250048
373.1	0
372.2	0
365.8	0
365.6	0
363.6	0
363.4	0
354.2	0.000113913
353.8	0.000146997
341.7	0
339.1	0
329.8	0.001163103
329.4	0.000625599
326.1	0
325.9	0
305.9	0.062634801
304.9	0.067922569
304.7	0
302.5	0
289.8	0
289.1	0
279.1	0
279.1	0
275.6	2.721545079
273.6	0.073732821
272.1	0.403223887
265.7	0.227438416
264	0
263.6	0
261.9	0
261.4	0
256.3	0.244825591
256.1	0
256	0.584995586
255.5	0
252.9	0
252.8	0
249.7	0
248	0.145315433
247.4	0
246.7	0
245.8	0.119380987
245	0
235.8	0.00182637
233.7	0.001715917

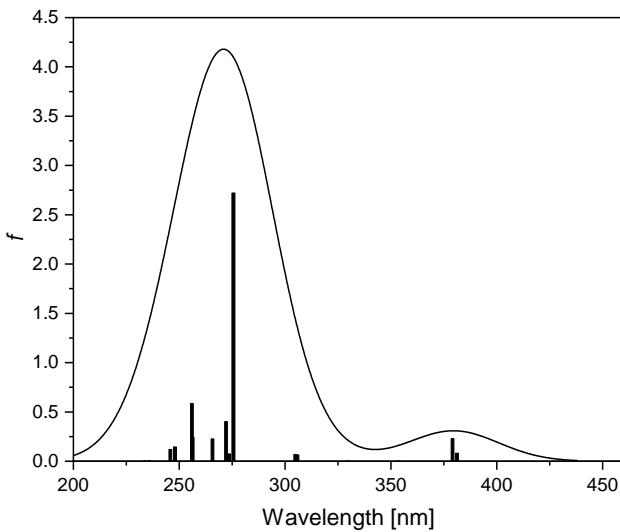


Figure S3.26. Simulated UV-vis spectrum of **1-HMPA** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. Two low-energy transitions have significant oscillator strength. The maximum was determined from the broadened spectrum to be $\lambda_{\max} = 379.5$ nm.

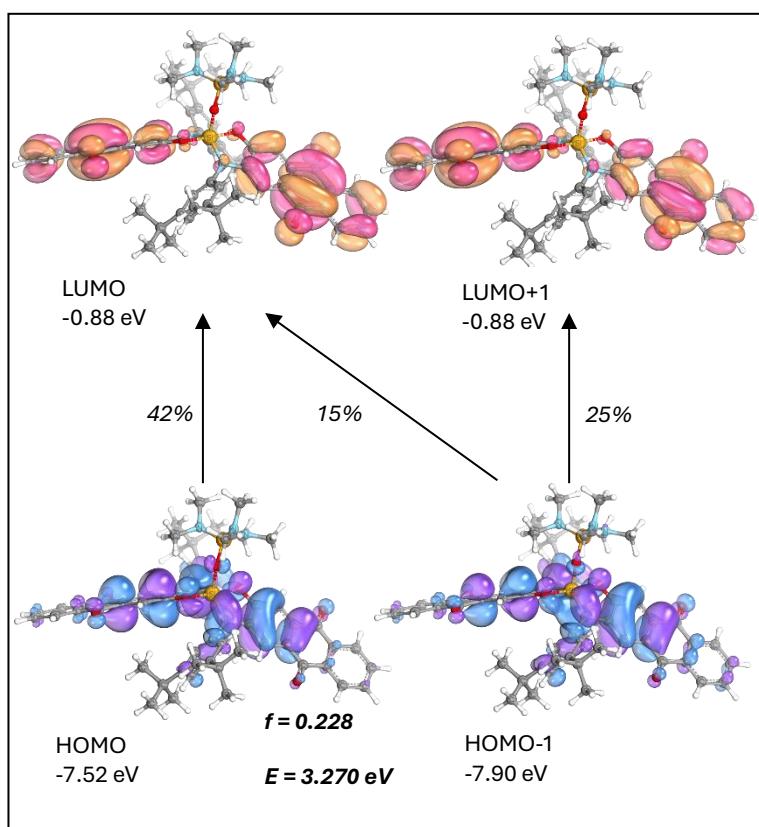


Figure S3.27. Molecular orbitals involved in the low energy CT transition of **1-HMPA**. HOMO \rightarrow LUMO 44%, HOMO-1 \rightarrow LUMO 15%, HOMO-1 \rightarrow LUMO+1 25%.

1-DMSO

Table S3.27. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-DMSO**.

Wavelength [nm]	Oscillator strength (f)
489.4	0
484.5	0
403.2	0
402.4	0
374.2	0.119425222
372.9	0
372.1	0
371	0.188668851
364.5	0
363.8	0
362.4	0
359.8	0
354.1	0.00014697
353.7	0.000283979
337.1	0
334.2	0
329.7	0.000977567
329.4	0.000407797
326.7	0
326	0
303.7	0
303.1	0.060518529
302.8	0
301.1	0.06435606
289.3	0
287.8	0
279.4	0
279.2	0
274.9	1.614711304
274	0.156276581
270.8	1.513276487
264	0.118359406
262.9	0
262.8	0
262	0
261.4	0
255.4	0.568223147
255.4	0.379537592
254.8	0
254	0
252.7	0
252.6	0
248.6	0
247.4	0.138359404
245.9	0
245.2	0
242.3	0
241.3	0.054251447
234.9	0.003805773
233.8	0.001047039

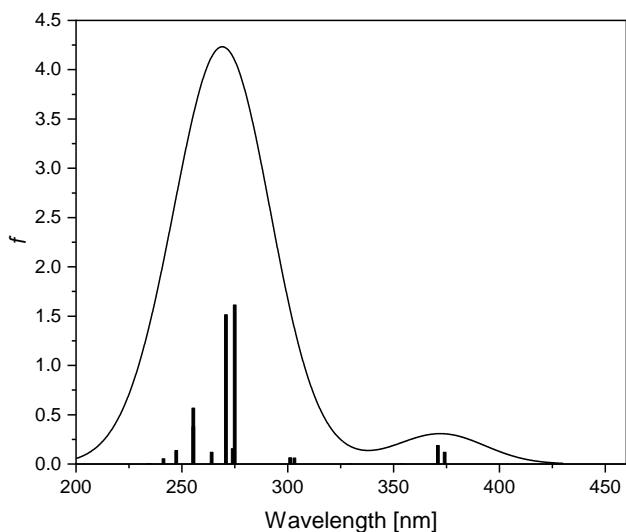


Figure S3.28. Simulated UV-vis spectrum of **1-DMSO** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. Two low-energy transitions have significant oscillator strength. The maximum was determined from the broadened spectrum to be $\lambda_{\max} = 372.1$ nm.

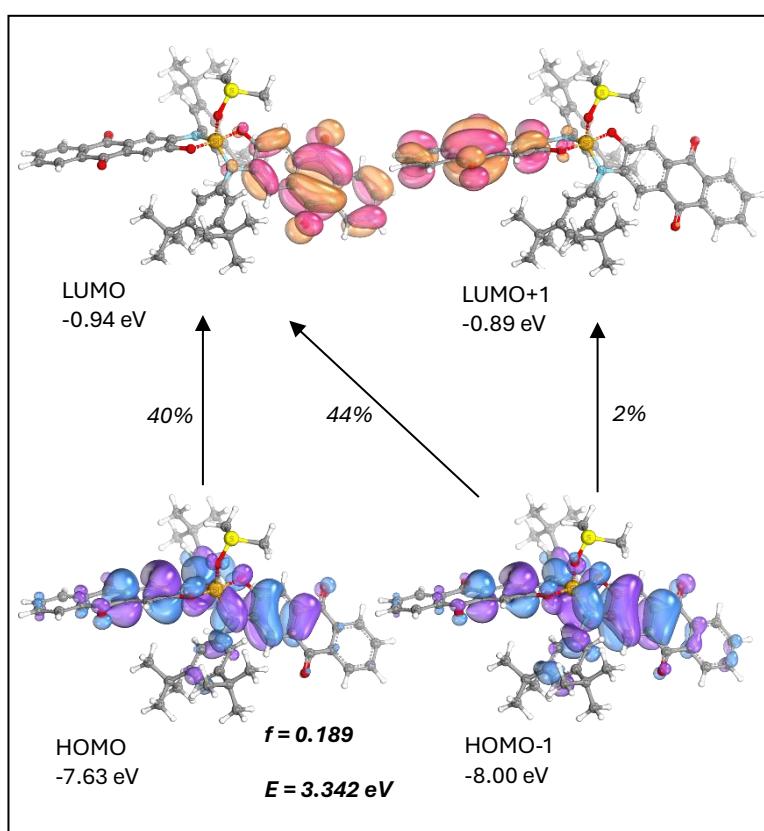


Figure S3.29. Molecular orbitals involved in the low energy CT transition of **1-DMSO**. HOMO \rightarrow LUMO 40%, HOMO-1 \rightarrow LUMO 44%, HOMO-1 \rightarrow LUMO+1 2%.

1-DIBA

Table S3.28. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-DIBA**.

Wavelength [nm]	Oscillator strength (f)
495.4	0
492.7	0
403	0
402.1	0
378.6	0.028609959
377.1	0.293895052
373.4	0
371.9	0
365.9	0
364.6	0
363.8	0
363.6	0
354.1	9.885E-6
353.6	0.000277779
342.1	0
340.9	0
329.9	0.001619901
329.3	0.000137246
326.3	0
326.1	0
310.3	0
307.4	0
306	0.064882755
304.7	0.046164268
303.1	0
291.4	0
290.4	0
279.2	0
279.2	0
278.6	0
277.4	2.890655567
274	0.098714576
272.6	0.064619981
268.4	0.291910811
267.7	0
266.9	0
265.5	0
265.1	0.000554392
263.1	0
262.4	0
261.5	0
258.4	0
256.7	0.27537162
255.7	0.553446569
255.5	0
255.1	0
253	0
252.8	0
251.1	0
249.5	0.156129122

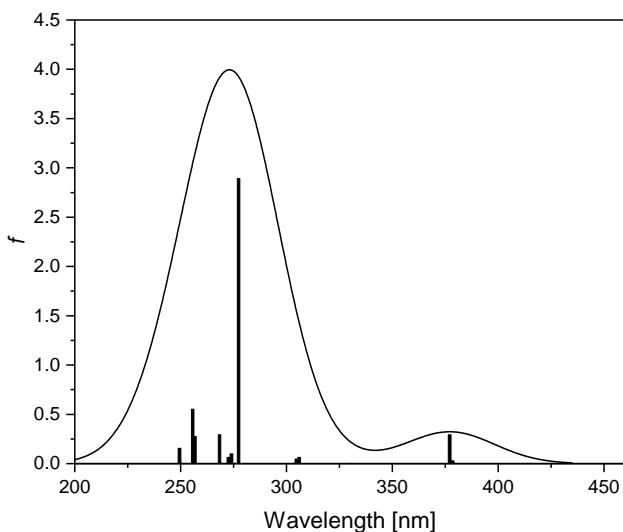


Figure S3.30. Simulated UV-vis spectrum of **1-DIBA** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM.

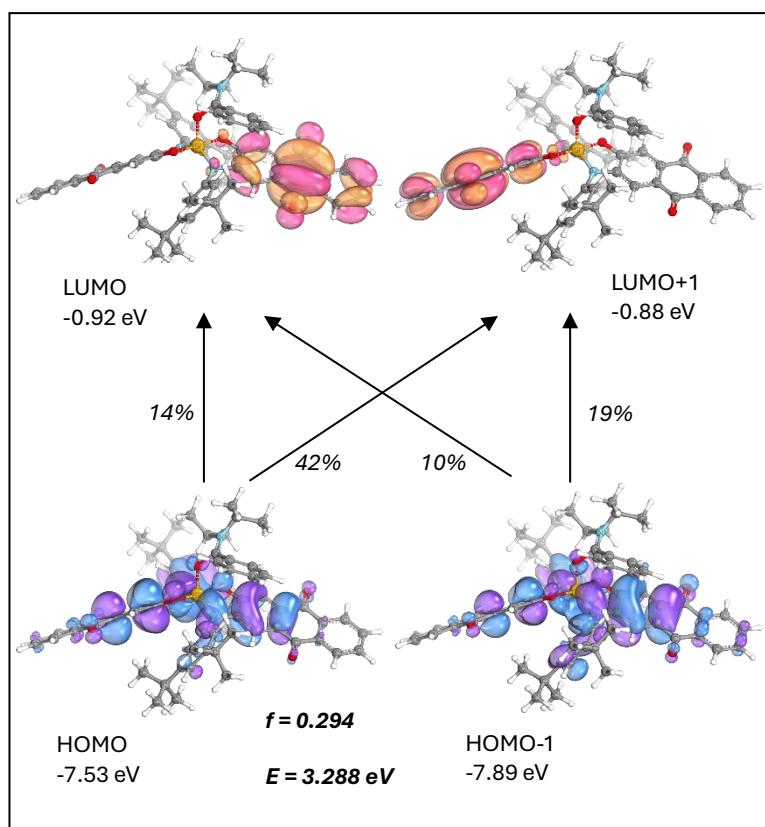


Figure S3.31. Molecular orbitals involved in the low energy CT transition of **1-DIBA**. HOMO->LUMO 14%, HOMO-1->LUMO 10%, HOMO-1->LUMO+1 19%, HOMO -> LUMO+1 42%.

1-dippNHC

Table S3.29. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-dippNHC**.

Wavelength [nm]	Oscillator strength (f)
506.4	0
499.8	0
402.8	0
402.8	0
386.3	0.117731782
381.9	0.194055568
372.5	0
372.5	0
372.4	0
368.3	0
363.3	0
363	0
354.3	0.00031863
354.2	0.000208581
342.4	0
339.7	0
329.9	0.000269825
329.8	0.000498366
325.9	0
325.7	0
311.6	0
310	0.079376332
308.4	0
306.9	0.076954866
304.4	0
304.3	0
303.9	0
291.6	0
289.2	0
278.9	0
278.8	0
277.7	2.783039868
273.5	0.181659116
272.5	0.009612058
269.4	0.280574247
267.4	0
264.7	0
264.6	0
264.2	0
263.5	0
263.4	0
263.3	0
262.4	0
259.5	0
258.9	0.074956339
257.8	0
257.6	0.40053585
255.7	0
254.9	0.329252292
254	0

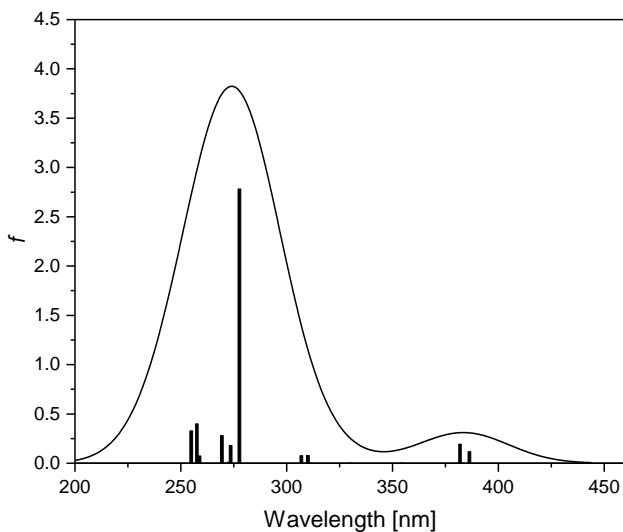


Figure S3.32. Simulated UV-vis spectrum of **1-dippNHC** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. Two low-energy transitions have significant oscillator strength. The maximum was determined from the broadened spectrum to be $\lambda_{\text{max}} = 383.5 \text{ nm}$.

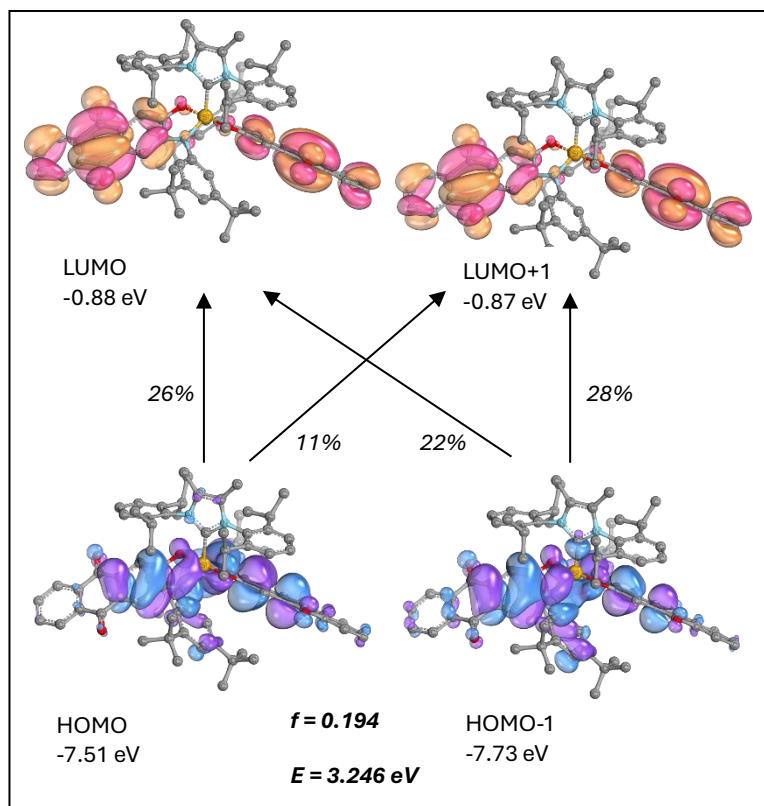


Figure S3.33. Molecular orbitals involved in the low energy CT transition of **1-dippNHC**. HOMO->LUMO 26%, HOMO-1->LUMO 22%, HOMO-1->LUMO+1 28%, HOMO -> LUMO+1 11%.

1-SIMes

Table S3.30. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-SIMes**.

Wavelength [nm]	Oscillator strength (f)
501.5	0
501.1	0
403.3	0
403.2	0
383.4	0.025102775
382.6	0.243971259
373.8	0
373.7	0
368.9	0
368.6	0
363.9	0
363.7	0
354.6	0.000696485
354.5	0.000833841
342.1	0
341.6	0
330.3	0.001567409
330.3	0.001546263
326.5	0
326.4	0
308.1	0.044309426
307.8	0.130724604
306.8	0
306.6	0
306.6	0
306.2	0
290.6	0
290.4	0
290	0
279.1	0
279	0
277.3	3.010884782
274.1	0.061292152
273.3	0.030784605
269	0.054541686
266.2	0
266.1	0
264.5	0
264.4	0
263.4	0
263.3	0
261.9	0
261.1	0
260.5	0
260	0
258.3	0.128191735
258	0.55954596
254.6	0
254.6	0
254.5	0.03759143

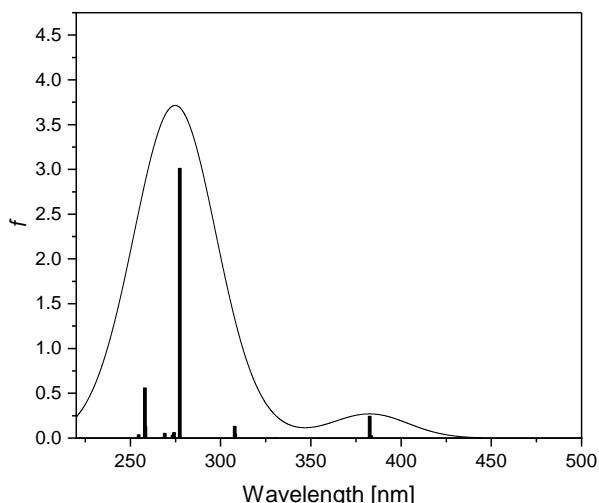


Figure S3.34. Simulated UV-vis spectrum of **1-SIMes** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. $\lambda_{\text{max}} = 382.5 \text{ nm}$.

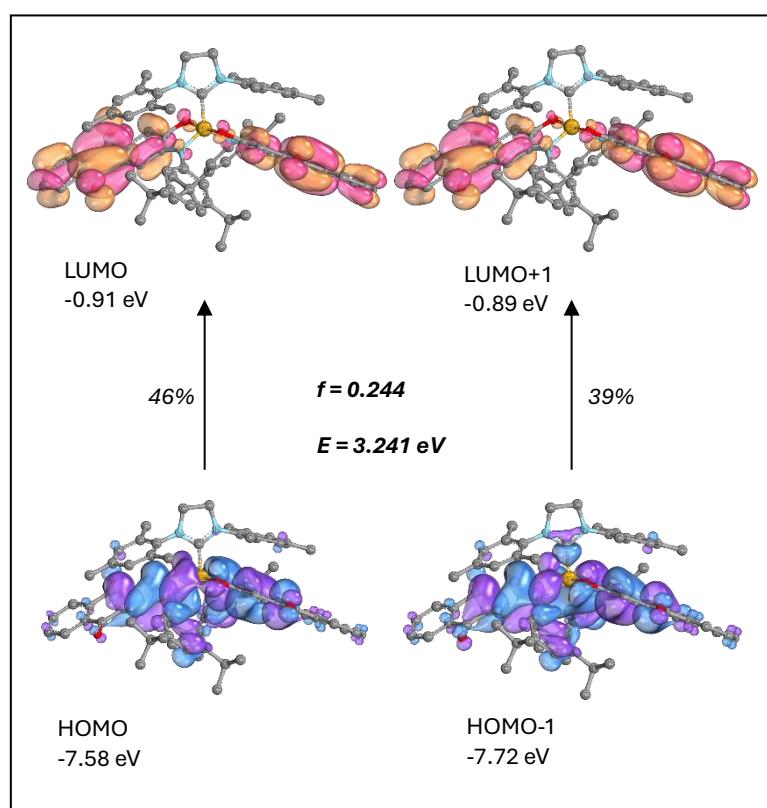


Figure S3.35. Molecular orbitals involved in the low energy CT transition of **1-SIMes**. HOMO \rightarrow LUMO 46%, HOMO-1 \rightarrow LUMO+1 39%.

1-iPrNHC

Table S3.31. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-iPrNHC**.

Wavelength [nm]	Oscillator strength (f)
500.9	0
499.9	0
402.3	0
402.3	0
381.9	0.04728963
380.8	0.251919463
372.4	0
372.3	0
363.7	0
363.7	0
363.5	0
363.5	0
353.8	0.000132128
353.8	0.000256024
342.7	0
342	0
329.5	0.000987088
329.5	0.000875121
325.9	0
325.8	0
306.1	0
305.7	0
305	0.013575356
304.8	0.085520644
288.8	0
288.8	0
279	0
279	0
276.6	2.98213764
273.8	0.130330215
272.9	0
272.5	0.19366517
267.7	0.247474881
262.5	0
262.4	0
260.8	0
260.3	0
256.9	0
256.4	0.201961164
256.3	0.606452701
256.2	0
253.1	0
253	0
252.8	0
252.3	0.079398086
252.1	0
251.5	0.222725174
250.9	1E-9
246.1	0
246	0

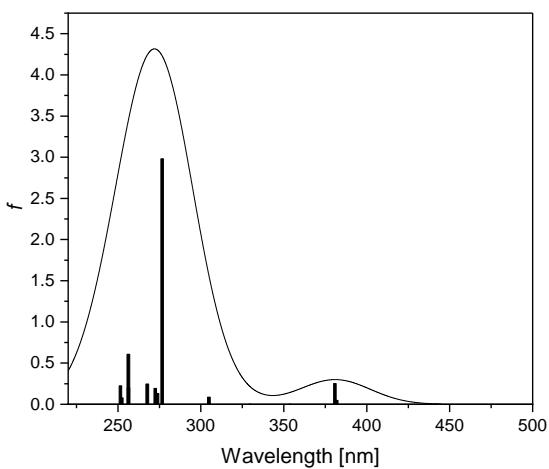


Figure S3.36. Simulated UV-vis spectrum of **1-iPrNHC** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. $\lambda_{\text{max}} = 380.9 \text{ nm}$.

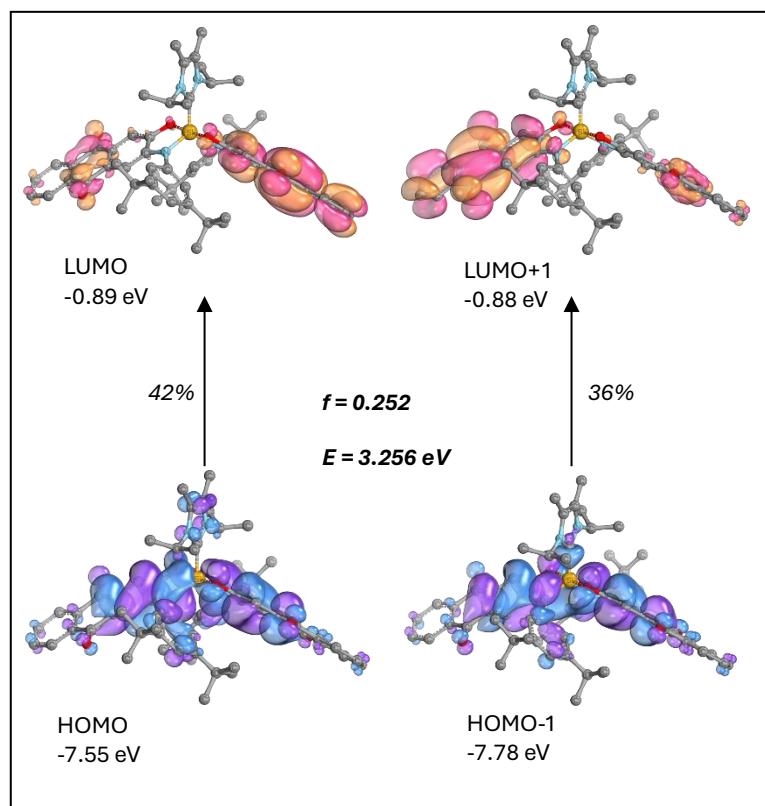


Figure S3.37. Molecular orbitals involved in the low energy CT transition of **1-iPrNHC**. HOMO- \rightarrow LUMO 42%, HOMO-1- \rightarrow LUMO+1 36%.

1-P(nBu)₃

Table S3.32. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-P(nBu)₃**.

Wavelength [nm]	Oscillator strength (f)
494	0
485.3	0
403.8	0
402.8	0
379.4	0.141867323
373.4	0
372.6	0.164280691
372.6	0
364.8	0
364.2	0
361.4	0
359.9	0
354.6	0.000329789
353.9	0.000182179
344.1	0
337.2	0
330.1	0.00119214
329.6	0.000809418
327	0
326.2	0
307.1	0
305	0.020531628
303.5	0
302.2	0.059609563
291.1	0
288.7	0
279.5	0
279.1	0
278.1	2.75876953
274.4	0.206443475
273.3	0.078805244
267.8	0.382284663
266.4	0
264.9	0
263.8	0
263	0
257.5	0
257.3	0.440870617
255.7	0
255.5	0.315260274
254.6	0.298176484
254.1	0
253.1	0
252.6	0
250.4	0
250.4	0.028559012
248.8	0
246.3	0
238.3	0.034856039
233.6	0

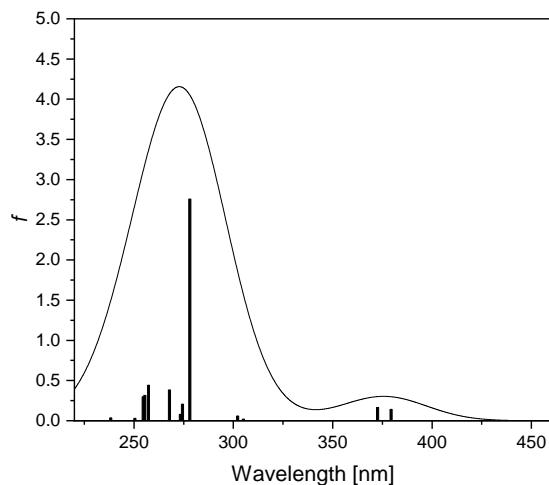


Figure S3.38. Simulated UV-vis spectrum of **1-P(nBu)₃** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. $\lambda_{\text{max}} = 375.6 \text{ nm}$.

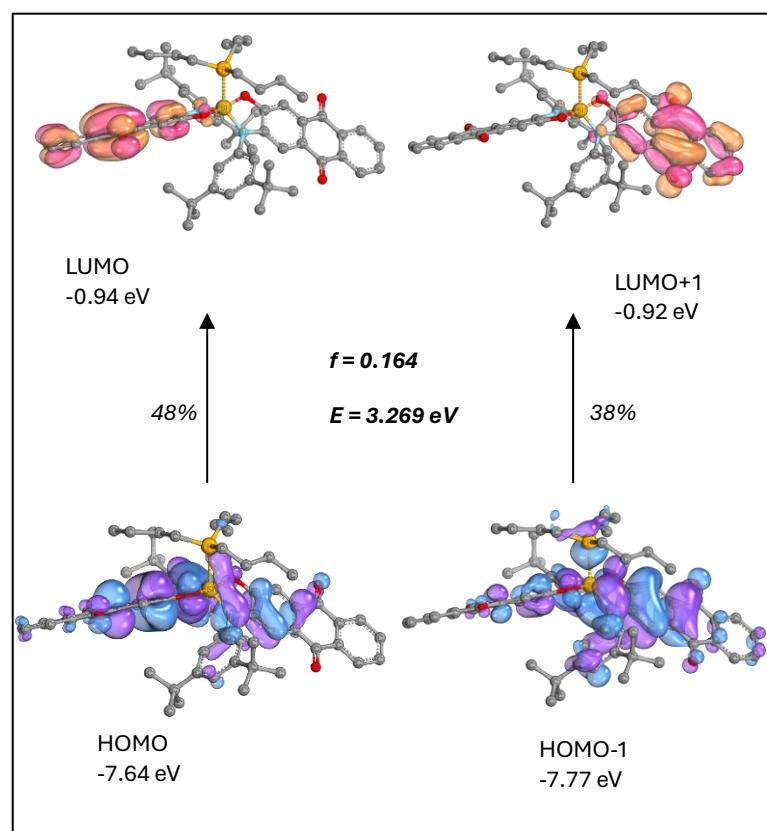


Figure S3.39. Molecular orbitals involved in the low energy CT transition of **1-P(nBu)₃**. HOMO->LUMO 48%, HOMO-1->LUMO+1 38%. Transition at 3.261 eV: HOMO->LUMO+1 46%, HOMO-1->LUMO 40%

1-PCy₃

Table S3.33. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-PCy₃**.

Wavelength [nm]	Oscillator strength (f)
496.7	0
488.9	0
402.4	0
402.3	0
379.7	0.129711599
374.9	0.168019321
372.3	0
372.2	0
363.9	0
363.9	0
361.4	0
360.9	0
353.8	0.000198702
353.6	0.000259009
341.7	0
338.6	0
329.5	0.000856829
329.5	0.000471743
326	0
326	0
306.7	0
303.9	0.031167341
303.8	0
303	0.061812832
288.9	0
288.5	0
279.1	0
278.8	0
276.2	2.801202883
274.1	0.101165958
272.4	0.418666565
266.8	0.187372058
263.1	0
262.9	0
262.5	0
260.6	0
256.7	0
256.1	0.465513288
255.3	0.541644726
254.9	0
254	0
253.8	0.060608545
252.8	0
252.7	0
251.7	0.034543077
251.5	0
246.3	0
245.7	0
234.8	0.006809999
234.5	0.009860696

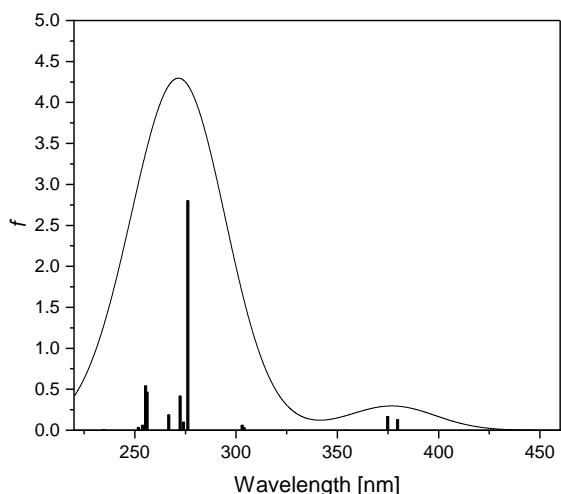


Figure S3.40. Simulated UV-vis spectrum of **1-PCy₃** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. $\lambda_{\text{max}} = 376.9$ nm.

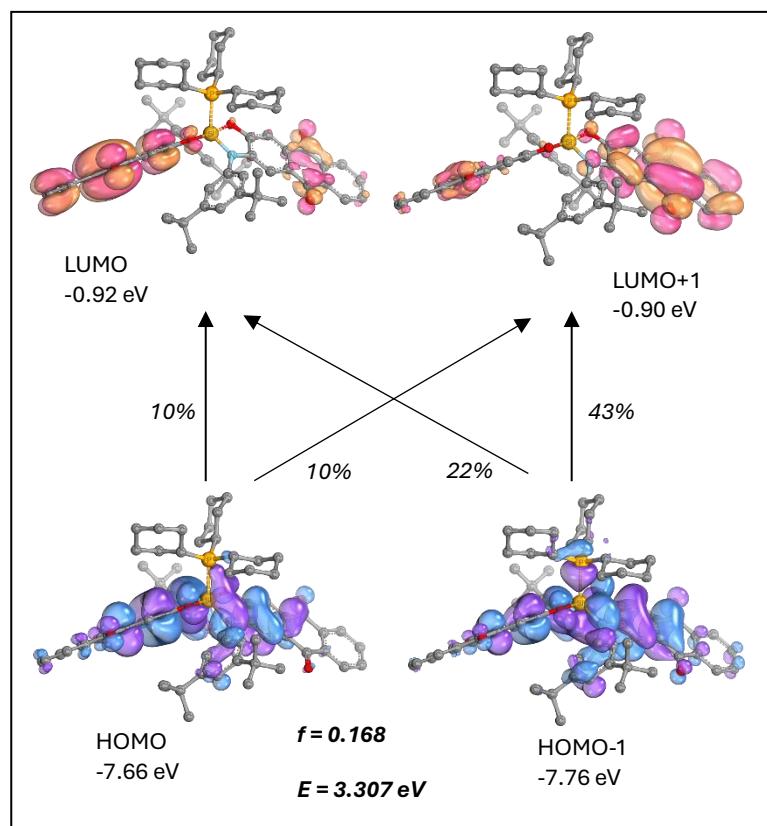


Figure S3.41. Molecular orbitals involved in the low energy CT transition of **1-PCy₃**. HOMO->LUMO 10%, HOMO-1->LUMO+1 43%, HOMO->LUMO+1 10%, HOMO-1->LUMO 22%. Transition at 3.265 eV: HOMO->LUMO 42%, HOMO-1->LUMO+1 4%, HOMO->LUMO+1 28%, HOMO-1->LUMO 12%.

1-DABCO

Table S3.34. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound **1-DABCO**.

Wavelength [nm]	Oscillator strength (f)
482.7	0
479.8	0
402.5	0
402.1	0
372.5	0
372.2	0
370	0.077217448
368.1	0.219262257
364.3	0
364.2	0
358.6	0
358.5	0
353.7	0.000428381
353.5	0.000344328
334.5	0
333.4	0
329.6	0.000676977
329.4	0.000677398
326.1	0
325.9	0
302.7	0
302.6	0
300.7	0.065293901
300.3	0.062208209
287.9	0
287.7	0
279.2	0
279	0
274.6	1.147756885
274.1	0.08778132
269.4	1.991034153
262.6	0.173247649
262.1	0
262	0
261.7	0
261.6	0
255.3	0.339116828
255	0.709680142
253.8	0
253.2	0
252.6	0
252.6	0
245.7	0
245.2	0
244.6	0
244.3	0.057167081
243.8	0
243.3	0.064359824
233.7	0.006838201
233.5	0.000735247

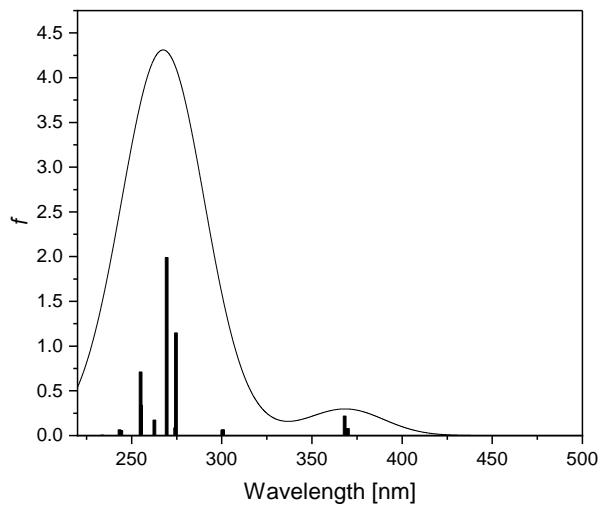


Figure S3.42. Simulated UV-vis spectrum of **1-DABCO** in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM. $\lambda_{\text{max}} = 368.4 \text{ nm}$.

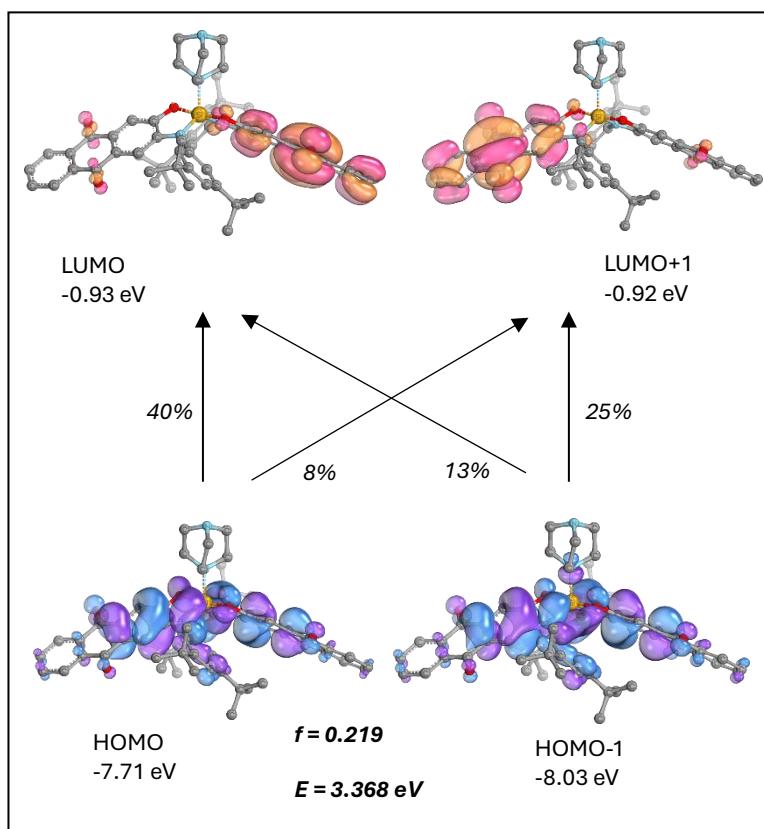


Figure S3.43. Molecular orbitals involved in the low energy CT transition of **1-DABCO**. HOMO \rightarrow LUMO 40%, HOMO-1 \rightarrow LUMO+1 25%, HOMO \rightarrow LUMO+1 8%, HOMO-1 \rightarrow LUMO 13%.

1-(pyridine)₂

Table S3.35. Calculated TD-DFT/TDA (ω B97X-D3/def2-TZVPP/SMD(DCM)) electronic transitions for compound 1-(pyridine)₂.

Wavelength [nm]	Oscillator strength (f)
530.7	0
530.5	0
404	0.002020827
402.5	0
402	0
401.9	0.430672072
371.6	0
371.2	0
368.2	0
368.1	0
363.6	0
363.4	0
360.3	0
360.2	0
353.8	7.1084E-5
353.5	7.1703E-5
329.2	7.7739E-5
328.9	9.0128E-5
325.7	0
325.6	0
311.2	0.076418422
310.9	0.000126911
310.3	0
310.2	0
289.5	0
289.4	0
288.7	0
288.5	2.749077853
288.3	0
288.2	0
284.3	0.035436076
283.1	0
280.6	0
279.6	0
278.8	0
278.5	0
276.8	0.071595775
275.5	0.353872657
273	0
272.8	0.002906425
272.6	0.043030254
271	0.002713319
270.2	0
269.8	0
268	0.092059306
267.1	0
265.9	0
263.9	0
262	0
261.2	0.002146404

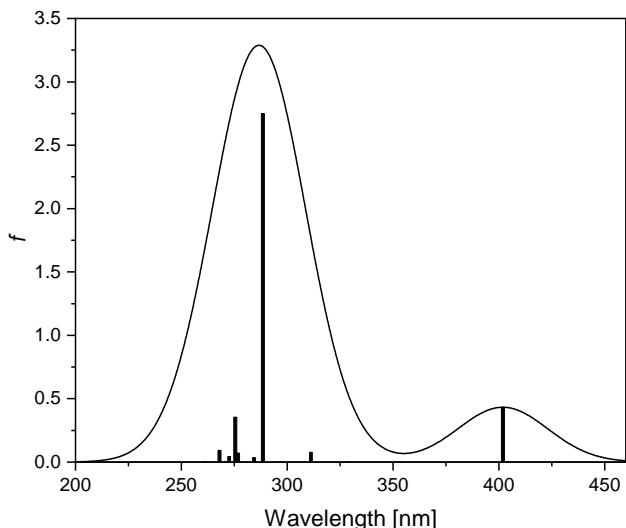


Figure S3.44. Simulated UV-vis spectrum of 1-(pyridine)₂ in DCM (ω B97X-D3/def2-TZVPP/SMD(DCM)). Gaussian line broadening with 50 nm FWHM.

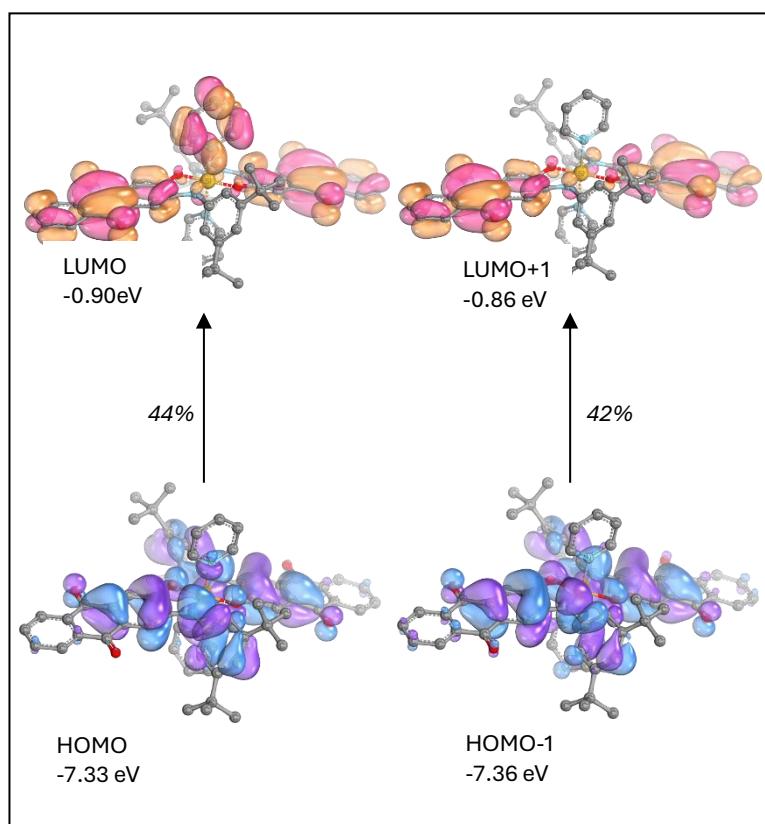


Figure S3.45. Molecular orbitals involved in the low energy CT transition of 1-(pyridine)₂. HOMO \rightarrow LUMO 44%, HOMO-1 \rightarrow LUMO+1 42%.

3.6 Correlation between TD-DFT/TDA excitations and experimental UV-vis spectra

For the correlation between experimental spectra and calculated transitions, all calculated energies were shifted by -0.56 eV to match λ_{\max} of **1**.

Table S3.36. Calculated and experimental absorption energies and maximum absorption wavelengths for **1** and adducts **1-X**.

Compound	E (at λ_{\max} , TD-DFT) [eV] <i>wB97X-D3/def2-TZVPP/SMD(DCM)</i>	E_{corr} (at λ_{\max} , TD-DFT, -0.56 eV) [eV] <i>wB97X-D3/def2-TZVPP/SMD(DCM)</i>	$\lambda_{\max, \text{corr}}(\text{TD-DFT})$ [nm] <i>wB97X-D3/def2-TZVPP/SMD(DCM)</i>	$\lambda_{\max}(\text{exp, DCM})$ [nm]	E (at λ_{\max} , exp, DCM) [eV]
1	3.602	3.042	407.6	407	3.05
[1-F]	3.114	2.554	485.4	495	2.50
[1-Cl]	3.129	2.569	482.6	489	2.54
[1-N₃]	3.122	2.562	483.9	496	2.50
[1-NCS]	3.169	2.609	475.2	484	2.56
[1-CN]	3.179	2.619	473.4	481	2.58
[1-Br]	3.124	2.564	483.5	479	2.59
1-pyridine	3.353	2.793	443.9	452	2.74
1-DMAP	3.307	2.747	451.3	466	2.66
1-OPEt	3.284	2.724	455.1	469	2.64
1-HMPA	3.267	2.707	458.0	474	2.62
1-DMSO	3.332	2.772	447.3	460	2.70
1-DIBA	3.288	2.728	454.5	467	2.65
1-dippNHC	3.233	2.673	463.8	475	2.61
1-SIMes	3.240	2.680	462.6	472	2.63
1-iPrNHC	3.254	2.694	460.2	468	2.65
1-P(nBu)₃	3.300	2.74	452.5	458	2.71
1-PCy₃	3.290	2.73	454.1	461	2.69
1-DABCO	3.363	2.803	442.3	449	2.76

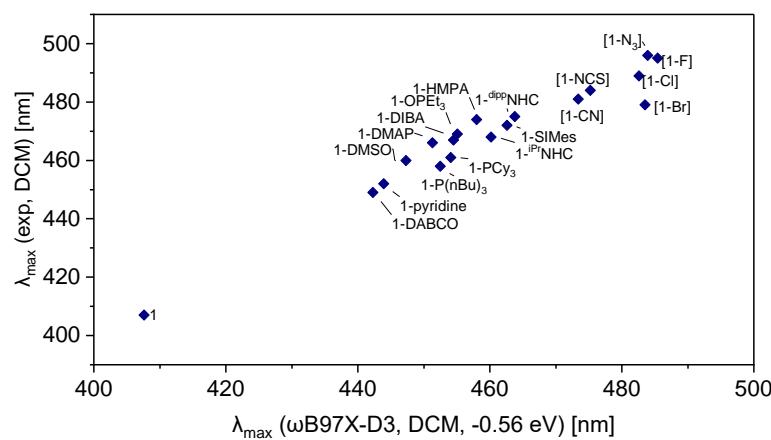


Figure S3.46. Correlation between calculated and experimental absorption spectra. Calculated λ_{\max} values were redshifted by 0.56 eV.

3.7 Correlations between UV-vis Data and Calculated Thermodynamics

Correlation between experimental UV-vis absorption energies and computed thermodynamics for 1

All data was fitted with OriginLab 2024b.

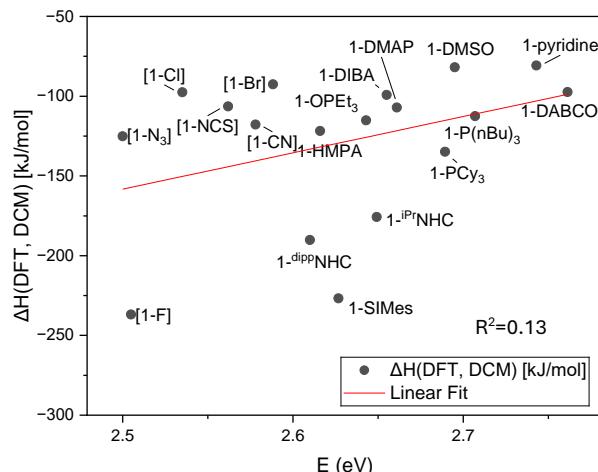


Figure S3.47. Correlation between experimental absorption energies and enthalpy of acid-base association for adducts **1-X** in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r2scan-3c).

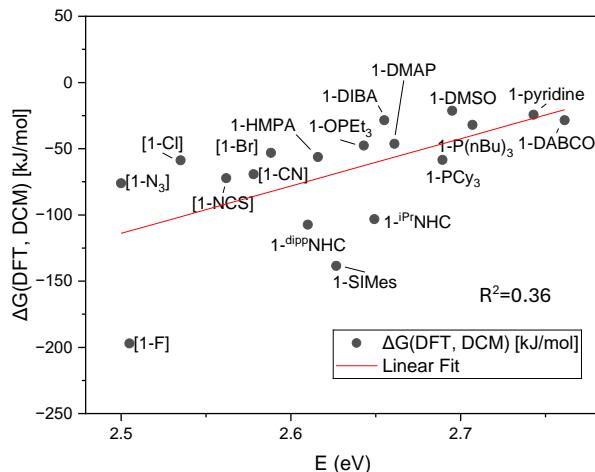


Figure S3.48. Correlation between experimental absorption energies and Gibbs free energies of acid-base association for adducts **1-X** in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r2scan-3c).

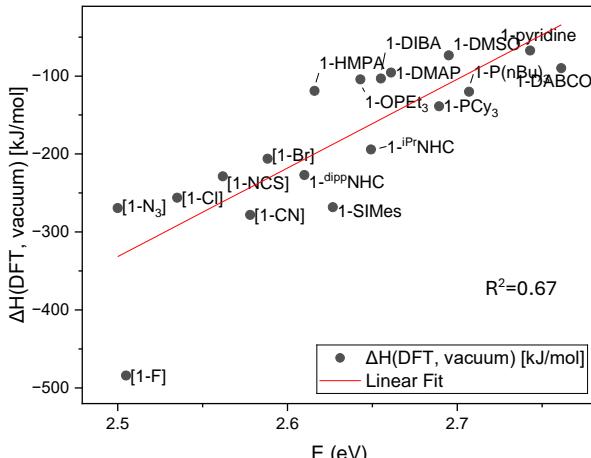


Figure S3.49. Correlation between experimental absorption energies and enthalpy of acid-base association for adducts **1-X** in vacuum (DSD-BLYP(D3BJ)/def2-TZVPP//r2scan-3c).

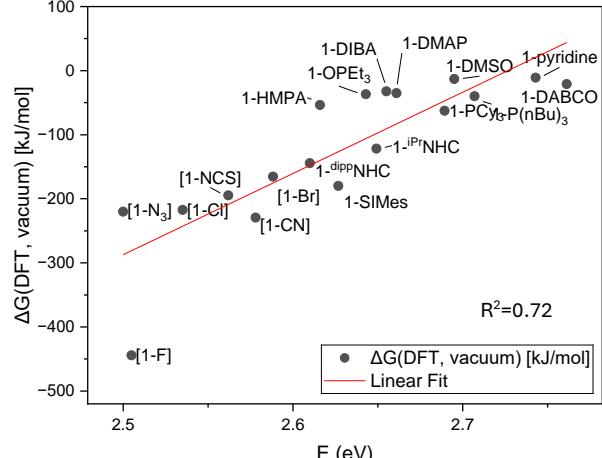


Figure S3.50. Correlation between experimental absorption energies and Gibbs free energies of acid-base association for adducts **1-X** in vacuum (DSD-BLYP(D3BJ)/def2-TZVPP//r2scan-3c).

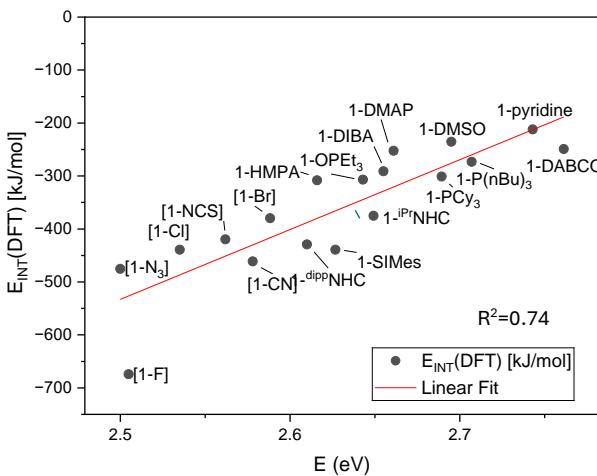


Figure S3.51. Correlation between experimental absorption energies and interaction energies between acid and base fragments adducts **1-X** (DSD-BLYP(D3BJ)/def2-TZVPP).

To compare the obtained data with alternative methods, COSMO-RS solvation correction and anions with explicit counterions were calculated on a subset of the bases.

SMD vs COSMO-RS solvation model

The COSMO-RS solvation model was used to obtain solvent corrected affinities ($\Delta H/\Delta G$). Correlations with experimental absorption energies are similarly weak as the SMD model.

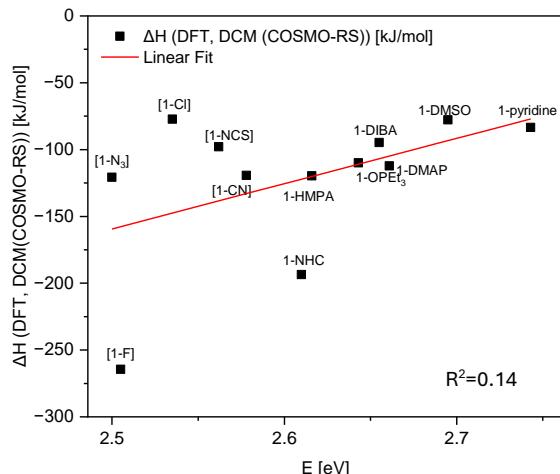


Figure S3.52. Correlation between experimental absorption energies and enthalpy of acid-base association for adducts **1-X** in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/COSMO-RS(DCM)//r2scan-3c).

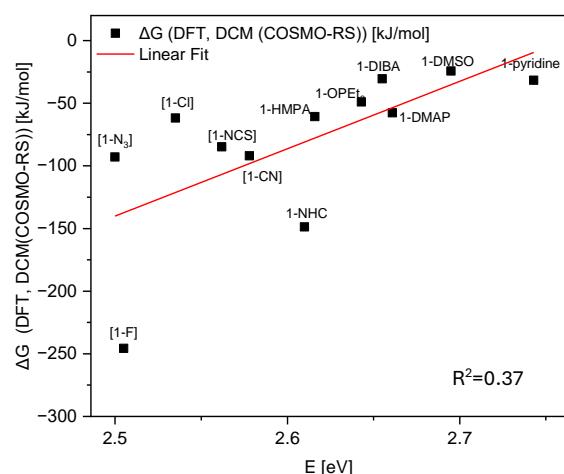


Figure S3.53. Correlation between experimental absorption energies and Gibbs free energies of acid-base association for adducts **1-X** in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/COSMO-RS(DCM)//r2scan-3c).

Explicit counterions

For calculations with explicit counterions, only the correlation of absorption energies with calculated reaction enthalpies is given due to entropic errors resulting from small imaginary frequencies.

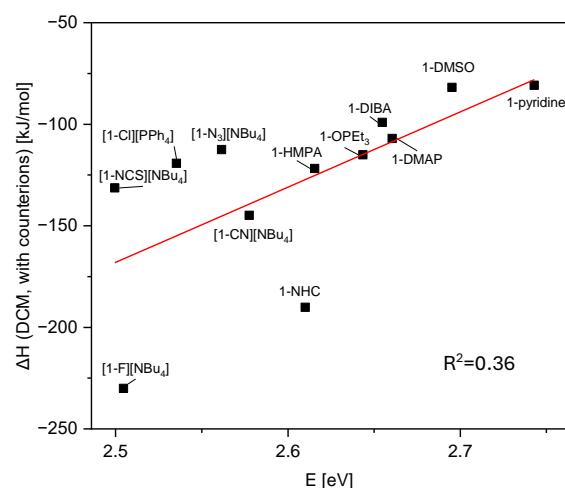


Figure S3.54. Correlation between experimental absorption energies and enthalpy of acid-base association for adducts **[1-X][Cation]** or **1-X** for neutral donors in DCM (DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r2scan-3c).

Correlation between $\Delta\Delta H$ (difference between first and second binding enthalpy) and eLB/gLB

The binding of a Lewis base results in a deactivation of the Lewis acidic silicon centre towards a second binding event. The extent of this deactivation is a function of the effect of a given Lewis base (effective Lewis basicity, eLB) and can be estimated by the difference in enthalpy between the two binding events. For example, the highest deactivation is observed for fluoride, where the second coordination is less favourable than the first by 170 kJ/mol.

To assess whether this deactivation is a function of effective Lewis basicity as proposed in this work, the deactivation was correlated with the absorption energy of the corresponding silicon complexes (eLB) and the enthalpy of the first base coordination (gLB). A much stronger correlation is observed for eLB ($R^2 = 0.88$, figure 3.43) than for gLB ($R^2 = 0.49$, figure 3.44).

It is important to note that steric effects play a much larger role in the second binding event. This is evidenced by failed convergence or dissociation during geometry optimisations for several of the bulkier Lewis base adducts. Therefore, the discussion is restricted to sterically undemanding bases.

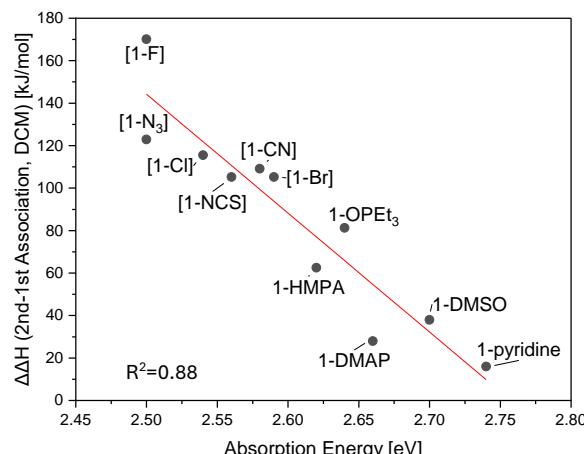


Figure S3.55. Correlation between experimental absorption energies (eLB) and $\Delta\Delta H$ in DCM (difference between first and second binding enthalpy, DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r2scan-3c).

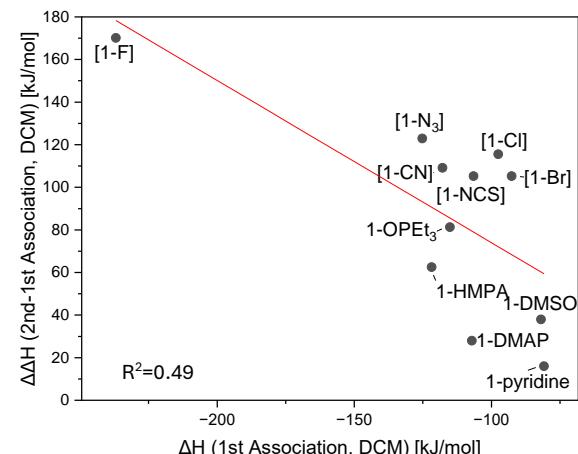


Figure S3.56. Correlation between the first binding enthalpy and $\Delta\Delta H$ in DCM (difference between first and second binding enthalpy, DSD-BLYP(D3BJ)/def2-TZVPP/SMD(DCM)//r2scan-3c).

3.8 Impact of Dipole Moments on Solvation Effects in Adducts 1-X

For solvation processes of free bases and the formed adducts in DCM, significant differences between neutral and anionic bases arise from charges and the emerging dipole moments in the formed adducts. For anionic adducts, dipole moments are oriented in the direction donor→Si. Local charge density and the solvent accessible surface area is reduced compared to the fragments. Consequently, the separated fragments receive higher ΔH_{Solv} than the adduct (Table S3.30, entries 1-5). In contrast, dipole moments are oriented in the direction Si→donor for neutral adducts, with an increase in dipole moment. This leads to a pronounced solvent stabilisation for the formed adducts compared to the separated fragments for small bases (Table S3.30, entries 6,7,8,10). Similar diverging effects of neutral and charged species have been elucidated for neutral amine-borane adducts and charged transition metal complexes.^{21, 22} This effect is only relevant for small bases. For large bases (e.g. NHC, DIBA), the fragments receive higher solvent stabilisation despite the increasing dipole moment, presumably because of a large reduction in surface area in the formed adduct.

For anionic bases, the solvation enthalpy of the adducts 1-X are almost identical (mean deviation 5 kJ/mol), suggesting that desolvation of the individual bases is the pivotal effect responsible for observed differences (mean deviation for ΔH_{Solv} of anionic bases 35 kJ/mol). Similarly, for neutral bases, the variance in ΔH_{Solv} is larger for individual bases (16 kJ/mol) than for the adducts (5 kJ/mol). Overall, the variance in solvation energies for bases is 90 kJ/mol and 22 kJ/mol for the adducts.

Table S3.37. Calculated dipole moments of complexes 1-X (DSD-BLYP/def2-TZVPP//r2scan-3c), ΔH_{Solv} of adducts (1-X), and ΔH_{Solv} of the separated acid and base fragments (1 + X). (SMD(DCM)).

Entry	Compound	Dipole Moment (1-X) [D]	$\Delta H_{\text{Solv}}(X)$ [kJ/mol]	$\Sigma \Delta H_{\text{Solv}}(1+X)$ [kJ/mol]	$\Delta H_{\text{Solv}}(1-X)$ [kJ/mol]	$\Delta \Delta H_{\text{Solv}}$ [kJ/mol]
1	1-F	6.24622	-347.87	-506.54	-259.35	247.19
2	1-Cl	6.68931	-261.28	-419.95	-261.46	158.49
3	1-Br	6.22181	-211.54	-370.21	-256.75	113.46
4	1-N ₃	7.12006	-229.22	-387.89	-243.81	144.08
5	1-NCS	8.47473	-214.39	-373.06	-250.77	122.29
6	1-CN	9.59049	-257.32	-415.99	-255.72	160.27
7	1-pyridine	7.27340	-29.60	-188.27	-201.81	-13.54
8	1-DMAP	11.95456	-43.50	-202.17	-213.55	-11.38
9	1-OPEt ₃	10.58696	-43.61	-202.28	-213.04	-10.76
10	1-HMPA	10.64565	-39.50	-198.17	-200.90	-2.73
11	1-DMSO	8.15066	-32.71	-191.38	-199.64	-8.26
12	1-DIBA	9.73996	-53.94	-212.61	-208.58	4.03
13	1-dippNHC	8.84686	-90.20	-248.87	-211.89	36.98
14	1-SIMes	6.96345	-92.09	-250.76	-209.34	41.42
15	1-iPrNHC	8.72549	-69.56	-228.23	-203.65	24.58
16	1-P(nBu) ₃	5.24454	-38.91	-197.58	-189.86	7.72
17	1-PCy ₃	6.88354	-52.64	-211.31	-207.37	3.94
18	1-DABCO	5.07830	-37.08	-195.75	-203.11	-7.36

3.9 Comparison of Absorption Energy and pK_A Values

Not for all bases pK_A values were available. No general correlation was found between the available pK_A values and the absorption energy of the corresponding adducts **1-X**. However, the order of the base strength within structurally similar bases is maintained. Similar observations were made for gLB (ΔH in DCM for the formation of **1-X**), which differs from pK_A (gLB towards H^+ in H_2O) due to the change in both reference Lewis acid (**1** vs. H^+) and solvation contributions (DCM vs. H_2O).



Table S3.38. Absorption energy of complexes **1-X** and available pK_A values (HX) in H_2O .

Base	Absorption energy (1-X) [eV]	pK_A (conjugated acid HX , H_2O)
[F]	2.505	3.2 ²³
[Cl]	2.535	-7.3 ²³
[Br]	2.588	-9.8 ²³
[N ₃]	2.500	4.65 ²⁴
[NCS]	2.562	-1.28 ²⁵
[CN]	2.578	9.36 ²⁶
pyridine	2.743	5.2 ²⁷
DMAP	2.661	9.87 ²⁸
OPEt ₃	2.643	-
HMPA	2.616	-0.97 ²⁹
DMSO	2.695	-1.54 ²⁹
DIBA	2.655	-
dippNHC	2.610	-
SIMes	2.627	-
iPrNHC	2.649	-
P(nBu) ₃	2.707	8.4 ³⁰
PCy ₃	2.689	9.7 ³⁰
DABCO	2.761	8.7 ³¹

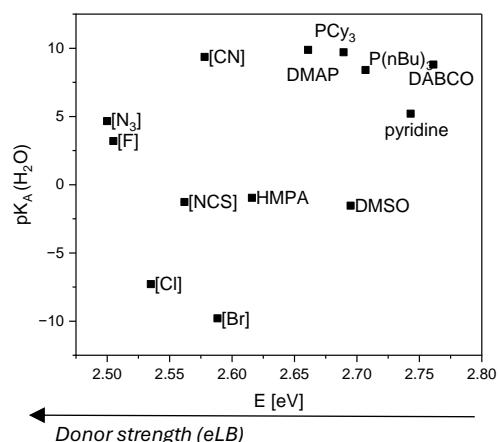


Figure S3.57. Experimental absorption energies (eLB) and available pK_A values.

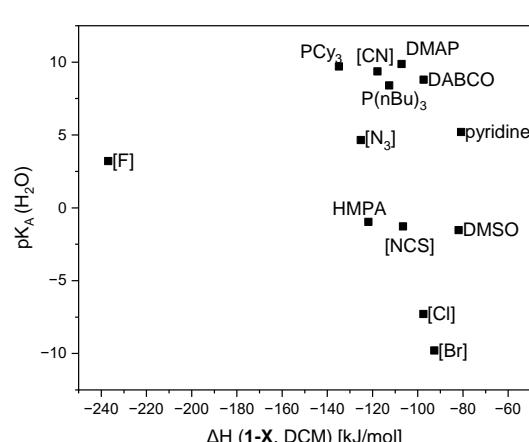


Figure S3.58. Computed enthalpy of adduct formation in DCM (gLB) and available pK_A values.

4 X-Ray Crystallography

Crystals were immersed in perfluorinated polyether oil and fixed on a cryo-loop. Data were collected from a shock-cooled single crystal at 100(2) K on a Bruker D8 VENTURE dual wavelength Mo/Cu four-circle diffractometer with a microfocus sealed X-ray tube using a mirror optics as monochromator and a Bruker PHOTON III detector. The diffractometer was equipped with an Oxford Cryostream 700 low temperature device and used MoK_α radiation ($\lambda = 0.71073 \text{ \AA}$). All data were integrated with SAINT and a multi-scan absorption correction using SADABS was applied.^{32, 33} The structure was solved by direct methods using SHELXT and refined by full-matrix least-squares methods against F^2 by SHELXL-2019/2.^{34, 35} All non-hydrogen atoms were refined with anisotropic displacement parameters. All hydrogen atoms were refined isotropic on calculated positions using a riding model with their U_{iso} values constrained to 1.5 times the U_{eq} of their pivot atoms for terminal sp³ carbon atoms and 1.2 times for all other carbon atoms. Crystallographic data for the structures reported here have been deposited with the Cambridge Crystallographic Data Centre.³⁶ CCDC 2403101-2403102 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures. This report and the CIF file were generated using FinalCif.³⁷

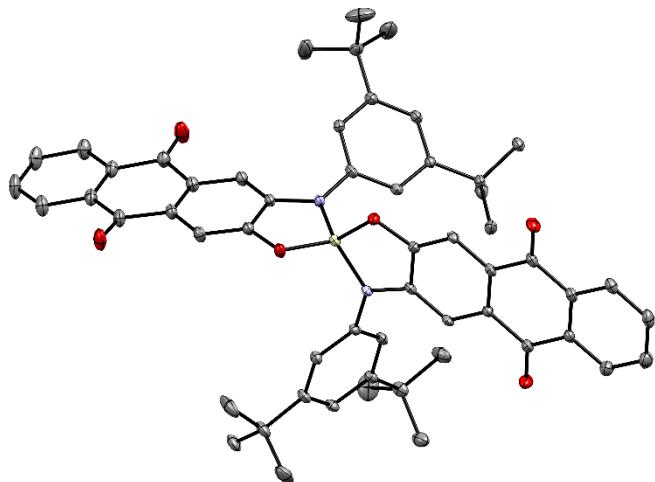


Figure S4.1. Solid-state structure of **1**, crystallised from DCM/*n*-pentane at room temperature. Thermal ellipsoids are displayed at the 50% probability level. Hydrogen atoms and cocrystallised DCM molecules are omitted for clarity.

CCDC number	2403101
Empirical formula	C ₅₆ H ₅₄ Cl ₀ N ₂ O ₆ Si
Formula weight	879.10
Temperature [K]	100(2)
Crystal system	triclinic
Space group (number)	P $\bar{1}$ (2)
<i>a</i> [\AA]	8.954(3)
<i>b</i> [\AA]	14.794(6)
<i>c</i> [\AA]	20.837(8)
α [$^\circ$]	106.67(2)
β [$^\circ$]	93.268(11)
γ [$^\circ$]	90.445(12)
Volume [\AA^3]	2639.0(17)
<i>Z</i>	2
ρ_{calc} [gcm^{-3}]	1.106
μ [mm^{-1}]	0.093
<i>F</i> (000)	932
Crystal size [mm^3]	0.885×0.214×0.182
Crystal colour	yellow
Crystal shape	block
Radiation	Mo K_α ($\lambda=0.71073 \text{\AA}$)
2 θ range [$^\circ$]	3.98 to 58.49 (0.73 \AA)
Index ranges	$-12 \leq h \leq 12$ $-20 \leq k \leq 20$ $-28 \leq l \leq 28$
Reflections collected	202726
Independent reflections	14321 $R_{\text{int}} = 0.0695$ $R_{\text{sigma}} = 0.0292$
Completeness to $\theta = 25.242^\circ$	99.9 %
Data / Restraints / Parameters	14321/0/601
Absorption correction	0.6714/0.7458
T _{min} /T _{max} (method)	(multi-scan)
Goodness-of-fit on F^2	1.074
Final <i>R</i> indexes	$R_1 = 0.0616$ $wR_2 = 0.1580$
[$\geq 2\sigma(I)$]	
Final <i>R</i> indexes	$R_1 = 0.0693$ $wR_2 = 0.1631$
[all data]	
Largest peak/hole [e\AA^{-3}]	0.73/-0.48

[1-Cl][PPh₄]

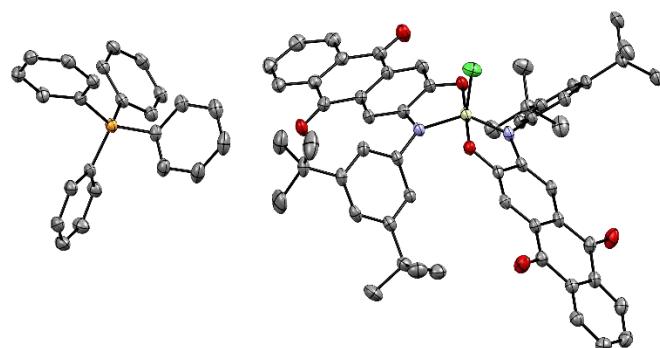


Figure S4.2. Solid-state structure of [1-Cl][PPh₄], crystallised from DCM/n-pentane at -40 °C. Thermal ellipsoids are displayed at the 50% probability level. Hydrogen atoms and cocrystallised DCM molecules are omitted for clarity.

CCDC number	2403102
Empirical formula	C _{83.92} H _{81.84} Cl _{8.84} N ₂ O ₆ PSi
Formula weight	1586.81
Temperature [K]	100(2)
Crystal system	monoclinic
Space group (number)	P2 ₁ /c (14)
a [Å]	23.7594(15)
b [Å]	12.8395(8)
c [Å]	27.1540(16)
α [°]	90
β [°]	101.651(2)
γ [°]	90
Volume [Å³]	8112.9(9)
Z	4
ρ_{calc} [gcm⁻³]	1.299
μ [mm⁻¹]	0.393
F(000)	3307
Crystal size [mm³]	0.22×0.10×0.05
Crystal colour	red
Crystal shape	plate
Radiation	MoK _α ($\lambda=0.71073\text{ \AA}$)
2θ range [°]	3.79 to 52.94 (0.80 Å)
Index ranges	-29 ≤ h ≤ 29 -16 ≤ k ≤ 16 -34 ≤ l ≤ 33
Reflections collected	511533
Independent reflections	16709 $R_{\text{int}} = 0.0981$ $R_{\text{sigma}} = 0.0259$
Completeness to θ = 25.242°	100.0 %
Data / Restraints / Parameters	16709/531/1034
Absorption correction	0.6796/0.7454
T_{min}/T_{max} (method)	(multi-scan)
Goodness-of-fit on F²	1.037
Final R indexes	$R_1 = 0.0507$
[I>2σ(I)]	wR ₂ = 0.1242
Final R indexes	$R_1 = 0.0672$
[all data]	wR ₂ = 0.1357
Largest peak/hole [eÅ⁻³]	0.71/-0.78

1-PCy₃

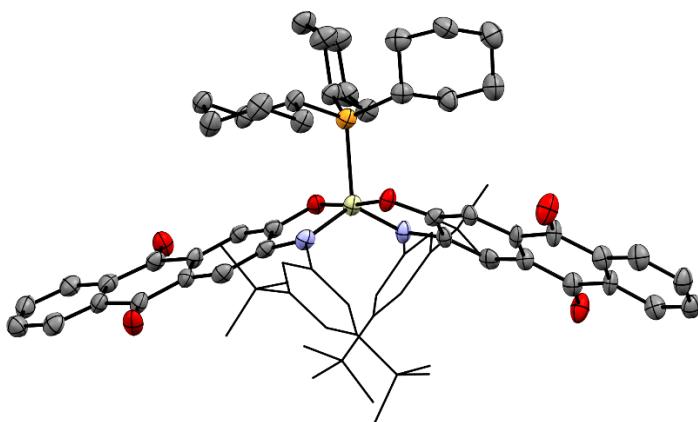


Figure S4.3. Solid-state structure of **1-PCy₃**, crystallised from DCM/n-pentane at -40 °C. Thermal ellipsoids are displayed at the 50% probability level. Hydrogen atoms and cocrystallised DCM and *n*-pentane molecules are omitted for clarity. Nitrogen substituents displayed as wireframe for clarity.

CCDC number	2466565
Empirical formula	C _{84.50} H ₁₁₁ Cl ₆ N ₂ O ₆ PSi
Formula weight	1522.51
Temperature [K]	100(2)
Crystal system	triclinic
Space group (number)	P $\bar{1}$ (2)
a [Å]	15.0843(17)
b [Å]	15.9059(17)
c [Å]	20.643(2)
α [°]	108.857(4)
β [°]	95.289(4)
γ [°]	114.939(3)
Volume [Å³]	4096.6(8)
Z	2
ρ_{calc} [gcm⁻³]	1.234
μ [mm⁻¹]	0.296
F(000)	1622
Crystal size [mm³]	0.12×0.13×0.27
Crystal colour	red
Crystal shape	block
Radiation	MoK α (λ =0.71073 Å)
2θ range [°]	4.05 to 50.05 (0.84 Å)
Index ranges	-17 ≤ h ≤ 17 -18 ≤ k ≤ 17 0 ≤ l ≤ 24
Reflections collected	14310
Independent reflections	14310 $R_{\text{int}} = 0.0923$ $R_{\text{sigma}} = 0.1327$
Completeness to	99.0 %
θ = 25.027°	
Data / Restraints / Parameters	14310 / 460 / 1049
Absorption correction T_{min}/T_{max}	0.5968 / 0.7453
(method)	(multi-scan)
Goodness-of-fit on F²	1.038
Final R indexes	$R_1 = 0.1148$
[$\text{l} \geq 2\sigma(\text{l})$]	$wR_2 = 0.2872$
Final R indexes	$R_1 = 0.1813$
[all data]	$wR_2 = 0.3177$
Largest peak/hole [eÅ⁻³]	0.61/-0.54

Unrefined structures of $\mathbf{1-dippNHC}$, $[\mathbf{1-NCS}][\mathbf{NBu_4}]$ and $\mathbf{1-HMPA}$

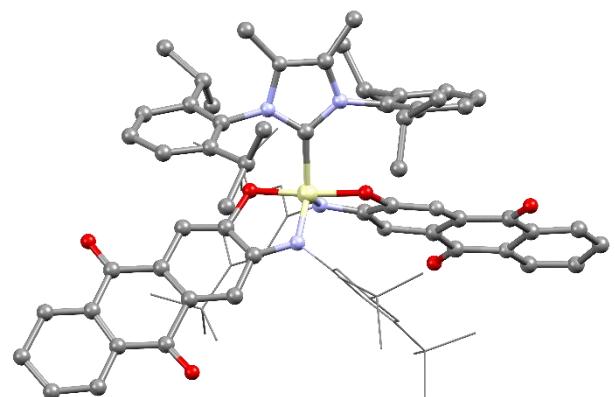


Figure S4.4. Unrefined solid-state structure of $\mathbf{1-dippNHC}$, crystallised from DCM at -40 °C.

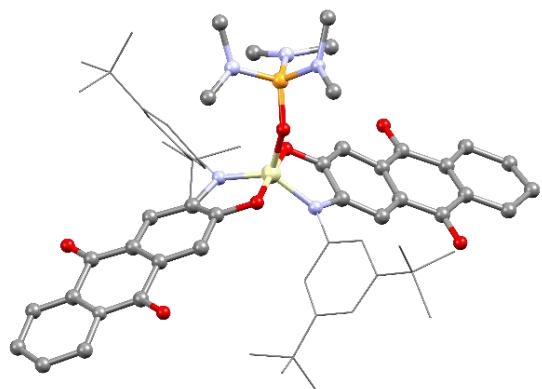


Figure S4.5. Unrefined solid-state structure of $\mathbf{1-HMPA}$, crystallised from THF/n-pentane at -40 °C. Cocrystallised THF molecules omitted for clarity.

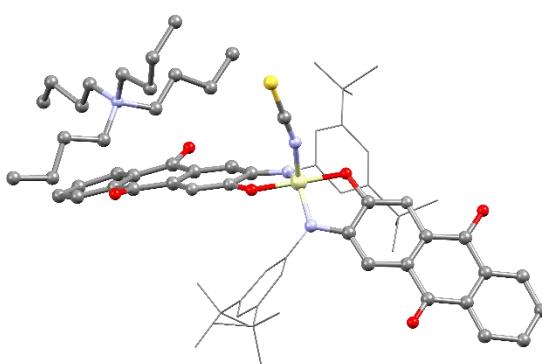


Figure S4.6. Unrefined solid-state structure of $[\mathbf{1-NCS}][\mathbf{NBu_4}]$, crystallised from 1,2-difluorobenzene/n-pentane at -40 °C. Cocrystallised oDFB molecules omitted for clarity.

5 IR Spectra of Isolated Compounds

ATR-IR spectra were measured on an *Agilent Cary 630* spectrometer inside a nitrogen-filled glovebox and subsequently plotted and analysed with OriginLab 2024.

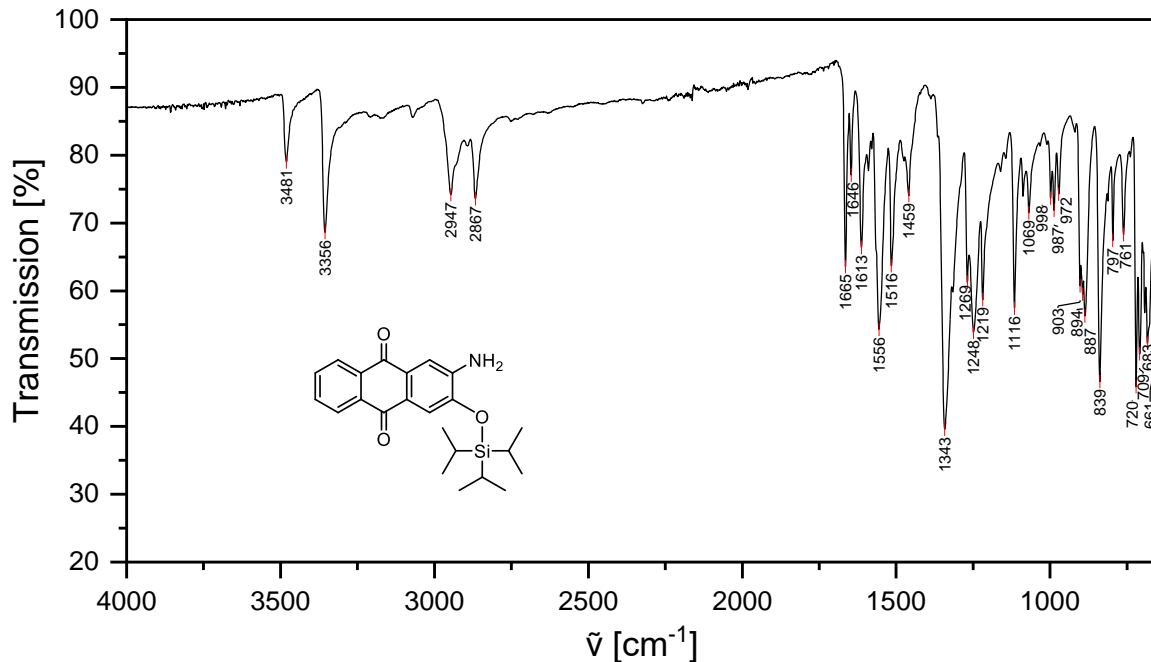


Figure S5.1. ATR-IR spectrum of 2-amino-3-(triisopropylsilyloxy)anthraquinone.

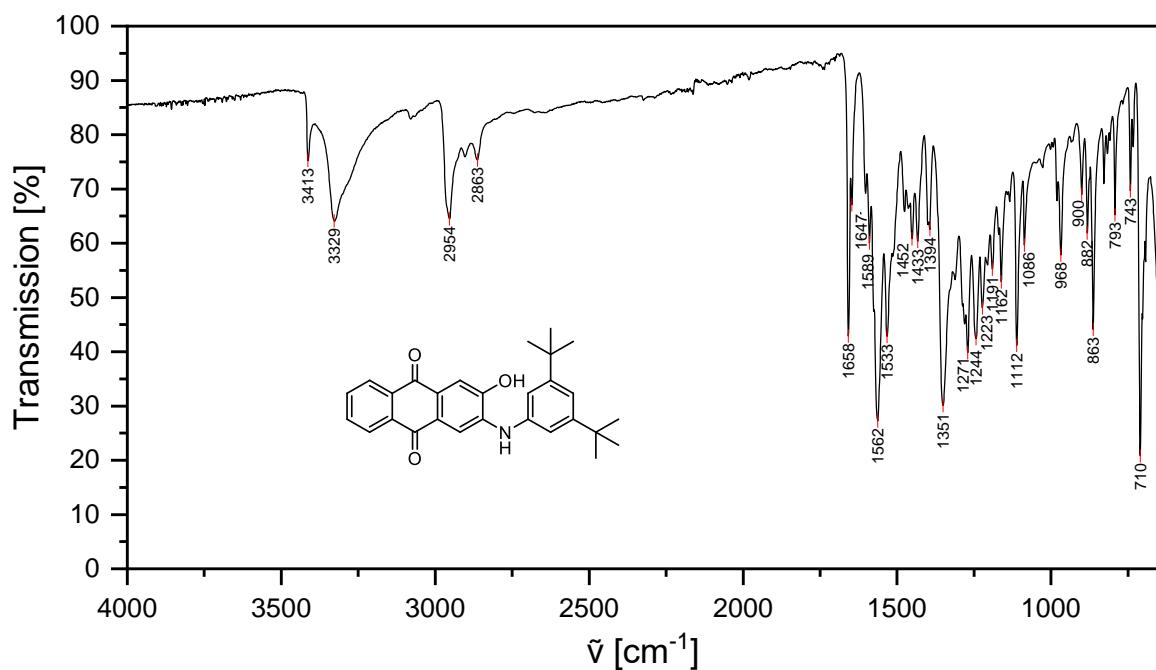


Figure S5.2. ATR-IR spectrum of L₁.

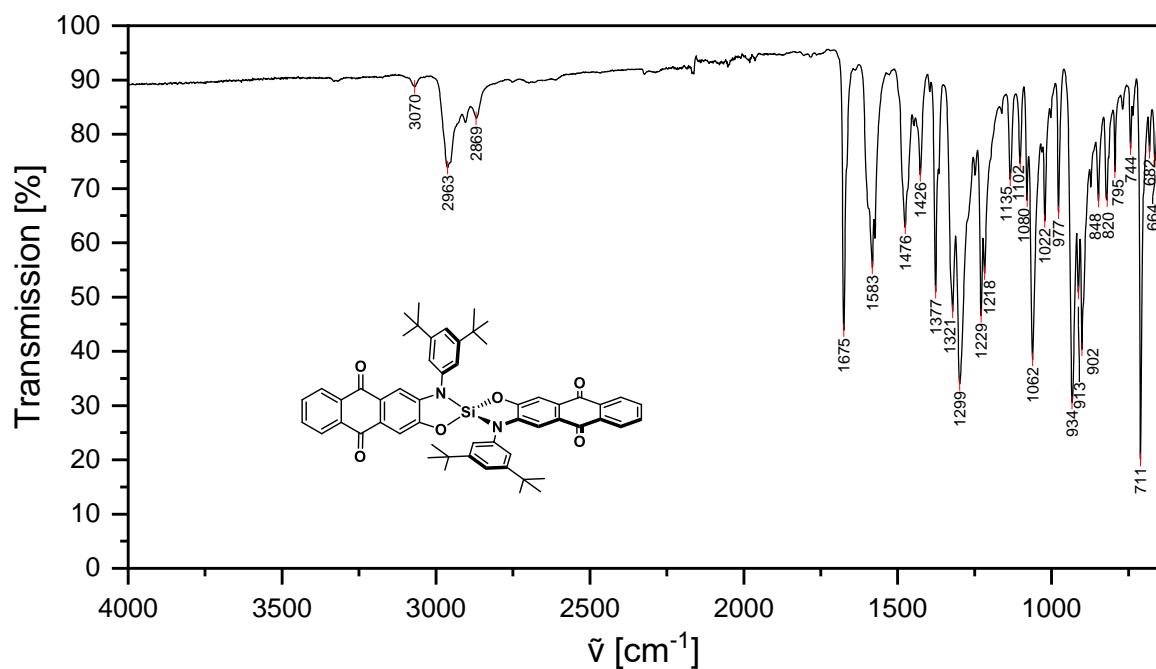


Figure S5.3. ATR-IR spectrum of **1**.

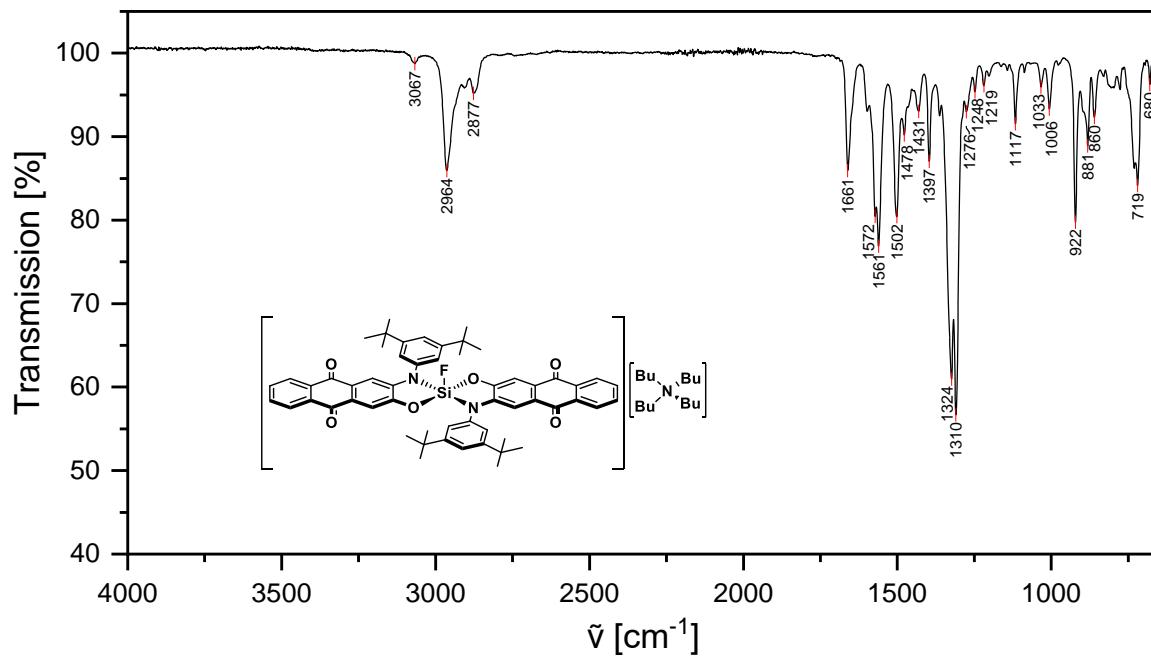


Figure S5.4. ATR-IR spectrum of **[1-F][NBu₄]**.

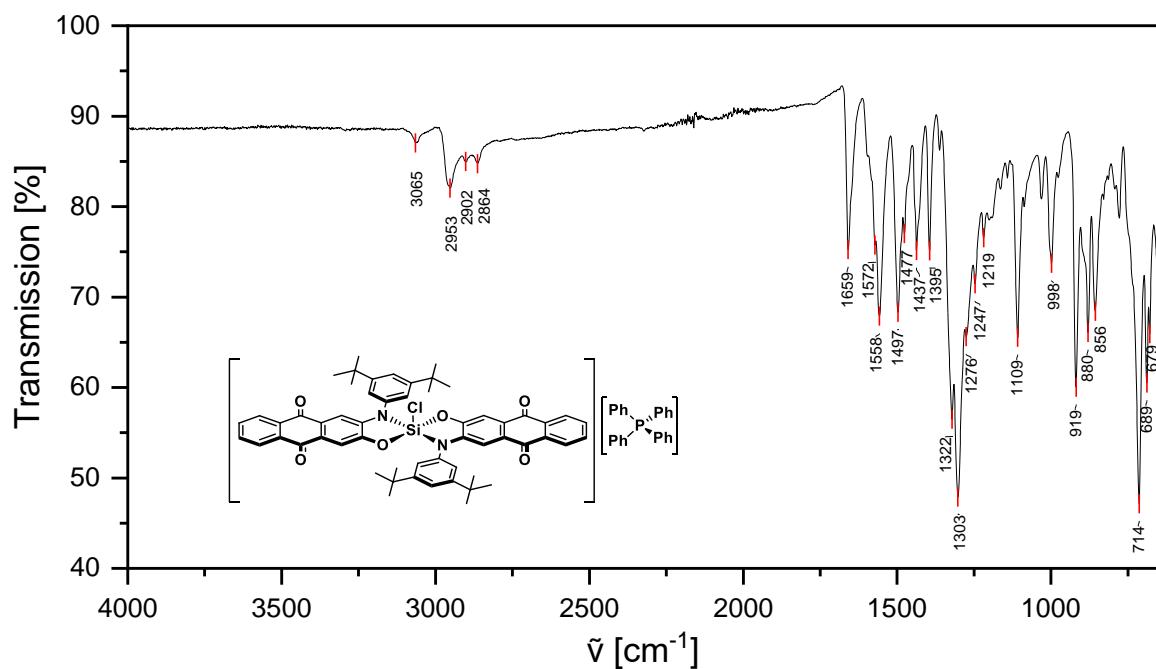


Figure S5.5. ATR-IR spectrum of **[1-Cl][PPh₄]**.

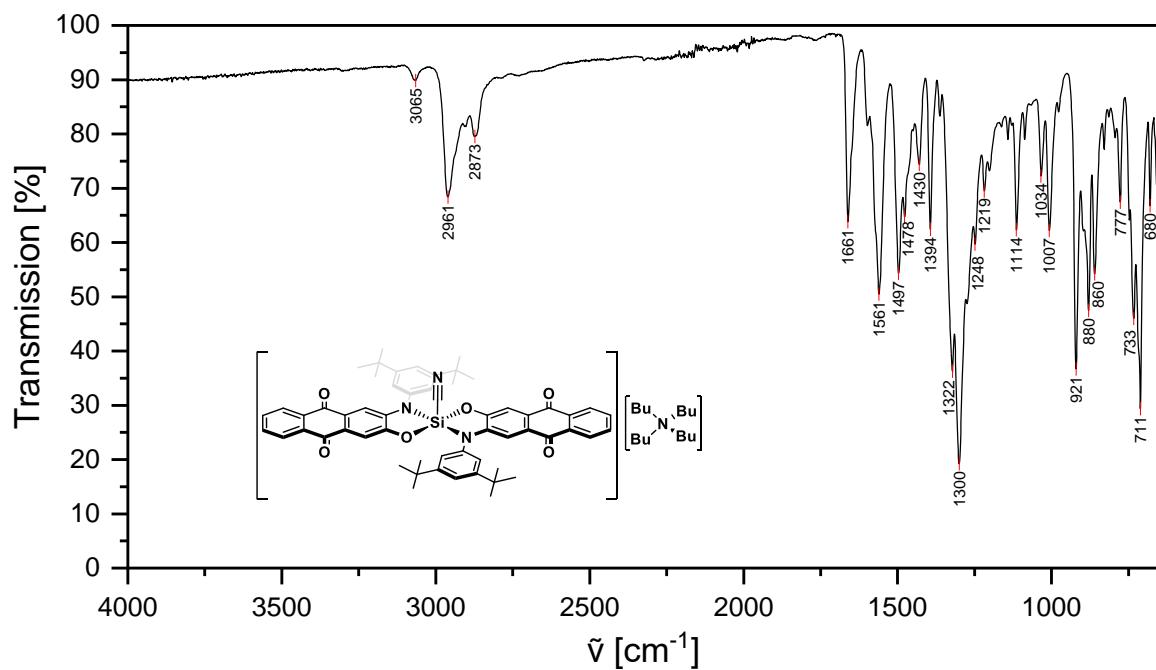


Figure S5.6. ATR-IR spectrum of **[1-CN][NBu₄]**.

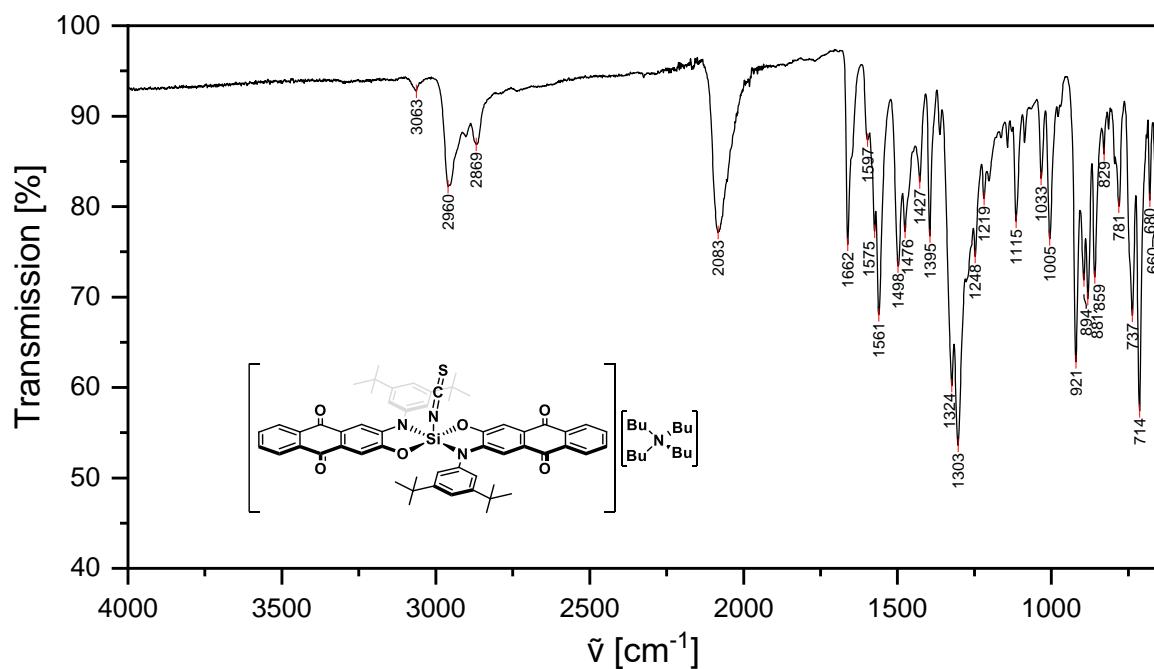


Figure S5.7. ATR-IR spectrum of **[1-NCS][NBu₄]**.

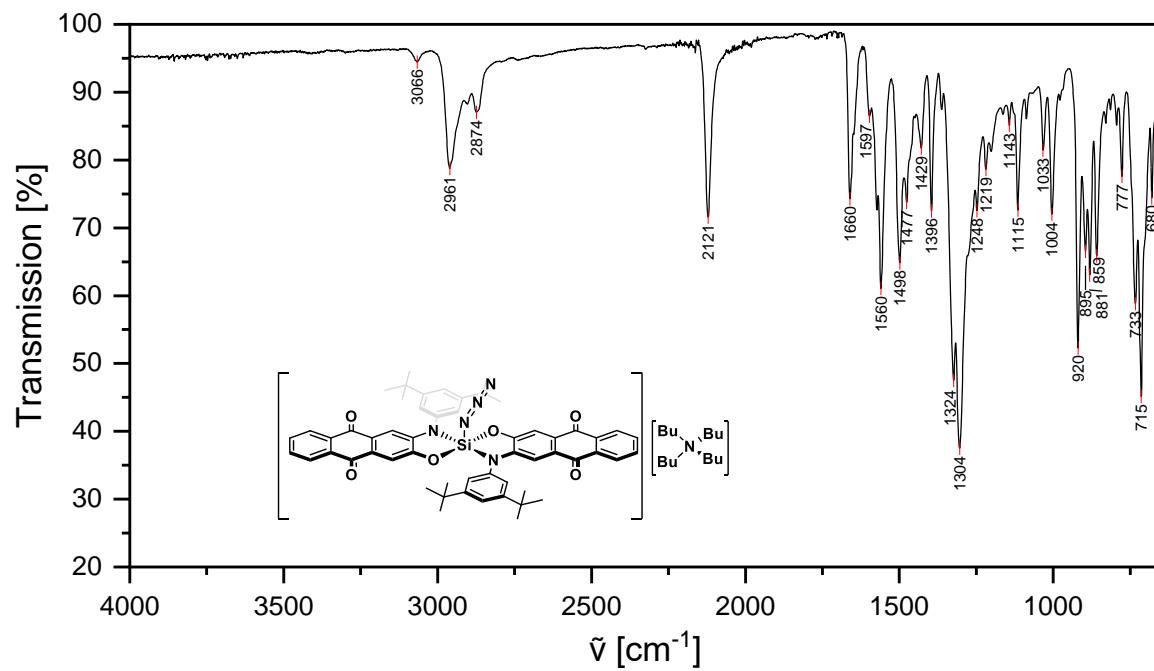


Figure S5.8. ATR-IR spectrum of **[1-N₃][NBu₄]**.

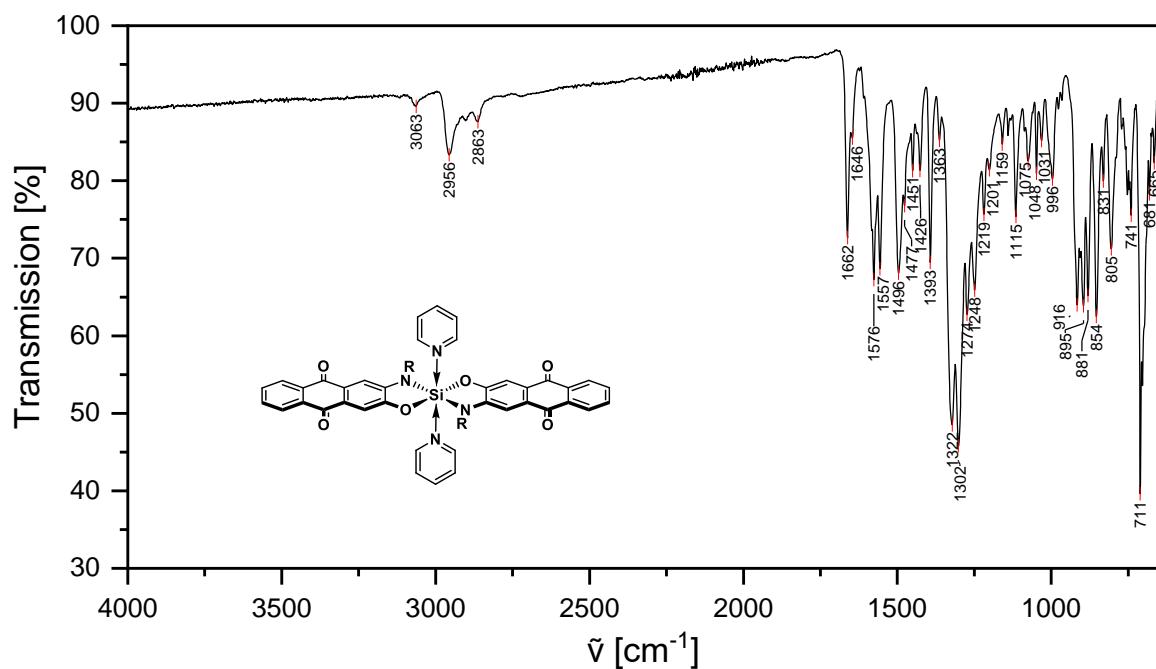


Figure S5.9. ATR-IR spectrum of **1-(pyridine)₂**.

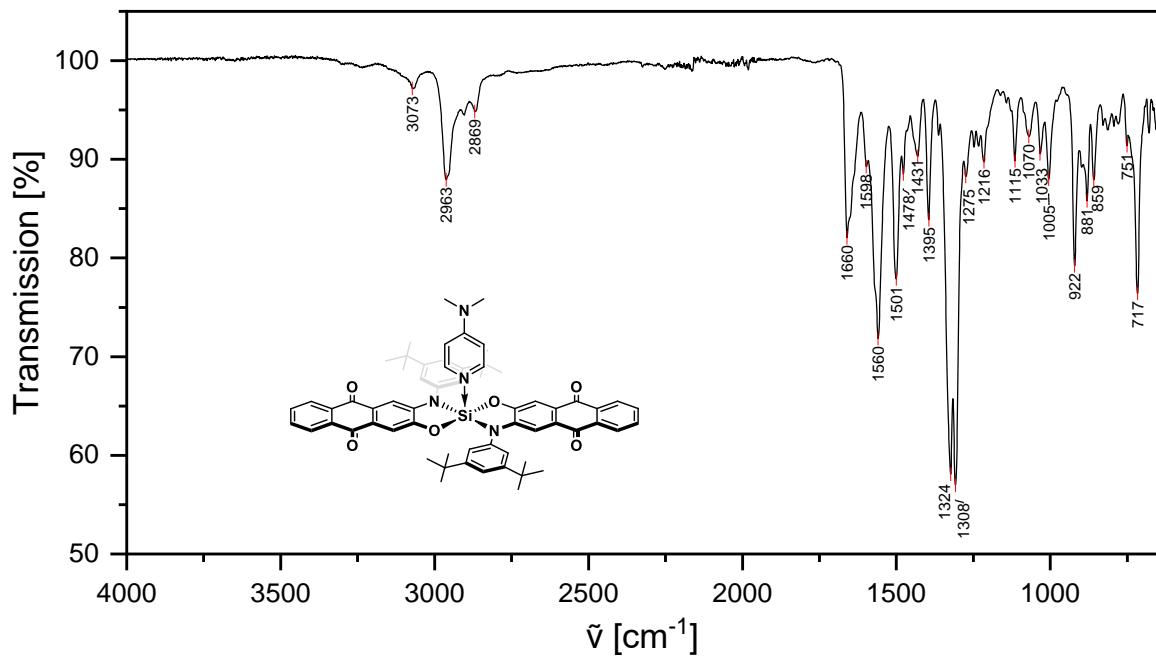


Figure S5.10. ATR-IR spectrum of **1-DMAP**.

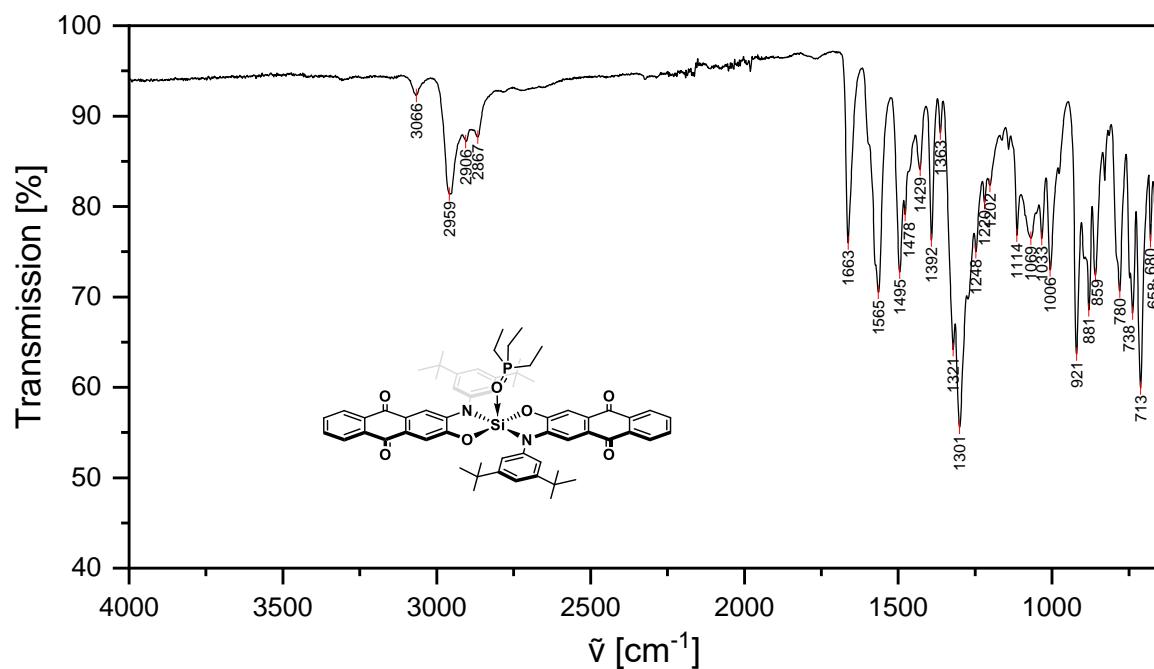


Figure S5.11. ATR-IR spectrum of **1-OPEt₃**.

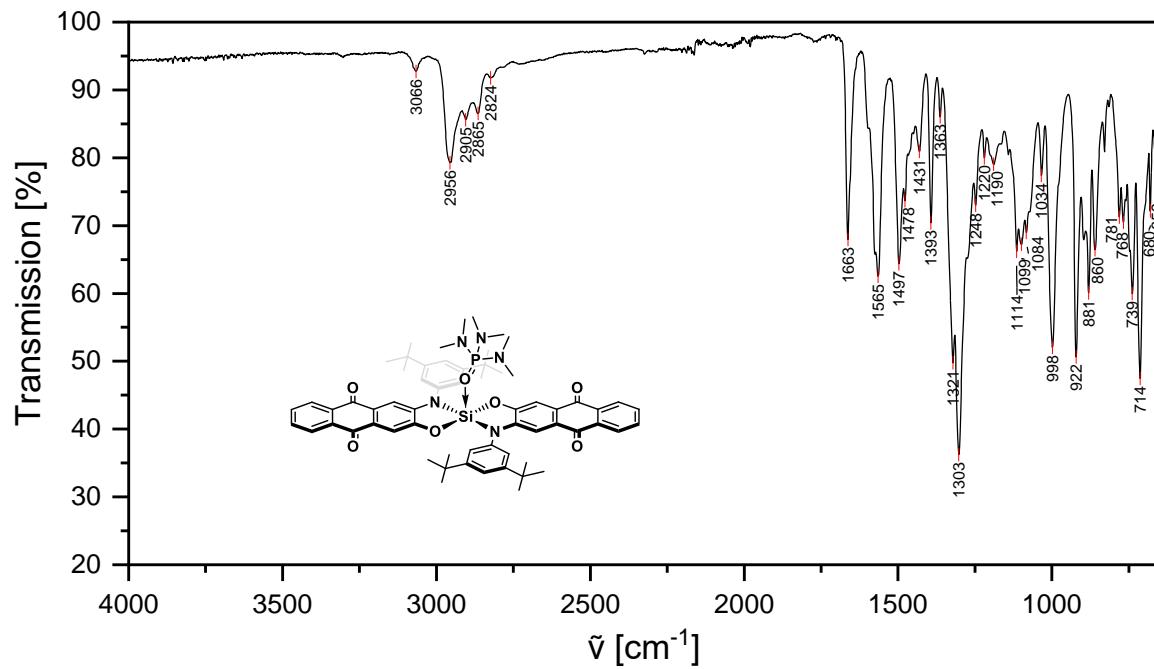


Figure S5.12. ATR-IR spectrum of **1-HMPA**.

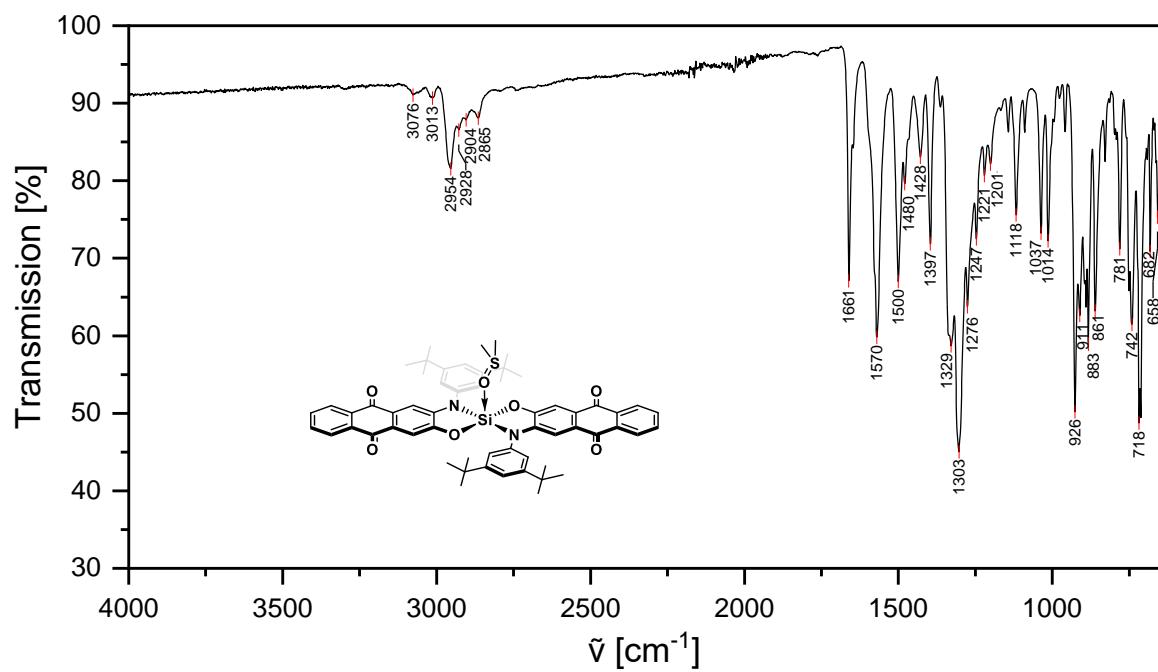


Figure S5.13. ATR-IR spectrum of **1-DMSO**.

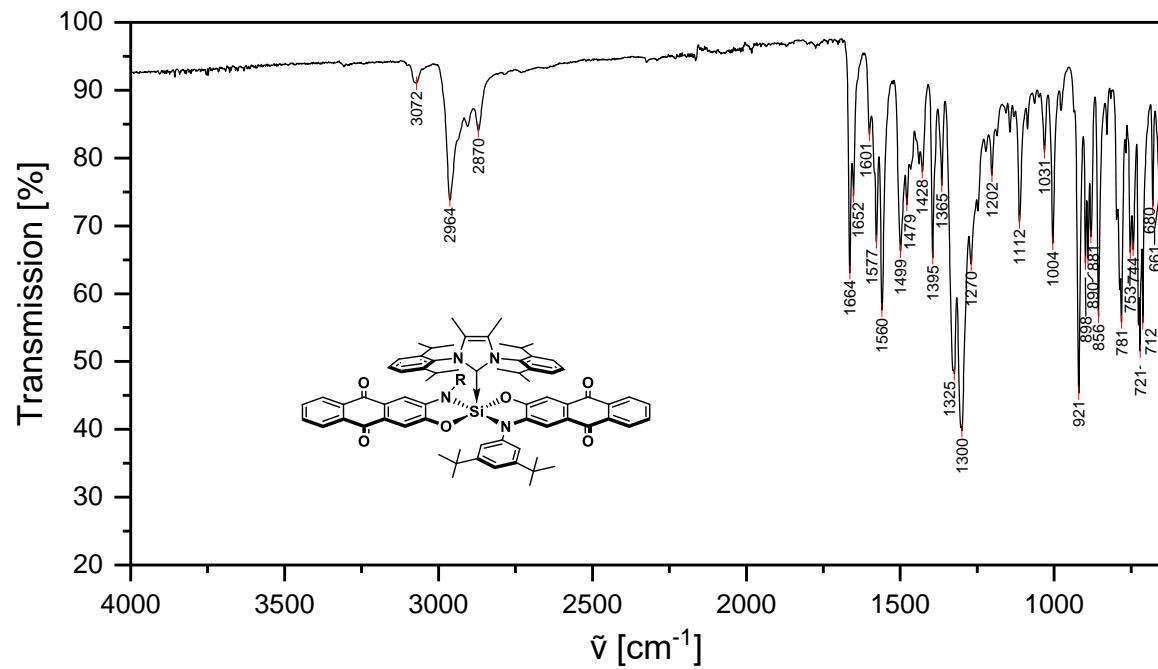


Figure S5.14. ATR-IR spectrum of **1-dippNHC**.

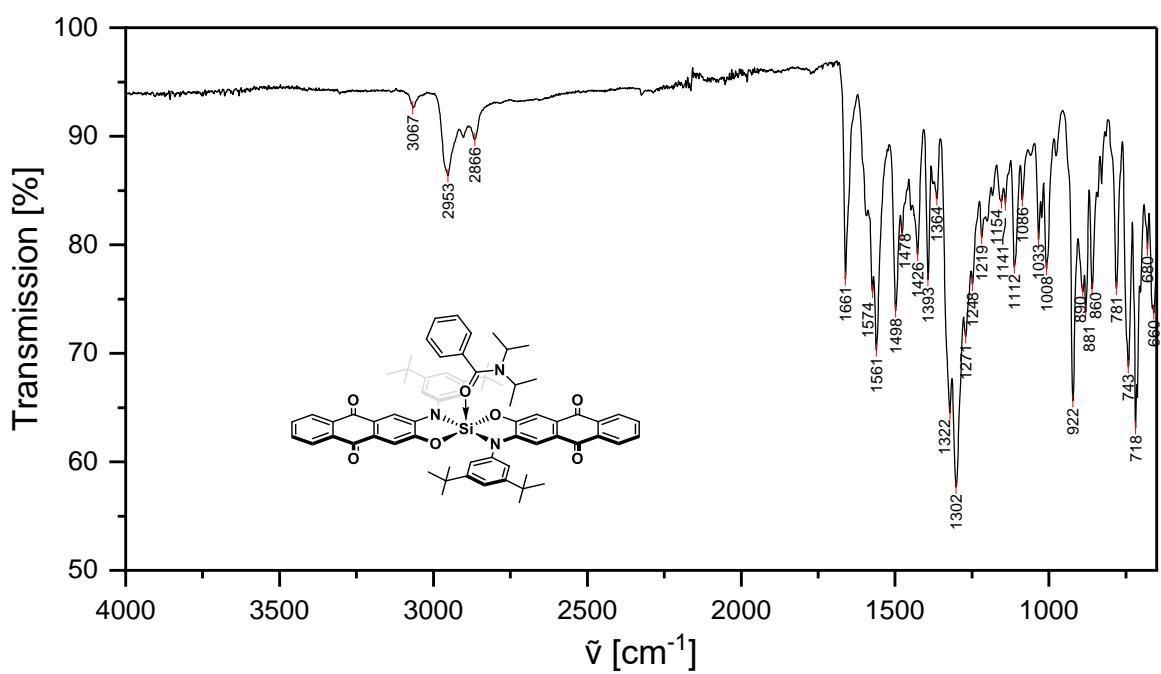
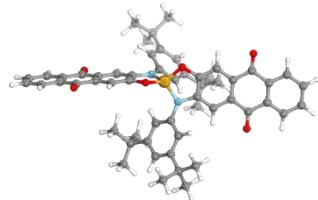


Figure S5.15. ATR-IR spectrum of **1**-DIBA.

6 Coordinates

1



119

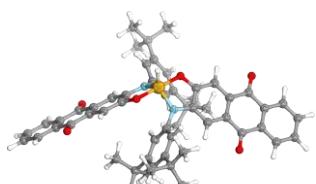
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 C 0.747531 1.562284 0.550664
 C -0.572498 1.637324 0.998582
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 H -2.549004 2.375731 0.602926
 H 0.452885 3.331241 -2.332833
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[1-H]

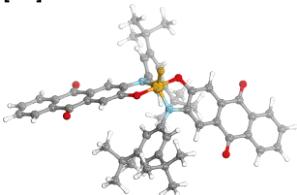


120

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[1-F]

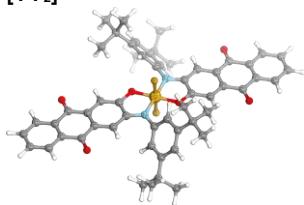


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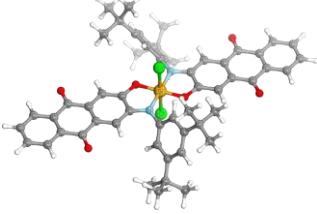
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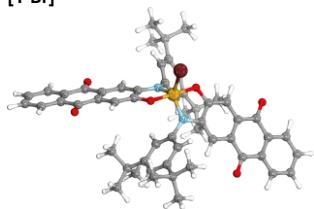
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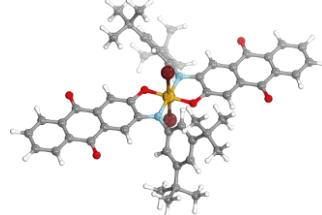
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C 6.628312 6.068976 -1.289982
C 7.607331 15.083391 -0.366216
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C 7.478727 14.011661 -1.240230
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H -3.093477 1.245748 1.462977
H -0.168591 4.341314 0.782885
H -1.250593 -0.278457 2.211766
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C 3.032382 2.067981 1.959599
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C 5.657174 -2.651652 2.729248
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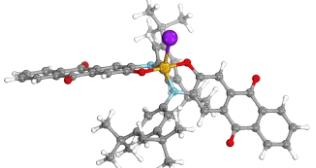
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 H 7.641263 -5.857155 6.203145
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 C 6.255485 -6.080605 1.856113
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 H 5.912121 -6.862301 1.165755
 H 6.605964 -6.566151 2.772610
 C 4.636861 -4.505906 0.807104
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 H 4.278774 -5.320387 0.166040
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 H 4.234295 -6.300433 3.739145
 H 3.556294 -6.603132 2.127223
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 C 11.029250 -1.128762 4.012574
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 O 8.739169 -0.576883 3.613776
 O 14.368778 2.214197 3.085992
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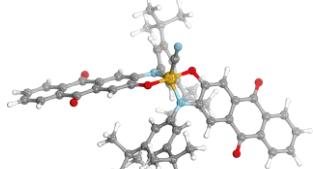


120

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 C 2.517017 -0.564038 0.635641
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 C 3.919238 1.666916 -0.569856
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 C 6.508577 4.557316 -3.149662
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 C 7.967971 6.236352 -1.433807
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 C 4.925198 11.963653 0.550573
 C 4.801584 10.824771 -0.369614
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 C 8.130355 8.576566 -0.545130
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 H 4.367132 6.285876 4.438211
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 H 10.383952 7.645012 -1.820655
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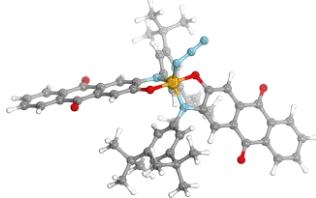
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 C 1.050680 1.991226 2.108303
 C -0.170067 1.975433 2.783920
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C	5.766911	1.055277	0.313108	H	16.717098	-2.646437	6.512432	C	6.045940	-6.127475	2.038385
C	5.514562	0.041927	1.281788	H	7.684055	6.540567	3.758653	H	6.878469	-5.633717	1.526036
C	4.310094	0.020207	1.968028	H	9.041955	5.816303	-1.592496	H	5.653169	-6.917912	1.385441
H	4.102389	-0.726237	2.727563	H	11.940725	5.764250	1.621589	H	6.442626	-6.599352	2.943080
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O	2.831646	3.954351	-0.366458	H	14.859058	-3.636624	5.161772	H	5.167970	-4.061708	0.462407
O	1.810046	0.057534	3.231210	H	16.938659	-0.179020	6.673778	H	3.582145	-3.830701	1.226142
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C	7.342666	-2.362681	4.711086	H	12.129604	5.613112	-1.462111	H	7.536790	7.754983	1.333314
C	7.324610	-1.455959	3.645893	H	12.238059	4.210624	-0.388003	H	5.936946	5.881482	1.973018
H	7.873968	-0.519351	3.699258	H	5.959692	6.401223	4.132834	H	9.442997	6.317760	-1.414982
H	6.617707	-4.260962	5.385199	H	11.926112	7.114664	0.464247	H	11.478930	6.904830	0.049171
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H	10.037505	-2.720984	5.170626	C	8.220632	-0.963245	-1.324550	C	9.316700	3.046561	2.349057
H	10.193915	-1.566466	6.513641	N	8.352735	-1.371866	-2.400724	C	10.364778	1.031300	3.161061
H	9.752991	-1.000896	4.894757	[1-CN ₂]				C	11.703763	1.429336	3.101409
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H	7.006384	-3.278944	7.365762					C	12.411722	-0.771704	3.898585
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H	5.557643	-7.296633	3.427592					C	8.598966	5.359418	2.402715
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C	4.373488	-5.329516	2.005657					C	8.599362	4.075313	2.949838
H	5.064761	-5.332935	1.156149					C	10.910380	4.894308	-0.619881
H	3.639123	-4.531492	1.851954					C	6.264117	6.000062	3.011611
H	3.835536	-6.283955	2.002720					C	7.867553	7.800798	2.376992
C	4.075517	-5.211849	4.474078					C	11.953767	5.981233	-0.297465
H	4.550529	-5.156522	5.458513					C	11.654633	3.666365	-1.158170
H	3.503692	-6.146675	4.426922					C	9.962742	5.404522	-1.722079
H	3.375987	-4.373151	4.390198					C	7.740382	6.438076	3.068538
H	11.049756	4.232914	-1.709455					C	9.362531	5.596253	1.260898
H	6.477984	6.552615	2.449098					C	10.102672	4.576349	0.642479
H	5.169546	4.417621	1.894604					N	9.248617	1.750394	2.888559
H	8.897408	6.973431	-0.261892					O	13.213221	-2.849345	4.717346
H	10.619511	6.942382	1.648914					O	8.769567	-0.588469	3.694417
H	7.126939	2.942283	4.759319					O	14.423289	2.114814	2.930267
C	13.239860	0.559710	3.971748					H	7.560145	7.377123	5.025680
C	11.008066	-2.034461	2.332620					H	18.017951	-2.402514	4.780927
C	10.092921	-1.206786	1.7114475					H	8.899658	8.170413	2.398805
C	9.013773	2.245526	1.289452					H	9.204046	4.648902	-1.951942
C	10.208327	0.209583	1.812289					H	12.633302	5.635741	0.489264
C	11.228710	0.774499	2.561776					H	6.110099	5.041481	3.515539
C	12.156713	-0.062612	3.197797					H	15.618940	-3.115059	4.910647
C	14.197012	-0.346178	4.673885					H	18.576641	-0.113603	3.992386
C	15.231439	0.208603	5.428297					H	9.380531	6.590374	0.827429
C	14.071782	-1.739631	4.582693					H	10.820163	-2.167681	4.322518
C	14.981930	-2.561396	5.248444					H	8.025036	3.840074	3.839565
C	9.820347	3.129728	0.574358					H	10.574541	2.466107	0.726480
C	12.053010	-1.467437	3.077524					H	16.731396	1.448418	3.332026
C	12.971526	-2.363383	3.788713					H	11.976155	2.430249	2.785578
C	16.134568	-0.615439	6.087214					H	12.222407	3.946968	-2.053322
C	16.009719	-2.002337	5.996836					H	12.361298	3.269321	-0.421251
C	7.869458	4.093517	2.352588					H	7.236334	8.532508	2.894725
C	7.130295	4.036433	4.741456					H	12.547222	6.217455	-1.190446
C	8.055150	2.718880	2.171678					H	5.622847	6.748912	3.493556
C	10.523840	5.519745	-0.019454					H	10.525719	5.624467	-2.638595
C	5.422518	4.057462	2.897254					H	9.222183	6.896335	4.603236
C	6.736010	6.109364	3.417241					Si	7.684260	0.706730	3.115487
C	11.296738	6.387718	0.992014					C	7.882933	0.034736	1.271587
C	11.541895	4.845544	-0.946576					C	7.485011	1.379206	4.959150
C	9.612238	6.418057	-0.876896					N	7.999260	-0.359154	0.184608
C	6.802726	4.579784	3.337057					N	7.366367	1.778144	6.044546
C	8.683953	4.960777	1.628721								
C	9.666457	4.502529	0.738057								
N	9.186249	0.838368	1.133984								
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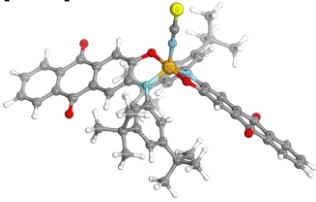
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C	-1.115674	2.939121	2.484967	C	14.210539	-0.355257	4.664633	O	2.245801	4.172794	1.482416
C	0.364384	3.969676	0.883038	C	15.247085	0.198113	5.417075	O	1.123407	-1.014140	2.552547
C	1.326991	2.992437	1.138367	C	14.078974	-1.748476	4.579546	O	6.657721	2.009418	2.819723
C	1.062650	1.984094	2.075917	C	14.985070	-2.571285	5.249500	N	6.208403	-0.406874	3.370624
C	-0.161429	1.964107	2.745342	C	9.829303	3.128970	0.580507	C	6.162182	-1.781818	3.654004
H	-1.599723	4.705462	1.348789	C	12.060951	-1.474190	3.074611	C	5.622601	-2.685612	2.727948
H	-2.067725	2.919668	3.008720	C	12.976394	-2.370744	3.787105	C	5.584682	-4.045933	3.003974
H	0.597851	4.739444	0.153862	C	16.146078	-0.627007	6.080512	C	6.126436	-4.499803	4.216659
H	-0.338183	1.171318	3.465772	C	16.014916	-2.013651	5.996291	C	6.686672	-3.620885	5.142282
C	2.630040	3.044584	0.410359	C	7.874757	4.080695	2.360280	C	6.680612	-2.254925	4.855199
C	3.602387	1.980510	0.679129	C	7.132023	4.008977	4.747738	H	7.098887	-1.528255	5.542727
C	3.352373	0.984584	1.651322	C	8.063071	2.707384	2.173131	H	6.112969	-5.563294	4.429367
C	2.071808	0.923179	2.369376	C	10.529897	5.523152	-0.000779	H	5.257985	-2.287850	1.788176
C	4.833505	2.012609	0.006785	C	5.426992	4.040134	2.901242	C	4.983100	-5.059184	2.024678
C	5.788277	1.060114	0.299290	C	6.738529	6.089568	3.435491	C	7.324980	-4.090544	6.452536
C	5.532264	0.040378	1.260679	C	11.301204	6.385903	1.016345	C	8.810792	-3.683029	6.454524
C	4.323035	0.015055	1.940188	C	11.545957	4.855316	-0.930789	H	9.339979	-4.151920	5.618000
H	4.111259	-0.735710	2.694281	C	9.618649	6.425740	-0.854091	H	9.294603	-3.995436	7.388587
H	5.035963	2.797928	-0.713366	C	6.806248	4.560503	3.346107	H	8.930276	-2.600675	6.351493
O	2.856244	3.964461	-0.376020	C	8.688277	4.953249	1.641467	C	6.610053	-3.422335	7.642031
O	1.815360	0.042098	3.188747	C	9.672675	4.500810	0.750112	H	6.684721	-2.332438	7.592555
O	7.002926	1.008998	-0.234161	C	9.198712	0.833572	1.128899	H	7.059031	-3.750667	8.588276
N	6.620585	-0.790372	1.404759	C	12.859447	-3.596209	3.757600	H	5.546280	-3.686366	7.650097
C	6.629302	-1.734993	2.472516	C	9.044508	-1.622478	1.022121	C	7.240570	-5.610485	6.637323
C	5.910911	-2.924580	2.357773	C	13.399396	1.771553	4.036210	H	7.765501	-6.142657	5.836104
C	5.896739	-3.836258	3.408326	C	6.384592	4.353380	5.472504	H	6.200948	-5.958032	6.662448
C	6.624034	-3.527443	4.567596	C	16.718835	-2.658723	6.515467	H	7.712005	-5.888969	7.587093
C	7.347100	-2.344577	4.701925	C	7.686050	6.519244	3.780320	C	6.061303	-6.076880	1.606668
C	7.333274	-1.444792	3.631074	C	9.049344	5.827705	-1.573595	H	6.904632	-5.565124	1.130706
H	7.883630	-0.508577	3.679995	C	11.944987	5.759253	1.642968	H	5.646772	-6.802952	0.895106
H	6.618353	-4.238066	5.386183	C	5.395064	2.946895	2.868662	H	6.449274	-6.629274	2.468465
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C	5.120956	-5.154344	3.339710	C	16.951652	-0.191572	6.665801	H	5.226577	-3.870763	0.204201
C	8.132525	-1.988712	5.966976	C	8.560876	6.021775	1.772275	H	3.648111	-3.667685	0.991017
C	9.617720	-1.792770	5.610776	C	10.909596	-3.117590	2.278392	H	4.010022	-5.154247	0.096374
H	10.039608	-2.704537	5.175167	C	7.452275	1.981671	2.704231	C	3.818813	-5.799385	2.710804
H	10.192533	-1.540188	6.510167	C	10.565636	2.715773	-0.099825	H	4.155113	-6.338399	3.602360
H	9.759242	-0.986356	4.885039	C	15.324099	1.280048	5.464643	H	3.369236	-6.526409	2.021803
C	7.572498	-0.678811	6.554459	C	11.345031	1.844371	2.656957	H	3.044511	-5.088276	3.018566
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H	8.113512	-0.414102	7.470856	C	12.245815	4.217940	-0.375126	H	7.571731	7.615845	1.045217
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C	8.033763	-3.077992	7.041344	C	11.930461	7.116607	0.493549	H	8.942026	5.570199	-1.639435
H	8.445311	-4.030146	6.688628	C	4.651747	4.378358	3.599600	H	11.171014	6.422843	-0.657974
H	6.998296	-3.240926	7.361638	C	10.220157	7.156587	-1.408387	H	8.358363	6.062868	5.012994
H	8.611154	-2.772927	7.920852	C	8.120482	4.346426	5.078706	C	14.123042	1.149764	3.844028
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H	6.859360	-6.316555	2.718986	C	8.287016	-0.975513	-1.240451	C	10.098160	-0.262977	3.896577
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H	6.615084	-6.296443	4.471096	C	7.356903	-0.256011	-3.272764	C	10.384514	1.097925	3.538436
C	4.385510	-5.326945	2.005521	C	11.713942	1.530687	3.525784	C	11.2744534	0.654445	3.877875
H	5.081116	-5.337838	1.159560	C	15.209663	0.198474	4.237984	C	16.534368	0.635540	4.232664
C	3.653426	-4.528744	1.842232	C	14.911731	-1.121679	4.603238	C	14.917344	-1.987292	4.959850
H	3.845890	-6.280492	2.007239	C	9.970243	3.089161	1.326899	C	9.970243	3.089161	1.326899
C	4.075815	-5.189133	4.471225	C	12.453717	-0.688917	4.240013	C	12.453717	-0.688917	4.240013
H	4.546362	-5.126867	5.457374	C	13.500486	-1.619548	4.626413	C	13.500486	-1.619548	4.626413
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H	3.378764	-4.349431	4.377397	C	8.673397	5.351162	2.346266	C	8.471442	6.896187	4.313602
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H	10.622679	6.935874	1.675830	C	12.229973	3.007939	-1.235675	C	12.229973	3.007939	-1.235675
H	7.128703	2.914754	4.759029	C	9.427802	4.609848	-1.841644	C	9.427802	4.609848	-1.841644
C	13.257669	0.551902	3.957662	C	7.908317	6.544827	2.923364	C	7.908317	6.544827	2.923364
C	11.012348	-2.039291	2.333126	C	9.302308	5.381509	1.103673	C	9.302308	5.381509	1.103673
C	10.099627	-1.211168	1.710956	C	9.952529	4.255171	0.575076	C	9.952529	4.255171	0.575076
C	9.023552	2.239094	1.289831	C	9.264247	1.807239	3.268325	C	9.264247	1.807239	3.268325
C	10.220732	0.206066	1.804353	C	13.281760	-2.789189	4.970905	C	13.281760	-2.789189	4.970905
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H	8.665609	3.825487	-1.784292	C	18.066816	-2.226518	5.231375	C	18.066816	-2.226518	5.231375
H	9.048914	8.110764	1.904270	C	9.048914	8.110764	1.904270	C	9.048914	8.110764	1.904270
H	8.665609	3.825487	-1.784292	C	12.413248	5.256798	-0.174262	C	12.413248	5.256798	-0.174262

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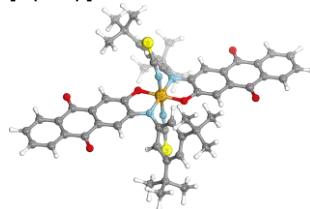
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[1-(NCS)₂]



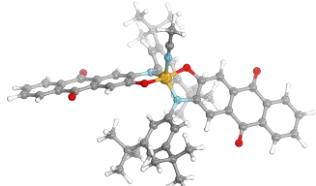
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1-MeCN



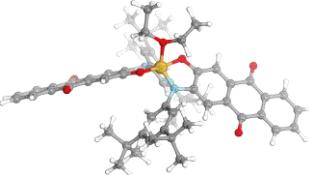
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1-Et₂O



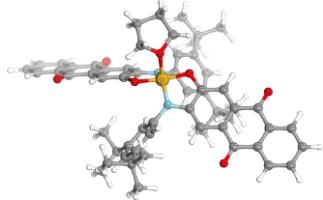
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H	-0.502578	8.568460	2.884923					H	10.814272	1.510501	-4.988615
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H	-2.155706	9.693661	-0.237671					H	12.068441	6.712396	-2.668300
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H	3.117983	2.840553	2.586973					H	10.106499	8.219036	-2.041063
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O	2.952913	5.734406	-4.824739					H	5.349548	5.216292	3.810078
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1-THF



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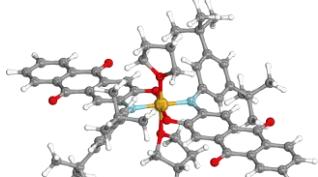
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1-(THF)₂



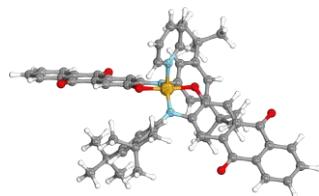
145

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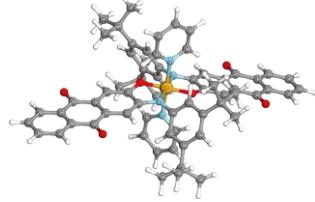
1-pyridine



130

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1-(pyridine)₂:trans



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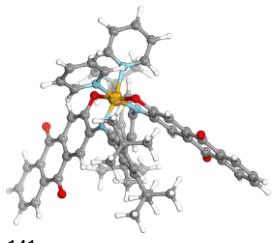
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1-(pyridine)₂,cis

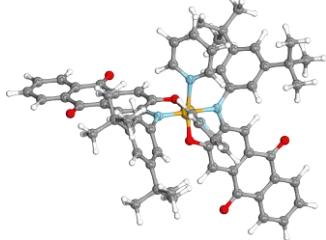


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1-(pyridine)₂cis 2

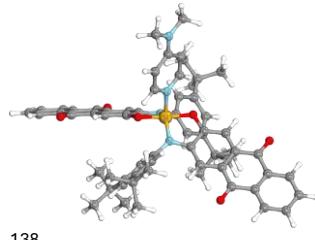


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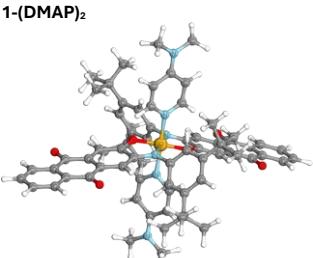
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C -6.393278 10.693854 0.653315
C -8.733333 9.479464 1.513821
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H -11.999906 17.224665 -0.301668
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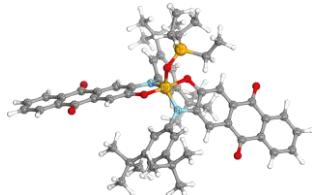
1-DMAP



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C -0.538249 1.703746 2.631861
H -1.999237 4.415812 1.203266
H -2.500146 2.564679 2.777794
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C 5.340813 0.107624 1.552135
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H 3.853403 -0.783017 2.846585
H 4.858405 2.894262 -0.382037
O 2.599286 3.906831 -0.231328
O 1.494297 -0.146921 3.157942
O 6.880502 1.190719 0.208834
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C 5.686784 -2.831998 2.590799
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C 7.237549 -2.494203 4.910060
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H 6.452368 -4.402465 5.478194
H 5.104024 -2.923917 1.680387
C 4.819161 -5.086032 3.443598
C 8.091753 -2.274649 6.161107
C 9.563832 -2.078458 5.750919
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H 10.185715 -1.929614 6.641303
H 9.694974 -1.206644 5.102752
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H 8.187381 -0.844954 7.799254
H 6.542836 -1.120577 7.182171
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H 4.692959 -5.118998 1.258160
H 3.335491 -4.282947 2.048157
H 3.438109 -6.047663 2.089069
C 3.813031 -5.138684 4.609481
H 4.319393 -5.168880 5.579013
H 3.186805 -6.034880 4.528383
H 3.160664 -4.259059 4.596563
H 11.319106 4.232937 -1.412724
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H 10.871205 7.026204 1.874570
H 7.008637 3.416956 4.887083
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C 11.002862 -1.885500 2.474930
C 10.056199 -1.032138 1.953839
C 9.022813 2.446011 1.545704

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C	14.494605	-0.271988	4.397140			C	6.210315	6.124631	4.002132
C	15.624331	0.259100	5.020836			C	7.825318	7.878895	3.274540
C	14.317072	-1.663382	4.344469			C	11.494905	6.012366	0.076954
C	15.271004	-2.505299	4.917501			C	10.972036	3.768676	-0.865144
C	9.908695	3.260937	0.840562			C	9.331336	5.622260	-1.146006
C	12.141669	-1.341572	3.085770			C	7.705113	6.489158	3.909868
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C	16.391843	-1.968834	5.536832			N	8.935235	1.742692	3.330945
C	7.903057	4.386926	2.454854			O	12.726506	-3.249154	4.188745
C	7.038955	4.507300	4.797981			O	8.405029	-0.569286	4.087895
C	8.039986	2.998577	2.350870			O	14.068879	1.729224	2.578249
C	10.780663	5.587736	0.220157			H	7.801372	7.332601	5.914339
C	5.429607	4.449802	2.867451			H	17.522447	-3.169038	3.535183
C	6.774162	6.497310	3.321343			H	8.870308	8.198744	3.195281
C	11.548303	6.417889	1.267185			H	8.508819	4.923159	-1.334663
C	11.804912	4.829345	-0.632163			H	12.235718	5.590871	0.764662
C	9.988081	6.531481	-0.703805			H	6.061949	5.151132	4.480989
C	6.800602	4.964764	3.345883			H	15.128374	-3.716265	3.998299
C	8.800378	5.184159	1.747899			H	18.143812	-0.877454	2.813645
C	9.814938	4.645424	0.942519			H	9.081486	6.682645	1.503208
N	9.132446	1.025753	1.445257			H	10.395491	-2.361893	4.277253
O	12.966495	-3.480521	3.661943			H	7.937697	3.852505	4.515363
O	8.902713	-1.422213	1.393068			H	10.076118	2.532941	1.074972
O	13.673844	1.868782	3.821536			H	16.369121	0.862206	2.553841
H	6.263903	4.919042	5.454701			H	11.688041	2.233672	2.829149
H	17.132127	-2.627697	5.981910			H	11.448036	4.068721	-1.804865
H	7.710315	6.927210	3.694674			H	11.737833	3.296677	-0.239797
H	9.421301	5.960426	-1.447258			H	7.298308	8.611414	3.895221
H	12.110874	5.763556	1.941636			H	12.000839	6.267448	-0.861711
H	5.360348	3.359151	2.922169			H	5.669630	6.879379	4.585571
H	15.109399	-3.577418	4.864209			H	9.804452	5.866628	-2.104433
H	17.446062	-0.168098	6.074574			H	9.373142	6.788275	5.286460
H	8.719073	6.262168	1.823305			Si	7.380352	0.764064	3.610802
H	10.866825	-2.961284	2.447028			N	7.480944	0.120946	1.683363
H	7.371733	2.330225	2.888083			C	7.661298	-1.183147	1.387876
H	10.667159	2.786221	0.227164			C	7.367485	0.981378	0.650293
H	15.738195	1.338316	5.047551			C	7.429098	0.591455	-0.666232
H	11.479986	1.989525	2.709646			C	7.615610	-0.772345	-0.991682
H	12.469216	5.545920	-1.126618			C	7.730312	-1.662452	0.101583
H	12.427940	4.165679	-0.022768			H	7.748448	-1.855047	2.235322
H	5.965183	6.856209	3.966091			H	7.881559	-2.724781	-0.040040
H	12.255239	7.093100	0.770887			H	7.223387	2.024121	0.913500
H	4.631898	4.866930	3.492989			H	7.321896	1.351403	-1.429411
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H	8.015582	4.848769	5.157840			C	7.849652	-2.619539	-2.567148
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N	8.086932	-0.739863	-0.909423			H	7.863864	-2.765236	-3.647766
C	8.251723	0.111663	-1.948891			H	7.024203	-3.215705	-2.154756
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C	8.199091	-2.574180	-2.440136			H	6.589880	0.259790	-3.365227
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N	8.532390	-2.161899	-4.802365			C	6.128496	1.861048	6.040787
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H	8.558258	0.431065	-4.009762			H	8.641593	1.660233	10.436347
H	7.950580	-2.721644	-0.316964			H	5.030271	2.460223	9.983478
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C	8.502336	-3.596789	-5.068675			C	8.364000	1.379855	6.317935
H	9.638087	-0.639127	-5.785231			H	5.049172	2.667623	7.667692
H	7.869907	-0.540282	-6.000004			H	9.285309	1.009925	5.880302
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1-OPEt₃

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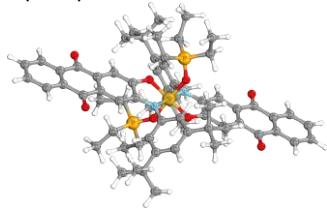
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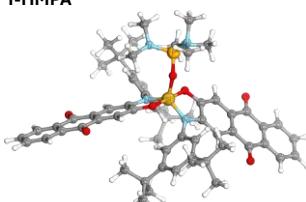
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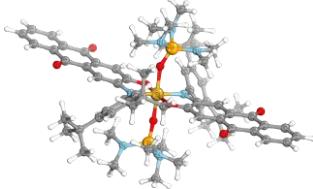
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H -0.847974 -0.415859 3.169139
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C 18.005896 -0.116415 3.027974
C 17.774687 -1.397231 3.530816

1-(OPEt₃)₂

C	8.659502	5.357999	2.201775
C	8.183701	6.523461	4.369101
C	8.735990	4.060439	2.711513
C	11.030129	5.119299	-0.801344
C	6.294735	5.850212	2.851441
C	7.787201	7.751356	2.241194
C	12.017301	6.238074	-0.416957
C	11.838669	3.948656	-1.374684
C	10.075691	5.629771	-1.897199
C	7.745029	6.368018	2.899978
C	9.414701	5.671131	1.073845
C	10.221233	4.718010	0.434998
N	9.567178	1.786566	2.638213
O	13.820103	-2.619482	4.177585
O	9.258040	-0.427037	3.738505
O	14.710005	2.337525	2.219640
H	7.541383	7.248231	4.883140
H	18.611950	-2.056684	3.741474
H	8.794246	8.182609	2.265829
H	9.352707	4.854304	-2.172363
H	12.703476	5.895011	0.364862
H	6.194430	4.874604	3.337962
H	16.269896	-2.823700	4.153112
H	19.022441	0.221043	2.846941
H	9.377471	6.676448	0.670458
H	11.429568	-1.943680	4.211277
H	8.148147	3.766149	3.573089
H	10.852992	2.660808	0.477201
H	17.089661	1.728132	2.364039
H	12.277461	2.578835	2.290843
H	12.400108	4.288891	-2.251290
H	12.561524	3.559232	-0.648645
H	7.118931	8.432700	2.778425
H	12.609890	6.537194	-1.289589
H	5.620530	6.553061	3.354660
H	10.642224	5.909525	-2.793193
H	9.220199	6.873092	4.430673
Si	8.110902	0.644218	2.919785
O	8.445042	-0.192147	1.264478
P	9.252964	-1.262473	0.509324
C	9.510790	-2.787166	1.477880
C	10.862365	-0.558773	-0.010329
C	8.256102	-1.666394	-0.968609
C	10.744354	-3.617639	1.116396
H	8.586018	-3.374869	1.398764
H	9.559780	-2.457757	2.520770
H	10.725051	-3.975739	0.083038
H	10.786135	-4.496570	1.766550
H	11.669272	-3.054912	1.275972
C	8.561551	-2.987798	-1.676145
H	8.333198	-0.812051	-1.653492
H	7.226093	-1.665599	-0.587837
H	9.565141	-3.010386	-2.107746
H	7.846374	-3.142474	-2.489680
H	8.466804	-3.835728	-0.992208
C	11.512751	-1.127846	-1.273300
H	11.541046	-0.629531	0.848864
H	10.646503	0.509268	-0.143115
H	11.758646	-2.187938	-1.173991
H	12.445909	-0.590666	-1.467619
H	10.872446	-1.001513	-2.151248
H	7.212762	-0.496521	6.333492
H	7.886787	-1.092177	8.668025
H	7.053000	0.460581	8.658019
C	7.958772	-0.055203	8.326401
C	8.091781	-0.032843	6.801559
H	6.142909	2.226373	7.035603
H	8.963673	-0.606914	6.462291
O	7.684755	1.533424	4.546395
C	7.073568	2.773368	6.835181
H	8.816563	0.398536	8.828905
P	8.197229	1.603000	5.992587
C	7.593345	3.477086	8.090376
H	7.834751	2.776614	8.893982
H	6.827801	4.161577	8.467180
H	10.821383	0.872081	7.504205
C	9.909579	2.233481	6.066945
H	8.485780	4.071890	7.874711
C	10.702766	1.947855	7.344866
H	10.419866	1.780573	5.208834
H	10.243719	2.381787	8.236942
H	9.857604	3.309642	5.851179
H	11.705775	2.372801	7.246528

H 6.828915 3.508968 6.057254
1-HMPA

148 xyz, charge: 0, multiplicity: 1
C -1.446351 3.098517 1.717122
C -1.662012 2.072221 2.637704
C 0.213650 3.224999 1.090767
C 0.814350 2.326762 1.379826
C 0.597627 1.294681 2.306516
C -0.644482 1.174053 2.930606
H -2.243976 3.800133 1.489664
H -2.627133 1.975491 3.126894
H -0.020254 4.014418 0.371189
H -0.786585 0.367229 3.642808
C 2.128555 2.481695 0.696687
C 3.184401 1.511532 1.024561
C 2.981322 0.490084 1.974374
C 1.673112 0.317159 2.637284
C 4.433533 1.651152 0.401640
C 5.444808 0.775206 0.727499
C 5.243833 -0.261331 1.674633
C 4.015080 -0.396741 2.302174
H 3.835191 -1.167819 3.043834
H 4.591248 2.466911 -0.296239
O 2.313125 3.393571 -0.107811
O 1.465938 -0.593295 3.433295
O 6.687853 0.820166 0.230770
N 6.401665 -0.995049 1.858097
C 6.453838 -1.974013 2.899039
C 5.759397 -3.174204 2.755071
C 5.792632 -4.125054 3.770213
C 6.551623 -3.845543 4.916395
C 7.253116 -2.653055 5.077638
C 7.181732 -1.708781 4.047547
H 7.703998 -0.758433 4.124422
H 6.587064 -4.587742 5.705275
H 5.199741 -3.337227 1.840432
C 5.033439 -4.507237 3.677812
C 8.068532 -2.332269 6.333039
C 9.529248 -2.040943 5.939978
H 9.980265 -2.902746 5.436693
H 10.121777 -1.816573 6.834669
H 9.607527 -1.183917 5.264058
C 7.465810 -1.089241 7.015856
H 7.496745 -0.214434 6.359061
H 8.029303 -0.847272 7.924515
H 6.420916 -1.266476 7.292845
C 8.064102 -3.486200 7.342221
H 4.986111 -4.397208 6.916166
H 7.052759 -3.712916 7.697626
H 8.666284 -3.208062 8.213486
C 6.029323 -6.621106 3.784093
H 6.770130 -6.572839 2.978705
H 5.498127 -7.577236 3.709459
H 6.568127 -6.610428 4.736328
C 4.266438 -5.590596 2.357633
H 4.940378 -5.567022 1.494017
H 3.518132 -4.799433 2.238511
H 3.739915 -6.550746 2.342268
C 4.017677 -5.532428 4.833579
H 4.511801 -5.494801 5.809180
H 3.455086 -6.471710 4.776899
H 3.307266 -4.700345 4.782917
H 11.380587 3.467588 -1.710344
H 6.624352 6.801995 1.421041
H 5.183394 4.677831 1.199709
H 9.361576 6.571749 -1.014762
H 10.879532 6.762800 1.060165
H 6.837116 3.701807 4.447757
C 13.299260 0.662639 3.892726
C 10.907378 -2.013336 2.658403
C 9.947418 -1.216961 2.071864
C 8.904271 2.207731 1.354735

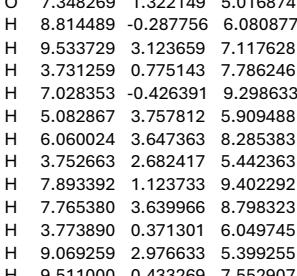
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C 11.157045 0.801944 2.673676
C 12.134434 -0.001190 3.276560
C 14.300959 -0.200020 4.579499
C 15.397255 0.396037 5.204149
C 14.160412 -1.596479 4.601867
C 15.117448 -2.378327 5.249796
C 9.851394 2.893030 0.590866
C 12.013271 -1.405764 3.269684
C 13.000411 -2.263617 3.946519
C 16.344539 -0.389053 5.847568
C 16.204537 -1.777335 5.870215
C 7.824709 4.294308 1.933402
C 6.919968 4.767359 4.213089
C 7.909213 2.899621 2.026658
C 10.817084 5.073948 -0.338054
C 5.345953 4.516154 2.270678
C 6.772025 6.548064 2.476634
C 11.564255 6.050804 0.589749
C 11.854353 4.168546 -1.012349
C 10.083053 5.865822 -1.436499
C 6.729217 5.032989 2.706916
C 8.772911 4.956073 1.155981
C 9.799575 4.277757 0.482152
N 8.978566 0.787018 1.443202
O 12.879798 -3.485746 3.983903
O 8.840918 -1.671474 1.475426
O 13.450496 1.880442 3.846287
H 6.153116 5.298422 4.788662
H 16.946924 -2.389606 6.374378
H 7.718351 6.985498 2.813782
H 9.537727 5.187737 -2.103385
H 12.083502 5.506149 1.385350
H 5.228702 3.446540 2.466546
H 14.984867 -3.455737 5.253552
H 17.195374 0.078864 6.334706
H 8.726955 6.035773 1.075340
H 10.805143 -3.093171 2.674652
H 7.195503 2.335802 2.621201
H 10.622812 2.313264 0.094581
H 15.482932 1.477711 5.171963
H 11.273012 1.880219 2.700893
H 12.558329 4.780746 -1.585728
H 12.431456 3.594741 -0.279071
H 5.966896 7.023365 3.046389
H 12.306128 6.622548 0.020059
H 4.556443 5.044282 2.817478
H 10.800638 6.436430 -2.037555
H 7.904640 5.111626 4.547384
Si 7.770992 -0.401069 0.903377
O 7.997459 -1.044942 -0.723862
P 8.282194 -0.478125 -2.134085
N 7.785641 -1.708946 -3.142971
N 7.491102 0.853705 -2.689539
N 9.888052 -0.055672 -2.160709
C 7.777859 -3.104151 -2.681438
H 7.023103 -3.655199 -3.254469
H 7.518712 -3.156091 -1.624098
H 8.752469 -3.591611 -2.834563
C 8.033395 -1.579335 -4.582162
H 7.260311 -2.131125 -5.130423
H 9.014235 -1.987825 -4.870434
H 7.983394 -0.531646 -4.890403
C 10.497988 0.579818 -3.331548
H 11.226505 1.330472 -2.998744
H 9.746098 1.082132 -3.943628
H 11.028249 -0.155184 -3.954697
C 10.866812 -0.735627 -1.299503
H 10.371267 -1.286386 -0.500056
H 11.532290 0.007353 -0.842024
H 11.476931 -1.438066 -1.884778
C 7.881512 2.215506 -2.280791
H 7.225649 2.576522 -1.480597
H 7.805108 2.882091 -3.148925
H 8.902768 2.235642 -1.899900
C 6.065224 0.770442 -3.050278
H 5.803182 -0.240425 -3.366101
H 5.869814 1.463547 -3.876960
H 5.435606 1.048740 -2.196339

1-(HMPA)₂

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xyz, charge: 0, multiplicity: 1

C	-2.380039	3.616201	2.709710	C	14.564942	0.220467	4.041274	C	4.429754	2.955478	6.260424
C	-2.698909	2.324124	3.128598	C	15.891145	0.650559	4.100906	C	7.077571	3.898179	7.978242
C	-1.052563	3.984453	2.532656	C	14.249924	-1.113590	4.344741	C	4.392351	0.594014	6.928203
C	-0.031487	3.064741	2.773343	C	15.266536	-1.998745	4.704377	C	8.671738	-0.157391	7.154566
C	-0.352577	1.764266	3.195193	C	9.668470	3.592513	1.435810	C	7.097057	0.581448	8.869220
C	-1.688830	1.402120	3.369810	C	11.805087	-0.646435	3.899241	C	8.833226	3.471707	6.342431
H	-3.171352	4.336373	2.521217	C	12.841812	-1.604497	4.295447	N	7.371704	0.473621	7.433213
H	-3.737901	2.038138	3.265843	C	16.896966	-0.236977	4.460106	N	7.431705	3.219852	6.727091
H	-0.778072	4.983021	2.207096	C	16.584044	-1.562885	4.761469	N	5.215988	1.787146	6.683440
H	-1.909643	0.390252	3.695193	C	8.072035	5.566144	2.605401	O	7.348269	1.322149	5.016874
C	1.386151	3.484165	2.577709	C	7.418411	6.656594	4.771010	H	8.814489	-0.287756	6.080877
C	2.427550	2.493646	2.857510	C	7.997259	4.240422	3.044848	H	9.533729	3.123659	7.117628
C	2.114814	1.180280	3.267289	C	10.814279	5.320389	-0.059560	H	3.731259	0.775143	7.786246
C	0.712756	0.756051	3.457860	C	5.691748	6.216044	3.013117	H	7.028353	-0.426391	9.298633
C	3.773831	2.863651	2.701276	C	7.382551	8.019634	2.686554	H	5.082867	3.757812	5.909488
C	4.768647	1.942784	2.945407	C	11.702409	6.460362	0.475429	H	6.060024	3.647363	8.285383
C	4.467186	0.606721	3.338834	C	11.737838	4.165004	-0.463969	H	3.752663	2.682417	5.442363
C	3.126545	0.248741	3.511867	C	10.053385	5.806092	-1.307446	H	7.893392	1.123733	9.402292
H	2.842045	-0.747425	3.832835	C	7.159403	6.613293	3.253138	H	7.765380	3.639966	8.798323
H	4.009665	3.877872	2.397281	C	8.975885	5.882184	1.594355	H	3.773890	0.371301	6.049745
O	1.652540	4.625503	2.196998	C	9.805117	4.909150	1.016207	H	9.069259	2.976633	5.399255
O	0.421936	-0.381588	3.817468	C	8.654028	1.889780	2.855768	H	9.511000	0.433269	7.552907
O	6.073801	2.214214	2.854491	O	12.579780	-2.771374	4.587421				
N	5.595160	-0.128706	3.556394	O	8.182986	-0.501783	3.443535				
C	5.488780	-1.523741	3.777127	O	13.794490	2.346935	3.352844				
C	4.689154	-2.305439	2.932419	H	6.789106	7.422388	5.240034				
C	4.539117	-3.668621	3.152019	H	17.372157	-2.256497	5.041122				
C	5.239510	-4.250686	4.219041	H	8.407727	8.367609	2.855254				
C	6.053242	-3.498992	5.062784	H	9.411728	5.009589	-1.702551				
C	6.159560	-2.123383	4.834795	H	12.235683	6.141034	1.377029				
H	6.774844	-1.498915	5.473114	H	5.475032	5.234991	3.446252				
H	5.136869	-5.315796	4.390559	H	14.996313	-3.024703	4.934368				
H	4.182285	-1.813747	2.109868	H	17.928166	0.101876	4.505607				
C	3.646532	-4.543103	2.267148	H	9.056443	6.907151	1.252196				
C	6.831229	-4.117454	6.226942	H	10.227723	-2.108945	4.073151				
C	8.338652	-3.926936	5.974258	H	7.318395	3.966208	3.842652				
H	8.643754	-4.438448	5.055245	H	10.265209	2.806749	0.991085				
H	8.924787	-4.336765	6.805258	H	16.107366	1.686734	3.859956				
H	8.596502	-2.870250	5.863041	H	11.385533	2.631962	3.063309				
C	6.431803	-3.411363	7.537359	H	12.464303	4.518528	-1.203344				
H	6.630690	-2.335823	7.491934	H	12.294728	3.772491	0.394378				
H	6.996220	-3.827317	8.380685	H	6.705336	8.723922	3.181270				
H	5.362725	-3.545951	7.737247	H	12.443461	6.745868	-0.280141				
C	6.555759	-5.616218	6.392336	H	5.012885	6.945773	3.469510				
H	6.861336	-6.183538	5.506670	H	10.756427	6.104602	-2.094222				
H	5.495737	-5.815298	6.585631	H	8.467275	6.895195	4.980290				
H	7.128663	-6.000312	7.243166	Si	7.130727	0.857008	3.164347				
C	4.522307	-5.567431	1.521754	O	7.003919	0.290206	1.373750				
H	5.262184	-5.058083	0.893614	P	6.848050	0.724656	-0.077585				
H	3.903104	-6.204979	0.879448	N	8.365925	1.19335	-0.657811				
H	5.065098	-6.213399	2.218951	N	5.910977	2.021932	-0.474223				
C	2.867198	-3.720854	1.233506	N	6.109785	-0.578329	-0.834572				
H	3.535049	-3.202753	0.536884	C	9.506752	0.301010	-0.226572				
H	2.222819	-2.975454	1.713176	H	9.348080	-0.084241	0.781392				
H	2.228279	-4.387049	0.643953	H	10.414544	0.916553	-0.215767				
C	2.620719	-5.285785	3.144659	H	9.678084	-0.544223	-0.911512				
H	3.104813	-5.948551	3.867954	C	8.552121	1.690619	-1.992189				
H	1.963178	-5.899261	2.517685	H	7.666788	2.247555	-2.305490				
H	2.001098	-4.572891	3.699072	H	8.765337	0.915459	-2.744417				
H	11.182786	3.338798	-0.921483	H	9.402082	2.384604	-1.972256				
H	7.173608	8.060300	1.612026	C	6.352033	-1.931916	-0.315605				
H	5.468873	6.169078	1.941263	H	6.378332	-1.926903	0.774702				
H	9.415335	6.665606	-1.078105	H	7.294753	-2.356788	-0.695425				
H	11.119505	7.352061	0.723719	H	5.531383	-0.581947	-0.642048				
H	7.182629	5.695149	5.239395	C	5.983983	-0.550047	-2.294207				
C	13.503916	1.189217	3.645656	H	5.125690	-1.164450	-2.593630				
C	10.469956	-1.076255	3.843643	H	6.879605	-0.949739	-2.796057				
C	9.480193	-0.182798	3.493599	H	5.809984	0.468736	-2.650978				
C	8.764261	3.241682	2.450470	C	6.370162	3.396080	-0.215734				
C	9.780664	1.174826	3.172355	H	6.110931	4.032137	-1.072445				
C	11.105005	1.603549	3.261220	H	7.450436	3.429076	-0.059275				
C	12.112131	0.700701	3.614129	H	5.892006	3.796052	0.686201				

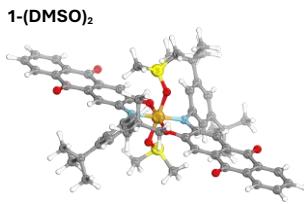


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xyz, charge: 0, multiplicity: 1

C	-1.157618	3.305286	0.978723	C	7.077571	3.898179	7.978242
C	-1.430581	2.337795	1.946685	C	4.392351	0.594014	6.928203
C	0.096357	3.363702	0.385704	C	8.671738	-0.157391	7.154566
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C	0.814539	1.482446	1.730485	C	8.082867	3.757812	5.909488
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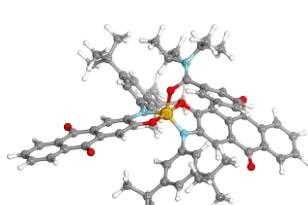
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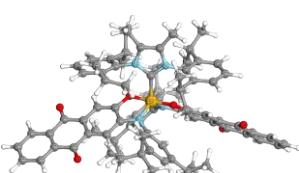
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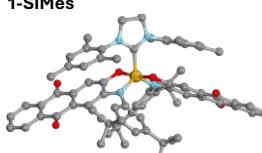
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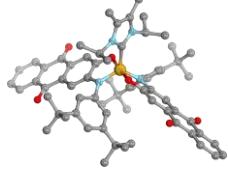
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H	4.246636	-4.097232	6.065543					C	-5.97633	-9.03369	-5.94264
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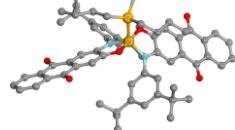
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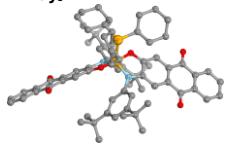
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 H -9.05499 -0.82916 -2.84646
 C -5.53277 1.33144 -3.09636
 H -6.51979 1.78661 -2.98502
 H -4.94679 1.57722 -2.20419
 H -5.04722 1.80753 -3.95037
 C -3.93835 -0.42695 -5.25283
 H -2.90934 -0.60577 -4.92171
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 H -4.04897 0.64758 -5.41200

1-PBu₃



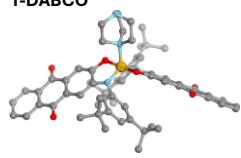
159

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 H 8.41979 -1.14191 5.68048
 C 7.58329 -3.14616 5.45563
 H 6.95276 -3.66858 4.72764
 H 6.96993 -3.01994 6.35792
 C 8.79716 -4.02384 5.76649
 H 9.41834 -3.53710 6.53107
 H 9.41522 -4.10185 4.86160
 C 5.49691 -0.31753 5.47816
 H 6.01544 -0.61451 6.40060
 H 5.48149 0.77829 5.46076
 C 4.06555 -0.85942 5.48275
 H 3.55062 -0.54199 4.57046
 H 3.53676 -0.38502 6.32110
 C 3.96767 -2.37650 5.61863
 H 4.34190 -2.68190 6.60667
 H 4.61753 -2.86033 4.87799
 C 7.48138 0.71139 3.53289
 H 7.61043 1.34781 4.41942
 H 8.47905 0.40286 3.19901
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 H 6.79255 0.84810 1.49906
 H 7.35523 2.36603 2.17672
 C 5.32630 1.86590 2.71220
 H 4.73342 0.97048 2.95066
 H 5.30348 2.50131 3.60939
 C 2.54232 -2.87797 5.41317
 H 2.19706 -2.63391 4.40324
 H 2.48297 -3.96459 5.51846
 H 1.85020 -2.42255 6.13101
 C 8.39565 -5.42276 6.22692
 H 9.27386 -6.02841 6.47050
 H 7.75648 -5.38350 7.11679
 H 7.84643 -5.94822 5.43650
 C 4.66547 2.59412 1.54461
 H 4.61478 1.94786 0.66129

H 3.64116 2.89032 1.78976	C 8.47186 2.27115 -1.94483	H 8.50950 5.06137 10.69666
H 5.22449 3.49735 1.27459	H 7.80496 2.44405 -1.09388	C 8.14393 3.11481 9.88102
C -2.70639 1.19787 -1.53147	H 8.69451 3.24396 -2.39836	C 8.01921 1.75939 10.18900
C -2.02215 1.41827 -2.72768	H 9.40600 1.84630 -1.56146	H 7.78024 1.02696 9.42564
C -2.03050 0.70913 -0.42163	C 6.50851 1.94388 -3.48338	C 6.07265 1.30711 14.45274
H -3.77007 1.40905 -1.46791	H 5.80210 2.10543 -2.66292	C 6.83544 0.54383 15.37635
C -0.66272 1.14992 -2.81229	H 6.02499 1.28483 -4.21208	C 8.26717 -1.44100 15.47311
C 0.02427 0.65773 -1.70175	H 6.69433 2.91208 -3.96273	C 9.57629 -1.53809 15.01327
C -0.66413 0.43642 -0.49811	C 8.78967 1.21803 -4.17937	H 9.87261 -0.93034 14.16495
H -2.53975 0.52877 0.51990	H 9.76080 0.80617 -3.88342	C 10.47478 -2.41326 15.62137
H -2.55366 1.80014 -3.59467	H 8.96515 2.21310 -4.60156	C 10.02358 -3.15347 16.71754
H -0.10779 1.31307 -3.73085	H 8.37254 0.59038 -4.97425	H 10.71378 -3.83751 17.20236
C 0.03866 -0.09024 0.70404	C 1.26125 -5.11636 1.13597	C 8.72129 -3.04989 17.21506
C 1.48338 -0.35360 0.58773	C 0.15373 -4.45655 1.98047	C 7.83896 -2.18253 16.57019
C 2.17144 -0.16394 -0.62677	H -0.06755 -5.04128 2.87950	H 6.80966 -2.07608 16.89600
C 1.48275 0.38083 -1.81653	H -0.76848 -4.36775 1.39455	C 4.63441 -1.65939 10.67743
C 2.16002 -0.86339 1.70583	H 0.45224 -3.45007 2.29333	C 3.64737 -2.10084 10.46681
C 3.49715 -1.16751 1.58885	C 1.49234 -4.25535 -0.11271	C 5.69337 -2.46074 9.90263
C 4.20392 -0.95336 0.37976	H 0.56908 -4.21341 -0.70008	H 5.68237 -3.51508 10.20126
C 3.53468 -0.46444 -0.73205	H 2.27726 -4.67281 -0.75348	H 6.69446 -2.07347 10.13589
H 1.61745 -1.02953 2.63041	H 1.76168 -3.22546 0.14429	C 5.45289 -2.35240 8.39370
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C 7.18960 -4.31026 2.20302	C 4.07580 -6.53629 6.39784	C 4.40319 -0.08734 8.71026
C 9.49024 -3.60218 1.79607	H 5.03932 -6.02237 6.32324	H 3.39868 -0.46054 8.46122
C 9.90449 -4.93533 1.92771	H 4.15236 -7.26614 7.21194	H 4.43075 0.96755 8.41096
C 8.15268 -3.30275 1.92843	H 3.31714 -5.79633 6.67411	C 4.62202 -0.18834 10.22277
C 8.95520 -5.94338 2.18064	C 4.77665 -8.30393 4.75730	H 5.57979 0.28198 10.48379
C 7.59624 -5.63152 2.30870	H 4.83167 -9.05330 5.55559	H 3.84500 0.37780 10.74923
C 11.33244 -5.25181 1.74943	H 5.76838 -7.85199 4.65097	C 8.92511 4.60711 13.33649
C 13.49436 -8.33603 1.84588	H 4.54397 -8.81649 3.81795	C 7.71028 5.54427 13.48109
C 12.56069 -9.33214 2.13495	C 2.36370 -7.97163 5.30201	H 7.49675 6.07610 12.54825
C 13.08834 -7.01392 1.72363	H 1.55616 -7.26734 5.53197	H 6.81455 4.98075 13.76346
C 11.22188 -9.00475 2.30166	H 2.45456 -8.66282 6.14651	H 7.90057 6.29131 14.26055
C 11.74443 -6.67623 1.88758	H 2.07388 -8.55961 4.42462	C 7.96691 3.64964 8.45770
C 9.36541 -7.35114 2.36529	1-PCy₃	
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H 12.88125 -10.36572 2.22944	O 7.66620 -2.29742 12.77597	
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H 14.54137 -8.59486 1.71626	C 12.86079 -2.87946 8.99371	
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Si 5.90321 -2.01505 2.10285	O 11.26822 -5.43317 11.02265	
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C 3.66859 -4.50349 1.73432	O 6.02055 0.78257 13.21926	
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C 2.52891 -6.10976 3.10058	H 15.23146 -5.85769 8.26785	
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H 6.65160 -3.40659 -0.64239	H 10.24816 0.13771 10.04826	
H 8.73805 -1.19691 -3.67002	C 8.94963 -0.99949 11.35427	
C 7.83328 1.33260 -2.98687	C 8.70601 -2.27386 11.92709	
C 8.33939 -3.78016 -2.75709	C 9.49937 -3.35945 11.63148	
C 9.29381 -3.57921 -3.93996	H 9.31960 -4.32772 12.08677	
H 8.79293 -3.10814 -4.79291	C 8.20447 1.32059 11.49609	
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 H 7.54380 2.96694 6.45183
 H 8.45187 1.78869 7.40796
 H 6.71202 2.02655 7.69612
 C 6.80716 4.66308 8.43230
 H 6.66226 5.04709 7.41579
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 H 6.73281 3.85762 20.82052

1-DABCO

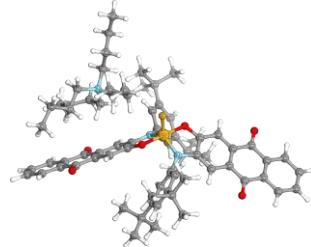


139
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 C -9.07852 0.21231 -4.08547
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 C -3.32251 1.80806 -1.55987
 C -3.67075 0.51483 -1.09540
 C -4.86861 -0.06094 -1.48579
 H -3.87591 3.50347 -2.76080
 H -5.15649 -1.05717 -1.16625
 H -8.04181 3.71556 -5.37594
 H -9.33978 -0.77048 -3.70576
 H -10.19586 2.69192 -6.11860
 H -10.84469 0.44739 -5.28381

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 O -7.34181 -1.06590 -2.29899
 O -2.11050 2.22285 -1.15496
 N -2.65777 -0.01987 -0.31143
 C -2.70979 -1.38699 0.10175
 C -1.81810 -2.29684 -0.46632
 C -1.80241 -3.62111 -0.04721
 C -2.73616 -4.01510 0.92421
 C -3.66010 -3.13127 1.47611
 C -3.62452 -1.79446 1.05809
 H -4.31075 -1.05977 1.47103
 H -1.12832 -1.93885 -1.22343
 H -2.73449 -5.04819 1.25137
 C -4.70215 -3.55975 2.51171
 C -6.10996 -3.30045 1.94115
 H -6.26876 -2.24168 1.71415
 H -6.87434 -3.60836 2.66405
 H -6.26296 -3.86486 1.01536
 C -4.51080 -2.73857 3.80125
 H -4.64058 -1.66664 3.62155
 H -3.50824 -2.89494 4.21426
 H -5.24617 -3.04186 4.55542
 C -4.59404 -5.04608 2.87093
 H -4.75482 -5.68618 1.99678
 H -5.36085 -5.29805 3.61100
 H -3.61880 -5.28928 3.30688
 C -0.80234 -4.63937 -0.59966
 C 0.09860 -4.03885 -1.68619
 H -0.48122 -3.68420 -2.54578
 H 0.79176 -4.80595 -2.04704
 H 0.70014 -3.20732 -1.30254
 C -1.56537 -5.83101 -1.20876
 H -2.22696 -5.49689 -2.01530
 H -2.17665 -6.34897 -0.46332
 H -0.85676 -6.55765 -1.62227
 C 0.09659 -5.13487 0.54970
 H 0.66206 -4.30671 0.98991
 H 0.81591 -5.87179 0.17400
 H -0.48750 -5.60918 1.34484
 H 2.20445 5.03977 -2.69460
 H 2.78946 6.07876 -4.00183
 C 2.03535 5.32872 -3.73658
 H 1.04628 5.79489 -3.80298
 H 3.71900 3.13523 -3.52854
 H 4.30462 4.18894 -4.82593
 C 3.52338 3.47043 -4.55183
 C 2.13216 4.11953 -4.68632
 O -0.32295 -0.05382 0.80416
 H 1.80763 2.85025 -2.28364
 C 1.07231 2.53936 -3.02104
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 H 1.01066 5.12466 -6.27664
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 H 3.61006 2.59874 -5.20894
 C 2.28052 -0.23368 -1.66702
 C 2.82616 -1.78768 0.13700
 H 0.04356 3.09641 -6.19100
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 O 0.33863 -3.39631 1.77362
 C 3.12649 -1.19978 -1.10799
 C 3.67153 -2.84926 0.71110
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 H -3.40659 0.50081 -4.44194
 C 4.34669 -1.57544 -1.85415
 C 4.88623 -3.25028 -0.04996
 H -3.58611 2.24656 -6.93425
 H 2.06703 3.81424 -6.85265
 C 5.20721 -2.64384 -1.27495
 C 5.71388 -4.24784 0.46699
 O 4.64709 -1.03281 -2.91221
 C -1.94575 1.28835 -5.87249

H 5.44329 -4.70155 1.41528
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 C 6.35247 -3.04196 -1.96584
 C 6.85091 -4.63841 -0.22726
 C 7.17046 -4.03510 -1.44447
 H -3.60826 -0.07151 -6.10043
 H 7.49283 -5.41519 0.17845
 H 6.57900 -2.55743 -2.91043
 H -2.30550 -0.73178 -5.10327
 H 8.06050 -4.34300 -1.98579
 C -1.26798 0.77695 -7.15783
 H -0.65124 1.54860 -7.62877
 H -2.02771 0.46750 -7.88463
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 C -1.86417 1.61024 2.52807
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 C 0.22084 3.69430 2.79743
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 H -2.93901 1.56536 2.32217
 H 0.68981 2.97792 0.75721
 H 0.68909 1.70115 1.98503
 H 0.48049 4.68255 2.40122
 H 0.97407 3.43098 3.54855
 H -0.75623 1.95905 4.38649
 H -2.43572 2.50961 4.43238
 C -2.03940 4.35089 2.46875
 H -2.96491 3.35315 0.73656
 H -1.43817 4.14263 0.34930
 H -3.04646 4.29498 2.89797
 H -1.80943 5.41035 2.30857
 N -1.09437 3.79381 3.44512

[1-F][NBu₄]



173
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 C 1.501460 2.343346 0.795862
 C 1.224021 1.374416 1.775041
 C -0.077262 1.231273 2.258568
 H -1.617610 3.643575 0.429733
 H -2.107775 1.927032 2.155307
 H 0.701234 3.903466 -0.434853
 H -0.275841 0.477167 3.013994
 C 2.878081 2.511403 0.253608
 C 3.918845 1.592673 0.729330
 C 3.649197 0.636406 1.733803
 C 2.297556 0.483824 2.294396
 C 5.207955 1.687757 0.177575
 C 6.192598 0.819773 0.606024
 C 5.915570 -0.162476 1.599690
 C 4.654423 -0.236979 2.172012
 H 4.425036 -0.973022 2.935603
 H 5.421461 2.432349 -0.582860
 O 3.116928 3.390547 -0.577865
 O 2.044539 -0.359602 3.156091
 O 7.439887 0.795180 0.154647
 N 7.020104 -0.938724 1.848912
 C 6.938834 -1.968930 2.837475
 C 6.434633 -3.218373 2.476352
 C 6.346490 -4.236981 3.421451
 C 6.782172 -3.967183 4.728934
 C 7.285276 -2.722607 5.107548
 C 7.349667 -1.713947 4.135844
 H 7.734104 -0.726032 4.376343
 H 6.725125 -4.760614 5.465329

H	6.133035	-3.372490	1.445780	H	5.209219	4.818633	3.398799	C	3.114830	0.407610	2.205125
C	5.804618	-5.627243	3.077152	H	10.603868	6.421260	-2.292839	C	1.758108	0.199039	2.736599
C	7.765873	-2.422571	6.530285	H	8.724339	5.151584	4.705498	C	4.661718	1.542313	0.699655
C	9.245638	-1.994876	6.486963	Si	8.470394	-0.438543	0.933588	C	5.655325	0.673150	1.101266
H	9.868588	-2.787700	6.056927	F	8.746192	-1.137840	-0.520998	C	5.403527	-0.326800	2.080065
H	9.607369	-1.788871	7.501487	N	4.268805	-2.08347	-2.120296	C	4.136111	-0.450902	2.631119
H	9.386967	-1.087959	5.890140	C	5.641696	-1.576063	-2.388113	H	3.915414	-1.208989	3.374837
C	6.924263	-1.273293	7.117161	C	4.328420	-3.723332	-2.267025	H	4.871237	2.320504	-0.026983
H	7.026573	-0.355943	6.528172	C	3.207753	-1.643731	-3.042634	O	2.567067	3.290335	0.063331
H	7.251286	-1.054048	8.140587	C	3.978864	-1.904621	-0.643383	O	1.507064	-0.699284	3.540884
H	5.861935	-1.542307	7.147369	C	2.784833	-2.569333	0.033376	O	6.905350	0.689129	0.651666
C	7.642307	-3.635713	7.458509	H	3.907769	-0.817193	-0.565556	N	6.540280	-1.052917	2.350240
H	8.243663	-4.481242	7.105653	H	4.895357	-2.213824	-0.131781	C	6.484355	-2.108907	3.310584
H	6.603041	-3.968454	7.560299	C	2.915119	-0.160801	-2.861118	C	5.740640	-3.254325	3.021494
H	8.003897	-3.364654	8.456491	H	2.306446	-2.240532	-2.881906	C	5.663106	-4.290637	3.947322
C	6.901257	-6.678782	3.329897	H	3.549949	-1.837240	-4.062369	C	6.367822	-4.155430	5.154818
H	7.783697	-6.478550	2.711166	C	5.989705	-1.230278	-3.829763	C	7.118446	-3.019726	5.457955
H	6.527812	-7.679322	3.080629	H	6.369609	-2.276519	-1.964410	C	7.156466	-1.982639	4.516216
H	7.217373	-6.690123	4.378066	H	5.676213	-0.661579	-1.788172	H	7.723491	-1.074756	4.704015
C	5.365040	-5.730897	1.611801	C	4.889958	-4.286644	-3.561681	H	6.321878	-4.966673	5.872364
H	6.200174	-5.555299	0.924204	H	3.302856	-4.076061	-2.129206	H	5.232043	-3.308923	2.064770
H	4.565840	-5.017244	1.380050	H	4.929965	-4.076435	-1.421533	C	4.845597	-5.559147	3.687624
H	4.979599	-6.737950	1.417782	C	4.748409	-5.812862	-3.545834	C	7.889752	-2.863157	6.772346
C	4.587881	-5.935926	3.969393	H	5.950775	-4.033986	-3.666695	C	9.384051	-2.651142	6.461912
H	4.854474	-5.936316	5.031234	H	4.360058	-3.887685	-4.434734	H	9.793434	-3.504572	5.908997
H	4.184004	-6.925415	3.724003	C	5.334791	-6.453107	-4.800636	H	9.951187	-2.543997	7.394429
H	3.794310	-5.194560	3.818322	H	3.687042	-6.079668	-3.454020	H	9.548708	-1.749006	5.864080
H	11.439229	3.511378	-1.978930	H	5.252108	-6.214503	-2.655749	C	7.351786	-1.634986	7.530696
H	6.966073	6.666470	1.722418	H	4.828970	-6.088833	-5.702481	H	7.479283	-0.714872	6.951142
H	5.661167	4.463514	1.718305	H	5.228268	-7.542727	-4.774286	H	7.890042	-1.512248	8.478210
H	9.324560	6.508399	-1.070811	H	6.402080	-6.223206	-4.898752	H	6.285032	-1.750776	7.755401
H	11.096688	6.877295	0.739621	C	7.455444	-0.787616	-3.900887	C	7.7555676	-4.089583	7.681433
H	7.752359	3.669020	4.760467	H	5.357717	-0.410820	-4.190854	H	8.143971	-4.996018	7.203241
C	13.539544	0.934225	4.560215	C	5.839317	-2.076013	-4.505890	H	6.714885	-4.270801	7.973236
C	11.368065	-1.871488	3.214366	C	7.854191	-0.367430	-5.312418	H	8.333113	-3.923125	8.597419
C	10.494322	-1.144987	2.431223	H	8.099660	-1.609506	-3.561243	C	5.780944	-6.782458	3.702726
C	9.473190	2.210941	1.427307	H	7.619098	0.046808	-3.205778	H	6.550938	-6.694540	2.927445
C	10.603864	0.270965	2.334928	H	7.724026	-1.193150	-6.021931	C	5.207115	-7.697437	3.512309
C	11.597975	0.939546	3.033967	H	8.903715	-0.568670	-5.348826	H	6.283253	-6.896241	4.668985
C	12.486588	0.206371	3.833578	H	7.244060	0.473978	-5.661264	C	4.124123	-5.518224	2.335409
C	14.447778	0.148398	5.440167	C	2.065173	0.348036	-4.028269	H	4.828453	-5.444496	1.498939
C	15.440076	0.809974	6.165109	H	3.843434	0.420879	-2.805449	H	3.425072	-4.675769	2.274566
C	14.319037	-1.246462	5.549163	H	2.373071	0.014848	-1.925715	H	3.547282	-6.440319	2.204153
C	15.184169	-1.956956	6.382499	C	1.670581	1.811060	-3.852635	C	3.784921	-5.715738	4.793754
C	10.261388	2.916928	0.518463	H	1.161902	-0.271629	-4.115255	H	4.241736	-5.811186	5.784117
C	12.376734	-1.198999	3.923738	H	2.623942	0.221770	-4.965548	H	3.184964	-6.615616	4.612532
C	13.270142	-1.980179	4.788265	H	1.074681	1.950778	-2.942886	H	3.109624	-4.852214	4.811300
C	16.297055	0.094641	6.992977	H	1.074858	2.165257	-4.700705	H	10.847960	3.235071	-1.964075
C	16.168918	-1.290049	7.101666	H	2.557006	2.451032	-3.773357	H	6.769780	6.624183	1.834681
C	8.382466	4.252034	2.129320	C	1.399958	-1.961630	-0.200353	H	5.390671	4.460047	1.959861
C	7.733328	4.731843	4.497061	H	3.011628	-2.461823	1.104327	H	9.172636	6.454240	-0.869755
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C	10.958151	5.127219	-0.579171	C	0.367215	-2.5454918	0.759927	H	7.549621	3.720717	4.963032
C	5.958891	4.329929	2.764890	H	1.062850	-2.124326	-1.230636	C	13.668000	0.670939	3.919278
C	7.260736	6.452335	2.756125	H	1.457819	-0.874999	-0.055821	C	11.134988	-2.049662	3.159027
C	11.756271	6.188941	0.201271	H	0.264923	-3.628937	0.620302	C	10.148428	-1.295047	2.555856
C	11.949148	4.267696	-1.371583	H	-0.617970	-2.093529	0.602468	C	9.092795	2.070494	1.627245
C	10.019024	5.830384	-1.577587	H	0.657313	-2.366934	1.801892	C	10.296786	0.109045	2.392521
C	7.347873	4.945325	3.020978					C	11.444030	0.742232	2.848202
C	9.173831	4.942735	1.212061					C	12.449521	-0.017739	3.462211
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N	9.613768	0.791252	1.532675					C	15.886746	0.475281	5.033527
O	13.165558	-3.204130	4.895767					C	14.558969	-1.530957	4.755568
O	9.484780	-1.658606	1.735548					C	15.564450	-2.271805	5.378335
O	13.674366	2.154247	4.457462					C	9.868862	2.711244	0.659854
H	7.006575	5.227658	5.151788					C	12.300068	-1.413733	3.616213
H	16.839180	-1.840907	7.748602					C	13.325922	-2.221846	4.287664
H	8.211095	6.959440	2.957786					C	16.883170	-0.270207	5.652094
H	9.429297	5.097307	-2.140270					C	16.722083	-1.645100	5.824054
H	12.424499	5.716433	0.930532					C	8.110190	4.178074	2.292658
H	5.941996	3.258318	2.989415					C	7.557558	4.773041	4.660509
H	15.072813	-3.034302	6.455643					C	8.231681	2.790719	2.440884
H	17.066904	0.615575	7.555278					C	10.642590	4.858609	-0.509445
H	9.056738	6.016037	1.115640					C	5.714766	4.368984	3.003200
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H	7.963140	2.278807	2.931679					C	11.601604	5.793776	0.252810
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H	15.527448	1.887910	6.070766					C	9.739218	5.702191	-1.427977
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H	12.520801	4.909280	-2.051148					C	8.899673	4.804882	1.329144
H	12.663197	3.759408	-0.713654					C	9.787830	4.094007	0.505625
H	6.503837	6.891666	3.415334					C	9.184895	0.653812	1.790537
H	12.367875	6.780966	-0.490068					C	13.184264	-3.433097	4.467923

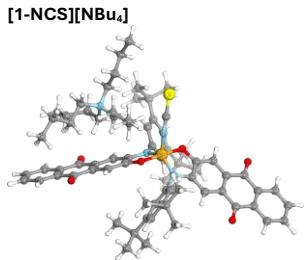
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xyz, charge: 0, multiplicity: 1

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C -1.651895 1.743674 2.272074
C -0.087

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O	13.825253	1.882587	3.764282		C	7.639778	4.816007	4.496201
H	6.868154	5.317366	5.317062		C	8.508433	2.937811	2.255231
H	17.501441	-2.227808	6.307078		C	10.986948	5.197861	-0.492767
H	8.062658	6.921397	3.018679		C	5.902229	4.395156	2.731746
H	9.022943	5.067758	-1.962843		C	7.195076	6.523223	2.734547
H	12.260517	5.220594	0.915516		C	11.763087	6.260172	0.308579
H	5.665237	3.311736	3.283328		C	11.998843	4.337575	-1.257444
H	15.424594	-3.340807	5.506189		C	10.075015	5.900208	-1.516788
H	17.787818	0.219552	6.001423		C	7.283204	5.018383	3.011271
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H	15.997709	1.546042	4.893136		O	9.450118	-1.590657	1.745677
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H	12.185647	3.328457	-0.798434		H	17.007727	-1.889910	7.497034
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H	12.228191	6.350177	-0.454865		H	9.499757	5.166578	-2.093532
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H	8.567859	5.166299	4.823116		H	15.192238	-3.050386	6.251159
Si	7.946960	-0.558396	1.363456		H	17.229894	0.577937	7.341739
Cl	8.152126	-1.535456	-0.557932		H	9.041458	6.087727	1.150673
C	4.517991	-0.970180	-7.885887		H	11.282816	-2.907416	3.227479
C	4.736011	-0.245020	-6.721225		H	7.898617	2.351869	2.938203
C	3.592349	-2.012029	-7.898881		H	10.979594	2.432831	-0.006985
C	2.882172	-2.332476	-6.745606		H	15.637984	1.878019	5.939302
C	3.089712	-1.612340	-5.573433		H	11.704300	2.065021	2.981925
C	4.015922	-0.566384	-5.562850		H	12.590345	4.978440	-1.920449
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H	5.465294	0.560911	-6.713502		H	6.421224	6.963179	3.373321
H	3.428836	-2.579024	-8.810994		H	12.393023	6.852190	-0.366038
H	2.166139	-3.148871	-6.752775		H	5.137778	4.888049	3.344345
H	2.539092	-1.873129	-4.674674		H	10.679062	6.489190	-2.217474
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C	2.534986	-1.833257	-0.465062		H	2.357691	-2.246076	-2.954970
C	2.218677	-0.385789	-2.368970		H	3.597932	-1.846313	-4.139308
C	3.900331	-1.895546	-0.728117		C	6.029385	-1.230069	-3.920869
C	3.593748	-0.451258	-2.630001		H	6.421214	-2.255724	-2.046988
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C	3.052727	2.494908	-5.476759		H	4.924831	-6.119193	-5.717375
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H 0.331239 -3.613511 0.555181
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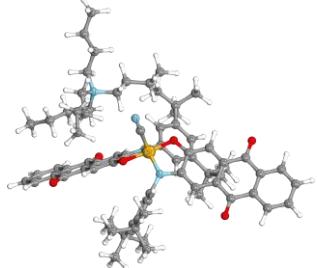
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C 1.192473 1.262956 1.837003
C -0.120245 1.122002 2.289342
H -1.604208 3.566299 0.456406
H -2.143105 1.833214 2.150519
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H -0.685843 -2.199883 0.476087
H 0.578201 -2.589296 1.655214
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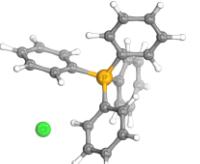


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C 1.007335 1.980274 1.116989
C 0.886428 0.845008 1.936097
C -0.364570 0.488571 2.442229
H -2.240273 2.980544 1.097215
H -2.456314 0.969878 2.536868
H -0.013189 3.617294 0.186323
H -0.444014 -0.392808 3.071130
C 2.321145 2.367746 0.536837
C 3.478762 1.511125 0.829455
C 3.369318 0.394303 1.685689
C 2.072252 0.006919 2.264420
C 4.715967 1.836118 0.253940
C 5.809042 1.036418 0.514917
C 5.701961 -0.104559 1.357160
C 4.488601 -0.405702 1.957873
H 4.381595 -1.259393 2.618805
H 4.809733 2.710991 -0.381075
O 2.419598 3.369321 -0.174163
O 1.954830 -0.983938 2.986494
O 7.032953 1.243515 0.033289
N 6.912697 -0.758442 1.467142
C 7.071228 -1.776646 2.460115
C 6.485108 -3.030857 2.282307
C 6.621138 -4.010437 3.263398
C 7.365776 -3.702633 4.413640
C 7.955006 -2.454277 4.609042
C 7.790133 -1.486175 3.610215
C 8.223328 -0.495105 3.718849
H 7.477861 -4.465158 5.176070
H 5.922383 -3.216435 1.374740

C	5.989229	-5.398598	3.128018	H	9.818981	7.774690	-0.738759	H	11.657062	-5.186341	-3.193453
C	8.748400	-2.098277	5.869240	H	7.832503	4.347216	5.515188	H	11.385980	-4.174020	-1.765257
C	10.181665	-1.696255	5.473600	Si	8.226795	0.040548	0.564586	H	12.358117	-3.561971	-3.112483
H	10.690574	-2.520115	4.960372	C	8.532527	-0.324107	-1.280875	C	8.333348	-0.490636	-4.065199
H	10.760842	-1.441249	6.369263	N	3.980856	-2.563152	-2.141096	H	10.348523	-1.008357	-4.661255
H	10.190206	-0.825835	4.808844	C	5.287150	-3.037288	-1.491877	H	9.636228	0.336152	-5.582087
C	8.066946	-0.909691	6.573524	C	3.145782	-3.741507	-2.622410	C	8.862346	0.363463	-2.908495
H	8.036772	-0.023152	5.931447	C	4.252337	-1.618428	-3.295339	H	8.000080	-1.460387	-3.672568
H	8.618443	-0.646319	7.483946	C	3.203790	-1.887301	-0.999751	H	7.455749	0.010146	-4.492928
H	7.037920	-1.160292	6.856694	C	1.737485	-1.522410	-1.207122	C	7.802301	0.598646	-1.837097
C	8.829035	-3.265090	6.859308	H	3.781541	-1.004535	-0.717364	H	9.213240	1.325511	-3.304644
H	9.332620	-4.135551	6.423648	H	3.272731	-2.606766	-0.177574	H	9.736695	-0.130251	-2.462257
H	7.836754	-3.575930	7.205746	C	4.905167	-0.301947	-2.901623	H	6.933362	1.123744	-2.250613
H	9.404441	-2.953548	7.738012	H	3.293319	-1.445387	-3.789918	H	8.197515	1.204726	-1.014697
C	7.097543	-6.467593	3.156925	C	4.890622	-2.162371	-3.995713	H	7.450856	-0.350367	-1.414951
H	7.803626	-6.320648	2.331089	C	6.480376	-3.312690	-2.395809	C	7.600445	0.021694	-7.516933
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C	5.202023	-5.555676	1.822356	C	3.811387	-4.752329	-3.541441	C	6.125935	0.277981	-7.841548
H	5.843134	-5.423753	0.943452	H	2.282425	-3.303410	-3.130394	H	8.182121	0.066921	-8.446252
H	4.373291	-4.840545	1.758697	H	2.793674	-4.245162	-1.715244	H	7.983542	0.817446	-6.866449
H	4.775702	-6.563780	1.774838	C	2.755265	-5.738313	-4.053965	C	5.923105	1.616096	-8.546169
C	5.022635	-5.634794	4.304167	H	4.584687	-5.312511	-3.004384	H	5.542140	0.257888	-6.911117
H	5.540402	-5.602533	5.268225	H	4.285692	-4.263019	-4.399918	H	5.739449	-0.535211	-8.471846
H	4.549676	-6.619602	4.209281	C	3.367670	-6.820245	-4.938526	H	6.283271	2.444530	-7.925201
H	4.231987	-4.875459	4.317142	H	1.988878	-5.188432	-4.616639	H	4.864312	1.794000	-8.763961
H	10.892472	4.968559	-1.227325	H	2.245178	-6.203717	-3.199498	H	6.471249	1.648508	-9.495028
H	6.101050	6.593260	3.006398	H	3.865671	-6.381092	-5.810860	C	10.190131	-2.330841	-8.482525
H	4.992356	4.410779	2.241063	H	2.601550	-5.712232	-5.303848	H	11.168872	-1.521056	-6.734377
H	8.509407	7.411544	0.397456	H	4.112960	-7.405755	-4.387702	H	10.148135	-0.366648	-7.576833
H	10.204657	7.367944	2.323480	C	7.624813	-3.871985	-1.541949	C	11.279972	-1.910783	-9.474710
H	6.972116	2.866760	5.052617	H	6.823794	-2.391667	-2.879630	H	9.215622	-2.282127	-8.983922
C	13.589583	1.127243	3.386903	H	6.242975	-4.028959	-3.185838	H	10.360431	-3.374299	-8.189863
C	11.527472	-1.521820	2.069049	C	8.850157	-4.214973	-2.382726	C	11.307048	-2.807227	-10.709038
C	10.478475	-0.747283	1.622079	H	7.278294	-4.767677	-1.008446	H	12.257457	-1.939060	-8.974245
C	9.006774	2.597581	1.601329	H	7.899426	-3.134885	-0.775422	H	11.111816	-0.868404	-9.776797
C	10.440088	0.652414	1.869785	H	8.610077	-4.977152	-3.133423	H	11.500660	-3.851169	-10.435252
C	11.453374	1.251522	6.2603046	H	9.660919	-4.602607	-1.756692	H	12.089826	-2.493535	-11.408006
C	12.519497	0.468505	3.067578	H	9.226622	-3.330725	-2.909306	H	10.349154	-2.772156	-11.241399
C	14.701617	0.279696	4.346706	C	5.453803	0.410233	-4.140596	F	9.975170	2.084757	-6.255524
C	15.728515	0.869764	5.085234	H	5.728997	-0.462553	-2.194591				
C	14.731294	-1.102230	4.093814	H	4.179596	0.351082	-2.404936				
C	15.786898	-1.871959	4.584681	C	5.987254	1.799507	-3.804452				
C	9.742995	3.594957	0.959948	H	4.662295	0.490038	-4.898916				
C	12.563929	-0.915677	2.796076	H	6.253255	-0.201023	-4.581030				
C	13.650920	-1.760100	3.309191	H	5.188259	2.443876	-3.418618				
C	16.776288	0.095869	5.569355	H	6.412384	2.285542	-4.689037				
C	16.805239	-1.276200	5.319077	H	6.770916	1.744024	-3.039799				
C	7.690049	4.259560	2.764735	C	1.411033	-0.252688	-1.994064				
C	6.888241	3.956607	5.118293	H	1.369645	-1.372083	-0.181427				
C	7.996914	2.919002	2.495709	H	1.161704	-2.370940	-1.592790				
C	10.245786	6.074602	0.549523	C	-0.078116	0.074373	-1.908400				
C	5.244046	3.990693	3.221893	H	1.696099	-0.351017	-3.047903				
C	6.371455	6.093088	3.943340	H	1.997296	0.581016	-1.586019				
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C	11.323154	5.558716	-0.410375	H	-0.308091	0.107357	-2.415921				
C	9.270770	6.961536	-0.248158	H	-0.399873	0.168424	-0.864081				
C	6.563500	4.585374	3.749789	N	8.743958	-0.491069	-2.408774				
C	8.436782	5.243679	2.118772								
C	9.468341	4.936057	1.216055								
N	9.299393	1.223320	1.342139								
O	13.677678	-2.975839	3.108970								
O	9.420245	-1.210402	0.953067								
O	13.574099	2.337592	4.061610								
H	6.092861	4.188938	5.836486								
H	17.624599	-1.880763	5.697822								
H	7.272271	6.571977	4.343593								
H	8.758565	6.378770	-1.022559								
H	11.628107	6.314288	2.225336								
H	5.306673	2.902423	3.120210								
H	15.796253	-2.938397	4.381793								
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H	8.219324	6.287040	2.316634								
H	11.544715	-2.591050	1.884240								
H	7.447164	2.112289	2.974340								
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PPh₄Cl

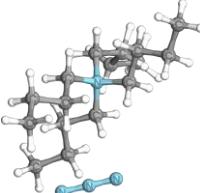


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C	11.474280	-4.158754	-2.856920
H	10.045901	-2.568161	-3.129872
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C 3.505374 1.588719 -2.854257
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C 3.496824 1.136903 -4.180516
H 2.264223 4.708693 -3.373019
H 3.074071 3.230100 -1.547152
H 2.238292 3.887927 -5.714301
H 3.857708 0.954387 -2.038302
H 3.018886 1.608654 -6.238904
P 4.111034 -0.525798 -4.535484
Cl 4.885558 0.102019 0.174257

NBu₄N₃



56

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C 7.824313 -1.268267 -6.814621
C 10.299736 -3.642279 -5.032339
H 8.813135 -3.636857 -6.622634
H 8.200799 -3.091772 -5.065922
C 10.196416 -3.690406 -3.505079
H 11.213543 -3.111989 -5.324590
H 10.427057 -4.666588 -5.404081
C 11.434368 -4.316196 -2.870959
H 10.048398 -2.681151 -3.102274
H 9.303585 -4.267477 -3.228704
H 11.592357 -5.337652 -3.237474
H 11.341954 -4.362001 -1.780675
H 12.333575 -3.734499 -3.106425
C 8.457957 -0.473476 -4.010831
H 10.443489 -1.091229 -4.617832
H 9.803596 0.280784 -5.516261
C 9.051078 0.300518 -2.828971
H 8.057621 -1.429701 -3.651814
H 7.622203 0.102477 -4.425563
C 7.992526 0.660326 -1.791941
H 9.527432 1.217005 -3.202684
H 9.842977 -0.301830 -2.363157
H 7.216592 1.298459 -2.230771
H 8.433359 1.202538 -0.948674
H 7.504422 -0.238290 -1.395897
C 7.721692 0.111349 -7.438620
H 7.100358 -1.376594 -6.002058
H 7.607683 -2.045362 -7.552774
C 6.268559 0.409410 -7.820345
H 8.346432 0.179058 -8.337050
H 8.061835 0.882410 -6.738766
C 6.130702 1.782988 -8.470718
H 5.641199 0.362035 -6.919911
H 5.899069 -0.367000 -8.504488
H 6.484316 2.570859 -7.794683
H 5.087598 1.999264 -8.725239
H 6.721530 1.843589 -9.392232
C 10.207151 -2.365361 -8.456064
H 11.230641 -1.610031 -6.708991
H 10.260988 -0.406623 -7.546829
C 11.321463 -2.003695 -9.444063
H 9.238256 -2.260852 -8.959649
H 10.318789 -3.417165 -8.166281
C 11.299042 -2.896042 -10.681352
H 12.294356 -2.090560 -8.941674
H 11.214129 -0.952369 -9.742443
H 11.429954 -3.950290 -10.410507
H 12.101148 -2.626784 -11.376649
H 10.346713 -2.802798 -11.216430
N 10.400890 2.583858 -6.588049
N 9.448952 2.840079 -5.939775

N 8.491906 3.082294 -5.294708

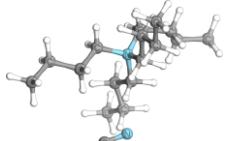
NBu₄NCS



56

xyz, charge: 0, multiplicity: 1
N -0.971152 1.191918 0.271562
C -0.953802 2.308670 1.308327
C -2.180203 1.305389 -0.645503
C -0.997818 -0.107496 1.066287
C 0.252633 1.258818 -0.631408
C -0.911706 3.725800 0.766298
H -0.086072 2.111918 1.944232
H -1.844463 2.160854 1.925016
C -0.866340 4.712692 1.938351
H -1.794713 3.941575 0.152279
H -0.025376 3.880852 0.139250
C 1.598476 1.197885 0.068705
H 0.159472 2.188051 -1.200569
H 0.151117 0.431169 -1.339037
C 2.722764 1.236002 -0.971502
H 1.692416 0.280518 0.661674
H 1.720480 2.043391 0.755520
C 4.101549 1.191691 -0.319231
H 2.628197 2.148002 -1.576645
H 2.610776 0.387655 -1.660340
H 4.246476 2.045405 0.352898
H 4.896471 1.218041 -1.072297
H 4.228458 0.276935 0.271358
C -0.999510 -1.391274 0.256562
H -1.886542 -0.054306 1.699974
H -0.130494 -0.074275 1.731389
C -1.058558 -2.588064 1.212788
H -0.093360 -1.471345 -0.355913
H -1.862364 -1.432684 -0.419402
C -1.049742 -3.919255 0.467489
H -1.963682 -2.513702 1.830401
H -0.205184 -2.542187 1.902714
H -1.909088 -3.998896 -0.208775
H -1.093646 -4.761949 1.165618
H -0.139215 -0.029348 -0.133175
C -3.539131 1.241224 0.029720
H -2.081471 0.500838 -1.379759
H -2.065195 2.251791 -1.181450
C -4.642824 1.379179 -1.024507
H -3.649396 2.045725 0.766414
H -3.669837 0.289403 0.558047
C -6.035537 1.326532 -0.403378
H -4.537956 0.577094 -1.767591
H -4.513379 2.327203 -1.563996
H -6.198191 0.377259 0.120283
H -6.813236 1.424253 -1.168303
H -6.174195 2.136801 0.321860
C -0.818754 6.162260 1.464975
H 0.012875 4.494668 2.559500
H -1.747084 4.558994 2.576284
H 0.069446 6.347082 0.849383
H -0.787075 6.853885 2.313565
H -1.701286 6.410491 0.863568
N 1.440396 2.127415 3.831519
C 0.462556 0.805463 4.349492
S -0.907418 0.231745 5.065139

NBu₄CN



55

xyz, charge: 0, multiplicity: 1
N 8.950802 -1.629629 -6.001321
C 8.804870 -1.895411 -7.494552
C 8.977017 -2.928049 -5.200520
C 10.249961 -0.849476 -5.843553

C 7.764151 -0.834303 -5.474257
C 10.115502 -3.903401 -5.506201

H 8.007405 -3.394991 -5.384643

H 8.997198 -2.634059 -4.147133

C 11.293127 -3.853307 -4.528033

H 10.480248 -3.785442 -6.533168

H 9.681760 -4.910664 -5.467701

C 12.335253 -4.921822 -4.844847

H 11.768984 -2.865608 -4.547333

H 10.913016 -3.996017 -3.507456

H 11.896452 -5.925897 -4.803034

H 13.165734 -4.889169 -4.131733

H 12.750898 -4.780752 -5.849655

C 10.514391 -0.269686 -4.464577

H 11.045550 -1.541187 -6.136713

H 10.222850 -0.052375 -6.592958

C 11.921519 0.334761 -4.425552

H 10.431970 -1.036588 -3.684519

H 9.786292 0.516541 -4.232451

C 12.237202 0.958163 -3.069292

H 12.010311 1.093918 -5.214384

H 12.658734 -0.446180 -4.656382

H 11.526774 1.757492 -2.828493

H 13.243642 1.389743 -3.056962

H 12.183606 0.209465 -2.269915

C 7.500939 0.489466 -6.168421

H 7.957181 -0.679041 -4.409165

H 6.892796 -1.489913 -5.556233

C 6.397269 1.251579 -5.428393

H 7.188505 0.327430 -7.206991

H 8.405228 1.107999 -6.198247

C 6.067846 2.577002 -6.108672

H 6.716117 1.435325 -4.393321

H 5.494639 0.627374 -5.373605

H 6.951015 3.224608 -6.155583

H 5.283371 3.115996 -5.566573

H 5.718116 2.416953 -7.135151

C 7.592398 -2.704558 -7.918320

H 9.726128 -2.393023 -7.809986

H 8.799355 -0.912085 -7.972841

C 7.556925 -2.798220 -9.447836

H 6.664591 -2.234111 -7.570986

H 7.628167 -3.717849 -7.500139

C 6.361108 -3.604312 -9.945604

H 8.488729 -3.257261 -9.804717

H 7.523859 -1.785407 -9.871078

H 6.386212 -4.629746 -9.558566

H 6.351088 -3.659633 -11.039380

H 5.417351 -3.147516 -9.625256

C 9.917535 2.229885 -8.529223

N 10.268594 1.169849 -8.883873

[CN]



2

xyz, charge: -1, multiplicity: 1

N -2.793857 1.071710 -0.090971

C -1.709193 1.449540 0.153971

[N₃]



3

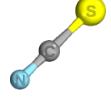
xyz, charge: -1, multiplicity: 1

N 4.592661 -0.560410 -5.980582

N 3.572312 -0.910174 -5.496611

N 5.613007 -0.210646 -6.464557

[NCS]



3

xyz, charge: -1, multiplicity: 1

C -4.147592 -0.235957 0.019214

N -3.693129 0.851265 0.004908

S -4.788149 -1.768409 0.039298

MeCN

6

xyz, charge: 0, multiplicity: 1

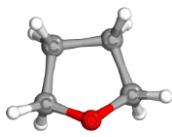
C 2.946713 8.186556 14.545857
 C 2.620733 7.731046 13.207968
 N 2.361164 7.368368 12.142796
 H 2.586516 9.208484 14.692760
 H 4.029800 8.168804 14.695082
 H 2.476983 7.536151 15.288816

Et₂O

15

xyz, charge: 0, multiplicity: 1

C 0.558388 7.522754 6.128818
 C 0.981066 7.115168 7.524461
 H 1.937659 6.585125 7.495444
 H 0.234286 6.453358 7.973006
 H 1.091307 7.998515 8.161426
 O 0.428787 6.348472 5.338832
 H 1.305891 8.200349 5.678331
 H -0.402780 8.066987 6.156741
 C 0.032903 6.632257 4.003709
 C -0.077543 5.323708 3.249973
 H 0.773678 7.296592 3.523365
 H -0.935060 7.165050 4.002032
 H 0.884604 4.803049 3.242326
 H -0.382341 5.505997 2.214578
 H -0.818585 4.671509 3.721518

THF

13

xyz, charge: 0, multiplicity: 1

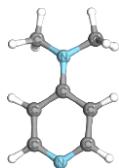
C 0.411222 7.811441 5.984014
 C 0.719111 6.312961 6.148021
 O -0.548355 7.889463 4.930663
 H 1.322375 8.368661 5.703944
 H -0.026272 8.276545 6.871939
 C -0.098110 6.948740 3.956542
 C 0.372396 5.721856 4.757324
 H 0.736469 7.379736 3.376241
 H -0.930385 6.744919 3.276905
 H 1.228294 5.233410 4.282194
 H -0.430430 4.984110 4.840198
 H 0.086909 5.872086 6.923680
 H 1.762536 6.143181 6.429686

Pyridine

11

xyz, charge: 0, multiplicity: 1

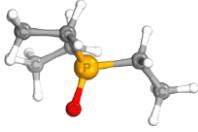
C 4.353718 -4.321812 -0.725173
 C 4.196662 -3.020725 -1.194232
 C 3.365849 -4.855346 0.094174
 C 2.265566 -4.066405 0.408564
 C 2.202432 -2.776796 -0.111485
 N 3.144417 -2.250052 -0.900548
 H 5.231396 -4.900180 -0.997998
 H 4.953840 -2.574189 -1.838488
 H 3.451839 -5.867149 0.480505
 H 1.468828 -4.439993 1.044885
 H 1.352662 -2.133713 0.116775

DMAP

19

xyz, charge: 0, multiplicity: 1

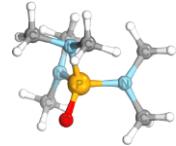
C 2.053708 -0.995674 -4.151777
 C 1.826924 -1.802107 -5.255463
 C 1.128694 -0.004608 -3.086890
 C 0.031159 -1.878912 -3.231036
 C -0.086152 -2.637559 -4.384533
 N 0.780136 -2.622699 -5.404662
 H 2.941520 -0.375223 -4.130110
 H 2.545111 -1.787477 -6.075798
 H -0.724724 -1.976302 -2.461106
 H -0.938867 -3.308956 -4.489708
 N 1.282499 -0.204650 -1.981598
 C 2.537480 0.504332 -1.784748
 H 3.396833 -0.177850 -1.681120
 H 2.467775 1.110122 -0.879232
 H 2.738495 1.184497 -2.620649
 C 0.427576 -0.417496 -0.823946
 H -0.628435 -0.285888 -1.087182
 H 0.669309 0.323745 -0.059797
 H 0.551134 -1.421178 -0.385839

OPt₃

23

xyz, charge: 0, multiplicity: 1

O 1.818422 1.451259 -4.131825
 P 2.334900 0.096244 -3.741574
 C 1.091377 -0.953858 -2.886562
 C 3.806892 0.170618 -2.626682
 C 2.869926 -0.937193 -5.166145
 C 3.736758 -0.146618 -6.149099
 H 1.958352 -1.292570 -5.664352
 H 3.396832 -1.823710 -4.786635
 H 3.208231 0.748825 -6.487226
 H 3.985177 -0.757210 -7.022695
 H 4.677456 0.175832 -5.690836
 C 3.810496 1.429528 -1.755123
 H 4.699936 0.157742 -3.264721
 H 3.841191 -0.743446 -2.018484
 H 3.732426 2.322798 -2.380137
 H 4.733575 1.487515 -1.169520
 H 2.966187 1.445748 -1.060473
 C 0.595931 -0.349798 -1.570486
 H 1.513707 -1.956161 -2.731273
 H 0.257497 -0.601016 -3.593297
 H 1.384846 -0.340896 -0.811780
 H -0.240646 -0.930494 -1.169523
 H 0.256183 0.679333 -1.721739

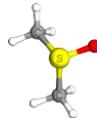
HMPA

29

xyz, charge: 0, multiplicity: 1

O 1.510391 1.696556 -3.866573
 P 2.200788 0.380099 -3.751109
 N 1.429855 -0.913421 -2.976741
 N 3.601578 0.535156 -2.820187
 N 2.472829 -0.272901 -5.287135
 C 2.633237 0.657254 -6.403974
 H 2.259358 0.188380 -7.324262
 H 3.688345 0.933719 -6.568816
 H 2.056493 1.563993 -6.210979
 C 4.262489 1.837477 -2.771711
 H 5.069887 1.921605 -3.518251

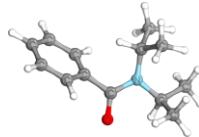
H 4.705169 1.986546 -1.777175
 H 3.529756 2.626506 -2.952530
 C 1.203321 -0.773251 -1.531417
 H 2.088164 -0.358187 -1.044726
 H 0.998392 -1.763472 -1.105578
 H 0.346127 -0.117300 -1.311739
 C 0.243732 -1.469150 -3.643511
 H 0.420594 -1.568595 -4.716231
 H -0.645254 -0.835816 -3.495416
 H 0.033819 -2.462114 -3.226570
 C 3.174672 -1.537108 -5.476590
 H 2.973759 -2.214602 -4.642423
 H 4.265969 -1.407337 -5.569507
 H 2.818578 -2.018933 -6.398082
 C 4.490145 -0.586257 -2.541696
 H 3.930806 -1.525388 -2.537738
 H 4.944752 -0.458257 -1.548734
 H 5.310009 -0.667360 -3.274578

DMSO

10

xyz, charge: 0, multiplicity: 1

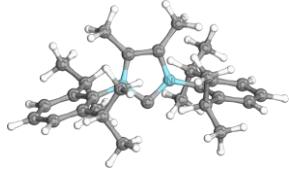
O 1.534856 -1.960244 -4.631953
 S 1.490977 -0.545107 -4.089562
 C 0.802866 -0.665515 -2.381992
 H 0.866892 0.305815 -1.881745
 H 1.360038 -1.431699 -1.834971
 H -0.242552 -0.965772 -2.484735
 C 3.211291 -0.138995 -3.561047
 H 3.813710 -0.079234 -4.470515
 H 3.579101 -0.946478 -2.921208
 H 3.226461 0.821740 -3.037041

DIBA

34

xyz, charge: 0, multiplicity: 1

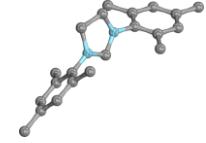
O -0.584038 -2.224236 -1.621757
 C -0.820852 -1.419445 -0.722410
 N -2.076002 -1.249778 -0.188508
 C 0.324416 -0.618472 -0.164401
 C 0.719422 -0.729567 1.170286
 C 1.089178 0.148640 -1.044561
 C 2.211911 0.827994 -0.586426
 C 2.599425 0.720259 0.747014
 C 1.857489 -0.069054 1.621201
 H 2.166881 -0.173173 2.657296
 H 0.143290 -1.349020 1.852048
 H 2.794196 1.434011 -1.274621
 H 3.483758 1.241896 1.101408
 H 0.799475 0.197760 -2.090283
 C -2.395454 -0.057340 0.626636
 C -3.421196 0.869846 -0.031108
 H -3.122924 1.127525 -1.051519
 H -3.485981 1.795191 0.551240
 H -4.422427 0.431402 -0.060348
 C -2.776409 -0.369060 2.079826
 H -1.457915 0.504516 0.665057
 H -2.672257 0.542049 2.679104
 H -2.118811 -1.131443 2.506367
 H -3.808549 -0.711979 2.179692
 C -3.107798 -2.165395 -0.740062
 C -4.219723 -2.509945 0.245715
 H -3.820593 -2.886397 1.191689
 H -4.833544 -3.300410 -0.198287
 H -4.883605 -1.665881 0.453316
 C -3.666609 -1.684359 -2.084018
 H -2.553333 -3.087568 -0.945441
 H -4.234464 -2.496166 -2.551729
 H -2.850674 -1.413965 -2.759039
 H -4.338311 -0.829016 -1.970389

dippNHC

71

xyz, charge: 0, multiplicity: 1

C 4.888097 0.666809 8.677576
 N 5.339119 -0.084512 7.582004
 C 4.033473 1.590623 8.165085
 N 3.998149 1.347423 6.783168
 C 4.790819 0.303771 6.389972
 C 6.236967 -1.192738 7.733521
 C 7.612548 -1.031691 7.497397
 C 8.434870 -2.140854 7.718020
 C 7.920329 -3.353940 8.151655
 C 6.556227 -3.491126 8.364122
 C 5.691269 -2.415643 8.159468
 H 8.584863 -4.196913 8.319195
 H 6.158240 -4.447897 8.687889
 H 9.503132 -2.042072 7.540653
 C 8.270383 0.225209 6.945272
 C 8.038617 0.330670 5.428864
 H 6.968985 0.413921 5.208207
 H 8.559148 1.208340 5.026897
 H 8.418703 -0.559753 4.917161
 C 7.926806 1.538906 7.658054
 H 9.347913 0.069904 7.096707
 H 6.904343 1.868634 7.463883
 H 8.059879 1.452330 8.741764
 H 8.601287 2.325938 7.302251
 C 4.191244 -2.600303 8.298183
 C 3.792980 -3.494758 9.476248
 H 4.074388 -4.539936 9.309018
 H 2.706154 -3.471665 9.608768
 H 4.259406 -3.164932 10.410892
 C 3.630893 -3.150347 6.976585
 H 3.736236 -1.616354 8.461163
 H 2.541932 -3.258958 7.035073
 H 4.063750 -4.134294 6.761146
 H 3.872272 -2.477980 6.147494
 C 5.324384 0.402828 10.072664
 H 4.964265 1.190501 10.739106
 H 6.416325 0.359173 10.147807
 H 4.939589 -0.555920 10.442439
 C 3.202048 2.125244 5.880453
 C 1.824830 1.868084 5.812267
 C 1.049551 2.676859 4.979813
 C 1.629787 3.698384 4.240526
 C 2.999859 3.915682 4.302395
 C 3.816800 3.131870 5.119069
 C 5.313054 3.375064 5.191218
 H -0.019036 2.501228 4.901738
 H 1.011765 4.321400 3.599925
 H 3.441263 4.704410 3.701198
 C 5.915336 3.810796 3.852590
 H 5.622783 3.131721 3.045565
 H 7.008109 3.805695 3.923801
 H 5.613838 4.827050 3.574670
 C 5.650246 4.393532 6.290511
 H 5.780014 2.421225 5.457317
 H 5.333866 4.038752 7.276399
 H 5.153683 5.351460 6.094938
 H 6.731299 4.568925 6.329700
 C 2.120708 0.683238 6.533599
 C -0.188964 0.964686 7.087906
 H -0.207667 1.875676 7.695854
 H -0.516943 0.127682 7.713299
 H -0.928474 1.077896 6.287970
 C 1.193923 -0.525506 5.583933
 H 1.857875 0.422536 7.379326
 H 0.564673 -0.314435 4.711421
 H 0.792106 -1.409599 6.091947
 H 2.204954 -0.752690 5.232044
 C 3.220615 2.646634 8.824444
 H 2.157674 2.376324 8.861985
 H 3.287810 3.598703 8.287265
 H 3.562980 2.805550 9.850163

SiMes

49

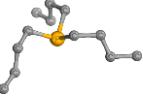
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 C -5.09921 -0.89364 -4.42544
 N -5.56627 -2.27216 -4.09847
 C -5.86538 -0.06126 -3.39859
 N -6.35643 -1.11765 -2.46674
 C -6.22887 -2.37511 -2.93083
 H -4.00929 -0.82791 -4.29663
 H -5.23432 0.66698 -2.87944
 H -6.71842 0.47483 -3.83879
 H -5.33850 -0.62589 -5.45941
 C -7.13570 -0.76300 -1.32783
 C -8.53050 -0.91461 -1.36855
 C -6.49741 -0.25045 -0.19252
 C -9.27140 -0.54011 -0.25051
 C -7.27936 0.11755 0.90420
 C -8.66481 -0.01579 0.89206
 H -10.35313 -0.66002 -0.27196
 H -6.78744 0.50452 1.79467
 C -5.08829 -3.38765 -4.84485
 C -4.07090 -4.19362 -4.31053
 C -5.62330 -3.64512 -6.11246
 C -3.60157 -5.26121 -5.07173
 C -5.11982 -4.72286 -6.84347
 C -4.11200 -5.54058 -6.34039
 H -2.81294 -5.89133 -4.66434
 H -5.53775 -4.93267 -7.82638
 C -3.50664 -3.91842 -2.94538
 H -4.26773 -4.07936 -2.17454
 H -2.65027 -4.56691 -2.74300
 H -3.18028 -2.87645 -2.84971
 C -3.60470 -6.71969 -7.12592
 H -4.05544 -7.65208 -6.76454
 H -3.84598 -6.62523 -8.18831
 H -2.51944 -6.82437 -7.02854
 C -6.73011 -2.79869 -6.67810
 H -7.30158 -3.36003 -7.42217
 H -7.41294 -2.46576 -5.89063
 H -6.33896 -1.90314 -7.17770
 C -9.20849 -1.48408 -2.58256
 H -8.91928 -0.94486 -3.49199
 H -8.91654 -2.52891 -2.73243
 H -10.29536 -1.42745 -2.48081
 C -9.49620 0.40552 2.07358
 H -10.28148 -0.32659 2.28752
 H -8.88289 0.52142 2.97142
 H -9.99004 1.36598 1.88192
 C -5.00051 -0.12022 -0.13278
 H -4.65585 -0.11633 0.90473
 H -4.51230 -0.94480 -0.66098
 H -4.65761 0.81571 -0.59212

iPrNHC

33

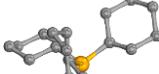
xyz, charge: 0, multiplicity: 1
 C -4.90695 -0.86512 -4.24948
 N -5.27402 -2.19202 -4.00577
 C -5.61027 -0.11138 -3.35348
 N -6.37192 -1.01519 -2.60662
 C -6.18801 -2.30954 -2.99889
 C -7.35495 -0.71629 -1.55455
 C -4.85483 -3.39554 -4.73946
 C -5.48047 -3.45319 -6.13282
 H -5.08999 -2.67190 -6.79252
 H -6.56615 -3.33856 -6.06470
 H -5.26135 -4.42022 -6.59790
 C -3.33859 -3.58224 -4.76596
 H -5.29092 -4.20171 -4.13860
 H -2.91345 -3.44355 -3.76717
 H -2.84529 -2.89221 -5.45628
 H -3.10655 -4.59920 -5.09839
 C -6.76824 0.09462 -0.40011

H -5.81377 -0.32979 -0.07383
 H -6.61352 1.14455 -0.66413
 H -7.46140 0.06899 0.44690
 C -8.62403 -0.08886 -2.13071
 H -7.61028 -1.71415 -1.18009
 H -8.44405 0.91568 -2.52655
 H -9.38502 -0.00540 -1.34750
 H -9.02177 -0.71271 -2.93636
 C -5.55347 1.36323 -3.14096
 H -6.54783 1.81221 -3.06010
 H -4.99537 1.62939 -2.23534
 H -5.05020 1.84134 -3.98507
 C -3.89793 -0.41011 -5.24835
 H -2.87297 -0.64433 -4.93682
 H -4.05412 -0.85817 -6.23410
 H -3.96014 0.67395 -5.37218

PBu₃

40

xyz, charge: 0, multiplicity: 1
 P 5.68471 -0.37930 4.78930
 C 5.85787 -2.20725 5.17415
 H 5.21472 -2.78062 4.48998
 H 6.89616 -2.48767 4.94974
 C 5.54337 -2.58146 6.62696
 H 5.65858 -3.66934 6.73866
 H 4.48812 -2.36939 6.84826
 C 6.42755 -1.88250 7.66005
 H 6.29232 -0.79581 7.58578
 H 7.48232 -2.07080 7.41588
 C 5.71584 -0.49183 2.91706
 H 4.90789 -1.16030 2.58409
 H 5.48468 0.50968 2.52903
 C 7.05505 -0.95512 2.33241
 H 6.95847 -1.00381 1.23809
 H 7.27131 -1.98204 2.65825
 C 8.23808 -0.05265 2.68411
 H 8.37002 -0.02600 3.77340
 H 8.00404 0.97859 2.38507
 C 3.82931 -0.22689 5.02219
 H 3.33655 -1.09616 4.56170
 H 3.63372 -0.28166 6.10193
 C 3.22579 1.07038 4.47211
 H 2.13993 1.04914 4.64441
 H 3.35351 1.11009 3.38143
 C 3.80005 2.34276 5.09564
 H 3.69788 2.28465 6.18826
 H 4.87842 2.39557 4.89804
 C 9.53322 -0.50635 2.01471
 H 9.43537 -0.50411 0.92290
 H 10.37085 0.14777 2.27687
 H 9.79709 -1.52530 2.32073
 C 6.13308 -2.34657 9.08456
 H 6.29851 -3.42498 9.19079
 H 6.77184 -1.83676 9.81279
 H 5.09048 -2.14394 9.35560
 C 3.11411 3.60404 4.57604
 H 3.22739 3.69297 3.48940
 H 2.04037 3.58891 4.79621
 H 3.53608 4.50631 5.03001

PCy₃

52

xyz, charge: 0, multiplicity: 1
 P 5.13584 -1.77786 12.57622
 C 4.75592 -1.68055 10.73485
 H 3.78307 -2.15114 10.52297
 C 5.82352 -2.40521 9.90008
 H 5.85060 -3.47236 10.14919
 H 6.81406 -1.99527 10.14862
 C 5.56611 -2.25005 8.39698
 H 4.61964 -2.74912 8.14150
 H 6.35515 -2.76101 7.83084
 C 5.48318 -0.77897 7.98975
 H 5.26731 -0.69010 6.91776

H 6.45946 -0.30126 8.15838
C 4.42017 -0.05109 8.81256
H 3.42957 -0.46720 8.57601
H 4.38856 1.01259 8.54480
C 4.68466 -0.20252 10.31351
H 5.63985 0.28146 10.56689
H 3.90797 0.32576 10.87928
C 3.57757 -0.99558 13.31035
H 3.78958 0.08047 13.18433
C 2.21056 -1.25780 12.66036
H 1.93265 -2.31146 12.78186
H 2.24353 -1.06421 11.58246
C 1.13003 -0.37739 13.30200
H 0.15679 -0.58088 12.83731
H 1.36169 0.67923 13.10078
C 1.05506 -0.59728 14.81392
H 0.73291 -1.63056 15.01159
H 0.29785 0.06024 15.25900
C 2.41704 -0.36375 15.47003
H 2.36245 -0.56625 16.54720
H 2.69845 0.69364 15.35951
C 3.49581 -1.24014 14.82551
H 4.47124 -1.05658 15.29453
H 3.24657 -2.29445 15.00945
C 5.05852 -3.63470 12.88952
H 5.78463 -4.01280 12.15020
C 5.64234 -4.00284 14.26531
H 4.96752 -3.66337 15.06209
H 6.59517 -3.48134 14.41706
C 5.83490 -5.51827 14.38863
H 6.22835 -5.76770 15.38221
H 6.58974 -5.84472 13.65798
C 4.52716 -6.26947 14.12988
H 3.80233 -6.01271 14.91680
H 4.68979 -7.35276 14.19169
C 3.93830 -5.89583 12.76801
H 2.98408 -6.41420 12.60867
H 4.61919 -6.23280 11.97234
C 3.74008 -4.37944 12.65120
H 3.32088 -4.12691 11.66936
H 3.00503 -4.06431 13.40409

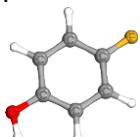
DABCO



20

xyz, charge: 0, multiplicity: 1
C 2.61634 3.22035 3.52392
C 1.97133 1.95160 2.88361
N 2.57689 3.12792 4.99420
H 2.08657 4.13387 3.22518
H 3.66483 3.33348 3.22021
C 1.17419 2.96996 5.41758
C 0.56435 1.67302 4.79965
H 0.62227 3.86344 5.09910
H 1.14996 2.94147 6.51441
N 1.55194 1.00569 3.93298
H -0.32420 1.89604 4.19539
H 0.26267 0.96151 5.57881
H 2.67631 1.43460 2.22018
H 1.08672 2.20928 2.28743
C 3.33091 1.93203 5.41021
C 2.72944 0.65404 4.74643
H 3.29133 1.87125 6.50525
H 4.38128 2.07512 5.12619
H 2.41669 -0.08000 5.49995
H 3.45886 0.16211 4.09044

p-F-PhOH

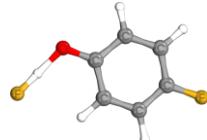


13

xyz, charge: 0, multiplicity: 1
C -5.012230 3.715799 -0.000000
C -5.023132 2.332543 0.000000
C -3.787010 4.377197 -0.000000
C -2.598702 3.647488 -0.000000

C -2.633212 2.253232 -0.000000
C -3.853362 1.589484 -0.000000
H -5.947796 4.265027 -0.000000
F -6.212876 1.684767 0.000000
H -3.762942 5.464734 -0.000000
H -1.699083 1.701113 -0.000000
H -3.902867 0.505637 0.000000
O -1.365280 4.242178 0.000000
H -1.474446 5.199542 0.000000

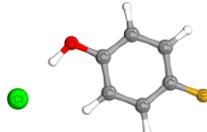
p-F-PhOH-F



14

xyz, charge: -1, multiplicity: 1
C -4.806586 3.807589 0.000000
C -4.968666 2.431750 -0.000000
C -3.525121 3.4346818 0.000000
C -2.362466 3.519240 0.000000
C -2.595655 2.113436 0.000000
C -3.879160 1.577454 -0.000000
H -5.683739 4.450598 -0.000000
F -6.243066 1.900147 -0.000000
H -3.380723 5.423891 0.000000
H -1.728141 1.458300 0.000000
H -4.039799 0.501812 -0.000000
O -1.154689 3.994613 -0.000000
H -1.075509 5.325628 0.000000
F -1.002571 6.376134 -0.000000

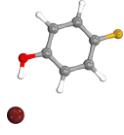
p-F-PhOH-Cl



14

xyz, charge: -1, multiplicity: 1
C -4.784228 3.826373 -0.000535
C -4.973609 2.454891 0.000320
C -3.491316 4.339098 -0.001207
C -2.384825 3.469092 -0.000963
C -2.613611 2.083056 -0.000004
C -3.908305 1.572498 0.000604
H -5.644923 4.489157 -0.000675
F -6.251555 1.954844 0.000931
H -3.305399 5.410829 -0.001698
H -1.756225 1.416302 0.000230
H -4.091934 0.501977 0.001328
O -1.117468 3.899019 -0.001855
H -1.077391 4.916418 -0.000462
Cl -1.045101 6.893857 0.003985

p-F-PhOH-Br



14

xyz, charge: -1, multiplicity: 1
Br 2.99902 0.49887 0.68326
O 0.07508 -0.30769 -0.28933
C -0.10541 -1.63438 -0.19601
C 0.88684 -2.51385 0.27059
C 0.63236 -3.87954 0.34158
C -0.60423 -4.36230 -0.05080
C -1.60124 -3.52261 -0.51441
C -1.34843 -2.15617 -0.58604
H 1.84783 -2.10210 0.57127
H -2.56025 -3.93472 -0.81424
H -2.10990 -1.47030 -0.94539
H 1.39164 -4.56824 0.70050
F -0.85042 -5.70905 0.02066
H 0.99810 -0.04617 0.01669

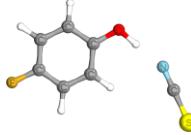
p-F-PhOH-N₃



16

xyz, charge: -1, multiplicity: 1
C 2.808972 3.489769 0.745826
C 1.465803 3.714114 0.498385
C 3.749909 3.843095 -0.216085
C 3.343417 4.423603 -1.434480
C 1.969186 4.637272 -1.647701
C 1.029539 4.283456 -0.684912
H 3.115706 3.041413 1.686552
F 0.541460 3.363120 1.449517
H 4.810110 3.680172 -0.046190
H 1.656713 5.086667 -2.585796
H -0.032012 4.446936 -0.846991
O 4.198372 4.778448 -2.393839
N 6.634195 4.210396 -1.679465
N 7.559267 4.932715 -1.889379
N 8.484958 5.614128 -2.081049
H 5.188545 4.576695 -2.118114

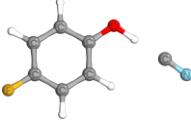
p-F-PhOH-NCS



16

xyz, charge: -1, multiplicity: 1
C 3.066512 3.552645 0.621295
C 1.684818 3.574750 0.548433
C 3.813558 4.072177 -0.431167
C 3.171816 4.615163 -1.559230
C 1.767711 4.620353 -1.598347
C 1.021333 4.100898 -0.546001
H 3.555641 3.132038 1.494802
F 0.951652 3.062689 1.586212
H 4.898744 4.063555 -0.392813
H 1.276303 5.040232 -2.471014
H -0.064383 4.102384 -0.570921
O 3.833417 5.129897 -2.602910
N 6.454739 4.983699 -2.275125
C 7.438832 4.696274 -1.690568
S 8.811830 4.300605 -0.885907
H 4.841613 5.074643 -2.470460

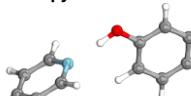
p-F-PhOH-CN



15

xyz, charge: -1, multiplicity: 1
C -4.531939 3.997149 0.000000
C -4.864962 2.653696 -0.000000
C -3.191768 4.369677 0.000000
C -2.171031 3.393346 0.000000
C -2.555572 2.037162 -0.000000
C -3.896035 1.665926 -0.000000
H -5.317895 4.747161 0.000001
F -6.188951 2.290118 -0.000000
H -2.913316 5.418844 0.000001
H -1.774081 1.282744 -0.000000
H -4.190596 0.620198 -0.000001
O -0.877496 3.690678 0.000000
C -0.378950 6.316052 0.000000
N -0.230629 7.473834 -0.000001
H -0.691980 4.753494 0.000000

p-F-PhOH-pyridine



24

xyz, charge: 0, multiplicity: 1

C -1.375291 5.486940 0.012714
C -1.424102 4.103038 0.053686
C -0.138077 6.118337 0.026092
C 1.041099 5.367426 0.079976
C 0.963062 3.970063 0.120093
C -0.275833 3.334170 0.107135
H -2.297295 6.057958 -0.028900
F -2.633803 3.484521 0.040834
H -0.066994 7.200926 -0.004847
H 1.873050 3.378242 0.160953
H -0.351465 2.251930 0.137753
O 2.216996 6.041475 0.090345
H 2.979781 5.412351 0.137294
C 5.100022 3.974432 -0.883655
C 6.253311 3.199506 -0.859027
N 4.468590 4.371608 0.226891
C 4.973874 4.003313 1.409648
C 6.121861 3.229726 1.531214
C 6.773439 2.820583 0.373283
H 4.659241 4.293387 -1.826913
H 6.730330 2.902066 -1.787384
H 4.432590 4.345204 2.290541
H 6.493830 2.956514 2.513486
H 7.673323 2.214983 0.430418

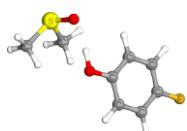
p-F-PhOH-DMAP



32

xyz, charge: 0, multiplicity: 1
C -1.204299 5.229344 -0.144140
C -1.189311 3.933416 0.344890
C 0.002846 5.861977 -0.413127
C 1.217757 5.201523 -0.193415
C 1.202369 3.891612 0.304529
C -0.006331 3.254629 0.573749
H -2.151939 5.731540 -0.310954
F -2.370431 3.312377 0.607785
H 0.023867 6.876729 -0.797927
H 2.139187 3.372640 0.485174
H -0.032198 2.240647 0.960199
O 2.358705 5.869920 -0.475820
H 3.158130 5.296693 -0.307960
C 5.009327 3.441641 -0.966115
C 6.167667 2.695746 -0.871804
N 4.632186 4.366361 -0.071704
C 5.452510 4.559752 0.970615
C 6.634449 3.872731 1.166856
C 7.032539 2.892880 0.228905
H 4.338819 3.293116 -1.811878
H 6.390644 1.974872 -1.648552
H 5.141328 5.315257 1.690752
H 7.234587 4.102530 2.038467
N 8.182912 2.177289 0.378605
C 9.085789 2.472979 1.481143
H 8.590479 2.333823 2.450889
H 9.470308 3.502514 1.433871
H 9.936129 1.790783 1.441474
C 8.596576 1.239699 -0.654850
H 7.843371 0.455642 -0.806841
H 9.524174 0.754347 -0.348389
H 8.773863 1.740091 -1.618298

p-F-PhOH-DMSO

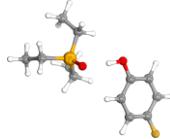


23

xyz, charge: 0, multiplicity: 1
C 1.583846 3.584272 1.070112
C 0.876483 2.810535 0.166977
C 2.461533 4.551867 0.588948
C 2.622243 4.731127 -0.789785
C 1.895244 3.935795 -1.680007
C 1.015252 2.972137 -1.201712
H 1.442796 3.434884 2.135803

F 0.019837 1.867140 0.635160
H 3.021060 5.176279 1.280067
H 2.024096 4.088556 -2.746966
H 0.440191 2.349282 -1.879284
O 3.470149 5.657131 -1.311348
H 4.014194 6.055939 -0.587182
O 5.275888 6.435701 0.546086
S 6.561657 5.646404 0.263282
C 6.903138 5.864035 -1.527012
H 6.000737 5.605900 -2.089669
H 7.748100 5.229907 -1.812293
H 7.158346 6.916032 -1.674902
C 6.077125 3.877426 0.169560
H 5.726625 3.590839 1.163956
H 6.954313 3.283258 -0.104531
H 5.272157 3.760943 -0.561806

p-F-PhOH-OPEt₃



36

xyz, charge: 0, multiplicity: 1
C 2.719569 2.350120 -0.537554
C 1.773851 2.613208 0.437370
C 3.108524 3.374446 -1.396898
C 2.550668 4.652900 -1.266577
C 1.595383 4.889230 -0.272335
C 1.201478 3.866377 0.581941
H 3.139026 1.353056 -0.627366
F 1.392646 1.613714 1.275410
H 3.843509 3.189987 -2.175490
H 1.164762 5.881841 -0.184919
H 0.458954 4.033911 1.355596
O 2.897986 5.685962 -0.207290
H 3.686593 5.433206 -2.617623
O 5.277854 5.152963 -3.209998
P 6.305158 5.842777 -2.330987
C 7.963545 5.741388 -3.117072
C 6.389592 5.048685 -0.676079
C 5.872999 7.613953 -2.098361
C 6.989710 8.593892 -1.731054
H 5.057725 7.638415 -1.363033
H 5.417715 7.897170 -3.056841
H 7.462012 8.350316 -0.775962
H 6.582022 9.606188 -1.647641
H 7.769399 8.619058 -2.498452
C 9.198420 5.872360 -2.222382
H 7.975433 6.489215 -3.921183
H 7.954695 4.762637 -3.614494
H 9.249241 6.842032 -1.720574
H 10.106519 5.764584 -2.823825
H 9.220357 5.092510 -1.455411
C 6.935972 5.864390 0.497980
H 6.949245 4.112118 -0.804187
H 5.350626 4.751464 -0.475477
H 7.974385 6.172644 0.350168
H 6.895767 5.265359 1.412968
H 6.334759 6.761683 0.671366

H 2.885259 5.900271 -1.731053
H 2.650689 1.799544 -0.506716

H 0.920558 2.412181 1.201554
O 3.816349 3.424205 -2.115377

H 4.225802 4.219741 -2.536254
O 5.121139 5.615339 -3.009609

P 6.286543 5.968627 -2.124387
N 7.672080 5.901683 -3.071269

N 6.699541 5.001494 -0.821738
N 6.014856 7.451100 -1.389304

C 6.925519 8.042099 -0.414724
H 7.584299 7.279625 0.007747

H 6.351811 8.491388 0.408791
H 7.546278 8.834378 -0.861964

C 8.967664 6.428920 -2.647004
H 9.019771 7.527806 -2.650665

H 9.752645 6.045900 -3.307730
H 9.190392 6.078778 -1.633739

C 5.846603 4.994159 0.375999
H 6.472724 4.839570 1.264634

H 5.100898 4.188474 0.328759
H 5.314622 5.941737 0.479999

H 7.467840 5.995261 -4.056847
C 7.284143 3.676496 -1.076315

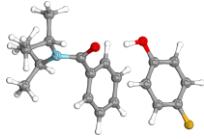
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H 6.500989 2.920590 -1.234637

H 7.885306 3.381494 -0.207022
C 5.106629 8.411923 -2.010930

H 4.443590 7.898600 -2.709975
H 5.654355 9.198591 -2.553943

H 4.495294 8.895964 -1.236865

p-F-PhOH-DIBA

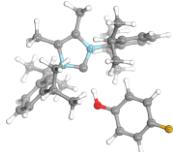


47

xyz, charge: 0, multiplicity: 1
C -0.209140 3.197678 0.836782
C 0.222286 2.863423 -0.435934
C 0.117604 4.447228 1.351018
C 0.864417 5.351206 0.590401
C 1.267543 4.996480 -0.701048
C 0.952970 3.742923 -1.214900
H -0.787008 2.484114 1.415460
F -0.083045 1.635813 -0.935432
H -0.194635 4.735597 2.349887
H 1.842019 5.701424 -1.294965
H 1.269042 3.448720 -2.210676
O 1.172113 6.548553 1.148767
H 1.905466 6.971117 0.649052
C 4.839719 4.772668 -0.847953
C 4.678011 3.401777 -0.686262
C 4.520551 5.643155 0.197016
C 4.043283 5.129897 1.403219
C 3.897566 3.757022 1.565281
C 4.208156 2.891827 0.520720
H 5.186826 5.173286 -1.796697
H 4.904998 2.730660 -1.509523
H 3.758475 5.810123 2.200605
H 3.515431 3.364037 2.502502
H 4.068860 1.821672 0.641340
C 4.509654 7.128057 -0.016086
N 5.673664 7.783396 -0.275913
C 5.530552 9.237508 -0.562905
C 5.433263 10.073250 0.717333
H 5.105568 11.086848 0.462793
H 6.394208 10.151620 1.232893
H 4.694682 9.643850 1.399171
C 6.578904 9.785230 -1.525299
H 4.558730 9.310894 -1.062106
H 6.618606 9.208773 -2.453505
H 7.581037 9.825505 -1.089916
H 6.296895 10.811741 -1.780295
C 6.977016 7.118784 -0.035093
C 7.806386 7.804676 1.052198
H 8.183436 8.781620 0.738395
H 8.673946 7.176628 1.279807
H 7.226269 7.931624 1.970806

C 7.800238 6.872636 -1.304624
H 6.707860 6.134276 0.357769
H 8.318857 7.768627 -1.651923
H 7.171780 6.500008 -2.118101
H 8.560063 6.113444 -1.090691
O 3.416346 7.718433 0.016640

p-F-PhOH-dippNHC



84

xyz, charge: 0, multiplicity: 1
C 7.045455 3.075197 -1.557678
C 5.855023 2.568030 -1.069708
C 8.199923 2.945071 -0.790552
C 8.161068 2.297074 0.453203
C 6.937623 1.795926 0.917711
C 5.780841 1.934578 0.160162
H 7.069221 3.576582 -2.520441
F 4.723336 2.698761 -1.814783
H 9.134171 3.360739 -1.153791
H 6.912622 1.296812 1.881471
H 4.828186 1.551155 0.512040
O 9.251590 2.149495 1.233615
H 10.114155 2.246752 0.688155
C 10.517397 4.709656 -3.752082
C 11.249254 3.752318 -0.49397
C 10.161726 5.913861 -3.157809
C 10.536184 6.185330 -1.850279
C 11.269717 5.260105 -1.104018
C 11.606168 4.048475 -1.724138
C 11.566317 2.411115 -3.683630
H 10.210786 4.506604 -4.773784
H 9.584357 6.643696 -3.718236
H 10.246754 7.128208 -1.395365
C 11.643623 5.578393 0.331310
C 12.453612 -1.240816 3.590329
C 12.121792 -1.637829 2.303666
C 12.612732 0.555876 1.494336
C 12.189334 -0.754421 1.221873
C 12.856620 0.064457 3.831760
C 12.944168 0.987427 2.789773
H 13.098352 0.372246 4.844313
C 11.718608 -1.276511 -0.129042
C 13.299069 2.435056 3.078290
H 12.391879 -1.951147 4.409868
H 11.793796 -2.658586 2.123276
C 10.331441 1.499396 -3.627958
H 12.354444 1.931294 -3.092264
C 12.084495 2.542278 -5.120486
C 12.493928 6.853318 0.417852
H 12.245716 4.750530 0.722309
C 10.388020 5.689987 1.207548
H 13.386872 6.782161 -0.212385
H 12.815595 7.021405 1.451425
H 11.928095 7.735486 0.099370
H 9.741926 6.506452 0.866096
H 10.668448 5.894595 2.246458
H 9.806166 4.763439 1.187270
H 9.977453 1.372679 -2.600958
H 10.566281 0.510814 -4.038545
H 9.510021 1.928547 -4.213256
H 11.301041 2.885096 -5.804389
H 12.427330 1.567544 -5.483143
H 12.919885 3.247564 -5.186266
C 12.041014 3.189579 3.538028
H 13.638672 2.901358 2.145991
C 14.429137 2.586568 4.102541
C 10.260927 -0.865556 -0.390609
H 11.729013 -2.370393 -0.023655
C 12.606742 -0.956083 -1.337082
H 9.613692 -1.151310 0.444142
H 9.890532 -1.348694 -1.302093
H 10.179340 0.217696 -0.513026
H 12.536766 0.091663 -1.638017
H 12.282857 -1.564014 -2.189009
H 13.657964 -1.192158 -1.140811

H 14.111145 2.287302 5.106732
H 14.738156 3.635453 4.161901
H 15.304339 1.985502 3.833213
H 11.238086 3.117345 2.798312
H 12.271505 4.248281 3.705210
H 11.667126 2.768168 4.478230
N 12.314810 3.059460 -0.964303
C 13.695429 2.845550 -1.042546
C 13.967933 1.856011 -0.148764
N 12.740055 1.513284 0.430136
C 11.707926 2.257899 -0.048342
C 14.573317 3.622744 -1.954443
H 14.339238 3.428157 -3.007832
H 15.620639 3.362954 -1.785582
H 14.456608 4.700238 -1.791486
C 15.245787 1.205983 0.238186
H 15.521167 1.444113 1.272678
H 16.054774 1.539450 -0.415634
H 15.174010 0.115416 0.168337

H -3.59658 1.49159 3.68114
H -3.34580 2.59993 2.32654
H -2.86051 3.06676 3.96847
C 0.40798 -0.31827 0.39273
H -0.39856 -0.99500 0.09186
H 1.31464 -0.90916 0.54707
H 0.58501 0.36572 -0.44558
C 1.86155 1.69573 4.75754
H 2.58423 0.87636 4.81733
H 1.46131 1.87838 5.75893
H 2.40880 2.59488 4.44746

p-F-PhOH-^{iPr}NHC



46

xyz, charge: 0, multiplicity: 1
N -0.09352 1.45295 0.35349
C 0.48719 0.26996 0.67487
C 0.75383 2.25272 -0.41660
C 1.90952 1.53850 -0.57328
N 1.71711 0.33457 0.10695
C 0.44415 3.62940 -0.89534
H -0.54294 3.69668 -1.36117
H 1.17802 3.93469 -1.64475
H 0.48048 4.36509 -0.08351
C 3.16371 1.94968 -1.26501
H 2.97078 2.81957 -1.89704
H 3.56043 1.16056 -1.90921
H 3.95141 2.22843 -0.55556
C -1.48656 1.71706 0.76037
C -1.63698 3.01988 1.54247
C 2.23663 4.25075 1.78883
C 2.89642 3.69005 0.70895
H 2.63160 2.70727 -1.19321
H -0.97309 3.86187 0.81693
H 0.37510 4.77538 2.70905
H 3.98145 3.65847 0.69630
F 2.96886 4.74671 2.82711
C -2.14632 1.30846 -0.37056
N -2.23006 0.79180 0.86389
C -3.31987 -0.20918 1.01111
C -4.13443 0.03349 -0.26094
H -3.89404 -0.04306 1.92705
H -2.88817 -1.21933 1.05086
N -3.17968 0.82953 -1.07972
H -5.04169 0.62634 -0.07590
H -4.42390 -0.89103 -0.76819
H -0.82118 2.26506 -0.99028
C -3.52764 1.27779 -2.38870
C -3.53455 0.35572 -3.44353
C -3.90416 0.80018 -4.71279
C -4.25305 2.12710 -4.95143
C -4.22909 3.02142 -3.88199
C -3.87185 2.62245 -2.59565
H -4.49755 4.06223 -4.05209
H -3.90352 0.09097 -5.53831
C -4.60935 2.59622 -6.33565
H -4.98152 1.77368 -6.95279
H -3.73088 3.01721 -6.83975
H -5.37430 3.37798 -6.30591
C -3.13285 -1.07873 -3.23601
H -2.35090 -1.16654 -2.47574
H -2.75767 -1.50885 -4.16826
H -3.98224 -1.69524 -2.91504
C -3.84509 3.62547 -1.47631
H -4.41542 4.51745 -1.74831
H -2.81611 3.93178 -1.25504
H -4.26245 3.21580 -0.55069
C -1.22276 0.95634 1.86302
C 0.06325 0.44223 1.64236
C 1.03965 0.67756 2.60780
C 0.76715 1.39046 3.77346
C -0.52968 1.85732 3.97701
C -1.53602 1.65871 3.03362
H 2.04739 0.30583 2.43423
H -0.75814 2.42089 4.87976
C -2.90694 2.23478 3.26089

p-F-PhOH-PBu₃



53

xyz, charge: 0, multiplicity: 1
P -2.16841 3.56209 2.22918
C -1.11364 2.03406 1.99758
C -3.61187 3.20724 1.10225

C -2.94001 3.19466 3.88566
 C -4.60560 4.36725 1.01508
 H -4.11685 2.29091 1.44231
 H -3.20529 2.99319 0.10556
 C -2.01753 3.53720 5.05743
 H -3.23786 2.13601 3.92219
 H -3.86115 3.78800 3.95950
 C 0.06827 2.23970 1.04335
 H -1.74922 1.20208 1.66031
 H -0.73643 1.76186 2.99156
 C -0.33756 2.54622 -0.39717
 H 0.69764 3.05822 1.41946
 H 0.69323 1.33536 1.05444
 C -2.64604 3.22001 6.41545
 H -1.75466 4.60307 5.01412
 H -1.06937 2.98848 4.96570
 C -5.77773 4.06887 0.07853
 H -4.08385 5.27085 0.66963
 H -4.99770 4.60656 2.01355
 C -6.76477 5.23028 -0.01426
 H -6.29766 3.16613 0.42737
 H -5.38742 3.83161 -0.92068
 H -7.18859 5.46548 0.96854
 H -7.59453 4.99768 -0.68902
 H -6.27276 6.13517 -0.38809
 C -1.72568 3.56982 7.58272
 H -2.90523 2.15301 6.45363
 H -3.59254 3.76969 6.51090
 H -1.47470 4.63636 7.57883
 H -2.19373 3.33915 8.54493
 H -0.78634 3.00862 7.52396
 C 0.86818 2.76481 -1.30685
 H -0.96184 1.72688 -0.78297
 H -0.95837 3.45233 -0.41492
 H 1.51147 1.87744 -1.32952
 H 0.56086 2.98578 -2.33383
 H 1.46910 3.60953 -0.95171
 H -0.58682 5.33406 2.12844
 O 0.28988 5.77761 2.07769
 C 0.57466 6.06979 0.78009
 C 1.91136 6.29299 0.43971
 C -0.41401 6.16897 -0.20435
 C -0.06786 6.46557 -1.51943
 C 1.26378 6.67854 -1.82905
 C 2.25974 6.60513 -0.86911
 H 2.66907 6.21638 1.21305
 H -1.45580 6.01681 0.06407
 H -0.82188 6.54211 -2.29625
 H 3.29315 6.78134 -1.14970
 F 1.60356 6.97156 -3.10978

65
 xyz, charge: 0, multiplicity: 1
 P -1.43157 3.32317 1.94789
 C -0.95683 1.65004 1.23301
 C -3.10344 3.75784 1.18271
 C -2.02850 3.00786 3.70625
 C -3.38721 5.24572 1.46261
 C -0.89786 3.11501 4.74262
 C -0.28531 1.85212 -0.14066
 C 0.11299 0.51646 -0.77558
 H -0.93088 2.40919 -0.82500
 H 0.61755 2.46365 0.00780
 C -1.45724 3.07124 6.16848
 H -0.33599 4.04323 4.58092
 H -0.18462 2.29068 4.61239
 C -4.77366 5.65726 0.96015
 H -2.62650 5.85155 0.95098
 H -3.29502 5.46945 2.53286
 C -4.91674 5.36726 -0.53399
 H -5.54561 5.10456 1.51608
 H -4.93779 6.72298 1.16120
 C -2.30401 1.81774 6.39603
 H -0.63559 3.11618 6.89391
 H -0.07856 3.96286 6.33861
 C 0.106125 -0.26290 0.13502
 H -0.79216 -0.08054 -0.96051
 H 0.57920 0.69478 -1.75249
 H -0.08537 5.25274 1.83818
 O 0.18748 6.20118 1.83792
 C -0.06553 6.74062 0.61638
 C -0.23753 8.12503 0.53331
 C -0.15080 5.96519 -0.54527
 C -0.43105 6.56271 -1.77112
 C -0.60128 7.93477 -1.82240
 C -0.50049 8.72798 -0.69087
 H -0.16757 8.71678 1.44049
 H 0.00615 4.89235 -0.48584
 H -0.50275 5.97460 -2.68063
 H -0.63620 9.80166 -0.77194
 F -0.86916 8.52191 -3.01673
 C -4.63149 3.89429 -0.82671
 H -4.20231 5.99179 -1.09046
 H -5.91972 5.63916 -0.88523
 C -3.24535 3.47898 -0.31843
 H -4.70321 3.69550 -1.90334
 H -5.39544 3.27187 -0.33746
 H -2.48186 4.04686 -0.86989
 H -3.08432 2.41550 -0.53068
 H -3.86711 3.15600 1.70098
 C 0.44912 -0.44268 1.52374
 H 2.00752 0.28994 0.22593
 H 1.30469 -1.23764 -0.30517
 C 0.03376 0.89835 2.13975
 H 1.15596 -0.95127 2.19122
 H -0.43730 -1.08976 1.44939
 H 0.92552 1.52608 2.28886
 H -0.39211 0.71921 3.13000

H -1.86155 1.03271 1.10870
 C -3.43427 1.72652 5.36917
 H -2.71321 1.81298 7.41380
 H -1.66381 0.92720 6.30798
 C -2.88493 1.75360 3.93748
 H -4.12065 2.57451 5.50920
 H -4.02247 0.81354 5.52584
 H -2.68547 3.87707 3.87932
 H -3.71359 1.71051 3.22051
 H -2.28521 0.84913 3.77534

p-F-PhOH-DABCO



33
 xyz, charge: 0, multiplicity: 1
 N 4.03078 2.34807 0.10203
 O 1.77643 1.26472 -1.00382
 C 1.36216 0.05984 -0.55052
 C 2.08344 -0.68908 0.38771
 C 1.61053 -1.92503 0.82085
 C 0.41789 -2.40219 0.30855
 C -0.31590 -1.68465 -0.62217
 C 0.15896 -0.45159 -1.05040
 H 3.02186 -0.30521 0.77785
 H -1.24745 -2.09116 -1.00285
 H -0.39458 0.13345 -1.77796
 H 2.16071 -2.51490 1.54698
 F -0.04647 -3.60781 0.72859
 C 3.82483 2.72709 1.51745
 C 4.09755 3.58094 -0.71708
 C 5.31555 1.62571 -0.01921
 C 5.03621 3.57076 2.01704
 H 3.70062 1.80858 2.10441
 H 2.88375 3.28665 1.57912
 C 5.22264 4.51187 -0.17552
 H 3.11356 4.06379 -0.68305
 H 4.28102 3.28025 -1.75557
 C 6.48733 2.56397 0.40142
 H 5.41801 1.28198 -1.05559
 H 5.26466 0.73623 0.62092
 N 5.96476 3.84405 0.90729
 H 5.59151 3.04218 2.80129
 H 4.70565 4.52883 2.43586
 H 4.80795 5.44533 0.22369
 H 5.93226 4.77862 -0.96796
 H 7.14715 2.77887 -0.44776
 H 7.09964 2.10502 1.18699
 H 2.62547 1.54522 -0.55077

p-F-PhOH-PCy₃



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