# **Electronic Supporting Information**

# Facile N-C Bond Cleavage and Arene Reduction by a Transient Uranium (II) Complex

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#### **A. Experimental Procedures**

**General Considerations**. Unless otherwise noted, all manipulations were carried out at ambient temperature under an argon atmosphere using Schlenk techniques and an MBraun glovebox equipped with a purifier unit. The water and oxygen levels were always kept at less than 0.1 ppm. Glassware was dried overnight at 140 °C before use. Syntheses were performed using glass coated stirrer bars. Volatiles transfer experiments to release and trap ammonia were performed using a two-bulb apparatus J. Young sealable distillation apparatus.

**NMR experiments** were carried out using NMR tubes adapted with J-Young valves. NMR spectra were recorded on a Bruker 400 MHz or 600 MHz spectrometers. NMR chemical shifts are reported in ppm with solvent as internal reference Elemental analyses were performed under nitrogen using a Thermo Scientific Flash 2000 Organic Elemental Analyzer at the Institute of Chemistry and Chemical Engineering at EPFL.

**EPR analyses** were performed on a Bruker Elexsys E500 spectrometer working at 9.4 GHz frequency with an Oxford ESR900 cryostat for 4-300 K operation. Spin quantification was performed using CuSO<sub>4</sub>× 5H<sub>2</sub>O (Cu<sup>2+</sup>, **S** =1/2, 5.36 mg, number of Cu<sup>2+</sup> paramagnetic centers in this standard sample is  $1.29 \times 10^{19}$ ) as a standard.

**Magnetic measurements** were performed using a Quantum Design MPMS3 superconducting quantum interference device (SQUID) magnetometer in a temperature range 2-300 K. The sample was restrained with eicosane (for complex **A**) or glass wool (for complexes **2** and **3**) and enclosed in a quartz capsule packed under argon and placed inside a plastic straw. The measurements were performed with applied magnetic field of 1 T in the zero-field cooled (ZFC) regime. Diamagnetic corrections were applied using Pascal's constants.<sup>1</sup> The magnetic moment per complex was calculated using the following formula:

$$\mu_{eff} = \sqrt{8\chi_{corr}T}$$

where  $\chi_{corr}$  =  $\chi_{measured}$  -  $\chi_{dia}$ ,  $\chi_{dia}$  was calculated using Pascal's constants.<sup>1</sup>

**UV/Vis/NIR spectra** were recorded as solutions using 1 mm cuvettes equipped with a J-Young valve and a Perkin Elmer 950 spectrometer.

**Electrochemical Methods.** Cyclic voltammetry data were carried out at room temperature in an argon-filled glovebox described above. Data were collected using a Biologic SP-300 potentiostat connected to a personal computer. All samples were saturated in complex with 0.1 M [Bu<sub>4</sub>N][BPh<sub>4</sub>] supporting electrolyte in THF solution. The experiments were carried out with a platinum disk (d = 5 mm) working electrode, a platinum wire counter electrode, and an Ag/AgCl reference electrode. Potential calibration was performed at the end of each data collection cycle using the ferrocene/ferrocenium [( $C_5H_5$ )<sub>2</sub>Fe]<sup>+/0</sup> couple as an internal standard.

**Starting materials**. Unless otherwise noted, reagents were purchased from commercial suppliers and used without further purification. Anhydrous solvents were purchased from Aldrich and further distilled from K/benzophenone (THF, diethyl ether), sodium sand/benzophenone (*n*-hexane). Deuterated solvents for NMR spectroscopy (*d*<sub>8</sub>-THF) were purchased from Cortecnet, freeze-degassed and distilled over K/benzophenone. *d*<sub>6</sub>-DMSO were freeze-degassed and dried over 3 Å molecular sieves for several days. KC<sub>8</sub>,  $[U(\kappa^6-{(^{IBu2}ArO)_2Me_2-cyclam})I]$  (**A**), K<sub>2</sub>{(<sup>IBu2</sup>PhO)\_2Me\_2-cyclam}(THF)<sub>2</sub> and KOSi(O<sup>r</sup>Bu)<sub>3</sub> were synthesized according to literature procedures.<sup>2-5</sup>

#### Synthesis of [U(κ<sup>6</sup>-{(<sup>tBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam})(OSi(O<sup>t</sup>Bu)<sub>3</sub>)] (1)

A colorless solution of KOSi(O'Bu)<sub>3</sub> (15.5 mg, 0.05 mmol, 1.0 equiv.) in THF (1 mL) at -40 °C was added to a violet suspension of  $[U(\kappa^6-{(H^{Bu2}ArO)_2Me_2-cyclam})]$  (A) (52.5 mg, 0.05 mmol, 1.0 equiv.) in THF (1.5 mL) at -40 °C. The reaction mixture darkened during the addition of the reagents, following which it was stirred at -40 °C overnight (20 h). The reaction mixture supernatant was dark red brown in colour and a white precipitate was observed. The volatiles were removed under vacuum and the resulting residue was triturated with diethyl ether (0.5 mL) and dried under vacuum for 15 mins. This procedure was repeated two times more. The resulting residue was suspended in hexane (1.5 mL) and stirred at r.t for two minutes, following which the reaction mixture was filtered over a porosity 4 filter frit to obtain a dark red brown solution, which was placed at -40 °C for one day. The maroon solid formed was collected over a cold (-40 °C) porosity 4 filter frit and dried under vacuum, affording 1 as a maroon solid (33.4 mg, 0.03 mmol, 56%). Dark red brown crystals of 1.(hexane) 0.5 suitable for X-Ray diffraction were obtained from slow evaporation of a concentrated reaction mixture in pentane at -40 °C. <sup>1</sup>H NMR (400 MHz, *d*<sub>8</sub>-THF, 233 K) δ 42.6 (s, 1H), 38.95 (s, 1H), 35.79 (s, 1H), 35.48 (s, 1H), 30.55 (s, 1H), 30.02 (s, 1H), 28.16 (s, 1H), 24.74 (s, 1H), 16.45 (s, 1H), 13.55 (s, 1H), 11.36 (s, 1H), 9.72 (s, 1H), 9.20 (s, 1H), 7.90 (s, 1H), 6.77 (s, 27H, O'Bu), 3.08 (s, 9H, Ar-'Bu), 0.76 (s, 1H), 0.58 (s, 9H, Ar-'Bu), -2.0 (s, 3H, NCH<sub>3</sub>), -4.65 (s, 1H), -5.74 (s, 1H), -6.41 (s, 1H), -8.71 (s, 9H, Ar-<sup>t</sup>Bu), -11.07 (s, 1H), -11.62 (s, 1H), -15.46 (s, 1H), -22.12 (s9H, Ar-<sup>t</sup>Bu), -25.18 (s, 1H), -27.17 (s, 3H, NCH<sub>3</sub>), -27.8 (s, 1H), -30.45 (s, 1H), -32.78 (s, 1H), -40.06 (s, 1H), -48.84 (s, 1H), -49.30 (s, 1H) ppm (Figure S4). Elemental analysis calcd (%) for  $[U(\kappa^{6}-\{(^{H_{2}}ArO)_{2}Me_{2}-cyclam\})(OSi(O'Bu)_{3})]$  (1) (1164.51 g mol<sup>-1</sup>): C<sub>54</sub>H<sub>97</sub>N<sub>4</sub>O<sub>6</sub>Si<sub>1</sub>U<sub>1</sub> : C 55.70, H 8.33, N 4.81; found: C 55.26, H 8.36, N 4.63. Complex **1** is stable in solution (*d*<sub>8</sub>-THF) for at least 7 days (Figure S5) at -40 °C and continues to persist after 15 days (Figure S6) at r.t.

#### Synthesis of [U(κ<sup>7</sup>-{(<sup>tBu2</sup>ArO)(<sup>tBu2</sup>ArO-κ<sup>2</sup>-N,C)Me<sub>2</sub>-cyclam})] (2)

A violet suspension of A (34.3 mg, 0.03 mmol, 1.0 equiv.) in THF (1.5 mL) at -40 °C was added to solid KC<sub>8</sub> (9.9 mg, 0.07 mmol, 2.2 equiv.) at -40 °C. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C for one day resulting in a dark brown yellow supernatant and a black precipitate (i.e. graphite) along with excess KC8. The reaction mixture was filtered over a cold (-40 °C) porosity 4 glass filter frit to obtain a dark brown yellow solution. The volatiles were evaporated under reduced pressure and the reaction mixture was dissolved in cold (-40 °C) hexane (0.2 mL) to obtain a dark brown yellow solution, which was placed at -40 °C for four days. The dark green vellow solid formed was collected over a cold (-40 °C) porosity 4 filter frit and dried under vacuum, affording 2 as a dark green yellow solid (13.4 mg, 0.02 mmol, 45 %). Yellow crystals of 2.(THF)<sub>2</sub> suitable for X-Ray diffraction were obtained from a concentrated reaction mixture in THF at -40 °C. <sup>1</sup>H NMR (400 MHz, d<sub>8</sub>-THF, 233 K) δ 229.50 (s, 1H, NCH<sub>2</sub>-amide), 223.68 (s, 1H, <sup>-</sup>NCH<sub>2</sub>-amide), 221.78 (s, 1H, <sup>-</sup>NCH<sub>2</sub>-amide), 216.67 (s, 1H, <sup>-</sup>NCH<sub>2</sub>-amide), 52.99 (s, 1H), 51.35 (s, 9H, <sup>t</sup>Bu), 39.31 (s, 1H), 36.83 (s, 1H), 31.24 (s, 1H), 17.21 (s, 1H), 15.26 (s, 9H, <sup>t</sup>Bu), 8.81 (s, 1H), 7.09 (s, 1H), -2.91 (s, 1H), -7.06 (s, 9H, <sup>t</sup>Bu), -8.52 (s, 1H), -10.31 (s, 1H), -16.50 (s, 1H), -22.08 (s, 9H, <sup>t</sup>Bu), -22.61 (s, 1H), -33.10 (s, 1H), -33.34 (s, 1H), -41.31 (s, 1H), -44.93 (s, 1H), -45.20 (s, 1H), -60.94 (s, 1H), -64.80 (s, 2H), -73.58 (s, 1H), -81.18 (s, 1H), -84.72 (s, 3H, NCH<sub>3</sub>), -94.25 (s, 1H), -97.56 (s, 1H), -112.47 (s, 3H, NCH<sub>3</sub>), -137.21 (s, 1H), -272.05 (s, 1H) ppm (Figure S13). Elemental analysis calcd (%) for [U(κ<sup>7</sup>-{(<sup>lBu2</sup>ArO)(<sup>lBu2</sup>ArO-κ<sup>2</sup>-N,C)Me<sub>2</sub>-cyclam})].(THF)<sub>0.5</sub> (**2.(THF)**<sub>0.5</sub>) (937.13 g mol<sup>-1</sup>): C44H74N4O2.5U1 : C 56.39, H 7.90, N 5.98; found: C 56.19, H 8.12, N 5.56. Complex **2** is stable in solution (*d*8-THF) for at least 13 days (Figure S14) at -40 °C whilst significant decomposition could be observed after seven days at r.t. (Figure S15). NOTE: When the reduction of A was carried out in THF in the presence of 2.2.2-cryptand, complex 2 was still observed as the major species, when all the other parameters where remained constant. Complex 2 was still obtained as the major species when the analogous reaction was performed under dinitrogen.

#### Synthesis of [{U( $\kappa^5$ -{(<sup>tBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam})}<sub>2</sub>( $\mu$ - $\eta^6$ : $\eta^6$ -benzene)] (3)

A violet suspension of  $[U(\kappa^6-{((^{Hbu2}ArO)_2Me_2-cyclam})]]$  (A) (129.8 mg, 0.1 mmol, 1.0 equiv.) in benzene (10.0 mL) at r.t. was added to solid KC<sub>8</sub> (34.1 mg, 0.3 mmol, 2.0 equiv.) at r.t. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at r.t. for one day resulting in a dark brown supernatant and a black precipitate (i.e. graphite) along with excess KC<sub>8</sub>. The reaction mixture was filtered over a porosity 4 glass filter frit to obtain a dark brown solution. The volatiles were evaporated under reduced pressure and the reaction mixture was dissolved in toluene (1.5 mL) and hexane (3.0 mL) to obtain a dark brown solution, which was placed at -40 °C for two days. The dark brown solid formed was filtered over a cold (-40 °C) porosity 4 filter frit, affording **3** as a dark brown solid (19.7 mg, 0.01 mmol). A second crop (18.7 mg, 0.01 mmol) could be obtained from a toluene/hexane (0.1 mL/0.6 mL) solution of the mother liquor at -40 °C. The combined yield of the crops is 32%. Brown crystals of **3.(toluene)**<sub>3.6</sub> suitable for X-Ray diffraction were obtained from a concentrated reaction mixture in toluene at -40 °C. <sup>1</sup>H NMR (400 MHz, *d*<sub>6</sub>-benzene, 298 K)  $\delta$  28.44 (s), 24.49 (s), 23.74 (s), 19.0 (s), 17.34 (s), 15.62 (s), 15.18 (s), 13.41 (s), 13.13 (s), 12.77 (s), 12.10 (s), 11.90 (s), 11.38 (s), 11.05 (s), 10.47 (s), 10.19 (s), 9.35 (s), 8.50 (s), 6.63 (s), 5.65 (s), 4.07 (s), 3.70 (s), 2.96 (s), 2.48 (s), 2.00 (s), 0.12 (s), -1.39 (s), -3.34 (s), -4.98 (s), -5.79 (s), -7.57 (s), -9.15 (s), -10.90 (m), -12.2 (s), -1.44 (m), -17.0 (s), -18.66 (s), -20.43 (s), -24.70 (s), -31.18 (s), -84.18 (s) ppm (**Figure S20**). Elemental analysis calcd (%) for [{U( $\kappa^5$ -{( $^{Hu2}ArO)_2Me_2-cyclam})}]$ *2* $(<math>\mu$ - $\eta^6$ : $\eta^6$ -benzene)] (**3.(toluene)**.5) (1926.34 g mol<sup>-1</sup>): C<sub>93.5</sub>H<sub>150</sub>N<sub>8</sub>O<sub>4</sub>U<sub>2</sub> : C 58.30, H 7.79, N 5.82;

found: C 58.46, H 8.06, N 6.02. Complex **3** has limited stability in solution ( $d_6$ -benzene) as full decomposition could be observed after two days (**Figure S21**) at r.t. *NOTE*: An insoluble dark brown solid was obtained when the analogous reduction of **A** was carried out in benzene in the presence of 2.2.2-cryptand (1.0 equiv.), thus precluding its further characterisation.

#### Titration of A with KC<sub>8</sub> (1.0 to 2.0 equiv.) in d<sub>8</sub>-THF at -40 °C

A violet suspension of  $[U(\kappa^{6}-\{(^{tBu2}ArO)_{2}Me_{2}-cyclam\})]]$  (**A**) (20.0 mg, 0.02 mmol, 1.0 equiv.) in *d*<sub>8</sub>-THF (0.5 mL) at -40 °C was added to solid KC<sub>8</sub> (2.6 mg, 0.02 mmol, 1.0 equiv.) at -40 °C. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C for overnight (14 h) resulting in a dark brown-purple supernatant and a black precipitate (i.e. graphite). Analysis of the <sup>1</sup>H NMR spectrum (Figure S9b)) of the crude reaction mixture at -40 °C revealed the presence of resonances corresponding to the starting material complex **A** and new resonances assigned to complex **2**. The resulting reaction mixture was added to solid KC<sub>8</sub> (2.6 mg, 0.02 mmol, 1.0 equiv.) and stirred at -40 °C overnight (14 h) resulting in a dark brown yellow supernatant and a black precipitate (i.e. graphite), along with excess KC<sub>8</sub>. Analysis of the <sup>1</sup>H NMR spectrum (Figure S9c)) of the crude reaction mixture at -40 °C revealed only the presence of complex **2** and indicating complete consumption of complex **A**.

#### Reaction of 1 with KC<sub>8</sub> (1.0-5.0 equiv.) in d<sub>8</sub>-THF at -40 °C

A dark red brown solution of **1** (7.5 mg, 0.006 mmol, 1.0 equiv.) in  $d_8$ -THF (0.5 mL) at -40 °C was added to solid bronze KC<sub>8</sub> (0.9 mg, 0.006 mmol, 1.0 equiv.) at -40 °C. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C overnight (14 h) resulting in a dark red brown supernatant and a black precipitate (i.e. graphite). Analysis of the <sup>1</sup>H NMR spectrum of the crude reaction mixture at -40 °C revealed the presence of resonances corresponding to the starting material complex **1** (**Figure S7**a)). The existing reaction mixture was added to solid bronze KC<sub>8</sub> (3.5 mg, 0.026 mmol, 4.0 equiv.) at -40 °C and the reaction mixture was stirred at -40 °C for 1 h, resulting in a dark red brown supernatant (in addition to excess bronze KC<sub>8</sub> and graphite). Analysis of the <sup>1</sup>H NMR spectrum of the crude reaction mixture **3** (**Figure S7**a)).

#### Reaction of 1 with KC<sub>8</sub> in d<sub>8</sub>-THF at -40 °C

A dark red brown solution of **1** (5.1 mg, 0.004 mmol, 1.0 equiv.) in  $d_8$ -THF (0.4 mL) at -40 °C was added to solid bronze KC<sub>8</sub> (1.3 mg, 0.01 mmol, 2.2 equiv.) at -40 °C. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C overnight (14 h) resulting in a dark red brown supernatant along with a black precipitate (i.e. graphite) and excess bronze KC<sub>8</sub>. Analysis of the <sup>1</sup>H NMR spectrum of the crude reaction mixture at -40 °C revealed the presence of resonances corresponding to the starting material complex **1** (Figure S8).

#### Reaction of K<sub>2</sub>(<sup>tBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam with KC<sub>8</sub> in d<sub>8</sub>-THF at -40 °C

A colorless solution of  $K_2({}^{tBu2}ArO)_2Me_2$ -cyclam (5.3 mg, 0.007 mmol, 1.0 equiv.) in  $d_8$ -THF (0.5 mL) at -40 °C was added to solid KC<sub>8</sub> (2.3 mg, 0.02 mmol, 2.2 equiv.) at -40 °C. No colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C for one day resulting in a colourless supernatant and unreacted bronze KC<sub>8</sub>. The reaction mixture was filtered over a cold (-40 °C) porosity 4 glass filter frit to obtain a colourless solution. Analysis of the <sup>1</sup>H NMR spectrum of the colourless solution at -40 °C revealed the presence of unreacted K<sub>2</sub>( ${}^{tBu2}ArO)_2Me_2$ -cyclam starting material and no other species were observed (**Figure S29**).

#### Reaction of A with KC<sup>8</sup> in hexane at r.t. followed by analysis of the reaction mixture in d<sup>8</sup>-THF

A violet suspension of **A** (10.3 mg, 0.01 mmol, 1.0 equiv.) in hexane (0.8 mL) at r.t. was added to solid bronze KC<sub>8</sub> (3.0 mg, 0.02 mmol, 2.2 equiv.) at r.t. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at r.t. for three days resulting in a dark red brown precipitate (in addition to excess bronze KC<sub>8</sub> and graphite) and a pale brown yellow supernatant. Analysis of the <sup>1</sup>H NMR spectrum (at -40 °C) of the crude reaction mixture (after removal of volatiles under reduced pressure) in cold (-40 °C) *d*<sub>8</sub>-THF (0.5 mL) revealed many resonances from which complex **2** was identified as the major species (**Figure S16**c) and **Figure S17**c)).

#### Reaction of A with KC<sub>8</sub> in hexane at r.t. followed by analysis of the reaction mixture in $d_6$ -benzene

A violet suspension of **A** (10.0 mg, 0.01 mmol, 1.0 equiv.) in hexane (0.8 mL) at r.t. was added to solid bronze  $KC_8$  (2.9 mg, 0.02 mmol, 2.2 equiv.) at r.t. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at r.t. for three days resulting in dark red brown precipitate (in addition to excess bronze  $KC_8$  and

graphite) and a pale brown yellow supernatant. Analysis of the <sup>1</sup>H NMR spectrum (at r.t.) of the crude reaction mixture (after removal of volatiles under reduced pressure) in  $d_6$ -benzene (0.5 mL) revealed the presence of many resonances from which complex **3** was the only identifiable species (**Figure S23**).

#### Reaction of A with KC<sub>8</sub> in hexane at r.t. followed by analysis by EPR spectroscopy

A violet suspension of **A** (11.5 mg, 0.01 mmol, 1.0 equiv.) in hexane (0.8 mL) at r.t. was added to solid bronze KC<sub>8</sub> (3.3 mg, 0.03 mmol, 2.2 equiv.) at r.t. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at r.t. for 3.5 days resulting in a dark red brown precipitate (in addition to excess bronze KC<sub>8</sub> and graphite) and a pale brown yellow supernatant. The precipitate formed was filtered over a porosity 4 filter frit and washed with hexane (0.5 mL) to obtain 7.2 mg of solid (*NOTE*: the solid collected is a mixture of uranium containing species, excess KC<sub>8</sub> and graphite). Analysis of the EPR spectrum (**Figure S40**) of the precipitate (7.2 mg) at 6 K revealed a signal *g*-values 2.91and 2.61 probably arising from a uranium containing species, in addition to a signal at *g* = 2 corresponding to the excess KC<sub>8</sub> present. The precipitate was then suspended in benzene (0.2 mL) and filtered over a porosity 4 filter frit to obtain a red brown solution. Analysis of the EPR spectrum (**Figure S42**) of the frozen red brown solution in benzene at 6 K revealed an EPR spectrum corresponding to that of complex **3**.

#### Titration of A with KC<sub>8</sub> (1.0 to 2.0 equiv.) in *d*<sub>6</sub>-benzene at r.t.

A violet suspension of **A** (9.5 mg, 0.009 mmol, 1.0 equiv.) in  $d_6$ -benzene (0.5 mL) at r.t. was added to solid bronze KC<sub>8</sub> (1.3 mg, 0.009 mmol, 1.0 equiv.) at r.t. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at r.t. for one day resulting in dark brown suspension. Analysis of the <sup>1</sup>H NMR spectrum of the crude reaction mixture at r.t. revealed the presence of new resonances assigned to complex **3**, and the presence of unreacted complex **A** (**Figure S22**a)). The resulting reaction mixture was added to solid KC<sub>8</sub> (1.3 mg, 0.009 mmol, 1.0 equiv.) and stirred at r.t. for one day resulting in dark brown suspension. Analysis of the <sup>1</sup>H NMR spectrum of the crude reaction mixture at r.t. for one day resulting in dark brown suspension. Analysis of the <sup>1</sup>H NMR spectrum of the crude reaction mixture at r.t. revealed complex **3** as the predominant species and the complete consumption of the starting material **A** (**Figure S22**b)).

#### Reaction of A with KC<sup>8</sup> in diethyl ether under dinitrogen at -40 °C followed by acid quenching

A violet suspension of **A** (10.3 mg, 0.01 mmol, 1.0 equiv.) in diethyl ether (0.8 mL) at -40 °C was added to solid bronze KC<sub>8</sub> (3.0 mg, 0.02 mmol, 2.2 equiv.) under dinitrogen. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C for three days resulting in dark purple precipitate (in addition to excess bronze KC<sub>8</sub> and graphite) and a pale brown yellow supernatant.

The volatiles were evaporated under reduced pressure and the resulting reaction mixture was cooled to -80 °C. A cold (-80 °C) colourless solution of 1 M HCl in diethyl ether (2.0 mL) was added to the reaction mixture at -80 °C resulting in the concomitant formation of a black precipitate and a colourless supernatant. The reaction mixture was brought to room temperature (r.t.) and stirred at r.t. overnight (16 h). The volatiles were evaporated under reduced pressure and the reaction mixture was dried under vacuum for 4 h. <sup>1</sup>H NMR spectrum (Figure S30) of the resulting sample in d<sub>6</sub>-DMSO (0.5 mL) revealed the presence of two peaks at 7.55 ppm and 7.43 ppm which correspond to the ammonium chloride triplet arising from the acid quenching of the product of dinitrogen reduction upon the reaction of A with KC8 (2.2 equiv.). The third signal corresponding to the ammonium chloride triplet is obscured by resonances corresponding to the cyclam ligand. The ammonium chloride triplet could however be observed and quantified upon performing a volatiles-transfer experiment (vide infra). Control experiments were performed (vide infra) to confirm the presence of ammonium chloride arising from dinitrogen reduction. Two control experiments were conducted wherein (1) the reduction reaction mixture conducted under argon was guenched with HCI in diethyl ether to detect the presence of the resonances at 7.55 ppm and 7.43 ppm and (2) the protonated ligand H<sub>2</sub>(<sup>IBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam was reacted with HCl in diethyl ether to rule out the possibility of ammonium chloride formation from the nitrogenous cyclam ligand (vide infra). No ammonium chloride was observed in either case (Figure S34 and Figure S36) thereby confirming dinitrogen reduction occurring when the reduction of 1 is carried out under dinitrogen.

#### Reaction of A with KC<sub>8</sub> in diethyl ether under argon at -40 °C

The <sup>1</sup>H NMR spectrum in toluene at -40°C (Figure **S24b**) of a crude reaction mixture resulting from **A** + 2.2 equiv. of KC<sub>8</sub> stirred for three days in diethyl ether at -40 °C showed resonances assigned to complex **2**.

Quantification of ammonium chloride generated from the reaction of A with KC<sub>8</sub> in diethyl ether under dinitrogen at -40 °C



Figure S1. Picture of the double bulb distillation apparatus used for the volatiles transfer experiment.

*NOTE*: The distillation apparatus used contains two bulbs (namely *A* and *B*), both of which are adapted with a J. Young valve and can be sealed on their own; both of which are connected to each other by a glass tube fitting at 90 ° with respect to each of them. The apparatus has one attachment/outlet closer to one of the bulbs (namely bulb *A*) which can be connected to the Schlenk line for Schlenk-line manipulations.

A violet suspension of A (12.3 mg, 0.01 mmol, 1.0 equiv.) in diethyl ether (0.8 mL) at -40 °C was added to solid bronze KC<sub>8</sub> (3.6 mg, 0.03 mmol, 2.2 equiv.) under dinitrogen. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C for three days resulting in dark purple precipitate (in addition to excess bronze KC<sub>8</sub> and graphite) and a pale brown yellow supernatant. The volatiles were evaporated under reduced pressure and the resulting reaction mixture was cooled to -80 °C. A cold (-80 °C) colourless solution of 1 M HCl in diethyl ether (2.0 mL) was added to the reaction mixture at -80 °C resulting in the concomitant formation of a black precipitate and a colourless supernatant. The reaction mixture was brought to room temperature (r.t.) and stirred at r.t. overnight (16 h). The volatiles were evaporated under reduced pressure to eliminate any unreacted HCI, and resuspended in diethyl ether (2.0 mL). Inside the dinitrogen glovebox, the J. Young valve-capped distillation apparatus was placed in a pre-chilled (-160 °C) glovebox-fitted cold well following which 1 M HCl in diethyl ether (2.0 mL) was pipetted in bulb B and frozen. The apparatus was connected to the glovebox vacuum line, and placed under vacuum. After 5 minutes, the J. Young valve leading to the bulb B was closed. The resulting reaction mixture in diethyl ether was transferred to bulb A of the distillation apparatus fitted with a stirrer bar (NOTE: bulb B is still closed and under static vacuum). The J. Young valves for both the bulbs were closed, the apparatus was taken out of the glovebox, attached to a Schlenk line and bulb A was frozen. Under a strong flow of dinitrogen, a colourless solution of NaO'Bu (11.5 mg, 0.1 mmol, 10.0 equiv.) in diethyl ether (2 mL) was added via a syringe, after which the bulb A was closed and thawed (NOTE: bulb B remains closed throughout this manipulation). The resulting black suspension was stirred at r.t. overnight (18 h). The bulb containing 1 M HCl in diethyl ether (2.0 mL) (i.e. bulb B) was frozen and the volatiles of bulb A (i.e. the reaction mixture) were vacuum-transferred to bulb B. Bulb B was thawed and the entire apparatus was entered into the dinitrogen glovebox. The volatiles in bulb B were evaporated under reduced pressure and the resulting white residue obtained was dried under vacuum for 30 mins. <sup>1</sup>H NMR spectrum of the white residue in *d*<sub>6</sub>-DMSO (0.4 mL) revealed the presence of ammonium chloride at 7.16 ppm (Figure S31). A quantitative <sup>1</sup>H NMR spectrum (Figure S32) in d<sub>6</sub>-DMSO (0.6 mL) with dimethylsulfone as the internal standard revealed the formation of ammonium chloride in 20% yield (0.2 equiv. of NH<sub>4</sub>Cl per complex A used).

# Quantification of ammonium chloride ( $^{15}NH_4CI$ ) generated from the reaction of A with KC<sub>8</sub> in diethyl ether under $^{15}N_2$ at -40 °C

A violet suspension of A (25.3 mg, 0.02 mmol, 1.0 equiv.) in diethyl ether (3.0 mL) was added to solid bronze KC<sub>8</sub> (33.3 mg, 0.25 mmol, 10.0 equiv.) under argon. No observable colour change was perceived immediately after addition. The reaction mixture was immediately brought outside the glovebox, attached to a Schlenk line, freeze/degassed (x3) and <sup>15</sup>N<sub>2</sub> (820 mbar) was added whilst the reaction mixture was frozen The resulting reaction mixture was stirred at -40 °C for three days resulting in dark purple precipitate (in addition to excess bronze KC<sub>8</sub> and graphite) and a pale brown yellow supernatant. The volatiles were evaporated under reduced pressure and the resulting reaction mixture was cooled to -80 °C. A cold (-80 °C) colourless solution of 1 M HCl in diethyl ether (6.0 mL) was added to the reaction mixture at -80 °C resulting in the concomitant formation of a black precipitate and a colourless supernatant. The reaction mixture was brought to room temperature (r.t.) and stirred at r.t. overnight (16 h). The volatiles were evaporated under reduced pressure to eliminate any unreacted HCI, and resuspended in diethyl ether (2.0 mL). Inside the dinitrogen glovebox, the J. Young valve-capped distillation apparatus was placed in a pre-chilled (-160 °C) glovebox-fitted cold well following which 1 M HCl in diethyl ether (2.0 mL) was pipetted in bulb B and frozen. The apparatus was connected to the glovebox vacuum line, and placed under vacuum. After 5 minutes, the J. Young valve leading to the bulb B was closed. The resulting reaction mixture in diethyl ether was transferred to bulb A of the distillation apparatus fitted with a stirrer bar (NOTE: bulb B is still closed and under static vacuum). The J. Young valves for both the bulbs were closed, the apparatus was taken out of the glovebox, attached to a Schlenk line and bulb A was frozen. Under a strong flow of dinitrogen, a colourless solution of aqueous KOH (30%, 2.5 mL) was added via a syringe, after which the bulb A was closed and frozen (NOTE: bulb B remains closed throughout this manipulation). The reaction mixture was thawed and the resulting black suspension was stirred at r.t. for 1 h). The bulb containing 1 M HCl in diethyl ether (2.0 mL) (i.e. bulb B) was frozen and the volatiles of bulb A (i.e. the reaction mixture) were vacuum-transferred to bulb B. Bulb B was thawed and all the volatiles present in bulb B were evaporated under reduced pressure and the resulting white residue obtained was dried under vacuum for 30 mins. A quantitative <sup>1</sup>H NMR spectrum (Figure S33) of the white residue in d<sub>6</sub>-DMSO (0.4 mL) with dimethylsulfone as the internal standard revealed the presence of <sup>15</sup>NH<sub>4</sub>Cl in 30% yield (0.3 equiv. of <sup>15</sup>NH<sub>4</sub>Cl per complex A used).

#### Reaction of A with KC<sub>8</sub> in diethyl ether under argon at -40 °C followed by acid quenching

A violet suspension of **A** (10.5 mg, 0.01 mmol, 1.0 equiv.) in diethyl ether (0.8 mL) at -40 °C was added to solid bronze KC<sub>8</sub> (3.0 mg, 0.02 mmol, 2.2 equiv.) under argon. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at -40 °C for three days resulting in dark brown supernatant along with excess bronze KC<sub>8</sub> and graphite precipitate. The volatiles were evaporated under reduced pressure and the resulting reaction mixture was cooled to -80 °C. A cold (-80 °C) colourless solution of 2 M HCl in diethyl ether (1.0 mL) was added to the reaction mixture at -80 °C resulting in the concomitant formation of a black precipitate and a colourless supernatant. The reaction mixture was brought to room temperature (r.t.) and stirred at r.t. overnight (16 h). The volatiles were evaporated under reduced pressure and the reaction mixture start in *d*<sub>6</sub>-DMSO (0.5 mL) revealed only the presence of resonances corresponding to the ligand and no peaks resembling the ammonium chloride triplet were observed, thereby confirming the activation and reduction of dinitrogen when the reaction is conducted under a dinitrogen atmosphere.

#### Reaction of A with KC<sub>8</sub> in hexane under dinitrogen at r.t. followed by acid quenching

A violet suspension of A (11.7 mg, 0.01 mmol, 1.0 equiv.) in hexane (0.8 mL) at r.t. was added to solid bronze KC8 (3.4 mg, 0.03 mmol, 2.2 equiv.) under dinitrogen. No observable colour change was perceived immediately after addition. The reaction mixture was stirred at r.t. for three days resulting in dark red brown precipitate (in addition to excess bronze KC8 and graphite) and a pale brown yellow supernatant. The volatiles were evaporated under reduced pressure and the resulting reaction mixture was cooled to -80 °C. A cold (-80 °C) colourless solution of 1 M HCl in diethyl ether (2.0 mL) was added to the reaction mixture at -80 °C resulting in the concomitant formation of a black precipitate and a colourless supernatant. The reaction mixture was brought to room temperature (r.t.) and stirred at r.t. overnight (16 h). The volatiles were evaporated under reduced pressure to eliminate any unreacted HCI, and resuspended in diethyl ether (2.0 mL). Inside the dinitrogen glovebox, the J. Young valve-capped distillation apparatus (vide supra, Figure S1) was placed in a pre-chilled (-160 °C) glovebox-fitted cold well following which 1 M HCl in diethyl ether (2.0 mL) was pipetted in bulb B and frozen. The apparatus was connected to the glovebox vacuum line, and placed under vacuum. After 5 minutes, the J. Young valve leading to the bulb B was closed. The resulting reaction mixture in diethyl ether was transferred to bulb A of the distillation apparatus fitted with a stirrer bar (NOTE: bulb B is still closed and under static vacuum). The J. Young valves for both the bulbs were closed, the apparatus was taken out of the glovebox, attached to a Schlenk line and bulb A was frozen. Under a strong flow of dinitrogen, a colourless solution of NaO'Bu (10.9 mg, 0.1 mmol, 10.0 equiv.) in diethyl ether (1.5 mL) was added via a syringe, after which the bulb A was closed and thawed (NOTE: bulb B remains closed throughout this manipulation). The resulting black suspension was stirred at r.t. overnight (18 h). The bulb containing 1 M HCl in diethyl ether (2.0 mL) (i.e. bulb B) was frozen and the volatiles of bulb A (i.e. the reaction mixture) were vacuumtransferred to bulb *B*. Bulb *B* was thawed and the entire apparatus was entered into the dinitrogen glovebox. The volatiles in bulb *B* were evaporated under reduced pressure and the resulting white residue obtained was dried under vacuum for 30 mins. Analysis of the <sup>1</sup>H NMR spectrum (**Figure S35**) of the remaining residue in  $d_6$ -DMSO (0.4 mL) did not reveal the presence of ammonium chloride.

#### Reaction of H<sub>2</sub>(<sup>tBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam with HCl in diethyl ether

A cold (-80 °C) colourless solution of 2 M HCl in diethyl ether (1.5 mL) was added to solid  $H_2(^{IBu2}ArO)_2Me_2$ -cyclam (9.1 mg, 0.01 mmol, 1.0 equiv.) at -80 °C. No observable colour change was perceived immediately after addition. The reaction mixture was brought to room temperature (r.t.) and stirred at r.t. overnight (16 h) resulting in a pink precipitate and a colourless supernatant. The volatiles were evaporated under reduced pressure and the reaction mixture was dried under vacuum for 4 h. <sup>1</sup>H NMR spectrum (**Figure S36**) of the resulting sample in *d*<sub>6</sub>-DMSO (0.5 mL) revealed the presence of resonances corresponding to the ligand.

## **B. NMR Spectroscopic Data**



**Figure S2.** Evolution of the <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>8</sub>-THF, 298 K) of a THF solution of isolated **A** after dissolution (sample was stored at r.t. at all times).



**Figure S3.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of the crude reaction mixture obtained by reacting **A** with KOSi(O<sup>4</sup>Bu)<sub>3</sub> in THF at -40 °C overnight (20 h) to yield **1**.



**Figure S4.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of isolated **1**.



**Figure S5.** Evolution of the <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a THF solution of isolated **1** after dissolution (sample was stored at -40 °C at all times).



(sample was stored at r.t. at all times).



**Figure S7.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a) the crude reaction mixture resulting from **1** + 1.0 equiv. of KC<sub>8</sub> overnight (14 h) in  $d_8$ -THF at -40 °C, b) the crude reaction mixture resulting from the addition of 4.0 equiv. of KC<sub>8</sub> to a) at -40 °C for 1 h.



**Figure S8.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a) complex **1** before addition of KC<sub>8</sub> and b) the crude reaction mixture resulting from **1** + 2.2 equiv. of KC<sub>8</sub> in  $d_8$ -THF overnight (14 h) at -40 °C.



220 200 180 160 140 120 100 80 60 40 20 0 -20 40 60 -80 -100 -120 -140 -160 -180 -200 -220 -240 -260 Chemical Shift (ppm)

**Figure S9.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a) complex **A** before addition of KC<sub>8</sub>, b) the crude reaction mixture resulting from **A** + 1.0 equiv. of KC<sub>8</sub> in  $d_8$ -THF overnight (14 h) at -40 °C and c) the crude reaction mixture resulting from **A** + 2.0 equiv. of KC<sub>8</sub> in  $d_8$ -THF overnight (14 h) at -40 °C.



**Figure S10.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> at -40 °C in  $d_8$ -THF (0.5 mL) under argon overnight (14 h) to yield **2**, in differing concentrations of **A** (partial window shown for clarity).



**Figure S11.** <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>8</sub>-THF, 233 K) of the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> at -40 °C in *d*<sub>8</sub>-THF for one day under dinitrogen.



**Figure S12.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> at -40 °C in  $d_8$ -THF for one day under a) argon and b) dinitrogen.



Figure S13. <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>8</sub>-THF, 233 K) of isolated 2.



**Figure S14.** Evolution of the <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a THF solution of isolated **2** after dissolution (sample was stored at -40 °C at all times).



 
 220
 200
 180
 160
 140
 120
 100
 80
 60
 40
 20
 0
 -20
 -40
 -60
 -80
 -100
 -120
 -140
 -180
 -200
 -220
 -240
 -260

 Ghemical Shift (ppm)

 Figure S15. Evolution of the <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>8</sub>-THF, 233 K) of a THF solution of isolated **2** after dissolution
(sample was stored at r.t. at all times).



**Figure S16.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a) isolated complex **2**, b) the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> at -40 °C in  $d_8$ -THF overnight (14 h) and c) the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> in hexane at r.t. for three days.



**Figure S17.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a) isolated complex **2**, b) the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> at -40 °C in  $d_8$ -THF overnight (14 h) and c) the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> in hexane at r.t. for three days (*NOTE*: the spectra depicted are the same as **Figure S16** and a smaller spectral width has been displayed for clarity).



**Figure S18.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -benzene, 298 K) of the crude reaction mixture obtained by reacting **A** and 2.2 equiv. of KC<sub>8</sub> at r.t. in benzene for one day to yield **3**.



**Figure S19.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -benzene, 298 K) of the crude reaction mixture obtained by reacting **A** and 2.2 equiv. of KC<sub>8</sub> at r.t. in  $d_6$ -benzene for one day under dinitrogen.



45 40 35 30 25 20 15 10 5 0 -5 -10 -15 -20 -25 -30 -35 -40 -45 -50 -55 -60 -65 -70 -75 -80 -85 Chemical Shift (ppm)

Figure S20. <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>6</sub>-benzene, 298 K) of isolated 3.



**Figure S21.** Evolution of the <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -benzene, 298 K) of a benzene solution of isolated **3** after dissolution (sample was stored at r.t. at all times).



**Figure S22.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -benzene, 298 K) of a) the crude reaction mixture resulting from **A** + 1.0 equiv. of KC<sub>8</sub> in  $d_6$ -benzene for one day at r.t. and b) the crude reaction mixture resulting from **A** + 2.0 equiv. of KC<sub>8</sub> in  $d_6$ -benzene for one day at r.t.



Figure S23. <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>6</sub>-benzene, 298 K) of the crude reaction mixture obtained by reacting **A** and

**Figure S23.** 'H NMR spectrum (400 MHz, *d*<sub>6</sub>-benzene, 298 K) of the crude reaction mixture obtained by reacting **A** and 2.2 equiv. KC<sub>8</sub> in hexane at r.t. for three days.



**Figure S24.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -toluene, 233 K) of a) the crude reaction mixture resulting from **A** + 2.2 equiv. of KC<sub>8</sub> for three days in diethyl ether at -40 °C under dinitrogen and b) the crude reaction mixture resulting from **A** + 2.2 equiv. of KC<sub>8</sub> for three days in diethyl ether at -40 °C under argon.



**Figure S25.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -toluene, 233 K) of the crude reaction mixture obtained by reacting complex **A** with 2.2 equiv. of KC<sub>8</sub> in THF at -40 °C overnight (14 h).



**Figure S26.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -benzene, 298 K) of the crude reaction mixture obtained when the **A** + 2.2 equiv. of KC<sub>8</sub> reaction carried out under dinitrogen in diethyl ether for three days was suspended in  $d_6$ -benzene.



**Figure S27.** Evolution of <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of the crude reaction mixture obtained when the **A** + 2.2 equiv. of KC<sub>8</sub> reaction carried out under dinitrogen in diethyl ether for three days was suspended in cold (-40 °C)  $d_8$ -THF (sample was stored at -40 °C).



Figure S28. <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>8</sub>-THF, 233 K) of K<sub>2</sub>(<sup>iBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam.


**Figure S29.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_8$ -THF, 233 K) of a) K<sub>2</sub>(<sup>IBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam before addition of KC<sub>8</sub> and b) the reaction mixture (after filtration to remove KC<sub>8</sub>) obtained by reacting K<sub>2</sub>(<sup>IBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam with 2.2 equiv. of KC<sub>8</sub> in  $d_8$ -THF at -40 °C for one day.



13.0 12.5 12.0 11.5 11.0 10.5 10.0 9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 0.0 -0. Chemical Shift (ppm)

**Figure S30.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -DMSO, 298 K) of the resulting reaction mixture obtained upon acid quenching (HCl in diethyl ether) of the reaction of **A** with KC<sub>8</sub> (2.2 equiv.) in diethyl ether at -40 °C for three days at under dinitrogen.



13.0 12.5 12.0 11.5 11.0 10.5 10.0 9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 0.0 Chemical Shift (ppm)

**Figure S31.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -DMSO, 298 K) of the white residue obtained by the volatiles-transfer experiment conducted on the reaction of **A** with KC<sub>8</sub> (2.2 equiv.) in diethyl ether at -40 °C for three days at under dinitrogen.



**Figure S32.** Quantitative <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -DMSO, 298 K) of the white residue obtained by the volatilestransfer experiment conducted on the reaction of **A** with KC<sub>8</sub> (2.2 equiv.) in diethyl ether at -40 °C for three days at under dinitrogen (dimethylsulfone used as the internal standard).



**Figure S33.** Quantitative <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>6</sub>-DMSO, 298 K) of the white residue obtained by the volatilestransfer experiment conducted on the reaction of **A** with KC<sub>8</sub> (2.2 equiv.) in diethyl ether at -40 °C for three days at under <sup>15</sup>N<sub>2</sub> (dimethylsulfone used as the internal standard).



**Figure S34.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -DMSO, 298 K) of the resulting reaction mixture obtained upon acid quenching (HCl in diethyl ether) of the reaction of **A** with KC<sub>8</sub> (2.2 equiv.) in diethyl ether at -40 °C for three days at under argon.

DMSO



**Figure S35.** <sup>1</sup>H NMR spectrum (400 MHz,  $d_6$ -DMSO, 298 K) of the resulting reaction mixture obtained upon acid quenching (HCl in diethyl ether) of the reaction of **A** with KC<sub>8</sub> (2.2 equiv.) in hexane at r.t. for three days at under dinitrogen.



**Figure S36.** <sup>1</sup>H NMR spectrum (400 MHz, *d*<sub>6</sub>-DMSO, 298 K) of the resulting reaction mixture obtained upon acid quenching (HCl in diethyl ether) of H<sub>2</sub>(<sup>tBu2</sup>ArO)<sub>2</sub>Me<sub>2</sub>-cyclam.

### C. X-Ray Crystal Structure Determination Details

Suitable crystals were selected and mounted on various Rigaku diffractometers (XtaLAB Synergy R, DW system, HyPix-Arc 150 detector or SuperNova, Dual, Cu at home/near, AtlasS type detectors). The crystals were kept at a steady T = 140.00(10) K during data collection. Data were measured using  $\omega$  scans with Cu K $\alpha$  radiation. The diffraction patterns were indexed and the total number of runs and images were based on the strategy calculation from the program CrysAlisPro 1.171.44.91a.<sup>6</sup> The unit cells were refined using CrysAlisPro 1.171.44.91a.<sup>6</sup> Data reduction, scaling and absorption corrections were performed using CrysAlisPro 1.171.44.91a.<sup>6</sup>

The structures were solved with the ShelXT solution program using dual methods and by using Olex2 1.5 as the graphical interface.<sup>7,8</sup> The models were refined with ShelXL 2019/3 using full matrix least squares minimisation on F2.<sup>9</sup> All non-hydrogen atoms were refined anisotropically. Some hydrogen atom positions were calculated geometrically and refined using the riding model, but most hydrogen atoms were refined freely.

The structures displayed problems dealing with disorder (disordered ligands or solvent) or twinning. In particular, the crystal structure of compound **1** was treated for twinning (3 BASF parameters (0.212(2), 0.142(2), 0.124(2)) were refined during the last cycles of least-squares). The refinement of all structures needed several restraints to adjust the atomic parameters of the disordered moieties. The solvent was squeezed in the last stages of the refinement (using Olex2) in the case of compound **2** and **3**.

**Table S1.** Crystal data and structural refinement parameters for complexes  $[U(\kappa^6-\{({}^{tBu2}ArO)_2Me_2-cyclam\})(OSi(O'Bu)_3)]$ .(hexane)<sub>0.5</sub> (**1.(hexane)**<sub>0.5</sub>),  $[U(\kappa^7-\{({}^{tBu2}ArO)({}^{tBu2}ArO-\kappa^2-N,C)Me_2-cyclam\})]$ .(THF)<sub>2</sub> (**2.(THF)**<sub>2</sub>) and  $[\{U(\kappa^5-\{({}^{tBu2}ArO)_2Me_2-cyclam\})\}_2(\mu-\eta^6:\eta^6-benzene)]$  (**3.(toluene)**<sub>3.6</sub>).

Compound	1.(hexane)₀.₅	2.(THF)₂	3.(toluene) <sub>3.6</sub>
Formula	C57H104N4O6SiU	C <sub>50</sub> H <sub>86</sub> N <sub>4</sub> O <sub>4</sub> U	C115.2H174.8N8O4U2
Crystal Size (mm)	0.23×0.11×0.09	0.37×0.08×0.07	0.20×0.06×0.04
Crystal System	triclinic	monoclinic	monoclinic
Space Group	<i>P</i> -1	P21/n	P2₁/c
Volume (Å <sup>3</sup> )	3015.04(16)	4980.6(3)	5387.5(4)
a (Å)	11.2069(3)	16.1897(4)	9.8734(7)
b (Å)	14.7053(5)	15.6587(6)	20.7159(8)
c (Å)	18.7776(6)	19.7929(8)	26.4929(3)
α (°)	84.700(3)	90	90
β(°)	80.165(2)	96.971(3)	96.162(3)
γ (°)	82.473(2)	90	90
Z	2	4	2
Formula Weight	1207.56	1045.25	2211.88
Density (g cm <sup>-3</sup> )	1.330	1.394	1.364
μ (mm <sup>-1</sup> )	8.138	9.514	8.801
F (000)	1256	2152	2276
Temperature (K)	199.99(10)	140.00(10)	140.00(10)
Total Reflections	13062	27066	57981
Unique Reflections	13062	9714	9157
R <sub>int</sub>		0.0975	0.0734
R Indices [I >2σ(I)]	<i>R</i> <sub>1</sub> = 0.0434 <i>wR</i> <sub>2</sub> = 0.0977	<i>R</i> <sub>1</sub> = 0.0633 <i>wR</i> <sub>2</sub> = 0.1386	<i>R</i> <sub>1</sub> = 0.0412 <i>wR</i> <sub>2</sub> = 0.1039
Largest Diff. Peak and Hole (e A⁻³)	3.762 and -1.696	1.946 and -2.135	1.456 and -1.936
GooF	0.975	1.002	1.050
CCDC	2411216	2411217	2411215

## D. EPR Spectroscopy Data



**Figure S37.** Solid state X-band (9.40 GHz) EPR spectrum of **A** (8.0 mg) at 6 K (*black line*, experiment; *red dashed line*, simulation). *g*-values:  $g_1 = 4.75$ ;  $g_2 = 0.82$ ;  $g_3 = 0.65$ .



**Figure S38.** Solid state X-band (9.40 GHz) EPR spectrum of **3** (8.0 mg) at 6 K (*black line*, experiment; *red dashed line*, simulation). *g*-values:  $g_1 = 2.80$ ;  $g_2 = 2.60$ ;  $g_3 = 0.87$ .



**Figure S39.** Frozen solution state X-band (9.40 GHz) EPR spectrum of **3** (10 mM in toluene) at 6 K (*black line*, experiment; *red dashed line*, simulation). *g*-values:  $g_1 = 2.78$ ;  $g_2 = 2.65$ ;  $g_3 = 1.13$ .



**Figure S40.** Solid state X-band (9.40 GHz) EPR spectrum of the precipitate obtained from  $\mathbf{A}$  + 2.2 KC<sub>8</sub> reaction in hexane (which contains the intermediate species  $\mathbf{B}$ ) at 6 K (\* corresponds to a signal arising from the unpaired electron from the excess KC<sub>8</sub> present).



**Figure S41.** Solid state X-band (9.40 GHz) EPR spectrum of a) isolated **A** (8.0 mg) at 6 K and b) the precipitate obtained from **A** + 2.2 KC<sub>8</sub> reaction in hexane (which contains the intermediate species **B**) at 6 K (\* corresponds to a signal arising from the unpaired electron from the excess KC<sub>8</sub> present).



**Figure S42.** Frozen solution state (benzene) X-band (9.40 GHz) EPR spectrum of the resulting red brown solution obtained when the precipitate from the  $\mathbf{A}$  + 2.2 KC<sub>8</sub> reaction in hexane is suspended in benzene and filtered (to remove graphite and excess KC<sub>8</sub>)(\* represents an organic radical impurity present in the sample).



Figure S43. Solid state X-band (9.40 GHz) EPR spectrum of KC<sub>8</sub> (13.0 mg) at 6 K.



Figure S44. Temperature dependent SQUID magnetisation data for A plotted as a function of  $\mu_{\text{eff}}$  vs. temperature, measured at 1 T.



**Figure S45.** Temperature dependent SQUID magnetisation data for **A** plotted as a function of  $\chi_M$  (molar magnetic susceptibility) vs. temperature, measured at 1 T.



**Figure S46.** Temperature dependent SQUID magnetisation data for **A** plotted as a function of  $1/\chi_M$  (where  $\chi_M$  is the molar magnetic susceptibility) vs. temperature, measured at 1 T.



Figure S47. Temperature dependent SQUID magnetisation data for A plotted as a function of  $\chi T$  vs. temperature, measured at 1 T.



Figure S48. Temperature dependent SQUID magnetisation data for 2 plotted as a function of  $\mu_{eff}$  vs. temperature, measured at 1 T.



**Figure S49.** Temperature dependent SQUID magnetisation data for **2** plotted as a function of  $\chi_M$  (molar magnetic susceptibility) vs. temperature, measured at 1 T.



**Figure S50**. Temperature dependent SQUID magnetisation data for **2** plotted as a function of  $1/\chi_M$  (where  $\chi_M$  is the molar magnetic susceptibility) vs. temperature, measured at 1 T.



Figure S51. Temperature dependent SQUID magnetisation data (per complex) for 3 plotted as a function of  $\chi T$  vs. temperature, measured at 1 T.



Figure S52. Temperature dependent SQUID magnetisation data (per complex) for 3 plotted as a function of  $\mu_{eff}$  vs. temperature, measured at 1 T.



**Figure S53.** Temperature dependent SQUID magnetisation data (per complex) for **3** plotted as a function of  $\chi_M$  (molar magnetic susceptibility) vs. temperature, measured at 1 T.



**Figure S54.** Temperature dependent SQUID magnetisation data (per complex) for **3** plotted as a function of  $1/\chi_M$  (where  $\chi_M$  is the molar magnetic susceptibility) vs. temperature, measured at 1 T.



Figure S55. Temperature dependent SQUID magnetisation data (per complex) for 3 plotted as a function of  $\chi$ T vs. temperature, measured at 1 T.



Figure S56. Temperature dependent SQUID magnetisation data (per U ion) for 3 plotted as a function of  $\mu_{\text{eff}}$  vs. temperature, measured at 1 T.



**Figure S57.** Temperature dependent SQUID magnetisation data (per U ion) for **3** plotted as a function of  $\chi_M$  (molar magnetic susceptibility) vs. temperature, measured at 1 T.



**Figure S58.** Temperature dependent SQUID magnetisation data (per U ion) for **3** plotted as a function of  $1/\chi_M$  (where  $\chi_M$  is the molar magnetic susceptibility) vs. temperature, measured at 1 T.



Figure S59. Temperature dependent SQUID magnetisation data (per U ion) for 3 plotted as a function of  $\chi T$  vs. temperature, measured at 1 T.



**Figure S60.** Temperature dependent SQUID magnetisation data (per U ion) for **A**, **2** and **3** plotted as a function of  $\chi$ T vs. temperature, measured at 1 T.

## F. Cyclic Voltammetry Data

**Table S2.** Electrochemical redox potentials (in V vs  $Fc/Fc^{+}$ ) for complex **A** including the  $E_{red}$ ,  $E_{ox}$ , and  $E_{1/2}$  values for events I and II in 0.1 M [NBu<sub>4</sub>][BPh<sub>4</sub>] electrolyte in THF.

	E <sub>red</sub>	E <sub>ox</sub>	E1/2
Wave I	-1.77	-1.61	-1.69
Wave II	-0.39	-0.28	-0.34



**Figure S61.** Cyclic voltammogram for complex **A** in THF (3 mM) with 0.1 M [NBu<sub>4</sub>][BPh<sub>4</sub>] as the supporting electrolyte, where arrows indicate the scan direction (Pt disk working electrode, 100 mV/s scan rate, referenced to the Fc/Fc<sup>+</sup> couple)(anodic scan).

# G. UV/Vis/NIR Spectra



Figure S62. UV/Vis/NIR spectrum of A in THF (0.2 mM).



Figure S63. UV/Vis/NIR spectrum of 1 in THF (0.4 mM).



Figure S64. UV/Vis/NIR spectrum of 2 in THF (0.4 mM).



Figure S65. UV/Vis/NIR spectrum of 3 in THF (0.7 mM).

### H. Computational Details

All the DFT calculations were performed using Gaussian09 suite of programs.<sup>10</sup> Hybrid DFT functional (B3PW91) along with SDD basis sets for U atoms and Pople (6-31G\*\*) basis set were used for the rest of the atoms are used for geometry optimization and frequency calculations to obtain thermal corrections.<sup>11–15</sup>



Scheme S1. Thermodynamic parameters calculated for the disproportionation mechanism starting from complex A.



Scheme S2. Thermodynamic parameters calculated for the reduction mechanism starting from complex A.

Table S3. Energetics of two spin states computed for L<sup>cyclam</sup>UC<sub>6</sub>H<sub>6</sub>UL<sup>cyclam</sup> complex (3)

	$\Delta H$ ( $\Delta G$ ), kcal/mol
s=2	0.0
s=3	4.8 (2.5)

Table S4. Comparison of selected bond parameters (Å) between DFT optimized and X-ray structures

Atom labels	DFT		X-ray	
	s=2	s=3		
U1-O2	2.20	2.21	2.21	
U1-O3	2.21	2.22	2.22	N137
U1-N5	2.93	2.91	2.91	NIB6
U1-N6	2.77	2.81	2.74	$\begin{array}{c} 02 \\ 036 \\ 0134 \end{array}$
U1-C13	2.67	2.77	2.60	C43
U1-C36	2.52	2.72	2.57	U1 U132
U1-C43	2.56	2.81	2.60	N4 C130 C130
U1-C126	2.55	2.79	2.62	
U1-C128	2.72	2.92	2.66	
U1-C130	2.64	2.74	2.62	
C13-C36	1.45	1.47	1.44	
C13-C43	1.45	1.39	1.42	
C36-C130	1.45	1.45	1.43	
C43-C128	1.46	1.45	1.43	
C126-C128	1.46	1.47	1.44	
C126-C130	1.45	1.39	1.42	
U132-C13	2.56	2.73	2.62	•

U132-C36	2.70	2.78	2.66
U132-C43	2.67	2.71	2.62
U132-C126	2.67	2.70	2.60
U132-C128	2.50	2.62	2.57
U132-C130	2.56	2.73	2.60
U132-O133	2.22	2.22	2.21
U132-O134	2.21	2.22	2.22
U132-N136	2.95	2.95	2.91
	0 70	0.05	0.74
U132-N136	2.95	2.95	2

Table S5. Computed natural charges for  $LUC_6H_6UL$  complex 3 (s=2)

Atom labels	Natural charges	
	s=2	s=3
U1	1.39949	1.42115
02	-0.75391	-0.79448
O3	-0.75235	-0.79319
N5	-0.52265	-0.53882
N6	-0.52895	-0.54482
C13	-0.46035	-0.36383
C36	-0.48488	-0.55808
C43	-0.46653	-0.34051
C126	-0.46255	-0.36325
C128	-0.49215	-0.55767
C130	-0.45315	-0.34339
O133	-0.76312	-0.78140
O134	-0.75775	-0.77846
N136	-0.52347	-0.53349
N137	-0.52713	-0.53573

Table S6. Computed Wiberg bond index between selected atoms in  $LUC_6H_6UL$  complex 3 (s=2, s=3)

Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U1	0.0000	U1	0.0000
02	0.6985	O3	0.7005	N5	0.1709
	0.6144		0.6122		0.1702
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U1	0.0000	U1	0.0000
N6	0.1928	C13	0.2763	C36	0.3955
	0.1806		0.1761		0.2656
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U1	0.0000	U1	0.0000
C43	0.3957	C126	0.3987	C128	0.2430
	0.1748		0.1890		0.2000
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U132	0.0000	U132	0.0000
C130	0.2772	C13	0.3771	C36	0.2556
	0.1702		0.2515		0.2838
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U132	0.0000	U132	0.0000	U132	0.0000
C43	0.2579	C126	0.2634	C128	0.4263
	0.2141		0.2443		0.3560
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U132	0.0000	U132	0.0000	U132	0.0000

C130	0.3788	O133	0.6667	O134	0.6813
	0.2264		0.6291		0.6288
Atom labels	Wiberg bond	Atom labels	Wiberg bond		
	index		index		
U132	0.0000	U132	0.0000		
N136	0.1645	N137	0.1892		
	0.1685		0.1826		

Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of LUC<sub>6</sub>H<sub>6</sub>UL complex **3** (s=2)

(0.99074) BD (1) U 1- O 2 (7.07%) 0.2658\* U 1 s(16.26%)p 0.02(0.29%)d 3.19(51.85%) f 1.94(31.54%)g 0.00(0.05%) (92.93%) 0.9640\* O 2 s(63.78%)p 0.57(36.22%)d 0.00(0.00%) (0.97313) BD (2) U 1- O 2 (7.21%) 0.2686\* U 1 s(0.66%)p 0.71(0.47%)d71.60(47.35%)f77.85(51.48%)g 0.04(0.03%) (92.79%) 0.9632\* O 2 s( 0.25%)p99.99(99.74%)d 0.06( 0.01%) (0.95673) BD (1) U 1-N 6 ( 5.64%) 0.2375\* U 1 s( 18.65%)p 0.05( 0.86%)d 3.07( 57.27%) f 1.25( 23.22%)g 0.00( 0.01%) (94.36%) 0.9714\* N 6 s(18.62%)p 4.37(81.36%)d 0.00(0.02%) (0.94132) BD (1) C 13- C 36 ( 49.80%) 0.7057\* C 13 s( 33.58%)p 1.98( 66.36%)d 0.00( 0.07%) (50.20%) 0.7085\* C 36 s( 33.23%)p 2.01( 66.71%)d 0.00( 0.07%) (0.94176) BD (1) C 13- C 43 (50.14%) 0.7081\* C 13 s( 34.18%)p 1.92( 65.75%)d 0.00( 0.07%) ( 49.86%) 0.7061\* C 43 s( 33.43%)p 1.99( 66.50%)d 0.00( 0.07%) (0.94211) BD (1) C 36- C130 (50.18%) 0.7084\* C 36 s( 33.12%)p 2.02( 66.81%)d 0.00( 0.07%) (49.82%) 0.7059\* C130 s(33.75%)p 1.96(66.18%)d 0.00(0.07%) (0.94187) BD (1) C 43- C128 (49.89%) 0.7063\* C 43 s( 33.52%)p 1.98( 66.41%)d 0.00( 0.07%) (50.11%) 0.7079\* C128 s(32.99%)p 2.03(66.94%)d 0.00(0.07%) (0.94180) BD (1) C126- C128 (49.78%) 0.7056\* C126 s( 33.44%)p 1.99( 66.49%)d 0.00( 0.07%) ( 50.22%) 0.7086\* C128 s( 33.31%)p 2.00( 66.62%)d 0.00( 0.07%) (0.93895) BD (1) C126- C130 ( 50.00%) 0.7071\* C126 s( 33.77%)p 1.96( 66.16%)d 0.00( 0.07%) ( 50.00%) 0.7071\* C130 s( 33.63%)p 1.97( 66.30%)d 0.00( 0.07%) (0.99157) BD (1) U132- O133 (7.14%) 0.2672\* U132 s(9.44%)p 0.03(0.25%)d 5.99(56.54%)f 3.57(33.72%)g 0.00(0.04%) (92.86%) 0.9636\* O133 s( 63.73%)p 0.57( 36.26%)d 0.00( 0.00%) (0.96789) BD (2) U132- O133 (6.09%) 0.2467\* U132 s(0.16%)p 5.29(0.84%)d99.99(61.87%)f99.99(37.08%)g 0.27(0.04%) (93.91%) 0.9691\* O133 s( 0.00%)p 1.00( 99.98%)d 0.00( 0.02%) (0.99143) BD (1) U132- O134 (7.24%) 0.2691\* U132 s(8.55%)p 0.02(0.18%)d 6.65(56.87%)f 4.02(34.35%)g 0.01(0.05%) (92.76%) 0.9631\* O134 s(63.97%)p 0.56(36.02%)d 0.00(0.01%) (0.97543) BD (2) U132- O134 (7.06%) 0.2657\* U132 s(0.42%)p 1.15(0.49%)d74.86(31.81%)f99.99(67.25%)g 0.06(0.03%) ( 92.94%) 0.9641\* O134 s( 0.29%)p99.99( 99.69%)d 0.08( 0.02%) (0.95597) BD (1) U132- N137 (5.57%) 0.2359\* U132 s(13.31%)p 0.08(1.03%)d 4.61(61.40%)f 1.82(24.25%)g 0.00(0.01%) (94.43%) 0.9718\* N137 s(18.30%)p 4.46(81.68%)d 0.00(0.02%)

Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of LUC6H6UL complex **3** (s=3) (0.98768) BD (1) U 1- O 2 (6.03%) 0.2455\* U 1 s(11.72%)p 0.03(0.38%)d 6.35(74.46%)f 1.14(13.37%)g 0.01(0.06%) (93.97%) 0.9694\* O 2 s(61.55%)p 0.62(38.44%)d 0.00(0.00%) (0.97177) BD (2) U 1- O 2 (6.24%) 0.2497\* U 1 s(0.32%)p 2.33(0.74%)d99.99(46.82%)f99.99(52.09%)g 0.09(0.03%) (93.76%) 0.9683\* O 2 s(0.09%)p99.99(99.89%)d 0.20(0.02%) (0.99026) BD (1) U 1- O 3 (6.06%) 0.2461\* U 1 s(13.98%)p 0.03(0.35%)d 5.20(72.63%)f 0.93(12.98%)g 0.00(0.06%) (93.94%) 0.9692\* O 3 s( 64.08%)p 0.56( 35.91%)d 0.00( 0.01%) (0.96979) BD (2) U 1-O 3 (5.79%) 0.2407\* U 1 s(0.80%)p 0.55(0.44%)d47.65(38.33%)f75.09(60.40%)g 0.03(0.03%) (94.21%) 0.9706\* O 3 s( 0.12%)p99.99(99.85%)d 0.24( 0.03%) (0.95751) BD (1) U 1-N 6 (5.14%) 0.2267\* U 1 s(13.38%)p 0.05(0.66%)d 4.65(62.17%)f 1.78(23.78%)g 0.00(0.01%) (94.86%) 0.9740\* N 6 s(18.89%)p 4.29(81.09%)d 0.00(0.02%) (0.97799) BD (1) U 1- U132 (69.37%) 0.8329\* U 1 s( 0.03%)p 0.56( 0.01%)d99.99( 5.52%)f99.99( 94.43%)g 0.07( 0.00%) (30.63%) 0.5534\* U132 s(0.00%)p 1.00(0.01%)d99.99(5.31%)f99.99(94.68%)g 0.06(0.00%) (0.95994) BD (1) C 13- C 36 (50.45%) 0.7103\* C 13 s( 33.45%)p 1.99( 66.49%)d 0.00( 0.06%) (49.55%) 0.7039\* C 36 s( 33.31%)p 2.00( 66.63%)d 0.00( 0.06%) (0.96145) BD (1) C 13- C 43 (49.68%) 0.7049\* C 13 s( 35.82%)p 1.79( 64.11%)d 0.00( 0.07%) (50.32%) 0.7094\* C 43 s(35.75%)p 1.80(64.18%)d 0.00(0.06%) (0.96093) BD (1) C 36- C130 (49.24%) 0.7017\* C 36 s( 33.71%)p 1.96( 66.23%)d 0.00( 0.06%) (50.76%) 0.7125\* C130 s( 33.80%)p 1.96( 66.14%)d 0.00( 0.06%) (0.96234) BD (1) C 43- C128 (50.97%) 0.7139\* C 43 s( 33.71%)p 1.96( 66.23%)d 0.00( 0.06%) (49.03%) 0.7002\* C128 s( 32.52%)p 2.07( 67.41%)d 0.00( 0.07%) (0.96087) BD (1) C126- C128 (50.65%) 0.7117\* C126 s(33.38%)p 1.99(66.56%)d 0.00(0.06%) (49.35%) 0.7025\* C128 s( 32.30%)p 2.09( 67.64%)d 0.00( 0.06%) (0.96028) BD (1) C126- C130 (49.76%) 0.7054\* C126 s(35.66%)p 1.80(64.28%)d 0.00(0.07%) (50.24%) 0.7088\* C130 s(35.65%)p 1.80(64.29%)d 0.00(0.06%) (0.98815) BD (1) U132- O133 (6.39%) 0.2527\* U132 s(12.21%)p 0.02(0.30%)d 5.84(71.31%)f 1.32(16.12%)g 0.00(0.05%) (93.61%) 0.9675\* O133 s(62.62%)p 0.60(37.38%)d 0.00(0.00%) (0.97124) BD (2) U132- O133 (6.39%) 0.2528\* U132 s(0.43%)p 1.53(0.66%)d99.99(49.69%)f99.99(49.19%)g 0.06(0.03%) (93.61%) 0.9675\* O133 s( 0.10%)p99.99(99.88%)d 0.18( 0.02%) (0.99028) BD (1) U132- O134 (6.31%) 0.2512\* U132 s(15.07%)p 0.02(0.30%)d 4.79(72.17%)f 0.82(12.40%)g 0.00(0.06%) (93.69%) 0.9679\* O134 s(64.11%)p 0.56(35.88%)d 0.00(0.01%) (0.97057) BD (2) U132- O134 (5.89%) 0.2426\* U132 s(0.76%)p 0.58(0.44%)d48.29(36.83%)f81.22(61.94%)g 0.03(0.03%) (94.11%) 0.9701\* O134 s( 0.19%)p99.99(99.78%)d 0.15( 0.03%) (0.95674) BD (1) U132- N137 (5.28%) 0.2299\* U132 s(14.28%)p 0.05(0.69%)d 4.19(59.84%)f 1.76(25.19%)g 0.00(0.01%) (94.72%) 0.9732\* N137 s(18.63%)p 4.37(81.35%)d 0.00(0.02%)

Donor NBO	Acceptor NBO	E(2)
		kcal/mol
(0.92040) LP ( 1) O 3	(0.09933) LV ( 5) U 1	58.90
s( 62.91%)p 0.59( 37.09%)d 0.00( 0.00%)	s( 22.39%)p 0.01( 0.27%)d 2.37( 53.01%)f 1.08(	
	24.28%)g 0.00( 0.05%)	
(0.90714) LP ( 2) O 3	(0.11566) LV ( 3) U 1	17.16
s( 0.19%)p99.99( 99.78%)d 0.12( 0.02%)	s( 0.03%)p11.03( 0.31%)d99.99( 35.42%)f99.99(	
	64.22%)g 0.87( 0.02%)	
(0.91372) LP ( 1) N 5	(0.08581) LV ( 6) U 1	15.94
s( 16.47%)p 5.07( 83.50%)d 0.00( 0.03%)	s( 25.50%)p 0.02( 0.62%)d 1.96( 49.96%)f 0.94(	
	23.91%)g 0.00( 0.01%)	
(0.61258) LP ( 1) C 13	(0.28128) LV ( 1) U 1	6.14
s( 0.38%)p99.99( 99.55%)d 0.18( 0.07%)		

Table S7. NBO Second order perturbation analysis (AMO) for LUC<sub>6</sub>H<sub>6</sub>UL complex 3 (s=2)

	s( 0.25%)p 0.91( 0.22%)d99.99( 31.43%)f99.99(	
	68.09%)g 0.02( 0.01%)	
(0.61258) LP ( 1) C 13	(0.10748) LV ( 4) U 1	6.81
s(0.38%)p99.99(99.55%)d0.18(0.07%)	s( 0.69%)p 1.19( 0.82%)d99 99( 76.80%)f31 47(	
	21 68%)g 0.01( 0.01%)	
(0.62644) LP (1) C 36	(0.23068)   V(2)   1	20.05
c(1,83%) $p53,76(.08,11%)$ $d(0,03(.0,06%))$	(0.20000) = (2)	20.00
s( 1.05%)p35.76( 96.11%)d 0.05( 0.00%)	S(-0.30%)p 0.43(-0.15%)u + 0.37(-27.39%) + 99.99(-27.39	
	72.08%)g 0.04( 0.01%)	0.01
(0.61494) LP (1) C 43	(0.28128) LV (1) U 1	8.61
s( 1.03%)p96.09( 98.91%)d 0.06( 0.06%)	s( 0.25%)p 0.91( 0.22%)d99.99( 31.43%)f99.99(	
	68.09%)g 0.02( 0.01%)	
(0.61494) LP ( 1) C 43	(0.23068) LV ( 2) U 1	11.44
s( 1.03%)p96.09( 98.91%)d 0.06( 0.06%)	s( 0.36%)p 0.43( 0.15%)d76.37( 27.39%)f99.99(	
	72.08%)g 0.04( 0.01%)	
(0.61534) LP ( 1) C126	(0.28128) LV ( 1) U 1	27.21
s(0.81%)p99.99(99.12%)d0.08(0.07%)	s( 0.25%)p 0.91( 0.22%)d99.99( 31.43%)f99.99(	
	68.09%)g 0.02( 0.01%)	
(0.60890) LP (1) C130	(0.23068)   V(2)     1	4.23
s(0.37%)p99.99(99.56%)d.0.19(0.07%)	s(0.36%) p 0.43(0.15%) d76.37(27.39%) f99.99(	
	72 08%)a 0.04( 0.01%)	
		7 57
(0.94132) BD $(1)$ C 13- C 30 (40.80%) 0.7057* C 12 c( 22.58%) = 1.08(	$(0.10740) \ge V(4) = 0$	1.51
(49.80%) 0.7057°C 13 S( 33.58%)p 1.98(	S(-0.89%)p - 1.19(-0.82%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)131.47(-0.46%)a99.99(-76.80%)a99.90(-76.80%)a99.99(-76.80%)a99.90(-76.80%)a99.90(-76.80%)a99.90%)a99.90(-76.80%)a99.90(-76.80%)a99.90%)a99.90(-76.80%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a99.90%)a90%)a99.90%)a90%)a90%)a90%)a90%)a90%)a90%)a90%)a	
	21.68%)g 0.01( 0.01%)	
(50.20%) 0.7085* C 36 s( 33.23%)p 2.01(		
66.71%)d 0.00( 0.07%)		
(0.94176) BD ( 1) C 13- C 43	(0.08079) LV ( 7) U 1	8.33
(50.14%) 0.7081* C 13 s( 34.18%)p 1.92(	s( 0.89%)p 0.49( 0.43%)d24.46( 21.76%)f86.41(	
65.75%)d 0.00( 0.07%)	76.89%)g 0.03( 0.02%)	
(49.86%) 0.7061* C 43 s( 33.43%)p 1.99(		
66.50%)d 0.00( 0.07%)		
(0.94211) BD (1) C 36- C130	(0.06202) LV ( 8) U 1	9.19
(50.18%) 0.7084* C 36 s( 33.12%)p 2.02(	s( 3.23%)p 0.28( 0.91%)d11.94( 38.59%)f17.70(	
66.81%)d 0.00( 0.07%)	57.21%)g 0.02( 0.06%)	
(49.82%) 0.7059* C130 s(33.75%)p 1.96(		
66 18%)d 0 00( 0 07%)		
(0.94187) BD ( 1) C 43- C128	(0.08581)   \/ ( 6)     1	5 36
(49.80%) 0.7063* C 43 s( 33.52%) n 1.98(	s(25.50%) = 0.02(-0.62%) d + 0.96%) f + 0.94(-0.000) f	0.00
66 41%)d 0.00( 0.07%)	3(20.00%) 0.02(-0.02%) 0.100(+0.00%)(-0.04(-0.04%))(-0.04(-0.04%)))	
(50.41%)(0.00(0.07%))	23.9176)g 0.00( 0.0176)	
( 50.11%) 0.7079 C128 S( 52.99%) p 2.03(		
		7.00
(0.94180) BD (1) C126- C128	(0.10748) LV $(4)$ U 1	7.98
(49.78%) 0.7056* C126 s( 33.44%)p 1.99(	s( 0.69%)p 1.19( 0.82%)d99.99( 76.80%)f31.47(	
66.49%)d 0.00( 0.07%)	21.68%)g 0.01( 0.01%)	
( 50.22%) 0.7086* C128 s( 33.31%)p 2.00(		
66.62%)d 0.00( 0.07%)		
(0.93895) BD ( 1) C126- C130	(0.06202) LV ( 8) U 1	8.33
( 50.00%) 0.7071* C126 s( 33.77%)p 1.96(	s( 3.23%)p 0.28( 0.91%)d11.94( 38.59%)f17.70(	
66.16%)d 0.00( 0.07%)	57.21%)g 0.02( 0.06%)	
(50.00%) 0.7071* C130 s( 33.63%)p 1.97(		
66.30%)d 0.00( 0.07%)		
(0.61258) LP ( 1) C 13	(0.27612) LV ( 1) U132	18.25
s(0.38%)p99.99(99.55%)d0.18(0.07%)	s(0,21%)p0,28(0,06%)d99,99(29,42%)f99,99(	
	70.31%)g 0.03( 0.01%)	
(0.61258) LP (1) C 13	(0.21766)   V ( 2)   132	5.38
(0.01200) = (1) 0 10 (0.01200) = (1) 0 10 (0.01200) = (1) 0 10	(2.21100) = 1(2) 0.102 s(2.14%) = 0.23(-0.49%) d14 04(-30.11%) f21.25(-0.10%)	0.00
	3(2.1470) $0.23(0.4370)$ $0.14.04(30.1170)$ $1.33(67.240)$ $0.01(0.020)$	
		<u> </u>
(0.01208) LP (1) U 13	(0.10407) LV (3) U132	5.55
s(_0.38%)p99.99(_99.55%)d_0.18(_0.07%)	s( 0.70%)p 1.22( 0.86%)d99.99( 81.93%)f23.47(	
	16.50%)g 0.01( 0.01%)	
(0.62644) LP ( 1) C 36	(0.27612) LV ( 1) U132	5.27
s( 1.83%)p53.76( 98.11%)d 0.03( 0.06%)		

	s(0.21%)p0.28(0.06%)d99.99(29.42%)f99.99(	
	70.31%)g 0.03( 0.01%)	
(0.61494) I P (1) C 43	(0.21766)   V(2)     132	7 13
(0.01404) = (1) = (1) = 40 (1) = (1) = (1) = (1)	s(214%) = 0.23(0.49%) d14.04(30.11%) f31.35(	7.10
s( 1.00 %)p30.03( 30.31 %)d 0.00( 0.00 %)	(2.14%) = 0.23(-0.43%) = 0.01(-0.02%)	
(0.61524) L D ( 1) C126	(0.10407) + 1/(2) + 1122	7 10
(0.01334) LF (1) C120	(0.10407) EV(3) 0132	7.19
s( 0.81%)p99.99( 99.12%)d 0.08( 0.07%)	s( 0.70%)p 1.22( 0.86%)d99.99( 81.93%)t23.47(	
	16.50%)g 0.01( 0.01%)	
(0.63105) LP (1) C128	(0.27612) LV ( 1) U132	24.01
s( 2.08%)p47.02( 97.86%)d 0.03( 0.06%)	s( 0.21%)p 0.28( 0.06%)d99.99( 29.42%)f99.99(	
	70.31%)g 0.03( 0.01%)	
(0.60890) LP ( 1) C130	(0.21766) LV ( 2) U132	14.10
s( 0.37%)p99.99( 99.56%)d 0.19( 0.07%)	s( 2.14%)p 0.23( 0.49%)d14.04( 30.11%)f31.35(	
	67.24%)g 0.01( 0.02%)	
(0.94132) BD ( 1) C 13- C 36	(0.10407) LV ( 3) U132	6.10
(49.80%) 0.7057* C 13 s( 33.58%)p 1.98(	s( 0.70%)p 1.22( 0.86%)d99.99( 81.93%)f23.47(	
66.36%)d 0.00( 0.07%)	16.50%)g 0.01( 0.01%)	
(50.20%) 0.7085* C 36 s( 33.23%)p 2.01(		
66.71%)d 0.00( 0.07%)		
(0.94176) BD (1) C 13- C 43	(0.10407) LV ( 3) U132	5.97
(50 14%) 0 7081* C 13 s( 34 18%)p 1 92(	s( 0,70%)p 1,22( 0,86%)d99,99( 81,93%)f23,47(	0.01
65 75%)d 0.00( 0.07%)	1650%) = 0.01(-0.01%)	
(49.86%) = 0.7061* C.43 s(33.43%) p 1.99(		
66 50%)d 0.00( 0.07%)		
(0.94211) BD $(1)$ C 36 C130	(0.00344) I.V. ( 4) 11132	6.95
(50.18%) 0.7084* C 36 s( 33.12%) p 2.02(	$(0.03044) \ge 0.03(-0.71\%) d 1.75(.43.01\%) f 1.21(-0.030) f 1.21(-$	0.95
66 91%) d 0.00( 0.07%)	3(25.05%) + 0.05(-0.11%) + 1.75(-45.51%) + 1.21(-20.25%) + 0.00(-0.01%)	
(40.82%) = 0.00(-0.07%)	30.35%)g 0.00( 0.01%)	
(49.82%) 0.7059 C150 S( 55.75%)p 1.90(		
	(0.40407) 1) ( ( 0) 11400	7.04
(0.94180) BD (1) C126- C128	(0.10407) LV (3) $0.132$	7.31
(49.78%) 0.7056° C126 s( 33.44%)p 1.99(	s( 0.70%)p 1.22( 0.86%)d99.99( 81.93%)t23.47(	
66.49%)d 0.00( 0.07%)	16.50%)g 0.01( 0.01%)	
(50.22%) 0.7086* C128 s( 33.31%)p 2.00(		
66.62%)d 0.00( 0.07%)		
(0.94180) BD ( 1) C126- C128	(0.09344) LV ( 4) U132	9.39
(49.78%) 0.7056* C126 s( 33.44%)p 1.99(	s( 25.03%)p 0.03( 0.71%)d 1.75( 43.91%)f 1.21(	
66.49%)d 0.00( 0.07%)	30.35%)g 0.00( 0.01%)	
(50.22%) 0.7086* C128 s( 33.31%)p 2.00(		
66.62%)d 0.00( 0.07%)		
(0.93895) BD ( 1) C126- C130	(0.10407) LV ( 3) U132	7.19
(50.00%) 0.7071* C126 s( 33.77%)p 1.96(	s( 0.70%)p 1.22( 0.86%)d99.99( 81.93%)f23.47(	
66.16%)d 0.00( 0.07%)	16.50%)g 0.01( 0.01%)	
(50.00%) 0.7071* C130 s(33.63%)p 1.97(		
66.30%)d 0.00( 0.07%)		
(0.93895) BD (1) C126- C130	(0.09344) LV ( 4) U132	9.81
(50.00%) 0.7071* C126 s( 33.77%)p 1.96(	s( 25.03%)p 0.03( 0.71%)d 1.75( 43.91%)f 1.21(	
66.16%)d 0.00( 0.07%)	30.35%)g 0.00( 0.01%)	
(50.00%) 0.7071* C130 s( 33.63%)p 1.97(		
66.30%)d 0.00( 0.07%)		
(0.91437) LP (1) N136	(0.05768) LV (5) U132	7.63
s(15,25%) = 5,56(84,72%) + 0.00(-0.03%)	s(18,80%) = 0.01(-0.21%) d = 0.73(-13,74%) f = 58(-0.01%) d = 0.73(-13,74%) f = 58(-0.01%) d = 0.73(-0.01%) d = 0.01%) d = 0.73(-0.01%) d = 0.73(-0.01%) d = 0.01%) d = 0.01%	1.00
	67.24%a = 0.00(-0.02%)	
	01.27/0/9 0.00( 0.02/0)	1

### Table S8. NBO Second order perturbation analysis (AMO) for $LUC_6H_6UL$ complex 3 (s=3)

Donor NBO	Acceptor NBO	E(2)
		kcal/mol
(0.91402) LP ( 1) N 5	(0.07547) LV ( 3) U 1	7.57
s( 19.41%)p 4.15( 80.56%)d 0.00( 0.02%)	s( 40.75%)p 0.02( 0.72%)d 1.04( 42.18%)f 0.40(	
	16.34%)g 0.00( 0.01%)	
(0.91402) LP ( 1) N 5	(0.05585) LV ( 4) U 1	8.57

s( 19.41%)p 4.15( 80.56%)d 0.00( 0.02%)	s( 7.96%)p 0.02( 0.14%)d 0.96( 7.66%)f10.58(	
	84.22%)g 0.00( 0.02%)	
(0.91423) LP ( 1) N136	(0.08336) LV ( 3) U132	7.00
s( 18.10%)p 4.52( 81.88%)d 0.00( 0.03%)	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
	19.67%)g 0.00( 0.01%)	
(0.91423) LP ( 1) N136	(0.05564) LV ( 4) U132	8.42
s( 18.10%)p 4.52( 81.88%)d 0.00( 0.03%)	s( 11.35%)p 0.00( 0.05%)d 0.95( 10.78%)f 6.86(	
	77.80%)g 0.00( 0.02%)	
(0.56669) LP ( 1) C 13	(0.09356) LV ( 2) U 1	5.53
s(0,16%)p99,99(99,78%)d 0,38(0,06%)	s(3.51%)p 0.17(0.58%)d22.91(80.43%)f 4.41(	
	15 47%)g 0.00(_0.00%)	
$(0.56669) \downarrow P (1) C 13$		3.04
(0.00009) EF(1) C 13	$(0.09003) \ge (2) 0.132$ c( 1.38%) p 0.53( 0.73%) d51.42( 70.88%) f10.50(	5.04
s( 0.10%)p39.99( 99.70%)d 0.30( 0.00%)	3(1.30%) 0.33(0.13%) 0.34(1.42(10.00%) 19.39(10.00%) 19.39(10.00%	
		11 54
(0.05887) LP (1) C 30		11.54
s( 0.02%)p99.99( 99.94%)d 2.37( 0.04%)	s( 0.25%)p 0.33( 0.08%)d99.22( 25.19%)f99.99(	
	74.47%)g 0.03( 0.01%)	
(0.65887) LP ( 1) C 36	(0.23573) LV ( 1) U132	6.47
s( 0.02%)p99.99( 99.94%)d 2.37( 0.04%)	s( 0.07%)p 0.75( 0.05%)d99.99( 32.89%)f99.99(	
	66.99%)g 0.09( 0.01%)	
(0.60887) LP ( 1) C 43	(0.04438) LV ( 5) U 1	3.54
s( 0.29%)p99.99( 99.66%)d 0.18( 0.05%)	s( 0.41%)p 2.71( 1.12%)d80.54( 33.31%)f99.99(	
	65.09%)g 0.14( 0.06%)	
(0.60887) LP ( 1) C 43	(0.04763) LV ( 5) U132	7.29
s( 0.29%)p99.99( 99.66%)d 0.18( 0.05%)	s( 1.36%)p 0.92( 1.25%)d25.45( 34.66%)f46.01(	
	62.67%)g 0.05( 0.06%)	
(0.56512) LP (1) C126	(0.23573)   V(1)   U132	3.07
s(0.42%)p99.99(99.52%)d 0.15(0.06%)	s( 0.07%)p 0.75( 0.05%)d99.99( 32.89%)f99.99(	0.01
	66 99%)g 0.09( 0.01%)	
(0.56512) L P ( 1) C126	(0.09685) 1 / ( 2) 11132	8.38
(0.00012) ET (1) 0120 c( 0.42%) p00.00( 00.52%) d.0.15( 0.06%)	$(0.00000) \ge (2) 0.002$ s( 1.38%) p.0.53( 0.73%) d51.42( 70.88%) f10.50(	0.00
3( 0.42 %)p35.55( 55.52 %)d 0.15( 0.00 %)	3( 1.50%)p 0.55( 0.75%)d51.42( 70.60%)15.55(	
(0.05820) LD ( 1) 0128		2.04
(0.05829) LP (1) C128	(0.14994) LV $(1)$ U 1	3.04
s( 2.76%)p35.25( 97.20%)d 0.01( 0.04%)	s( 0.25%)p 0.33( 0.08%)d99.22( 25.19%)f99.99(	
	74.47%)g 0.03( 0.01%)	40.47
(0.65829) LP (1) C128	(0.23573) LV (1) U132	19.47
s( 2.76%)p35.25( 97.20%)d 0.01( 0.04%)	s( 0.07%)p 0.75( 0.05%)d99.99( 32.89%)f99.99(	
	66.99%)g 0.09( 0.01%)	
(0.65829) LP ( 1) C128	(0.08336) LV ( 3) U132	3.59
s( 2.76%)p35.25( 97.20%)d 0.01( 0.04%)	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
	19.67%)g 0.00( 0.01%)	
(0.61149) LP ( 1) C130	(0.04438) LV ( 5) U 1	4.97
s( 0.13%)p99.99( 99.82%)d 0.42( 0.05%)	s( 0.41%)p 2.71( 1.12%)d80.54( 33.31%)f99.99(	
	65.09%)g 0.14( 0.06%)	
(0.61149) LP (1) C130	(0.23573) LV ( 1) U132	4.04
s(0,13%)p99,99(99,82%)d 0,42(0,05%)	s( 0.07%)p 0.75( 0.05%)d99.99( 32.89%)f99.99(	
	66.99%)g 0.09( 0.01%)	
(0.61149)   P ( 1) C130	(0.04763)   V ( 5) U132	4.77
s(0,13%)p99,99(99,82%)d 0,42(0,05%)	s(1.36%) p 0.92(-1.25%) d25 45(-34.66%) f46 01(-1.25%) d25 45(-34.6%) d25	
	62 67%)g 0.05(_0.06%)	
0 95994) BD ( 1) C 13- C 36	(0.07547)   V(3)   1	9.08
$(50.45\%)$ 0.7102* C 12 c (22.45%) $\sim$ 1.00(	$(0.070 + 7) \ge (0.00 + 1)$ s(A0.75%) = 0.02(-0.72%) d 1.04(A2.100/) f 0.40(	3.30
66 40%)d 0 00( 0 06%)	3(+0.75%) + 0.02(-0.72%) + 1.04(+2.10%) + 0.40(-16.34%) + 0.00(-0.04%)	
(40.55%) = 0.00(-0.00%)		
(49.55%) U.7U39 U 30 S( 33.31%)P 2.00(		
	(0.00005) () ( ( 0) () (00	5.04
U.95994) BD (1) C 13- C 36	(U.U9685) LV (2) U132	5.24
(50.45%) 0.7103* C 13 s( 33.45%)p 1.99(	s( 1.38%)p 0.53( 0.73%)d51.42( 70.88%)f19.59(	
66.49%)d 0.00( 0.06%)	27.00%)g 0.00( 0.01%)	
(49.55%) 0.7039* C 36 s( 33.31%)p 2.00(		
66.63%)d 0.00( 0.06%)		
	(0 07547) I V ( 3) U 1	6 78

(49.68%) 0.7049* C 13 s( 35.82%)p 1.79(	s( 40.75%)p 0.02( 0.72%)d 1.04( 42.18%)f 0.40(	
64.11%)d 0.00( 0.07%)	16.34%)g 0.00( 0.01%)	
(50.32%) 0.7094* C 43 s( 35.75%)p 1.80(		
64.18%)d 0.00( 0.06%)		
(0.96145) BD (1) C 13- C 43	(0.04438) I V ( 5) U 1	3.63
(49.68%) 0.7049* C 13 s( 35.82%)p 1.79(	s(0,41%)p 2,71(1,12%)d80,54(33,31%)f99,99(	0.00
64 11%)d 0.00( 0.07%)	65 09%)g 0 14( 0 06%)	
(50.32%) 0.7094* C.43 s( 35.75%) n 1.80(	00.00 %)g 0.14( 0.00 %)	
	(0.00226) 1) ( ( 2) 11422	2.00
	(0.08330) LV $(3)$ 0132	3.20
(49.68%) 0.7049° C 13 s( 35.82%)p 1.79(	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
64.11%)d 0.00( 0.07%)	19.67%)g 0.00( 0.01%)	
(50.32%) 0.7094* C 43 s( 35.75%)p 1.80(		
64.18%)d 0.00( 0.06%)		
(0.96145) BD ( 1) C 13- C 43	(0.04763) LV ( 5) U132	4.90
(49.68%) 0.7049* C 13 s( 35.82%)p 1.79(	s( 1.36%)p 0.92( 1.25%)d25.45( 34.66%)f46.01(	
64.11%)d 0.00( 0.07%)	62.67%)g 0.05( 0.06%)	
(50.32%) 0.7094* C 43 s( 35.75%)p 1.80(		
64.18%)d 0.00( 0.06%)		
(0.96093) BD (1) C 36- C130	(0.07547) LV ( 3) U 1	7.89
(49.24%) 0.7017* C 36 s( 33.71%)p 1.96(	s(40.75%)p 0.02(0.72%)d 1.04(42.18%)f 0.40(	
66.23%)d 0.00( 0.06%)	16.34%)g 0.00( 0.01%)	
(50.76%) 0.7125* C130 s( 33.80%)p 1.96(		
66 14%)d 0 00( 0 06%)		
(0.96093) BD ( 1) C 36- C130	(0.04438) I.V. (5) II. 1	3.65
(0.90093) BD (1) C 30- C 130 (40.24%) 0.7017* C 36 c(33.71%) p 1.96(	$(0.04430) \ge (0.04430) \ge (0.0440) $	5.05
	5(0.41%)p 2.71(1.12%)d00.54(55.51%)135.55(	
(50.25%)(0.00(-0.00%))	65.09%)g 0.14( 0.06%)	
(50.76%) 0.7125°C130 S(33.80%)p 1.96(		
66.14%)d 0.00( 0.06%)		5.00
(0.96093) BD (1) C 36- C130	(0.08336) LV ( 3) U132	5.33
(49.24%) 0.7017* C 36 s( 33.71%)p 1.96(	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
66.23%)d 0.00( 0.06%)	19.67%)g 0.00( 0.01%)	
( 50.76%) 0.7125* C130 s( 33.80%)p 1.96(		
66.14%)d 0.00( 0.06%)		
(0.96093) BD ( 1) C 36- C130	(0.04763) LV ( 5) U132	4.12
(49.24%) 0.7017* C 36 s( 33.71%)p 1.96(	s( 1.36%)p 0.92( 1.25%)d25.45( 34.66%)f46.01(	
66.23%)d 0.00( 0.06%)	62.67%)g 0.05( 0.06%)	
(50.76%) 0.7125* C130 s(33.80%)p 1.96(		
66.14%)d 0.00( 0.06%)		
(0.96234) BD ( 1) C 43- C128	(0.07547) LV ( 3) U 1	3.38
(50.97%) 0.7139* C 43 s( 33.71%)p 1.96(	s(40.75%)p 0.02(0.72%)d 1.04(42.18%)f 0.40(	
66 23%)d 0 00( 0 06%)	16.34%)g 0.00(_0.01%)	
(49.03%) = 0.7002* C128 s(32.52%) p 2.07(	10.047.07g 0.00( 0.047.07	
(40.00%) $(0.0002)$ $(0.002.002.002.002.000)$		
	(0.04429) 1) ( ( 5) 11 4	2.25
(0.90234) BD $(1)$ C 43- C 120 ( $(50.079/)$ 0 7120* C 42 $(22.719/)$ 1 0 C	$(0.04430) \ge 0.01$	3.35
	S(0.41%)p 2.71(-1.12%)d60.54(-55.51%)199.99(-55.00%) = 0.14(-0.06%)	
	65.09%)g 0.14( 0.06%)	
(49.03%) 0.7002* C128 s( 32.52%)p 2.07(		
67.41%)d 0.00( 0.07%)		
(0.96234) BD ( 1) C 43- C128	(0.08336) LV ( 3) U132	7.30
(50.97%) 0.7139* C 43 s( 33.71%)p 1.96(	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
66.23%)d 0.00( 0.06%)	19.67%)g 0.00( 0.01%)	
(49.03%) 0.7002* C128 s( 32.52%)p 2.07(		
67.41%)d 0.00( 0.07%)		
(0.96234) BD (1) C 43- C128	(0.04763) LV ( 5) U132	5.05
(50.97%) 0.7139* C 43 s( 33.71%)p 1.96(	s( 1.36%)p 0.92( 1.25%)d25.45( 34.66%)f46.01(	
66.23%)d 0.00( 0.06%)	62.67%)g 0.05( 0.06%)	
(49.03%) 0.7002* C128 s( 32.52%)n 2.07(		
67 41%)d 0 00( 0 07%)		
(0.96087) BD ( 1) C126 C129	(0.00356) 1 \/ ( 2) 11.1	1 82
(0.30007) DD (1) 0120- 0120		4.03
(50.65%) 0.7117* C126 s( 33.38%)p 1.99(	s( 3.51%)p 0.17( 0.58%)d22.91( 80.43%)f 4.41(	
--	--	------
66.56%)d 0.00( 0.06%)	15.47%)g 0.00( 0.00%)	
(49.35%) 0.7025* C128 s( 32.30%)p 2.09(		
67.64%)d 0.00( 0.06%)		
(0.96087) BD ( 1) C126- C128	(0.09685) LV ( 2) U132	4.16
(50.65%) 0.7117* C126 s( 33.38%)p 1.99(	s( 1.38%)p 0.53( 0.73%)d51.42( 70.88%)f19.59(	
66.56%)d 0.00( 0.06%)	27.00%)g 0.00( 0.01%)	
(49.35%) 0.7025* C128 s( 32.30%)p 2.09(		
67.64%)d 0.00( 0.06%)		
(0.96087) BD ( 1) C126- C128	(0.08336) LV ( 3) U132	9.90
(50.65%) 0.7117* C126 s( 33.38%)p 1.99(	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
66.56%)d 0.00( 0.06%)	19.67%)g 0.00( 0.01%)	
(49.35%) 0.7025* C128 s( 32.30%)p 2.09(		
67.64%)d 0.00( 0.06%)		
(0.96028) BD ( 1) C126- C130	(0.09356) LV ( 2) U 1	3.86
(49.76%) 0.7054* C126 s( 35.66%)p 1.80(	s( 3.51%)p 0.17( 0.58%)d22.91( 80.43%)f 4.41(	
64.28%)d 0.00( 0.07%)	15.47%)g 0.00( 0.00%)	
(50.24%) 0.7088* C130 s(35.65%)p 1.80(		
64.29%)d 0.00( 0.06%)		
(0.96028) BD ( 1) C126- C130	(0.07547) LV ( 3) U 1	3.51
(49.76%) 0.7054* C126 s( 35.66%)p 1.80(	s( 40.75%)p 0.02( 0.72%)d 1.04( 42.18%)f 0.40(	
64.28%)d 0.00( 0.07%)	16.34%)g 0.00( 0.01%)	
( 50.24%) 0.7088* C130 s( 35.65%)p 1.80(		
64.29%)d 0.00( 0.06%)		
(0.96028) BD ( 1) C126- C130	(0.04438) LV ( 5) U 1	4.33
(49.76%) 0.7054* C126 s( 35.66%)p 1.80(	s( 0.41%)p 2.71( 1.12%)d80.54( 33.31%)f99.99(	
64.28%)d 0.00( 0.07%)	65.09%)g 0.14( 0.06%)	
( 50.24%) 0.7088* C130 s( 35.65%)p 1.80(		
64.29%)d 0.00( 0.06%)		
(0.96028) BD ( 1) C126- C130	(0.08336) LV ( 3) U132	8.25
(49.76%) 0.7054* C126 s( 35.66%)p 1.80(	s( 35.55%)p 0.02( 0.64%)d 1.24( 44.13%)f 0.55(	
64.28%)d 0.00( 0.07%)	19.67%)g 0.00( 0.01%)	
( 50.24%) 0.7088* C130 s( 35.65%)p 1.80(		
64.29%)d 0.00( 0.06%)		
(0.96028) BD ( 1) C126- C130	(0.04763) LV ( 5) U132	3.35
(49.76%) 0.7054* C126 s( 35.66%)p 1.80(	s( 1.36%)p 0.92( 1.25%)d25.45( 34.66%)f46.01(	
64.28%)d 0.00( 0.07%)	62.67%)g 0.05( 0.06%)	
( 50.24%) 0.7088* C130 s( 35.65%)p 1.80(		
64.29%)d 0.00( 0.06%)		



**Figure S66.** Computed MOs for LUC<sub>6</sub>H<sub>6</sub>UL complex **3** (s=2). (a)AMO-HOMO-5 (b)AMO-HOMO-4 (c)AMO-HOMO-3 (d)AMO-HOMO-2 (e)AMO-HOMO-1 (f)AMO-HOMO (g)AMO-LUMO (h)BMO-HOMO-1 (i)BMO-HOMO (j)BMO-LUMO (k) spin density plot



**Figure S67.** Computed MOs for LUC<sub>6</sub>H<sub>6</sub>UL complex **3** (s=3). a)AMO-HOMO-6 (b)AMO-HOMO-5 (c)AMO-HOMO-4 (d)AMO-HOMO-3 (e)AMO-HOMO-2 (f)AMO-HOMO-1 (g)AMO- HOMO (h) AMO- LUMO (i)BMO-HOMO (j)BMO-LUMO (k) spin density plot

Table S9. Energetics of two spin states computed for  $^{Ad}LUC_{6}H_{5}CH_{3}UL^{Ad}$  complex

	$\Delta H (\Delta G)$ , kcal/mol
s=2	0.0
s=3	15.3(16.9)



Table S10. Comparison of selected bond parameters (Å) between DFT optimized and X-ray structures for  $^{\rm Ad}LUC_6H_5CH_3UL^{\rm Ad}\,complex$ 

Table S11. Computed natural charges for selected atoms in <sup>Ad</sup>LUC<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>UL<sup>Ad</sup> complex (s=2)

Atom labels	Natural charges
U1	1.24560
U2	1.25222
N3	-0.72521
N4	-0.71393
N5	-0.72082
N6	-0.72498
C7	-0.45072
C8	-0.43667
C9	-0.48909
C10	-0.47808
C11	-0.41635
C12	-0.22285
C77	-0.69044

Table S12. Computed Wiberg bond index between selected atoms in <sup>Ad</sup>LUC<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>UL<sup>Ad</sup> complex (s=2)

Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U1	0.0000	U1	0.0000
N3	0.6414	N4	0.6476	C7	0.3455
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U1	0.0000	U1	0.0000
C8	0.4194	C9	0.2542	C10	0.4521
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U1	0.0000	U1	0.0000	U1	0.0000
C11	0.3253	C12	0.2689	C77	0.0122
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U2	0.0000	U2	0.0000	U2	0.0000
N5	0.6510	N6	0.6590	C7	0.3111
Atom labels	Wiberg bond	Atom labels	Wiberg bond	Atom labels	Wiberg bond
	index		index		index
U2	0.0000	U2	0.0000	U2	0.0000

Atom labelsWiberg bond indexAtom labelsWiberg bond indexMiberg bond indexWiberg bond indexU20.0000U20.0000U20.0000C110.3411C120.3375C770.0110Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of $^{Ad}LUC_6H_5CH_3UL^{Ad}$ comple (0.88011) BD (1) U 1-N 3C120.3375C770.0110Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of $^{Ad}LUC_6H_5CH_3UL^{Ad}$ comple (0.88011) BD (1) U 1-N 31.98%0.3461* U 1 s(0.67%)p 0.51(0.34%)d72.56(48.25%) f76.28(50.73%)g 0.02(0.01%)(88.02%)0.9382* N 3 s(2.76%)p35.23(97.22%)d 0.01(0.02%)0.87678) BD (1) U 1-N 41.195%)0.3456* U 1 s(0.81%)p 0.44(0.35%)d61.23(49.37%) f61.35(49.46%)g 0.02(0.01%)(88.05%)0.9384* N 4 s(3.29%)p29.36(96.68%)d 0.01(0.02%)0.87636) BD (1) U 2-N 51.155%)0.3399* U 2 s(0.20%)p1.87(0.37%)d99.99(52.03%) f99.99(47.38%)g 0.08(0.02%)(0.88262) BD (1) U 2-N 51.19(0.39%)d99.99(54.59%) f99.99(44.68%)g 0.04(0.01%)88.18%)0.9390* N 6 s(2.93%)p33.07(97.05%)d 0.01(0.02%)(0.82005) BD (1) U 2-C 92.50%)0.5079* U 2 s(5.03%)p 0.09(0.43%)d 8.25(41.50%) f10.55(53.03%)g 0.00(0.01%)(74.20%)0.8614* C 9 s(6.33%)p14.78(93.61%)d 0.01(0.05%)0.00(0.01%)(74.20%)				
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$				
U20.0000U20.0000U20.0000C110.3411C120.3375C770.0110Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of $^{Ad}LUC_6H_5CH_3UL^{Ad}$ comple0.88011) BD (1) U 1- N 3(11.98%)0.3461* U 1 s( 0.67%)p 0.51( 0.34%)d72.56( 48.25%) f76.28( 50.73%)g 0.02( 0.01%)(88.02%)0.9382* N 3 s( 2.76%)p35.23( 97.22%)d 0.01( 0.02%)(0.87678) BD (1) U 1- N 4(11.95%)0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23( 49.37%) f61.35( 49.46%)g 0.02( 0.01%)(88.05%)0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%)(0.87836) BD (1) U 2- N 5(11.55%)0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%)(88.45%)0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%)(0.88262) BD (1) U 2- N 6(11.82%)0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%)(88.18%)0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%)(0.82005) BD (1) U 2- C 9(25.80%)0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%)(74.20%)0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%)(0.9300) BD (1) C 7- C 8(150.507)0.60114.72 (0.2076)(150.507)0.60114.72 (0.2076)				
C110.3411C120.3375C770.0110Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of $^{Ad}LUC_{6}H_5CH_3UL^{Ad}$ comple (0.88011) BD (1) U 1-N 3 (11.98%) 0.3461* U 1 s( 0.67%)p 0.51( 0.34%)d72.56( 48.25%) f76.28( 50.73%)g 0.02( 0.01%) (88.02%) 0.9382* N 3 s( 2.76%)p35.23( 97.22%)d 0.01( 0.02%) (0.87678) BD (1) U 1-N 4 (11.95%) 0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23( 49.37%) f61.35( 49.46%)g 0.02( 0.01%) (88.05%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2-N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2-N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2-C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7-C 8 (12.80%) 0.971* U 2 s( 5.03%)p 1.400 (55.04%) 0.01( 0.05%) (0.93300) BD (1) C 7-C 8				
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Bonding orbitals (Alpha Molecular orbitals, AMO) from NBO analysis for ground state spin of <sup>Ad</sup> LUC <sub>6</sub> H <sub>5</sub> CH <sub>3</sub> UL <sup>Ad</sup> completed (0.88011) BD (1) U 1-N 3 (11.98%) 0.3461* U 1 s( 0.67%)p 0.51( 0.34%)d72.56( 48.25%) f76.28( 50.73%)g 0.02( 0.01%) (88.02%) 0.9382* N 3 s( 2.76%)p35.23( 97.22%)d 0.01( 0.02%) (0.87678) BD (1) U 1-N 4 (11.95%) 0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23( 49.37%) f61.35( 49.46%)g 0.02( 0.01%) (88.05%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2-N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2-N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2-C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.9300) BD (1) C 7-C 8				
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(11.98%) 0.3461* U 1 s( 0.67%)p 0.51( 0.34%)d72.56(48.25%) f76.28(50.73%)g 0.02( 0.01%) (88.02%) 0.9382* N 3 s( 2.76%)p35.23( 97.22%)d 0.01( 0.02%) (0.87678) BD (1) U 1- N 4 (11.95%) 0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23(49.37%) f61.35(49.46%)g 0.02( 0.01%) (88.05%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2- N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99(52.03%) f99.99(47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99(54.59%) f99.99(44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
<pre>(1.1000) 0.9382* N 3 s( 2.76%)p35.23( 97.22%)d 0.01( 0.02%) (0.87678) BD (1) U 1- N 4 (11.95%) 0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23( 49.37%) f61.35( 49.46%)g 0.02( 0.01%) (88.05%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2- N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8</pre>				
(0.87678) BD (1) U 1-N 4 (11.95%) 0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23( 49.37%) f61.35( 49.46%)g 0.02( 0.01%) (88.05%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2-N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2-N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2-C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7-C 8				
(11.95%) 0.3456* U 1 s( 0.81%)p 0.44( 0.35%)d61.23( 49.37%) f61.35( 49.46%)g 0.02( 0.01%) (88.05%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2- N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
<pre>(11.50%) 0.9384* N 4 s( 3.29%)p29.36( 96.68%)d 0.01( 0.02%) (0.87836) BD (1) U 2- N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7-C 8</pre>				
(0.87836) BD (1) U 2- N 5 (11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
(11.55%) 0.3399* U 2 s( 0.20%)p 1.87( 0.37%)d99.99( 52.03%) f99.99( 47.38%)g 0.08( 0.02%) (88.45%) 0.9405* N 5 s( 2.89%)p33.57( 97.09%)d 0.01( 0.02%) (0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
$\begin{array}{l} (11.55\%) & 0.0305 \text{ for } 12.5((0.20\%)\text{ p} 1.51((0.31\%)\text{ d} 35.05((0.20\%)\text{ f} 155.05((0.02\%)\text{ g} 0.00((0.02\%)\text{ f} 155.05((0.02\%)\text{ g} 0.00((0.02\%)\text{ g} 0.00((0.01\%)\text{ g} 0.00((0.02\%)\text{ g} 0.00((0.01\%)\text{ g} 0.00((0.0$				
(0.88262) BD (1) U 2- N 6 (11.82%) 0.3438* U 2 s( 0.33%)p 1.19( 0.39%)d99.99( 54.59%) f99.99( 44.68%)g 0.04( 0.01%) (88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
$\begin{array}{l} (11.82\%) & 0.3438^{*} \ U \ 2 \ s( \ 0.33\%)p \ 1.19( \ 0.39\%)d99.99( \ 54.59\%) \ f99.99( \ 44.68\%)g \ 0.04( \ 0.01\%) \\ (88.18\%) & 0.9390^{*} \ N \ 6 \ s( \ 2.93\%)p \ 33.07( \ 97.05\%)d \ 0.01( \ 0.02\%) \\ (0.82005) \ BD \ (1) \ U \ 2-C \ 9 \\ (25.80\%) & 0.5079^{*} \ U \ 2 \ s( \ 5.03\%)p \ 0.09( \ 0.43\%)d \ 8.25( \ 41.50\%) \ f10.55( \ 53.03\%)g \ 0.00( \ 0.01\%) \\ (74.20\%) & 0.8614^{*} \ C \ 9 \ s( \ 6.33\%)p \ 14.78( \ 93.61\%)d \ 0.01( \ 0.05\%) \\ (0.93300) \ BD \ (1) \ C \ 7-C \ 8 \\ (59.65\%) & 0.7474^{*} \ O \ 2 \ (25.80\%) \ 1.020\% \\ (59.65\%) & 0.7474^{*} \ O \ 2 \ (25.80\%) \ 1.020\% \\ \end{array}$				
(88.18%) 0.9390* N 6 s( 2.93%)p33.07( 97.05%)d 0.01( 0.02%) (0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
(0.82005) BD (1) U 2- C 9 (25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD (1) C 7- C 8				
(25.80%) 0.5079* U 2 s( 5.03%)p 0.09( 0.43%)d 8.25( 41.50%) f10.55( 53.03%)g 0.00( 0.01%) (74.20%) 0.8614* C 9 s( 6.33%)p14.78( 93.61%)d 0.01( 0.05%) (0.93300) BD ( 1) C 7- C 8				
$(22.80\%)$ $(0.3079 \ 0.2 \ s(-3.03\%) \ 0.09(-0.45\%) \ 0.25(-41.50\%) \ 10.55(-53.05\%) \ 0.00(-0.07\%)$ $(74.20\%)$ $0.8614^{*}$ C 9 s(-6.33\%) p14.78(-93.61\%) \ 0.01(-0.05\%) (0.93300) BD (-1) C 7-C 8 $(52.55\%)$ $0.3547^{*}$ O 7 (-0.400\%) $-1.00(-0.50\%)$				
(0.93300) BD $(1)$ C 7-C 8 (59350') C 7-C 8				
(50.65%) 0.7117°C 7 S(34.09%)p 1.93(65.84%)d 0.00( 0.06%)				
(49.35%) 0.7025°C 8 S( 33.26%)p 2.00( 66.67%)d 0.00( 0.07%)				
(0.84869) BD $(2)$ C 7-C 8 (50.60%) = 0.74454 C 7-C 8				
(50.62%) 0.7115°C 7 s( 0.01%)p 1.00( 99.89%)d 0.00( 0.10%)				
(49.38%) 0.7027°C 8 s( 0.49%)p99.99( 99.40%)d 0.22( 0.11%)				
(0.93521) BD (1) C 7- C 12				
(49.42%) 0.7030°C 7 s( 34.13%)p 1.93( 65.80%)d 0.00( 0.07%)				
( 50.58%) 0.7112* C 12 s( 33.16%)p 2.01( 66.79%)d 0.00( 0.05%)				
(0.93010) BD (1) C 8- C 9				
(49.94%) 0.7067* C 8 s( 33.44%)p 1.99( 66.49%)d 0.00( 0.07%)				
(50.06%) 0.7075*C 9 s( 31.26%)p 2.20( 68.67%)d 0.00( 0.07%)				
(0.92842) BD (1) C 9- C 10				
(49.86%) 0.7061* C 9 s( 30.78%)p 2.25( 69.15%)d 0.00( 0.07%)				
( 50.14%) 0.7081* C 10 s( 32.85%)p 2.04( 67.08%)d 0.00( 0.06%)				
(0.93316) BD ( 1) C 10- C 11				
( 49.92%) 0.7065* C 10 s( 33.15%)p 2.01( 66.78%)d 0.00( 0.07%)				
( 50.08%) 0.7077* C 11 s( 33.92%)p 1.95( 66.02%)d 0.00( 0.06%)				
(0.93584) BD ( 1) C 11- C 12				
( 49.14%) 0.7010* C 11 s( 34.30%)p 1.91( 65.63%)d 0.00( 0.07%)				
( 50.86%) 0.7132* C 12 s( 33.14%)p 2.02( 66.81%)d 0.00( 0.05%)				
(0.84424) BD ( 2) C 11- C 12				
( 49.52%) 0.7037* C 11 s( 0.02%)p99.99( 99.88%)d 5.58( 0.10%)				
(50.48%) 0.7105* C 12 s( 0.15%)p99.99( 99.76%)d 0.64( 0.09%)				
(50.48%) 0.7105* C 12 s( 0.15%)p99.99( 99.76%)d 0.64( 0.09%) (0.98670) BD (1) C 12- C 77				
(50.48%) 0.7105* C 12 s( 0.15%)p99.99( 99.76%)d 0.64( 0.09%) (0.98670) BD (1) C 12- C 77 (52.61%) 0.7253* C 12 s( 33.41%)p 1.99( 66.55%)d 0.00( 0.04%)				

Table S13. NBO Second order	perturbation analysis	(AMO) for AdLUC <sub>6</sub> H <sub>5</sub> CH <sub>3</sub>	UL <sup>Ad</sup> complex (s=2)
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Donor NBO	Acceptor NBO	E(2)
		kcal/mol
(0.87145) LP ( 1) N 3	(0.15873) LV ( 3) U 1	34.32
s( 23.38%)p 3.28( 76.60%)d 0.00( 0.02%)	s( 3.56%)p 0.07( 0.24%)d23.57( 83.86%)f 3.47(	
	12.33%)g 0.00( 0.01%)	
(0.87145) LP ( 1) N 3	(0.11744) LV ( 5) U 1	12.36
s( 23.38%)p 3.28( 76.60%)d 0.00( 0.02%)	s( 0.16%)p 0.57( 0.09%)d99.99( 79.64%)f99.99(	
	20.10%)g 0.04( 0.01%)	
(0.87145) LP ( 1) N 3	(0.06751) LV ( 6) U 1	9.46

s( 23.38%)p 3.28( 76.60%)d 0.00( 0.02%)	s( 66.20%)p 0.00( 0.32%)d 0.36( 23.59%)f 0.15(	
	9.88%)g 0.00( 0.01%)	
(0.86925) LP ( 1) N 4	(0.11744) LV ( 5) U 1	11.26
s( 23.40%)p 3.27( 76.58%)d 0.00( 0.02%)	s( 0.16%)p 0.57( 0.09%)d99.99( 79.64%)f99.99(	
	20.10%)g 0.04( 0.01%)	
(0.86925) LP (1) N 4	(0.06751) LV ( 6) U 1	6.73
s( 23.40%)p 3.27( 76.58%)d 0.00( 0.02%)	s( 66.20%)p 0.00( 0.32%)d 0.36( 23.59%)f 0.15(	
	9.88%)g 0.00( 0.01%)	
(0.63550) LP ( 1) C 10	(0.32605) LV ( 1) U 1	19.00
s( 1.84%)p53.39( 98.10%)d 0.03( 0.06%)	s( 0.48%)p 0.12( 0.06%)d73.01( 35.09%)f99.99(	
	64.36%)g 0.02( 0.01%)	
(0.63550) LP ( 1) C 10	(0.19233) LV ( 2) U 1	10.21
s( 1.84%)p53.39( 98.10%)d 0.03( 0.06%)	s( 7.54%)p 0.00( 0.03%)d 5.03( 37.91%)f 7.23(	
	54.52%)g 0.00( 0.01%)	
(0.63550) LP ( 1) C 10	(0.13594) LV ( 4) U 1	12.11
s( 1.84%)p53.39( 98.10%)d 0.03( 0.06%)	s( 0.01%)p 1.00( 1.02%)d65.42( 66.66%)f31.71(	
	32.31%)g 0.00( 0.00%)	
(0.93300) BD ( 1) C 7- C 8	(0.13594) LV ( 4) U 1	19.10
(50.65%) 0.7117* C 7 s( 34.09%)p 1.93(	s( 0.01%)p 1.00( 1.02%)d65.42( 66.66%)f31.71(	
65.84%)d 0.00( 0.06%)	32.31%)g 0.00( 0.00%)	
(49.35%) 0.7025* C 8 s( 33.26%)p 2.00(		
66.67%)d 0.00( 0.07%)		
(0.84869) BD (2) C 7- C 8	(0.13594) LV ( 4) U 1	11.26
(50.62%) 0.7115* C 7 s( 0.01%)p 1.00(	s( 0.01%)p 1.00( 1.02%)d65.42( 66.66%)f31.71(	
99.89%)d 0.00( 0.10%)	32.31%)g 0.00( 0.00%)	
(49.38%) 0.7027* C 8 s( 0.49%)p99.99(		
99.40%)d 0.22( 0.11%)		
(0.84869) BD (2) C 7- C 8	(0.19233) LV ( 2) U 1	7.59
(50.62%) 0.7115* C 7 s( 0.01%)p 1.00(	s( 7.54%)p 0.00( 0.03%)d 5.03( 37.91%)f 7.23(	
99.89%)d 0.00( 0.10%)	54.52%)g 0.00( 0.01%)	
(49.38%) 0.7027* C 8 s( 0.49%)p99.99(		
99.40%)d 0.22( 0.11%)		
(0.93521) BD ( 1) C 7- C 12	(0.13594) LV ( 4) U 1	6.33
(49.42%) 0.7030* C 7 s( 34.13%)p 1.93(	s( 0.01%)p 1.00( 1.02%)d65.42( 66.66%)f31.71(	
65.80%)d 0.00( 0.07%)	32.31%)g 0.00( 0.00%)	
( 50.58%) 0.7112* C 12 s( 33.16%)p 2.01(		
66.79%)d 0.00( 0.05%)		
(0.93521) BD ( 1) C 7- C 12	(0.11744) LV ( 5) U 1	9.36
(49.42%) 0.7030* C 7 s( 34.13%)p 1.93(	s( 0.16%)p 0.57( 0.09%)d99.99( 79.64%)f99.99(	
65.80%)d 0.00( 0.07%)	20.10%)g 0.04( 0.01%)	
(50.58%) 0.7112* C 12 s( 33.16%)p 2.01(		
66.79%)d 0.00( 0.05%)		
(0.93010) BD ( 1) C 8- C 9	(0.11744) LV ( 5) U 1	13.56
(49.94%) 0.7067* C 8 s( 33.44%)p 1.99(	s( 0.16%)p 0.57( 0.09%)d99.99( 79.64%)f99.99(	
66.49%)d 0.00( 0.07%)	20.10%)g 0.04( 0.01%)	
(50.06%) 0.7075* C 9 s( 31.26%)p 2.20(		
68.67%)d 0.00( 0.07%)		
(0.92842) BD ( 1) C 9- C 10	(0.13594) LV ( 4) U 1	6.84
(49.86%) 0.7061* C 9 s( 30.78%)p 2.25(	s( 0.01%)p 1.00( 1.02%)d65.42( 66.66%)f31.71(	
69.15%)d 0.00( 0.07%)	32.31%)g 0.00( 0.00%)	
(50.14%) 0.7081* C 10 s( 32.85%)p 2.04(		
67.08%)d 0.00( 0.06%)		
(0.92842) BD ( 1) C 9- C 10	(0.11744) LV ( 5) U 1	12.08
(49.86%) 0.7061* C 9 s( 30.78%)p 2.25(	s( 0.16%)p 0.57( 0.09%)d99.99( 79.64%)f99.99(	
69.15%)d 0.00( 0.07%)	20.10%)g 0.04( 0.01%)	
( 50.14%) 0.7081* C 10 s( 32.85%)p 2.04(		
67.08%)d 0.00( 0.06%)		
(0.93316) BD ( 1) C 10- C 11	(0.13594) LV ( 4) U 1	20.27
(49.92%) 0.7065* C 10 s( 33.15%)p 2.01(	s( 0.01%)p 1.00( 1.02%)d65.42( 66.66%)f31.71(	
66.78%)d 0.00( 0.07%)	32.31%)g 0.00( 0.00%)	

(50.08%) 0.7077* C 11 s( 33.92%)p 1.95(		
66.02%)d 0.00( 0.06%)		10.77
(0.93584) BD $(1)$ C 11- C 12 (40.14%) 0.7010* C 11 c (24.20%) b 1.01(	(0.11744) LV(5) U 1	10.77
65 63%)d 0.00( 0.07%)	S(0.10%)p(0.57(-0.09%)d99.99(-79.04%)199.99(-20.10%)p(0.04(-0.01%))	
(50.86%) 0.7132* C 12 s(.33.14%)p 2.02(	20.10 /0)g 0.04( 0.01/0)	
66.81%)d 0.00( 0.05%)		
(0.84424) BD (2) C 11- C 12	(0.11744)   V ( 5) U 1	5.81
(49.52%) 0.7037* C 11 s( 0.02%)p99.99(	s(0,16%)p0,57(0,09%)d99,99(79,64%)f99,99(	0.01
99.88%)d 5.58( 0.10%)	20.10%)g 0.04( 0.01%)	
(50.48%) 0.7105* C 12 s( 0.15%)p99.99(		
99.76%)d 0.64( 0.09%)		
(0.87007) LP (1) N 5	(0.20993) LV (1) U 2	15.02
s( 23.13%)p 3.32( 76.85%)d 0.00( 0.02%)	s( 0.25%)p 0.76( 0.19%)d99.99( 33.90%)f99.99(	
	65.66%)g 0.01( 0.00%)	
(0.87007) LP ( 1) N 5	(0.14943) LV ( 3) U 2	19.38
s( 23.13%)p 3.32( 76.85%)d 0.00( 0.02%)	s( 2.00%)p 0.26( 0.51%)d42.16( 84.47%)f 6.49(	
	13.00%)g 0.01( 0.01%)	
(0.87007) LP (1) N 5	(0.11688) LV ( 4) U 2	14.76
s( 23.13%)p 3.32( 76.85%)d 0.00( 0.02%)	s( 0.04%)p 1.41( 0.06%)d99.99( 79.43%)f99.99(	
	20.46%)g 0.16( 0.01%)	
(0.87007) LP ( 1) N 5	(0.06796) LV ( 5) U 2	8.01
s( 23.13%)p 3.32( 76.85%)d 0.00( 0.02%)	s( 74.83%)p 0.01( 0.38%)d 0.21( 15.88%)f 0.12(	
	8.89%)g 0.00( 0.01%)	
(0.86957) LP (1) N 6	(0.20993) LV (1) U 2	8.96
s( 23.00%)p 3.35( 76.99%)d 0.00( 0.01%)	s( 0.25%)p 0.76( 0.19%)d99.99( 33.90%)f99.99(	
	65.66%)g 0.01( 0.00%)	21.62
(0.80957) LP (1) N 6	(0.16594) LV (2) U 2	21.63
s(23.00%)p 3.35(78.99%)d 0.00(-0.01%)	S(2.20%)p(0.24(-0.52%)d24.94(-54.96%)r19.16(-42.29%)a(-0.00(-0.01%))	
(0.86957) I.D.( 1) N. 6	42.29%)g 0.00( 0.01%)	15 77
(0.00307) LF (1) 1 (0 s(23.00%) n 3.35(76.99%) d 0.00( $0.01\%$ )	$(0.14943) \ge (0.5) = 2$ s( 2.00%)p 0.26( 0.51%)d42 16( 84.47%)f 6.49(	13.77
	13 00%)g 0.01( 0.01%)	
(0.86957)   P ( 1) N 6	(0.11688)   V (4) U 2	10.97
s(23.00%)p 3.35(76.99%)d 0.00(0.01%)	s( 0.04%)p 1.41( 0.06%)d99.99( 79.43%)f99.99(	10.01
	20.46%)g 0.16( 0.01%)	
(0.86957) LP (1) N 6	(0.06796) LV ( 5) U 2	9.58
s( 23.00%)p 3.35( 76.99%)d 0.00( 0.01%)	s( 74.83%)p 0.01( 0.38%)d 0.21( 15.88%)f 0.12(	
	8.89%)g 0.00( 0.01%)	
(0.63550) LP ( 1) C 10	(0.13259) BD*( 1) U 2- C 9	13.96
s( 1.84%)p53.39( 98.10%)d 0.03( 0.06%)	(74.20%) 0.8614* U 2 s( 5.03%)p 0.09(	
	0.43%)d 8.25( 41.50%)f10.55( 53.03%)g 0.00(	
	0.01%)	
	(25.80%) -0.5079* C 9 s( 6.33%)p14.78(	
	93.61%)d 0.01( 0.05%)	
(0.93300) BD (1) C 7- C 8	(0.11688) LV (4) U 2	14.20
(50.65%) 0.7117°C 7 s( 34.09%)p 1.93(	s( 0.04%)p 1.41( 0.06%)d99.99( 79.43%)f99.99(	
(40.25%) = 0.00(-0.00%)	20.46%)g 0.16( 0.01%)	
66 67%)d 0.00( 0.07%)		
(0.84869) BD (2) C 7-C 8	(0 13250) BD*( 1) IL 2- C 9	10.02
(50.62%) 0.7115* C. 7 s( 0.01%)p 1.00(	(74.20%) 0.8614*11.2 s( 5.03%)p 0.09(	10.02
99 89%)d 0 00( 0 10%)	(44.26%) $(50014$ $(622)$ $(5007%)$ $(5007%)$ $(5007%)$ $(5007%)$ $(6007%$	
(49.38%) 0.7027* C 8 s( 0.49%)p99.99(	0.01%)	
99.40%)d 0.22( 0.11%)	(25.80%) -0.5079* C 9 s( 6.33%)p14.78(	
	93.61%)d 0.01( 0.05%)	
(0.93521) BD ( 1) C 7- C 12	(0.06796) LV ( 5) U 2	5.38
(49.42%) 0.7030* C 7 s( 34.13%)p 1.93(	s( 74.83%)p 0.01( 0.38%)d 0.21( 15.88%)f 0.12(	
65.80%)d 0.00( 0.07%)	8.89%)g 0.00( 0.01%)	
( 50.58%) 0.7112* C 12 s( 33.16%)p 2.01(		
66.79%)d 0.00( 0.05%)		

(0.93010) BD ( 1) C 8- C 9	(0.14943) LV ( 3) U 2	8.23
(49.94%) 0.7067* C 8 s( 33.44%)p 1.99(	s( 2.00%)p 0.26( 0.51%)d42.16( 84.47%)f 6.49(	
66.49%)d 0.00( 0.07%)	13.00%)g 0.01( 0.01%)	
(50.06%) 0.7075* C 9 s( 31.26%)p 2.20(		
68.67%)d 0.00( 0.07%)		
(0.92842) BD ( 1) C 9- C 10	(0.14943) LV ( 3) U 2	7.14
(49.86%) 0.7061* C 9 s( 30.78%)p 2.25(	s( 2.00%)p 0.26( 0.51%)d42.16( 84.47%)f 6.49(	
69.15%)d 0.00( 0.07%)	13.00%)g 0.01( 0.01%)	
( 50.14%) 0.7081* C 10 s( 32.85%)p 2.04(		
67.08%)d 0.00( 0.06%)		
(0.93316) BD ( 1) C 10- C 11	(0.11688) LV ( 4) U 2	13.95
(49.92%) 0.7065* C 10 s( 33.15%)p 2.01(	s( 0.04%)p 1.41( 0.06%)d99.99( 79.43%)f99.99(	
66.78%)d 0.00( 0.07%)	20.46%)g 0.16( 0.01%)	
( 50.08%) 0.7077* C 11 s( 33.92%)p 1.95(		
66.02%)d 0.00( 0.06%)		
(0.93584) BD ( 1) C 11- C 12	(0.16594) LV ( 2) U 2	13.63
( 49.14%) 0.7010* C 11 s( 34.30%)p 1.91(	s( 2.20%)p 0.24( 0.52%)d24.94( 54.98%)f19.18(	
65.63%)d 0.00( 0.07%)	42.29%)g 0.00( 0.01%)	
( 50.86%) 0.7132* C 12 s( 33.14%)p 2.02(		
66.81%)d 0.00( 0.05%)		
(0.84424) BD ( 2) C 11- C 12	(0.16594) LV ( 2) U 2	13.35
(49.52%) 0.7037* C 11 s( 0.02%)p99.99(	s( 2.20%)p 0.24( 0.52%)d24.94( 54.98%)f19.18(	
99.88%)d 5.58( 0.10%)	42.29%)g 0.00( 0.01%)	
( 50.48%) 0.7105* C 12 s( 0.15%)p99.99(		
99.76%)d 0.64( 0.09%)		





**Figure S68.** Computed MOs for <sup>Ad</sup>LUC<sub>6</sub>H<sub>5</sub>CH<sub>3</sub>UL<sup>Ad</sup> complex (s=2). (a)AMO-HOMO-5 (b)AMO-HOMO-4 (c)AMO-HOMO-3 (d)AMO-HOMO-2 (e)AMO-HOMO-1 (f)AMO-HOMO (g)AMO-LUMO (h)BMO-HOMO-1 (i)BMO-HOMO (j)BMO-LUMO (k)spin density plot

Optimized coordinates

А			
U	-0.023122000	-0.746079000	0.106174000
I I	0.509148000	-0.164877000	3.161881000
0	1.633109000	0.690753000	-0.265595000
0	-1.771759000	0.626849000	0.109847000
Ν	2.249981000	-2.227842000	-0.542058000
Ν	-0.106177000	-3.562867000	1.106542000
Ν	-2.345984000	-2.071912000	-1.045483000
Ν	-0.176337000	-0.701013000	-2.738668000
С	2.100309000	-3.630446000	-0.112797000
Н	3.100306000	-4.085378000	0.006426000
Н	1.617287000	-4.183866000	-0.923464000
С	1.341740000	-3.835784000	1.191794000
Н	1.752999000	-3.191334000	1.974566000
Н	1.511007000	-4.877177000	1.522274000
С	-0.740151000	-4.633606000	0.315598000
Н	-0.634265000	-5.583509000	0.871246000
Н	-0.185737000	-4.762055000	-0.618934000
С	-2.205830000	-4.430492000	-0.044287000
Н	-2.597401000	-5.417538000	-0.320385000

Н	-2.796896000	-4.121820000	0.822537000
С	-2.461834000	-3.528624000	-1.245293000
Н	-1.767282000	-3.826556000	-2.038508000
Н	-3.476943000	-3.744323000	-1.626694000
С	-2.540968000	-1.431686000	-2.357106000
Н	-2.816516000	-0.389402000	-2.191459000
н	-3.386705000	-1.889400000	-2.899736000
С	-1.299423000	-1.517034000	-3.232661000
н	-0.961425000	-2.553954000	-3.315696000
н	-1.567099000	-1.200807000	-4.256752000
С	1.061064000	-1.071245000	-3.459139000
н	1.770385000	-0.249299000	-3.322763000
н	0.836802000	-1.114236000	-4.539595000
С	1.742101000	-2.375211000	-3.026744000
н	2,265819000	-2.770864000	-3.905370000
н	0.999637000	-3 140470000	-2 765685000
С	2,786197000	-2.218958000	-1.917861000
н	3 547127000	-3 012265000	-2 014776000
н	3 314325000	-1 272277000	-2 059181000
C	-0.657059000	-3 616386000	2.000101000
ц	1 722280000	3 377852000	2.472100000
	-1.722209000	2 972617000	2.407155000
	-0.101734000	-2.873017000	2 00007000
	-0.520908000	-4.021102000	2.9099997000
	-0.453702000	0.714200000	-3.036036000
п	0.377319000	1.323983000	-2.687040000
н	-1.354828000	1.049426000	-2.524293000
Н	-0.579209000	0.873182000	-4.122214000
C	3.198909000	-1.554600000	0.401805000
н	2.664503000	-1.396245000	1.349390000
Н	4.027461000	-2.249304000	0.616764000
С	3.790872000	-0.242633000	-0.028401000
С	5.178420000	-0.133410000	-0.090992000
Н	5.771337000	-1.032050000	0.071019000
С	5.810654000	1.086611000	-0.328735000
С	4.976214000	2.199591000	-0.456272000
Н	5.440333000	3.166203000	-0.599233000
С	3.576370000	2.156084000	-0.398474000
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Н	7.554893000	-0.794442000	-1.370256000
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н	7.662311000	-0.324582000	1.178916000
н	7.606334000	1.343576000	1.758825000
С	7.839752000	2.588990000	-0.686622000
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н	4 259158000	4 729067000	-1 467021000
н	4.2506/10000	4.725007000	0.307066000
н	3 033544000	5 603534000	-0 545305000
с С	1 0/1022000	3 506629000	0.04500000
С Ц	1.341332000	3.390020000 4.515252000	0.040010000
п	1.343813000	4.313252000	0.019010000
п	2.598328000	3.040002000	1.720840000
П		2.149145000	0.979204000
	1.841015000	3.475453000	-1.009485000
н	2.411180000	3.353018000	-2.597895000

Н	1.293104000	4.422797000	-1.729893000
Н	1.114143000	2.669185000	-1.589912000
С	-3.382776000	-1.659801000	-0.044612000
Н	-2.935761000	-1.805489000	0.949403000
Н	-4.233456000	-2.355786000	-0.110909000
С	-3.938283000	-0.265689000	-0.129780000
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С	-2.818193000	3.390220000	0.440679000
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Ĥ	-4.406687000	4.609530000	1.348294000
н	-4.187115000	4,900830000	-0.391384000
н	-3 006426000	5 513833000	0 769179000
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ц	-2.81/635000	3 093260000	2 602466000
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н	-1.4/4093000	2 329226000	1 784718000
C	-1.793084000	3 607/05000	-0.685754000
ц	1 100251000	4 400705000	-0.003734000
п	-1.190201000	4.499705000	-0.463130000
п	-2.290800000	3.752454000	-1.040903000
	-1.119092000	2.754260000	-0.764678000
	-0.140390000	0.526279000	0.710564000
п	-7.870753000	0.912405000	1.697560000
п	-7.898329000	-0.537129000	0.000312000
н	-9.235999000	0.617660000	0.602792000
C	-7.847343000	0.742790000	-1.775329000
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н	-7.587022000	-0.315939000	-1.873442000
Н	-7.354471000	1.284461000	-2.589691000
С	-7.884179000	2.772717000	-0.334943000
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Ν	-2.326405000	-2.240590000	0.073623000
Ν	0.037145000	-3.500375000	-1.597981000
Ν	2.240610000	-2.101275000	0.644290000
Ν	0.040894000	-0.824497000	2.430974000
С	-2.163698000	-3.629786000	-0.389717000
Н	-3.158887000	-4.095073000	-0.514680000
Н	-1.664401000	-4.196601000	0.402606000
С	-1.405462000	-3.787025000	-1.700464000
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Н	2,755840000	-4.039130000	-1.302360000
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С	1.182006000	-1.644317000	2.867206000
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	0.111437000	-2.490034000	-3.443314000
н	0.382978000	-4.282598000	-3.553987000
C	0.289358000	0.582098000	2.775408000
н	-0.554432000	1.190159000	2.448535000
н	1.180828000	0.955014000	2.267864000
Н	0.422377000	0.711213000	3.863645000
С	-3.251836000	-1.539864000	-0.872186000
Н	-2.675819000	-1.355547000	-1.797745000
Н	-4.076224000	-2.225570000	-1.131279000
С	-3.842235000	-0.237514000	-0.421235000
С	-5.226337000	-0.131512000	-0.292256000
Н	-5.823129000	-1.032471000	-0.425586000
С	-5.849949000	1.085530000	-0.017317000
С	-5.014142000	2.202465000	0.076578000
Н	-5.472561000	3.167136000	0.250577000
С	-3.619987000	2.161220000	-0.051059000
С	-3.005443000	0.894054000	-0.253668000
С	-7.374781000	1.158714000	0.136605000
С	-7.823592000	0.222569000	1.276079000
Н	-7.369031000	0.520259000	2.227059000
Н	-7.539996000	-0.816739000	1.082080000
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н	-7 427812000	2 946723000	1 403452000
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ĉ	3 680360000	4 713152000	0.131085000
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п	-4.241011000	4.725520000	0.606215000
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Н	4.158650000	-2.379994000	-0.225374000
С	3.873007000	-0.287815000	-0.178337000
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Н	5.828340000	-1.021264000	0.267042000
С	5.837124000	1.128667000	0.092305000
С	5.008545000	2.219986000	-0.189608000
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C	3 637357000	2 124270000	-0 457225000
c	3 042824000	0.831906000	-0.427531000
c	7 338521000	1 262829000	0 379597000
н	7.6517//000	3 189888000	-0.6170/3000
с С	2 915910000	2 200527000	0.792961000
C	2.010019000	3.360327000	-0.762601000
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н	4.463928000	4.615759000	-1.549251000
н	4.125531000	4.865581000	0.177700000
Н	3.028494000	5.511615000	-1.045157000
C	2.177235000	3.237477000	-2.1/8543000
н	2.951670000	3.145837000	-2.948183000
Н	1.573566000	4.122937000	-2.410506000
Н	1.533838000	2.358415000	-2.233135000
С	1.719721000	3.567818000	0.280247000
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Н	2.162516000	3.700458000	1.274610000
Н	1.053218000	2.705330000	0.303920000
С	8.138076000	0.488965000	-0.687403000
Н	7.943531000	0.891283000	-1.687003000
Н	7.873700000	-0.573196000	-0.699925000
Н	9.214889000	0.562424000	-0.493016000
С	7.652725000	0.678463000	1.770987000
н	8.724336000	0.756513000	1.990641000
н	7.376633000	-0.378691000	1.838287000
н	7.105221000	1.216266000	2.552338000
С	7.809537000	2.723322000	0.361073000
Н	7.293393000	3.327040000	1.115207000
н	8.882238000	2.768747000	0.578492000
С			
U	-0.027382000	-0.657384000	0.033817000
Í.	-1.589870000	-0.573010000	-2.668592000
0	-1.656603000	0.682147000	0.390774000
0	1 597375000	0.669474000	0 161805000
Ň	-2 369648000	-1 929670000	1 360975000
N	-2.000040000	-3 677722000	-0.271644000
N	2 203754000	1 706/02000	1 612/00000
N	0.146952000	-1.790492000	2.015265000
	0.140052000	-0.104913000	1 262674000
	-2.208797000	-3.393448000	1.362674000
н	-3.179067000	-3.867860000	1.599045000
Н	-1.529663000	-3.661076000	2.179124000
C	-1./20191000	-3.984172000	0.054447000
н	-2.339536000	-3.637567000	-0.775740000
н	-1.859938000	-5.078851000	0.100693000
С	0.537126000	-4.562208000	0.554157000
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Ν	0.137717000	-3.080561000	1.966045000
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н	2 819/8/000	-3.811227000	-0 225039000
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ц	1 352013000	-1.706240000	-1.2-3317000
н Ц	3 302131000	-1.1 30240000	1 401621000
с С	3.302134000 2.564064000	2 170526000	2 2007000
U LI	2.004904000	-2.170020000	-2.300910000
	2.524092000	-3.243100000	-2.202010000
П	3.149292000	-2.0331/8000	-3.315490000
	0.453079000	-2.5/6/25000	-3.519192000
н	1.018594000	-2.623941000	-4.46/020000
Н	-0.500434000	-2.085773000	-3./3998/000
C .	0.175975000	-3.995221000	-3.032460000
н	1.095/31000	-4.486548000	-2.09261/000

Н	-0.145791000	-4.557133000	-3.919383000
С	-0.907483000	-4.169382000	-1.969818000
н	-1.865816000	-3.795119000	-2.379781000
н	-1.046161000	-5.259057000	-1.859764000
С	-1.260640000	-4.387321000	0.361315000
Н	-0.829294000	-5.407510000	0.382129000
н	-2.337384000	-4.537743000	0.160987000
С	-1.163264000	-3.750206000	1.746600000
н	-1.930916000	-2.980354000	1.851019000
н	-1.338852000	-4.514368000	2.524338000
С	0.098557000	-2.378076000	3.253506000
н	1.032986000	-1.839078000	3.430111000
н	-0.058719000	-3.078172000	4.090714000
н	-0.716827000	-1.652969000	3.240682000
С	3.144091000	-0.726223000	1.073576000
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Н	4.157805000	-0.987036000	1.419764000
С	3.155425000	0.711408000	0.631658000
Ĉ	4.335940000	1.451236000	0.740936000
н	5 232677000	0.937505000	1 076265000
C	4 370040000	2 810337000	0 442998000
c	3 160860000	3 398367000	0.041687000
н	3 165710000	4 459815000	-0 179256000
c	1 949218000	2 715287000	-0.086085000
c	1 959938000	1 321379000	0 199248000
ĉ	5.640844000	3 663073000	0.199240000
c	6 852676000	2 830731000	1 008432000
L L	6.604704000	2.039731000	1.000432000
	7 726071000	2.399000000	1.999139000
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	7.003310000	2.031190000	1 57402000
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н	5.189540000	4.386490000	2.563978000
	5.970686000	4.278251000	-0.823156000
н	6.152365000	3.496474000	-1.568411000
н	6.869421000	4.903030000	-0.758875000
Н	5.154759000	4.906873000	-1.192569000
C	0.659685000	3.457119000	-0.475483000
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н	-0.605893000	2.266972000	0.856747000
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С	0.898342000	4.958848000	-0.699601000
Н	1.277327000	5.457052000	0.199496000
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С	1.221595000	-0.343038000	-3.205486000
Н	0.200792000	0.023284000	-3.317991000
Н	1.707966000	-0.380416000	-4.194018000
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Н	-4.989066000	2.463036000	1.378634000
С	-4.409035000	2.054228000	-0.637098000
С	-3.612480000	1.223908000	-1.436248000
Н	-3.634695000	1.323456000	-2.520109000

С	-2.797382000	0.233943000	-0.890992000
С	-1.952543000	-0.704730000	-1.693270000
Н	-1.902775000	-0.380191000	-2.741585000
н	-2.441228000	-1.698432000	-1.698915000
С	-3.551018000	0.708561000	2.886882000
С	-4.569465000	1.621267000	3.587540000
н	-4 351855000	2 682338000	3 425767000
ц.	4.534168000	1 440460000	4 667030000
	-4.554100000	1.440400000	4.007939000
	-0.093005000	0.742020000	3.232343000
	-3.919014000	-0.743847000	3.245994000
н	-4.929081000	-0.982132000	2.895282000
н	-3.894085000	-0.889998000	4.333210000
Н	-3.226389000	-1.447225000	2.781674000
С	-2.159100000	1.057651000	3.448647000
Н	-1.380020000	0.472460000	2.956537000
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С	-5.310547000	3.111198000	-1.291639000
С	-6.081816000	3.947013000	-0.260804000
н	-6 749569000	3 327490000	0 347141000
н	-6 699628000	4 692028000	-0 774628000
н	-5 406139000	4.002020000	0.413802000
$\hat{c}$	6 338357000	2 /13208000	2 20/082000
ц	-0.330337000	1 922024000	2.204903000
	-5.646310000	1.032931000	-2.992696000
н	-6.992717000	3.149855000	-2.687719000
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Н	-5.089040000	4.825407000	-2.626705000
н	-3.904246000	3.543641000	-2.920867000
LU	C6H6UL complex	(s=2)	
LU U	C6H6UL complex ( 3.830446000	(s=2) 8.827230000	11.685739000
LU U O	C6H6UL complex 3.830446000 4.010036000	(s=2) 8.827230000 6.827908000	11.685739000 12.589393000
LU U O O	C6H6UL complex 3.830446000 4.010036000 5.529797000	(s=2) 8.827230000 6.827908000 9.249462000	11.685739000 12.589393000 10.344299000
LU U O N	C6H6UL complex 3.830446000 4.010036000 5.529797000 0.692672000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000	11.685739000 12.589393000 10.344299000 11.642751000
LU U O N N	C6H6UL complex 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000
LU U O N N	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000
	C6H6UL complex 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202222000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 0.184024000
	C6H6UL complex 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000
	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000
	C6H6UL complex 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000 10.965093000
LU U O O N N N C N C H	C6H6UL complex 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000 10.965093000 10.917935000
LU U O O N N N C N C H H	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000 10.965093000 10.917935000 11.743496000
U O O N N N C N C H H C	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000
U O O N N N C N C H H C C	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000
U O O N N N C N C H H C C H	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000
U O O N N N C N C H H C C H C	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000 7.518024000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.184024000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000
U O O N N N C N C H H C C H C C	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000 7.518024000 7.129830000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000
U O O N N N C N C H H C C H C C C	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000
Ц О О О С С С С С С С С С С С С С	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2 154590000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000
Ц О О О С Z C C C C C C C C C C C C C	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000
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L O O Z Z Z C Z C H H C C H C C Z Z C O C C L	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8 160312000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.088747000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 2.021556000
С О О О О О О О С С С С С С С С С С С С	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.529888000 10.267704000 9.395361000 8.581028000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.088747000 10.915472000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 7.021556000
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Ξοοοχχοντηρητουκαράτους.	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000 1.239183000 0.438185000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.088747000 10.915472000 6.055752000 5.901465000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 7.021556000 12.991360000 13.742057000
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	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000 1.239183000 0.438185000 1.506948000 -0.403532000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.988747000 10.915472000 6.055752000 5.901465000 7.114306000 6.731693000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 7.021556000 12.991360000 13.742057000 13.038881000 11.372203000
Ц Ооохххохностоонностистоли Соосс	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000 1.239183000 0.438185000 1.506948000 -0.403532000 -1.081177000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.915472000 6.055752000 5.901465000 7.114306000 6.731693000 6.811970000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 7.021556000 12.991360000 13.742057000 13.038881000 11.372203000
Ú J O O Z Z Z C Z C I L C C L C C C I C I C I C I C I C I	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000 1.239183000 0.438185000 1.506948000 -0.403532000 -1.081177000 -1.007297000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.529888000 10.267704000 9.395361000 8.581028000 8.581028000 8.913005000 11.589670000 8.768077000 8.768077000 8.981647000 9.468824000 10.915472000 6.055752000 5.901465000 7.114306000 6.731693000 6.811970000 6.328074000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 7.021556000 12.991360000 13.742057000 13.742057000 13.038881000 11.372203000 12.243589000
Ο Ο ΤΟ Ο Ζ Ζ Ζ Ζ Ζ Ζ Ζ Ι Ο Ι Ο Ο Ι Ο Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι Ι Ο Ι	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000 1.239183000 0.438185000 1.506948000 -0.403532000 -1.081177000 -1.007297000 6.149779000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.915472000 6.055752000 5.901465000 7.114306000 6.731693000 6.811970000 6.328074000 7.979746000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 12.991360000 13.742057000 13.038881000 11.372203000 12.243589000 10.552399000 8.451720000
Ο Ο Ο Ο Ι Ι Ο Ι Ι Ο Ι Ο Ι Ο Ο Ι Ι Ο Ι Ο	C6H6UL complex ( 3.830446000 4.010036000 5.529797000 0.692672000 4.048522000 1.949805000 6.202337000 1.017903000 0.083523000 -0.777047000 0.742201000 6.965799000 3.581591000 3.581591000 3.715413000 7.518024000 7.129830000 2.755252000 2.154590000 3.596326000 7.584273000 8.160312000 1.239183000 0.438185000 1.506948000 -0.403532000 -1.081177000 -1.007297000 6.149779000 5.487798000	(s=2) 8.827230000 6.827908000 9.249462000 5.800207000 6.889877000 9.035983000 9.183289000 4.266403000 8.126224000 8.803802000 8.529888000 10.267704000 9.395361000 8.581028000 8.913005000 11.589670000 8.768077000 8.981647000 9.468824000 10.915472000 6.055752000 5.901465000 7.114306000 6.731693000 6.811970000 6.328074000 7.979746000 6.769745000	11.685739000 12.589393000 10.344299000 11.642751000 9.364226000 9.521817000 9.045191000 10.965093000 10.917935000 11.743496000 8.661921000 14.279090000 14.279090000 14.981233000 6.663895000 9.424565000 8.310459000 7.406946000 8.310319000 7.422052000 7.021556000 12.991360000 13.742057000 13.038881000 11.372203000 12.243589000 10.552399000 8.451720000 9.043323000

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H H C	3.523091000 2.664333000 3.537088000 2.734717000 0.916293000	16.265757000 16.943025000 15.916958000 15.870268000 14.261637000	$\begin{array}{c} 18.283030000\\ 17.632488000\\ 17.550906000\\ 18.672346000\\ 12.393308000\\ 11.694960000\\ \end{array}$
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H H C C H	3.523091000 2.664333000 3.537088000 2.734717000 0.916293000 1.338200000	16.265757000 16.943025000 15.916958000 15.870268000 14.261637000 13.454402000	$\begin{array}{c} 17.632488000\\ 17.632488000\\ 17.550906000\\ 18.672346000\\ 12.393308000\\ 11.694960000\\ 12.295478000\\ 12.295478000\\ \end{array}$
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HHCCHHHC	3.523091000 2.664333000 3.537088000 2.734717000 0.916293000 1.338200000 -0.166821000 1.336560000	16.265757000 16.943025000 15.916958000 15.870268000 14.261637000 13.454402000 14.104206000 14.202331000	17.632488000 17.632488000 17.550906000 18.672346000 12.393308000 11.694960000 12.295478000 11.620820000 10.684517000
H H C C H H H C	3.523091000 2.664333000 3.537088000 2.734717000 0.916293000 1.338200000 -0.166821000 1.336560000 0.489022000	16.265757000 16.943025000 15.916958000 15.870268000 14.261637000 13.454402000 14.104206000 14.202331000 16.687024000	17.632488000 17.632488000 17.550906000 18.672346000 12.393308000 11.694960000 12.295478000 11.620820000 10.684517000 11.460997000
ННССНННСН	3.523091000 2.664333000 3.537088000 2.734717000 0.916293000 1.338200000 -0.166821000 1.336560000 0.489022000 0.828456000	$\begin{array}{c} 16.265757000\\ 16.943025000\\ 15.916958000\\ 15.870268000\\ 14.261637000\\ 13.454402000\\ 14.104206000\\ 14.202331000\\ 16.687024000\\ 16.677205000\\ \end{array}$	17.632488000 17.632488000 17.550906000 18.672346000 12.393308000 11.694960000 12.295478000 11.620820000 10.684517000 11.460997000 10.419572000
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