

SUPPLEMENTARY INFORMATION

Conformational Switching Modulates Excited-State Pathways in a Cofacial Perylene Dimer

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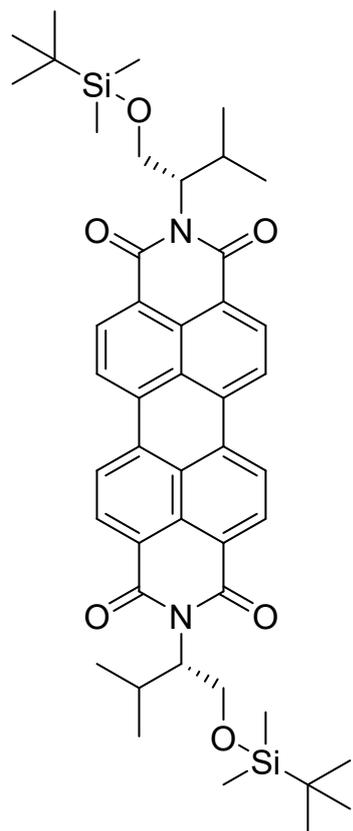


Figure S 1 Molecular structure of the **refPDI** monomer

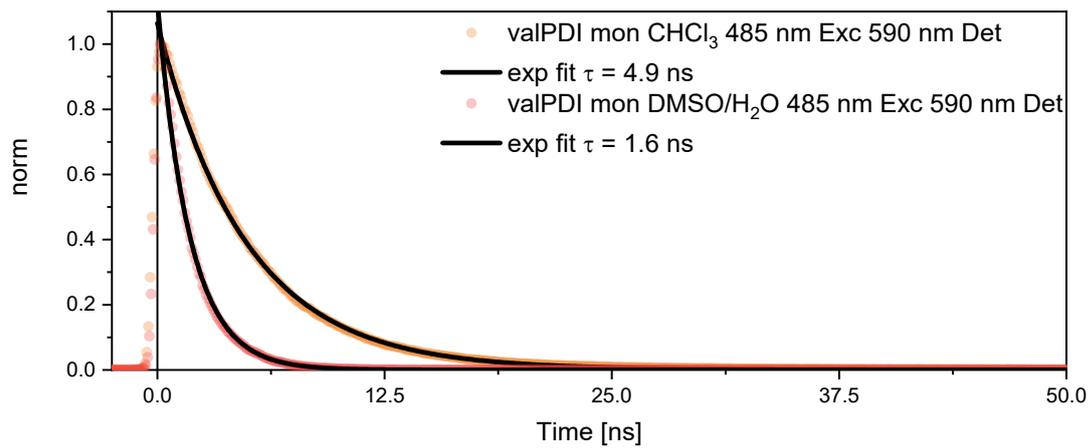


Figure S 2 ns fluorescence traces and fit (black) of **refPDI** in CHCl₃ (orange) and 1:1 DMSO/water (red)

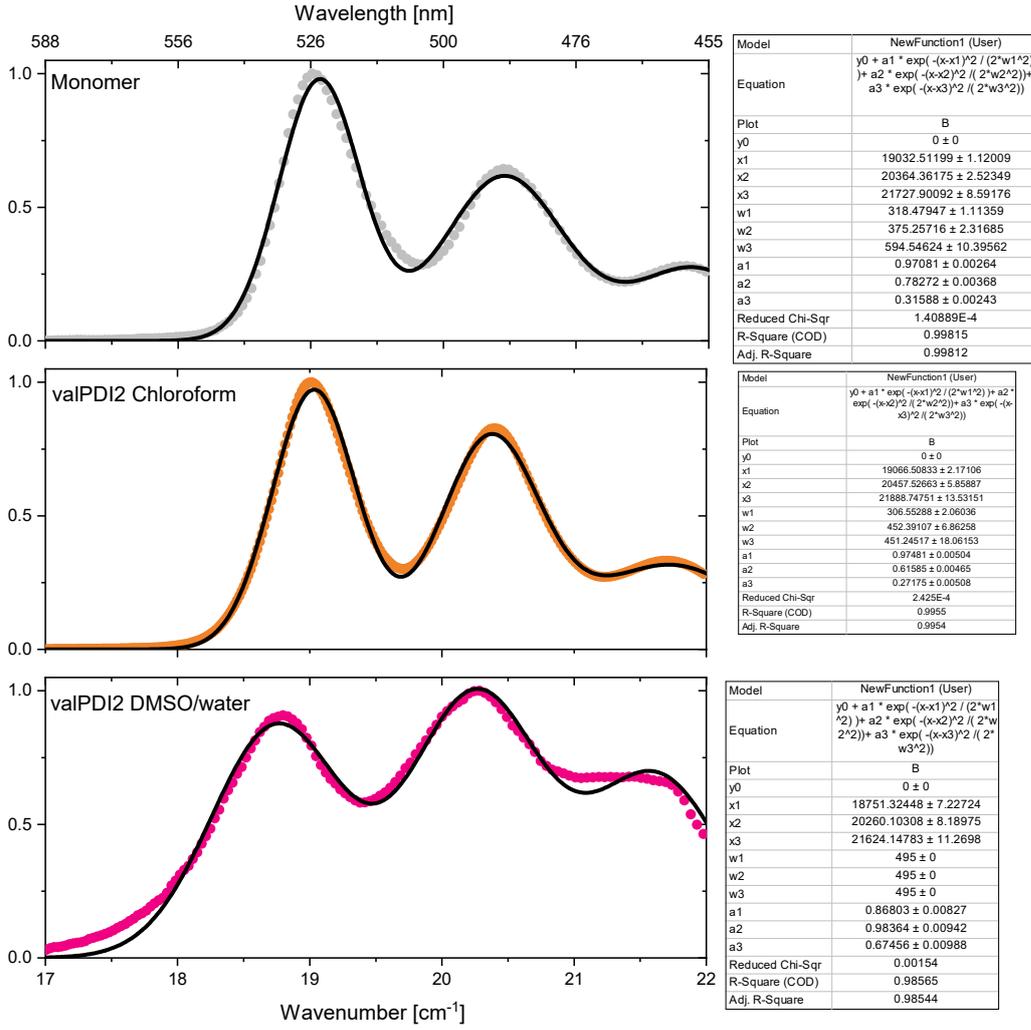


Figure S 3 steady-state fits for Spano's excitonic coupling model

Details of the excitonic coupling calculations

The excitonic coupling strength for valPDI₂ in chloroform and DMSO/water can be determined from their steady-state absorption spectra, if the Huang-Rhys factor λ^2_{mon} of the reference monomer refPDI is known.¹⁻³ λ^2_{mon} is defined as:

$$\lambda^2_{mon} = (R_{0-1}/R_{0-0}) \quad (S1)$$

Where R_{0-0} and R_{0-1} are the areas of the 0-0 and 0-1 vibronic bands of its steady-state absorption spectrum (see Figure S3). $(R_{0-1}/R_{0-0}) = 0.68$ for refPDI. Equation S1 can then substituted into Equation S2 which gives the total excitonic coupling when solved for J_{Tot} :

$$R_{abs} = (\lambda^2_{mon})^{-1} \left[\frac{1 - G(0, \lambda^2) e^{-\lambda^2_{mon} J_{Tot} / \omega_0}}{1 - G(1, \lambda^2) e^{-\lambda^2_{mon} J_{Tot} / \omega_0}} \right]^2 \quad (S2)$$

Where $R_{abs} = (\lambda^2_{dim})^{-1} = (R_{0-0}/R_{0-1})$ of **valPDI**₂ in chloroform or DMSO/water and ω_0 is the spacing of the vibronic progression (1350 cm⁻¹). Finally, G is the vibrational function, as defined in Equation S3:

$$G(v_t; \lambda^2) = \sum_{\substack{u=0,1,\dots \\ u \neq v_t}} \frac{(\lambda^2)^u}{u!(u-v_t)!}, v_t \in N_0 \quad (\text{S3})$$

The values of J_{Tot} extracted using the method outlined above are reported in the main manuscript.

The long-range Coulombic coupling for both conformers was calculated using the transition electrostatic potential method (trESP) as implemented in Multiwfn. This approach relies on point atomic charges rather than TDMs and yields more reliable estimation of coupling strengths when dipole-dipole distances are comparable or smaller to the molecular dimensions. The sum over the electrostatic interactions between transition charges of the PDI pair yields the long-range coupling J_{Coul} , as per Equation S4:

$$J_{Coul} = \frac{1}{4\pi\epsilon_0} \sum_i \sum_j \frac{q_i^{(1)} q_j^{(2)}}{|r_i^{(1)} - r_j^{(2)}|} \quad (\text{S4})$$

In which ϵ_0 is the permittivity of free space, $q_i^{(1)}$ is the transition charge of the i -th atom and $r_i^{(1)}$ is the position vector associated to the i -th charge. We used time dependent (TD-)DFT at the ω -B97XD level of theory and dgdzvp as basis set in Gaussian 16 to calculate the transition densities of the two conformers of **valPDI**₂ populated in chloroform and DMSO/water. The transition densities were fit to atomic partial charges and the Coulomb coupling terms were calculated at the optimised ground state structures. The atomic partial charges were all rescaled by $\sqrt{2}$ as indicated in the Multiwfn manual.

The short-range couplings were calculated by using the charge transfer integral package CATNIP, the molecular orbitals (MO) coefficients were obtained by DFT calculations at the ω -B97XD/dgdzvp level of theory in Gaussian 16.

Both the long- and short-range coupling calculations required to determine transition charges or MO coefficients of each PDI chromophore. Such fragments were obtained by disconnecting the dimer structures at two opposite alpha carbons between the pair of C=O of each ligand to yield two identical ‘‘halves’’, followed by capping with H-atoms at the broken bonds positions to keep the S_0 multiplicity. The added H atoms were not included in the fragments used for the long-range coupling calculation in Multiwfn. Both dimer structures were geometry-optimized followed by frequency calculations to prove the existence of true minima, as determined by the absence of imaginary frequencies. All relevant coordinates are reported at the end of the present document.

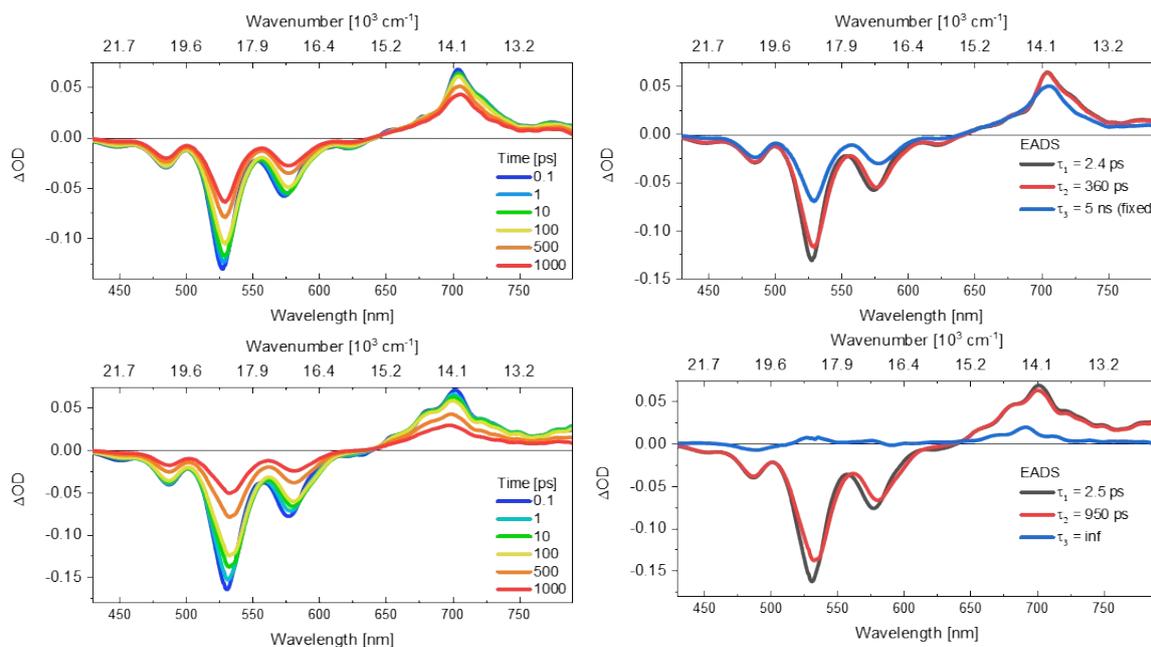
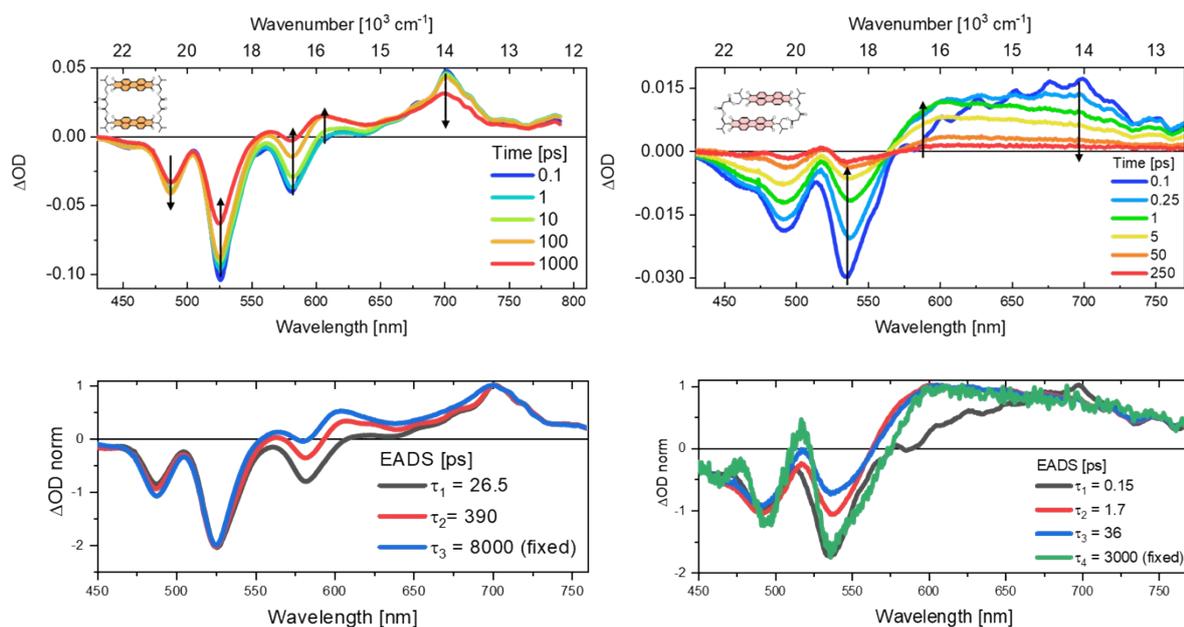


Figure S 4 fsTA spectra at selected pump probe delay times (left) and EADS from global fitting (right) of **refPDI** monomer in CHCl_3 (top) and DMSO/water (bottom)



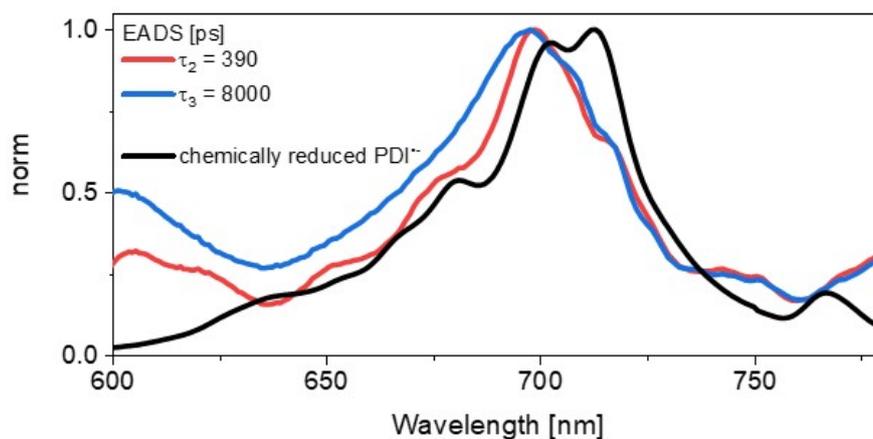


Figure S 5 (top) fsTA spectra at selected pump probe delay times of **valPDI₂** in CHCl₃ (left) and DMSO/water (right).

(middle) normalised evolution associated decay spectra (EADS) obtained by applying a sequential model to the fsTA data of **valPDI₂** in CHCl₃ (left) and DMSO/water (right). Data were fit in GloTarAn.⁴ Global fit required 3 and 4 components -indicated in the legend- to accurately fit the data in chloroform and DMSO/water, respectively.

(bottom) comparison between ESA ($S_n \leftarrow S_1$) region of normalised EADS2 and EADS3 of **valPDI₂** in CHCl₃ (maxima at 700 nm) and steady-state absorption of the chemically reduced **refPDI** radical anion (maximum at 710 nm). The radical anion was produced adding 0.3 ml of a saturated solution of sodium dithionite (Na₂S₂O₄) to 3 ml of a 0.2 mM THF solution of **valPDI₂**

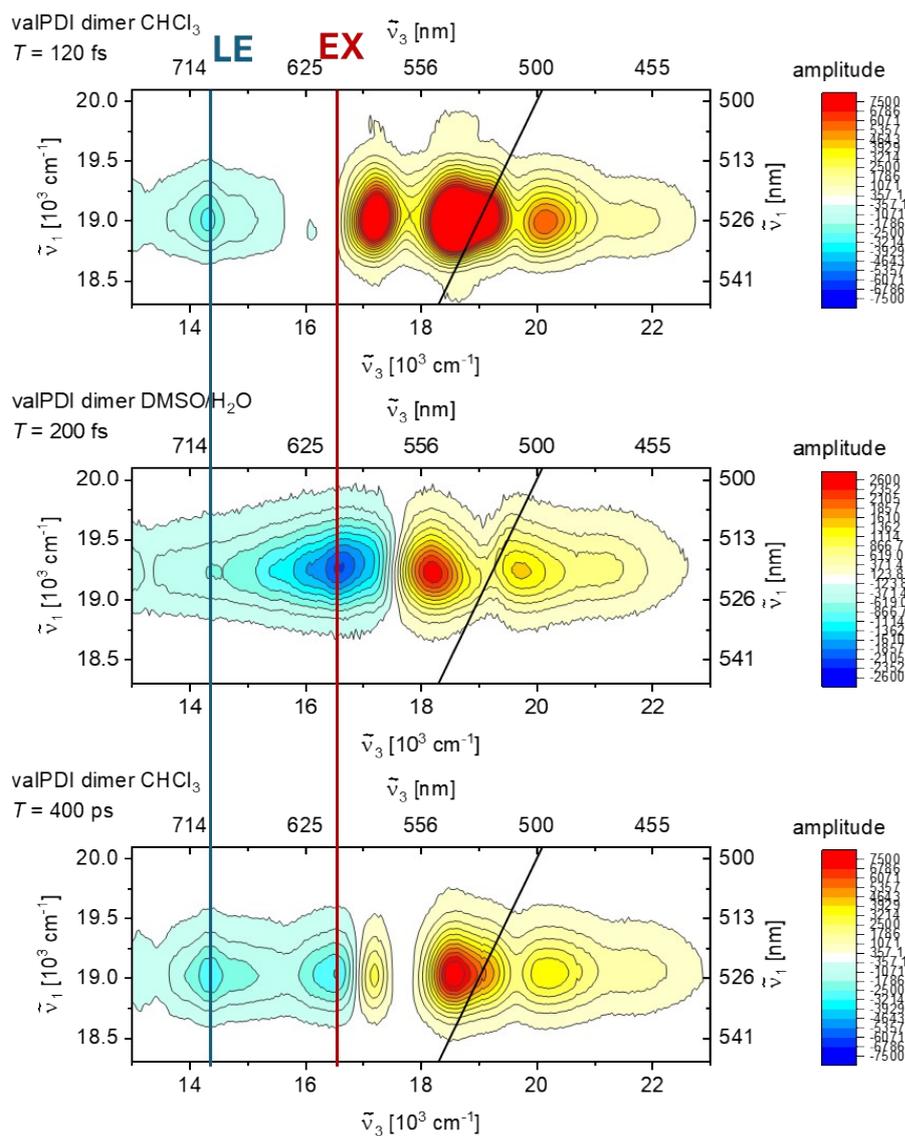


Figure S 6 Absorptive HB2DES of **valPDI**₂ in CHCl_3 and DMSO/water showing the superposition of EX (red) and LE (blue) states in the open conformation.

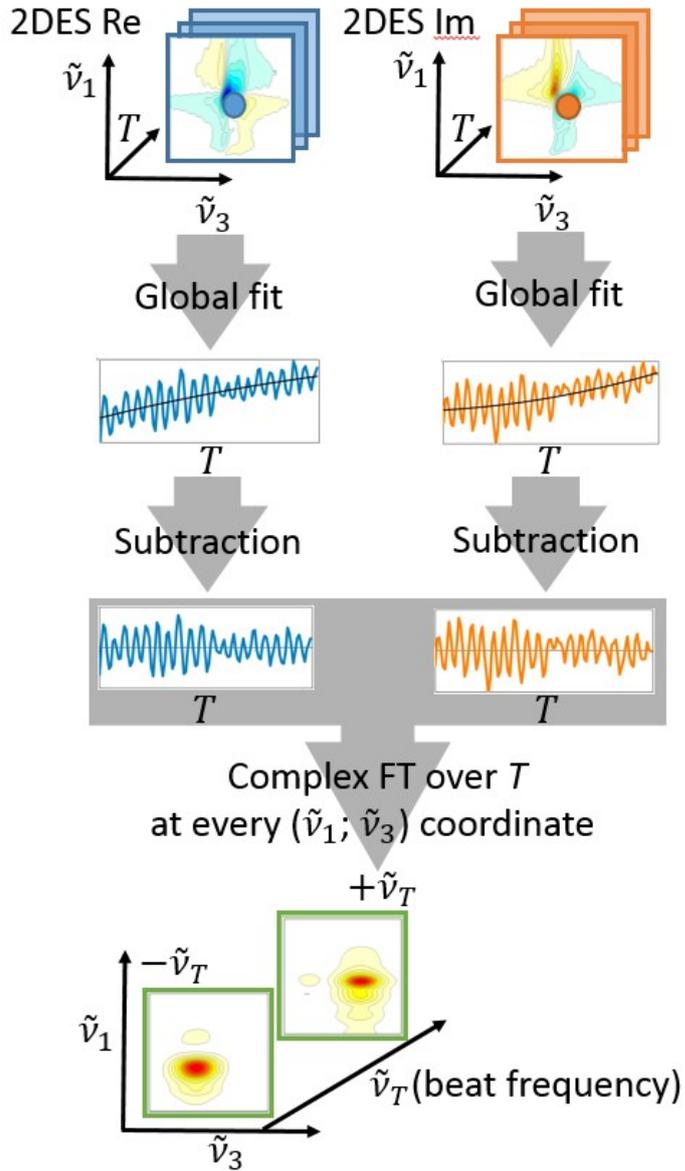


Figure S 7 Scheme of the method to extract complex valued coherent beatmaps from a HB2DES dataset. Adapted from Bressan et al.⁵

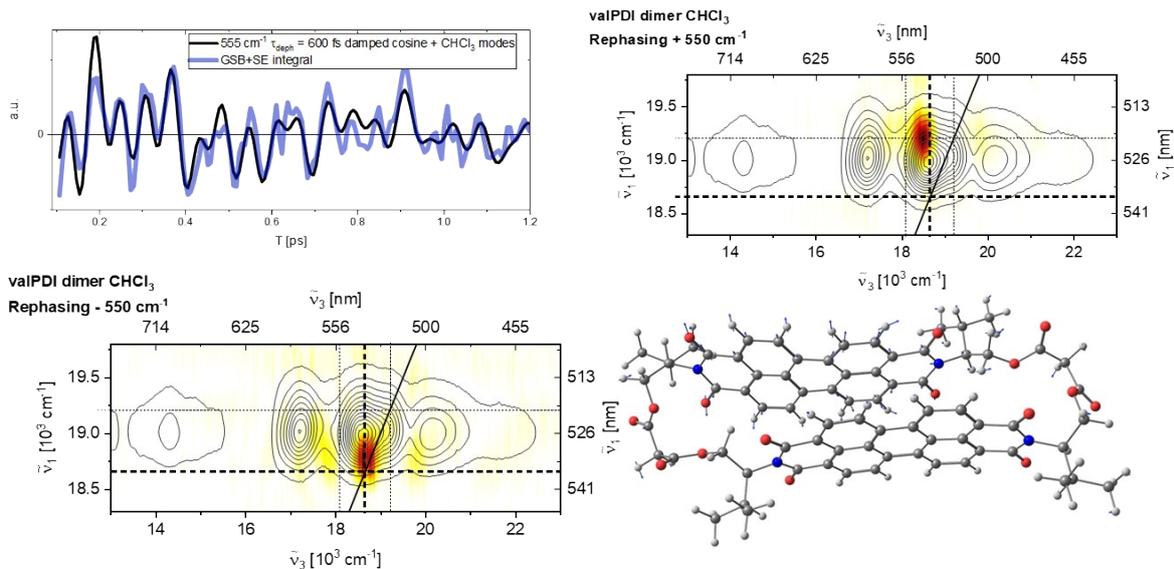


Figure S 8 Integrated oscillatory residuals (blue) in the GSB+SE region of the **valPDI**₂ response in CHCl₃. Data were fit to a sum of damped cosines at 369 cm⁻¹ with fixed dephasing time of 1 ps to account for the strong CHCl₃ solvent response and to molecular modes at 180 and 550 cm⁻¹ with 650 fs dephasing time. Rephasing positive and negative beatmaps at 550 cm⁻¹ showing the pattern expected for a two electronic level displaced harmonic oscillator comprising of GSB and SE vibrational coherence.⁶ Beatmaps are shown as white-yellow-red heat maps and amplitude normalized to 1. Black dashes indicate the position of the 0-0 transition; thin dashes are drawn at ± 550 cm⁻¹ from the 0-0 transition. DFT structure showing the displacement vectors associated with the ring breathing at 530 cm⁻¹

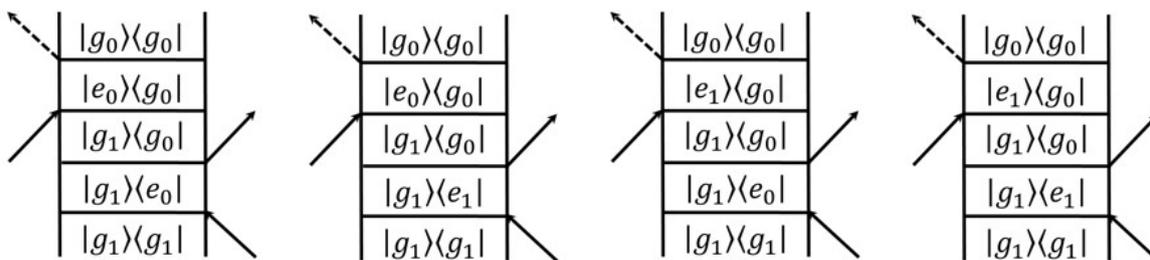


Figure S 9 Rephasing double-sided Feynman diagrams accounting for hot ground state bleach pathways oscillating at negative frequencies during T . g and e are the electronic ground and excited states and the subscripts indicate the number of quanta of a Raman active low frequency (below 200 cm^{-1}) molecular vibration coupled to the electronic transition.

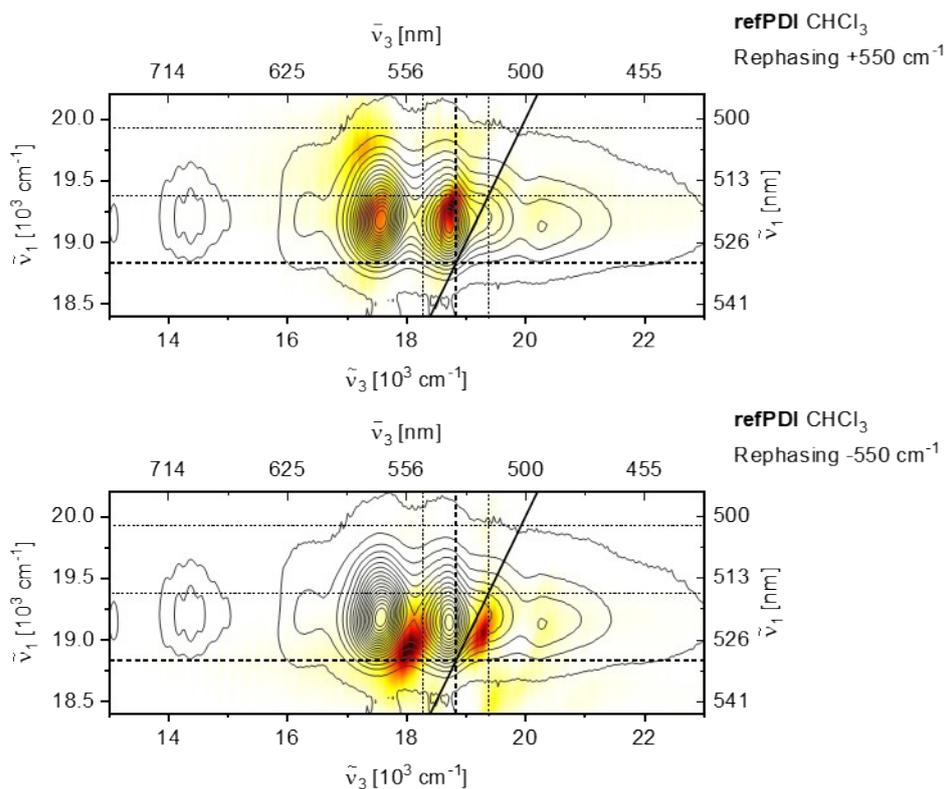
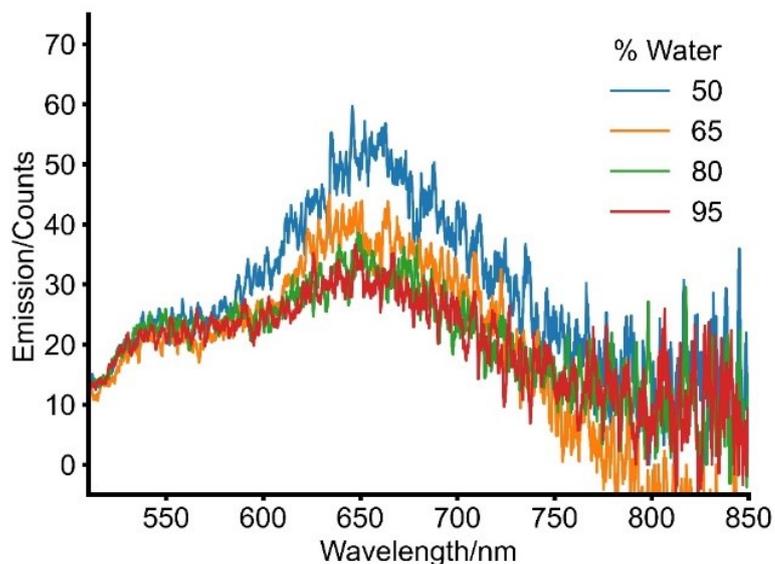


Figure S 10 Rephasing positive and negative beatmaps at 550 cm^{-1} of **refPDI** in CHCl_3 showing the pattern expected for a two electronic level displaced harmonic oscillator comprising of GSB and SE vibrational coherence. Beatmaps are shown as white-yellow-red heat maps and amplitude normalized to 1. Black dashes indicate the position of the 0-0 transition; thin dashes are drawn at $\pm 550 \text{ cm}^{-1}$ from the 0-0 transition. The redshifted signals at detection frequencies of ~ 17500 (positive) and 18000 cm^{-1} (negative) are due to anharmonic coupling between the 550 cm^{-1} PDI breathing and the 1350 cm^{-1} stretching mode giving rise to the strong vibronic progression evident in the absorption and emission spectra of **refPDI**. Anharmonic coupling between low frequency modes and the vinyl stretch was previously reported by us for other PDI derivatives in chlorinated solvents.⁷



| Solvent ratio (DMSO:H ₂ O) | valPDI ₂ FQY (%) |
|--|--------------------------------|
| 50:50 | 6.5 |
| 35:65 | 4 |
| 20:80 | 5 |
| 5:95 | 4 |

Figure S 11 Emission spectra and FQY (table) of 1 μ M solutions of valPDI₂ in different DMSO/water volume fraction solvent mixtures. Data were measured on an Edinburgh Instruments FLS1000 spectrometer with 0.1 nm step size and 250 ms dwell time using a PMT-980 photomultiplier tube and an integrating sphere (for FQY measurements). All measurements were taken with 3 nm excitation bandwidth and 0.95 nm emission bandwidth and samples excited at 493 nm.

Coordinates

valPDI2 in H₂O

| | | | |
|---|----------|-----------|----------|
| O | 14.31065 | -2.84217 | 12.64182 |
| O | 11.52014 | -7.55509 | 11.75193 |
| O | 17.19156 | -4.50085 | 15.76230 |
| O | 13.98778 | -8.98419 | 15.33146 |
| O | 16.07215 | -6.77954 | 12.41802 |
| O | 14.98067 | -9.67667 | 12.54505 |
| N | 15.82792 | -3.62861 | 14.16436 |
| O | 12.66662 | 6.28916 | 19.11902 |
| O | 9.41787 | 2.12398 | 17.11630 |
| N | 12.68966 | -8.32513 | 13.56519 |
| O | 15.99528 | 4.83191 | 21.87413 |
| O | 17.77983 | -8.22625 | 12.79791 |
| C | 14.84400 | -2.72031 | 13.73777 |
| N | 14.46131 | 5.64537 | 20.38974 |
| C | 16.00154 | -2.54447 | 16.34494 |
| C | 14.54163 | -1.60494 | 14.63993 |
| C | 11.21825 | -2.26039 | 15.42577 |
| C | 13.58527 | 4.16833 | 18.64928 |
| C | 13.81281 | 1.80578 | 17.11757 |
| C | 14.90267 | -0.38956 | 16.73867 |
| N | 10.54937 | 1.35501 | 18.94980 |
| C | 16.81878 | -7.86382 | 12.15179 |
| C | 14.03290 | 0.63662 | 16.28025 |
| C | 14.45922 | 3.14727 | 19.06440 |
| C | 13.36402 | -8.09327 | 14.76729 |
| C | 16.54950 | -2.48306 | 17.61196 |
| H | 17.18939 | -3.28266 | 17.95089 |
| C | 13.26507 | -6.74390 | 15.34587 |
| O | 14.46086 | -10.01852 | 10.36773 |
| C | 11.50292 | -3.50246 | 14.72634 |

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|---|----------|----------|----------|
| C | 16.38512 | -3.63184 | 15.44171 |
| C | 12.60263 | -3.01906 | 17.33971 |
| C | 14.54233 | 1.93225 | 18.33056 |
| C | 12.89616 | -4.25780 | 16.63557 |
| C | 12.80678 | 3.97876 | 17.52317 |
| H | 12.11933 | 4.75238 | 17.21936 |
| C | 10.99500 | -3.77470 | 13.46030 |
| H | 10.40314 | -3.04072 | 12.93997 |
| C | 16.31571 | -1.38870 | 18.42967 |
| H | 16.77822 | -1.37894 | 19.40233 |
| C | 11.92332 | -5.99303 | 13.46736 |
| C | 15.53427 | -0.31629 | 18.00901 |
| C | 13.70901 | -5.24862 | 17.17891 |
| H | 14.19812 | -5.09522 | 18.12615 |
| C | 12.02524 | -7.31549 | 12.84447 |
| C | 10.41270 | -1.26867 | 14.87386 |
| H | 10.01073 | -1.38090 | 13.88114 |
| C | 12.58303 | -9.68642 | 12.98049 |
| H | 12.36114 | -9.48967 | 11.93877 |
| C | 13.70060 | -0.59342 | 14.21638 |
| H | 13.24368 | -0.66135 | 13.24192 |
| C | 16.29684 | -8.57346 | 10.93338 |
| H | 17.09562 | -9.17203 | 10.50284 |
| H | 15.96576 | -7.86405 | 10.18037 |
| C | 15.39532 | 0.89661 | 18.79789 |
| C | 13.45761 | 0.50952 | 15.02003 |
| H | 12.81325 | 1.28113 | 14.63435 |
| C | 13.50313 | 5.43860 | 19.38802 |
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| C | 13.88157 | -6.47466 | 16.55398 |

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| H | 14.48457 | -7.23686 | 17.02171 |
| C | 15.14647 | -1.51685 | 15.90711 |
| C | 11.39404 | -0.88836 | 17.45012 |
| C | 12.92908 | 2.82191 | 16.76562 |
| H | 12.32955 | 2.73919 | 15.87453 |
| C | 15.13902 | -9.49871 | 11.22734 |
| C | 11.73679 | -2.06805 | 16.73474 |
| C | 11.19811 | -4.99764 | 12.84019 |
| H | 10.76681 | -5.18758 | 11.87003 |
| C | 16.02903 | 2.29196 | 20.67938 |
| H | 16.61795 | 2.42983 | 21.57250 |
| C | 17.80330 | -3.97912 | 12.71141 |
| H | 18.46848 | -4.06214 | 13.57141 |
| C | 13.09155 | -2.75852 | 18.61587 |
| H | 13.74889 | -3.45878 | 19.10236 |
| C | 16.42464 | -4.51195 | 13.13814 |
| H | 15.74911 | -4.43067 | 12.29288 |
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| C | 12.49824 | -5.74291 | 14.72496 |
| C | 10.09164 | -0.11768 | 15.57478 |
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| C | 16.09509 | 1.09367 | 19.98459 |
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| C | 11.35878 | -10.43827 | 13.53526 |
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| O | 12.40098 | 6.54763 | 22.05954 |
| C | 10.13558 | 1.25993 | 17.61060 |

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| C | 11.10656 | -11.70999 | 12.73216 |
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| H | 11.84069 | -12.48215 | 12.95651 |
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| C | 12.73569 | -1.61447 | 19.31335 |
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| O | 10.49694 | 4.52040 | 20.69193 |
| C | 11.87914 | -0.68541 | 18.75488 |
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| H | 18.58362 | -5.84467 | 11.88970 |
| H | 19.32774 | -4.39545 | 11.25277 |
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| H | 13.58841 | 8.32513 | 19.68458 |

| | | | |
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| O | 11.54127 | 8.58664 | 21.58305 |
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| H | 14.13922 | 6.41695 | 23.03849 |
| C | 7.70836 | 3.03014 | 20.86838 |
| H | 8.12230 | 3.38413 | 21.80977 |
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| H | 7.63996 | 3.87572 | 20.18585 |
| O | 10.10433 | 4.67094 | 22.92162 |
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| H | 8.00548 | 1.52747 | 19.40281 |
| C | 10.79575 | 3.09881 | 20.74884 |
| H | 11.83690 | 3.01158 | 20.46786 |
| H | 10.67273 | 2.74413 | 21.76204 |
| C | 15.52248 | 7.90464 | 18.87432 |
| H | 16.53918 | 7.70474 | 19.21524 |
| H | 15.55654 | 8.77787 | 18.22595 |
| H | 15.20347 | 7.06204 | 18.26636 |
| C | 15.03958 | 9.42781 | 20.78629 |
| H | 14.34868 | 9.71796 | 21.57329 |
| H | 15.09268 | 10.25746 | 20.08407 |
| H | 16.02887 | 9.30386 | 21.22849 |
| C | 8.65375 | 0.74110 | 21.26632 |
| H | 9.13990 | -0.12671 | 20.82788 |
| H | 7.65937 | 0.43111 | 21.58330 |
| H | 9.21057 | 1.02296 | 22.15801 |
| C | 11.41973 | 7.38949 | 21.72095 |
| C | 10.22519 | 5.17963 | 21.82547 |
| C | 10.10685 | 6.65532 | 21.55793 |
| H | 9.75219 | 6.84441 | 20.54852 |
| H | 9.39675 | 7.09307 | 22.25426 |

| | | | |
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| O | 17.33779 | -6.04092 | 18.12792 |
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| H | 16.07750 | -5.69291 | 19.34600 |
| H | 14.89567 | -6.52033 | 19.90095 |
| O | 18.60287 | 3.81867 | 21.91097 |
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| H | 17.75284 | 4.27483 | 22.02816 |
| O | 19.13170 | 2.95412 | 19.38675 |
| H | 18.85539 | 3.29224 | 20.26344 |
| H | 20.08249 | 3.10499 | 19.35296 |
| O | 14.02753 | 0.42938 | 22.42520 |
| H | 13.37318 | 0.45427 | 21.71122 |
| H | 13.51627 | 0.10006 | 23.19683 |
| O | 13.55678 | -6.92483 | 9.76862 |
| H | 13.23469 | -6.52469 | 8.95248 |
| H | 12.76631 | -7.18438 | 10.25895 |
| O | 12.11749 | -1.41893 | 11.31612 |
| H | 12.82078 | -1.97251 | 11.68636 |
| H | 12.24891 | -1.42736 | 10.36417 |
| O | 19.38125 | 0.19205 | 18.89485 |
| H | 19.11093 | 1.09787 | 19.11765 |
| H | 19.11631 | -0.37178 | 19.64730 |
| O | 10.85623 | 2.57433 | 13.83312 |
| H | 10.12449 | 2.78441 | 14.42272 |
| H | 10.97344 | 3.34070 | 13.26270 |
| O | 18.97320 | -0.59431 | 16.35491 |
| H | 19.86640 | -0.56801 | 15.99905 |
| H | 19.07335 | -0.35466 | 17.30452 |
| O | 17.13989 | 1.27265 | 15.55890 |

| | | | |
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| H | 17.42048 | 2.07909 | 16.02662 |
| O | 10.92568 | 2.28802 | 24.17838 |
| H | 10.53662 | 3.12335 | 23.87643 |
| H | 10.86350 | 2.29918 | 25.13810 |
| O | 16.36266 | -8.57717 | 17.15251 |
| H | 17.19729 | -8.68366 | 16.65152 |
| H | 15.66639 | -8.90391 | 16.56595 |
| O | 16.31835 | 0.55660 | 13.01279 |
| H | 16.44450 | 1.31495 | 12.43463 |
| H | 16.59746 | 0.85030 | 13.89956 |
| O | 9.07107 | -2.63662 | 19.75985 |
| H | 8.22086 | -2.92075 | 20.11223 |
| H | 8.92385 | -2.53882 | 18.79647 |
| O | 18.60038 | -8.58998 | 15.56285 |
| H | 18.99489 | -7.69528 | 15.51233 |
| H | 18.40420 | -8.81317 | 14.64539 |
| O | 16.14449 | -1.37856 | 22.20584 |
| H | 15.63298 | -2.15759 | 21.90749 |
| H | 15.48609 | -0.67273 | 22.32276 |
| O | 10.90782 | 6.63062 | 16.92022 |
| H | 10.92164 | 7.52838 | 16.57677 |
| H | 11.41949 | 6.64031 | 17.74451 |
| O | 8.53663 | -2.59463 | 17.08567 |
| H | 8.56839 | -3.42903 | 16.58336 |
| H | 7.97684 | -1.99684 | 16.55779 |
| O | 18.04944 | 3.53013 | 16.94603 |
| H | 18.83275 | 3.89211 | 16.51779 |
| H | 18.32741 | 3.30319 | 17.85612 |
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| H | 20.34781 | -5.70811 | 15.61127 |
| O | 6.75766 | 1.56018 | 16.64818 |
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| H | 6.30531 | 2.30468 | 16.23984 |
| O | 15.79954 | 5.17527 | 16.73297 |
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| H | 9.05211 | 4.02126 | 17.01356 |
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| O | 13.39938 | 3.09288 | 22.77630 |
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| H | 12.62986 | 2.92783 | 23.34151 |
| O | 6.88595 | -0.93094 | 15.55572 |
| H | 5.97320 | -1.23463 | 15.56277 |
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| O | 9.95710 | 0.34531 | 12.23418 |
| H | 10.35339 | 1.05004 | 12.77343 |
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valPDI2 in chloroform

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| N | 10.24487679908361 | 0.35767381428421 | 16.39019718735362 |
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| N | 2.59179014806421 | 2.22313011046789 | 17.80560288940572 |
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| O | 5.69277353353916 | 0.34095477828395 | 16.87602216630160 |
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| C | 7.99183568091953 | 7.05506155526533 | 18.90393743509393 |
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| H | -0.81108811885812 | 12.78897890556928 | 27.25502710535811 |
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