

## EXPERIMENTAL DATA

**FUNCTIONAL UNIT:** This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

### Question 1

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET on page 7** in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

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Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453.10		
Carbon dioxide	677.35		233.45
Sulfuric acid	571.55		14.95
activated carbon	211.60		211.60
water			209.70
sodium sulfate			805.00
salicylic acid		1150	242.65

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

2.5

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.11		
Oil of wintergreen	14,886.1		
water			178.63
salicylic acid		1150	
methanol			266.66
Sodium sulfate			5636.85

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28
acetic acid	1001.87		
methyl acetate		616.67	32.07

1.5

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2** and calculate:

- a. in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 3 [2]
- b. in row 3 the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is **1359 kg**.  
 For route 3 theoretical yield of salicylic acid is **1191 kg**.
- c. in row 4 the E-factor (total waste) (in kg) [2]
- d. Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	<del>38.60%</del>	<del>39.98%</del>
<b>% Yield</b> How efficient the chemical reactions are.	<del>78.90%</del>	<del>94.72%</del>
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	<del>61.41%</del>	<del>29.23%</del>
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 3. Route 3 has a lower waste production (E-factor) and has a higher % reaction mass efficiency and % yield thus making it more green.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	<del>Cradle to gate</del> gate to gate ✓	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	No change as the plants in route 3 can be farmed while the oil in route 1 is non-renewable	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	Cost would go up as supply and demand states that a lower supply results in a higher demand hence a higher cost	

### Question 3

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable? [2]

Reusing waste for other processes and ensuring that the disposal of waste is not near populated areas as that would decrease the quality of life in that area. Reuse of waste products can decrease expenditure of raw materials for new processes.

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable? [2]

Using renewable energy sources rather than non-renewable. Less carbon emissions from renewable resources will increase quality of life as air quality has improved. Using renewable resources increase brand image which leads to higher sales and profit.

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  
Give 2 examples?

labor, distribution and packaging

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]

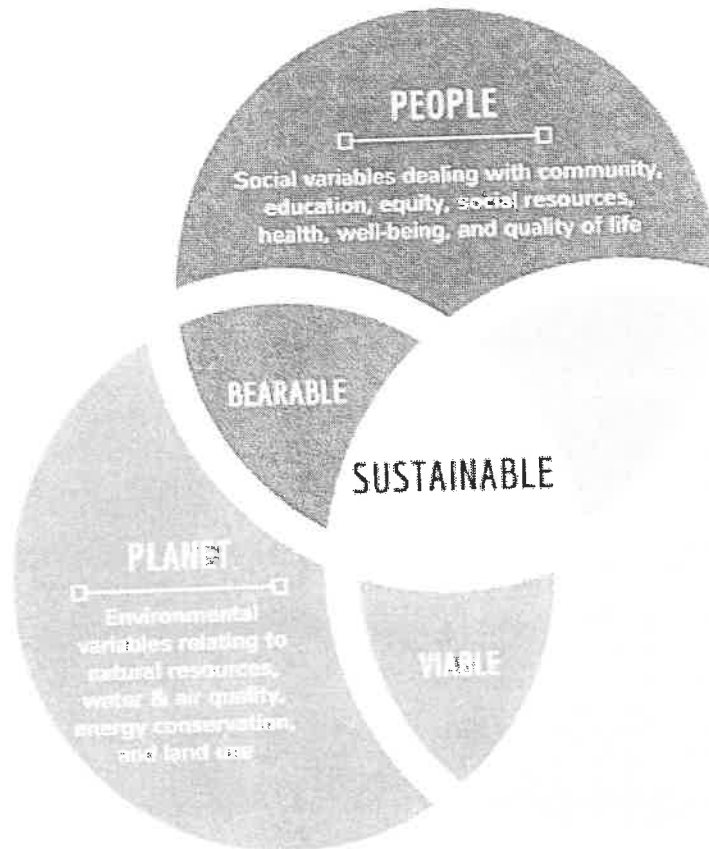
Refinement of chemical processes ensures efficiency and less waste production which leads to an increase of global sustainability.

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



#### Sustainability strives for:

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

#### ROLE PLAY ACTIVITY

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

Finance manager

1. Maximize profit
2. decrease costs
3. increase automation
4. ensure sustainability
5. ensure business growth
6. effective marketing strategies
7. increase production rate
8. tax ground.

Asprin Manufacturing

Community representative

1. Create jobs for local people
2. ~~hire~~ employ local people
3. decrease cost of product
4. don't monopolize
5. do not decrease quality of life by dumping nature
6. ensure quality of life through high air quality
7. ensure equal representation of minorities in the business.
8. Community Outreach Programs

Environmental Officer

1. Minimize deforestation
2. minimize carbon emissions
3. minimize water use
4. ~~do not~~ minimize contamination of water source
5. minimize methane emissions
6. reduce the use of non-renewable raw materials
7. use of renewable raw materials
8. ~~increase~~ ~~recycle~~ ~~recycle~~ ~~recycle~~ ~~more~~ Recycle more

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Have a community recycling division
4. Implement a community base water purification division
5. Ensure local acquisition of raw materials
6. Lower prices for involved community members
7. 0% carbon emissions
8. Panda habitat (endangered species)
9. Pursues towards involved community members
10. Automate processes to ensure more time is put toward sustainable development.

Asprin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?

No

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Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

### Question 1

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. Do not calculate the shaded blocks in the tables.

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET on page 7** in the aspirin manufacture document.

From Table 1 on page 7:

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$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453,10 ↗		
Carbon dioxide	677,35 ↗		233,45 ↗
Sulfuric acid	571,55 ↗		14,95 ↗
activated carbon	211,60 ↗		211,60 ↗
water			204,70 ↗
sodium sulfate			805,00 ↗
salicylic acid		1150	242,65 ↗

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945,11 ↗		
Oil of wintergreen	1488,61 ↗		
water			178,63 ↗
salicylic acid		1150	
methanol			266,66 ↗
Sodium sulfate			5636,85 ↗

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149,39 ↗
acetic acid	1001,87 ↗		
methyl acetate		616.67	32,07 ↗

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2** and calculate:

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 3. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is **1359 kg**.  
For route 3 theoretical yield of salicylic acid is **1191 kg**.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150}{2980,16} \times 100 = 38,60\%$	$\frac{1150}{5021,57} \times 100 = 22,90\%$
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100 = 84,62\%$	$\frac{1150}{1191} \times 100 = 96,56\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	1829,65 kg	6082,14 kg
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1, less waste is generated and it has a greater mass efficiency.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	No, it won't change because the green matrix results will remain the same.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	It will increase the price of buying aspirin, because the demand of crude oil will increase whilst the supply decreases thus crude oil becomes expensive.	

### Question 3

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable? [2]

Environmental: Reforestation and afforestation to replace the trees used to extract salicylic acid.  
 Social: the sulphuric acid could be used to lower pH of the land, so that it could be easy to apply more agricultural activities  
 Economic: because there is more agricultural activities, more job opportunities arise

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable? [2]

Using solar panels to reduce the cost of energy and to produce energy in an environmentally friendly way required by the plant

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? Give 2 examples? [2]

The cost of disposing of waste materials.  
 The cost of buying and maintaining the equipment used.

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]

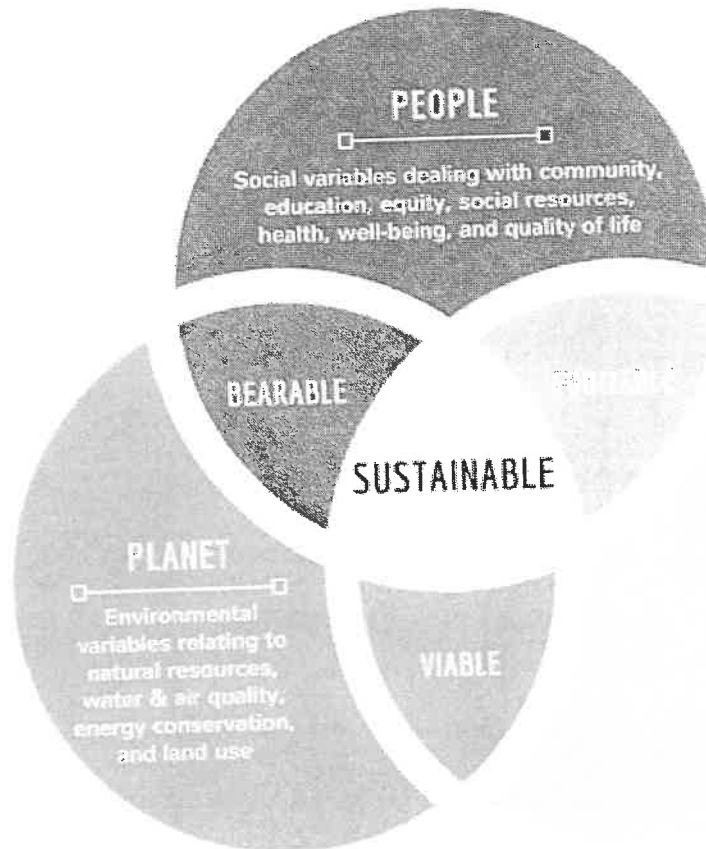
It enables us to choose a route which produces more desired outcomes and less waste products. This is beneficial for the environment

### SYSTEMS THINKING IN SUSTAINABILITY

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the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



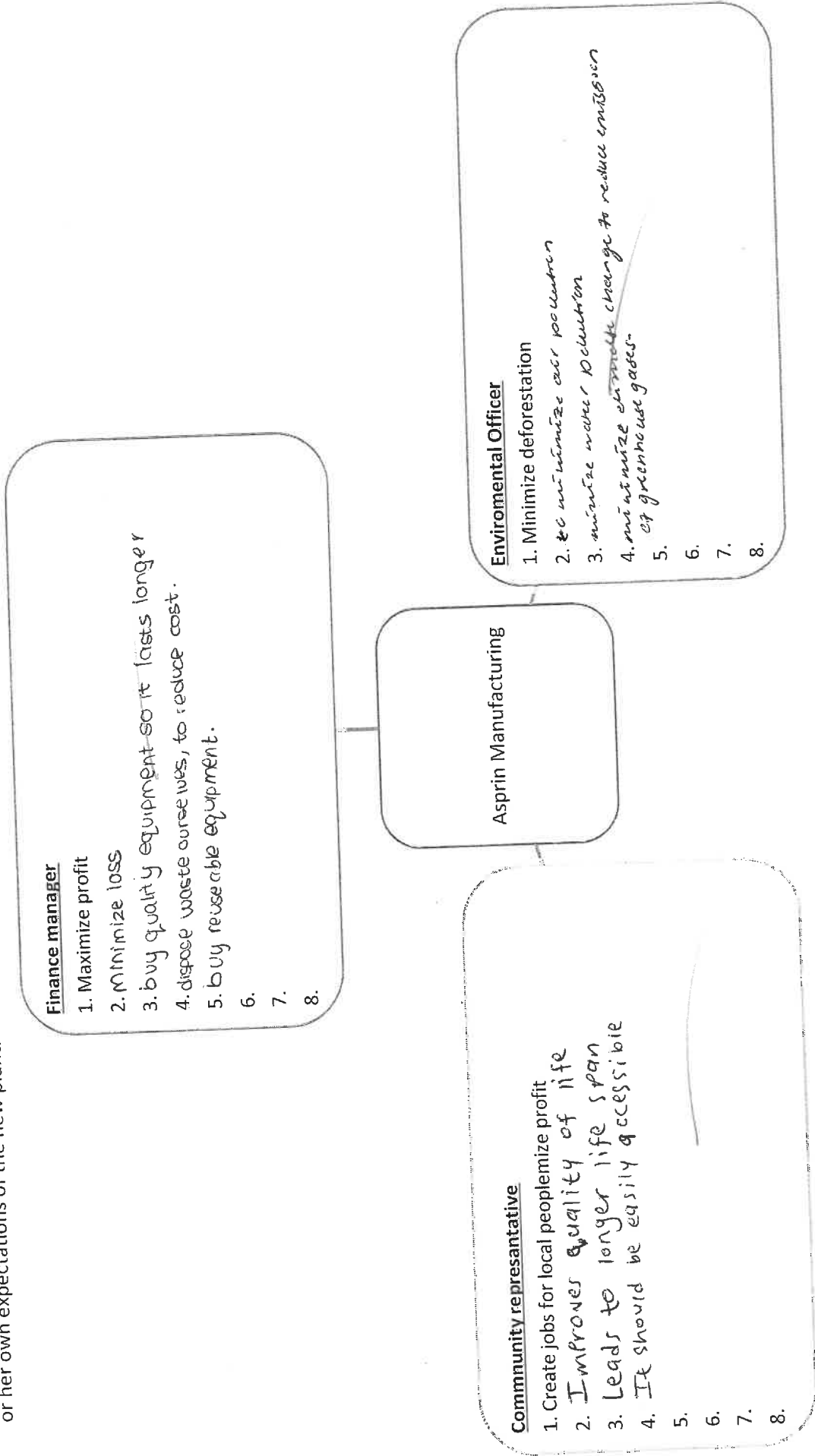
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- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
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#### ROLE PLAY ACTIVITY

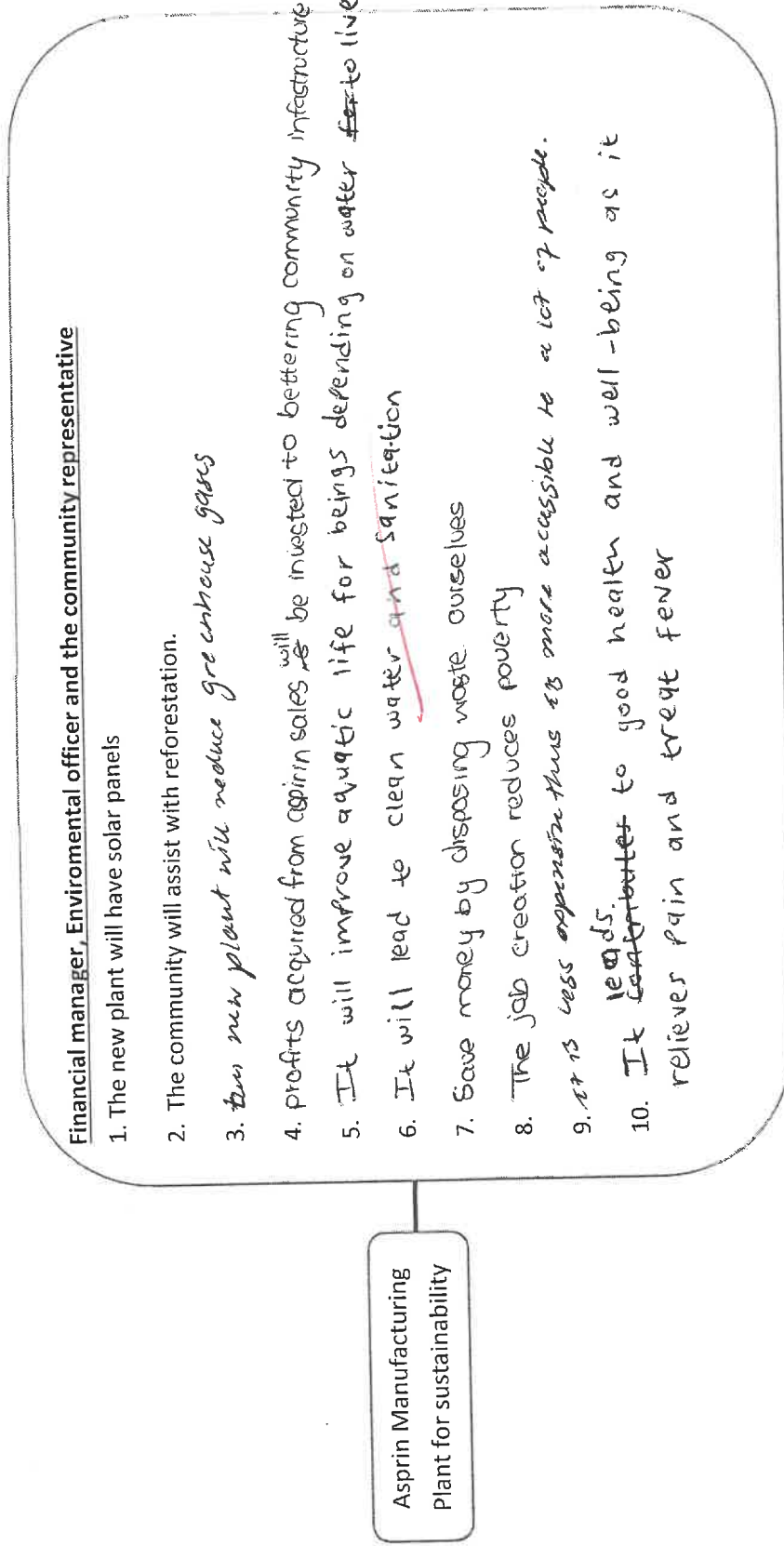
- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

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- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.



<b>Are all group members happy with the new plant design?</b>	<del>No</del> <b>Yes</b>
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In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05	—	117.30
Sodium hydroxide	453,10	—	—
Carbon dioxide	677,35	—	233.45
Sulfuric acid	571,55	—	14.95
activated carbon	2.11,60	—	211.60
water	—	—	204.70
sodium sulfate	—	—	805.00
salicylic acid	—	1150	242.65

4 53.10

5

1150 x 100

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen  
(Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945,11 ✓		
Oil of wintergreen	1488,61 ✓		
water			178,63 ✓
salicylic acid		1150	
methanol			266,66 ✓
Sodium sulfate			5636,85

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas  
(Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149,28 ✓
acetic acid	1001,87 ✓		
methyl acetate		616.67	32,07 ✓

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2** and calculate:

- a. in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- b. in row 3 the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is **1359 kg**.  
 For route 3 theoretical yield of salicylic acid is **1191 kg**.
- c. in row 4 the E-factor (total waste) (in kg) [2]
- d. Answer the four questions that follow. [7]

1829,65

13

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150}{2979,65} \times 100$ = 38,60%	$\frac{1150}{521,57} \times \frac{100}{1}$ 22,90%
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100\%$ = 84,62%	$\frac{1150}{1191} \times 100\%$ = 96,56%
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	233,45 + 117,30 + 14,95 + 211,60 + 204,70 + 805,00 + 242,65 = 1829,65	178,63 + 266,66 + 5636,25 = 6082,14 kg
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1. Because the atom economy is higher for route 1 which means that there is less waste and its better for the environment (ecofriendly)	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes, because the raw material extraction process of route 3 is more sustainable than route 1.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	The depletion of crude oil would mean less availability of phenol since oil is a non renewable resource. Therefore the production of it would rise, hence the price of aspirin would increase.	

### Question 3

Answer the following questions in the space provided:

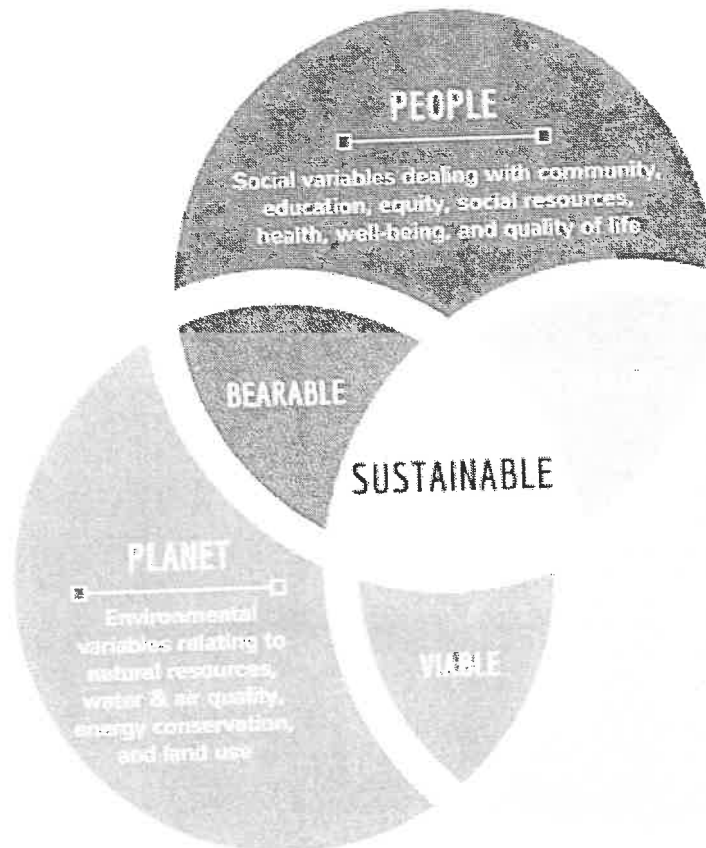
A	<p>How can the impact (<b>environmental, social or economic</b>) of the inputs and output be minimized or managed to make the process sustainable? [2]</p>	<ul style="list-style-type: none"> <li>By recycling by-products leads to a rise in job opportunities, <del>also</del> as well as minimize waste ✓</li> <li>Areas of extraction processes would be less polluted, if renewable processes were employed. ✓</li> </ul>
B	<p>How can the impact (<b>environmental, social or economic</b>) of energy be minimized or managed to make the process sustainable? [2]</p>	<ul style="list-style-type: none"> <li>Cost of transportation ✓</li> <li>Cost of labour/work force ✓</li> </ul>
C	<p>Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2] Give 2 examples?</p>	<ul style="list-style-type: none"> <li>Less usage of non-renewable processes, as well as the use of more renewable ones, would ensure minimal energy, which is then more sustainable ✓</li> <li>Larger work forces would make the labour less intensive ✓</li> </ul>
D	<p>How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]</p>	<p>Through chemistry, we are able to determine sustainable processes and improvements, to ensure <sup>that</sup> we have a more sustainable approach ✓</p>

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



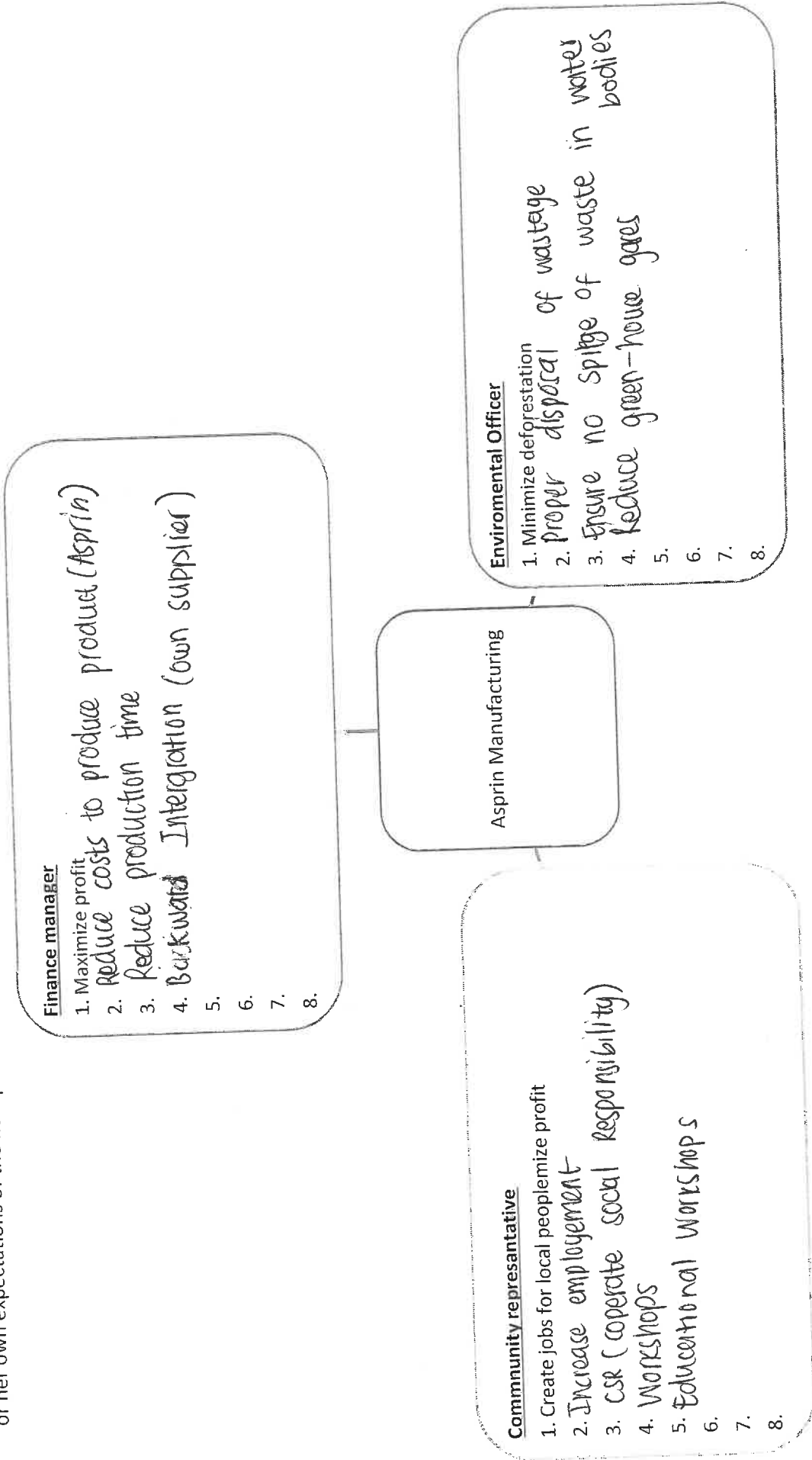
#### Sustainability strives for:

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

#### ROLE PLAY ACTIVITY

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.



- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Safe disposal of Aspirin waste
4. Ensure efficient Aspirin processes (time and cost effect)
5. Community workshops
6. Water filtration.
7. Responsible consumption & production of Aspirin
8. Affordable and clean energy. (to minimize wastage)
- 9.
- 10.

Aspirin Manufacturing Plant for sustainability

Are all group members happy with the new plant design?

Yes	No
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**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. Do not calculate the shaded blocks in the tables.

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol.

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453,70 SF		
Carbon dioxide	677,35		233,45
Sulfuric acid	571,55		14,95
activated carbon	211,60		211,67
water			204,71
sodium sulfate			8057
salicylic acid		1150	242,65

[5]  
4,15

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]  
 Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document. (2)

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen  
 (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945,11 ↑		
Oil of wintergreen	1488,606 ↑ <sub>st</sub>		
water			178,63 ↑
salicylic acid		1150	
methanol			266,66 ↑
Sodium sulfate			5636,89

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]  
 Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document. (1.5)

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas  
 (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149,08 ↑
acetic acid	1001,87 ↑		
methyl acetate		616.67	32,07 ↑

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in **Table 4 row 1**. Refer to the **calculated mass in Table 1 and 2** and calculate:

- in **row 2** the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- in **row 3** the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is **1359 kg**.  
For route 3 theoretical yield of salicylic acid is **1191 kg**.
- in **row 4** the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150}{2979,27} \times 100 = 38,60\%$	$\frac{1150}{2864,97} \times 100 = 40,14\%$
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100 = 84,62\%$	$\frac{450616}{1191} \times 100 = 51,78\%$ (3)
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	1829,65	449,03
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 3 less waste is produced and it has a higher reaction mass efficiency	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Cradle to grave. X	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes, because if the method used to extract the raw materials is harmful to the environment, then it is not as sustainable	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	It would increase the cost of buying aspirin as more costs will go into searching for phenol from crude oil which is being depleted as a non-renewable resource. A high demand and low supply would increase the cost.	

**Question 3**

Answer the following questions in the space provided:

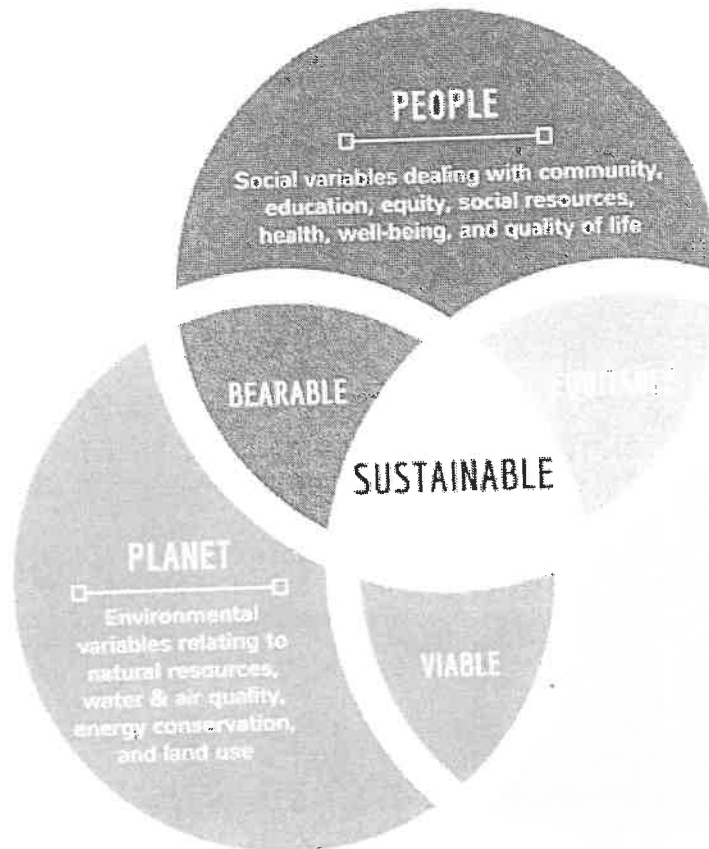
<p>How can the impact (<b>environmental, social or economic</b>) of the inputs and output be minimized or managed to make the process sustainable?</p> <p>[2]</p>	<ul style="list-style-type: none"> <li>• We can discard waste properly to prevent water from being contaminated and <del>to reduce land and</del> the soil.</li> <li>• We can re-use the outputs so that we can be able to save money.</li> <li>• After removing the trees, another trees should be planted to avoid climate change.</li> </ul>
<p>How can the impact (<b>environmental, social or economic</b>) of energy be minimized or managed to make the process sustainable?</p> <p>[2]</p>	<ul style="list-style-type: none"> <li>• Companies can invest more in greener sources of energy.</li> <li>• Companies who use fossil fuels can try to</li> </ul>
<p>Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur?</p> <p>Give 2 examples?</p> <p>[2]</p> <p>①</p> <p>②</p>	<ul style="list-style-type: none"> <li>• Repairing of machines and waste disposal</li> </ul>
<p>How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues?</p> <p>[2]</p>	<ul style="list-style-type: none"> <li>• It helped us to understand the roots of aspirin synthesis.</li> <li>• It helped us to be able to determine the most sustainable route</li> <li>• It allows us to make products that are beneficial</li> </ul>

**SYSTEMS THINKING IN SUSTAINABILITY**

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Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



#### Sustainability strives for:

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

#### ROLE PLAY ACTIVITY

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

Finance manager

1. Maximize profit
2. Reduce the cost of production
3. Market brand efficiency
4. Get investors to contribute to Capital
5. Charity work to reduce tax paid by company
6. Find most affordable method to discard of waste
- 7.
- 8.

Asprin Manufacturing

Community representative

1. Create jobs for local people to maximize profit
2. Don't dump waste in community during disposal
3. Product must be beneficial for users
4. Make affordable to all community members
5. Educate people on the product
6. Make product accessible to all rural and urban communities
7. Sell to local businesses (Support small businesses)
8. Charity work, provide medicine to underprivileged.

Environmental Officer

1. Minimize deforestation
2. Try minimize pollution
3. Extract crudel in areas with little aquatic life
4. Don't contaminate water during extraction (avoid oil)
5. Don't overuse water
6. Recycle waste products if possible
7. Use cleaner sources of energy
8. Reduce over extraction of raw materials.

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Get investors that specifically focus on sustainability
4. Affordable and efficient waste management
5. Create jobs for local ~~peop~~ people
6. Support small businesses by working with them to distribute our product
7. Reduce over ~~extraction~~ extraction of raw materials
8. Use more capital to go with a more sustainable production route
9. Recycle waste products
10. Prevent dumping in communal areas to reduce risk of diseases

Asprin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?

Yes	No
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**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid from phenol**  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453.10 ↑		
Carbon dioxide	677.35 ↑		233.45 ↑
Sulfuric acid	571.55 ↑		14.95 ↑
activated carbon	211.60 ↑		211.60 ↑
water			204.70 ↑
sodium sulfate			805.00
salicylic acid		1150	242.65

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen  
(Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.117		
Oil of wintergreen	1488.617		
water			178.637
salicylic acid		1150	
methanol			266.667
Sodium sulfate			5636.857

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas  
(Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.287
acetic acid	1001.877		
methyl acetate		616.67	32.077

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2** and calculate:

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is **1359 kg**.  
For route 3 theoretical yield of salicylic acid is **1191 kg**.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150}{2979.65} \times 100\%$ $= 38.60\%$	$\frac{1150}{5021.57} \times 100\%$ $= 22.90\%$
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100\%$ $= 84.62\%$	$\frac{1150}{1191} \times 100\%$ $= 96.56\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	$233.45 + 14.95 + 211.60 + 204.77 + 805 + 242.65$ $= 1829.65 \text{ kg}$	$178.63 + 266.66 + 5636.85$ $= 6082.14 \text{ kg}$
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1. This route produces less waste.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Cradle to grave	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes. An efficient raw material extraction process will result in the saving of energy, which is more sustainable.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	The cost of buying aspirin would increase, because the supply of aspirin is limited to the production of crude oil, so if the crude oil is depleted then the supply and demand of aspirin will not be equal leading to an increase in the price of aspirin.	

### Question 3

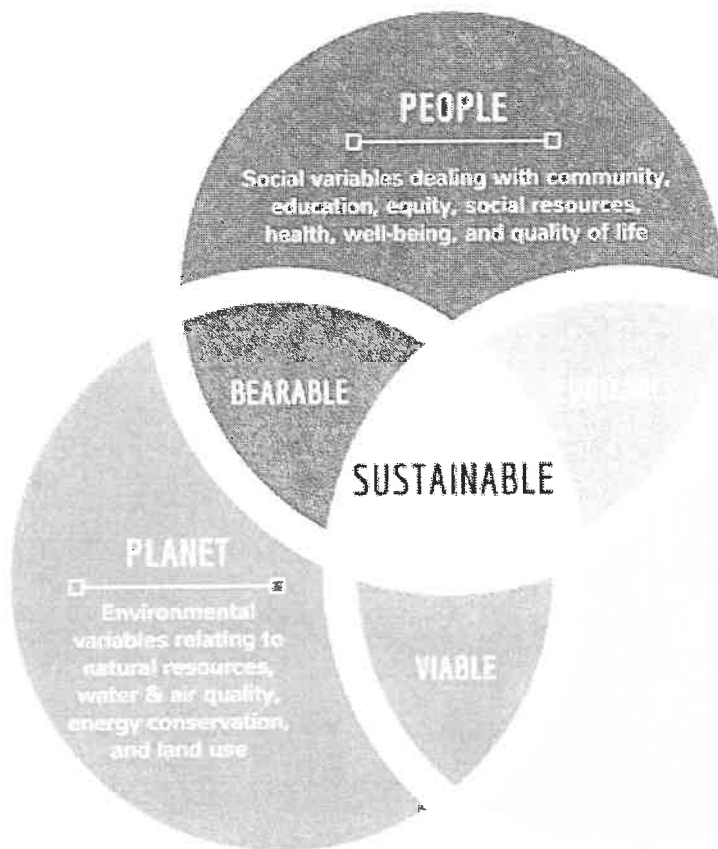
Answer the following questions in the space provided:	
How can the impact ( <b>environmental, social or economic</b> ) of the inputs and output be minimized or managed to make the process sustainable? [2]	<u>Environmental</u> • Using resources efficiently. • Maximizing the use of the reactants in order to reduce waste.
How can the impact ( <b>environmental, social or economic</b> ) of energy be minimized or managed to make the process sustainable? [2]	• By reproducing the resources such as planting more trees (wintergreen) or finding alternative methods to produce other inputs (synthetically).
Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  Give 2 examples?	• Cost of packaging. • Distribution costs.
How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]	The exercise educates us more about sustainability which shows that chemistry is putting in necessary measures, such as the program of Green Chemistry. Which aims to minimize the environmental impact during the production and use of chemicals.

### SYSTEMS THINKING IN SUSTAINABILITY

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**ROLE PLAY ACTIVITY**

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- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

**Finance manager**

1. Maximize profit
2. Contributes to the tax
3. Contributes to the gross of the country.
4. Increase the supply and decrease demand leading
5. the low price value.
- 6.
- 7.
- 8.

Asprin Manufacturing

**Community representative**

1. Create jobs for local people to profit
2. To ensure safety of the local people
3. Advocate for the necessary human rights
4. Promote the education of local people
5. about the plant and how it works.
- 6.
- 7.
- 8.

**Environmental Officer**

1. Minimize deforestation
2. Minimize pollution
3. Educate people on global warming and climate change.
- 4.
5. Promote the habit of recycling and reusing.
- 6.
- 7.
- 8.

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be levers, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. The community will form a local recycling program.
4. The community representatives will educate the locals on sustainability.
5. The reforestation will improve air quality therefore reduce air pollution.
- 6.
- 7.
- 8.
- 9.
- 10.

Asprin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?

<del>Yes</del>	No
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## EXPERIMENTAL DATA

**FUNCTIONAL UNIT:** This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

### Question 1

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

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4.5

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(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
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Sodium hydroxide	453.10 ✓		
Carbon dioxide	677.35 ✓		233.45 ✓
Sulfuric acid	571.55 ✓		14.95 ✓
activated carbon	211.60 ✓		211.60 ✓
water			204.70 ✓
sodium sulfate			805.00
salicylic acid		1150	242.65 ✓

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7) 2.5

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.11 ✓		
Oil of wintergreen	1488.61 ✓		
water			178.63 ✓
salicylic acid		1150	
methanol			286.66 ✓
Sodium sulfate			5636.85 ✓

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7) 1.5

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28 ✓
acetic acid	1001.87 ✓		
methyl acetate		616.67	32.07 ✓

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2 and calculate:**

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 3. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is **1359 kg**.  
For route 3 theoretical yield of salicylic acid is **1191 kg**.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	38.60%	40.14%
<b>% Yield</b> How efficient the chemical reactions are.	84.62%	51.78%
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	<del>61.40%</del> 1829.65	<del>29.23%</del> 449.05
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 3 because it produces less waste	
What type of system boundary ( <b>Cradle to gate or Gate to grave or Cradle to grave or Gate to gate</b> ) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	It wouldn't because Route 3 uses wintergreen oil from trees which can be replanted unlike Route 1 which uses crude oil which depletes	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	Aspirin prices would increase because more costs would be incurred to access crude oil due to its scarcity	

### Question 3

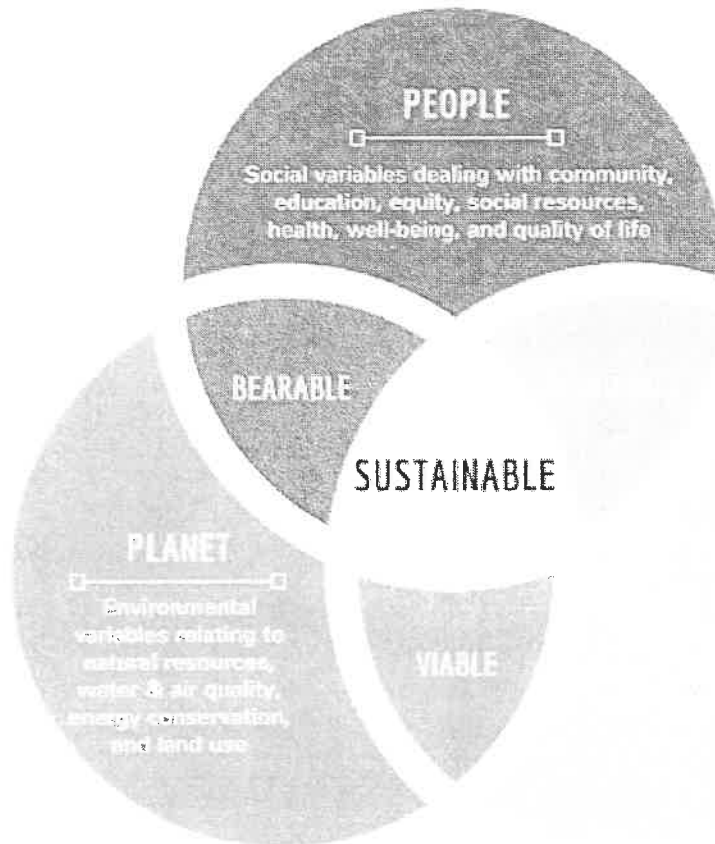
Answer the following questions in the space provided:	
How can the impact ( <b>environmental, social or economic</b> ) of the inputs and output be minimized or managed to make the process sustainable? [2]	recycling, reusing and reducing awareness campaigns ✓ hefty fines for those that are ignorant to sustainable use of resources ✓ investment in catalytic conversion ✓
How can the impact ( <b>environmental, social or economic</b> ) of energy be minimized or managed to make the process sustainable? [2]	use of renewable energy sources that have minimal impacts e.g solar, wind turbines etc ✓
Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  Give 2 examples?	labour costs waste management ✓ disaster management e.g spills or explosions within the factory ✓
How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]	we can find alternative aspirin production methods with higher carbon efficiency and lower waste production ✓

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



#### Sustainability strives for:

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

#### ROLE PLAY ACTIVITY

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

Finance manager

1. Maximize profit
2. High demand leading to high supply
3. Cheaper labour
4. longer lasting equipment
5. minimize waste production to minimise
6. plan for unexpected costs
7. maximize production
- 8.

Asprin Manufacturing

Community representative

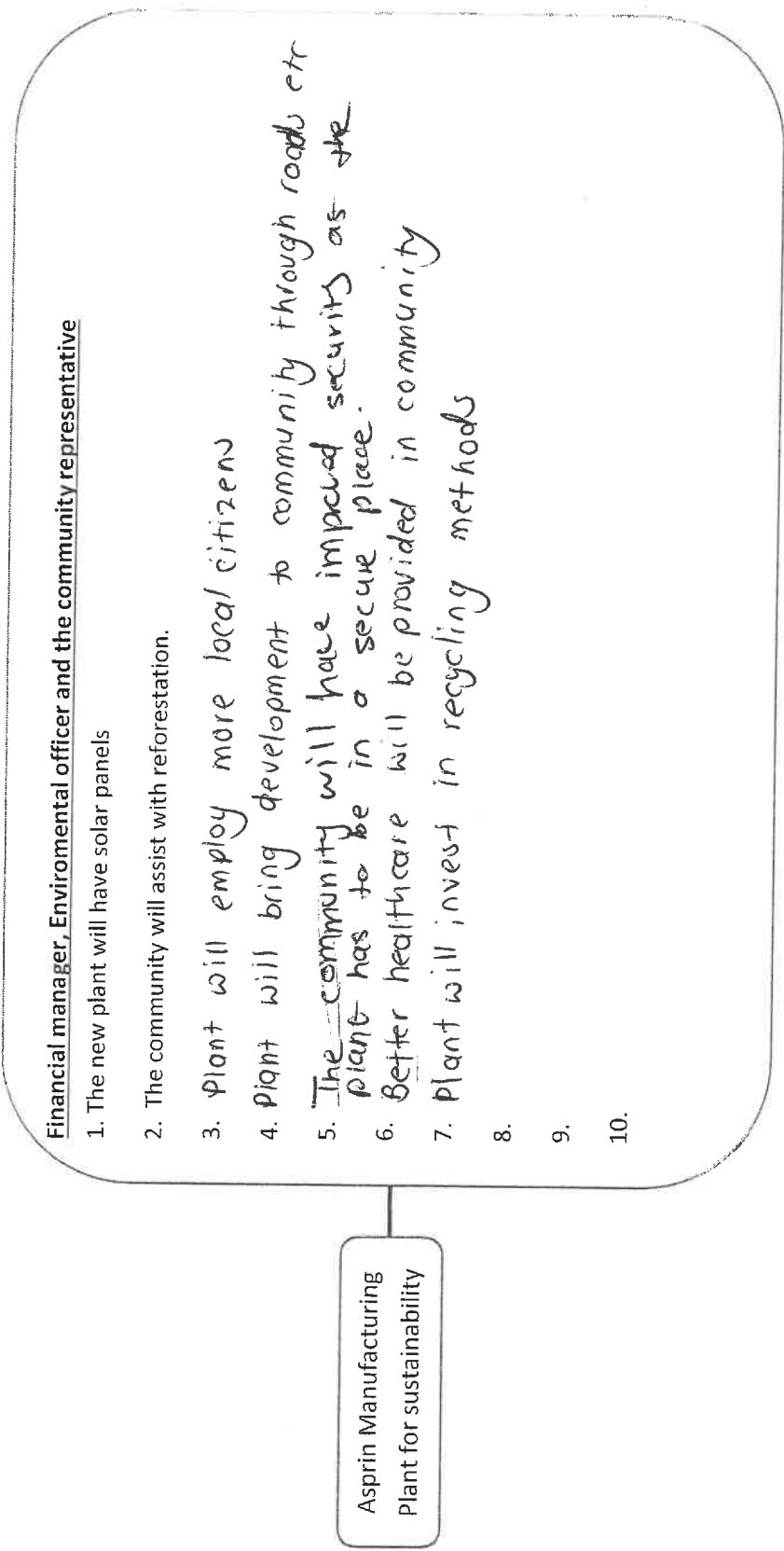
1. Create jobs for local people to profit
2. Improve living standards
3. urbanisation of local community.
4. improved security
5. improved quality of education
6. gender equality
- 7.
- 8.

Environmental Officer

1. Minimize deforestation
2. minimize land pollution
3. minimize water pollution
4. use renewable energy source
5. Invest in reusing and recycling methods
6. Efficient & sustainable waste management
7. reduce carbon emission
8. use biodegradable packaging

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.



Asprin Manufacturing Plant for sustainability

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Plant will employ more local citizens
4. Plant will bring development to community through roads etc
5. The community will have improved security as the plant has to be in a secure place.
6. Better healthcare will be provided in community
7. Plant will invest in recycling methods
- 8.
- 9.
- 10.

Are all group members happy with the new plant design?

Yes  No

**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

*Refer to the aspirin manufacture information document where necessary.*

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

**Refer to the Table 1 DATA SHEET on page 7** in the aspirin manufacture document.

**From Table 1 on page 7:**

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453.10		
Carbon dioxide	677.35		
Sulfuric acid	1945.11		
activated carbon	211.60		
water			204.70
sodium sulfate			805.00
salicylic acid		1150	242.65

5

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

2.5

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.11		
Oil of wintergreen	1488.61		
water			178.63
salicylic acid		1150	
methanol			266.66
Sodium sulfate			4262.27

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

1.5

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28
acetic acid	1001.87		
methyl acetate		616.67	32.07

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in **Table 4 row 1**. Refer to the **calculated mass in Table 1 and 2** and calculate:

- a. in **row 2** the % reaction mass efficiency of salicylic acid for the route 1 and route <sup>3</sup>2. [2]
- b. in **row 3** the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is **1359 kg**.  
 For route 3 theoretical yield of salicylic acid is **1191 kg**.
- c. in **row 4** the E-factor (total waste) (in kg) [2]
- d. Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	<del>27,78%</del>	<del>5,10%</del>
<b>% Yield</b> How efficient the chemical reactions are.	<del>84,62%</del>	<del>96,56%</del>
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	<del>7,38%</del>	<del>0%</del>
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 3 because there is no waste ✓✓	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate ✓	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes as adding raw materials to route 3 would result in waste.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	Price of Aspirin would increase as only two routes can be used to produce Aspirin.	

### Question 3

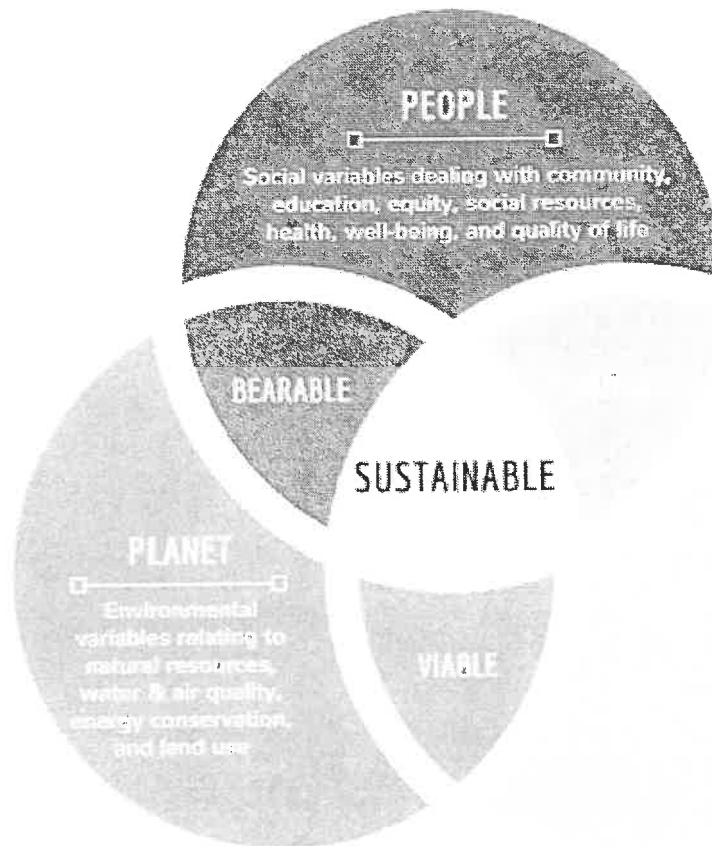
Answer the following questions in the space provided:	
How can the impact ( <b>environmental, social or economic</b> ) of the inputs and output be minimized or managed to make the process sustainable? [2]	By recycling the byproducts used produced in the process. These byproducts can be used to produce other products and in doing so it is not necessary to discard it as waste. This reduces the environmental impact as there is less waste and can also influence the economic impact because you get raw materials for free.
How can the impact ( <b>environmental, social or economic</b> ) of energy be minimized or managed to make the process sustainable? [2]	By making use of renewable energy instead of relying on fossil fuels we can reduce the pollution caused by burning fossil fuels and minimize the environmental impact.
Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  Give 2 examples?	The cost of waste management and the cost of the employees salary.
How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]	Chemistry can be used to find new ways to produce certain products or to maximize the efficiency of the process, and in doing so increase the sustainability.

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

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the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



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- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

#### ROLE PLAY ACTIVITY

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

Finance manager

1. Maximize profit
2. Maximize efficiency of the system
3. Maximize prices of Aspirin
4. Increase demand and supply
- 5.
- 6.
- 7.
- 8.

Asprin Manufacturing

Community representative

1. Create jobs for local people to profit
2. Increase tax contribution
3. Increase quality of life
4. Increase longevity of life
5. Allocate resources into the community
6. Decrease drug dependency and abuse
- 7.
- 8.

Environmental Officer

1. Minimize deforestation
2. Minimize waste produced
3. Minimize water pollution
4. Minimize air pollution
5. Minimize extraction of raw materials
- 6.
- 7.
- 8.

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be levers, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Asprin Manufacturing  
Plant for sustainability

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Allocate funds and resources for educational purposes in the community
4. New plant will use renewable resources.
5. The plant will support and partnership with local businesses.
6. Ensure gender equality in job opportunities.
7. Provide medical aid for all employees.
8. Company and community will have environmental programmes to ensure clean water/environment.
- 9.
- 10.

Are all group members happy with the new plant design?

Yes

No

## EXPERIMENTAL DATA

**FUNCTIONAL UNIT:** This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

### Question 1

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

**From Table 1 on page 7:**

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid from phenol**  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05	<del>927</del>	117.30
Sodium hydroxide	453.10 ↑		
Carbon dioxide	677.35 ↑		233.45 ↑
Sulfuric acid	571.55 ↑		14.95 ↑
activated carbon	211.60 ↑		211.60 ↑
water			204.70 ↑
sodium sulfate			805.00 ↑
salicylic acid		1150	242.65 ↑

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the **Table 2 DATA SHEET on page 7** in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of **1150 kg of salicylic acid from oil of wintergreen**  
(Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.11		
Oil of wintergreen	1488.61		
water			<del>155.33</del> 178.63
salicylic acid		1150	
methanol			266.66
Sodium sulfate			5636.89

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **616.67 kg of methyl acetate**. [1.5]

Refer to the **Table 3 DATA SHEET on page 7** in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of **616.67 kg of methyl acetate from natural gas**  
(Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28
acetic acid	1001.87		
methyl acetate		616.67	32.07

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2 and calculate:**

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is 1359 kg.  
For route 3 theoretical yield of salicylic acid is 1191 kg.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150\text{kg}}{2768.09\text{kg}} \times 100 = 41.55\%$	$\frac{1150\text{kg}}{6609.12\text{kg}} \times 100 = 17.40\%$
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150\text{kg}}{1359\text{kg}} \times 100 = 84.62\%$	$\frac{1150\text{kg}}{1191\text{kg}} \times 100 = 96.56\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	<del><math>\frac{1829.6\text{kg}}{2768.09\text{kg}} \times 100 = 65.79\%</math></del> <del><math>\frac{1829.6\text{kg}}{1359\text{kg}} \times 100 = 134.63\%</math></del>	<del><math>\frac{6082.11\text{kg}}{6609.12\text{kg}} \times 100 = 92.02\%</math></del> <del><math>\frac{6082.11\text{kg}}{1191\text{kg}} \times 100 = 510.67\%</math></del>
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1, because it has a better atom economy than route 2, as well as less waste as well.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	No, it does not reduce the waste by improving waste management.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	A decrease in the oil supply would increase the cost of production as the price of oil increases. Therefore the price of aspirin will increase and less sales.	

### Question 3

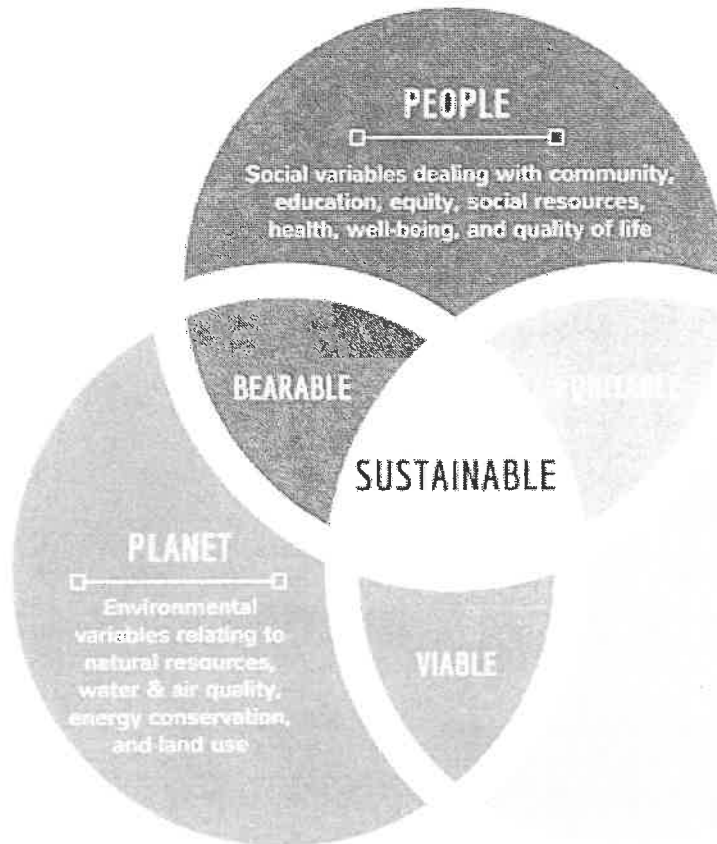
Answer the following questions in the space provided:	
How can the impact ( <b>environmental, social or economic</b> ) of the inputs and output be minimized or managed to make the process sustainable? [2]	• Implement a better waste management system, to benefit the environment and economy.
How can the impact ( <b>environmental, social or economic</b> ) of energy be minimized or managed to make the process sustainable? [2]	• Find a renewable energy source to replace burning fossil fuels to benefit the environment and social aspects, because of less pollution and rely less on the power grid.
Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  Give 2 examples?	• Costs of protective equipment • Costs of training personnel to follow procedures, when dealing with hazardous chemicals.
How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]	<del>When</del> We use chemistry to calculate the most efficient routes and most sustainable options beforehand.

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the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



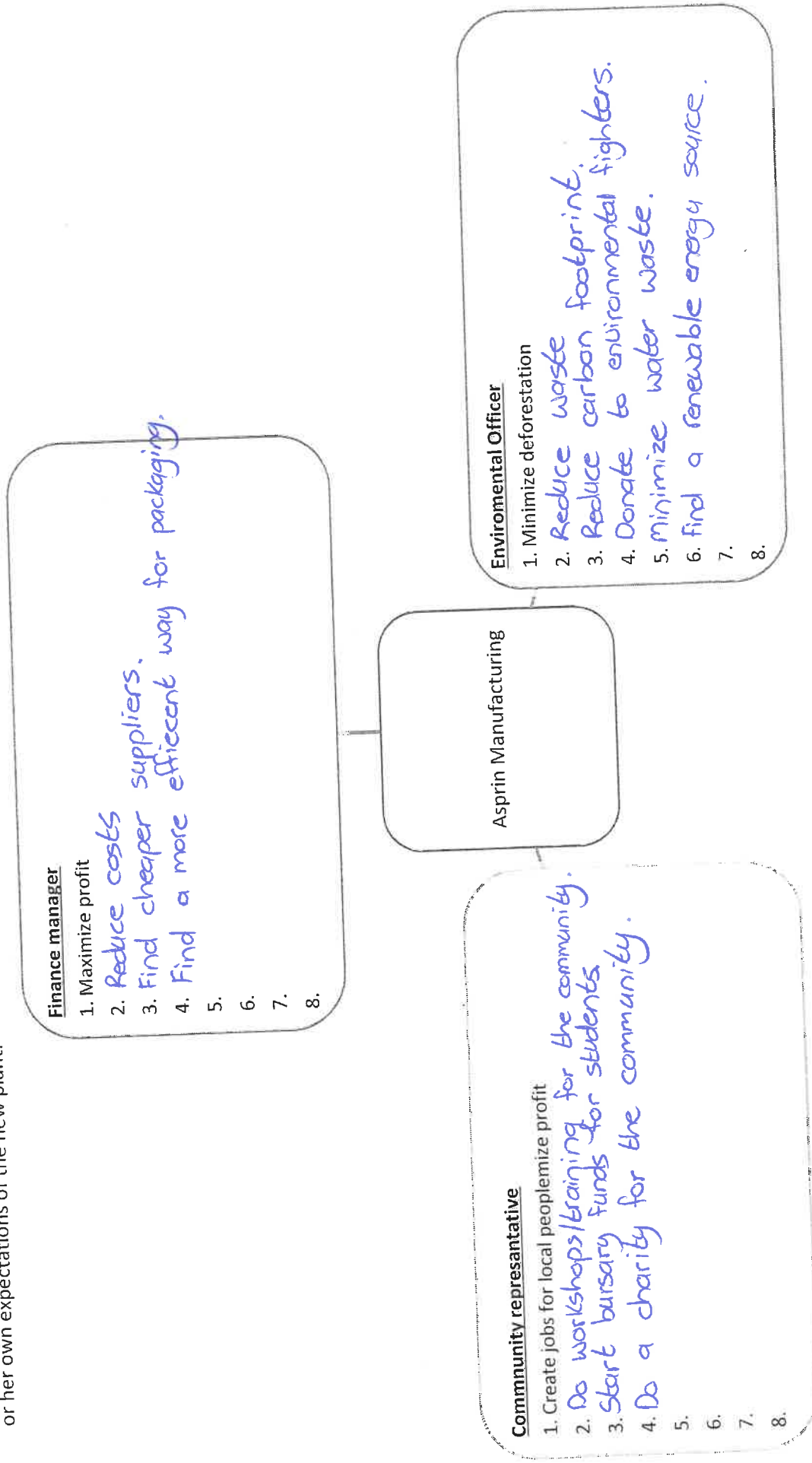
**Sustainability strives for:**

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
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**ROLE PLAY ACTIVITY**

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.



- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Asprin Manufacturing  
Plant for sustainability

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. *no discrimination depending on your sex*
4. *Employ the community.*
5. *Fair labour regulations.*
6. *Implement a waste management system, for water.*
7. *Use local suppliers to support them.*
8. *Create a community fund to feed community.*
- 9.
- 10.

Are all group members happy with the new plant design?

Yes

No

**EXPERIMENTAL DATA**

**FUNCTIONAL UNIT:** This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. Do not calculate the shaded blocks in the tables.

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$\frac{927 \text{ kg } C_6H_5OH}{1000 \text{ kg SA}} = \frac{1000 \text{ kg SA}}{1000 \text{ kg SA}}$$

$$\frac{x \text{ kg } C_6H_5OH}{1150 \text{ kg SA}} = \frac{1150 \text{ kg SA}}{1000 \text{ kg SA}}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05	—	117.30
Sodium hydroxide	453.10	—	—
Carbon dioxide	677.35	—	233.45
Sulfuric acid	571.55	—	14.95
activated carbon	211.60	—	211.60
water	—	—	204.70
sodium sulfate	—	—	805.00
salicylic acid	—	1150	242.65

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. (2.5)
- Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen  
(Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85	—	—
Sulfuric acid	1946.11 ↗	—	—
Oil of wintergreen	1488.61 ↗	—	—
water	—	—	178.63 ↗
salicylic acid	—	1150	—
methanol	—	—	266.66 ↗
Sodium sulfate	—	—	5636.85 ↗

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. (1.5)
- Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas  
(Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33	—	267.68
water	—	—	149.28 ↗
acetic acid	1001.87 ↗	—	—
methyl acetate	—	616.67	32.07 ↗

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the calculated mass in Table 1 and 2 and calculate:

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 3. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is 1359 kg.  
For route 3 theoretical yield of salicylic acid is 1191 kg.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{\text{actual mass of desired product}}{\text{total mass of all reactants}} \times 100$ $= \frac{1150}{2767.05} \times 100 = 41.55\%$ $= \frac{1150}{2979.65} \times 100 = 38.60\%$	$= \frac{1150}{5021.57} \times 100 = 22.90\%$
<b>% Yield</b> How efficient the chemical reactions are.	$= \frac{\text{actual mass of desired product}}{\text{expected mass of desired product}} \times 100$ $= \frac{1150}{1359} \times 100 = 84.62\%$	$= \frac{1150}{1191} \times 100 = 96.56\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	$= \frac{\text{mass of all reactants used}}{\text{mass of desired product}} \times 100$ $= \frac{1829.65}{1150} \times 100 = 159.05$	$= \frac{6072.14}{1150} \times 100 = 5280.12$
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	We would choose Route 1, as there is less waste produced and there is better utilization of reactants, as shown by % atom economy.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	No, as you will be adding in a step which will contradict sustainability of an extra step means more room for waste production.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	The price of aspirin will increase as the raw natural used to produce it will be scarce and	

### Question 3

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable? [2]

The impact can be minimized or managed by shortening the process by eliminating unnecessary steps because you will be limiting waste, thus producing more product. Start using renewable resources so that fuel, ~~non~~ renewable fuel, is reduced and the environment is saved from unnecessary toxicity.

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable? [2]

Cost efficient energies or fuels can be used example, local energies and fuels reducing the cost improving, benefiting economy by job creation and ~~low~~ carbon footprint without transport.

Renewable energy

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]

Give 2 examples?

~~the best~~ Example 1 would be the cost of moving your waste to waste plant/factory to discard of turn  
Example 2 would be labor costs

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]

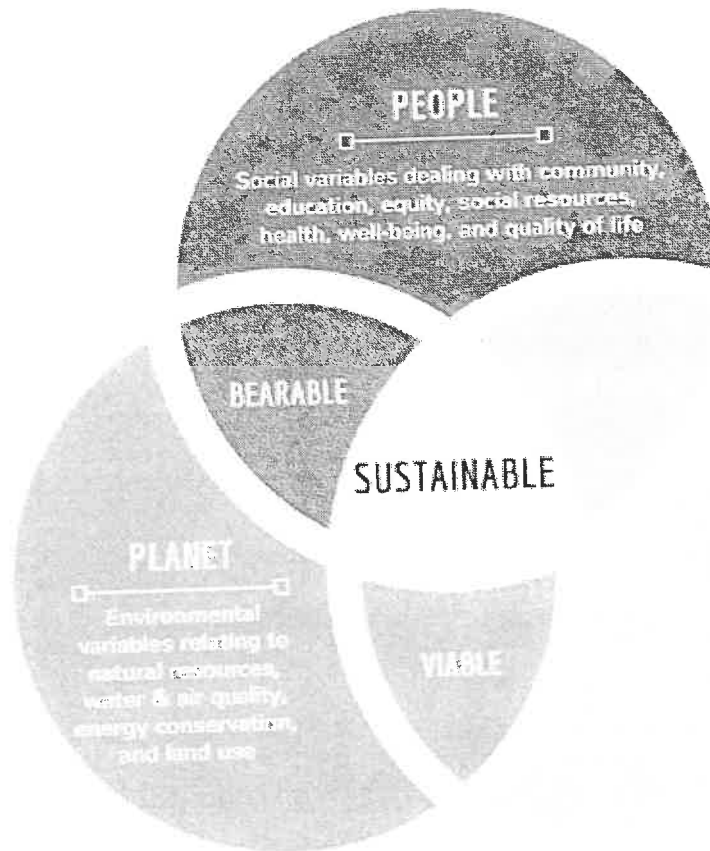
It shows us that in ~~archemical~~ reaction a chemical production the use of an expedited process can result in a large sum of wastage which is contradictory to sustainability as with wastage dilution is a result.

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



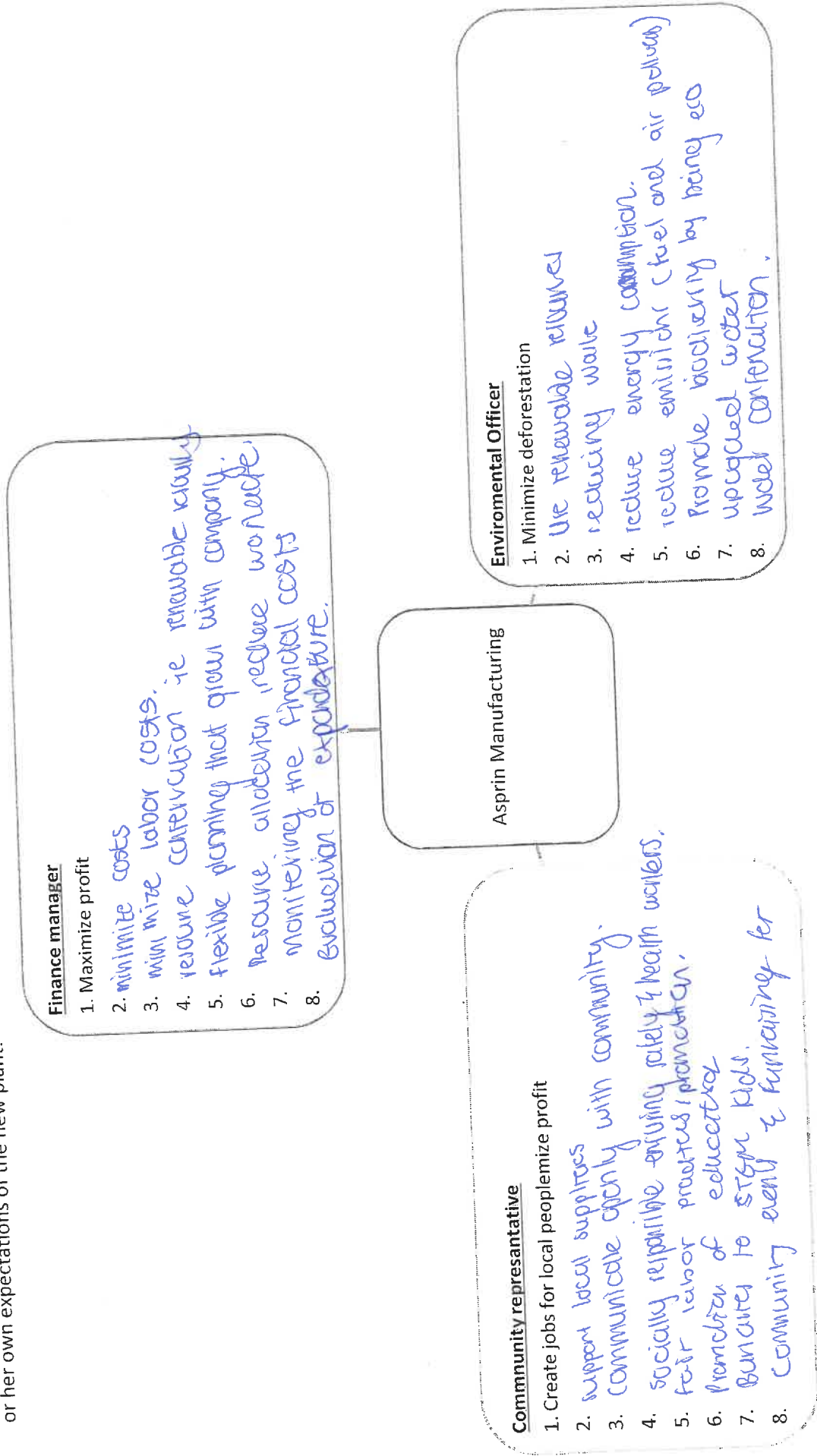
#### Sustainability strives for:

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#### ROLE PLAY ACTIVITY

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- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be levers, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. The new plant will reduce the emission of greenhouse gases which will be climate action.
4. The new plant reaches job creation; increase in economic growth, decrease poverty
5. The plant promotes gender equality ie fair labor practices
6. The plant will promote education as it will be used as research plant for tertiary education
7. The new plant will promote the reduction of greenhouse which will reduce acidic rain
8. The new plant recycles water, usage of up cycled water.
9. The new plant recycles old plastic tubes and used recycled goods.
10. The new plant has a hand in more profitable organizations help with origin and funding of schools.

Aspirin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?	
Yes <input checked="" type="checkbox"/>	No <input type="checkbox"/>

**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. Do not calculate the shaded blocks in the tables.

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05	948.75	117.30
Sodium hydroxide	453.10	453.10	0
Carbon dioxide	677.35	443.90	233.45
Sulfuric acid	571.55	556.60	14.95
activated carbon	211.60	0	211.60
water	0	204.70	204.70
sodium sulfate	0	805.00	805.00
salicylic acid	0	1150	242.65

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

2.5

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85	1587.85	0
Sulfuric acid	1945.11	1945.11	0
Oil of wintergreen	1488.61	1488.61	0
water	0	<del>178.63</del> 178.63	178.63
salicylic acid	0	1150	1150
methanol	0	266.66	266.66
Sodium sulfate	0	5636.85	5636.85

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

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$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

1.5

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33	266.65	267.68
water	0	149.28	149.28
acetic acid	1001.87	1001.87	0
methyl acetate	0	616.67	32.07

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in **Table 4 row 1**. Refer to the **calculated mass in Table 1 and 2** and calculate:

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is **1359 kg**.  
For route 3 theoretical yield of salicylic acid is **1191 kg**.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Table 4		
Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150}{1587.85+1945.11+1488.61} \times 100 = 22.90\%$ ✓	$\frac{1150}{1066.05+453.10+677.35+571.55} \times 100 = 38.60\%$ ✓
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100 = 84.62\%$ ✓	$\frac{1150}{1191} \times 100 = 96.56\%$ ✓
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	<del>242.65</del> 1712.35 kg ✓	<del>7232.14</del> kg ✓
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1. ✓ It produces less waste than Route 3	
What type of system boundary ( <b>Cradle to gate</b> or <b>Gate to grave</b> or <b>Cradle to grave</b> or <b>Gate to gate</b> ) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate ✓	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes, collection of ✓ oil of winter-green is more environmentally friendly than the collection of crude oil	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	It would increase the price of Aspirin because the supply would decrease therefore the demand would increase which increases the price.	

### Question 3

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable? [2]

~~Route 3~~ Oil from route 3 is more sustainable than that of the one in route 1.  
- use wintergreen oil rather than crude oil because it is more sustainable and environmentally friendly.

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable? [2]

- Use greener energy like solar power, wind, hydropower rather than fossil fuels.

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  
Give 2 examples?

\* Salaries to workers  
\* Sustaining the maintenance of the machines

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]

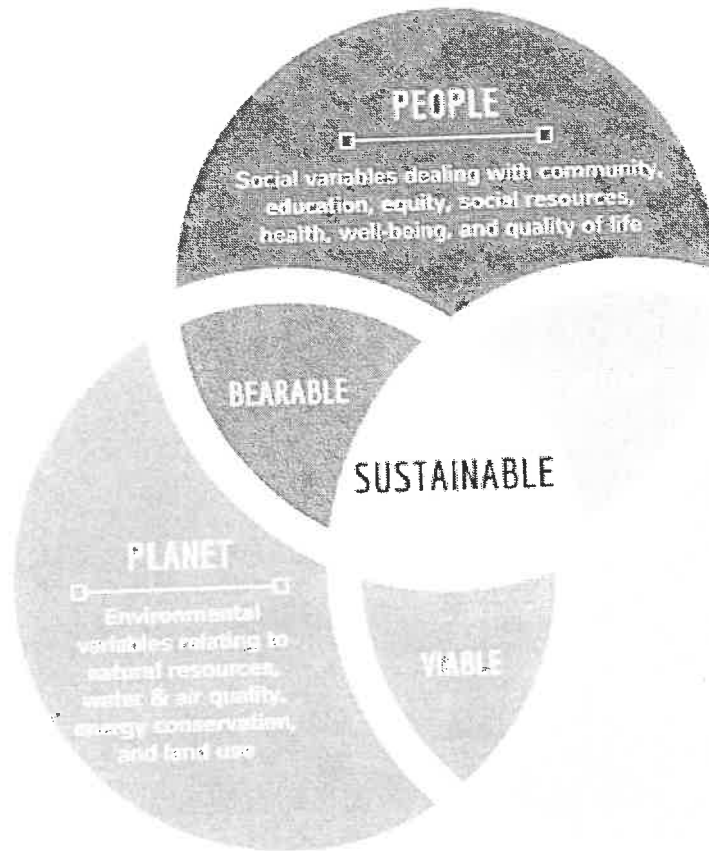
- To create more sustainable systems for the future.

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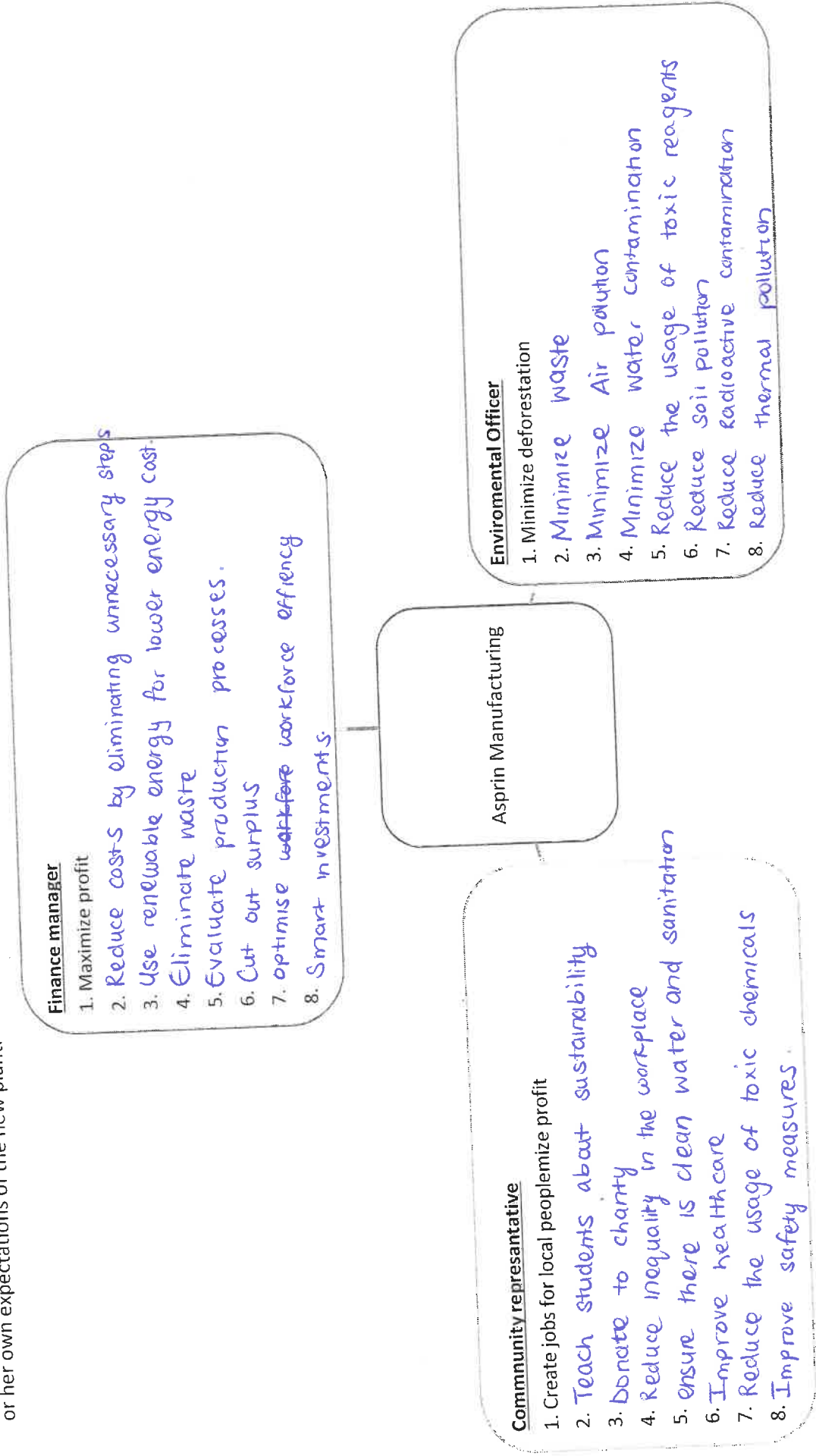
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Hint: You are welcome to refer to the **17 sustainable development goals**.

Asprin Manufacturing  
Plant for sustainability

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. The community will spread awareness about gender ~~inequality~~
4. Companies will ~~start~~ invest in sustainable & systems.
5. Companies will be forced to use greener energy
6. Use safer processes to ensure the health of workers.
7. Open more recycling companies around local communities
8. Spread awareness of drug addiction
9. Donate to charity
10. Donate money to help clean the oceans.

Are all group members happy with the new plant design?

~~Yes~~

No

**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

*Refer to the aspirin manufacture information document where necessary.*

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

**Refer to the Table 1 DATA SHEET on page 7** in the aspirin manufacture document.

**From Table 1 on page 7:**

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid from phenol**  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453,10 ↑		
Carbon dioxide	677,35 ↑		233,45 ↑
Sulfuric acid	571,55 ↑		14,95 ↑
activated carbon	211,60 ↑		211,60 ↑
water			204,70 ↑
sodium sulfate			805,00 ↑
salicylic acid		1150	242,65 ↑

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.17		
Oil of wintergreen	1488.61		
water			178.63
salicylic acid		1150	
methanol			266.66
Sodium sulfate			5636.85

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28
acetic acid	1001.87		
methyl acetate		616.67	32.06684

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2** and calculate:

- a. in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]  
 b. in row 3 the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is **1359 kg**.  
 For route 3 theoretical yield of salicylic acid is **1191 kg**.  
 c. in row 4 the E-factor (total waste) (in kg) [2]  
 d. Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$= \frac{1150}{913.6} \times 100$ $= 60,10\%$	$= \frac{1150}{5021,57} \times 100$ $= 22,90\%$ (1)
<b>% Yield</b> How efficient the chemical reactions are.	$= \frac{1150}{1359} \times 100$ $84,62\%$	$= \frac{1150}{1191} \times 100$ $96,56\%$ (3)
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	1829,65 kg	6082,14 kg (2)
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1 because less waste was produced.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Cradle to <del>gate</del> Grave.	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes as the material extraction processes results in less sustainable extraction	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	Cost of aspirin would inflate because the resources required would be less available	

**Question 3**

Answer the following questions in the space provided:

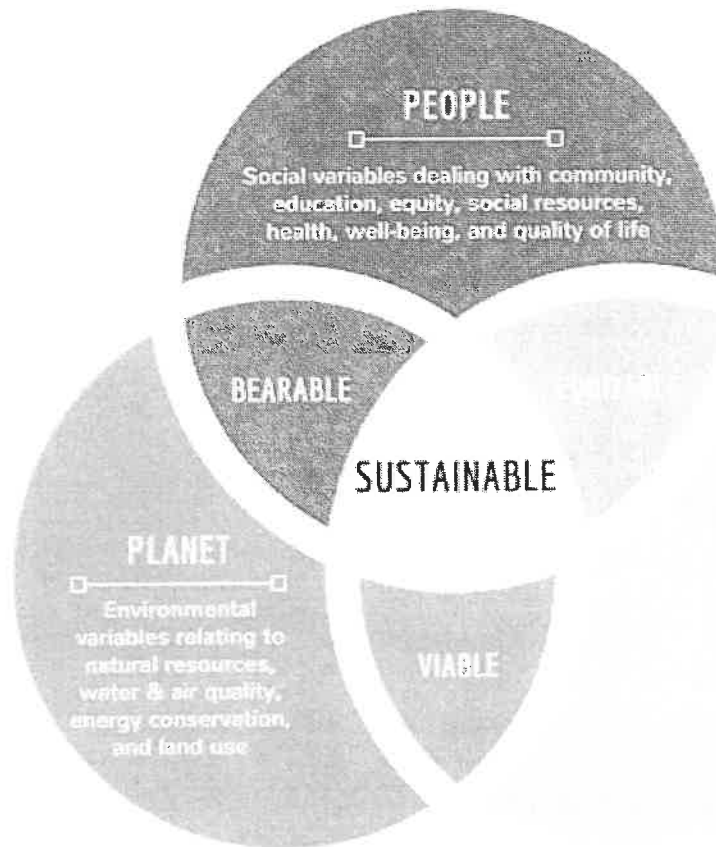
<p>How can the impact (<b>environmental, social or economic</b>) of the inputs and output be minimized or managed to make the process sustainable? [2]</p>	<p><b>Environmental</b>: The process whereby we obtain crude oil utilises extensive amounts of energy which is terrible for waste management. Using trees as an energy source will result in rapid deforestation which would be harmful to the environment.</p>
<p>How can the impact (<b>environmental, social or economic</b>) of energy be minimized or managed to make the process sustainable? [2]</p>	<p><b>Environmental</b>: We should prioritise the use of renewable resources as this would drastically decrease the energy required to obtain the non-renewable resources.</p>
<p>Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]</p> <p>Give 2 examples?</p>	<ul style="list-style-type: none"> <li>- The cost of machinery and its maintenance.</li> <li>- The cost of labour to oversee the process and ensure that the machinery is maintained.</li> </ul>
<p>How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]</p>	<p>The synthesis of aspirin is not dependant on a single route of manufacture, as there are multiple different routes to obtain the resources needed to synthesis the aspirin. This allows for the recuperation of materials, natural resources expenditure.</p>

**SYSTEMS THINKING IN SUSTAINABILITY**

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



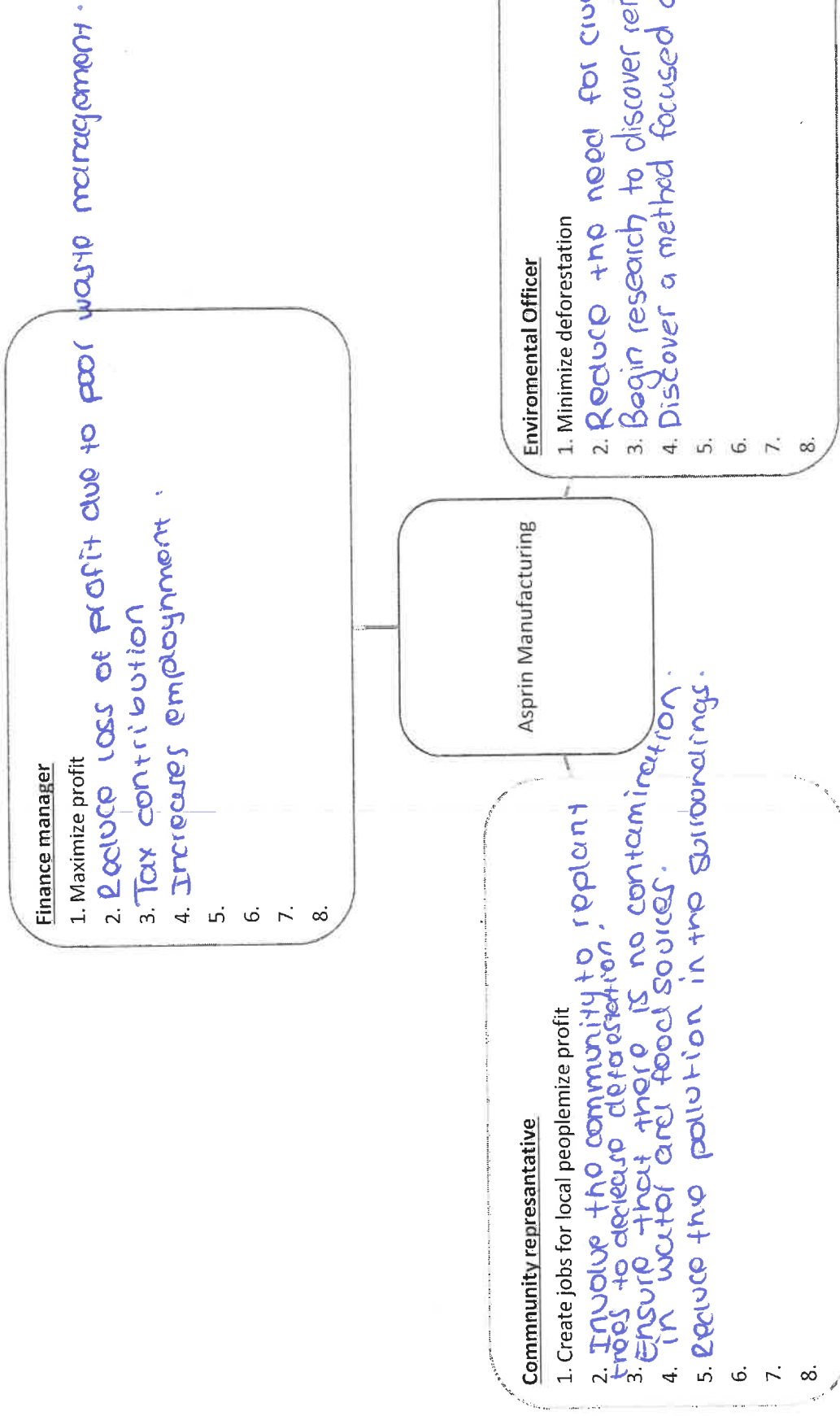
**Sustainability strives for:**

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

**ROLE PLAY ACTIVITY**

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.



- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. The new plants should result in an increase in revenue.
4. The new plants should be ~~part~~ on additive towards tax contribution
5. The new plants should create jobs for people.
6. The new plants should maximize profit.
- 7.
- 8.
- 9.
- 10.

Asprin Manufacturing Plant for sustainability

Are all group members happy with the new plant design?

<input checked="" type="radio"/> Yes	<input type="radio"/> No
--------------------------------------	--------------------------

## EXPERIMENTAL DATA

**FUNCTIONAL UNIT:** This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

### Question 1

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol.

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

[5]

④

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453.10		<del>233.45</del>
Carbon dioxide	677.35		233.45
Sulfuric acid	571.65		14.95
activated carbon	211.60		211.60
water			205.7
sodium sulfate			<del>805.805</del>
salicylic acid		1150	243.

2768.05

51

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	11945.11		
Oil of wintergreen	1488.61		
water			178.63
salicylic acid		1150	
methanol			266.66
Sodium sulfate			5636.85

- c. Complete Table 3 below by calculating the exact mass (in kg to two-decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28
acetic acid	1001.87		
methyl acetate		616.67	32.07

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1. Refer to the **calculated mass in Table 1 and 2** and calculate:

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is 1359 kg. A  
 For route 3 theoretical yield of salicylic acid is 1191 kg. ↑
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$\frac{1150}{1066.05 + 453.1 + 677.35 + 571.55} \times 100 = 41.55\%$	$\frac{1150}{1587.25 + 1945.11 + 1488.61} \times 100 = 22.90\%$
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100 = 84.62\%$	$\frac{1150}{1191} \times 100 = 96.5\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	$17.30 + 233.45 + 1495 + 211.6 + 204.7 + 805 = 1830$	$178.63 + 266.66 + 5636.85 = 6082.14$
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1, it has a higher efficiency and less waste	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	NO, route 1 is the more sustainable route because it produces less waste and higher reaction mass efficiency	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	The cost of aspirin would increase as the crude oil availability would decrease in the production of aspirin	

### Question 3

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable?

[2]

~~Environmental~~! replace toxic reagents and products with environmentally friendly products  
Reduce the usage or generation of toxic compounds substances

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable?

[2]

~~Economic~~! reduce the energy required to produce the desired product either by the use of faster processes or by the use of renewable energies involving lower energy cost with equal efficiency

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur?

[2]

Give 2 examples?

~~Cost of labour in production~~  
~~cost of equipments maintenance~~

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues?

[2]

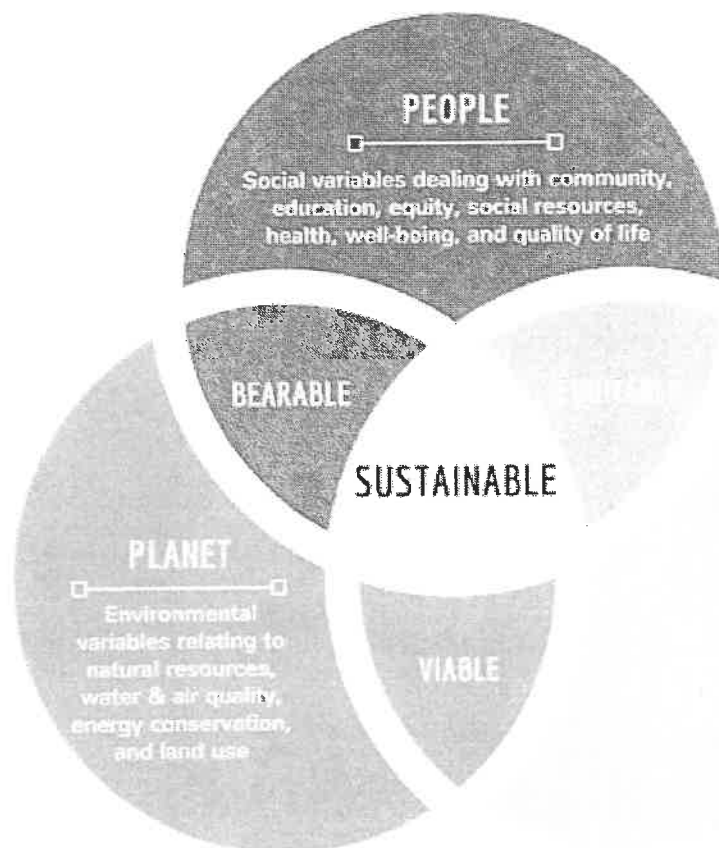
~~Replace toxic reagents and products with environmentally friendly~~  
~~Reduce waste generated in any preparation and handling of chemicals~~

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

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the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



#### Sustainability strives for:

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

#### ROLE PLAY ACTIVITY

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

Finance manager

1. Maximize profit
2. Reduce cost by eliminating unnecessary steps
3. High Tex contribution
4. Prices of chemicals and drugs to decrease
5. Meet all supply and demand
6. expectations
- 7.
- 8.

Asprin Manufacturing

Community representative

1. Create jobs for local people to profit
2. No contamination of any water sources
3. No contamination of any food sources
4. (crops)
5. Longevity of life increase
6. Increase quality of life
- 7.
- 8.

Environmental Officer

1. Minimize deforestation
2. Reduce the generation of toxic compound substance
3. Replace toxic products with environmentally friendly products
4. Minimize contamination of any water sources
5. No contamination of any water sources
6. Reduce the waste produced
- 7.
- 8.

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. High Tax contribution
4. Maximize profit
5. Increase employment
6. Increase quality of life
7. No contamination of any water resources
8. Reduce the waste produced
- 9.
- 10.

Asprin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?

Yes

No

## EXPERIMENTAL DATA

**FUNCTIONAL UNIT:** This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

### Question 1

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453.10 ✓		
Carbon dioxide	677.35 ✓		233.45 ✓
Sulfuric acid	571.55 ✓		14.95 ✓
activated carbon	211.60 ✓		211.60 ✓
water			204.70 ✓
sodium sulfate			805.00 ✓
salicylic acid		1150	242.65 ✓

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen 2.5  
(Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.11 ✓		
Oil of wintergreen	1488.61 ✓		
water			178.63 ✓
salicylic acid		1150	
methanol			266.66 ✓
Sodium sulfate			5636.85 ✓

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas  
(Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			149.28 ✓
acetic acid	1001.87 ✓		
methyl acetate		616.67	

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1.

Refer to the **calculated mass in Table 1 and 2** and calculate:

- a. in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2? [2]
- b. in row 3 the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is **1359 kg**.  
 For route 3 theoretical yield of salicylic acid is **1191 kg**.
- c. in row 4 the E-factor (total waste) (in kg) [2]
- d. Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$= \frac{1150 \text{ kg}}{2979 \text{ kg}} \times 100$ $= 38.60\%$	$= \frac{1150 \text{ kg}}{5021.57} \times 100$ $= 22.90\%$
<b>% Yield</b> How efficient the chemical reactions are.	$= \frac{1150 \text{ kg}}{1359 \text{ kg}} \times 100$ $= 84.62\%$	$= \frac{1150 \text{ kg}}{1191 \text{ kg}} \times 100$ $= 96.56\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	1829.65 kg	6082.14 kg
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. <span style="float: right;">[2]</span>	Route 1, because it produces less waste	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. <span style="float: right;">[1]</span>	It was not mentioned how do they get raw materials. Cradle to gate	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief <span style="float: right;">[2]</span>	No, the route 1 and cradle to gate are not harmful to the environment	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. <span style="float: right;">[2]</span>	* Raw material cost * Transport and logistics cost	

**Question 3**

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable? [2]

- \* Consider using renewable resources, eco-friendly materials and non-toxic substances which reduces the negative to environment. Reuse and recycling.
- \* Consider social aspects of the process including labor conditions, worker safety and community well-being.
- \* Collaborate with suppliers and encourage sustainable practices throughout the supply chain.

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable? [2]

- \* Investing in renewable energy such as hydro and solar power helps reduce greenhouse gas emissions.
- \* Provide insulators and efficient equipment which reduces waste energy.
- \* By shifting energy-intensive activities to off-peak.

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  
Give 2 examples?

- Labor costs ✓
- Energy expenses ✓
- Facility maintenance costs ✓

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]

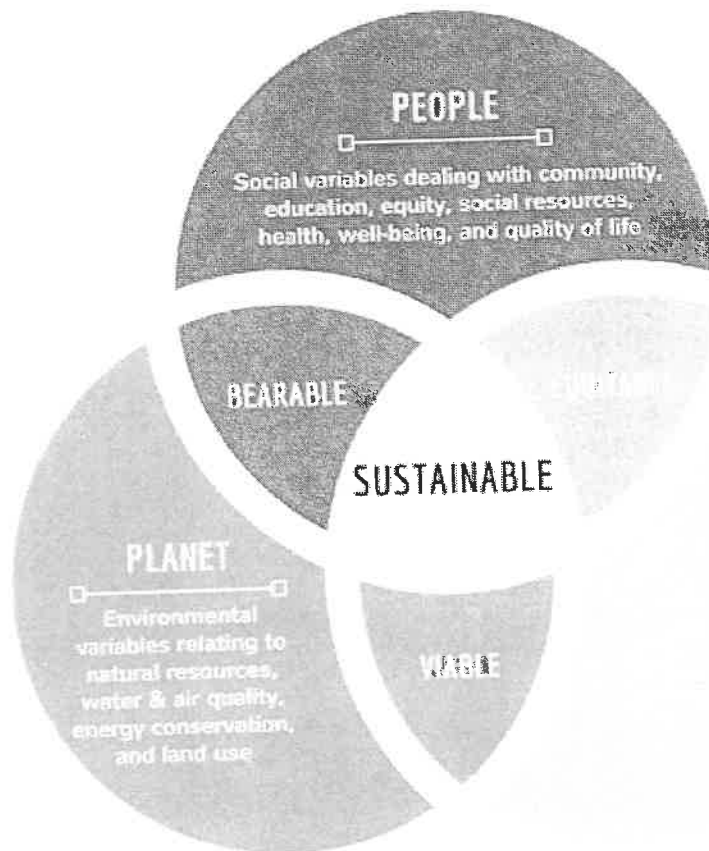
- > Green chemistry principles promote the development of environmentally friendly processes such as minimize the use of hazardous materials.
- > By carefully selecting reaction conditions, catalyst and purification methods can be used to minimize the production of unwanted byproducts or toxics.

**SYSTEMS THINKING IN SUSTAINABILITY**

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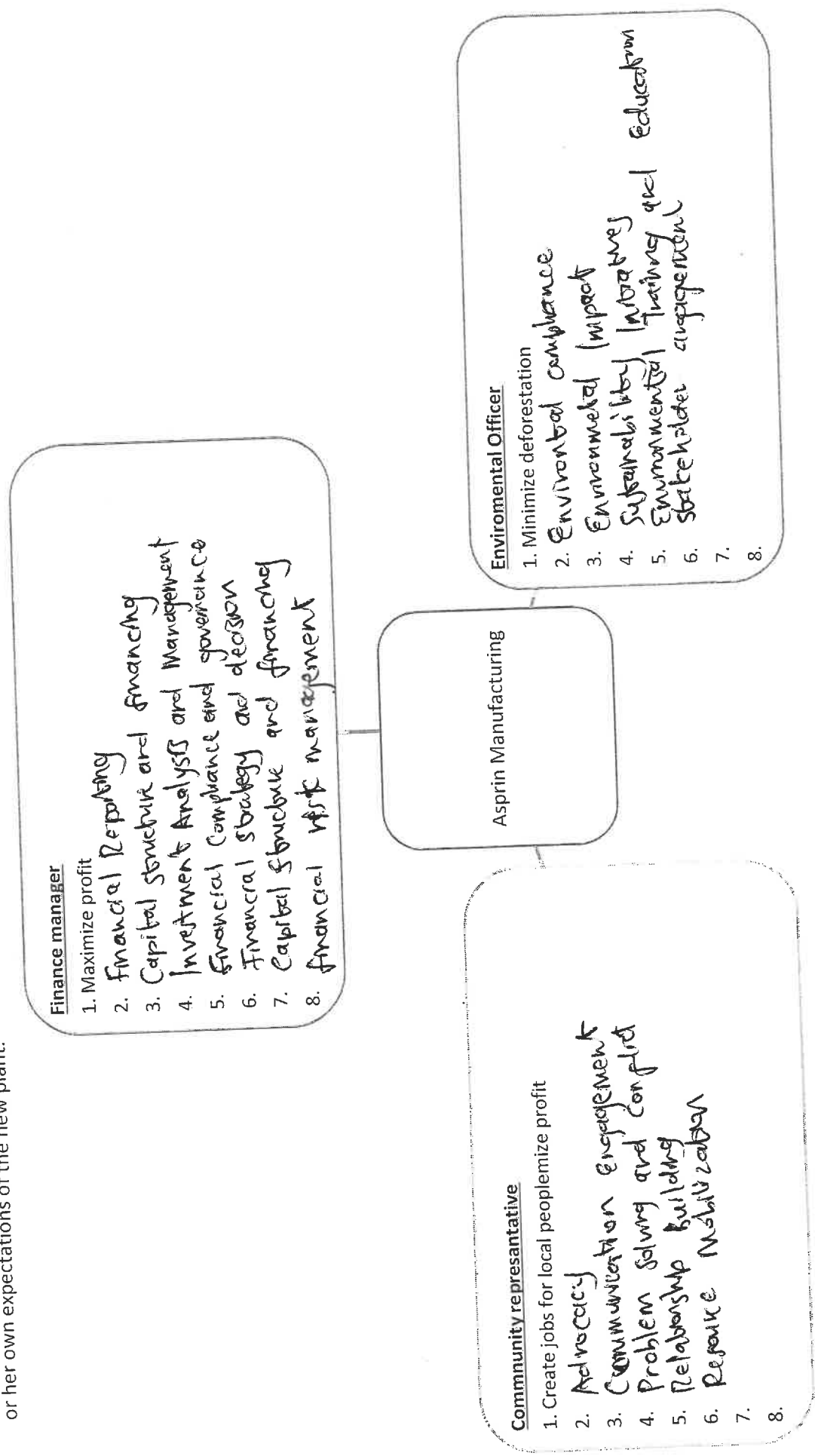
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#### ROLE PLAY ACTIVITY

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- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.



Finance manager

1. Maximize profit
2. Financial Reporting
3. Capital Structure and financing
4. Investment Analysis and Management
5. Financial Compliance and governance
6. Financial Strategy and decision
7. Capital Structure and financing
8. financial risk management

Community representative

1. Create jobs for local people/mize profit
2. Advocacy
3. Communication engagement
4. Problem Solving and conflict
5. Relationship Building
6. Resource Mobilization
- 7.
- 8.

Asprin Manufacturing

Environmental Officer

1. Minimize deforestation
2. Environmental Compliance
3. Environmental Impact
4. Sustainability Initiatives
5. Environmental Training and Education
6. Stakeholder engagement
- 7.
- 8.

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Green chemistry
4. Energy Efficiency
5. Water conservation
6. Waste Management
7. Sustainable Raw materials
8. Renewable energy sources
9. Continuous Improvement and Innovation
10. Environmental Monitoring and reporting

Asprin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?	Yes ✓	No
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**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. **Do not calculate the shaded blocks in the tables.**

- a. Complete Table 1 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize **1150 kg of salicylic acid** from phenol. [5]

Refer to the **Table 1 DATA SHEET** on page 7 in the aspirin manufacture document.

From Table 1 on page 7:

**927 kg phenol was put in to produce 1000 kg of Salicylic acid**

$$927 \text{ kg } C_6H_5OH = 1000 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = 1150 \text{ kg SA}$$

$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

4

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453,10 ✓		
Carbon dioxide	677,35 ✓		233,45 ✓
Sulfuric acid	571,55 ✓		14,95 ✓
activated carbon	211,60 ✓		211,60 ✓
water			204,70 ✓
sodium sulfate			536,67 ✓
salicylic acid		1150	161,77 ✓

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

2

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1944.65 ✓		
Oil of wintergreen	7488.61 ✓		172
water			178.63 ✓
salicylic acid		1150	
methanol			266.66 ✓
Sodium sulfate			5636.85 ✓

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

1.5

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68
water			1949.28 ✓
acetic acid	1001.87 ✓		
methyl acetate		616.67	32.07 ✓

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in Table 4 row 1.

Refer to the **calculated mass in Table 1 and 2** and calculate:

- in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 2. [2]
- in row 3 the % yield of salicylic acid for the two routes. [2]  
For route 1 theoretical yield of salicylic acid is 1359 kg.  
For route 3 theoretical yield of salicylic acid is 1191 kg.
- in row 4 the E-factor (total waste) (in kg) [2]
- Answer the four questions that follow. [7]

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	41,55% <	17,40% <
<b>% Yield</b> How efficient the chemical reactions are.	$\frac{1150}{1359} \times 100 = 84,62\%$	$\frac{1150}{1191} \times 100 = 96,56\%$
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	$233,47 + 14,95 + 21,60 = 269,02$ $204,70 + 536,67 = 741,37$ <del>= 1465,91 kg</del>	$178,3 + 266,60 + 5836,85 = 6681,75$ = 6682,114 kg
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1, there is a high chance to quantify the efficiency and environmental performance of chemical processes.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Cradle to grave disposal & extraction of raw material was not mentioned.	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes, the extraction process itself may result in more environmental damage than good, defeating the whole aim.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	- Crude oil will become hard to obtain, hence the supply will be lower, driving the price up. Crude oil becomes more expensive, resulting in an increase in the price of producing aspirin, and ultimately a higher price to buy for the consumers.	

### Question 3

Answer the following questions in the space provided:

How can the impact (**environmental, social or economic**) of the inputs and output be minimized or managed to make the process sustainable? [2]

- We could improve the handling of chemicals and ~~how?~~  
 - We could reduce waste by recycling and reusing  
 - Advanced technology could be used.

How can the impact (**environmental, social or economic**) of energy be minimized or managed to make the process sustainable? [2]

- We could make use of solar energy; it's a cleaner form of energy and much more cheaper/affordable in the long run.

Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]  
 Give 2 examples?

~~Unnecessary steps in the process such as such as~~  
 - Labour (salaries)  
 - Cost of waste disposal.

How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]

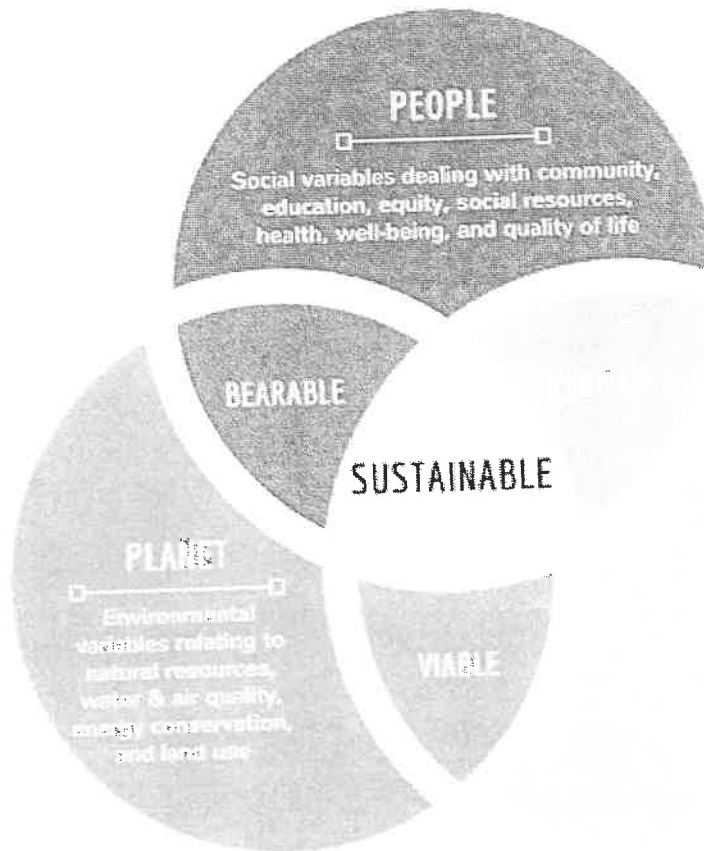
- Chemistry can be used to develop new water purification processes and treatment. This reduces pollution and ~~protects~~ protects environment.  
 - Chemistry can be used to develop new energy sources such as solar cells and batteries. These energy sources help reduce reliance on fossil fuels.

### SYSTEMS THINKING IN SUSTAINABILITY

Sustainability is all about systems and making sure we are thinking about the entire picture so we can address a problem from all angles. It is important that we start seeing the Earth as a system. Within this system, there are other systems such as People, Planet and Profit. All three systems are tightly interconnected and initiatives to address one often overlap with the other. Just like, there would not be people without a planet, there would not be a business without people, and there would not be prosperity without business.

Sustainable systems thinking is a mindset that helps us to untangle and work within the complexity of life on Earth. It helps us to better understand the impact of our work on the world around us and be responsible for what we create and how we innovate. An important step in developing a systemic approach is to find

the source of the problems, as these increases the chances of identifying the leverage. Leverage is a small, focused action, that can produce significant changes with a minimum effort. Leverage can also be achieved by weakening the symptomatic response to the problem while strengthening the fundamental solution. Leverages results in viable, bearable and equitable options during problem solving. The aspirin manufacture forms a part of the aspirin system, which includes engineering, the planet, people, human health, and economics.



**Sustainability strives for:**

- **Viable** environmental-economic impact: Business is executed with the environment and resources in mind.
- **Bearable** socio-environmental impact: Develop healthy habits that benefit the society and the environment.
- **Equitable** socio-economic impact: The generation of economic opportunity for both businesses and individuals.

**ROLE PLAY ACTIVITY**

- An Engineer in partnership with a chemist decides that there is a need to upgrade the aspirin manufacturing plant for sustainability. They takes up the idea to the board, which includes the **Finance manager, the Environmental officer and the Community representative.**

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

Finance manager

1. Maximize profit
2. Reduce Costs by cutting unnecessary expenditure.
3. Obtain funding for project
4. Manage financial risks
5. Conduct financial analysis
6. Provide financial report to management
7. Develop a financial plan for projects
8. Cater to unexpected costs

Aspirin Manufacturing

Community representative

1. Create jobs for local people to maximize profit
2. Improve quality of life of individuals in community
3. Update the community about the plant
4. Monitor the plant to make sure it doesn't negatively affect community
5. Work with company to identify needs of community
6. Get involved early in the planning process
7. Be persistent

Environmental Officer

1. Minimize deforestation.
2. Minimize air pollution by using clean energy
3. Conduct environmental impact statements
4. Developing environmental sustainability
5. Promoting environmental regulations
6. Comply with environmental regulations
7. Recycle waste and byproducts
8. Dispose of toxic waste properly

- Lastly, all group members should discuss amongst themselves and write down their ideal sustainable plant. There will be leverages, compromises, adjustments, and amendments as **viable, bearable and equitable spheres of sustainability** should be taken into consideration when designing the new sustainable plant.

Hint: You are welcome to refer to the **17 sustainable development goals**.

Financial manager, Environmental officer and the community representative

1. The new plant will have solar panels
2. The community will assist with reforestation.
3. Supply people in community with jobs
4. Development of infrastructure in community
5. Improve standard of living.
6. Economic development
7. Plant made from recycled material
8. Use means of production that lead to less carbon emission.
9. ~~Use~~ waste to have designated waste disposal site.
10. Money generated from plant used to provide bursaries.

Asprin Manufacturing  
Plant for sustainability

Are all group members happy with the new plant design?	<input checked="" type="checkbox"/> Yes	<input type="checkbox"/> No
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**EXPERIMENTAL DATA**

*FUNCTIONAL UNIT:* This LCI is for the manufacture of **1500 kg** of Aspirin.

In order to produce **1500 kg of Aspirin**, we have to start with **1150 kg salicylic acid** and **616.67 kg** of methyl acetate.

Salicylic acid can be produced using 3 routes.

LCI will be used to find the greener route of producing **1150 kg of salicylic acid**.

Refer to the aspirin manufacture information document where necessary.

**Question 1**

Complete Table 1-3 below by calculating the mass of the components needed to synthesize the product per table. Do not calculate the shaded blocks in the tables.

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$$x \text{ kg } C_6H_5OH = \frac{927 \text{ kg } C_6H_5OH \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1066.05 \text{ kg}$$

Table 1: Material balance for the production of **1150 kg of salicylic acid** from phenol  
(Refer to Table 1 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
phenol	1066.05		117.30
Sodium hydroxide	453.10 ✓		2
Carbon dioxide	677.35 ✓		233.45 ✓
Sulfuric acid	591.55 ✓		14.95 ✓
activated carbon	211.60 ✓		211.60 ✓
water			204.70 ✓
sodium sulfate		1	805.00 ✓
salicylic acid		1150	242.65 ✓

- b. Complete Table 2 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 1150 kg of salicylic acid from oil of wintergreen. [2.5]

Refer to the Table 2 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 2 on page 7:

1380.74 kg sodium hydroxide was put in to produce 1000 kg of Salicylic acid

$$1380.74 \text{ kg NaOH} = 1000 \text{ kg SA}$$

$$x \text{ kg NaOH} = 1150 \text{ kg SA}$$

$$x \text{ kg NaOH} = \frac{1380.74 \text{ kg NaOH} \times 1150 \text{ kg SA}}{1000 \text{ kg SA}} = 1587.85 \text{ kg}$$

Table 2: Material balance for the production of 1150 kg of salicylic acid from oil of wintergreen (Refer to Table 2 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
Sodium hydroxide	1587.85		
Sulfuric acid	1945.11 ✓		
Oil of wintergreen	1488.61 ✓		
water			178.63 ✓
salicylic acid		1150	
methanol			266.66 ✓
Sodium sulfate			9636.85 ✓

- c. Complete Table 3 below by calculating the exact mass (in kg to two decimal places) of the components needed to synthesize 616.67 kg of methyl acetate. [1.5]

Refer to the Table 3 DATA SHEET on page 7 in the aspirin manufacture document.

From Table 3 on page 7:

866.48 kg methanol was put in to produce 1000 kg of methyl acetate

$$866.48 \text{ kg CH}_3\text{OH} = 1000 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = 616.67 \text{ kg MA}$$

$$x \text{ kg CH}_3\text{OH} = \frac{866.48 \text{ kg methanol} \times 616.67 \text{ kg MA}}{1000 \text{ kg MA}} = 534.33 \text{ kg}$$

Table 3: Material balance for the production of 616.67 kg of methyl acetate from natural gas (Refer to Table 3 page 7)

COMPONENT	INPUT (kg)	PRODUCT (kg)	WASTE (kg)
methanol	534.33		267.68 ✓
water			149.28 ✓
acetic acid	1001.87 ✓		
methyl acetate		616.67	32.07 ✓

## Question 2

The % atom economy (calculated in the pre-practical) for route 1 and route 3 is shown in **Table 4 row 1**. Refer to the **calculated mass in Table 1 and 2 and calculate:**

- a. in row 2 the % reaction mass efficiency of salicylic acid for the route 1 and route 3 [2]  
 b. in row 3 the % yield of salicylic acid for the two routes. [2]  
 For route 1 theoretical yield of salicylic acid is **1359 kg**.  
 For route 3 theoretical yield of salicylic acid is **1191 kg**.  
 c. in row 4 the E-factor (total waste) (in kg) [2]  
 d. Answer the four questions that follow. [7]

12

Green matrix and meaning	Route 1	Route 3
<b>% Atom economy</b> How efficient the process is in terms of atoms	65.22 %	54.88 %
<b>% Reaction mass efficiency</b> How efficient the process is in terms of amount of inputs to product	$= \frac{1150}{2979.65} \times 100$ <del>38.60%</del>	$= \frac{1150}{5021.75} \times 100$ 22.90%
<b>% Yield</b> How efficient the chemical reactions are.	$= \frac{1150}{1359} \times 100$ 84.62%	$= \frac{1150}{1191} \times 100$ 96.56%
<b>E-factor (Total waste) (in kg)</b> How efficient the chemical reactions in terms of waste produced.	$117.60 + 233.45 + 14.95 + 211.60 + 204.70 + 800.00 + 242.65$ = 1829.65 kg	$178.69 + 266.66 + 966.85$ = 6082.14 kg
Consider the calculated green metrics and decide which SA synthetic route you would select for your company. Justify your choice. [2]	Route 1 as it produces less waste than route 3.	
What type of system boundary (Cradle to gate or Gate to grave or Cradle to grave or Gate to gate) is implemented in this inventory? Refer to Figure 3, Page 4 of the Aspirin Manufacture document. [1]	Gate to Gate.	
Would expansion of the boundaries to include the raw material extraction processes likely result in your change of choice of the more sustainable route? Explain in brief [2]	Yes, as trees (oil of wintergreen) serve as a renewable resource unlike crude oil which isn't.	
Phenol is obtained from crude oil. Crude oil is a non-renewable resource. What effect would depletion of crude oil reserves have on the cost of buying aspirin? Briefly explain. [2]	It would increase the cost of Aspirin as less demand would increase while supply decreases. Less people would get access.	

**Question 3**

6

Answer the following questions in the space provided:

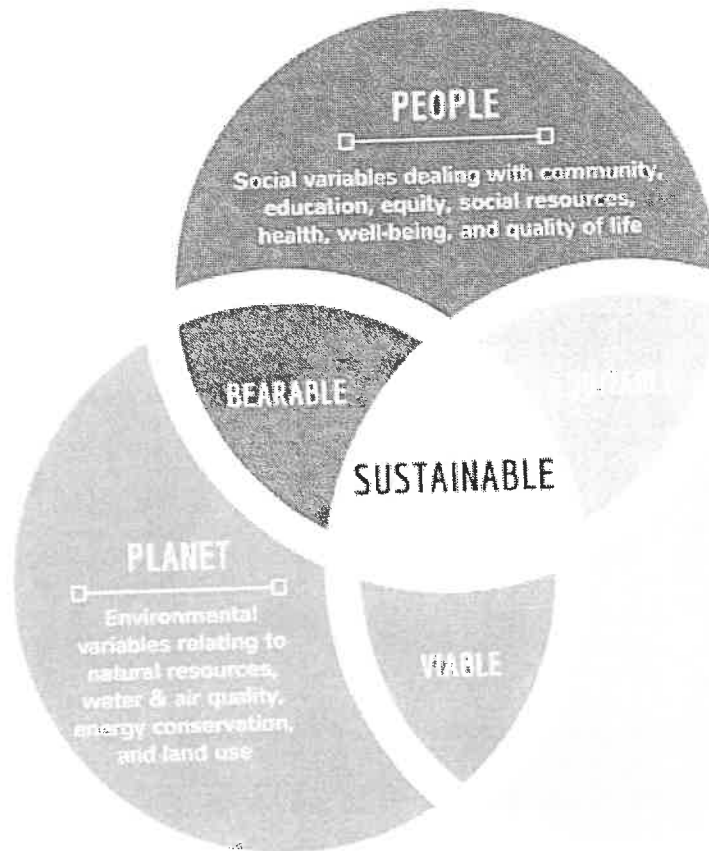
<p>How can the impact (<b>environmental, social or economic</b>) of the inputs and output be minimized or managed to make the process sustainable? [2]</p>	<p>Teaching workers the proper disposal <del>methods</del> and manufacturing methods in order to decrease the total wastage will reduce the impact of substance on environment. ✓          Being able to be best productivity by ensuring efficiency which can reduce costs.</p>
<p>How can the impact (<b>environmental, social or economic</b>) of energy be minimized or managed to make the process sustainable? [2]</p>	<p>The reduction of pollution and impact on environment can be attained by use more use of renewable resources and decrease usage of non-renewable resources. ✗</p>
<p>Besides the energy costs and the buying all the raw materials, what other cost implications do you anticipate the aspirin manufacturing process to incur? [2]          Give 2 examples?</p>	<ul style="list-style-type: none"> <li>* Lost of <del>out</del> catalyst substances such as <math>H_2SO_4</math>. ✓</li> <li>* Packaging of the product. ✓</li> <li>* Transportation of the product.</li> </ul>
<p>How does the exercise you just carried out on synthesis of aspirin display the central role of chemistry in the mitigation of global sustainability issues? [2]</p>	<p>* Chemistry can be one of the most impactful factor to sustainability issues if one is aware of the efficiency and sustainable management of production.</p>

**SYSTEMS THINKING IN SUSTAINABILITY**

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#### ROLE PLAY ACTIVITY

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Hint: You are welcome to refer to the **17 sustainable development goals**.

Asprin Manufacturing  
Plant for sustainability

- Financial manager, Environmental officer and the community representative
1. The new plant will have solar panels
  2. The community will assist with reforestation.
  3. Recycling of waste to useful products
  4. More job opportunities will be created for local community
  5. Provide food hamper and clothing for disadvantaged local community residents
  6. Treat waste ~~to protect~~ and correct waste disposal to protect nature, aquatic life etc
  7. Major contributor to the Country's Gross Domestic Product
  8. Partnering with big ~~int~~ world ~~level~~ trade partners
  9. Building a health facility for workers and local community residents
  10. Partnering with water sanitation companies

Are all group members happy with the new plant design?

Yes	No
-----	----

- Carry out a role play in which each group member assumes the stated role in the meeting. In the space provided, each member should write down his or her own expectations of the new plant.

**Finance manager**

1. Maximize profit
2. Financially stable for the next decade
3. Cost efficiency
4. Capital expenditure control
5. Budget control
6. Reduce financial risk takes.
7. efficient workers better than more workers (and is cost effective).
8. Have high net-worth value

**Community representative**

1. Create jobs for local people/profit
2. Attract tourists to tour the new plant.
3. Prioritising fair working conditions for local people
4. Offer learning or educational programmes.
5. building a health-care facility for locals and workers
6. Charitable work, through providing food hampers
7. Don't poison the water.
8. Offer bursaries (from profits) for local students

Aspirin Manufacturing

**Environmental Officer**

1. Minimize deforestation
2. Minimize air pollution
3. Minimize water pollution
4. Use of waste for energy recycle waste to <sup>energy</sup> avoid energy
5. Minimize depletion of non-renewable resources.
6. reduce water consumption.
7. reduce chemical wastes.
8. reduce carbon footprint