

Supplementary Information

A 7.5 V window dual-functional planar micro-device based on biopolymer ionogel electrolyte for charge storage and neuromorphic computing

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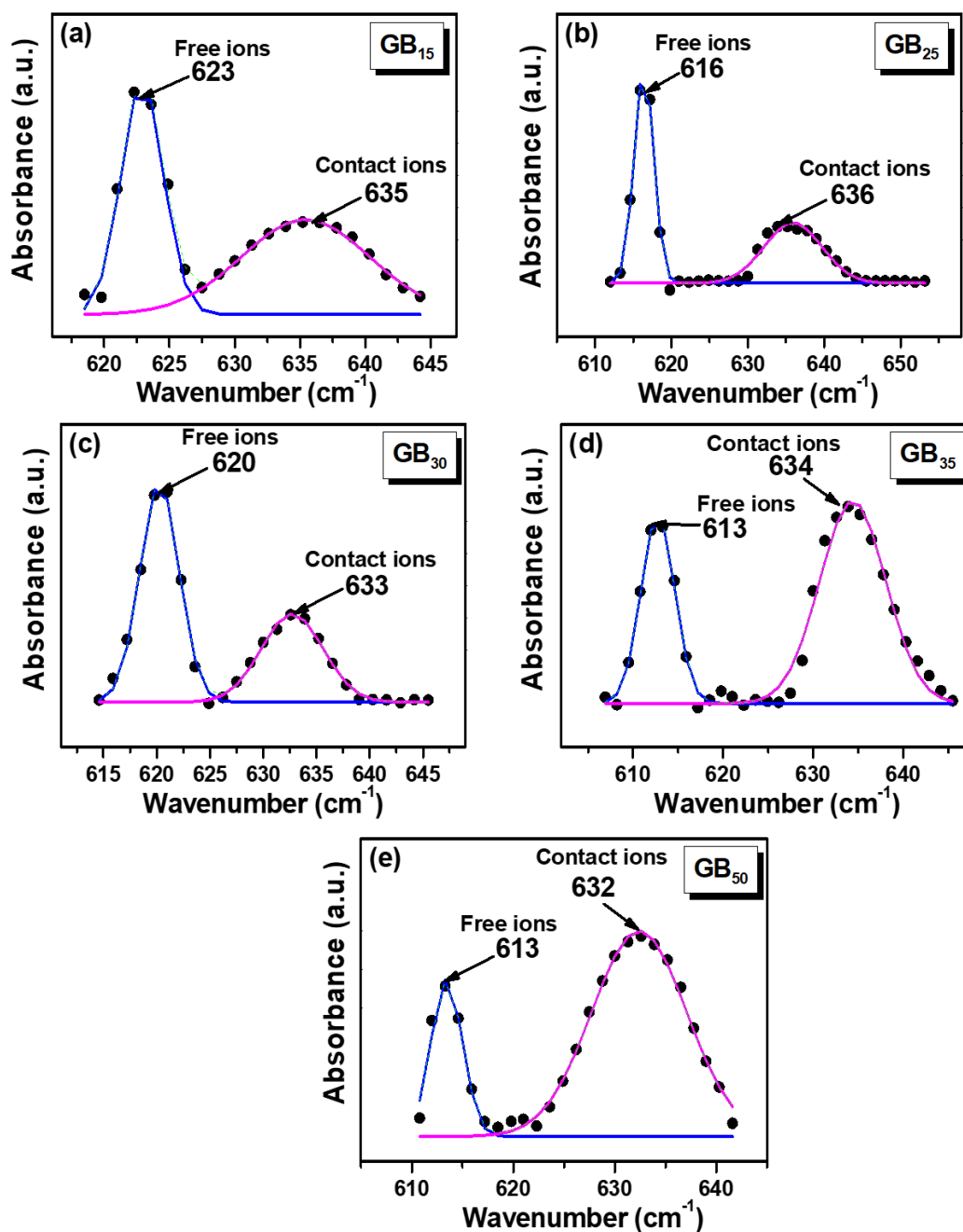


Figure S1. (a-e) FTIR absorption spectra in the region of 600 to 655 cm^{-1} for GB_x ($15 \leq x \leq 50$) blended biopolymer ionogel electrolytes with different BMIMBF_4 compositions. Deconvoluted FTIR absorption spectra based on Gaussian fitting function for the GB_x ionogels are represented by solid lines.

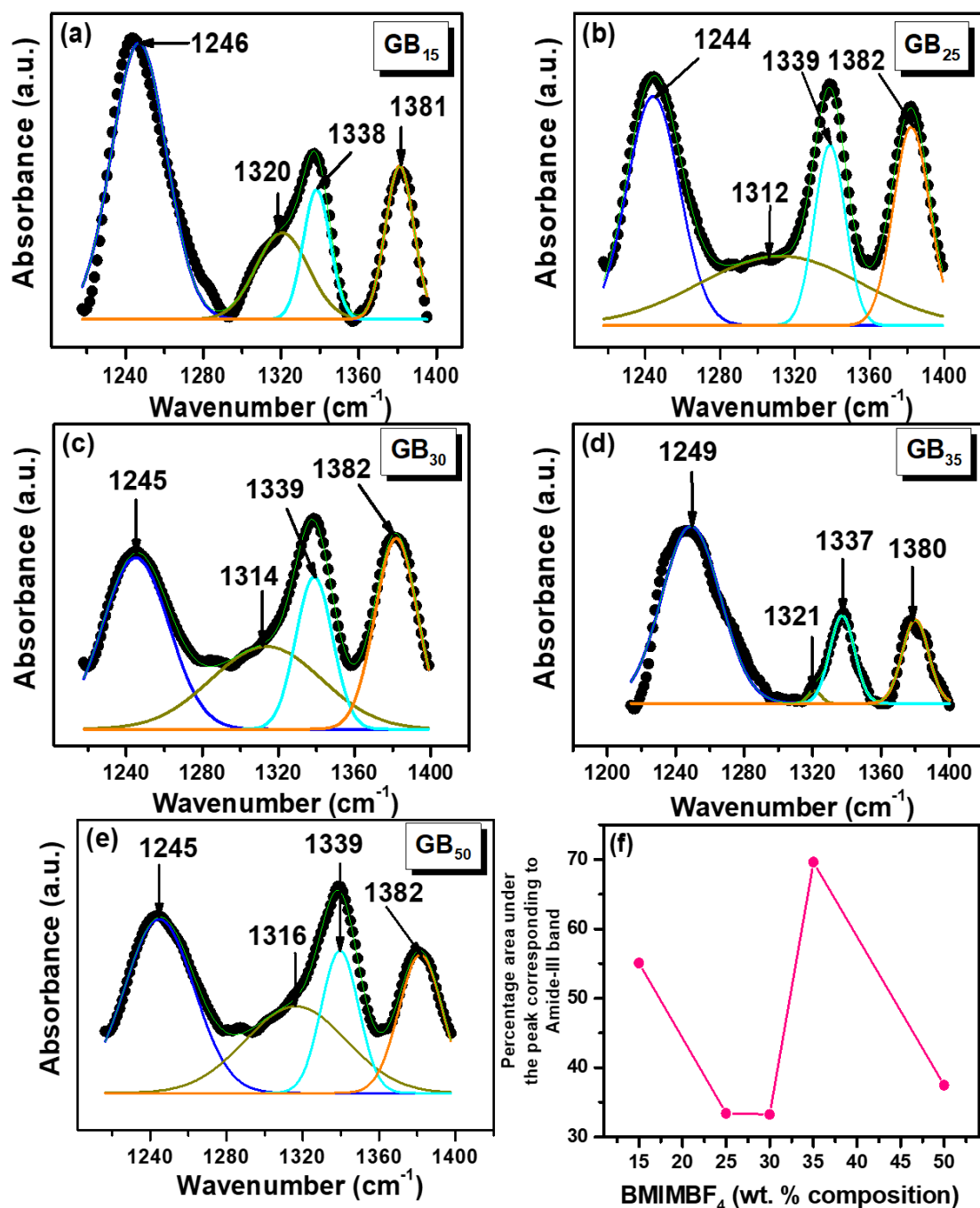


Figure S2. (a-e) FTIR absorption spectra in the region, 1200–1400 cm⁻¹ for GB_x (15 ≤ *x* ≤ 50) blended biopolymer ionogel electrolytes with different BMIMBF₄ compositions. Solid lines show the Gaussian fitting of the characteristic absorption peaks. (f) Percentage area under the peak corresponding to amide-III band is shown as a function of BMIMBF₄ doping concentrations in GB_x (15 ≤ *x* ≤ 50) ionogels.

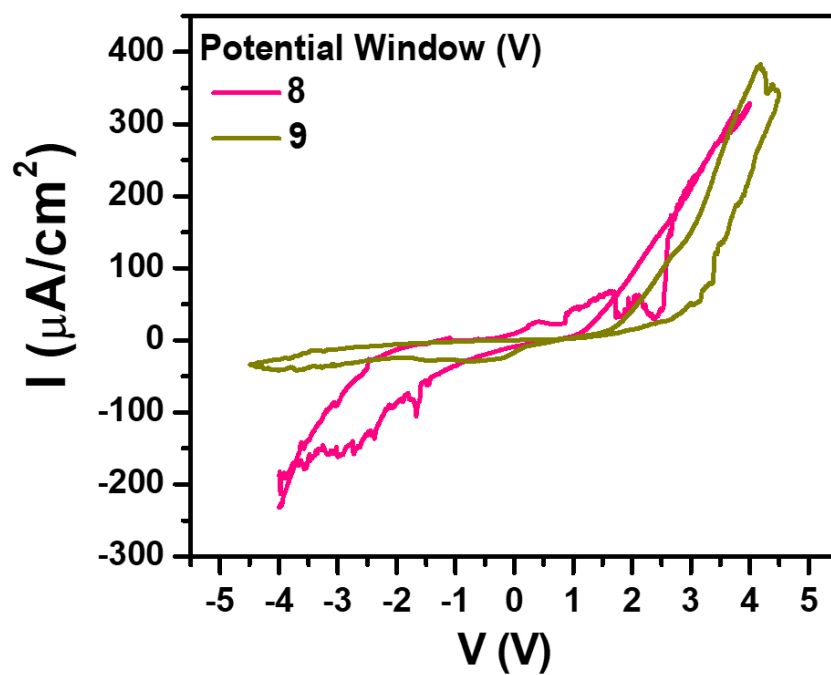


Figure S3. Cyclic voltammetry of the device measured at a scan rate of 0.1 V/s at voltage windows of 8 V and 9 V.

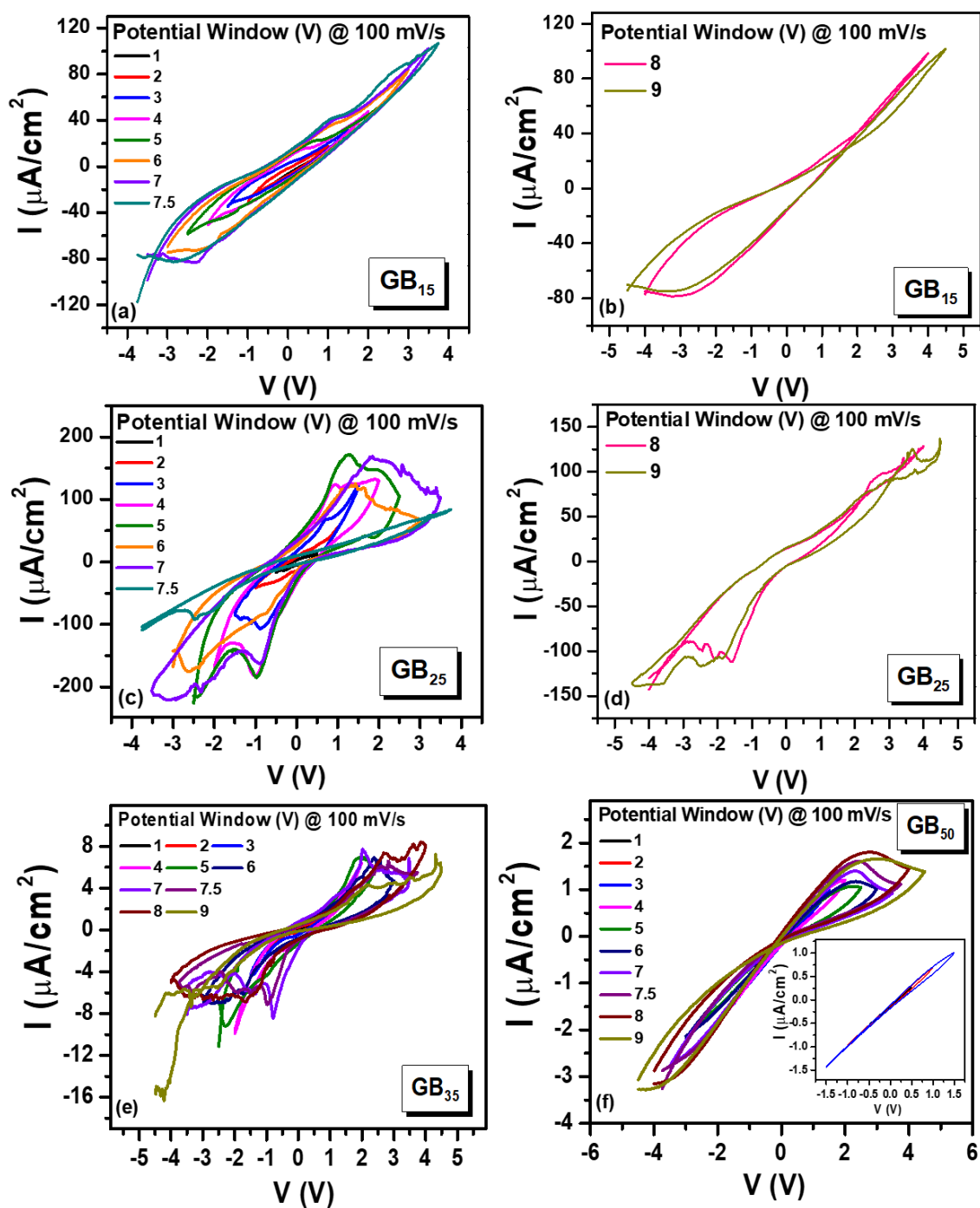


Figure S4. (a) Cyclic voltammetry of the micro-devices measured at a scan rate of 0.1 V/s at different voltage windows (1-9 V) developed with ionogels (a,b) GB₁₅, (c,d) GB₂₅, (e) GB₃₅, and (f) GB₅₀.

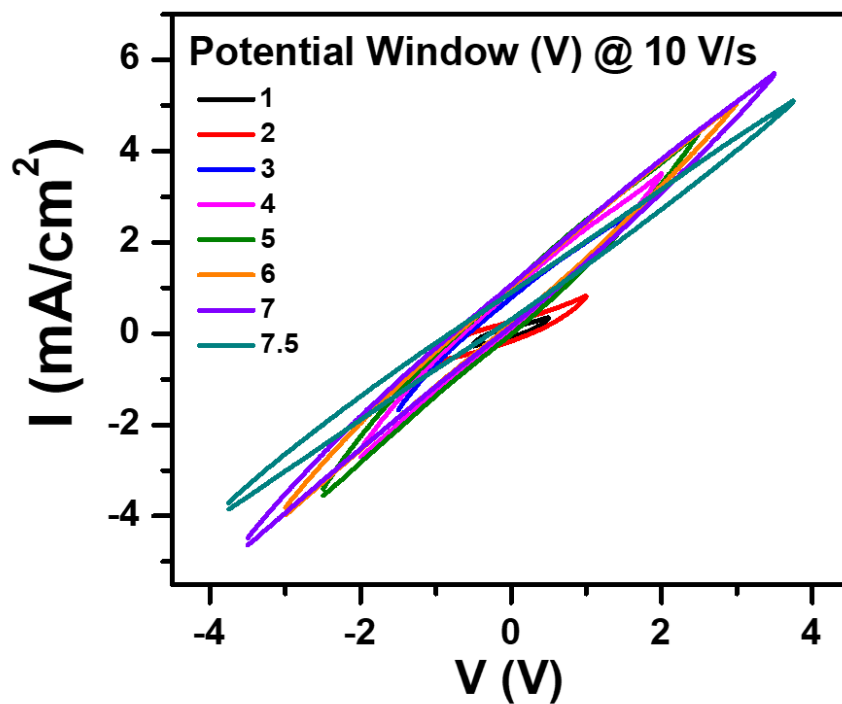


Figure S5. Cyclic voltammetry of the device measured at a scan rate of 10 V/s at different voltage windows (1-7.5 V).

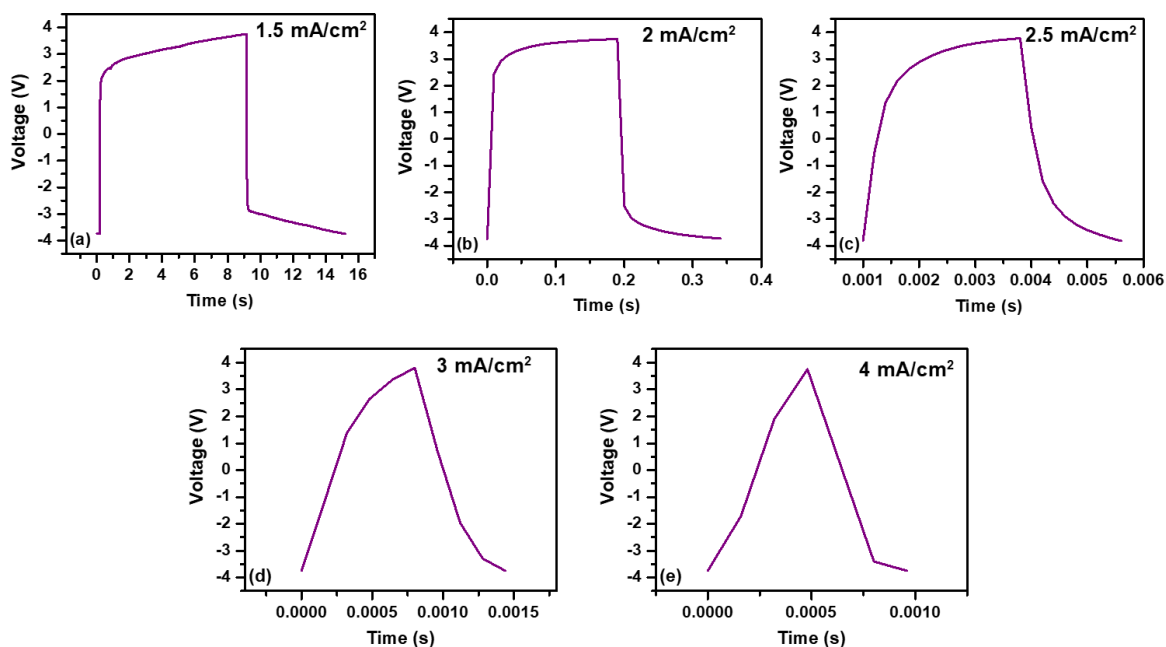


Figure S6. Galvanostatic charge-discharge for the device measured at a voltage window of 7.5 V under varying current densities, (a) 1.5, (b) 2, (c) 2.5, (d) 3, and (e) 4 mA/cm².

Table 1. Supercapacitive performance of CS/GL blended biopolymer ionogel electrolyte compared with other ionic liquid-based electrolyte materials.

Current Collector	Electrode Coating material	Ionic liquid-based electrolytes	ESW (V)	Capacitance	References
Aluminum foil	Polyacrylonitrile-derived porous carbon fibers (PPCF)	PVA-PAA Ionogel	2	615 mF/cm ²	1
Graphite paper	Activated Carbon	PAAm/ChCl/[EMIM][TFSI]IL	2.4	43.81 F/g	2
Au	Vertically oriented graphene ionogel (GI)	EMImBF ₄	4	149 μ F/cm ²	3
Silver ink	Printed Carbon	PVA-[DPTA][BF ₄]	1.2	98.8 mF/cm ²	4
Graphite paper	Activated Carbon	PAAM-PVA/BMIMBr IL	1.6	284.8 mF/cm ²	5
Au	Activated Carbon	PIP13FSI/ PYR14FSI/ SiO ₂	3	90 F/g	6
Aluminum foil	Carbon	AAM/[EMIM] ⁺ [NO ₃] ⁻ /PEGDA/	2	37.35 F/g	7
-	3D interconnected large mesoporous carbon	PVDF-HFP/[EMI][BF ₄]	4	323 F/g	8
None	Ionic liquid pre-intercalated MXene (PM)	EMIMBF ₄ /PVDF-HFP ionogel	3	44 mF/cm ²	9
Au	None	CS/GL/LiClO₄/BMI MBF₄ ionogel	7.5	5.78 F/cm²	Present work

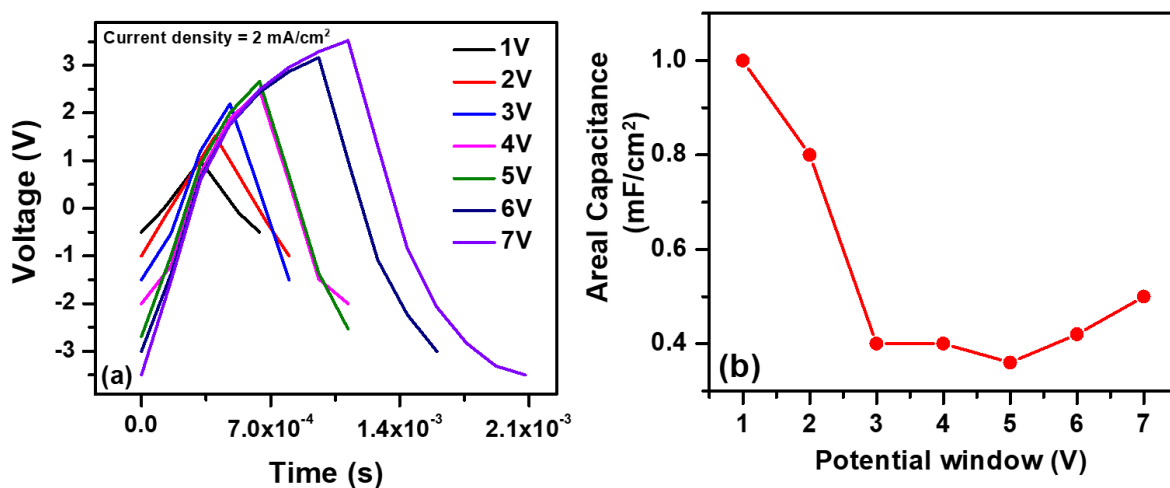


Figure S7. (a) Galvanostatic charge-discharge for the device measured at a fixed current density of 2 mA/cm^2 under varying potential windows (1-7 V), (b) Variation of areal capacitance with the potential window at a fixed current density of 2 mA/cm^2 .

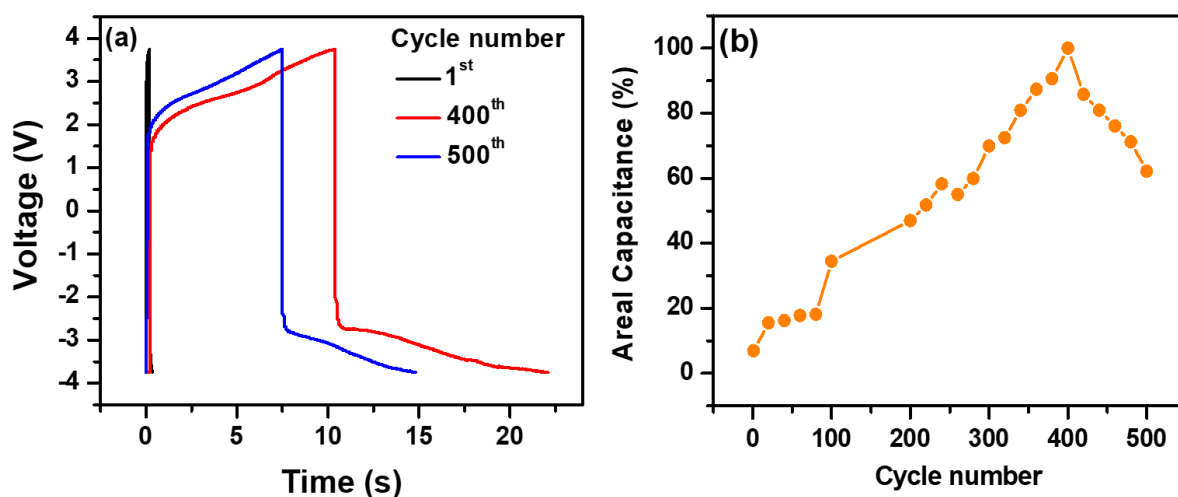


Figure S8. (a) Galvanostatic charge-discharge for the device measured at a fixed current density of 2 mA/cm^2 for 1st, 400th and 500th cycle, (b) Cycling stability for the device over 500 cycles.

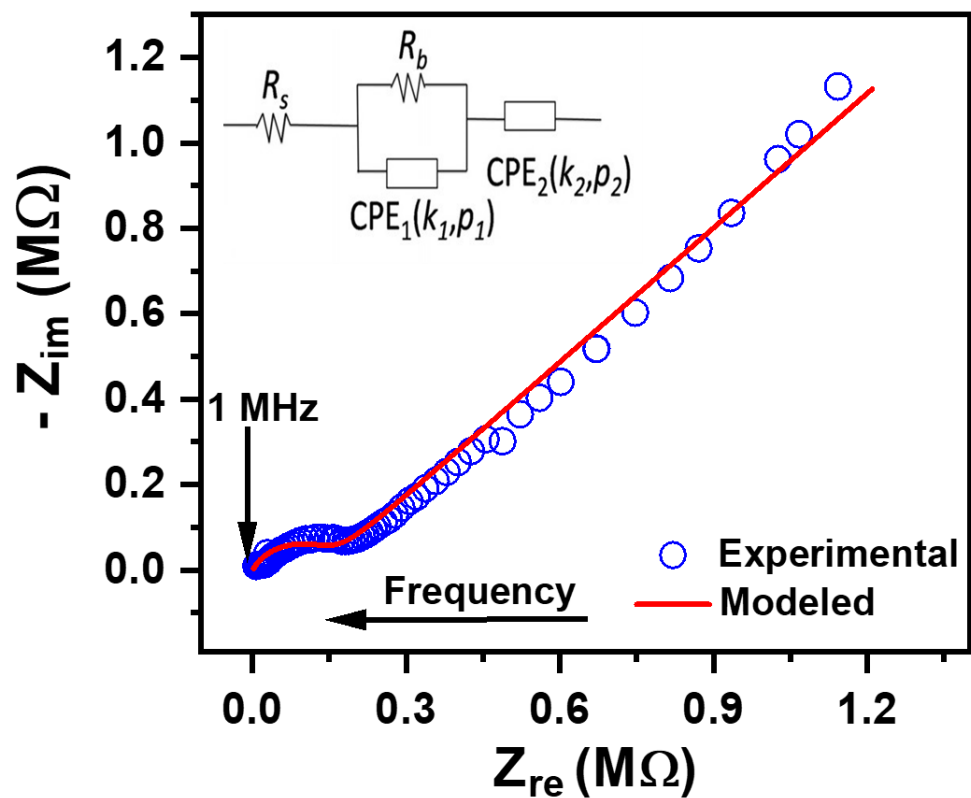


Figure S9. Nyquist plot of the device over a frequency range of 1MHz to 1 Hz, solid line represents the modelled plots based on the equivalent circuit fitting as shown in the inset.

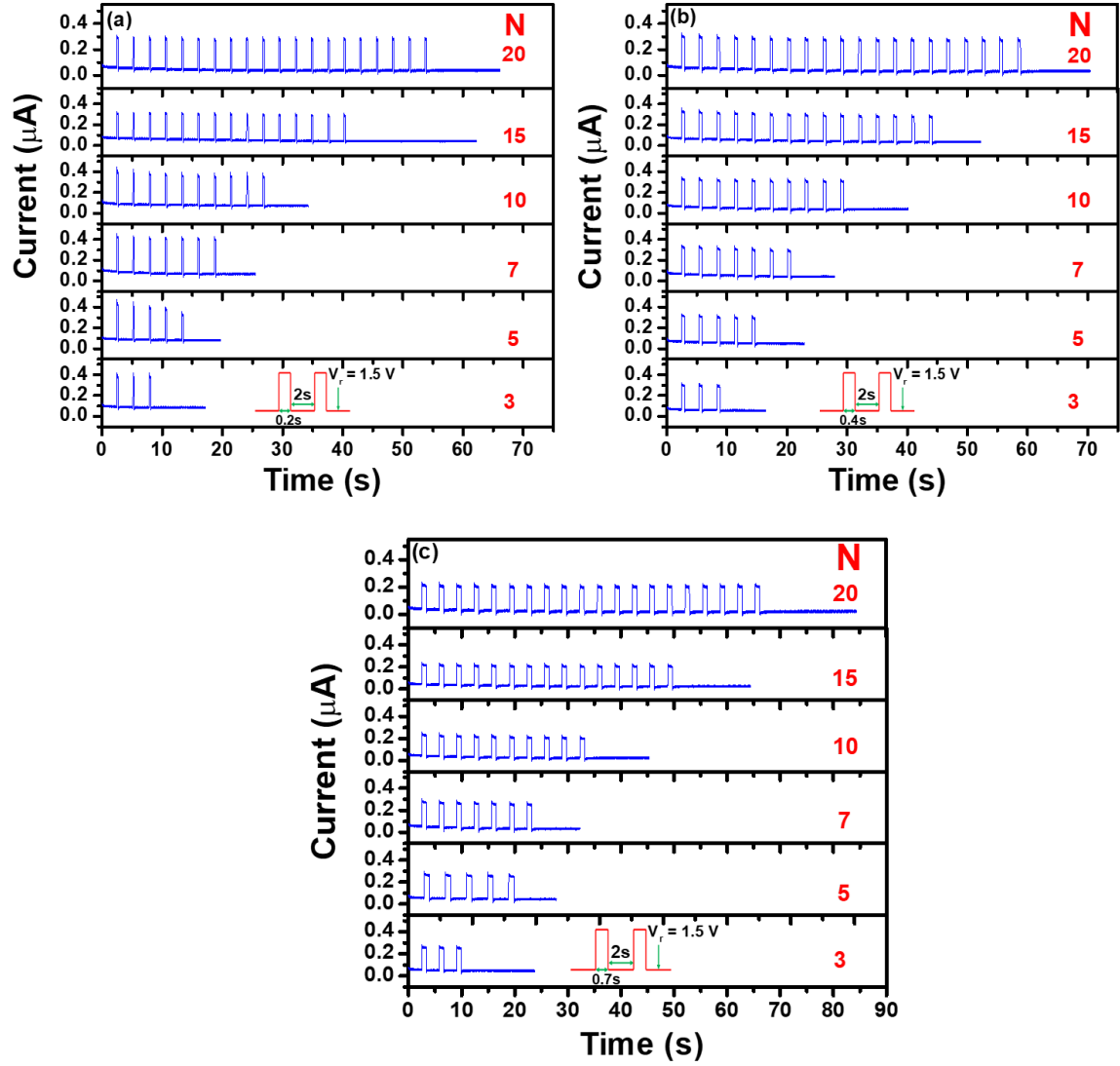


Figure S10. Synaptic behaviour emulated with pulsed voltage signals and fixed pulse interval ($t_i = 2$ s) with varying number of pulses ($N = 3, 5, 7, 10, 15, 20$) at a pulse width (t_w), (a) 0.2 s, (b) 0.4 s, and (c) 0.7 s.

References

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