## **Supplementary Information for:**

## Zwitterion Additive promotes (100)-textured Zinc

## Anodes for deep cycling Zinc Ion Batteries

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Figure S1. Comparison of adsorption energies of Zn atom,  $H_2O$  and PPS molecule on the Zn(002) plane, Zn(100) plane, and Zn(101) plane.



Figure S2. (a) Plots of capacitive currents versus scan rate for calculating the EDL capacitances of the Zn electrodes in different electrolytes and CV curves of Zn//Zn symmetric cells with (b) ZnSO<sub>4</sub> electrolytes and (c) ZnSO<sub>4</sub>+ PPS electrolytes.

The calculation of electric double layer capacitance was based on the equation C = i/v, where i was defined half of the difference between positive and negative scanning current at 0 V vs.  $Zn^{2+}/Zn$  of each scanning rate, C was the EDL capacitance and v was the scanning rate<sup>1</sup>.



Figure S3. Corresponding  $RTC_{(100)}/RTC_{(101)}$  ratios of cycled zinc anode in electrolytes with and without 5 mM PPS at various current density.

Table S1. Corrosion potential and corrosion current density of Zn anode in electrolyte with and without PPS.

| Electrode                              | $E_{\rm corr.}$ (V) | $J_{\rm corr.}$ (mA cm <sup>-2</sup> ) |
|--|---------------------|--|
| In ZnSO <sub>4</sub> electrolyte       | -0.978              | 2.012×10 <sup>-3</sup>                 |
| In ZnSO <sub>4</sub> + PPS electrolyte | -0.975              | 8.644×10 <sup>-4</sup>                 |

Table S2. The fitted  $R_{ct}$  and  $R_s$  of Zn//Zn symmetric cells.

| Electrolyte                          | $R_{ m ct}\left(\Omega ight)$ | $R_{ m s}\left(\Omega ight)$ |
|--------------------------------------|-------------------------------|------------------------------|
| Initial in ZnSO <sub>4</sub>         | 935.8                         | 2.23                         |
| Initial in ZnSO <sub>4</sub> + PPS   | 1034                          | 1.93                         |
| 50 cycles in ZnSO <sub>4</sub>       | 47.31                         | 2.06                         |
| 50 cycles in ZnSO <sub>4</sub> + PPS | 133                           | 1.34                         |

 Table S3. Comparison of electrochemical performances of this work with previously reported symmetric Zn-based cells.

| Stuate ging [Ref.]   | Cumunt dansity   | Analaanaity                            | Time          |
|--|------------------|--|---------------|
| Strategies   | $(mA \ am^{-2})$ | Areal capacity $(mAh \text{ am}^{-2})$ | 1 line<br>(b) |
|  | (IIIA CIII )     |  | (11)          |
| I his work   | 1                | 1                                      | 4100          |
|  | 2                | 2                                      | 4500          |
|  | 5<br>10          | 5                                      | 2300          |
| $G_{1}$ $1$ $1$ $1$ $M_{7}$ $G_{2}$ $2$                                      | 10               | 10                                     | <b>390</b>    |
| Sorbitol in 1 M $ZnSO_4^2$   | 1                | 1                                      | 1000          |
|  | 5                | 5                                      | 480           |
| $\alpha$ -Cyclodextrins in 3 M<br>ZnSO <sub>4</sub> <sup>3</sup>             | 5                | 5                                      | 200           |
|  | 10               | 1                                      | 160           |
| 2-Methylimidazole in 2 M $ZnSO_4^4$  | 1                | 1                                      | 1500          |
|  | 10               | 5                                      | 300           |
| Amphiphilic<br>dibenzenesulfonimide in 1<br>M ZnSO <sub>4</sub> <sup>5</sup> | 1                | 1                                      | 2700          |
|  | 2                | 2                                      | 1700          |
|  | 5                | 5                                      | 1180          |
| ammonium dihydrogen<br>phosphate in 1 M ZnSO <sub>4</sub> <sup>6</sup>       | 1                | 1                                      | 2100          |
| 1 1  | 5                | 5                                      | 300           |
| Glycine in 2 M ZnSO <sub>4</sub> <sup>7</sup>                                | 1                | 1                                      | 3100          |
|  | 10               | 10                                     | 300           |
| Cysteine in 2 M ZnSO <sub>4</sub> <sup>8</sup>                               | 5                | 5                                      | 620           |
| L-Cysteine in 2 M ZnSO <sub>4</sub> <sup>9</sup>                             | 2                | 2                                      | 1600          |
|  | 10               | 10                                     | 200           |
| propylene glycol in 1 M<br>ZnSO <sub>4</sub> <sup>10</sup>                   | 2                | 2                                      | 1000          |
| Betaine in 1 M ZnSO <sub>4</sub> <sup>11</sup>                               | 1                | 1                                      | 1800          |
|  | 2                | 2                                      | 830           |

## **Supplementary References**

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