

Achieving High-Performance OER Catalysis with Dual-Site Modulated Fe-Based Perovskites

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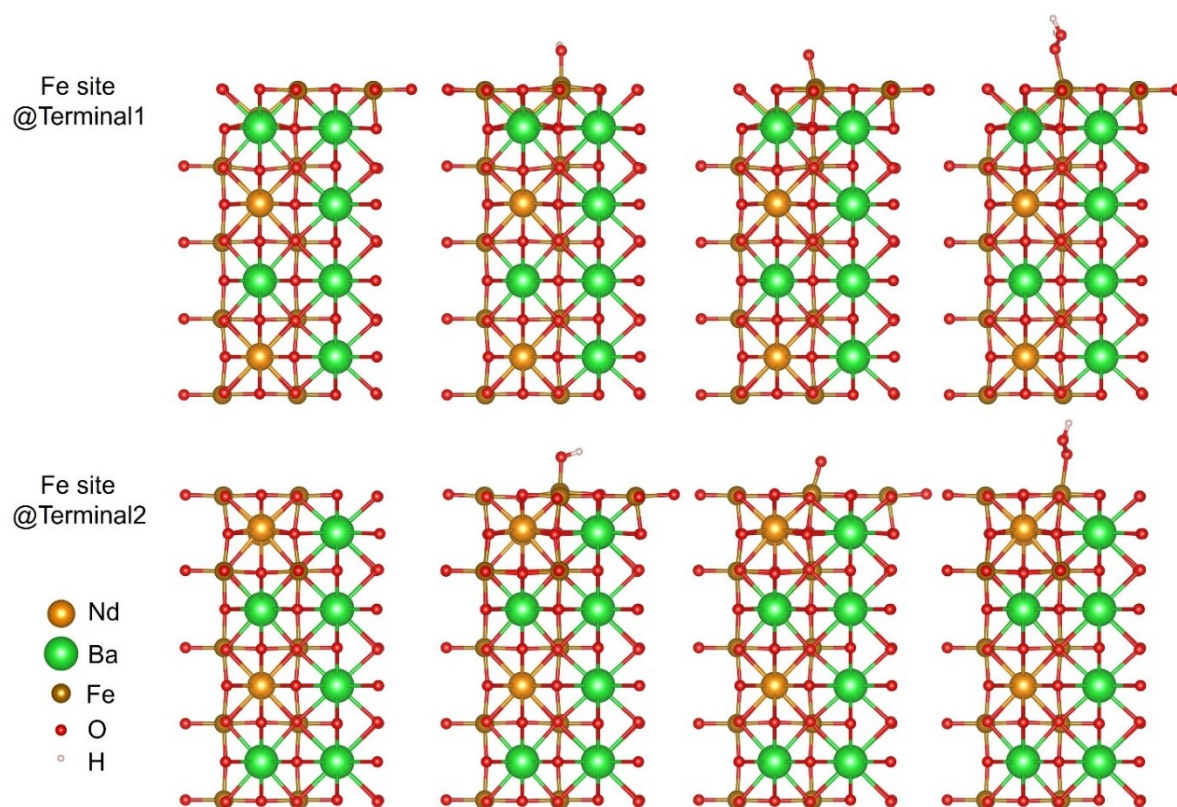


Figure S1 Schematic diagram illustrating the surface adsorption model for each step of OER on NBF.

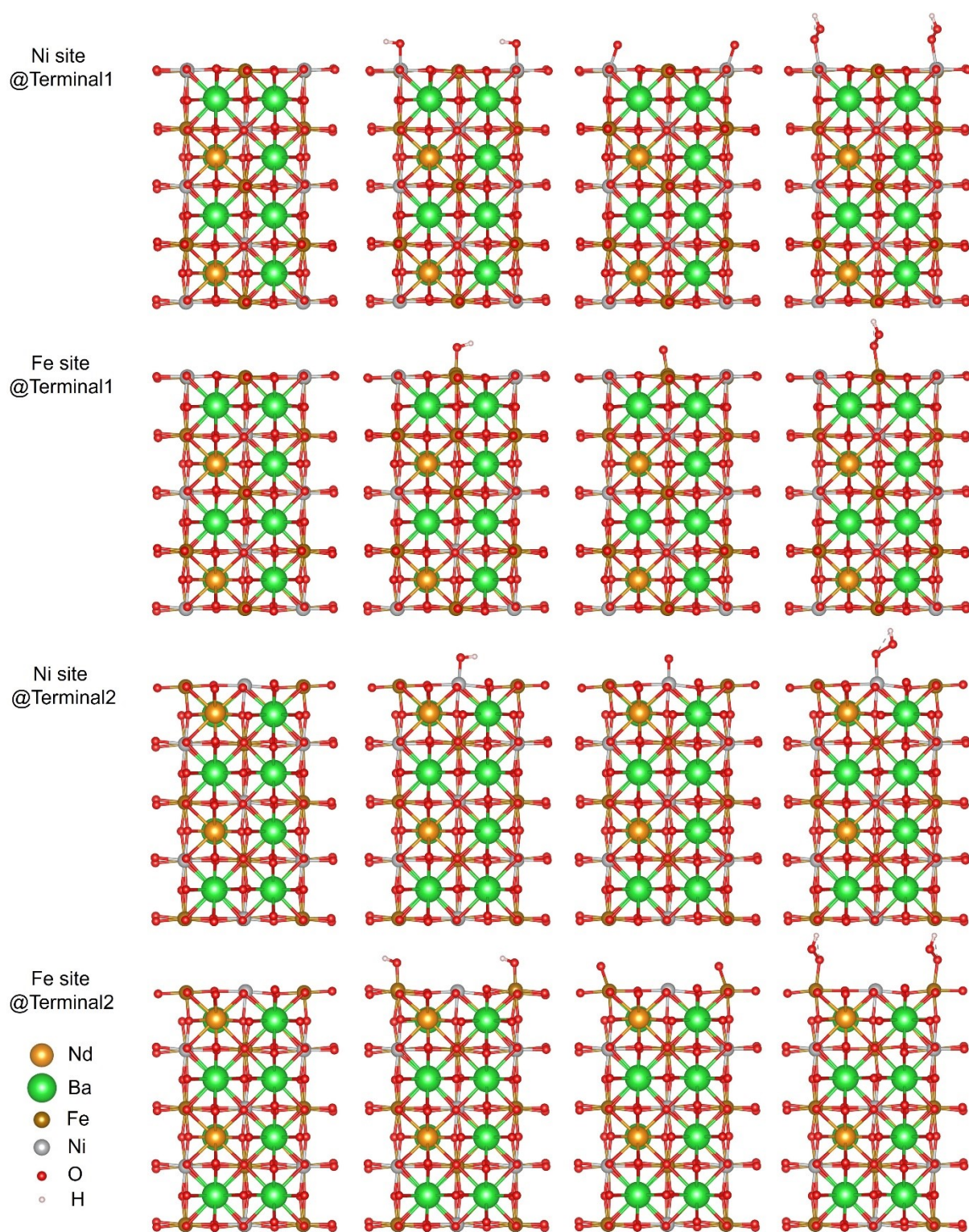


Figure S2 Schematic diagram illustrating the surface adsorption model for each step of OER on NBFN.

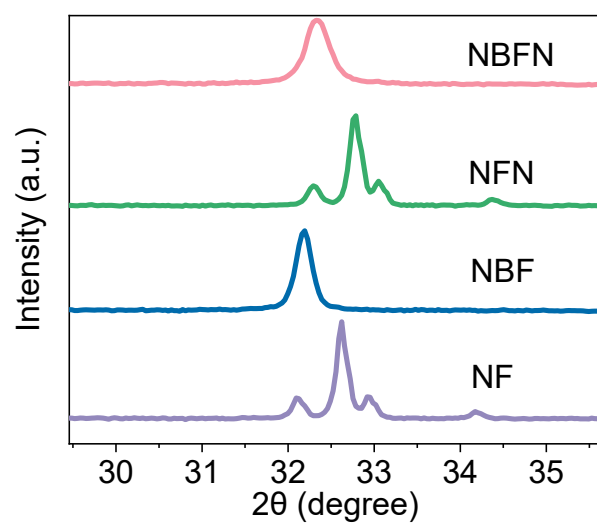


Figure S3 Magnified XRD pattern.



Figure S4 SEM-EDS results showing elements composition of NBFN.

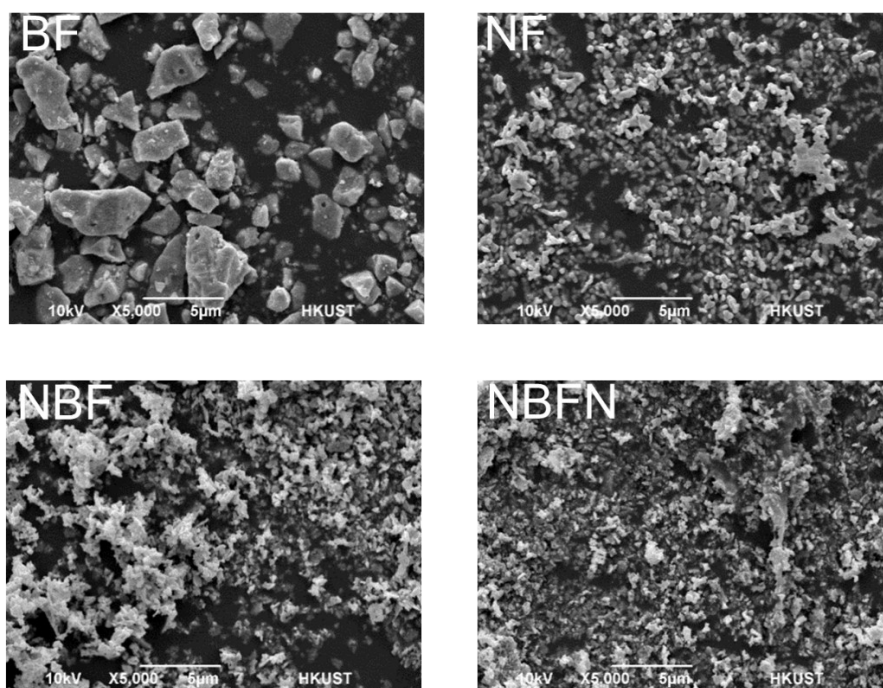


Figure S5 Morphology study of BF, NF, NBF and NBFN.

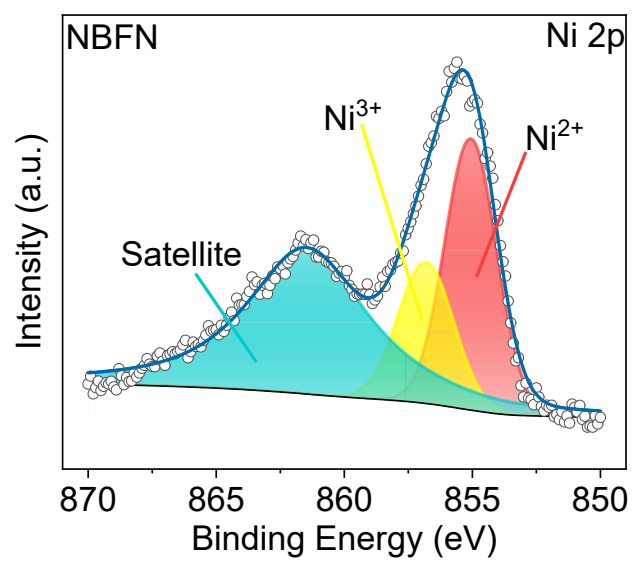


Figure S6 Ni 2p XPS spectra of NBFN.

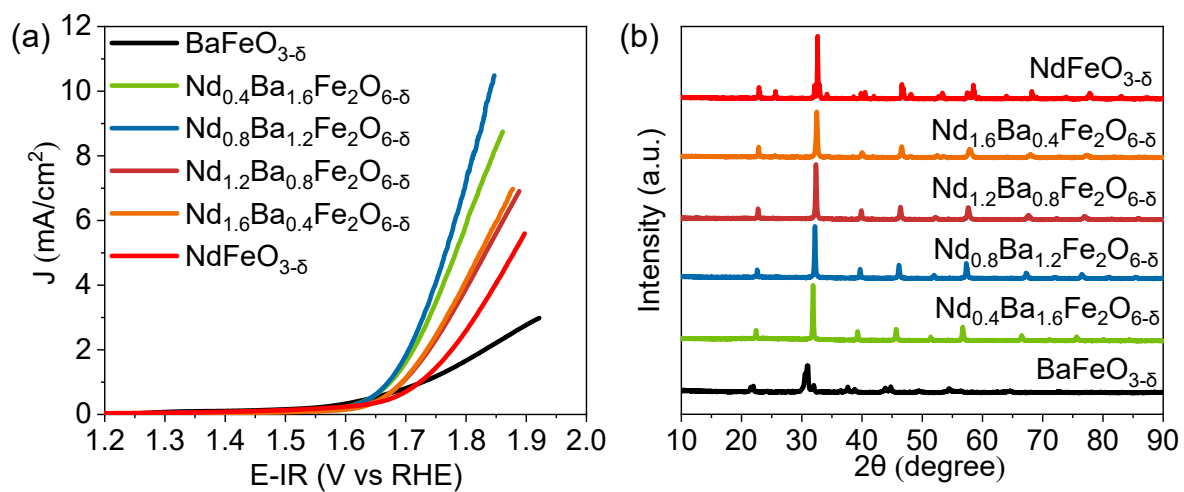


Figure S7 (a) LSV curves, (b) XRD pattern of $\text{Nd}_x\text{Ba}_{2-x}\text{Fe}_2\text{O}_{6-\delta}$.

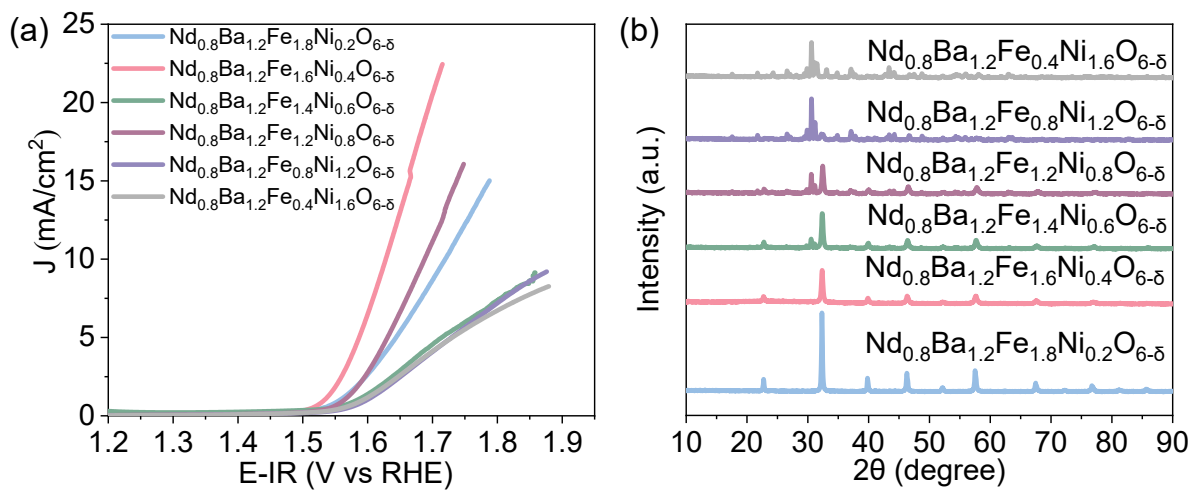


Figure S8 (a) LSV curves, (b) XRD pattern of $\text{Nd}_{0.8}\text{Ba}_{1.2}\text{Fe}_y\text{Ni}_{2-y}\text{O}_{6-\delta}$.

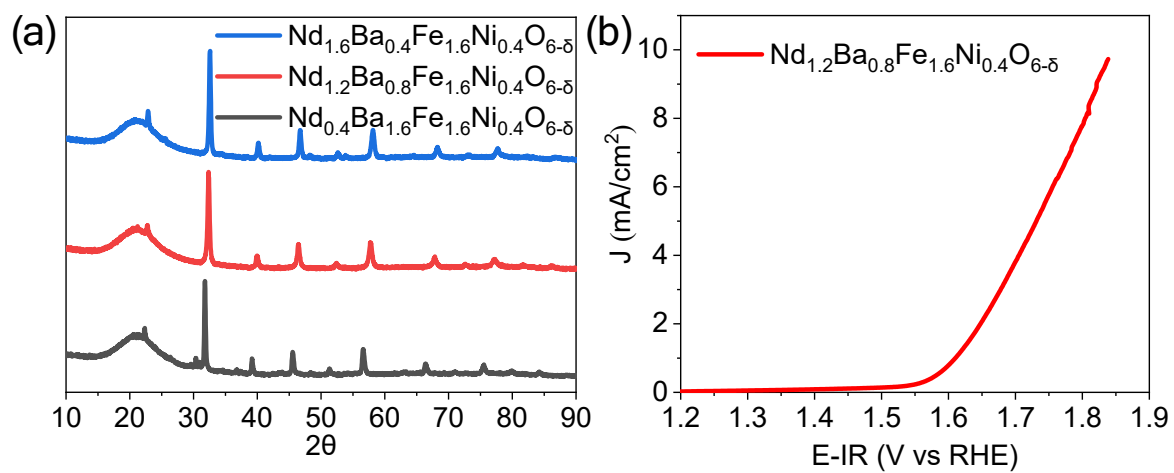


Figure S9 (a) XRD pattern of $\text{Nd}_x\text{Ba}_{2-x}\text{Fe}_{1.6}\text{Ni}_{0.4}\text{O}_{6-\delta}$, (b) LSV curve of $\text{Nd}_{1.2}\text{Ba}_{0.8}\text{Fe}_{1.6}\text{Ni}_{0.4}\text{O}_{6-\delta}$.

δ.

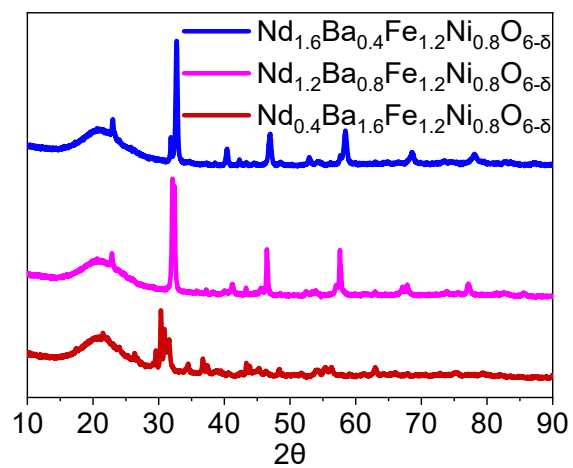


Figure S10 XRD pattern of $\text{Nd}_x\text{Ba}_{2-x}\text{Fe}_{1.2}\text{Ni}_{0.8}\text{O}_{6-\delta}$.

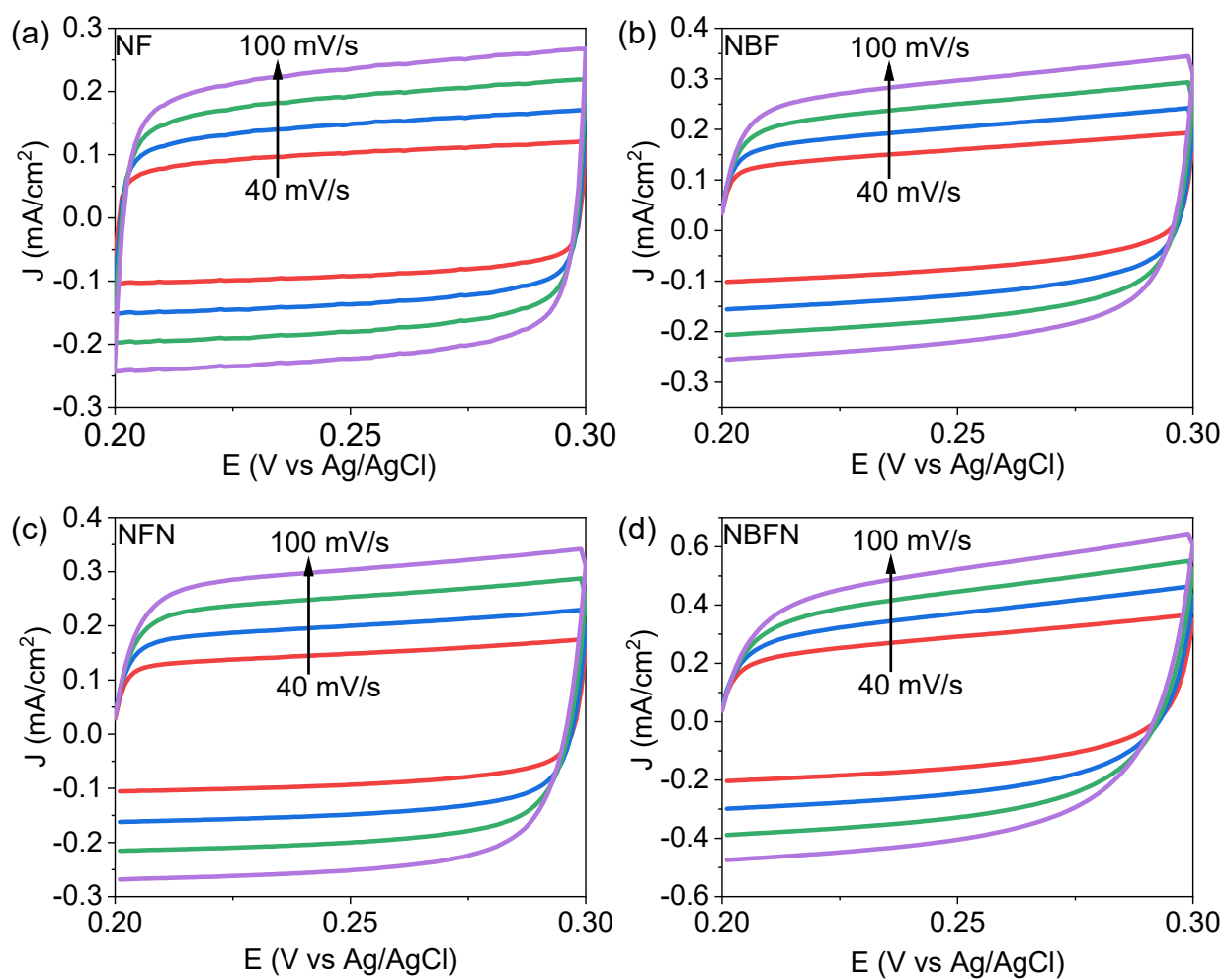


Figure S11 CVs of (a) NF, (b) NBF, (c) NFN and (d) NBFN at difference scan rates from 40 to 100 mV/s.

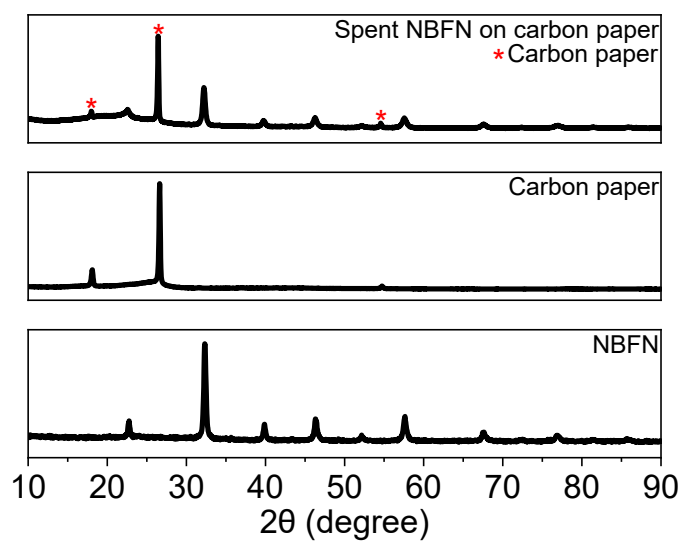


Figure S12 XRD pattern of fresh NBFN, carbon paper and NBFN loaded carbon paper after stability test.

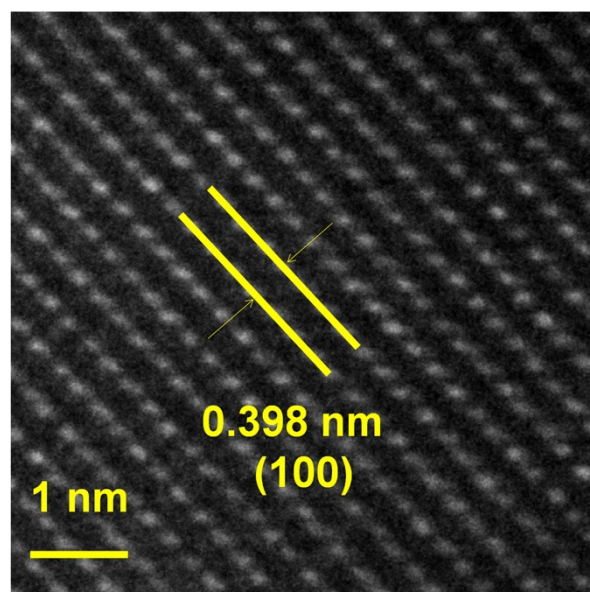


Figure S13 HR-TEM image of NBFN after stability.

Table S1 Composition study of NBFN through ICP.

Element	Atomic%
Nd	19.31%
Ba	29.22%
Fe	41.40%
Ni	10.07%

Table S2 The relative amounts of different surface oxygen species of NF, NBF, NFN and NBFN.

Composition	O ²⁻ (%)	O ²⁻ /O ⁻ (%)	OH ⁻ /O ₂ (%)	H ₂ O (%)
NF	36.7396	13.9236	39.5124	9.8244
NBF	35.8741	29.4212	29.2694	5.4353
NFN	33.8339	21.7364	41.1509	3.2788
NBFN	26.3	23.3081	46.5457	3.8462

Table S3 The Fe valence state of NF, NBF, NFN and NBFN.

Fe valence state	
NF	3.1891
NBF	3.2143
NFN	3.0146
NBFN	3.0017

Table S4 Comparison of overpotential@10 mA/cm² with the literature.

Catalysts	Overpotential@10 mA/cm ² (mV)	
LaNiO ₃	449	[1] https://doi.org/10.1039/D1TA09306A
La _{0.95} FeO _{3-δ}	410	[2] https://doi.org/10.1021/acs.chemmater.5b04457
BSCF	440	[3] https://doi.org/10.1002/cjoc.202100215
SrNb _{0.1} Co _{0.7} Fe _{0.2} O _{3-δ}	420	[4] https://doi.org/10.1002/anie.201408998
PrBaCo ₂ O _{5+δ}	520	[5] https://doi.org/10.1002/chem.201700507
Sr ₂ Fe _{1.5} Mo _{0.5} O _{6-δ}	550	[6] https://doi.org/10.1016/j.seppur.2022.122316
PrBa _{0.85} Ca _{0.15} FeMnO _{5+δ}	400	[7] https://doi.org/10.1021/acs.chemmater.7b01114
LaCr _{0.5} Fe _{0.5} O ₃	390	[8] https://doi.org/10.1038/s41598-020-70283-9
RuO ₂	390	[9] https://doi.org/10.1021/acscatal.7b02650
IrO ₂	450	[10] https://doi.org/10.1039/C9TA01404G
This work	320	/

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