

Supporting information

Facet-Dependent Photocatalytic Reduction of Nitroaromatics Using Tailored SrTiO₃ Crystals: Mechanism and Reactivity Enhancement

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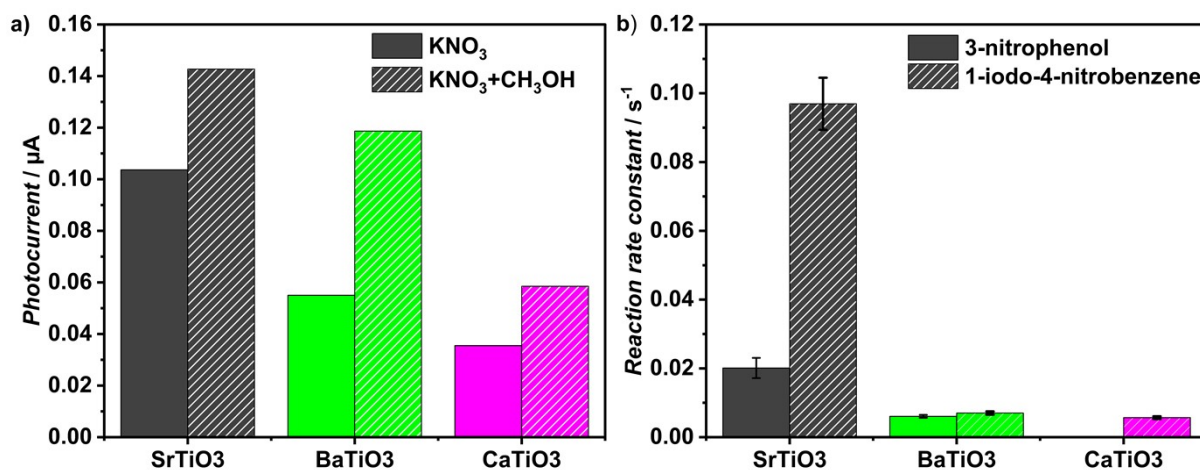


Fig. S1. The value of photocurrents (a) and reaction rate constant for 3-NP and INB reduction reaction in the presence of commercial titanates (SrTiO₃, BaTiO₃, CaTiO₃).

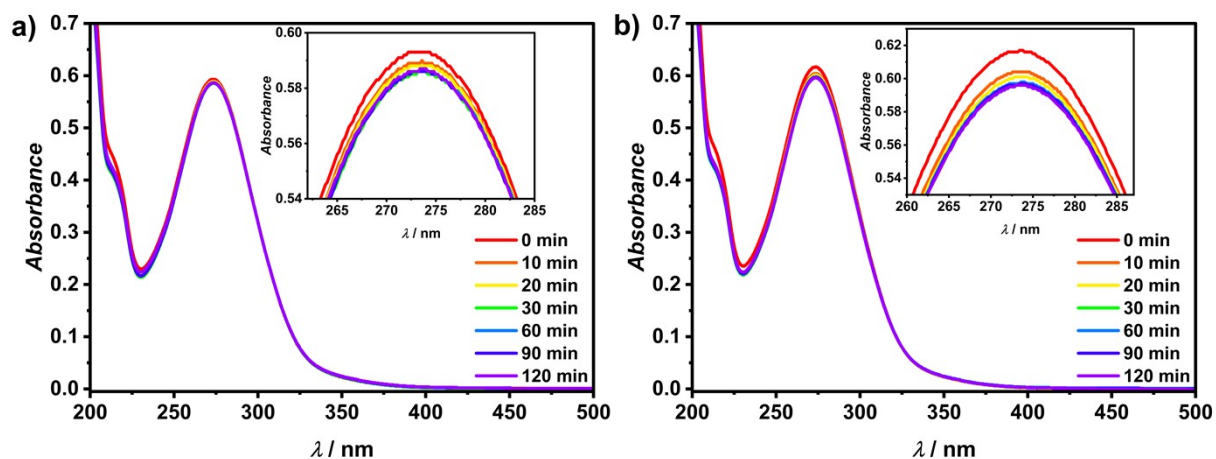


Fig. S2. UV-Vis spectra for the reduction reaction of PNBA in water in the presence of cubes (a) and rhombic dodecahedrons (b) SrTiO_3 crystals.

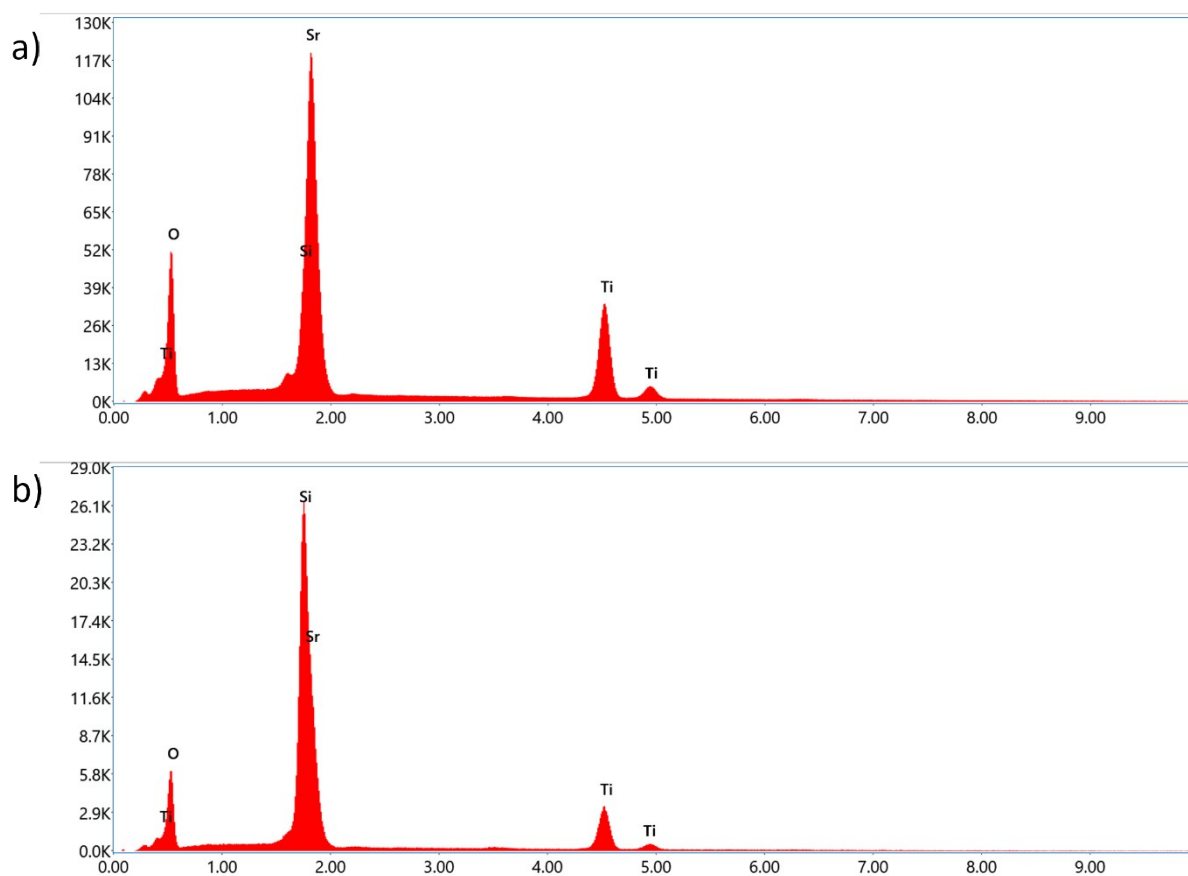


Fig. S3. EDS spectrum for cubes (a) and rhombic dodecahedrons (b) SrTiO_3 crystals.

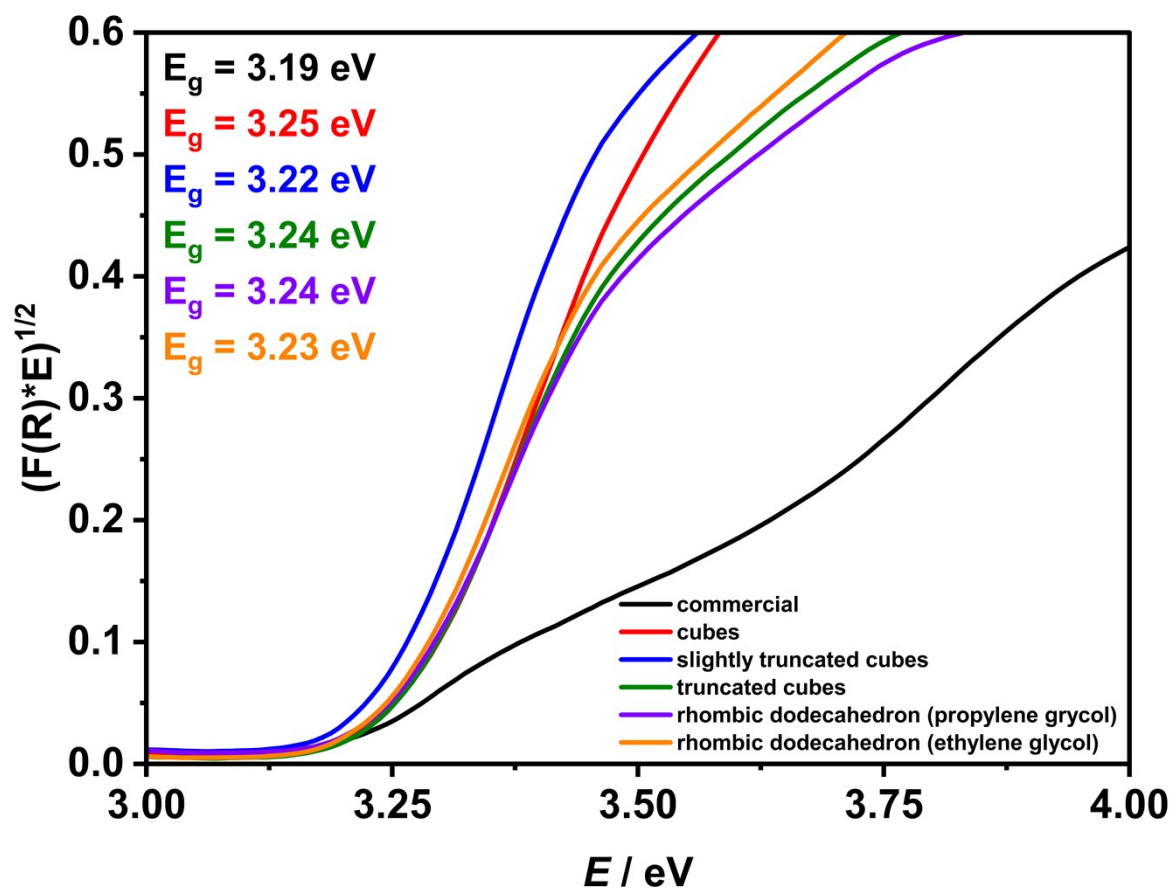


Fig. S4. Tauc plots and calculated band gap energies for tailored SrTiO₃ crystals.

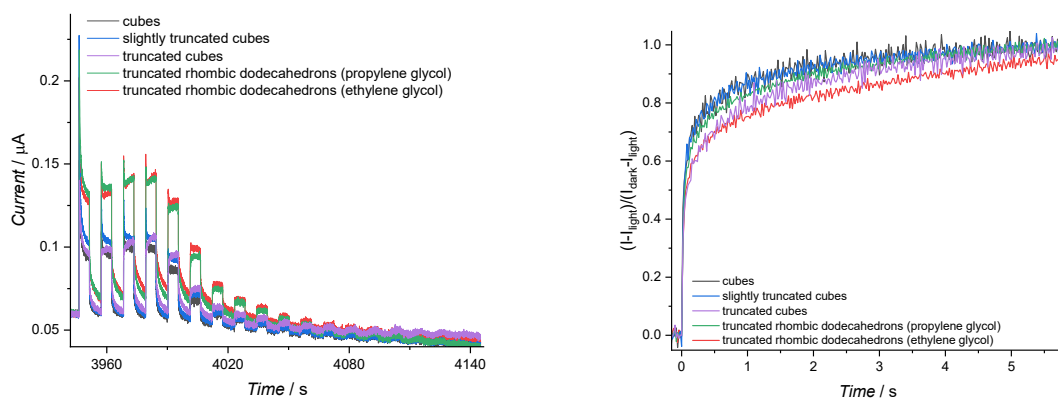


Fig. S5. Transient photocurrent for SrTiO₃ materials collected at 1000 mV vs Ag/AgCl (left), and normalised photocurrent decay for SrTiO₃ materials after irradiation with monochromatic light (340 nm), where I represents the current of the electrode after switching off the light. I_{light} and I_{dark} denote the steady-state current values upon illumination and in the dark, respectively.

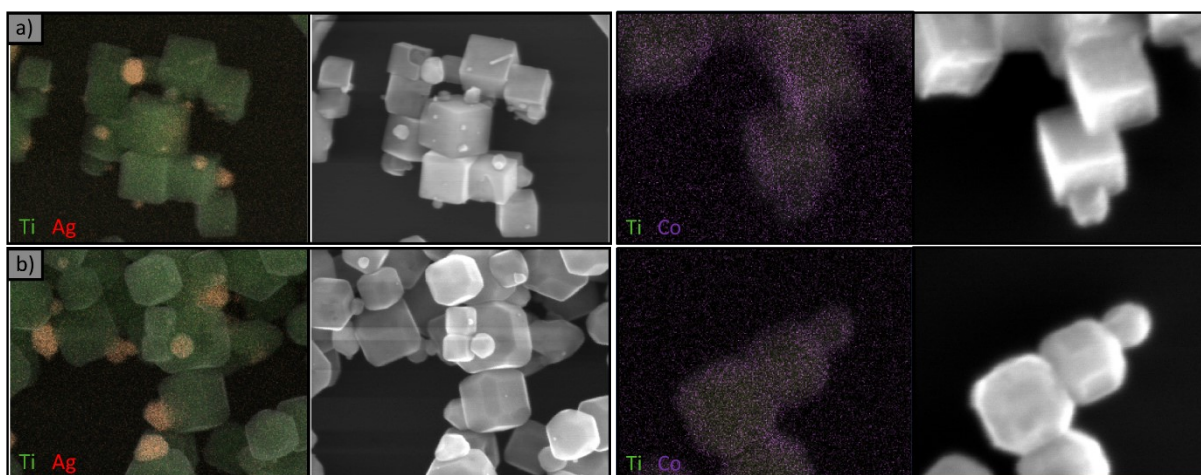


Fig. S6. EDS map of titanium (Ti), silver (Ag) and cobalt (Co) distribution after photo-deposition on cubes (a) and rhombic dodecahedrons (b).

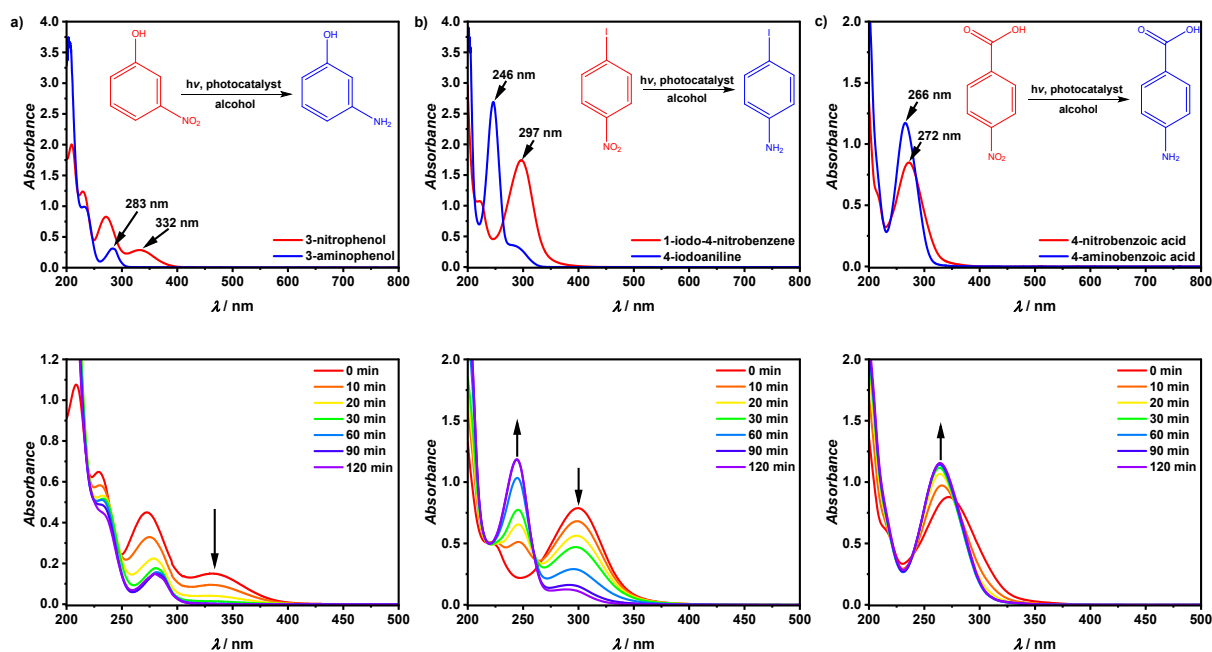


Fig. S7. Standard absorption spectra of substrates and products and evolution of absorption spectra of the tested solution upon irradiation with photocatalyst for 3-NP (a), INB (b) and PNBA (c) as a substrate.

Table S1. Comparison of reaction rate constant and conversion for 3NP, INB and PNBA reduction for tailored SrTiO₃ crystals.

	3NP		INB		PNBA	
	k / s^{-1}	$X / \%$	k / s^{-1}	$X / \%$	k / s^{-1}	$X / \%$
commercial	0.015	18.0	0.040	33.3	0.695	85.5
cubes	0.040	9.9	0.022	15.2	0.231	71.3
slightly truncated cubes	0.003	6.7	0.017	9.4	0.281	74.0
truncated cubes	0.545	67.5	0.445	87.1	0.716	88.7
truncated rhombic dodecahedron	0.578	57.9	0.441	86.4	0.793	93.1
(propylene glycol)						
truncated rhombic dodecahedron	0.344	62.1	0.151	71.0	0.534	96.3
(ethylene glycol)						

k / s^{-1} – reaction rate constant; $X / \%$ - conversion (after 2h of the reaction).

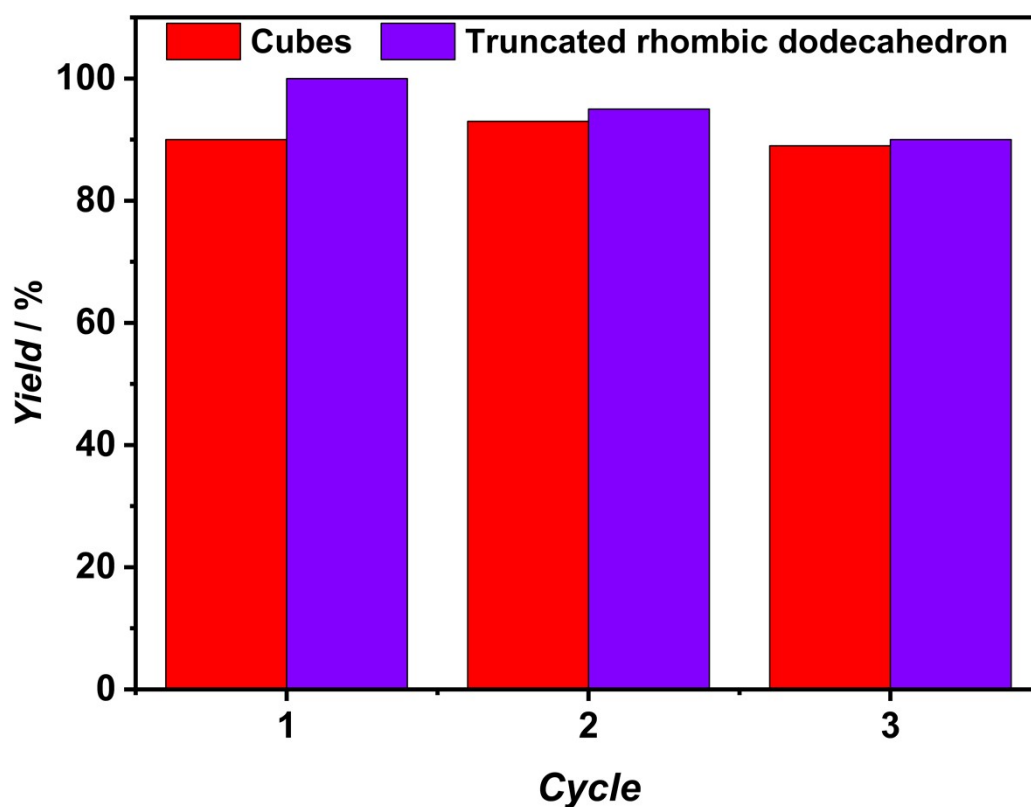


Fig. S8. Cycling tests of photocatalyst in the reduction of 1-iodo-4-nitrobenzene.