Supporting Information

ZIF-L derived carbon flower with in-situ grown CNTs accelerates the reaction kinetics of Li-Se batteries

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Fig. S1 SEM images of (a) Fe-ZIF-L-10, (b) CNTs@HPC-10 and (c) Se/CNTs@HPC-10.



Fig. S2 (a-d) HAADF-STEM image and element distributions of C, Se and N in Se/HPC.



Fig. S3 (a) XRD patterns of the precursors. (b) XRD patterns after carbonization. (c) N_2 adsorption-desorption isotherms after Se encapsulation.



Fig. S4 (a) C 1s, (b) N 1s and (c) Se 3d XPS spectra of Se/HPC. (d) Fe2p XPS spectrum of Se/CNTs@HPC-5.



Fig. S5 Cycling performance of Se/CNTs@HPC-5 cathodes with different areal mass loading of Se.



Fig. S6 CV curves of Se/CNTs@HPC-10 cathode at the scan rate of 0.2 mV s⁻¹.



Fig. S7 Capacitive contribution at the scan rate of 0.5 mV s⁻¹ for (a) Se/HPC-5 and (b) Se/HPC, respectively.



Fig. S8 Linear fitting of peak currents versus square root of the scan rates.



Fig. S9 (a) GITT curves and Li⁺ diffusion coefficient of Se/CNTs@HPC-10. (b) EIS spectra of different cathodes after cycling 10 cycles at 1 C.



Fig. S10 SEM image of Se/CNTs@HPC-5 electrode after 100 cycles at 1 C.

Matariala	Current	Cycle	Reversible	Areal	Deference
Matchais	Tale	number	capacity	loading	Kelelelice
	(C)	(n)	$(mAh g^{-1})$	$(mg \ cm^{-2})$	
Se/CNTs@HPC-5	0.5	350	606	1.0	This work
	5	400	355	1.0	
Se/C	0.15	250	430	1.2	1
Se/HPNC	1	500	410	-	2
Se@NPC-NS	0.5	225	585	0.72	3
Se@NHCS	0.5	1000	443	1.0	4
Se@CNTs@MPC	0.1	100	596	-	5
Se/CNTs microsphere	1	500	440	-	6
APPC/Se@PDA	5	1400	500	0.8-4.0	7
Se/Co-NC	1	200	480	1.0	8
Se@LHPC	0.5	450	500	0.47	9
CSe@HNCNFs	0.2	100	699	27.12	10

 Table S1
 Electrochemical performances of the reported Se/C composites

Table S2EDX elemental analysis of Se/CNTs@HPC-5

Z Eleme		t Family	Atomic Fraction	Atomic Error	Mass Fraction	Mass Error
	Element		(%)	(%)	(%)	(%)
6	С	K	47.12	6.78	15.48	1.48
7	Ν	K	9.85	2.43	3.77	0.84
8	0	K	6.49	1.60	2.84	0.63
30	Zn	K	2.72	0.53	4.86	0.80
34	Se	K	33.83	6.55	73.05	11.75

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