

Supporting Information

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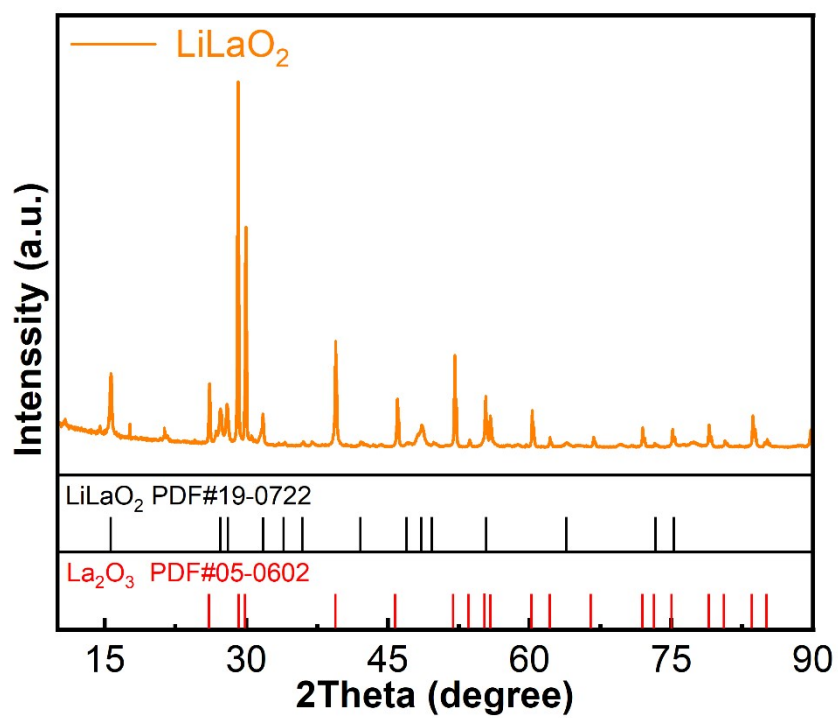


Fig. S1 XRD pattern of as-prepared LiLaO_2 with standard PDF cards.

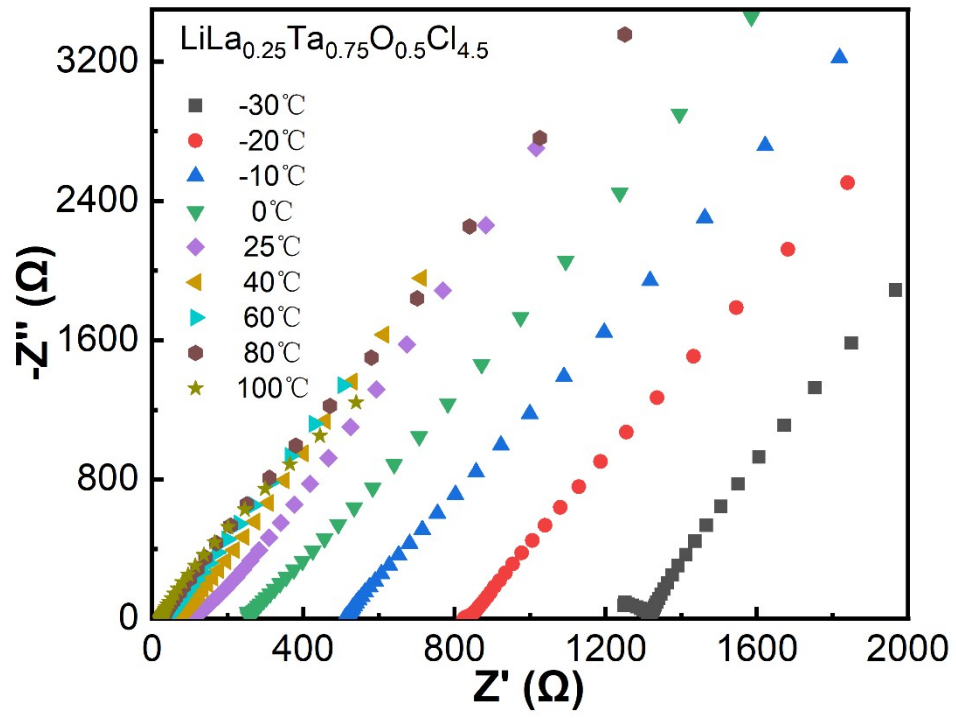


Fig. S2 Nyquist profiles for $\text{LiLa}_{0.25}\text{Ta}_{0.75}\text{O}_{0.5}\text{Cl}_{4.5}$ in the temperature range of -30 °C and 100 °C.

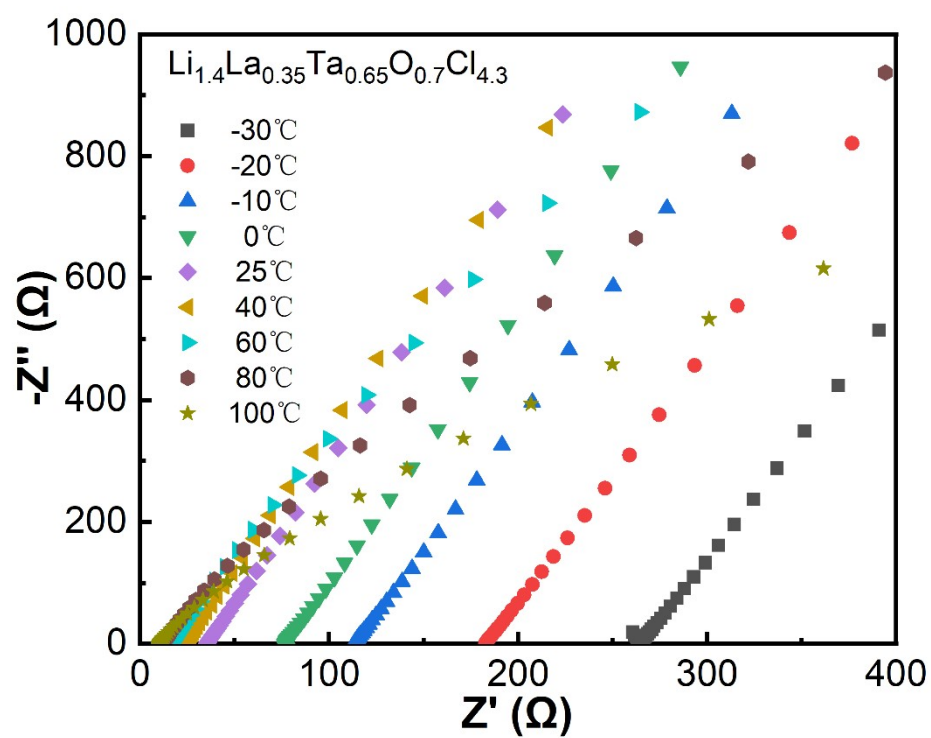


Fig. S3 Nyquist profiles for $\text{Li}_{1.4}\text{La}_{0.35}\text{Ta}_{0.65}\text{O}_{0.7}\text{Cl}_{4.3}$ in the temperature range of -30°C and 100°C .

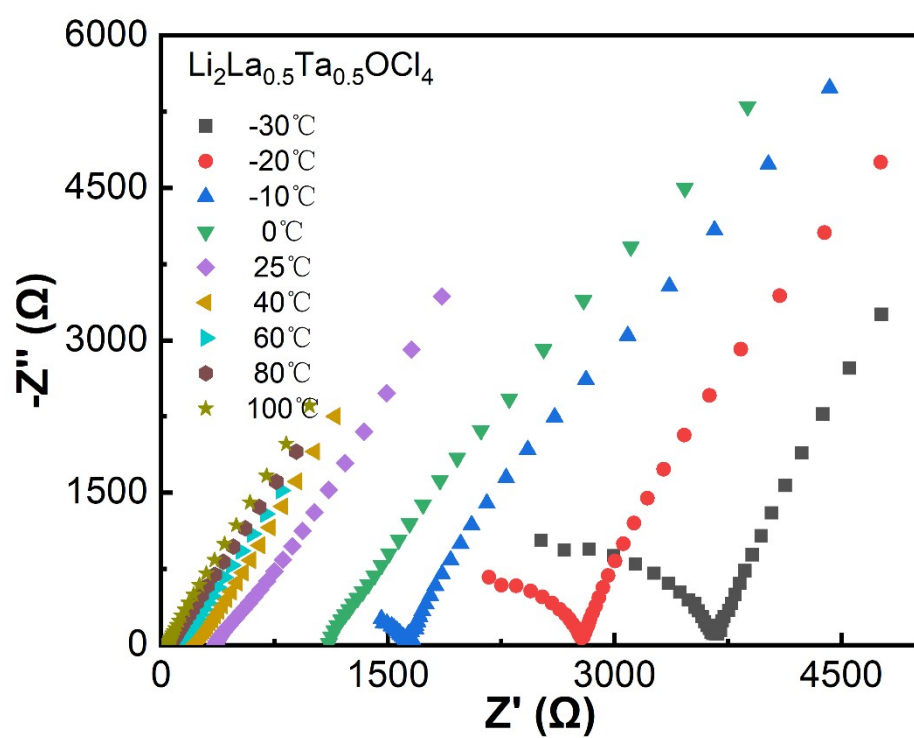


Fig. S4 Nyquist profiles for $\text{Li}_2\text{La}_{0.5}\text{Ta}_{0.5}\text{OCl}_4$ in the temperature range of -30°C and 100°C .

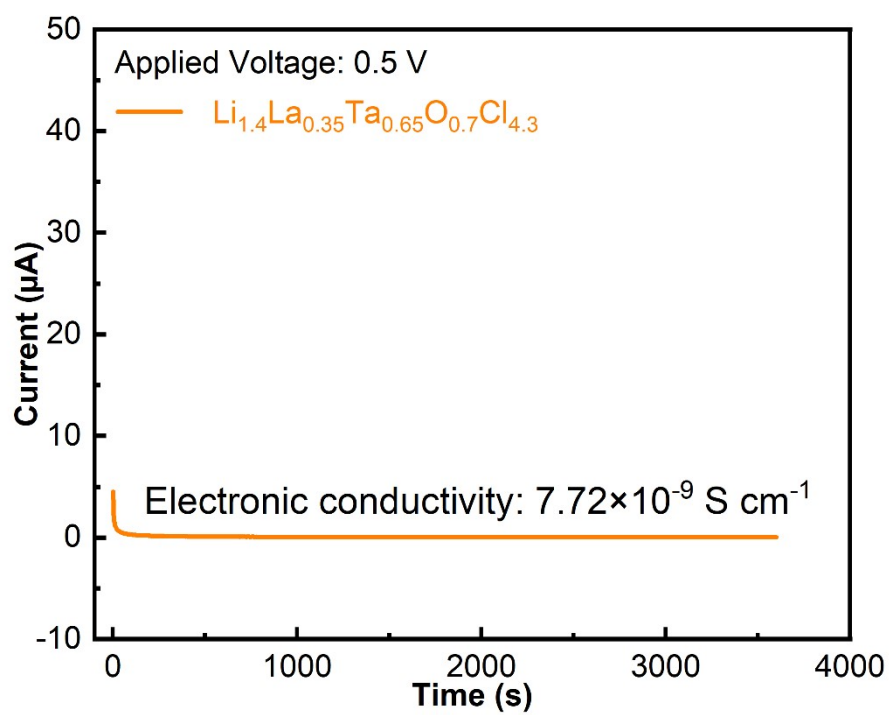


Fig. S5 Direct-current polarization curve for $\text{Li}_{1.4}\text{La}_{0.35}\text{Ta}_{0.65}\text{O}_{0.7}\text{Cl}_{4.3}$.

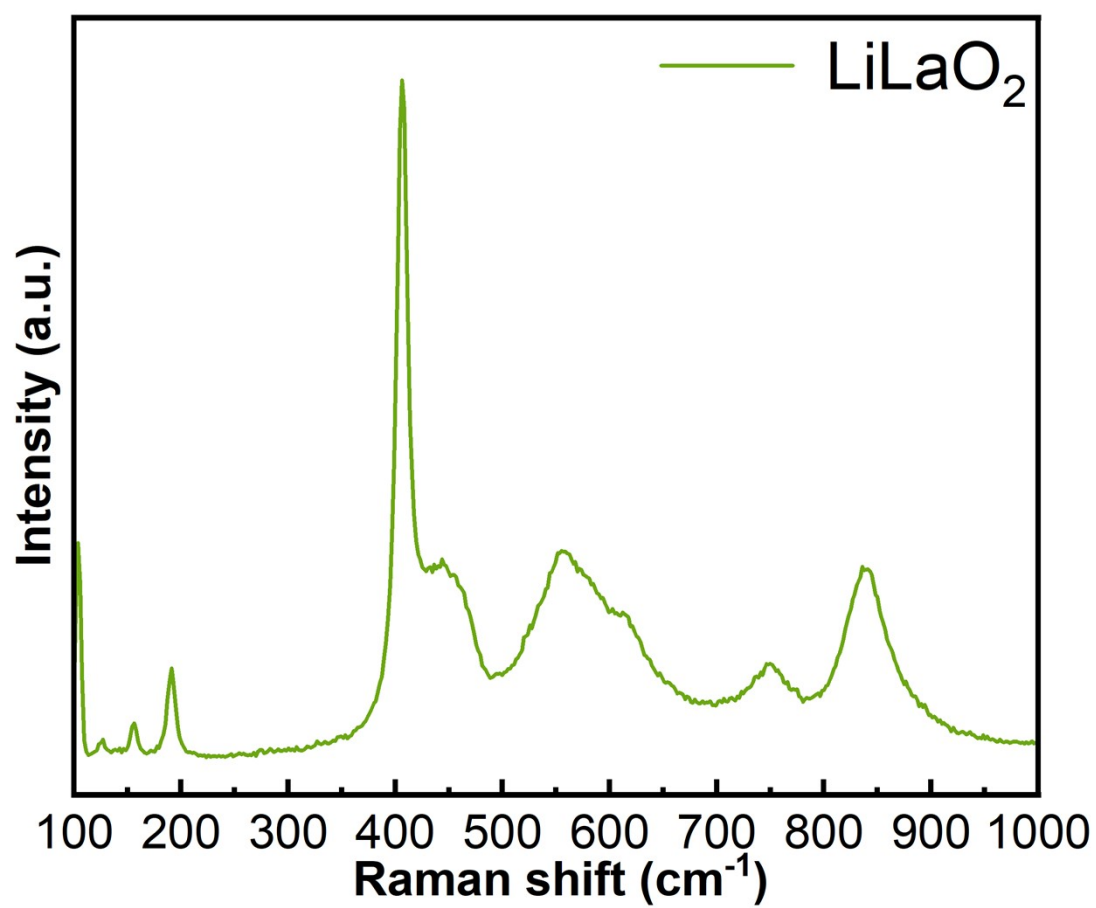


Fig. S6 Raman spectrum of as-prepared LiLaO_2 .

Table S1. Comparison of ionic conductivities and symmetric Li cells with the representative halide SEs

Materials	Conductivity (mS cm⁻¹)	Critical current density (mA cm⁻²)	Galvanostatic cycling: current density (mA cm⁻²), fixed capacity (mAh cm⁻²), duration (h)	Overpotential (mV)	Ref.
Li ₃ YCl ₆	0.345	none	0.13, 0.13, 500	600	1
Li ₃ InCl ₆	1.29	none	0.05, 0.05, 10 (at 80 °C)	surpassing 500	2
Li ₂ OHCl/Li ₃ InCl ₆ /Li ₂ OHCl	0.027	none	0.2, 0.2, 100 (at 80 °C)	approaching 500	
Li ₃ YBr _{5.7} F _{0.3}	1.8	0.9	0.1, 0.1, 1000	~50	3
Li ₃ YBr ₆	2.1	0.3	0.1, 0.1, ~300	~100	
Li ₃ ScCl ₆	3.0	none	0.1, 0.083, 2500	620	4

Li_2ZrCl_6	0.28	0.6	0.1, 0.025, ~300	~2000	5
$\text{Li}_2\text{ZrCl}_{5.5}\text{F}_{0.5}$	0.16	0.7	0.1, 0.025, 800	~1000	
Li_4YI_7	1.04	none	0.1, 0.1, 1000	12	6
$\text{Li}_{1.8}\text{Fe}_{1.1}\text{Cl}_4$	0.042	0.5	0.05, 0.05, 1600	approachin g 2000	7
LiAlOCl-681	1.0	none	none	none	8
AlOCl	1.02	none	none	none	9
$\text{Li}_2\text{O-Li}_2\text{ZrCl}_6$	1.1	none	none	none	10
$\text{Li}_{0.388}\text{Ta}_{0.238}\text{La}_{0.47}$ 5Cl_3	3.02	4	0.2, 1.0, 5000	~100	11
$\text{Li}_{1.4}\text{La}_{0.35}\text{Ta}_{0.65}\text{O}_{0.7}\text{Cl}_{4.3}$	4.9	40	10, 10, 1214	430	This work

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