

Supporting Information

Construction of aqueous supercapacitors with oxidation suppression of nickel phosphide via interfacial engineering and electric field modulation for enhanced secondary energy storage

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Results and discussion

1. Crystallographic, morphological, and elemental composition

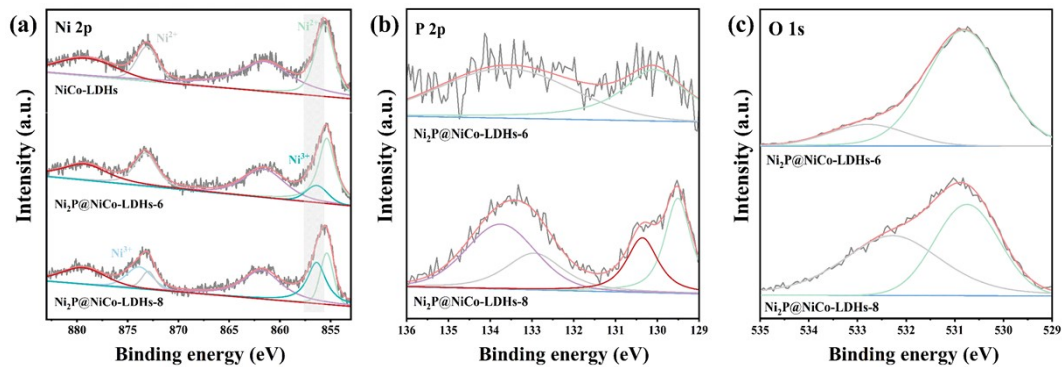


Figure S1. XPS analysis and mapping: (a) Ni 2p; (b) P 2p; (c) O 1s.

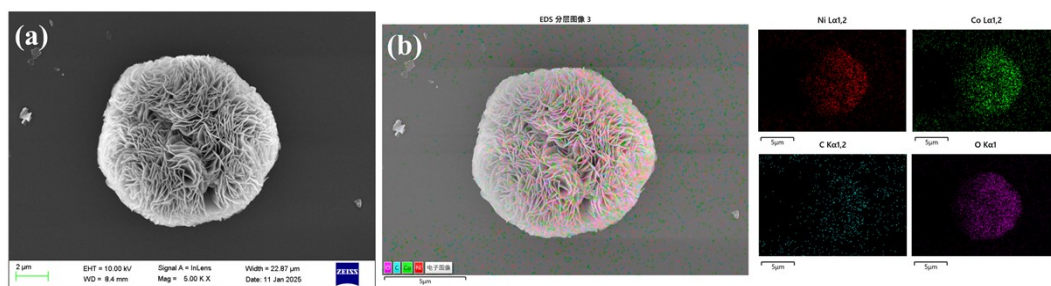


Figure S2. (a) and (b) Elemental mapping graphs of nickel-cobalt hydroxide.

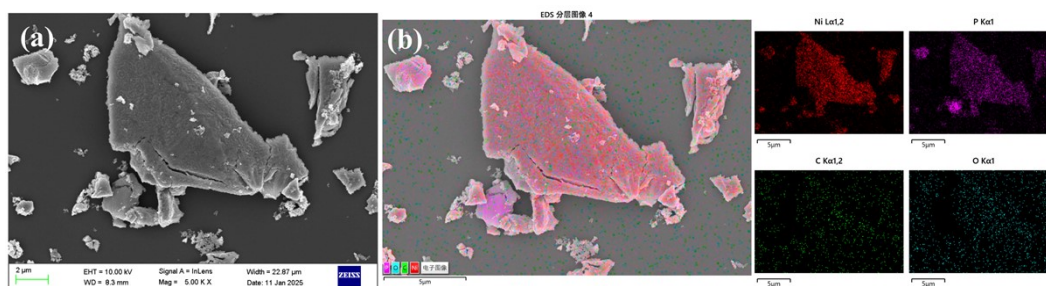


Figure S3. (a) and (b) Elemental mapping graphs of Ni_2P .

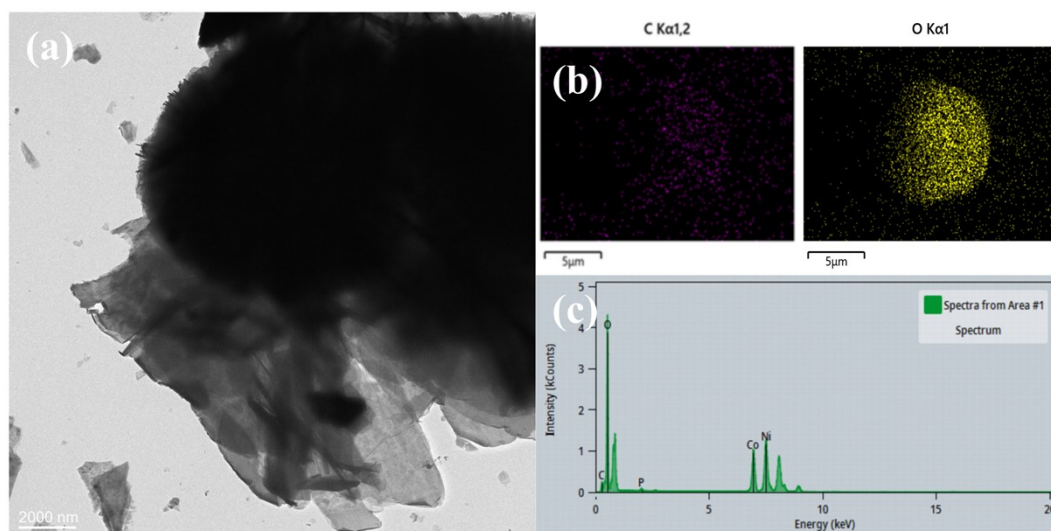


Figure S4. (a) Low-magnification morphologies of $\text{Ni}_2\text{P}@ \text{NiCo-LDHs-6}$; (b) Elemental mapping graphs of $\text{Ni}_2\text{P}@ \text{NiCo-LDHs-6}$; (c) Elemental content diagram of $\text{Ni}_2\text{P}@ \text{NiCo-LDHs-6}$.

2. First principle calculations

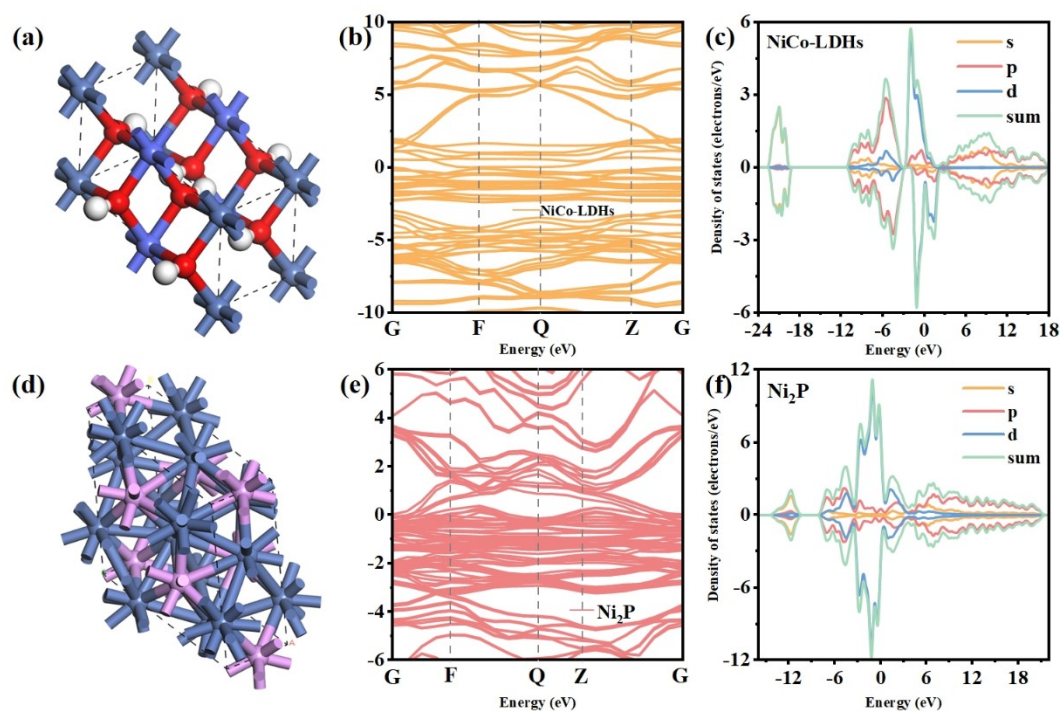


Figure S5. The following were calculated for nickel cobalt layered double hydroxide and nickel phosphide: (a) and (d) structural models; (b) and (e) energy level diagrams; (c) and (f) density of states diagrams.

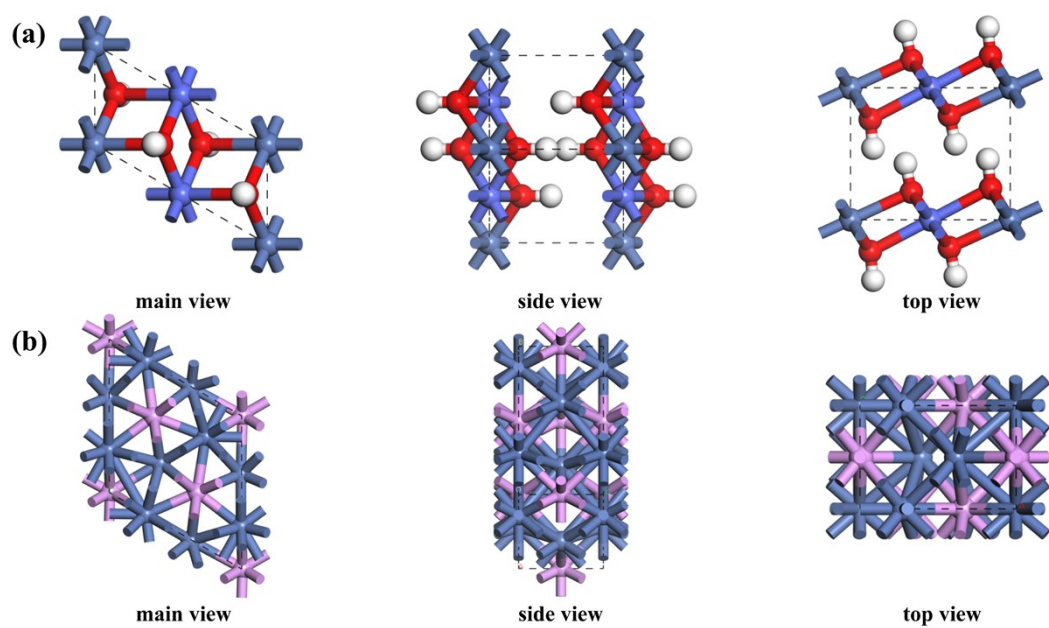


Figure S6. (a) Three-view diagrams of the nickel cobalt hydrotalcite model; (b) Three-view diagrams of the nickel phosphide model.

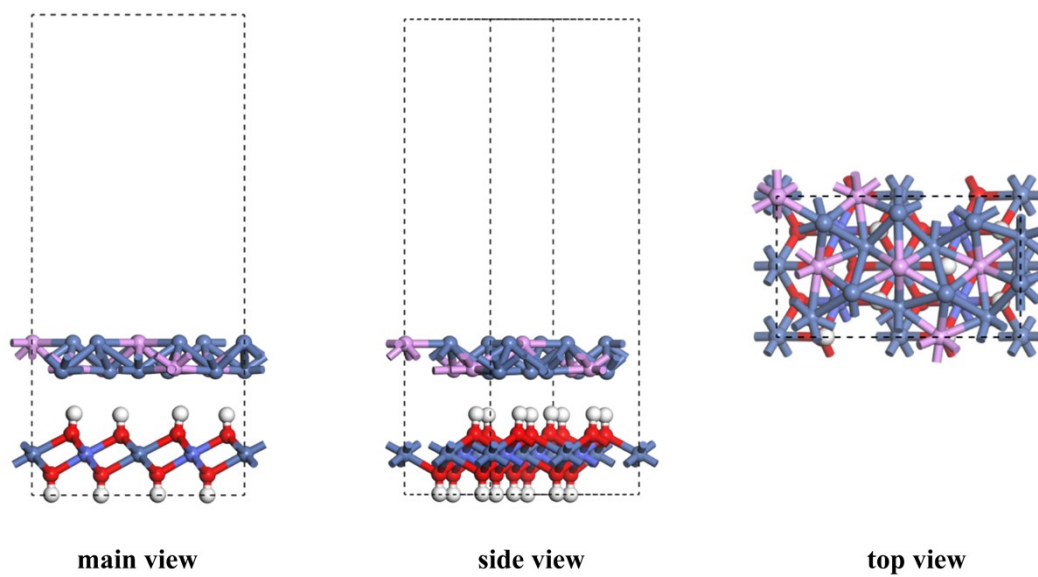


Figure S7. Three-view diagrams of the constructed $\text{Ni}_2\text{P}@\text{NiCo-LDHs}$ heterostructure model.

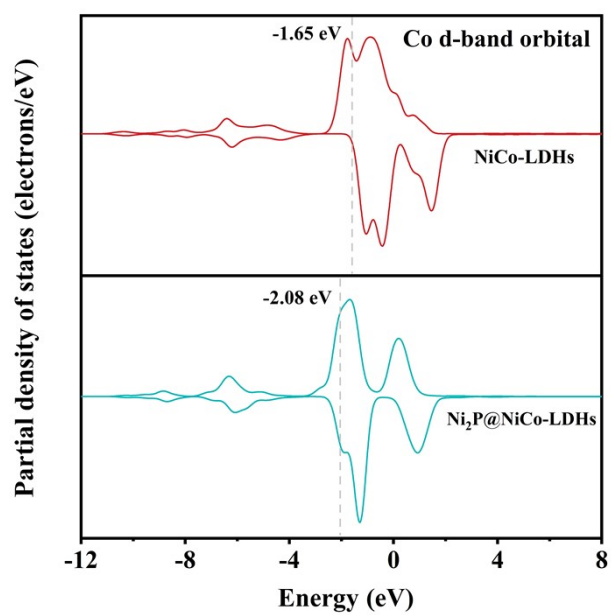


Figure S8. The d-orbital center of Co in the calculated Ni₂P@NiCo-LDHs heterostructure model.

3. Three-Electrode System

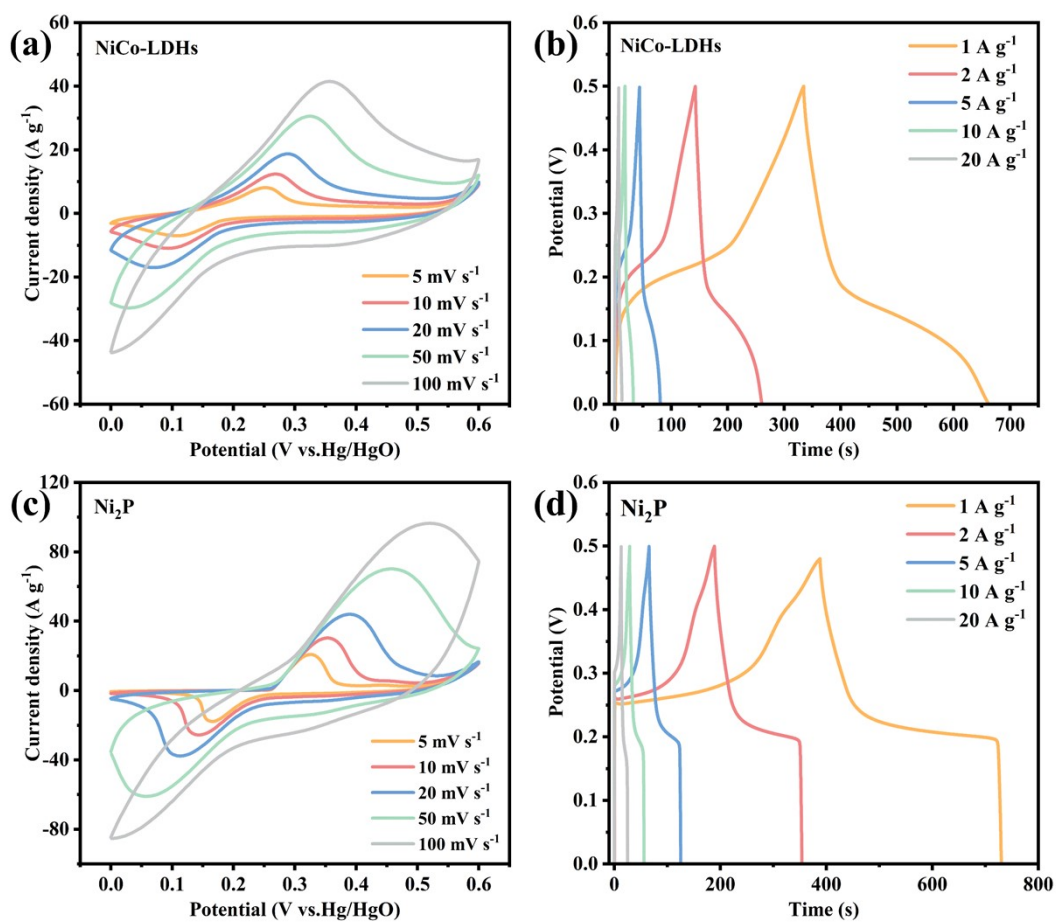


Figure S9. (a) CV curve of NiCo-LDHs; (b) GCD curve of NiCo-LDHs; (c) CV curve of Ni₂P; (d) GCD curve of Ni₂P.

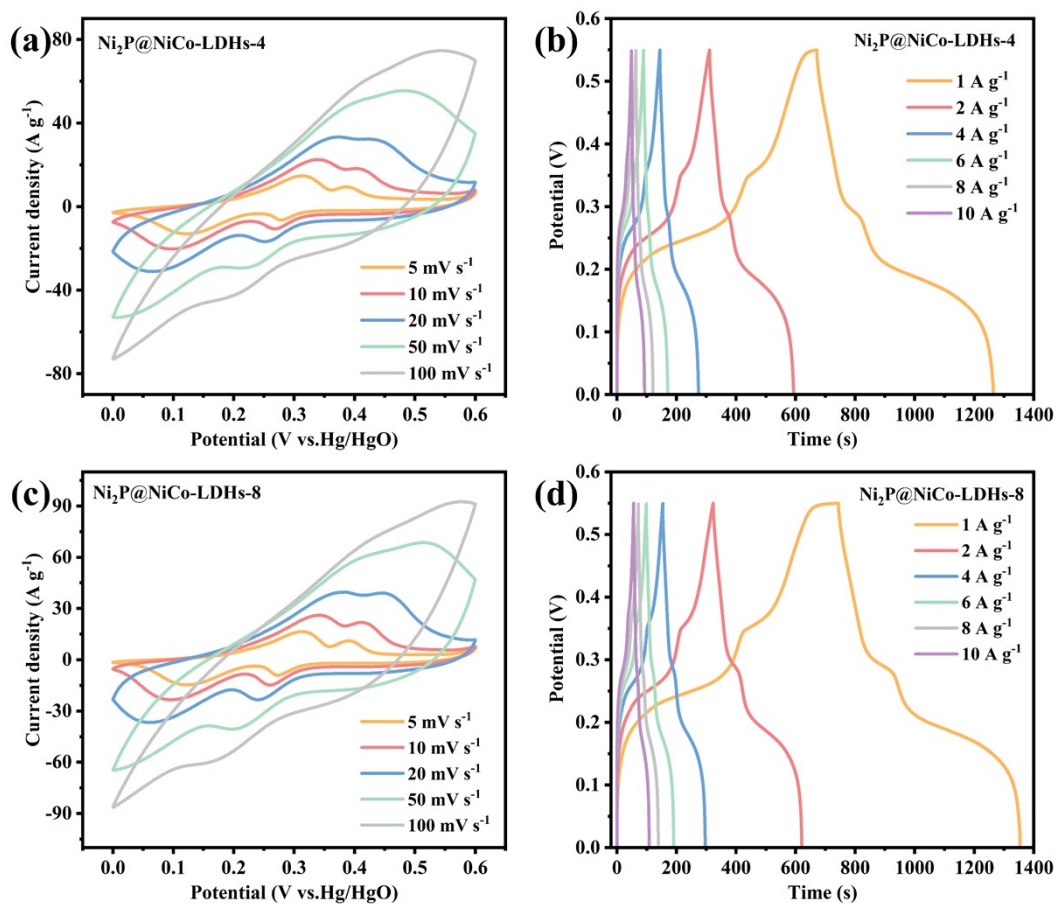


Figure S10. (a) CV curve of $\text{Ni}_2\text{P@NiCo-LDHs-4}$; (b) GCD curve of $\text{Ni}_2\text{P@NiCo-LDHs-4}$; (c) CV curve of $\text{Ni}_2\text{P@NiCo-LDHs-8}$; (d) GCD curve of $\text{Ni}_2\text{P@NiCo-LDHs-8}$.

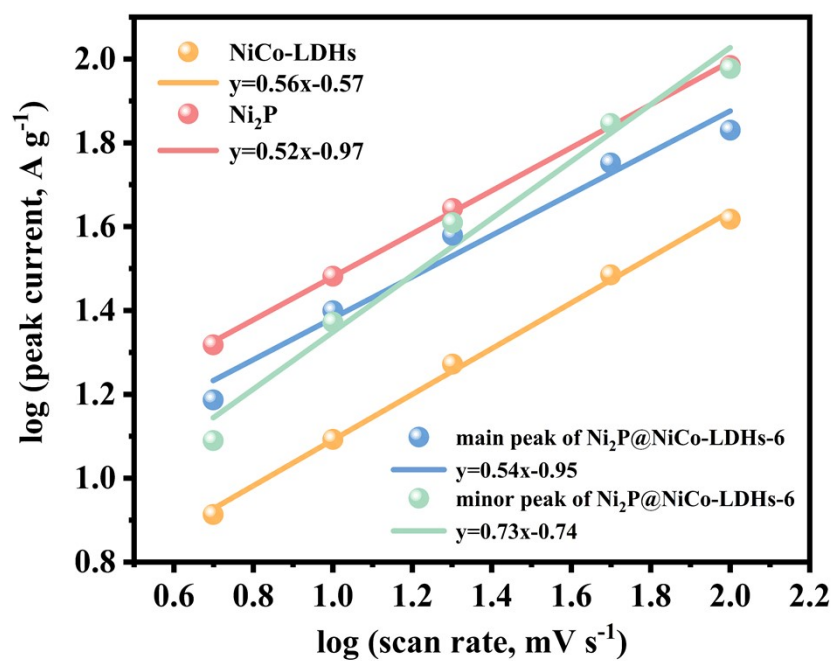


Figure S11. The relationship between the logarithm of the peak-to-peak current of anodes composed of different materials and the logarithm of the scan rate.

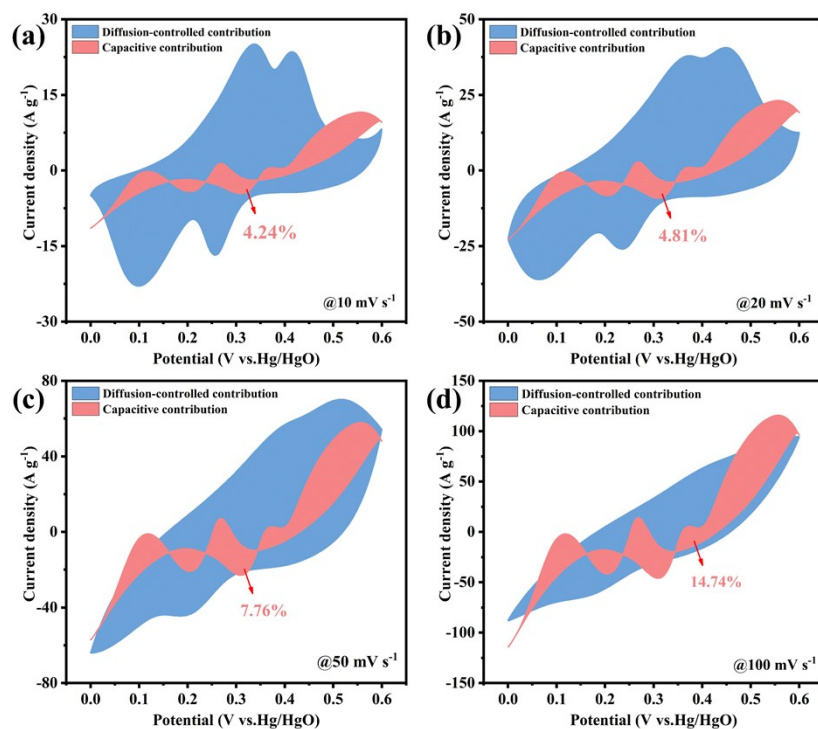


Figure S12. The proportion of the double-layer capacitance control to the total capacitance control at different scanning rates in $\text{Ni}_2\text{P}@\text{NiCo-LDHs-6}$.

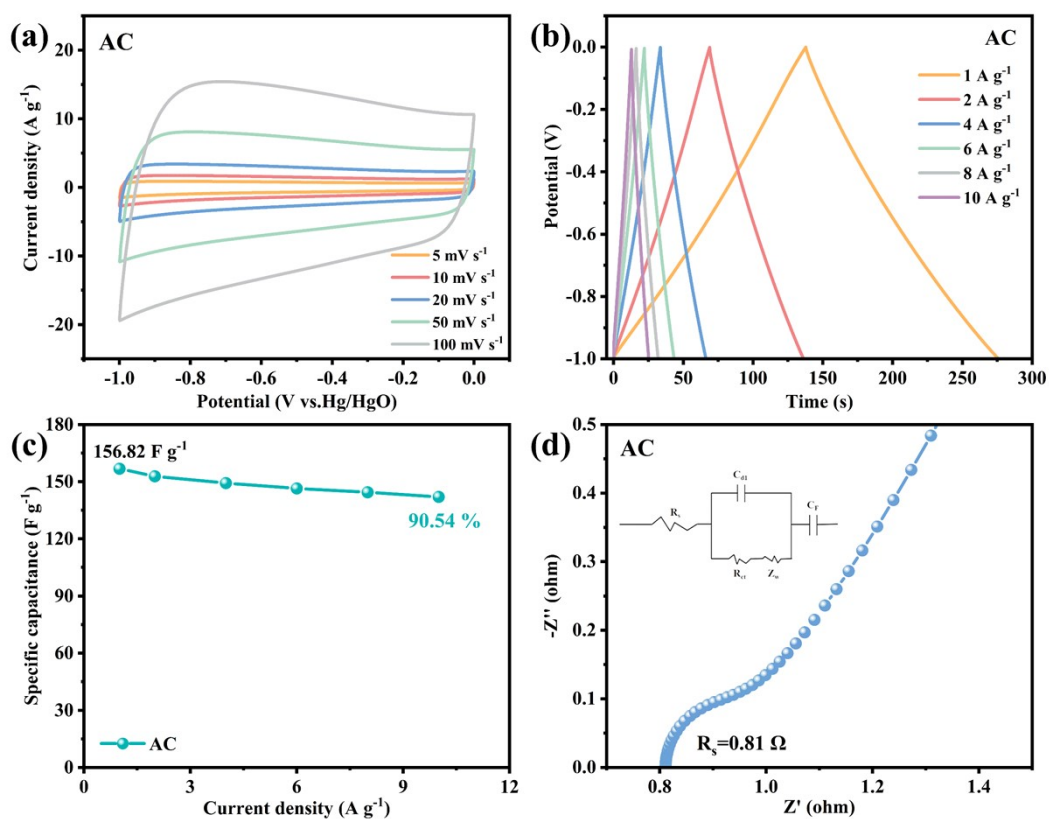


Figure S13. The following of AC materials employed in the assembly of asymmetric supercapacitors: **(a)** CV curves; **(b)** GCD curves; **(c)** specific capacitance and capacitance retention rate; **(d)** EIS curves.

Table S1. Electrochemical performance comparison of electrode materials in this work with previously published works.

Electrode materials	Specific capacitance	Rate capability	Capacitance retention	Reference
Ni ₂ P@NiCo-LDHs	1001.46 F g ⁻¹ at 1 A g ⁻¹	75.63% at 10 A g ⁻¹	92.86% after 4000 cycles	This work
NiCo MOF-2	927.1 F g ⁻¹ at 1 A g ⁻¹	69.7% at 10 A g ⁻¹	-	1
NiCoS/ACC	413.3 F g ⁻¹ at 1 A g ⁻¹	78.39% at 5 A g ⁻¹	86% after 5000 cycles	2
MXene-LDH	983.98 F g ⁻¹ at 2 A g ⁻¹	54.53% at 50 A g ⁻¹	76% after 5000 cycles	3
NiCo/NMCS-50	585 F g ⁻¹ at 1 A g ⁻¹	36.6% at 10 A g ⁻¹	-	4
CoS _x /NiCo LDH	1903.8 F g ⁻¹ at 1 A g ⁻¹	76.1% at 10 A g ⁻¹	-	5
NiCo/NHCS-23	536 F g ⁻¹ at 1 A g ⁻¹	64.7% at 10 A g ⁻¹	-	6
NiCo LDHMs/NF-100	1156 F g ⁻¹ at 1 A g ⁻¹	72% at 10 A g ⁻¹	95.37% after 5000 cycles	7
MO/NMO/NiCo LDH	815 F g ⁻¹ at 1 A g ⁻¹	23.72% at 20 A g ⁻¹	-	8
Ni-Co-S-10	640 F g ⁻¹ at 1 A g ⁻¹	-	84% after 10000 cycles	9
NiCoP/HMCS-2	856.06 F g ⁻¹ at 1 A g ⁻¹	87.07% at 20 A g ⁻¹	86.45% after 5000 cycles	10

Table S2. the Nyquist impedance value of the electrode in the three electrodes.

Electrode	Mass content (mg)	R_{ct} (Ω)
Ni ₂ P	1.7	3.78
NiCo-LDHs	1.56	0.67
Ni ₂ P@NiCo-LDHs-4	1.63	0.24
Ni ₂ P@NiCo-LDHs-6	1.61	0.21
Ni ₂ P@NiCo-LDHs-8	1.68	0.22

Table S3. Energy density for the Ni₂P@NiCo-LDHs-6//AC ASC device with other ASC devices reported preciously. The reference numbers are consistent with the main text.

Electrode materials	Power density (W kg ⁻¹)	Energy density (W h kg ⁻¹)	Ref.
Ni ₂ P/NiSe ₂ /MoSe ₂ //AC	400	23.5	67
NiP/NCHTs//AC	1125	46.53	68
NiCo-LDH@Ni(OH) ₂ //AC	799.9	30.6	69
Mo-Ni ₂ P/Co ₂ P//AC	800	35.56	70
WS ₂ /NiCo(PO ₄) ₃ /MXenes//AC	703	38	71
NiCo-LDH/Ni/CC//AC	750	49.3	72
Ni ₂ P/CNT//AC	749.8	25.2	73
Ni ₂ P@NiCo-LDHs-6//AC	766.81	51.88	This work

Table S4. the Nyquist impedance values of assembled asymmetric supercapacitors.

Asymmetric supercapacitors	R_{ct} (Ω) 5,500 cycles before	R_{ct} (Ω) 5,500 cycles after
Ni ₂ P//AC	0.47	5.84
Ni ₂ P@NiCo-LDHs-6//AC	0.25	2.14

References

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