

## Supplemental Material

# Enhanced negative electrocaloric effect in Sn-doped $\text{PbHfO}_3$ films through phase transition induction

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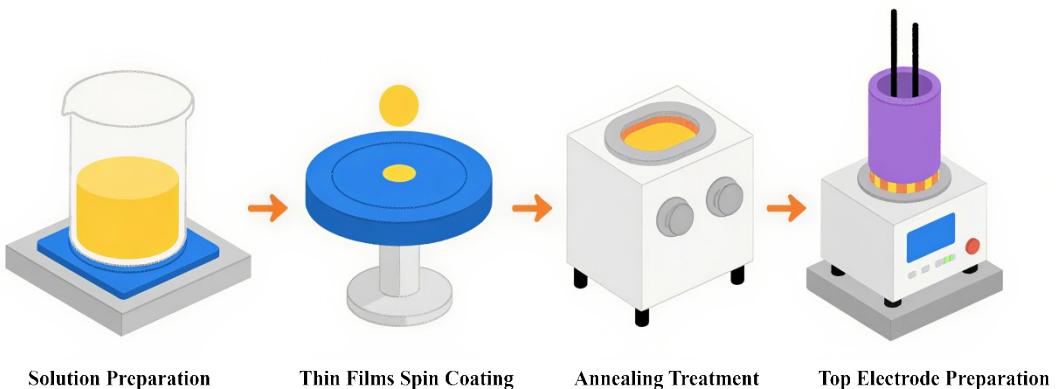


Fig. S1 Flow chart of PHS-100x films preparation by sol-gel method.

Initially, hafnium acetylacetone ( $\text{Hf}(\text{CH}_3\text{COCHCOCH}_3)_4$ , 97%) was dissolved in a mixed solution of glacial acetic acid ( $\text{CH}_3\text{COOH}$ ) and ethylene glycol monomethyl ether ( $\text{C}_3\text{H}_8\text{O}_2$ ) under heating until complete dissolution. Subsequently, lead acetate trihydrate ( $\text{Pb}(\text{CH}_3\text{COO})_2$ , 99.5%) and tin tetrachloride pentahydrate ( $\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$ , 99.0%) were mixed and dissolved in ethylene glycol methyl ether ( $\text{C}_3\text{H}_8\text{O}_2$ ). After this solution cooled, the hafnium acetylacetone ( $\text{Hf}(\text{CH}_3\text{COCHCOCH}_3)_4$ , 97%) solution was added. Finally, a small amount of acetylacetone ( $\text{C}_5\text{H}_8\text{O}_2$ ) was added to the mixed solution as a stabilizer, and it was stirred using a magnetic stirrer. A clear PHS-100x solution, free of precipitation and with a concentration of 0.2 mol/L, was obtained after 40 minutes of stirring. To preclude potential errors in the preparation of film components arising from lead loss during the solution preparation stage, it is imperative to ensure that the lead acetate trihydrate ( $\text{Pb}(\text{CH}_3\text{COO})_2$ ) addition exceeds 10% during solution configuration. Following an aging period of 3 days, the precursor solution achieved the requisite stability and uniformity, which is critical for the formation of crack-free films during the subsequent annealing process. Ultimately, the prepared PHS-100x solution was dropped onto the center of the FTO/glass. Spin-coating was performed at a low speed of 800 rpm for 10 s, followed by a high speed of 3000 rpm

for 20 s. The resulting coating underwent 10 repeated cycles, each comprising a 5 min pyrolysis at 200°C and a 15 min sintering at 400°C, followed by annealing at 650°C for 15 min. To facilitate subsequent film testing, gold electrodes were sputtered onto the films using magnetron sputtering for a duration of 110 seconds.

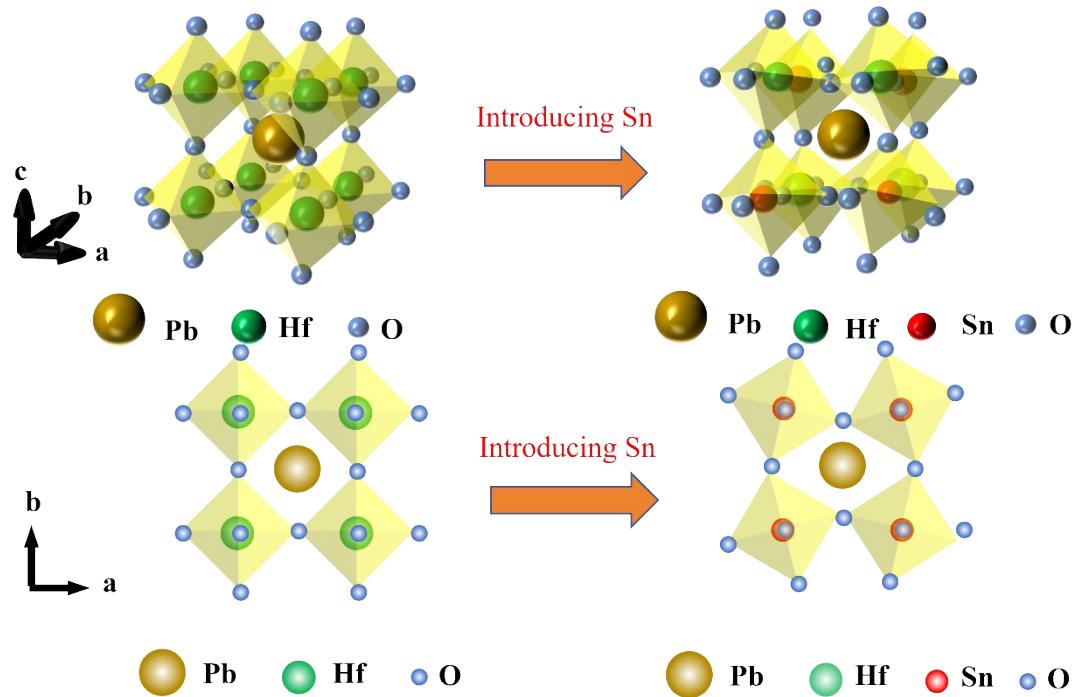


Fig.S2 Schematic of  $\text{PbHfO}_3$  Crystal Structure.

Table S1. Interplanar spacing ( $d$ ) for different  $\text{Sn}^{4+}$  doping concentrations

concentration of $\text{Sn}^{4+}$ (%)	$2\theta$ (°)	$\theta$ (°)	$\sin\theta$	$d$ (Å)
0.0	54.24	27.12	0.455823	<b>1.69150</b>
0.5	54.26	27.13	0.455876	<b>1.69142</b>
1.0	54.29	27.145	0.455982	<b>1.69126</b>
2.0	54.34	27.17	0.456088	<b>1.69084</b>

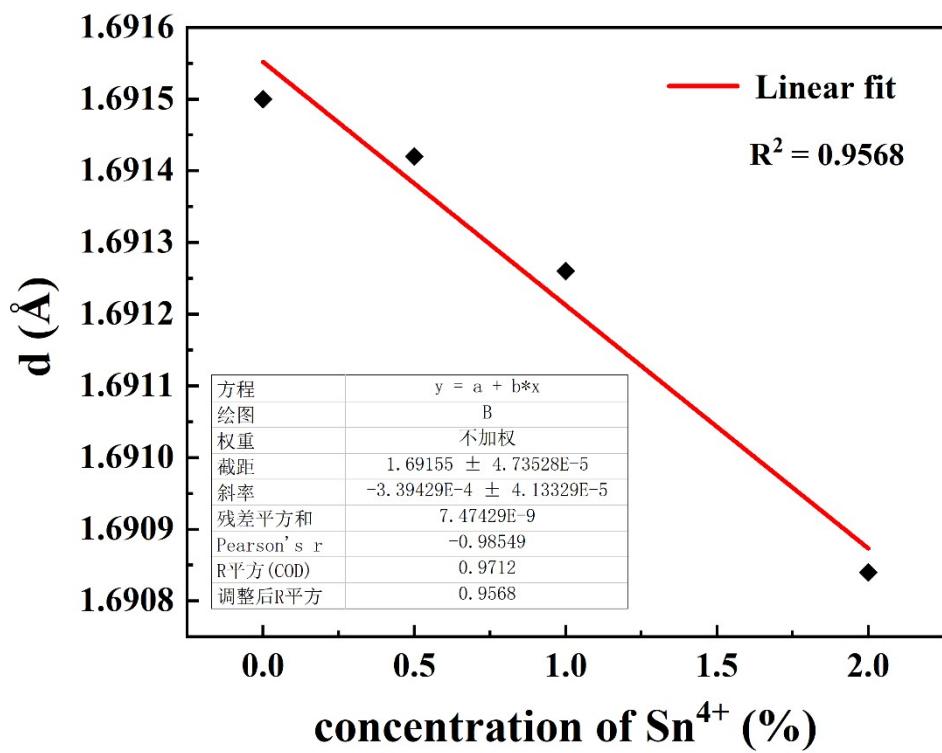


Fig. S3 Interplanar spacing ( $d$ ) versus  $\text{Sn}^{4+}$  ion concentration with linear fit.