

Supplementary Information

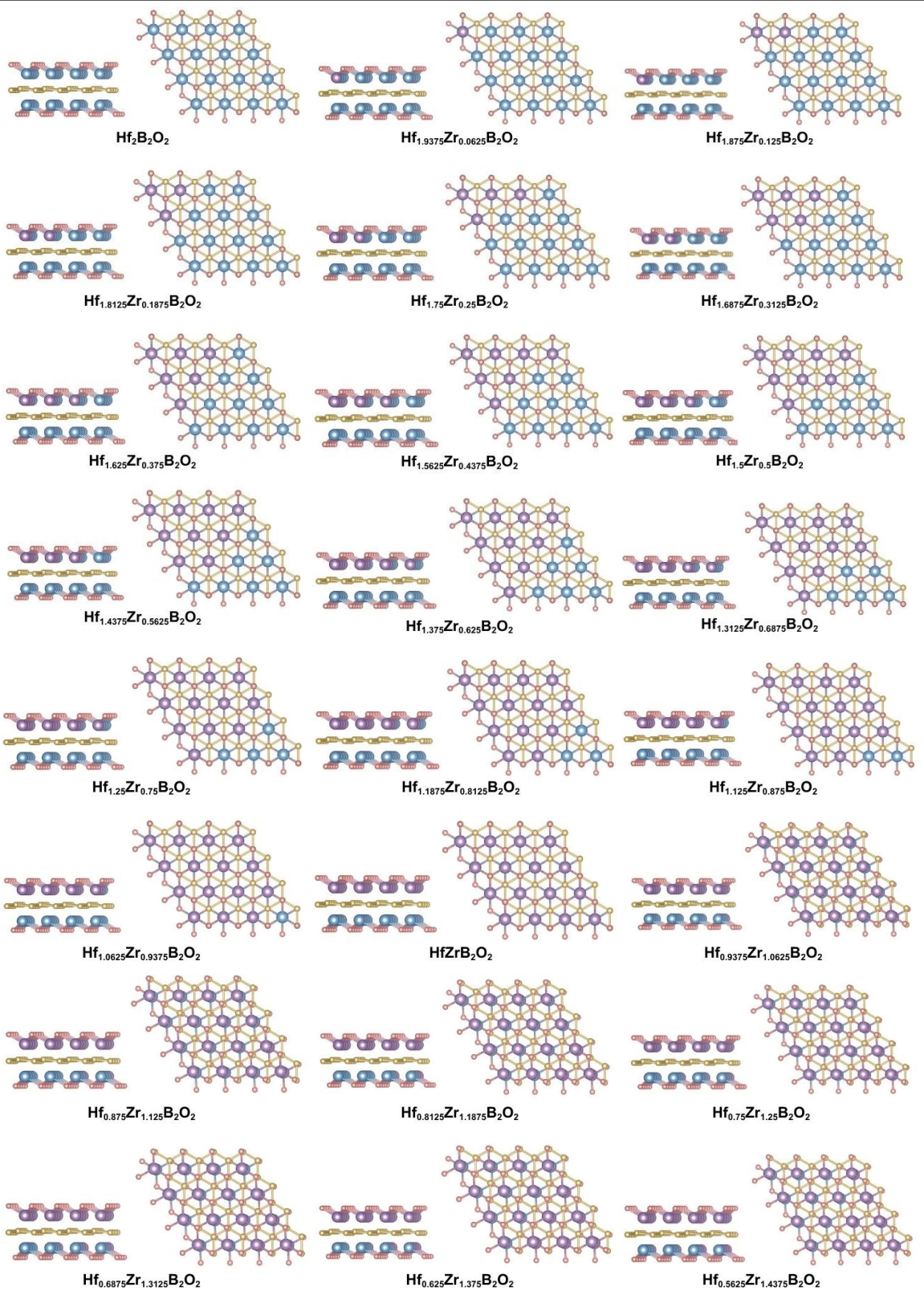
First-principles study of the Pt@($\text{Hf}_{2-x}\text{Zr}_x$) B_2O_2 bifunctional ORR/OER catalysts

Miao Xue^a, Ya Gao^a, Erpeng Wang^b, Jian Zhou^{a,*} and Zhimei Sun^{a,*}

^aSchool of Materials Science and Engineering, Beihang University, Beijing 100191, China

^bInstitute for Advanced Studies in Precision Materials, Yantai University, Yantai, 264005, China

*Address correspondence to: jzhou@buaa.edu.cn (Jian Zhou), zmsun@buaa.edu.cn (Zhimei Sun).



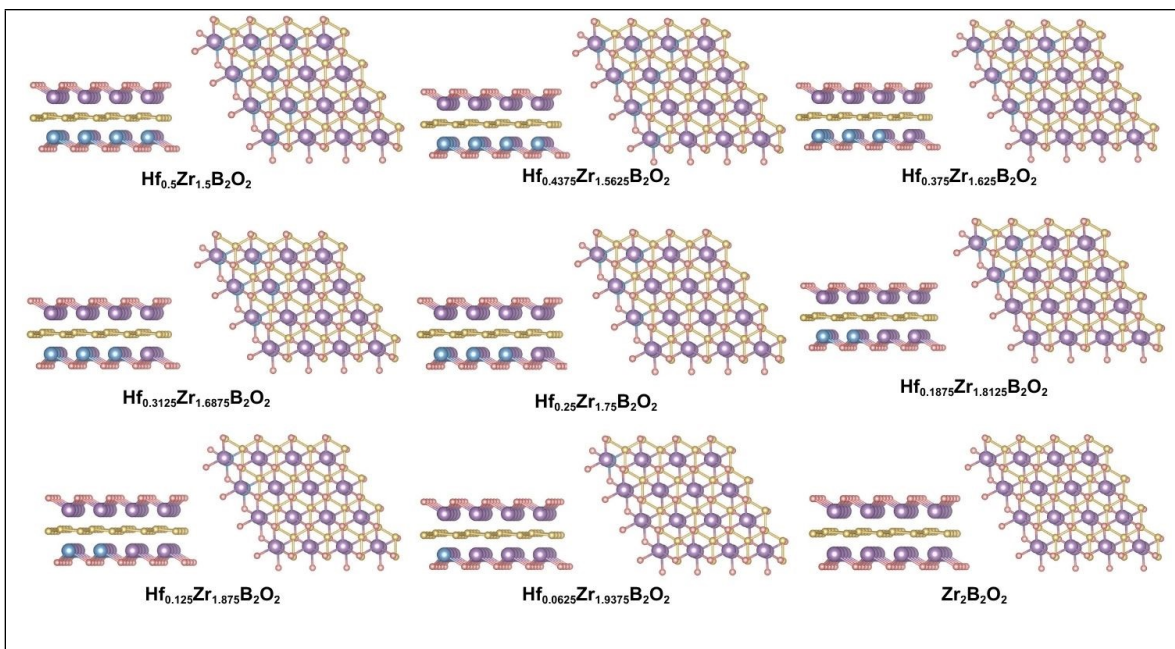
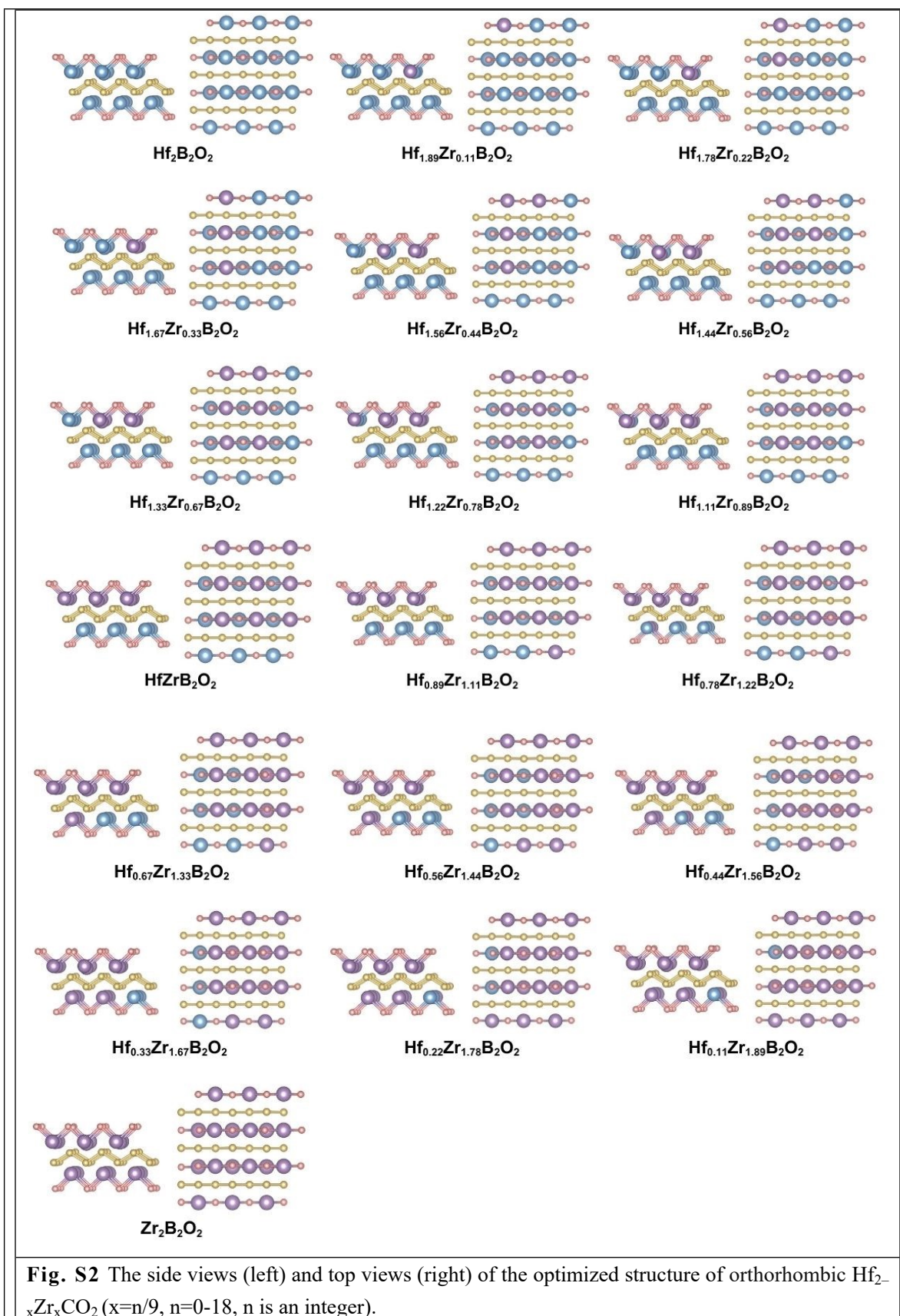


Fig. S1 The side views (left) and top views (right) of the optimized structure of hexagonal $\text{Hf}_{2-x}\text{Zr}_x\text{CO}_2$ ($x=n/16$, $n=0-32$, n is an integer).



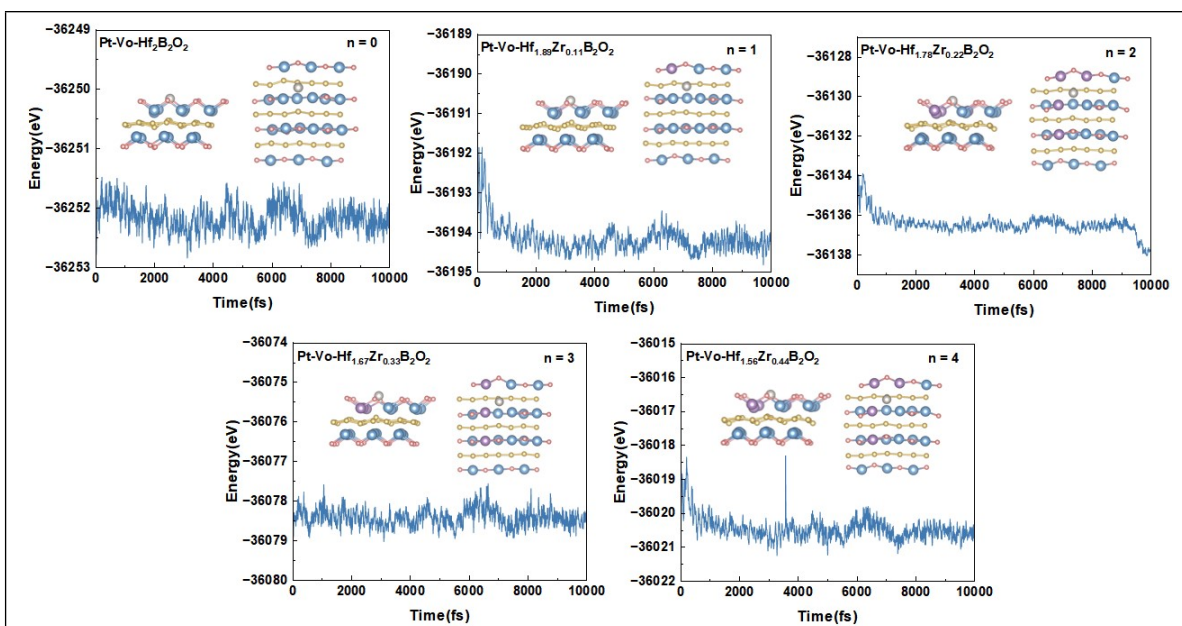


Fig. S3 Energy fluctuations of the orthorhombic Pt-V_O-Hf_{2-x}Zr_xB₂O₂ in 300 K AIMD simulations (10 ps).

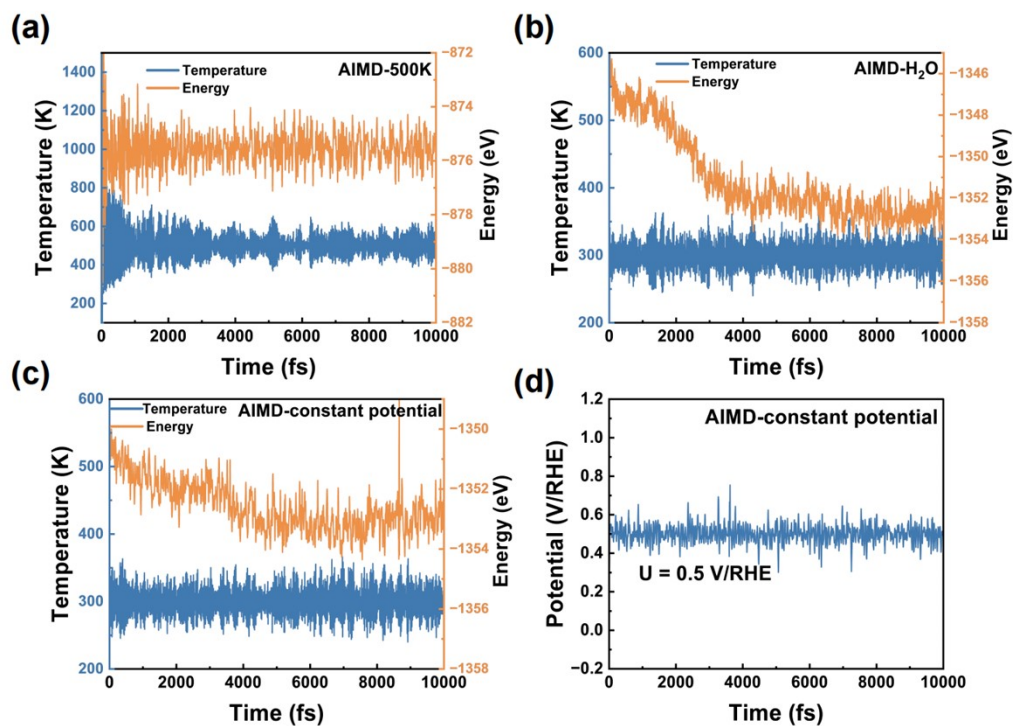


Fig. S4 AIMD simulation results (10 ps) of $\text{Pt@Hf}_{1.375}\text{Zr}_{0.625}\text{B}_2\text{O}_2$ under various conditions: (a) at 500 K, (b) in explicit aqueous solution, and (c) under constant potential. (d) The applied potential during the constant-potential AIMD simulations.

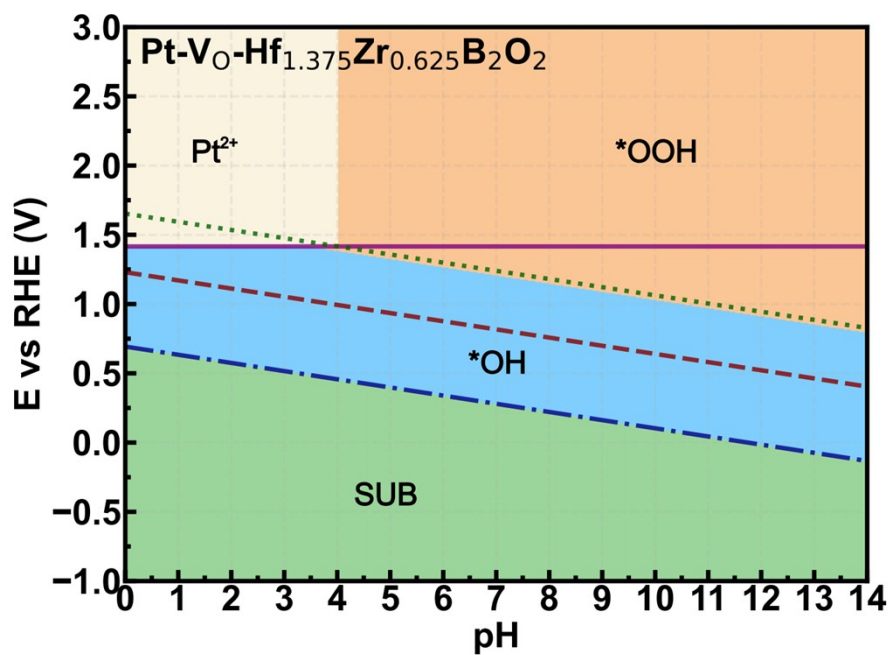


Fig. S5 Pourbaix diagram for $\text{Pt@Hf}_{1.375}\text{Zr}_{0.625}\text{B}_2\text{O}_2$, the blue, green and red dashed line represent the working potentials of ORR, OER, and standard equilibrium potential (1.23 V) at varying pH values. SUB denotes the clean substrate surface, *O and *OH denotes *O and *OH adsorbed surface.

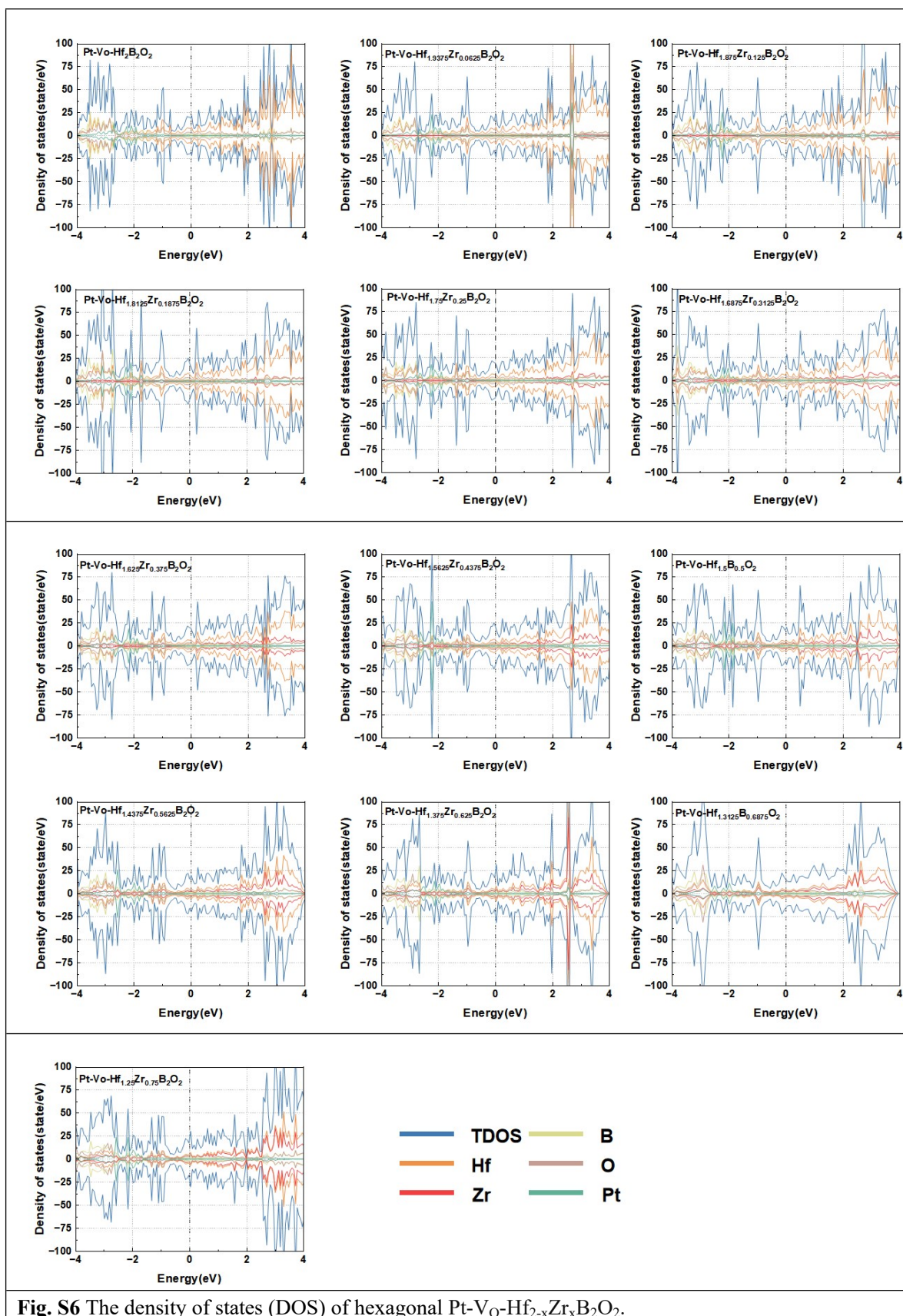


Fig. S6 The density of states (DOS) of hexagonal $\text{Pt-Vo-Hf}_{2-x}\text{Zr}_x\text{B}_2\text{O}_2$.

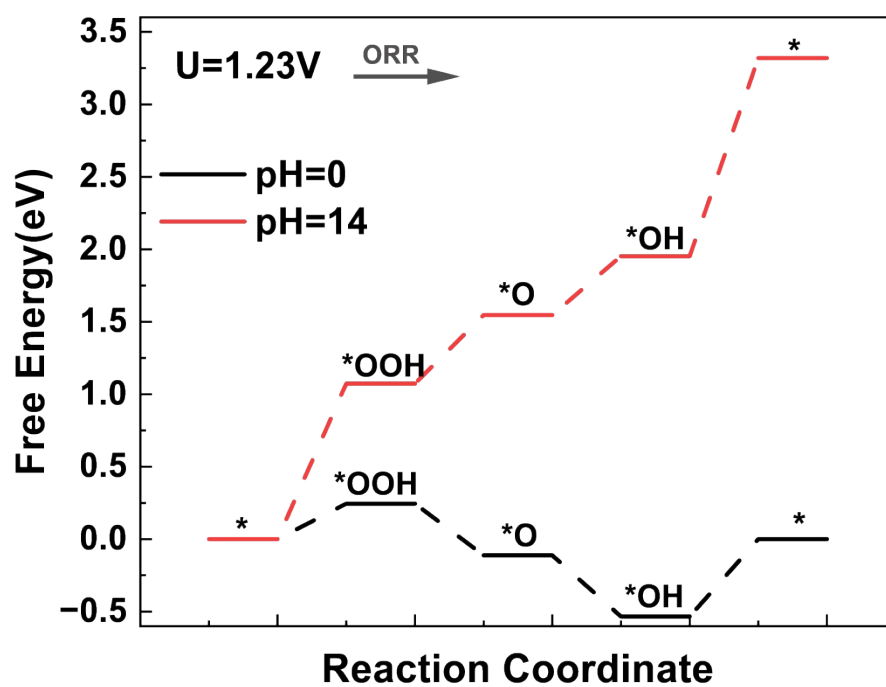


Fig. S7 ORR/OER Gibbs free energy diagram for Pt@Hf_{1.375}Zr_{0.625}B₂O₂ under conditions of pH = 0 and pH = 14, U is 1.23 V in both cases.

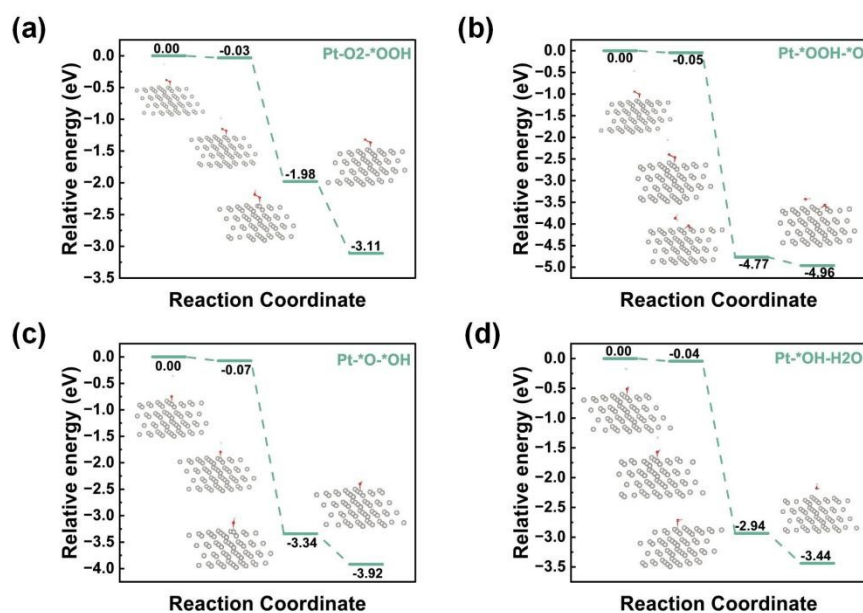


Fig. S8 The four-step NEB process of ORR on the Pt (111) catalyst: (a) $O_2 \rightarrow *OOH$, (b) $*OOH \rightarrow *O$, (c) $*O \rightarrow *OH$, (d) $*OH \rightarrow H_2O$.

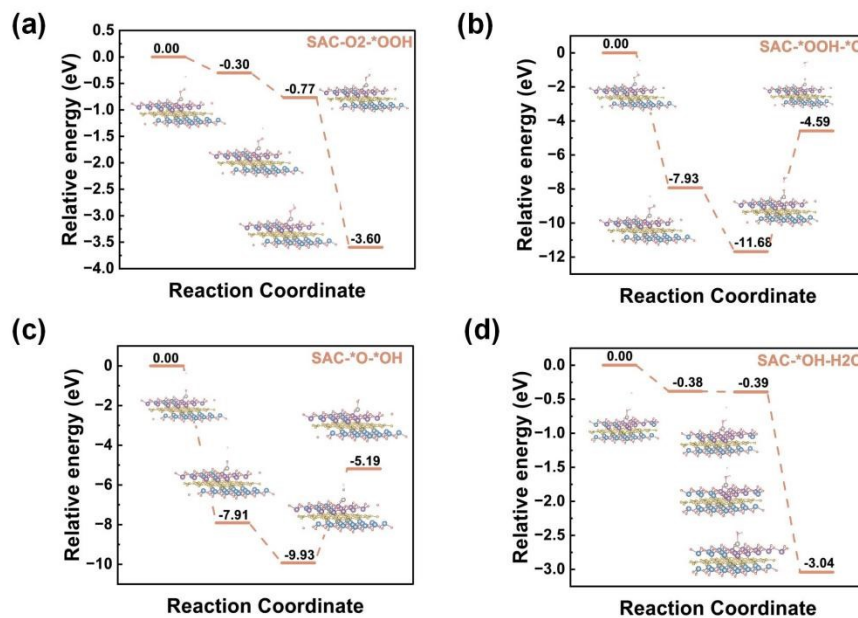


Fig. S9 The four-step NEB process of Pt@Hf_{1.375}Zr_{0.625}B₂O₂ on the Pt (111) catalyst: (a) $O_2 \rightarrow ^*OOH$, (b) $^*OOH \rightarrow ^*O$, (c) $^*O \rightarrow ^*OH$, (d) $^*OH \rightarrow H_2O$.

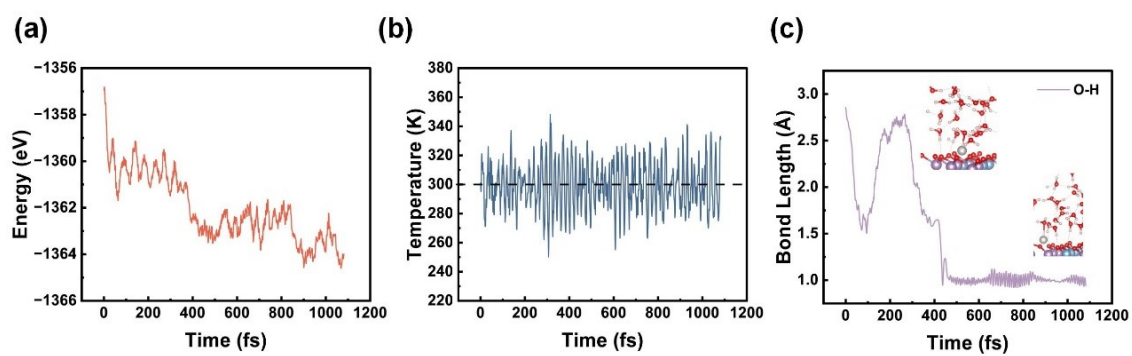


Fig. S10 (a) and (b) AIMD simulations on $\text{Pt@Hf}_{1.375}\text{Zr}_{0.625}\text{B}_2\text{O}_2$ for $^*\text{OH} \rightarrow ^*\text{H}_2\text{O}$ hydrogenation process at 300 K, (c) the dynamic evolution of the O-H bond length between $^*\text{OH}$ and H_2O .

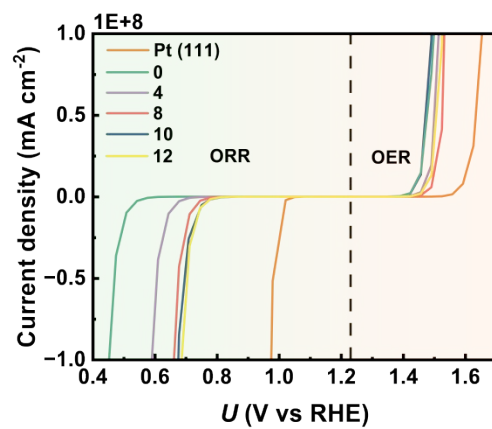


Fig. S11 Theoretical polarization curves for Pt-V_O-Hf_{2-x}Zr_xB₂O₂ (x = n/16, n = 0, 4, 8, 10, 12) and Pt (111).

Table S1 The cohesive energies (E_{coh}) comparison for preferential Zr substitution sites in hexagonal and orthorhombic structures (*position*1, 2 can be seen in Figure 1 (a,b), *hex* represents the hexagonal structure and *orth* represents the orthorhombic structure).

<i>position</i>	<i>hex</i> (eV)	<i>orth</i> (eV)
1	7.44112	7.06911
2	7.44102	7.06917

Table S2 The lattice constant of a_1 and a_2 of the $4\times 4\times 1$ cell for hexagonal and orthorhombic $\text{Hf}_{2-x}\text{Zr}_x\text{B}_2\text{O}_2$, respectively, and the vertical distance between the Pt atom and the nearest Hf/Zr layer on the surface of hexagonal $\text{Pt-V}_\text{O}\text{-Hf}_{2-x}\text{Zr}_x\text{B}_2\text{O}_2$ SACs (hexagonal: $\text{hex-}x=n/16$, $n=0\text{-}32$; orthorhombic: $\text{orth-}x=n/9$, $n=0\text{-}18$; where n is an integer).

n	$\text{hex-}x$	$\text{orth-}x$	a_1 (Å)	a_2 (Å)	distance (Å)
0	0	0	12.648	9.566	1.83
1	0.0625	0.11	12.673	9.570	1.83
2	0.125	0.22	12.677	9.573	1.82
3	0.1875	0.33	12.682	9.578	1.82
4	0.25	0.44	12.686	9.586	1.81
5	0.3125	0.55	12.690	9.583	1.82
6	0.375	0.67	12.695	9.595	1.82
7	0.4375	0.78	12.700	9.590	1.82
8	0.5	0.89	12.704	9.591	1.83
9	0.5625	1	12.709	9.597	1.84
10	0.625	1.11	12.714	9.598	1.84
11	0.6875	1.22	12.717	9.602	1.84
12	0.75	1.33	12.720	9.614	1.84
13	0.8125	1.44	12.723	9.618	1.84
14	0.875	1.56	12.728	9.616	1.85
15	0.9375	1.67	12.735	9.620	1.85
16	1	1.78	12.741	9.622	1.85
17	1.0625	1.89	12.745	9.626	1.84
18	1.125	2	12.748	9.630	1.84
19	1.1875	/	12.752	/	1.84
20	1.25	/	12.756	/	1.84
21	1.3125	/	12.760	/	1.84
22	1.375	/	12.764	/	1.84

23	1.4375	/	12.769	/	1.84
24	1.5	/	12.772	/	1.84
25	1.5625	/	12.776	/	1.84
26	1.625	/	12.778	/	1.83
27	1.6875	/	12.782	/	1.83
28	1.75	/	12.783	/	1.84
29	1.8125	/	12.788	/	1.84
30	1.875	/	12.791	/	1.84
31	1.9375	/	12.798	/	1.84
32	2	/	12.802	/	1.84

Table S3 The binding energies ($hex-E_b$, $orth-E_b$) and the formation energies ($hex-E_{f-Pt}$, $orth-E_{f-Pt}$) of the single Pt atom embedded in the oxygen vacancies of hexagonal and orthorhombic $Hf_{2-x}Zr_xB_2O_2$.

n	$hex-x$	$orth-x$	$hex-E_b$ (eV)	$orth-E_b$ (eV)	$hex-E_{f-Pt}$ (eV)	$orth-E_{f-Pt}$ (eV)
0	0	0	-6.482	-5.927	6.058	-0.386
1	0.0625	0.11	-6.571	-5.912	6.056	-0.370
2	0.125	0.22	-6.560	-5.780	6.057	-0.238
3	0.1875	0.33	-6.554	-5.788	6.057	-0.246
4	0.25	0.44	-6.537	-5.803	6.053	-0.261
5	0.3125	0.55	-6.468	-5.673	5.955	-0.132
6	0.375	0.67	-6.457	-5.686	5.953	-0.145
7	0.4375	0.78	-6.441	-5.693	5.945	-0.152
8	0.5	0.89	-6.359	-5.631	5.836	-0.089
9	0.5625	1	-6.278	-5.640	5.724	-0.098
10	0.625	1.11	-6.263	-5.658	5.719	-0.117
11	0.6875	1.22	-6.257	-5.663	5.717	-0.121
12	0.75	1.33	-6.245	1.398	5.710	6.939
13	0.8125	1.44	-6.235	-5.661	5.707	-0.119
14	0.875	1.56	-6.220	-5.656	5.690	-0.114
15	0.9375	1.67	-6.203	-5.650	5.675	-0.108
16	1	1.78	-6.180	-4.086	5.661	1.455
17	1.0625	1.89	-6.176	-4.109	5.660	1.432
18	1.125	2	-6.173	-4.091	5.660	1.450
19	1.1875	/	-6.174	/	5.660	/
20	1.25	/	-6.185	/	5.660	/
21	1.3125	/	-6.194	/	5.682	/
22	1.375	/	-6.203	/	5.680	/
23	1.4375	/	-6.204	/	5.678	/

24	1.5	/	-6.196	/	5.655	/
25	1.5625	/	-6.188	/	5.631	/
26	1.625	/	-6.193	/	5.635	/
27	1.6875	/	-6.206	/	5.654	/
28	1.75	/	-6.199	/	5.625	/
29	1.8125	/	-6.202	/	5.643	/
30	1.875	/	-6.207	/	5.641	/
31	1.9375	/	-6.209	/	5.639	/
32	2	/	-6.230	/	5.646	/

Table S4 The dissolution potentials of hexagonal ($hex-U_{diss}$ (V)) and orthorhombic ($orth-U_{diss}$ (V)) Pt-V_O-Hf_{2-x}Zr_xB₂O₂ SACs.

n	$hex-x$	$orth-x$	$hex-U_{diss}$ (V)	$orth-U_{diss}$ (V)
0	0	0	1.650	1.373
1	0.0625	0.11	1.695	1.365
2	0.125	0.22	1.689	1.299
3	0.1875	0.33	1.686	1.303
4	0.25	0.44	1.678	1.311
5	0.3125	0.55	1.643	1.246
6	0.375	0.67	1.638	1.252
7	0.4375	0.78	1.630	1.256
8	0.5	0.89	1.589	1.225
9	0.5625	1	1.549	1.229
10	0.625	1.11	1.541	1.238
11	0.6875	1.22	1.538	1.241
12	0.75	1.33	1.532	-2.290
13	0.8125	1.44	1.527	1.240
14	0.875	1.56	1.520	1.237
15	0.9375	1.67	1.511	1.234
16	1	1.78	1.500	0.452
17	1.0625	1.89	1.497	0.464
18	1.125	2	1.496	0.455
19	1.1875	/	1.497	/
20	1.25	/	1.502	/
21	1.3125	/	1.506	/
22	1.375	/	1.511	/
23	1.4375	/	1.511	/

24	1.5	/	1.507	/
25	1.5625	/	1.503	/
26	1.625	/	1.506	/
27	1.6875	/	1.512	/
28	1.75	/	1.509	/
29	1.8125	/	1.510	/
30	1.875	/	1.513	/
31	1.9375	/	1.514	/
32	2	/	1.524	/

Table S5 The overpotentials of Pt-V_O-Hf_{2-x}Zr_xB₂O₂ SACs toward ORR, OER.

<i>n</i>	pH = 0		
	η^{ORR} (V)	η^{OER} (V)	η^{Bi} (V)
0	0.777	0.412	1.189
1	0.672	0.494	1.166
2	0.653	0.498	1.150
3	0.651	0.514	1.165
4	0.646	0.511	1.157
5	0.610	0.480	1.090
6	0.598	0.471	1.069
7	0.609	0.502	1.111
8	0.578	0.469	1.047
9	0.555	0.443	0.998
10	0.537	0.423	0.960
11	0.521	0.446	0.967
12	0.529	0.477	1.005

Table S6 Comparison of ORR/OER energetics calculated using PBE+D3 and SCAN for Pt@Hf_{1.375}Zr_{0.625}B₂O₂.

<i>Reaction</i>	<i>Elementary step</i>	ΔG (PBE+D3) (eV)	ΔG (SCAN) (eV)
<i>ORR</i>	$* \rightarrow *OOH$	-0.99	-1.21
	$*OOH \rightarrow *O$	-1.59	-1.59
	$*O \rightarrow *OH$	-1.65	-1.69
	$*OH \rightarrow *$	-0.69	-0.43
<i>OER</i>	$* \rightarrow *OH$	0.69	0.43
	$*OH \rightarrow *O$	1.65	1.69
	$*O \rightarrow *OOH$	1.59	1.59
	$*OOH \rightarrow *$	0.99	1.21

Table S7 The charge transfer of Pt, O and H of Pt-V_O-Hf_{2-x}Zr_xB₂O₂ before and after intermediate adsorption.

Catalysts	Charge transfer (e)											
	sup		*OOH				*O		*OH			
	Pt	O ₁	O ₂	H	OOH	Pt	O	Pt	O	H	OH	Pt
0	-0.13	0.47	0.34	-0.33	0.48	-0.04	0.71	-0.09	0.70	-0.20	0.50	-0.17
1	-0.33	0.45	0.42	-0.35	0.52	-0.29	0.79	-0.48	0.72	-0.22	0.50	0.01
2	-0.27	0.49	0.07	-0.06	0.50	-0.18	0.81	-0.30	0.85	-0.30	0.55	-0.01
3	-0.24	0.47	0.23	-0.22	0.48	-0.16	0.66	-0.17	0.69	-0.19	0.50	-0.14
4	-0.22	0.42	0.37	-0.31	0.48	-0.08	0.63	-0.14	0.66	-0.18	0.48	-0.11
5	-0.38	0.45	0.25	-0.27	0.43	0.06	0.65	0.01	0.60	-0.11	0.49	0.05
6	-0.33	0.41	0.42	-0.35	0.48	0.01	0.67	-0.02	0.62	-0.14	0.48	0.02
7	-0.32	0.47	0.23	-0.20	0.50	-0.11	0.67	-0.03	0.62	-0.13	0.49	-0.002
8	-0.25	0.51	0.33	-0.31	0.53	-0.08	0.81	-0.22	0.67	-0.23	0.44	0.01
9	-0.35	0.50	0.41	-0.43	0.48	-0.11	0.72	-0.17	0.68	-0.24	0.44	-0.02
10	-0.34	0.55	-0.33	0.35	0.57	-0.30	0.78	-0.21	0.78	-0.29	0.49	-0.07
11	-0.27	0.56	-0.30	0.34	0.60	-0.32	0.79	-0.19	0.70	-0.24	0.46	0.06
12	-0.08	0.46	0.27	-0.25	0.48	-0.12	0.74	-0.14	0.71	-0.26	0.45	0.14

More calculation details

The dissolution potential (U_{diss}) can be obtained by the following formula:

$$U_{diss} = U_{diss-bulk} - E_{cluster} / n_e$$

where $U_{diss-bulk}$ is the standard electrode potential, as found in the handbook of chemistry and physics. (Acree, W. E. NIST Chemistry Webbook, NIST Standard Reference Database No. 69; National Institute of Standards and Technology: Gaithersburg, MD, 2018; <http://webbook.nist.gov/chemistry> (accessed May 10, 2021)). n_e is the number of transferred electrons in the dissolution reaction.

The adsorption energy (E_{ads}) of intermediates can be calculated by the following formulas:

$$E_{ads}(O) = E_{*O} - E_{*} - (E_{H_2O} - E_{H_2})$$

$$E_{ads}(OH) = E_{*OH} - E_{*} - (E_{H_2O} - \frac{1}{2}E_{H_2})$$

$$E_{ads}(OOH) = E_{*OOH} - E_{*} - (2E_{H_2O} - \frac{3}{2}E_{H_2})$$

$$E_{ads}(H) = E_{*H} - E_{*} - \frac{1}{2}E_{H_2}$$

$$E_{ads}(O_2) = E_{*O_2} - E_{*} - E_{O_2}$$

The E_{*O} , E_{*OH} , E_{*OOH} , E_{*H} , E_{*O_2} , and E_{*} are the total energies of catalysts with O, OH, OOH, H,

O₂ and without adsorbents. The E_{H_2O} , E_{H_2} , and E_{O_2} are total energies of free H₂O, H₂ and O₂ molecules in liquid and gas phases. Furthermore, we use 1/2 H₂ to represent the free energy of (H⁺ + e⁻).

The input files for VASP calculation:

INCAR

Global Parameters

LREAL = Auto
ENCUT = 600
PREC = Accurate
ADDGRID = .TRUE.
ISMear = 0
SIGMA = 0.05
NELM = 120
NELMIN = 4
EDIFF = 1E-05
EDIFFG = -2E-02
IVDW = 11
ISYM = 0

Structural optimization

ALGO = Fast
LWAVE = .FALSE.
LCHARG = .FALSE.
NSW = 100
IBRION = 2

The input files for CP2K calculation: cp2k.inp

&GLOBAL

PROJECT cp2k

PRINT_LEVEL LOW

RUN_TYPE MD

&END GLOBAL

&FORCE_EVAL

METHOD Quickstep

&SUBSYS

&CELL

A	12.71624900	0.00000000	0.00000000
B	-6.35885466	11.01225738	0.00000000
C	-0.00895307	-0.00165284	20.62451899

PERIODIC XYZ #Direction(s) of applied PBC (geometry aspect)

&END CELL

&COORD

Hf	-0.00024805	0.00294227	8.49474438
Hf	3.17543590	0.00366986	8.48490649
Hf	-3.18193491	11.00642156	12.08386443
Hf	-1.58902836	2.75477416	8.48325653
Hf	1.58769285	2.76121266	8.52908421
Hf	6.35522941	0.00225912	8.49959115
Hf	6.35160049	0.00111286	12.08456566
Hf	9.53199241	0.00170894	8.49416690
Hf	9.53654550	-0.00087027	12.09607414
Hf	4.76251886	2.75513819	8.47541921
Hf	4.76762204	2.75220485	12.09908532
Hf	7.94296961	2.75258142	8.49950865
Hf	7.94214429	2.75607132	12.08452441
Hf	-3.18069409	5.50796946	8.49816805
Hf	0.00390784	5.50242820	8.52425807
Hf	-4.77183835	8.25942481	8.51310021
Hf	-1.59339258	8.25753190	8.49814743
Hf	3.16975583	5.50159087	8.52916671
Hf	6.35154559	5.50544871	8.48492711
Hf	6.36055665	5.50900358	12.08376130
Hf	1.58673169	8.25579316	8.48325653
Hf	4.76434564	8.25602350	8.49476501
Zr	-6.35718131	11.00521038	12.08180198
Zr	-1.58982964	2.75210435	12.11636867
Zr	1.58729069	2.71653500	12.11335749
Zr	-3.18691332	5.51392572	12.08792746
Zr	-0.03500416	5.52484695	12.12453598

Zr	-4.77154680	8.25911622	12.08902056
Zr	-1.60186885	8.25977701	12.08825745
Zr	3.20817187	5.52426418	12.11354311
Zr	1.58863861	8.25774849	12.11651304
Zr	4.77401050	8.25931504	12.08149261
B	1.59176687	0.91582832	10.38168163
B	-0.00593658	1.83469561	10.18766678
B	4.76900530	0.91733720	10.37922731
B	3.18115527	1.83520322	10.17457021
B	-0.00594761	3.66661150	10.40195553
B	-1.59528851	4.58534812	10.18958486
B	3.18429417	3.66634756	10.39752126
B	1.58815991	4.58781003	10.24995282
B	7.94345666	0.91880161	10.38188787
B	6.35539439	1.83541295	10.16836223
B	11.11855835	0.91374703	10.38139288
B	9.53177748	1.83511518	10.17391022
B	6.35722862	3.66840832	10.37924793
B	4.76848859	4.58466656	10.17452896
B	9.52713458	3.66737305	10.38077415
B	7.94314609	4.58692437	10.17087842
B	-1.59398136	6.42506233	10.38333159
B	-3.18128693	7.34107785	10.18434623
B	1.58890652	6.42899222	10.40189365
B	-0.00161691	7.34596687	10.18960548
B	-3.18162785	9.18267494	10.38083602
B	-4.77189388	10.09736942	10.17644704
B	-0.00369755	9.18498751	10.38093914
B	-1.59228874	10.09669811	10.17091967
B	4.76979653	6.42091085	10.38161975
B	3.17514458	7.34504199	10.18772865
B	7.94003420	6.42049247	10.38054728
B	6.35280698	7.34051687	10.17622017
B	3.17669336	9.18402939	10.38151663
B	1.58521442	10.09710532	10.17395147
B	6.35367353	9.17803928	10.37464867
B	4.76475944	10.09507896	10.17529206
O	-1.63849694	4.57263540	13.13231185
O	-0.03368107	1.80907655	13.13008440
O	3.25282559	1.79013848	13.09256840
O	-0.00534804	3.67217426	7.49670329
O	1.58478935	0.92798666	7.48591666
O	3.17364742	3.67251623	7.48940221
O	4.76492651	0.91916581	7.48637040

O	4.84331212	4.54504186	13.09265090
O	7.90644789	4.55357669	13.08858787
O	6.36161062	1.83198658	13.07355260
O	9.47866761	1.81486098	13.09271277
O	6.35380466	3.67139322	7.48634978
O	7.94709408	0.91669950	7.48086365
O	9.53960473	3.67079822	7.49057780
O	11.12323229	0.92071798	7.49330024
O	-4.72991187	10.11353198	13.10488124
O	-1.58168156	10.14546884	13.08856725
O	-3.18623911	7.34403566	13.12602137
O	-0.01226914	7.38971399	13.13235310
O	-3.18426124	9.17474579	7.49926073
O	-1.58175709	6.41781688	7.48995907
O	-0.00034381	9.17259909	7.49053655
O	1.58437563	6.42575618	7.49672391
O	1.57624397	10.15339733	13.09269215
O	4.72220360	10.11967698	13.10296316
O	3.18360549	7.38200560	13.13008440
O	6.35974700	7.29584773	13.10473687
O	3.17302008	9.17643113	7.49336211
O	4.75587032	6.42083456	7.48604041
O	6.35549554	9.17683839	7.49583706
O	7.94567162	6.42686821	7.49934322
Pt	1.59888751	4.58168659	13.96779049

&END COORD

&VELOCITY #You can set initial atomic velocities in this section

&END VELOCITY

&KIND Hf

ELEMENT Hf

BASIS_SET DZVP-MOLOPT-SR-GTH-q12

POTENTIAL GTH-PBE

&END KIND

&KIND Zr

ELEMENT Zr

BASIS_SET DZVP-MOLOPT-SR-GTH-q12

POTENTIAL GTH-PBE

&END KIND

&KIND B

ELEMENT B

BASIS_SET DZVP-MOLOPT-SR-GTH-q3

POTENTIAL GTH-PBE

&END KIND

&KIND O

```

ELEMENT O
BASIS_SET DZVP-MOLOPT-SR-GTH-q6
POTENTIAL GTH-PBE
&END KIND
&KIND Pt
ELEMENT Pt
BASIS_SET DZVP-MOLOPT-SR-GTH-q18
POTENTIAL GTH-PBE
&END KIND
&END SUBSYS

&DFT
BASIS_SET_FILE_NAME BASIS_MOLOPT
POTENTIAL_FILE_NAME POTENTIAL
# WFN_RESTART_FILE_NAME cp2k10-RESTART.wfn
CHARGE 0 #Net charge
MULTIPLICITY 1 #Spin multiplicity
&QS
EPS_DEFAULT 1.0E-10 #Set all EPS_xxx to values such that the energy will be correct
up to this value
EXTRAPOLATION ASPC #Extrapolation for wavefunction during e.g. MD. ASPC is
default, PS can also be used
EXTRAPOLATION_ORDER 3 #Order for PS or ASPC extrapolation. 3 is default
&END QS
&POISSON
PERIODIC XYZ #Direction(s) of PBC for calculating electrostatics
PSOLVER PERIODIC #The way to solve Poisson equation
&END POISSON
&XC
&XC_FUNCTIONAL PBE
&END XC_FUNCTIONAL
&VDW_POTENTIAL
POTENTIAL_TYPE PAIR_POTENTIAL
&PAIR_POTENTIAL
PARAMETER_FILE_NAME dftd3.dat
TYPE DFTD3
REFERENCE_FUNCTIONAL PBE
#CALCULATE_C9_TERM T #Calculate C9-related three-body term, more accurate
for large system
&END PAIR_POTENTIAL
&END VDW_POTENTIAL
&END XC
&MGRID
CUTOFF 500

```

```

REL_CUTOFF 55
&END MGRID
&SCF
MAX_SCF 128 #Maximum number of steps of inner SCF
EPS_SCF 1.0E-05 #Convergence threshold of density matrix of inner SCF
# SCF_GUESS RESTART #Use wavefunction from WFN_RESTART_FILE_NAME file as
initial guess
# IGNORE_CONVERGENCE_FAILURE #Continue calculation even if SCF not converged,
works for version >= 2024.1
&OT
PRECONDITIONER FULL_ALL #Usually best but expensive for large system.
Cheaper: FULL_SINGLE_INVERSE and FULL_KINETIC
MINIMIZER DIIS #CG is worth to consider in difficult cases
LINESEARCH 2PNT #1D line search algorithm for CG. 2PNT is default. 3PNT is more
expensive but may be better. GOLD is best but very expensive
ALGORITHM STRICT #Algorithm of OT. Can be STRICT (default) or IRAC
&END OT
&OUTER_SCF
MAX_SCF 20 #Maximum number of steps of outer SCF
EPS_SCF 1.0E-05 #Convergence threshold of outer SCF
&END OUTER_SCF
&PRINT
&RESTART OFF #Do not generate wfn file to suppress meaningless I/O cost
&END RESTART
&END PRINT
&END SCF
&END DFT
&END FORCE_EVAL

&MOTION
&MD
ENSEMBLE NVT
STEPS 5000 #Number of steps to run
TIMESTEP 2.0 #Step size in fs. Decrease it properly for high temperature simulation
TEMPERATURE 500 #Initial and maintained temperature (K)
# COMVEL_TOL 0 #Uncomment this can remove translation motion of center-of-mass every
step
&THERMOSTAT
TYPE CSV
&CSV
TIMECON 200 #Time constant in fs. Smaller/larger results in stronger/weaker
temperature coupling
&END CSV
&END THERMOSTAT

```

```

&PRINT
  &PROGRAM_RUN_INFO
    &EACH
      MD      1 #Output frequency of MD information, 0 means never
    &END EACH
  &END PROGRAM_RUN_INFO
&END PRINT
&END MD
&PRINT
  &TRAJECTORY
    &EACH
      MD      1 #Output frequency of coordinates, 0 means never
    &END EACH
    FORMAT xyz
  &END TRAJECTORY
  &VELOCITIES
    &EACH
      MD      0 #Output frequency of velocities, 0 means never
    &END EACH
  &END VELOCITIES
  &FORCES
    &EACH
      MD      0 #Output frequency of forces, 0 means never
    &END EACH
  &END FORCES
  &RESTART
    BACKUP_COPIES 0 #Maximum number of backing up restart file, 0 means never
    &EACH
      MD      1 #Frequency of updating last restart file, 0 means never
    &END EACH
  &END RESTART
  &RESTART_HISTORY OFF
  &END RESTART_HISTORY
&END PRINT
&END MOTION

```

The optimized atomic coordinates of Pt-V_O-Hf_{1.375}Zr_{0.625}B₂O₂ (4×4×1 supercell):

CONTCAR

Hf22 Zr10 B32 Pt1 O31

1.0000000000000000

11.0122965673159356	-6.3586384391362651	-0.0027536341925885
-0.0002072600427606	12.7163221602735383	0.0009384463758615
-0.0042997458502392	0.0015241554405765	20.6245199825259853

Hf Zr B O Pt

22 10 32 31 1

Selective dynamics

Direct

0.0004354015824362	0.0003285495601304	0.4118759496709004	T	T	F
0.2502021221613120	0.0003945433177464	0.4113994655763520	T	T	F
0.2500225919009651	0.9995576309210676	0.5858983621382308	T	T	T
0.0004522249134595	0.2502167980335344	0.4113187377450629	T	T	F
0.2505619340498200	0.2508023937119361	0.4135411707827998	T	T	F
0.5001962628937235	0.0002672436309581	0.4121105188834662	T	T	F
0.4999938365621972	0.0001886979302128	0.5859320102725221	T	T	T
0.7499904305446563	0.0002167023564539	0.4118483022046107	T	T	F
0.7503664993705783	0.0000090670021535	0.5864899707634379	T	T	T
0.4999508573330971	0.2502503605667741	0.4109391134730842	T	T	F
0.5003555726537527	0.2500102715759454	0.5866358564316201	T	T	T
0.7499450078435359	0.2500179241370546	0.4121071939256282	T	T	F
0.7501737250627798	0.2503607178014704	0.5859298549821261	T	T	T
0.0003051295546470	0.5002292073074699	0.4120422975070639	T	T	F
0.2504899005232843	0.4997259183636018	0.4133070697908465	T	T	F
0.0001203890636958	0.7500825684579908	0.4127662060365225	T	T	F
0.2499854113729398	0.7499106883190834	0.4120412443863586	T	T	F
0.4994126148524174	0.4996498268839034	0.4135454931637952	T	T	F
0.7498006324596815	0.4999998435479256	0.4113995002625757	T	T	F
0.7508069852204073	0.5003491436654173	0.5858933752487232	T	T	T
0.4999888884218180	0.7497529975249364	0.4113187568322942	T	T	F
0.7498857852724470	0.7497736167250864	0.4118771493512909	T	T	F
0.0002684344296426	0.9994484059694386	0.5857975803728124	T	T	T
0.0004050922303307	0.2500010922183620	0.5874737281753610	T	T	T
0.2486371997989494	0.2467713704174770	0.5873277702210373	T	T	T
0.0002220783758347	0.5007962051810182	0.5860947337994986	T	T	T
0.2485842528593096	0.5017875787264927	0.5878698199131236	T	T	T
0.0002641306010531	0.7500808728087023	0.5861475211743183	T	T	T
0.2495563213715002	0.7501406821501746	0.5861108705769382	T	T	T
0.5035989390434281	0.5017354329943302	0.5873373178100820	T	T	T
0.5003650708869287	0.7499572456500871	0.5874806390886320	T	T	T
0.7509306829474340	0.7500989943954437	0.5857828207085234	T	T	T
0.1671551261869908	0.0832400348029481	0.5033657252699939	T	T	T

0.0832295128899005	0.1666790010048516	0.4939588413917377	T	T	T
0.4170803731197594	0.0833773878096622	0.5032472355634496	T	T	T
0.3338837469628899	0.1667245264301584	0.4933238321880324	T	T	T
0.1664230941107405	0.3330334809209532	0.5043491340347472	T	T	T
0.0831487206042496	0.4164596876562427	0.4940519584800214	T	T	T
0.4172900086633788	0.3330090099203318	0.5041340927828557	T	T	T
0.3336083467977957	0.4166839621514455	0.4969789134437477	T	T	T
0.6667838149501009	0.0835097329536723	0.5033756579858419	T	T	T
0.5835136157537235	0.1667437887523988	0.4930226731386966	T	T	T
0.9162430946255284	0.0830510723388969	0.5033522167346121	T	T	T
0.8332901326220110	0.1667168858887180	0.4932917224493707	T	T	T
0.6669007634903252	0.3331961362194207	0.5032479020502194	T	T	T
0.5835618485076708	0.4163976912268552	0.4933221179508607	T	T	T
0.9161337193470587	0.3331016455689522	0.5033216668509866	T	T	T
0.8333175041839738	0.4166026405987253	0.4931454036675618	T	T	T
0.1667988966981042	0.5835222386654948	0.5034460546662416	T	T	T
0.0835624403431297	0.6667016956808141	0.4937976943776476	T	T	T
0.4172791264907971	0.5838792968716149	0.5043455581371958	T	T	T
0.3338322187982712	0.6671457509287961	0.4940525904991588	T	T	T
0.1671682032009656	0.8339352765952199	0.5033254174046107	T	T	T
0.0836381194112903	0.9169946307242043	0.4934148559299132	T	T	T
0.4171842887088673	0.8341448549073291	0.5033299900162476	T	T	T
0.3336497219590910	0.9169345063061911	0.4931465801758677	T	T	T
0.6670551702170258	0.5831447262190679	0.5033631650562285	T	T	T
0.5836091691223189	0.6670621790745415	0.4939619988268475	T	T	T
0.9163422477077390	0.5831070894677737	0.5033113273966805	T	T	T
0.8332926687970996	0.6666508122491521	0.4934040971334568	T	T	T
0.6672446966652856	0.8340576577940695	0.5033584026506475	T	T	T
0.5835457369204207	0.9169714685159249	0.4932943958474425	T	T	T
0.9168089651277569	0.8335144475502929	0.5030246839461512	T	T	T
0.8334915137836560	0.9167872202569995	0.4933587032985614	T	T	T
0.0792851831778378	0.4153273809982068	0.6367328056955657	T	T	T
0.0799956723047046	0.1643737579871143	0.6366249226341267	T	T	T
0.3375840486472015	0.1626535970261074	0.6348064484605729	T	T	T
0.1666133389502065	0.3335167693773471	0.3634852771392900	T	T	F
0.1670487630966591	0.0843226150139529	0.3629615322753565	T	T	F
0.4166226041806951	0.3335480333764309	0.3631313277313666	T	T	F
0.4167328660261091	0.0835224674729389	0.3629838452783432	T	T	F
0.5877569859107510	0.4128211256785193	0.6348099205717830	T	T	T
0.8290277156332664	0.4135964589232657	0.6346131293581792	T	T	T
0.5839567910296779	0.1664539340630853	0.6338839344101999	T	T	T
0.8283041399163338	0.1648989087531234	0.6348130863940469	T	T	T
0.6666576113939584	0.3334459616680405	0.3629827124323270	T	T	F
0.6668652615786286	0.0832981783476896	0.3627169628876175	T	T	F

0.9171606385749698	0.3333919323328587	0.3631883240794167	T	T	F
0.9168182998794592	0.0836626813498356	0.3633200611020584	T	T	F
0.0877836133063497	0.9184843790459993	0.6354028995603898	T	T	T
0.3368084952155144	0.9213835530288534	0.6346120835839741	T	T	T
0.0834183236164705	0.6669916233243711	0.6364276323019666	T	T	T
0.3350917370201643	0.6711399570890464	0.6367348891783280	T	T	T
0.1664918697491302	0.8331942044074765	0.3636092557200143	T	T	F
0.1673224806134144	0.5828428632000566	0.3631580695747090	T	T	F
0.4167756692777331	0.8329994157458103	0.3631860553540278	T	T	F
0.4166657783980270	0.5835636364052164	0.3634864240805484	T	T	F
0.5855072442496834	0.9221041154659915	0.6348124185962334	T	T	T
0.8313728754057621	0.9190423124655780	0.6353096701598702	T	T	T
0.5860635350888046	0.6704399939590218	0.6366252222362050	T	T	T
0.8319205399846581	0.6626162643791957	0.6353960938957997	T	T	T
0.6665015980052900	0.8333465435082701	0.3633233774714952	T	T	F
0.6658471589652706	0.5831174359337012	0.3629682627645252	T	T	F
0.9167892582812627	0.8333838872504415	0.3634429127008829	T	T	F
0.9169659669706505	0.5836650823168483	0.3636126231158556	T	T	F
0.3343135591569393	0.4161549559791240	0.6772418060978680	T	T	T