

Supplementary Material

Anti-freezing, high-voltage, and ultra-stable micro-supercapacitors enabled by phase engineering of 2D SiP₂ in LiCl-DMSO electrolyte

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Table S1 Comparison of electrochemical performance of representative nano-electrode materials for MSCs

Electrode materials*	C_A (mF cm ⁻²)	Energy density (μWh cm ⁻²)	Power density (mW cm ⁻²)	Capacitance retention (%)	Cycle stability	Ref.
B ₄ C@C	7.33	1.02	1.09	100.94	50,000	1
PG-MSCs	9.8	-	-	89.5	2000	2
LIG-MSCs	7.12	0.63	-	100	10,000	3
MoS ₂	8	-	-	92	1000	4
PEDOT MSCs	9	-	-	80	10,000	5
G-CNTs	9.81	1.36	0.25	95.5	10,000	6
3D GA	12.6	-	-	-	-	7
MXene@EG	3.26	-	-	82	2500	8
TP@EG	3.42	-	-	86	10,000	9
Graphene	1.324	0.64	20	98	10,000	10
(Ta, Ti, Nb)C/C	41.6	5.8	0.04	92.6	9000	11
LSG/Ni-CAT	15.2	4.1	7	87	5000	12
MOF						
rGO/Ag-NW	8.6	0.669	0.73	-	-	13
Pnma SiP ₂	12.39	6.86	18.26	97.89	20,000	This
				82.37	30,000	work

*PG: Phosphorene and graphene, LIG: Laser-induced graphene, PEDOT: Poly(3,4-

ethylenedioxythiophene) electrodes, G-CNTs: Graphene-CNT, 3D GA: 3D-printed graphene aerogel, MXene@EG: $\text{Ti}_3\text{C}_2\text{T}_x$ @electrochemically exfoliated graphene, TP@EG: Thiophene nanosheets and electrochemically exfoliated graphene, LSG/Ni-CAT MOF: Ni-catecholate-based MOF on 3D laser-scribed graphene, rGO/Ag-NW: Reduced graphene oxide /silver nanowire hybrid electrodes.

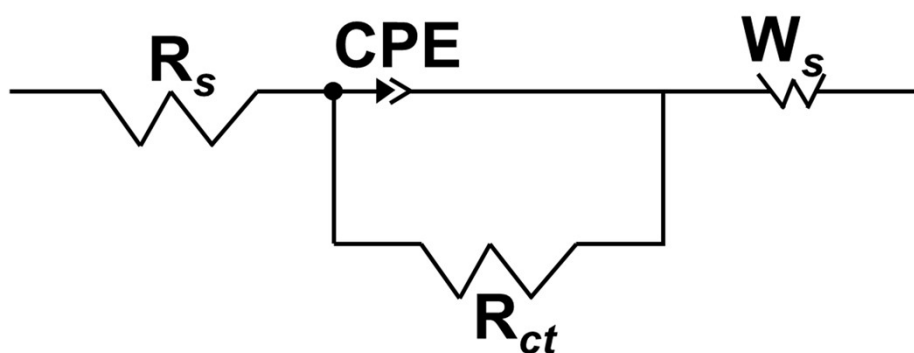


Fig. S1 Equivalent circuit model for micro-supercapacitor testing.

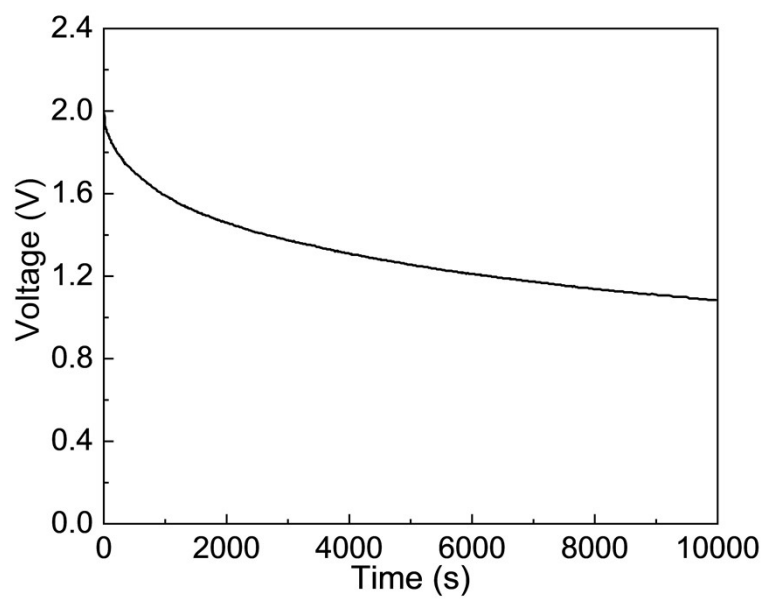


Fig. S2 Self-discharge curve under a 2 V voltage window.

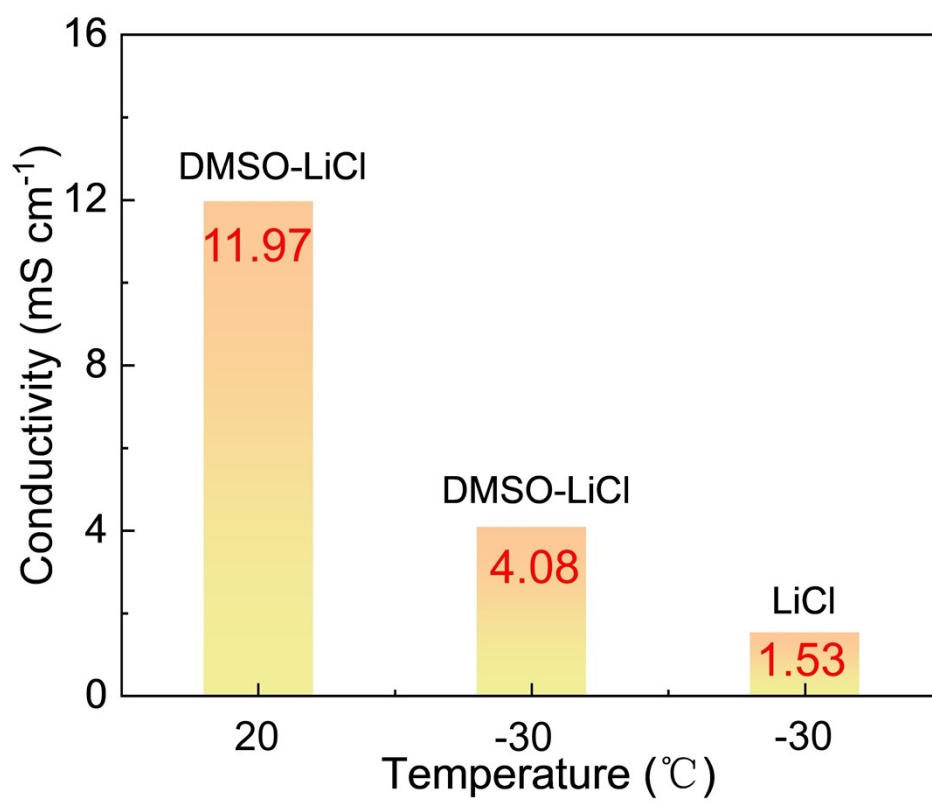


Fig. S3 Ionic conductivities of different gel electrolytes at various temperatures.

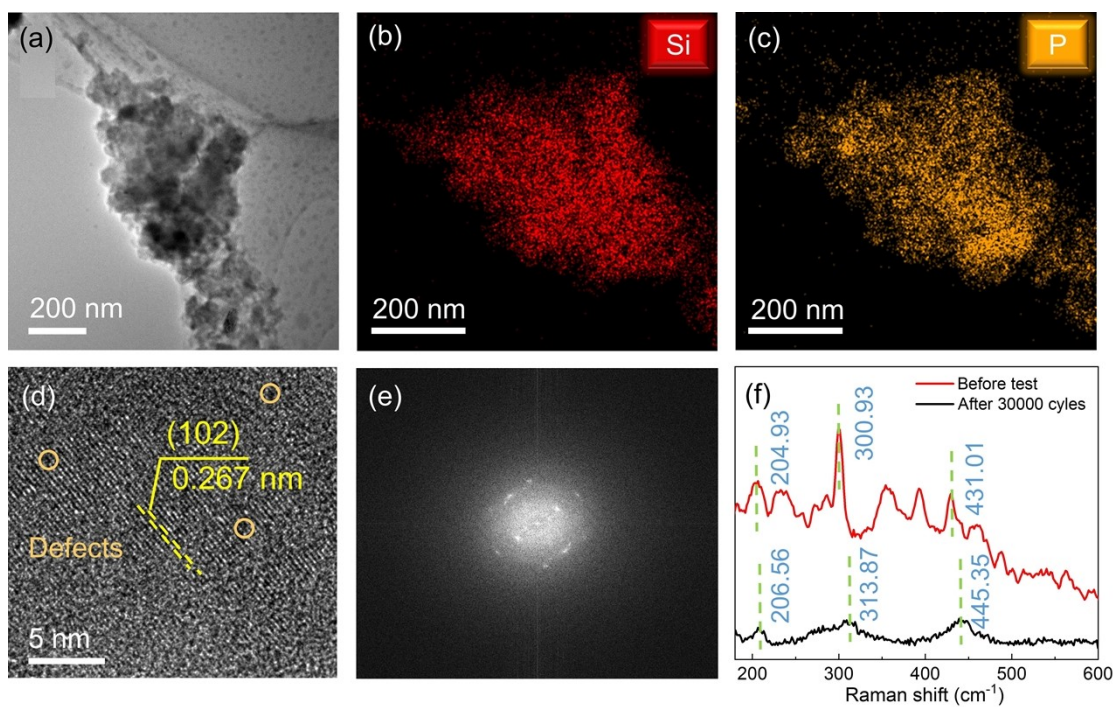


Fig. S4 (a) TEM images of Pnma-SiP₂ after cycling, along with the corresponding elemental distribution maps of (b) Si and (c) P. (d) High-resolution TEM images of Pnma-SiP₂ after cycling. (e) Fourier transform pattern of Pnma-SiP₂. (f) Raman spectra of Pnma-SiP₂ before and after cycling.

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