

Supporting Information

Boosting the Syngas Production of a SrTiO₃ Photocathode via Plasmonic Hot Electron Injection from an Ag Underlayer

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Figure S1: KPFM Histograms.

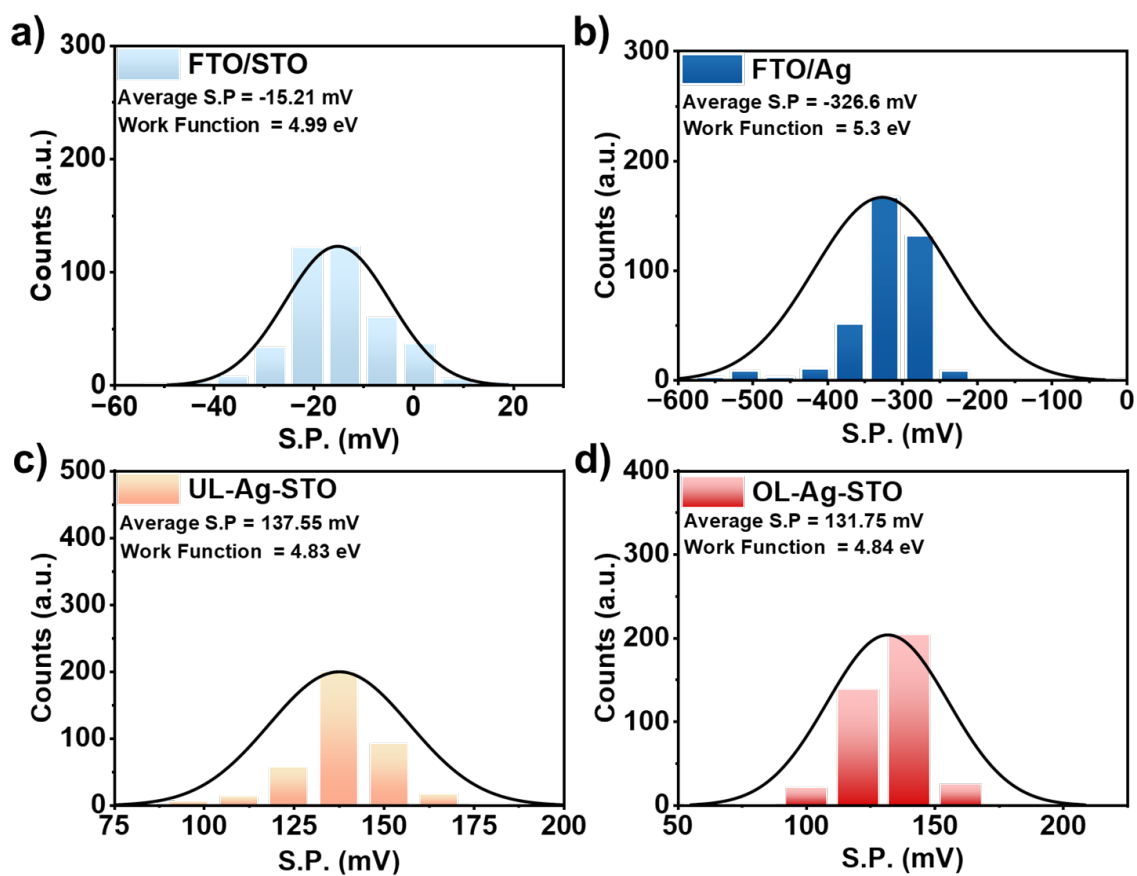


Figure S1 (a-d): KPFM histograms show the Gaussian distributions in surface potentials across FTO/STO, FTO/Ag, UL-Ag-STO and OL-Ag-STO samples. The subsequent averages were used to calculate consequent work functions for each sample.

Figure S2: SEM surface Imaging.

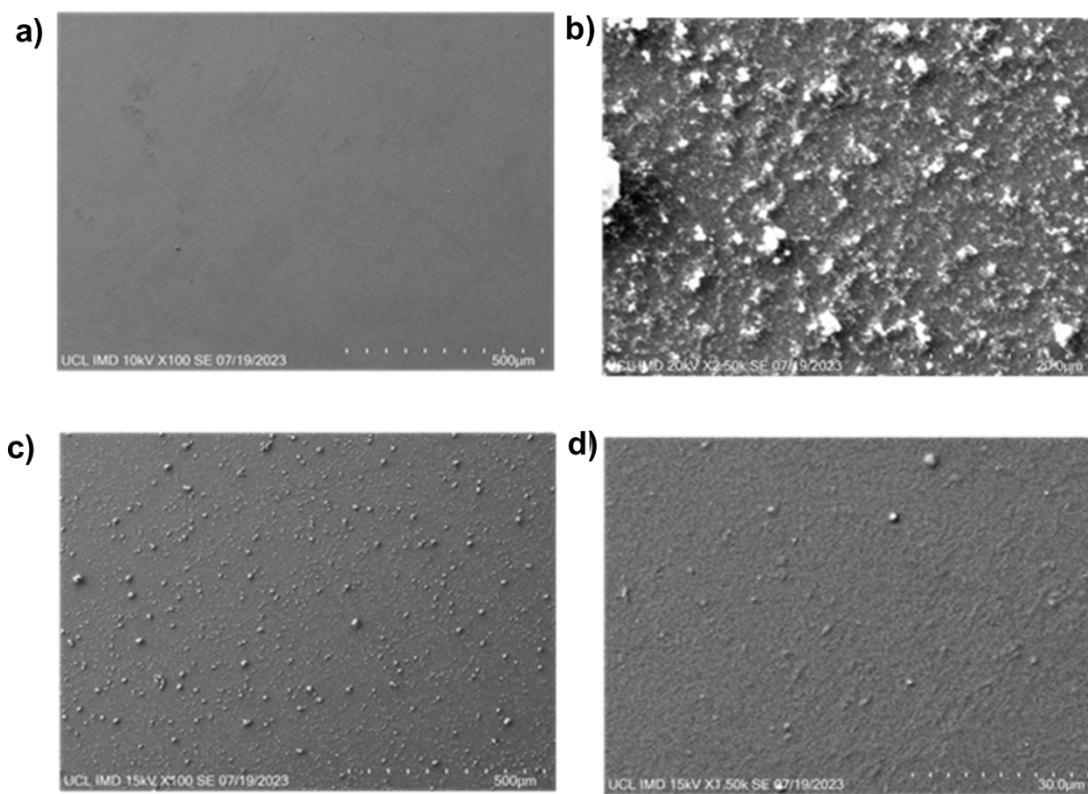


Figure S2: SEM images of different catalysts and reference films fabricated. (a) FTO/Ag, (b) FTO/STO, (c) UL-Ag-STO, and (d) OL-Ag-STO.

Calculation of Faradaic efficiency for the ECR products

$$F. E_g = \frac{q_{\text{output}}}{q_{\text{input}}} \dots\dots\dots \text{Equation 1}$$

$$q_{\text{output}} = N_p \times n_e \times F \dots\dots\dots \text{Equation 2}$$

$$N_p = C_p(\text{ppm}) \times 10^{-6} \times \frac{PV}{RT} \dots\dots\dots \text{Equation 3}$$

$$N_p = 40.89 \times 10^{-12} \times C_p(\text{ppm}) \dots\dots\dots \text{Equation 4}$$

$$q_{\text{input}} = i \times t \dots\dots\dots \text{Equation 5}$$

$$t = \frac{V (\text{cm}^3)}{r_F (\text{cm}^3 \text{ min}^{-1})} \dots\dots\dots \text{Equation 6a}$$

$$t = \frac{1 \text{ cm}^3 \times 60 \text{ s}}{20 \text{ cm}^3} = 3 \text{ s} \dots\dots\dots \text{Equation 6b}$$

$$F. E_g = \frac{1.32 \times 10^{-6} \times n_e \times C_p(\text{ppm})}{i} \dots\dots\dots \text{Equation 7}$$

where, q_{output} is the total charge consumed to generate products, q_{input} is the total charge supplied to the system, N_p is the number of moles of the product, n_e the number of electrons required for generating one molecule of the product, F (96485 C) is the Faraday constant, $x_p(\text{ppm})$ is the concentration of product in ppm (obtained from GC), P is the pressure (101325 N m⁻²), V (cm³) the loop volume, R (8.314 J K⁻¹ mol⁻¹) universal gas constant, T (298 K) is the absolute temperature, t is the time required to fill a 1 cm³ sample loop, and r_F (sccm= cm³min⁻¹) is the flow rate.

Figure S3: Fitted EIS spectra and light-dark charge transfer comparison.

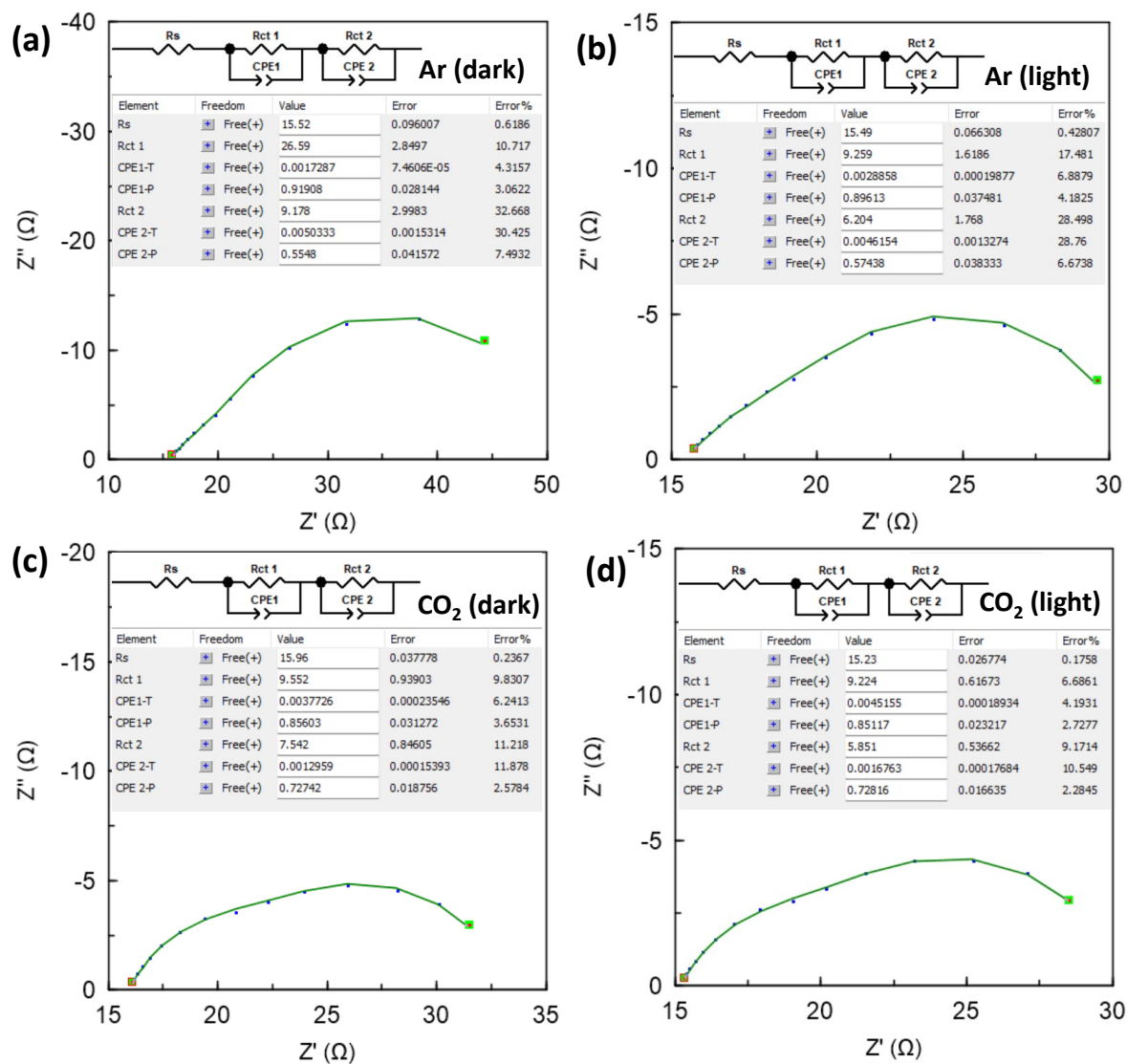


Figure S3: Fitted EIS spectra demonstrating fitted two-time constant equivalent circuit (R_s -($R_{ct1}/CPE1$)-($R_{ct2}/CPE2$)) under light and dark conditions for UL-Ag-STO.

Table S1: EIS Fitted Parameters - comparison of EIS parameters for UL-Ag-STO under different conditions obtained after fitting.

Conditions	$R_s (\Omega)$	$R_{ct1}(\Omega)$	$R_{ct2} (\Omega)$	$CPE_1 (F)$	$CPE_2 (F)$
Ar (dark)	15.50	9.20	26.60	0.55	0.92
Ar (light)	15.50	6.20	9.30	0.57	0.89
CO ₂ (dark)	16.00	7.50	9.60	0.72	0.86
CO ₂ (light)	15.20	5.90	9.10	0.85	0.72

Where R_s is the contact resistance, R_{ct1} is the charge transfer resistance at the Ag/STO interface, and R_{ct2} is the charge transfer resistance at the STO/PBS for CO₂ reduction, while CPE_1 and CPE_2 are the corresponding capacitance values.

Figure S4: Faradic efficiency of photocathodes.

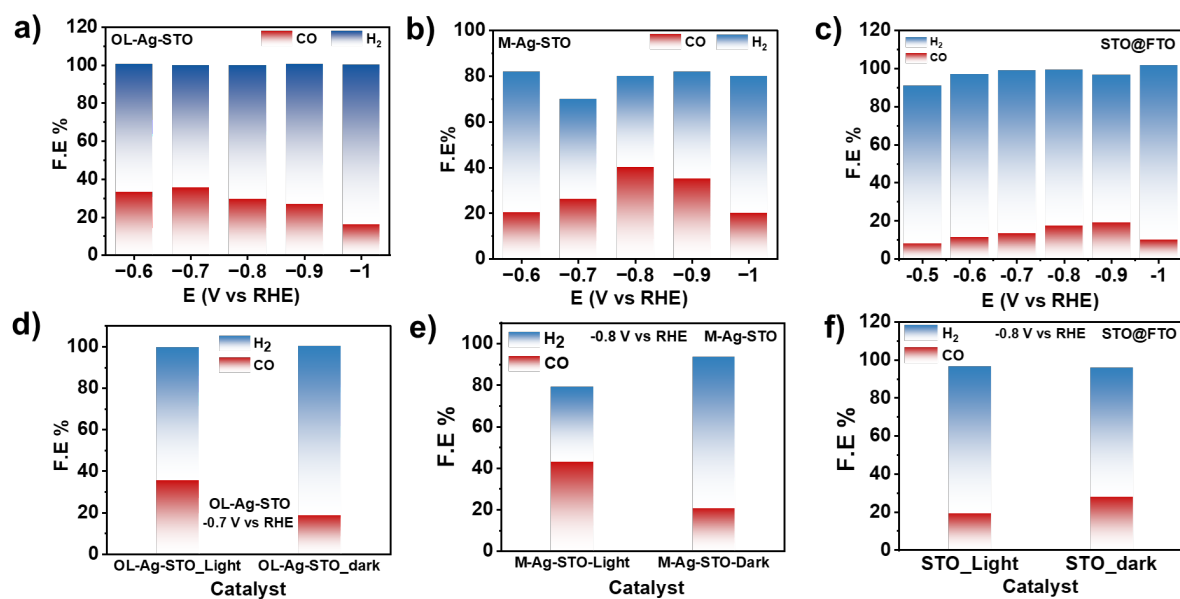


Figure S4: %F.E. plots for various photocathodes at various potentials: (a) OL-Ag-STO, (b) AgTiO₂-STO, (c) STO, and F.E. % comparison plots under dark and light conditions. (d) OL-Ag-STO, (e) M-Ag-STO, and (f) STO in CO₂ saturated PBS.

Figure S5: Time-dependent cyclic stability.

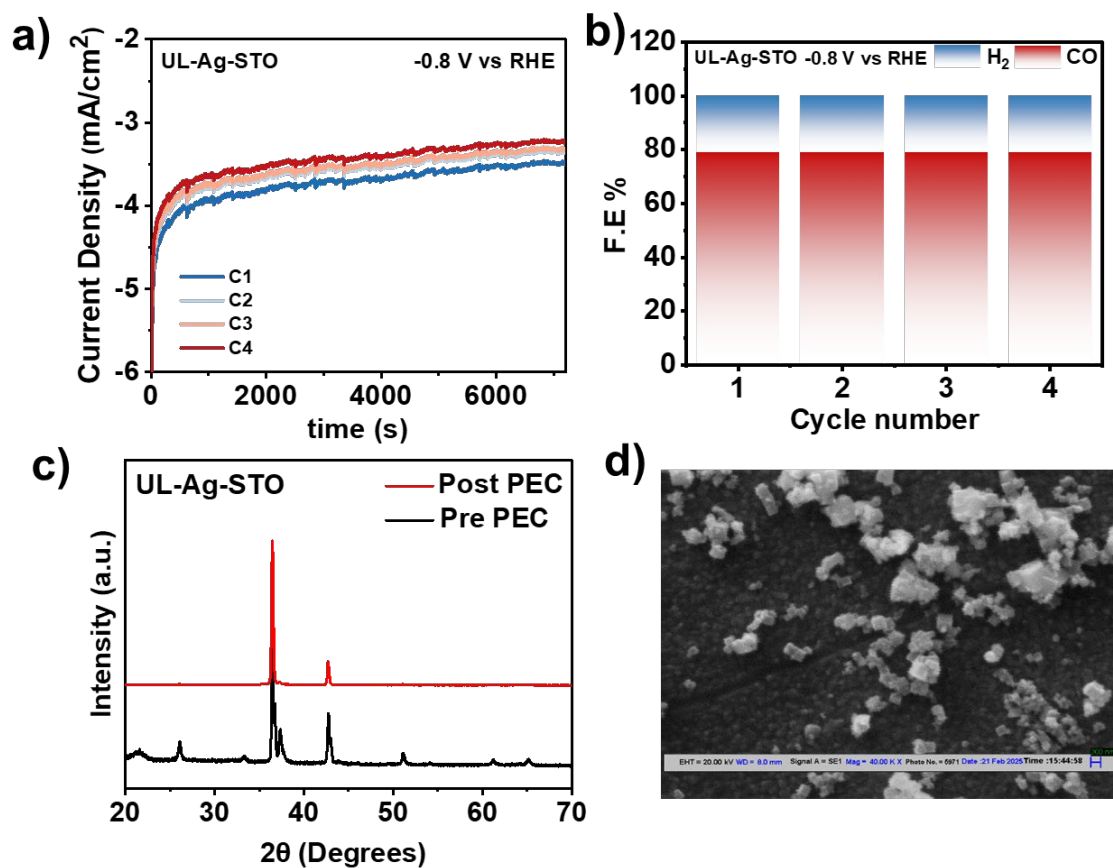


Figure S5: (a-b) Cyclic stability of UL-Ag-STO at -0.8 V vs RHE in CO_2 saturated PBS, (c) post-XRD analysis, and (d) post SEM analysis after 10 hours of continuous operation.

Figure S6: Mott-Schottky Plots and Flat band calculations.

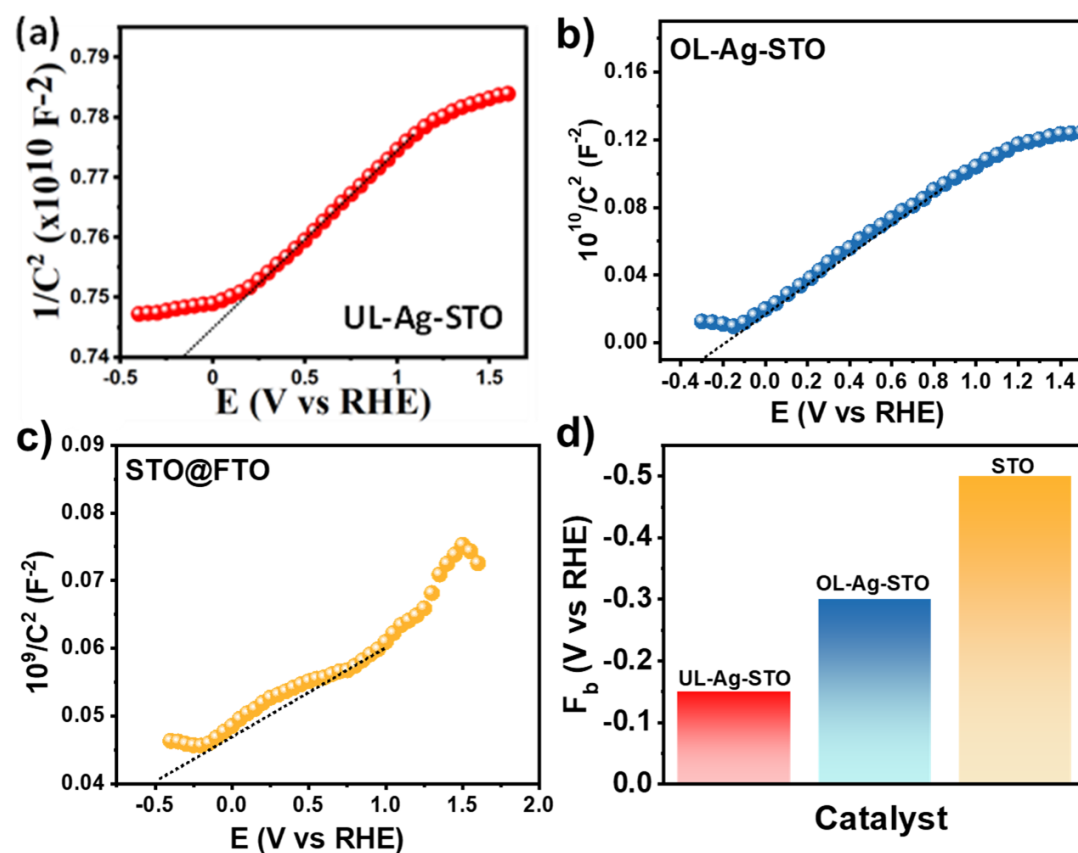


Figure S6: Mott-Schottky plots for (a) UL-Ag-STO, (b) OL-Ag-STO, (c) STO, and comparison of flat band potentials (F_b).

Table S2: Photoelectrocatalyst Benchmarking based on Ag and TiO₂.

Catalyst	E (V vs RHE)	Selectivity	F.E %	References
Cu-ZnO/GaN/n ⁺ -p Si	+0.18	CO	70	<i>Angew. Chem. Int. Ed.</i> 2016, 55, 14262 – 14266
p-Si/SiO ₂ -Cu-Ag	-1.0	CO	79.8	<i>ACS Appl. Mater. Interfaces</i> 2021, 13, 16, 18905–18913
	-1.4	CH ₄	59.3	
Ag/TiO ₂ /p-Si	-0.6	CO	47	<i>Appl. Sci.</i> 2020, 10(10), 3487
SD-Ag/Si	+0.26	CO	95.2	<i>ACS Appl. Mater. Interfaces</i> 2024, 16, 17, 21868–21876
Ag/Cu ₂ O/ZS/FTO	-1.2 (Fc ⁺ /Fc)	C ₂ H ₄	60	<i>Nature Energy</i> , 6, (2021), 1124–1132
SC-TiO ₂ /CP	-1.8 (Ag/AgCl)	C ₂ H ₄	46.6	<i>J. Env. Chem. Eng.</i> 10 (2022) 107441
Ag/Cu-TS-1	-0.6	C ₂ H ₅ OH	47	<i>Catal. Sci. Technol.</i> , 2022, 12, 2490
Au ₃ Cu / Si NW	-0.2	CO	80	<i>Nano Lett.</i> 2016, 16, 9, 5675–5680
Ag ₅₀ Cu ₅₀ /p-Si	-0.72	CO	26	<i>ACS Appl. Energy Mater.</i> 2023, 6, 16, 8397–8409
	-0.87	CH ₄	18.2	
Si/Ag	-0.5	CO	90	<i>J. Mater. Chem. A</i> , 2018, 6, 21906-21912
Au-TiO ₂	-0.11	CO	84.2	<i>ACS Catal.</i> 2021, 11, 18, 11416–11428
Sn/TiO ₂ /Si	-1.0	HCOOH	68.82	<i>A European Journal</i> , 10.1002/chem.202303552, 30, 14, (2024)
UL-Ag-STO@FTO	-0.8	CO	60 (front illumination)	This work
			80 (back illumination)	

Table S3: State-of-the-art photoelectrocatalyst for CO₂ reduction.

Photocathode	j (mA cm ⁻²)	E (V vs RHE)	References
p-Si/SiO ₂ -Cu-Ag	-3.0	-0.8	<i>ACS Appl. Mater. Interfaces</i> 2021, 13, 16, 18905-18913
Si/Ag ₁₂₀ -EM ₆₀	-2.0	-1.0 vs. Ag/AgCl	<i>ACS Appl. Nano Mater.</i> 2023, 6, 2, 821-826
CPB-F-N-Au	-0.4	-0.8 vs. Ag/AgCl	<i>Energy Fuels</i> 2023, 37, 23, 18163-18172
Au/TiO ₂ /n ⁺ p-Si	-5.5	-0.8	<i>Small</i> 2022, 18, 20, 2201882
Ni-In/SiNWs	-1.8	-0.8	<i>J. of Energy Chem.</i> 2021, 54, 422-428
Bi@ZFO NTs	-1.5	-0.8	<i>J. Phys. Chem. Lett.</i> 2022, 13, 29, 6867-6874
CuS/GaN/Si	-4.5	-0.8	<i>J. Am. Chem. Soc.</i> 2021, 143, 27, 10099-1010
CZTS/CdS (HA)	-1.8	-0.8	<i>Small</i> 2021, 17, 31, 210049
In _{0.4} Bi _{0.6} + PV + light on	-5.0	-0.6	<i>ACS Energy Lett.</i> 2019, 4, 6, 1279-1286
Sn/TiO ₂ /Si	-4.7	-1.0	<i>ChemSusChem</i> 2022, 15, 8, e202200188
Si TiO ₂ CNT Co ^{II} (BrqPy)	-1.5	-0.11	<i>Angew. Chem. Int. Ed.</i> 2022, 61, 24, e202201086
RA-Au	-2.94	-0.11	<i>Adv. Energy Mater.</i> 2017, 7, 3, 1601103
meso-TiO ₂ CoPcP	-1.53	-1.09 vs SHE	<i>ACS Catal.</i> 2021, 11, 3, 1868-1876
Ag-8.2/SiNWs	-4.1	-1.0	<i>Nanoscale</i> , 2019, 11, 26, 12530-12536
STA-GO/CoPc	-2.6	-0.7	<i>Angew. Chem. Int. Ed.</i> 2023, 62, 4, e202215213
UL-Ag-STO	-1.4	-0.2	<i>This work</i>
	-1.8	-0.6	
	-2.6	-0.7	
	-4.0	-0.8	