

## Supporting information

# Thermally Stable Polysulfone Nanofiltration/Reverse Osmosis Membranes via Amino Grafting

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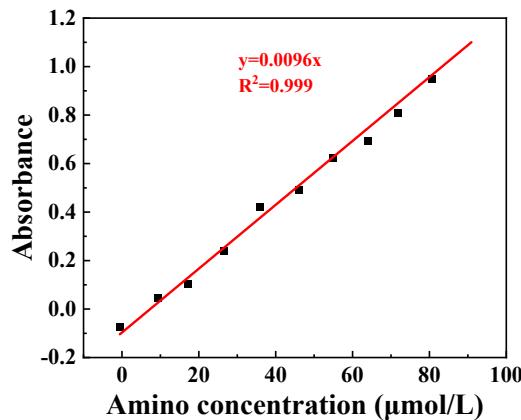
## S1 Surface density of amino group

The surface amination reaction was carried out following a nucleophilic substitution mechanism<sup>1</sup>. The dye adsorption method was used to measure the surface density of amino group ( $-\text{NH}_2$ ) on substrates prepared under different amination conditions<sup>2</sup>. In detail, a substrate sample ( $3 \times 3 \text{ cm}^2$ ) was immersed into an aqueous solution of acid orange 7 (500  $\mu\text{mol/L}$ , pH=3) at 25°C for 24 h to ensure adequate adsorption. The dissociative acid orange 7 was removed by thoroughly washing with HCl solution at pH=3. The adsorbed acid orange 7 was desorbed in NaOH solution at pH=12 at 25°C for 24 h. The absorbance of acid orange 7 in elution was measured by UV spectrophotometer (UV-1800, Shimadzu, Shanghai, China) and the concentration of acid orange 7 was calculated from standard curve illustrated in Figure S1 and Eq.\\* MERGEFORMAT (1). The surface density of amino group ( $C_{\text{NH}_2}$ ) was determined by Eq.\\* MERGEFORMAT (2),

$$y = 0.0096x \quad \text{\* MERGEFORMAT (1)}$$

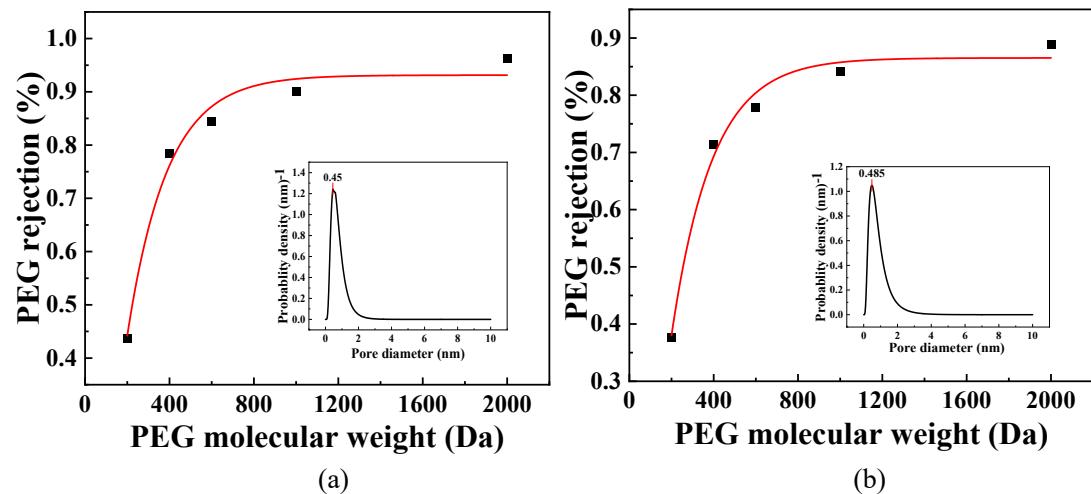
$$C_{\text{NH}_2} = \frac{C_{\text{dye}} \times V_{\text{elution}}}{A} \quad \text{\* MERGEFORMAT (2)}$$

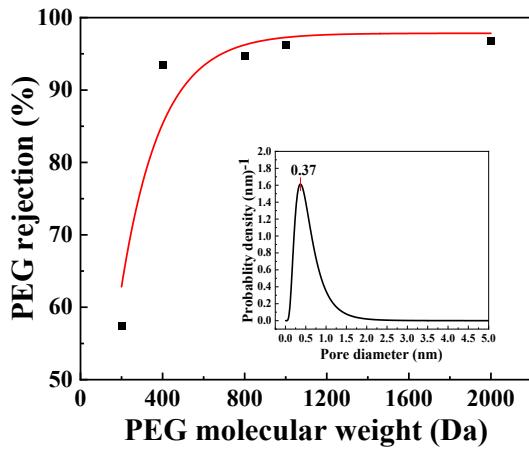
where  $C_{\text{dye}}$  is the concentration of acid orange 7 in the elution,  $V_{\text{elution}}$  is the volume of elution, and  $A$  is the surface area of the substrate sample.



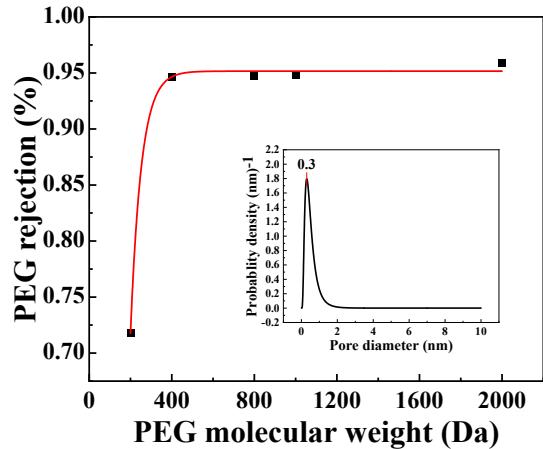
**Figure S1** Standard curve of acid orange 7.

## S2 Pore Size of Membranes Under Different Amination Conditions

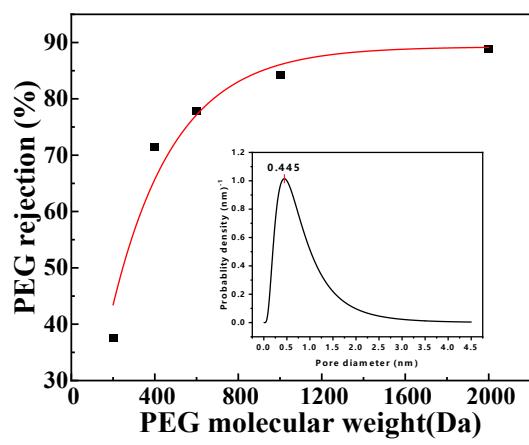




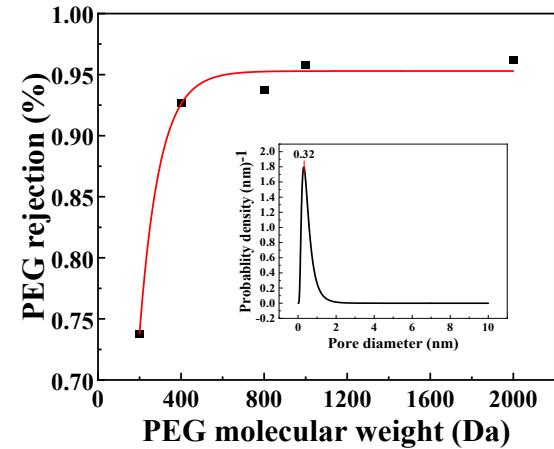
(c)



(d)



(e)



(f)

**Figure S2** Pore Size of Membranes Under Different Amination Conditions. (a-c): The amination time was set at 24 h, with DETA concentrations of 20 wt.%, 30 wt.%, and 40 wt.% respectively, (d-e): The DETA concentration was fixed at 25 wt.%, while the amination time was set at 12 h, 24 h, and 36 h respectively.

## References

1. L. Wang, Y. Cui, N. Wang, H. Zhang, B. Zhu, L. Zhu and Y. Xu, *Polymer Degradation and Stability*, 2014, 103, 69-74.
2. V. Hoseinpour, A. Ghaee, V. Vatanpour and N. Ghaemi, *Carbohydrate Polymers*, 2018, 188, 37-47.