

Supporting Information

Formaldehyde Electrolysis in a Membrane-Free Electrolyzer: Low-Energy Hydrogen and Formate Co-Production with Cu-Based Boride Electrocatalysts

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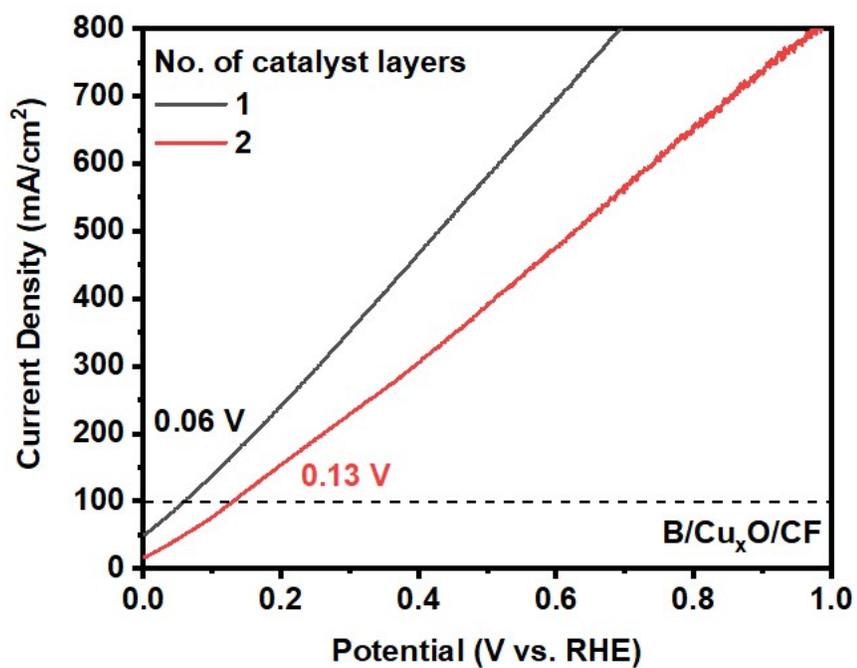


Fig. S1: Comparison of FOR activity for different catalyst loading

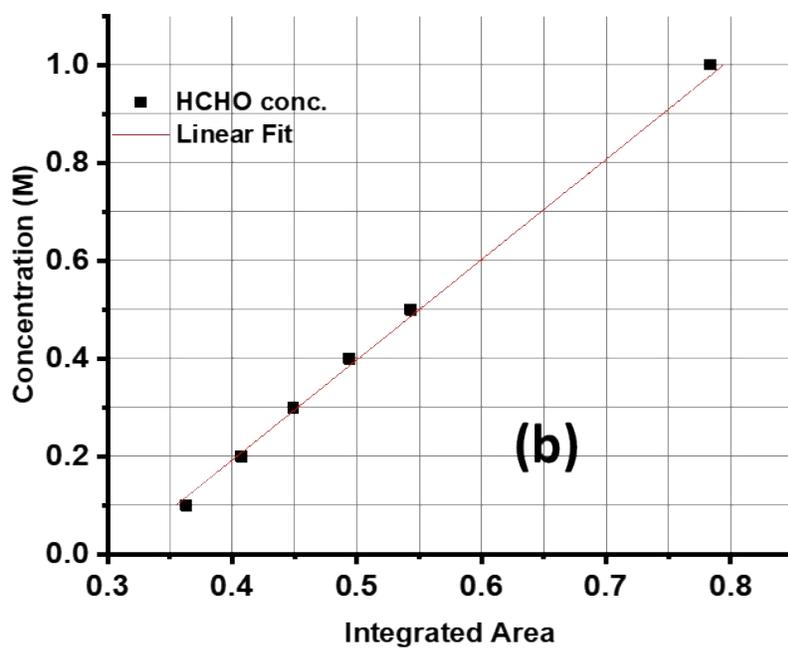
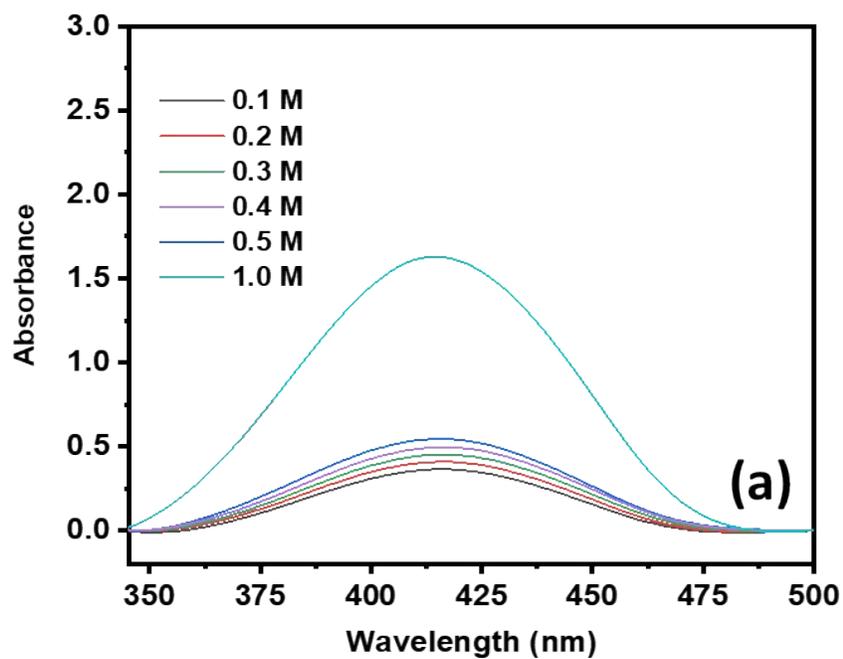
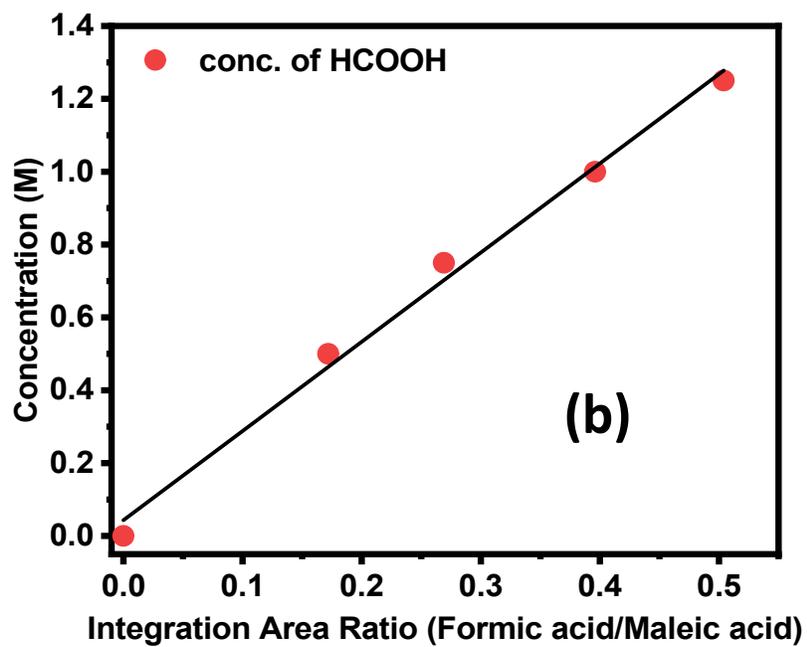
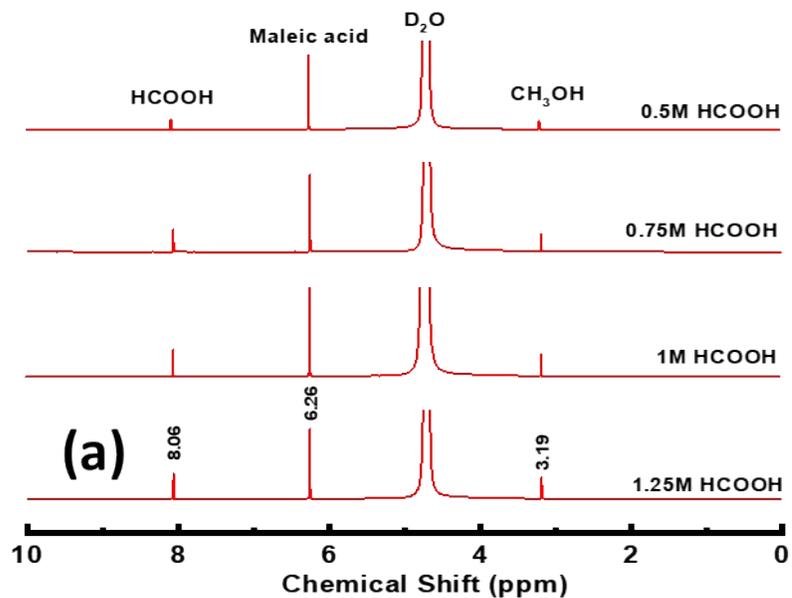


Fig. S2. (a) UV absorption spectra for known HCHO concentrations and (b) Calibration curve obtained using peak intensity of $\lambda = 413$ nm.



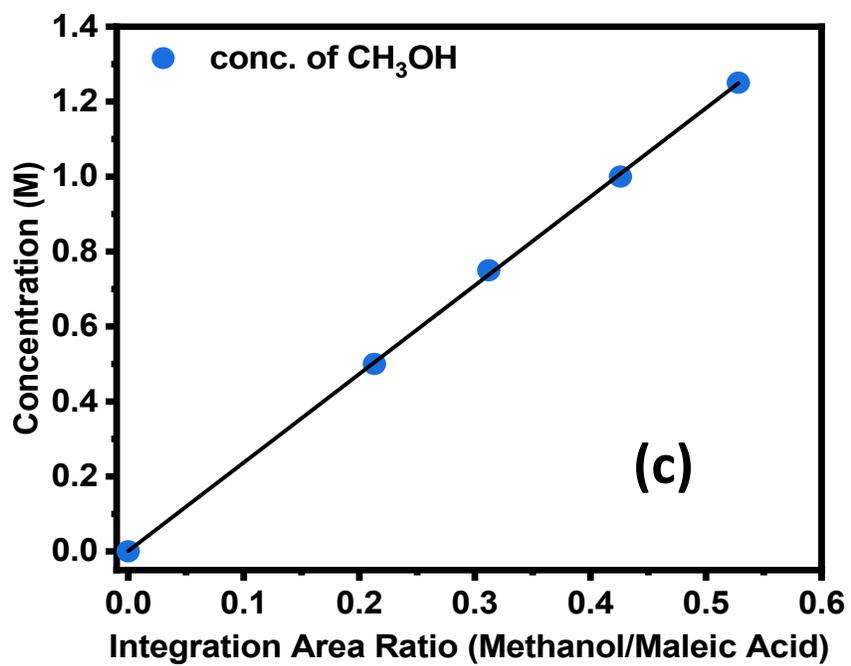


Fig. S3. (a) ¹H NMR spectra for a known concentration of HCOOH and Methanol; and Calibration curves obtained for (b) HCOOH and (c) Methanol.

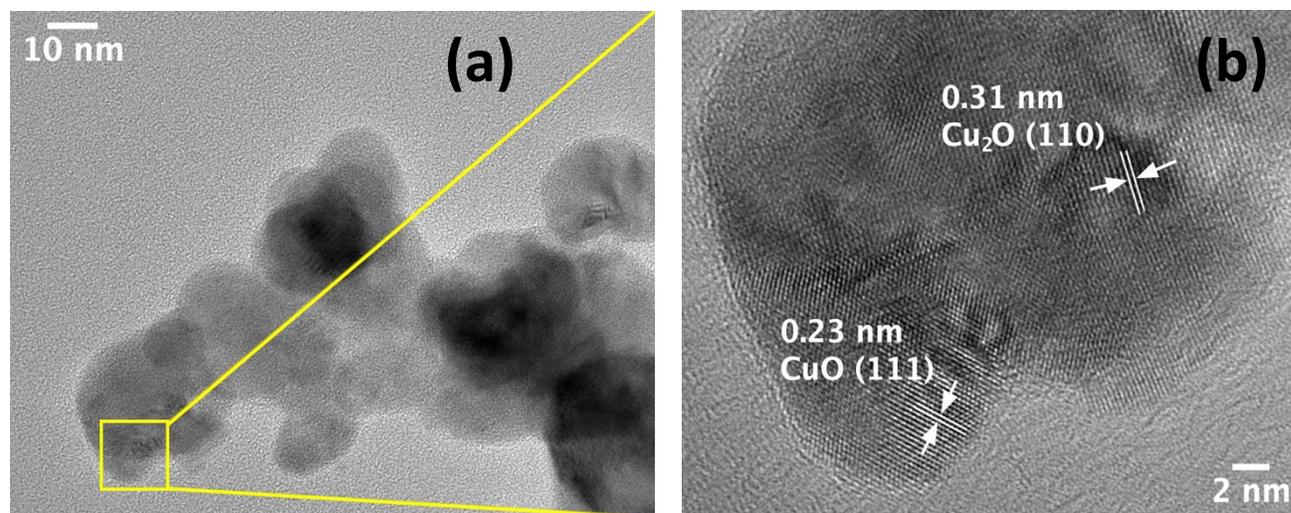
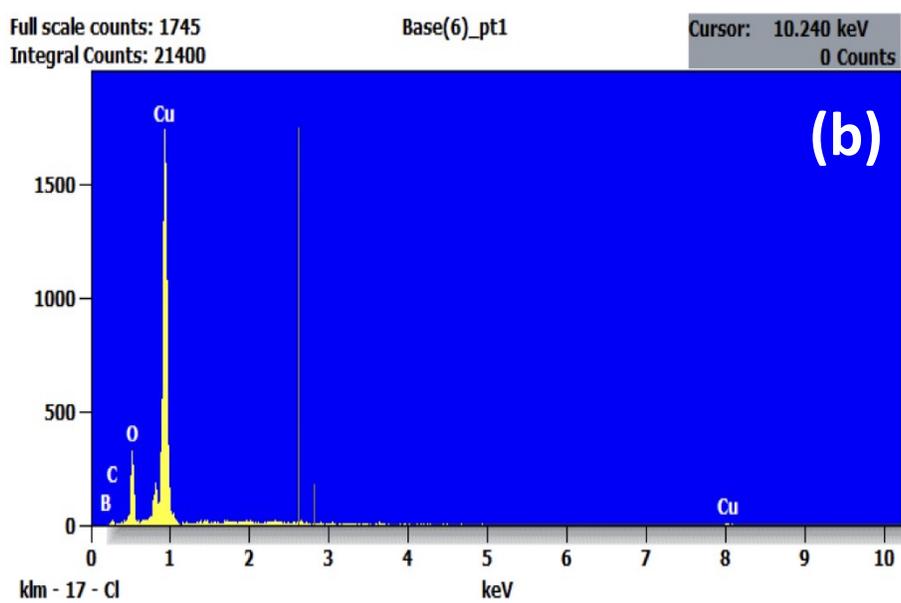
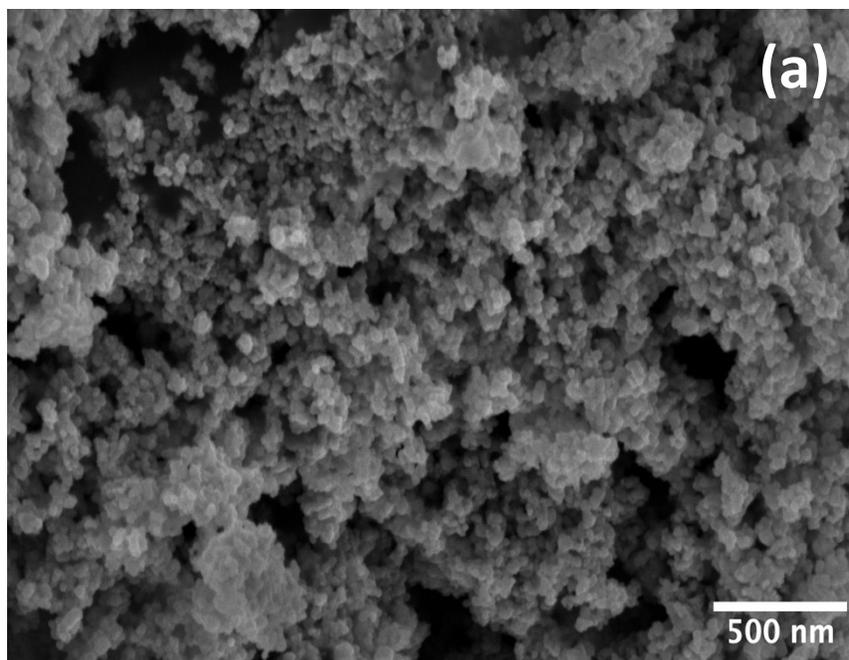
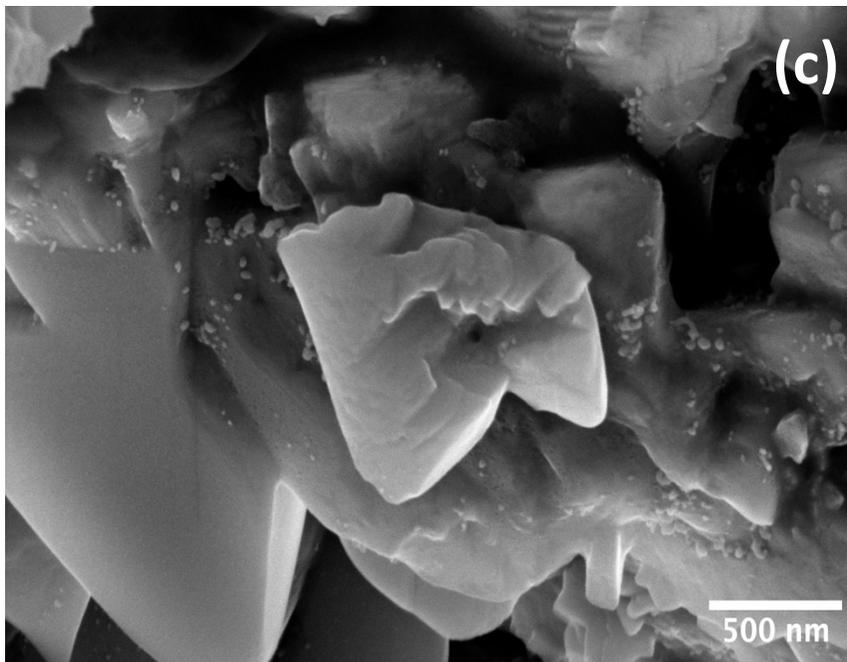


Fig. S4: (a) TEM and (b) HRTEM micrographs of B/Cu_xO/CF

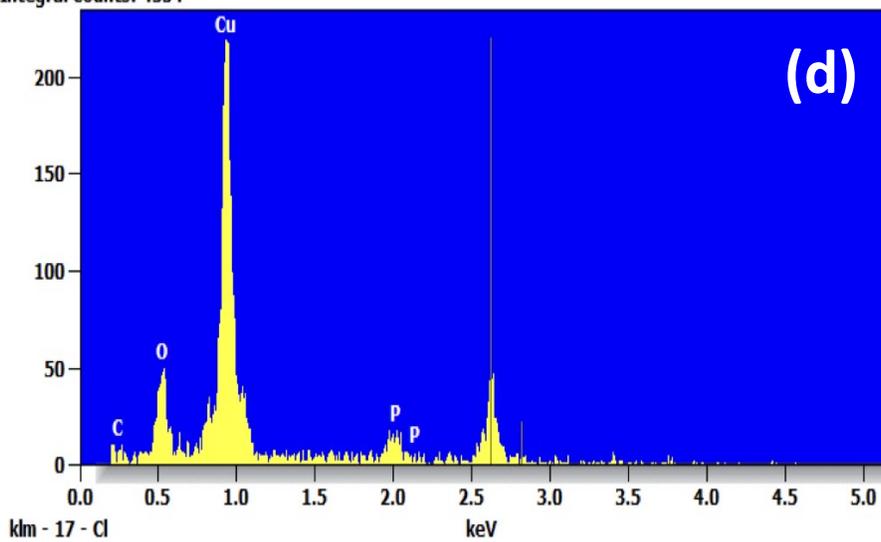


Element	B	C	O	Cu
Weight %	0	0.93	8.66	90.41



Full scale counts: 220
Integral Counts: 4354

Base(1)_pt1



Element	P	C	O	Cu
Weight %	8.79	0.82	8.53	81.86

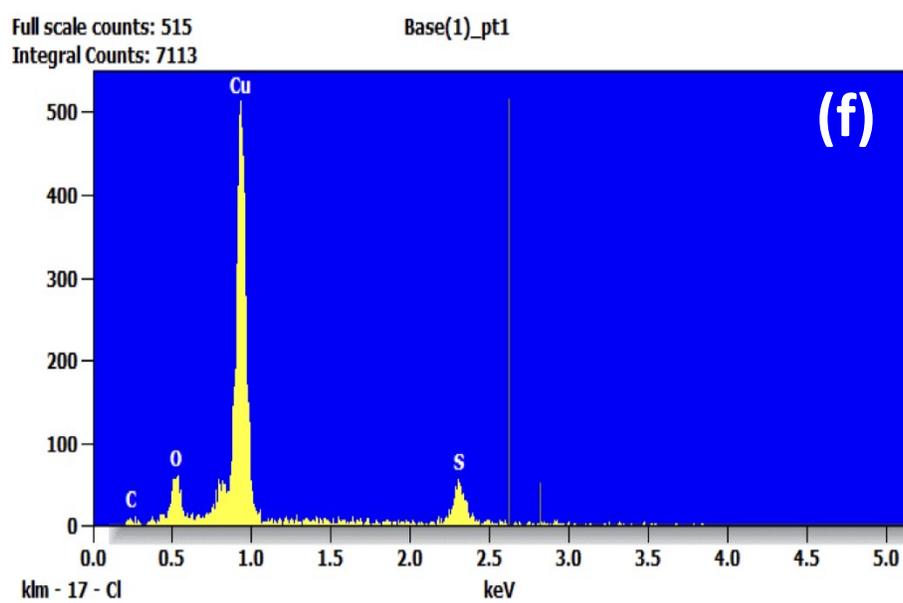
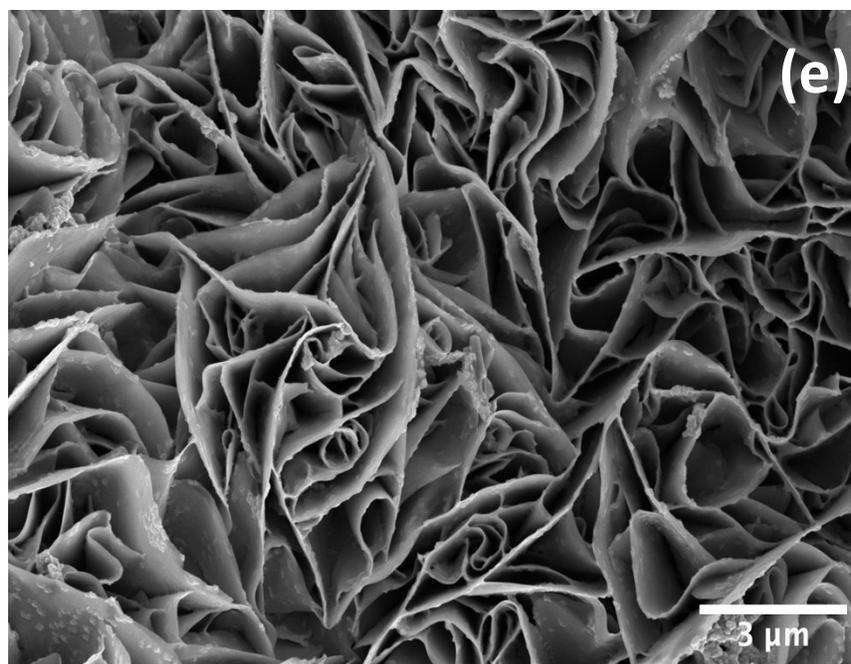
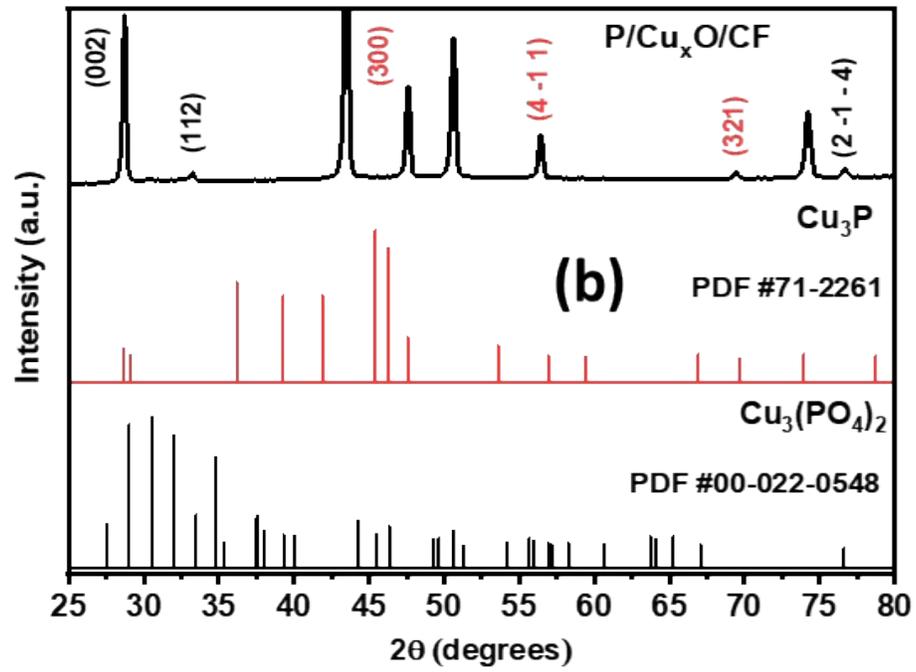
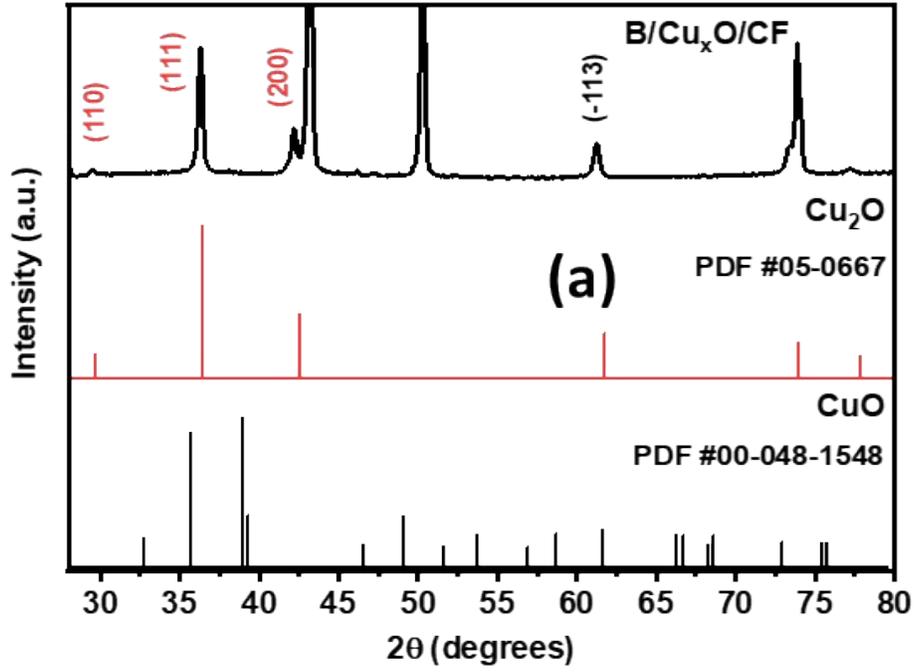


Fig. S5. SEM images of (a) B/Cu_xO/CF, (b) P/Cu_xO/CF, and (c) S/Cu_xO/CF, and (b, d, and f) corresponding EDS spectra.



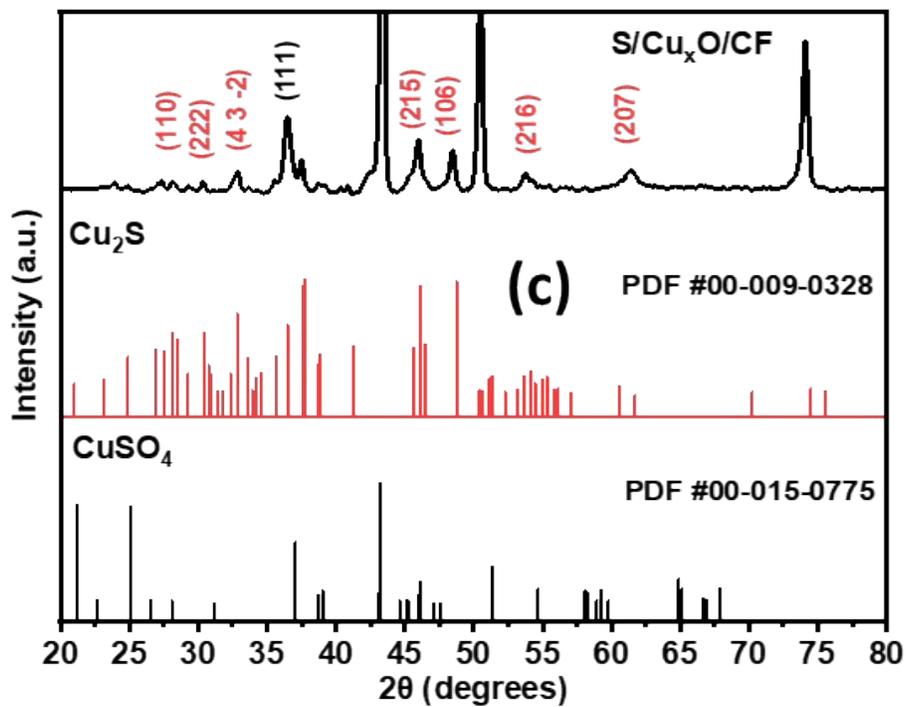


Fig. S6. XRD spectra of (a) B/Cu_xO/CF, (b) P/Cu_xO/CF, and (c) S/Cu_xO/CF electrocatalysts.

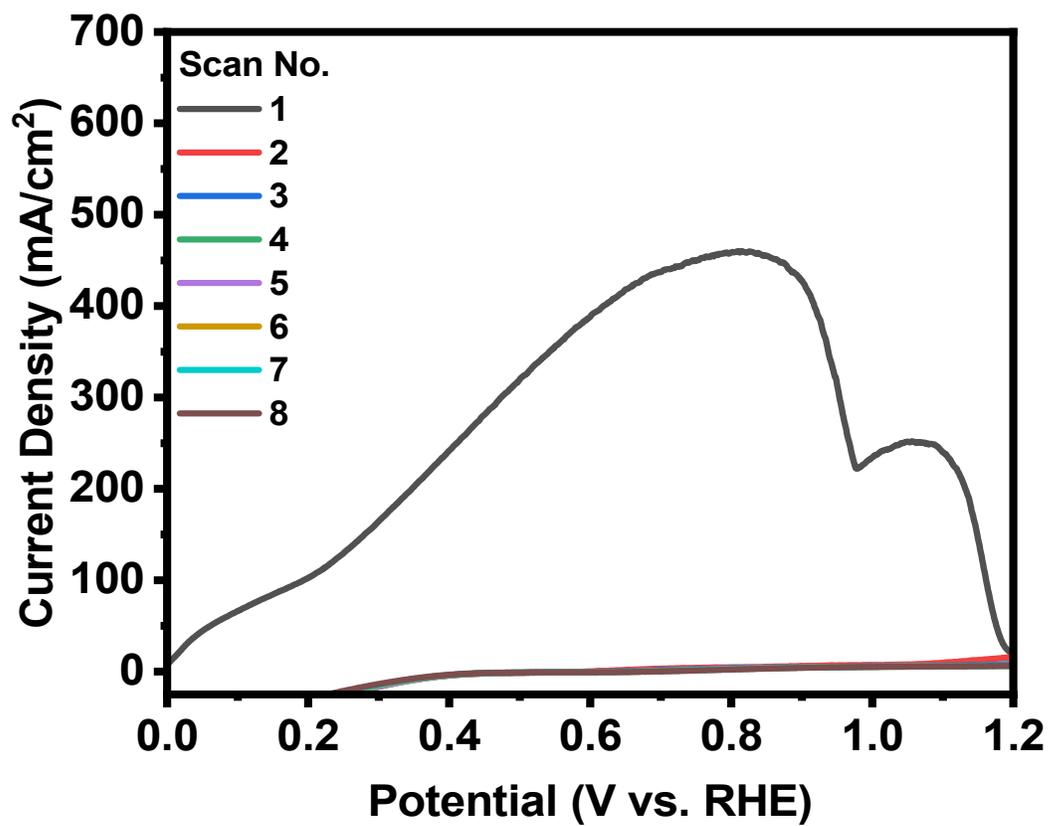


Fig. S7; Subsequent Linear Polarization Curves for FOR of P/Cu_xO/CF

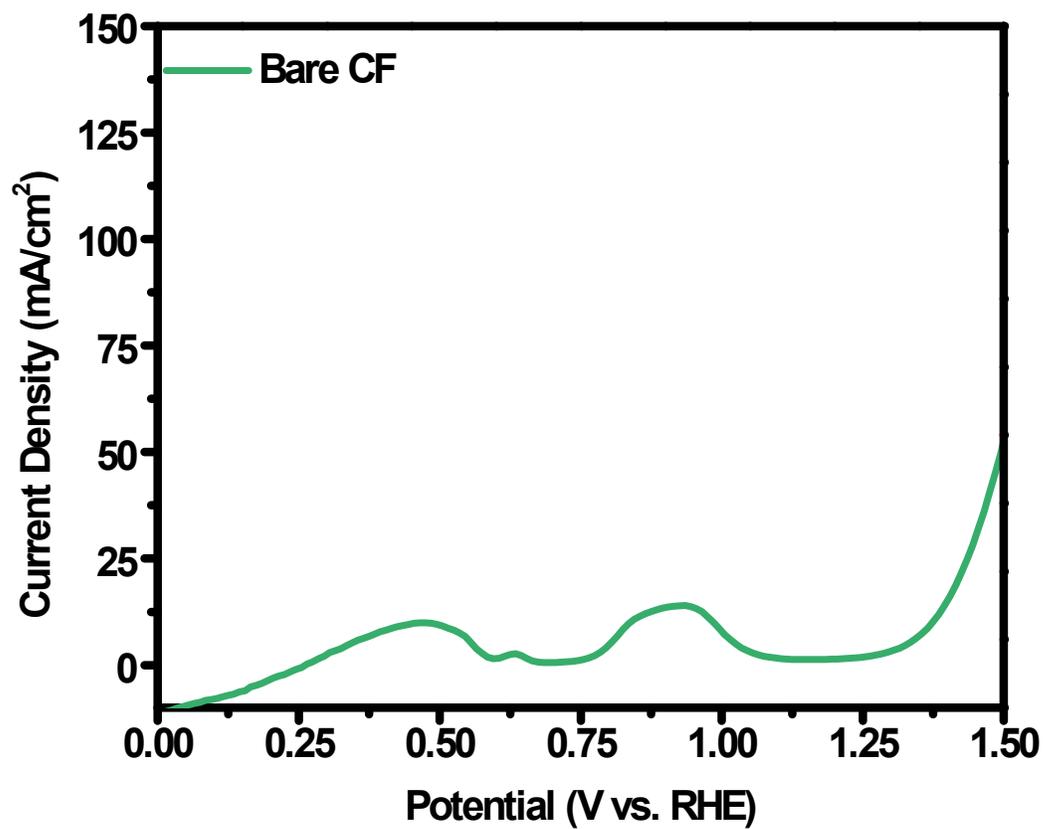


Fig. S8: Linear Polarization Curves for FOR of Cu foam.

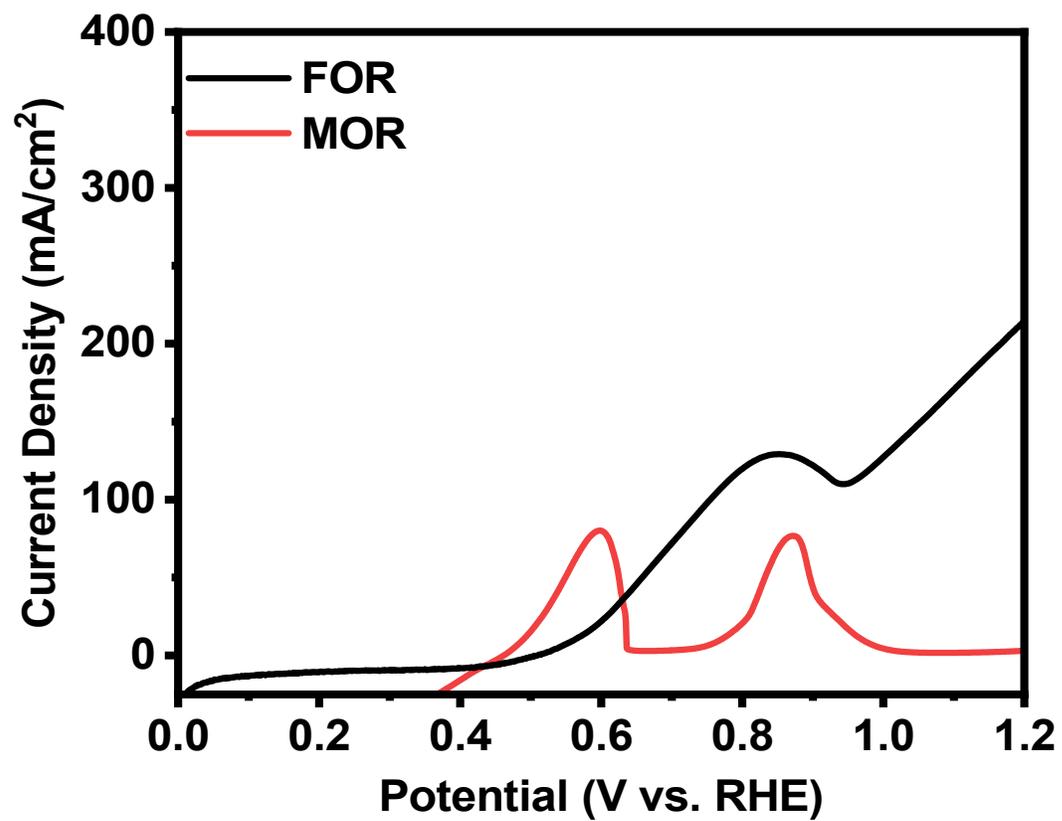


Fig. S9; Comparison of anodic activity for FOR and MOR of S/Cu_xO/CF catalyst.

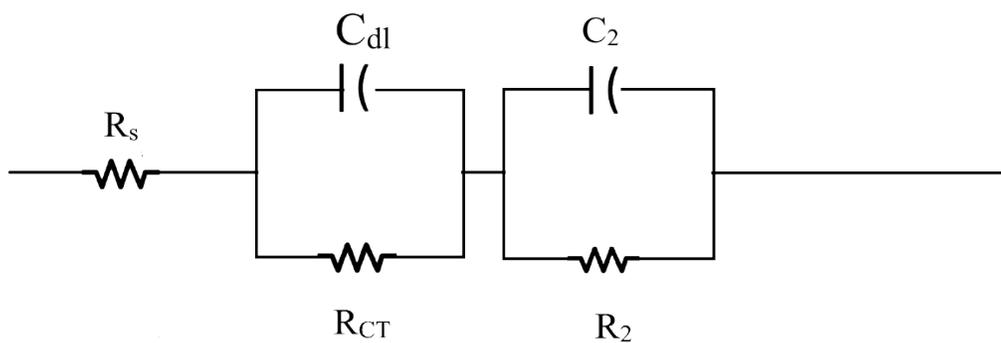
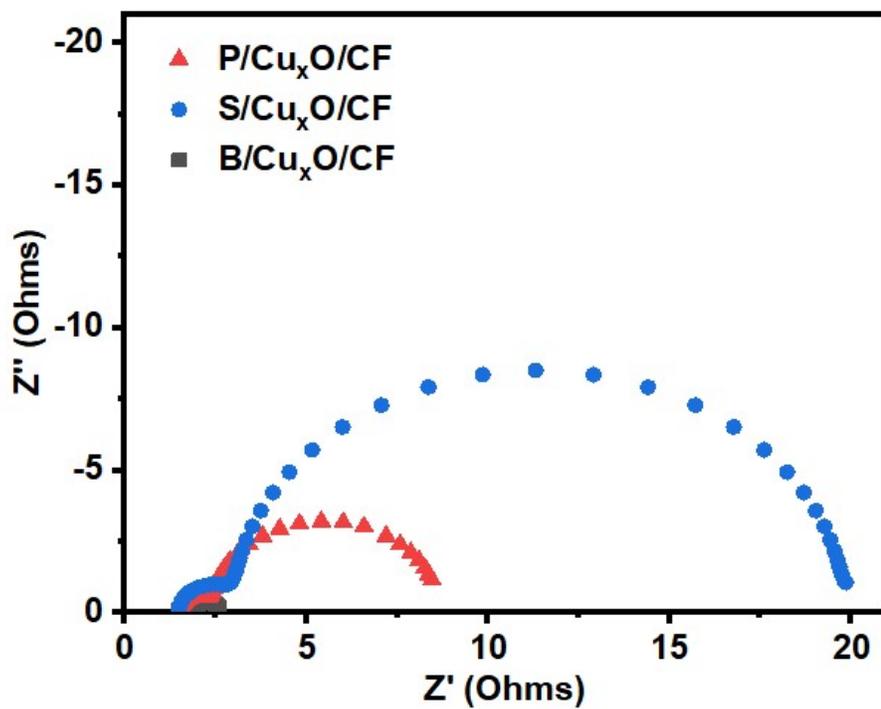
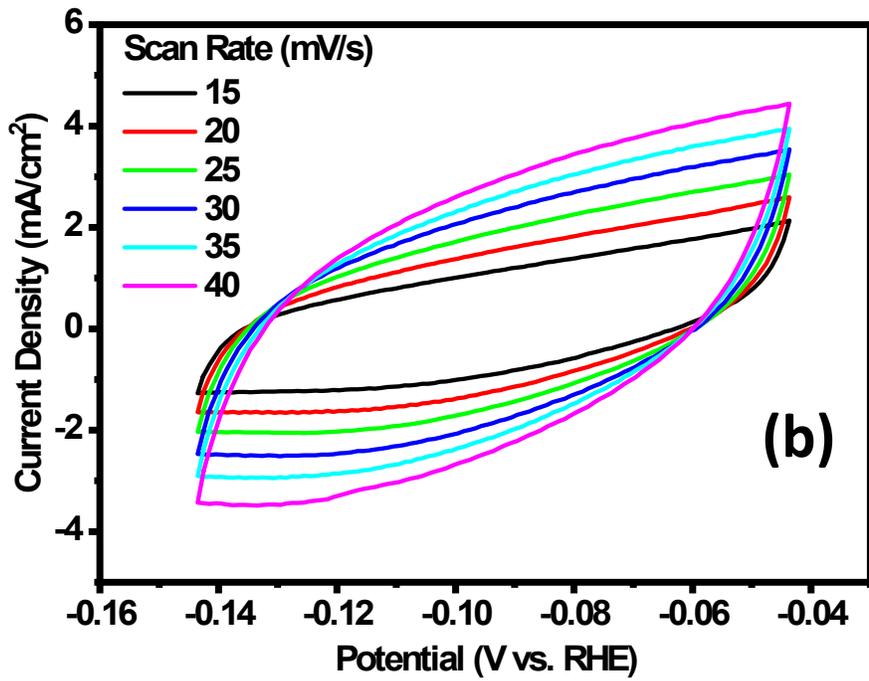
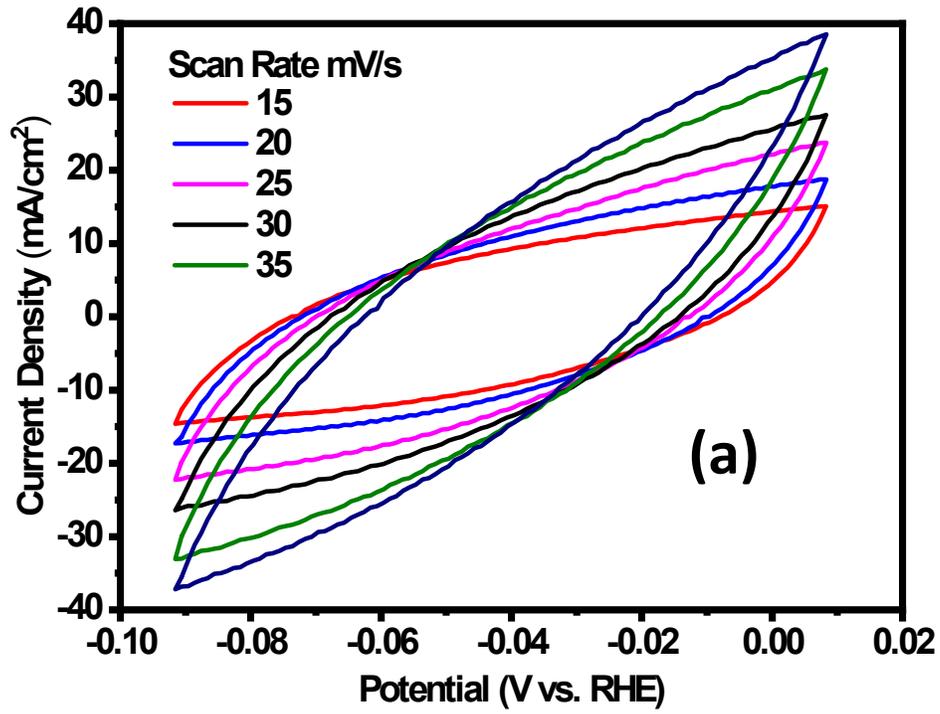


Fig. S10: EIS spectra for FOR of B, P, and S/Cu_xO/CF electrocatalysts and equivalent circuit for fitting and interpretation of data.



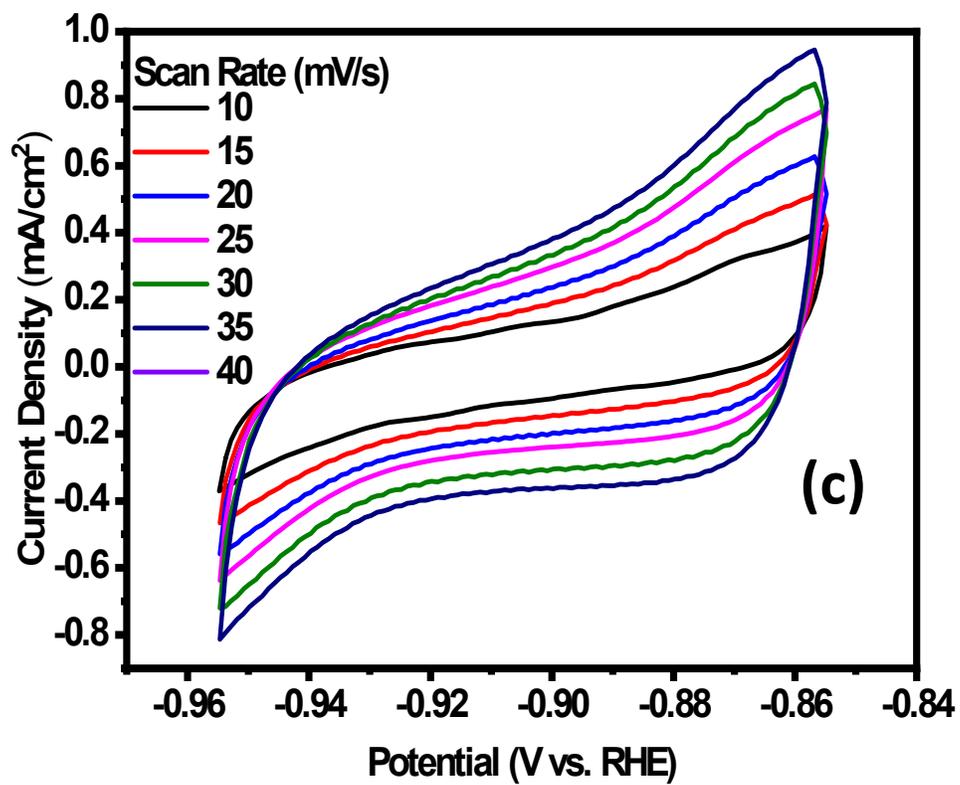


Fig. S11: Cyclic voltammetry curves for (a) B/Cu_xO/CF, (b) P/Cu_xO/CF, and (c) S/Cu_xO/CF electrocatalysts.

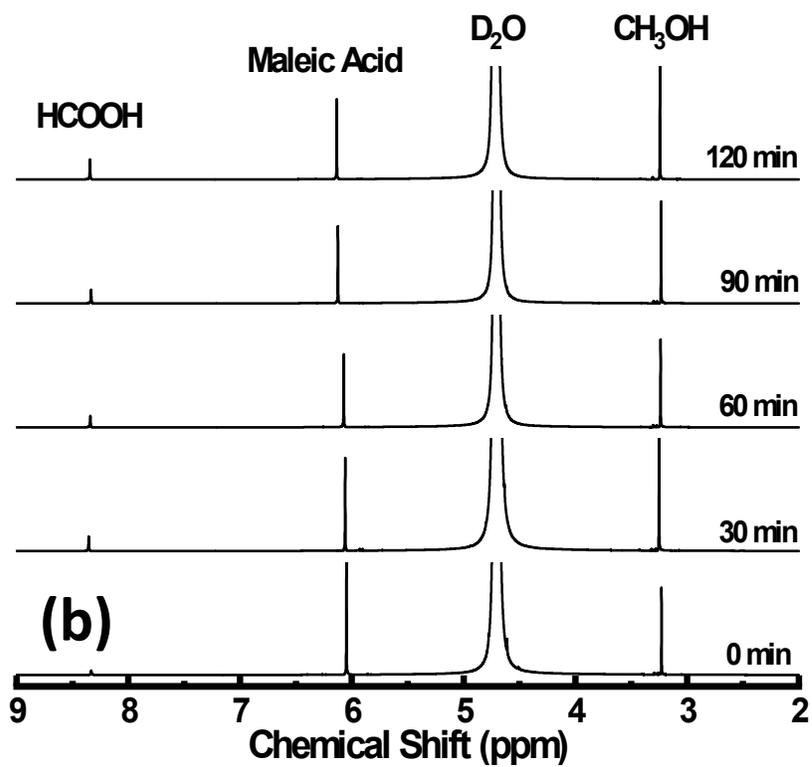
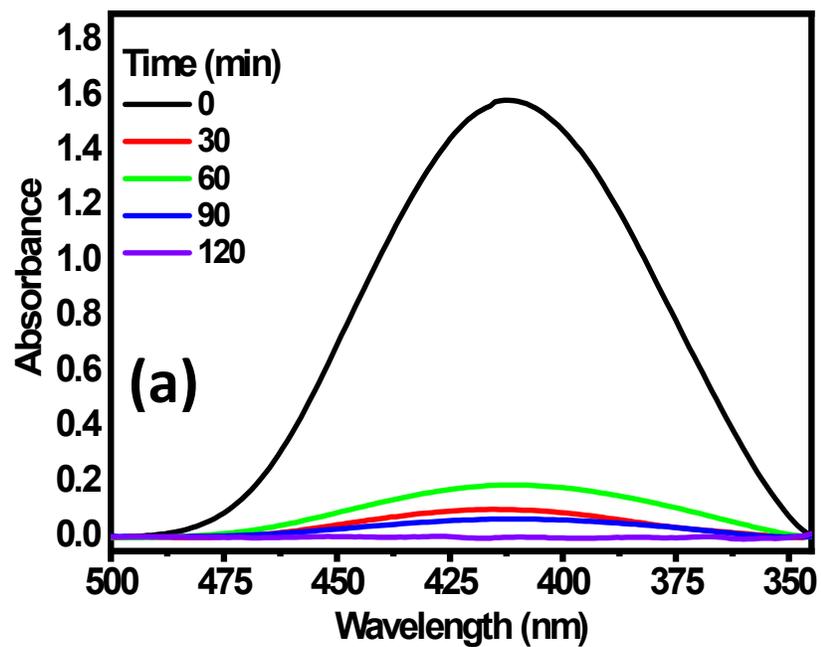


Fig. S12 (a): UV-Vis spectroscopy and (b) ¹H NMR spectroscopy at simultaneous periodic intervals during electrolysis test.

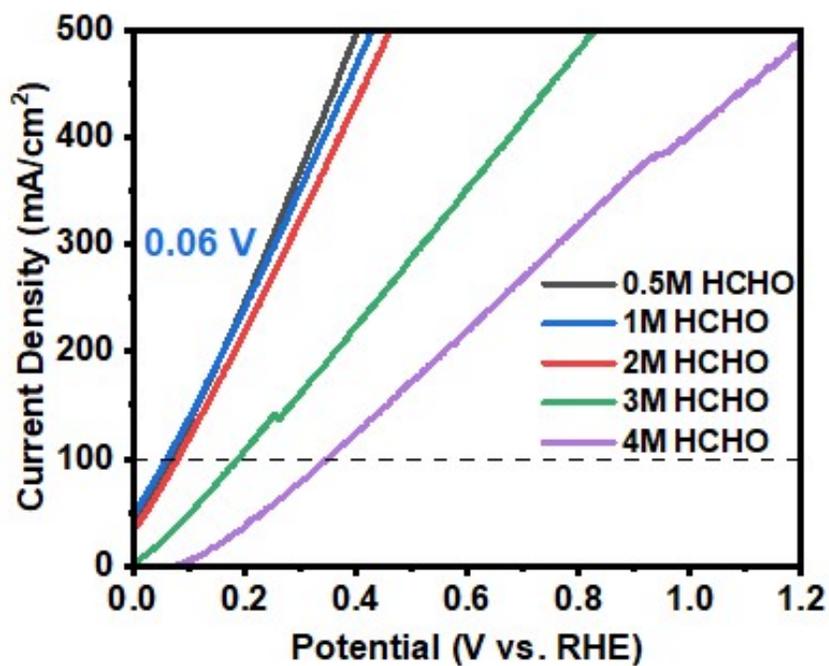


Fig. S13; FOR activity of B/Cu_xO/CF for different HCHO concentrations.

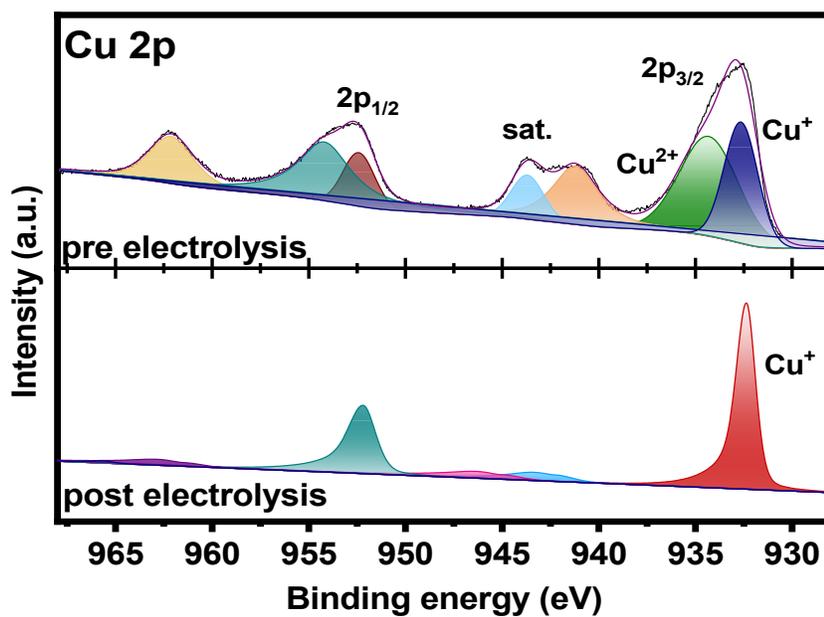


Fig. S14: Pre and Post FOR XPS spectra of B/Cu_xO/CF electrocatalyst

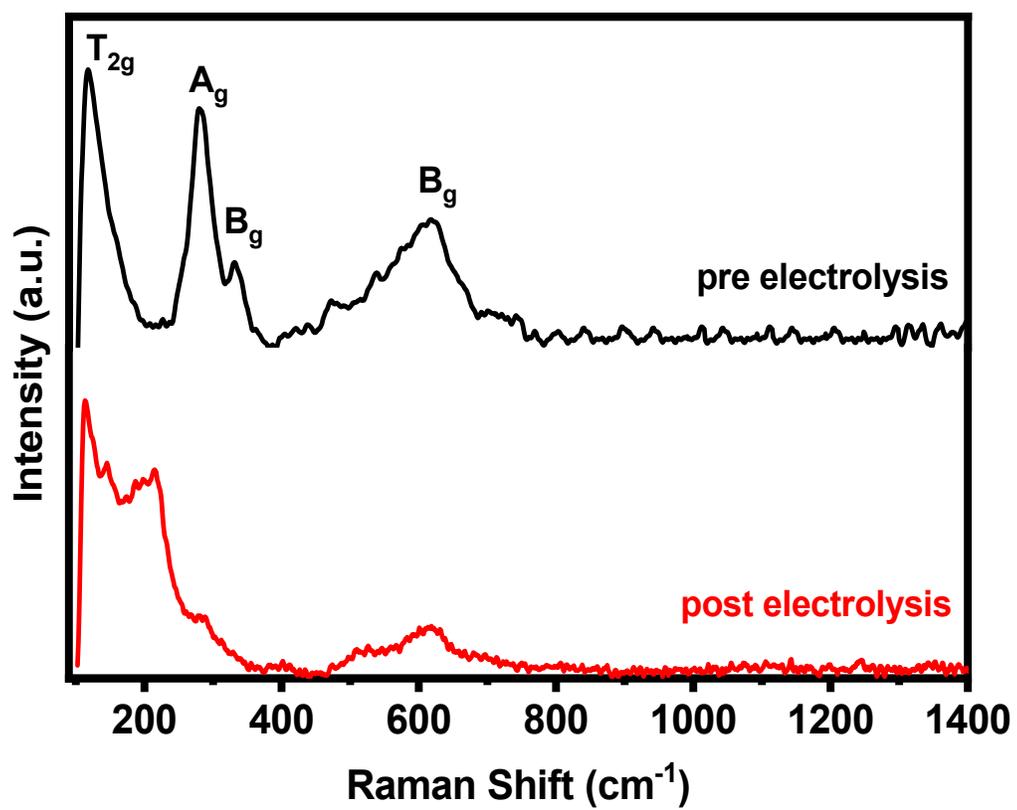


Fig. S15: Pre and Post FOR Raman spectra of B/Cu_xO/CF electrocatalyst.

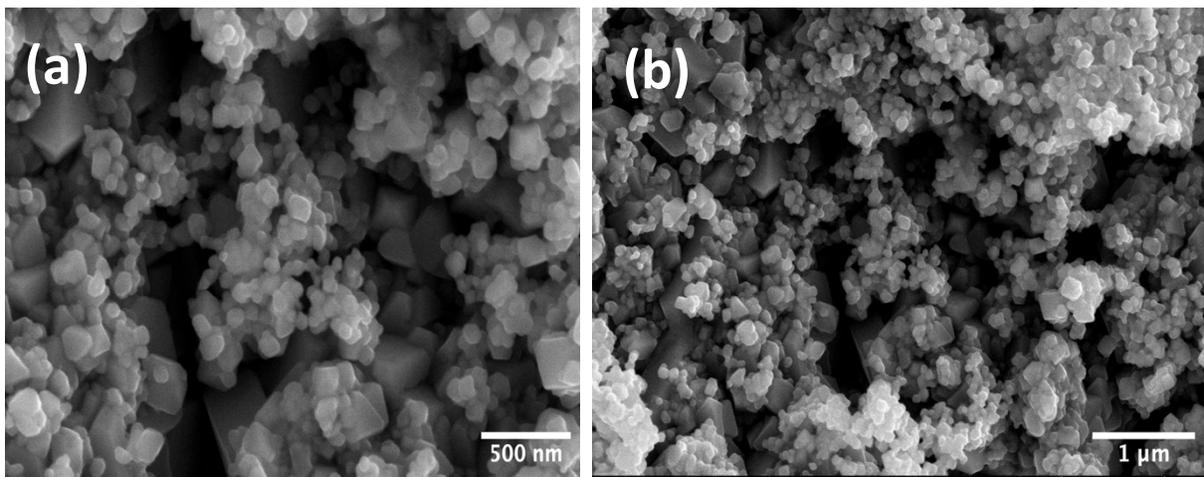


Fig. S16: Post FOR SEM images of B/Cu_xO/CF electrocatalyst.

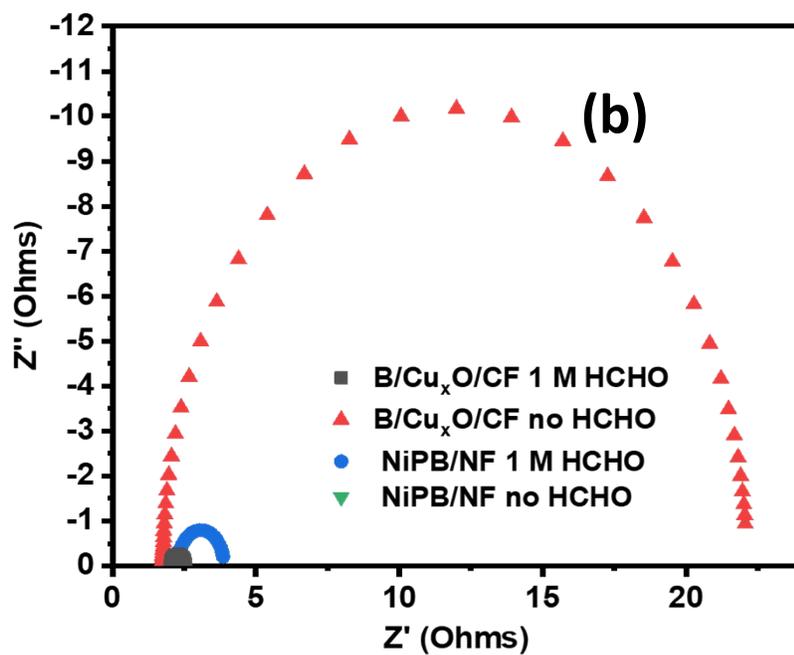
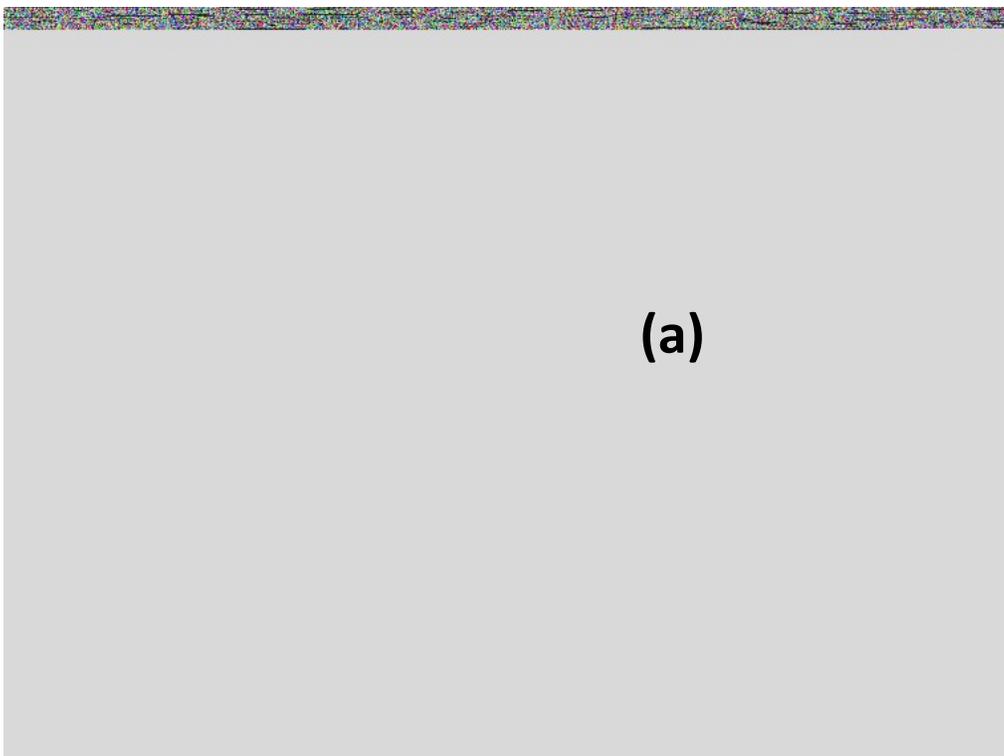


Fig. S17: (a) Nyquist plots of B, S, and P/Cu_xO/CF electrocatalysts for HER in 1 M HCHO + 1 M KOH. (b) Nyquist plots of B/Cu_xO/CF and NiPB/NF electrocatalysts for HER with and without the presence of HCHO.

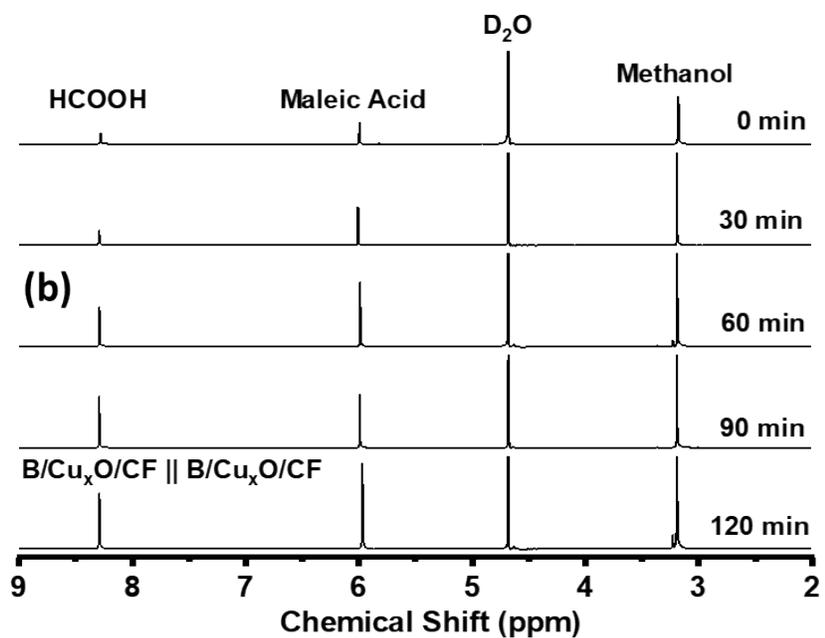
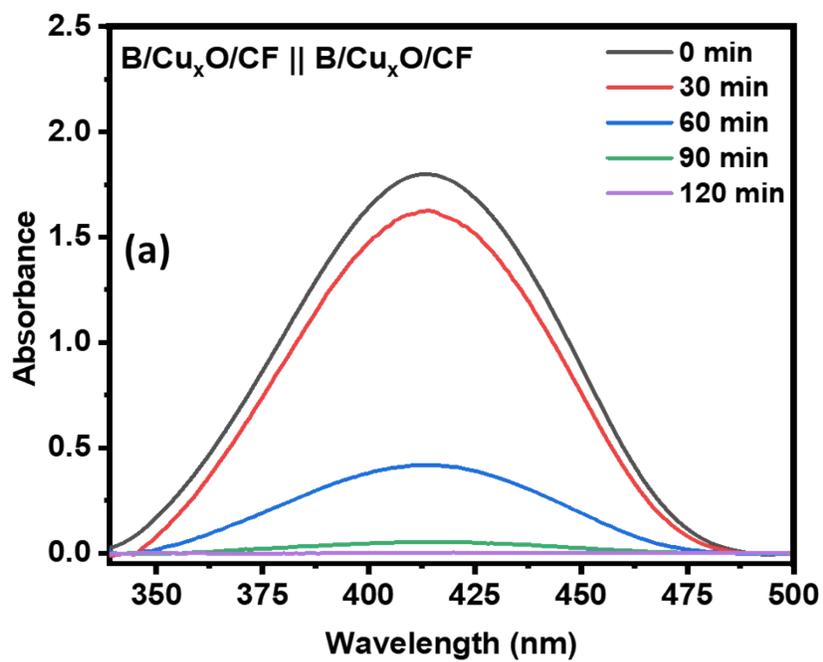


Fig. S18; (a) UV-Vis spectroscopy and (b) ¹H NMR spectroscopy at simultaneous periodic intervals during electrolysis test for B/Cu_xO/CF || B/Cu_xO/CF

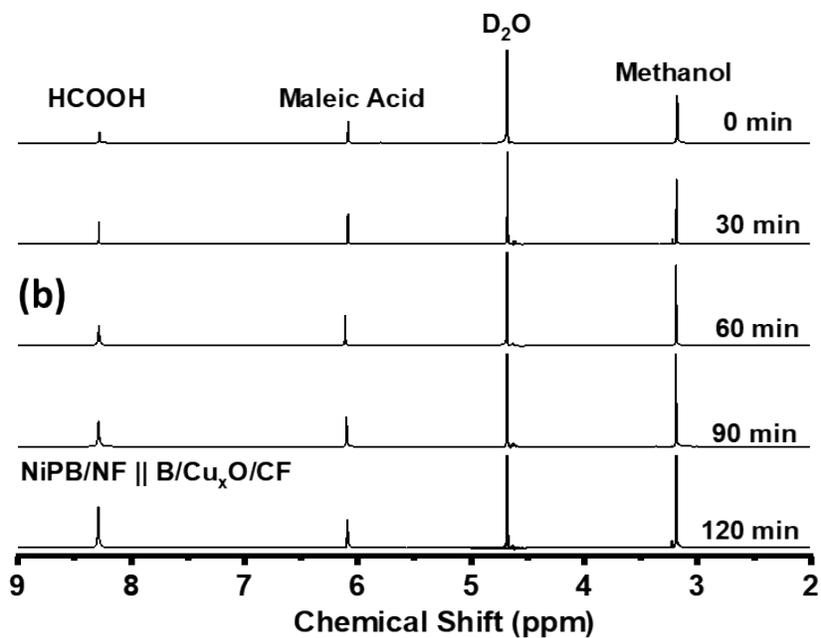
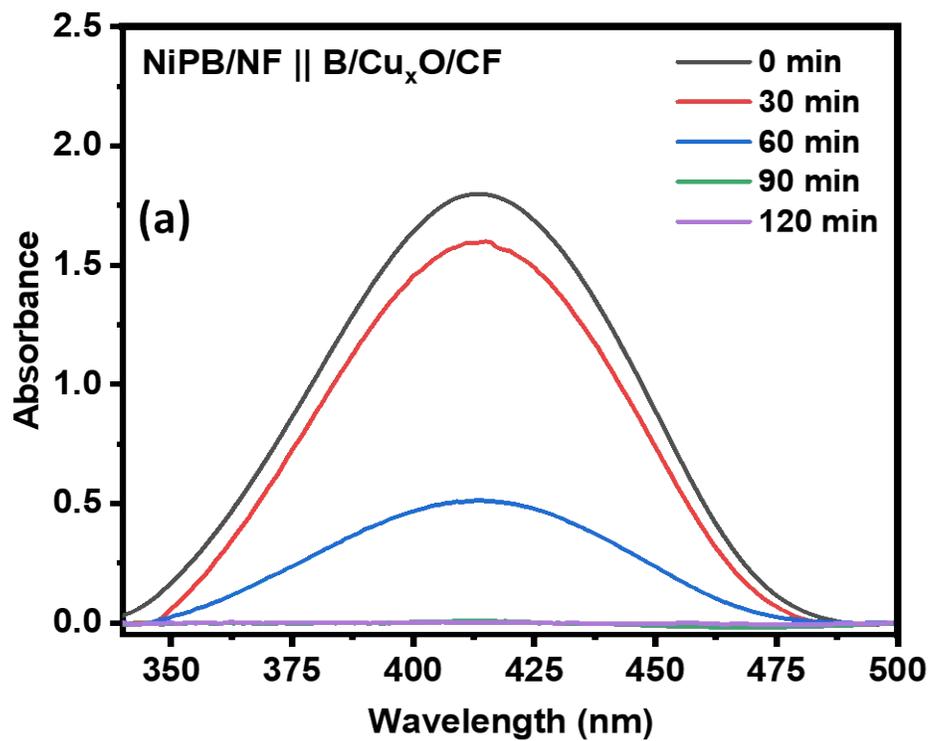


Fig. S19; (a) UV-Vis spectroscopy and (b) ¹H NMR spectroscopy at simultaneous periodic intervals during electrolysis test for NiPB/NF || B/Cu_xO/CF

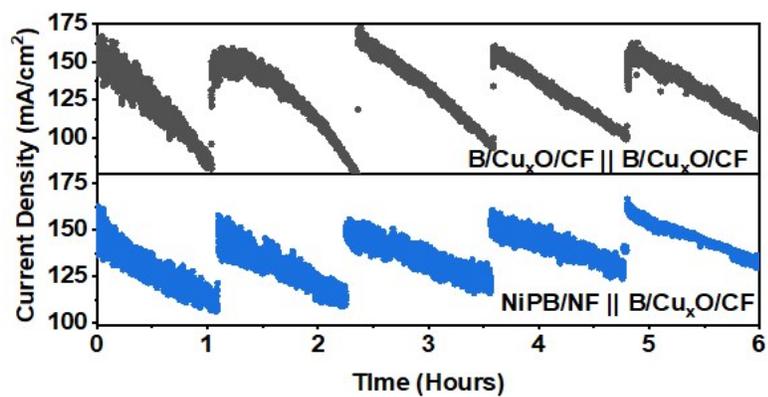


Fig. S20; Chronoamperometry tests using a two electrode system (B/Cu_xO/CF || B/Cu_xO /CF and NiPB/NF || B/Cu_xO /CF) with hourly periodic replenishment of HCHO.

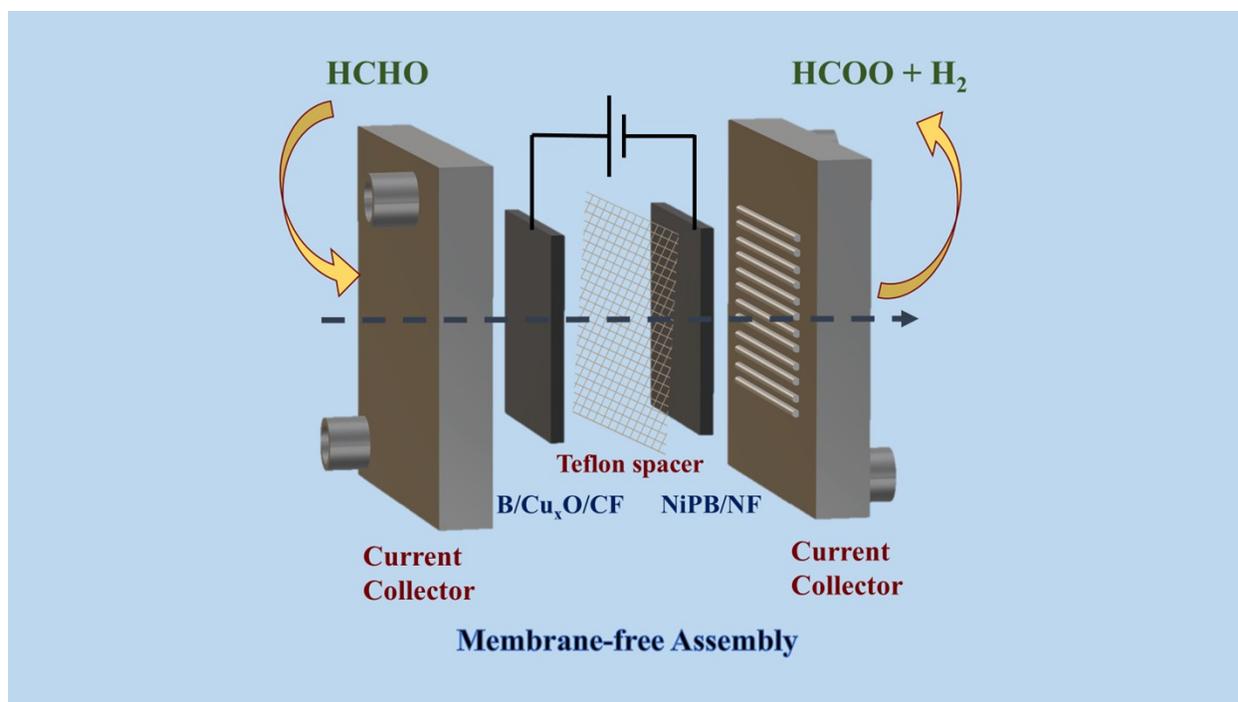


Fig. S21; Schematic of membrane-free assembly of asymmetric setup (NiPB/NF || B/Cu_xO/CF)

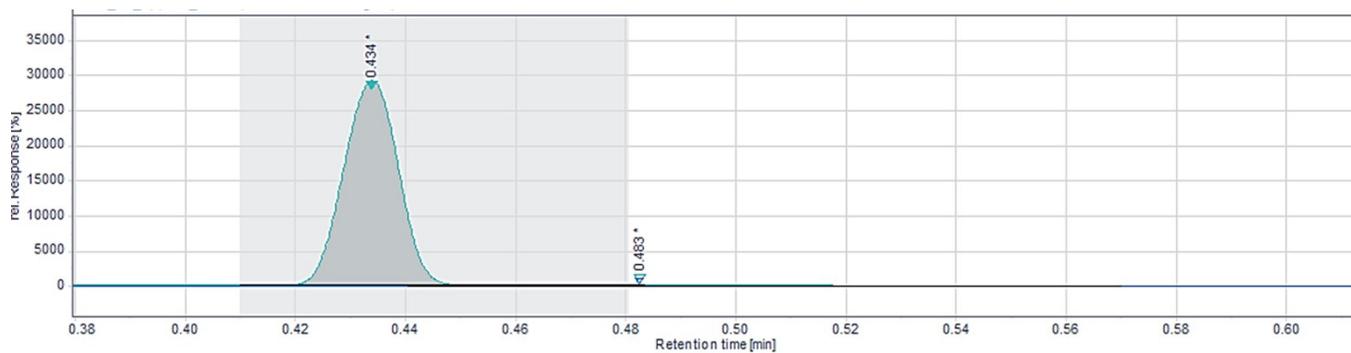


Fig S22; Gas chromatographic analysis of the product stream, for NiPB/NF || B/Cu_xO/CF system

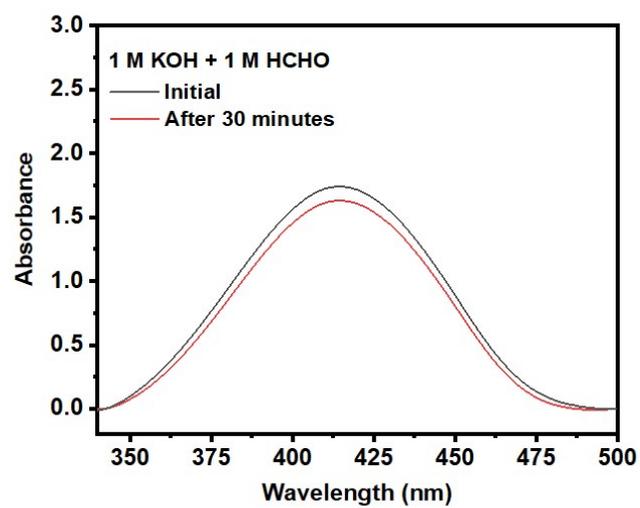


Fig. S23; UV- Vis spectra of spontaneous HCHO degradation over 30 min