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Supporting Information

Synergistic Regulation of Dual Monomers for Constructing High Bandgap and High Polarity Polyimide with High-Temperature Capacitive Energy Storage Performance

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1 **Characterization:**

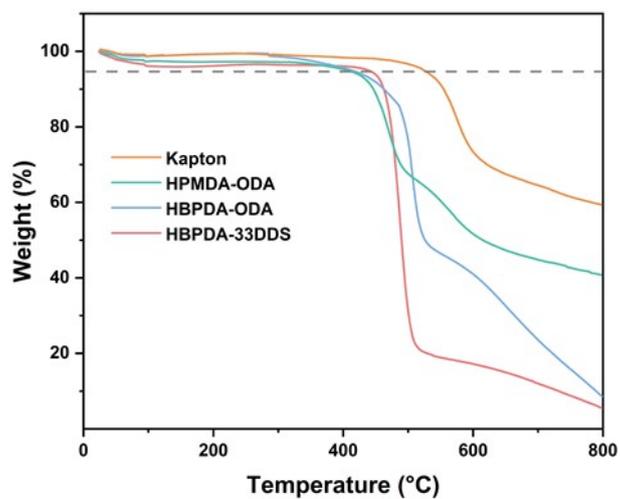
2 The Fourier infrared spectroscopy (FT-IR) curves were measured on a Bruker alpha.
3 The UV-vis spectra were measured on a PerkinElmer Lambda3S, Fluorescence spectra
4 were measured by using a research grade FLSP 920 fluorescence spectrometer
5 (Edinburgh Instruments, UK). Thermal gravimetric analyzer (TGA) curves were
6 carried out on a Mettler Toledo TGA2 with a heating rate of 10 °C/min under a nitrogen
7 atmosphere. Differential Scanning Calorimetry curves were carried on a TA DSC
8 Q2000 with a heating rate of 10 °C/min under a nitrogen atmosphere. First, the
9 temperature was raised from 50 °C to 250-400 °C, held for 5 minutes, and then cooled
10 back to 50 °C at the same rate to eliminate the thermal history. Finally, the temperature
11 was increased again to 250-400 °C, and the second heating data was used to determine
12 the glass transition temperature. XRD tests were performed using Bruker D8
13 ADVANCE. with measuring angle from 5 ° to 70 °, equipped with a Cu K α source (λ
14 = 1.54 Å). Mechanical properties of the specimens were characterized on an Instron
15 5869 universal testing machine (Instron, USA) at a strain rate of 20 mm/min. Prior to
16 electrical testing, copper (Cu) electrodes were deposited on both surfaces of the PI films
17 via thermal evaporation. For the measurement of electric displacement–electric field
18 (*D-E*) loops and dielectric breakdown strength, circular electrodes with a diameter of
19 3.4 mm were used, corresponding to an effective area of 9.08 mm². For the
20 measurement of leakage currents, direct current conductivity, and thermally stimulated
21 depolarization current (TSDC) profiles, electrodes with a diameter of 7 mm were
22 employed, providing an effective area of 38.48 mm². Dielectric properties were
23 performed through an Agilent LCR instrument (4294A). Electric displacement electric
24 field (*D-E*) loops were obtained by PolyK (PK CPE20B) using a modified Sawyer-
25 Tower circuit, in which the dielectric films were exposed to a triangular unipolar wave
26 at a frequency of 100 Hz. Dielectric breakdown strength was recorded on a Trek 610C
27 at a voltage rise rate of 500 V/s. The experiment was conducted in a silicone oil bath to
28 reduce surface corona discharge. The leakage currents, direct current conductivity, and
29 TSDC profiles were evaluated using an electrometer (Keithley 6517A). The leakage

1 current was collected with an electric field gradient of 25 MV/m. The TSDC
2 characterization procedures are as follows: after a 30-minute polarization at 250 °C and
3 rapid cooling to -10 °C for minutes, the specimens were short-circuited, and the current
4 was recorded as the temperature increased from -10 to 250 °C with the heating rate of
5 5 °C/min.

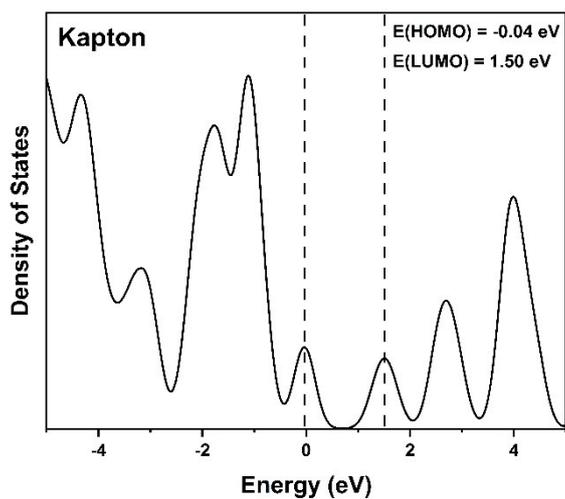
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7 **Density functional theory (DFT) calculation:**

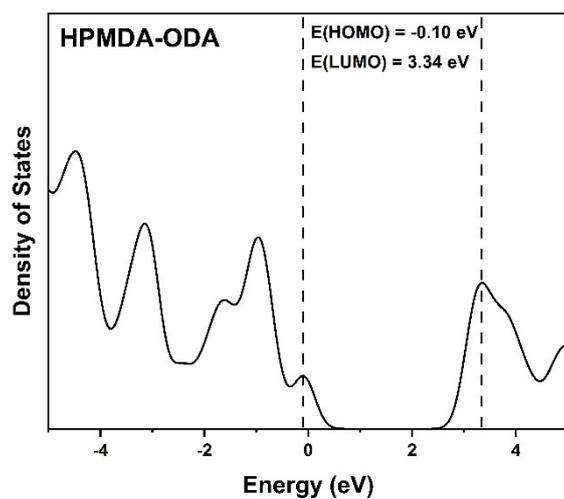
8 Amorphous cells were constructed using Materials Studio (MS) software, with each
9 cell containing 5 molecular chains and a polymerization degree of 12. Molecular
10 dynamics simulations were performed on the amorphous cells of all polymers under
11 isothermal (NVT) and isoenergetic equilibration (NVE) conditions, and at least five
12 annealing processes were applied to each. Density functional theory (DFT) calculations
13 were conducted using the Materials Studio (MS). The electron exchange and correlation
14 energy were treated within the generalized gradient approximation (GGA) with the
15 Perdew-Burke-Ernzerhof (PBE) functional, and a plane-wave basis set with a cutoff
16 energy of 500 eV was employed. The energy calculation and geometry optimization
17 were terminated when the electronic energy tolerance reached 1×10^{-5} eV/atom and the
18 maximum force component on any atom was less than 0.02 eV/Å, respectively. The
19 band gap of the materials was determined by calculating the highest occupied molecular
20 orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) from the density of
21 states. To account for van der Waals interactions, the empirical correction from the
22 Grimme scheme (DFT-D3) was adopted, and the charge density difference was
23 visualized using VMD software. Additionally, the Gaussian16 package was used to
24 optimize the geometric structures of different polymer molecules at the B3LYP/6-31G*
25 level of theory in a vacuum, and the wavefunction results were analyzed using Multiwfn
26 software.



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 2 **Fig. S1.** Thermal gravimetric analysis curves of Kapton, HPMDA-ODA, HBPDA-
 3 ODA and HBPDA-33DDS.

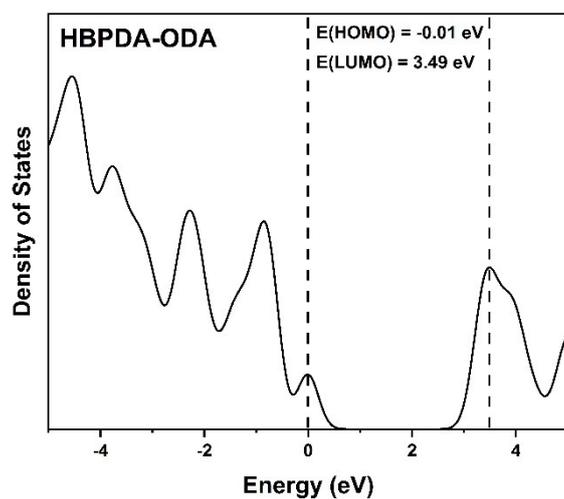


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 5 **Fig. S2.** The results of the density of states (DOS) calculations for Kapton.



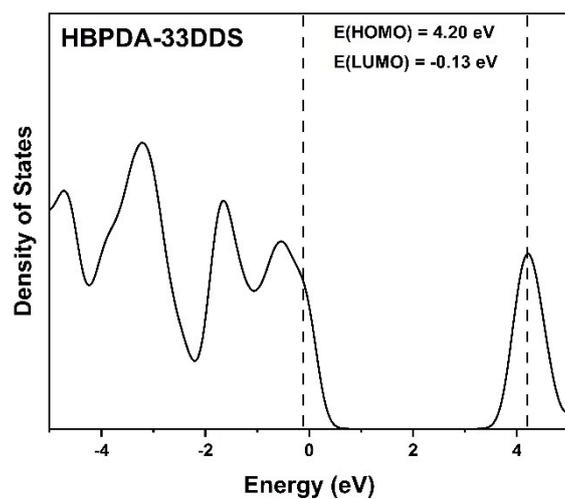
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2 **Fig. S3.** The results of the density of states (DOS) calculations for HPMDA-ODA.



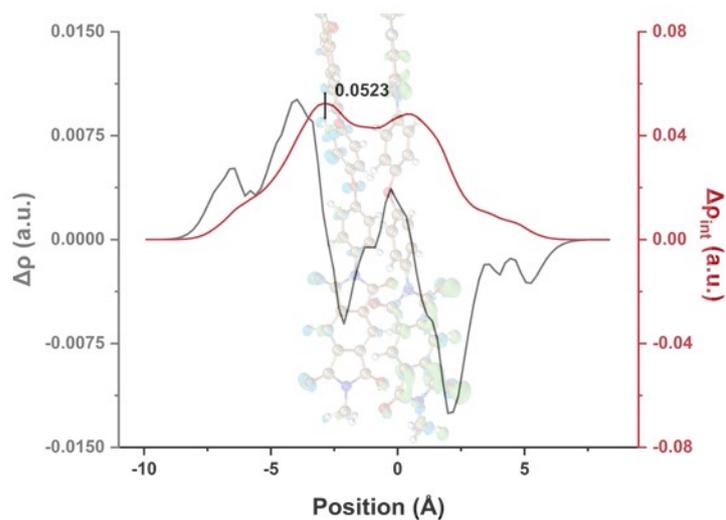
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4 **Fig. S4.** The results of the density of states (DOS) calculations for HBPDA-ODA.



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2 **Fig. S5.** The results of the density of states (DOS) calculations for HBPDA-33DDS.

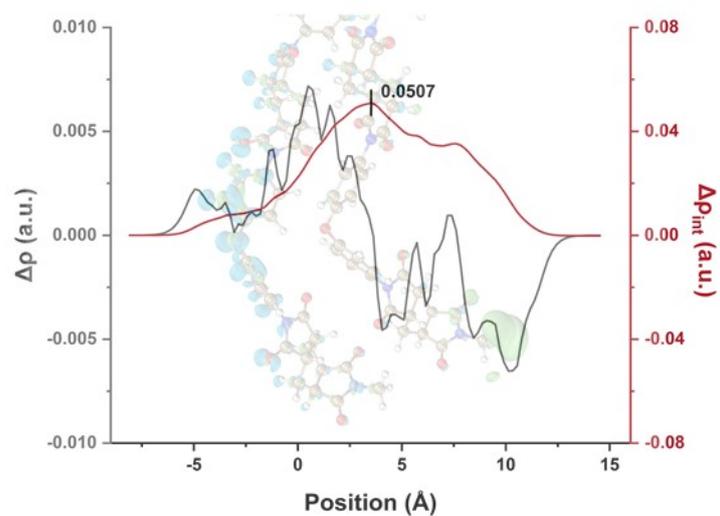


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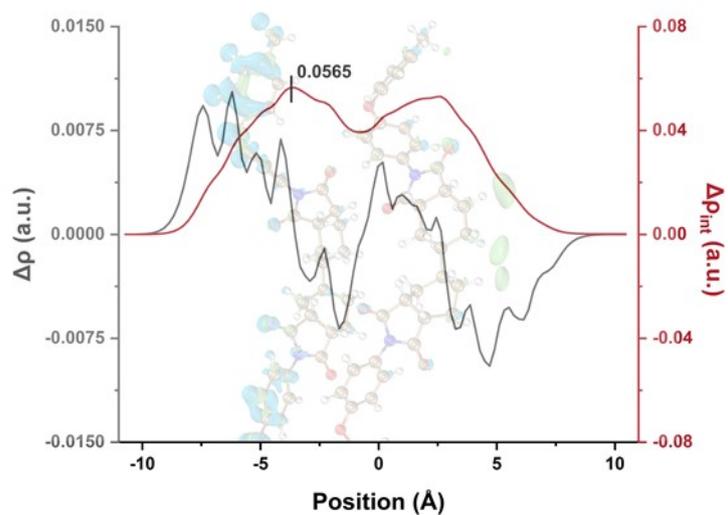
4 **Fig. S6.** The differential charge densities of Kapton before and after applying an

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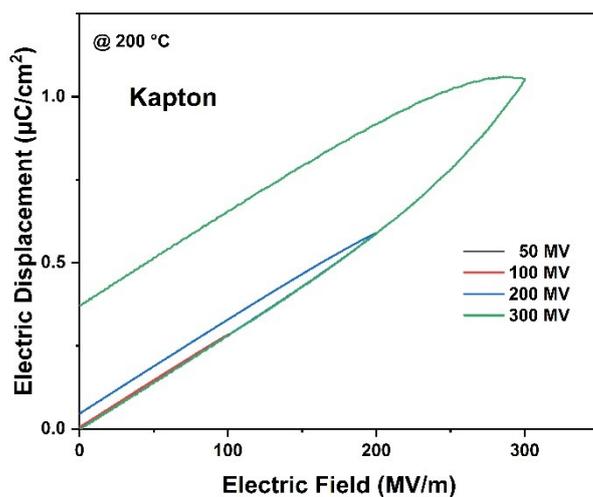
electric field of 400 MV/m.



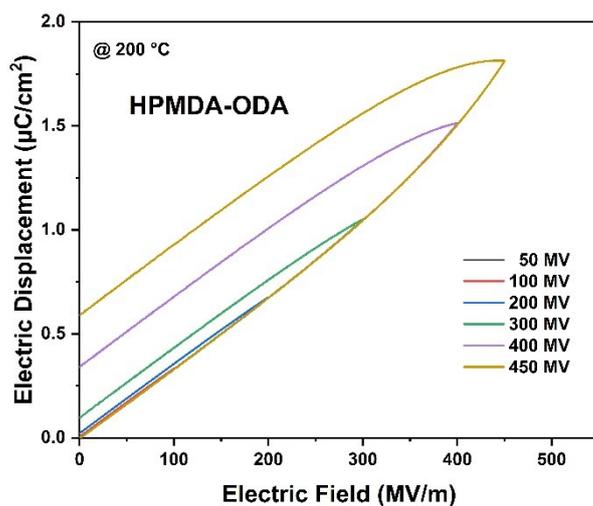
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 2 **Fig. S7.** The differential charge densities of HPMDA-ODA before and after applying
 3 an electric field of 400 MV/m.



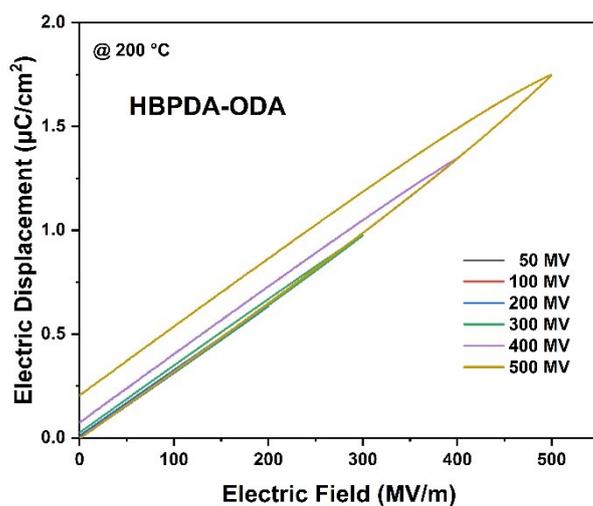
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 5 **Fig. S8.** The differential charge densities of HBPDA-ODA before and after applying
 6 an electric field of 400 MV/m.



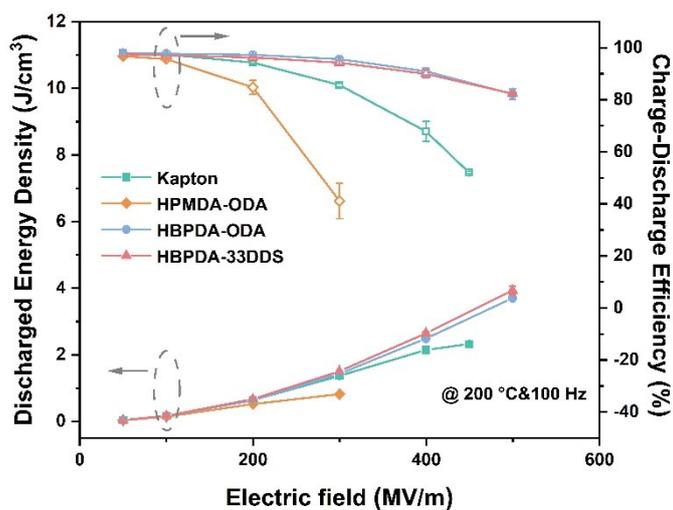
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 2 **Fig. S9.** Electric displacement electric field (D-E) loops of Kapton at 200 °C and 100
 3 Hz.



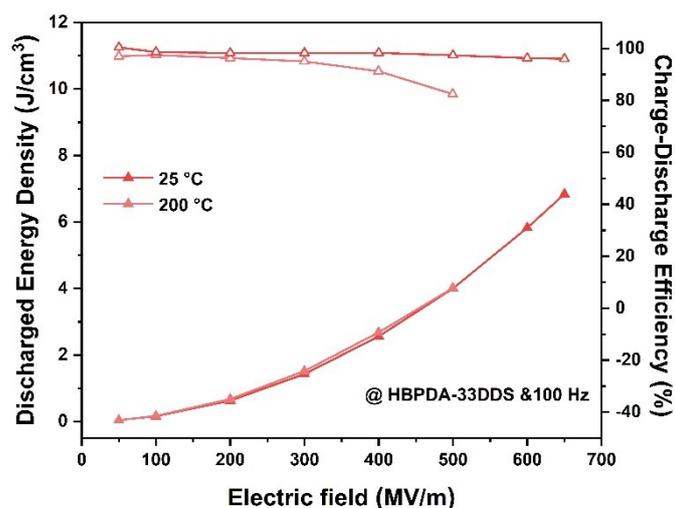
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 5 **Fig. S10.** Electric displacement electric field (D-E) loops of HPMDA-ODA at 200 °C
 6 and 100 Hz.



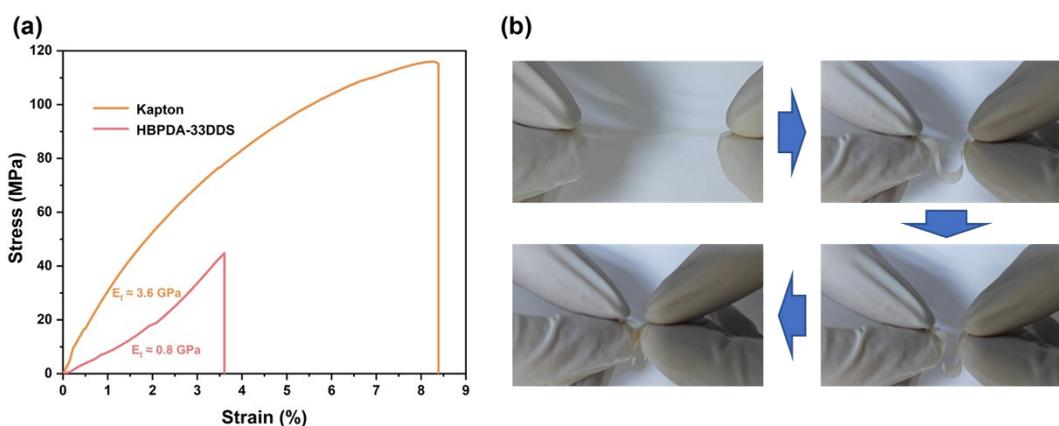
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 2 **Fig. S11.** Electric displacement electric field (D-E) loops of HBPDA-ODA at 200 °C
 3 and 100 Hz.



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 5 **Fig. S12.** Energy storage density and charge–discharge efficiency of the tested films
 6 at 200 °C and 100 Hz. Data points and error bars represent the mean values and
 7 standard deviations calculated from at least three independent samples, respectively.



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2 **Fig. S13.** Discharged energy density and charge-discharge efficiency of HBPDA-
3 33DDS at 25 °C and 200 °C (100 Hz).



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5 **Fig. S14.** Mechanical properties and flexibility of the polyimide films. (a) Tensile
6 stress–strain curves of the commercial Kapton and synthesized HBPDA-33DDS films.
7 (b) Digital photographs demonstrating the flexibility and structural integrity of the
8 standalone HBPDA-33DDS film under manual bending and stretching.