

Supplementary Materials for
Unraveling the Deactivation Mechanism and Stability Enhancement of Cobalt-
based Heterogenous Co₃O₄/g-C₃N₄ Catalysts in CO₂ Hydrogenation

The PDF file includes:

Figs. S1 to S23

Tab. S1

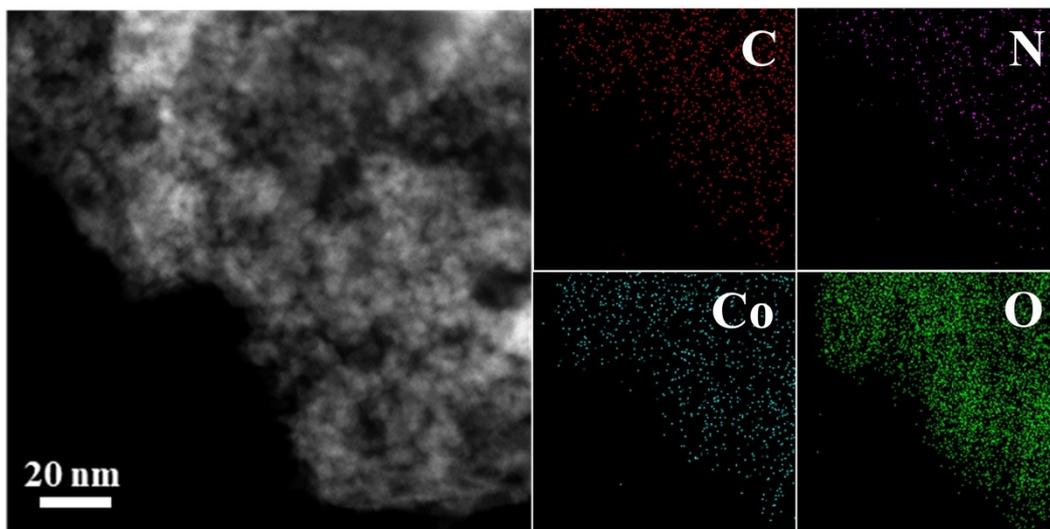


Figure S1. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-10$.

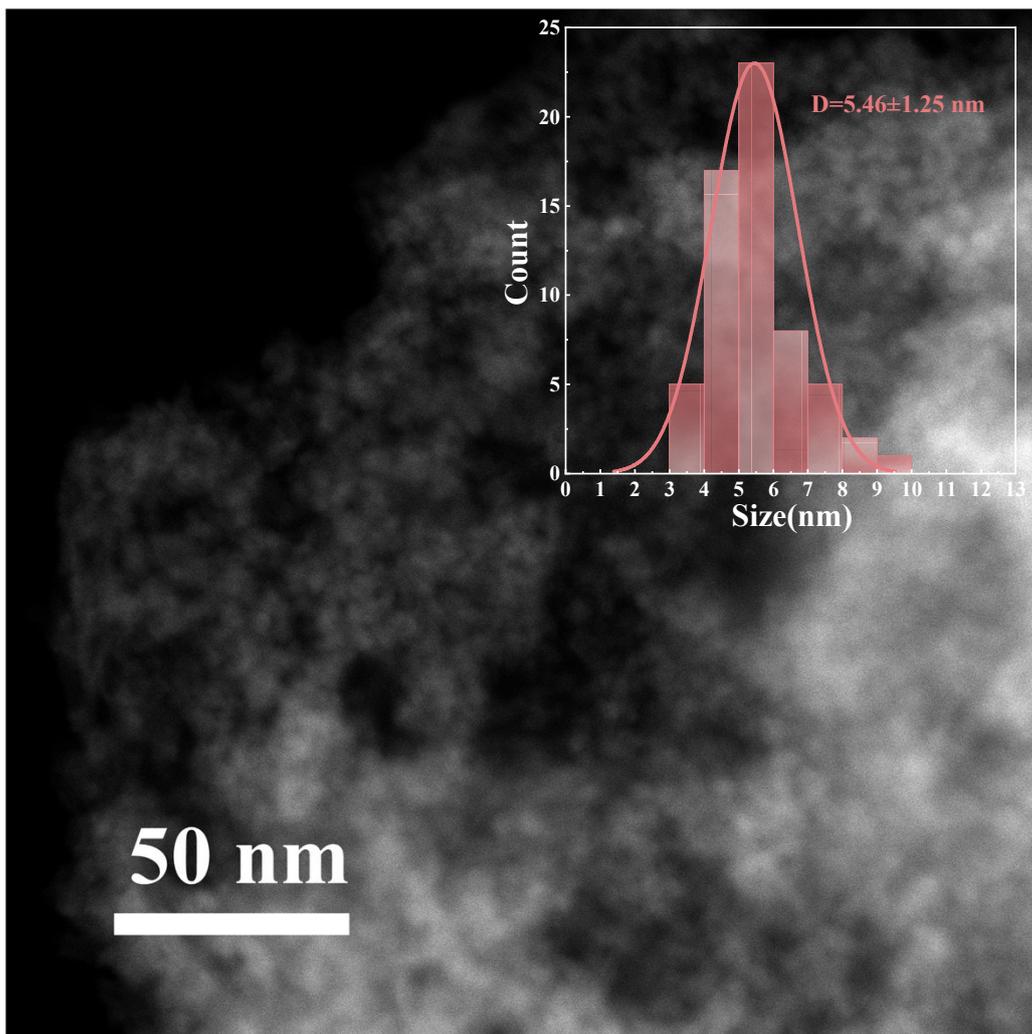


Figure S2. Particle size distribution of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-10$.

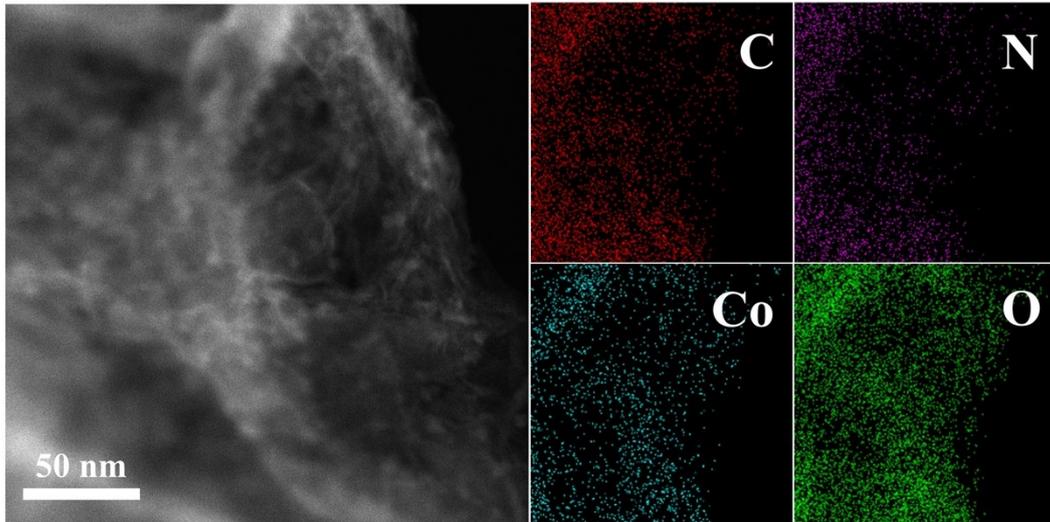


Figure S3. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4\text{-50}$.

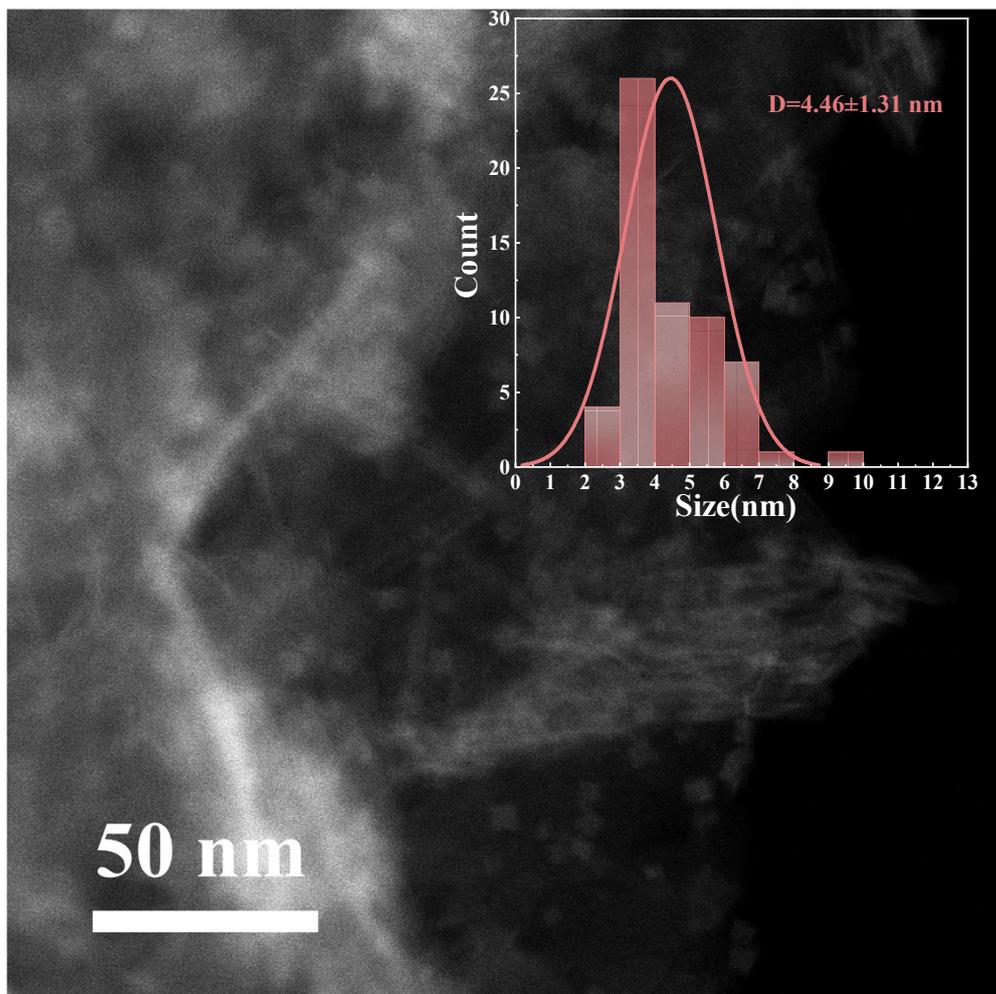


Figure S4. Particle size distribution of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-50$.

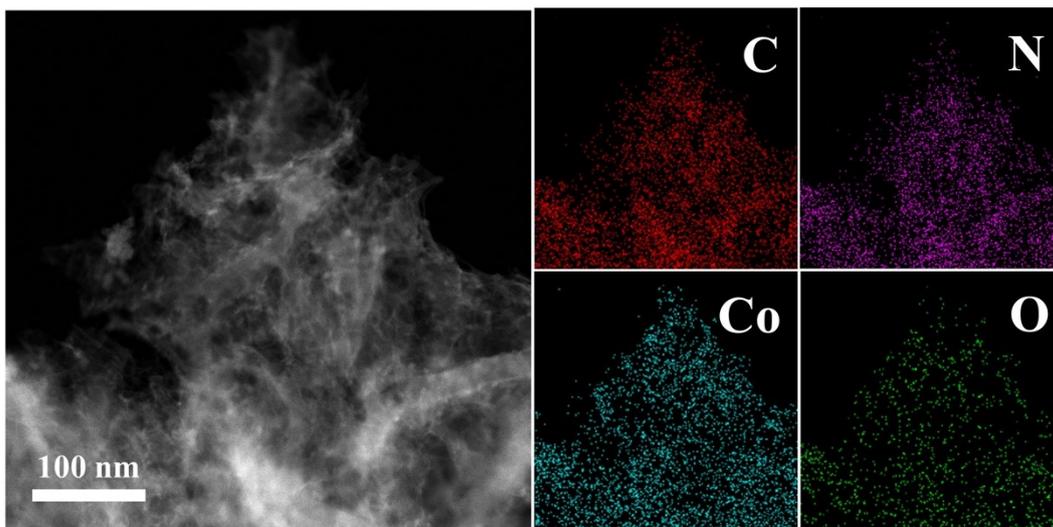


Figure S5. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4\text{-90}$.

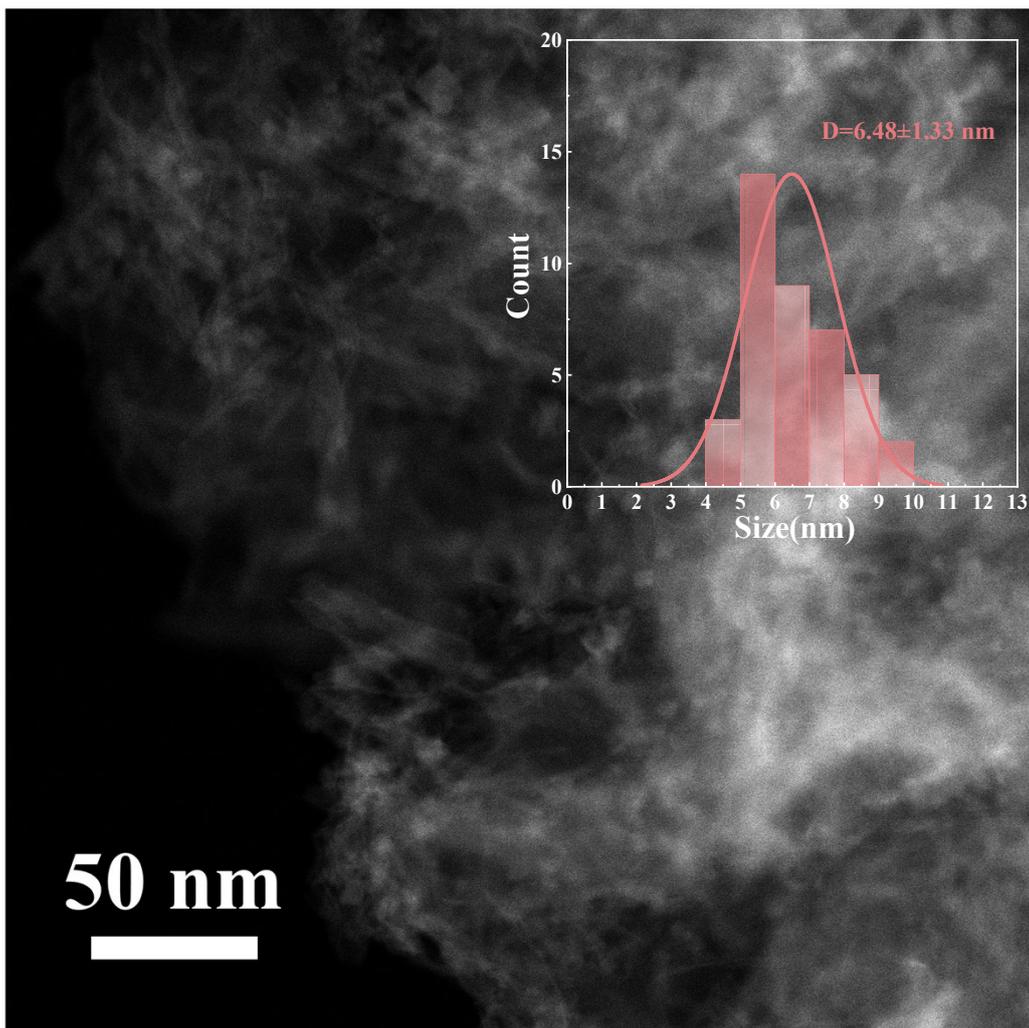


Figure S6. Particle size distribution of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-90$.

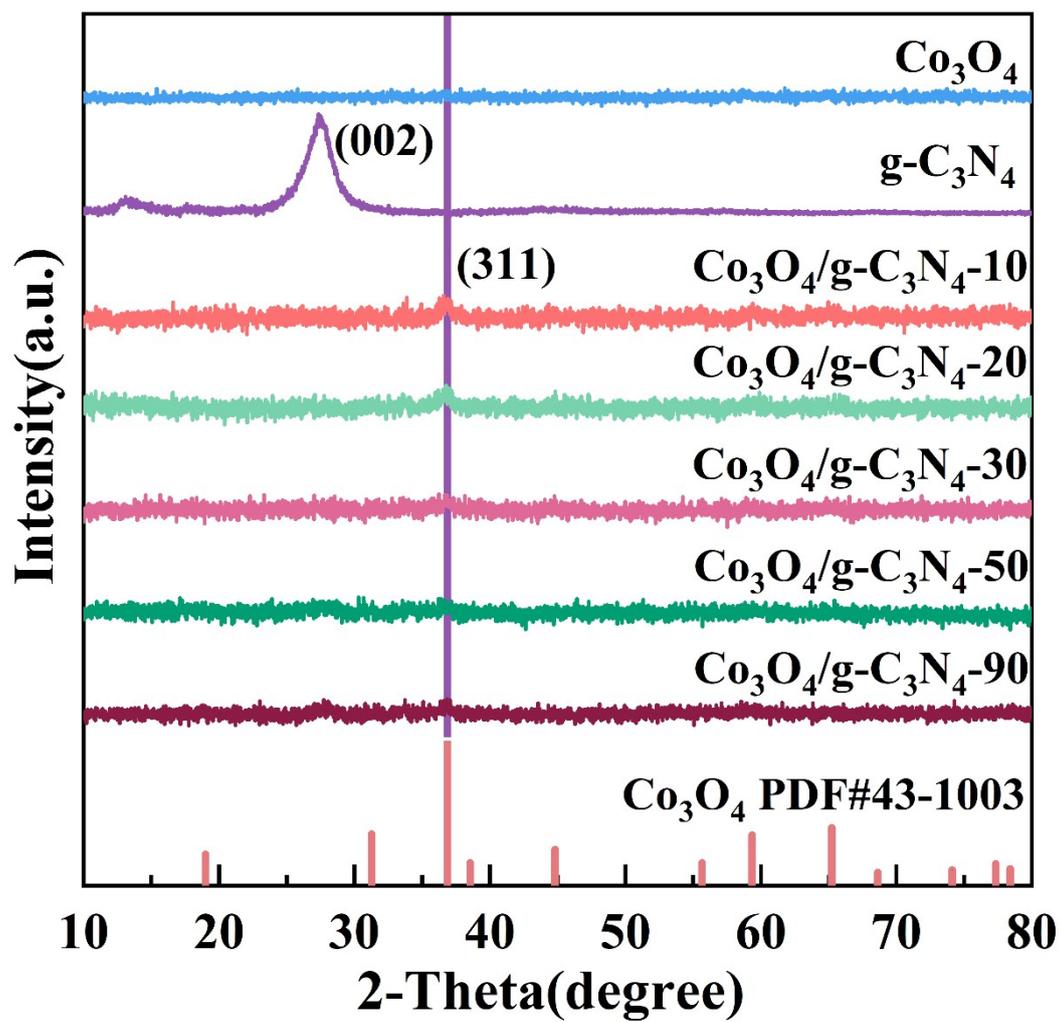


Fig. S7 XRD patterns of catalysts with different g-C₃N₄ loadings.

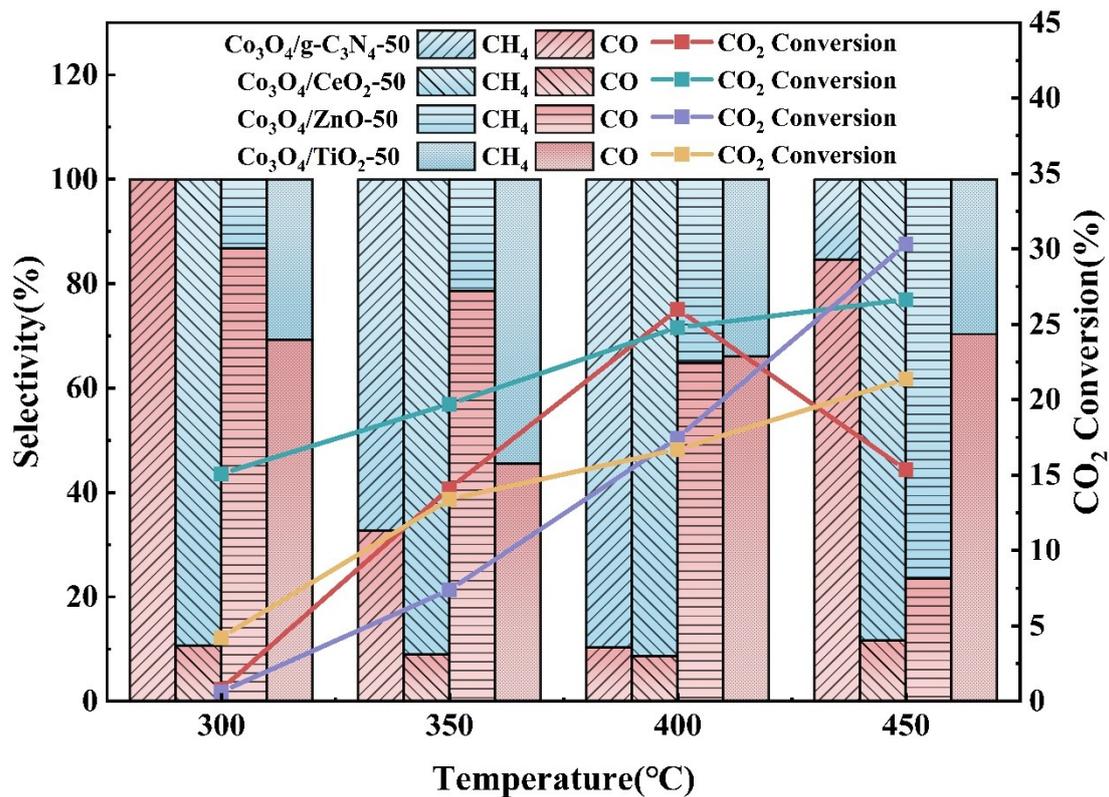


Figure S8. Catalytic performance of Co_3O_4 supported on $\text{g-C}_3\text{N}_4$, CeO_2 , ZnO and TiO_2 at different temperatures.

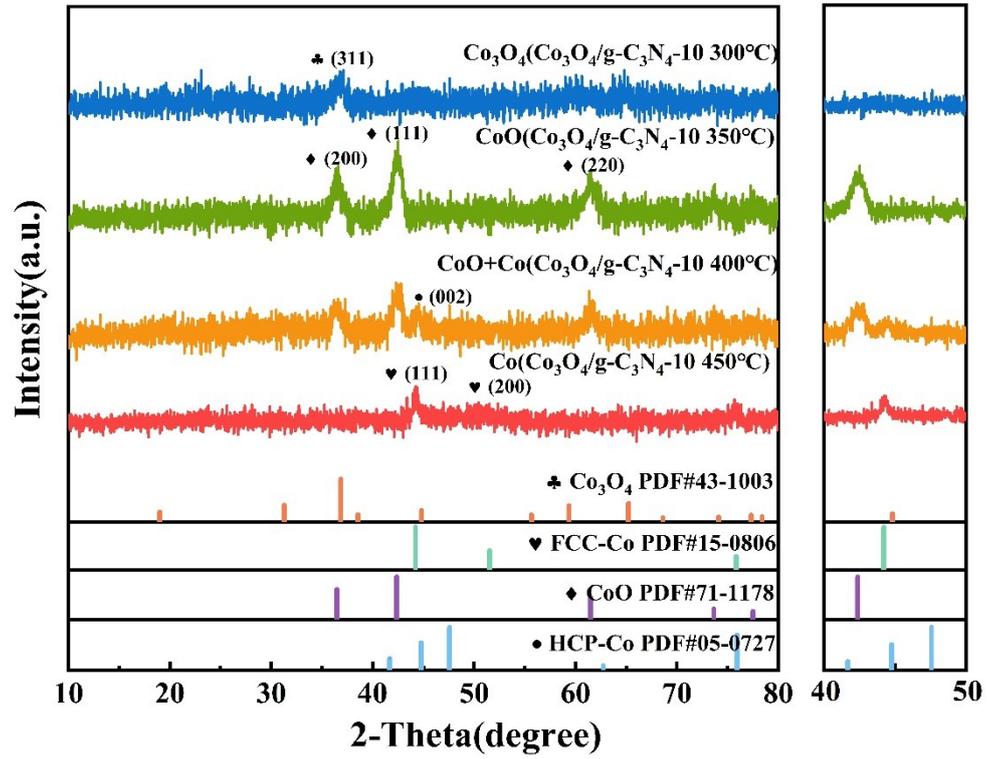


Figure S9. XRD of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4\text{-10}$ at different reaction temperatures.

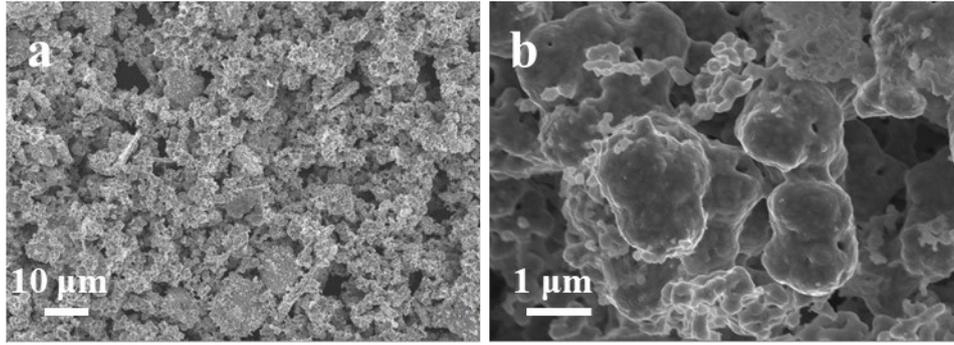


Figure S10. SEM image of the Co_3O_4 NPs after 20 h cycling stability at 450 °C

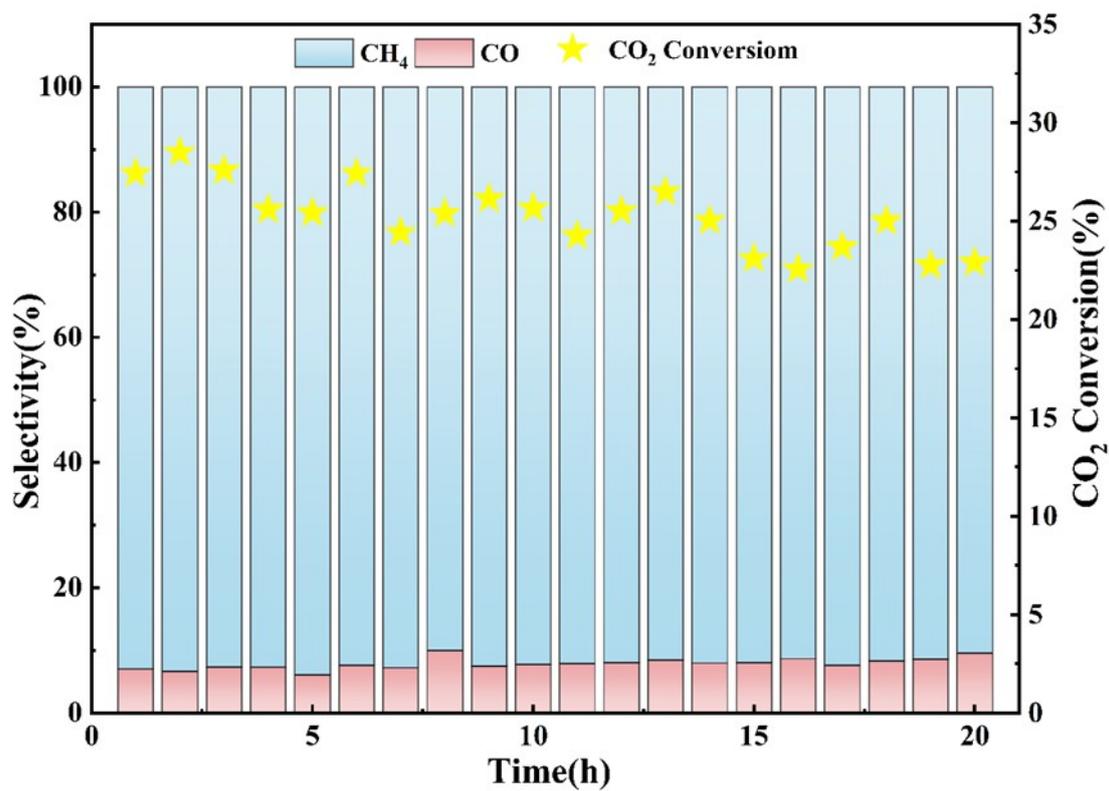


Figure S11. Cycling stability of Co₃O₄ NPs at 350 °C over 20 h.

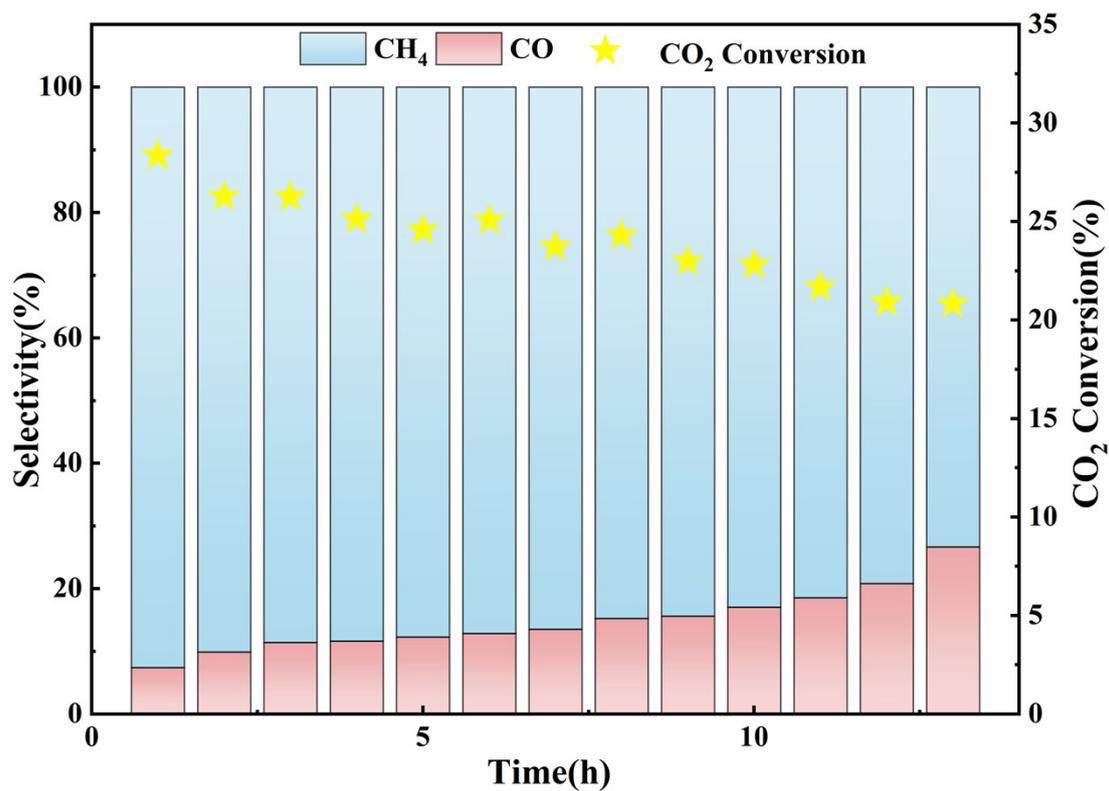


Figure S12. Cycling stability of Co₃O₄/g-C₃N₄-10 at 400 °C over 13 h.

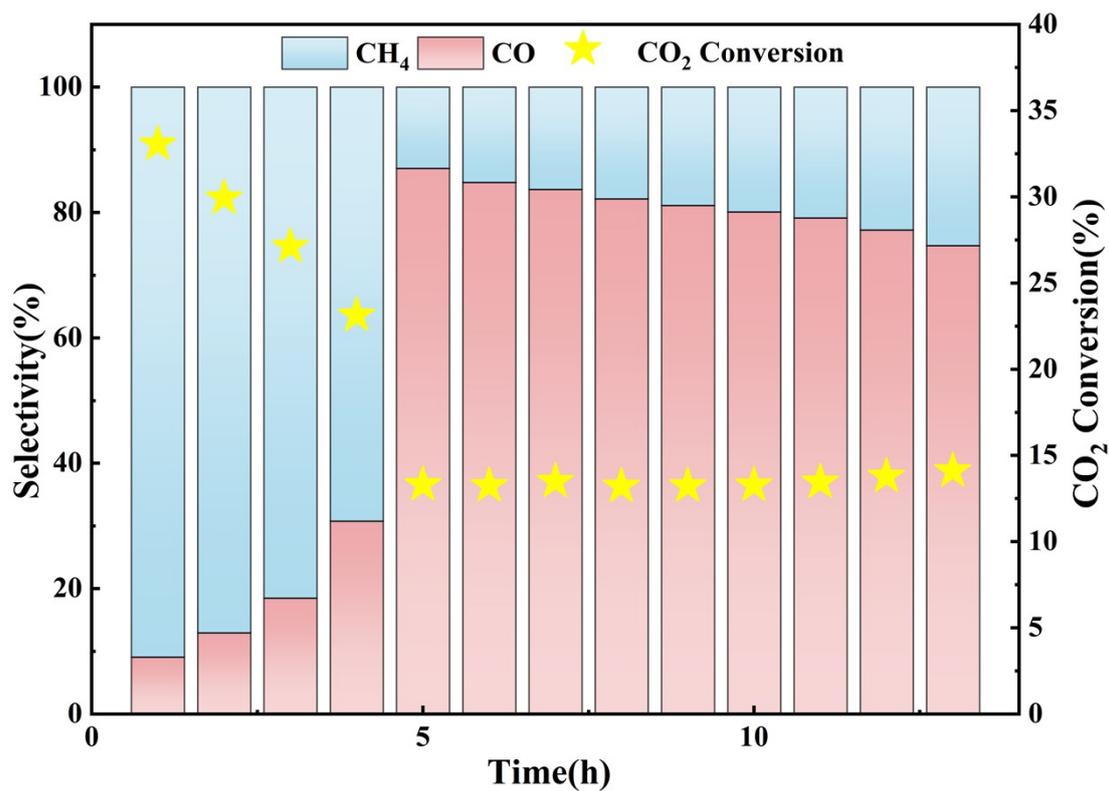


Figure S13. Cycling stability of Co₃O₄/g-C₃N₄-20 at 400 °C over 13 h.

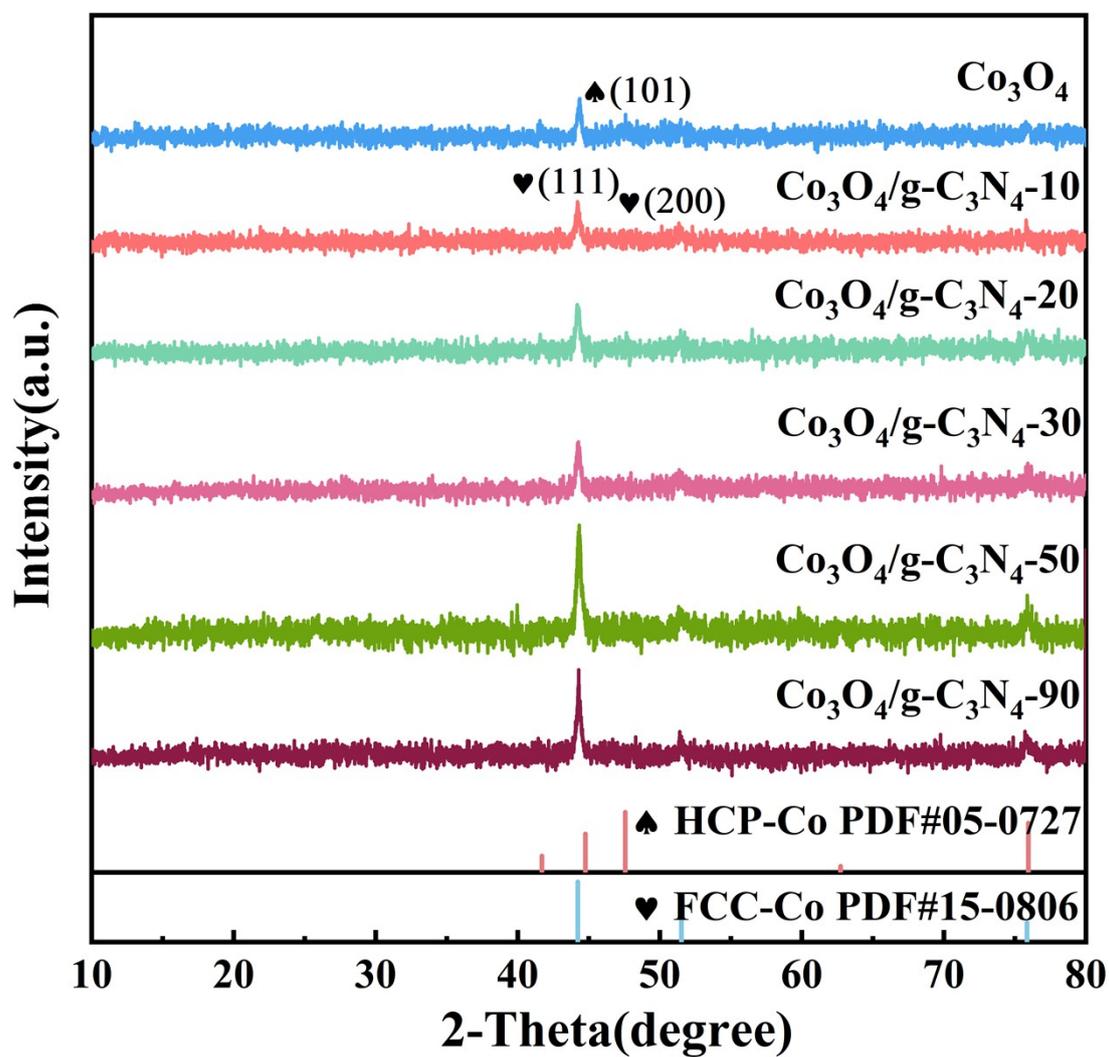


Figure S14. XRD patterns of the catalyst after performance evaluation.

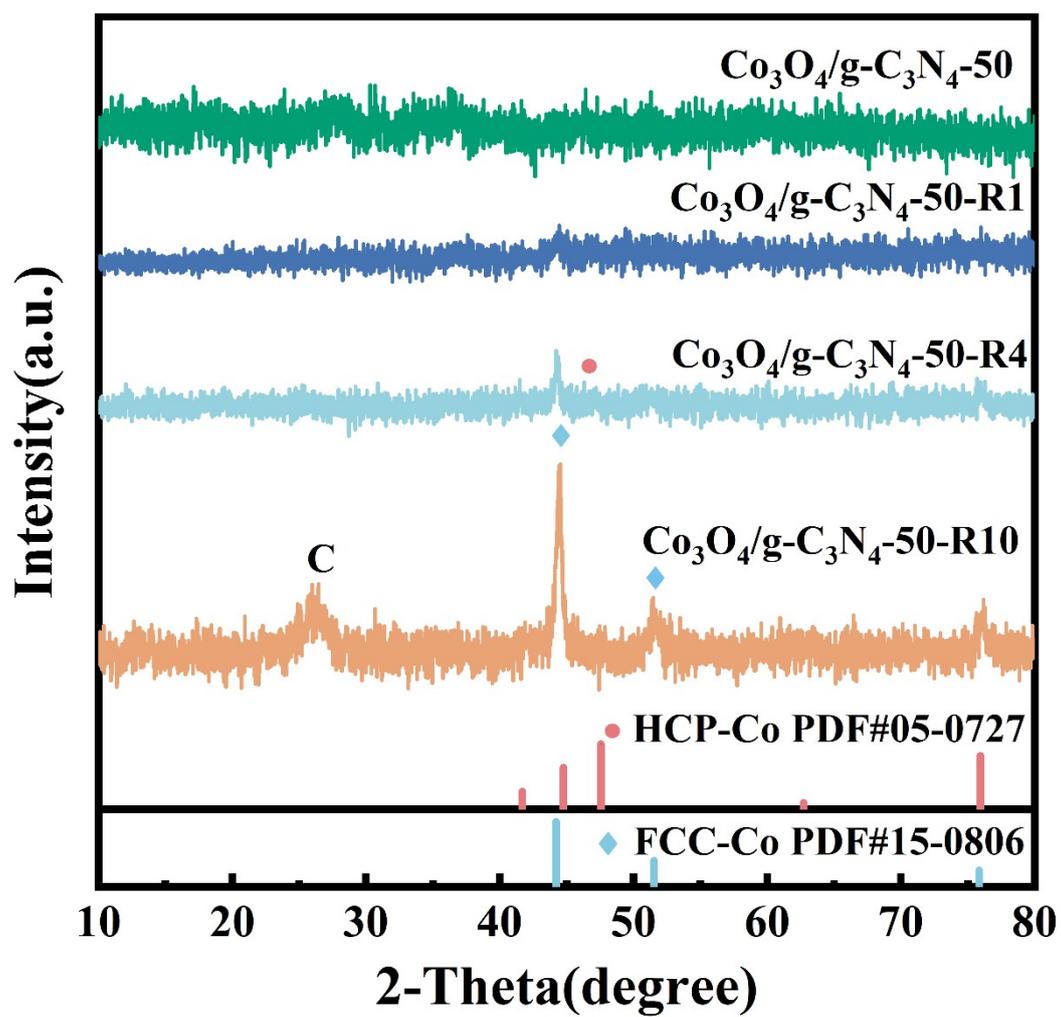


Figure S15. XRD patterns of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-50$ after 1 h, 4h and 10 h of reaction.

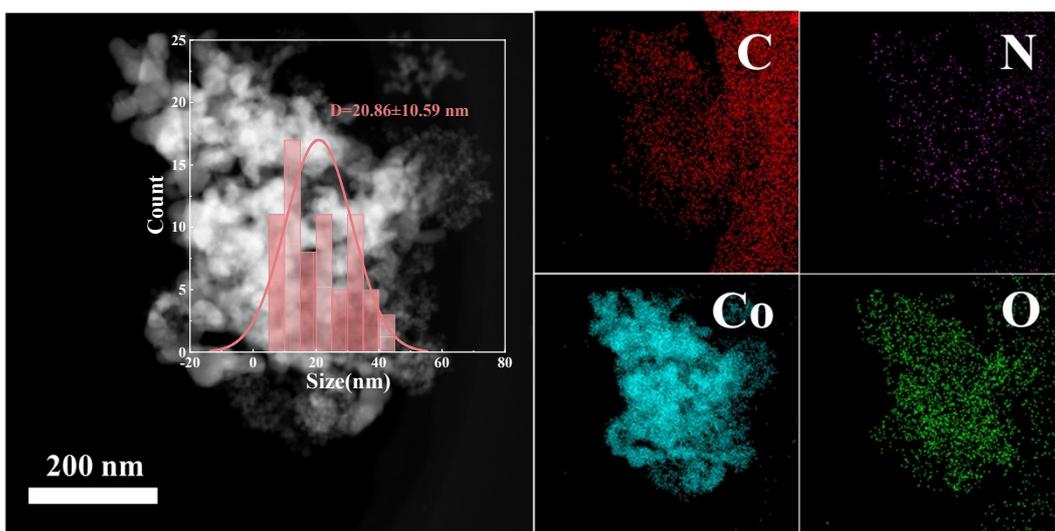


Figure S16. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4\text{-50}$ after 1 h of reaction.

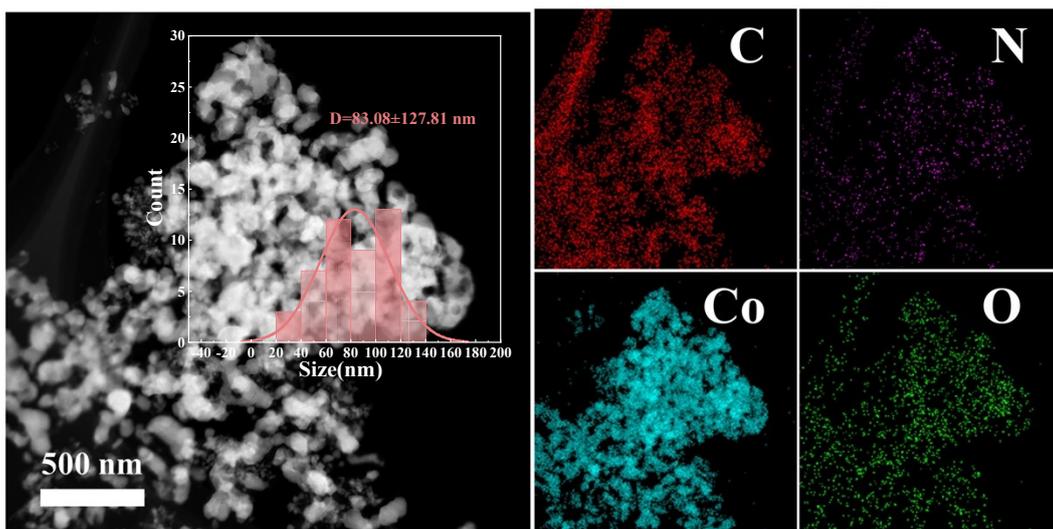


Figure S17. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-50$ after 4 h of reaction.

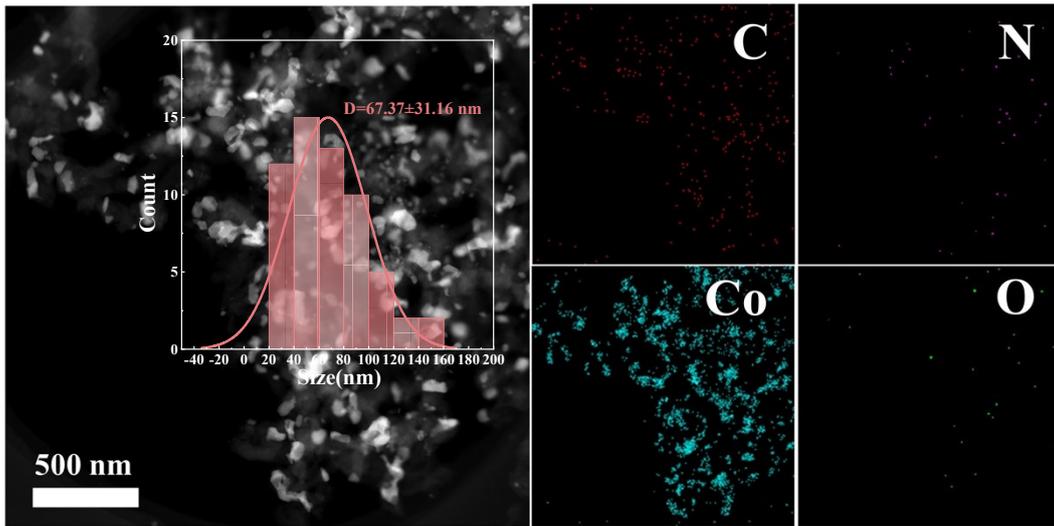


Figure S18. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4\text{-50}$ after 10 h of reaction.

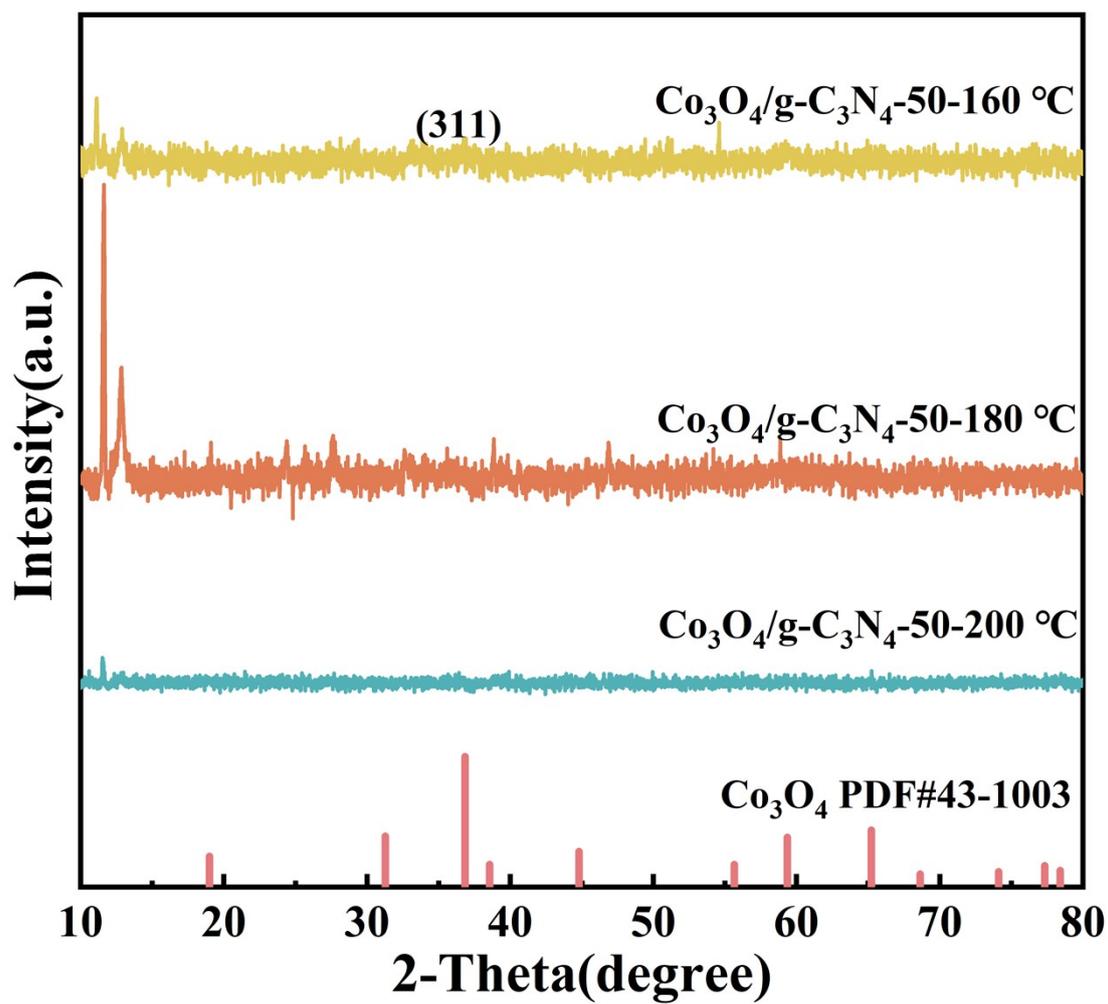


Figure S19. XRD patterns of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4$ -50-160 °C, $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4$ -50-180 °C and $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4$ -50-200 °C.

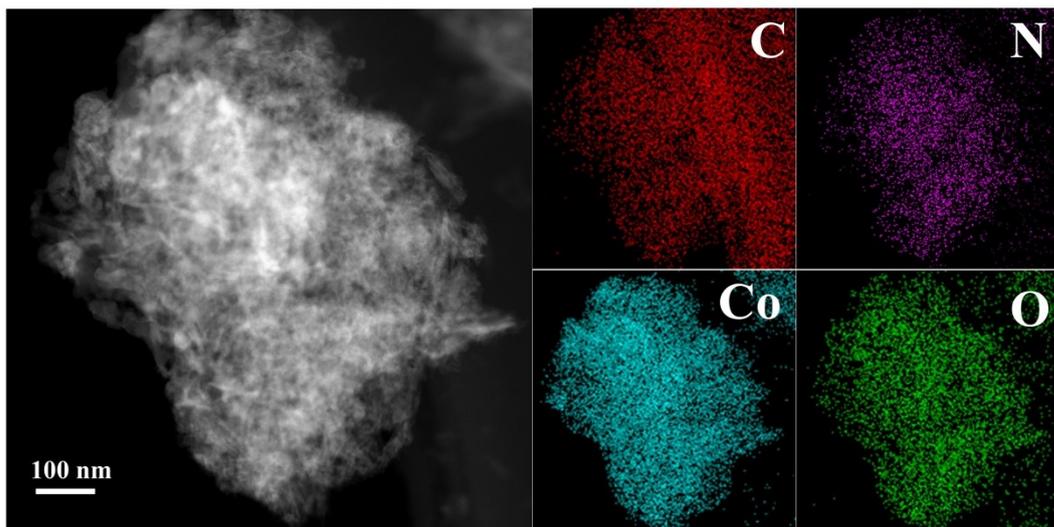


Figure S20. EDS mapping of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4$ -50-180 °C.

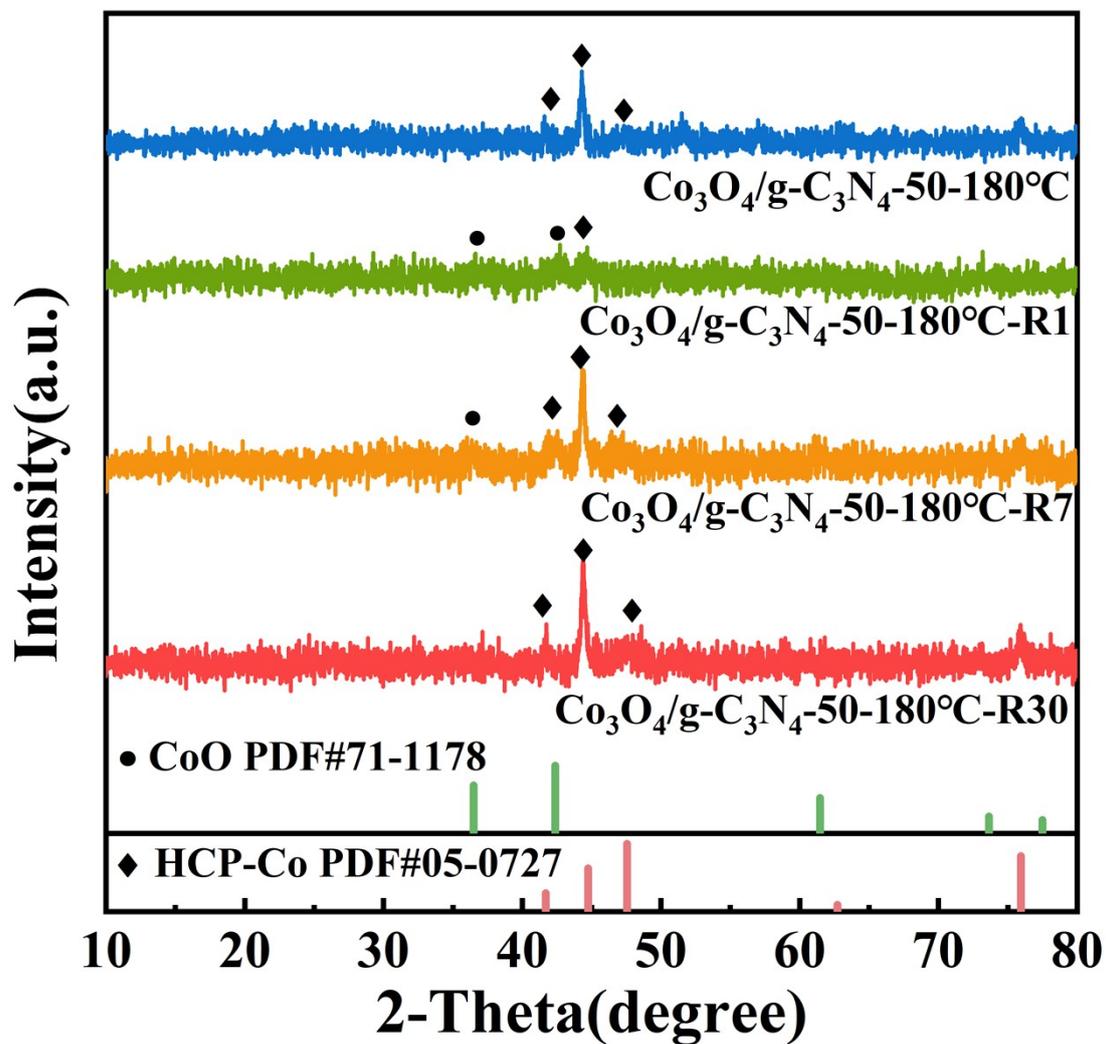


Figure S21. XRD patterns of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4-50-180^\circ\text{C}$ after performance evaluation at different temperatures and after the 1 h, 7 h and 30 h stability test at 350°C .

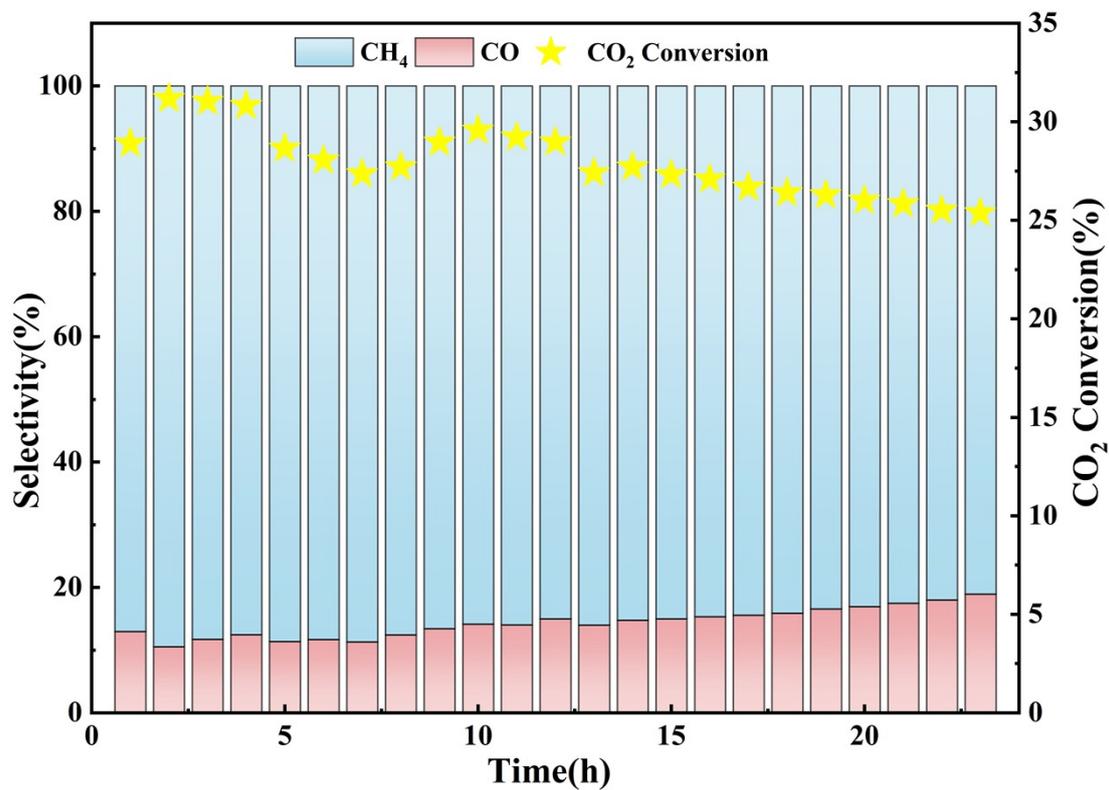


Figure S22. Cycling stability of Co₃O₄/g-C₃N₄-50 180°C at 380 °C over 23 h.

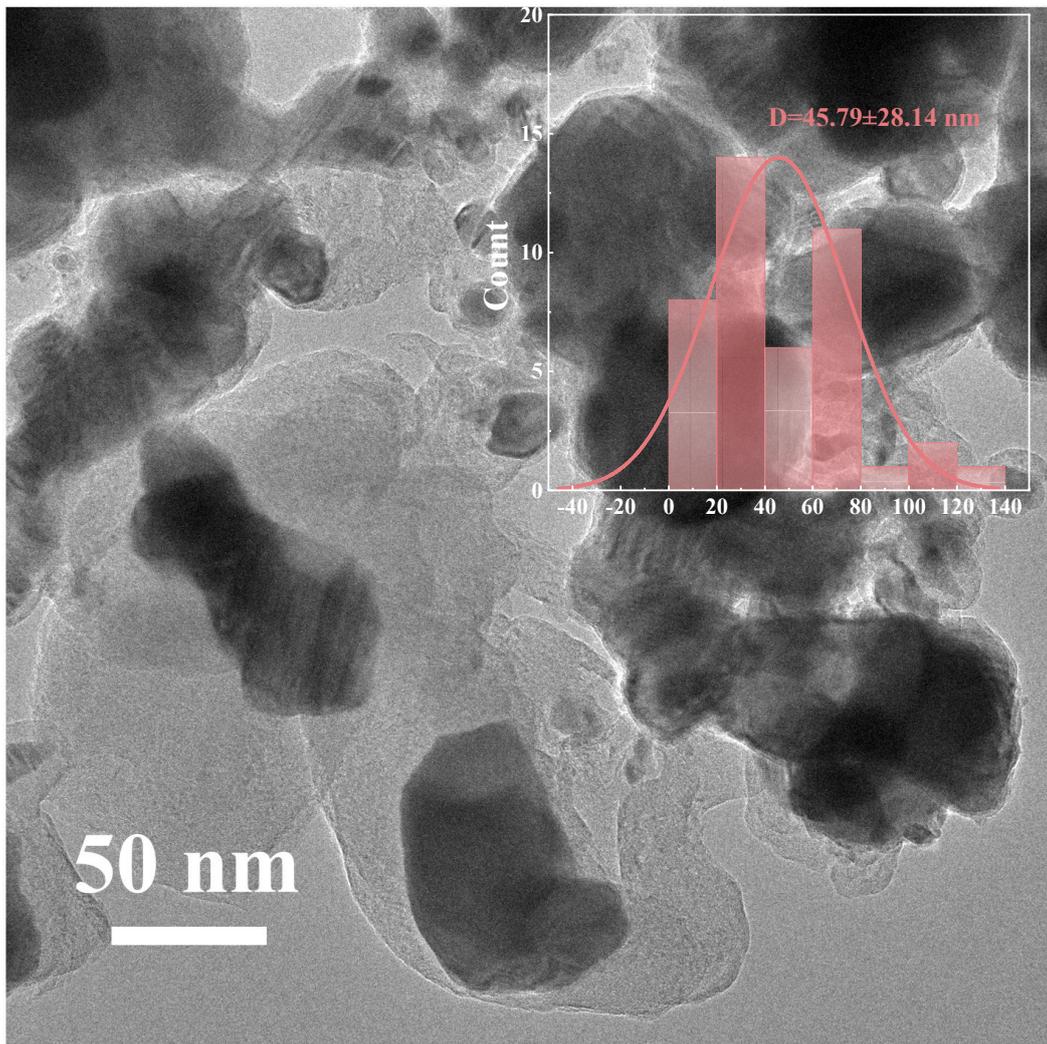


Figure S23. Particle size distribution of $\text{Co}_3\text{O}_4/\text{g-C}_3\text{N}_4$ -50- 180°C after the cycling stability test.

Table S1. Electronic structural properties of different catalysts

	Co ₃ O ₄ NPs		Co ₃ O ₄ /g-C ₃ N ₄ -50		Co ₃ O ₄ /g-C ₃ N ₄ -50-180 °C	
	Binding Energy(eV)	Relative(%)	Binding Energy(eV)	Relative(%)	Binding Energy(eV)	Relative(%)
Co ³⁺ (2p ^{3/2})	779.61	49.63	779.46	19.50	779.45	24.23
Co ²⁺ (2p ^{3/2})	781.61	50.37	781.04	80.50	780.6	75.77
O _v	531.11	28.22	531.43	41.93	531.23	26.73
O _{latt}	529.41	64.36	529.72	36.46	529.71	58.06