1 upplemental Information

2 Study on the properties of biomass film with different compositions

3 1. Sample preparation

4 Mainly prepared by solution casting method can absorb membrane. Chitosan was dissolved in 2%v/v acetic acid at room temperature to prepare 1.5%w/v chitosan 5 solution, and gelatin was dissolved in water at 80 °C to prepare 1.5%w/v gelatin 6 7 solution. The mixture was mixed according to the proportions and 30%wt glycerol was added to mix well. The composite solution was made by stirring at 40 °C for 60 minutes, 8 9 and the bubbles were completely removed by ultrasonic treatment for 30 minutes for use. The chitosan and gelatin solutions were divided into the following proportions: 10 11 1:0, 9:1, 8:2, 7:3, 6:4, and 1:1, The composite solutions of different ratios were cast in a smooth mold and then dried at 65°C for 24 hours to release the mold. The smooth 12surface films are denoted as SCM1-0, SCM9-1, SCM8-2, SCM7-3, SCM6-4, and 13SCM1-1 (smooth construction membrane), respectively. A dry and ventilated place was 1415 stored for more than 48 hours to allow the acetic acid to fully evaporate before subsequent testing. 16

17 2. Characterization of biomass absorbable membrane with different 18 compositions

19 2.1 Physical characterization of biomass absorbable membrane with different20 compositions

21 2.1.1 Morphology characterization

To study the surface topology, the surface morphology of the film was characterized. With a conductive adhesive film on the sample stage, using an ion sputtering apparatus (HITACHI E – 1010, HITACH, Japan) with gold sputtering coating surface, then under 15.00 kV acceleration voltage surface observed microstructure by scanning electron microscope (SEM, SUPRA 35, LEO, Germany). 1 At the same time use an EDS spectrometer measuring surface element distribution.

2 2.1.2 Functional group and phase analysis

3 Using Fourier infrared spectrometer (FTIR, Cary 630, Agilent, Germany)) 4 analysis of the sample composition and internal crosslinking, the scanning range is 4000 5 cm⁻¹-400 cm⁻¹, with a resolution of 2 cm⁻¹.

6 The phase and crystallization properties of each sample were tested and
7 characterized by X-ray diffraction (XRD, D8 Advance, Bruker, Germany), scanning in
8 the Angle range of 5-80° (2θ). Wherein SCM0-1 is a pure gelatin film made of 1.5%w/v
9 gelatin solution according to the preparation process described in 2.2.

10 2.1.3 Degradation property

11 The films were cut into 90 mm diameter discs, immersed in phosphate buffer 12 (PBS) at 37 °C, and shaken in a 37 °C constant temperature gas bath oscillator. In 7 13 days and 14 days, dry surface moisture absorption, after completely dry weighs its 14 degradation rate, expressed in the weightlessness rate measured three times average in 15 each sample. The weightlessness rate formula is expressed as:

16
$$Degradation = \frac{(m_0 - m_n)}{m_0} \times 100\%$$

17 Where:

18 M_0 - the quality of the samples before soaking; M_n - the quality of the sample soaked in 19 n days

20 2.1.4 Hydrophilia

For smooth films with six components, the hydrophilicity of the different films was tested using a contact Angle measuring instrument. Fasten the films to a flat surface, using a contact Angle measurement instrument to test samples of static water contact Angle, averaging three times each sample test.

1 2.1.5 Mechanical property

The tensile properties of the different samples were tested using an electronic tensile testing machine (AG-X, Shimadzu, Japan), and the samples were cut into 25×75mm strips with 3 samples in each group. The thickness of three samples was measured using a spiral micrometer before testing, and each sample was measured three times and averaged. The strain was stretched at 1mm/min until the fracture experiment was stopped. The Young's modulus, tensile strength, and elongation at the break of each sample were obtained by drawing tensile stress-strain curves.

9 2.2 The biocompatibility of the different compositions of biomass absorbable 10 membrane

11 Each film will be cut into 16 mm diameter wafers after UV irradiation sterilization and fully placed in 24-Well Cell Culture Plates, the Mouse calvarial preosteoblast cell 12line MC3T3-E1 to 1.5×10^4 / hole density in the thin film surface. The cells were 13cultured in a constant temperature incubator (5%CO₂, 37 °C) for 5 days, and the Cell 14 Counting Kit-8 assay (CCK-8, Beyotime, China) was used to evaluate cell proliferation 15 every day. In brief, cells were washed twice with PBS and then treated with 550 μ L 16 17CCK-8 working solution (550 µL culture medium containing 10%CCK-8 working solution (v/v)). The cell plate at 37 °C under 2.5 hours after incubation, with each hole 18 19 150 µL of the solution to 96-well plates, the absorbance (OD) at 450 nm was read using a microplate reader (Spectra Max M5, Molecular Devices, USA). 20

21 **3. Results**

22 **3.1 Properties of biomass films with different compositions**

As shown in **Fig. S1**, 500 times the magnification of biomass flat membrane surface is smooth, without impurities and larger dissolved polymer particles, the chitosan and gelatin in the system are completely dissolved and dispersed

1 homogeneously and did not produce obvious physical separation. The biomass films of 2 each component were prepared by the same process and had a dense smooth surface 3 structure. Spectrum EDS electron microscopy is used to determine different 4 composition distributions of the material surface elements, all membrane mainly 5 consists of two elements of C, and O, atomic ratio is close to 6:4.

6 Fig. S2 shows all the films in 3100-3500 cm⁻¹ show a broad peak, this is due to the stretching vibration of intermolecular hydrogen bond O-H (3200-3500 cm⁻¹) and 7 N - H stretching vibration absorption (3100-350cm⁻¹), the composite films in 1645 cm 8 ⁻¹ (amide I, C-O stretching vibration), 1540 cm ⁻¹ (amide II, N-H bending and C-N 9 tensile) and 1155 cm⁻¹ (amide III, C-N, N-H stretching vibration) has the characteristics 10 of the belt, which attributed to the formation of hydrogen bonds between chitosan and 11 gelatin and electrostatic interactions. In the FTIR of pure chitosan, there are 895 cm⁻¹ 12 β -glycosidic bond configuration absorption peak, 1155 cm⁻¹ C-O absorption peak on 13the ring, and 1655 cm⁻¹ C=O stretching vibration peak¹. In the FTIR of pure gelatin 14 film, the absorption peaks of the amide I band, amide II band, and amide III band were 15 1637 cm⁻¹, 1541 cm⁻¹, and 1240 cm⁻¹, respectively². After gelatin was blended with the 16 chitosan membrane, the absorption peak of the amide I band at 1637cm⁻¹ shifted to a 17higher wave number, and the absorption band became wider, indicating that there was 18 a strong hydrogen bond between chitosan and gelatin molecules. 19

20 Fig. S3 shows the parcel peak can be seen that all the material for the amorphous structure, no apparent crystalline phase, and no fine spectrum peak structure. By pure 21substance XRD diagram can be seen that chitosan in 2 theta is 10.2° and 20.6° 22 respectively in the crystallization, at about 15° appear smaller without water of 23 crystallization peak ³. Gelatin shows amorphous diffraction peaks around 19.9° and 24 257.5° at 2 θ , which are characteristic diffraction peaks of gelatin ⁴. If there is no interaction between chitosan and gelatin molecules or the interaction is very weak, they 26 will have their crystallization area in the composite membrane, the diffraction pattern 27 is characterized by chitosan and gelatin diffraction pattern in proportion to the simple 28

1 superposition. 20.6° in composite membrane of chitosan characteristic peak gradually 2 with the addition of gelatin to left shift, and not in 2 theta 7.5° and 10.2° and 15° 3 occurrence characteristics of diffraction peak, these instructions between chitosan and 4 gelatin strong interaction, disrupting the crystallization of chitosan and gelatin in their 5 respective situations.

6 It can be seen from **Fig. S4** that the degradation of pure chitosan is slower than 7 that of gelatin. With the increase in the amount of gelatin added, the degradation rate 8 of biomass film is gradually accelerated. When the ratio of the two components is 1:1, 9 the degradation rate of 7 days reaches about 60%. Further degradation was observed at 10 14 days compared to 7 days.

11 The hydrophilic and hydrophobic properties of the material surface are one of the important factors affecting cell adhesion and proliferation. According to relevant 12 reports, both hydrophilic and too hydrophobic surfaces are not suitable for cell adhesion 13and proliferation ⁵. Cell adhesion usually depends on protein adhesion in the cells 14 initially, while hydrophobic surfaces are more conducive to protein adhesion, which 15 can bring favorable effects on cell adhesion and proliferation. Material of hydrophilic 16 and hydrophobic is to describe the nature of the interaction between fluid and solid 17surface, usually with Water Contact Angle measurement. If the droplets on the material 18 19 surface form a smaller Contact Angle (less than 90 degrees Celsius), suggests that the material has good hydrophilicity, on the contrary, when the water contact Angle is 20 21greater than 90° it shows that the material surface is hydrophobic. Fig. S5 illustrates the 22 static water contact Angle plots of the six component biomass films. It can be seen from 23 Fig. S6 that the water contact Angle of the 6 smooth biomass films formed by chitosan/gelatin composite with different components is about 120°, which is a 24 25hydrophobic surface, consistent with the surface hydrophobicity of cell adhesion and proliferation, and can be used as a biological material to promote cell proliferation and 26 27 differentiation.

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3 Mechanical properties are one of the important indexes to evaluate whether the

biomass film can be used as a tendon-bone interface graft. To ensure that the material acting on the affected site can compensate for the insufficient mechanical properties of the original interface and encapsulate the tendon, it is necessary to have good mechanical properties of the membrane. It can be seen from Fig. S7 and Fig. S8 that chitosan/gelatin biomass films with different compositions belong to brittle materials, with elongation at break of about 5% and high tensile strength, all of which can reach more than 20 MPa, which can meet the requirements of tendon-bone interface.

8 As promotes tendon-to-bone interface healing tissue engineering material, the 9 most basic requirements of biomass film are a non-toxic effect in vivo, good biocompatibility, and promotion of cell proliferation. The films of all components were 10 therefore co-cultured with the Mouse calvarial preosteoblast cell line MC3T3-E1, 11 through the Cell Counting Kit-8 measurement to evaluate cell proliferation. As shown 12 in Fig. S9, all material surface inoculation cells, trained in five days, with the extension 13of incubation time, the absorbance on the surface of the material all showed a trend of 14 15 increase, indicating that the material was not toxic to cells, had good biocompatibility, can accommodate cell growth and proliferation. There was no significant difference in 16 the first 3 days of co-culture, but after 5 days of co-culture, the group with a 17chitosan/gelatin ratio of 9:1 had a higher absorbance and better ability to promote cell 18 19 proliferation. On the fourth and fifth day, the absorbance of SCM9-1 material was 20 significantly different from that of the other five groups of materials. As you can see, 21chitosan and gelatin both are good biocompatibility materials, proper proportion can improve the promoting effect, and is conducive to a large number of cell proliferation. 22

SCM1-0	SCM9-1	SCM8-2
SCM7-3	SCM6-4	SCM1-1
		50 μm

Fig. S1 SEM images and EDS energy spectra of biomass films with different compositions.





substance (b).









Fig. S5 Static water contact Angle plots of biomass films with different compositions.

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(a)SCM0-1. (b)SCM9-1. (c)SCM8-2. (d)SCM7-3. (e)SCM6-4. (f)SCM1-1.





biomass films with different compositions.



Fig. S9 CCK 8 cell proliferation of biomass films with different compositions. Data are presented





Fig. S10 EDS energy spectra of biomass films with different compositions.











Fig. S12 The length of the long diameter and short diameter distribution of PCM.





Fig. S13 EDS energy spectra of biomass films with different morphologies.





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(a)SCM. (b)MCM. (c)PCM.



3 surface of the material. (a)The fluorescence intensity of RUNX2 secreted by MC3T3-E1 cells.
4 (b)The fluorescence intensity of α-SMA secreted by NIH3T3 cells.

1 Supplemental References

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