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## Supporting information

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## **Conductive Nano Nickel Oxide/Hydroxide Paper Electrochemical Sensor for Serotonin Detection in Genetically Engineered *Drosophila***

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19 The RMS and mean roughness values increased from  $10.04 \pm 2.9$  nm and  $8.3 \pm 2.6$  nm (bare  
 20 PPE) to  $13.3 \pm 5.2$  nm and  $10.4 \pm 3.7$  nm (Ppy-rGO), and significantly to  $49.5 \pm 18.1$  nm and  
 21  $38.5 \pm 13.2$  nm after NiO/Ni(OH)<sub>2</sub> deposition, respectively. The values are presented as mean  
 22  $\pm$  standard deviation, calculated from multiple measurements across different surface regions  
 23 using Gwyddion software. The substantial increase in roughness following each modification  
 24 step suggests statistically relevant changes in surface morphology. These morphological  
 25 features, including mountain-like ridges and increased thickness, contribute to an enhanced  
 26 electroactive surface area, which is advantageous for electrochemical sensing. Detailed  
 27 roughness parameters and thickness values, calculated using Gwyddion software, are provided  
 28 in **Table S1**.

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Modified electrode	RMS Roughness (nm)	Mean Roughness (nm)
Bare PPE	$10.04 \pm 2.9$	$8.3 \pm 2.6$
PPE-Ppy-rGO	$13.3 \pm 5.2$	$10.4 \pm 3.7$
PPE-Ppy-rGO-NiO/Ni(OH) <sub>2</sub>	$49.5 \pm 18.1$	$38.5 \pm 13.2$

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**Table S1.** AFM Data.

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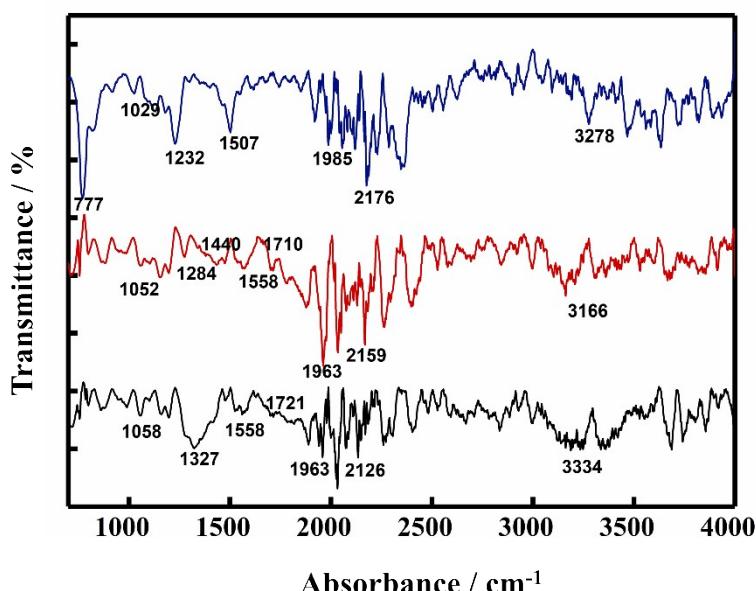
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51 **Figure S1.** FT-IR spectra obtained for PPE (**black**), PPE-Ppy-rGO (**red**), and PPE-Ppy-rGO-  
 52 NiO/Ni(OH)<sub>2</sub> (**blue**)

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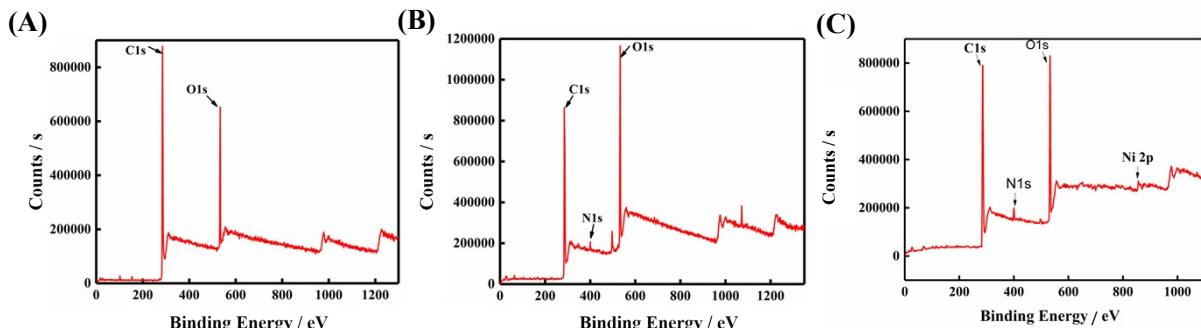
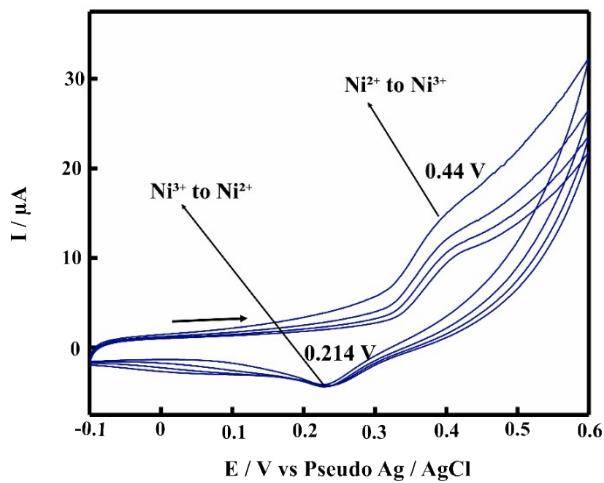


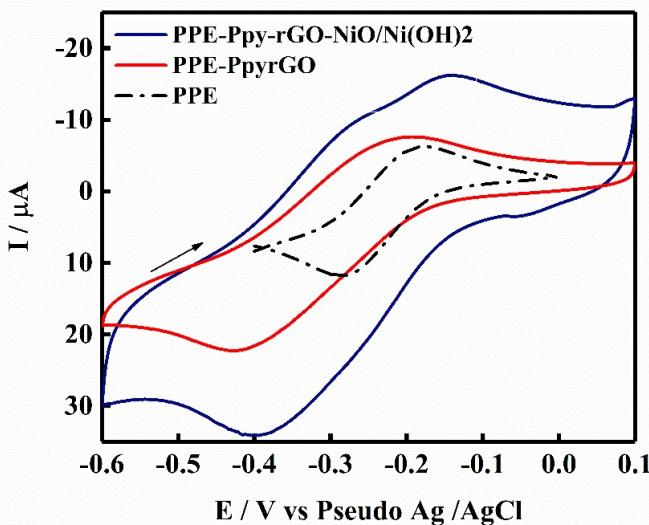
Figure S2. XPS Survey scan for (A) PPE (B) PPE-Pny-RGO (C) PPE-Pny-rGO-NiO/Ni(OH)<sub>2</sub>

Electrode	Atomic %				Atomic ratio		
	C1s	O1s	N1s	Ni2p	O1/C1	N1/C1	Ni/C1
PPE	82.99	17.01	-	-	0.0249	-	-
PPE-Ppy-rGO	71.71	26.91	1.38	-	0.375	0.0192	-
PPE-Ppy-rGO-NiO/Ni(OH) <sub>2</sub>	73.61	23.62	1.99	0.79	0.227	0.026	0.011

**Table S2.** Atomic % and Atomic ratio calculated from the XPS data obtained for various electrodes.



**Figure S3.** CV curves for the PPE-Ppy-rGO-NiO/Ni(OH)<sub>2</sub> nanocomposites electrodes, recorded over a potential range of -0.1 to 0.6 V at a scan rate of 0.05 V/s for 8 cycles in 0.1 M KOH



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86 **Figure S4.** Cyclic voltammetry (CV) analysis of PPE, PPE-Ppy-rGO, and PPE-Ppy-rGO-  
87 NiO/Ni(OH)<sub>2</sub> in 1 mM Ru(NH<sub>3</sub>)<sub>6</sub>Cl<sub>3</sub> and 0.1 M KCl, measured at a scan rate of 0.05 V s<sup>-1</sup>.

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91 Electrochemical active surface area calculated using the Randles–Sevcik equation, where D  
92 calculated as from previous reports <sup>1,2</sup>

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<i>Electrode</i>	<i>Electroactive probe</i>	<i>D/cm<sup>2</sup>/s<sup>-1</sup></i>	<i>A<sub>geo</sub>/cm<sup>2</sup></i>	<i>A<sub>real</sub>/cm<sup>2</sup></i>	<i>% Real</i>
Bare PPE	Ru(NH <sub>3</sub> ) <sub>6</sub> Cl <sub>3</sub>	9.1 × 10 <sup>-6</sup>	0.07	0.37	528.6
PPE-Ppy-rGO	Ru(NH <sub>3</sub> ) <sub>6</sub> Cl <sub>3</sub>	9.1 × 10 <sup>-6</sup>	0.07	0.46	657.1
PPE-Ppy-rGO-NiO/Ni(OH) <sub>2</sub>	Ru(NH <sub>3</sub> ) <sub>6</sub> Cl <sub>3</sub>	9.1 × 10 <sup>-6</sup>	0.07	0.99	1414.3

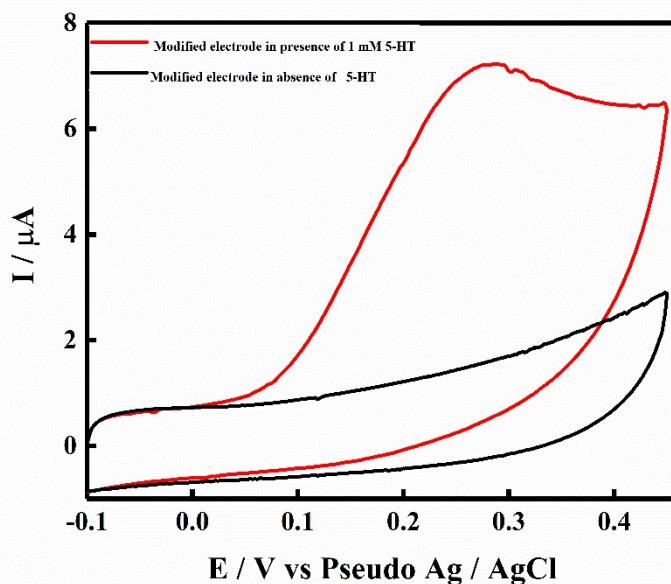
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96 **Table S3.** Electrochemical active surface area analysis of modified electrodes

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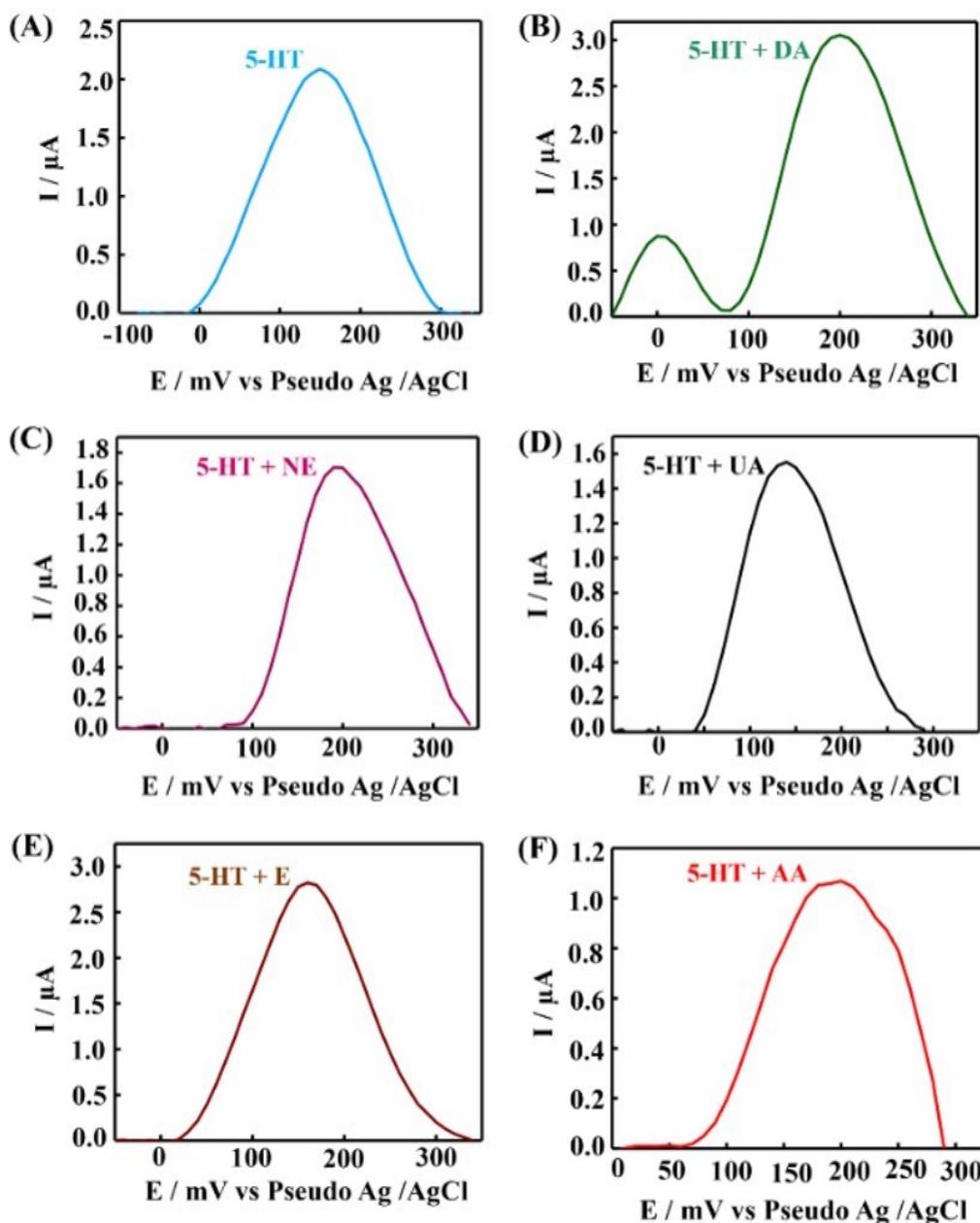
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106 E / V vs Pseudo Ag / AgCl

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125 **Figure S5.** Cyclic voltammetry (CV) analysis of PPE-Ppy-rGO-NiO/Ni(OH)<sub>2</sub> in the absence  
126 (black) and in presence of 1 mM 5-HT 0.1 M PB solution (pH 7.4) (red) measured at a scan  
127 rate of 0.05 V s<sup>-1</sup>.  
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159 **Figure S6.** SWV for interference studies of the developed PPE-Ppy-rGO-NiO/Ni(OH)<sub>2</sub> sensor  
 160 against various interfering species, (A) 5-HT (B) 5-HT + DA (C) 5-HT + NE (D) 5-HT + UA  
 161 (E) 5-HT + E (F) 5-HT + AA each at a concentration of 100 μM with 5-HT.

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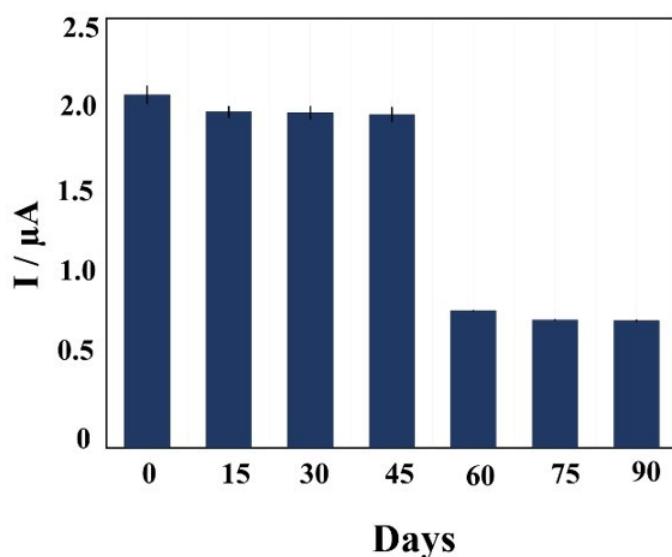
<b>Sensing materials</b>	<b>Method of detection</b>	<b>Linear range</b>	<b>LOD nM</b>	<b>LOD ng/mL</b>	<b>Real samples analysed</b>	<b>Ref.</b>
<i>Nb<sub>2</sub>CT<sub>x</sub>/PCN</i>	DPV	1 μM – 100 μM	63.24	11.14	Human serum	3
<i>Nafion-CNT/EC CFMEs</i>	CV	100 μM – 800 μM	140	24.6	Artificial urine sample	4
<i>FeC-AuNPs-MWCNT</i>	SWV	0.05 μM – 20 μM	17	2.99	Urine sample	5
<i>Aptamer-based biosensor</i>	EIS	25 μM - 150 nM	5.6	0.99	-	6
<i>Au-CNT membrane electrode</i>	CV	0.5 μM – 10 μM	30	5.29	Cell media	7
<i>Dy<sub>2</sub>MoO<sub>6</sub> nanosheets</i>	DPV	10 μM - 130 nM	340	59.9	Human blood and urine	8
<i>RGO/p-PDA/β-Ni(OH)<sub>2</sub></i>	Amperometry	5 μM - 325 μM	46.2	8.14	Urine	9
<i>PPE-Ppy-rGO-Fe<sub>2</sub>O<sub>3</sub></i>	DPV	0.01 μM - 500 μM	22	3.88	<i>Drosophila melanogaster</i> brain sample	10
<i>PPE-Ppy-rGO-Ni/Ni(OH)<sub>2</sub></i>	SWV	0.007 nM - 0.48 nM 0.48 nM – 500 μM	0.024 383.7 nM	0.0042 67.61	Genetically modified <i>Drosophila melanogaster</i> brain sample	Present work

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173 **Table S4.** Comparison of electrochemical performance of various reported electrodes.

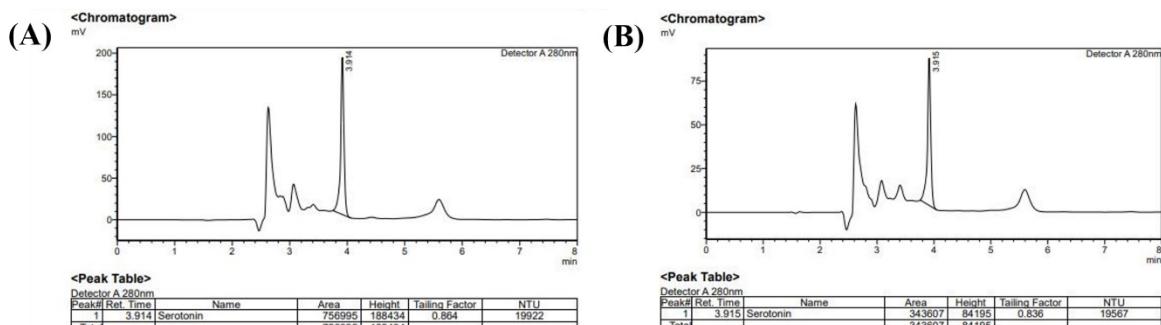
174 Normal serotonin (5-HT) concentrations typically range from 270 to 1490 nM in serum,  
 175 approximately 300 to 1650 nM  
 176 in urine, and are markedly lower  
 177 in cerebrospinal fluid (CSF), often  
 178 below 0.0568 nM.<sup>11-12</sup>



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**Figure S7.** Histogram presenting the repeatability study for three PPE-Ppy-rGO- NiO/Ni(OH)<sub>2</sub> electrode.

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**Figure S8.** HPLC response for genetically engineered *Drosophila melanogaster* brain samples (A) TRH-Gal4 (B) TRH-Gal4/UAS-TeTxLC (C) TRH-Gal4/UAS-rpr

Sample (Genetically engineered <i>Drosophila melanogaster</i> brain sample)	Spikes	Added amount of 5-HT ( $\mu$ M)	Found amount Of 5-HT ( $\mu$ M)	Recovery (%)
TRH-Gal4	-	0.0	27.6	-
	1st spike	8	35	98.3
	2nd spike	7.0	42	98.6
	3rd spike	5.0	46.5	97.7
TRH-Gal4/UAS -TeTxLC	-	0.0	11.9	-
	1st spike	9.0	21	100.5
	2nd spike	7.0	27	96.8
	3rd spike	5.0	32.2	97.9
TRH-Gal4/UAS -rpr	-	0.0	7.59	-
	1st spike	8.3	15.6	98.1
	2nd spike	6.3	21.9	98.7
	3rd spike	5	27.3	100.4

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**Table S5.** Spiked real sample analysis**References**

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