

SUPPORTING INFORMATION

Dual NIR-I Excitation and Emission-Based Thermometry and pH-Responsive Drug Delivery Using NaYF₄:Yb,Er@SiO₂-Folic Acid Conjugates

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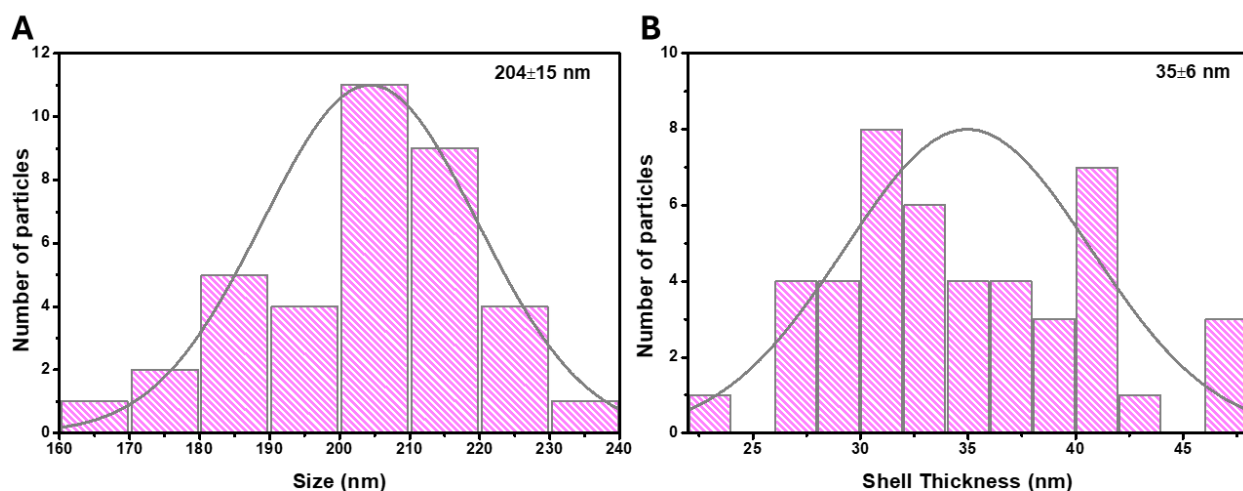


Figure S1. (A) Particle size distribution histograms for NaYF₄:Yb,Er and (B) SiO₂ shell thickness histogram for NaYF₄:Yb,Er@SiO₂.

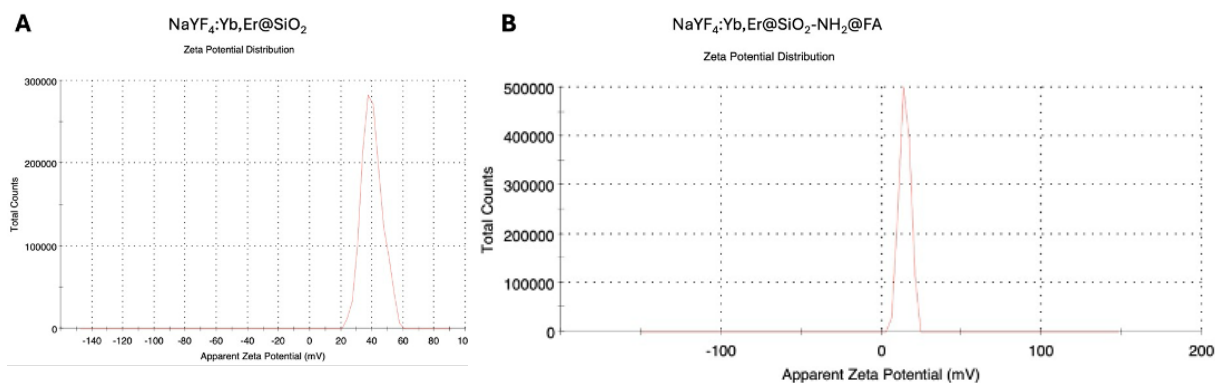


Figure S2. Zeta potential measurements of (A) $\text{NaYF}_4:\text{Yb,Er}@ \text{SiO}_2$ and (B) $\text{NaYF}_4:\text{Yb,Er}@ \text{SiO}_2\text{-NH}_2@ \text{FA}$, showing surface charge changes after functionalization.

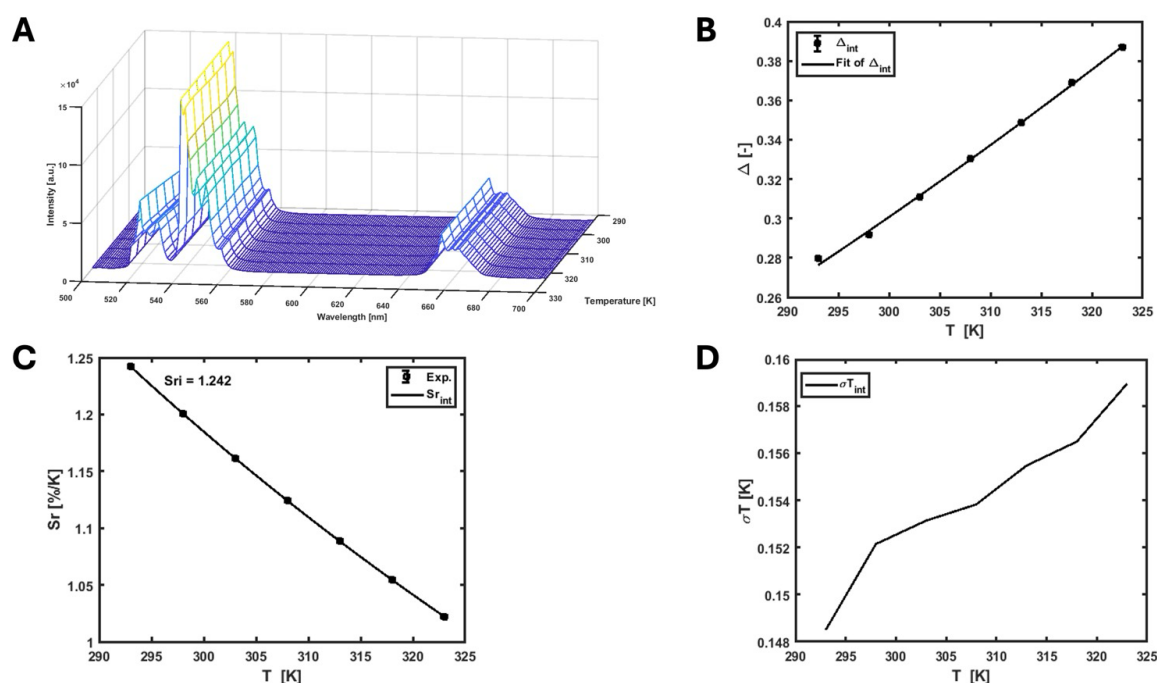


Figure S3. (A) Emission map of $\text{NaYF}_4:\text{Yb,Er}$ dispersed in cyclohexane, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the visible region under 975 nm excitation, (B) graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

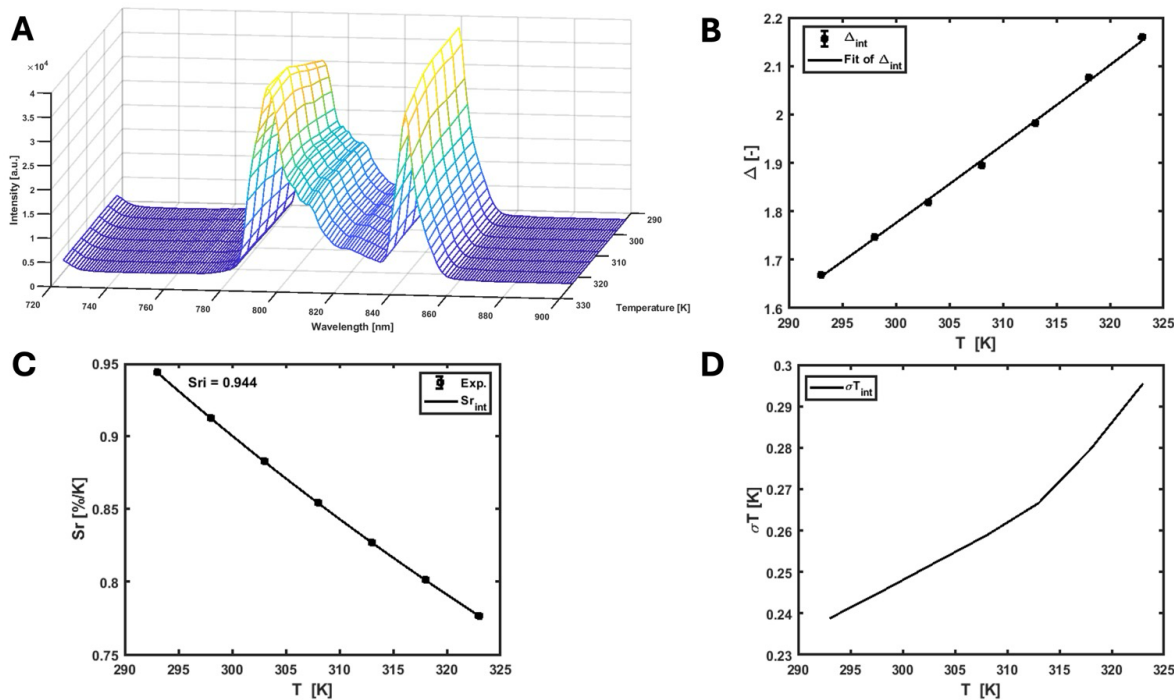


Figure S4. (A) Emission map of NaYF₄:Yb,Er dispersed in cyclohexane, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the NIR-I region under 975 nm excitation, (B) graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

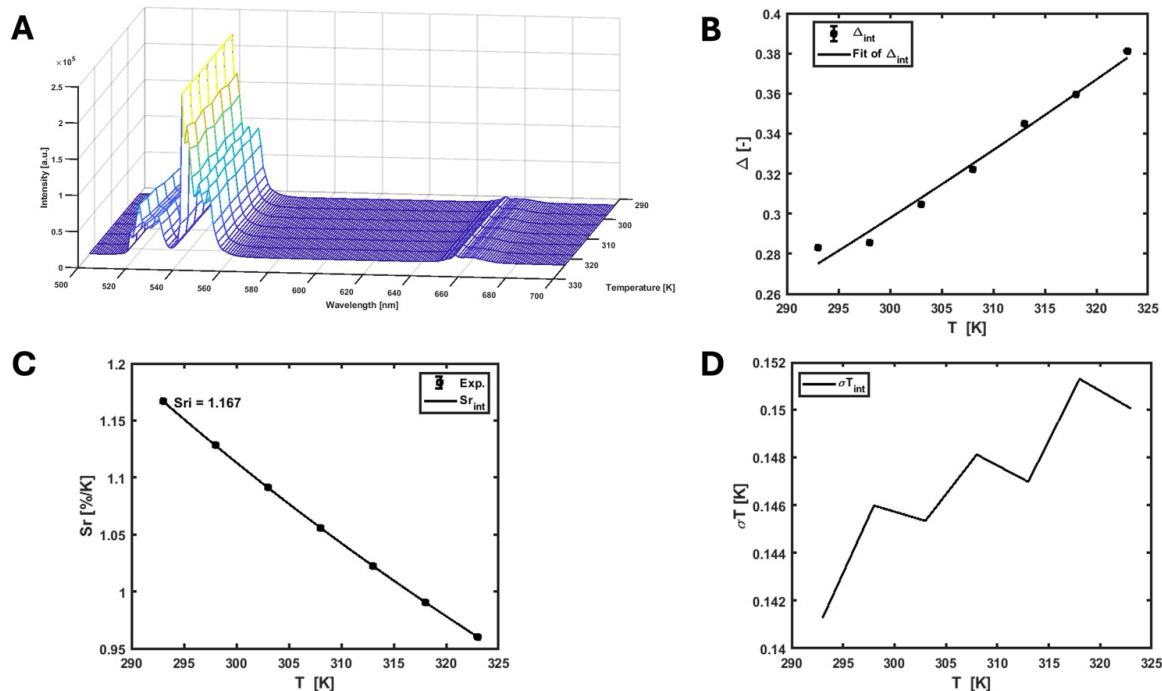


Figure S5. (A) Emission map of NaYF₄:Yb,Er dispersed in cyclohexane, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the visible region under 940 nm excitation, (B) graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

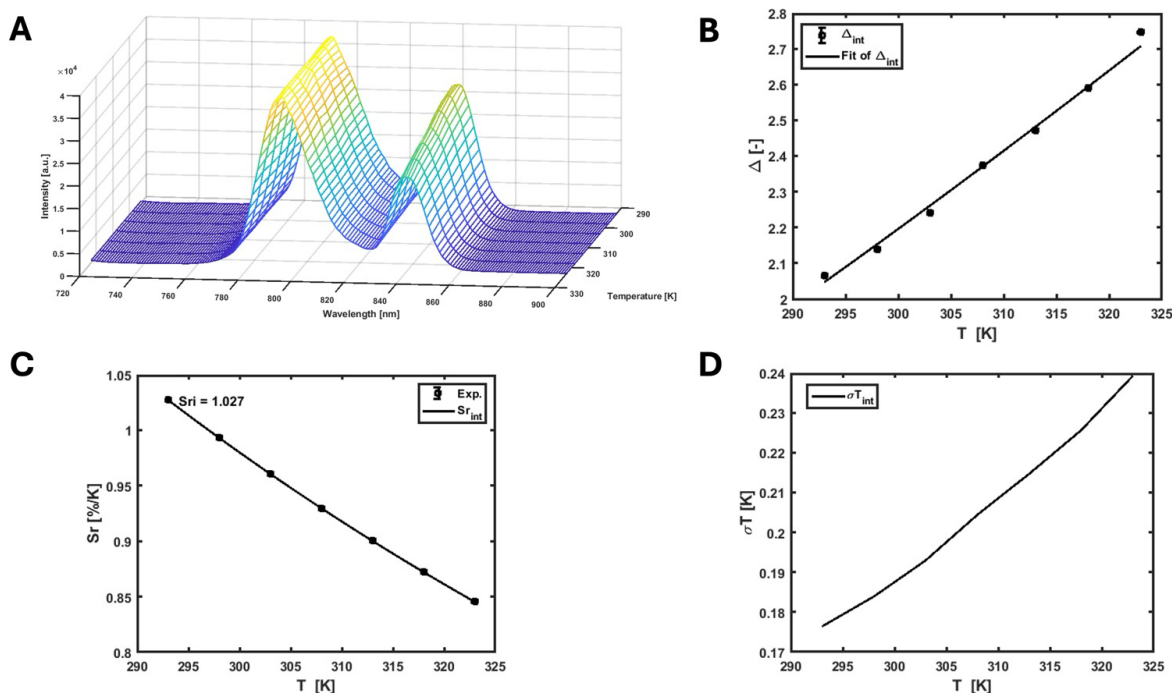


Figure S6. (A) Emission map of NaYF₄:Yb,Er dispersed in cyclohexane, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the NIR-I region under 940 nm excitation, (B) graph showing the Δ parameters versus temperature. graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

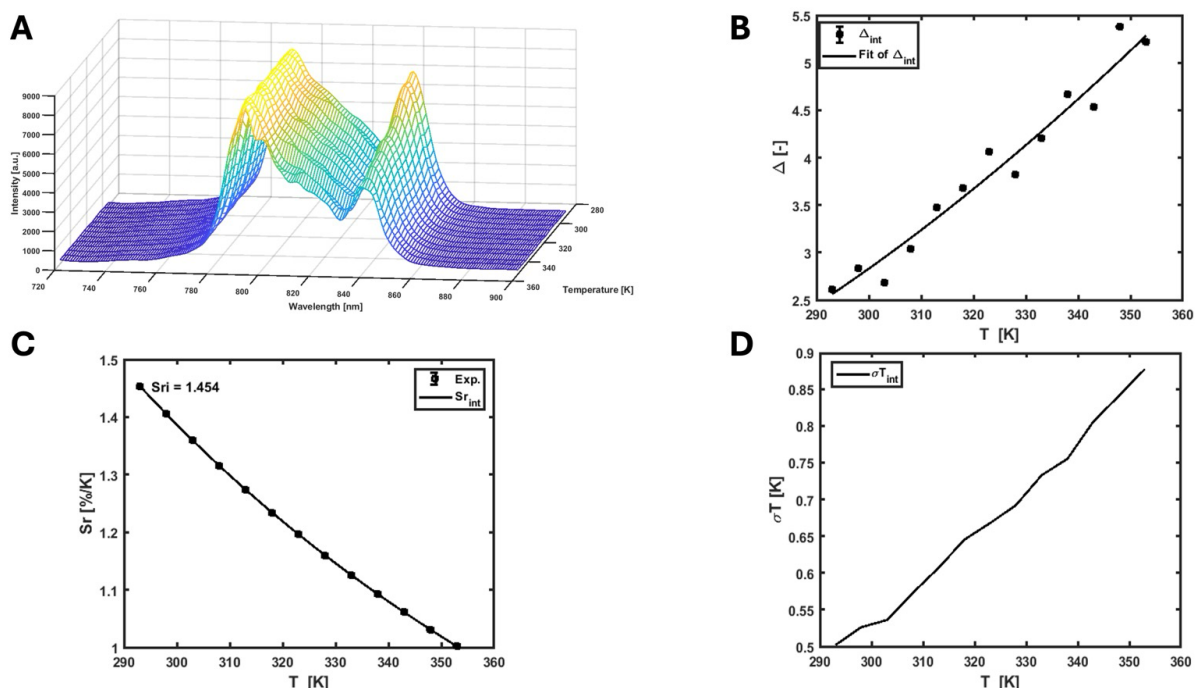


Figure S7. (A) Emission map of NaYF₄:Yb,Er@SiO₂ dispersed in water, recorded across the temperature range of 293.15–353.15 K (20–80 °C) in the NIR-I region under 940 nm excitation, (B) graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

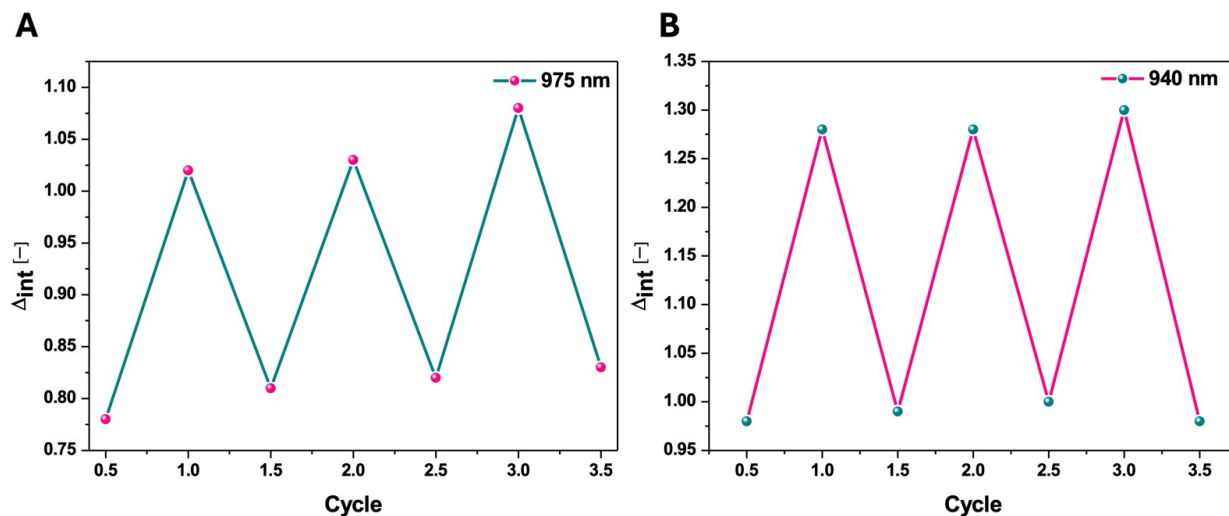


Figure S8. Cycle test for NaYF₄:Yb,Er@SiO₂-NH₂@FA in DI water under two heating–cooling cycles: 293.15 K–323.15 K, using excitation wavelengths of 975 nm and 940 nm.

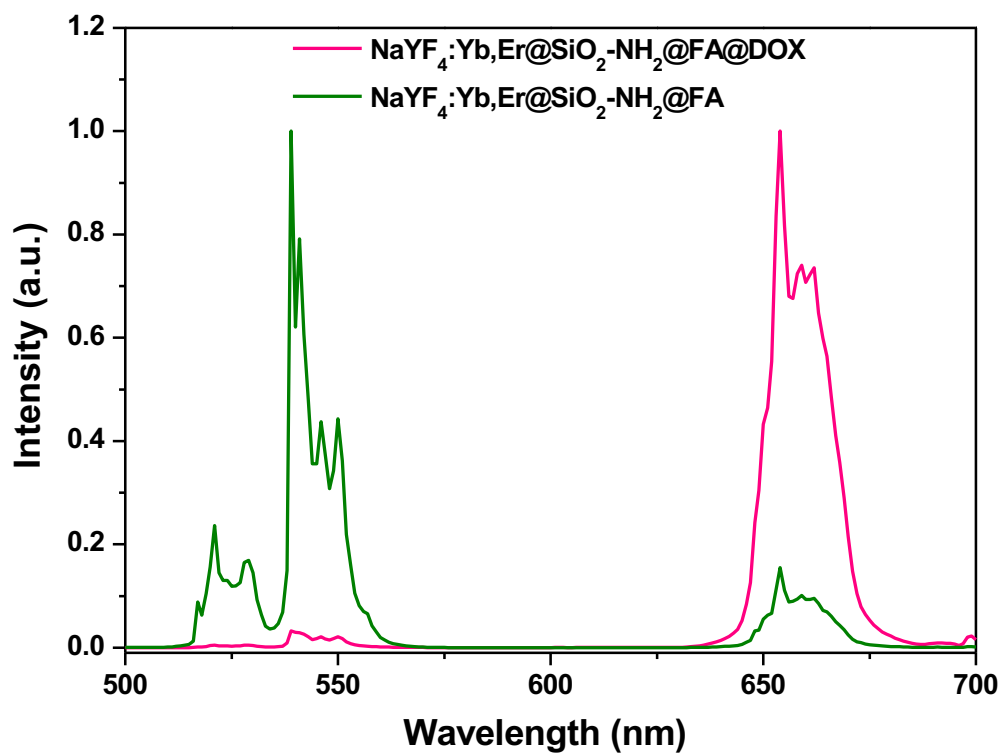


Figure S9. Upconversion emission spectra of $\text{NaYF}_4:\text{Yb,Er@SiO}_2\text{-NH}_2\text{@FA@DOX}$ after DOX loading under 975 nm excitation. A clear drop in green emission is observed in the DOX loaded material, due to overall with DOX absorption in that region. This confirms close proximity of the DOX molecules to the Er^{3+} ions.

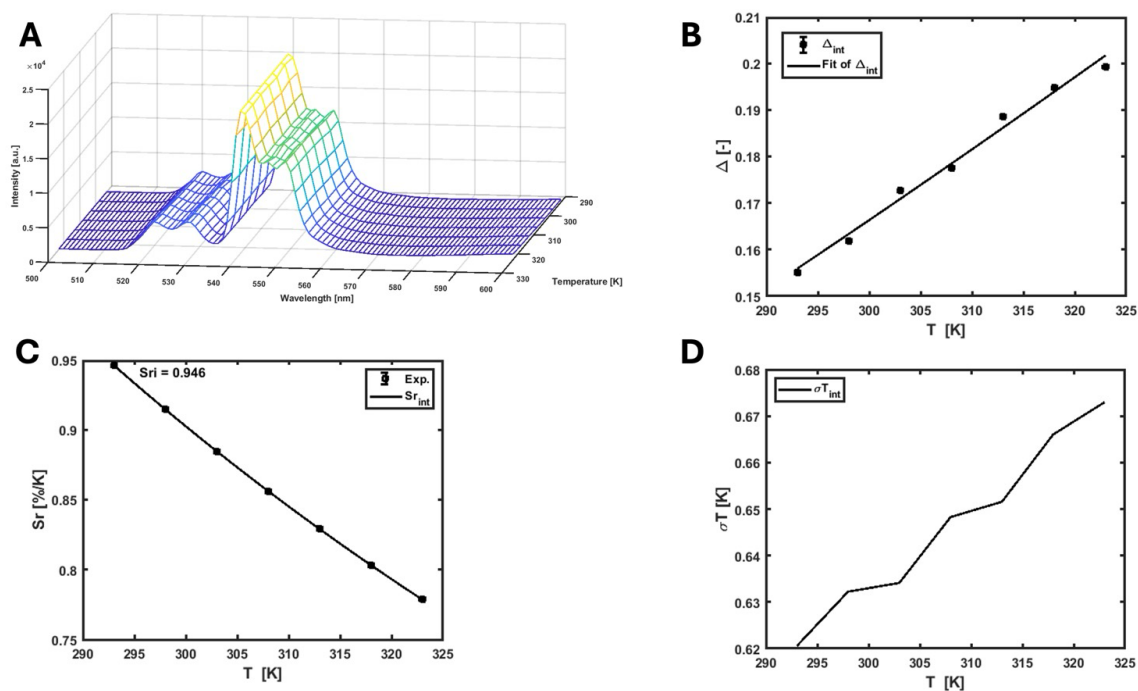


Figure S10. (A) Emission map of NaYF₄:Yb,Er@SiO₂-NH₂@FA@DOX dispersed in water, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the visible region under 975 nm excitation, (B) graph showing the Δ parameters versus temperature. graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

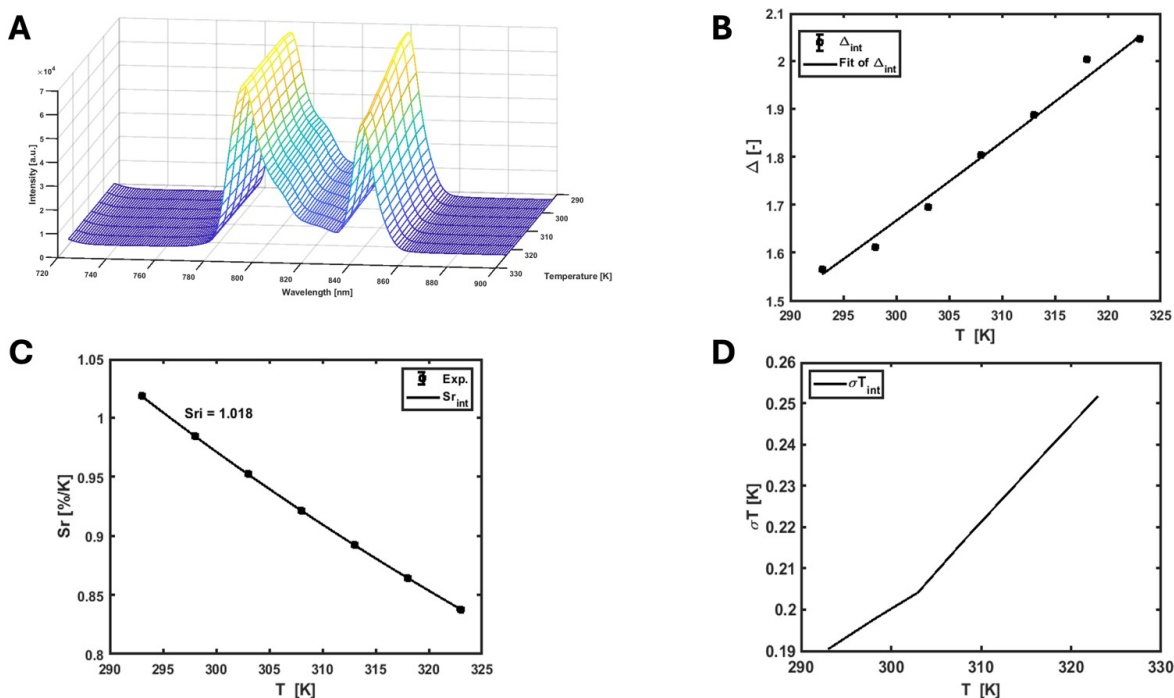


Figure S11. (A) Emission map of NaYF₄:Yb,Er@SiO₂-NH₂@FA@DOX dispersed in water, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the NIR-I region under 975 nm excitation, (B) graph showing the Δ parameters versus temperature. graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σ_T) at varying temperatures.

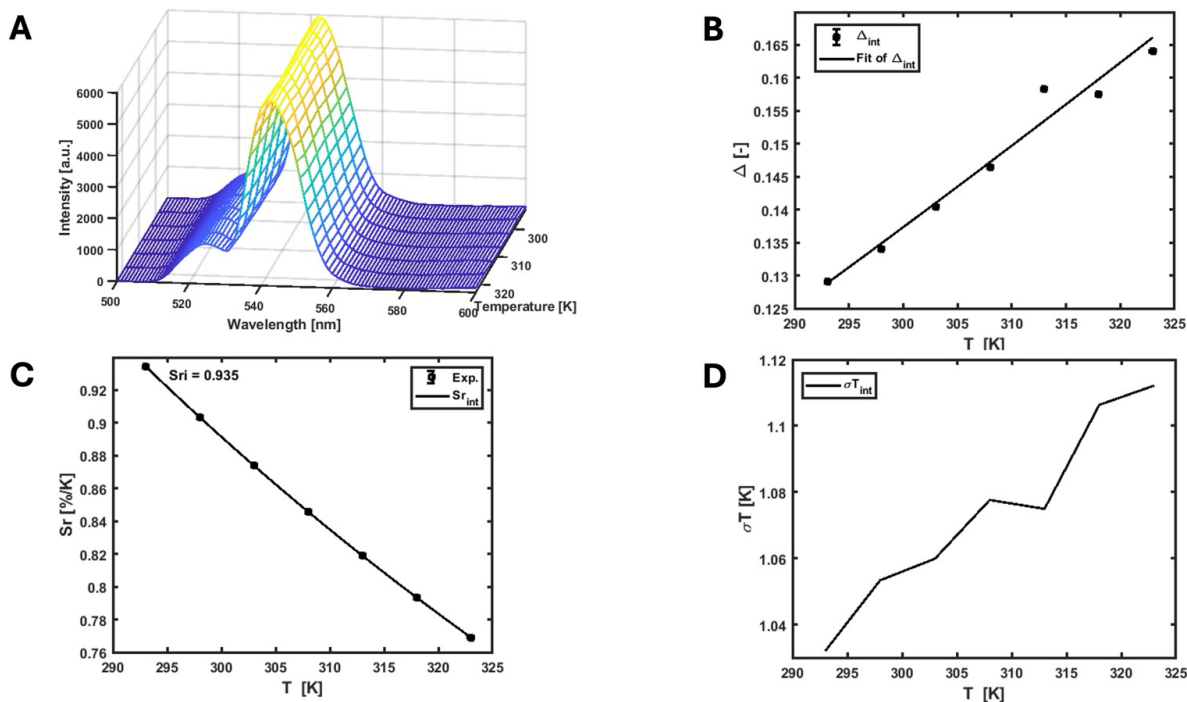


Figure S12. (A) Emission map of NaYF₄:Yb,Er@SiO₂-NH₂@FA@DOX dispersed in water, recorded across the temperature range of 293.15–323.15 K (20–50 °C) in the visible region under 940 nm excitation, (B) graph showing the Δ parameters versus temperature. graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

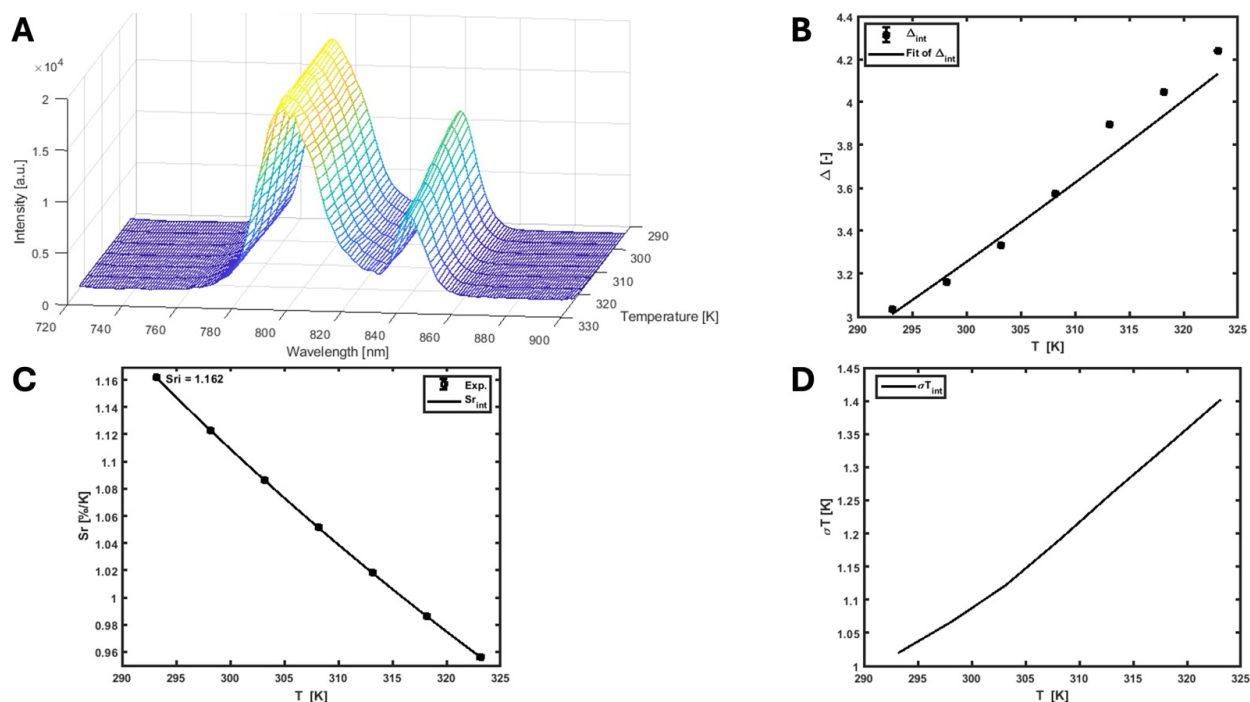


Figure S13. Emission map of NaYF₄:Yb,Er@SiO₂-NH₂@FA@DOX dispersed in water, recorded through chicken breast tissue (5 mm thickness at light entrance and exit) across the temperature range of 293.15–323.15 K (20–50 °C) in the NIR-I region ($^2H_{11/2} \rightarrow ^4I_{13/2}$ and $^4S_{3/2} \rightarrow ^4I_{13/2}$) under 975 nm excitation, (B) graph showing the Δ parameter versus temperature. The points show the experimental Δ values, and the solid line shows the least squares fit to the experimental points. (C) Plot of the relative sensitivity (S_r) at varying temperatures. (D) Plot of the temperature uncertainty (σT) at varying temperatures.

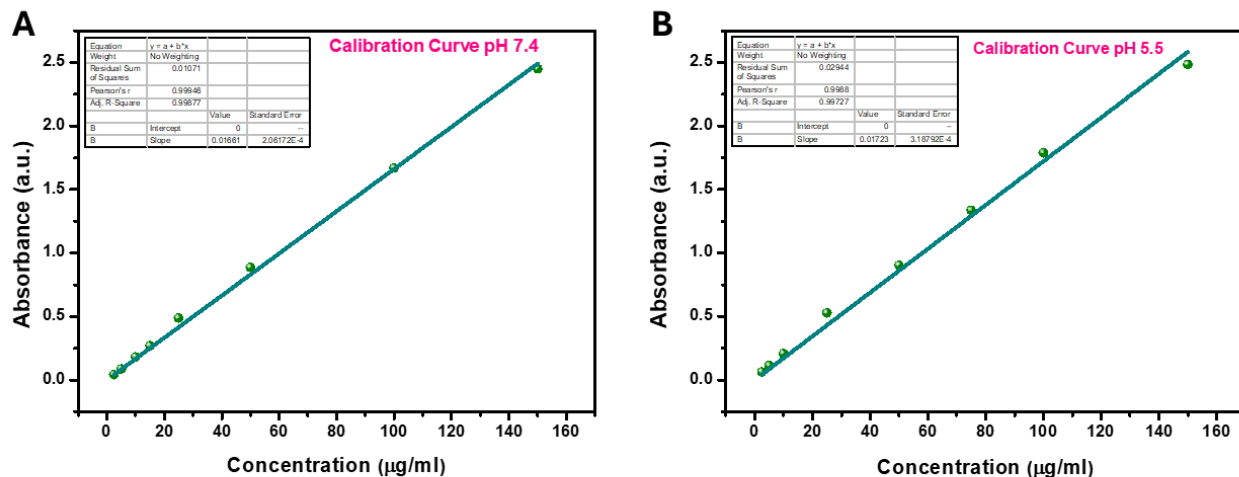


Figure S14. DOX calibration curves in PBS at (A) pH 7.4, and (B) pH 5.5.

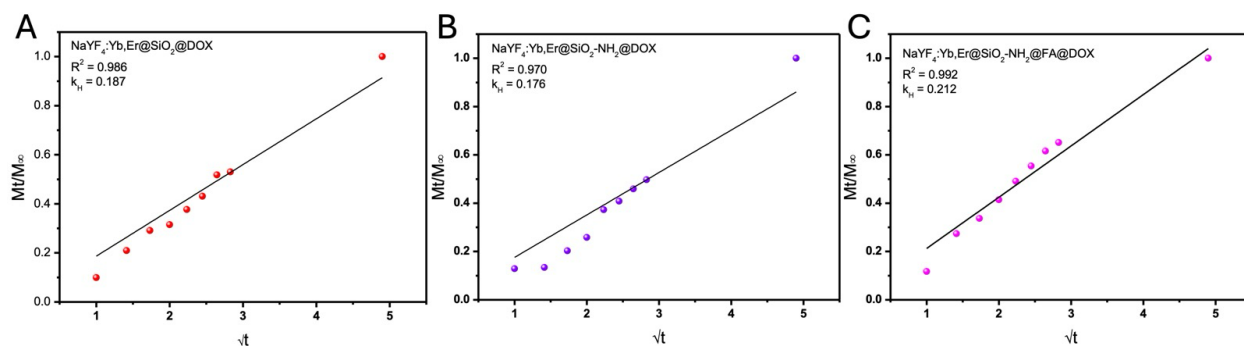


Figure S15. Higuchi kinetic model fitting of DOX release from (A) NaYF₄:Yb,Er@SiO₂, (B) NaYF₄:Yb,Er@SiO₂-NH₂, and (C) NaYF₄:Yb,Er@SiO₂-NH₂@FA particles at pH 5.5.

Table S1. Higuchi kinetic parameters (k_H, R²) for DOX release from NaYF₄:Yb,Er@SiO₂, NaYF₄:Yb,Er@SiO₂-NH₂, and NaYF₄:Yb,Er@SiO₂-NH₂@FA particles at pH 5.5.

Sample	k _H	R ²
NaYF ₄ :Yb,Er@SiO ₂ @DOX	0.187	0.986
NaYF ₄ :Yb,Er@SiO ₂ -NH ₂ @DOX	0.176	0.970
NaYF ₄ :Yb,Er@SiO ₂ -NH ₂ @FA@DOX	0.212	0.992

Table S2. Korsmeyer–Peppas kinetic parameters (n , R^2) for DOX release from $\text{NaYF}_4\text{:Yb,Er@SiO}_2$, $\text{NaYF}_4\text{:Yb,Er@SiO}_2\text{-NH}_2$, and $\text{NaYF}_4\text{:Yb,Er@SiO}_2\text{-NH}_2\text{@FA}$ particles at pH 5.5.

Sample	n (diffusional exponent)	R^2
$\text{NaYF}_4\text{:Yb,Er@SiO}_2\text{@DOX}$	0.78	0.978
$\text{NaYF}_4\text{:Yb,Er@SiO}_2\text{-NH}_2\text{@DOX}$	0.74	0.910
$\text{NaYF}_4\text{:Yb,Er@SiO}_2\text{-NH}_2\text{@FA@DOX}$	0.84	0.967

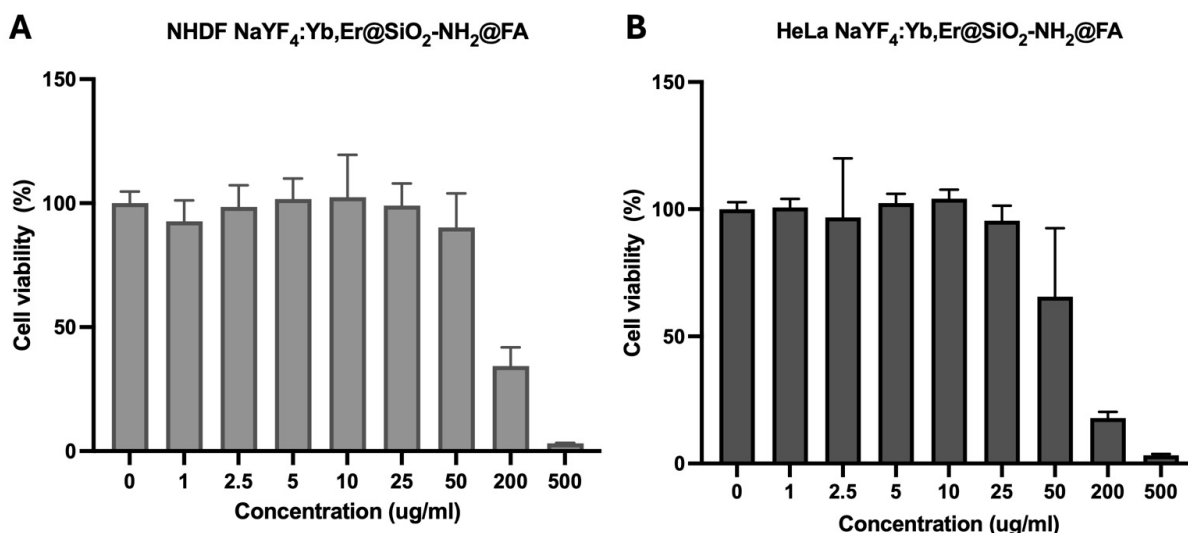


Figure S16. Cytotoxicity test results plotted as bar graph (mean + standard deviation) of the tested concentrations of $\text{NaYF}_4\text{:Yb,Er@SiO}_2\text{-NH}_2\text{@FA}$ particles, ranging from 0-500 ug/mL on the chosen (A) NHDF and (B) HeLa cell line models. Statistical analysis applying the Mann-Whitney-Wilcoxon non-parametric test yielded no statistically significant differences between the medians of the control group and the median of every tested concentration group.