

Electronic Supplementary Information

Enhancing thermal stability of n-type conduction in carbon nanotubes via cation replacement mediated by bicyclic guanidinium salts

Kaho Kawasaki,^a Mayuko Nishinaka,^a Yasuko Koshiba,^{a,b} Azumi Akiyama,^{a,b} Qingshuo Wei,^{c,d} Masahiro Funahashi,^{*,a,b} Shohei Horike,^{*,a,b,c,e}

^a Department of Chemical Science and Engineering, Graduate School of Engineering, Kobe University, 1-1 Rokkodai-cho, Kobe 657-8501, Japan

^b Research Center for Membrane and Film Technology, Kobe University, 1-1 Rokkodai-cho, Kobe 657-8501, Japan

^c Research Institute of Core Technology for Materials Innovation, Department of Materials and Chemistry, National Institute of Advanced Industrial Science and Technology (AIST), 1-1-1 Higashi, Tsukuba, Ibaraki 305-8565, Japan

^d Graduate School of Pure and Applied Science, University of Tsukuba, 1-1-1 Tennodai, Tsukuba, Ibaraki 305-8577, Japan

^e Center for Environmental Management, Kobe University, 1-1 Rokkodai-cho, Kobe, 657-8501, Japan

*Corresponding author.

Tel: +81-78-803-6150. Email: funahashi.masahiro@phoenix.kobe-u.ac.jp (Masahiro Funahashi).

Tel: +81-78-803-6194. Email: horike@crystal.kobe-u.ac.jp (Shohei Horike).

Table S1. List of chemicals and materials used in this study.

	purity (%)	supplier	remark
Carbon nanotube	99	OCSiAl	Single-walled CNT. Tuball. Diameter: 1.6±0.4 nm.
Brij L4	80	Sigma	Nonionic surfactant
Acetone	>99.0	Nacalai	
Acetonitrile	99.5	Nacalai	
TBD	≥98.0	TCI [†]	
HCl	–	FUJIFILM [‡]	35.0–37.0 wt% aqueous solution
HNO ₃	–	Nacalai	60–61 wt% aqueous solution
Li-TFSI	>98.0	TCI	
CoCp* ₂	–	Sigma	

[†]TCI: Tokyo Chemical Industry; [‡]FUJIFILM: FUJIFILM Wako Pure Chemical

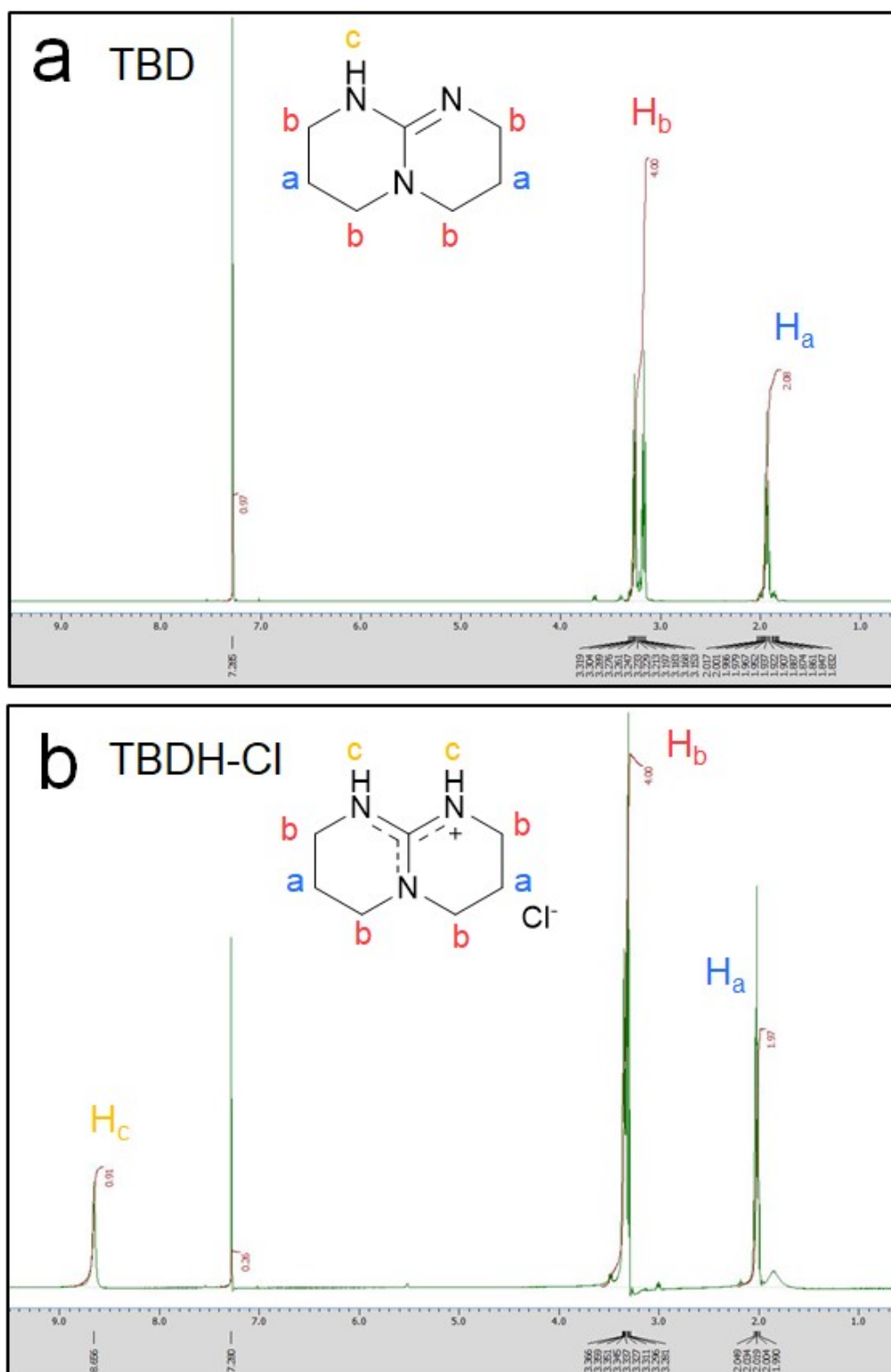


Fig. S1. 1H NMR spectra of (a) TBD (as comparison) and (b) synthesized TBDH-Cl. Spectra of TBDH- NO_3 and TBDH-TFSI can be found in our previous study.^{S1}

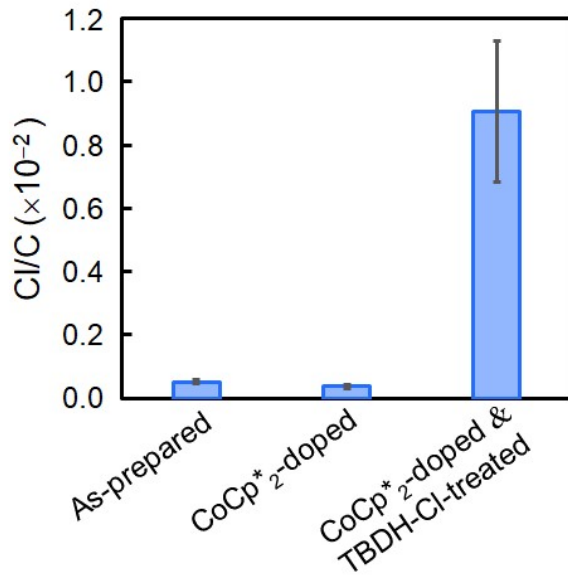


Fig. S2. Compositional ratios of chlorine to carbon (Cl/C) in as-prepared, CoCp*₂-doped, and CoCp*₂-doped followed by TBDH-Cl treatment of CNT films, derived from the data in Figure 5a of the main article. Data statistics: $n = 3$; error bars represent standard deviations (SDs).

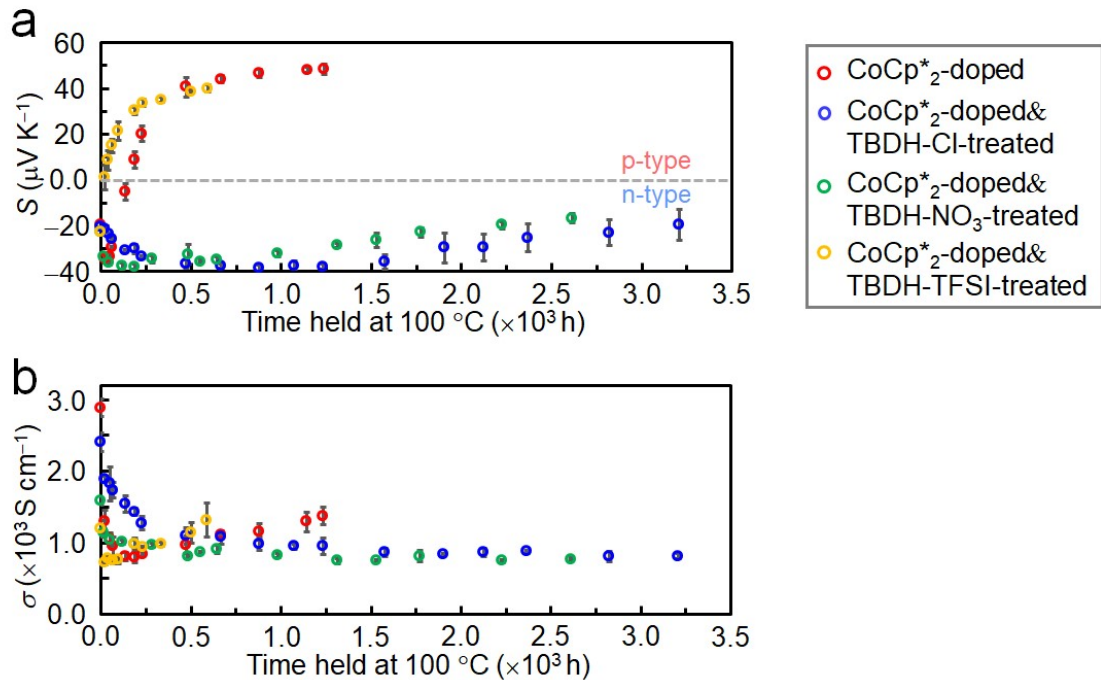


Fig. S3. Variation with time in (a) Seebeck coefficient (S) and (b) electrical conductivity (σ) of CoCp*₂-doped and TBDH-X- ($X^- = \text{Cl}^-$, NO_3^- or TFSI^-) treated CNT films during incubation at 100 °C in air. Data statistics: $n = 3$; error bars represent SDs.

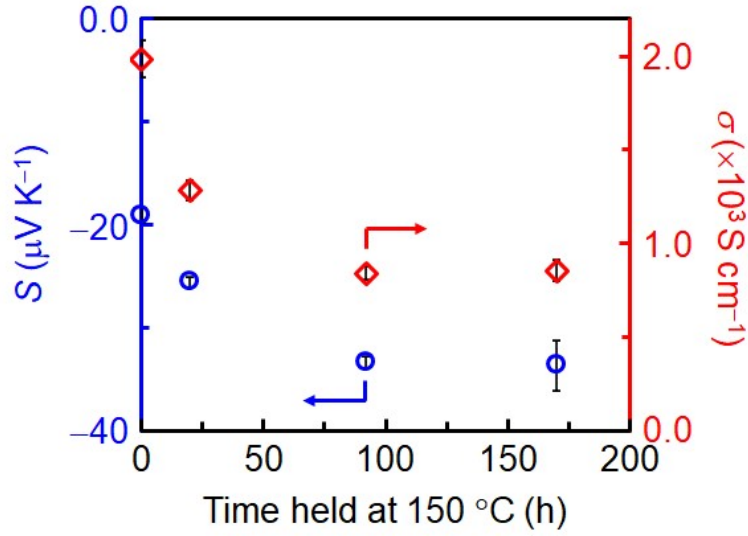


Fig. S4. Variation with time in Seebeck coefficient (S) and electrical conductivity (σ) of TBD-doped CNT films during incubation at 150 °C in air. Data statistics: $n = 3$; error bars represent SDs. Considering the faster depletion behaviors in these parameters than those during incubation at 100 °C in air reported in our previous papers,^{S1,S2,S3} the retention ability of n-doped states of CNTs becomes poor at the higher temperatures. On the other hand, our previous study revealed that the TBD-doped CNTs showed decrease in electrical conductivity while retaining the negative Seebeck coefficient in high-humid conditions.^{S3} The once decreased conductivity recovered by evacuating. The adsorbed water molecules seemed to contribute to the decrease of electron mobility rather than dedoping. Based on these considerations, we set the incubation temperature as 100 °C throughout the study to confirm the effect of cation replacement on the stability of n-doped states.

Table S1 Comparison of power factors of n-type materials. Note that the power factor is a metric for comparing power output of materials; dimensionless figure of merit (including thermal conductivity) is important to comparatively evaluate the heat-to-electricity energy conversion efficiency.

Material	Power factor ($\mu\text{W m}^{-1} \text{K}^{-2}$)	Ref.
CNT	150	25
CNT	100–150	26
CNT	82–86	27
CNT	230	28
CNT	261	34
CNT	258	50
poly(benzodifurandione)	90	48
CNT	62–101	This study

Reference

- (S1) M. Nishinaka, I. Harada, K. Akaike, Q. Wei, Y. Koshiha, S. Horike, K. Ishida, Electrochemical charge-carrier modulation of carbon nanotubes using ionic liquids derived from organic superbases for stable thermoelectric materials, *Carbon*, 2024, **218**, 118667.
- (S2) S. Horike, Q. Wei, K. Akaike, K. Kirihaara, M. Mukaida, Y. Koshiha and K. Ishida, *Nat. Commun.*, 2022, **13**, 3517.
- (S3) M. Nishinaka, Q. Wei, Y. Koshiha and S. Horike, *Energy Mater. Adv.*, 2024, **5**, 0123.