

Supporting information

Oxygen vacancies and CDs synergistically mediated CDs/TiO₂ composite materials: Investigation of electronic energy band structure modulation, photocatalytic performances, and mechanisms

Dahui Feng^a, Han Zhao^a, Yanhua Song^a, Ruyan Xie^a, Xiaozhen Zhang^a, Shuzhe Zhang^a, Peng Chen^{*b} and Haifeng Zou^{*a}

^a College of Chemistry, Jilin University, Qianjin Street 2699, Changchun 130012, China.

E-mail: haifengzou0431@sohu.com

^B The Second Hospital of Jilin University, Jilin University, Yatai Street 4026, Changchun 130022, China.

E-mail: c_p@jlu.edu.cn

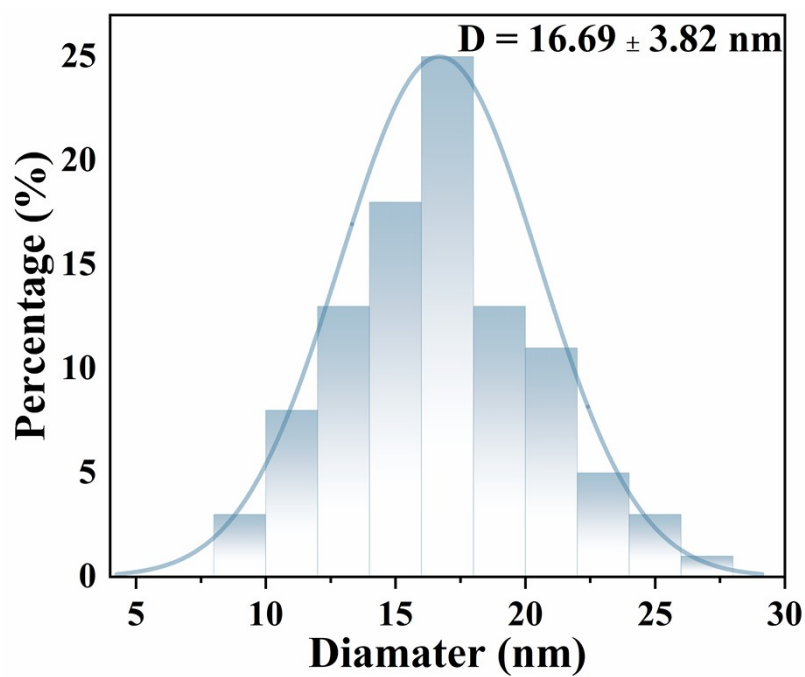


Fig. S1. Column diagram of DT dimension analysis in TEM.

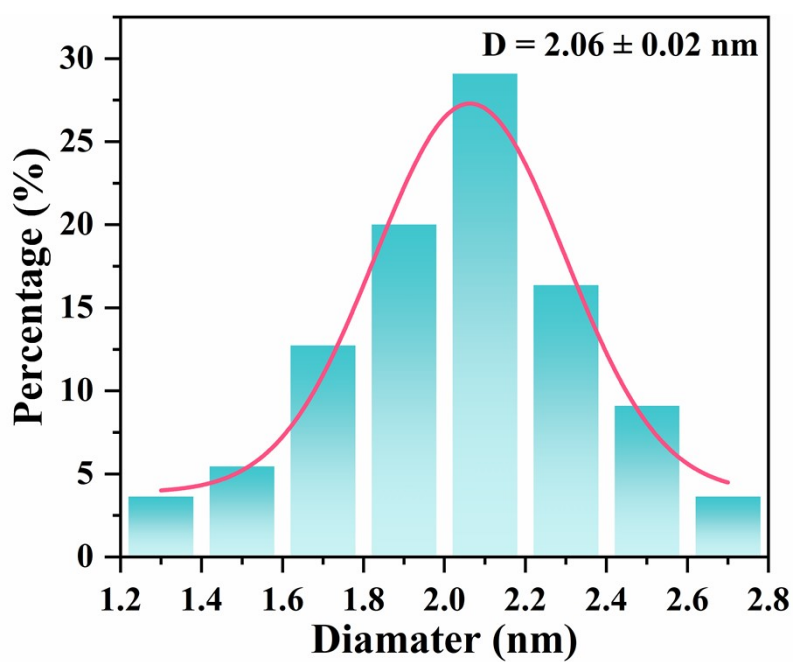


Fig. S2. Particle size analysis of CDs on the surface of CDT in transmission electron microscopy.

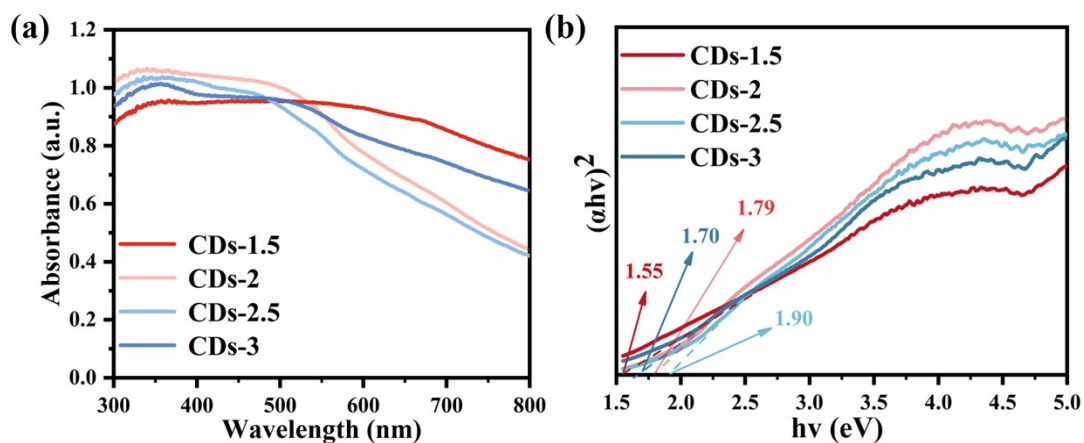


Fig. S3. (a) UV-visible diffuse reflection spectra of CDs synthesized with different amounts of citric acid (1.5, 2.0, 2.5 and 3.0 g); (b) Plots of the band-gap values.

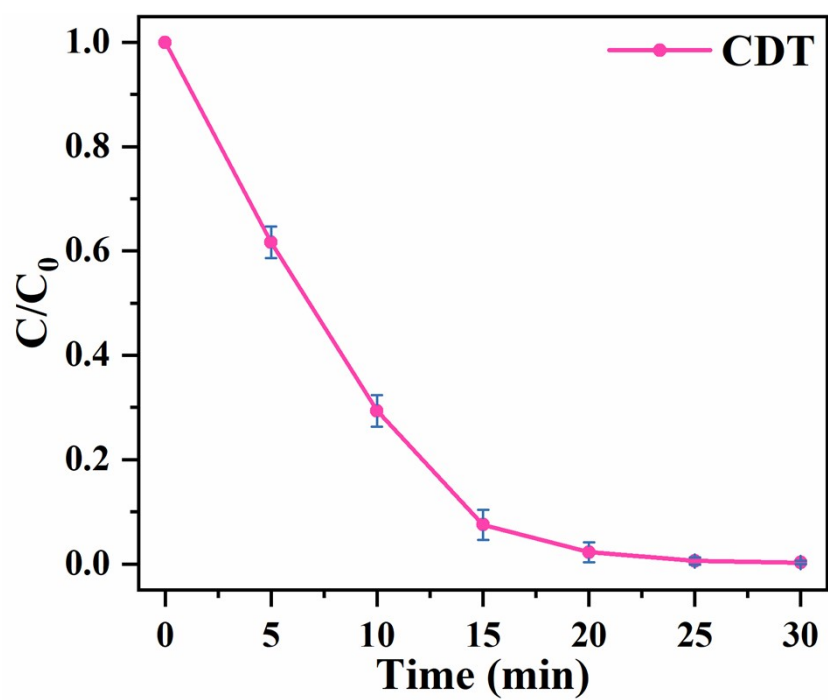


Fig. S4. Visible light degradation curve of RhB by CDT.

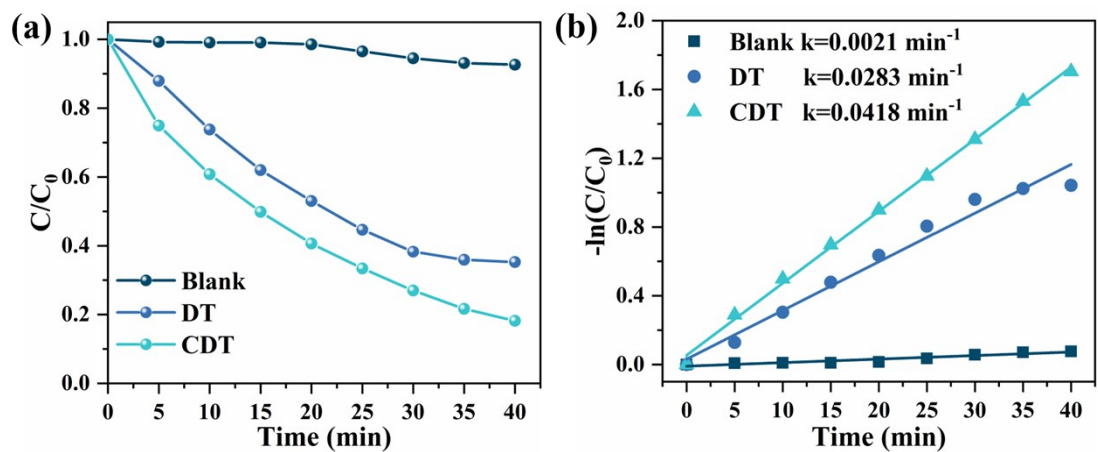


Fig. S5. (a) Degradation curves of Blank, DT, and CDT on 20 mg/L TC under visible light illumination; (b) Reaction kinetics curve.

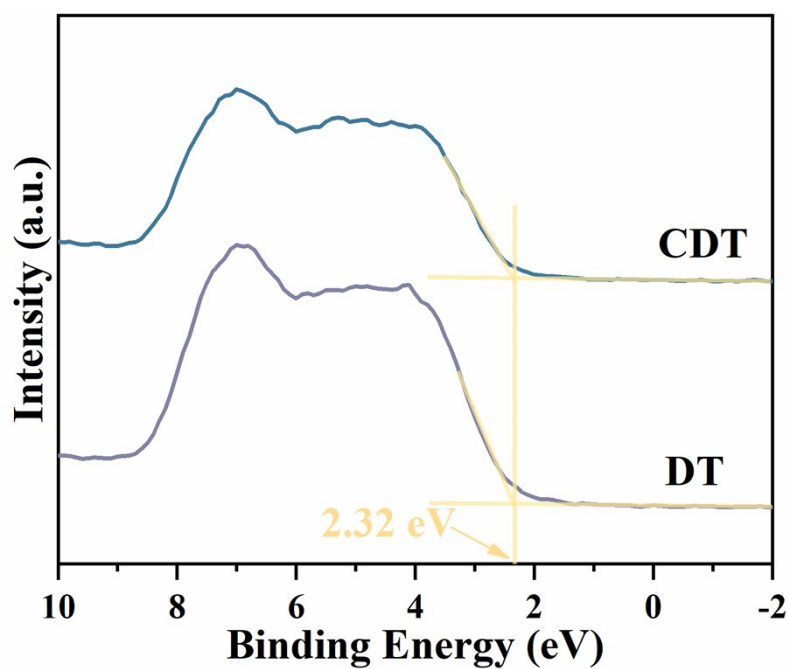


Fig. S6. XPS valence band spectra of DT and CDT.

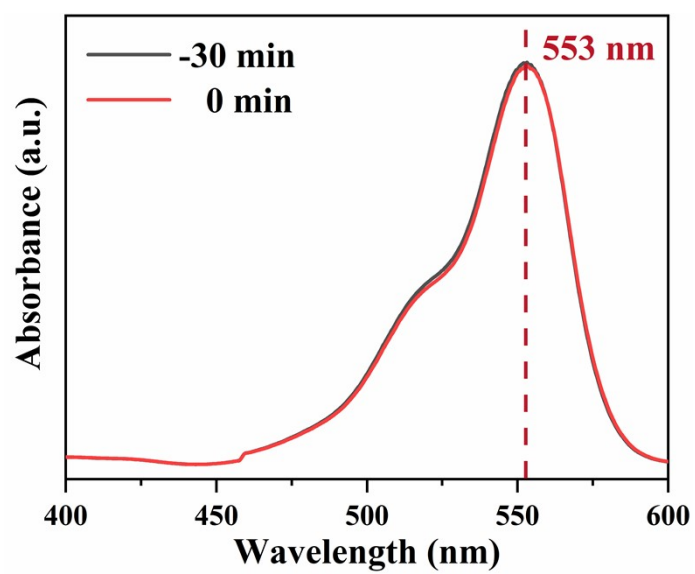


Fig. S7. Changes of RhB absorbance before and after dark state adsorption.