Electronic Supplementary Information (ESI)

Polymer-Assisted Synthesis of Mixed-Halide Quasi-2D Perovskites for Tunable Blue-Green Lasers

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Experimental Section

- 1. Materials: Lead bromide (PbBr₂, 99.99%), cesium bromide (CsBr, 99.99%), butylammonium bromide (BABr, 99.5%), cesium chloride (CsCl, 99.5%), and lead chloride (PbCl₂, 99.999%) were purchased from Xi'an Polymer Light Technology Corp and used as received without further purification. Dimethyl sulfoxide (DMSO), and Polyvinyl pyrrolidone (PVP) were purchased from Beijing InnoChem Science & Technology Co., Ltd.
- 2. PVP-modified Quasi-2D Perovskite thin films preparation: The PVP-modified quasi-2D perovskite $BA_2Cs_{n-1}Pb_nX_{3n+1}$ (n = 5) films were synthesized using a one-step spin-coating method in a nitrogen-filled glovebox. The BABr, CsBr, PbBr₂, CsCl and PbCl₂ was dissolved in DMSO solvent to prepare starting solution (maintain Pb²⁺ concentration of 0.15 mol/L). 10 mg/mL PVP was added to the precursor solution. The obtained perovskite precursor solution was spin-coated onto the substrates via spin coating process at 3000 rpm for 60 s. Then the perovskite films were annealed at 60 °C for 5 min. Mixed-halide perovskite films were obtained by varying the precursor solution constitution following a similar strategy.
- **3. Morphology and spectra measurements:** The morphologies using field emission scanning electron microscopy (FESEM, Thermo Fisher Scientific) at acceleration voltages of 10 kV. UV-vis absorption spectra were obtained from a Thermo Fisher Scientific Evolution 201. PL spectra were measured on an Edinburgh FLS1000 spectrometer. The bonding energy of Pb, Br and Cl was tested by an X-ray photoelectron spectrometer (XPS, K-Alpha+). X-ray diffraction patterns were measured by a D/max 2500 X-ray diffractometer with Cu K α radiation ($\lambda = 1.54050$ Å) operated in the 2 θ range from 5 to 40.

- **4. TA Measurements:** A femtosecond laser system (Pharos, Light Conversion)delivered laser pulses at-1030 nm (180 fs, 6 kHz), which were then divided into two components by using a 9:1 beam splitter. The major component was sent to an optical parametric amplifier. (Orpheus, Light Conversion) to generate the pump pulses (370 nm, 6 kHz). The minor component was further attenuated and focused into a 3-mm sapphire plate to generate the probe pulses. Both the pump and probe pulses were guided into a Harppa spectrometer and time resolved spectral data were recorded. A short-pass filter was inserted into the probe beam to cut off the fundamental light of 1030 nm. The time delay between the pump and probe beams were regulated through a computer-controlled motorized translation stage in the probe beam. The temporal resolution between the pump and the probe pulses was determined to be ~200 fs (FWHM). The transmitted light was detected by a CMOS linear image sensor. Analysis of the kinetic traces derived from time-resolved spectra was performed using nonlinear least-square fitting to a general sum-of-exponentials function after deconvolution of instrument response function (IRF). All the spectroscopic measurements were carried out at room temperature.
- 5. ASE Measurements: For the ASE measurements, the laser pulses from the femtosecond laser system (Pharos, Light Conversion) delivered at 1030 nm(180 fs, 6 kHz), which were then divided into two components using a 9:1 beam splitter. The major component was sent to an optical parametric amplifier (Orpheus, Light Conversion) to generate the pump pulses (400 nm, 6 kHz). These pump pulses were then focused vertically onto the PVP-modified BA₂Cs_{n-1}Pb_nX_{3n+1} film with dimension of ~0.1×1.5 mm² via a cylindrical lens, and the emission from the edge of the film serving as the optical waveguide was vertically collected and detected with a liquid-nitrogen-cooled CCD (Isoplane320, Rrineenton Instuments) attached to a polychromator (Spectropro-550i, Acton).

6. Laser Measurements: The PVP-modified BA₂Cs_{n-1}Pb_nX_{3n+1} micro-ring arrays were performed on a home-made inverted fluorescence microscopy equipped with a 40× objective. The optical. parametric amplifier (Orpheus, Light Conversion, 400 nm,6 kHz) was focused to a 200-μmdiameter spot. The PL spectra were collected in a reflection mode with a movable aperture on the optical path at the front focal plane, collected and detected with a liquid-nitrogen-cooled CCD (nanosecond-pulsed laser system, excitation wavelength: 355 nm, pulse width: 7 ns, repetition rate: 20 Hz).

7. Addition sections



Fig. S1 (a-b) Steady-state absorption and PL spectra of PVP-modified $BA_2Cs_{n-1}Pb_nX_{3n+1}$ (n = 5) films with varying Cl/Br ratio. (d-e) Absorption and PL spectra of pristine $BA_2Cs_{n-1}Pb_nX_{3n+1}$ films. Correlation between wavelength of PL peak, band edge absorption and chlorine content of PVP-modified (c) and pristine (f) $BA_2Cs_{n-1}Pb_nX_{3n+1}$ films, respectively.



Fig. S2 The XRD patterns of pristine and PVP-modified $BA_2Cs_{n-1}Pb_nX_{3n+1}$ films.



Fig. S3 PL image of pristine and PVP-modified quasi-2D perovskite $BA_2Cs_{n-1}Pb_nX_{3n+1}$ thin films under 365 nm UV excitation.



Fig. S4 (a) The 3D TOF-SIMS profile mapping of the CNO⁻ distribution. (b) TOF-SIMS depth profiles of functional groups in the Cl_{0.15}Br_{0.85} film. (c) The 3D TOF-SIMS profile mapping of the CsBrCl⁻, PbBrCl₂⁻ and Br⁻ distributions.



Fig. S5 (a-c) XPS spectra (Cl 2p, Br 3d and Pb 4f) of the pristine sample and PVP-modified films (green: 5 mg/mL PVP, blue: 10 mg/mL PVP).



Fig. S6 (a-c) The SEM images of pristine and PVP-modified $BA_2Cs_{n-1}Pb_nX_{3n+1}$ films.



Fig. S7 (a-c) The AFM images of pristine and PVP-modified BA₂Cs_{n-1}Pb_nX_{3n+1} films.



Fig. S8 Optical images of PVP-modified $BA_2Cs_{n-1}Pb_nX_{3n+1}$ films.

Fig. S9





Fig. S9 Pump-fluence dependent PL intensity of different Cl⁻/Br⁻ ratios films at 400 nm excitation and the plot of integrated PL intensities versus pump density of different Cl⁻/Br⁻ ratios films. (there are six groups of Cl_{0.45}Br_{0.55}, Cl_{0.30}Br_{0.70}, Cl_{0.22}Br_{0.78}, Cl_{0.15}Br_{0.85}, Cl_{0.11}Br_{0.89} and pure Br).



Fig. S10 (a) Optical images of precursor solution of mixed-halide perovskite ($Cl_{0.66}Br_{0.34}$ could not dissolve completely). (b) Correlation between Cl^- content and ASE/PL wavelength of PVP-modified BA₂Cs_{n-1}Pb_nX_{3n+1} films. (c-h) Pump-fluence-dependent PL intensity of X= $Cl_{0.66}Br_{0.34}$, $Cl_{0.56}Br_{0.44}$ and $Cl_{0.52}Br_{0.48}$ films.



Fig. S11 Summary of the ASE thresholds of the PVP-modified $BA_2Cs_{n-1}Pb_nX_{3n+1}$ films.



Fig. S12 PDMS-confined solution-growth strategy presentation diagram.



Fig. S13 The spectra of ASE and laser with $Cl_{0.45}Br_{0.55}$ and $Cl_{0.22}Br_{0.78}$.



Fig. S14 Spectral stability test by extending the duration of optical pumping from PVP-modified $BA_2Cs_{n-1}Pb_nX_{3n+1}$ micro-ring.



Fig. S15 (a) Nanosecond-pulsed pump-fluence dependent PL intensity of $X = Cl_{0.15}Br_{0.85}$, $Cl_{0.11}Br_{0.89}$ and pure Br films at 355 nm excitation. (b-d) The plot of integrated PL intensities versus pump density of $X = Cl_{0.15}Br_{0.85}$, $Cl_{0.11}Br_{0.89}$ and pure Br films.



Fig. S16 (a) Pump-fluence-dependent PL intensity from $Cl_{0.22}Br_{0.78}$ micro-rings. (b) The plot of integrated PL intensities versus pump density.

	GSB _{n=2}	$\mathbf{GSB}_{n=4}$	GSB _n ~∞ (decay)	GSB _n ∼∞ (generate)
Peak/nm	420	475	485	485
τ_1/ps	0.11	0.35	92.37	0.51
$ au_1$ %	59.41%	25.67%	9.28%	100%
τ_2/ps	1.74	6.07	1660.36	-
τ2%	40.59%	74.33%	90.72%	-
$ au_{ave}/ps$	0.77	4.60	1514.86	0.51

Table S1. The fitting parameters for the decay kinetics of GSB features with different *n*-valuesin PVP-modified quasi-2D perovskite $BA_2Cs_{n-1}Pb_nX_{3n+1}$.

Spectral region	Material	Pump source	λ/nm	Threshold /µJ cm ⁻²	Reference
<450nm	$BA_2Cs_{n-1}Pb_nCl_{3n+1}$	340 nm, 250 fs, 1 kHz	425	~25	20211
450-500nm	CsPb(Cl/Br)3	400 nm, 100 fs, 1 kHz	~475	~50	2015 ²
	CsPbX3	400 nm, 100 fs, 1 kHz	~480	~9	2015 ³
	CsPbCl _{1.6} Br _{1.4}	400 nm, 100 fs, 1 kHz	470	35.6	2020^{4}
	CsPbBr2.25Cl0.75	355 nm, 70 fs, 15 Hz	498	45	20215
	$DPEA_2Cs_{n-1}Pb_n(Cl/Br)_{3n+1}$	355 nm, 80 fs, 1 kHz	480	6.5	20226
	$OA_2(Rb/Cs)_{n-1}Pb_nBr_{3n+1}$	400 nm, 200 fs, 1 kHz	484	13.5	20247
	BA2Csn-1PbnX3n+1	400 nm, 180 fs, 6 kHz	477	38.39	This work
	BA2Csn-1PbnX3n+1	400 nm, 180 fs, 6 kHz	489	23.57	This work
	BA2Csn-1PbnX3n+1	400 nm, 180 fs, 6 kHz	496	16.74	This work

Table S2. The summary of different blue-green lasers.

Spectral region	Material	Pump source	λ/nm	Threshold /µJ cm ⁻²	Reference
500-550nm	BA ₂ Cs _{n-1} Pb _n Cl _{3n+1}	340 nm, 250 fs, 1 kHz	~520	/	20211
	$DPEA_2Cs_{n-1}Pb_n(Cl/Br)_{3n+1}$	355 nm, 80 fs, 1 kHz	~513	/	2022 ⁶
	$OA_2(Rb/Cs)_{n-1}Pb_nBr_{3n+1}$	400 nm, 200 fs, 1 kHz	~510	/	20247
	NMA ₂ FA _{n-1} Pb _n X _{3n+1}	400 nm, 150 fs, 1 kHz	530	8.5	2018 ⁸
	BA2MA _{n-1} Pb _n Br _{3n+1}	400 nm, 150 fs, 1 kHz	~542	13.6	2018 ⁹
	OA2MA _{n-1} Pb _n Br _{3n+1}	400 nm, 100 fs, 1 kHz	548	17.8	201810
	PEA ₂ FA _{n-1} Pb _n Br _{3n+1}	400 nm, 150 fs, 1 kHz	542	4	202011
	$BA_2Cs_{n-1}Pb_nBr_{3n+1}$	340 nm, 250 fs, 1 kHz	~535	7.2	20211
	BA2Csn-1PbnX3n+1	400 nm, 180 fs, 6 kHz	505	16.82	This work
	BA2Csn-1PbnX3n+1	400 nm, 180 fs, 6 kHz	510	16.89	This work
	BA2Csn-1PbnX3n+1	400 nm, 180 fs, 6 kHz	530	12.21	This work

Cl0.45Br0.55	Threshold ∕µJ∙cm⁻²	Clo.30Bro.70	Threshold ∕µJ·cm ⁻²	Cl0.22Br0.78	Threshold ∕µJ·cm ⁻²
1	37.06	1	23.18	1	17.14
2	38.92	2	24.68	2	17.27
3	37.55	3	23.97	3	18.22
4	38.68	4	22.89	4	16.47
5	39.12	5	23.9	5	16.97
6	38.39	6	23.57	6	16.74
Average	38.28666667	Average	23.69833333	Average	17.11833333
SD	0.813428956	SD	0.63508792	SD	0.607783673
Cl _{0.45} Br _{0.55}	Threshold /μJ·cm ⁻²	Cl _{0.11} Br _{0.89}	Threshold ∕µJ·cm ⁻²	Br	Threshold ∕µJ·cm ⁻²
1	17.26	1	15.86	1	11.92
2	17.55	2	17.17	2	12.96
3	16.14	3	17.05	3	12.9
4	17.08	4	17.25	4	12.44
5	17.95	5	16.14	5	11.85
6	16.82	6	16.89	6	12.21
Average	17.13333333	Average	16.72666667	Average	12.38
SD	0.623591747	SD	0.582569023	SD	0.475773055

Table S3. Summary of the ASE thresholds of the quasi-2D perovskite $BA_2Cs_{n-1}Pb_nX_{3n+1}$ thin films.

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