

Supporting Information

Cr³⁺-Induced Broadband Near-infrared I Combined with Near-infrared II Emission via Rare Earth Co-Doping in Cs₂NaInCl₆ for Multifunctional Detection

Hui Xie^{a,b}, Hui Fu^{b}, Zhentao Du^{a*}, Linjie Tong^{a,b}, Jinliang Jiang^b, Xue Jiang^c,*

Jialong Zhao^c, Weiyu Yang^b, and Jinju Zheng^{b}*

^a School of Resources, Environment and Materials, Guangxi University, Nanning
530004, China

^b Institute of Micro/Nano Materials and Devices, Ningbo University of Technology,
Ningbo 315211, China

^c School of Physical Science and Technology, MOE Key Laboratory of New
Processing Technology for Non-ferrous Metals and Materials, Guangxi University,
Nanning 530004, China

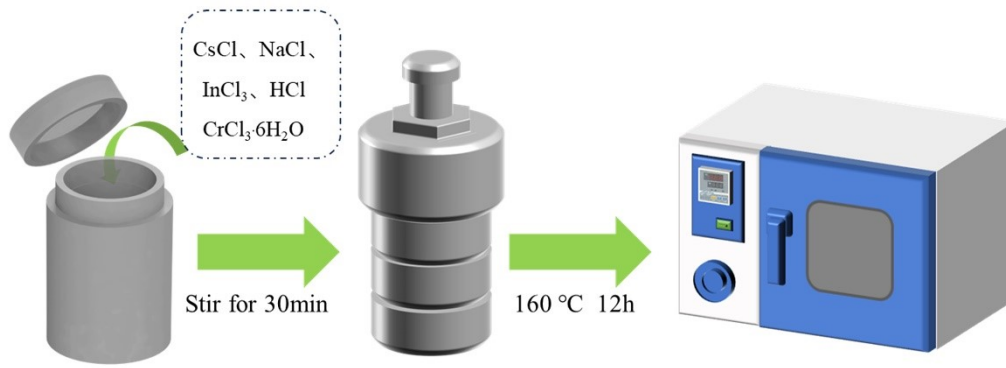


Figure S1. Schematic diagram of the hydrothermal synthesis for the samples.

Table S1. The ICP-OES data of Cr^{3+} doped CNIC DPs. The feeding ratios are determined based on the equation of $[\text{Cr}]/[\text{In}] \times 100\%$ and the actual doping concentrations are $[\text{Cr}]/([\text{In}] + [\text{Cr}]) \times 100\%$.

Samples	Feeding ratio [%]	Actual content [%]
CNIC: Cr^{3+}	Cr^{3+}	Cr^{3+}
	10	0.21
	20	0.34
	30	0.58
	40	1.15
	50	2.90

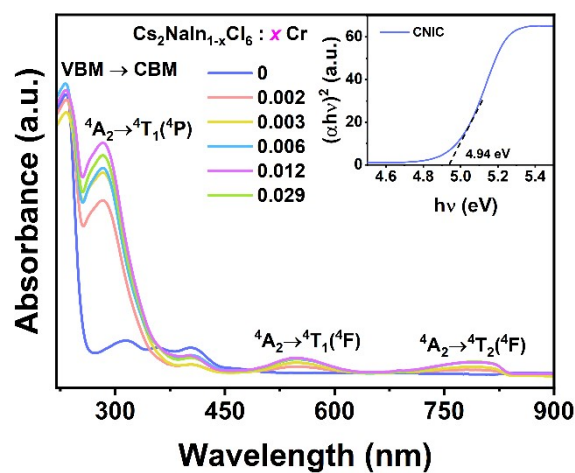


Figure S2. Absorption spectra of CNIC: Cr^{3+} with different Cr^{3+} doping concentrations.

Table S2. A summary of the NIR emission performance of different materials.

Samples	λ_{ex} [nm]	λ_{em} [nm]	FWHM [nm]	PLQY [%]	Thermal stability	Ref.
$\text{K}_2\text{LiGaF}_6: \text{Cr}^{3+}$	430	766	133	82.5	$I_{425 \text{ K}} = 85\%$	1
$\text{Cs}_2\text{AgInCl}_6: \text{Cr}^{3+}/\text{Mn}^{2+}$	350	1000	188	49.3	/	2
$\text{Cs}_2\text{NaScCl}_6: \text{Cr}^{3+}$	300	950	162	100	$I_{423 \text{ K-20 h}} = 91.9\%$	3
$\text{Cs}_2\text{AgInCl}_6: \text{Cr}^{3+}/\text{Er}^{3+}$	360/583/ 815	1010/ 1540	162	57.5	$I_{423 \text{ K-960 h}} = 85\%$	4
$\text{MgTiO}_3: \text{Cr}^{3+}$	514	866	154	31.55	$I_{350 \text{ K}} = 40\%$	5
$\text{Cs}_2\text{TeCl}_6: \text{Cr}^{3+}$	380	590	~100	81.5	$I_{473 \text{ K-1h}} = 87\%$ $I_{373 \text{ K-1h}} = 94\%$	6
$\text{Ca}_{0.5}\text{Cd}_{0.5}\text{Ti}_4(\text{PO}_4)_6: \text{Cr}^{3+}$	470	832	165	46.63	$I_{373 \text{ K}} = 43.48\%$	7
$\text{Sr}_2\text{ScTa}_{0.6}\text{Nb}_{0.4}\text{O}_6: \text{Cr}^{3+}$	337	954	179.7	86.3	$I_{373 \text{ K}} = 54\%$	8
$\text{Cs}_2\text{SnCl}_6: \text{W}^{4+}$	365	965	~130	/	$I_{300 \text{ K}} = 64\%$	9
$\text{CaLaMgSbO}_6: \text{Ni}^{2+}$	440	1546	229	32.94	$I_{373 \text{ K}} = 40\%$	10
$\text{Sr}_2\text{ScTaO}_6: \text{Ni}^{2+}$	420	1441	254	49.23	$I_{373 \text{ K}} = 45.65\%$	11
$\text{ZnAl}_2\text{O}_4: \text{Mn}^{2+}$	450	830	135	85.8	$I_{423 \text{ K}} = 60\%$	12
$\text{Li}_5\text{La}_3\text{Sb}_2\text{O}_{12}: \text{Mn}^{4+}$	341	709	~50	56.3	$I_{423 \text{ K}} = 54.7\%$	13
$\text{Ca}_3\text{Sn}_2\text{Ga}_2\text{GeO}_{12}: \text{Fe}^{3+}$	278/406	760	~80	54.1	$I_{423 \text{ K}} = 72\%$	14
$\text{Cs}_2\text{NaLuCl}_6: \text{Fe}^{3+}$	395	1265	309	17.29	$I_{373 \text{ K}} = 23.35\%$	15
$\text{NaBaScSi}_2\text{O}_7: \text{Fe}^{3+}$	301	815	116	16.2	$I_{423 \text{ K}} = 46\%$	16
$\text{Cs}_2\text{NaInCl}_6: \text{Cr}^{3+}$	290	960	160	92.41	$I_{423 \text{ K-480 h}} = 87\%$ $I_{340 \text{ K}} = 50\%$	This work

Note: I_T refers to the ratio of the emission intensity at a certain temperature T to that at the initial temperature.

Table S3. The ICP-OES data of Cr³⁺/Ho³⁺ doped CNIC DPs. The feeding ratios are determined based on the equation of [Cr]/[In]×100% and [Ho]/[In]×100%, while the actual doping contents are determined by [Cr]/([In]+[Cr]+[Ho])×100% and [Ho]/([In]+[Cr]+[Ho])×100%.

Sample	Feeding ratio [%]		Actual content [%]	
	Cr ³⁺	Ho ³⁺	Cr ³⁺	Ho ³⁺
CNIC: Cr ³⁺ /Ho ³⁺	40	20	0.99	0.61
	40	40	1.12	1.77
	40	80	0.89	2.99
	40	100	0.71	4.0
	40	120	0.34	7.78

Table S4. The ICP-OES data of Cr³⁺/Tm³⁺ doped CNIC DPs. The feeding ratios are determined based on the equation of [Cr]/[In]×100% and [Tm]/[In]×100%, while the actual doping contents are determined by [Cr]/([In]+[Cr]+[Tm])×100% and [Tm]/([In]+[Cr]+[Tm])×100%.

Sample	Feeding ratio [%]		Actual content [%]	
	Cr ³⁺	Tm ³⁺	Cr ³⁺	Tm ³⁺
CNIC: Cr ³⁺ /Tm ³⁺	40	20	0.90	0.58
	40	40	1.32	0.99
	40	80	1.18	1.52
	40	100	0.87	1.88
	40	120	0.43	2.93

Table S5. The ICP-OES data of Cr³⁺/Er³⁺ doped CNIC DPs. The feeding ratios are determined based on the equation of [Cr]/[In]×100% and [Er]/[In]×100%, while the actual doping contents are determined by [Cr]/([In]+[Cr]+[Er])×100% and [Er]/([In]+[Cr]+[Er])×100%.

Sample	Feeding ratio [%]		Actual content [%]	
	Cr ³⁺	Er ³⁺	Cr ³⁺	Er ³⁺
CNIC: Cr ³⁺ /Er ³⁺	40	20	1.02	0.32
	40	40	1.16	0.89
	40	80	0.98	1.86
	40	100	0.81	2.69
	40	120	0.44	4.16

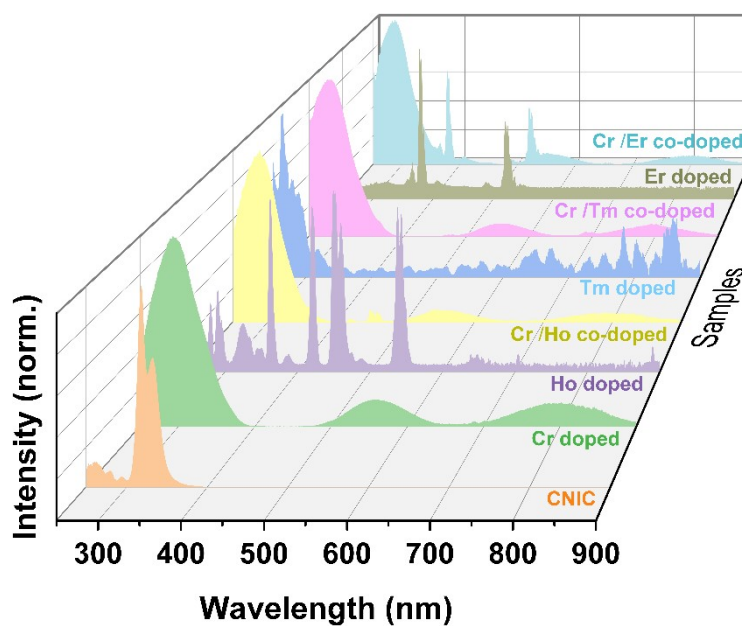


Figure S3. PL spectra of CNIC ($\lambda_{em}=445$ nm), CNIC: Cr³⁺ ($\lambda_{em}=960$ nm), CNIC: Ho³⁺ ($\lambda_{em}=1180$ nm), CNIC: Ho³⁺ ($\lambda_{em}=1180$ nm), CNIC: Tm³⁺ ($\lambda_{em}=1220$ nm), CNIC: Cr³⁺, Tm³⁺ ($\lambda_{em}=1220$ nm), CNIC: Er³⁺ ($\lambda_{em}=1540$ nm) and CNIC: Cr³⁺, Er³⁺ ($\lambda_{em}=1540$ nm).

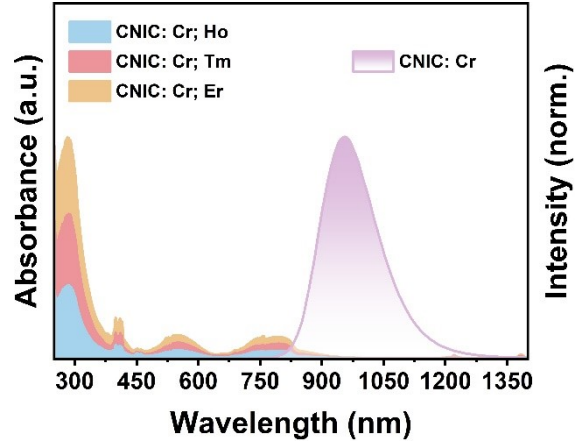


Figure S4. The absorption spectra of CNIC: Cr³⁺, Ln³⁺ (Ln³⁺ = Ho³⁺, Tm³⁺, Er³⁺) and the emission of CNIC: Cr³⁺ under 290 nm excitation.

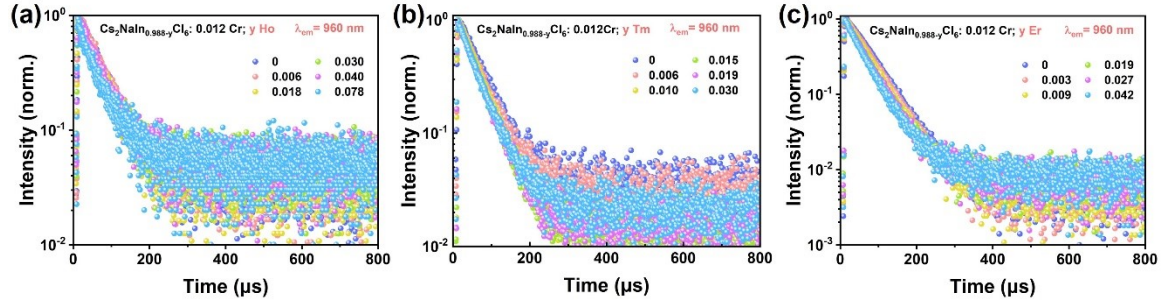


Figure S5. NIR PL decay curves at 960 nm of Cs₂NaIn_{0.988-y}Cl₆: 0.012 Cr³⁺, y Ln³⁺ (Ln³⁺ = Ho³⁺, Tm³⁺, Er³⁺) under 290 nm excitation.

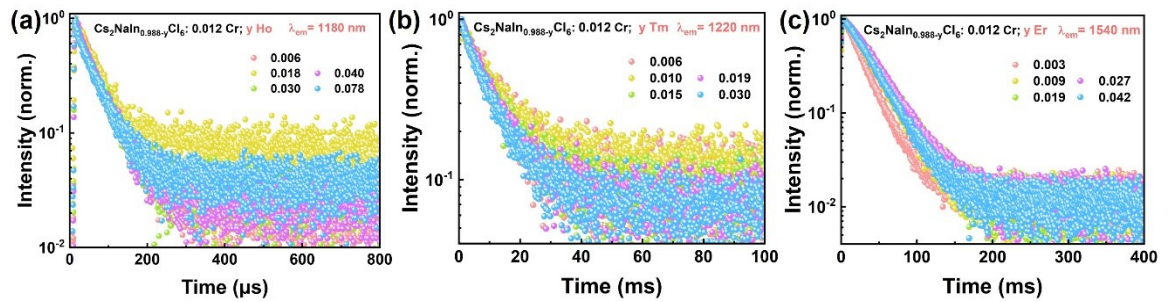


Figure S6. NIR PL decay curves of Cs₂NaIn_{0.988-y}Cl₆: 0.012 Cr³⁺, y Ho³⁺ (λ_{em} = 1180 nm), Cs₂NaIn_{0.988-y}Cl₆: 0.012 Cr³⁺, y Tm³⁺ (λ_{em} = 1220 nm), Cs₂NaIn_{0.988-y}Cl₆: 0.012 Cr³⁺, y Er³⁺ (λ_{em} = 1540 nm) under 290 nm excitation.

Table S6. Decay lifetimes (τ) and energy-transfer efficiency (η_{ET}) of Cs₂NaIn_{0.988-y}Cl₆: 0.012 Cr³⁺, y Ho³⁺ under 290 nm excitation.

Samples	τ [μ s] (λ_{em} =960 nm)	τ [μ s] (λ_{em} =1180 nm)	η_{ET} [%]
0%	56.01		
20%	53.58	55.71	4.3%
40%	50.15	53.05	10.5%
80%	49.06	52.33	12.4%
100%	48.55	51.45	13.3%
120%	47.8	48.89	14.7%

Table S7. Decay lifetimes (τ) and energy-transfer efficiency (η_{ET}) of Cs₂NaIn_{0.988-y}Cl₆: 0.012 Cr³⁺, y Tm³⁺ under 290 nm excitation.

Samples	τ [μ s] (λ_{em} =960 nm)	τ [μ s] (λ_{em} =1220 nm)	η_{ET} [%]
0%	56.01	-	-
20%	51.62	9.74	7.8%
40%	49.66	9.99	11.3%
80%	47.51	9.24	15.1%
100%	47.20	9.09	15.7%
120%	46.77	8.34	16.5%

Table S8. Decay lifetimes (τ) and energy-transfer efficiency (η_{ET}) of $\text{Cs}_2\text{NaIn}_{0.988-y}\text{Cl}_6$: 0.012 Cr^{3+} , y Er^{3+} under 290 nm excitation.

Samples	τ [μs] ($\lambda_{\text{em}}=960$ nm)	τ [μs] ($\lambda_{\text{em}}=1540$ nm)	η_{ET} [%]
0%	56.01	-	-
20%	54.77	23.46	2.2%
40%	51.61	29.89	7.9%
80%	49.10	32.15	12.3%
100%	48.79	31.64	12.9%
120%	46.62	30.13	16.8%

Note: the equation of ET efficiency (η_{ET}) is as follow:

$$\eta_{ET} = 1 - \frac{\tau_x}{\tau_0} \quad (1)$$

where τ_0 and τ_x are the decay lifetimes monitored at 960 nm in the Cr^{3+} singly-doped DPs and the $\text{Cr}^{3+}/\text{Ln}^{3+}$ co-doped DPs, respectively.

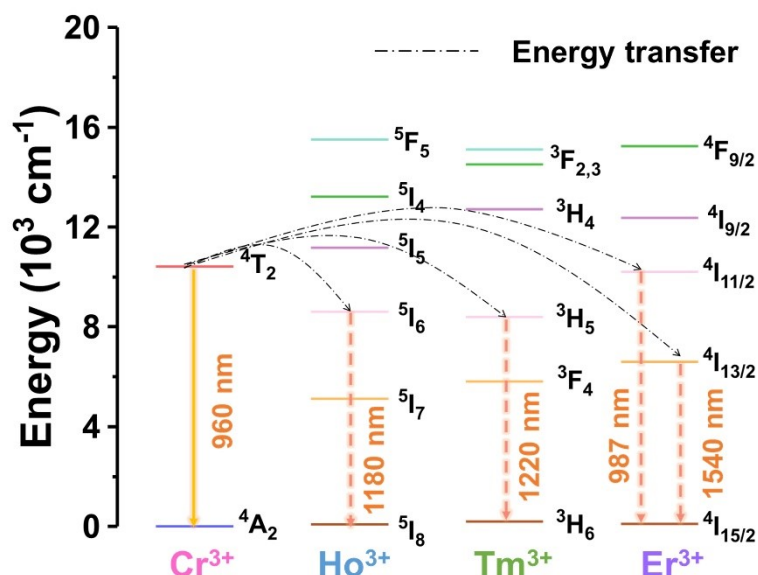


Figure S7. Schematic diagram of the photophysical process of energy transfer from Cr^{3+} to Ln^{3+} in CNIC: $\text{Cr}^{3+}/\text{Ln}^{3+}$ ($\text{Ln}^{3+} = \text{Ho}^{3+}, \text{Tm}^{3+}, \text{Er}^{3+}$).

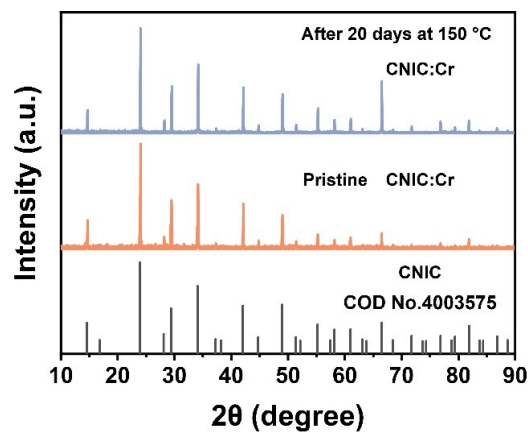


Figure S8. XRD patterns of CNIC: Cr³⁺ after heating at 150 °C for 20 days.

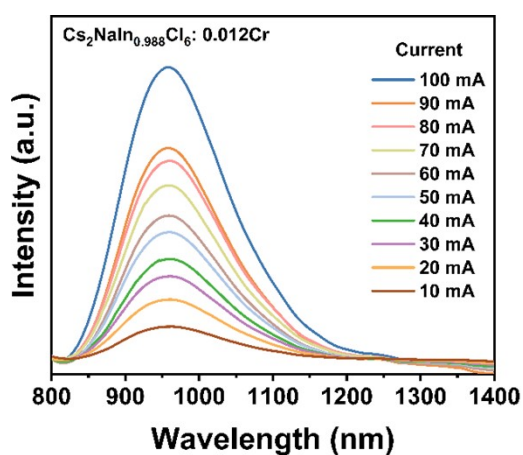


Figure S9. EL spectra of the fabricated NIR pc-LED measured under a voltage of 24V and varying driving currents from 10 mA to 100 mA.

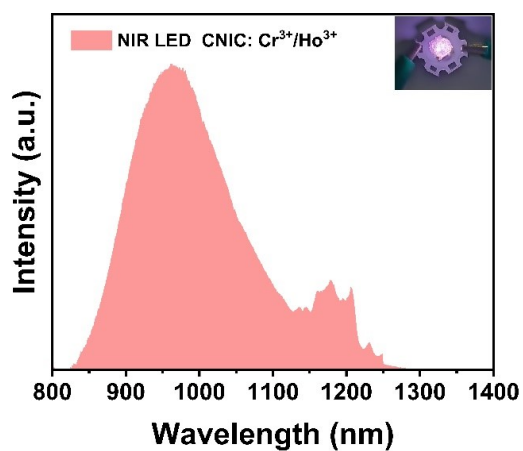


Figure S10. Emission spectrum of the LED fabricated by CNIC: Cr³⁺/Ho³⁺ DPs under 310 nm chip excitation.

References

1. Y. Li, J. Zuo, T. Zheng, S. Yang, P. He, J. Zhang, S. Liu, F. Yang, J. Peng and X. Ye, *Adv. Opt. Mater.*, 2024, **12**, 2400171.
2. D. Chen, C. Wu, H. Li, L. Zhou, P. Chen, Q. Pang, X. Zhang and J. Z. Zhang, *J. Mater. Chem. C*, 2023, **11**, 12649-12657.
3. C. Zhao, Y. Gao, J. Wang and J. Qiu, *Laser Photonics Rev.*, 2023, **18**, 2300952.
4. W. Gan, L. Cao, S. Gu, H. Lian, Z. Xia and J. Wang, *Chem. Mater.*, 2023, **35**, 5291-5299.
5. Y. Wang, C. Deng, J. Zhang, Y. Liu, J. Chen, Y. Jin, Z. Fan and D. Fan, *Ceram. Int.*, 2025, **51**, 15466-15474.
6. L. Zi, W. Xu, Z. Song, R. Sun, S. Liu, T. Xie, J. Zhu, S. Lu and H. Song, *J. Mater. Chem. C*, 2023, **11**, 2695-2702.
7. B. Yan, J. Wen, Y. Zhou, Q. Sha, X. Huang, Z. Dong, C. Wang, Q. Wang, L. Ning and C.-K. Duan, *Ceram. Int.*, 2025, **51**, 11018-11025.
8. M. Han, Z. Gao, Y. Zhang, S. Chen, J. Li, Y. Shen, Y. Tian and D. Deng, *J. Alloys Compd.*, 2024, **995**, 174780.
9. B. Mondal, S. Mehta, A. Mondal and A. Nag, *J Phys Chem Lett*, 2025, **16**, 4937-4943.
10. W. Wang, S. Su, H. Xu, W. Yang, L. Zhao, W. Teng, X. Yang, X. Sun, H. Zhang, J. Cui, Y. Zhang, B. Teng and D. Zhong, *J. Lumin.*, 2025, **282**, 121237.
11. F. Huang, Y. Yan, G. Zhu, S. Zhao, H. Wang, D. Deng, S. Xu and H. Yu, *Appl. Mater. Today*, 2025, **44**, 102731.
12. C. Zhan, H. Zhu, S. Liang, W. Nie, Z. Wang and M. Hong, *Adv. Opt. Mater.*, 2024, **12**, 2400574.
13. B. Han, L. Shi, Y. Lv, W. Li, J. Hu, Y. Liu, Y. Wang and X. Chen, *Ceram. Int.*, 2025, DOI: 10.1016/j.ceramint.2025.05.331.
14. P. Wang, A. Chen, B. Lou, Q. Chen, D. Zhang, C. K. Duan and C. G. Ma, *Laser Photonics Rev.*, 2025, e00030.
15. F. Zhu, Y. Gao and J. Qiu, *Adv. Opt. Mater.*, 2025, 2500549.
16. C. Gong, Z. Wang, J. Wang, X. Xue and X. Wang, *Sci. China Mater.*, 2025, **68**, 1074-1081.