Vertical Nb₂O₅ Micro-Petal Array Photoelectrochemical Deep-UV

Photodetectors Towards Underwater Weak-Light Photodetection

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The preparation of the KNb₃O₈ precursor

The preparation of KNb₃O₈ was achieved via a hydrothermal method according to previous work. (*Journal of Alloys and Compounds*, 2015, 627, 117–122) A 3 M KOH solution (30 mL) was prepared, and 1 g of Nb₂O₅ powder was added and stirred thoroughly using a magnetic stirrer. The mixture was then transferred to a hydrothermal synthesis reactor and heated at 230 °C for 2 h. After cooling to room temperature, HCl was added dropwise to the solution in a fume hood until the pH reached 7 and a white precipitate formed. The precipitate was separated using a centrifuge and washed three times with deionized water. It was then dried in an oven at 60 °C overnight. Finally, the dried solid was ground into fine particles and further dried at 100 °C for 2 h to obtain KNb₃O₈ powder.

The preparation of simulated seawater

6.684 g of NaCl, 0.050 g of NaHCO₃, 0.873 g of Na₂SO₄, 0.181 g of KCl, 1.213 g of MgCl₂·H₂O, and 0.813 g of CaCl₂ were accurately weighed using an electronic balance. These were added to a beaker containing 250 mL of deionized water and stirred vigorously on a magnetic stirrer for 30 min. The solution was then sealed and stored in a dark place for later use.

The preparation of the Nb₂O₅ microcrystals (MCs)-based PEC PDs

2 mg of PVDF and 20 mg of Nb₂O₅ were dissolved in 10 mL of DMF to form a Nb₂O₅ solution with a concentration of 2 mg mL⁻¹ for drop-casting. A volume of 33 μ L of this solution was evenly dropped onto an FTO substrate with an area controlled at 1 cm², and the substrate was placed in a 60 °C oven until the solvent evaporated. This process was repeated six times, resulting in a total drop-cast volume of 200 μ L, to obtain Nb₂O₅ MCs on the FTO glass.



Figure S1. Cross-view SEM images of a) pure FTO glass, b) 1 h, c) 2 h, and d) 3 h- Nb_2O_5 samples.



Figure S2. Length distribution diagrams of a) 1 h, b) 2 h, and c) 3 h-Nb₂O₅ samples.



Figure S3. Top-view SEM image of pure FTO substrate.



Figure S4. Optical images of a) the three-electrode system and b) the working electrode.



Figure S5. SEM image of Nb₂O₅ MCs powder.



Figure S6. *J*-t curve of Nb₂O₅ MC PEC PDs under 254 nm irradiation with different power intensities.



Figure S7. a) EIS curves and b) CV curves of Nb₂O₅ MPAs and MCs.



Figure S8. SEM image of the long-term Nb₂O₅ MPA sample.

Table S1. The *P* for 254 nm and 365 nm wavelength of light.

$P_{\lambda} (\mu \mathrm{W} \mathrm{cm}^{-2})$	Ι	II	III	IV	V
254 nm	19.9	39.8	71.6	118	250
365 nm	4600				

Table S2. The $J_{\rm ph}$ of Nb₂O₅ MPA PEC SBUV PDs under 254 nm irradiation with different power intensities.

$J_{\rm ph}$ (µA cm ⁻²) Sample	Ι	II	III	IV	V
1 h	0.5673	0.7459	0.9097	1.2867	2.6909

2 h	2.9141	3.6789	5.1856	7.3747	13.7340
3 h	0.7413	0.9582	1.3541	2.1949	3.9415

Table S3 The *R* of Nb_2O_5 MPA PEC SBUV PDs under 254 nm irradiation with different power intensities.

$R (mA W^{-1})$					
Sample	Ι	II	III	IV	V
1 h	28.5075	18.7412	12.7053	10.9042	10.7636
2 h	146.4372	92.4347	72.4246	62.4975	54.9356
3 h	37.2513	24.0704	18.9106	18.6008	15.76

Table S4. Comparison table with other PEC PDs.

Materials	Wavelength	Power intensity	Deinstin metic	Bias	R	Els stus lata	t _r	$t_{\rm f}$	Ref.
	(nm)	(µW cm ⁻²)	Rejection ratio	(V)	(mA W ⁻¹)	Electrolyte	(ms)	(ms)	
Nb ₂ O ₅ MPAs	254	19.9	22306(254/365)	0	146.44	Seawater	80	90	Ours
α -Ga ₂ O ₃	254	79	126.4(254/365)	0	101.5	0.01 M Na ₂ SO ₄	1000	5000	1
β -Ga ₂ O ₃	213	318	N/A	0	1.51	Seawater	220	140	2
α -Ga ₂ O ₃ /CFP	254	100	N/A	0	12.9	$0.5 \text{ M} \text{ Na}_2 \text{SO}_4$	150	130	3
Ga ₂ O ₃ /Al ₂ O ₃	254	2000	34.8 (264/400)	0	12.45	0.1 M NaOH	100	100	4
Ga ₂ O ₃ /ZnO	266	127000	N/A	0	7.97	$0.5 \text{ M} \text{ Na}_2 \text{SO}_4$	150	1100	5
In_2O_3	254	39	1319(254/365)	0	86.15	1 M KOH	15	18	6
In_2O_3	254	40	1567(254/455)	0.4	172.36	$0.5 \text{ M} \text{ Na}_2 \text{SO}_4$	800	2200	7
Ga-In	254	79	262.45(254/455)	0	50.04	$0.01 \text{ M} \text{ Na}_2 \text{SO}_4$	450	380	8
ZnAl-LDH	254	40	1037(254/455)	0	29.25	0.01 M KOH	100	100	9
SnO_2	254	40	158.73(254/365)	0	68.08	1 M Na ₂ SO ₄	60	60	10
ZnS NPF	254	40	1343(254/455)	0	241.74	0.01 M Na ₂ SO ₄	15	15	11
AlGaN/GaN	254	100	N/A	0	20.9	Seawater	121	151	12
AlGaN: Ru	254	1500	N/A	0	48.8	0.01 M H ₂ SO ₄	83	19	13

Table S5. The D^* of Nb₂O₅ MPA PEC SBUV PDs under 254 nm irradiation with different power

intensities.					
D*(10 ¹⁰ Jones) Sample	Ι	Π	III	IV	V
1 h	2.8172	1.9799	1.3669	1.1603	1.1474
2 h	17.4527	10.5476	8.3517	6.7996	6.3896
3 h	3.9354	2.4549	1.9631	1.7501	1.6417

Table S6. The calculated I_{ph} , *R* and D^* values of 2 h-Nb₂O₅ PEC PDs illuminated by 365 nm at level I in simulated seawater and Nb₂O₅ MCs PEC SBUV PDs for 254 nm light at level I in simulated seawater, respectively.

Parameter Irradiation	$J_{\rm ph}(\mu{ m A~cm^{-2}})$	<i>R</i> (mA W ⁻¹)	$D^* (10^{10} \text{ Jones})$
2 h (365 nm, I)	0.0302	0.006565	0.027115
MCs (254 nm, I)	0.1165	5.855	0.3517

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Sample	1 h	2 h	3 h	MCs
R _s	10.27	8.97	9.819	14.48
R _{ct}	13.41	8.1	12.85	29.17

Ref.

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