

## Supplementary Materials

### **Synergy of $\text{sp}^2$ -Hybridized Carbon Doping and Photogenerated Surface Oxygen Vacancies for Enhancing Photocatalytic Performance of $\text{BiOCl}$ and Solvent Effects**

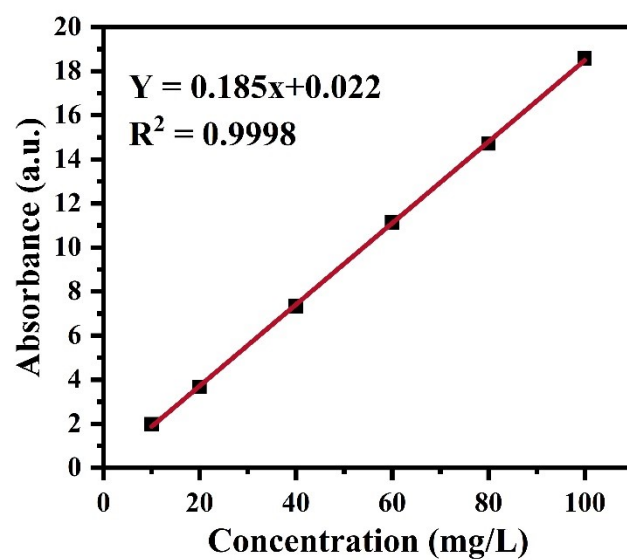
Jingyao Li,<sup>a, b</sup> Libo Wang,<sup>a, b</sup> Yiqian Li,<sup>a, b</sup> Miaomiao Tian,<sup>a, b</sup> Ya Wang,<sup>a, b</sup> Gang Liu,<sup>\*a, b</sup>, Tao Wang,<sup>a</sup> Qingxin Wang,<sup>a</sup> Zhixin Zhang,<sup>a</sup> and Wenhong Su<sup>a</sup>

<sup>a</sup> Institute of Chemical and Industrial Bioengineering, Jilin Engineering Normal University, Changchun 130052, China

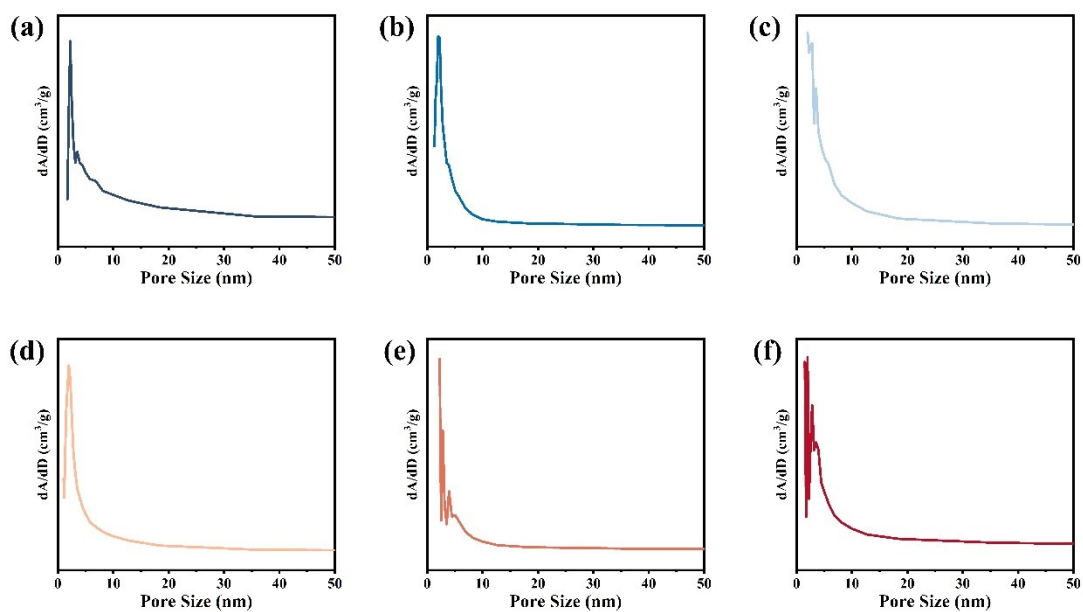
<sup>b</sup> Jilin Science and Technology Innovation Center of Green Synthesis and New Materials Research and Development

**Corresponding Author:**

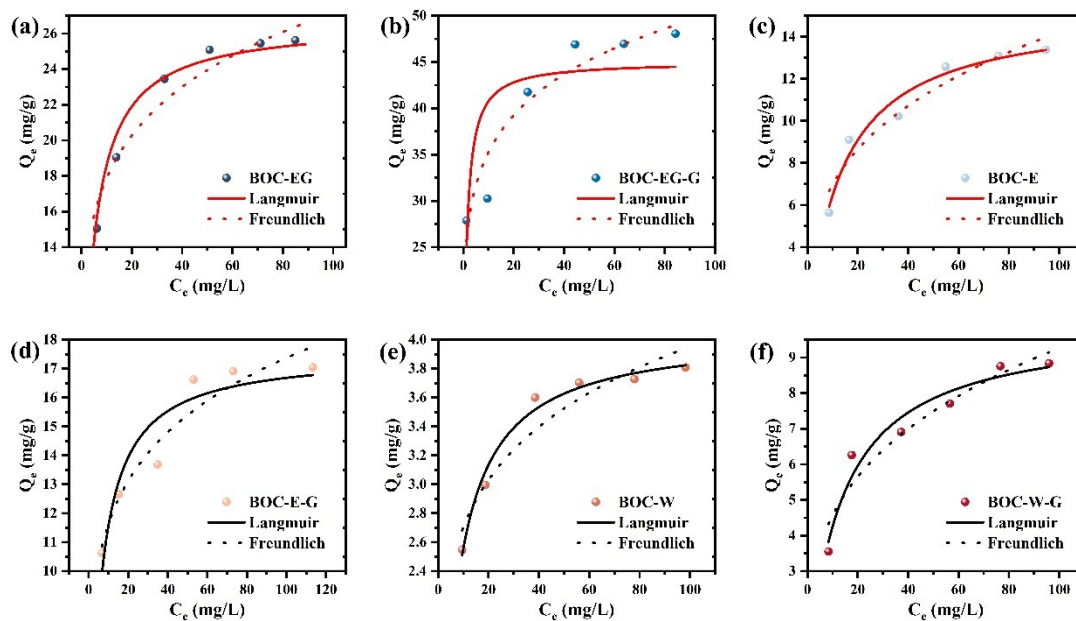
\*E-mail: 2297474857@qq.com



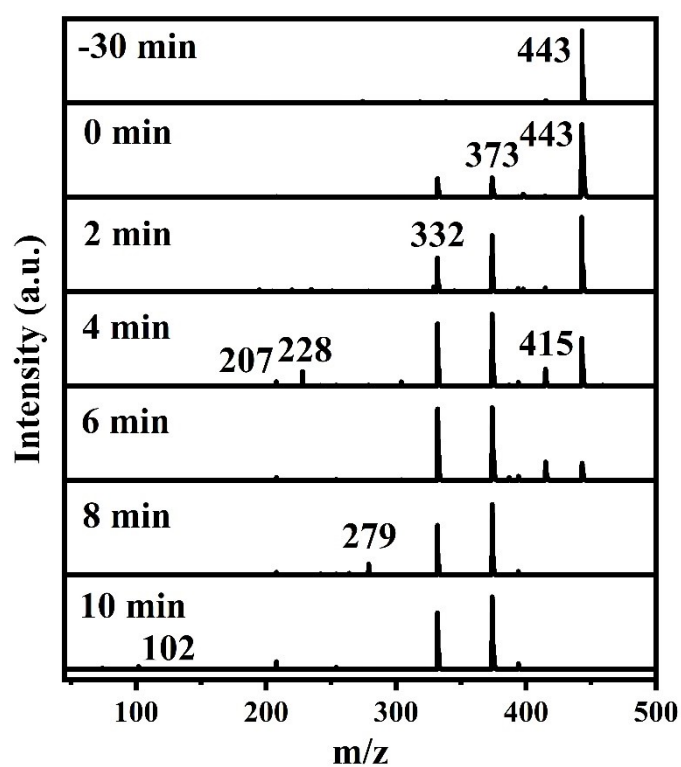
**Figure S1.** Standard curves of RhB.



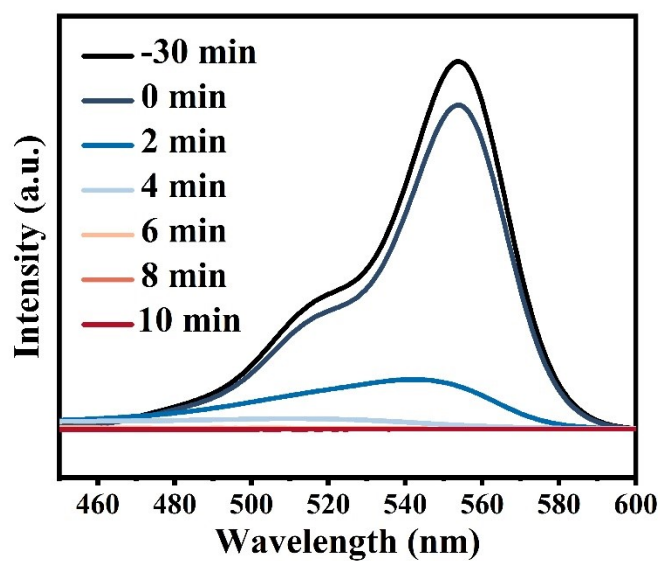
**Figure S2.** Pore size distribution curves of (a) BOC-EG, (b) BOC-EG-G, (c) BOC-E, (d) BOC-E-G, (e) BOC-W and (f) BOC-W-G.



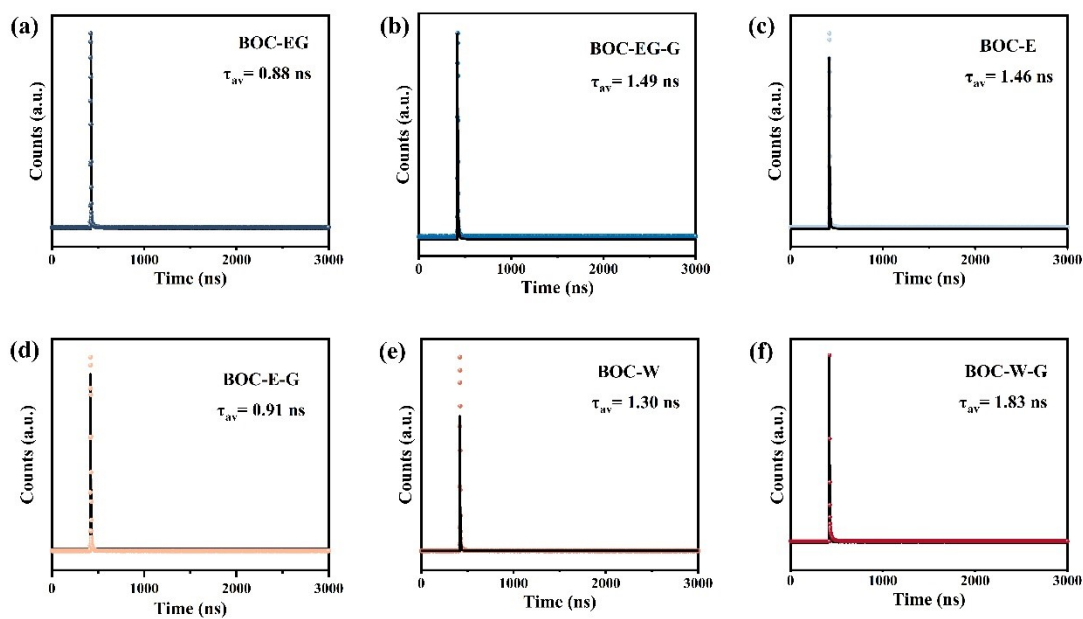
**Figure S3.** Langmuir and Freundlich fitting of adsorption isotherm of (a) BOC-EG, (b) BOC-EG-G, (c) BOC-E, (d) BOC-E-G, (e) BOC-W and (f) BOC-W-G.



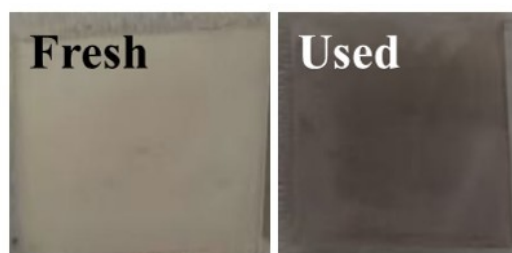
**Figure S4.** Mass spectra of RhB at different degradation time.



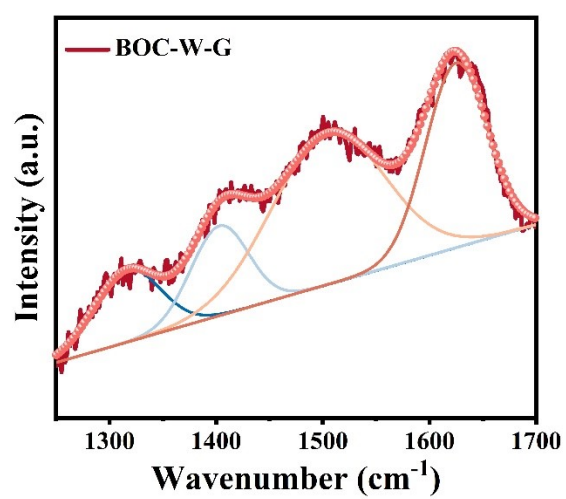
**Figure S5.** UV-vis spectra of RhB solution under different irradiation time using BOC-W-G.



**Figure S6.** Decay curves of (a) BOC-EG, (b) BOC-EG-G, (c) BOC-E, (d) BOC-E-G, (e) BOC-W and (f) BOC-W-G.



**Figure S7.** Digital photo of fresh and used BOC-W-G.



**Figure S8.** Raman spectra of BOC-W-G.

**Table S1.** Langmuir and Freundlich adsorption isotherm parameters for RhB adsorption.

| Isotherm models | Parameters              | BOC-EG | BOC-EG-G | BOC-E | BOC-E-G | BOC-W | BOC-W-G |
|-----------------|-------------------------|--------|----------|-------|---------|-------|---------|
| Langmuir        | $R^2$                   | 0.961  | 0.633    | 0.967 | 0.89    | 0.985 | 0.929   |
|                 | Q <sub>max</sub> (mg/g) | 26.63  | 45.04    | 15.3  | 17.54   | 4.05  | 9.95    |
|                 | K <sub>L</sub> (L/mg)   | 0.23   | 0.95     | 0.07  | 0.19    | 0.17  | 0.07    |
| Freundlich      | $R^2$                   | 0.955  | 0.91     | 0.935 | 0.929   | 0.921 | 0.958   |
|                 | K <sub>F</sub> (L/g)    | 11.74  | 21.71    | 3.43  | 7.76    | 1.85  | 2.25    |
|                 | n                       | 5.49   | 6.49     | 3.24  | 5.71    | 6.06  | 3.25    |

**Table S2.** Adsorption kinetics parameters for RhB adsorption.

| Kinetics models              | Parameters                          | Values |
|------------------------------|-------------------------------------|--------|
| Pseudo-first-order kinetics  | $R^2$                               | 0.979  |
|                              | Q <sub>e,cal</sub> (mg/g)           | 137.41 |
|                              | K <sub>1</sub> (min <sup>-1</sup> ) | 0.004  |
| Pseudo-second-order kinetics | $R^2$                               | 0.998  |
|                              | Q <sub>e,cal</sub> (mg/g)           | 2.85   |
|                              | K <sub>2</sub> (min <sup>-1</sup> ) | 0.235  |

**Table S3.** Comparison of RhB degradation by various photocatalysts

| Photocatalysts                          | Dosage<br>(g/L) | Concentration<br>(mg/L) | Efficiency<br>(%) | Time<br>(min) | Reference |
|-----------------------------------------|-----------------|-------------------------|-------------------|---------------|-----------|
| BOC-W-G                                 | 0.66            | 10                      | 100               | 6             | This work |
| BiOCl-CTAB                              | 0.5             | 10                      | 98.6              | 20            | [1]       |
| CQDs/BiOCl                              | 0.5             | 10                      | 95.76             | 60            | [2]       |
| Bi/BiOCl-1                              | 2               | 20                      | 98                | 120           | [3]       |
| CoFe <sub>2</sub> O <sub>4</sub> /BiOCl | 1               | 10                      | 92.9              | 120           | [4]       |

## References

- [1] T. Guo, X. Fan, X. Jiang, Y. Qi, J. Du, A. Zhang, H. Wang, Engineering shape of BiOCl nanosheets with improved visible-light response for superior photocatalytic degradation of Rhodamine B, *Journal of Alloys and Compounds* 948 (2023). <https://doi.org/10.1016/j.jallcom.2023.169586>.
- [2] Y. Zhao, H. Guo, J. Liu, Q. Xia, J. Liu, X. Liang, E. Liu, J. Fan, Effective photodegradation of rhodamine B and levofloxacin over CQDs modified BiOCl and BiOBr composite: Mechanism and toxicity assessment, *Journal of Colloid and Interface Science* 627 (2022) 180-193. <https://doi.org/10.1016/j.jcis.2022.07.046>.
- [3] Y. Yu, Z. Yang, Z. Shang, X. Wang, One-step solution combustion synthesis of Bi/BiOCl nanosheets: Reaction mechanism and photocatalytic RhB degradation, *Journal of Physics and Chemistry of Solids* 174 (2023). <https://doi.org/10.1016/j.jpcs.2022.111172>.
- [4] J. Liu, H. Wang, M.-J. Chang, M. Sun, Z.-W. He, C.-M. Zhang, W.-Y. Zhu, J.-L. Chen, H.-L. Du, L.-G. Peng, Z.-M. Luo, L. Zhang, Magnetically separatable CoFe<sub>2</sub>O<sub>4</sub>/BiOCl: Controllable synthesis, superior photocatalytic performance and mechanism towards decomposing RhB, NOR and Cr(VI) under visible light, *Colloids and Surfaces A: Physicochemical and Engineering Aspects* 648 (2022). <https://doi.org/10.1016/j.colsurfa.2022.129299>.
- [5] H. Ighnih, R. Haounati, R.E. Malekshah, H. Ouachtak, A. Jada, A.A. Addi, Photocatalytic degradation of RhB dye using hybrid nanocomposite BiOCl@Kaol

under sunlight irradiation, Journal of Water Process Engineering 54 (2023).

<https://doi.org/10.1016/j.jwpe.2023.103925>.