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## **Supporting Information**

## Carbon dot engineered trimetallic hydroxyl carbonates: A

## strategy for enhanced Redox-Diffusion Coupled charge storage

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**Figure S1:** UV-Vis spectrum of carbon dots representing characteristic absorption under UV range.



**Figure S2:** HR-TEM images of carbon dots synthesized from *Trachyspermum ammi* (Ajwain) leaves (a) at magnification of 5nm and (b) at magnification of 10nm.



Figure S3: XPS full scan survey of CDMHC.



Figure S4. Comparative electrochemical performance of different compositions of CDMHC synthesized (a) CV at scan rate of 10 mV s<sup>-1</sup>, (b) GCD at current density of 5 A g<sup>-1</sup> and (c) Nyquist plot at 0 V with lowest  $R_s$  value shown by 5CDMHC.



**Figure S5.** Electrochemical performance of commercial activated carbon used as anodic electrode material (a) CV curves at different scan rates of 10, 25, 50, 100 and 200 mV s<sup>-1</sup>; (b) GCD at different current density of 1, 2, 4, 6, 8 and 10 A g<sup>-1</sup>.

Anode mass optimization by charge balance equation for device fabrication:

$$\frac{m_+}{m_-} = \frac{C_-\Delta V_-}{C_+\Delta V_-} \tag{S1}$$

where C is the specific capacitance,  $\Delta V$  is the potential window, and  $m_+$  and  $m_-$  are the active masses of positive and negative electrodes, respectively.



**Figure S6:** (a) CV curves at 50 mV s<sup>-1</sup> for anode (AC) and cathode (CDMHC), (b) CV curves at varied potential window for AC//CDMHC hybrid device.