

Supporting Information

Triarylamine-BODIPYs exhibiting record hole mobility: synthesis, photophysical, electrochemical, spectroelectrochemical, and charge carrier mobility studies

Amiy Krishna,^[a] Tarun,^[b] Upendra Kumar Pandey,^{*,[b]} and Rajneesh Misra^{*,[a]}

[a] Department of Chemistry, Indian Institute of Technology Indore

Indore 453552, India. E-mail: rajneeshmisra@iiti.ac.in

[b] Organic & flexible electronics laboratory, Department of Electrical Engineering, School of Engineering, Shiv Nadar Institution of Eminence, Gautam Buddha Nagar, Uttar Pradesh 201314, India. E-mail: upendra.pandey@snu.edu.in

Synthesis of intermediate BTI₂.....	S2
Solvatochromic studies.....	S3
Normalized absorption spectra in DCM and thin film.....	S4
Differential pulse voltammograms.....	S5
Spectroelectrochemical studies.....	S6–S7
Charge mobility studies and the corresponding data.....	S8–S12
Low-angle XRD patterns of spin-coated thin films of BTA1–BTA4.....	S13
Dielectric Constant.....	S14
Molecular electrostatic potential surface plots.....	S14
DFT Calculation data.....	S15–S28
Copies of ¹H-NMR, ¹³C-NMR, ¹¹B-NMR, ¹⁹F-NMR and HRMS/MALDI Spectra.....	S29–S40
Reference.....	S41

Synthesis of intermediate BTI₂

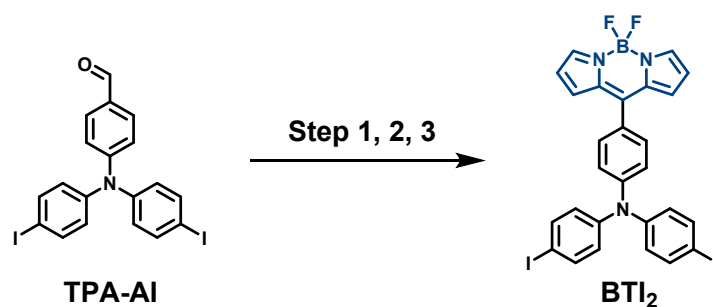


Fig. S1 Synthetic route of intermediate BTI₂.

(Note: Step1: pyrrole, TFA, r.t. Step2: DDQ, DCM, r.t. Step3: Et₃N, BF₃·OEt₂, DCM, r.t.).

Procedure: Pyrrole (2.5 mL, 36 mmol) and 4-N,N-bis(4-iodophenyl)-amino-benzaldehyde (0.756 g, 1.44 mmol) were added to a dry 25 mL round-bottomed flask and degassed with a stream of N₂ for 5 min. TFA (11.5 μ L, 0.15 mmol) was added, and the solution was stirred for 20 minutes under N₂ at room temperature. The excess pyrrole was removed on a rotary evaporator under vacuum, and the crude compound was passed through flash silica gel column chromatography with a mixture of ethyl acetate and petroleum ether as eluent (1:6 by volume). The resultant compound was dissolved in freshly distilled dichloromethane and oxidized with DDQ (1.2 equivalent) for 30 min at room temperature. The reaction mixture was then treated with Et₃N (40 equivalent) followed by BF₃·OEt₂ (50 equivalent), and stirred for 1 h at room temperature. The solvent was removed on a rotary evaporator, and the resultant crude compound was purified by silica gel column chromatography with hexane/ethyl acetate (3:1) and afforded pure compound BTI₂ in 28% yield; ¹H NMR (500 MHz, CDCl₃, 25 °C): δ 7.92 (s, 2H), 7.64 (d, J = 8.7 Hz, 4H), 7.47 (d, J = 8.5 Hz, 2H), 7.13 (d, J = 8.6 Hz, 2H), 7.02 (d, J = 3.9 Hz, 2H), 6.94 (d, J = 8.7 Hz, 4H), 6.56 (d, J = 2.4 Hz, 2H) ppm; ¹¹B NMR (160 MHz, CDCl₃, 25 °C) δ 0.46, 0.28, 0.10 ppm; ¹⁹F NMR (470 MHz, CDCl₃, 25 °C) δ -145.04, -145.10, -145.16, -145.23 ppm; MALDI-TOF-MS: m/z calculated for chemical formula C₂₇H₁₈BF₂I₂N₃ [M]⁺ 686.96; observed 686.9.

Solvatochromic studies

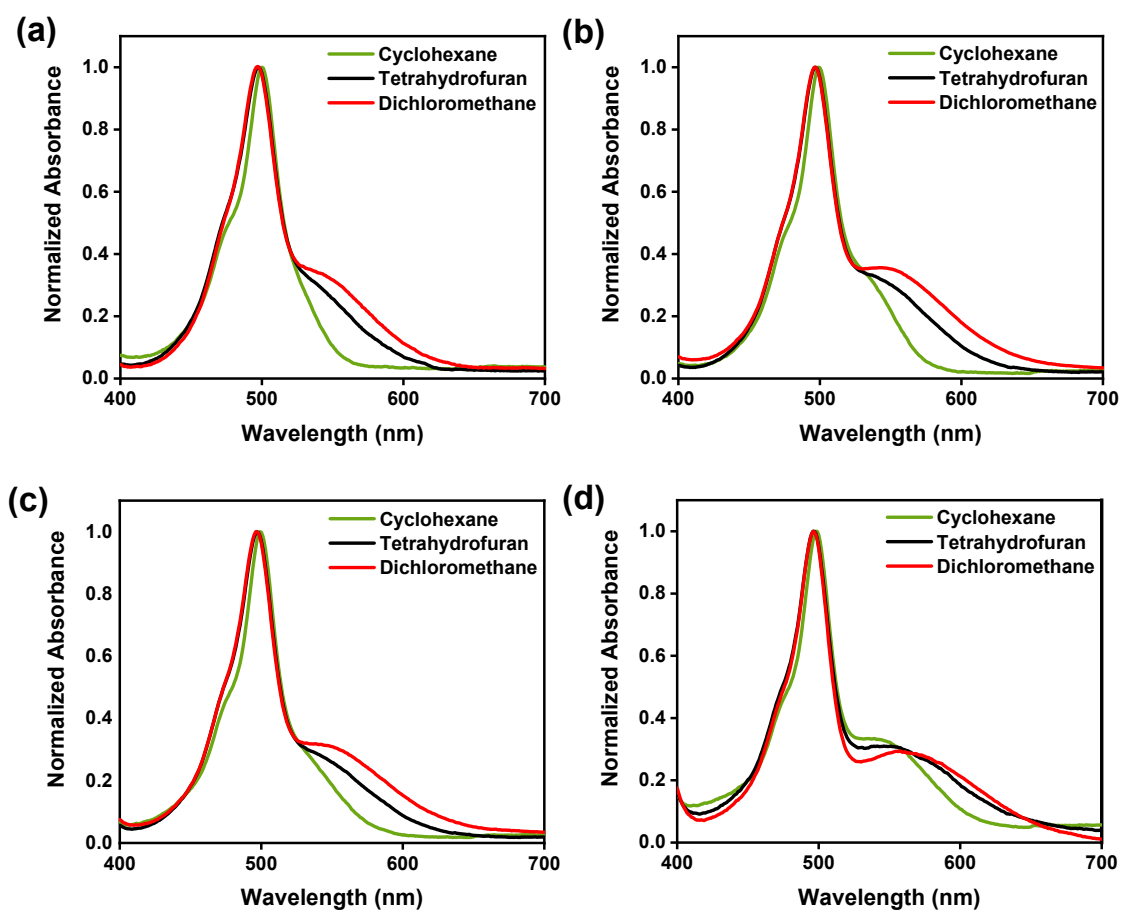


Fig. S2 Normalized electronic absorption spectra of the compounds **BTA1–BTA4** in solvents of different polarity.

Normalized absorption spectra in DCM and thin film

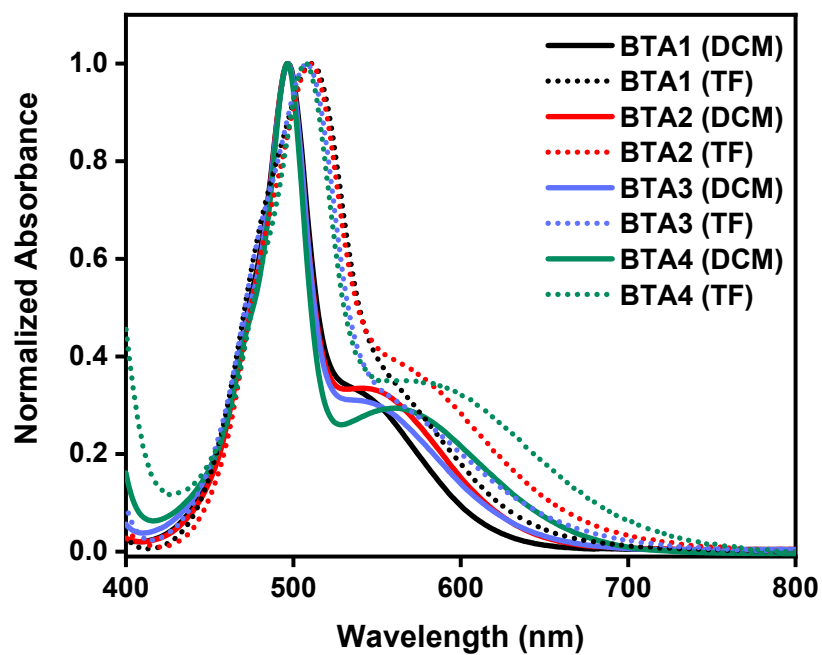


Fig. S3 Normalized electronic absorption spectra of the compounds **BTA1–BTA4** recorded in DCM and thin film.

Differential pulse voltammograms

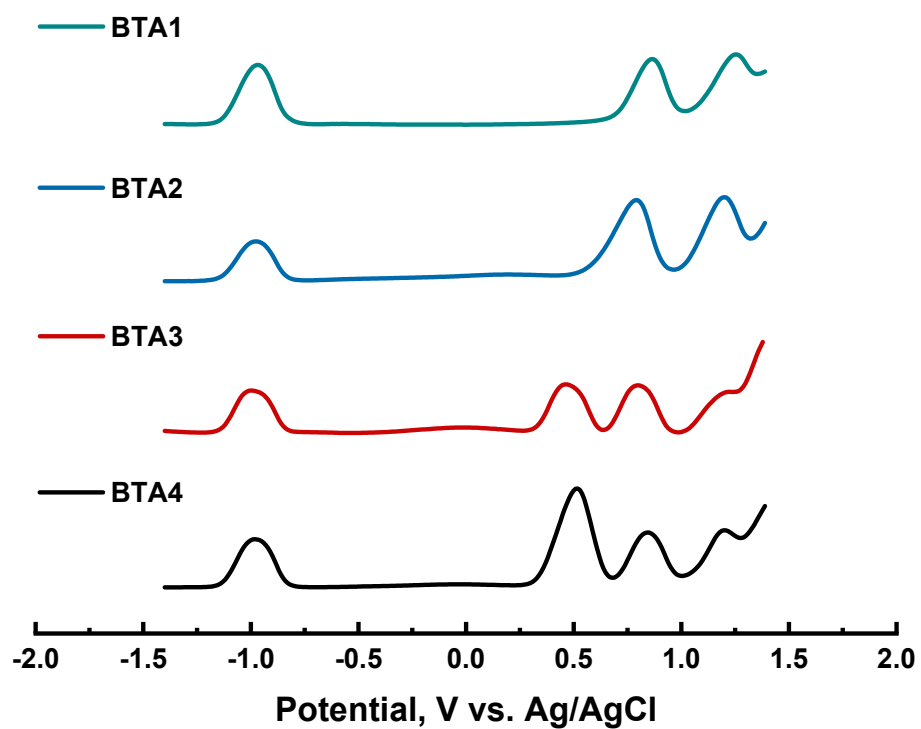


Fig. S4 DPVs of the compounds **BTA1–BTA4** in dichloromethane containing 0.1 M TBAClO₄ as the supporting electrolyte, recorded at a scan rate of 50 mV/s vs. Ag/AgCl.

Spectroelectrochemical studies

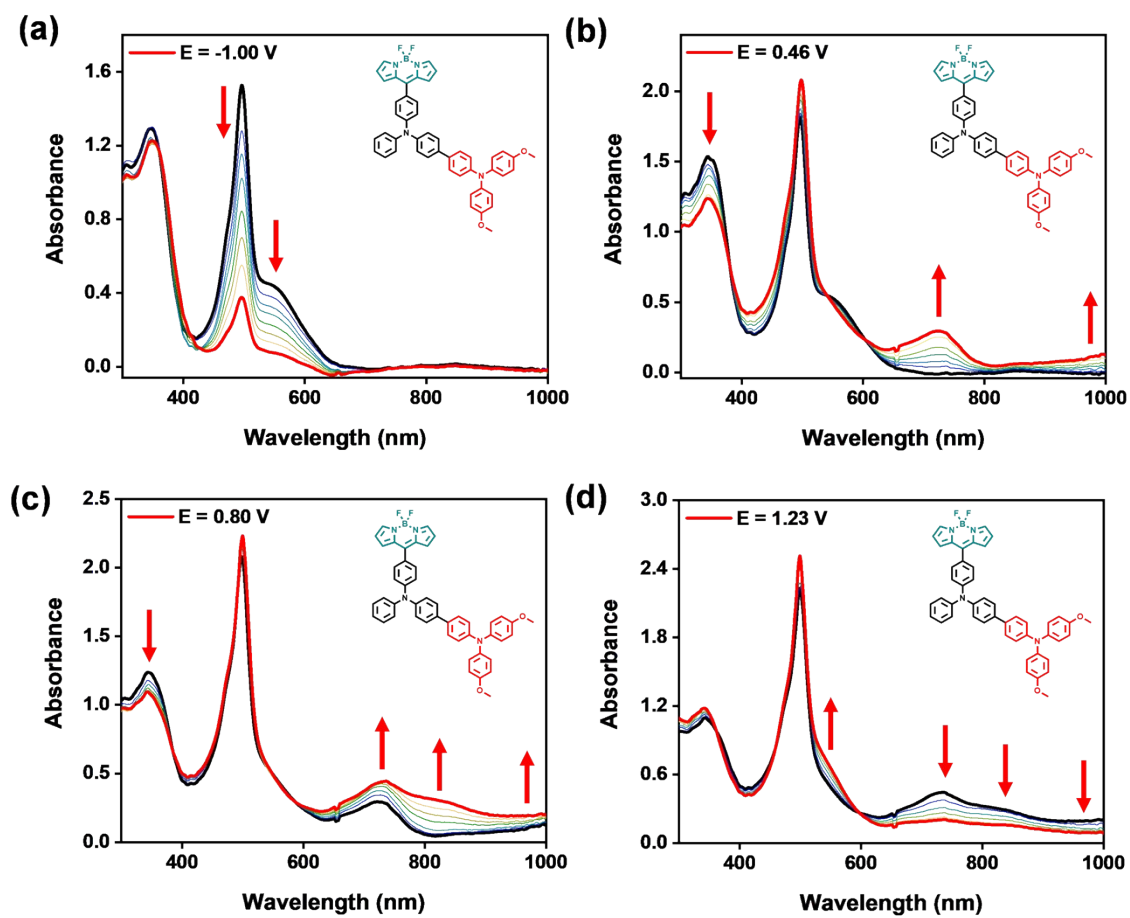


Fig. S5 Spectral changes observed for the compound **BTA3** at the indicated redox potentials.

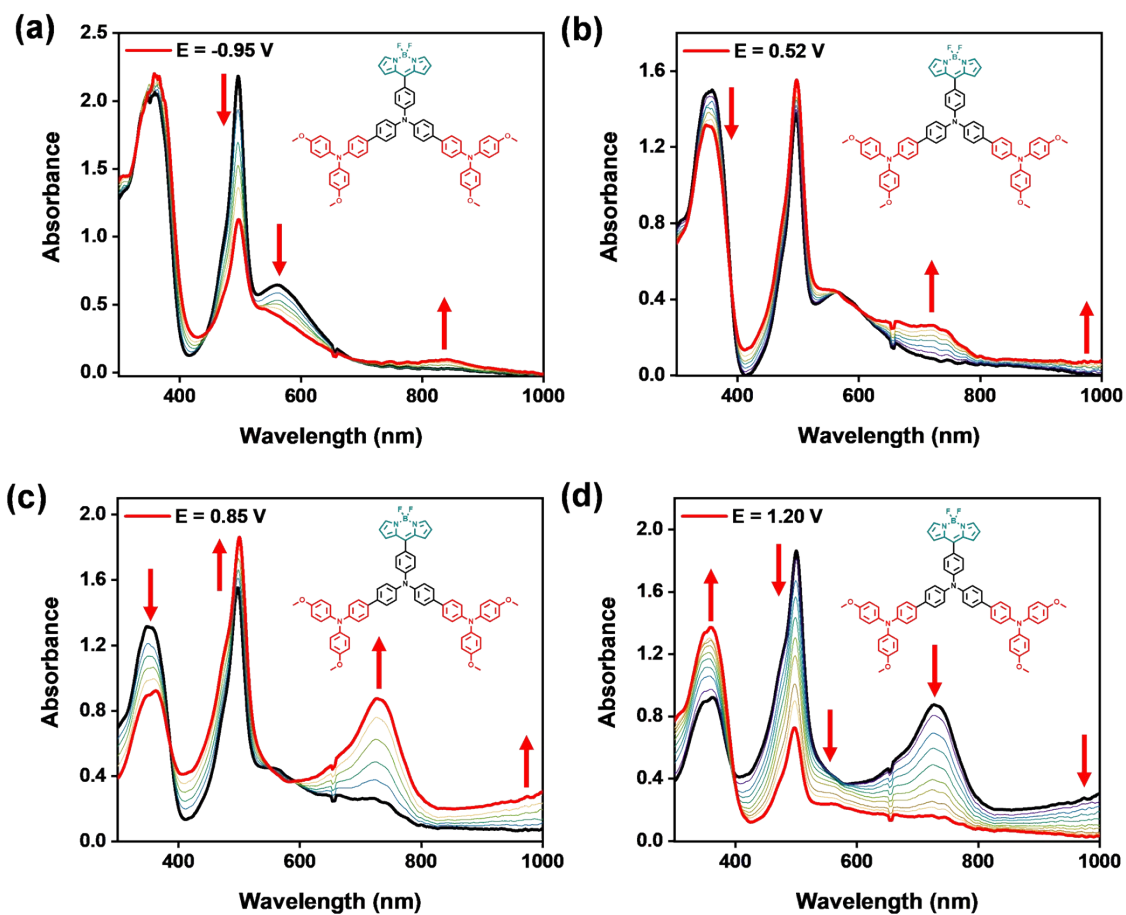


Fig. S6 Spectral changes observed for the compound **BTA4** at the indicated redox potentials.

Charge mobility studies and the corresponding data

ITO substrates ($15 \text{ } \Omega/\text{cm}$) were first patterned using 35M HCl solution. Subsequently, patterned ITO substrates were sequentially cleaned for 20 min each with 3% soap solution (Hellamanex III, Sigma Aldrich), then deionized water, acetone, and isopropyl alcohol in an ultrasonicator. Cleaned substrates were then dried under the gentle flow of ultra-pure nitrogen gas. The ITO substrates were then exposed to UV Ozone cleaner (BioBee, Bangalore, India) for 20 minutes at 70°C .

Hole-only devices were fabricated in ITO/2PACz ($\sim 5\text{-}10 \text{ nm}$)/ **BTA1–BTA4**($\sim 100 \text{ nm}$)/ MoO_3 ($\sim 10 \text{ nm}$)/Ag ($\sim 100 \text{ nm}$) architecture. For hole-only devices, $100 \text{ } \mu\text{l}$ of 2PACz (1 mg/ml in ethanol) was first spin-coated at 1500 rpm for 45 seconds on ITO substrates. The spin-coated substrates were then annealed at 80°C for 10 minutes under ambient conditions. Subsequently, **BTA1–BTA4** series compounds in chloroform (10 mg/ml) were spin-coated onto 2PACz coated substrate at 3000 rpm for 30 seconds to obtain $\sim 100 \text{ nm}$ thick films duly verified using dektek profilometer (Bruker). The samples were then placed in thermal evaporator (Hind High Vacuum, India) to deposit 7 nm of MoO_3 and 100 nm of Ag sequentially at $6 \times 10^{-6} \text{ mbar}$ base pressure to complete the hole-only devices.

For fabricating electron-only devices ITO/ ZnO ($\sim 30\text{-}35 \text{ nm}$)/ **BTA1–BTA4**($\sim 100 \text{ nm}$)/LiF ($\sim 1 \text{ nm}$)/Al ($\sim 100 \text{ nm}$), the ZnO sol-gel solution was first prepared by vigorously stirring 200 mg of zinc acetate dihydrate in 52 mg of ethanolamine and 2 ml of 2-methoxyethanol overnight at room temperature. Subsequently, the ZnO sol-gel solution was spin-coated onto cleaned UV ozone-treated ITO substrates at 4000 rpm for 60 seconds. The spin-coated substrates were then annealed at 180°C for 1 hour. The active layer solution (same as hole only device) was spin-coated at 1500 rpm for 30 seconds to obtain 100 nm thick films. The devices were then completed by evaporating 1 nm of LiF layer and subsequently 100 nm of aluminium under $6 \times 10^{-6} \text{ mbar}$ base pressure. The samples for both hole-only and electron-only devices

have an active area of 6.0 mm², calculated using the overlap area of both electrodes. All the devices were fabricated inside a glove box (GS Germany, O₂ < 100 PPM, H₂ < 0.2 PPM). The mobility data of **BTA1–BTA4** materials for the best five devices are given in Table S1.

Table S1 Best five Hole and Electron mobility devices data of **BTA1–BTA4** materials under dark conditions.

Material	BTA1 mobilities (cm ² V ⁻¹ s ⁻¹)	
	Hole (× 10 ⁻³)	Electron (× 10 ⁻³)
	5.20	1.30
	4.00	1.60
	6.80	1.70
	5.30	1.40
	6.60	2.00
Average	5.58	1.60
STD	1.14	0.27

Material	BTA2 mobilities (cm ² V ⁻¹ s ⁻¹)	
	Hole (× 10 ⁻²)	Electron (× 10 ⁻³)
	1.60	4.60
	2.70	1.30
	2.10	2.30
	1.50	4.10
	1.00	2.20
Average	1.78	2.90
STD	0.64	1.39

Material	BTA3 mobilities (cm ² V ⁻¹ s ⁻¹)	
	Hole (× 10 ⁻²)	Electron (× 10 ⁻³)
	1.90	1.65
	1.10	1.70
	2.20	2.00
	1.10	2.40
	1.60	1.20
Average	1.58	1.79
STD	0.49	0.44

Material	BTA4 mobilities (cm ² V ⁻¹ s ⁻¹)	
	Hole (× 10 ⁻²)	Electron (× 10 ⁻⁴)

	0.81	6.40
	0.81	4.10
	1.20	6.00
	1.30	4.10
	1.32	5.40
Average	1.09	5.20
STD	0.26	1.07

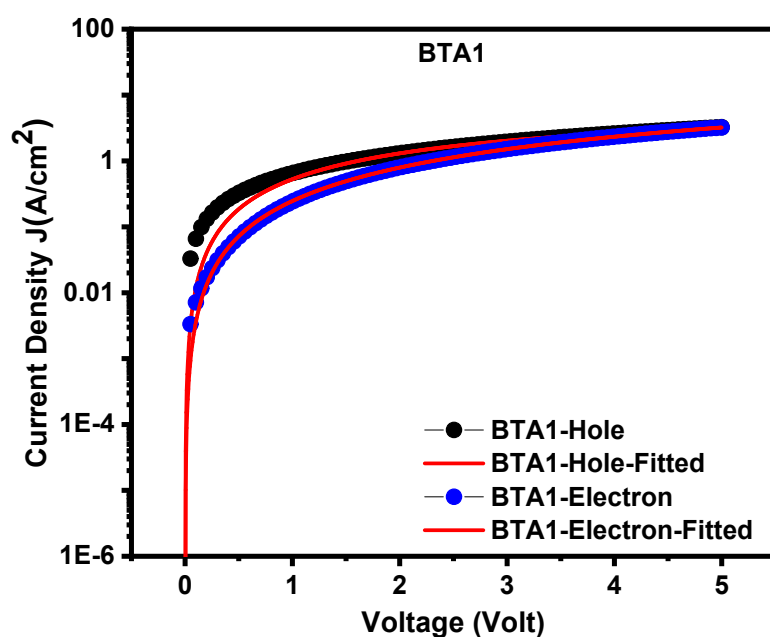


Fig. S7 Semi-logarithmic J - V characteristics of **BTA1** single carrier devices (either hole-only or electron-only) measured under dark conditions. The data were fitted using the modified Mott-Gurney law to extract charge carrier mobilities.

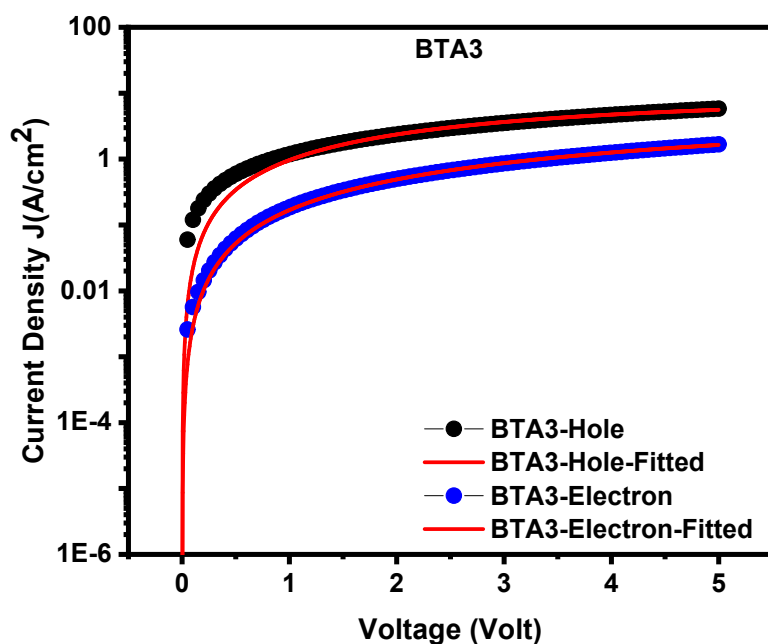


Fig. S8 Semi-logarithmic J - V characteristics of **BTA3** single carrier devices (either hole-only or electron-only) measured under dark conditions. The data were fitted using the modified Mott-Gurney law to extract charge carrier mobilities.

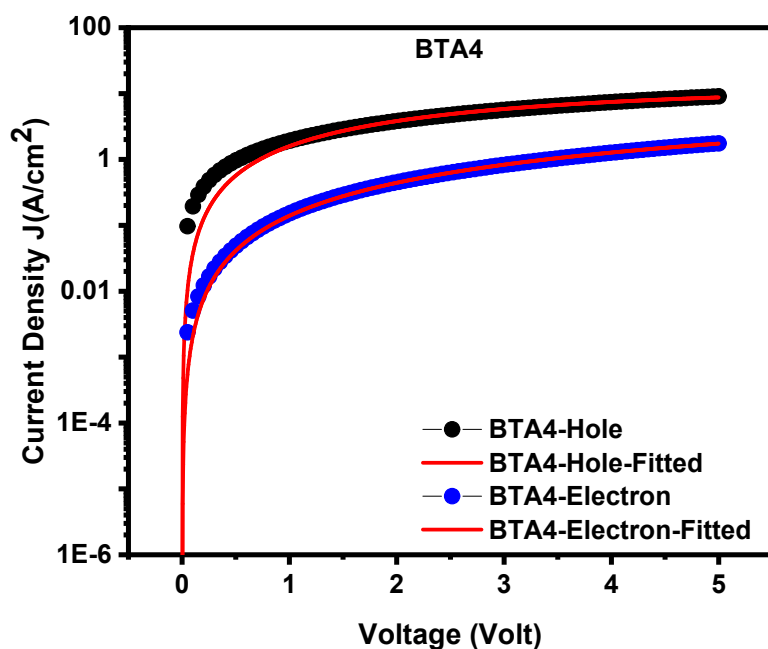


Fig. S9 Semi-logarithmic J - V characteristics of **BTA4** single carrier devices (either hole-only or electron-only) measured under dark conditions. The data were fitted using the modified Mott-Gurney law to extract charge carrier mobilities.

Table S2 Charge carrier mobilities of selected BODIPY-based conjugated small molecules evaluated using the SCLC method.

Entry	Compounds	Charge carrier mobility (cm ² V ⁻¹ s ⁻¹)		Ratio μ_e/μ_h	References
		Hole (μ_h)	Electron (μ_e)		
1	BTA1	5.58×10^{-3}	1.60×10^{-3}	0.29	This work
2	BTA2	1.78×10^{-2}	2.90×10^{-3}	0.16	This work
3	BTA3	1.58×10^{-2}	1.75×10^{-3}	0.11	This work
4	BTA4	1.09×10^{-2}	5.20×10^{-4}	0.05	This work
5	TM-01	7.61×10^{-5}	–	–	20
6	TM-02	4.32×10^{-4}	–	–	20
7	TM-03	7.74×10^{-5}	–	–	20
8	TM-04	1.69×10^{-4}	–	–	20
9	DIMER	1.16×10^{-3}	0.90×10^{-3}	0.78	21
10	Br-T-BO	4.71×10^{-4}	2.90×10^{-4}	0.62	21
11	H-T-BO	4.27×10^{-5}	1.00×10^{-5}	0.23	21
12	Py-BDP-1	6.54×10^{-5}	–	–	22
13	Py-BDP-2	6.04×10^{-5}	–	–	22
14	Py-BDP-3	6.41×10^{-5}	–	–	22
15	Py-BDP-4	5.76×10^{-5}	–	–	22
16	Py-BDP-5	4.86×10^{-5}	–	–	22
17	aza-BODIPY 1	4.60×10^{-4}			19
18	aza-BODIPY 2	3.20×10^{-4}	–	–	19
19	aza-BODIPY 3	1.20×10^{-4}	–	–	19
20	PTZ-PTZ-BDP	2.02×10^{-5}	–	–	23
21	CF3-BDP-Cz	2.45×10^{-4}	1.56×10^{-4}	0.64	24
22	CF3-BDP-TPA	2.18×10^{-4}	0.99×10^{-4}	0.45	24
23	D-A1-D	7.01×10^{-5}	–	–	25
24	D-A2-D	3.63×10^{-4}	–	–	25

Hole and electron mobilities (μ_h and μ_e) measured by the SCLC technique are listed for BODIPY-based CSMs from this work and the literature. Electron mobility is not reported for all compounds; where not available/applicable, a dash (–) is used.

Low-angle XRD patterns of spin-coated thin films of BTA1–BTA4

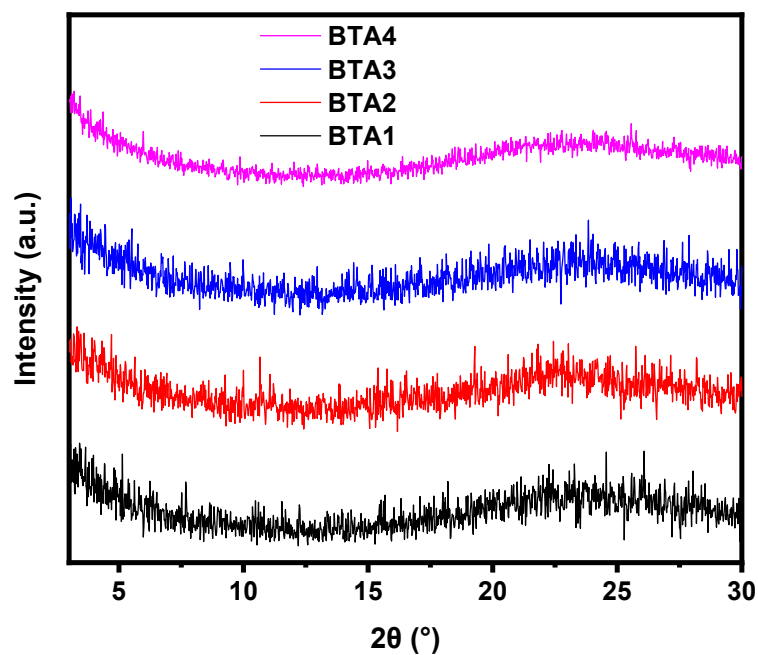


Fig. S10 Low-angle XRD patterns of spin-coated thin films of **BTA1–BTA4** measured on glass substrates.

Dielectric Constant

The relative dielectric constants (ϵ_r) of the **BTA1–BTA4** series materials were extracted from capacitance measurements using C-V measurements. The increasing dielectric constant from **BTA1** to **BTA4** reflects the enhanced donor-acceptor (D-A) interactions in the latter compounds, typically correlating with higher conductivity.¹

Table S3 Dielectric constant of **BTA1–BTA4** materials.

Material	BTA 1	BTA 2	BTA 3	BTA 4
Dielectric (ϵ_r)	3.27	3.5	3.8	4.1

Molecular electrostatic potential surface plots

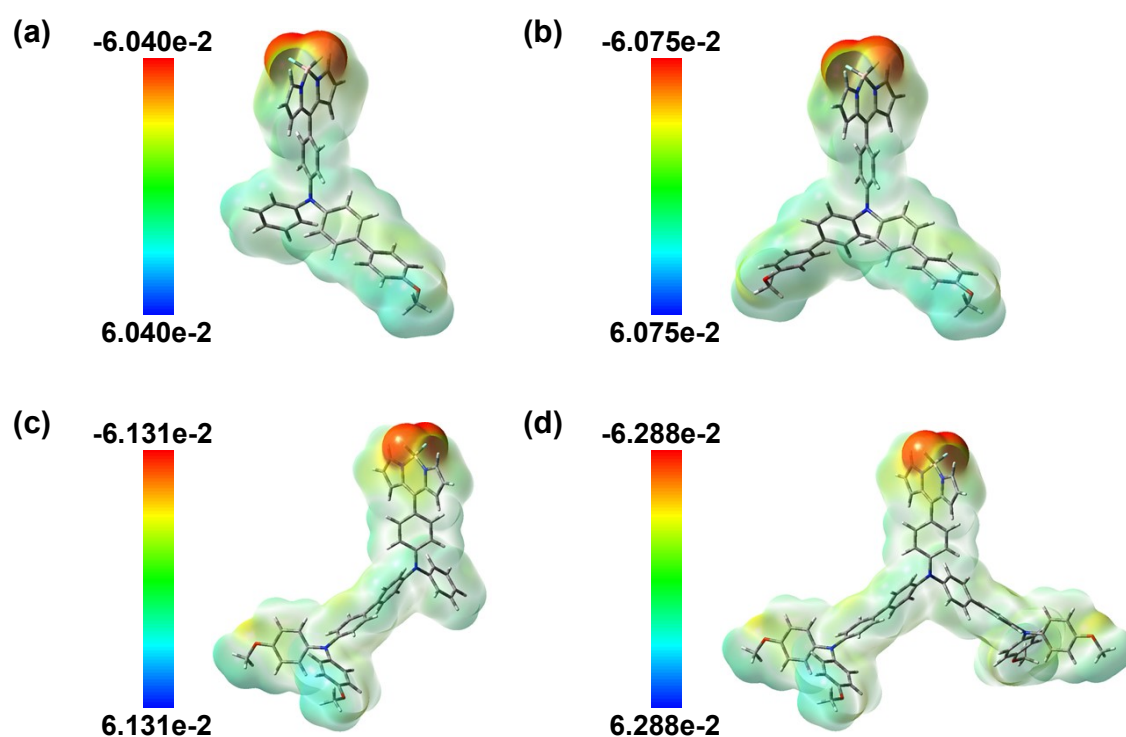


Fig. S11 Molecular electrostatic potential surface plots of the compounds (a) **BTA1**, (b) **BTA2**, (c) **BTA3**, and (d) **BTA4**.

DFT Calculation data

Calculation method: B3LYP/6-31G (d,p)

BTA1

Standard orientation:

Center	Atomic	Atomic	Coordinates (Angstroms)			
Number	Number	Type	X	Y	Z	

1	6	0	-0.332914	1.363213	-0.032388	
2	6	0	-1.446294	1.974893	0.575451	
3	6	0	-0.510886	0.098670	-0.626097	
4	6	0	-2.683803	1.346028	0.585105	
5	1	0	-1.335344	2.948317	1.039147	
6	6	0	-1.753296	-0.520690	-0.616936	
7	1	0	0.332370	-0.396181	-1.094005	
8	1	0	-3.529083	1.844491	1.047756	
9	1	0	-1.859690	-1.501527	-1.067898	
10	6	0	2.126793	1.235153	0.003008	
11	6	0	2.274809	0.181545	0.916974	
12	6	0	3.195625	1.537632	-0.852499	
13	6	0	3.453950	-0.554952	0.958663	
14	1	0	1.457627	-0.065055	1.586935	
15	6	0	4.379370	0.810315	-0.786444	
16	1	0	3.099388	2.356001	-1.558289	
17	1	0	3.531046	-1.384131	1.655198	
18	1	0	5.202559	1.085887	-1.438522	
19	6	0	1.015638	3.421594	-0.110135	
20	6	0	0.246083	4.143880	-1.034604	
21	6	0	1.892369	4.111693	0.739609	
22	6	0	0.344899	5.532605	-1.094716	

23	1	0	-0.424813	3.612596	-1.701614
24	6	0	1.998077	5.498973	0.659702
25	1	0	2.487275	3.555420	1.456305
26	1	0	-0.257109	6.078761	-1.815048
27	1	0	2.681550	6.020212	1.323575
28	7	0	0.923630	1.999171	-0.044644
29	6	0	-2.869969	0.086193	-0.013330
30	6	0	-4.191864	-0.571534	-0.004809
31	6	0	-4.908289	-0.698507	1.202895
32	6	0	-4.738735	-1.088192	-1.197197
33	7	0	-6.169612	-1.288807	1.234264
34	6	0	-4.537135	-0.397992	2.540832
35	7	0	-5.976067	-1.729408	-1.201890
36	6	0	-4.253725	-1.073005	-2.532325
37	6	0	-6.569286	-1.348597	2.513611
38	5	0	-6.979715	-1.808098	-0.000541
39	6	0	-5.585899	-0.799577	3.358787
40	1	0	-3.597015	0.036605	2.846145
41	1	0	-3.327061	-0.624283	-2.857229
42	6	0	-5.195748	-1.723233	-3.319882
43	1	0	-7.524740	-1.785685	2.768870
44	9	0	-8.061672	-0.980229	-0.245128
45	9	0	-7.378589	-3.114443	0.211289
46	1	0	-5.644445	-0.728893	4.435424
47	1	0	-5.152000	-1.901527	-4.384713
48	6	0	-6.244209	-2.105683	-2.461570
49	1	0	-7.169690	-2.613408	-2.695865
50	6	0	1.222937	6.217608	-0.252478
51	6	0	4.537911	-0.257409	0.114216
52	6	0	5.797871	-1.036211	0.173668

53	6	0	6.537567	-1.316956	-0.983147
54	6	0	6.299582	-1.523102	1.396597
55	6	0	7.726592	-2.046681	-0.939576
56	1	0	6.167961	-0.982462	-1.947950
57	6	0	7.478506	-2.250846	1.457684
58	1	0	5.768595	-1.303886	2.318061
59	6	0	8.204293	-2.519587	0.287771
60	1	0	8.259125	-2.247733	-1.861420
61	1	0	7.866836	-2.614915	2.403125
62	8	0	9.349761	-3.242138	0.452223
63	6	0	10.129138	-3.536203	-0.696837
64	1	0	10.988083	-4.106163	-0.340190
65	1	0	9.570229	-4.142121	-1.421346
66	1	0	10.482635	-2.622347	-1.191202
67	1	0	1.303127	7.298763	-0.307510

Electronic Energy: -1775.503573 Hartree

BTA2

Standard orientation:

Center	Atomic	Atomic	Coordinates (Angstroms)		
Number	Number	Type	X	Y	Z

1	6	0	-0.662820	-0.028401	-0.008625
2	6	0	-1.408050	1.005453	0.591124
3	6	0	-1.365631	-1.092346	-0.607232
4	6	0	-2.795660	0.971527	0.591848
5	1	0	-0.891676	1.837143	1.056175
6	6	0	-2.753530	-1.115362	-0.607119
7	1	0	-0.815832	-1.903213	-1.070822

8	1	0	-3.345924	1.786985	1.049275
9	1	0	-3.269251	-1.955425	-1.059973
10	6	0	1.499316	-1.208033	0.032687
11	6	0	1.167382	-2.233384	0.930281
12	6	0	2.603421	-1.386378	-0.812913
13	6	0	1.909966	-3.408930	0.963321
14	1	0	0.317248	-2.109909	1.593081
15	6	0	3.353732	-2.556124	-0.755509
16	1	0	2.878917	-0.597137	-1.504561
17	1	0	1.613155	-4.198917	1.646317
18	1	0	4.222011	-2.655781	-1.399589
19	6	0	1.444239	1.244027	-0.047977
20	6	0	1.078715	2.246570	-0.957968
21	6	0	2.524397	1.479284	0.814225
22	6	0	1.765592	3.455669	-0.987587
23	1	0	0.245823	2.078614	-1.632845
24	6	0	3.219595	2.682821	0.760761
25	1	0	2.824634	0.708207	1.516047
26	1	0	1.443370	4.226739	-1.680560
27	1	0	4.071257	2.827894	1.418248
28	7	0	0.744761	0.001315	-0.009160
29	6	0	-3.503592	-0.086192	-0.008556
30	6	0	-4.979484	-0.111889	-0.003391
31	6	0	-5.682698	0.087820	1.202510
32	6	0	-5.694128	-0.352282	-1.194962
33	7	0	-7.075255	0.097774	1.231632
34	6	0	-5.220858	0.203326	2.540955
35	7	0	-7.086915	-0.402323	-1.200066
36	6	0	-5.248606	-0.558030	-2.527947
37	6	0	-7.464220	0.219800	2.509872

38	5	0	-8.027449	-0.024982	-0.004492
39	6	0	-6.341874	0.294547	3.356749
40	1	0	-4.185956	0.191160	2.848624
41	1	0	-4.218989	-0.551157	-2.852649
42	6	0	-6.377830	-0.750605	-3.314413
43	1	0	-8.515334	0.236832	2.762666
44	9	0	-8.638179	1.190512	-0.260624
45	9	0	-8.958372	-1.023325	0.213127
46	1	0	-6.366049	0.386289	4.433122
47	1	0	-6.413771	-0.940688	-4.377502
48	6	0	-7.489547	-0.639942	-2.457808
49	1	0	-8.543130	-0.705505	-2.692010
50	6	0	2.854996	3.703917	-0.134151
51	6	0	3.590136	4.990518	-0.178480
52	6	0	4.100353	5.580124	0.986163
53	6	0	3.802103	5.670091	-1.394003
54	6	0	4.793713	6.791080	0.957326
55	1	0	3.932285	5.100297	1.945740
56	6	0	4.487583	6.874730	-1.440487
57	1	0	3.445002	5.232180	-2.321263
58	6	0	4.990960	7.447113	-0.262943
59	1	0	5.160700	7.213898	1.884892
60	1	0	4.657405	7.389673	-2.380276
61	6	0	3.023517	-3.599853	0.126439
62	6	0	3.817331	-4.850870	0.175146
63	6	0	4.374212	-5.408139	-0.984050
64	6	0	4.040059	-5.528337	1.389941
65	6	0	5.122391	-6.585927	-0.950566
66	1	0	4.200646	-4.929953	-1.943482
67	6	0	4.779638	-6.700316	1.440957

68	1	0	3.647414	-5.113799	2.313563
69	6	0	5.328792	-7.240785	0.268791
70	1	0	5.524453	-6.984686	-1.874240
71	1	0	4.957110	-7.213319	2.380399
72	8	0	6.039774	-8.394866	0.423284
73	8	0	5.650647	8.631725	-0.413414
74	6	0	6.187866	9.253320	0.743803
75	1	0	6.669337	10.168871	0.397642
76	1	0	5.403359	9.510145	1.467055
77	1	0	6.934448	8.616978	1.235857
78	6	0	6.623514	-8.983620	-0.728579
79	1	0	7.139382	-9.879365	-0.380154
80	1	0	5.863447	-9.269708	-1.466762
81	1	0	7.349235	-8.311278	-1.203704

Electronic Energy: -2121.092608 Hartree

BTA3

Standard orientation:

Center	Atomic	Atomic	Coordinates (Angstroms)		
Number	Number	Type	X	Y	Z

1	6	0	-3.896326	1.553998	-0.669179
2	6	0	-5.122519	2.090018	-0.228723
3	6	0	-3.763938	0.152180	-0.715493
4	6	0	-6.167059	1.257178	0.147690
5	1	0	-5.252257	3.165163	-0.184501
6	6	0	-4.815648	-0.673020	-0.342089
7	1	0	-2.829258	-0.286813	-1.044497
8	1	0	-7.105511	1.696567	0.469018

9	1	0	-4.680375	-1.748968	-0.371539
10	6	0	-1.478911	1.980124	-0.871491
11	6	0	-1.045583	1.433801	0.345103
12	6	0	-0.549755	2.129286	-1.910401
13	6	0	0.276189	1.031822	0.506057
14	1	0	-1.752584	1.313439	1.159555
15	6	0	0.775176	1.746549	-1.731551
16	1	0	-0.868619	2.559832	-2.853996
17	1	0	0.577393	0.583045	1.447548
18	1	0	1.480025	1.901046	-2.542559
19	6	0	-3.084635	3.673633	-1.627880
20	6	0	-4.025965	3.819215	-2.658175
21	6	0	-2.373571	4.798469	-1.184931
22	6	0	-4.259290	5.072317	-3.221476
23	1	0	-4.568866	2.949176	-3.012425
24	6	0	-2.600683	6.044309	-1.766514
25	1	0	-1.644969	4.687652	-0.388827
26	1	0	-4.991044	5.170452	-4.018038
27	1	0	-2.042701	6.906724	-1.413702
28	7	0	-2.833385	2.394175	-1.048482
29	6	0	-6.042718	-0.143784	0.098793
30	6	0	-7.158646	-1.022307	0.500351
31	6	0	-7.816899	-0.809852	1.729344
32	6	0	-7.565074	-2.085298	-0.332426
33	7	0	-8.891105	-1.606289	2.119431
34	6	0	-7.532591	0.085862	2.794591
35	7	0	-8.599467	-2.939877	0.044189
36	6	0	-7.112897	-2.491942	-1.616045
37	6	0	-9.263871	-1.227668	3.351616
38	5	0	-9.529133	-2.774076	1.294871

39	6	0	-8.448937	-0.173595	3.806303
40	1	0	-6.730556	0.808572	2.806677
41	1	0	-6.334801	-2.008992	-2.187781
42	6	0	-7.867050	-3.598503	-1.986767
43	1	0	-10.079430	-1.728273	3.854932
44	9	0	-10.805030	-2.430184	0.883194
45	9	0	-9.536673	-3.933014	2.048881
46	1	0	-8.524356	0.315952	4.766641
47	1	0	-7.790678	-4.169790	-2.900767
48	6	0	-8.775674	-3.836795	-0.938273
49	1	0	-9.544224	-4.593141	-0.857899
50	6	0	-3.546844	6.190183	-2.782809
51	6	0	1.222471	1.181025	-0.523960
52	6	0	2.631213	0.759862	-0.345047
53	6	0	3.379179	0.228768	-1.410116
54	6	0	3.280612	0.871146	0.896634
55	6	0	4.702753	-0.163086	-1.252404
56	1	0	2.912497	0.105425	-2.382967
57	6	0	4.598937	0.469142	1.072265
58	1	0	2.748285	1.293836	1.743648
59	6	0	5.341009	-0.054497	-0.002834
60	1	0	5.248646	-0.564191	-2.098869
61	1	0	5.064790	0.566210	2.046433
62	7	0	6.681838	-0.457400	0.166703
63	6	0	7.520550	0.188883	1.122242
64	6	0	7.229275	-1.514699	-0.618493
65	6	0	7.601446	1.591329	1.179162
66	6	0	8.298616	-0.560571	2.010124
67	6	0	6.549512	-2.737577	-0.757008
68	6	0	8.467229	-1.364598	-1.251338

69	6	0	8.423018	2.216883	2.105698
70	1	0	7.009995	2.187484	0.492013
71	6	0	9.145272	0.060955	2.930333
72	1	0	8.247270	-1.643970	1.976808
73	6	0	7.088501	-3.766079	-1.516396
74	1	0	5.591151	-2.873362	-0.266560
75	6	0	9.027478	-2.403080	-1.998032
76	1	0	9.003710	-0.426467	-1.153175
77	6	0	9.206746	1.457624	2.987477
78	1	0	8.487829	3.298818	2.155161
79	1	0	9.735928	-0.552624	3.599994
80	6	0	8.335386	-3.610707	-2.140384
81	1	0	6.567340	-4.711201	-1.627588
82	1	0	9.990839	-2.252049	-2.470079
83	8	0	9.989172	2.170520	3.851742
84	8	0	8.782083	-4.687476	-2.853431
85	1	0	-3.725748	7.163370	-3.229246
86	6	0	10.797270	1.452153	4.769622
87	1	0	11.331904	2.203057	5.353101
88	1	0	10.192304	0.834227	5.445782
89	1	0	11.524638	0.810961	4.254954
90	6	0	10.035846	-4.581933	-3.508043
91	1	0	10.197178	-5.538362	-4.007386
92	1	0	10.036308	-3.779789	-4.257421
93	1	0	10.852143	-4.405234	-2.795606

Electronic Energy: -2407.489791 Hartree

BTA4

Standard orientation:

Center	Atomic	Atomic	Coordinates (Angstroms)		
Number	Number	Type	X	Y	Z

1	6	0	0.005146	3.319302	-0.012898
2	6	0	1.055038	4.039202	0.590981
3	6	0	-1.038860	4.049248	-0.615157
4	6	0	1.055201	5.427014	0.592408
5	1	0	1.872142	3.502492	1.058739
6	6	0	-1.026986	5.437029	-0.615102
7	1	0	-1.861708	3.520247	-1.081641
8	1	0	1.882648	5.956512	1.052782
9	1	0	-1.852384	5.973336	-1.071040
10	6	0	-1.230478	1.189215	0.007598
11	6	0	-2.254989	1.537023	0.899990
12	6	0	-1.430022	0.101244	-0.853797
13	6	0	-3.450420	0.826522	0.912132
14	1	0	-2.114817	2.375151	1.574772
15	6	0	-2.620452	-0.617036	-0.818183
16	1	0	-0.641486	-0.186337	-1.541425
17	1	0	-4.238555	1.136159	1.591454
18	1	0	-2.737139	-1.472946	-1.475702
19	6	0	1.222927	1.179405	-0.031774
20	6	0	2.249797	1.515781	-0.925762
21	6	0	1.414546	0.094179	0.834795
22	6	0	3.440300	0.796983	-0.934262
23	1	0	2.115533	2.351817	-1.604347
24	6	0	2.600075	-0.632380	0.803018
25	1	0	0.623914	-0.184613	1.523645
26	1	0	4.230752	1.097909	-1.614813
27	1	0	2.710931	-1.485781	1.464803

28	7	0	-0.000989	1.912535	-0.013655
29	6	0	0.018004	6.161714	-0.012108
30	6	0	0.028839	7.637201	-0.007319
31	6	0	0.247643	8.336426	1.197942
32	6	0	-0.194806	8.357544	-1.199110
33	7	0	0.292894	9.728277	1.226091
34	6	0	0.354777	7.872728	2.536292
35	7	0	-0.211072	9.751123	-1.204740
36	6	0	-0.408904	7.916755	-2.532186
37	6	0	0.428169	10.114853	2.503903
38	5	0	0.183364	10.683530	-0.008679
39	6	0	0.476844	8.991649	3.351189
40	1	0	0.317135	6.838691	2.844754
41	1	0	-0.425580	6.887175	-2.856562
42	6	0	-0.572941	9.050107	-3.319441
43	1	0	0.472013	11.165373	2.755908
44	9	0	1.407679	11.276152	-0.266172
45	9	0	-0.799744	11.630251	0.212066
46	1	0	0.572392	9.014005	4.427279
47	1	0	-0.760025	9.090129	-4.382928
48	6	0	-0.436853	10.158987	-2.463157
49	1	0	-0.476425	11.213761	-2.697810
50	6	0	3.646370	-0.297161	-0.074815
51	6	0	4.913613	-1.064011	-0.094799
52	6	0	5.462731	-1.605881	1.080239
53	6	0	5.622556	-1.279778	-1.289197
54	6	0	6.649506	-2.328553	1.068154
55	1	0	4.958775	-1.447395	2.029129
56	6	0	6.817708	-1.988228	-1.313234
57	1	0	5.225428	-0.895996	-2.224351

58	6	0	7.355336	-2.531526	-0.131953
59	1	0	7.041613	-2.734319	1.994041
60	1	0	7.339219	-2.131229	-2.253047
61	7	0	8.565407	-3.257460	-0.149411
62	6	0	8.932741	-4.029052	-1.290642
63	6	0	9.449304	-3.221322	0.968832
64	6	0	8.016663	-4.912746	-1.888024
65	6	0	10.219980	-3.938737	-1.829572
66	6	0	9.807319	-1.999648	1.565727
67	6	0	9.997076	-4.400488	1.483875
68	6	0	8.377699	-5.666458	-2.995456
69	1	0	7.016011	-5.000244	-1.477681
70	6	0	10.598347	-4.710878	-2.929866
71	1	0	10.938030	-3.261493	-1.378918
72	6	0	10.672921	-1.966703	2.649462
73	1	0	9.397306	-1.074690	1.173685
74	6	0	10.887128	-4.373885	2.559528
75	1	0	9.730810	-5.351095	1.033366
76	6	0	9.673812	-5.576395	-3.524612
77	1	0	7.673979	-6.348575	-3.461024
78	1	0	11.605816	-4.616839	-3.317270
79	6	0	11.225168	-3.153364	3.154254
80	1	0	10.950979	-1.026624	3.114406
81	1	0	11.294938	-5.307390	2.928477
82	6	0	-3.664187	-0.270591	0.058379
83	6	0	-4.936106	-1.029507	0.083402
84	6	0	-5.488248	-1.576877	-1.087677
85	6	0	-5.646494	-1.232576	1.279177
86	6	0	-6.678968	-2.292923	-1.070475
87	1	0	-4.983368	-1.428276	-2.037673

88	6	0	-6.845649	-1.934007	1.308236
89	1	0	-5.247297	-0.844490	2.211659
90	6	0	-7.386169	-2.483141	0.130960
91	1	0	-7.073032	-2.703429	-1.993429
92	1	0	-7.367921	-2.067081	2.249062
93	7	0	-8.600110	-3.202349	0.153552
94	6	0	-8.974246	-3.959629	1.302200
95	6	0	-9.482628	-3.171774	-0.966000
96	6	0	-8.064847	-4.842864	1.910380
97	6	0	-10.261717	-3.855343	1.838022
98	6	0	-9.833729	-1.953877	-1.574576
99	6	0	-10.036165	-4.352801	-1.470488
100	6	0	-8.432395	-5.582277	3.025281
101	1	0	-7.064156	-4.941232	1.502607
102	6	0	-10.646730	-4.613190	2.945924
103	1	0	-10.974682	-3.178335	1.379024
104	6	0	-10.698112	-1.926537	-2.659454
105	1	0	-9.419239	-1.027442	-1.190813
106	6	0	-10.925059	-4.331614	-2.547203
107	1	0	-9.775288	-5.300534	-1.010867
108	6	0	-9.728675	-5.478126	3.551448
109	1	0	-7.733795	-6.263887	3.499221
110	1	0	-11.654169	-4.508587	3.330667
111	6	0	-11.256055	-3.114965	-3.153724
112	1	0	-10.970794	-0.989405	-3.133446
113	1	0	-11.337447	-5.266380	-2.907771
114	8	0	-12.103897	-2.977106	-4.216738
115	8	0	-9.992198	-6.257146	4.643142
116	8	0	12.074895	-3.009968	4.215047
117	8	0	9.930632	-6.368916	-4.608176

118	6	0	12.657951	-4.179037	4.766835
119	1	0	13.291971	-3.841573	5.587995
120	1	0	11.896884	-4.866395	5.158438
121	1	0	13.274622	-4.711381	4.030943
122	6	0	11.224502	-6.311110	-5.185758
123	1	0	11.213050	-7.010456	-6.022989
124	1	0	11.457653	-5.305681	-5.559521
125	1	0	12.001595	-6.615805	-4.472683
126	6	0	-11.286270	-6.184039	5.218571
127	1	0	-11.280385	-6.873719	6.063836
128	1	0	-11.513053	-5.172792	5.580384
129	1	0	-12.064627	-6.491737	4.508169
130	6	0	-12.692973	-4.148226	-4.757733
131	1	0	-13.324206	-3.815238	-5.582855
132	1	0	-11.935384	-4.843655	-5.141734
133	1	0	-13.313401	-4.669837	-4.017330

Electronic Energy: -3385.065059 Hartree

Copies of ^1H -NMR, ^{13}C -NMR, ^{11}B -NMR, ^{19}F -NMR and HRMS/MALDI spectra

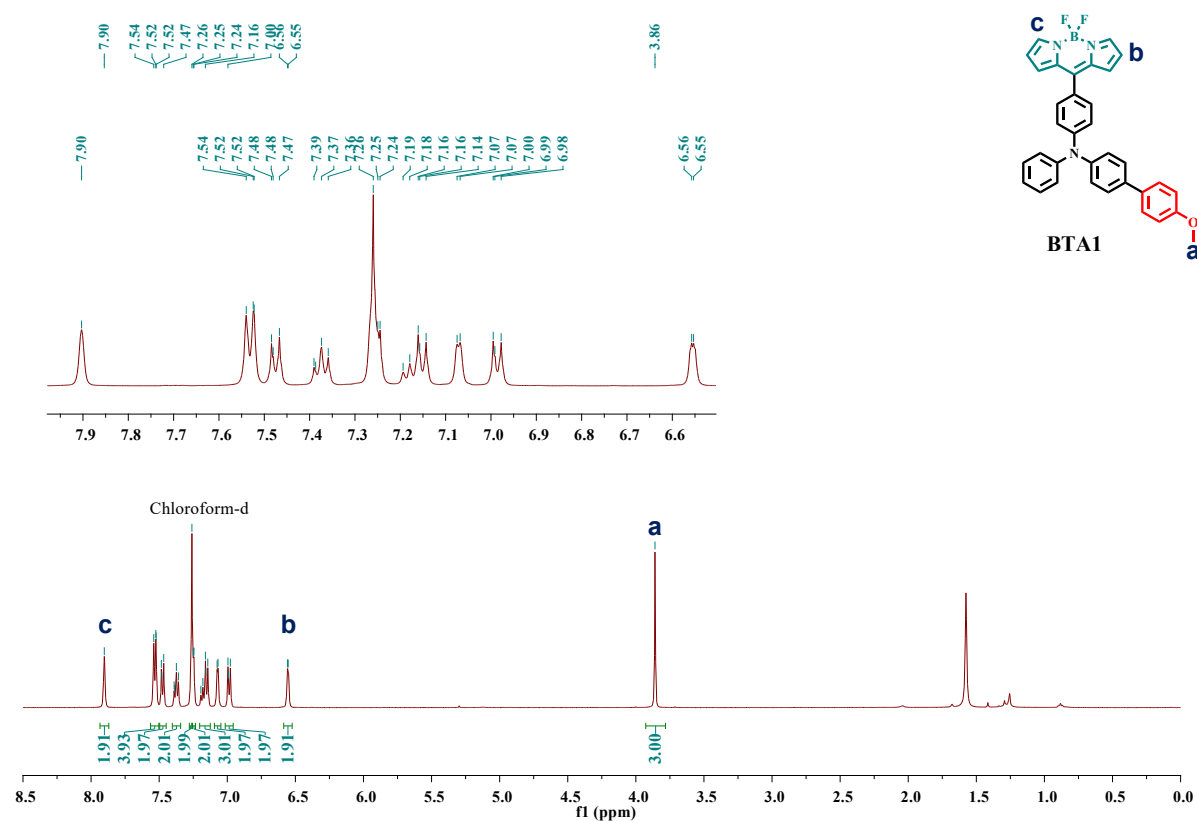


Fig. S12 ^1H NMR (CDCl_3) spectra of BTA1.

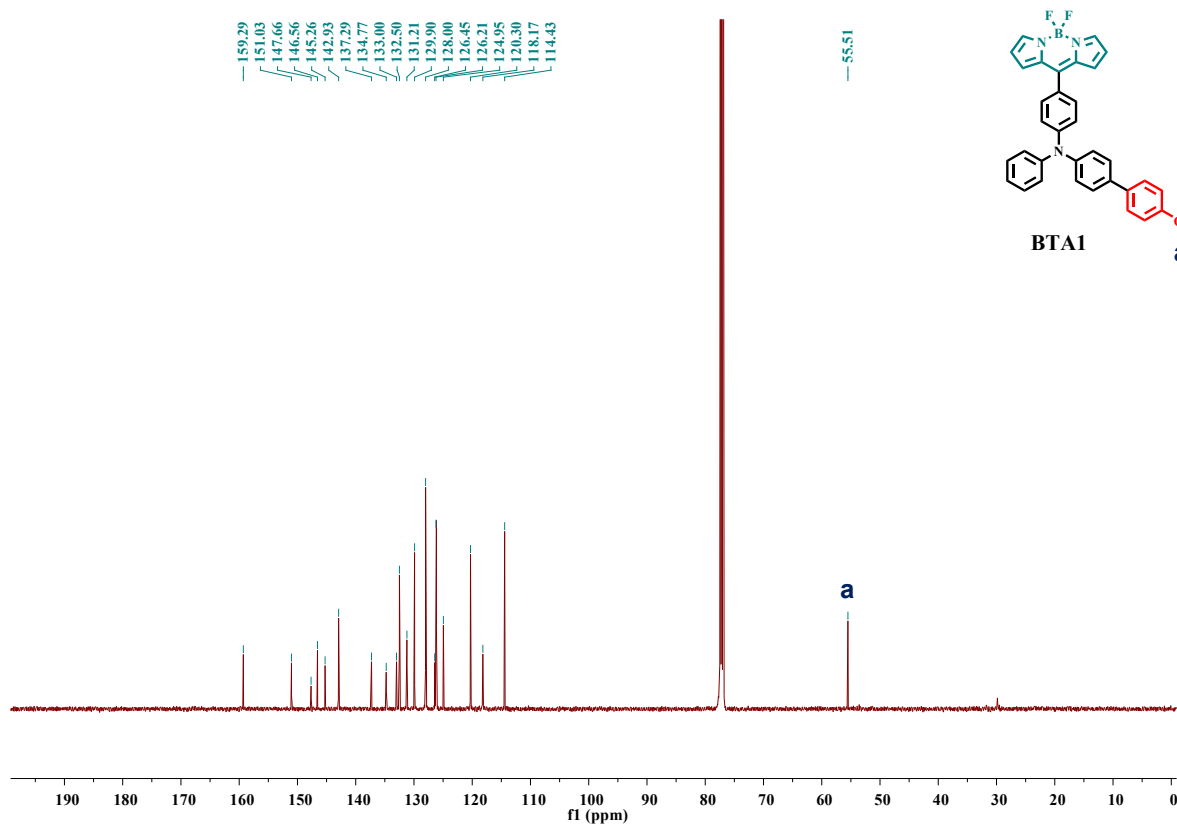


Fig. S13 $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3) spectra of BTA1.

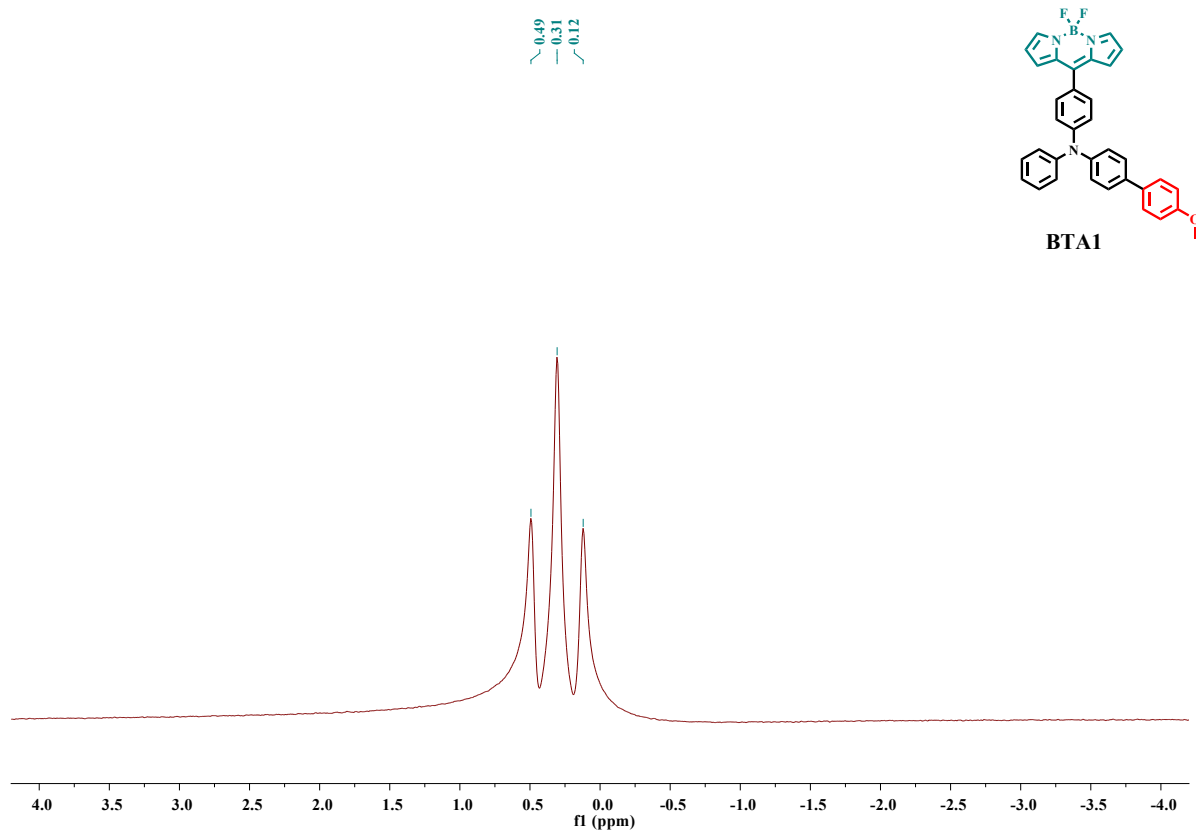


Fig. S14 ¹¹B NMR (CDCl₃) spectra of BTA1.

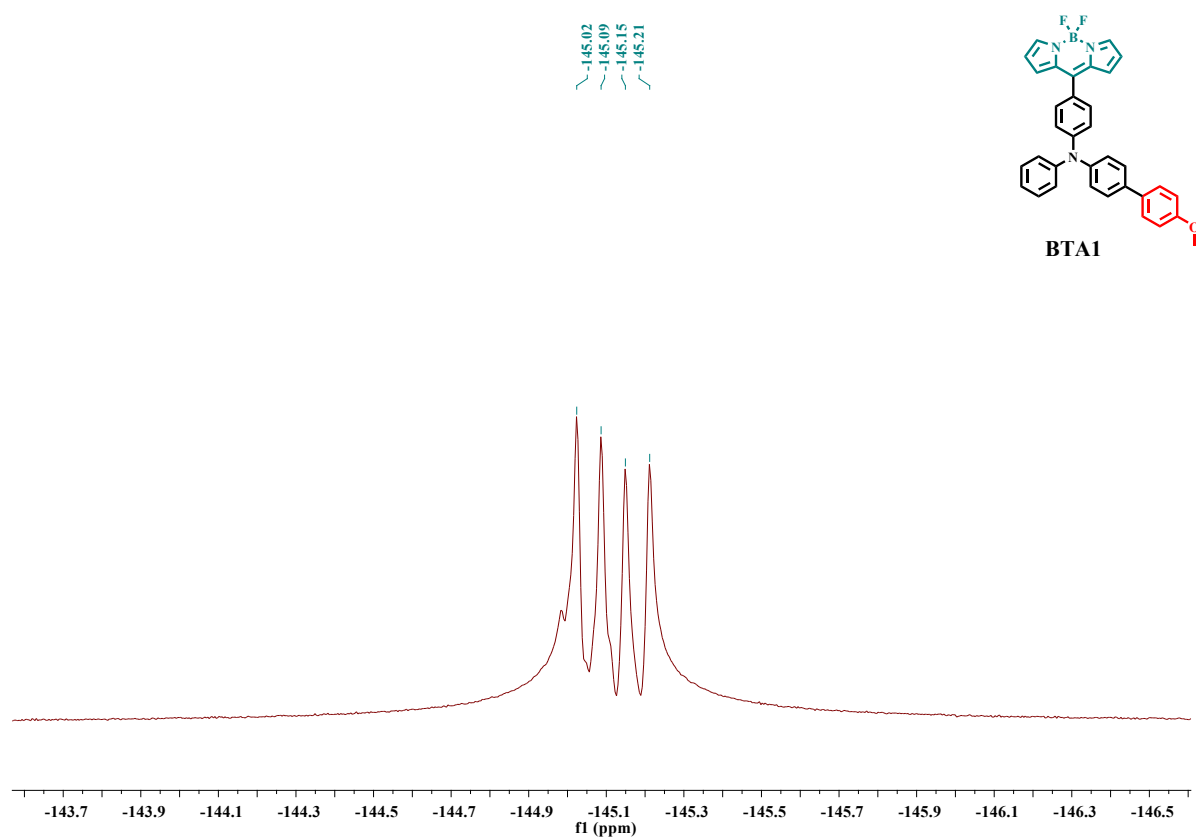


Fig. S15 ¹⁹F NMR (CDCl₃) spectra of BTA1.

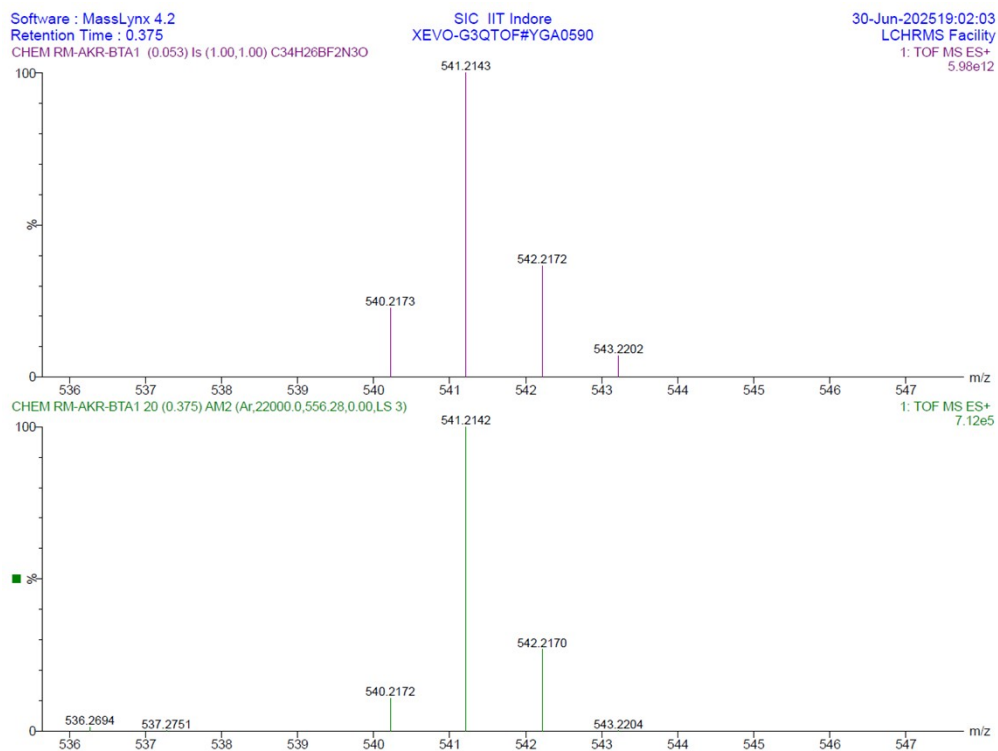


Fig. S16 HRMS of **BTA1**.

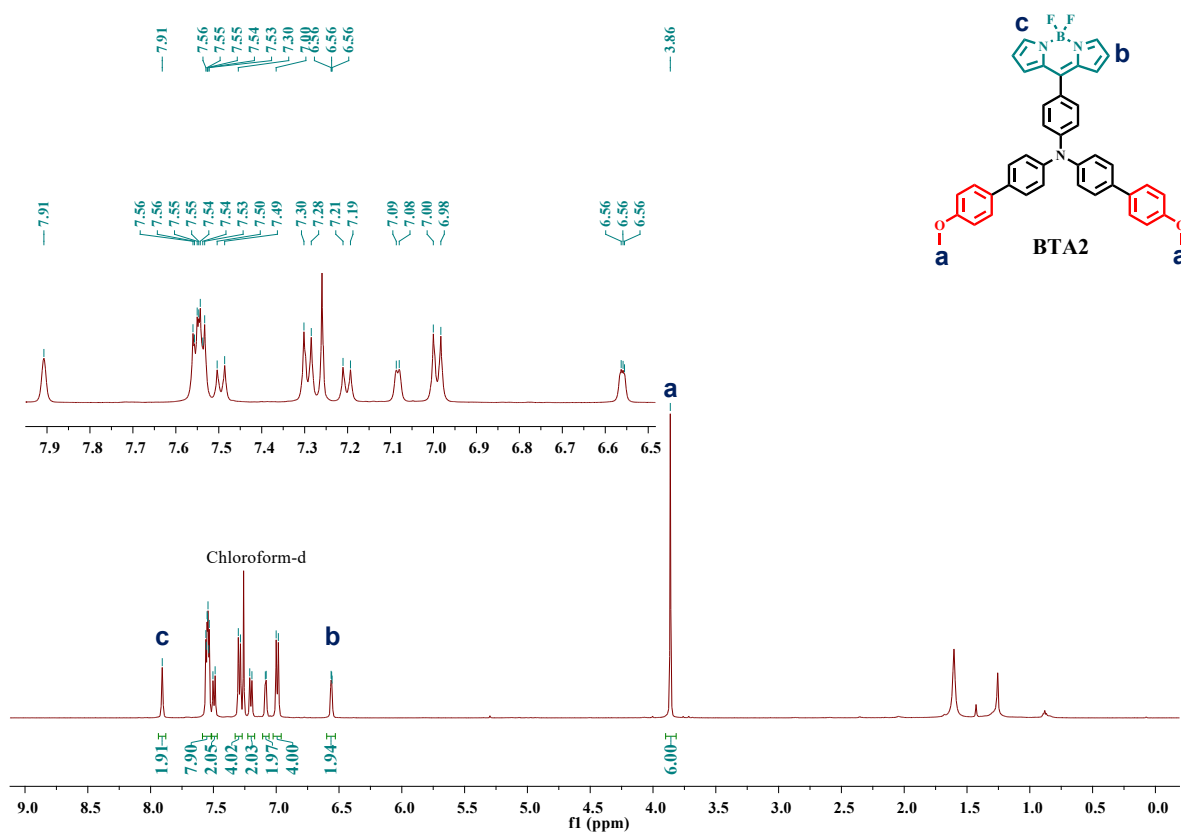


Fig. S17 ¹H NMR (CDCl₃) spectra of **BTA2**.

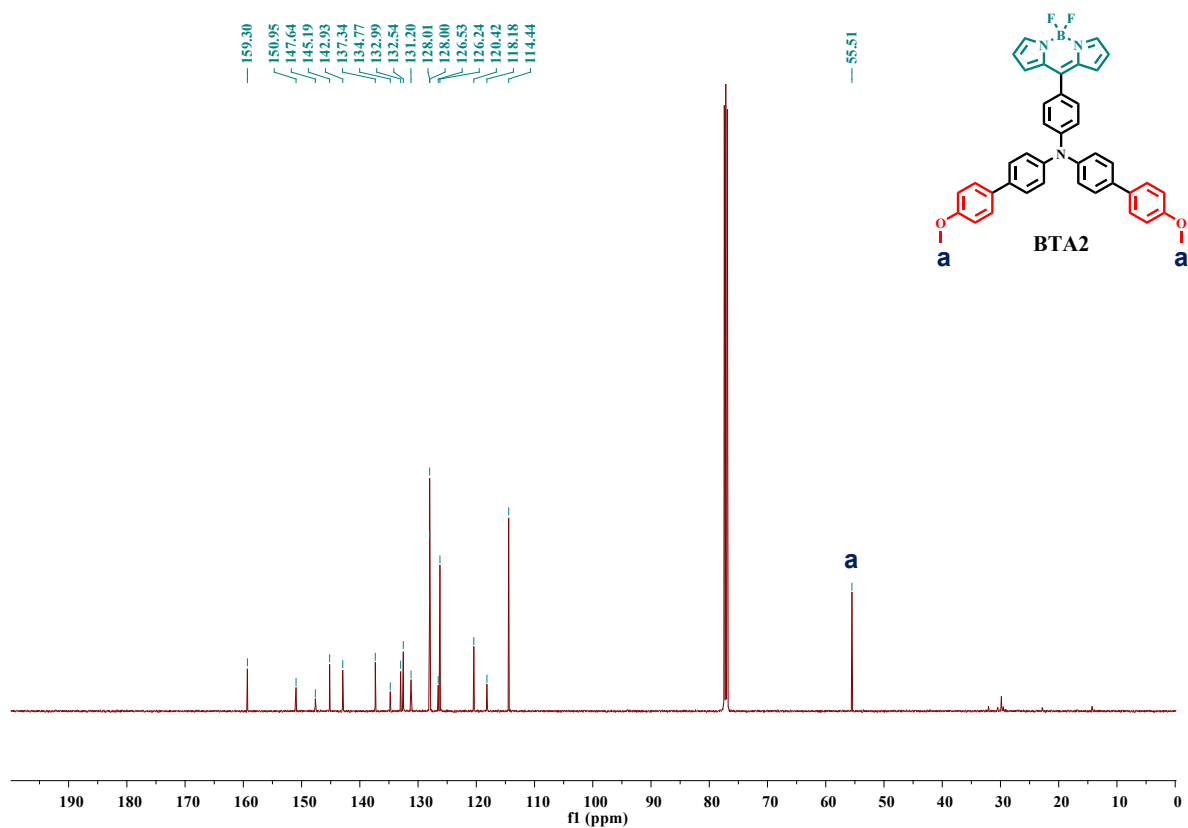


Fig. S18 $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3) spectra of BTA2.

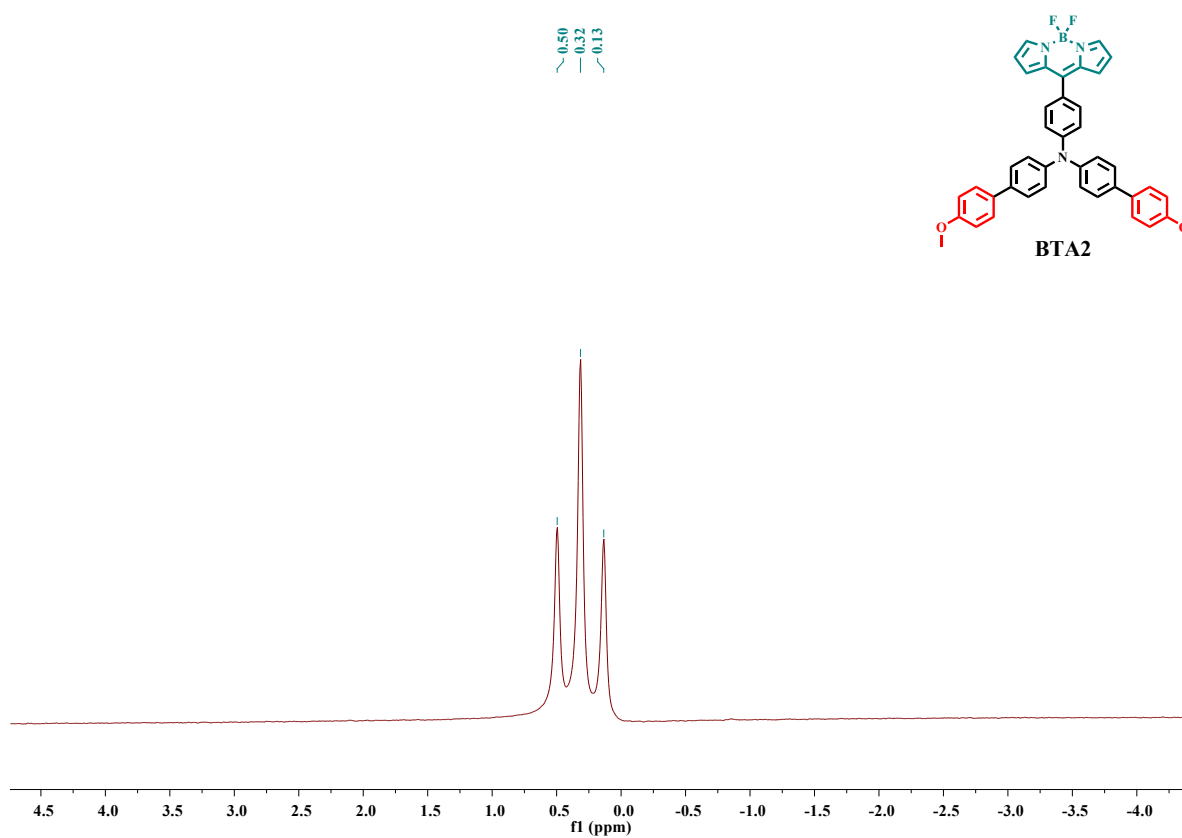


Fig. S19 ^{11}B NMR (CDCl_3) spectra of BTA2.

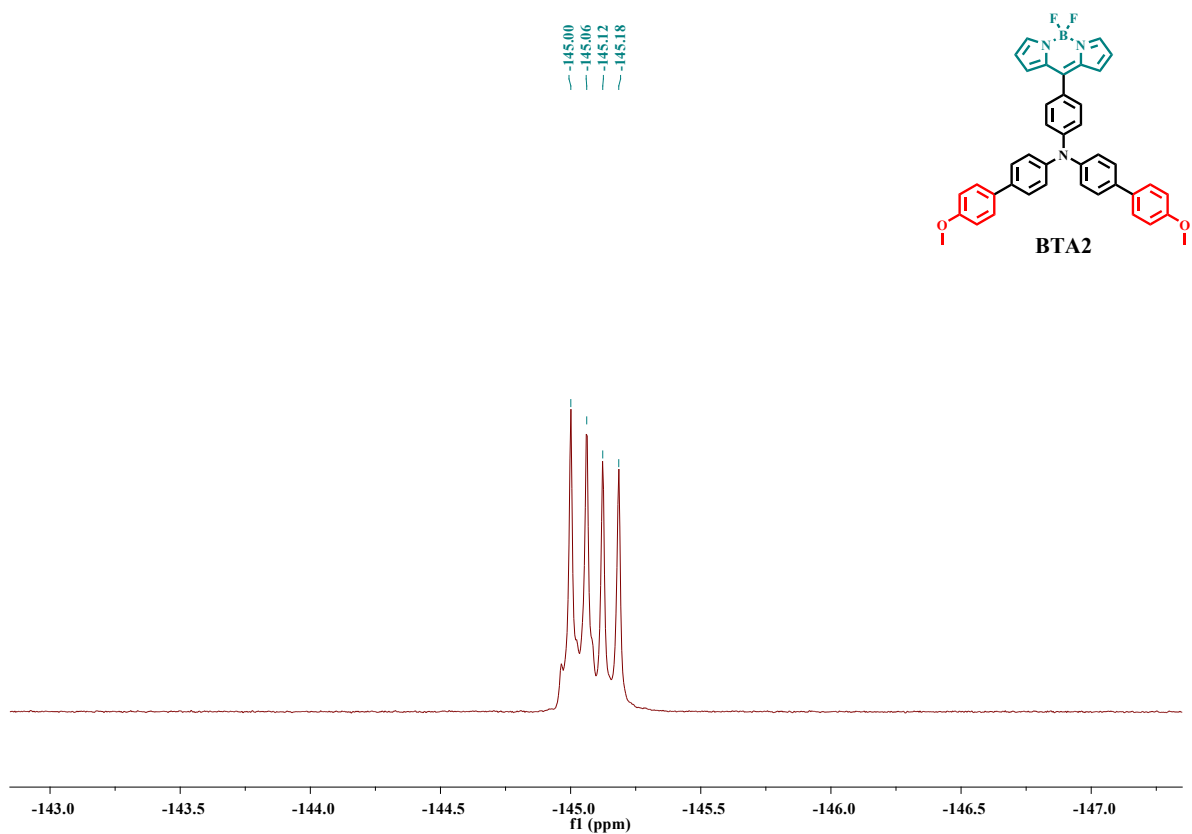


Fig. S20 ^{19}F NMR (CDCl_3) spectra of **BTA2**.

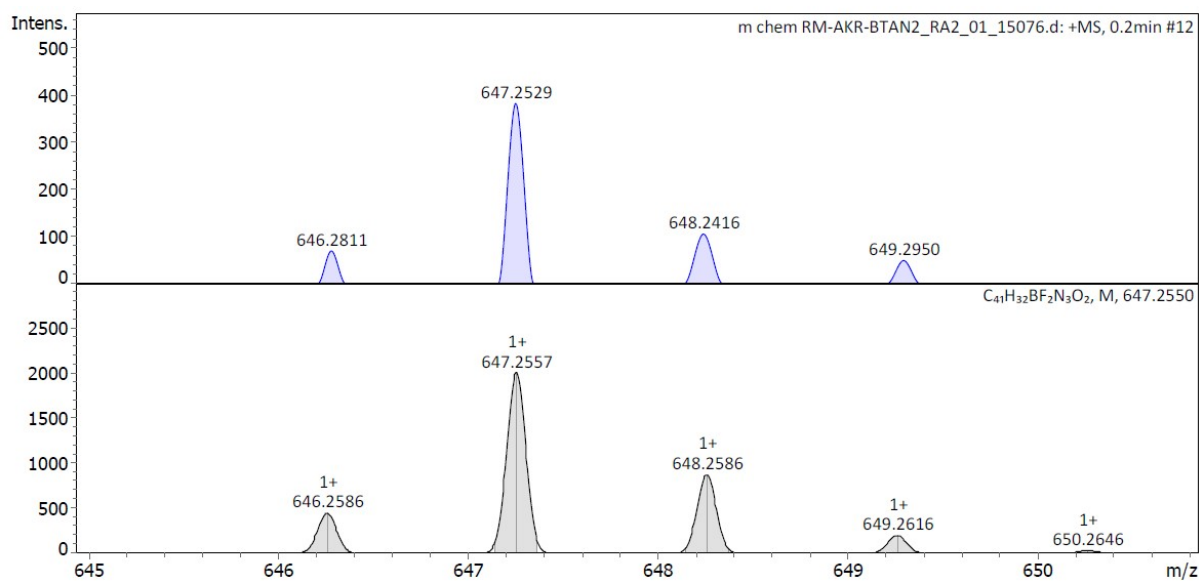


Fig. S21 HRMS of **BTA2**.

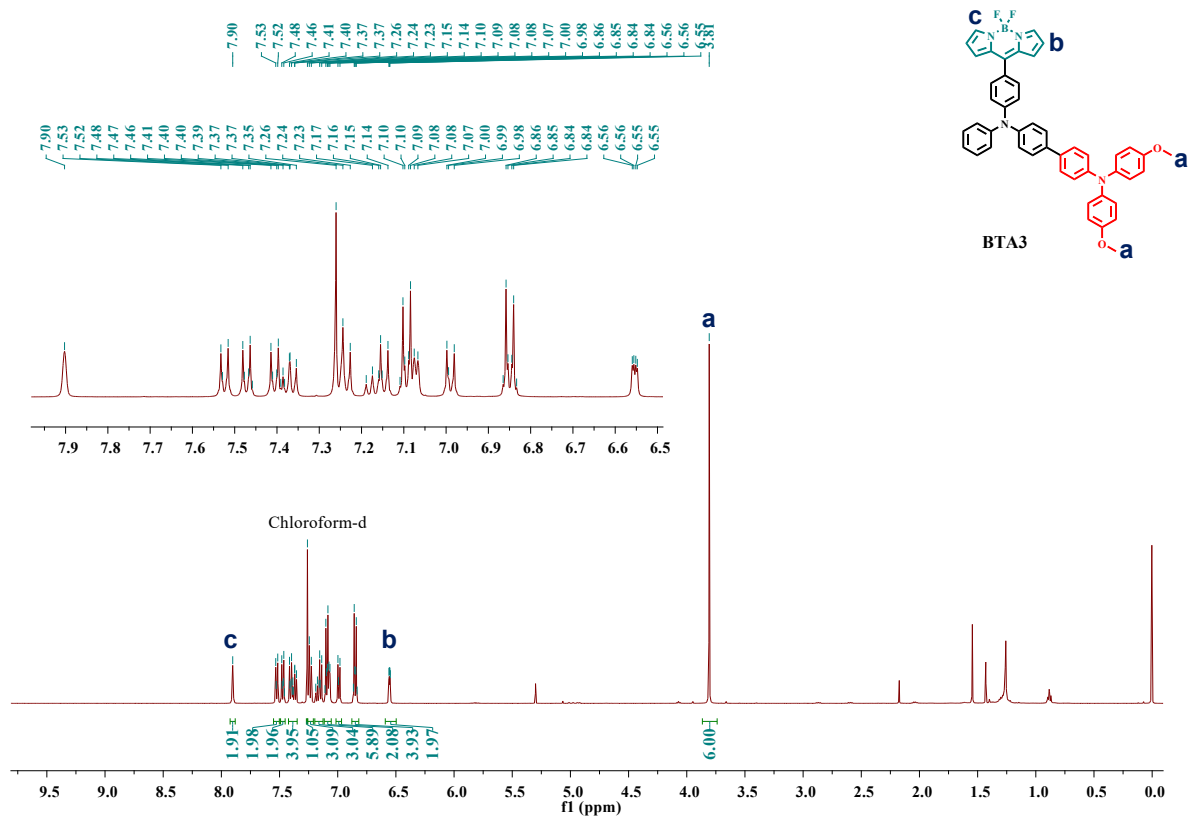


Fig. S22 ^1H NMR (CDCl_3) spectra of BTA3.

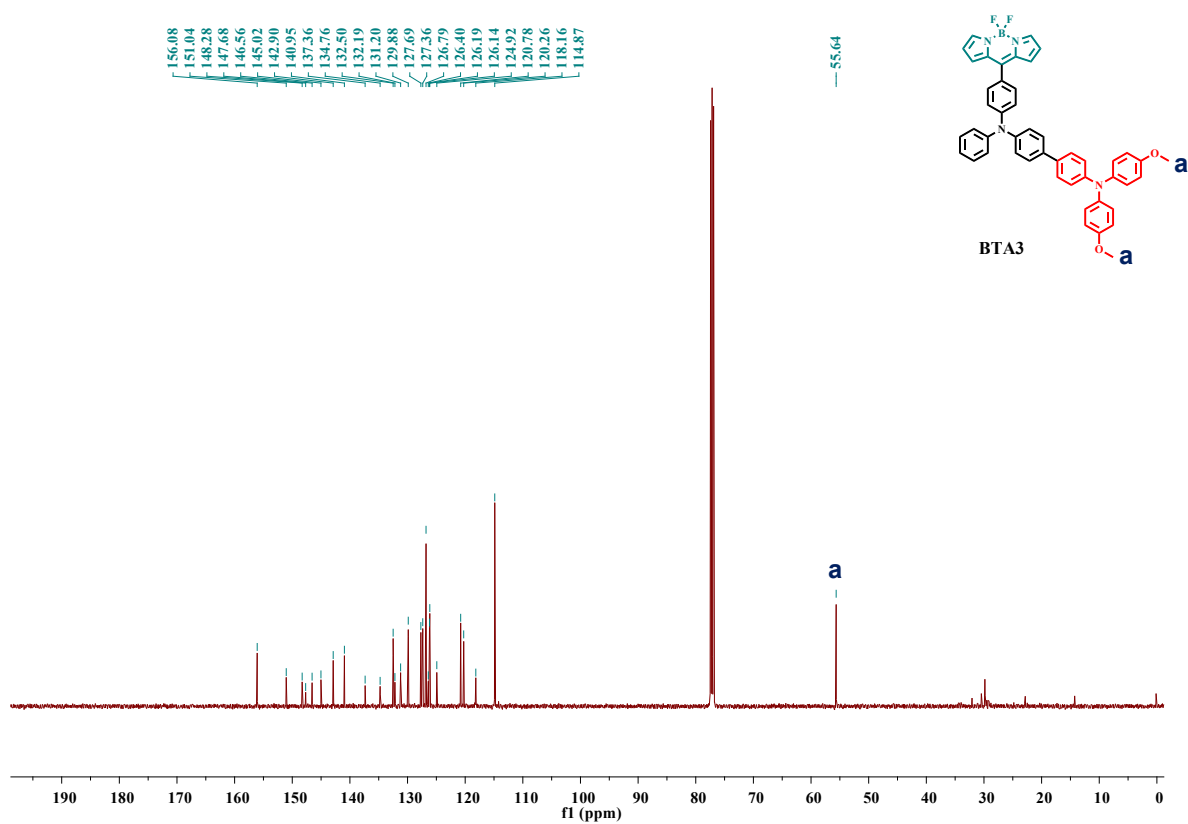


Fig. S23 $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3) spectra of BTA3.

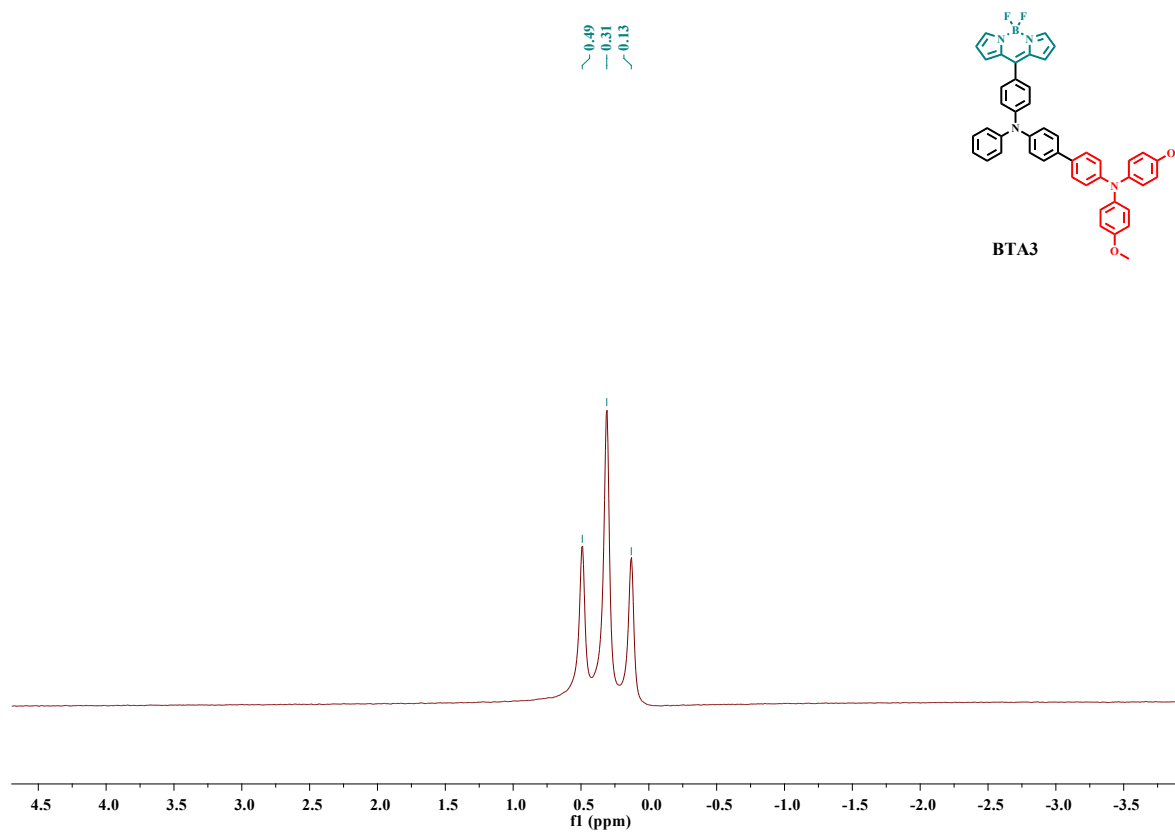


Fig. S24 ^{11}B NMR (CDCl_3) spectra of **BTA3**.

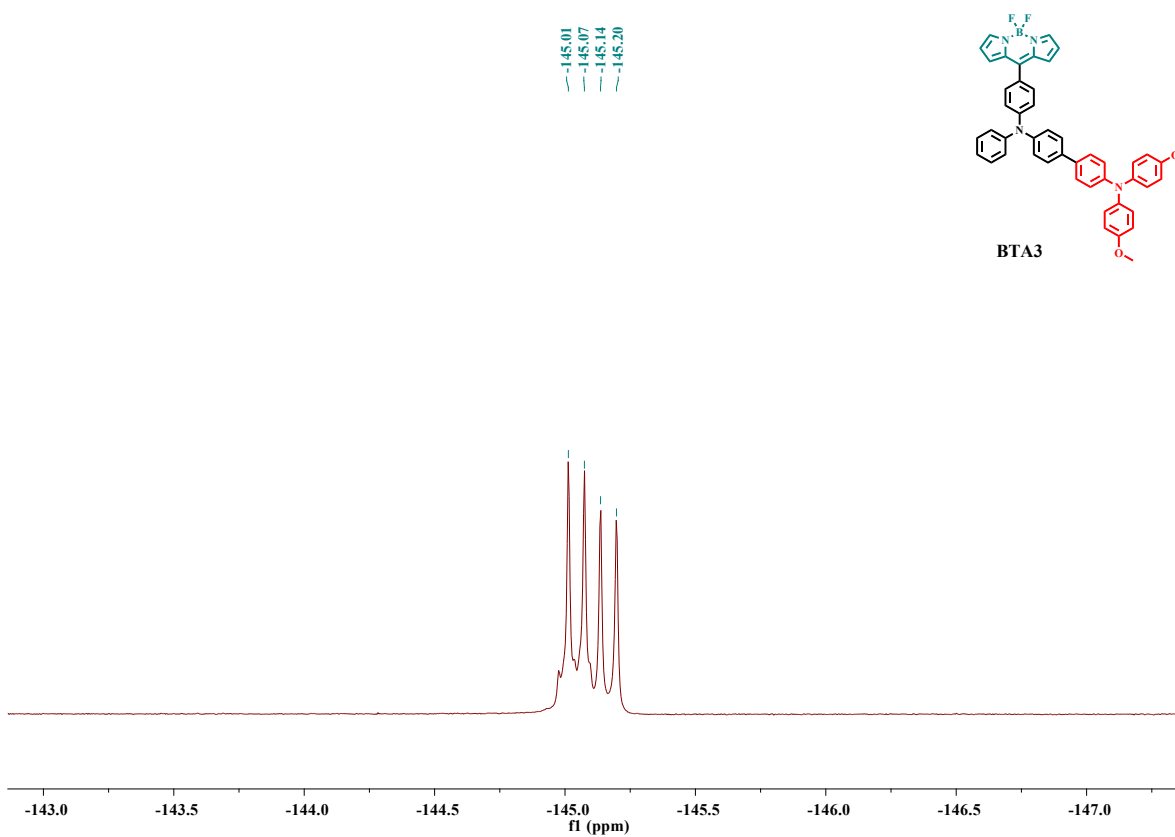


Fig. S25 ^{19}F NMR (CDCl_3) spectra of **BTA3**.

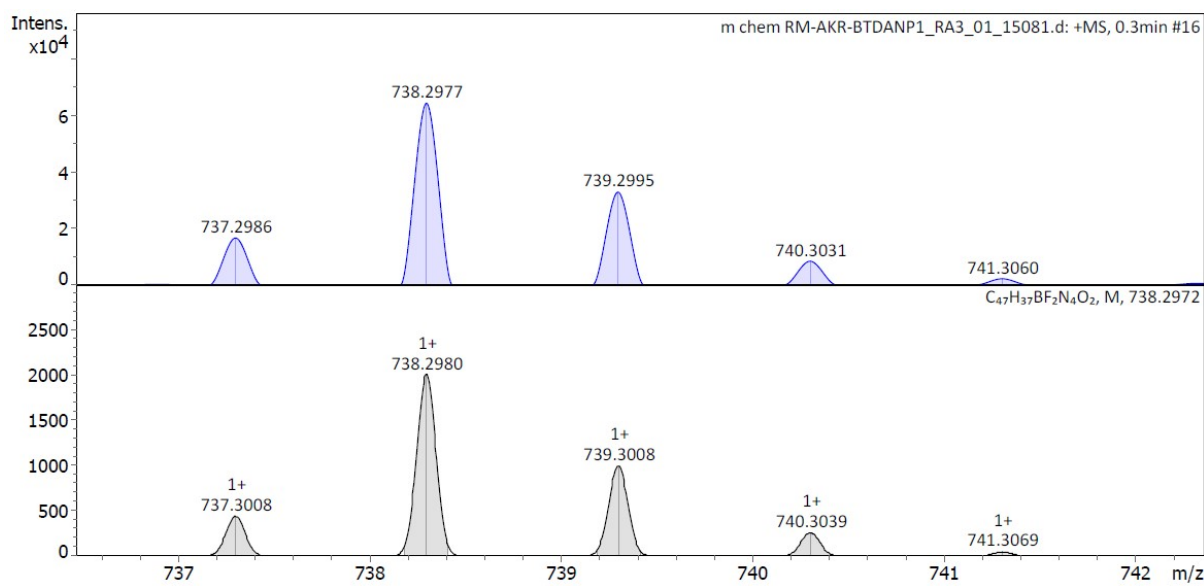


Fig. S26 HRMS of BTA3.

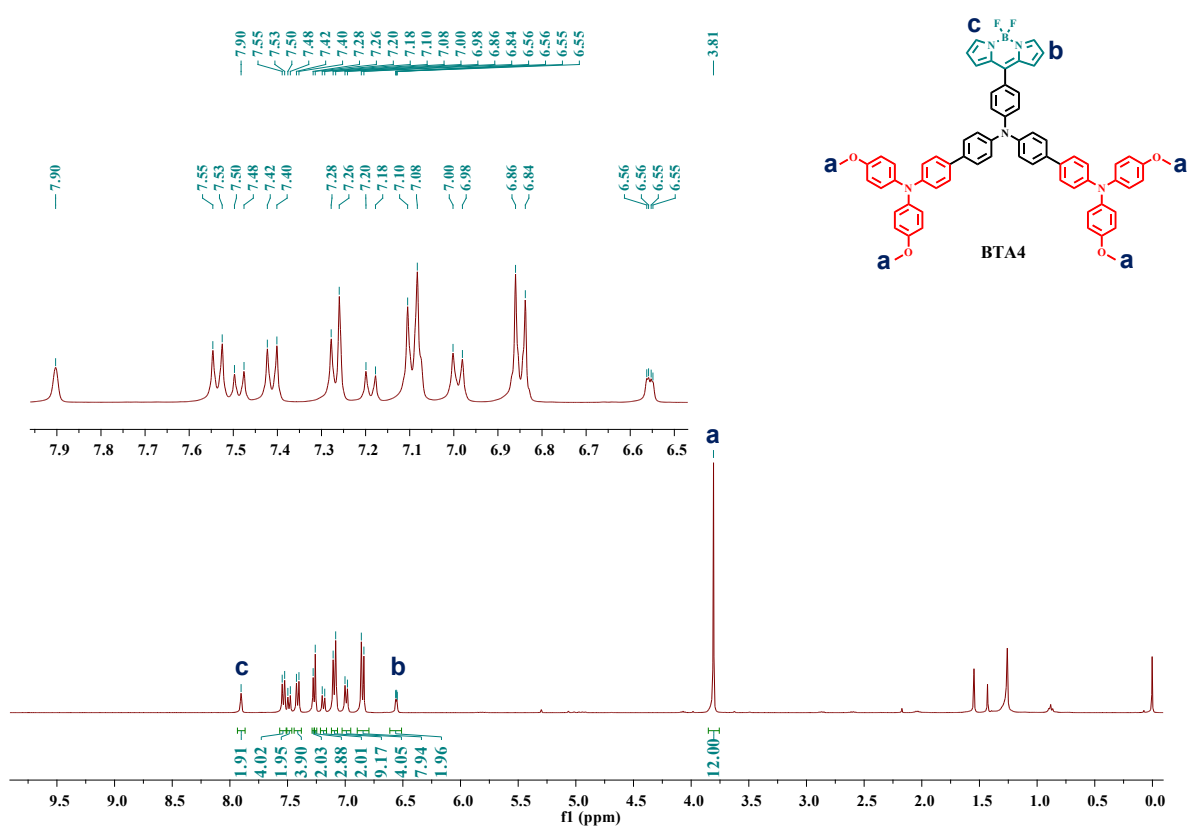


Fig. S27 1H NMR ($CDCl_3$) spectra of BTA4.

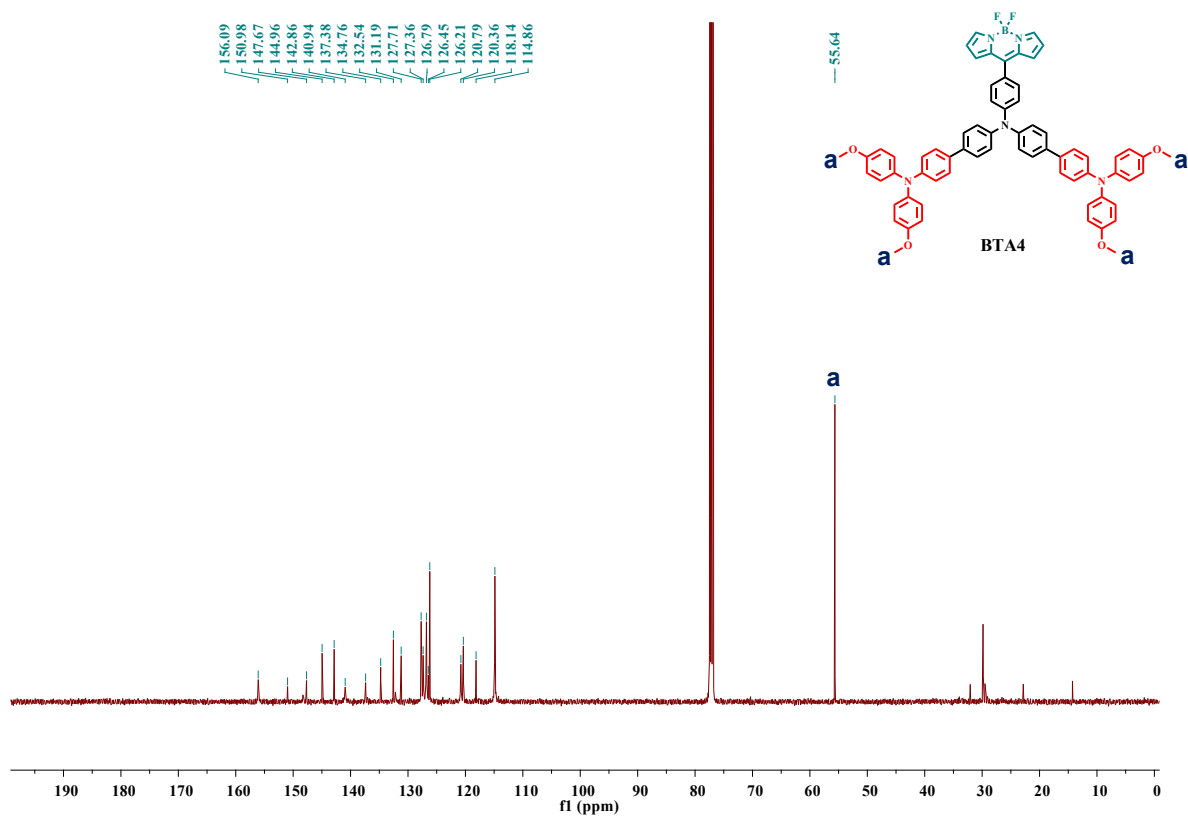


Fig. S28 $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3) spectra of **BTA4**.

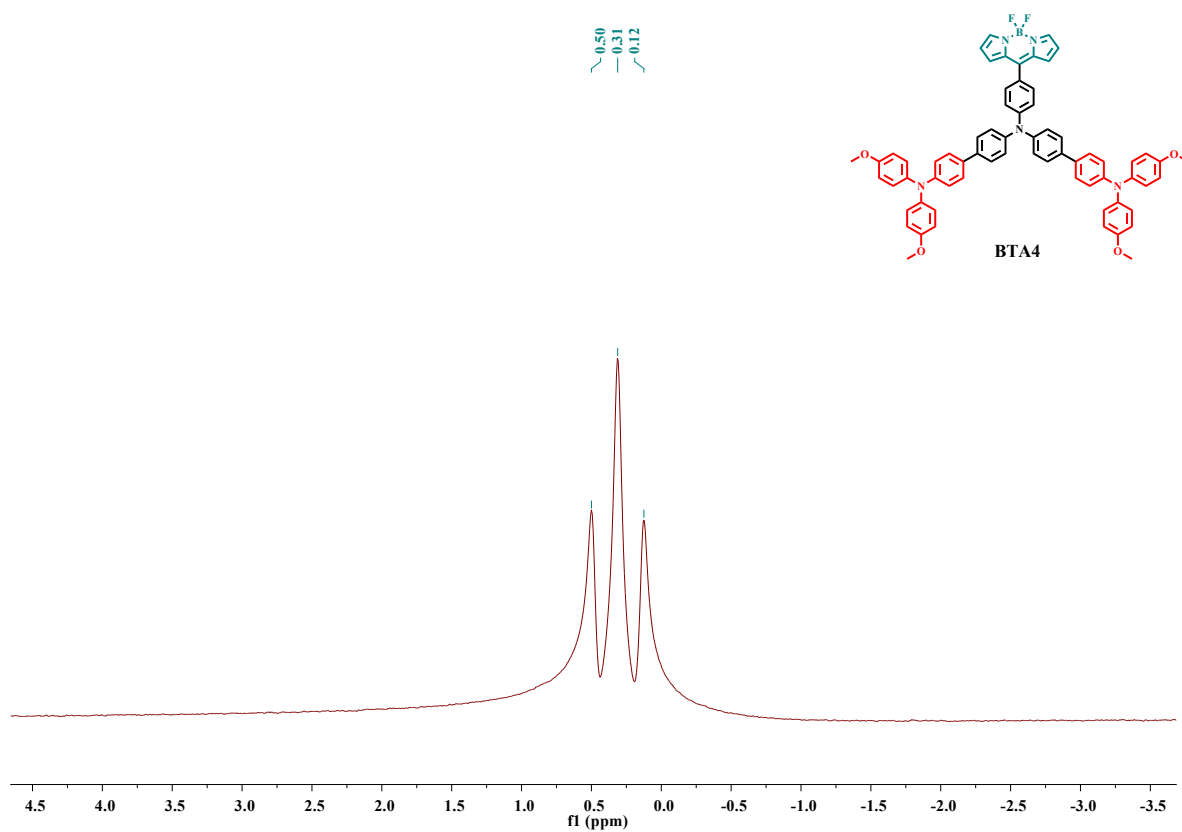


Fig. S29 ^{11}B NMR (CDCl_3) spectra of **BTA4**.

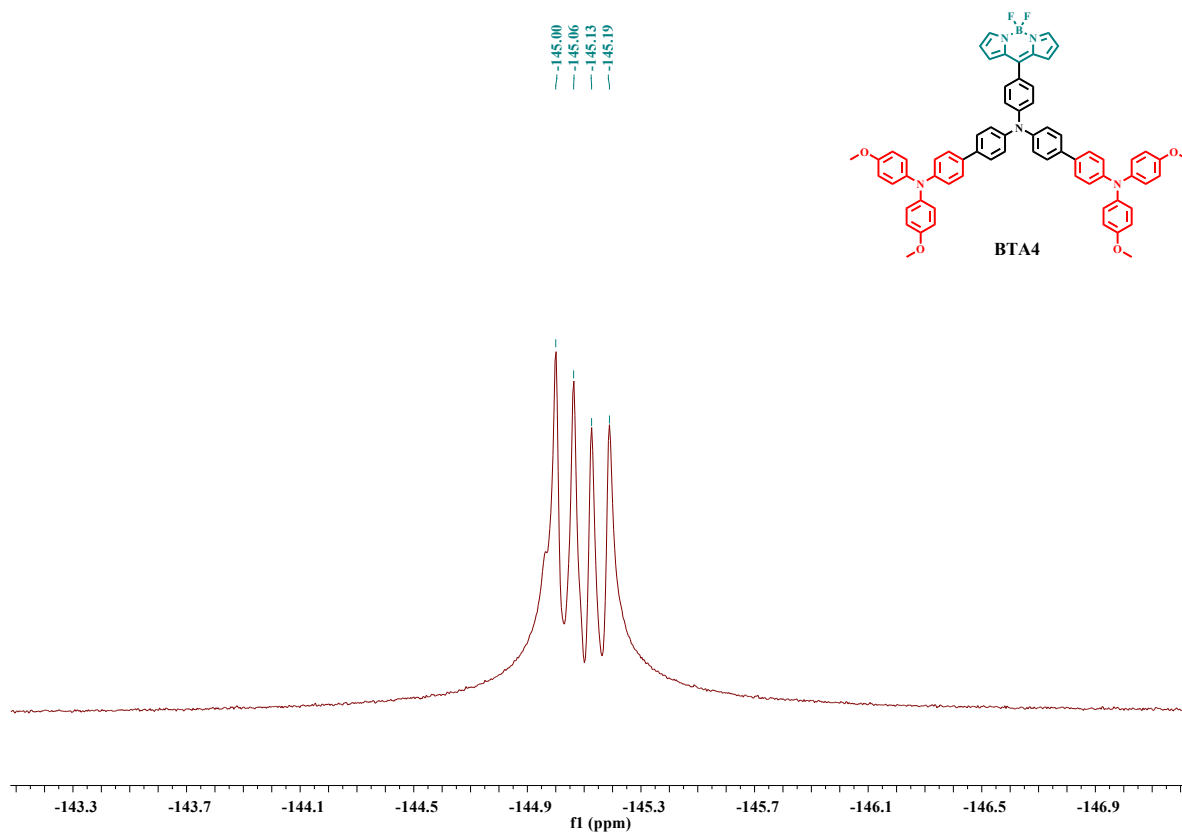


Fig. S30 ^{19}F NMR (CDCl₃) spectra of BTA4.

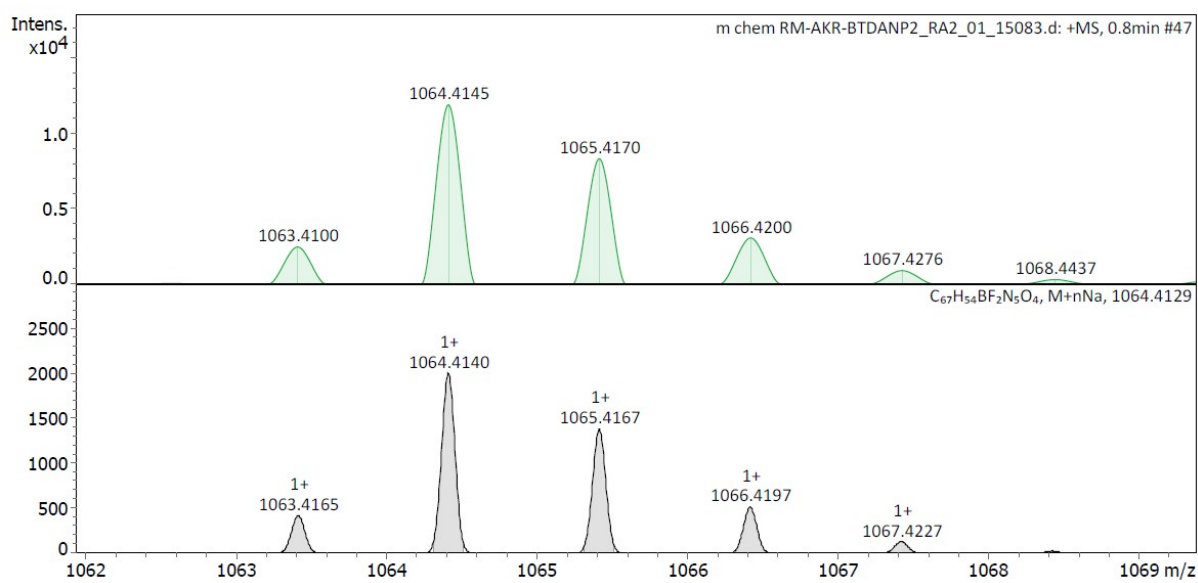


Fig. S31 HRMS of BTA4.

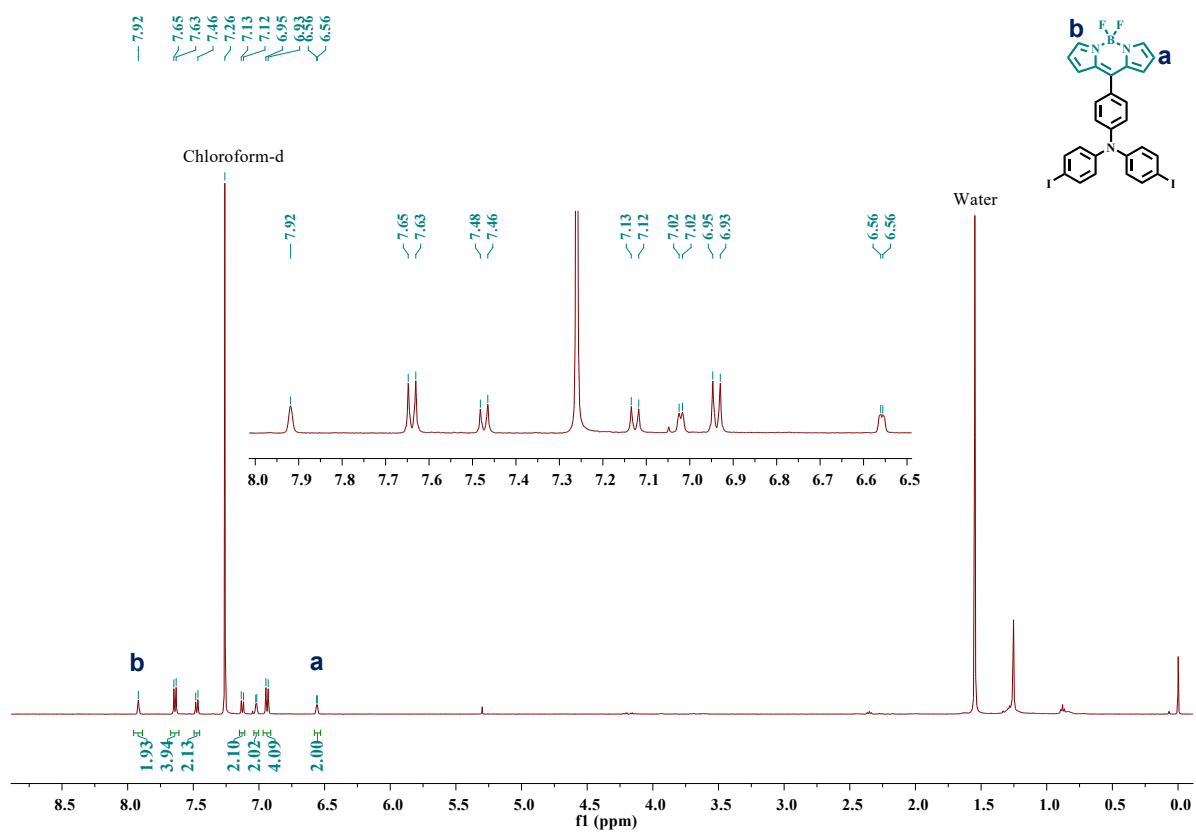


Fig. S32 ¹H NMR (CDCl₃) spectra of BTI₂.

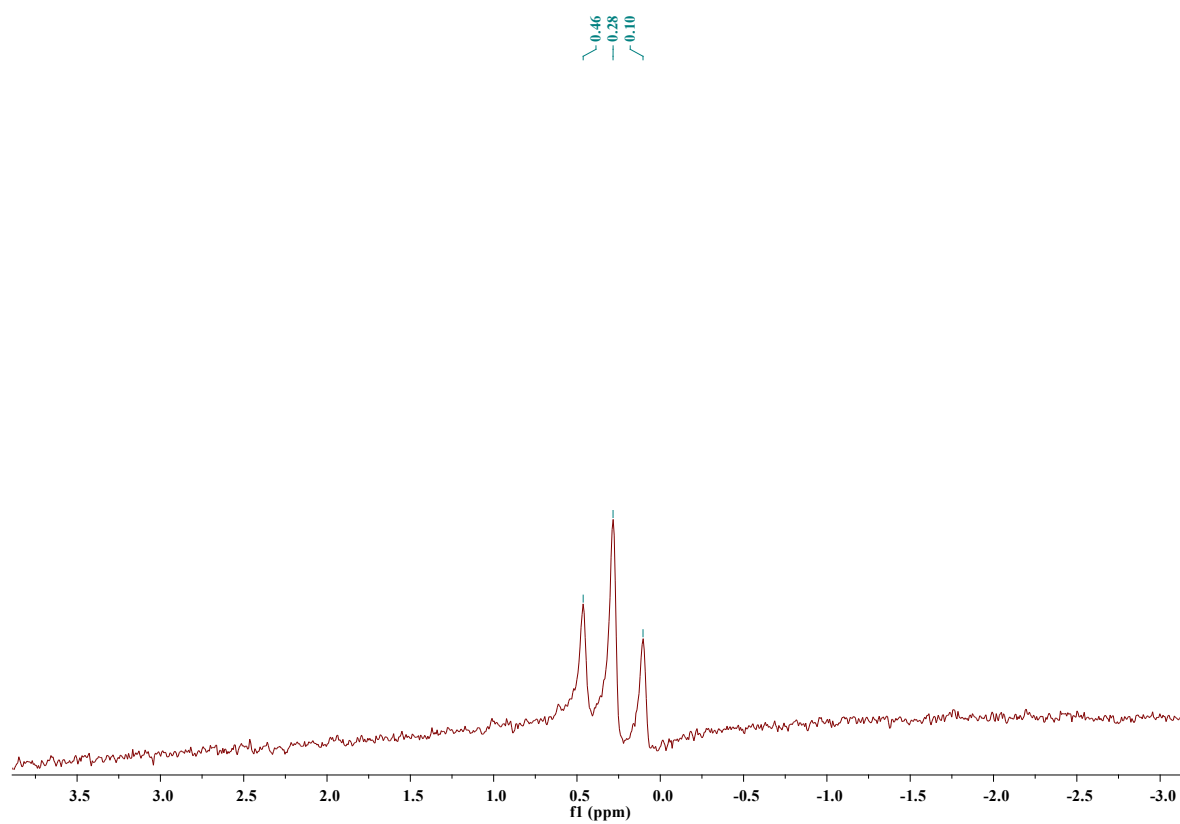


Fig. S33 ¹¹B NMR (CDCl₃) spectra of BTI₂.

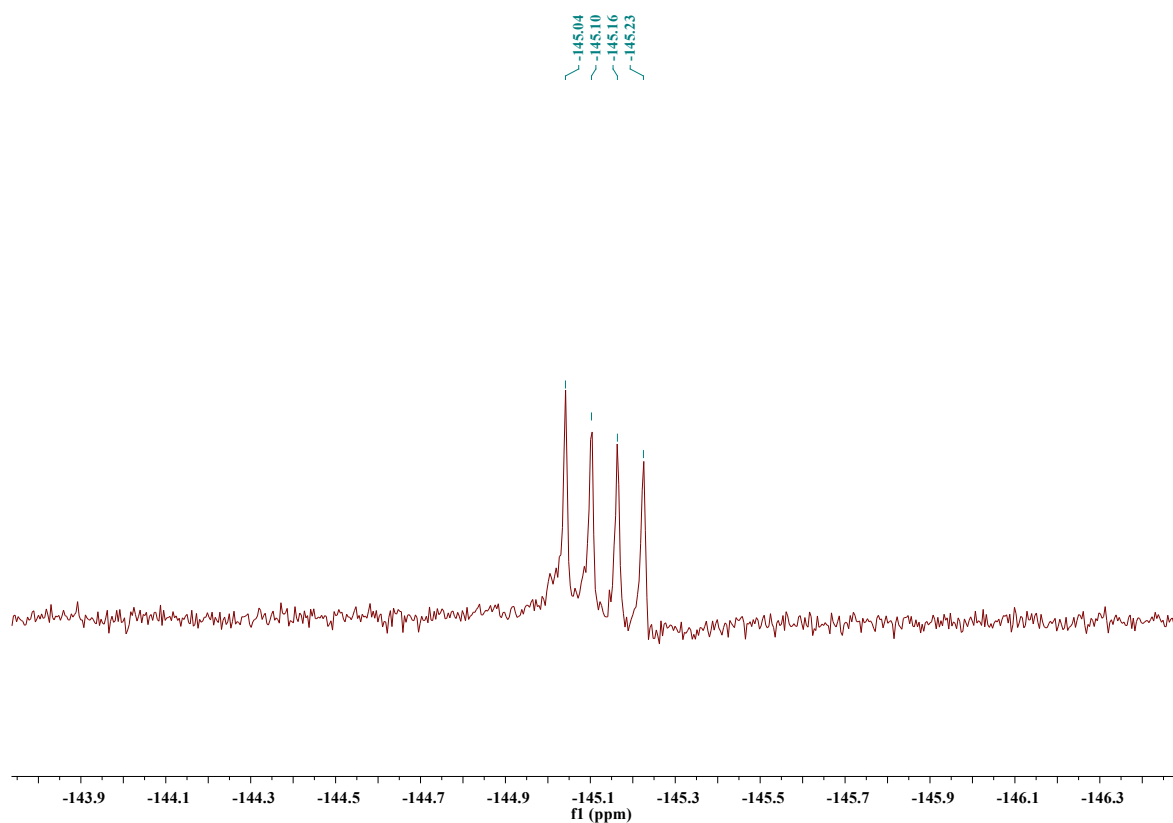


Fig. S34 ^{19}F NMR (CDCl_3) spectra of BTI_2 .

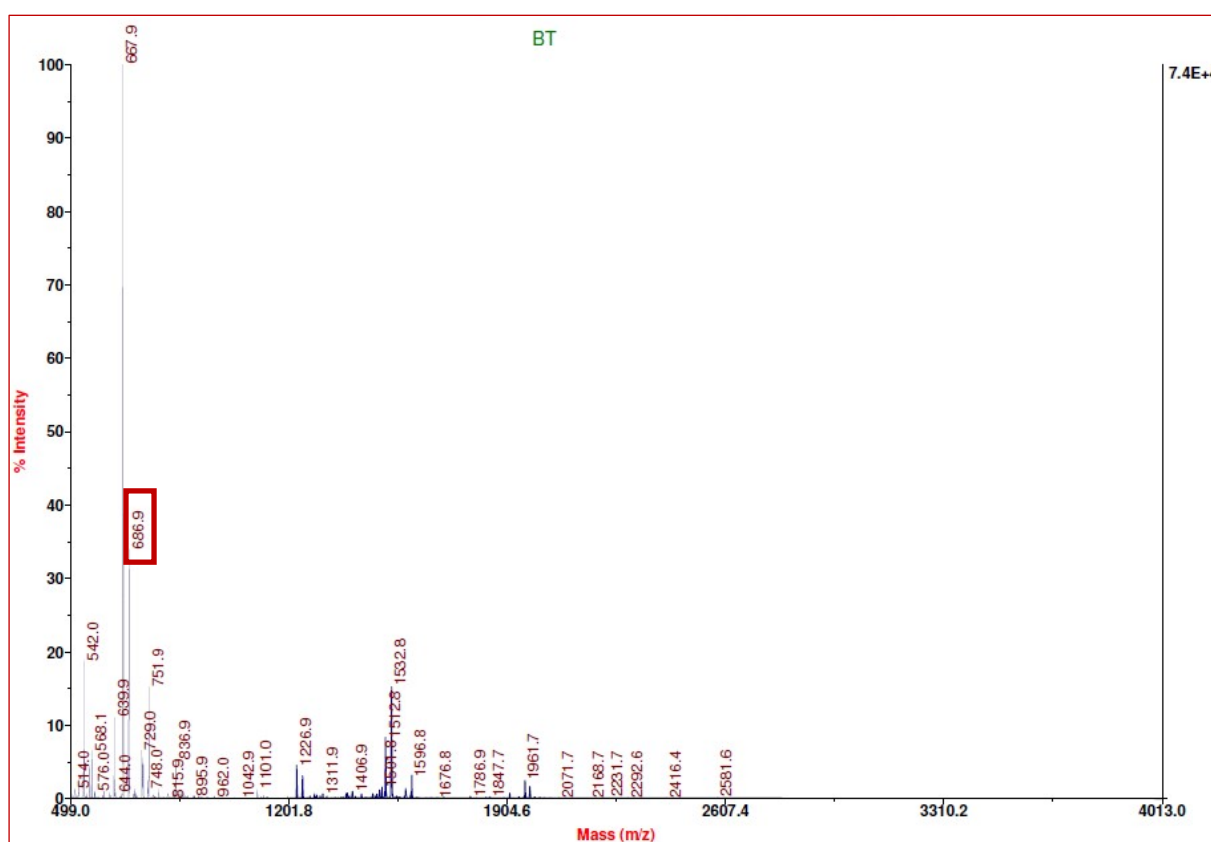


Fig. S35 MALDI-TOF-MS of BTI_2 .

Reference

- 1 B. G. Kim, X. Ma, C. Chen, Y. Ie, E. W. Coir, H. Hashemi, Y. Aso, P. F. Green, J. Kieffer and J. Kim, *Adv. Funct. Mater.*, 2013, **23**, 439–445.