

Structure–properties correlation in $(1-x)\text{Bi}_{0.90}\text{Pr}_{0.10}\text{FeO}_3 - (x)\text{Ca}_{0.5}\text{Sr}_{0.5}\text{TiO}_3$: Phase evolution, optical, and electrical properties

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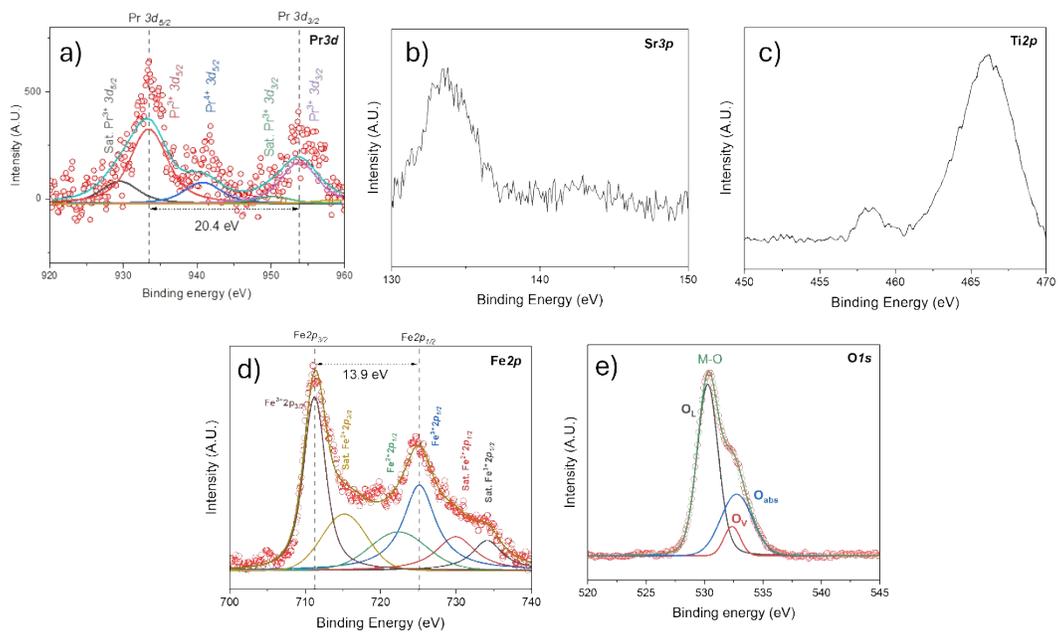


Fig. SF1: High-resolution XPS spectrum of the $(1-x)\text{Bi}_{0.9}\text{Pr}_{0.1}\text{FeO}_3-x\text{Ca}_{0.5}\text{Sr}_{0.5}\text{TiO}_3$ sample $x = 0.15$ measured for (a) Pr 3d, (b) Sr 3p (c) Fe 2p, (d) Ti 2p and (e) O 1s core levels

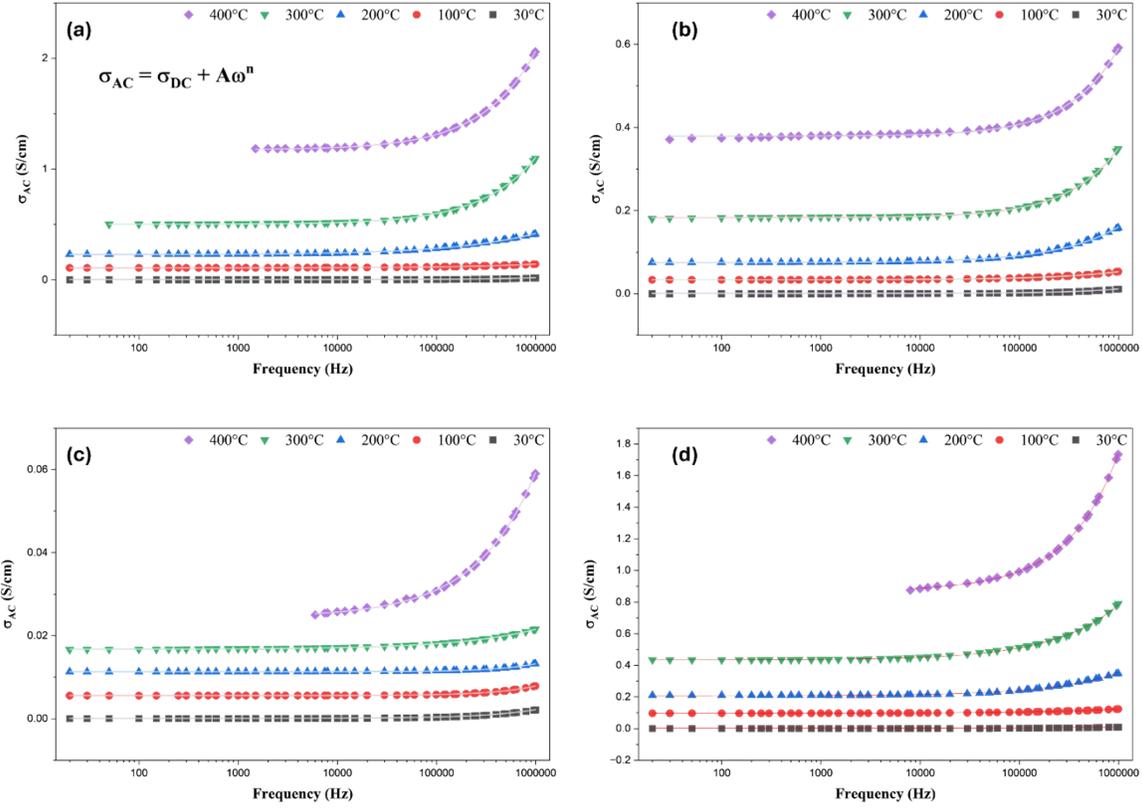


Fig. SF2: Fitting of the conductivity σ_{AC} by the Jonscher power law for (1-x)BPFO-xCSTO, $x = 0.0, 0.05, 0.10,$ and 0.15 .

Table S1: Binding energies from Bi 4f, Fe 2p, and Pr 3d core levels for $x = 0.00$

Element	Level	Oxidation number	B.E (eV)
Bi	4f _{7/2}	3	159.4
	4f _{5/2}	3	164.7
Fe	2p _{3/2}	3	711.1
	Sat. 2p _{3/2}	3	718.8
	Sat. 2p _{3/2}	2	714.2
	2p _{1/2}	3	724.8
	Sat. 2p _{1/2}	2	727.7
Pr	Sat. 3d _{5/2}	3	930.2
	3d _{5/2}	3	933.8
	3d _{5/2}	4	940.1
	Sat. 3d _{3/2}	3	949.7
	3d _{3/2}	3	953.9
	3d _{3/2}	4	957.7

Table S2: Binding energies from Bi 4f and Fe 2p core levels.

Element	Level	Oxidation number	B.E (eV)
Bi	4f _{7/2}	3	159.4
	4f _{5/2}	3	164.7
Fe	2p _{3/2}	3	711.2
	Sat. 2p _{3/2}	2	715.2
	2p _{1/2}	2	722.3
	2p _{1/2}	3	725.1
	Sat. 2p _{1/2}	2	729.9
	Sat. 2p _{1/2}	3	734.1
	Pr	Sat. 3d _{5/2}	3
3d _{5/2}		3	933.4
3d _{5/2}		4	940.7
Sat. 3d _{3/2}		3	950.2
3d _{3/2}		3	953.8
3d _{3/2}		4	958.3

Table S3. Fitting parameters obtained from Jonscher's power law fitting at different temperatures.

Temperature (°C)	x = 0.0			x = 0.05		
	$\sigma_{D.C}$ (S/cm)	A	n	$\sigma_{D.C}$ (S/cm)	A	n
30	3.16E-04	6.05E-08	0.7989	3.43E-05	2.43E-08	0.83066
100	0.00137	6.84E-06	0.5445	5.17E-05	8.79E-07	0.63926
200	0.03519	5.58E-05	0.5202	0.0047	1.42E-06	0.70282
300	0.5565	2.52E-06	0.7909	0.12572	2.69E-08	0.85202
400	4.10544	2.90E-07	0.807	1.1309	2.46E-07	0.8727
Temperature (°C)	x = 0.10			x = 0.15		
	$\sigma_{D.C}$ (S/cm)	A	n	$\sigma_{D.C}$ (S/cm)	A	N
30	1.77E-05	6.13E-09	0.8138	-6.83E-06	7.49E-07	0.60073
100	7.40E-06	3.16E-10	1.0093	-2.21E-04	4.35E-06	0.55868
200	5.97E-05	2.08E-10	1.0233	-0.00113	7.15E-06	0.63377
300	4.33E-04	9.04E-07	0.548	0.04246	9.95E-06	0.66842
400	0.12542	4.11E-07	0.7249	2.13887	1.52E-06	0.84646

Table S4. Physical properties comparison with reported literature.

Properties	In this study	Reported in lit.	Comments
ϵ_r	~79.09 – 84.00 (0.00-0.15) (RT, 100KHz)	~85 (from 1KHz to 1GHz), ~90 (from 100KHz to 800MHz) (BFO-BaZrO ₃ , BFO-CaTiO ₃ , respectively) [1]	BFO-ABO ₃ -type ceramics were fabricated with a conventional method.
		~26-36 (observed at 100KHz and above) [2]	BFO films were prepared by a chemical solution route.
		~120 (100KHz) [3]	Dielectric properties of pure polycrystalline BFO obtained by the solid-state reaction route, heating at 800°C for 48 h.
		~60 (at 100KHz) [4]	Bi _{1-x} Ba _x Fe _{1-y} Co _y O ₃ was synthesized by the solid-state route method.
		46 and 73 (BFO and BFO-BiMnO ₃ , respectively, at 100KHz) [5]	BFO/BiMnO ₃ composite ceramics obtained by the sol-gel technique.
E_g (eV)	2.09-2.21	2.02–2.27 [6]	(x-1)BFO-xSmFeO ₃ ceramics synthesized by solid-state reaction. The E_g value decreased as x decreased.
		1.93 [7]	Ca-Hf co-doped BFO was synthesized by using the sol-gel technique.
		1.6 [8]	Ce-doped BFO-PbTiO ₃ composite ceramics were synthesized by the solid-state reaction method.
		2.26 [9]	0.8BFO-0.2CTO composite synthesized by the sol-gel method
		3.95 [10]	Structured BFO-PbTiO ₃ thin films synthesized by the Pechini method for multifunctional photovoltaic applications.

Pr and Ti co-substituted BiFeO₃ ceramics with $x \leq 0.20$ were synthesized, with structural phase transition from rhombohedral (*R3c*) with $x \leq 0.10$ to orthorhombic (*Pnma*) phase with $0.10 < x \leq 0.20$ [11].

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